

## **Volatile organic compounds (VOCs) allow sensitive differentiation of biological soil quality**

Brown, Robert; Ian D. Bull; Toby Journeaux; Chadwick, Dave; Jones, Davey L.

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1 **Volatile organic compounds (VOCs) allow sensitive differentiation of biological soil quality**

2 Robert W. Brown<sup>a</sup>, Ian D. Bull<sup>b</sup>, Toby Journeaux<sup>b</sup>, David R. Chadwick<sup>a,c</sup>, Davey L. Jones<sup>a,d</sup>

3 <sup>a</sup>*School of Natural Sciences, Bangor University, Bangor, Gwynedd, LL57 2UW, UK*

4 <sup>b</sup>*School of Chemistry, University of Bristol, Cantock's Close, Bristol, BS8 1TS, UK*

5 <sup>c</sup>*Interdisciplinary Research Centre for Agriculture Green Development in Yangtze River Basin,*

6 *Southwest University, Chongqing, China*

7 <sup>d</sup>*School of Agriculture and Environment, The University of Western Australia, Perth, WA 6009,*

8 *Australia*

9

10 *Corresponding Author: Robert Brown*

11 *Corresponding Author Address: School of Natural Sciences,*

12 *Bangor University, Gwynedd, LL57 2UW*

13 *Corresponding Author Tel: +44 07399 564 591*

14 *Corresponding Author Email: rob.brown@bangor.ac.uk*

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## 23 **ABSTRACT**

24 Understanding the change in function of the biological community under different soil conditions is  
25 key to effective soil quality monitoring and mitigation of soil degradation. Current measures of  
26 biological soil quality suffer from drawbacks with most techniques having high expense, low  
27 throughput or a narrow focus on one component of the community. The aim of this study was to  
28 assess the use of volatilomics as a method to profile the soil microbial community and compare the  
29 technique to phospholipid fatty acid (PLFA) profiling as a measure of biological soil quality. An  
30 agricultural grassland soil (Eutric Cambisol) was subjected to a range of stresses in replicate  
31 laboratory mesocosms. Treatments included the imposition of hypoxia/anoxia by flooding with  
32 freshwater or saltwater in the presence or absence of plant residues. The volatile organic compound  
33 (VOC) and PLFA profile of each treatment was then compared to unamended mesocosms. We  
34 hypothesized that the VOC fingerprint of soil would be highly responsive to changes in microbial  
35 metabolic status/functioning and thus provide a complementary approach to PLFAs for evaluating soil  
36 biological health. We also hypothesized that the VOC profile would have greater discriminatory  
37 power than PLFAs for determining differences between soil treatments. A headspace solid phase  
38 microextraction (HSSPME) method coupled with gas chromatography quadrupole-time of flight mass  
39 spectrometry (GC/Q-TOFMS) was used to analyse the broad spectrum of VOCs produced by each  
40 soil. Across all soil treatments 514 unique VOC peaks were detected. Overall, VOCs showed greater  
41 sensitivity than the PLFA analysis in separating soil quality treatments. Eighteen individual VOCs  
42 were identified which were primarily responsible for this separation (e.g. indole,  $\alpha$ -ionone,  
43 isophorone, 3-octanone, *p*-cresol, 2-ethyl-phenol). Anaerobic soils amended with residues showed the  
44 greatest separation from other treatments, with most of this differentiation associated with ten  
45 individual VOCs. The anaerobic soils also showed a significant reduction in the number of VOCs  
46 emitted but an increase in total VOC emissions. In conclusion, our findings provide evidence that soil  
47 VOCs rapidly respond to changes in soil quality and therefore hold great potential as a novel  
48 functionally relevant diagnostic measure of biological soil quality.

49 *Keywords:* soil function, metabolomics, method, microbial communities, soil quality indicator

## 50 **1. Introduction**

51 Soils are key to providing a wide range of ecosystem services crucial for earth system function and  
52 stability (Adhikari and Hartemink, 2016). However, most of the ice-free soils on the planet have been  
53 exploited either directly or indirectly by humans. Anthropogenic activity has subsequently contributed  
54 to the global-scale degradation of around 6 million ha of agricultural land (UNFAO and ITPS, 2015).  
55 Projections estimate that 95% of the land area on Earth could become degraded by 2050  
56 (Montanarella et al., 2018). Therefore, it is essential that we continually assess the quality of our soils  
57 so that the provision of ecosystem services (e.g. nutrient cycling, water purification, food  
58 provisioning, climate regulation) can be sustained. Effective soil monitoring is fundamental to  
59 understanding the causes of degradation, which in turn could decrease the economic burden of soil  
60 degradation, which is estimated to be \$231 bn globally and \$2 bn in the UK (Graves et al., 2015;  
61 Nkonya et al., 2016).

62 Soil quality is often broadly defined as the capacity of a soil to function (Karlen et al., 1997).  
63 Although a range of soil quality indicators have been proposed, these mainly focus on the  
64 measurement of chemical attributes of the soil (e.g. pH, plant macronutrients, organic matter) and the  
65 physical characteristics of the soil (e.g. texture, bulk density, aggregate stability, hydrophobicity)  
66 (Bünemann et al., 2018; Schloter et al., 2018). However, soil fertility and productivity are not merely  
67 a function of soil physical and chemical characteristics. Soil biology is a crucial mediator in many  
68 processes linked to nutrient cycling, plant health and soil productivity, and is highly responsive to  
69 changes in management and environmental conditions, often being correlated to functional change  
70 (Lal, 2016; Lemanceau et al., 2015). Common soil biological indicators include measures of microbial  
71 activity (e.g. basal or substrate-induced respiration, enzyme activity) and the size and composition of  
72 the microbial community (e.g. metagenomics-metabarcoding, mesofaunal counts, CHCl<sub>3</sub>-fumigation-  
73 extraction) (Bending et al., 2004). However, these methods all suffer from major drawbacks,  
74 including: (i) problems defining critical thresholds of 'good' or 'bad' soil quality, (ii) low sample  
75 throughput, (iii) high labour or equipment costs, (iv) narrow focus on one component of the  
76 community, and (v) limited spatial resolution/integration.

77 Volatile organic compounds (VOCs) are relatively low-molecular weight organic compounds  
78 (typically <250 amu) with high vapour pressures that give the soil its odour and can be produced via  
79 both biotic and abiotic processes (though biological production of soil VOCs far exceeds the  
80 production of VOCs by abiotic processes) (Insam and Seewald, 2010). Farmers and land managers  
81 have, for centuries, used soil odour to infer soil quality (Semple, 1928). Study of soil odour first began  
82 to develop with the documentation of geosmin, the odour of moist soil in 1891 (Berthelot and André,  
83 1891). Since then, studies have identified a large number of VOCs produced from soil but the full  
84 extent of the functional significance of these compounds still remains largely unknown (Peñuelas et  
85 al., 2014). Biological VOCs are secondary metabolites, therefore not directly involved in organismal  
86 growth, development or reproduction. However, soil VOC emissions are highly dynamic, responding  
87 rapidly to changes in soil conditions and thus giving the soil system a unique VOC emission profile  
88 depending on soil conditions, the taxa present in soil, and their metabolic activities (Insam and  
89 Seewald, 2010). The type and amount of VOC compounds emitted from soil are dependent on a range  
90 of factors including; nutrient availability (Wheatley et al., 1996), oxygen status (McNeal and Herbert,  
91 2009), moisture availability (Asensio et al., 2007), organic matter inputs (Seewald et al., 2010),  
92 temperature (Schade and Custer, 2004), pH (Insam and Seewald, 2010) and interactions (sorption)  
93 with the solid phase (Serrano and Gallego, 2006). This sensitivity of VOCs to soil conditions makes  
94 them a good candidate as an indicator of soil biological quality.

