



Year: 2021

Superconductivity and charge density wave formation in lithium-intercalated 2H-TaS₂

Liu, Huanlong ; Huangfu, Shangxiong ; Zhang, Xiaofu ; Lin, Hai ; Schilling, Andreas

Abstract: We systematically investigated the superconducting properties and the interplay between charge density waves (CDWs) and superconductivity (SC) in lithium-intercalated 2H-TaS₂. By gradually increasing the lithium content x , the CDW formation temperature is continuously suppressed, and the onset temperature of SC is increased with a maximum transition temperature $T_c=3.5\text{K}$ for $x=0.096$. The bulk nature of SC is confirmed by a superconducting shielding fraction of the order of unity for this composition. The electronic contribution to the specific heat and Hall resistivity data demonstrates that the CDW weakens with lithium intercalation, thereby indirectly increasing the carrier density and boosting SC. While the sign of the charge carriers in undoped 2H-TaS₂ changes from electronlike to hole type near the CDW formation temperature 75K , the lithium-intercalated Li_xTaS_2 shows predominantly hole-type carriers in the CDW phase even for very low lithium content.

DOI: <https://doi.org/10.1103/physrevb.104.064511>

Posted at the Zurich Open Repository and Archive, University of Zurich

ZORA URL: <https://doi.org/10.5167/uzh-210954>

Journal Article

Published Version



The following work is licensed under a Creative Commons: Attribution 4.0 International (CC BY 4.0) License.

Originally published at:

Liu, Huanlong; Huangfu, Shangxiong; Zhang, Xiaofu; Lin, Hai; Schilling, Andreas (2021). Superconductivity and charge density wave formation in lithium-intercalated 2H-TaS₂. *Physical review B*, 104(6):064511.

DOI: <https://doi.org/10.1103/physrevb.104.064511>

Superconductivity and charge density wave formation in lithium-intercalated 2H-TaS₂Huanlong Liu,¹ Shangxiong Huangfu,^{1,2} Xiaofu Zhang^{ⓧ,1,3,4,*} Hai Lin,¹ and Andreas Schilling^{ⓧ1}¹*Department of Physics, University of Zurich, Winterthurerstrasse 190, CH-8057 Zurich, Switzerland*²*Laboratory for High Performance Ceramics, Empa, Überlandstrasse 129, CH-8600 Dübendorf, Switzerland*³*State Key Laboratory of Functional Materials for Informatics, Shanghai Institute of Microsystem and Information Technology, Chinese Academy of Sciences (CAS), Shanghai 200050, China*⁴*CAS Center for Excellence in Superconducting Electronics, Shanghai 200050, China*

(Received 5 May 2021; revised 16 July 2021; accepted 10 August 2021; published 25 August 2021)

We systematically investigated the superconducting properties and the interplay between charge density waves (CDWs) and superconductivity (SC) in lithium-intercalated 2H-TaS₂. By gradually increasing the lithium content x , the CDW formation temperature is continuously suppressed, and the onset temperature of SC is increased with a maximum transition temperature $T_c = 3.5$ K for $x = 0.096$. The bulk nature of SC is confirmed by a superconducting shielding fraction of the order of unity for this composition. The electronic contribution to the specific heat and Hall resistivity data demonstrates that the CDW weakens with lithium intercalation, thereby indirectly increasing the carrier density and boosting SC. While the sign of the charge carriers in undoped 2H-TaS₂ changes from electronlike to hole type near the CDW formation temperature ~ 75 K, the lithium-intercalated Li _{x} TaS₂ shows predominantly hole-type carriers in the CDW phase even for very low lithium content.

DOI: [10.1103/PhysRevB.104.064511](https://doi.org/10.1103/PhysRevB.104.064511)**I. INTRODUCTION**

Layered transition-metal chalcogenides (TMDs) are typical quasi-two-dimensional electronic systems with multifarious phases [1–3], which show intriguing electronic and magnetic properties, including charge density waves (CDWs) and superconductivity (SC) [4–8]. SC and CDWs are two very different collective electronic phenomena, but they can coexist in the TMDs with the 2H structure variant. Upon substituting Se by S, for example, the superconducting critical temperature (T_c) is enhanced from 2H-NbSe₂ (2H-TaSe₂) through 2H-NbS₂ (2H-TaS₂) while weakening the CDW state [4,9]. The results of angle-resolved photoemission spectroscopy experiments demonstrate that these two orders (CDWs and SC) in the above TMDs may be cooperative phenomena rather than competitive [10,11]. By contrast, it is usually found that the increase of T_c is accompanied by the disappearance of CDWs for intercalated 2H-TaS₂ [12–14]. The interplay between the superconducting and CDW states has also been investigated by high-pressure experiments on bulk 2H-TaS₂ [15,16]. The T_c is enhanced dramatically to a maximal value, and the CDW is weakened and collapses at a critical pressure. At the same time, the electron-phonon coupling strength drastically decreases, and the electron density of states at the Fermi level suddenly increases beyond this critical pressure [17]. Increasing the pressure even further, T_c reaches a maximum. Thus, the simple picture of the two orders in competition may fall too short. Near the collapse of the CDW, a quantum critical point is approached, and the associated

quantum fluctuations may even enhance SC [16], which is reminiscent of the situation in the high-temperature superconductor YBa₂Cu₃O_{6+ δ} upon hole doping [18]. Therefore, the understanding of such spontaneous charge orders in two dimensions is essential to reveal the nature of SC in layered materials [19,20].

