

Article

Ammonium to Total Nitrogen Ratio Interactive Effects with Salinity Application on *Solanum lycopersicum* Growth, Physiology, and Fruit Storage in a Closed Hydroponic System

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Abstract: Using high-salinity water for plant fertigation may have negative consequences for plant growth, overall yield and crop quality. In the present study, the effects of NaCl-salinity in conjunction with three different ammonium to nitrate ratios (Nr) on tomato (*Solanum lycopersicum* Mill.) plant growth, nutritional status, yield, fruit quality and postharvest storage were examined. The electrical conductivity (EC) was increased by adding NaCl into the nutrient solution and three different Nr ratios were applied, Nr0.05, Nr0.10 and Nr0.15, while the other macronutrient and micronutrient concentrations were constant in all treatments. The EC of the nutrient solution supplied to the plants was 2.2 mS cm⁻¹ at the low salinity treatments and 7.5 mS cm⁻¹ at the high salinity treatments. Increased salinity resulted in decreased plant growth factors and fruit yield, despite the Nr. An increased Nr reduced the pH value of the nutrient solution, while the fruits of the plants cultivated under high salinity obtained reduced Ca, K, Mg, P and N content. Reduced Ca content can lead to blossom end rot disorder and this was evidenced on tomato cultivation under high salinity. However, in the present study, this disorder appeared at the very late stages and did not affect the marketability of the fruits. On the other hand, both salinity and Nr0.15 increased fruit firmness at harvest or following storage, while citric acid, total soluble solid and vitamin C contents were also increased by salinity. Total phenols were increased by salinity at harvest, and lycopene and β -carotene content were increased by salinity at harvest and/or following storage at 12 °C or 25 °C. Salinity enhances fruit quality and improves the organoleptic characteristics of the crop, while an appropriate Nr ratio may restrict the detrimental effects of salinity on the nutritional status of plants by regulating the pH in hydroponic systems.

Keywords: minerals; fruit quality; salinity stress; nitrogen form; soilless culture; storage; tomato

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1. Introduction

Salinity and a lack of high-quality irrigation water are two key side effects of ongoing climate change that have a negative impact on plant physiological processes, plant growth and agronomic performance, with serious implications for global horticulture crop output [1,2]. According to reports, an increase in water and soil salinity is causing problems in 20% of agricultural and 33% of irrigated lands around the world [3,4]. Increased salt levels in soil solution are known to raise osmotic pressure in the rhizosphere, resulting in a reduction in water availability and nutrient uptake by plants [5]. Salinity stress has also been shown to have a deleterious impact on plant physiological and biochemical processes, affecting both the primary and secondary metabolism of plants [6]. Increasing sodium chloride (NaCl) levels has been demonstrated to have a negative effect on plant

height and vigor [7–9], leaf area [10,11] and leaf number [12,13]. High salinity also reduces plant photosynthetic rate [14], which can alter protein content [15], which in turn alters quality and activates many plant secondary metabolic processes [16], as well as antioxidant mechanisms to combat salinity impacts [17,18].

Nitrogen (N) is the sole nutrient that can be delivered to plants in both anionic (nitrate; NO_3^-) and cationic (ammonium; NH_4^+) form by fertigation, and both N forms have different absorption rates depending on their external concentrations [19]. Plant responses to NH_4^+ vary by species and are influenced by environmental factors such as temperature, light intensity, pH and nutrient concentrations in the growth media [20]. The form of N used by a plant may affect its morphology and chemical composition, with NH_4^+ likely to cause lower levels of potassium (K), calcium (Ca) and magnesium (Mg) and higher levels of phosphorus (P), sulphur (S) and organic N [21]. Several studies have found that the form of nitrogen, rather than the total nitrogen amount, is more important, with major implications not only for the overall yield but also for the marketable quality of the final vegetable products [22–24]. As a result, changing the composition of the nutrient solution (NS) can be a cost-effective way to improve vegetable quality by reducing antinutritional components [25] or increasing the beneficial compounds levels [26]. Most plants exhibit growth reduction and severe toxicity symptoms if NH_4^+ is the dominant or sole form of the supplied N, due to the intracellular concentration of ammonia, which is highly toxic for the plant cells [19], being increased as result. Despite the fact that the energetic cost of NO_3^- absorption and assimilation is significantly higher than that of NH_4^+ , the majority of plant species prefer NO_3^- to NH_4^+ [27]. This is because large concentrations of NH_4^+ in the solution (> 0.5 mM) can be harmful to plants (particularly at low pH levels), reducing biomass production [27,28]. The concentration of nitrate in edible tissues, on the other hand, is an essential quality attribute of vegetables. Nitrate levels in vegetables are frequently addressed as a potential health hazard. Furthermore, nitrite can react with amines and amides in the body to generate nitrosamines and nitrosamides, both of which have been linked to cancer [22].

Although various studies have demonstrated that high salinity has a negative impact on plant growth and fruit yield, tomato is regarded as a species which is fairly resilient to salt stress [29,30]. However, there is still a gap in our understanding of the plant physiology parameters, biochemical alterations and postharvest performance of saline-grown tomato plants in combination with nitrogen forms and ratios, for example, the ratio of ammonium ions to total nitrogen (Nr: NH_4^+ /total N). Several studies have involved a broad spectrum of Nr ranging from 0 to 1 [31] but this has not enabled an accurate optimization of the Nr responses. Therefore, interest has been focused on a narrower range of Nr [32]. As a result, this study's main objectives were to: a) investigate the effects of the application of ammonium to total nitrogen ratios as well as salinity conditions via nutrient solution on tomato plant growth, mineral content and fruit quality; b) assess the postharvest performance of fruits when plants were subjected to Nr or saline conditions under storage and shelf-life conditions. The findings of this study could lead to a better knowledge of plant responses to salinity and/or Nr application for the purpose of using hydroponic nutrient solution to aid in tomato plant development and fruit shelf-life.

2. Materials and Methods

2.1. Plant Material and Growing Conditions

The present study was carried out in a run-to-waste system hydroponic Venlo double-roof type glasshouse with a North-South orientation, located at the Mediterranean Agronomic Institute of Chania, Greece, at the latitude of 35.35° N, longitude of 24.02° E and altitude of 8 m. The study took place during the autumn–spring growing period (from the middle of October until the end of March in 2014–2015). Air temperature and relative humidity were averaged from 15.3°C to 24.6°C and 62.1–89.5%, respectively (Figure S1).

Tomato seedlings (*Solanum lycopersicum* Mill. cv. Belladonna) grown in standard potting mix were purchased from a local nursery. Seedlings were transplanted into perlite filled bags (three seedlings per bag, spaced at 33 cm) at the two true-leaf stage.

2.2. Experimental Set Up

Seedlings were given a seven-day period to become acclimatized to the new growing media (perlite). Six treatments were examined, considering three different $\text{NH}_4\text{-N}/\text{N}_{\text{total}}$ supply ratios (Nr) in conjunction with two salinity levels (three N ratio \times two salinity \times four replication plots). The treatments were named as plants grown with (i) Nr0.05 + LS ($\text{NH}_4\text{-N}/\text{N}_{\text{total}}$: 0.05, low salinity of 2.2 mS cm^{-1}), (ii) Nr0.10 + LS ($\text{NH}_4\text{-N}/\text{N}_{\text{total}}$: 0.10, low salinity of 2.2 mS cm^{-1}), (iii) Nr0.15 + LS ($\text{NH}_4\text{-N}/\text{N}_{\text{total}}$: 0.15, low salinity of 2.2 mS cm^{-1}), (iv) Nr0.05 + HS ($\text{NH}_4\text{-N}/\text{N}_{\text{total}}$: 0.05, high salinity of 7.5 mS cm^{-1}), (v) Nr0.10 + HS ($\text{NH}_4\text{-N}/\text{N}_{\text{total}}$: 0.10, high salinity of 7.5 mS cm^{-1}) and (vi) Nr0.15 + HS ($\text{NH}_4\text{-N}/\text{N}_{\text{total}}$: 0.15, high salinity of 7.5 mS cm^{-1}). For the purposes of our study, we opted to use the moderate salinity level of 50 mM NaCl in order to avoid subjecting plants to severe stress that could not allow fruit formation and the harvesting of fruit due to early plant death. Therefore, the term salinity is used to differentiate the experimental treatments (0 and 50 mM NaCl). To achieve the targeted salinity conditions in treatments (iv), (v) and (vi), electrical conductivity (EC) was gradually increased with the addition of NaCl over a nine-days period (from 2.20 for standard NS up to 7.50 mS cm^{-1} for salinity treatment) to avoid osmotic shock. The different Nr supply was on molar basis. Six tanks containing the six different nutrient solutions used for fertigation, with each tank being 300 L in volume, were established in the glasshouse. The composition of macronutrients in standard NS (1:100 *v/v*) was: $\text{NO}_3\text{-N} = 14.96 \text{ mmol L}^{-1}$, $\text{K} = 8.75 \text{ mmol L}^{-1}$, $\text{PO}_4\text{-}^3\text{-P} = 1.07 \text{ mmol L}^{-1}$, $\text{Ca} = 7.22 \text{ mmol L}^{-1}$, $\text{Mg} = 3.04 \text{ mmol L}^{-1}$, $\text{SO}_4\text{-}^2\text{-S} = 3.37 \text{ mmol L}^{-1}$ and sodium (Na) = 0.34 mmol L^{-1} ; while the micronutrient contents were: boron (B) = 25.00 $\mu\text{mol L}^{-1}$, iron (Fe) = 20.00 $\mu\text{mol L}^{-1}$, manganese (Mn) = 10.00 $\mu\text{mol L}^{-1}$, copper (Cu) = 0.77 $\mu\text{mol L}^{-1}$, zinc (Zn) = 4.00 $\mu\text{mol L}^{-1}$ and molybdenum (Mo) = 0.50 $\mu\text{mol L}^{-1}$. The pH of NS treatment was adjusted at optimal levels (5.6) for both types of solutions with the addition of HNO_3 (0.9% *v/v*) and the ratio of N/P/K was constant in all treatments. Fresh NS was prepared twice weekly, according to fertigation needs, while pH and EC levels were checked and adjusted whenever a new solution was prepared. Fertigation was applied via a drip irrigation system seven times per day (at full development stage) for three minutes per application, with a total volume of 777 mL of solution applied daily per plant. The fertigation schedule was adjusted from time to time depending on the plant developmental stage and the current environmental conditions. In general, an effort was made to collect 25% of the drainage solution from the nutrient solution that was applied.

