

## MICROALGAE POTENTIAL IN THE CAPTURE OF CO<sub>2</sub> EMISSION

**Francesca Frongia**

Department of Life Sciences, University of Modena and Reggio Emilia,  
via G. Campi 103, 41125 - Modena, Italy

 <https://orcid.org/0000-0002-1218-6661>

**Laura Arru**

Department of Life Sciences, University of Modena and Reggio Emilia  
via G. Amendola 2, 42122 - Reggio Emilia, Italy

 <https://orcid.org/0000-0002-1807-1689>

**Maria Rita Cramarossa**

Department of Life Sciences, University of Modena and Reggio Emilia  
via G. Campi 103, 41125 - Modena, Italy

 <https://orcid.org/0000-0002-6066-7355>

**Luca Forti\***

Department of Life Sciences, University of Modena and Reggio Emilia  
via G. Campi 103, 41125 - Modena, Italy, [luca.forti@unimore.it](mailto:luca.forti@unimore.it)

 <https://orcid.org/0000-0002-5662-7756>

*Article history: Received 27 September 2021, Accepted 3 October 2021, Available online 4th October 2021*

### Abstract

In a perspective projected to reduce the atmospheric concentration of greenhouse gases, in which carbon dioxide is the master, the use of microalgae is an effective and decisive response. The review describes the bio circularity of the process of abatement of carbon dioxide through biofixation in algal biomass, highlighting the potential of its reuse in the production of high value-added products.

### Keywords

Microalgae, CO<sub>2</sub> biofixation, circular bioeconomy

### Introduction

#### Greenhouse effect and decarbonisation strategies

Earth's temperature is rising almost 0.15°- 0.20°C per decade since 1975, causing an increase of 1° C since 1880 [1,2]. Scientists believe that this trend cannot be explained uniquely by natural changes, but it has to consider the influence of other factors, first of all the effect of the anthropogenic emission of large quantities of greenhouse gases (GHGs). GHGs include CO<sub>2</sub>, methane, nitrous oxide, hydrofluorocarbons, chlorofluorocarbons, etc. Considering that many GHGs can remain in the atmosphere for tens or hundreds of years, creating serious consequences even in the long term, the situation is even more critical. In order to counteract this trend, and to protect the environment, many Countries pledged to enter into agreements such as the Kyoto Protocol (1979) and the Paris Climate Agreement (2015). Among the GHGs, CO<sub>2</sub> is considered to have the greatest negative impact on global warming. The rise of atmospheric CO<sub>2</sub> concentration has been about 2 ppm/year in the last ten years, and in 2019 was almost 40% higher (399 ppm) than that measured during the Industrial Revolution (280 ppm) [3,4]. Since CO<sub>2</sub> is one of those gases with exceptional persistence in the atmosphere (even if also non-CO<sub>2</sub> greenhouse gases could have a negative role as well) [5], transported by the wind and spreading all over the world, it can be responsible for global warming virtually irreversible for more than 1,000 years. Although still controversial and debated [6,7], contribution of fossil fuels-burning power plants seems to be about 40% of the total CO<sub>2</sub> global emission [8], to which burning of fossil fuels for transport must be added [9]. In heavy industries, CO<sub>2</sub> emissions are a by-product produced through chemical reactions that do not involve combustion, but also CO<sub>2</sub> emissions indirectly produced by electricity generation must be taken into account [9]. Some predictive studies show how the failure to reduce the GHGs emissions will affect the

atmospheric temperature in coastal areas by 2°C by 2050 and by 4°C by 2100, while in inland areas the temperature will increase by 4°C by 2050 and by 7°C by 2100.