95 Phospholipid fatty acid (PLFA) analysis has become a standard method for profiling the soil  
96 microbial community; giving a quantitative description of the microbial community within a sample  
97 (Frostegård et al., 2011). PLFA analysis provides information of the size of the microbial biomass,  
98 biomarkers of bacterial and fungal community structure, and an insight into the functional  
99 composition of the community (Willers et al., 2015). Microbial community composition and structure  
100 is responsive to management or naturally induced changes driven by soil physico-chemistry (Chang et  
101 al., 2017; Cobb et al., 2017; Hardy et al., 2019). For example, soil pH and organic matter content can  
102 greatly affect the PLFA composition of the microbial community, particularly in regard to  
103 fungal:bacterial ratios (Rousk et al., 2009; Welc et al., 2012). Despite the development of alternative

104 methods in soil biological analysis, for example metabarcoding (Orwin et al., 2018) or community  
105 level physiological profiling (CLPP; Ramsey et al., 2006), PLFA analysis remains a rapid, sensitive  
106 and reproducible method of detecting differences in community composition between treatments  
107 (Frostegård et al., 2011).

108         This laboratory mesocosm study aims to critically test the relationship between microbial  
109 community structure (PLFAs) and VOC production under a range of soil stresses  
110 (anoxia/waterlogging, salinity) and organic matter regimes (e.g. urine, plant residue addition). It tests  
111 the hypothesis that there will be significant differences in both the PLFA profile of the microbial  
112 community and the number and amount of VOCs emitted between treatments. However, as VOC  
113 production largely reflects the metabolism of the active microbial community, we hypothesize that it  
114 will have greater discriminatory power to resolve differences between treatments in comparison to  
115 PLFAs which reflects both the active, inactive and dead microbial biomass pools.

## 116 **2. Materials and methods**

### 117 *2.1. Experimental setup*

118 A Eutric Cambisol ( $n = 5$ , depth = 0 - 10 cm, Ah horizon) was collected from a post-harvest maize  
119 field located at the Henfaes Agricultural Research Station, Abergwyngregyn, North Wales (53°14'N,  
120 4°01'W). The site is characterised by a temperate-oceanic climate regime with a mean annual  
121 temperature of 11 °C and annual rainfall of 960 mm. On collection, the soil was sieved to pass 2 mm  
122 to remove stones and plant material and to ensure sample homogeneity. Field-moist soil (200 g) was  
123 then placed in 300 cm<sup>3</sup> polypropylene sample containers and treatments applied. Treatment consisted  
124 of the following, i) aerobic - control, ii) aerobic – amended with lysed grass residue (5 % w/w), iii)  
125 aerobic – amended with sheep urine (equivalent of 5 L m<sup>-2</sup>), iv) anaerobic – with river water, v)  
126 anaerobic – with river water, amended with lysed grass residue (5 % w/w), vi) anaerobic – with sea  
127 water, and vii) anaerobic – with sea water, amended with lysed grass residue (5 % w/w) (Fig. 1).  
128 There were five independent replicates of each treatment. Grass residues were lysed by freezing for 1  
129 h at -80 °C before being incorporated. This ensured that the plant material was metabolically inactive

130 prior to addition (i.e. no *de novo* biotic plant emissions). Sheep urine was collected from Welsh  
131 Mountain ewes (*Ovis aries* L.) as described in Marsden et al. (2018) and was loaded onto soil at rates  
132 equivalent to a typical sheep urination event (Selbie et al., 2015). To induce anaerobiosis, treatments  
133 were saturated with either freshwater or saline water and the containers hermetically sealed. These  
134 conditions reflected recent coastal and river flooding events which have occurred close to the  
135 sampling site (Sánchez-Rodríguez et al., 2018). Aerobic treatments remained unsealed and kept at  
136 constant weight throughout the 10-day incubation through the daily addition of deionised water. All  
137 mesocosms were subsequently incubated at 24 °C to stimulate the soil microbial community. This  
138 reflected summer soil temperatures at the field site. Duplicate mesocosms of each treatment replicate  
139 were created to allow for half of the mesocosms to be destructively sampled for soil properties and  
140 PLFAs, and half to be analysed for VOC's.

## 141 2.2. Soil sampling and analysis

142 At the end of the 10-day incubation, excess water was drained from the anaerobic mesocosms and soil  
143 was homogenised thoroughly by hand with a spatula and analysis undertaken immediately. Soil  
144 moisture content was determined gravimetrically by oven drying (105 °C, 24 h) and soil organic  
145 matter was determined by loss-on-ignition (450 °C, 16 h) (Ball, 1964). Soil pH and electrical  
146 conductivity (EC) were measured using standard electrodes submerged in 1:5 (w/v) soil-to-deionised  
147 water suspensions. The oxidation–reduction potential (ORP) was measured directly in the soil using a  
148 SenTix® ORP-T 900 combination electrode (Xylem Analytics, Weilheim, Germany) connected to a  
149 mV reader. Total C and N was determined on oven-dried, ground soil using a TruSpec® Analyzer  
150 (Leco Corp., St. Joseph, MI).

151 At the end of the incubation, 1:5 (w/v) soil-to-0.5 M K<sub>2</sub>SO<sub>4</sub>, 1:5 (w/v) soil-to-0.5 M AcOH  
152 (acetic acid) and 1:5 (w/v) soil-to-deionised (DI) H<sub>2</sub>O extractions were performed. TOC (total organic  
153 carbon) and TN (total nitrogen) were determined on K<sub>2</sub>SO<sub>4</sub> extracts using a Multi N/C 2100S  
154 Analyzer (AnalytikJena, Jena, Germany). Nitrate (NO<sub>3</sub>-N) and ammonium (NH<sub>4</sub>-N) concentrations  
155 within the K<sub>2</sub>SO<sub>4</sub> extracts were determined by the colorimetric VCl<sub>3</sub> method of Miranda et al. (2001)  
156 and the salicylic acid method of Mulvaney (1996), respectively. Available P was measured on the DI

157 H<sub>2</sub>O extracts using the molybdate blue colorimetric method of Murphy and Riley (1962). Cations (Na,  
158 K and Ca) were determined in the AcOH extracts using a Sherwood Model 410 Flame Photometer  
159 (Sherwood Scientific Ltd, Cambridge, UK). Soil characteristics are summarised in Table 1.

160 Characterisation of the soil microbial community was performed by PLFA analysis.  
161 Homogenised soil was stored at -20 °C prior to, and post lyophilisation using a Modulyo Freeze Dryer  
162 with RV pump (Edwards, Crawley, UK). 15 g samples were shipped, on dry ice, to Microbial ID Inc.  
163 (Newark, DE, USA), extracted, fractionated, and transesterified using the high throughput method of  
164 Buyer and Sasser (2012). Subsequently, samples were analysed using an Agilent (Agilent  
165 Technologies, Wilmington, DE, USA) 6890 gas chromatograph (GC) equipped with autosampler,  
166 split–splitless inlet, and flame ionization detector. The system was controlled by MIS Sherlock®  
167 (MIDI, Inc., Newark, DE, USA) and Agilent ChemStation software. GC-FID specification, analysis  
168 parameters and standards can be found in Buyer and Sasser (2012).

### 169 *2.3. VOC extraction, collection, analysis and data processing*

170 Duplicate soil samples, as aforementioned in section 2.1, were stored at 4 °C before analysis. VOCs  
171 were collected using headspace solid-phase microextraction (HSSPME) due to the method's  
172 suitability for evaluating complex sample matrices. A multi-component solid-phase microextraction  
173 (SPME) fibre (50/30 µm DVB/CAR/PDMS; Supelco, Bellefonte, USA), was selected due to both its  
174 sensitivity to a wide range of non-polar compounds and the thickness of the fibre (Cserháti 2010),  
175 which allowed transfer of the fibre from the extraction vial into the GC without the likelihood of  
176 significant loss of adsorbed volatile compounds. This was attached to a manual SPME holder  
177 (Supelco, Bellefonte, USA) for sampling.

