Two New Acylated Flavanone Glycosides from the Leaves and Branches of *Phyllanthus emblica*

Ying-Jun Zhang, Tomomi Abe, Takashi Tanaka, Chong-Ren Yang, and Isao Kouno*, a

^a Faculty of Pharmaceutical Sciences, Nagasaki University; 1–14 Bunkyo-Machi, Nagasaki 852–8521, Japan: and ^b Laboratory of Phytochemistry, Kunming Institute of Botany, Chinese Academy of Sciences; Kunming 650204, P. R. China. Received November 28, 2001; accepted February 18, 2002

Two new acylated flavanone glycosides, (S)-eriodictyol 7-O-(6"-O-trans-p-coumaroyl)- β -D-glucopyranoside (1) and (S)-eriodictyol 7-O-(6"-O-galloyl)- β -D-glucopyranoside (2) were isolated from the leaves and branches of *Phyllanthus emblica* together with a new phenolic glycoside, 2-(2-methylbutyryl)phloroglucinol 1-O-(6"-O- β -D-apiofuranosyl)- β -D-glucopyranoside (3), as well as 22 known compounds. Their structures were determined by spectral and chemical methods.

Key words Phyllanthus emblica; Euphorbiaceae; acylated flavanone glycoside

Phyllanthus emblica L. (Euphorbiaceae) is a shrub or tree native to subtropical and tropical areas of China, India, Indonesia and the Malay Peninsula. The fruit has been widely used for antiinflammatory and antipyretic treatment. The root, leaves and bark are also used for the treatment of indigestion, diarrhea or dysentery, eczema and wart. As a continuation of investigation on the constituents of this plant, leaves and two new acylated flavanone glycosides, a new phenolic glycoside, and 22 known compounds were isolated. This paper describes the isolation and structural elucidation of these compounds.

Results and Discussion

As described in the previous paper,²⁾ the EtOH extract of the fresh leaves and branches of P. emblica was suspended in water and then extracted with Et₂O. The Et₂O layer was partitioned between hexane and MeOH, and the MeOH layer was further chromatographed successively over Sephadex LH-20, silica gel, MCI-gel CHP 20P and Chromatorex ODS to afford 1 and 2, as well as 17 known compounds. The known ones were identified as naringenin,⁶⁾ eriodictyol,⁶⁾ kaempferol,⁷⁾ dihydrokaempferol,^{8,9)} quercetin,⁷⁾ naringenin 7-O-glucoside (prunin), 10) naringenin 7-O-(6"-O-galloyl)-glucoside, 10) naringenin 7-O-(6"-O-trans-p-coumaroyl)-glucoside,¹¹⁾ kaempferol 3-*O*-rhamnoside,⁷⁾ quercetin 3-*O*-rhamnoside,⁷⁾ myricetin 3-*O*-rhamnoside,⁷⁾ 2-(2-methylbutyryl)phloroglucinol 1-O-β-D-glucopyranoside (multifidol glucoside) (**5**), ¹²) (-)-epigallocatechin 3-*O*-gallate, ¹³) 1,2,3,6-tetra- $O^{-,14}$ 1,2,4,6-tetra- $O^{-,15}$ and 1,2,3,4,5-penta-O-galloyl- β -Dglucose, 16) and decarboxyellagic acid 17) by comparison of the physical and spectral data with literature values.

The water layer was separated first by Sephadex LH-20 column chromatography,²⁾ and the obtained fraction 1 was subjected to MCI-gel CHP 20P, Chromatorex ODS, and silica gel to afford **3**, together with 8 known constituents identified as eriodictyol 7-*O*-glucoside (**4**),¹⁸⁾ kaempferol 3-*O*-rhamnoside, quercetin 3-*O*-rhamnoside, quercetin 3-*O*-glucoside,⁷⁾ myricetin 3-*O*-rhamnoside, rutin,¹⁹⁾ 3-*O*-methylellagic acid 4'-O- α -L-rhamnopyranoside²⁰⁾ and tuberonic (12-OH-jasmonic) acid glucoside.²¹⁾ The last compound was first isolated from the potato leaves as a tuber-inducing stimulus²²⁾ and found to be a leaf-closing substance of *Albizzia julib*-

rissin.²¹⁾ This is the first time that tuberonic acid glucoside was obtained from *P. emblica*, which has a nyctinastic movement