The experiment was laid out as a completely randomized design, with four replication plots for each of the six treatments and six plants per replication plot, giving us a total of twenty-four plants per treatment, out of which sixteen individual plants (four plants per replication) were considered as experimental units and subjected to further measurements.

Plants were grown in bags filled with perlite as a growing media. The bags were placed on the metal channels, forming planting rows which had a distance of 0.8 m. The central corridor was 1.3 m wide. Three tomato plants (*Solanum lycopersicum* Mill. cv. Belladonna) were planted in each bag, spaced 0.33 m apart. The bags had slits on the bottom to allow drainage. The first fruit set was 4 weeks and the first harvest took place after 12 weeks after transplanting.

A vertical single stem pruning system was applied to the plants and pruning was performed almost every week. Sulfur was spread on the glasshouse floor during the cultivation period in order to protect the crop from pathogen infection. Crop protection was performed with the use of approved insecticides (AFFIRM, Decis) against *Tuta absoluta* (considered the most serious pest enemy for tomato cultivation in Crete) every 10 days. Additional sprayings with Switch and Mepam were performed three times during the

cultivation period against *Botrytis cinerea* with the appearance of the first symptoms (middle January). Additionally, yellow sticky traps were hung for protection from pests (*Bemisia tabaci*).

Pollination was performed with bumble bees that were established in the glasshouse for the whole growing period. The bumble bees were applied twice during cultivation. The insecticides used against *T. absoluta* were always applied during the late afternoon; by that time the bees had already entered the beehive. The beehive remained closed until the next day to protect the bumble bees from the insecticide. The beehive was removed from the glasshouse before the spraying of fungicides.

2.3. Plant Growth and Physiological Parameters

Forty-eight tomato plants (eight plants per treatment) were marked and used for plant growth measurements. Plant growth characteristics were recorded once, at the end of the study (12 weeks after transplanting). The recorded parameters included plant height (cm), number of leaves, stem diameter (cm), number of fruits, fresh fruit weight (g) and total yield (kg).

Leaf chlorophyll was extracted with dimethyl sulfoxide (DMSO) and Chlorophyll a (Chl a), Chlorophyll b (Chl b) and total Chlorophylls (total Chl) were assayed and expressed as $\mu\text{g g}^{-1}$ Fw [33]. Leaf photosynthetic rate (P_n), stomatal conductance (gs) and internal leaf concentration of CO_2 (C_i) were recorded with a Li-Cor gas analyzer (Li-6400, Li-Cor, Inc., Lincoln, NE, USA). All gas exchange measurements started 3 h after the onset of the photoperiod with the following instrumental adjustments: leaf chamber (6 cm^2), temperature $26 \pm 2 \text{ }^\circ\text{C}$ and photon flux density $1300 \mu\text{mol m}^{-2} \text{ s}^{-1}$ at ambient CO_2 concentration. For each treatment, measurements were taken from two fully expanded, healthy and sun-exposed leaves per plant from four plants in total.

2.4. Fruit Harvesting

For yield estimation, fruit were harvested from the first four clusters of selected tomato plants. These were then weighed and yield was expressed as kg per plant. For chemical analyses, twenty-four fruits from each treatment were selected from the last (fourth) cluster at the end of March and further divided into three batch samples. One batch of fruit was evaluated immediately after harvest and the other two batches were used for the assessment of storage performance. All the selected fruit were visually inspected prior to harvesting in order to make sure they were healthy and free of injuries, bruises and pathogen infestations. Moreover, the fruits affected by the physiological disorder "blossom-end rot" (BER) were also counted at each harvesting date.

2.5. Analysis of the Drainage Solution

Once a week, samples of the drainage solution were collected from every experimental plot. The samples were immediately taken to the laboratory and the pH and the EC were measured. The nitrates of the drainage solution were determined by means of a spectrophotometer according to the ultraviolet spectrophotometric screening method [34]. This technique is used for screening samples that have low organic matter contents. Measurement of UV absorption at 220 nm enables the rapid determination of NO_3 . Due to the fact that dissolved organic matter may also absorb at 220 nm and NO_3 does not absorb at 275 nm, a second measurement made at 275 nm and was sometimes used to correct the NO_3 value.

2.6. Mineral Composition Analysis

After harvesting, leaf and fruit mineral content was determined. Sub-samples of 0.2–0.3 g were placed in the oven ($500 \text{ }^\circ\text{C}$ for 6 h) for ashing and were acid digested with 10 mL 2N HCl. The content of P, K, Ca, Mg, Na, aluminium (Al), Fe, Zn, Mn, B and Cu was determined with an atomic emission spectrometer (ICP-AES PSFO 2.0, Leeman Labs INC.,

USA). The content of N was determined by using the Kjeldahl method (Kjeldahl nitrogen determination apparatus, Gerhardt Ltd, UK). Data were expressed as g kg⁻¹ and mg kg⁻¹ dry weight for macro and micro-nutrients, respectively [17,35].

2.7. Fruit Macroscopic Examination and Quality Analysis

Fruit color was measured with CR-300 chromatometer (Minolta, Japan), with two measurements per fruit being taken on the equatorial axis. The values of L^* , a^* and b^* were measured (where: L^* = lightness, a^* = redness and b^* = yellowness), and color indices (chroma, hue value, whitening index-WI and color index-CI) were computed [36]. Firmness was measured in one batch of the samples with a GY-2 penetrometer (Zhejiang Top Instrument, China) equipped with a plunger of 3.5 mm in diameter. Fruits were probed randomly in two sides and data were expressed in Newton.

The other two batches of fruit were further used for postharvest performance, one batch of fruits was stored at room temperature at 25 °C, and the other group was stored in a fridge at 12 °C for 7 days. Following the 7 days of storage, the fruits were weighed again, evaluated macroscopically and measured for quality changes.

Fruit quality-related attributes, namely total soluble solids (TSS), titratable acidity (TA) and ascorbic acid (AA) content were measured [37]. TSS were measured with a digital refractometer (PR-100; ATAGO, Japan) and TSS content was expressed as °Brix. For TA determination, tomato juice (5 mL) was diluted with distilled water to a final volume of 50 mL, titrated with 0.1N NaOH, in the presence of phenol-phtalein as an indicator, and expressed as citric acid percentage [37]. The ascorbic acid content was determined using the 2,6-Dichloroindophenol titrimetric method [38].

2.8. Determination of Carotenoids and Total Phenolics

Total phenolic composition was determined according to previous reports by the authors [33,37]. Briefly, methanolic extracts of blended tomato (2.5 g) following the addition of 1.5 mL of deionized water, 125 µL of Folin–Ciocalteu’s reagent (Merck, Germany) and 1.25 mL of 7% (w/v) sodium carbonate and incubation for 90 min were measured at 755 nm (Specord 205, Analytikjena, Jena, Germany). Results were expressed as gallic acid equivalents (µmol GAE g⁻¹ fresh weight).

Carotenoids (lycopene and β-carotene) content was measured according to the Nagata and Yamashita method with adjustments [39]. Briefly, 1 g of blended tomatoes was placed in 50 mL tubes and stored at -20 °C. After de-freezing, 16 mL of acetone:hexane 4:6 (v:v) were added to each tube and shaken vigorously for phase separation. From each tube, an aliquot was taken with a syringe from the upper phase and measured with a spectrophotometer at λ = 663, 645, 505 and 453 nm, using as reference a solution of acetone:hexane (4:6) ratio. Carotenoids content was calculated according to the following Nagata and Yamashita [39] Equations (1) and (2):

$$\text{Lycopene (mg 100 mL}^{-1}\text{ of extract)} = -0.0458 * A_{663} + 0.204 * A_{645} + 0.372 * A_{505} - 0.0806 * A_{453}. \quad (1)$$

$$\beta\text{-carotene (mg 100 mL}^{-1}\text{ of extract)} = 0.216 * A_{663} - 1.22 * A_{645} - 0.304 * A_{505} + 0.452 * A_{453}. \quad (2)$$

2.9. Organoleptic Control

The main idea was to study the effects of ammonium or salinity impacts on fruit quality (organoleptic test) immediately after harvest and to detect differences in the quality characteristics according to the treatments. Codded tomato samples from the six treatments (ammonium levels of 0.05, 0.10 and 0.15; salinity of low and high levels) were given as a blind test to a total of ten people (five men and five women) of various age groups and nationalities [37]. Participants were asked to rate the appearance, color, aroma, sweetness and texture of fruit on a scale of 1 to 5 (where 1 was for poor/soft/acidic/no, 2 for medium/little soft/less acidic/probably no, 3 for good/good/mild/sometimes, 4 for very good/little hard/little sweet/probably yes and 5 for excellent/hard/sweet/yes).

Participants' preferences, the satisfaction index and the intention of the participants to buy the product were recorded too. The visual appearance evaluation was performed on intact fruit while the evaluation of aroma, texture and sweetness was performed on freshly cut fruit. All tests took place on lab benches with full light and without any disturbance during the testing procedure.

