Knowledge-based potentials in protein fold recognition

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ABSTRACT

An accurate potential function is essential for protein folding problem and structure prediction. Two different types of potential energy functions are currently in use. The first type is based on the law of physics and second type is referred to as statistical potentials or knowledge based potentials. In the latter type, the energy function is extracted from statistical analysis of experimental data of known protein structures. By increasing the amount of three dimensional protein structures, this approach is growing rapidly. There are various forms of knowledge based potentials depending on how statistics are calculated and how proteins are modeled. In this review, we explain how the knowledge based potentials are extracted by using known protein structures and briefly compare many of the potentials in theory.

Keywords: Knowledge Based Potentials; Reference State; Accessible Surface; Protein Structure; Decoy Structure.

INTRODUCTION

Proteins are macromolecules that are formed by amino acids and linked together peptide bonds. biological with These macromolecules perform a wide variety of functions in organisms such as catalysis reactions and transporting. Also, almost all diseases can be related to the function or malfunction of proteins. The function of proteins is a consequence of their unique three dimensional structures and through their binding to other molecules such as DNA, RNA or proteins.

In 1973, Anfinsen [1] showed that the structure of protein is dictated by its amino acid sequence. In fact, he showed that an unfolded protein could refold to its biologically active conformation. Therefore, the main problem in protein research is modeling and predicting the relationship between sequence and structure. Anfinsen's results led to the thermodynamic hypothesis of folding, which demonstrates that the native structure of proteins fold in the lowest potential energy function. Therefore, based on this hypothesis all studies of proteins including structure prediction, folding simulation and protein design depend on an accurate energy function.

Two different types of potential energy functions are currently in use. The first type is

based on the law of physics. In physical energy function, a molecular mechanics force field is used. Molecular mechanics force fields such as AMBER [2-6], CHARMM [7-8], GROMOS [9], ECEPP [10-12] and OPLS [13-141 are parameterized from ab-initio calculation and small molecular structural data. They are essentially summation of pair electrostatic and Vander wise Waals interaction energies, bonds, angles and dihedral angles terms. In addition, terms such as entropy and solvent effect are implicitly included [15-16]. These potential functions are very time consuming, so these functions have been out of favor in protein structure prediction [17].

To reduce computational complexity, second type of potential energy function is used. These types are referred to as statistical potentials, knowledge based potentials, scoring functions or empirical potentials [17]. In this type, the energy function is extracted from statistical analysis of experimental data of known protein structures [18-25]. In the last decades, this approach was rapidly growing as a consequence of the increasing amount of the experimentally determined three dimensional protein structures. There are various forms of statistical energy functions depending on how statistics are calculated and how proteins are

modeled [26], e.g. distance independent contact energies[27-35], solvent accessible surface potential[22,25,31], packing density potentials, distance dependent potentials[20,27,36-50] angular and dependence[31, 51-54]. Recently, combination of these statistics such as distance and orientation are widely used [55-64]. Initially, statistical potentials were based on statistical mechanics and Boltzmann law [27,55,65], but recently employ many other ideas, such as conditional probabilities [39], linear and quadratic programming on various decoy sets [66-68] and information theory [69-72]. The dependence of statistical potentials on structural data base is also studied [73-74].

Most often, statistical potentials use the Boltzmann law to convert the observed frequencies into potentials. These potentials are obtained as the ratio of observed and expected frequencies, where the observed frequencies are the number of occurrence in known protein structures and expected frequencies are the number of occurrence in the absence of any interaction which serve as reference states. Therefore, depending on this consideration, there are various forms of statistical potential functions.

In this review, we explain how the statistical potentials are extracted by using known protein structures and briefly compare many of the potentials in theory.

Derivation of knowledge based potentials

For derivation of knowledge based potentials, at first, the structural representative of protein reduced to the coordinates of C_{α} , C_{β} or side chain centers or all atoms. Once the amino acid sequence and reduced structure are given, the protein descriptors are extracted. A descriptor can be, e.g. the distance between pair of atoms, solvent accessible surface area, backbone or side chain dihedral angle, packing density or any other features of protein. Therefore, a protein structure can be represented by a vector $c = (c_1, c_2, ..., c_d)$ where each c_i is a descriptor.