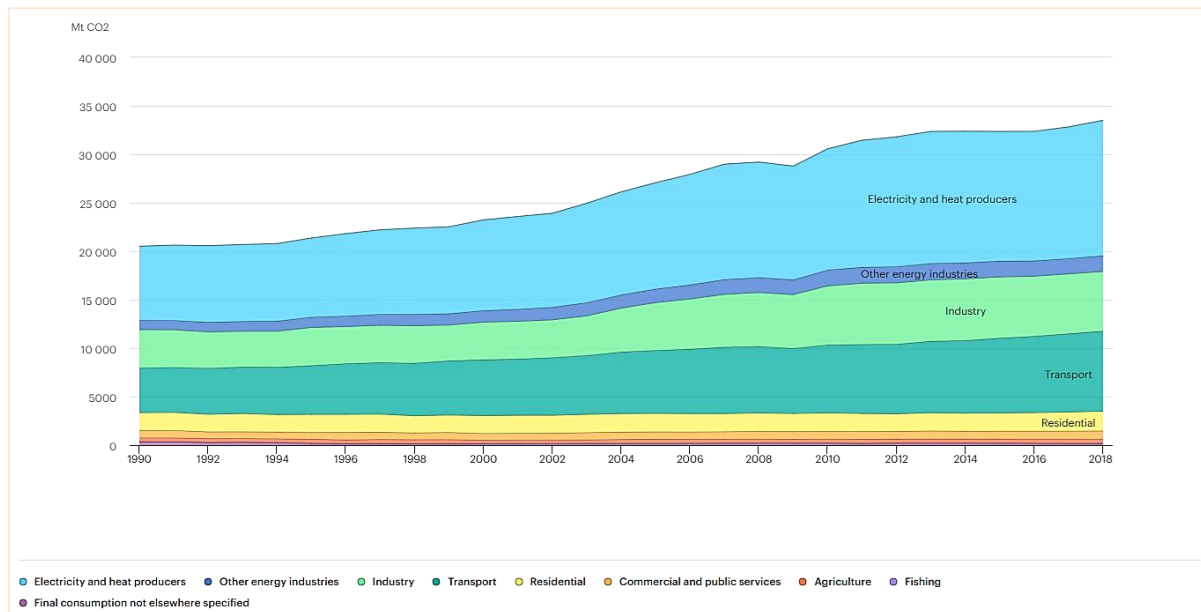


Fig.1. CO<sub>2</sub> emissions by sector. Source: <https://www.iea.org/articles/global-CO2-emissions-in-2019>

Although the combustion of fossil fuels is currently the cheapest form of energy production, it is one of the main factors contributing to CO<sub>2</sub> emissions into the atmosphere [10].

Several studies focus on finding solutions both to reduce atmospheric CO<sub>2</sub> pollution (by removing it from atmosphere or by reducing industrial emission) and to give alternatives to fossil fuels [10,11].

The main chemical processes to reduce CO<sub>2</sub> presence in the atmosphere capture are absorption by amino solvents to treat industrial air flows [11,12] and adsorption of CO<sub>2</sub> molecules to a solid phase [13].

However, both chemical processes are economically disadvantageous, due to the energy consumption for solvent regeneration in the first case, and for separation of pollutants from adsorbents in the second one [13].

#### Microalgae CO<sub>2</sub> capture and utilization

In recent years, the concept of circular bioeconomy has emerged, focusing on the sustainable valorisation and transformation of biomass in production chains converting agro-industrial wastes into high added value products and use of renewable resources into products with a high added value [14]. The use of versatile and environmentally friendly photosynthetic organisms such as microalgae represents a promising approach in the development of such closed loop systems [15,16].

In Nature, Microalgae play a key role in the mitigation of environmental carbon and in bioremediation thanks to their high photosynthetic efficiency -about 40% more than terrestrial plants-, and to the significant sequestration of CO<sub>2</sub>: about 1 kg of microalgae consumes 1.83 kg of CO<sub>2</sub> and represents 40% of the global CO<sub>2</sub> sequestration [16].

Microalgae has been studied not only to reduce CO<sub>2</sub> from the atmosphere or from flue gas emissions [17], but also to be applied in wastewater treatments [18] to generally lower pollutants and converting them into organic biomass rich in lipids, proteins, and other high value-added compounds [19] for energetic applications (biodiesel, biogas), food (human and animal feed), pharmaceuticals and cosmetics production [20].

Optimisation of carbon fixation efficiency by microalgae should take into account many variables.

It should be considered the use of the most suitable strain in relation with the different mediums to be treated, adjusting operating conditions as physicochemical and hydrodynamic parameters [8]. Good characteristics are tolerance to high CO<sub>2</sub> concentrations, high temperatures, and presence of toxic compounds such as NO<sub>x</sub>, SO<sub>x</sub>, hydrogen sulfide. For this reason the search for appropriate microalgae strain is one of the main concerns regarding the improvement of CO<sub>2</sub> capture processes [21]. Several microalgae such as *Chlorella spp.*,

*Scenedesmus spp.*, *Chlorococcum spp.*, *Nannochloropsis spp.* have shown good ability to capture the CO<sub>2</sub> present in effluents similar to those emitted by industrial activities [22–25].

The supply of nutrients plays a fundamental role in the regulation of key metabolic processes related to both CO<sub>2</sub> fixation and biomass synthesis. Nutrients for microalgae cultivation include carbon, nitrogen, phosphorus, magnesium, sulfur and trace element [26].

When microalgae are grown in autotrophy, light is crucial for photosynthetic activity, being the energy source. The growth of microalgae and the fixation of CO<sub>2</sub> depend on both the light-dark cycle and the intensity of the light, but this is not a universal rule. CO<sub>2</sub> fixation by *Aphanothece microscopica Nägeli* and by *Nannochloropsis sp.* achieves approximately 100% efficiency with continuous illumination of the culture [27]. There is also evidence that shorter lighting periods lead to a reduction in biomass production and carbon dioxide fixation [27]. Another crucial parameter to enhance both CO<sub>2</sub> capture and cell growth is temperature. The solubility of CO<sub>2</sub> depends on temperature and is reduced at high temperatures. Besides, also the affinity of RuBisCO (the key enzyme for CO<sub>2</sub> fixation) for CO<sub>2</sub> decreases as the temperature increases. In any case, the effect of temperature on the reaction metabolic rate depends on the strain being considered [28].