2H-TaS₂ is a typical TMD material with coexisting CDWs and SC. Stoichiometric 2H-TaS₂ is reported to show an in-plane CDW formation at ~ 75 K, as well as a transition to SC at $T_c \sim 0.8$ K [21,22]. The 2H-TaS₂(pyridine)_{1/2}, as the first organic intercalation compound, shows an enhanced T_c up to 3.5 K [23]. To further increase T_c , various alkali-metal or transition-metal atoms have been introduced into the structure to modify the electronic states of 2H-TaS₂ by doping or intercalation [24–27]. The superconducting transition temperature in 2H-Cu _{x} TaS₂, for example, is found to increase with copper intercalation [26], while the CDW is gradually weakened until it disappears at the maximum value of T_c , which is a very common feature of some intercalated TaS₂ or other two-dimensional materials. However, there have still been only a few systematic studies on the intercalation of alkali metals and the resulting induced SC and suppressed CDW state in 2H-TaS₂. The sodium intercalated 2H-TaS₂ (2H-Na _{x} TaS₂) shows superconducting transitions with T_c of 2.5 K in Na_{0.05}TaS₂ and 4.4 K in 2H-Na_{0.1}TaS₂ [14,25]. The CDW state is weakened, which has been proven by the anisotropy of resistivity and spectroscopic signatures on sodium intercalation [14,28]. Additionally, there are reports of lithium-intercalated TaS₂ by solid-phase reaction or using *n*-butyllithium solution soaking strategies [24]. The corresponding T_c can reach up to 4.5 K, where the CDW state fully disappears. In general, the CDW phase and its transition are

*zhangxf@mail.sim.ac.cn

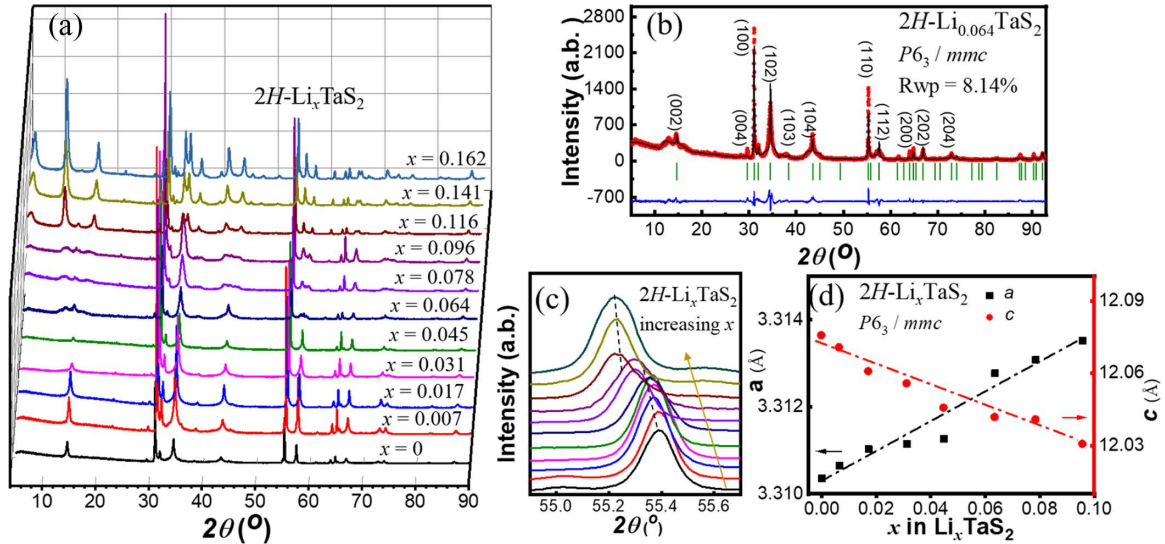


FIG. 1. (a) The powder x-ray diffraction (PXR) pattern at ambient temperature for all samples of $2H-Li_xTaS_2$ ($0 \leq x \leq 0.162$). (b) The PXR pattern of $2H-Li_{0.064}TaS_2$. The red dots are the observed data, while the black solid line represents the calculated intensities. The bottom blue solid line is the difference between the observed and calculated intensities. (c) The enlarged (110) reflections with increasing x , indicating the variation of cell parameters. (d) The change of the cell parameter for $2H-Li_xTaS_2$ samples ($0 \leq x \leq 0.096$).

not only related to the change of the Fermi level in the Ta-5d derived electron band upon alkali intercalation, but they are also affected by the superlattice formation due to interlayered alkali metal in $1T-TaS_2$ [29]. Therefore, the details of the effects of alkali metal intercalation on SC and CDWs are not yet fully understood.

To further clarify the nature of these two electronic states in intercalated $2H-TaS_2$, we performed a systematic investigation on lithium-intercalated $2H-TaS_2$ including magnetization, heat-capacity, and transport measurements. The lithium was chosen because the valence electrons are not expected to participate in chemical bonds. Moreover, the lithium atom is only weakly paramagnetic and is not expected to be detrimental to SC. We find that the T_c in $2H-Li_xTaS_2$ first increases with lithium intercalation, in which the CDW state is concurrently suppressed. As the lithium content exceeds $\approx 10\%$, the T_c moderately decreases, which is associated with a structural change.

II. EXPERIMENTS

Polycrystalline $2H-Li_xTaS_2$ samples were synthesized by solid-state reaction methods. Stoichiometric amounts of raw materials (99.9% Li_2S , 99.99% Ta, and 99.9% S powders) were mixed, ground, pressed into tablets, and sealed in evacuated silica tubes. The tubes were then loaded into a muffle furnace and annealed at $800^\circ C$ for 12 h. Finally, they were cooled down to room temperature along with the cooling of the furnace.

Single crystals for $x = 0$ and 0.007 were obtained by extending the annealing time. The tubes were heated slowly up to $800^\circ C$, sintered for 72 h, and then slowly cooled down to $600^\circ C$ with a cooling rate of $2^\circ C/h$. Finally, the tubes were cooled down in the furnace by autoswitching off the power.