2.10. Statistical Methods

Statistical analysis was performed using the SPSS v22.0 (SPSS Inc., Chicago, Ill.) program. Data were evaluated for normality before being subjected to analysis of variance (ANOVA). The significance of the differences between mean values was tested using the Duncan's multiple range test (DMRT) at $p \leq 0.05$, following one- and two-way ANOVA. For testing the panel's ratings, significant differences in rates were expressed in percentage values after a logarithmic transformation prior to analysis. Results were expressed as mean values \pm SE ($n = 8$ for plant growth/physiological/fruit quality and $n = 4$ for chemical analyses).

3. Results

3.1. Plant Growth and Physiological Parameters

Table 1 presents the impacts of different ammonium to total nitrogen ratios as well as the different levels of salinity in the nutrient solution on plant growth-related attributes on hydroponically grown tomato plants. Two-way analysis of variance showed that Nr affected leaf photosynthesis and internal CO₂ concentration ($p < 0.05$), whereas salinity affected tomato stem diameter, leaf photosynthesis and internal CO₂ concentration ($p < 0.05$). The interaction of Nr \times salinity application affected fruit number ($p < 0.001$).

Table 1. The effect of salinity (Sal) levels (LS: low salinity of 0 mM NaCl; HS: high salinity of 50 mM NaCl) and the application of NH₄/N_{total} ratios (Nr0.05, Nr0.10 and Nr0.15) on the plant growth, yield and physiological parameters of hydroponically grown tomato plants.

	Low Salinity			High Salinity			Nr	Sal	Nr \times Sal
	Nr0.05 + LS	Nr0.10 + LS	Nr0.15 + LS	Nr0.05 + HS	Nr0.10 + HS	Nr0.15 + HS			
Plant height (cm)	180.00 \pm 6.87a	183.37 \pm 5.16a	170.62 \pm 5.18ab	167.87 \pm 6.33ab	149.00 \pm 7.92c	160.85 \pm 3.46bc	ns	ns	ns
Number of leaves	22.50 \pm 0.70a	21.62 \pm 0.59a	22.75 \pm 0.70a	20.87 \pm 0.39a	20.75 \pm 0.90a	21.42 \pm 0.52a	ns	ns	ns
Stem diameter (mm)	8.25 \pm 0.63a	8.31 \pm 0.56a	7.87 \pm 0.78a	5.50 \pm 0.42b	5.12 \pm 0.45b	5.28 \pm 0.24b	ns	*	ns
Fruit number	11.25 \pm 1.35b	15.33 \pm 1.66a	12.14 \pm 1.10ab	14.12 \pm 0.89ab	11.71 \pm 1.04ab	14.00 \pm 1.09ab	ns	ns	***
Fruit fresh weight (g)	202.38 \pm 24.08a	180.83 \pm 13.16ab	188.16 \pm 8.74a	125.83 \pm 2.96c	144.64 \pm 7.51bc	127.92 \pm 7.25c	ns	ns	ns
Yield (kg per plant)	2.30 \pm 0.43ab	2.71 \pm 0.24a	2.29 \pm 0.26ab	1.77 \pm 0.11b	1.67 \pm 0.13b	1.77 \pm 0.13b	ns	ns	ns
Chlorophyll a (mg g ⁻¹)	14.60 \pm 1.86a	9.18 \pm 1.62a	13.91 \pm 0.96a	13.14 \pm 2.27a	14.52 \pm 2.58a	11.79 \pm 2.00a	ns	ns	ns
Chlorophyll b (mg g ⁻¹)	6.23 \pm 0.80a	3.61 \pm 0.57a	5.52 \pm 0.51a	6.30 \pm 1.10a	5.85 \pm 0.96a	5.39 \pm 1.24a	ns	ns	ns
Total Chlorophylls a (mg g ⁻¹)	20.82 \pm 2.63a	12.80 \pm 2.19a	19.43 \pm 1.39a	19.42 \pm 3.32a	20.37 \pm 3.54a	17.17 \pm 3.24a	ns	ns	ns
Photosynthetic rate ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	11.46 \pm 1.99b	14.84 \pm 1.11ab	17.82 \pm 1.14a	17.52 \pm 2.14a	18.17 \pm 1.58a	15.60 \pm 0.85ab	*	*	ns
Stomatal conductance ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	0.611 \pm 0.054a	0.687 \pm 0.052a	0.683 \pm 0.122a	0.529 \pm 0.055a	0.660 \pm 0.040a	0.455 \pm 0.157a	ns	ns	ns
Internal CO ₂ ($\mu\text{mol mol}^{-1}$)	280.66 \pm 31.59b	341.00 \pm 18.07ab	385.00 \pm 30.09ab	385.50 \pm 42.29ab	395.50 \pm 31.33a	313.00 \pm 35.79ab	*	*	ns

Results are expressed as means \pm SE ($n = 8$). Values in rows followed by the same Latin letter are not significantly different according to Duncan's multiple range test at $p \leq 0.05$. ns, *, ** and *** indicate non-significant or significant differences at $p \leq 5\%$, 1% and 0.1%, respectively, following two-way ANOVA.

A salinity of 50 mM NaCl at a medium (Nr0.10) and high (Nr0.15) ammonium to total nitrogen ratio decreased (up to 18.7%) plant height in comparison to low salinity and low-medium Nr (Table 1). High salinity decreased stem diameter, fruit fresh weight and therefore crop yield in comparison to low saline-treated plants, while the fruit number produced remained unaffected. Plants grown under a low Nr ratio (Nr0.05) revealed the

lowest photosynthetic rates in comparison with the plants grown in Nr0.15 + LS, Nr0.05 + HS and Nr0.10 + HS, and this counts as evidence for a higher internal CO₂ concentration in the case of the latter (Nr0.10 + HS).

Blossom end rot appeared above the third cluster and ranged from 7–9.4% of the fruits where salinity was applied, with Nr0.05 giving the lowest value, while the percentage of the fruits with blossom end rot was increased at higher Nr ratios.

3.2. Drainage Solution and Leaf and Fruit Mineral Content

Increasing Nr to 0.15 markedly suppressed the nutrient solution pH in the root environment of tomatoes as indicated by the values measured in the drainage water (Figure S2). This effect was more profound after the 13th of December and until the middle of February due to the low air temperature and light intensity on that period. Even an Nr of 0.15 resulted in pH values below 5.0 during that season. Conversely, when Nr was as low as 0.05, the pH in the root zone was raised to values above 7.0 during the winter. However, salinity did not have any effect on the drainage solution pH, but salinity increased the EC of the drainage solution (varying from 8 to 12 mS cm⁻¹), with a more pronounced increase at the lower Nr (Figure S2). Increased salinity and Nr resulted in decreased nitrate accumulation in the drainage nutrient solution (Figure S2).

The impacts of salinity and/or ammonium to total nitrogen levels on mineral content in tomato leaves and fruits is presented in Table 2. Two-way ANOVA revealed that Nr significantly affected the content of P ($p < 0.05$) in leaves, while salinity significantly affected the content of Mg and Na ($p < 0.001$), specifically K, P and B ($p < 0.01$) and N and Zn ($p < 0.05$), in leaves. The interaction of Nr x salinity affected N content ($p < 0.001$) in leaves. In fruit, salinity significantly affected Na ($p < 0.01$), specifically N, K, and Mn ($p < 0.05$), whereas the interaction of Nr x salinity affected Mn content ($p < 0.05$). No changes on fruit mineral content were found in terms of the different Nr applications (Table 2).

In leaves, N content increased at low Nr levels (i.e., Nr0.05 + LS), but decreased at \geq Nr0.10 + LS, while the opposite effects were observed in saline-treated plants, with N content increasing at \geq Nr0.10 + HS (Table 2). In contrast, leaf K content decreased (up to 53%) in saline-treated plants, independently of the Nr levels. The content of Mg, Na and Zn was increased in saline-treated plants, and the levels were more pronounced at the high Nr ratio of 0.15 (Nr0.15 + HS). The content of Fe and Mn was greater as a result of the Nr0.15 + HS treatment, while the greater B content was found as a result of the Nr0.10 + HS treatment. No differences were observed in leaf Ca, P and Cu content as averaged in 25.93 g kg⁻¹, 4.01 g kg⁻¹ and 1.74 mg kg⁻¹, respectively.

Table 2. The effect of salinity (Sal) levels (LS: low salinity of 0 mM NaCl; HS: high salinity of 50 mM NaCl) and the application of NH₄/N_{total} ratios (Nr0.05, Nr0.10 and Nr0.15) on mineral content of tomato leaves and fruit of hydroponically grown tomato plants.

