Multiphoton Ionization of Uracil at 355 nm

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Abstract We present the experimental results from ionization and dissociation by multiphoton absorption (MPI) of uracil and a mixture of uracil with Ar using a Reflectron time of flight spectrometer along with radiation from 355 nm at pulsed Nd:YAG laser . We focus on the light ions production. The MPI mass spectra show that the presence and intensity of the resulting ions depend on the density power of the laser. The resulting ions in the mass spectra are identified and found similar behavior in the case of H⁺ and C⁺ as when multiple charged ions are used. Different results were found in contrast with those, recently reported, when electrons or photons of other wavelength were used. The number of 355nm absorbed photons was calculated accordingly to Keldysh theory and similar results were fond using pure uracil or uracil-Ar mixture. Our results are compared with those obtained in other laboratories under different experimental conditions, some of them show only partial agreement and differences are discussed.

Keywords: Uracil, Multiphoton, Dissociation

1. INTRODUCTION

Uracil is one of the four nitrogen bases that belong to RNA and is one of the building blocks of RNA. The last years special interest has been concentrated on the study of the effects under UV radiation in nucleic acids also it has been considered the role that this molecule could have played in the origin and development of life on our planet [1-3]. Although aspects of astrochemical, medical or biological are of great interest, the objective of the present study is the radiation effects on the uracil molecules, processes of dissociation and ionization. Recently, it was shown that radiation can produce secondary ions/ fragments with kinetic energies from thermal up to several hundred eV in a

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Prieto, E. Guerrero, A. Martínez, D. Álvarez, I. Cisneros, C. biological medium, and in the subsequent scattering events these energetic ions/fragments can also cause severe damage to DNA [3-4].

The Monte Carlo simulations for radiation damage studies on DNA and RNA [6], account for ionization. However, the probability of simultaneous ionization and dissociation and high energy deposition (known as dissociative ionization) has not been considered in these simulations, mainly due to lack of data until very recently [7]. In particular processes of dissociation and ionization are useful to study possible different molecular mechanisms caused by the 355 nm laser radiation. Most of the available experimental data by the interaction with electrons, photons from laser are in the range of 220-290 nm [8-9], synchrotron light of energy from 6 to 22 eV [10]and 260nm [11] or multiple charged ions [4]. In the present study, the photodissociation and photoionization of uracil in the multiple photon absorption regimes were investigated at the wavelength of 355 nm and intensities of radiation in the range 10⁸ to 10⁹ W·cm⁻². Radiation interacts with a cooled molecular beam of uracil or Ar-uracil mixture produced by the adiabatic expansion of vapors into a high vacuum chamber at 10^{-8} torr. The resulting ions were analyzed using a home-assembled Jordan R-TOF mass analyzer. On the basis of the detected ions at 355 nm, the processes were identified as a Dissociation-Ionization. The number of photons required to form a ion was calculated accordingly with the Keldysh approximation [12] and is reported in the Multiphoton Ionization Spectra (MPI). The MPI show that the presence and intensity of the ions depends on the density power as well as the wavelength and the fact of the use Ar as carrier gas. The results are compared with those obtained in other laboratories under different experimental conditions [8-10]. This work gives new insights on behavior of the uracil molecules when they interact with photons and the products that can emerge as result of photoionization and photodissociation, as the well as the dependence of the amount or ions produced when Ar is used as carrier gas in contrast of when pure uracil is used.

2. EXPERIMENTAL

The photofragment spectra for uracil multiphoton absorption were obtained by a high resolution time of flight mass spectrometer, Reflectron (R-TOF) figure 1. The Reflectron is a commercial spectrometer (Jordan) modified in the laboratory [13], coupled to a vacuum chamber 60 cm diameter, housing the interaction zone to produce the fragment ions. A sample of uracil (Sigma-Aldrich, purity ~99 %) was introduced by pulsed valve, pure or seeded with Ar, into the ionization chamber in gas phase. The pulse valve has an extension with a conical tip termination inside the chamber that allows the adiabatic gas



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1. Pulsed valve, **2.** Sample reservoir, **3.** Valve, **4.** Skimmer, **5.** Electrostatic lenses, **6.** Extraction and acceleration plates, **7.** RTOF Electrostatic lenses, **8.9** Detectors

Figure 1. Experimental setup

expansion closer to the skimmer for the generation of the supersonic molecular beam. A cooled molecular beam of uracil or with uracil/Ar was produced by adiabatic expansion in a high-vacuum chamber at 2×10^{-8} torr. To sublimate the uracil sample, a controlled coiled resistive heater in the reservoir was used, the temperature was monitored and held at 110° C. The pulsed valve was synchronously coupled with the laser pulses with an opening time of 250 µs to ensure that the uracil molecules and laser light coincide at the center of the interaction zone. The operating pressure was 2×10^{-6} torr. The 355 nm laser radiation was produced from the third harmonic of a Nd:YAG laser (Spectra Physics), operating at 30 Hz repetition rate. The temporal laser pulse width is 5.5 ns and the energies per pulse from 1 to 30 mJ.