Compound 1, a yellow amorphous powder, had a molecular formula C₃₀H₂₈O₁₃ on the basis of its ¹³C-NMR spectral data (Table 1), negative-ion FAB-MS $[m/z 595, (M-H)^{-}]$ and elemental analysis. The ¹H- and ¹³C-NMR spectra of 1 were closely related to those of eriodictyol 7-O-glucoside (4), ¹⁸⁾ except for the appearance of additional signals [δ 7.52, 6.35 (each d, $J=16.0\,\mathrm{Hz}$) and δ 7.49, 6.79 (each 2H, d, J=8.5 Hz)] arising from a trans-p-coumaroyl group. Acidic hydrolysis of 1 in aqueous MeOH yielded eriodictyol, D-glucose and coumaric acid methyl ester, confirming the components of the molecule 1. The location of the p-coumaroyl was determined to be the glucose C-6" position on the basis of the downfield shift of the glucose C-6" (δ 63.3) and H-6" [δ 4.42 (dd, J=2.0, 12.0 Hz) and 4.14 (dd, J=6.5, 12.0 Hz)] by comparison with those of eriodictyol 7-O-glucoside [$\delta_{\rm C}$ 60.8; $\delta_{\rm H}$ 4.03, 3.85]. The linkage of the glucopyranosyl moiety at the 7-hydroxyl group of eriodictyol was confirmed by the HMBC correlation between the glucose H-1" (δ 5.05, d, $J=7.5 \,\mathrm{Hz}$) and C-7 (δ 165.1). The absolute stereochemistry at C-2 was assigned to be S by observation of a positive Cotton effect at 337 nm and a negative one at 294 nm in the cir-

842 Vol. 50, No. 6

Table 1. ¹³C-NMR Data of Compounds **1—5**

	$1^{a)}$	$2^{a)}$	$4^{b)}$		$3^{c)}$	$5^{d)}$
C-2	78.6	78.7	78.9	C-1	161.8	161.7
3	42.2	42.2	44.4		106.9	106.5
4	197.1	197.1	197.4	2 3	165.5	166.6
5	163.1	163.1	163.2	4	95.6	95.7
6	95.7	95.3	95.7	5	167.3	166.4
7	165.1	165.1	165.5	6	98.4	98.6
8	96.3	96.3	96.7	1'	211.8	211.7
9	162.6	162.8	162.9	2'	46.9	46.8
10	103.4	103.4	103.5	3′	28.3	28.2
1'	129.3	129.3	129.4	4′	12.1	12.0
2'	114.3	114.4	114.7	5′	16.8	16.8
3′	145.0	145.3	145.4	glu-1"	101.7	101.5
4'	145.8	145.9	146.0	2"	74.8	74.7
5'	115.4	115.4	115.6	3"	$78.6^{f)}$	78.6
6'	118.1	118.3	118.4	4"	71.3	71.1
glu-1"	99.4	99.4	99.8	5"	$78.0^{f)}$	78.2
2"	73.0	73.1	73.2	6"	68.6	62.4
3"	76.2	76.1	$77.3^{e)}$	api-1‴	110.9	
4"	69.9	69.3	69.7	2‴	77.2	
5"	73.9	73.8	$76.5^{e)}$	3‴	80.5	
6"	63.3	63.0	60.8	4‴	75.0	
Acyl group				5‴	65.7	
1‴	125.1	119.4				
2"',6"'	130.3	108.8				
3‴,5‴	115.9	145.6				
4‴	159.9	138.6				
7‴	145.3	165.9				
8‴	114.0					
9‴	166.5					

a) 125 MHz, in DMSO-d₆. b) 75 MHz, in DMSO-d₆. c) 75 MHz, in CD₃OD. d) 100.6 MHz, in CD₃OD. e, f) Assignments may be interchanged in each column.

cular dichroism (CD) spectrum.^{23,24)} Therefore, the structure of **1** was established as (S)-eriodictyol 7-O-(6"-O-trans-p-coumaroyl)- β -D-glucopyranoside.