The value for pH determines the form in which dissolved inorganic carbon (DIC) exists in water. CO<sub>2</sub>, HCO<sub>3</sub><sup>-</sup>, CO<sub>3</sub><sup>2-</sup> and H<sub>2</sub>CO<sub>3</sub>, can all be found in water, but only CO<sub>2</sub> and HCO<sub>3</sub><sup>-</sup> can be used by microalgal cells. The acidic pH favors the formation of H<sub>2</sub>CO<sub>3</sub>, whilst the alkaline one allows the assimilation of NO<sub>3</sub><sup>-</sup> and HCO<sub>3</sub><sup>-</sup>. It is generally preferred to cultivate microalgae in alkaline conditions due to the positive effect on CO<sub>2</sub> solubilization [29].

#### Microalgae cultivation systems for CO<sub>2</sub> capture

Microalgae can be produced either in open (outdoor) or closed systems (photobioreactor).

Open systems for microalgae growth are the cheapest, but they are also the most prone to the effects of external factors and contamination. Closed cultivation systems, also known as photobioreactors (PBR), despite being more expensive, allow the strict control of cultivation parameters, favoring the most suitable conditions for the growth of microalgae [30].

Typical configurations for CO<sub>2</sub> capture systems are tubular or flat PBRs.

Tubular PBRs are commonly used for CO<sub>2</sub> capture due to good scalability and low contamination risk. They are divided into horizontal and vertical tubular reactors [30]. The main advantages of horizontal PBRs are the large surface exposed to light and the relatively low CO<sub>2</sub> losses [31]; on the other hand, an important disadvantage is the accumulation of oxygen in the culture medium, that can lead to a decrease in biomass production and CO<sub>2</sub> uptake [29]. Vertical PBRs, on the other hand, are advantageous for their high mass transfer and good mixing, which has made them suitable systems for biomass production and CO<sub>2</sub> sequestration; however, they have a small illumination area, which can induce a decrease in the growth rate [32].

Another configuration commonly used for CO<sub>2</sub> capture with microalgae are flat plate PBRs. An important advantage is the short light path and high illumination area. An important drawback is the low mixing and high shear stress [32].

#### Effect of flue gas compounds on microalgae

In order to apply the biofixation of microalgae to industrial power plants or fuel gases it is necessary to understand the influence of combustion gas compounds on microalgae (Tab. 1).

In fact, in addition to the CO<sub>2</sub> contained in about 10/15% in coal-fired power plants and 5/6% in natural gas-fired power plants, nitric oxide (NO) and nitrogen dioxide (NO<sub>2</sub>) are present in flue gases, as well as SO<sub>x</sub> [33].

In combustion gases the level of NO<sub>x</sub> emission varies from 90/95% of NO and 5/10% of NO<sub>2</sub>. If the NO concentration is very low, it is transformed into NO<sub>2</sub> and absorbed as a nitrogen source. However, the increase in NO concentration may lead to a decrease in the growth rate for some microalgal species [8].

SO<sub>x</sub> are produced by burning hydrogen sulfide, sulfur, or organosulfur compounds. It is well known that the presence of SO<sub>2</sub> strongly inhibits microalgae growth. Inhibitory effects of SO<sub>2</sub> on microalgae growth can be attributed to increased acidity, which leads to cell death. A pH control in the growth medium would maintain algal growth unvarying in the presence of SO<sub>2</sub> [8].

Concerning the interaction between microalgae and other compounds as unburnt hydrocarbons, O<sub>2</sub>, N<sub>2</sub>, C<sub>x</sub>H<sub>x</sub>, H<sub>2</sub>O, CO, aerosols, heavy metals, and particulate matter, they have yet to be studied in detail.

Table 1. Inhibition effects of microalgae species cultivated using flue gas with SO<sub>x</sub> and NO<sub>x</sub> compounds. Source: [4]

Microalgal species	CO <sub>2</sub> % (v/v)	NO <sub>x</sub> (ppm)	SO <sub>x</sub> (ppm)	Source	Inhibitory effect	Cultivation system	References
Nannochloropsis limnetica	10	-	25	Real flue gas from rice husk emission	Inhibited	Bubble column	[34]
	3	-	11	Real flue gas from rice husk emission	Inhibited	Bubble column	[34]
Chlorella sp.	8-10	38	3.8	Real flue gas from co-generator units	No inhibition	Bubble column	[35]
	6-8	37	-	Real flue gas from combustion of natural gas from boiler	No inhibition	Open thin layer PBR	[36]
	23	78	87	Real flue gas from coke oven of steel plant	No inhibition	Double set PBR	[37]
Chlorella sp. MTF-15 6-8	25	70-80	80-90	Real flue gas from coke oven of steel plant	Slight inhibition	Column-type glass-fabricated PBR	[22]
	26	8-10	15-20	Real flue gas from coke oven of steel plant	Slight inhibition	Column-type glass-fabricated PBR	[22]
	24	25-30	15-20	Real flue gas from coke oven of steel plant	Slight inhibition	Column-type glass-fabricated PBR	[22]
Scenedesmus sp.	18	150	200	Real flue gas from combustion chamber of coke oven	No inhibition	Airlift	[38]
Mixed culture of Scenedesmus sp., Chlorella sp., Nitzschia sp., Chlamydomonas sp., Oocystis sp. & Protoderma sp.	7.5	77	-	Real flue gas from combustion of natural gas in manure-drying motors	No inhibition	High-rate algal pond	[39]