178 Briefly, the SPME fibre was conditioned in the GC injection port at 270 °C prior to each soil  
179 being sampled, until no interfering peaks were obtained in blank injections. Fresh soil (3.5 g) was  
180 deposited into a glass vial (10 mL) (Supelco, Bellefonte, USA) alongside a magnetic stirrer bar for  
181 agitation, and subsequently capped with a SPME compatible PTFE/silicone septum (Supelco,  
182 Bellefonte, USA). To equilibrate the samples, they were agitated in a thermostat bath (80 °C) for 60



183 min, to decrease the partition coefficient between the sample and the headspace, and increase the  
184 vapour pressure (Zhang and Pawliszyn, 1993). A preconditioned SPME fibre was then manually  
185 inserted through the septum into the vial and exposed to the headspace for 20 min. The SPME fibre  
186 was then withdrawn into the holding sheath, removed from the vial and directly (< 30 s) desorbed into  
187 a GC-MS injection port at 250 °C in split mode (1:10 split ratio) for 2 min. A gas chromatograph  
188 (7890B; Agilent Technologies, Palo Alto, USA) interfaced to a quadrupole time-of-flight mass  
189 spectrometer (7200B; Agilent Technologies, Palo Alto, USA) (GC/Q-TOFMS) was used for  
190 compound identification. Chromatographic separation was obtained on a non-polar column (HP-1 50  
191 m length x 0.32 mm id x 0.17 µm D<sub>r</sub>; Agilent Technologies, Palo Alto, USA) using a He carrier gas  
192 with a flow of 2 mL min<sup>-1</sup>. The oven temperature was programmed from 60 °C to 250 °C at 4 °C min<sup>-1</sup>.  
193 The transfer line and ion source temperatures were both set to 300 °C. Using the electron ionisation  
194 (EI) mode at 70 eV, mass spectra were acquired across an *m/z* range of 25-400. Compound  
195 identification was attained using the NIST Mass Spectrometry library (Stein et al., 2014).

#### 196 2.4. Treatment analysis

197 The lysed grass treatments' moisture content was determined gravimetrically by oven drying (80 °C,  
198 24 h). Subsequently, grass C:N ratios were determined on the oven-dried and ground samples using a  
199 TruSpec<sup>®</sup> Analyzer. The sheep urine treatment was analysed for dissolved organic C (DOC) and total  
200 dissolved N (TDN) using a Multi N/C 2100S Analyzer. River and sea water samples were analysed  
201 for pH and EC using standard electrodes in addition to colorimetric determination of NO<sub>3</sub>-N, NH<sub>4</sub>-N,  
202 and P as per the methods described above, TOC and TN were also measured using the Multi N/C  
203 2100S Analyzer. Treatment characteristics are summarised in Table S1.

#### 204 2.5. Data and statistical analysis

205 The raw VOC data files were processed using MassHunter Workstation Profinder version B.08.00  
206 (Agilent Technologies, Palo Alto, CA, USA). Feature extraction was achieved using the Batch  
207 Feature Extraction algorithm with the subsequent processing parameters: peak filters = 500, ion count  
208 threshold = 5, retention time tolerance = 0.3 min, absolute height = 10000 counts, *m/z* range = 25-250,

209 retention time range = 5-35 min. This processing step produced a data output for each independent  
210 entity in the form of [intensity × retention time × mass]. These data were then input into MassHunter  
211 Mass Profiler Professional version B.14.5 (Agilent Technologies, Palo Alto, CA, USA) and log<sub>10</sub>  
212 transformed and normalised using an external scalar. To moderate the number of features subjected to  
213 statistical analysis, data filtering was executed. Briefly, a frequency filter (entities present in >90% of  
214 samples in at least one sample group), fold change filter (14.0) and sample variability filter  
215 (coefficient variable < 25%) were applied. This approach identified 18 discriminatory compounds,  
216 summarised in Figure 2. Total relative VOC production was also quantified by summing the areas of  
217 all peaks considered from each sample's chromatogram, examples of which are shown in Figure S2.

218 All of the following statistical analysis was performed in R v3.5.3 (R Core Team, 2019). In  
219 the analysis of discriminatory VOCs, relative peak heights were used to effectively compare samples  
220 based on the chemical diversity of compounds. A non-metric dimensional scaling (NMDS) approach  
221 was used to condense the multivariate VOC data in a comprehensible number of dimensions and  
222 visualize the relative degree of similarity among samples using the whole VOC dataset, which was  
223 performed using the 'vegan' package. NMDS was also used to analyse PLFA data. All PLFAs  
224 detected were used in the analysis, to represent the whole microbial community. Heatmap analysis  
225 and an ANOVA with Tukey post-hoc test was performed on log<sub>10</sub> transformed and pareto-scaled  
226 discriminatory VOC compound data in 'metaboanalyst 4.0' (Chong et al., 2018). This package was  
227 also used to perform hierarchical cluster analysis on log<sub>10</sub> transformed and pareto-scaled PLFA and  
228 VOC data. An ANOVA was also used to test log<sub>10</sub> transformed total relative VOC production as well  
229 as total number of peaks identified between treatments. Significant differences were further explored  
230 using a Tukey HSD post-hoc test. The relationships between total VOCs and number of VOCs to key  
231 soil parameters were tested using Spearman correlation analysis, using the 'corrplot' R package (Wei  
232 and Simko, 2017). For all analyses the significance threshold was set at  $p \leq 0.05$ .

### 233 **3. Results**

#### 234 *3.1. Treatment driven changes in VOCs and PLFAs*

235 NMDS analysis was used to show the clustering for all VOC and PLFA compounds, respectively,  
236 produced under the soil treatments. Both, VOC (Fig. 3A) and PLFA (Fig. 3B) analysis show  
237 separation between anaerobic + residues treatments compared to other treatments, implying a different  
238 microbial community and production of VOCs under these conditions. Using VOC analysis,  
239 anaerobic treatments without residues also grouped closely. Hierarchical clustering analysis (Fig. 4A  
240 and 4B) further illustrated the relationships between treatment levels, with results largely supporting  
241 NMDS findings. Overall, VOC analysis was more able to differentiate between anaerobic + residues  
242 treatments and aerobic treatments, with anaerobic treatments without residues and urine treatments  
243 being more closely related than they appeared using NMDS analysis. PLFA analysis was less able to  
244 separate treatments from one another.

### 245 *3.2. Relative VOC production between treatments*

246 Total relative VOC production was significantly higher from anaerobic + residues treatments ( $p <$   
247  $0.001$ ) compared to other treatments. Aerobic residue treatments also produced significantly more  
248 VOCs than anaerobic without residues and urine treatments ( $p < 0.001$ ). Across all soil treatments 514  
249 unique VOC peaks were detected. The total number of VOC compounds detected was significantly  
250 lower for both anaerobic + residues treatments ( $p < 0.05$ ). Control and urine treatments had a  
251 significantly higher number of VOC compounds detected ( $p < 0.05$ ).

### 252 *3.3. Soil factors affecting VOC production*

253 Several key soil properties were found to be significant predictors of both, the relative total  
254 concentration of VOCs and the number of VOC compounds emitted from soil samples. Particularly,  
255 organic matter content, dissolved organic carbon and PLFA microbial biomass were significantly  
256 correlated with total VOC concentration and number of VOCs emitted (summarised in Table 2). ORP  
257 and soil moisture were also significantly correlated to the number of VOCs emitted from soil.