Compound 2 was obtained as a yellow amorphous powder and gave dark blue coloration with the ferric chloride reagent. Comparison of the ¹H- and ¹³C-NMR spectral data (Table 1) with those of 1 showed that their structures were very similar except for the presence of a galloyl group in 2 instead of the coumaroyl group in 1. The presence of the galloyl group in 2 could be easily recognized from the characteristic two-proton singlet at δ 6.91 and seven sp^2 carbon signals [δ 145.6 (2C), 138.6, 119.4, 108.8 (2C), 165.9] in the ¹H- and ¹³C-NMR spectra. Acidic hydrolysis of **2** yielded eriodictyol, glucose and gallic acid, confirming that 2 was a galloyl ester of eriodictyol glucoside. The location of the ester was determined to be at glucose C-6" on the basis of the long-range correlations between the two methylene protons of glucose H-6" and the galloyl carboxyl carbon (δ 165.9) signals in the heteronuclear multiple bond connectivity (HMBC) spectrum. In addition, the HMBC correlation of the glucose anomeric proton (δ 5.08, d, J=7.5 Hz) with C-7 (δ 165.1) confirmed that the 6"-O-galloylglucopyranosyl moiety was attached to the 7-hydroxyl group of eriodictyol. The absolute stereochemistry at C-2 was also assigned as S based on a positive Cotton effect at 336 nm and a negative one at 291 nm in the CD spectrum. Thus, the structure of 2 was determined as (S)-eriodictyol 7-O-(6"-O-galloyl)- β -D-glucopyranoside.

Compound **3** had a molecular formula $C_{22}H_{32}O_{13}$ on the basis of the ¹³C-NMR spectral data (Table 1), the negative-ion FAB-MS $[m/z\ 503,\ (M-H)^-]$ and elemental analysis.

The 1 H- and 13 C-NMR spectral data of **3** were similar to those of 2-(2-methylbutyryl)phloroglucinol (multifidol) glucoside (**5**)¹²⁾ except for the appearance of a set of signals arising from a pentose moiety. The 1 H- and 13 C-NMR spectral data (Table 1) due to the sugar moieties revealed the existence of a 6-O- β -apiofuranosyl- β -glucopyranosyl moiety in **3**.²⁵⁾ Furthermore, hydrolysis of **3** yielded 2-(2-methylbutyryl)phloroglucinol (multifidol), 12 D-glucose ([α]_D +24.3°) and D-apiose ([α]_D +6.5°). On the basis of the above results, **3** was characterized as 2-(2-methylbutyryl)phloroglucinol 1-O-(6"-O- β -D-apiofuranosyl)- β -D-glucopyranoside. The absolute configuration of C-2' was assigned as *S* according to the positive Cotton effect at 280 nm in the CD spectrum. 12

Up to now, we have studied the chemical constituents of the roots, fruit juice, and leaves and branches of *P. emblica*, ^{1—5)} and obtained a number of polyphenols and sesquiterpenoids which might be the potent bioactive principles of the plant. Studies of the antioxidative and antiproliferative activities of these components are now in progress.

Experimental

The instruments used for obtaining physical data and experimental conditions for chromatography were the same as we described previously.^{2—5)}

Plant Material The leaves and branches of *Phyllanthus emblica* were collected at Xishuangbanna, Yunnan, the People's Republic of China. A voucher specimen is deposited in the Herbarium of Kunming Institute of Botany, Chinese Academy of Sciences.