#### Applications of CO<sub>2</sub> capture and utilization by microalgae

To date, there are few examples of commercial applications of microalgae for the capture of CO<sub>2</sub> in a biorefinery concept, due to the high process costs [40].

Here are some examples of successful pilot scale applications.

The first company in the world to use the exhaust gases of a power plant for seaweed farming was Seabiotic, in Israel. The first company in the world to use the exhaust gases of a power plant for seaweed farming was Seabiotic, in Israel. In 2006, this company, in collaboration with a coal-burning power plant in the city of

Ashkelon, developed a pilot plant with a pond area of 1000 m<sup>2</sup>, to test algae growth using CO<sub>2</sub> from flue gases. The plant produced around 7 tons of biomass per year from flue gases containing 12% vol CO<sub>2</sub>. Subsequently, the Hearol project, by Seambiotic, Yantai Hairong Electricity Technology and Penglai Weiyuan Science & Trading Ltd was developed, with the aim of using the exhaust gases generated by the Penglai coal-fired power plant to grow microalgae on a commercial scale [21].

In Germany, RWE has started a project in which combustion gases from the Niederaussem power plant are fed into an algae plant near the plant to convert CO<sub>2</sub> into biomass. The plant has been operational since 2008 on an area of 600 square meters and supplies about 6000 kg of algal biomass using about 12000 kg of CO<sub>2</sub> per year [41].

At the University of Kentucky in the United States, researchers cultivated *Scenedesmus acutus* in an 18,000-liter pilot-scale PBR system using exhaust gas derived from Duke Energy's East Bend Power Plant, Kentucky. The exhaust gas was initially pre-treated to reduce SO<sub>x</sub> and NO<sub>x</sub> and then pumped into the culture systems. The mean growth rate recorded during the study was 32.9 g m<sup>-2</sup> d<sup>-1</sup> [42].

The Daqi project in China is capable of capturing 110 tons of CO<sub>2</sub> with microalgae and producing respectively 20 tons of biodiesel and 5 tons of protein per year [21].

Eni, an Italian multinational active in fuel and natural gas sectors, started in 2019 the experimental plant for the CO<sub>2</sub> biofixation from microalgae thanks to the aid of artificial led light. The process, through CO<sub>2</sub> biofixation by microalgae, allow to enhance CO<sub>2</sub> as a raw material and to transform it in high value products such as algal flour for food and nutraceutical markets or biooil, which can sequentially be used as feedstock in biorefineries. The pilot plant consisting of 4 PBRs is integrated with renewable energy sources, and has achieved daily productivity data of biomass that could lead to 1 hectare plant producing 500 tons of biomass per hectare per year, trapping about 1000 tons of CO<sub>2</sub> [43].

Between 2011 and 2013, the Green Mission project (a collaboration between the State of Brandenburg, the European Union and Vattenfall) followed by the Green Vision project, tested an algal farm facility using the combustion gas obtained from the Senftenberg power plant (Brandenburg). The facility is one of the largest closed algal cultivation systems globally with a volume of 48000 L, with an increased biomass productivity using raw combustion gas [21].

#### Environmental and economic impacts aspects of CO<sub>2</sub> capture and utilization by microalgae

Microalgae are receiving increasing attention due to their potential application to the capture and use of CO<sub>2</sub> in the renewable energy sector. The use of microalgae has several advantages over the use of other plant raw materials, including a high photosynthetic conversion, a high capacity to produce different raw materials for biofuels, a high environmental bioremediation capacity (CO<sub>2</sub> fixation from atmosphere or from combustion gases, water purification) and the non-competitiveness for the use of land for food crops. Furthermore, net CO<sub>2</sub> emissions are assumed to be essentially zero if the CO<sub>2</sub> released from the biofuel from microalgae can be recycled and reused for microalgae cultivation. Consequently, these advantages and potential make microalgae suitable for solving CO<sub>2</sub> and energy reduction problems [33].

CO<sub>2</sub> capture through a biorefinery approach with microalgae cultivation is economically feasible, as waste from power plants or other industrial plants is reused [44] and residual microalgae biomass, rich in proteins and carbohydrates, can be used as a carbon source for the production of bioelectricity, biohydrogen and also fatty acids and other molecules, which can in turn be used to produce bioplastics [45].