The powder x-ray diffraction (PXR) data of all the samples were collected by using a Stoe STADIP diffractometer at room temperature ($Cu K\alpha_1$ radiation, $\lambda = 1.54051 \text{ \AA}$). The inductively coupled plasma mass spectrometry (ICP-MS) measurements were performed with an Agilent QQQ 8800 Triple quad ICP-MS spectrometer. The transport measurements were performed with a Physical Property Measurement System (PPMS, Quantum Design Inc.) and a standard four-probe technique. The heat capacity was measured with the heat-capacity option of the PPMS. The magnetic properties were studied in a Magnetic Properties Measurement System (MPMS 3 from Quantum Design Inc.)

III. RESULTS AND DISCUSSION

The influence of lithium intercalation on the structure of $2H-TaS_2$ is shown in Fig. 1. The identity and phase purity of the samples were determined by PXR. As we can see in Fig. 1(a), the diffraction patterns of all samples of lithium-intercalated $2H-TaS_2$ solid solutions can be well fitted to the $2H$ -type structures ($P6_3/mmc$) for lithium content below $x = 0.096$. The Rietveld refinement for $2H-Li_{0.064}TaS_2$ is shown in Fig. 1(b) as an example, demonstrating high phase purity. The lithium content of these $2H-Li_xTaS_2$ powders have been measured by ICP-MS. The slight systematic variation of the (110) peaks with increasing lithium content x is shown in Fig. 1(c). For $x \leq 0.096$, the cell parameters of a and c show only small but systematic changes [Fig. 2(d)], resulting in a shrinking of the unit cell volume. The c axis shows a regular decrease, which is like the trend observed in sodium-intercalated NbS_2 and $2H-TaS_2$ [14,30]. This can be taken as evidence that the lithium atoms are not substituting Ta atoms but are intercalated into the interlayer regions of $2H-TaS_2$. As the lithium content is increased further beyond 0.096, the crystal structure changes. The distance between the

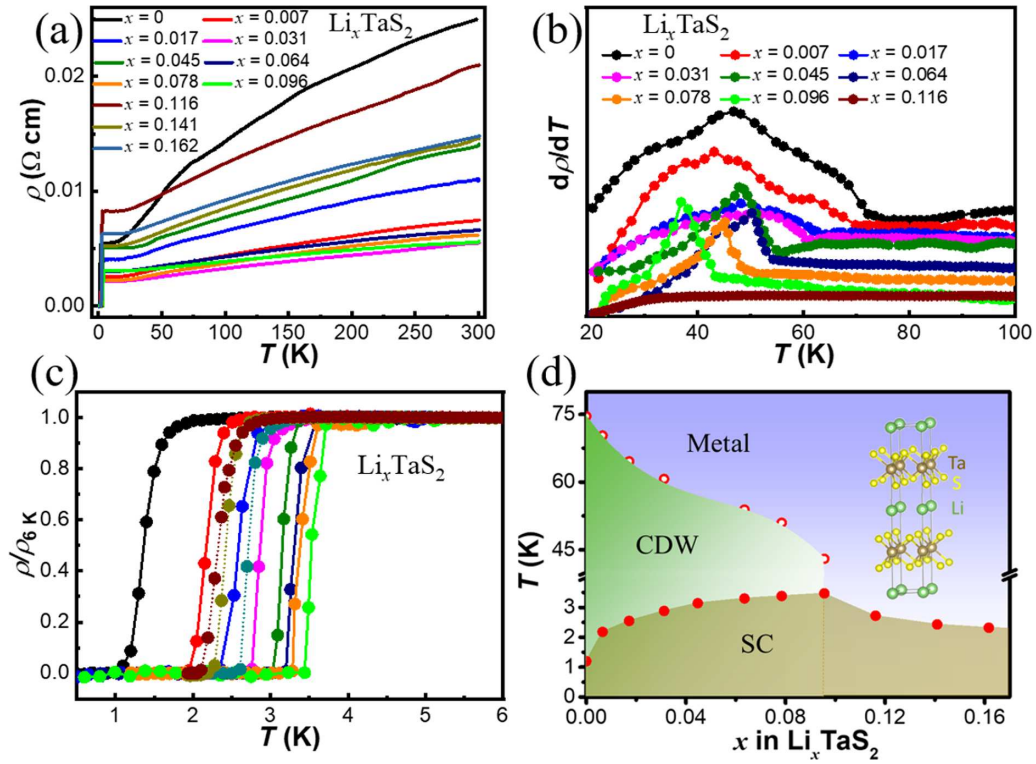


FIG. 2. The physical properties for all the samples of nominal composition $2H\text{-Li}_x\text{TaS}_2$ ($0 \leq x \leq 0.162$). (a) Resistivity in a temperature range between 0.5 and 300 K. (b) Temperature dependence of derivative of resistance, ranging from 20 to 100 K. Some curves have been offset for clarity. The data for $x = 0.141$ and 0.162 are like those for $x = 0.116$ and are therefore not shown. (c) Normalized resistivity $\rho/\rho_{(6\text{K})}$, between 0.5 and 6 K. (d) The electronic phase diagram of Li_xTaS_2 . Open circles represent the charge density wave (CDW) transition temperature, and the filled circles correspond to the superconducting transition temperature. Inset: Crystal structure of $2H\text{-Li}_x\text{TaS}_2$.

two adjacent layers can be determined to $\sim 4.4 \text{ \AA}$, which is much larger than for Li_xTaS_2 with $x < 0.096$ ($\sim 3.4 \text{ \AA}$). It is therefore unreasonable to assume that, at high lithium content, only lithium ions exist between the layers. As the samples of Li_xTaS_2 are air sensitive, it has been speculated that trace amounts of moisture enter the interlayer regions of TaS_2 for chelation with lithium ions due to the unstable chemical nature of lithium, forming $\text{Li}_x(\text{H}_2\text{O})_y\text{TaS}_2$ [31,32]. We could indeed verify this conjecture for $x \geq 0.22$ by a Rietveld refinement of corresponding PXRD data.