Extraction and Isolation As described in the previous paper,²⁾ the EtOH extract of the fresh leaves and branches (15 kg) of *P. emblica* was suspended in water and then partitioned with Et₂O. After being concentrated to dryness, the Et₂O layer (198.0 g) was further partitioned between hexane and MeOH. The MeOH layer (100.34 g) was chromatographed over Sephadex LH-20 (80—100% MeOH, and then 50% acetone) to give four fractions

June 2002 843

(fractions 1—4). Fraction 2 (68.3 g) was subjected to a silica gel (CH₃Cl-MeOH- H_2O , 100:0:0-5:5:1) column and afforded seven fractions (fractions 2-1-2-7), among which fraction 2-5 (5.05g) was identified as quercetin 3-O-rhamnoside. Fractions 2-1—2-4 and 2-6—2-7 were separately chromatographed over Sephadex LH-20 (40-90% MeOH) and MCIgel CHP 20P (0-100% MeOH) to give kaempferol (6.5 mg), dihydrokaempferol (12.9 mg), naringenin (25.2 mg) and eriodictyol (17.5 mg) from fraction 2-1, quercetin (63 mg) from fraction 2-2, naringenin 7-O-(6"-O-trans-p-coumaroyl)-glucoside (167 mg) from fraction 2-3, 1 (160 mg), decarboxyellagic acid (105 mg), 5 (221 mg), naringenin 7-O-glucoside (77 mg), gallic acid (1.37 g) and kaempferol 3-O-rhamnoside (846 mg) from fraction 2-4, 2 (95 mg), naringenin 7-O-(6"-O-galloyl)-glucoside (35 mg), myricetin 3-O-rhamnoside (20 mg), (+)-gallocatechin (7.9 mg), (-)-epicagallocatechin (6.6 mg) and (-)-epigallocatechin 3-O-gallate (17.6 mg) from fraction 2-6, and 1,2,3,6-tetra-O-galloyl- β -D-glucose (176 mg), 1,2,4,6-tetra-O-galloyl- β -D-glucose (75 mg), and 1,2,3,4,6-penta-O-galloyl- β -D-glucose (45 mg) from fraction 2-7, respectively.

The water layer was treated as described²⁾ to give five fractions (fractions 1—5). Fraction 1 was subjected to MCI-gel CHP 20P (0—100% MeOH), Chromatorex ODS (40—100% MeOH) and Si gel (CHCl₃—MeOH–H₂O, 9:1:0.1—6:4:1) to afford 3 (32.7 mg), 3'-O-methylellagic acid 4-O- α -L-rhamnopyranoside (91 mg), 4 (156 mg), kaempferol 3-O-rhamnoside (200 mg), quercetin 3-O-rhamnoside (402 mg), quercetin 3-O-glucoside (176 mg), myricetin 3-O-rhamnoside (17 mg), rutin (12 mg) and tuberonic acid glucoside (556 mg).

(S)-Eriodictyol 7-O-(6"-O-trans-p-Coumaroyl)-β-D-glucopyranoside (1): Yellow amorphous powder, $[\alpha]_D^{28} - 92.8^\circ$ (c=0.36, MeOH). UV λ_{max} (EtOH) nm (log ε): 286 (4.27), 315 (4.48). CD λ_{max} (EtOH) nm (Δ ε): 337 (+4.1), 294 (-29.2). ¹H-NMR (DMSO- d_6): δ: 2.71 (1H, dd, J=3.0, 17.0 Hz, H-3b), 3.21 (1H, dd, J=12.0, 17.0 Hz, H-3a), 3.27 (1H, dd, J=7.5, 8.5 Hz, H-2"), 3.33 (1H, dd, J=8.5, 9.0 Hz, H-3"), 3.36—3.58 (1H, overlapping with solvent, H-4"), 3.75 (1H, ddd, J=2.0, 6.5, 9.0 Hz, H-5"), 4.14 (1H, dd, J=6.5, 12.0 Hz, H-6"b), 4.42 (1H, dd, J=2.0, 12.0 Hz, H-6"a), 5.05 (1H, d, J=7.5 Hz, H-1"), 5.37 (1H, dd, J=3.0, 12.0 Hz, H-2), 6.12 (1H, d, J=2.5 Hz, H-6), 6.18 (1H, d, J=2.5 Hz, H-8), 6.35 (1H, d, J=16.0 Hz, H-8"'), 6.73 (2H, br s, H-5', 6'), 6.79 (2H, d, J=8.5 Hz, H-3"', 5"), 6.87 (1H, d, J=2.0 Hz, H-2'), 7.49 (2H, d, J=8.5 Hz, H-2", 6"'), 7.52 (1H, d, J=16.0 Hz, H-7"'), 12.05 (1H, s, 5-OH). ¹³C-NMR: see Table 1. FAB-MS m/z: 595 (M-H)⁻. Anal. calcd for C₃₀H₂₈O₁₃· 2H₂O: C, 56.96; H, 5.10. Found: C, 57.12; H, 5.12.