Leaves	Low Salinity			High Salinity			Nr	Sal	Nr x Sal
	Nr0.05 + LS	Nr0.10 + LS	Nr0.15 + LS	Nr0.05 + HS	Nr0.10 + HS	Nr0.15 + HS			
N (g kg ⁻¹)	31.17 ± 0.41a	26.82 ± 1.02b	27.10 ± 0.49b	28.77 ± 0.72b	33.35 ± 0.63a	33.50 ± 1.03a	ns	*	***
K (g kg ⁻¹)	33.17 ± 1.45a	32.11 ± 1.55a	34.74 ± 1.00a	24.43 ± 1.51b	21.70 ± 0.67b	16.32 ± 1.72c	ns	**	ns
Ca (g kg ⁻¹)	26.46 ± 12.05a	23.63 ± 7.21a	19.11 ± 2.19a	23.71 ± 4.66a	29.76 ± 5.81a	32.92 ± 10.67a	ns	ns	ns
P (g kg ⁻¹)	3.02 ± 0.25a	2.86 ± 0.45a	3.90 ± 0.45a	5.31 ± 1.09a	4.05 ± 0.44a	4.91 ± 1.66a	*	**	ns
Mg (g kg ⁻¹)	3.98 ± 0.33c	3.93 ± 0.34c	4.25 ± 0.26c	8.17 ± 0.62b	7.51 ± 0.88b	10.85 ± 1.36a	ns	***	ns
Na (g kg ⁻¹)	2.08 ± 0.18c	2.23 ± 0.15c	2.41 ± 0.40c	10.04 ± 1.10b	8.55 ± 1.19b	13.95 ± 2.02a	ns	***	ns
Fe (mg kg ⁻¹)	80.17 ± 6.05ab	71.49 ± 3.26b	73.36 ± 3.53ab	76.98 ± 2.13ab	78.38 ± 5.26ab	89.22 ± 7.92a	ns	ns	ns
Zn (mg kg ⁻¹)	4.87 ± 0.82c	5.73 ± 2.03c	6.27 ± 2.67c	17.69 ± 1.96b	16.89 ± 3.83b	37.53 ± 8.25a	ns	*	ns
Cu (mg kg ⁻¹)	1.04 ± 0.24a	1.35 ± 0.46a	0.55 ± 0.21a	2.05 ± 0.54a	1.48 ± 0.49a	3.98 ± 0.83a	ns	ns	ns
Mn (mg kg ⁻¹)	24.21 ± 2.36b	29.77 ± 7.82ab	40.97 ± 13.22ab	47.28 ± 10.47ab	51.37 ± 7.97ab	79.98 ± 33.47a	ns	ns	ns
B (mg kg ⁻¹)	72.47 ± 3.62ab	63.16 ± 7.69b	69.34 ± 3.74ab	84.17 ± 5.56ab	89.81 ± 5.80a	79.52 ± 12.57ab	ns	**	ns
Fruits									
N (g kg ⁻¹)	29.90 ± 7.01a	29.35 ± 5.28a	19.12 ± 1.61ab	16.87 ± 0.95b	16.75 ± 0.72b	18.55 ± 1.97ab	ns	*	ns
K (g kg ⁻¹)	27.60 ± 1.56a	24.10 ± 2.82ab	24.86 ± 1.01ab	21.22 ± 1.90bc	19.77 ± 1.68bc	16.95 ± 1.72c	ns	*	ns
Ca (g kg ⁻¹)	2.65 ± 0.30a	1.70 ± 0.11b	1.62 ± 0.02b	1.47 ± 0.08b	1.54 ± 0.09b	1.32 ± 0.05b	ns	ns	ns
p (g kg ⁻¹)	1.86 ± 0.18a	1.26 ± 0.08b	1.56 ± 0.27ab	1.13 ± 0.22b	1.12 ± 0.15b	1.15 ± 0.15b	ns	ns	ns
Mg (g kg ⁻¹)	0.73 ± 0.07a	0.55 ± 0.01ab	0.70 ± 0.14a	0.44 ± 0.06b	0.49 ± 0.06ab	0.56 ± 0.03ab	ns	ns	ns
Na (g kg ⁻¹)	0.77 ± 0.08b	0.45 ± 0.09b	0.57 ± 0.09b	2.10 ± 0.39a	2.17 ± 0.15a	1.86 ± 0.29a	ns	**	ns
Al (g kg ⁻¹)	1.36 ± 0.05a	0.95 ± 0.20ab	0.71 ± 0.20b	1.11 ± 0.09ab	1.08 ± 0.10ab	0.82 ± 0.12b	*	ns	ns
Fe (mg kg ⁻¹)	47.72 ± 6.66ab	48.60 ± 2.35ab	67.53 ± 23.42a	31.72 ± 3.36b	28.10 ± 3.37b	29.22 ± 2.19b	ns	ns	ns
Zn (mg kg ⁻¹)	3.85 ± 0.46a	7.53 ± 2.77a	6.38 ± 3.27a	3.10 ± 1.23a	1.05 ± 0.38a	3.53 ± 1.62a	ns	ns	ns
Cu (mg kg ⁻¹)	1.88 ± 0.36a	1.57 ± 0.77ab	0.55 ± 0.20ab	nd	nd	0.28 ± 0.01b	ns	ns	ns
Mn (mg kg ⁻¹)	14.32 ± 1.74a	4.88 ± 0.18b	7.57 ± 0.91b	4.01 ± 1.61b	4.57 ± 3.13b	3.23 ± 0.36b	ns	*	*
B (mg kg ⁻¹)	18.20 ± 3.36a	14.46 ± 3.83a	18.19 ± 9.15a	12.07 ± 3.57a	13.36 ± 3.51a	8.39 ± 1.59a	ns	ns	ns

Results are expressed as mean values \pm SE ($n = 4$). Values in rows followed by the same Latin letter are not significantly different according to Duncan's multiple range test at $p \leq 0.05$; *nd-not detectable. ns, *, ** and *** indicate non-significant or significant differences at $p \leq 5\%$, 1% and 0.1%, respectively, following two-way ANOVA.

In the fruits, the content of N, K, Ca, P and Mg was increased in non-saline treated plants and at a low Nr ratio, while Mg was also accumulated in case of a higher Nr ratio in non-saline treated tomatoes (Table 2). Indeed, Na was accumulated in saline treated plants, independently of the Nr ratio. Copper and Mn content was greatest at low Nr ratios in non-saline treated plants. Iron content was increased at Nr0.15 + LS but decreased in saline-treated plants. Both Zn and B remained unchanged in all examined treatments (Table 2).

3.3. Quality Parameters at Harvested Fruits

Harvested fruits were analyzed for quality related attributes. The analysis of variance of the fruit quality data revealed that fruit dry matter content, TSS, Vitamin C content ($p < 0.001$), citric acid content, color index ($p < 0.01$), fruit firmness, color b^* and Hue ($p < 0.05$) were all influenced by salinity. The Nr ratio did not affect fruit quality parameters, while the interaction Nr \times salinity significantly affected the color index ($p < 0.01$), Hue and whitening index ($p < 0.05$), as presented in Table 3.

The content of TSS in tomato fruit was significantly higher in plants grown under salinity (50 mM NaCl), regardless of the Nr ratio, as shown in Table 3. Citric acid content was lower in non-saline grown plants. Fruit firmness was the greatest in the saline-treated plants with a high Nr ratio (Nr0.15 + HS), while a low Nr ratio resulted in decreased firmness. Plants grown in saline conditions produced fruits with increased dry matter content, which significantly differed at Nr0.10 + HS compared to the fruits harvested from the non-saline grown plants. Vitamin C content in tomato fruit was increased at a salinity of 50 mM NaCl at \geq Nr0.10 (i.e., Nr0.10 + HS and Nr0.15 + HS) in comparison to the fruits harvested from the non-saline grown plants.

Fruit lightness was decreased in saline-treated plants compared to the non-saline grown plants (Table 3). A higher L^* value in fruit was found in the case of the low Nr ratio in non-saline grown plants, which resulted in the Hue and whitening index being increased and the color index being decreased. The color a^* and b^* values, and therefore the fruit chroma, was unchanged in the examined NaCl and Nr applications.

The total phenols content increased at the Nr0.05 + HS, whereas the lycopene and β -carotene content was the same for all the treatments, with an average of 0.058 and 0.046 mg g^{-1} fresh weight for lycopene and β -carotene, respectively (Table 3).

3.4. Organoleptic Test

Sensory evaluation took place through a panel test of the examined samples at the day of harvest for fruit harvested within a 5-days harvesting period (Table 4). Considering the salinity effect, all testers were able to determine variations between the tested samples. Fifty percent of the testers preferred the fruits that were harvested from saline-grown plants, 40% of the testers preferred fruits harvested from non-saline grown plants, whereas only 10% of the participants preferred none of the examined samples (data not presented). Higher aroma and sweetness scores were found in fruit subjected to salinity, while fruit appearance, color, texture, overall satisfaction and marketability did not differ significantly with a salinity of 50 mM NaCl (despite the increased scores in saline treated tomatoes compared to the non-saline treated fruit).

Considering the Nr effect, 90% of the testers could determine differences between samples. Fifty percent of the testers preferred the fruits that were harvested from high Nr ratio (Nr0.15) treated plants, 30% of the testers preferred fruits harvested from Nr0.10 treated plants and 20% of the testers preferred fruits harvested from Nr0.05 treated plants (data not presented). The greatest scores for fruit color were found in the cases of Nr0.10 and Nr0.15 treated plants, while lower color scores were found for the Nr0.05 treatment. All the other fruit quality related attributes revealed similar scores by the testers.

Table 3. The effect of salinity (Sal) levels (LS: low salinity of 0 mM NaCl; HS: high salinity of 50 mM NaCl) and the application of NH₄/N_{total} ratios (Nr0.05, Nr0.10, and Nr0.15) on the tomato fruit quality related attributes of hydroponically grown tomato plants.