To investigate the effects of lithium intercalation on the CDW state and SC, we have performed comprehensive transport measurements. Figure 2(a) shows the temperature dependence of the resistivity for the low-lithium intercalation samples for temperatures ranging from 0.5 to 300 K. In the normal state, the resistivity decreases with temperature, showing a weakly metallic behavior. The parent $2H\text{-TaS}_2$ exhibits a CDW phase transition at a temperature $\sim 75 \text{ K}$, which is consistent with corresponding literature values [14,15], demonstrating the high quality of our compounds. The evolution of the CDW transition temperature with lithium intercalation was inferred from the maxima in the temperature derivative of the resistance $d\rho(T)/dT$, and it turns out to be strongly reduced with increasing lithium content. As we can see in Fig. 2(b), the CDW transition temperature is effectively suppressed by the lithium intercalation from 75 K for $x = 0$ to 41 K for $x = 0.096$ and vanishes with a further increase of the

lithium content above $x = 0.1$. Figure 2(c) shows the transition to SC for Li_xTaS_2 on an expanded scale. The undoped $2H\text{-TaS}_2$ shows a transition at $T_c \approx 1.2 \text{ K}$ (T_c defined by a 50% criterion), which is like the reported results of T_c with zero resistance $\sim 0.8 \text{ K}$ [33]. Upon lithium intercalation, T_c increases, and the CDW ordering temperature T_{CDW} decreases with increasing x , demonstrating that there is a correlation between coexisting SC and CDW phases. The T_c reaches a maximum of $\approx 3.5 \text{ K}$ for $x = 0.096$ and then decreases for higher lithium content, which we attribute to the formation of $\text{Li}_x(\text{H}_2\text{O})_y$ as a new intercalator, as we have discussed above. The resulting phase diagram is shown in Fig. 2(d), which is like the electronic phase diagram of $2H\text{-TaSe}_{2-x}\text{S}_x$ [34].

Both CDWs and SC are closely related to the conduction electrons. To further reveal the electronic effect of lithium intercalation on CDWs and SC, the Hall resistivity has been measured to estimate the carrier type and density in $2H\text{-Li}_x\text{TaS}_2$. As the samples with large lithium content became quite unstable, it was difficult to obtain enough large crystals. We therefore only measured the Hall resistivity on single crystals with low lithium content, as shown in Figs. 3(a) and 3(b) for undoped $2H\text{-TaS}_2$ and $2H\text{-Li}_{0.007}\text{TaS}_2$ crystals, respectively. The magnetic field-dependent Hall resistivity ρ_{xy} shows an ideal linear dependence, allowing us to calculate the carrier density in a simple single-band model. These results are summarized in Fig. 3(c). The temperature-dependent carrier densities of undoped $2H\text{-TaS}_2$ crystals show

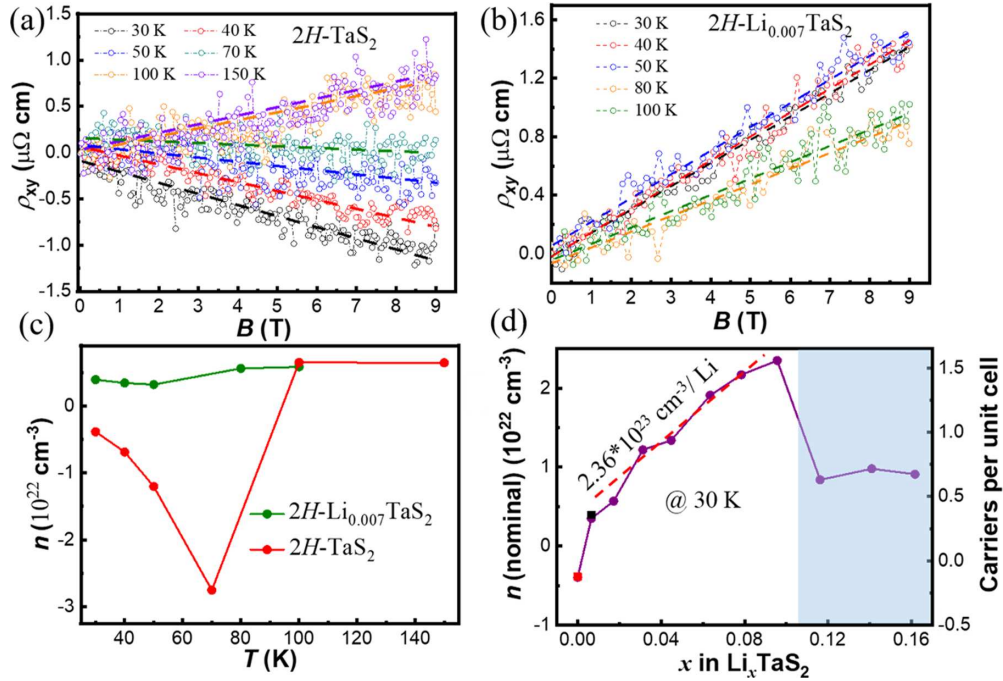


FIG. 3. (a) Magnetic field-dependent Hall resistance of $2H\text{-TaS}_2$ and (b) $2H\text{-Li}_{0.007}\text{TaS}_2$ crystals. (c) The corresponding temperature-dependent carrier densities of $2H\text{-TaS}_2$ and $2H\text{-Li}_{0.007}\text{TaS}_2$ single crystals. (d) Nominal charge carrier density and corresponding carriers per unit cell as obtained at $T = 30$ K from $2H\text{-Li}_x\text{TaS}_2$ polycrystalline samples. The dashed line is linear fit to these data for $0 < x < 0.1$. The red and the green data points are from the data for single crystals shown in Fig. 3(c).