Hydrolysis of 1 A solution of **1** (20 mg) in MeOH (1 ml) was treated with 1 m H₂SO₄ (3 ml) at 80 °C for 3 h. The reaction mixture was extracted with Et₂O and the Et₂O layer was applied to Sephadex LH-20 and silica gel to afford *p*-coumaric acid methyl ester (1 mg), 1 H-NMR (CD₃OD): δ: 3.62 (3H, s, OMe), 6.23 (1H, d, J=16.2 Hz, H-8), 6.71 (2H, d, J=8.7 Hz, H-3, 5), 7.36 (2H, d, J=8.7 Hz, H-2, 6), 7.52 (1H, d, J=16.2 Hz, H-7); and eriodictyol¹¹⁾ (6.5 mg), 1 H-NMR (CD₃OD): δ: 2.69 (1H, dd, J=3.0, 17.1 Hz, H-3a), 3.07 (1H, dd, J=12.6, 17.1 Hz, H-3b), 5.28 (1H, dd, J=3.0, 12.6 Hz, H-2), 5.87 (1H, d, J=2.1 Hz, H-6), 5.89 (1H, d, J=2.1 Hz, H-8), 6.78 (2H, br s, H-5′, 6′), 6.91 (1H, br s, H-2′). The water layer was neutralized with Amberlite IRA-400 (OH⁻ form) resin, concentrated to dryness and applied to a silica gel column to afford D-glucose (2.4 mg): [α]_D²⁵ +16.3° (c=0.19, H₂O).

(S)-Eriodictyol 7-O-(6''-O-Galloyl)- β -D-glucopyranoside (2): Yellow amorphous powder, [α]_D \sim 79.4° (c=0.41, MeOH). UV $\lambda_{\rm max}$ (EtOH) nm ($\log \varepsilon$): 281 (4.38). CD $\lambda_{\rm max}$ (EtOH) nm ($\Delta \varepsilon$): 336 (+3.1), 291 (-33.3). ¹H-NMR (DMSO- d_6): δ : 2.74 (1H, dd, J=3.5, 17.0 Hz, H-3b), 3.24 (1H, dd, J=12.5, 17.0 Hz, H-3a), 3.30 (1H, dd, J=7.5, 8.5 Hz, H-2"), 3.35 (1H, t, J=8.5 Hz, H-3"), 3.38—3.48 (1H, overlapping with solvent, H-4"), 3.77 (1H, br s, H-5"), 4.27 (1H, dd, J=4.5, 12.0 Hz, H-6"), 4.41 (1H, br d, J=12.0 Hz, H-6"), 5.08 (1H, d, J=7.5 Hz, H-1"), 5.43 (1H, dd, J=3.5, 12.5 Hz, H-2), 6.14 (1H, d, J=2.5 Hz, H-6), 6.15 (1H, d, J=2.5 Hz, H-8), 6.75 (2H, br s, H-5', 6'), 6.88 (1H, br s, H-2'), 6.91 (2H, s, H-2"), 6"), 12.00 (1H, s, 5-OH). ¹³C-NMR: see Table 1. FAB-MS m/z: 601 (M-H) $^-$; Anal. calcd for $C_{28}H_{26}O_{15} \cdot 5/2H_{2}O$: C, 51.94; H, 4.83. Found: C, 51.64; H, 4.70.