	Low Salinity			High Salinity			Nr	Sal	Nr x Sal
	Nr0.05 + LS	Nr0.10 + LS	Nr0.15 + LS	Nr0.05 + HS	Nr0.10 + HS	Nr0.15 + HS			
Dry matter content (%)	4.48 ± 0.32b	5.02 ± 0.14b	4.78 ± 0.12b	6.73 ± 0.10ab	8.98 ± 2.02a	6.92 ± 0.28ab	ns	***	ns
Firmness (N)	5.18 ± 0.35c	4.77 ± 0.20c	5.58 ± 0.31bc	5.31 ± 0.36c	6.31 ± 0.30ab	6.75 ± 0.26a	ns	*	ns
TSS (°Brix)	4.00 ± 0.23b	3.84 ± 0.16b	4.12 ± 0.14b	5.46 ± 0.20a	5.76 ± 0.21a	5.22 ± 0.25a	ns	***	ns
Citric acid (%)	0.21 ± 0.01b	0.20 ± 0.02b	0.21 ± 0.01b	0.27 ± 0.03a	0.24 ± 0.02ab	0.27 ± 0.01a	ns	**	ns
Ripening (TSS/TA)	18.95 ± 1.17a	20.00 ± 2.53a	20.58 ± 2.36a	20.89 ± 2.57a	24.12 ± 2.07a	19.26 ± 1.62a	ns	ns	ns
Color (L* value)	35.70 ± 0.83a	34.30 ± 0.44ab	34.77 ± 0.68ab	33.79 ± 0.23b	33.83 ± 0.31b	33.89 ± 0.33b	ns	ns	ns
Color (a* value)	18.65 ± 1.53a	20.08 ± 1.07a	20.26 ± 1.26a	21.02 ± 0.64a	18.39 ± 0.53a	20.55 ± 0.99a	ns	ns	ns
Color (b* value)	12.75 ± 0.66a	12.26 ± 0.62a	12.61 ± 0.36a	11.74 ± 0.33a	11.45 ± 0.24a	11.89 ± 0.24a	ns	*	ns
Chroma	22.61 ± 1.60a	23.53 ± 1.22a	23.89 ± 1.21a	24.08 ± 0.64a	21.67 ± 0.51a	23.75 ± 0.97a	ns	ns	ns
Hue	0.60 ± 0.01a	0.54 ± 0.00bc	0.56 ± 0.02ab	0.51 ± 0.01c	0.55 ± 0.01abc	0.53 ± 0.01bc	ns	*	*
Whitening index	31.74 ± 0.42a	30.17 ± 0.32b	30.48 ± 0.50b	29.53 ± 0.33b	30.36 ± 0.40b	29.72 ± 0.33b	ns	ns	*
Color index	40.82 ± 1.19c	47.82 ± 1.38b	46.21 ± 2.25b	53.11 ± 1.72a	47.56 ± 1.65b	50.94 ± 1.71b	ns	**	**
Vitamin C (%)	0.59 ± 0.08bc	0.52 ± 0.05c	0.74 ± 0.07bc	0.92 ± 0.13ab	1.15 ± 0.18a	1.14 ± 0.11a	ns	***	ns
Lycopene (mg g⁻¹ Fw)	0.054 ± 0.014a	0.056 ± 0.009a	0.039 ± 0.009a	0.073 ± 0.013a	0.071 ± 0.013a	0.054 ± 0.005a	ns	ns	ns
β-carotene (mg g⁻¹ Fw)	0.052 ± 0.007a	0.044 ± 0.004a	0.038 ± 0.010a	0.041 ± 0.004a	0.049 ± 0.004a	0.049 ± 0.005a	ns	ns	ns
Total phenols (μmol GAE g⁻¹ Fw)	0.35 ± 0.02ab	0.29 ± 0.02ab	0.28 ± 0.01b	0.38 ± 0.04a	0.35 ± 0.02ab	0.32 ± 0.01ab	ns	ns	ns

Results are expressed as means ± SE (*n* = 8). Values in rows followed by the same Latin letter are not significantly different according to Duncan's multiple range test at *p* ≤ 0.05. ns, *, ** and *** indicate non-significant or significant differences at *p* ≤ 5%, 1% and 0.1%, respectively, following two-way ANOVA.

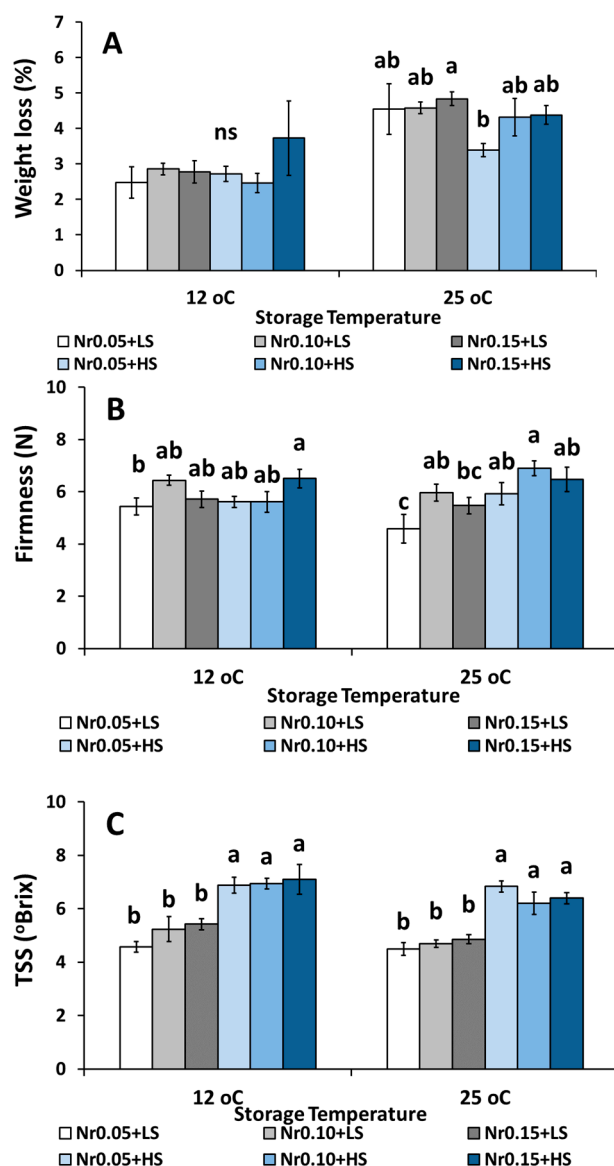
Table 4. Quantitative analysis, expressed as percentage values (%) of quality attributes in relation to salinity levels (0 and 50 mM NaCl) and/or the application of $\text{NH}_4/\text{N}_{\text{total}}$ ratios (Nr0.05, Nr0.10 and Nr0.15) on hydroponically grown tomatoes. Assessments were reported by 10 panelists per treatment.

Treatment		Appearance	Color	Aroma	Texture	Sweetness	Overall satisfaction	Marketability
$\text{NH}_4/\text{N}_{\text{total}}$ ratios	Nr0.05	66.00 ± 4.26a	64.00 ± 2.66b	62.00 ± 6.28a	84.00 ± 5.81a	58.00 ± 4.66a	64.00 ± 4.98a	68.00 ± 4.42a
	Nr0.10	78.00 ± 4.66a	76.00 ± 4.00a	76.00 ± 4.98a	82.00 ± 6.28a	70.00 ± 5.37a	74.00 ± 3.05a	80.00 ± 5.16a
	Nr0.15	76.00 ± 4.00a	75.00 ± 3.33a	70.00 ± 6.14a	84.00 ± 4.98a	66.00 ± 4.26a	69.33 ± 2.99a	74.00 ± 5.20a
Salinity	0 mM NaCl	66.00 ± 3.05a	78.00 ± 3.59a	56.00 ± 4.98b	70.00 ± 6.14a	74.00 ± 3.05b	66.00 ± 4.26a	74.00 ± 5.20a
	50 mM NaCl	66.00 ± 5.02a	68.00 ± 5.33a	76.00 ± 5.81a	82.00 ± 7.57a	82.00 ± 2.53a	70.00 ± 6.14a	76.00 ± 6.53a

Results are expressed as mean ± SE ($n = 10$). In each treatment, values in columns followed by the same Latin letter are not significantly different according to Duncan's multiple range test at $p \leq 0.05$.

3.5. Postharvest Fruit Quality Attributes

Harvested fruits from plants grown under a salinity of 0 or 50 mM NaCl and/or an ammonium to total nitrogen ratio of Nr0.05, Nr0.10 or Nr0.15 were stored for 7 days at 12 °C and 25 °C, reflecting storage conditions in packinghouse or shelf-life conditions. At 12 °C, fruit firmness was maintained for the Nr0.15 + HS plants in comparison to the Nr0.05 + LS ones (Figure 1B). A salinity of 50 mM NaCl increased fruit TSS, independently of the different Nr (Figure 1C), while the fruit weight loss, TA and ripening index (TSS/TA) were not affected by neither salinity nor Nr (Figure 1A,D,E). At 25 °C, fruit firmness was maintained for the saline-treated plants at all Nr levels as well as for the Nr0.10 + LS treatment in comparison to Nr0.05 + LS (Figure 1B). Fruit weight loss decreased for the Nr0.05 + HS plants compared to the Nr0.15 + LS ones (Figure 1A). Similar to the results at 12 °C, at 25 °C, fruit TSS was increased in saline-treated plants, independent of the different Nr ratios (Figure 1C). Fruit titratable acidity was increased at \geq Nr0.10 in saline-treated plants compared to the non-saline treated plants (Figure 1D). The highest ripening index was observed for the Nr0.05 + HS plants in comparison to the other treatments (Figure 1E).