### 258 *3.4. Identities of discriminatory VOCs produced*

259 Across the 7 treatments, software was able to identify 18 VOC compounds that were able to  
260 differentiate between the control and treated samples; these compounds are subsequently referred to

261 as discriminatory compounds (Fig. 2). Of the 18 compounds, 16 were found to have significant  
262 differences between treatments when tested statistically (*p*-cresol and hexanoic acid were not  
263 significantly different between treatments). Additionally, 10 compounds were found in significantly  
264 higher concentrations in anaerobic + residues treatments compared to the other treatments ( $p < 0.001$ ),  
265 as highlighted in Figure 2.

## 266 **4. Discussion**

### 267 *4.1. Identification of VOC profile trends*

268 NMDS and hierarchical clustering of VOC data (Fig. 3A and 4A) illustrate the clustering of anaerobic  
269 + residues treatments, suggesting that nutrient-rich anaerobic conditions had the greatest impact upon  
270 the VOC profile of the soils relative to the control soils. Anaerobic + residues treatments also had the  
271 highest levels of total VOCs, but the lowest number of individual compounds detected. This suggests  
272 that addition of plant residues under anaerobic conditions caused a large increase in the concentration,  
273 but not number, of VOCs detected. It is generally understood that VOCs are found in greater  
274 concentrations under anaerobic conditions, due to the production of metabolic end-products by  
275 anaerobic fermentation and extracellular degradation of complex organic metabolites (Insam and  
276 Seewald, 2010; Seewald et al., 2010; Stotzky et al., 1976). However, the results of this study show  
277 this to be conditional on the amount of available nutrients and C under anaerobic conditions, as soils  
278 under anaerobic conditions but with no plant residues produced significantly less total VOCs than  
279 anaerobic soils with plant residues. This is likely due to a lack of easily assimilable C in non-residue  
280 amended anaerobic samples, as well as lower levels of microbial activity and thus VOC production.  
281 However, although anaerobic + residues treatments resulted in a greater total VOC concentration  
282 compared to other treatments, the number of compounds contributing to the signal produced was  
283 significantly less. This suggests a limited range of metabolic reactions are responsible for producing  
284 the majority of the soil VOC profile, as microbial metabolism is the dominant source of VOC  
285 emissions from soil (Leff and Fierer, 2008).

286 Control and urine amended soils produced a significantly greater number of VOC compounds  
287 compared to other treatments. Under the control treatment, it is likely that the soil microbial  
288 community will be highly adapted to the prevailing conditions (i.e. low stress) and thus more likely to  
289 have a streamlined metabolism and high C use efficiency. Most of the C used for energy production  
290 will produce only CO<sub>2</sub> as an end-product rather than respiratory-derived VOCs (e.g. ethanol, volatile  
291 organic acids) or those used in secondary metabolism for stress alleviation (Insam and Seewald,  
292 2010). Under these low stress conditions, the large number of different VOC compounds are likely to  
293 be characteristic of a diverse soil microbial community (McNeal and Herbert, 2009). Conversely,  
294 under the urine treatment, the input of nutrients into the system, particularly K and soluble N (e.g.  
295 urea NH<sub>4</sub><sup>+</sup>), is likely to induce a rapid change in soil microbial metabolism (Waldrop and Firestone,  
296 2004; Williams et al., 2000). However, it is also likely that the urine itself will contain some unique  
297 VOCs (Deev et al., 2020; Mozdiak et al., 2019). The experimental set up used here, however, was  
298 unable to distinguish between the direct and indirect effects of urine on the soil VOC profile.

#### 299 *4.2. Sources of discriminatory VOC compounds*

300 Following the positive identification of 18 discriminatory VOCs, the abundance of which could be  
301 used to determine several soil characteristics key in determining a soils quality (e.g. SOM, levels of  
302 oxygen content), prospective sources and functions of the compounds were examined using the  
303 existing literature. Notably, of these compounds, 10 were found in significantly higher relative  
304 concentrations in the anaerobic + residues treatment, several of which can be identified as breakdown  
305 products of the residues.

306 For example, the biodegradation of carotenoids is the most likely source of β-ionone and α-  
307 ionone. Carotenoids form part of a group of terpenoid organic pigments that play a major role in  
308 photosynthesis in addition to the photoprotection of photosystems in plants and photosynthetic  
309 bacteria (Lobo et al., 2012). These compounds were only produced under anaerobic + residues  
310 treatments suggesting that either under aerobic conditions these molecules were further catabolised or  
311 were only produced under nutrient-rich anaerobic conditions.

312 Benzeneacetic acid, an auxin, was solely identified in anaerobic soils with added grass residues. This  
313 suggests that under aerobic conditions all the benzeneacetic acid was mineralised. Alternatively,  
314 benzeneacetic acid may only be produced by anaerobic bacteria. For example, it has been reported  
315 that *Azoarcus evansii*, an endophytic facultative anaerobic denitrifying bacterium found in several  
316 grass species, is a significant producer of benzeneacetic acid (Schulz and Dickschat, 2007; Sun et al.,  
317 2019).

318         Despite many possible sources of the discriminatory compounds, linking a VOC to specific  
319 processes, functions or microorganisms is challenging due to the variety of degradation pathways  
320 which a VOC can take within the soil, dependant on environmental conditions (i.e. oxygen and  
321 nutrient status). Furthermore, the necessity to increase the temperature of the sample during the  
322 extraction procedure in order to increase the partition coefficient, generates a degree of ambiguity as  
323 to the true source of the compounds in the headspace, as under such conditions there is a possibility of  
324 the breakdown of thermolabile compounds within the sample (Kaspar et al., 2008). Accordingly, it is  
325 uncertain whether the compounds extracted are a samples' intrinsic VOCs; released by the  
326 microorganisms within the soil or as a direct result of an amendment (e.g. urine), or VOCs produced  
327 during the extraction procedure due to thermal degradation of the sample. For example, lignin is  
328 widely described as stable below temperatures of 100 °C, however, the thermal stability of many soil  
329 components is poorly reported within the literature (Brebu and Vasile, 2010).

330         Additionally, there is potential discrepancy between the production of a volatile compound  
331 and its emission from matrix. The emission of VOCs from the soil matrix, is highly dependent on the  
332 soil's structure and moisture as well as the rate at which the VOCs are being emitted. These factors  
333 ultimately control the rate of VOC production, dispersal and consumption (Aochi and Farmer, 2005).  
334 Similarly to greenhouse gases, it is likely that soils act as sources or sinks for VOCs depending on  
335 environmental conditions (Insam and Seewald, 2010; Oertel et al., 2016). For example, VOCs  
336 produced further down the soil profile could be consumed or degraded before reaching the soil  
337 surface. Particularly in aerobic systems, when in a steady state equilibrium these production-  
338 consumption systems may cause emission of very few VOCs, however, on disturbance, emissions

339 may increase. Heating during analysis, to improve the partition co-efficient (the ratio of analyte in  
340 matrix-to-analyte in headspace) and drive VOCs into the headspace, is likely to increase the emissions  
341 from the soil sample (Turner et al., 2019).

#### 342 *4.3. Relationship between soil properties and VOC emission*

343 Overall, levels of soil moisture seemed to be the best predictor of the number of VOCs produced from  
344 samples, with higher levels of soil moisture reducing the number of VOCs emitted. Similarly, ORP  
345 was positively correlated to the number of VOCs produced. Under anaerobic, reducing conditions it is  
346 likely that alcoholic sugar fermentation predominates (Pezeshki and DeLaune, 2012). Previous  
347 studies, e.g. Stotzky et al. (1976) and Seewald et al. (2010), have shown that anaerobic conditions  
348 increase the diversity and amount of VOCs emitted. However, this study showed that this was only  
349 the case if conditions were not nutrient limited. Both anaerobic treatments + residues additions  
350 resulted in an increase in total VOCs emitted but in a reduction in the number of VOC compounds  
351 emitted. As shown previously, substrate availability and quality are key in the production and  
352 emission of VOCs (Wheatley et al., 1996). However, there was no correlation between total dissolved  
353 nitrogen and total VOCs or number of VOCs produced, and only a weak correlation between  
354 available inorganic N ( $\text{NO}_3$  and  $\text{NH}_4$ ) and total VOCs emitted.