A very promising algae for capturing CO<sub>2</sub> from flue gases is chlorella. Studies have shown that Chlorella could grow in an atmosphere containing up to 40% (v / v) CO<sub>2</sub>, with a CO<sub>2</sub> fixation rate between 0.73 and 2.22 g/L/day. [4,45].

A very important aspect concerns the fact that the NO<sub>x</sub> and SO<sub>x</sub> compounds present in the fed CO<sub>2</sub> stream do not affect the production of *Chlorella* biomass. [4,46]. In fact, some studies reported how these pollutants are metabolized at the cellular level by microalgae in culture [22,47]. Some microalgal species could therefore be potentially useful for bioremediation of CO<sub>2</sub>, but also of other greenhouse gases [47].

Therefore, the main purpose is the conversion of CO<sub>2</sub> into different products, thus closing the carbon cycle and contributing to the bioeconomy of the process [48].

The EU emission trading scheme (ETS) is a milestone of EU policy to tackle climate change and a key tool for cost-effectively reducing greenhouse gas emissions. It is the world's leading CO<sub>2</sub> market and continues to be the largest. The ETS is a free trade program where, one the state has set the limit for the environmental load of carbon dioxide that can be emitted distributes to companies an amount of exchangeable certificates capable of covering the fixed quantity. Those who are unable to cover their emissions incur the payment of financial penalties. The most important parameter of all is therefore the method of assignment of certificates. In order to

achieve climate neutrality in the EU by 2050, including the interim target of a net greenhouse gas emission reduction of at least 55% by 2030, the Commission proposes to review and possibly extend the scope of the EU ETS system.

The main impact produced by the ETS is represented by the cost that companies will have to face to obtain the necessary permits to cover their emissions, so being the cost of innovative technologies for the reduction of CO<sub>2</sub> emissions presumably lower than the expected cost of purchasing new certificates on the market, companies will feel encouraged to adopt new technologies.

The possibility of acquiring a valid and efficient technological innovation that allows to reduce, at least in part, the polluting CO<sub>2</sub> emissions by channelling the latter into photobioreactors to produce biomass, must make us reflect and think about a whole series of other important benefits that can be drawn from the use of this technology. Firstly, part of the cost currently incurred only for the "virtual" compensation of CO<sub>2</sub>, which continues to flow into the atmosphere, would be invested. On the other hand, the biosynthesis operated by microalgae intervenes in this process by sequestering and transforming inorganic carbon (CO<sub>2</sub>) into organic carbon and returning molecular oxygen to the environment.

### Impact

Microalgae are capable to convert CO<sub>2</sub> from the atmosphere and from flue gas, leading to a reduction of GHGs emissions. Thanks to this, the greenhouse effect will be reduced, and therefore also global warming, achieving a healthier environment. Worldwide emissions of CO<sub>2</sub>, about 40 Gt per year, are too high compared to about 14000 tons of microalgae biomass commercialized (about 27000 tons of CO<sub>2</sub> biofixed). This incredibly low contribution highlights the need to boost productivity and improve existing technologies in order to generate more microalgal biomass capable of capturing more CO<sub>2</sub> [29]. One of the most important aspects related to the capture of CO<sub>2</sub> from microalgae is the reuse of biofixed biomass for energy production, considering the need to meet global energy demand. Moreover, CO<sub>2</sub> biofixation using microalgae is combined with other processes like wastewater treatment: this is advantageous to offer more economical feasibility and environmentally sustainability.

The transition from pilot to industrial scale is difficult to apply as microalgal cells are exposed to hostile circumstances, resulting in a reduction in CO<sub>2</sub> biofixation and product yield. Therefore, it is necessary to integrate the use of promising algal strains, optimized process parameters, targeted cultivation systems, to ensure economic and environmental feasibility on a large scale.

### Conflict of interest

There are no conflicts to declare.

### Acknowledgments

This research has not been supported by any external funding.

### References

- [1] T.M.L. Wigley, P.D. Jones, P.M. Kelly, Global warming?, *Nature*. 291 (1981) 285. <https://doi.org/10.1038/291285a0>.
- [2] N.J.L. Lenssen, G.A. Schmidt, J.E. Hansen, M.J. Menne, A. Persin, R. Ruedy, D. Zyss, Improvements in the GISTEMP Uncertainty Model, *J. Geophys. Res. Atmos.* 124 (2019) 6307–6326. <https://doi.org/10.1029/2018JD029522>.
- [3] M. Molazadeh, H. Ahmadzadeh, H.R. Pourianfar, S. Lyon, P.H. Rampelotto, The use of microalgae for coupling wastewater treatment with CO<sub>2</sub> biofixation, *Front. Bioeng. Biotechnol.* 7 (2019). <https://doi.org/10.3389/fbioe.2019.00042>.
- [4] W.Y. Cheah, P.L. Show, J.S. Chang, T.C. Ling, J.C. Juan, Biosequestration of atmospheric CO<sub>2</sub> and flue gas-containing CO<sub>2</sub> by microalgae, *Bioresour. Technol.* 184 (2015) 190–201. <https://doi.org/10.1016/j.biortech.2014.11.026>.
- [5] S. Solomon, J.S. Daniel, T.J. Sanford, D.M. Murphy, G.K. Plattner, R. Knutti, P. Friedlingstein, Persistence of climate changes due to a range of greenhouse gases, *Proc. Natl. Acad. Sci. U. S. A.* 107 (2010) 18354–18359. <https://doi.org/10.1073/pnas.1006282107>.
- [6] J. Singh, D.W. Dhar, Overview of carbon capture technology: Microalgal biorefinery concept and state-of-the-art, *Front. Mar. Sci.* 6 (2019) 1–9. <https://doi.org/10.3389/fmars.2019.00029>.
- [7] A. Boretti, Covid 19 impact on atmospheric CO<sub>2</sub> concentration, *Int. J. Glob. Warm.* 21 (2020) 317–323. <https://doi.org/10.1504/IJGW.2020.108686>.

