

研 究 主 論 文 抄 録

論文題目

Synthesis of Fe and Sn, In system nanoparticles by pulsed plasma in liquid method
(液体中パルスプラズマ法による Fe と Sn、In 系ナノ粒子の合成)

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主論文要旨

The purpose of this study in this dissertation was to synthesize Fe, magnetite (Fe_3O_4), $\text{Au}@\text{Fe}_3\text{O}_4$, FePt, In, Sn and oxide (In_2O_3 , SnO_2) nanoparticles using a pulsed plasma in liquid method and characterize their structure, morphology and properties. The pulsed plasma in liquid, in which nanoparticles are formed from repetitive pulsed plasma, has the potential to meet the increasing demand for the direct preparation of well-dispersed nanostructures and offers a simple, non-toxic and low-cost alternative to conventional techniques. Brief conclusions of each chapter are described below.

In the **Chapter 1**, the general introduction to Nanotechnology, importance and main properties of magnetic Fe-system nanoparticles, Sn- and In- system nanoparticles and their applications, pulsed plasma in liquid method, purpose of the work and summary of the dissertation were given.

In the **Chapter 2**, pure α -Fe nanoparticles with ≤ 10 nm in size were synthesized using pulsed plasma in liquid method. This was the first time that pure metallic nanoparticles were prepared by arc discharge method using water-toluene interface as a medium. Several experiments made evident, that toluene-water ratio in emulsion influences the purity and size of Fe nanoparticles. The purity of the nanoparticles gradually increased from 48 to 98%, while particle size decreased from 21 to 9.5 nm with smaller toluene volume fraction (from 40 to 5%) in the microemulsions. Finally, toluene:water with 95:5 (%) ratio was found to be the most appropriate medium for pure Fe nanoparticle formation. Lattice parameters of the obtained Fe samples calculated from XRD found to be larger ($a=0.2927$ nm) than those previously reported Fe ($a_{\text{BCC-Fe}}=0.2866$ nm). HRTEM showed spherical-shaped Fe nanoparticles with partial aggregation. Vibrating sample magnetometer indicated superparamagnetic properties of particles with high-saturation magnetization ($M_s=125$ emu g^{-1}) at room temperature.

In the **Chapter 3**, spherical ferromagnetic Fe_3O_4 (magnetite) nanoparticles with an average diameter of 19 nm were synthesized by pulsed plasma in liquid at a voltage of 200 V, a current of 6

A, a frequency of 60 Hz, and a single discharge duration of 10 μ s. Water with different concentrations of 1-hexadecylpyridinium bromide (CPyB) was applied as a liquid, and several experiments made evident that the surfactant concentration affects the phase compositions of the produced materials. The purity of magnetite phase in the sample increased (from 65 to 98%) with increasing CPyB concentration (from 0.10 to 0.84 g) in 200 ml of water. The crystal structure of magnetite nanoparticles with the Fd3m space group and a lattice parameter of $a=0.8393$ nm was evident from X-ray diffraction results.

Ferromagnetic Au-Fe₃O₄ nanoparticles with coercivity (Hc) of 210 Oe were synthesized from Fe electrodes submerged in aqueous solution of HAuCl₄ using pulsed plasma. Cytotoxicity of spherical Au-Fe₃O₄ samples, with average size of 18 nm for Au and 14 nm for Fe₃O₄, was evaluated using a mammalian endothelial cell line, HeLa cells. Both Fe₃O₄ and Au-Fe₃O₄ nanoparticles showed low cytotoxic effects on the cancer cells (cell viability 85-92%), demonstrating their biomedical applicable possibilities.

In the **Chapter 4**, ferromagnetic FePt nanoparticles were successfully synthesized from FePt (50:50) alloy rods using the pulsed plasma in ethanol. A1 phase of as synthesized FePt nanoparticles which has chemically disordered face-centered-cubic structure has been transformed into the ordered face-centered-tetragonal L1₀-FePt after annealing at 400°C for 1h. Along with phase change, the size and magnetic properties of FePt nanoparticles were changed. The average diameter of fcc -FePt was about 3 nm and the mean diameter of fct-L1₀ type FePt nanoparticles was 6 nm. VSM and SQUID analysis indicated the ferromagnetic behavior of A1 and L1₀-type FePt nanoparticles with coercivity of 750 and 1100 Oe, respectively.

In the **Chapter 5**, synthesis and characterization of tin (Sn), indium (In), tin dioxide (SnO₂) and indium oxide (In₂O₃) nanoparticles produced from Sn-Sn and In-In electrode using the pulsed plasma in water were reported. The tetragonal Sn and In nanoparticles showed good crystallinity, with a spherical shape and an average diameter of 3-4 and 4.5-6.0 nm, respectively. Thermogravimetric analysis revealed insignificant weight loss from the particles (2.4%), indicating their thermal stability. DSC measurements revealed a decrease in the melting temperature of the synthesised nanoparticles to 226.5°C (for Sn) and 152.7°C (for In), lower than that of the bulk material by 2.7%. SnO₂ and In₂O₃ nanoparticles were successfully obtained via further annealing (800°C and 400°C) of the as-synthesized Sn and In nanoparticles. Structural and morphological analysis indicated the full oxidation of Sn and In and size growth of oxide nanoparticles up to 15-25 nm. Despite the lower sample concentration of SnO₂ catalyst (0.1 g/L) in dye solution, the full degradation of methylene blue and methyl orange completed in 90 and 180 min, respectively. This shows a good photocatalytic activity of SnO₂ nanoparticles obtained by us.

In the **Chapter 6**, general conclusion, awards, main papers of the presented work and acknowledgements were summarized.