a sudden change both in the magnitude and sign as the temperature approaches T_{CDW} , which is like reports on $2H\text{-TaS}_2$, NbSe_2 , and $\text{YBa}_2\text{Cu}_3\text{O}_{6+\delta}$ [27,35–37]. Two kinds of charge carriers, electrons and holes, are dominating the transport behavior below and above the CDW phase transition, respectively. For the $2H\text{-Li}_{0.007}\text{TaS}_2$ crystals, however, the positive Hall resistivity indicates that the dominant carriers are holes in this system. An associated comparably small change in the carrier density between 60 and 80 K also confirms the CDW transition, in a similar way as it has been observed in $2H\text{-Cu}_{0.03}\text{TaS}_2$ [38] and $2H\text{-In}_{0.5}\text{TaS}_2$ [39]. The carrier density measured on single-crystalline $2H\text{-Li}_{0.007}\text{TaS}_2$ is somewhat larger ($\approx 3.95 \times 10^{21}$ holes cm^{-3} at $T = 30$ K) than the undoped $2H\text{-TaS}_2$ ($\approx 3.86 \times 10^{21}$ electrons cm^{-3} at $T = 30$ K), which would correspond to ≈ 0.23 and 0.22 charge carriers per unit cell, respectively. In the CDW state above T_{CDW} , n is of the order of $\approx 5 \times 10^{21}$ cm^{-3} for both compositions, which corresponds to ≈ 0.3 holelike carriers per unit cell. To obtain a systematic trend as a function of lithium content x , we also performed Hall effect measurements on polycrystalline samples at $T = 30$ K. The absolute values of the corresponding numbers must be taken with the reservation that Hall measurements on polycrystals can be affected by anisotropy [40] and grain boundary effects [41]. Nevertheless, as all the samples were prepared in a similar manner, we can still identify a clear trend. The data shown in Fig. 3(d) indicate that there is an almost linear increase in n as a function of lithium content up to $x = 0.096$ and a sudden drop for larger values of x , which is not unexpected due to the changes in the crystal structure, as discussed above. Using these estimates based on the linear fitting [see Fig. 3(d)], we find that the

hypothetical intercalation of one lithium atom would correspond to a change in n by ≈ 14 carriers per formula unit. Therefore, the lithium intercalation does not primarily act as a mechanism for carrier doping but most probably leads to a weakening of the CDW state, which indirectly results in an increase of the density of mobile charge carriers with a resulting boost to SC.

To study the physical properties of lithium-intercalated $2H\text{-TaS}_2$ superconductors in more detail, we have chosen to compare intrinsic $2H\text{-TaS}_2$ ($T_c \approx 1.2$ K) with $2H\text{-Li}_{0.096}\text{TaS}_2$ (with a maximum $T_c \approx 3.5$ K). Detailed measurements of the field dependence of the resistive transition to SC and the magnetization are presented in Fig. 4. The effect of applying a magnetic field on T_c for the $x = 0$ and 0.096 samples are shown in Figs. 4(a) and 4(b), respectively. As expected, T_c is gradually suppressed, and the width of the superconducting transition increases as the magnetic field is increased. The resulting temperature dependences of the upper critical fields $\mu_0 H_{c2}(T)$ are shown in Fig. 4(c). The extrapolated slopes are $dH_{c2}/dT = -1.41$ and -1.78 T/K for $x = 0$ and 0.096 , respectively. The upper critical fields at zero temperature $\mu_0 H_{c2}(0)$ can be estimated using the Werthamer-Helfand-Hohenberg (WHH) approximation in the dirty limit [25,42]:

$$\mu_0 H_{c2}^{\text{WHH}}(0) = -0.69 T_c \left(\frac{dH_{c2}}{dT} \right)_{T=T_c}, \quad (1)$$

to $\mu_0 H_{c2}^{\text{WHH}}(0) \approx 1.17$ T for $x = 0$, and $\mu_0 H_{c2}^{\text{WHH}}(0) \approx 4.24$ T for $x = 0.096$, respectively. From $\mu_0 H_{c2} = \Phi_0 / 2\pi \xi(0)^2$, we can estimate the corresponding superconducting coherence lengths $\xi(0) = 16.8$ and 8.8 nm, respectively.