Knowledge based potentials are a simple consequence of the Boltzmann distribution. According to the Boltzmann law, the distribution of protein molecules among the microscopic states at the equilibrium state is related to potential function that means for a microstate C (descriptor) the probability of occupancy P(c) connects to potential function E(c) as follows:

$$P(c) = \frac{\exp(-\frac{E(c)}{RT})}{Z} \quad (1)$$

Where R is Boltzmann constant and T is the absolute temperature measured in Kelvin and Z is the partition function:

$$Z = \sum_{c} \exp(\frac{-E(c)}{RT})$$

From equation (1):

$$E(c) = -RT\ln(c) - RT\ln Z$$

The efficient knowledge based potential must consider the sequence-structure relation of protein, so the energetic interactions that are independent of the protein sequence and protein structure must be removed. This energetic contribution is referred to as *reference state*. That means

$$\Delta E(c) = E(c) - E'(c)$$

So, the efficient potential energy is

$$\Delta E(c) = -RT \ln \left(\frac{P(c)}{P'(c)}\right) - RT \ln \left(\frac{Z}{Z'}\right)$$

where, P'(c) is the probability of the descriptor *c* in the reference state. *Z* and *Z*' are both constant and usually assumed that Z=Z'. So

$$\Delta E(c) = -RT \ln(\frac{P(c)}{P'(c)})$$

By assuming that the probability distribution to each descriptor is independent, we have:

$$\frac{P(c)}{P'(c)} = \prod_{i} \frac{P(c_i)}{P'(c_i)}$$

Therefore:

$$\Delta E(c) = -RT \sum_{i} \ln(\frac{P(c_i)}{P'(c_i)})$$

where, $P(c_i)$ and $P'(c_i)$ are the probability of the i^{th} descriptor in native proteins and the reference state, respectively. $P(c_i)$ can be estimated by counting frequency of i^{th} descriptor in data base of native protein structures and $P'(c_i)$ is the probability of the i^{th} descriptor in reference state. Therefore, the choice of reference state is critical and effective for knowledge based potentials. The portion of i^{th} descriptor energy ΔE_i is

$$\Delta E_i = -\mathrm{RT} \ln \frac{P(c_i)}{P'(c_i)} \quad (2)$$

Distance dependent potentials

Descriptor for distance dependent potentials is distance of interactions such as distance between pair of atoms or residues. The distance of interactions is usually divided into a number of small intervals. We use (i,j,d) to represent the k^{th} descriptor c_k . The effective energy function is equal to

$$-RT \ln \frac{N_{obs}(i, j, d)}{N_{exp}(i, j, d)}$$

where, $N_{obs}(i, j, d)$ represent the observed number of (i, j, d) interacting pairs which can be estimated from database of known protein structures and $N_{exp}(i, j, d)$ represents the expected number of (i, j, d) interacting pairs in the reference state which typically results from calculations or simulations.

The major difference of these types of potentials is in the atom type definition and derivation of the reference state. Sippl [20] first proposed a model of reference state which is known as the "uniform density" [75]. He assumed that each pair of contacting atom types in reference state is uniformly distributed along the distance between them. Based on this assumption Samudrala and Moult [39] calculate the $N_{exp}(i, j, d)$ as

$$\frac{N_{obs}(i,j)}{N_{total}}N_{obs}(d)$$

where, $N_{obs}(d)$ is the number of occurrences of interacting of any atom type at distance dand N_{total} is the total number of interacting pairs observations.

Lu and Skolnick [40] employed a quasichemical approximation and estimated $N_{exp}(i, j, d)$ as

$$N_{exp}(i, j, d) = \chi_i \chi_j N_{obs}(d)$$

where χ_k is the mole fraction of atom type *k*. The higher population of hydrophobic residues than that of hydrophilic residues at the core of proteins led to unphysical long range repulsion between hydrophobic residues in statistical potential based on Sippl's Assumption [76]. Zhou and Zhou [46] proposed a reference state

$$N_{exp}(i,j,d) = N_{obs}(i,j,d_{cut}) \frac{V_d}{d_{cut}}$$

by assuming that atom types can be modeled as ideal gas molecules (called DFIRE for *distance scaled finite ideal gas reference state*). In fact in this model, it is assumed that the distribution of interaction pairs follows the uniform distribution in whole volume of the protein [75]. The DFIRE reference state is as where $d_{cut} = 14.5$ and V_d is the volume of a spherical shell of width Δd at distance d from the center.

In 2006, Shen and Sali used no interacting atoms in a homogeneous sphere (called DOPE for *Discrete Optimized Protein Energy*) as reference state [48]. DOPE and DFIRE were derived from a non-interacting ideal gas reference state with the difference that in DOPE the size effect of proteins takes in to account.

Side chain packing is very important determinant for protein structure [77]. Described knowledge based potentials are limited in their ability to describe side chain packing. Recently, the orientation dependent and all atom potential (called OPUS-PSP) have been proposed. In this potential function a set of 19 rigid body blocks were extracted from the chemical structures of 20 amino acids to capture the feature of side chain packing. [63]

Protein is a continuous sequential chain of amino acids and reference state should correctly reflect and counteract the chain connectivity effect. The ideal gas reference state cannot be able to capture this feature [78]. Recently, Cheng etal, proposed a more physical reference state that is based on free rotating and self avoiding chain model [79].