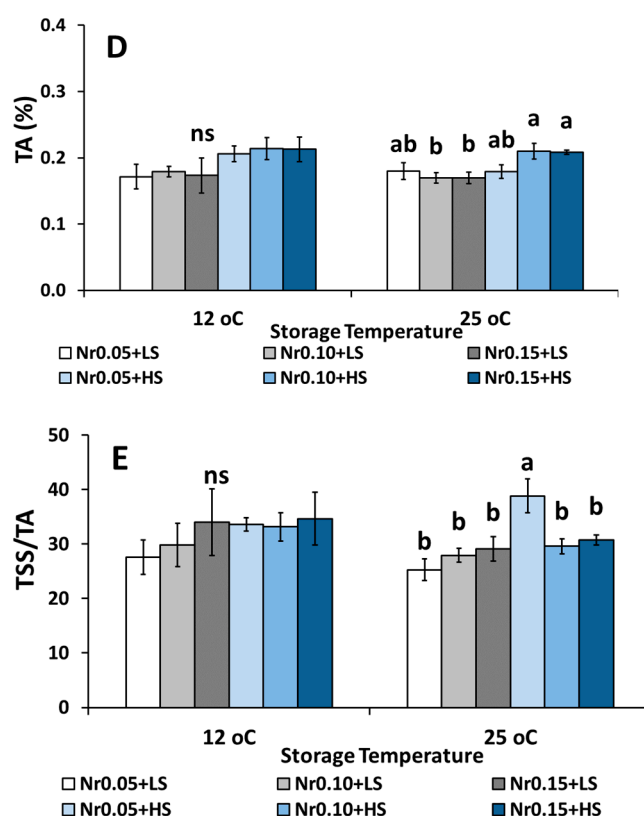
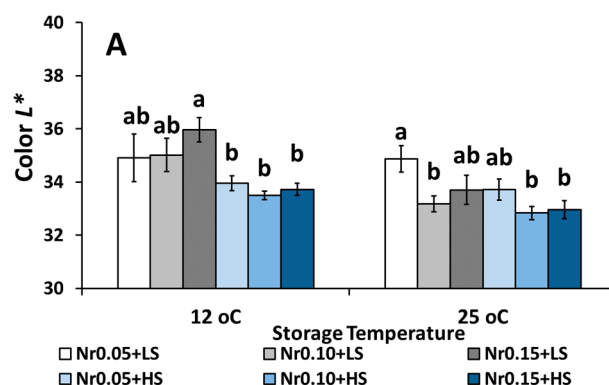
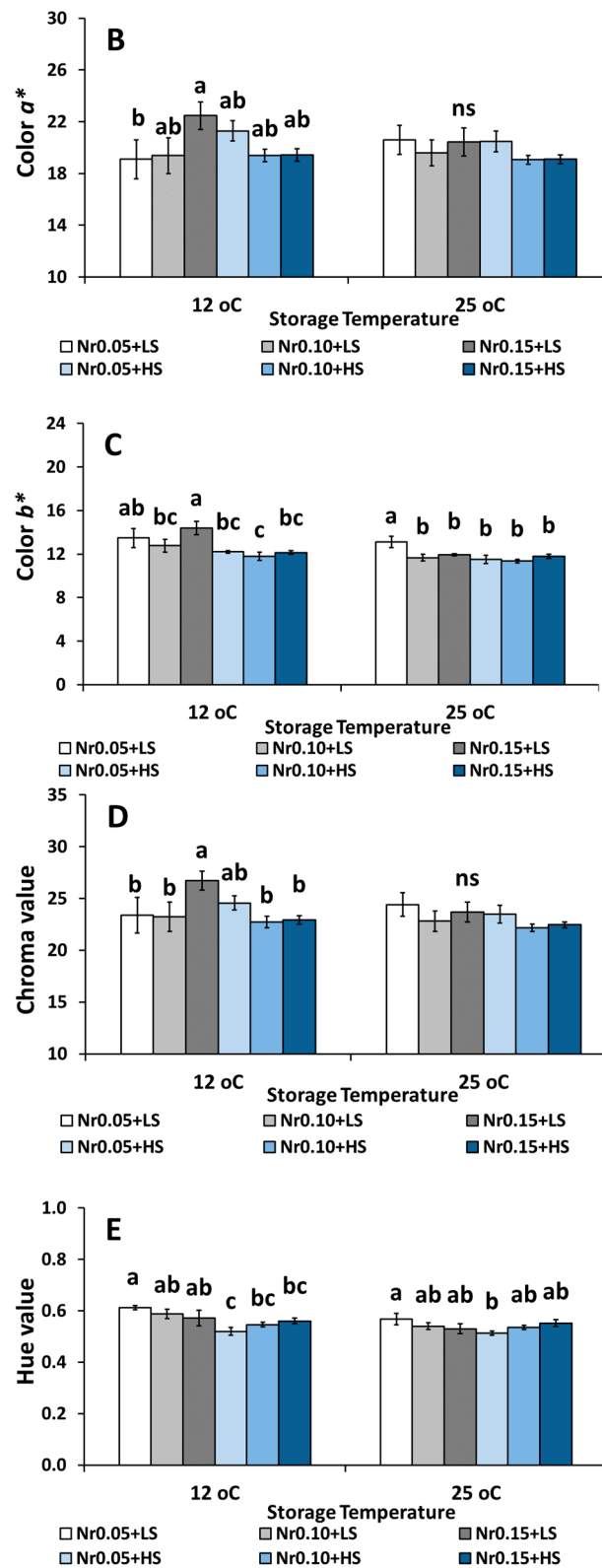


Figure 1. The effect of salinity (Sal) levels (LS: low salinity of 0 mM NaCl; HS: high salinity of 50 mM NaCl) and the application of $\text{NH}_4/\text{N}_{\text{total}}$ ratios (Nr0.05, Nr0.10 and Nr0.15) on tomato plants and the harvested fruit after storage for 7 days at 12 °C and 25 °C for (A) weight loss, (B) firmness, (C) total soluble solids-TSS, (D) titratable acidity-TA and (E) TSS/TA ratio. Non-significant differences are indicated by ns, while significant differences ($p < 0.05$) among treatments are indicated by different Latin letters according to Duncan's multiple range test. Error bars show SE ($n = 8$).

Fruit color attributes were more affected in commodities stored at 12 °C than in commodities stored at 25 °C (Figure 2). At 12 °C, Nr0.15 + LS treatment revealed fruits with increased lightness (L^*), redness (a^*), yellowness (b^*) and chroma (Figure 2A–D). The greatest Hue value was found for the Nr0.05 + LS plants, the greatest WI was found for Nr0.10 + LS plants, while the greatest color index was observed for Nr0.05 + HS plants (Figure 2E–G). At 25 °C, the Nr0.05 + LS treatment revealed fruits with increased lightness (L^*), yellowness (b^*) and Hue values (Figure 2A–E) but with a lower color index (Figure 2G). Indeed, no differences were found in color a^* , chroma and WI among the saline and/or Nr treatments at 25 °C storage conditions (Figure 2B,D,F).





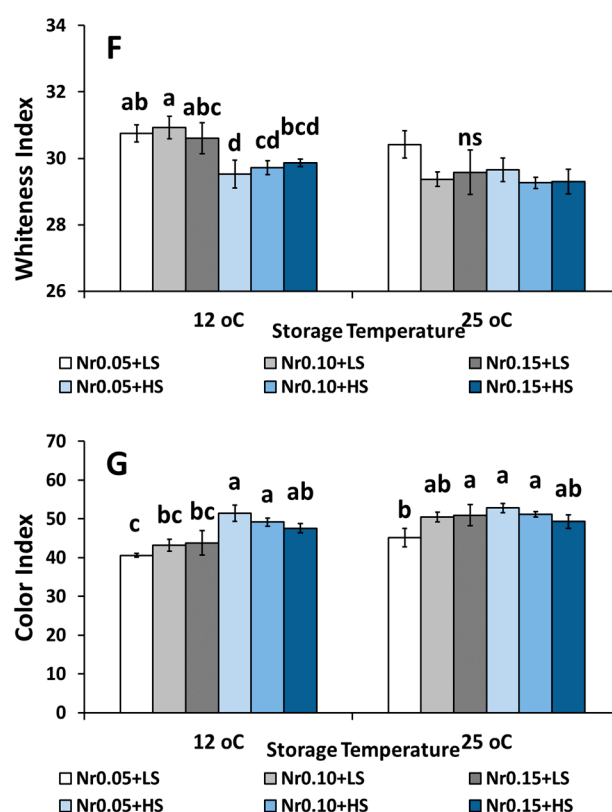
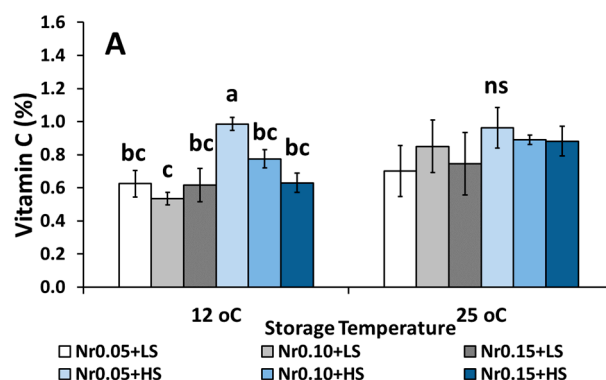


Figure 2. The effect of salinity (Sal) levels (LS: low salinity of 0 mM NaCl; HS: high salinity of 50 mM NaCl) and the application of $\text{NH}_4/\text{N}_{\text{total}}$ ratios (Nr0.05, Nr0.10, and Nr0.15) on tomato plants and the harvested fruit after storage for 7 days at 12 ° and 25 °C for fruit color parameters. (A) color L^* , (B) color a^* , (C) color b^* , (D) chroma, (E) Hue, (F) whiteness index, and (G) color index. Non-significant differences are indicated by ns, while significant differences ($p < 0.05$) among treatments are indicated by different Latin letters according to Duncan's multiple range test. Error bars show SE ($n = 8$).

At 12 °C, fruit vitamin C was increased in the case of Nr0.05 + HS (Figure 3A). Lycopene content was increased for Nr0.10 + HS plants in comparison to Nr0.05 + LS, Nr0.10 + LS, Nr0.15 + LS and Nr0.15 + HS plants (Figure 3C). The content of β -carotene was increased in the cases of Nr0.10 + HS and Nr0.15 + HS in comparison to the Nr0.05 + LS and Nr0.10 + LS plants (Figure 3D). The total phenols content in fruits was similar in both of the storage conditions in terms of temperature for both the saline and Nr treated plants (Figure 3B). At 25 °C, the content of β -carotene in fruits was increased in the saline-treated plants compared to the non-saline treated plants, independently of the Nr (Figure 3D). Similar to the total phenols content at 25 °C, the vitamin C and lycopene content remained unchanged for all treatments (Figure 3A,C).