- [8] B. Zhao, Y. Su, Process effect of microalgal-carbon dioxide fixation and biomass production: A review, *Renew. Sustain. Energy Rev.* 31 (2014) 121–132. <https://doi.org/10.1016/j.rser.2013.11.054>.
- [9] Intergovernmental Panel on Climate Change, *Climate Change 2014 Mitigation of Climate Change*, Cambridge University Press, Cambridge, 2014. <https://doi.org/10.1017/cbo9781107415416>.
- [10] X. Li, A. Kraslawski, Conceptual process synthesis: Past and current trends, *Chem. Eng. Process. Process Intensif.* 43 (2004) 583–594. <https://doi.org/10.1016/j.cep.2003.05.002>.
- [11] K. Damen, M. Van Troost, A. Faaij, W. Turkenburg, A comparison of electricity and hydrogen production systems with CO<sub>2</sub> capture and storage. Part A: Review and selection of promising conversion and capture technologies, *Prog. Energy Combust. Sci.* 32 (2006) 215–246. <https://doi.org/10.1016/j.pecs.2005.11.005>.
- [12] J.N. Knudsen, J.N. Jensen, P.J. Vilhelmsen, O. Biede, Experience with CO<sub>2</sub> capture from coal flue gas in pilot-scale: Testing of different amine solvents, in: *Energy Procedia*, 2009: pp. 783–790. <https://doi.org/10.1016/j.egypro.2009.01.104>.
- [13] Z. Qiao, Z. Wang, C. Zhang, S. Yuan, Y. Zhu, J. Wang, PVAm–PIP/PS composite membrane with high performance for CO<sub>2</sub>/N<sub>2</sub> separation, *AIChE J.* 59 (2012) 215–228. <https://doi.org/10.1002/aic>.
- [14] P. Stegmann, M. Londo, M. Junginger, The circular bioeconomy: Its elements and role in European bioeconomy clusters, *Resour. Conserv. Recycl.* X. 6 (2020) 100029. <https://doi.org/10.1016/j.rcrx.2019.100029>.
- [15] A. Shahid, A. Zafar Khan, T. Liu, S. Malik, I. Afzal, M.A. Mehmood, Production and Processing of Algal Biomass, in: *Algae Based Polym. Blends, Compos. Chem. Biotechnol. Mater. Sci.*, Elsevier, 2017: pp. 273–299. <https://doi.org/10.1016/B978-0-12-812360-7.00007-0>.
- [16] C.W.W. Ng, R. Tasnim, J.L. Coe, Effects of atmospheric CO<sub>2</sub> concentration on soil-water retention and induced suction in vegetated soil, *Eng. Geol.* 242 (2018) 108–120. <https://doi.org/10.1016/j.enggeo.2018.06.001>.
- [17] M.K. Lam, K.T. Lee, Renewable and sustainable bioenergies production from palm oil mill effluent (POME): Win-win strategies toward better environmental protection, *Biotechnol. Adv.* 29 (2011) 124–141. <https://doi.org/10.1016/j.biotechadv.2010.10.001>.
- [18] C. Grandclément, I. Seyssiecq, A. Piram, P. Wong-Wah-Chung, G. Vanot, N. Tiliacos, N. Roche, P. Doumenq, From the conventional biological wastewater treatment to hybrid processes, the evaluation of organic micropollutant removal: A review, *Water Res.* 111 (2017) 297–317. <https://doi.org/10.1016/j.watres.2017.01.005>.
- [19] J. Masojídek, G. Torzillo, M. Koblížek, Photosynthesis in Microalgae, in: *Handb. Microalgal Cult. Appl. Phycol. Biotechnol.* Second Ed., John Wiley & Sons, Ltd, Oxford, UK, 2013: pp. 21–36. <https://doi.org/10.1002/9781118567166.ch2>.
- [20] I. Afzal, A. Shahid, M. Ibrahim, T. Liu, M. Nawaz, M.A. Mehmood, Microalgae: A Promising Feedstock for Energy and High-Value Products, in: *Algae Based Polym. Blends, Compos. Chem. Biotechnol. Mater. Sci.*, Elsevier, 2017: pp. 55–75. <https://doi.org/10.1016/B978-0-12-812360-7.00003-3>.
- [21] A.K. Vuppaladadiyam, J.G. Yao, N. Florin, A. George, X. Wang, L. Labeeuw, Y. Jiang, R.W. Davis, A. Abbas, P. Ralph, P.S. Fennell, M. Zhao, Impact of Flue Gas Compounds on Microalgae and Mechanisms for Carbon Assimilation and Utilization, *ChemSusChem.* 11 (2018) 334–355. <https://doi.org/10.1002/cssc.201701611>.
- [22] C.Y. Kao, T.Y. Chen, Y. Bin Chang, T.W. Chiu, H.Y. Lin, C. Da Chen, J.S. Chang, C.S. Lin, Utilization of carbon dioxide in industrial flue gases for the cultivation of microalga *Chlorella sp.*, *Bioresour. Technol.* 166 (2014) 485–493. <https://doi.org/10.1016/j.biortech.2014.05.094>.
- [23] P. Kandimalla, S. Desi, H. Vurimindi, Mixotrophic cultivation of microalgae using industrial flue gases for biodiesel production, *Environ. Sci. Pollut. Res.* 23 (2016) 9345–9354. <https://doi.org/10.1007/s11356-015-5264-2>.
- [24] M. Ota, M. Takenaka, Y. Sato, R. Lee Smith, H. Inomata, Effects of light intensity and temperature on photoautotrophic growth of a green microalga, *Chlorococcum littorale*, *Biotechnol. Reports.* 7 (2015) 24–29. <https://doi.org/10.1016/j.btre.2015.05.001>.
- [25] L. Meier, R. Pérez, L. Azócar, M. Rivas, D. Jeison, Photosynthetic CO<sub>2</sub> uptake by microalgae: An attractive tool for biogas upgrading, *Biomass and Bioenergy.* 73 (2015) 102–109. <https://doi.org/10.1016/j.biombioe.2014.10.032>.
- [26] Y. Chisti, Biodiesel from microalgae, *Biotechnol. Adv.* 25 (2007) 294–306.
- [27] E. Jacob-Lopes, C.H.G. Scoparo, L.M.C.F. Lacerda, T.T. Franco, Effect of light cycles (night/day) on CO<sub>2</sub> fixation and biomass production by microalgae in photobioreactors, *Chem. Eng. Process. Process Intensif.*

