1 Article

2 Effects of Waste Activated Sludge Extracellular

3 Polymeric Substances (EPS) Composition on

4 **Biosorption for Primary Carbon Diversion**

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8 Abstract: Extracellular polymeric substances make up a significant fraction of the organic matter in 9 waste activated sludge, strongly influence the sludge's physicochemical properties, as well as serve 10 biological functions. Biosorption is a physicochemical process in which biomass, such as waste 11 activated sludge, acts as a sorbent which uptakes external nutrients and compounds from a sorbate 12 such as raw wastewater into its cell matrix. This paper characterizes the composition of waste 13 activated sludge extracellular polymeric substances from various treatment processes and analyzes 14 their ability to conduct biosorption when mixed with raw wastewater. Biosorption effectiveness was 15 characterized as the ability to remove the soluble fraction of the organic matter from the wastewater. 16 Due to different waste activated sludge bioaggregate structures, the amount of total organics 17 available in the sludge could not be used as an indicating factor for how effective the sludge was at 18 conducting biosorption. However, a positive correlation was found between the amount of 19 extracellular polymeric substances present in the organic fraction of the sludges and the normalized 20 removal of soluble organic matter. The extracellular polymeric substances, regardless of its chemical 21 composition, removed nearly the same amount of soluble and truly soluble organics removing 22 1.43±0.15 (n=16) mg of soluble chemical oxygen demand, and 1.20±0.18 (n=16) mg of truly soluble 23 chemical oxygen demand, per mg of cation exchange resin extracted extracellular polymeric 24 substances. Additionally, it was observed that utilizing multiple extraction methods in series could 25 increase extracellular polymeric substances extraction yields and provide a more holistic insight 26 into its composition as well as its biosorption properties.

Keywords: biosorption; extracellular polymeric substances; cation exchange resin; sulfide
 extraction; base extraction; high-rate biological contactor, primary wastewater treatment

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30 1. Introduction

31 Waste activated sludges (WAS) are bioaggregates of cells as well as organic matter exteriors 32 known as extracellular polymeric substances (EPS). EPS production is dependent on microbial 33 growth phases and conditions though the mechanisms are not well understood. The composition of 34 EPS consists of a combination of secretions, cell lysis products, hydrolyzed macromolecules, and 35 sorbed organics and compounds. EPS plays a vital role in the physicochemical properties of the 36 sludge including their structure, surface charge, flocculation, settling and dewatering properties, and 37 adsorption ability [1]. EPS also serves as means for biological functions such as cell-to-cell recognition 38 and communication, though the physicochemical properties are more pertinent for this study-39 particularly adsorption ability.

The literature suggests that a large fraction of the organic matter contained in WAS consists of EPS, but due to the dynamic nature of sludges and the variance between sludges from different treatment processes, there is no consensus about what exactly constitutes EPS. EPS are often categorized as loosely-bound EPS, tightly-bound EPS, and soluble EPS, also referred to as soluble microbial products (SMP). Physicochemical properties of the sludge are typically attributed to the bound portion of the EPS though some studies suggest that the SMP also may play a role [1]. The SMP and bound EPS can be separated by centrifugation, but the extraction of bound EPS requires physical or chemical treatment in order to be released from the cell matrix before centrifugation [2].

48 Biosorption is a physicochemical process where WAS acts as a sorbent which uptakes external 49 nutrients and compounds from a sorbate-such as raw, untreated wastewater (WW)-into its cell 50 matrix. In this process, particulate and soluble organic matter contained in the WW adsorbs onto the 51 WAS flocs. Conventional wastewater primary treatments such as screens and gravity sedimentation 52 are effective at removing the particulate organic fraction from wastewater however, they do not 53 effectively remove the soluble fraction. The high-rate biological contactor (HRBC) is a primary 54 treatment process that removes particulate organic matter and a portion of the soluble organic matter 55 by means of biosorption. This process operates in short hydraulic retention time (HRT) and low 56 dissolved oxygen (DO) concentration in promoting more carbon diversion to the anaerobic digesters 57 for methane gas production as well as reducing aeration requirements in secondary treatment 58 processes. High biomethane potential (BMP) values of the float in a pilot test using a trickling 59 filter/solids contact (TF/SC)'s WAS as a biosorbent revealed that oxidation of the sorbed organic 60 matter was at a minimum [3].