The laser radiation (with a Gaussian profile and vertically polarized) was focused into the interaction region using a 15 cm focal length lens. The diameter at the focal point was $\sim 80.0 \ \mu\text{m}$. Under these experimental conditions, radiation intensities between 10⁹ and 10¹⁰ W·cm⁻² were achieved. The cooled molecular beam interacted orthogonally with the laser radiation at a point located between two parallel plates continuously polarized at 5.0 KeV and 4.5 KeV, corresponding to the extraction and the acceleration potentials, respectively.

The distance between the plates was 0.6 cm. Holes of 10 mm diameter at the center of each plate with a fine metal mesh were used to extract the positively charged ions from the interaction region. The ions were driven along the field-free region of the Reflectron and eventually reached a dual Chevron microchannel plate detector after they were refocused. The ions arrived to the detector, according to the ratio charge/masses (e/m). The resolution achived was of the order of 1000. The current signal was pre-amplified, voltageconverted, digitized and sampled in time using a picosecond time analyzer from EG&G Ortec. Prieto, E. Guerrero, A. Martínez, D. Álvarez, I. Cisneros, C.

3. RESULTS AND DISCUSSION

Uracil molecules were multi-photon ionized at the wavelength of 355 nm with fluence between $10^8 - 10^9$ W cm⁻². That corresponds to the multiphoton absorption regime accordingly to the Keldysh theory [12,14], since γ is >> 1. Also from Keldysh theory the relation between the ion counts per pulse (I), the laser pulse energy (E) and the photon order (α), the number of photons involved is $I = cE^{\alpha}$, where c is a constant. We measured the ion yields for the most prominent ions and calculated the number of photons involved, the results are displayed in the Table [1].

Photon order					
m/z	Argon	Without Ar			
1	2.04	1.90			
12	1.09	0.89			
13	2.02	1.75			
14	1.99	1.88			
15	1.99	2,07			
17	3.54	3,14			
18	3.36	2.21			
24	1.36	1.17			
25	1,56	1,17			
26	1.44	1,08			
27	1.01	0.68			
28	1,67	1,43			
29	1.89	1.46			
30	1,62	1.31			

Table 1: Number of calculated photons

As it is shown the values are slightly higher for the case where Ar is used in conjunction with uracil. Figure [2] shows the case for m/z = 28, CH_2N^+ . Overall uncertainty is estimated as 20%

Furthermore, the corresponding energies calculated from the number of photons reported in [8] for m/z = 28,14 y 12 are higher, especially for the case of m/z = 12, C⁺, where the result is three times higher than in our measurements. This fact deserves further investigation. It can be interpreted as in this process, a transition to an excited molecular orbital becomes saturated in such way that a small number of photons can ionize the highly excited state.

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Figure 2: Power dependence of CH₂N⁺ from MFI on laser intensity

The shapes and photon fluence dependence do not suggest that the resulting ions are produced by a different pathways from other experiments apart from, J. DeVries [4].

Table [2] presents the relative intensities of the m/z peaks taken from a ratio of each individual intensity to the 95% of the total ion current in uracil and Ar-uracil spectrum, at 280 mW. In Fig [3] a comparison of MPI mass spectra for the cases with and without of Ar, it seems that the Ar makes the ions number to be slightly lower.

Table 2: Relative intens	sity for	diferent	m/z	ions
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Relative intensity					
km	w/Argon	Without Argon			
1	13,47	14,51			
12	31,60	29,26			
13	5,69	5,87			
14	2,93	3,03			
15	5,67	6,18			
18	3,19	2,54			
24	12,60	12,93			
25	5,00	4,86			
26	7,26	7,57			
27	1,90	2,14			
28	1,22	1,39			
29	4,88	4,80			

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Figure 3: Comparison of time of fight spectra for the interaction of 280 W/cm² with pure uracil and uracil-Ar mixture.

The most import and interesting feature in the present measurements is the different structure of the our TOF spectra. Small ions prevail with maximum intensities of H⁺ and C⁺ only observed in experiments with ionization using highly charged ions (Xe^{q+} q = 5 - 25)[4] Figure [3].