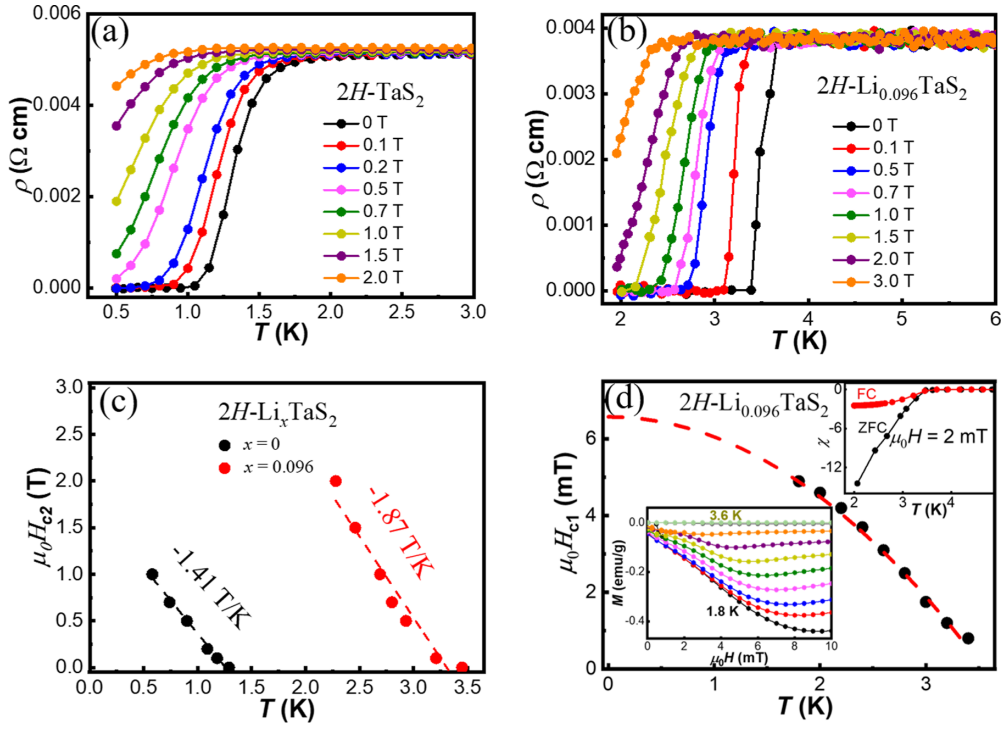


FIG. 4. (a) and (b) Field-dependent resistivity measurements for the samples $x = 0$ and 0.096 . (c) The dashed lines indicate the extrapolated slopes dH_{c1}/dT used for the Werthamer-Helfand-Hohenberg (WHH) approximation. (d) The temperature dependence of the lower critical field H_{c1} for $x = 0.096$. Lower inset: the zero-field-cooled (ZFC) field dependence of the magnetization $M(H)$ for temperatures between 1.8 and 3.6 K (in 0.2 K steps) in magnetic fields $\mu_0 H$ between 0 and 15 mT, and upper inset: ZFC and field-cooled (FC) magnetic susceptibility.

The zero-field-cooled (ZFC) field dependence of the magnetization $M(H)$ for temperatures between 1.8 and 3.6 K is shown in Fig. 4(d) for the $2H\text{-Li}_{0.096}\text{TaS}_2$ sample, exhibiting typical type-II superconducting behavior. The ZFC and the field-cooled (FC) magnetic susceptibilities measured in 2 mT are shown in the lower inset of Fig. 4(d). A large superconducting shielding fraction of $\sim 120\%$, as determined from comparing the measured magnetization with theoretical expectations, is an indication of the bulk nature of SC, thereby confirming complete magnetic flux expulsion, as required by the Meissner-Ochsenfeld effect (an overestimated value $> 100\%$ is most probably caused by demagnetization effects). By defining the lower critical field H_{c1} as the minimum on the $M(H)$ curves, its temperature dependence can be well fitted using an empirical formula [43]:

$$\mu_0 H_{c1}(T) = \mu_0 H_{c1}(0) \left[1 - \left(\frac{T}{T_c} \right)^2 \right]. \quad (2)$$

A resulting estimate of the corresponding lower critical field is $\mu_0 H_{c1}(0) \approx 6$ mT. Together with $\mu_0 H_{c1}(T) = \frac{\Phi_0}{4\pi\lambda^2} \ln \frac{\lambda}{\xi}$, we obtain an estimate for the London penetration depth $\lambda(0) \simeq 310$ nm and $\kappa = \frac{\lambda(0)}{\xi(0)} \simeq 35.2$, confirming that the material is a strongly type-II superconductor.

The low-temperature specific heats of the $2H\text{-Li}_x\text{TaS}_2$ samples ($x = 0.045, 0.064, \text{ and } 0.096$) are shown in Fig. 5. As expected, the specific heat data show a peak at $T_c \approx 2.9, 3.3, \text{ and } 3.5$ K, respectively. These results are consistent with the data from our transport measurements. The normal state specific heat can be fitted by a standard expression at low

temperatures:

$$\frac{C_p}{T} = \gamma + \beta T^2, \quad (3)$$

where γ is the Sommerfeld constant, which is proportional to the electron density of states $D(E_F)$ at the Fermi level. The fitted values of β for all selected samples are ~ 0.4 mJ/mol K⁴ (see Table I), as reported for Cu_xTaS_2 and Cu_xTiSe_2 [26,44], corresponding to a Debye temperature of ~ 244 K, which we have calculated from the corresponding three-dimensional lattice Debye model. The resulting electronic contributions to the

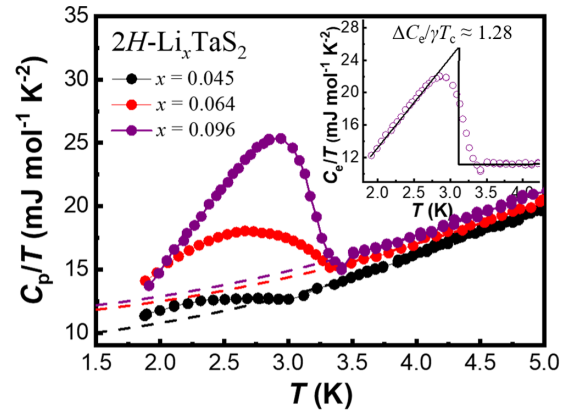


FIG. 5. Specific-heat C/T for $2H\text{-Li}_x\text{TaS}_2$ with different x values. Right inset: The solid line shows an entropy-conserving construction to obtain $\Delta C_e/\gamma T_c$ for an optimally intercalated $2H\text{-Li}_{0.096}\text{TaS}_2$ sample.

TABLE I. Electronic (γ) and lattice (β) contributions according to a fit to Eq. (3). We obtained the calculated values γ_c from the free-electron model using the measured carrier densities n .

x	β (mJ/mol K ⁴)	Carrier density n (cm ⁻³)	Calculated γ_c (mJ/mol K ²)	Experimental γ (mJ/mol K ²)	Effective mass
0	0.35 [45]	3.5×10^{21}	0.88	8.5 [45,46]	9.66
0.045	0.41(3)	7.6×10^{21}	1.14	9.0	7.90
0.064	0.39(1)	1.05×10^{22}	1.27	10.9	8.59
0.096	0.40(2)	1.2×10^{22}	1.33	11.2	8.35