#### 355 *4.4. Critical analysis of VOCs and PLFAs as soil quality indicators*

356 Whilst both VOCs and PLFAs can be related directly to the soil biological community, each offer  
357 very different types of insight. PLFAs focus on the membrane lipid composition, from which links  
358 can be made to microbial community composition (Mann et al., 2019). However, use of isotopic  
359 substrates can enable PLFA to be used as indicators of soil function, i.e. Bull et al. (2000). VOCs  
360 focus on secondary metabolism of the soil biological community, which relates to community  
361 function. Advantages and disadvantages of each method are summarised in Table 3.

362           Sensitivity is a key attribute in biological quality monitoring, as methods must be able to  
363 detect subtle changes in soil biochemistry to give an accurate representation of soil quality. Both  
364 PLFA and VOC analysis are, in theory, sensitive, as both sets of compounds degrade rapidly under

365 environmental conditions (Li et al., 2019; Zhang et al., 2019). However, this experiment has  
366 demonstrated that, in terms of discriminatory power, VOCs can more robustly separate the impact of  
367 short-term soil treatments, which correspond to different soil qualities (Fig. 3 and 4).

368 Both PLFAs and VOCs relate to the biological function of soil. However, the turnover of  
369 VOCs is more rapid and more functionally relevant than PLFAs. The impact of many environmental  
370 factors e.g. pH, heat and moisture content are similar between both types of analysis. One advantage  
371 of VOC analysis over PLFA analysis is the lack of pre-treatment and multi-stage chemical extraction  
372 required, reducing the amount of inherent bias within the method. However, identification of  
373 individual VOCs is difficult; while databases exist, they are by no means extensive and identification  
374 may not be absolute without the use of confirmatory standards.

#### 375 4.5. Future research direction

376 This study highlights several potential future research areas within the soil and environmental  
377 sciences. Specifically: Is it possible to increase the rapidity of VOC analysis, by reducing extraction  
378 and/ or analysis time, while retaining the resolution and sensitivity to enable compound identification?  
379 How do *ex-situ* sampling techniques demonstrated in this study compare to *in-situ*, non-destructive,  
380 methods, for example, sorbent tube sampling? To what extent are abiotic VOCs contributing to the  
381 overall soil VOC profile compared to biotic VOCs? Under different soil types and cropping regimes,  
382 does the core microbiome have similar metabolism and therefore produce similar VOCs, i.e. is  
383 analysis of VOCs applicable over large spatial scales? And what is the temporal frequency of analysis  
384 required to accurately assess a soils quality? Addressing these questions will further advance sampling  
385 and analysis of VOCs in soil and elucidate the role of VOCs as indicators of changes in soil quality.  
386 Another key aspect is defining the boundary values for VOC concentrations and profiles for 'healthy'  
387 soils. Arguably, the control and aerobic residues treatment in this study have started to provide some  
388 data to represent healthy soils.

## 389 5. Conclusions



390 This study applied a HSSPME extraction and analysis method for the determination of VOCs in soils.  
391 The method was applied to seven soil treatments representing a range of soil qualities and a number of  
392 discriminatory compounds were identified. When compared to PLFA analysis, VOC analysis was  
393 better able to differentiate between soil treatments. 18 discriminatory compounds were identified, 10  
394 of which were associated with anaerobic treatments with residues. Contrary to previous findings,  
395 VOCs under anaerobic treatments with residue additions exhibited a relative increase in concentration  
396 of total VOCs emitted but a reduction in the abundance of specific compounds emitted. These results  
397 demonstrate the potential of secondary metabolites as an indicator of soil quality and highlights the  
398 need for further research into soil VOC analysis to understand nutrient cycling and metabolism as  
399 well as the effect of *in-situ* vs. *ex-situ* sampling.

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406

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584

585 **Table and figure captions**

586 **Fig. 1.** Flow diagram illustrating the seven treatments applied in the study.

587 **Fig. 2.** Heatmap of eighteen VOC compounds identified as driving changes between soil treatments.  
588 Compounds highlighted in purple are found in significantly higher concentrations in anaerobic +  
589 residues treatments. Compounds highlighted with † were not significantly different between  
590 treatments ( $p > 0.05$ ). Darker red = large positive relative difference between treatments, darker blue  
591 = large negative relative difference between treatments. Samples are ordered using group averages  
592 and clustered using a Ward algorithm. Compound structures are shown in Fig. S1.

593 **Fig. 3.** A) NMDS plot of VOC profiles of each soil treatment. B) NMDS plot of PLFA profile for  
594 each soil treatment. Ellipses represent 95% confidence intervals for each treatment. The legend is the  
595 same for both panels.

596 **Fig. 4.** A) Dendrogram, using Euclidean distance measure and a ward clustering algorithm, of VOC  
597 profiles of each soil treatment ( $n = 5$ ). B) Dendrogram, using Euclidean distance measure and a ward  
598 clustering algorithm, of PLFA profiles of each soil treatment replicate ( $n = 5$ ).

599 **Table 1.** Characteristics of treated soils used in this study. Values are expressed on a mean dry soil  
600 weight basis  $\pm$  SEM ( $n = 5$ ). Letters denote significant differences between treatments using a  
601 Kruskal-Wallis test with Dunn Post-hoc test and Bonferroni correction ( $p < 0.05$ ).

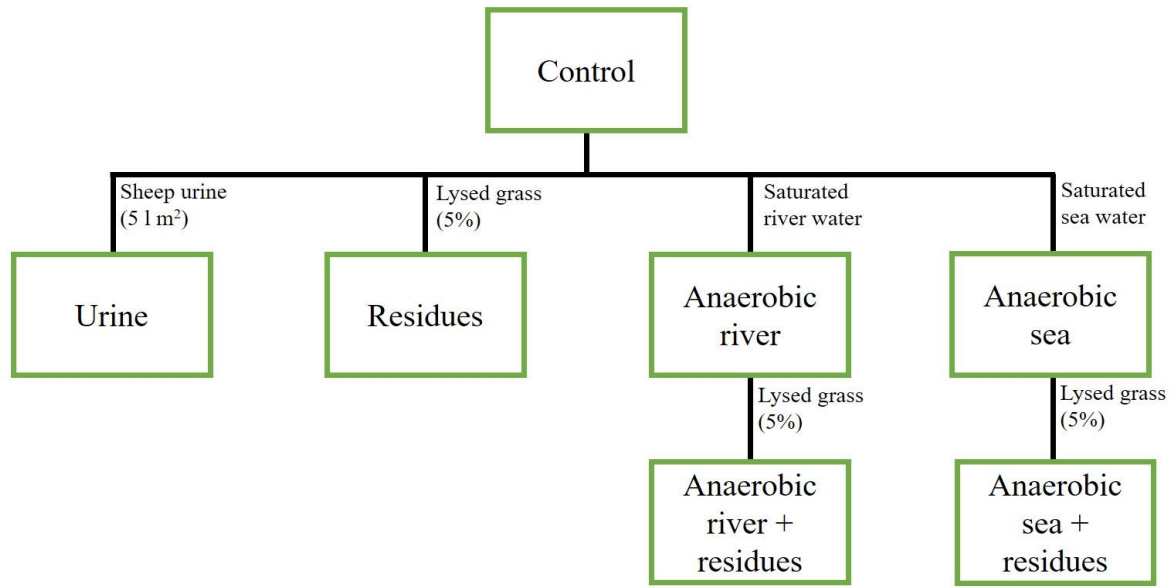
602 **Table 2.** Pearson's correlation coefficients for key soil properties in relation to both total amount of  
603 VOCs emitted and the number of VOCs emitted from samples. *NS* signifies not significant (i.e.  $p >$   
604  $0.05$ ).