Hydrolysis of 2 A solution of **2** (1 mg) in 1 M $_{2}$ SO₄ (1 ml) was heated at 80 °C for 3 h, and the reaction mixture was treated in the manner described for hydrolysis of **1**. TLC analysis indicated the presence of p-glucose [CHCl₃–MeOH–H₂O, 7:3:0.5 (*Rf* 0.1) and benzene–ethyl formate–formic acid, 1:7:1 (*Rf* 0.1)] and gallic acid [CHCl₃–MeOH–H₂O, 7:3:0.5 (*Rf* 0.25) and benzene–ethyl formate–formic acid, 1:7:1 (*Rf* 0.8)] in the water layer, and eriodictyol [CHCl₃–MeOH–H₂O, 8:2:0.2 (*Rf* 0.7)] in the Et₂O layer by comparison with the authentic samples.

2-(2-Methylbutyryl)phloroglucinol 1-O-(6"-O- β -D-Apiofuranosyl)- β -D-glucopyranoside (3): Yellow amorphous powder, $[\alpha]_1^{18}$ -60.3° (c=0.30,

MeOH). UV λ_{max} (MeOH) nm (log ε): 285 (4.02), 320 (3.55) (sh). CD λ_{max} (EtOH) nm (Δε): 279 (+1.27), 307 (-2.58). ¹H-NMR (CD₃OD): δ: 0.80 (1H, t, J=7.2 Hz, H-4′), 1.03 (1H, d, J=6.9 Hz, H-5′), 1.28 (1H, m, H-3′a), 1.72 (1H, m, H-3′b), 3.23—3.44 (3H, m, H-2″, 3″, 4″), 3.49 (2H, s, H-5″), 3.51 (1H, m, H-5″), 3.54 (1H, dd, J=3.3, 9.6 Hz, H-6″a), 3.66 (1H, d, J=9.9 Hz, H-4″a), 3.79 (1H, m, H-2′), 3.82 (1H, d, J=2.4 Hz, H-2″), 3.89 (1H, d, J=9.9 Hz, H-4″b), 3.94 (1H, dd, J=3.9, 9.6 Hz, H-6″b), 4.88 (1H, d, J=2.4 Hz, H-1″), 5.91 (1H, d, J=7.5 Hz, H-1″), 5.87 (1H, d, J=2.4 Hz, H-4, 6.08 (1H, d, J=2.4 Hz, H-6). ¹³C-NMR: see Table 1. FAB-MS m/z: 1007 (2M-H)⁻, 503 (M-H)⁻, 209 (M-glucose-apiose)⁻. Anal. Calcd for C₂₂H₃₂O₁₃: C, 52.38; H, 6.39. Found: C, 52.21; H, 6.37.

Acid Hydrolysis of 3 A solution of 3 (5 mg) in $1 \text{ M } \text{H}_2\text{SO}_4$ (2 ml) was heated at 80 °C for 3 h. The reaction mixture was extracted with Et₂O three times and the combined organic phase was evaporated to dryness under reduced pressure to furnish 2-(2-methylbutyryl)phloroglucinol ¹²⁾ (2.4 mg). ¹H-NMR (CD₃OD): δ : 0.90 (1H, t, J=7.2 Hz, H-4'), 1.11 (1H, d, J=6.6 Hz, H-5'), 1.38 (1H, m, H-3'a), 1.79 (1H, m, H-3'b), 3.85 (1H, m, H-2'), 5.80 (2H, s, H-4, 6). The water layer was neutralized with Amberlite IRA-400 (OHform) resin, concentrated to dryness and subjected to silica gel chromatography [CHCl₃–MeOH–H₂O (6:4:1)] to afford D-glucose (1.5 mg): $[\alpha]_D^{25}$ +24.3° (c=0.12, H₂O) and D-apiose (1.2 mg): $[\alpha]_D^{25}$ +6.5° (c=0.09, H₂O).