specific heat for these lithium-intercalated samples increases with lithium content, and they are slightly larger than that of the parent compound $2H\text{-TaS}_2$ ($\gamma = 8.5$ mJ/mol K²) [45,46]. These results demonstrate that the lithium intercalation increases, along with the charge carrier density n , also the electron density of states at the Fermi level [47]. We state here that the measured values for γ of the order of 10 mJ/mol K⁻² are far larger than one can expect from a simple free-electron model. Assuming one charge carrier per unit cell, we obtain with $\gamma_c = \pi^2 k_B^2 D(E_F)/3$ and $D(E_F) = (3n/\pi)^{1/3} m_e / (\pi \hbar^2)$ a calculated value γ_c of the order of 1.5 mJ/mol K⁻² only, which may hint at an enhanced effective mass in superconducting $2H\text{-Li}_x\text{TaS}_2$ (see Table I). Here, k_B is the Boltzmann constant, m_e the electron mass, and \hbar the reduced Planck constant. The corresponding effective mass enhancements estimated from the measured carrier densities and the specific heat data are all of the order of $\approx 8\text{--}9$ (see Table I). The right inset of Fig. 5 shows the discontinuity in the electronic specific heat (C_e/T_c) at the superconducting transition with the phonon contribution subtracted and with a Bardeen-Cooper-Schrieffer (BCS) entropy-conserving construction. The obtained ratio $\Delta C_e/\gamma T_c = 1.28$ is very close to the standard BCS value 1.43, thereby qualifying $2H\text{-Li}_x\text{TaS}_2$ as weakly coupled superconductors [48].

IV. CONCLUSIONS

Figure 2(d) summarizes the electronic phase diagram of $2H\text{-Li}_x\text{TaS}_2$, showing the evolution of the superconducting, CDW, and metallic phases with varying lithium content x . The superconducting and CDW states are interrelated and coexist with each other. The lithium intercalation gradually enhances the superconducting transition temperature and weakens the CDW state. At $x \approx 0.096$, the CDW phase is fully suppressed, and SC reaches its maximum critical temperature $T_c = 3.5$ K, with a fully developed discontinuity in the specific heat which is compatible with a weak-coupling scenario. The changes upon lithium intercalation are accompanied by an increase of the hole-type carrier density. However, the measured changes in the charge carrier densities are too large to be explained by doping alone. We therefore suggest that lithium intercalation leads primarily to a weakening of the CDW state, which then indirectly causes an increase of the density of mobile hole-type charge carriers.

ACKNOWLEDGMENT

This paper was supported by the Schweizerische Nationalfonds zur Förderung der Wissenschaftlichen Forschung (Grants No. 20-175554 and No. 206021-150784).

- [1] M. Chhowalla, H. S. Shin, G. Eda, L.-J. Li, K. P. Loh, and H. Zhang, *Nat. Chem.* **5**, 263 (2013).
- [2] B. Radisavljevic, A. Radenovic, J. Brivio, V. Giacometti, and A. Kis, *Nat. Nanotechnol.* **6**, 147 (2011).
- [3] Q. H. Wang, K. Kalantar-Zadeh, A. Kis, J. N. Coleman, and M. S. Strano, *Nat. Nanotechnol.* **7**, 699 (2012).
- [4] A. H. Castro Neto, *Phys. Rev. Lett.* **86**, 4382 (2001).
- [5] D. W. Shen, B. P. Xie, J. F. Zhao, L. X. Yang, L. Fang, J. Shi, R. H. He, D. H. Lu, H. H. Wen, and D. L. Feng, *Phys. Rev. Lett.* **99**, 216404 (2007).
- [6] J. A. W. Straquadine, F. Weber, S. Rosenkranz, A. H. Said, and I. R. Fisher, *Phys. Rev. B* **99**, 235138 (2019).
- [7] D. Shin, N. Tancogne-Dejean, J. Zhang, M. S. Okyay, A. Rubio, and N. Park, *Phys. Rev. Lett.* **126**, 196406 (2021).
- [8] A. Fang, A. G. Singh, J. A. W. Straquadine, I. R. Fisher, S. A. Kivelson, and A. Kapitulnik, *Phys. Rev. Research* **2**, 043221 (2020).
- [9] J. A. Wilson and A. D. Yoffe, *Adv. Phys.* **18**, 193 (1969).
- [10] T. Valla, A. V. Fedorov, P. D. Johnson, P. A. Glans, C. McGuinness, K. E. Smith, E. Y. Andrei, and H. Berger, *Phys. Rev. Lett.* **92**, 086401 (2004).
- [11] T. Valla, A. V. Fedorov, P. D. Johnson, J. Xue, K. E. Smith, and F. J. DiSalvo, *Phys. Rev. Lett.* **85**, 4759 (2000).
- [12] L. J. Li, X. D. Zhu, Y. P. Sun, H. C. Lei, B. S. Wang, S. B. Zhang, X. B. Zhu, Z. R. Yang, and W. H. Song, *Physica C* **470**, 313 (2010).
- [13] X. Zhu, Y. Sun, S. Zhang, J. Wang, L. Zou, L. E. DeLong, X. Zhu, X. Luo, B. Wang, G. Li, Z. Yang, and W. Song, *J. Phys. Condens. Matter* **21**, 145701 (2009).
- [14] L. Fang, Y. Wang, P. Y. Zou, L. Tang, Z. Xu, H. Chen, C. Dong, L. Shan, and H. H. Wen, *Phys. Rev. B* **72**, 014534 (2005).
- [15] D. C. Freitas, P. Rodière, M. R. Osorio, E. Navarro-Moratalla, N. M. Nemes, V. G. Tissen, L. Cario, E. Coronado, M. García-Hernández, S. Vieira, M. Núñez-Regueiro, and H. Suderow, *Phys. Rev. B* **93**, 184512 (2016).
- [16] X.-M. Zhao, K. Zhang, Z.-Y. Cao, Z.-W. Zhao, V. V. Struzhkin, A. F. Goncharov, H.-K. Wang, A. G. Gavriliuk, H.-K. Mao, and X.-J. Chen, *Phys. Rev. B* **101**, 134506 (2020).
- [17] K. Zhang, H. Jiang, J. Yang, J. Zhang, Z. Zeng, X. Chen, and F. Su, *Appl. Phys. Lett.* **117**, 101105 (2020).