- 48 (2009) 306–310. <https://doi.org/10.1016/j.cep.2008.04.007>.
- [28] A. Kumar, S. Ergas, X. Yuan, A. Sahu, Q. Zhang, J. Dewulf, F.X. Malcata, H. van Langenhove, Enhanced CO<sub>2</sub> fixation and biofuel production via microalgae: Recent developments and future directions, *Trends Biotechnol.* 28 (2010) 371–380. <https://doi.org/10.1016/j.tibtech.2010.04.004>.
- [29] M. Morales, L. Sánchez, S. Revah, The impact of environmental factors on carbon dioxide fixation by microalgae, *FEMS Microbiol. Lett.* 365 (2018). <https://doi.org/10.1093/femsle/fnx262>.
- [30] H.W. Yen, I.C. Hu, C.Y. Chen, J.S. Chang, Design of Photobioreactors for Algal Cultivation, in: *Biofuels from Algae*, Second, Elsevier Inc., 2013: pp. 23–45. <https://doi.org/10.1016/B978-0-444-59558-4.00002-4>.
- [31] L. Moraes, G.M. Rosa, A. Morillas España, L.O. Santos, M.G. Morais, E. Molina Grima, J.A.V. Costa, F.G. Acien Fernández, Engineering strategies for the enhancement of *Nannochloropsis gaditana* outdoor production: Influence of the CO<sub>2</sub> flow rate on the culture performance in tubular photobioreactors, *Process Biochem.* 76 (2019) 171–177. <https://doi.org/10.1016/j.procbio.2018.10.010>.
- [32] G. de M. Michele, K. da S. Cleber, A.H. Adriano, A.V.C. Jorge, Carbon dioxide mitigation by microalga in a vertical tubular reactor with recycling of the culture medium, *African J. Microbiol. Res.* 9 (2015) 1935–1940. <https://doi.org/10.5897/ajmr2015.7632>.
- [33] W. Klinthong, Y.H. Yang, C.H. Huang, C.S. Tan, A Review: Microalgae and their applications in CO<sub>2</sub> capture and renewable energy, *Aerosol Air Qual. Res.* 15 (2015) 712–742. <https://doi.org/10.4209/aaqr.2014.11.0299>.
- [34] S.R. Ronda, C. Kethineni, L.C.P. Parupudi, V.B.S.C. Thunuguntla, S. Vemula, V.S. Settaluri, P.R. Allu, S.K. Grande, S. Sharma, C.V. Kandala, A growth inhibitory model with SO<sub>x</sub> influenced effective growth rate for estimation of algal biomass concentration under flue gas atmosphere, *Bioresour. Technol.* 152 (2014) 283–291. <https://doi.org/10.1016/j.biortech.2013.10.091>.
- [35] F. Kaštánek, S. Šabata, O. Šolcová, Y. Maléterová, P. Kaštánek, I. Brányiková, K. Kuthan, V. Zachleder, In-field experimental verification of cultivation of microalgae *Chlorella* sp. using the flue gas from a cogeneration unit as a source of carbon dioxide, *Waste Manag. Res.* 28 (2010) 961–966. <https://doi.org/10.1177/0734242X10375866>.
- [36] J. Doucha, F. Straka, K. Lívanský, Utilization of flue gas for cultivation of microalgae (*Chlorella* sp.) in an outdoor open thin-layer photobioreactor, *J. Appl. Phycol.* 17 (2005) 403–412. <https://doi.org/10.1007/s10811-005-8701-7>.