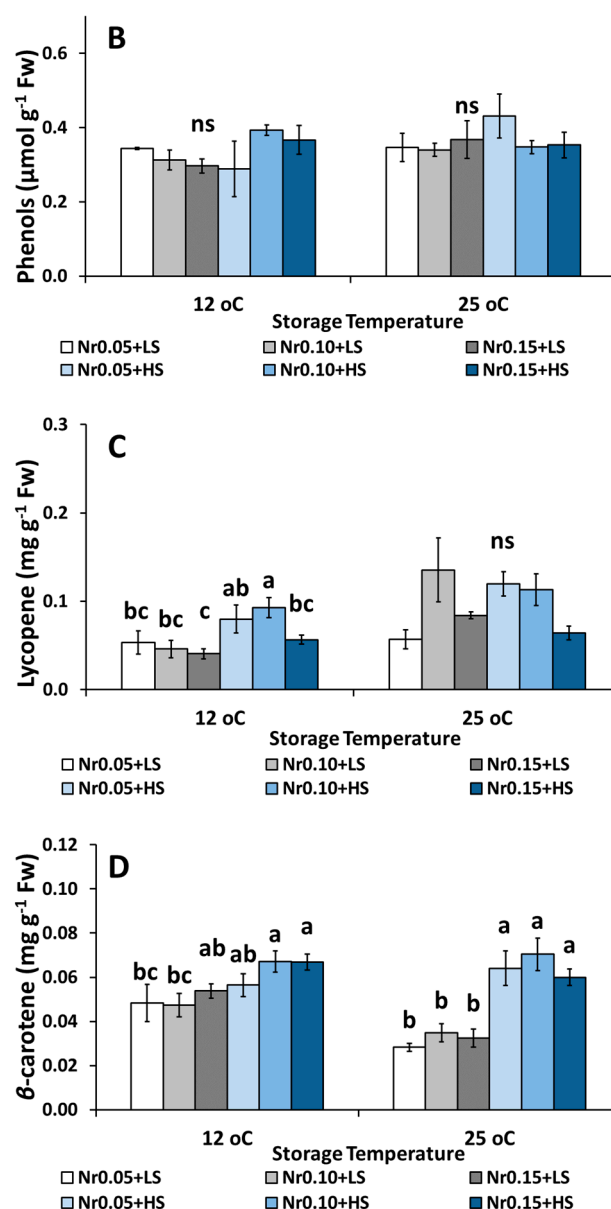


Figure 3. The effect of salinity (Sal) levels (LS: Low Salinity of 0 mM NaCl; HS: High Salinity of 50 mM NaCl) and the application of $\text{NH}_4/\text{N}_{\text{total}}$ ratios (Nr0.05, Nr0.10, and Nr0.15) on tomato plants and the harvested fruit after storage for 7 days at 12 °C and 25 °C for (A) vitamin C, (B) total phenols, (C) lycopene and (D) β -carotene content. Non-significant differences are indicated by ns, while significant differences ($p < 0.05$) among treatments are indicated by different Latin letters according to Duncan's multiple range test. Error bars show SE ($n = 8$).

4. Discussion

In the present study, which examined the impact of salinity and/or the ammonium–nitrate ratio (Nr0.05; Nr0.10 and Nr0.15) on tomato cultivation, the results revealed that NaCl-salinity significantly reduced the vegetative growth of tomato plants to the degree that the plant height, stem diameter, fruit fresh weight and total yield were directly affected. Other parameters such as the number of leaves per plant were not significantly influenced. The reduction in vegetative growth and yield under high salinity treatment is consistent with the results of Psaras et al. [40], Stamatakis et al. [41] and Costan et al. [35] concerning tomato crops. However, a different response was observed in an eggplant crop under salinity, where the vegetative parts and the leaf area were not significantly reduced by moderate levels of salinity and the most detrimental effects were observed on mean

fruit weight and yield [42]. These findings further support the notion that the eggplant species is less sensitive to salinity stress than the tomato species. Furthermore, our findings showed that growing tomato plants in saline conditions increased the occurrence of BER symptoms by up to 9.4 percent, decreasing the marketability of the fruit significantly. Zhai et al. [43] found that high salinity increased the incidence of BER because high salinity is related to impaired Ca^{2+} uptake, which is the primary cause of BER symptoms.

The increased NH_4^+ levels in the present study did not greatly affect the growth and yield of tomatoes, possible due to the low NH_4^+ levels used. However, in the context of a higher Nr ratio in cucumber plants, plant yield was significantly decreased [20]. This decrease in yield due to the increased Nr within the N supply might be attributed to the inhibition of the uptake of K^+ and other cations [44]. In an early spring trial, providing 30% of total N in the form of NH_4^+ significantly enhanced both the fresh and dry shoot weight of lettuce [19].

With respect to the nutrient solution, pH was both affected by the Nr ratio as well as the environmental conditions. A more intense drop in the pH value was observed under a high Nr and under the cold weather conditions of winter, while higher spring temperatures spring might alleviate the high Nr effect. In general, an increased Nr leads to pH reduction due to the release of H^+ ion into the nutrient/soil solution [19,22]. The impacts of a particular $\text{NH}_4^+\text{-N}/\text{total-N}$ supply ratio may be quite different in an opened hydroponic system than in a closed one [19]. It is well known that too high or too low pH values in the root environment of plants may quickly result in nutrient deficiencies or toxicities, even when the supply of nutrients is optimal [32].

Nitrate accumulation either in the drainage solution or in the plants is related to low light density and low-cool temperatures, whereas these conditions were observed in the glasshouse from December to the middle of February. At the same time, the accumulation of nitrates in the drainage solution reached its peak during this period. The positive correlation of nitrate accumulation in the drainage solution and the plants with the weather conditions occurs mainly due to the incapability of the plants to absorb and/or metabolize the nitrates due to reduced photosynthesis rates and the reduced activity of the enzyme nitrate reductase under these conditions (poor light density, cool temperatures) [45]. Furthermore, ammonium increases the acidity of the cells and decreases the permeability of the cell membrane, leading to an inhibition of nitrate absorption and reduction. This is consistent with the present study, where the highest accumulation of nitrate in the drainage solution was observed under high ammonium rates.

In terms of leaf nutrient content, the results showed that the levels of Ca in the leaves did not change regardless of the treatments used. This is consistent with the findings of Tuna et al., who claimed that Ca is replaced by Na in cell membranes during salt stress [46], and previous studies on tomato crops subjected to salinity [35]. As in the case of Ca, p and Cu levels also remained unaffected by the tested treatments. Several previous studies have shown that salinity either increases [35] or has no effect on p uptake [47,48]. Additionally, it has been reported that p levels increased in a mountainous population of *Cichorium spinosum* but were unaffected in the seaside population when plants were subjected to the application of increased Nr [22]. As a result, it is reasonable to believe that a variety of factors influence p uptake, including temperature, water availability in the root environment, species and even cultivar [49]. Salinity decreased K levels in the present study, and these data indicate that K^+ and Na^+ are in competition. The decrease in K levels in leaf tissues is consistent with the findings of del Amor et al. and Atluntas et al., who revealed that a decrease in K content is balanced out by an increase in Na content [50,51]. Similarly, when plants were cultivated in a saline environment, Na^+ levels are much greater, which is confirmed by the results of del Amor et al. [50] and Costan et al. [35]. The Zn levels in the present study were increased with the application of 50 mM NaCl and with high Nr levels, in comparison to saline with low Nr treatments and/or non-saline conditions, indicating the interactive role between ammonium and sodium ions. Indeed, salinity itself had previously been shown to increase Zn levels in tomato leaves [35] and

lavender leaves [52]. Finally, the fact that Fe and Mn content increased with the application of saline and high Nr is in agreement with the findings of Turan et al., who have stated that salinity increases the concentration of Mn in the shoot and root tissues of maize [53]. It has also been reported that increased Nr levels result in decreased Zn, Cu and Mn levels in *C. spinosum* [22].

Regarding fruit nutrient status, Ca content was negatively influenced with an increasing Nr due to $\text{NH}_4^+/\text{Ca}^{2+}$ competition, while Na^+/K^+ and $\text{Na}^+/\text{Ca}^{2+}$ competition also occurred under high salinity. Reduced Ca content in fruits can lead to BER disorder, which is quite common in the case of tomato crops. In the present study, this disorder appeared above the third cluster, and ranged from 7–9.4% in the fruits where salinity was applied, with Nr0.05 giving the lowest value, while the percentage of the fruits with BER was increased at higher Nr ratios. Potassium content decreased with salinity, especially at increased Nr ratios, with the application of Nr0.15 under high salinity giving the lowest value. This indicates the strong competition of K^+ with both NH_4^+ and Na^+ in this case. Similar results were found by Adler and Wilcox in a study with respect to salt-stressed muskmelon plants, suggesting that the influx of Na is regulated by K/Na selectivity and exchanges across cell membranes [54]. Additionally, an increased Nr ration resulted in the decreased content of several minerals in cucumber fruits [44]. Phosphorus content was reduced by salinity at low Nr. In general, salinity decreases the concentration of p in plant tissue [35,55], with several contradictions being present in previous studies, as it has been indicated that salinity either increased or has no effect on p uptake. Plant growing conditions, the plant type and even the cultivar play an important role in p accumulation [49]. In the present study, the reduced Mg content in the context of high salinity at Nr0.05 indicates competition between Mg and Na but not with ammonium ions. The nitrogen content of the fruits decreased with salinity at low and moderate Nr levels, which could be attributed to the reduced nitrate uptake and translocation into the plant under high salinity conditions. In the present study, the nitrogen content of the fruits was restored when nitrogen was supplied in ammonium form (Nr0.15). The application of an increased Nr ratio to hydroponically grown cucumber resulted in decreased Ca, Mg, K Cu, Mn and NO_3 content in cucumber fruits [44].