61 EPS contained in the WAS has been shown to have an influence on the biosorption process in 62 past studies and in the crucial role that they play in the biosorption of heavy metals [4]. Additionally, 63 while some biological activity occurs in the raw wastewater, it has been shown that properties of the 64 sludge have a larger influence on the biosorption than the properties of the influent wastewater [3]. 65 Though it is suspected that higher EPS concentrations within the WAS correlates to more biosorption, 66 the study seeks to identify, if any, key compositional fractions of the EPS responsible for biosorption. 67 To evaluate the objective of this study the EPS from different structural sources and treatment 68 processes were extracted and related to the ability of the WAS to remove soluble organic matter from 69 WW in a simulated HRBC process.

6) WW III a simulated TINDC proces

70 2. Materials and Methods

71 2.1 Procedure for Biosorption

WW and WAS were collected and transported at ambient conditions and were tested within two hours of collection. Samples were taken at three different wastewater treatment plants (WWTPs) on the island of O'ahu with varying treatment methods and design capacities. Honouliuli WWTP in Ewa Beach uses a trickling filter and solid contact (TF/SC) process with a design capacity of 38 million gallons per day (MGD). Wahiawa WWTP utilizes a membrane bioreactor (MBR) and is designed to treat up to 2.49 MGD. Waimanalo WWTP is a conventional activated sludge (CAS) plant with a design capacity 0.6 MGD.

A primary treatment HRBC biosorption contact tank was simulated by using a bench scale setup.
WW and WAS were mixed at a 5% WAS by volume mixing ratio in a five-liter contact tank while
being constantly stirred and aerated to 1 mg/L dissolved oxygen for a 30-minute contact time.

82 To simulate the solid-liquid separation of the treated effluent of the biosorption contactor, a 83 bench scale dissolved air floatation (DAF) system was utilized. Tap water was pressurized to 60 psi 84 in a vessel which was vigorously shaken. 150 ml of pressurized tap water was added to 850 ml of 85 biosorption effluent in a one-liter graduated cylinder in which a float formed separately from the 86 subnatant. After a 3-minute contact time, the subnatant was extracted. DAF processes use pressurized 87 DAF subnatant instead of tap water, thus the DAF separation was conducted a second time using 88 pressurized extracted subnatant instead of tap water. This secondary subnatant was used for analysis 89 of biosorption.

WW, WAS, and DAF effluent were analyzed to determine the effects of biosorption. Total chemical oxygen demand (COD), soluble COD (sCOD), filtered and flocculated COD (ffCOD), colloidal COD (cCOD), total suspended solids (TSS), and volatile suspended solids (VSS) were measured based on Standard Methods [5] and according to previous studies [6]. Normalized removals of the various types of COD were determined by dividing the difference between WW COD and DAF effluent COD by the mass of VSS from the WAS added to the biosorption contactor such 98 removed per g VSS, and mg ffCOD removed per g VSS, respectively.

99 2.2 Procedure for Extracellular Polymeric Substances Extraction and Characterization

100 The WAS used for biosorption was analyzed in parallel for its EPS properties so that the 101 removals observed could be compared to the EPS in the WAS. In order to extract EPS from WAS, 102 three extraction methods were employed. The primary extraction method used was the cation 103 exchange resin (CER) method [7] which was run for 24 hours. CER was predominantly used due to 104 its ubiquity in literature and its relatively higher yields. Since the CER method targets calcium and 105 magnesium bound EPS, it does not extract all of the EPS bound in the cell matrix. The base extraction 106 method and sulfide extraction method described by Park et. al. [8] were also utilized to gain further 107 insight into WAS EPS as they target other cations within the sludge flocs. CER, base, and sulfide 108 extractions were conducted in parallel and in series for analysis. Sludge pellets were rinsed and 109 resuspended to original volumes with deionized water between extractions when conducted in 110 series.

111 The extracted EPS were characterized and measured using colorimetric methods according to 112 wastewater literature [9] and specific EPS literature [10] for concentrations of proteins [11], 113 carbohydrates [12], humic acids [7], and uronic acids [13].

114In order to produce EPS values that could be compared, measured weights of centrifuged WAS115pellets from known volumes of WAS were added to the set volume of extraction solutions such that116EPS could be expressed in mg EPS per L of WAS and even further as mg EPS per g WAS VSS. These

117 normalized EPS values were used to compare with the normalized biosorption values in order to 118 determine the relationship between EPS and soluble organic removal.