605 **Table 3.** Summary of the advantages and disadvantages of VOC and PLFA analysis in the analysis of  
606 soil quality.

607

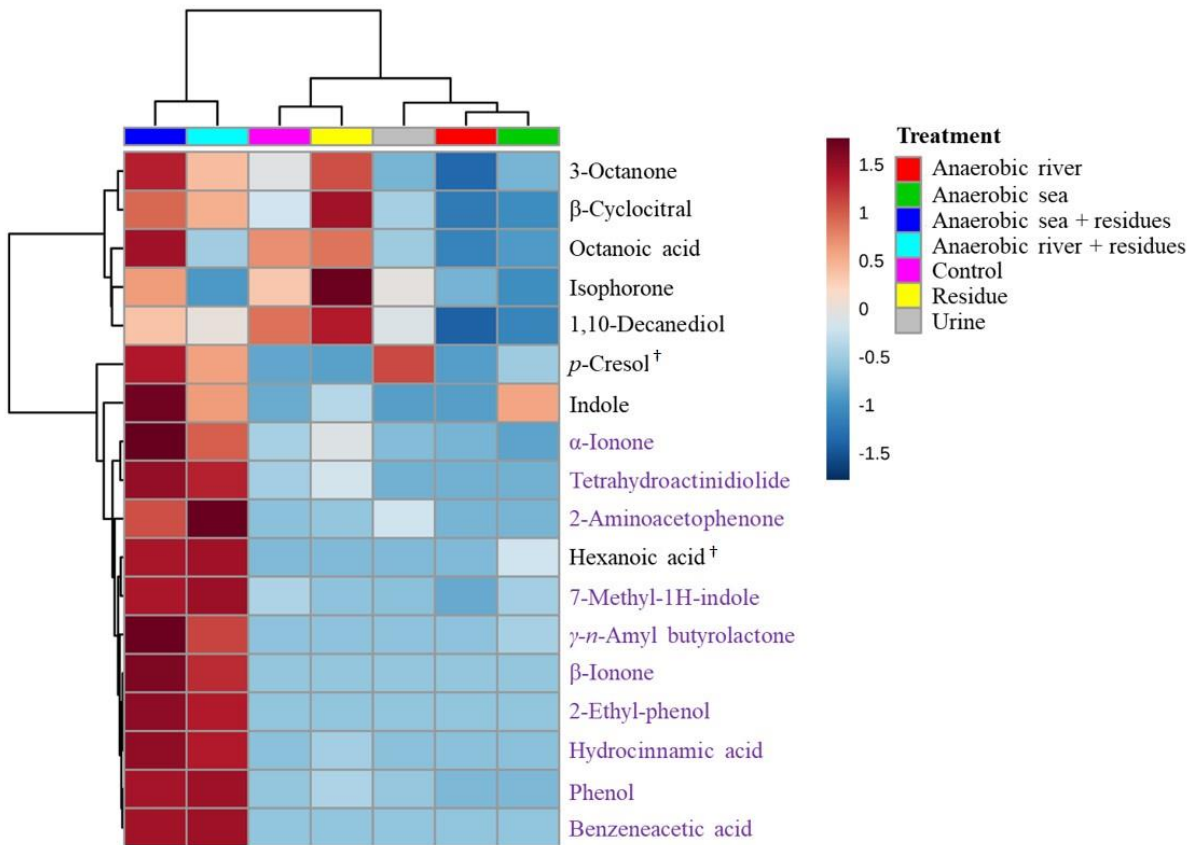
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609 **Figure 1**