Low pH values are considered to be responsible for increased Al released from the perlite into the drainage solution; this is consistent with the results of the present study, which indicated the Al increment at low pH values (increased Nr). Iron content was increased by increasing the Nr under low salinity. This is consistent with Flores et al. [56], who, in the context of a tomato crop, indicated an increased correlation between nitrogen assimilation and the levels of Fe by adding nitrogen in ammonium form, resulting in reduced energy requirements for plant growth. Salt stress reduced the Cu content, and this has also been evidenced in salt-stressed maize grown in both soil [57] and solution cultures [58]. Manganese content was suppressed by salinity and a high Nr ratio (Nr0.10; Nr0.15). Salinity reduces Mn content in the shoot tissue of horticultural crops, for example, squash [59] and corn [57], according to most research, regardless of whether they were grown in soils or in solution cultures. However, some studies on tomato crops have shown that salinity either has no effect [60] or increases the Mn content [59] of leaves or shoot tissue.

Several quality characteristics of the tomato are influenced by salinity and the Nr ratio. Citric acid content in general significantly increased with salinity. The increased acidity of salt-stressed vegetable crops has been reported for tomatoes [35,41] and eggplants [42] and has been associated with an increase in the level of organic acids which are involved in the osmoregulation mechanism of plants under high salinity conditions [61]. The total soluble solid contents were also significantly higher under high salinity, and were not affected by either the Nr or by the storage temperatures. The beneficial effects of salinity on soluble solids content are consistent with prior research [40,62], though it should be noted that when results are expressed on a fresh weight basis, such an increase could be due to a concentration effect rather than an increase in total soluble solids [63].

Additionally, Ben and Kafkafi [64] previously reported that TSS in melon fruit remained the same when the melon plants exposed to different Nr levels. Fruit firmness was influenced by both salinity and the Nr ratio. More specifically, salinity had a positive effect on fruit firmness at the point of harvest, while the Nr ratio had a positive effect only under high salinity conditions or following storage. As expected, the dehydration due to weight loss of the tomatoes stored at room temperature was much more intense than the ones that were stored at 12 °C and remained unaffected by salinity or Nr.

The vitamin C content of the fruits was higher due to salinity at the point of harvest, while an increased Nr ratio decreased the vitamin C content under high salinity conditions following storage for a week at 12 °C. Storage at room temperature diminished the effect of both salinity and Nr on vitamin C content. These enhancing effects of salinity on fruit firmness and vitamin C content of tomato are in agreement with the results of other investigators [65,66]. The effect on concentration impacts originates from reduced water content in the fruit due to plant adaptation to salinity. Other investigators [32] obtained fruits with lower firmness (softer fruits) when the Nr was increased by more than 0.15.

Plant stress factors such as salinity play an important role in terms of the antioxidant content of fruits. The total phenol content was increased by salinity at low Nr at the point of harvest. Moreover, the content of lycopene and β -carotene increased in fruits derived from the saline treated plants following storage at 12 °C. However, the Nr levels seemed not to affect the fruits' antioxidant-related measures (vitamin C, lycopene, β -carotene and total phenolics) at the harvesting stage, but decreased vitamin C and lycopene levels were found in fruits stored at 12 °C for 1 week in relation to high salinity and Nr ratio treatments. Tomato fruit has been found to contain more total phenols when grown in saline conditions [62], while lavender plants have been found to contain more total phenols when grown in salinity levels of more than 50 mM NaCl [52]. Similar to the findings of our study, Serio et al. found that lycopene content was unchanged in tomato fruit when plants were exposed to salinity [67]; however, Moya et al. found the opposite [62]. These contradictory outcomes could be related to the severity of salinity stress, since different NaCl levels were applied in these studies and mild stress (4.5 dS m⁻¹) appears to be favorable to tomato fruit lycopene content when compared to more severe stress (6.0 dS m⁻¹).

Moderate salinity enhances fruit carotenoid content in tomato [66], possibly via the activation of the biosynthetic pathway, inducing up-regulation of the genes encoding the enzymes related to lycopene levels [68]. Concerning the tomatoes that were stored at room temperature, only β -carotene content was increased due to the increased EC of the nutrient solution. Similar results regarding salinity and increased antioxidant activity were observed in pepper fruit but only in mature red fruits while at the green (immature) stage, the results were controversial [69]. Furthermore, in previous studies, the antioxidant status of tomatoes increased after ten days of storage in similar conditions (15 °C and 25 °C respectively), whereas the lycopene content was 1.8-fold higher than that of refrigerated tomatoes (7 °C). This could be explained due to a reduction in the water content of the fruits stored as well as the slowdown of the fruits' metabolism in chilled temperatures [70].

Regarding the color of tomatoes, salinity decreased fruit lightness at the point of harvest, as reported in previous studies on tomato [66], while the values a^* (red colour) and b^* (yellow colour) were not affected. After storage at 12 °C for a week, all color factors were improved with increasing Nr, while salinity in general had a negative effect. Following storage at 25 °C, salinity decreased the lightness of the tomatoes at high and low Nr. This is not consistent with research in which salinity in general improved color factors due to the increase of tomato pigments (lycopene and β -carotene) [30], although, in our study, only β -carotene increased at high salinity levels and lycopene was affected to a lesser extent, with this being in agreement with previous reports on tomatoes under saline conditions [35].

Following organoleptic tests, the increased Nr improved the fruits' appearance, color, marketability and aroma, with significant higher scores associated with fruit color

according to the panelists, and with the fruits cultivated under Nr0.10 and Nr0.15 getting more positive records than the Nr0.05 sample. The fruit of the 50 mM NaCl treated plants performed better in terms of aroma and sweetness scores. In addition, the texture, satisfaction level and marketability of the salinity treated fruit received the best numerical score. The results of the questionnaires back up previously stated findings that salinity has a positive effect on some quality measures of tomato fruit. According to research, increased salt levels in NS result in increased fruit dry matter, sugars and organic acid content, as well as improving flavor and maintaining firmness in hydroponically produced tomato plants [71,72]. The findings of our study highlight the relevance of organoleptic tests in addition to chemical composition investigations because consumer approval is critical for fruit marketability.

The storage temperature did not greatly affect fruit metabolism due to a short duration of 7 days and the ripening stage of the fruits (red fruits). Indeed, fruits stored at chilled conditions at 12 °C, in comparison to those stored at 25 °C, revealed lower fruit weight loss, as chilled conditions delayed the plant's metabolism and the evapotranspiration of the plant tissue. Neither storage at 12 °C or 25 °C, nor Nr levels affected the total soluble solids content of the fruits, while salinity increased TSS. The vitamin C content at 12 °C was increased at low Nr levels with the application of salinity, which is in agreement with previous studies which reported an increase in vitamin C content in tomato fruit grown hydroponically under saline conditions [41]. However, Vitamin C was unaffected in the case of fruits stored at 25 °C. Storage for 7 d showed that the content of vitamin C, lycopene, β -carotene and total phenols was almost the same or slightly higher than prior to storage, regardless of storage temperature, Nr level and salinity level (with exception of lycopene and β -carotene content, which were increased with the application of salinity at 25 °C), a finding which is very important for the preservation of fruit quality. Similar to the present results, it has been reported in previous studies that lycopene content in tomatoes stored at 15 °C and 25 °C was higher (1.8-fold) than in refrigerated tomatoes (7 °C) [70]. Fruit quality is influenced by genotype. Manzo et al. found substantial differences in post storage quality between short and long shelf-life cultivars [73], while Petric et al. emphasized the need of maintaining good post-storage quality at the right temperature [74]. Following storage at 25 °C, no differences were observed between the salinity levels concerning color attributes (L^* , a^* , b^*), owing to the fact that all the tested fruit were harvested at the mature red stage and that no further color development should be expected after this stage, but also the fact that lycopene content, which is responsible for the red color of the fruits, remained unaffected for all the tested treatments [73].

5. Conclusions

This study explored the effects of salinity on hydroponically grown tomato plants and evaluated whether the application of different Nr levels affected the negative effects posed by NaCl stress. Moreover, the postharvest performance of fruit stored either in chilled conditions or at shelf life was examined. Despite the negative effects of salinity on plant growth as well as yield, the fruit fresh weight was only decreased significantly with the application of high Nr. Salinity and high ammonium ratios (high Nr) induced nutrient imbalances in the fruit, such as Ca, and affected BER in fruits, the symptoms of which appeared on the fruits above the third cluster after five months of tomato cultivation (March) and did not affect the marketability of the examined fruits. It was profound that salinity improved and/or maintained various fruit quality related attributes, including total acid content, total soluble solids content, firmness and vitamin C content, while an increased Nr also preserved fruit firmness after storage. These characteristics are considered favorable by consumers and are important in terms of fruit storability and postharvest life. Lycopene, β -carotene and vitamin C were increased by salinity during harvesting or following storage. The organoleptic control showed that tomato fruits cultivated in an increasing Nr or under high salinity were more preferable to consumers due to the improvement of some organoleptic characteristics. Salinity enhances fruit quality and

improves the organoleptic characteristics of the crop, while an appropriate Nr ratio may restrict the negative effects of salinity on the nutritional status of plants by regulating the pH in hydroponic systems. More research on this topic is needed in order to look into the interactions between different Nr ratios and NaCl salinity in other fruits and vegetables, and also to determine whether the application of minerals and which application methods are effective tools to reduce the harmful impacts of salt.

Supplementary Materials: The following supporting information can be downloaded at: www.mdpi.com/article/10.3390/agronomy12020386/s1, Figure S1. Average, minimum and maximum air relative humidity (RH) and temperature (°C) during the experiment. Figure S2. The effect of salinity (Sal) levels (LS: Low Salinity of 0 mM NaCl; HS: High Salinity of 50 mM NaCl) and the application of NH₄/N_{total} ratios (Nr0.05, Nr0.10, and Nr0.15) on the fluctuation of the electrical conductivity (EC; mS cm⁻¹), pH and nitrates content of the drainage solution during cultivation.

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