From the ions observed, the fragmentation process corresponds to a ladder switching mechanism. The fragmentation of parent ion, not detected in the present study, can give rise m/z = 69 ion, $C_3H_3NO^+$, reported in [10], which in turn it can lose H₂O producing the m/z = 51 ion, C₂HN⁺. This ion was reported for first time in electron experiments [15, 17] using electrons and Ar ions as projectiles [16]. The ion m/z = 45 has not been reported, it is suggested corresponds to CO₂H⁺ or CH₂NO⁺ ions. With regard to the ion of m/z = 43, HNCO⁺, it is produced along with the fragment m/z = 69 which in turn dissociates as m/z = 41, HC_2O^+ . Other fragmentation pathway of m/z = 69 leads to the formation of m/z = 28, HCNH⁺. The ion m/z = 29, corresponds to N₂H⁺ HCO⁺ or CH₂N⁺. A deuterated ion comparison supports to the last one. The m/z = 27 could be $C_2H_2^+$ or HCN⁺, Reflectron impact experiments suggest HCN⁺. For m/z = 26: C₂H₂ [10], it was interpreted as formed via loss of 2 HNCO molecules directly from the parent ion, a not ladder switching mechanism. As in [10] we assign m/z = 18, H_2O^+ , m/z= 17, NH₃⁺, m/z = 16, O⁺, m/z = 15: NH⁺, not reported before. The ions m/z = 14: N⁺, m/z = 12: C⁺, m/z = 1:H⁺, have been reported in [4]. The ratio of intensities of peaks m/z = 12 and 1 agree better with Ref [4,]. Coupier, B., [18], measured both using proton and electron impact and discussed the structure of the various of the ions observed, such as m/z = 69, 42, 29 and 28. Appearance Energies (AE) for certain of the ions detected have been already

published and they have been used mainly to propose different dissociation pathways in some cases to justified the appearance of such ions [10].

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Unfortunately, although computational work can simulate quite accurately the behavior of molecules such as uracil, fully *ab initio* simulations of molecular photodynamics are far beyond reach even for the most advanced supercomputers. This makes the need of experimental more crucial since still there are several open questions to answer related to uracil photodissociation.

Table 3: is the summary of different TOF data for uracil along with that obtained in the present experiment.

Present wor	k	Riszka et al.	Bare et al.	Iinhoff etal.	Rice et al.	Jochims et.
ion			m			
H^+	1		1	1		
H_2^+	2			2		
C^+	12	12	12	12		
CH^+	13	13	13	13		
$\mathrm{N^{+}, CH_{2}^{+}}$	14	14	14	14		14
CH_3^+	15		15	15		
$\mathrm{O^{+},NH_{2}^{+}}$	16			16		
$\mathrm{NH}_3^+,\mathrm{OH}^+$	17			17		17
H_2O^+	18			18		18
C_2^+	24	24	24	24		
C_2H^+	25		25	25		
$C_2H_2^+, CN^+$	26	26	26	26	26	26
CNH^+	27	27	27	27	27	27
$\rm NHCH^{+}, \rm CO^{+}$	28	28	28	28	28	28
HNCH_2^+	29	29	29	29	29	29
$\rm NO^+, \rm H_2 \rm CO^+$	30			30		
$\rm COH_3^+, \rm NOH^+$	31					
NOH_2^+	32					
C_3^+	36					
$C_{3}H^{+}$	37					
$C_2 N^+$	38	38	38	38	38	
$C_2 NH^+$	39	39	39	39	39	39
$C_2 NH_2^+$	40	40	40	40	40	40

Prieto, E.	$C_2NH_3^+$	41	41	41	41	41	11
Guerrero, A. Martínez D	$C_2H_2O^+$	42	42	42	42	42	42
Álvarez, I.	HNCO+	43	43	43	43	43	43
Cisneros, C.	$\mathrm{CO}_2^+,\mathrm{CH}_2\mathrm{NO}^+$	44	44	44	44	44	44
	CO ₂ H ⁺ , CH ₃ NO ⁺	45					
	C_3HN^+	51			51	51	••••

Ryszka[9], Barck[8], Imhoff[16], Rice[17] Jochims[10]

CONCLUSIONS

Multiphoton studies are an excellent tool to address the main process that take place when photons interact with neutral species. In the present work, it is reported the analysis of photodissociation of uracil and uracil seeded with Ar gas using laser radiation of 355 nm in a cooled molecular beam. TOF-MS spectra are presented. Table 3 is the summary of different TOF data for uracil along with that obtained in the present experiment

At 355 nm, photon absorption is followed by a D-I. Some new dissociation channels were proposed based on cations detected, the assignment of the m/z peaks in our MPI absorption spectra are mostly based on previous data. We have focused in the range of low m/z regime, our results are different from the ones found when electrons or photons of different energies were employed, mainly with respect to the H⁺ and C⁺ production. With regard to those ions, our results are like those found from the interaction with highly charged ions. MPI was found similar with and without Ar as carrier gas. For low mass ions the number of absorbed photons was calculated accordingly to the Keldysh theory in order to compare them with some theoretical results. However, are necessary more experiments and calculations to provide a deeper understanding of the uracil photodissociation.

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