- [18] B. J. Ramshaw, S. E. Sebastian, R. D. McDonald, J. Day, B. S. Tan, Z. Zhu, J. B. Betts, R. Liang, D. A. Bonn, W. N. Hardy, and N. Harrison, *Science* **348**, 317 (2015).
- [19] J. Chang, E. Blackburn, A. T. Holmes, N. B. Christensen, J. Larsen, J. Mesot, R. Liang, D. A. Bonn, W. N. Hardy, A. Watenphul, M. v. Zimmermann, E. M. Forgan, and S. M. Hayden, *Nat. Phys.* **8**, 871 (2012).
- [20] Y. I. Joe, X. M. Chen, P. Ghaemi, K. D. Finkelstein, G. A. de la Peña, Y. Gan, J. C. T. Lee, S. Yuan, J. Geck, G. J. MacDougall, T. C. Chiang, S. L. Cooper, E. Fradkin, and P. Abbamonte, *Nat. Phys.* **10**, 421 (2014).
- [21] L. F. Mattheiss, *Phys. Rev. B* **8**, 3719 (1973).
- [22] R. Delaplace, P. Molinie, and D. Jerome, *J. Phys. Lett.* **37**, 13 (1976).
- [23] F. R. Gamble, F. J. DiSalvo, R. A. Klemm, and T. H. Geballe, *Science* **168**, 568 (1970).
- [24] Y. Nozuyama and S. Iida, *Jpn. J. Appl. Phys.* **23**, L319 (1984).
- [25] A. Lerf, F. Sernetz, W. Biberacher, and R. Schöllhorn, *Mater. Res. Bull.* **14**, 797 (1979).
- [26] K. E. Wagner, E. Morosan, Y. S. Hor, J. Tao, Y. Zhu, T. Sanders, T. M. McQueen, H. W. Zandbergen, A. J. Williams, D. V. West, and R. J. Cava, *Phys. Rev. B* **78**, 104520 (2008).
- [27] N. Z. Wang, M. Z. Shi, C. Shang, F. B. Meng, L. K. Ma, X. G. Luo, and X. H. Chen, *New J. Phys.* **20**, 023014 (2018).
- [28] W. Z. Hu, G. Li, J. Yan, H. H. Wen, G. Wu, X. H. Chen, and N. L. Wang, *Phys. Rev. B* **76**, 045103 (2007).
- [29] C. Pettenkofer and W. Jaegermann, *Phys. Rev. B* **50**, 8816 (1994).
- [30] W. P. F. A. M. Omloo, and F. Jellinek, *J. Less Common Met.* **20**, 121 (1970).
- [31] D. C. Johnston, *Mater. Res. Bull.* **17**, 13 (1982).
- [32] D. C. Johnston and S. P. Frysinger, *Phys. Rev. B* **30**, 980 (1984).
- [33] S. Nagata, T. Aochi, T. Abe, S. Ebisu, T. Hagino, Y. Seki, and K. Tsutsumi, *J. Phys. Chem. Solids* **53**, 1259 (1992).
- [34] L. Li, X. Deng, Z. Wang, Y. Liu, M. Abeykoon, E. Dooryhee, A. Tomic, Y. Huang, J. B. Warren, E. S. Bozin, S. J. L. Billinge, Y. Sun, Y. Zhu, G. Kotliar, and C. Petrovic, *npj Quantum Materials* **2**, 11 (2017).
- [35] H. N. S. Lee, H. McKinzie, D. S. Tannhauser, and A. Wold, *J. Appl. Phys.* **40**, 602 (1969).
- [36] L.-j. Li, Z.-a. Xu, J.-q. Shen, L.-m. Qiu, and Z.-h. Gan, *J. Phys. Condens. Matter* **17**, 493 (2005).
- [37] P. M. C. Rourke, A. F. Bangura, C. Proust, J. Levallois, N. Doiron-Leyraud, D. LeBoeuf, L. Taillefer, S. Adachi, M. L. Sutherland, and N. E. Hussey, *Phys. Rev. B* **82**, 020514(R) (2010).
- [38] X. D. Zhu, Y. P. Sun, X. B. Zhu, X. Luo, B. S. Wang, G. Li, Z. R. Yang, W. H. Song, and J. M. Dai, *J. Cryst. Growth* **311**, 218 (2008).
- [39] Y. Li, Z. Wu, J. Zhou, K. Bu, C. Xu, L. Qiao, M. Li, H. Bai, J. Ma, Q. Tao, C. Cao, Y. Yin, and Z.-A. Xu, *Phys. Rev. B* **102**, 224503 (2020).
- [40] T. K. Xia and D. Stroud, *Phys. Rev. B* **37**, 118 (1988).
- [41] A. K. Ghosh, A. Rose, H. P. Maruska, D. J. Eustace, and T. Feng, *Appl. Phys. Lett.* **37**, 544 (1980).
- [42] F. von Rohr, R. Nesper, and A. Schilling, *Phys. Rev. B* **89**, 094505 (2014).
- [43] E. H. Brandt, *Phys. Rev. B* **68**, 054506 (2003).
- [44] E. Morosan, H. W. Zandbergen, B. S. Dennis, J. W. G. Bos, Y. Onose, T. Klimczuk, A. P. Ramirez, N. P. Ong, and R. J. Cava, *Nat. Phys.* **2**, 544 (2006).
- [45] M. Abdel-Hafiez, X. M. Zhao, A. A. Kordyuk, Y. W. Fang, B. Pan, Z. He, C. G. Duan, J. Zhao, and X. J. Chen, *Sci. Rep.* **6**, 31824 (2016).
- [46] M. H. Van Maaren, *Phys. Lett. A* **40**, 353 (1972).
- [47] J. Pan, C. Guo, C. Song, X. Lai, H. Li, W. Zhao, H. Zhang, G. Mu, K. Bu, T. Lin, X. Xie, M. Chen, and F. Huang, *J. Am. Chem. Soc.* **139**, 4623 (2017).
- [48] D. C. Johnston, *Supercond. Sci. Technol.* **26**, 115011 (2013).