119 3. Results and Discussion

120 3.1. Extracellular Polymeric Substances Extraction and Characterization

121 The three WWTPs studied in this research use different types of biological processes for 122 treatment and thus generate different types of sludge. The EPS compositions of the WAS reflected 123 that. Table 1 shows the chemical composition of the WAS from the various plants using the CER 124 extraction method. 25-30% of the organic fraction of the WAS was extracted as EPS from the TF/SC 125 plant, 28-35% for the CAS plant, and 14-15% for the MBR. The MBR had a significantly smaller EPS 126 fraction likely due to the nature of the sludge's growth conditions. TF/SC systems require the 127 sludge to develop as a biofilm, and CAS systems require sludge that has good settling properties. 128 Both biofilm development and settleability are processes that are heavily associated with EPS 129 composition. In an MBR, there is no biological need for these properties and is likely why a lower 130 EPS fraction is observed. The largest fraction of the TF/SC sludge was humic acids which could 131 likely be characterized by the fact that the wasted sludge produced by biofilm systems are humus 132 sludges. Proteins were the major fraction of the other two sludges though humic acids also had a 133 strong presence. A large humic portion is not atypical. Studies done by Wilen et. al. [14] and 134 Frolund et. al. [15] characterize large portions of the EPS as humic acids—both characterizing it as 135 either the second largest, or sometimes the largest fraction, of EPS. It is important to recognize that 136 when analyzing the properties of EPS, it is only in regard to the fraction that was extracted. It was 137 likely that only a fraction of the total amount of EPS was extracted due to the nature of the CER 138 extraction, thus other extraction methods in parallel and in series were looked into.

			mg/gVSS			
WWTP	Process	EPS/VSS (%)	Carbohydrate	Protein	Humic Acid	Uronic Acid
Honouliuli	TF/SC	28±1.7	32±2.0	89±11.2	151±11.7	4±0.7
Waimanalo	CAS	30±2.9	46±5.6	124±12.3	128±27.2	5±0.8
Wahiawa	MBR	14±0.4	27±1.5	57±1.3	53±2.0	6±0.2

140 **Table 1.** EPS composition of WAS using CER extraction (Honouliuli; n=6, Waimanalo; n=5, Wahiawa; n=5)

141 As standalone extraction methods, CER yielded the most amount of EPS, then the base,

142 followed by the sulfide method. These findings align with the literature [8]. When comparing

143 methods in series, CER values in the base-to-sulfide-to-CER and sulfide-to-base-to-CER extractions

144 were significantly lower than standalone CER values whereas base and sulfide values were

145 comparable to their standalone values when CER was conducted first. Two series, CER-to-base-to-

sulfide and CER-to-sulfide-to-base were tested in duplicate to validate which series of methods

147 would yield the most EPS. Figure 1 shows the results of maximizing the amount of EPS extracted

148 from WAS using these methods. The CER extraction needed to be conducted first followed by the

base method then the sulfide method. About 20% more EPS was extracted from the CER-to-base-to-

150 sulfide series compared to the other one. It is not clear why this phenomenon was observed. It is a

151 possibility that conducting certain extractions first could alter the structure of the flocs in a way that

152 either increases or decreases sequential yields though more research is needed to verify this.



153

154 Figure 1. Average EPS extracted by respective series (n=2)

155 The TF/SC and MBR plants were further analyzed using the CER-to-base-to-sulfide series 156 extraction in order to analyze changes in EPS yields. The sum of the three extraction methods was defined as the total amount of extracted EPS. Table 2 shows that there was a significant increase in 157 158 EPS extracted beyond what the CER alone could extract though the CER was the major fraction of 159 the total amount. The fraction of organics as EPS nearly doubled for each plant. Figure 2 shows the 160 chemical compositions breakdown of the extracted EPS from each extraction method at each plant 161 and also compares it to the breakdown of the total EPS extracted. Minimal compositional differences 162 between the CER and total EPS distributions were observed when comparing them though this is 163 likely attributed to the CER constituting more than half of the total EPS extracted. Park et. al. 164 suggested that lectin-like proteins linked to polysaccharides were bridged by calcium and 165 magnesium, targeted by the CER, organic materials were linked with aluminum, which can be solubilized at high pH with the base extraction, and biopolymers were bound to iron, which could 166 167 be removed from the flocs by formation of FeS in the sulfide extraction [8]. The differing composition

- 168 of the sludges using the same extraction method implies that the materials bound in the specific ways
- 169 stated are not compositionally similar and are likely unique to the sludge.







172 Figure 2. Chemical composition of each extraction at the two studied plants expressed as percent of173 respective extracted amount (n=5 for all tests)

174 3.2 Biosorption Correlation

175 Using the extraction data from the previous section with the results of the biosorption testing, 176 relationships between the ability of the WAS to conduct biosorption and its respective EPS 177 characteristics could be observed. For the first comparison, normalized soluble and truly soluble 178 organic removals were compared with the normalized amount of total EPS generated via the CER 179 method across the three WWTP and can be seen in Figure 3 and Figure 4, respectively. In terms of 180 both removals and EPS extracted, Wahiawa, the MBR, exhibited a very tight spread of data. 181 Individually, each plant exhibited a loose trend that more EPS suggests more removal. When looked 182 at as one data set rather than analyzing them by plants a stronger relationship could be developed 183 for the trend suggesting that more EPS available in the organic fraction of the WAS leads to better 184 biosorption properties.