610

611 **Figure 2**



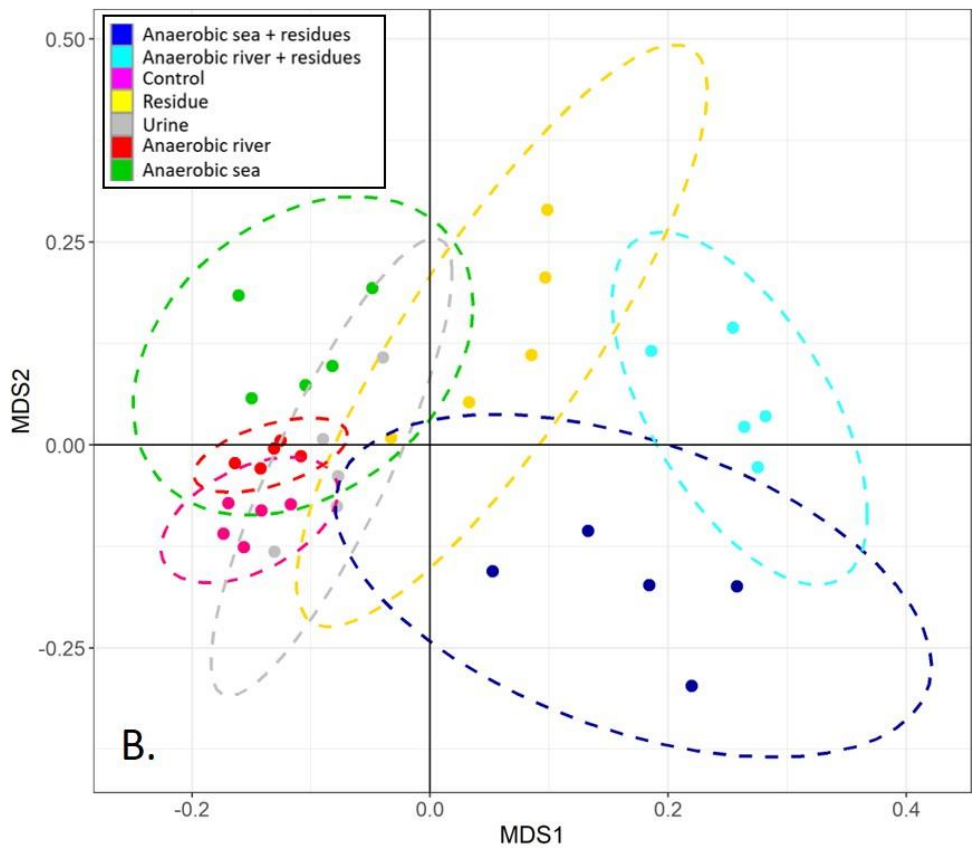
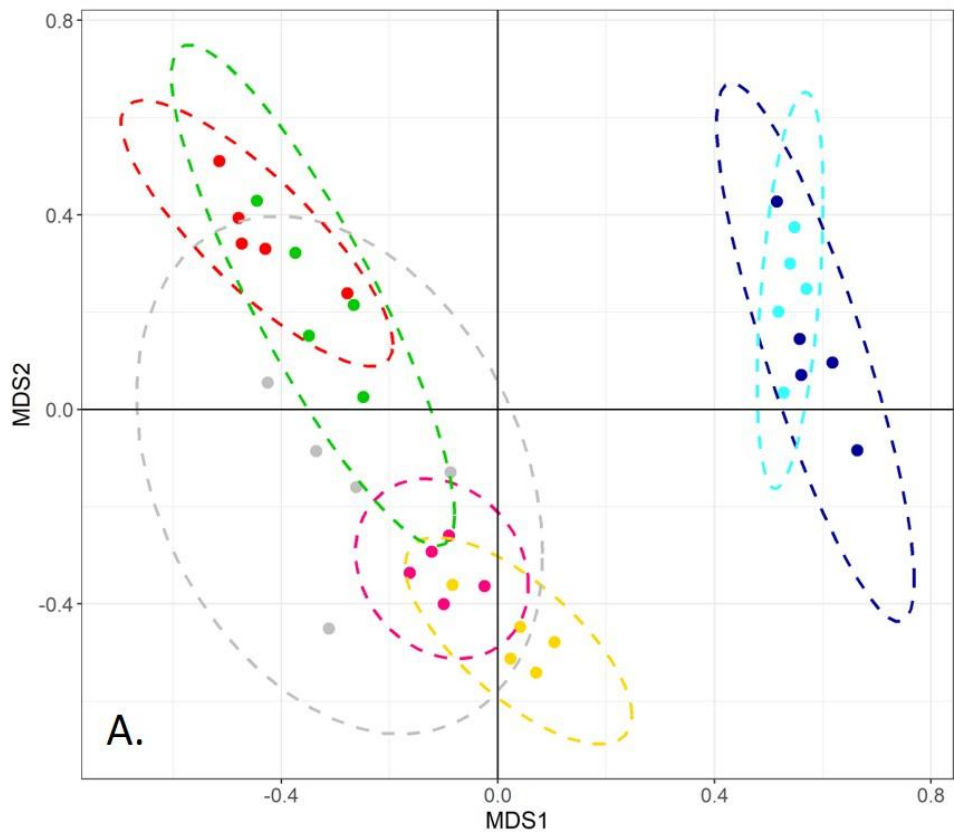
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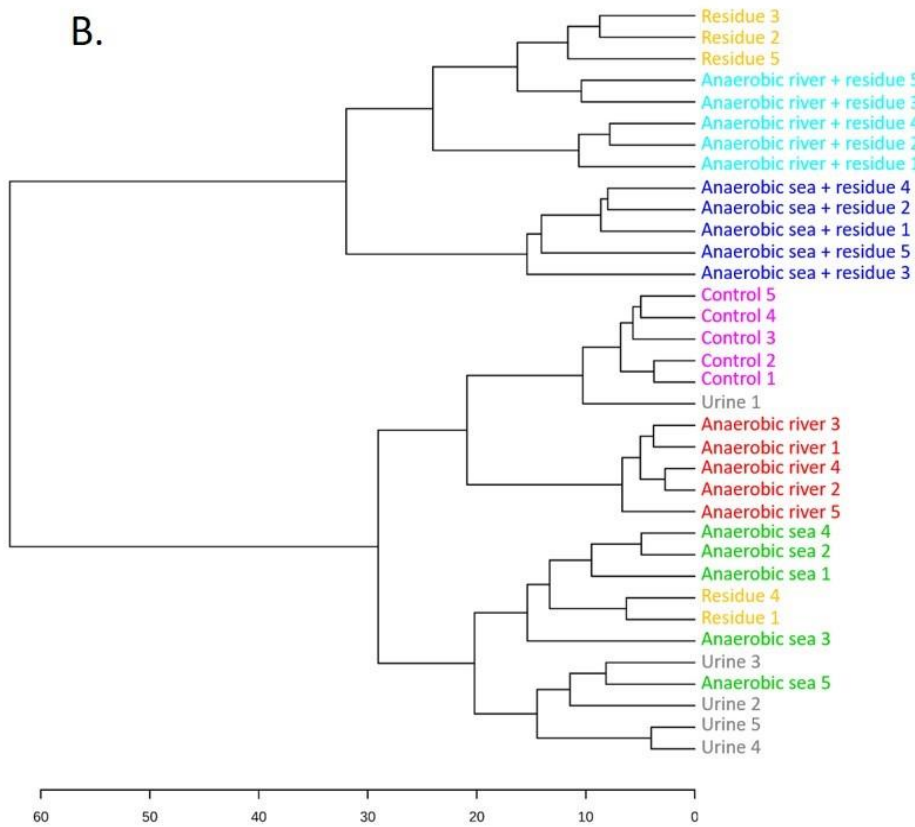
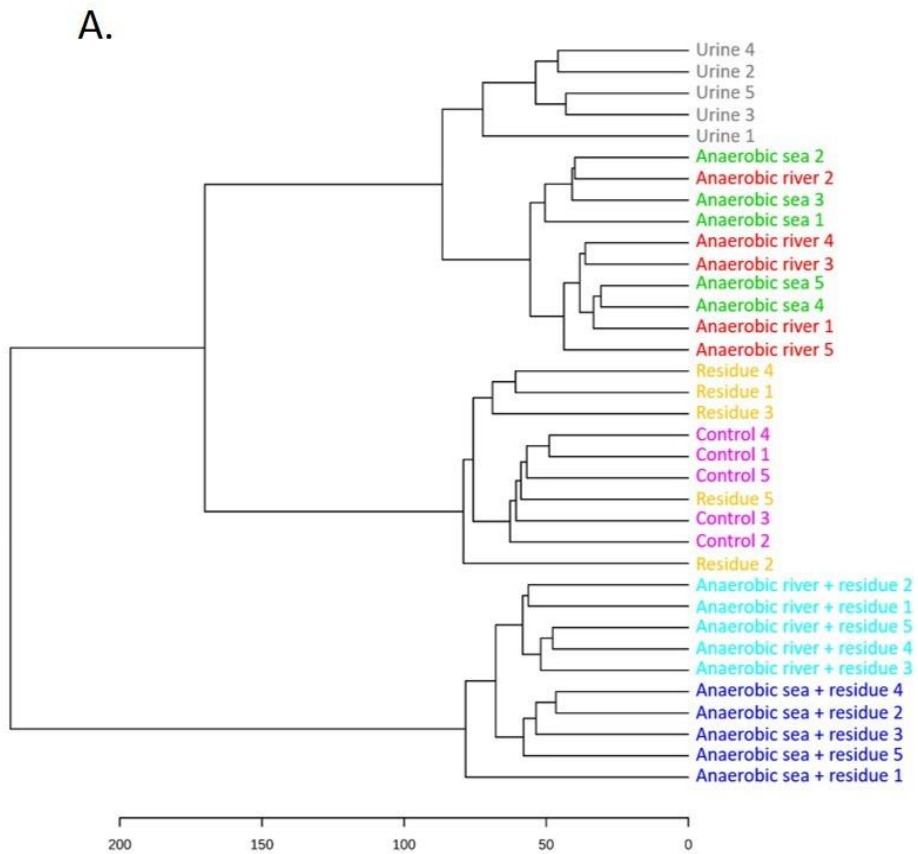
615

616 **Figure 3**



617

618



621 **Table 1.** Characteristics of treated soils used in this study. Values are expressed on a mean dry soil weight basis  $\pm$  SEM ( $n = 5$ ). Letters denote significant differences  
 622 between treatments using a Kruskal-Wallis with Dunn Post-hoc test and Bonferroni correction ( $p < 0.05$ ).