In 2007, Rykunov and Fiser proposed a reference state which atoms assumed to be random [80]. Consequently, a good model to approximate such model would be a system with randomized particles.

Zhang and Zhang, proposed a random walk ideal chain as the reference state [78]. This reference state was derived from a linear freely jointed chain model, which can be considered as the segment of an ideal polymer chain performing a random walk in three dimensional spaces.

Some other definitions, such as the use of decoys were also suggested [66-68].

Recently, Mirzaie et al introduced a knowledge based distance dependent force

extracted from knowledge based potentials [50].

Interaction center in statistical potentials

Various representations for interaction centers have been introduced. Two major representations are residue or atom level. In residue level, CA, CB or side chain centroids are used and then extended alphabet based on occurring amino acids extracted [27, 42-43, 45, 81-82]. Side chain to backbone and side chain to side chain residue potential is also introduced [54].

In successful energy function two interaction centers per residue namely CA atoms and the side chain center of mass (CA atom in Glycine) were considered [83]. CB atoms were also used for representation [84].

Another representation is all heavy atoms. In a strict physicochemical point of view, all the atoms with different environments, connectivity and chemical nature would be different. Among all the 20 amino acids the total number of heavy atoms is 167 and the number of nonequivalent heavy atoms is 98[50]. Some models such as DFIRE use 167 residue specific atomic types [47], but to reduce this number and raise observed frequencies, various atom type definitions have been considered [44, 61, 50, 85]. For example, four atom types containing carbon, nitrogen, oxygen and sulfur were considered by Mirzaie, et al [50], a total of 19 different atom types were used by Ferrada and Melo [86], a total of 40 atom types were considered for all heavy atoms in the 20 amino acids by Melo and Feytmans [44]. All pairs of nonhydrogen atoms in each of the 20 amino acids ignoring the N-terminal and C-terminal nitrogen and oxygen atoms containing 158 residue dependent atom types were considered by Shen and Sali [48] and Mingyang, et al [63].

Also, a clustering algorithm was used to group atom types by similarity. In fact, from an initial 167 atom types, 12 different atom types were extracted [32].

Accessible surface potentials

Solvent effects and hydrophobic interactions are known as important characteristic for protein folding [87]. So, calculation of the solvent accessible surface area is very important to estimate solvation energies [88, 89]. Zarei, et al presented a method for prediction of protein surface accessible with information theory [90], and Naderi Manesh, et al presented a method based on residue type and conformational states [91].

The accessible surface of an interaction center is defined as the number of interaction centers within a sphere around it and the radius of the sphere is the distance range of the potential. This type of potentials has been described by Sippl method [22, 25].

Contact potentials

Contact potentials are the simplest description of pair-wise potentials. They are similar to the distance dependent potentials. These types of potentials have two values; below the contact distance threshold, energy is assigned to interaction and above it energy is zero.

In the contact potential, the distance between the centers of interacting pairs are calculated and the observed frequency of contacts between residues converts to free energy using Boltzmann equation. In this way, two problems may be encountered. First, when an atom or center of mass is selected for each residue, calculated potential is independent of orientation of the side chains and when the distance between two atoms of two residues is equal to the distance of two atoms of other residues in other positions, the same potentials are assigned to them although the orientation of two residue side chains may be quite different. Second, the atoms of two residues may not have direct contact with each other and some atoms may be located in an interval close to them [92].

Delaunay tessellation is a geometrical tool that is used in protein structure analysis. In fact, voronoi tessellation partitions the space convex poly-topes called into voronoi polyhedra. For a given protein, the voronoi polyhedra is the region of space around an atom, such that all points of this region are closer to this atom than to any other atoms of the protein. A group of four atoms, whose voronoi polyhedra meet at one vertex, forms another basic topological object called, the Delaunay tessellation simplex. So we can consider two atoms are in contact, if they are two vertices of an edge in a simplex [50].

In 2010, Arab et al [92] developed a new approach to calculate a knowledge-based potential energy using pair-wise residue contact area. They calculate the parts of each pair-wise residue area which are in contact in 2Å by rolling a probe ball of different sizes around the atoms of a residue to determine the contact area of each pair. This pair-wise contact area is used to determine statistical contact area preference between each residue pairs, when a contact area preference estimates a sum of energetic interactions and structural constraints.