186 Figure 3. Normalized soluble organics removal correlation with the amount of CER extracted EPS present187 in the organic fraction of WAS from the three studied WWTP



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190 present in the organic fraction of WAS from the three studied WWTP

Regardless of the treatment process the WAS originated from, the EPS fraction within them all exhibited similar removal capabilities. Table 3 shows quantified values of removal in relation to the amount of EPS present in the organics of the WAS. Since there was a large difference in total organics in the different WAS samples (7,500-8,000 mg/l VSS for the MBR and ~3,000 mg/l for the other two plants) having similar removals when normalized in this way highly suggests that the EPS is the fraction of the organics responsible for biosorption and that the total amount of organics present is not a good indicator of the biosorption capability of the sludge.

Table 3. Amount of soluble and truly soluble organic matter removed during biosorption per amount of EPS
 extracted via the CER method (Honouliuli; n=6, Waimanalo; n=5, Wahiawa; n=5)

	mg Removed/mg CER EP		/mg CER EPS
WWTP	Process	sCOD	ffCOD
Honouliuli	TF/SC	1.41±0.16	1.33±0.15
Waimanalo	CAS	1.50 ± 0.18	1.14 ± 0.21
Wahiawa	MBR	1.39 ± 0.08	1.11 ± 0.08
Total		1.43 ± 0.15	1.20 ± 0.18

Further analysis to see if any of the individual chemical constituents of the total EPS played more of a role in the biosorption process was also conducted. Figure 5 and 6 shows the individual chemical constituents of the EPS as well as the total EPS in relation to the normalized removals. Minimal correlation between uronic acids and removal was observed though the other analyzed fractions had stronger, yet still varying, relations. It is likely that the chemical composition of the EPS is not the driving factor behind the ability of a sludge to conduct biosorption.



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Figure 6. Normalized truly soluble organics removal correlation with the chemical constituent fractions of
 CER extracted EPS present in the organic fraction of WAS from the three studied WWTP

212 There was the potential that the trend of larger EPS fractions indicating higher removals only 213 existed when looking at the CER extracted EPS thus, the series extractions that were conducted on 214 the two plants were compared to their respective biosorption removals in order to see if the trends 215 persisted with more extracted EPS available and with respect to the other extraction methods. Figure 216 7 and Figure 8 show these comparisons. The series extractions allowed for higher identification 217 percentages of organics as EPS and was closer to the true fraction of EPS in the WAS than the 218 standalone CER extraction. As seen in the figures, with higher amounts of extracted EPS, the trend 219 persisted nearly equally or better than the CER relationship and all of the other extractions. This data 220 suggests that EPS from one type of extraction is not necessarily more related to biosorption properties 221 than the others but rather that all EPS from any of the studied sources had a strong correlation with 222 biosorption capabilities. Moreover, having a larger fraction of the organics identified as EPS and 223 exhibiting good correlation with biosorption properties further emphasizes the trend that biosorption 224 is related to the amount of organics as EPS regardless of its origin within the cell matrix or chemical 225 composition.



Figure 7. Normalized soluble organics removal correlation with extracted EPS from the three extractionsas well as the total extracted amount from the two WWTP studied in series



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231 extractions as well as the total extracted amount from the two WWTP studied in series

4. Conclusions

The studied WAS exhibited varying chemical compositions and expressed differing fractions of their organic material as EPS. Significantly larger fractions of the EPS could be extracted from the WAS by using the CER method, the base method, and the sulfide method in series. The differing chemical compositions of the EPS extracted by the same extraction technique at different plants suggest that EPS bound in the matrix in similar ways are not necessarily compositionally similar.

242 A positive correlation was found between the amount of EPS present in the organic fraction of 243 the sludges and the normalized removal of soluble organic matter. Furthermore, the EPS, regardless 244 of its chemical composition, remove nearly the same amount of soluble and truly soluble organics 245 removing 1.43±0.15 (n=16) mg of soluble chemical oxygen demand, and 1.20±0.18 (n=16) mg of truly 246 soluble chemical oxygen demand, per mg of cation exchange resin extracted extracellular polymeric 247 substances. When analyzing the series extractions with biosorption trend was shown to persist 248 beyond just the CER extracted fraction suggesting that EPS, regardless of how it was bound within 249 the cell matrix, influences the biosorption capabilities.

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