	Aerobic			Anaerobic			
	Control	Residues	Urine	River water	River water + residues	Sea water	Sea water + residues
Texture	Sandy clay loam						
Soil moisture (%)	29.3 <sup>A</sup> $\pm$ 1.2	36.2 $\pm$ 1.3	29.8 <sup>B,C</sup> $\pm$ 0.5	74.1 $\pm$ 1.8	83.1 <sup>A,B</sup> $\pm$ 1.5	68.2 $\pm$ 1.1	86.5 <sup>C</sup> $\pm$ 1.3
pH	6.14 <sup>A</sup> $\pm$ 0.15	5.71 <sup>B,C</sup> $\pm$ 0.06	7.53 <sup>A,B,D</sup> $\pm$ 0.11	6.51 $\pm$ 0.08	6.90 <sup>C</sup> $\pm$ 0.05	6.23 <sup>D</sup> $\pm$ 0.09	6.88 $\pm$ 0.06
EC ( $\mu$ S cm <sup>-1</sup> )	27.1 $\pm$ 11.7	54.0 $\pm$ 6.3	72.3 $\pm$ 2.0	3.3 <sup>A,B</sup> $\pm$ 0.1	21.0 $\pm$ 1.8	563.6 <sup>A</sup> $\pm$ 7.1	661.6 <sup>B</sup> $\pm$ 18.9
ORP (mV)	413.5 <sup>A,C</sup> $\pm$ 4.9	325.6 $\pm$ 9.8	216.5 $\pm$ 2.8	24.5 <sup>B,D</sup> $\pm$ 4.7	-73.7 <sup>A,B</sup> $\pm$ 3.6	141.5 $\pm$ 11.3	-144.9 <sup>C,D</sup> $\pm$ 16.2
Total C (%)	2.63 $\pm$ 0.05	2.97 $\pm$ 0.18	2.44 $\pm$ 0.05	2.44 $\pm$ 0.11	2.90 $\pm$ 0.12	2.59 $\pm$ 0.08	2.6 $\pm$ 0.10
Total N (%)	0.28 $\pm$ 0.01	0.36 <sup>A,B</sup> $\pm$ 0.02	0.32 <sup>C</sup> $\pm$ 0.01	0.25 <sup>A</sup> $\pm$ 0.01	0.3 $\pm$ 0.01	0.25 <sup>B,C</sup> $\pm$ 0.01	0.3 $\pm$ 0.01
C:N ratio	9.40 $\pm$ 0.31	8.22 <sup>C</sup> $\pm$ 0.15	7.74 <sup>A,B,D</sup> $\pm$ 0.17	9.21 <sup>A</sup> $\pm$ 0.57	9.53 <sup>B</sup> $\pm$ 0.16	9.97 <sup>C,D,E</sup> $\pm$ 0.22	9.42 <sup>E</sup> $\pm$ 0.44
Dissolved organic C (mg C kg <sup>-1</sup> )	51.0 <sup>A,C</sup> $\pm$ 2.1	87.3 $\pm$ 14.3	171.6 $\pm$ 10.5	67.1 $\pm$ 4.8	240.3 <sup>A,B</sup> $\pm$ 8.6	58.4 <sup>B,D</sup> $\pm$ 4.7	303.3 <sup>C,D</sup> $\pm$ 45.1
Extractable NO <sub>3</sub> <sup>-</sup> (mg N kg <sup>-1</sup> )	36.3 $\pm$ 1.7	162.8 <sup>A,B,D</sup> $\pm$ 17.0	70.8 <sup>C</sup> $\pm$ 6.4	0.65 <sup>A</sup> $\pm$ 0.16	1.02 $\pm$ 0.23	0.37 <sup>B,C</sup> $\pm$ 0.24	0.77 <sup>D</sup> $\pm$ 0.18
Extractable NH <sub>4</sub> <sup>+</sup> (mg N kg <sup>-1</sup> )	3.33 <sup>A,C</sup> $\pm$ 1.24	51.32 $\pm$ 6.57	487.04 <sup>A,B</sup> $\pm$ 31.59	2.70 <sup>B</sup> $\pm$ 0.50	81.97 $\pm$ 4.66	3.28 $\pm$ 0.56	98.72 <sup>C</sup> $\pm$ 5.78
Extractable P (mg P kg <sup>-1</sup> )	2.61 $\pm$ 0.33	2.47 $\pm$ 0.24	3.12 <sup>C</sup> $\pm$ 0.11	4.21 <sup>A,B,D</sup> $\pm$ 0.27	1.90 <sup>A</sup> $\pm$ 0.08	2.05 <sup>B</sup> $\pm$ 0.16	1.74 <sup>C,D</sup> $\pm$ 0.04
Exchangeable Na (mg Na kg <sup>-1</sup> )	13.8 <sup>A,C</sup> $\pm$ 1.5	28.3 $\pm$ 2.2	80.2 $\pm$ 5.9	17.9 <sup>B,D</sup> $\pm$ 1.1	22.3 $\pm$ 2.5	5477 <sup>A,B</sup> $\pm$ 344	6490 <sup>C,D</sup> $\pm$ 99
Exchangeable K (mg K kg <sup>-1</sup> )	89.5 <sup>A,C</sup> $\pm$ 11.1	406.8 $\pm$ 33.9	646.9 <sup>A,B</sup> $\pm$ 35.9	121.2 <sup>B,D</sup> $\pm$ 8.6	321.7 $\pm$ 74.3	376.2 $\pm$ 25.1	671.8 <sup>C,D</sup> $\pm$ 20.0
Exchangeable Ca (mg Ca kg <sup>-1</sup> )	1099 <sup>A</sup> $\pm$ 77	1083 $\pm$ 90	1169 <sup>B</sup> $\pm$ 133	1455 $\pm$ 117	1689 <sup>A,B</sup> $\pm$ 116	1262 $\pm$ 61	1587 $\pm$ 1028
Bacterial/Fungal PLFA ratio	0.06 $\pm$ 0.00	0.10 $\pm$ 0.01	0.07 <sup>A</sup> $\pm$ 0.01	0.07 $\pm$ 0.00	0.07 $\pm$ 0.01	0.13 <sup>B</sup> $\pm$ 0.06	0.04 <sup>A,B</sup> $\pm$ 0.01
Microbial biomass ( $\mu$ mol PLFA kg <sup>-1</sup> )	110.2 <sup>A,B,D</sup> $\pm$ 2.6	177.6 <sup>A</sup> $\pm$ 11.0	145.0 $\pm$ 6.1	128.7 <sup>C</sup> $\pm$ 3.6	215.6 <sup>B,C</sup> $\pm$ 14.1	158.7 $\pm$ 20.0	209.4 <sup>D</sup> $\pm$ 27.3

EC - electrical conductivity, ORP - redox potential.

623 **Table 2.** Pearson's correlation coefficients for key soil properties in relation to both total amount of  
624 VOCs emitted, and the number of VOCs emitted from samples. *NS* signifies not significant (i.e.  $p >$   
625 0.05).

	Total VOCs	No. VOCs
626 pH	<i>NS</i>	<i>NS</i>
627 ORP (mV)	<i>NS</i>	0.576
628 Soil moisture (%)	<i>NS</i>	-0.64
629 Organic matter (%)	0.361	-0.425
630 Total dissolved N (mg kg <sup>-1</sup> )	<i>NS</i>	<i>NS</i>
631 Total dissolved organic C (mg kg <sup>-1</sup> )	0.385	-0.391
632 Microbial biomass (μmol PLFA kg <sup>-1</sup> )	0.388	-0.489

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634 **Table 3.** Summary of the advantages and disadvantages of VOC and PLFA analysis in the analysis of  
 635 soil quality.

	<b>VOC analysis</b>	<b>PLFA analysis</b>
<b>Advantages</b>	<p>Functionally relevant</p> <p>Rapid extraction and analysis.</p> <p>Microbial culture not required.</p> <p>Rapid degradation - offers snapshot of current state of biological activity.</p> <p>Possibility of identifying characteristic biomarkers for specific conditions.</p> <p>Non-destructive.</p>	<p>Sensitive and reproducible.</p> <p>Measure of biomass and community structure.</p> <p>Microbial culture not required.</p> <p>Rapid degradation - offers snapshot of current state of biology.</p> <p>Lack of group or species-specific FA's</p>
<b>Disadvantages</b>	<p>Difficult to separate plant and microbial derived VOCs in soil.</p> <p>Lack of fundamental understanding of VOCs in soil.</p> <p>SPME requires destructive sampling.</p> <p>Most relationships have been investigated using laboratory/pure culture studies – may not reflect diverse soil community response.</p> <p>Medium sample throughput with possibility of automation.</p> <p>High analytical capital costs.</p>	<p>Time consuming and complex extraction required.</p> <p>Little functional relevance.</p> <p>Extraction of entire microbial community – not only the active fraction.</p> <p>Most relationships have been investigated using laboratory/pure culture studies – may not reflect diverse soil community response.</p> <p>Low sample throughput.</p> <p>High capital and labour costs.</p> <p>Destructive.</p>

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