Acknowledgements The authors are grateful to Mr. K. Inada and Mr. N. Yamaguchi (Nagasaki University) for NMR and MS measurements. One of the authors (Y. J. Zhang) is grateful to the Japanese Government for the scholarship awarded her.

References and Notes

- Zhang Y.-J., Tanaka T., Yang C.-R., Kouno I., Chem. Pharm. Bull., 49, 537—540 (2001).
- Zhang Y.-J., Abe T., Tanaka T., Yang C.-R., Kouno I., J. Nat. Prod., 64, 1527—1532 (2001).
- 3) Zhang Y.-J., Tanaka T., Iwamoto Y., Yang C.-R., Kouno I., *Tetrahedron Lett.*, **41**, 1781—1784 (2000).
- Zhang Y.-J., Tanaka T., Iwamoto Y., Yang C.-R., Kouno I., J. Nat. Prod., 63, 1507—1510 (2000)
- Zhang Y.-J., Tanaka T., Iwamoto Y., Yang C.-R., Kouno I., J. Nat. Prod., 64, 870—873 (2001).
- Wagner H, Chari V. M., Sonnenbichler J., *Tetrahedron lett.*, 21, 1799— 1802 (1976).
- Markham K. R., Ternal B., Stanley R., Geiger H., Mabry T. J., *Tetrahedron*, 34, 1389—1397 (1978).
- Grande M., Piera F., Cuena A., Torres P., Bellido I. S., *Planta Med.*, 51, 414—419 (1985).
- Manez S., Paya M., Terencio C., Villar A., *Planta Med.*, 54, 187—188 (1988).
- Rahman W., Ishratullah K., Wagner H., Seligmann O., Chari V. M., Osterdahl B.-G., *Phytochemistry*, 17, 1064—1065 (1978).
- El Sissi H. I., Saleh N. A. M., El Negoumy S. I., Wagner H., Iyengar M. A., Seligmann O., *Phytochemistry*, 13, 2843—2844 (1974).
- Kosasi S., Sluis W. G. V. D., Labadie R. P., *Phytochemistry*, 28, 2439—2441 (1989).
- Coxon D. T., Holmes A., Ollis W. D., Vora V. C., *Tetrahedron*, 28, 2819—2826 (1972).
- Nishizawa M., Yamagishi Y., Nonaka G., Nishioka I., J. Chem. Soc., Perkin Trans. 1, 1983, 961—965 (1983).
- Haddock E. A., Gupta P. K., Al-Shafi S. M. K., Haslam E., J. Chem. Soc., Perkin Trans. 1, 1982, 2515—2524 (1982).
- Nishizawa M., Yamagishi Y., Nonaka G., Ageta M., J. Chem. Soc., Perkin Trans. 1, 1982, 2963—2968 (1982).
- 17) Lin T.-C., Chemical Studies on Tannins and Related Compounds from Combretaceae. Ph.D. Thesis, Kyushu University, Fukuoka, Japan, 1990, pp. 20—21.
- Moretti C., Sauvain M., Lavaud, C., Massiot G., Bravo J.-A., Munoz V., J. Nat. Prod., 61, 1390—1393 (1998).
- 19) Wenkert E., Gottlieb H. E., *Phytochemistry*, **16**, 1811—1816 (1977).
- 20) Yazaki Y., Hillis W. E., *Phytochemistry*, **15**, 1180—1182 (1976).
- Ueda M., Okazaki M., Ueda K., Yamamura S., Tetrahedron, 56, 8101—8105 (2000).
- Yoshihara T., Omer E.-S. A., Koshino H., Sakamura S., Kikuta Y., Koda Y., Agric. Biol. Chem., 53, 2835—2837 (1989).
- 23) Gaffield W., Tetrahedron, 26, 4093-4108 (1970).
- 24) Bohm B. A., "The Flavonoids," ed. by Harborne J. B., Mabry T. J., Mabry H., Academic, New York, 1975, part I, pp. 594—595.
- Jiang Z.-H., Tanaka T., Kouno I., Chem. Pharm. Bull., 47, 421—422 (1999).