- [37] S.Y. Chiu, C.Y. Kao, T.T. Huang, C.J. Lin, S.C. Ong, C. Da Chen, J.S. Chang, C.S. Lin, Microalgal biomass production and on-site bioremediation of carbon dioxide, nitrogen oxide and sulfur dioxide from flue gas using *Chlorella* sp. cultures, *Bioresour. Technol.* 102 (2011) 9135–9142. <https://doi.org/10.1016/j.biortech.2011.06.091>.
- [38] F.F. Li, Z.H. Yang, R. Zeng, G. Yang, X. Chang, J.B. Yan, Y.L. Hou, Microalgae capture of CO<sub>2</sub> from actual flue gas discharged from a combustion chamber, *Ind. Eng. Chem. Res.* 50 (2011) 6496–6502. <https://doi.org/10.1021/ie200040q>.
- [39] I. de Godos, J.L. Mendoza, F.G. Acien, E. Molina, C.J. Banks, S. Heaven, F. Rogalla, Evaluation of carbon dioxide mass transfer in raceway reactors for microalgae culture using flue gases, *Bioresour. Technol.* 153 (2014) 307–314. <https://doi.org/10.1016/j.biortech.2013.11.087>.
- [40] W. Zhou, J. Wang, P. Chen, C. Ji, Q. Kang, B. Lu, K. Li, J. Liu, R. Ruan, Bio-mitigation of carbon dioxide using microalgal systems: Advances and perspectives, *Renew. Sustain. Energy Rev.* 76 (2017) 1163–1175. <https://doi.org/10.1016/j.rser.2017.03.065>.
- [41] M. Dębowski, M. Zieliński, J. Kazimierowicz, N. Kujawska, S. Talbierz, Microalgae cultivation technologies as an opportunity for bioenergetic system development—advantages and limitations, *Sustain.* 12 (2020) 1–37. <https://doi.org/10.3390/su12239980>.
- [42] M.H. Wilson, J. Groppo, A. Placido, S. Graham, S.A. Morton, E. Santillan-Jimenez, A. Shea, M. Crocker, C. Crofcheck, R. Andrews, CO<sub>2</sub> recycling using microalgae for the production of fuels, *Appl. Petrochemical Res.* 4 (2014) 41–53. <https://doi.org/10.1007/s13203-014-0052-3>.
- [43] S.D. Milanese, Eni sviluppa una innovativa tecnologia per la biofissazione della CO<sub>2</sub> con luce artificiale, (2020).
- [44] L. Brennan, P. Owende, Biofuels from microalgae—a review of technologies for production, processing, and extractions of biofuels and co-products, *Renew. Sustain. Energy Rev.* 14 (2010) 557–577.
- [45] S. Venkata Mohan, J.A. Modestra, K. Amulya, S.K. Butti, G. Velvizhi, A Circular Bioeconomy with Biobased Products from CO<sub>2</sub> Sequestration, *Trends Biotechnol.* 34 (2016) 506–519. <https://doi.org/10.1016/j.tibtech.2016.02.012>.
- [46] S.P. Singh, P. Singh, Effect of CO<sub>2</sub> concentration on algal growth: A review, *Renew. Sustain. Energy Rev.*



- 38 (2014) 172–179. <https://doi.org/10.1016/j.rser.2014.05.043>.
- [47] J.H. Duarte, L.S. Fanka, J.A.V. Costa, Utilization of simulated flue gas containing CO<sub>2</sub>, SO<sub>2</sub>, NO and ash for *Chlorella fusca* cultivation, *Bioresour. Technol.* 214 (2016) 159–165. <https://doi.org/10.1016/j.biortech.2016.04.078>.
- [48] F. Bauer, Narratives of biorefinery innovation for the bioeconomy: Conflict, consensus or confusion?, *Environ. Innov. Soc. Transitions.* 28 (2018) 96–107. <https://doi.org/10.1016/j.eist.2018.01.005>.