Composite potentials

Some potential combine various terms, including hydrogen bonding, torsion angle, solvation, pair-wise potential in residue level. or all atom level and some terms of physical energy function. For example, ROSETTA scoring function [93,94] combines sequence dependent and sequence independent features of protein. A composite residue level potential was introduced that combined contact and local sequence structure descriptions [95]. In 2007, a hydrophobic potential of mean force, generalized Born function and Amber 99 force field were combined [96]. The distribution of pseudo-bonds, pseudo-angle, pseudo-dihedrals and distances between centers of interactions was converted into potentials of mean force in PC2CA model [83]. The QMEAN scoring function is a successful composite potential which consist of 6 different terms; distance dependent pair-wise potentials, solvation potential, torsion angle potential, secondary structure specific and solvent accessibility[97-98].Another composite potential, in residue level [62] and all atom version [63] combines distance dependent pair-wise potential with orientation preference, hydrogen bonding, packing, three body interactions and salvation terms.

Validation of potentials functions

For assessing potential function, it must be able to distinguish protein native structure from protein-like decoys. In fact the energy of native structure must be lowest among energy of decoy structures. Another challenging test of energy function is discrimination of near native structures that means, in absence of the native structure, the energy of structure with minimum RMSD to native structure must be minimum.

In addition, the correlation coefficient between the energy of a model and RMSD should be close to 1, i.e., proteins with high RMSD have high energy.

The *Z*-score of the native structure in the decoys set is equal to

$$Z - score = \frac{\langle E_{decoys} \rangle - E_{native}}{\sigma_{decoys}}$$

In which E_{native} is the energy calculated for native structure and $\langle E_{decoys} \rangle$ and σ_{decoys} are the average and the standard deviation of energy distributions of decoys proteins, respectively. For a good energy function, *Zscore* should be high.

One of the most popular decoy data sets is available in the Decoys'R'us database under the category

'multiple'(<u>http://dd.compbio.washington.edu</u>). This data set contains the ig_structal_hires, 4state_reduced, fisa_casp3, fisa, vhp_mcmd, semfold, hg_structal, lmds, ig_structal and lattice_ssfit. These decoys are obtained with different methods and are very appropriate for assessment of model.

The 4state-reduced decoy set contains 7 different proteins. For each protein, 632 to 689 native like conformations are present in the dataset. This decoy set was generated using a four-state off lattice model with a conformational relaxation method [99].

The fisa and fisa_casp3 decoy sets with four and six targets (500–1400 models per target), respectively, were obtained using a combination of a Bayesian scoring function and a simulated annealing protocol [100,101].

The ig_structal, hg_structal and ig_structal_hires decoy sets contain immunoglobulins (ig) or globins (hg) created by homology modeling.

The largest lattice_ssfit decoy set, containing 2000 decoys for each of the eight targets, was generated using a tetrahedral lattice model with the all-atom ENCAD energy function [99]. The ranges of RMSD from native for all 8 proteins in the set are larger than 4Å.

The lmds set includes decoys with RMSD's less than 10 Å. lmds decoy set with 215–500 models for each one of 10 primarily short targets, was obtained by a local optimization method and a reduced ENCAD energy function [103].

The semfold set includes a very large number of decoys for each of the 6 proteins. In some cases RMSD from native are in range 3 Å to 5 Å. This decoy set was generated by fragment insertion method. This decoy set is the most challenging, since it has more than 10000 decoys for each of the six targets. The vhp_mcmd decoy set has been generated by molecular dynamics simulations.

The ROSETTA decoy set presented by Baker and coworkers [104,105] contains 20

random models and 100 lowest scoring models from 10,000 decoys, which were generated for 58 small proteins using ROSETTA de novo structure predictions followed by all-atom refinement. This data set is downloadable at http://depts.washington.edu/bakerpg.

The I-TASSER decoy set includes the atomic structure decoys generated for 56 non-homologous small proteins. The backbone structures were first generated by the I-TASSER ab initio modeling by Wu et al. [106], where for each protein target 12,500–32,000 conformations were taken from the trajectories of 3 lowest-temperature replicas of the simulations. A full set of I-TASSER decoys is downloadable at http://zhanglab.ccmb.med.umich.edu/decoys.

DISCUSSION

The results of energy functions depend on the presence or absence of the native structure [80]. Often, all atom potential on the presence of native structure have very good performance, but in the absence of native structure, residue based, all atom and composite potentials perform competitively [80]. For example, the composite potential QMEAN6 [97,98] and residue level potential [84] are the best performing potentials, where the native structure were absent while some potentials have better performance in the presence of native structure.

Recently, some different approaches for discrimination of native structure from decovs have been presented. For example, graph theoretic properties including average degree and clustering coefficient of protein graph have been used to perceive the difference between correctly folded proteins and decoys [107,108]. Decoy discrimination method using wavelet analysis [109] and a simple approach based on network pattern of conserved hydrophobic residues have been also presented [110]. An approach by using machine learning and neural network and evolutionary information [111], and a physical method based on force is also presented [50]. The majority of knowledge based potentials are pair-wise, but multi body potentials were also reported [112-117].

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