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Theoretical Studies on Radialenes and Related Molecules

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Received July 10, 1972

For certain classes of molecules it is possible to obtain a general solution of the Hückel problem, *i. e.* to derive expressions for the orbital energy, orbital coefficients, total π -electron energy, etc. in a closed analytical form¹⁻⁵. General solutions are important because a large amount of numerical labour can be saved. Besides, they show the dependence of HMO quantities on the molecular topology, which has been recently investigated by various authors.⁶⁻⁸

We have investigated molecules denoted by Q_n^m (see Fig. 1a). Atoms are numbered by two indexes for simplicity. AO's are labelled by the same two

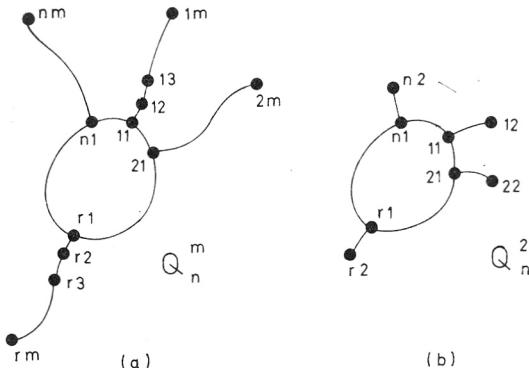


Fig. 1.

indexes. The special case where $m = 2$ (see Fig. 1b) represents $[n]$ radialenes. They have been investigated previously by various authors⁹⁻¹³. Q_n^1 are the $[n]$ annulenes, which are considered in detail elsewhere¹⁴⁻¹⁷.

Although the molecules Q_n^m are not claimed to possess any definite geometry, we can treat them as if they belong to the D_{nh} symmetry group. This is a proper consequence of the fact that the symmetry of the Hückel Hamiltonian is determined by the group of automorphisms of the graph corresponding to the given molecule¹⁸⁻²¹.

Orbital coefficients for Q_n^1 can be easily obtained using symmetry considerations,¹ and are well known^{22,23}. Thus,

$$\Psi_{t1} = (n)^{-1/2} \sum_{r=1}^n \exp \left(i \frac{2\pi t r}{n} \right) \chi_{11} \quad (1)$$

are the HMO's for $[n]$ annulene.

Any symmetry operation of the D_{nh} group transforms the sets of atoms

$$\{1p, 2p, 3p, \dots, np\} \quad p = 1, 2, \dots, m$$

into itself. Therefore they form different symmetry classes and the MO of the whole molecule is of the form²⁴

$$\Psi_t = \sum_{p=1}^m c_{tp} \Psi_{tp} \quad (2)$$

From the topology of the molecule Q_n^m it can be deduced that the action of the Hückel Hamiltonian on atomic orbitals χ_{rp} is given by

$$\begin{aligned} H\chi_{r1} &= \alpha\chi_{r1} + \beta(\chi_{r-1,1} + \chi_{r+1,1} + \chi_{r,2}) \\ H\chi_{rp} &= \alpha\chi_{rp} + \beta(\chi_{r,p-1} + \chi_{r,p+1}) \quad p = 2, 3, \dots, m-1 \\ H\chi_{rm} &= \alpha\chi_{rm} + \beta\chi_{r,m-1} \end{aligned} \quad (3)$$

where by definition,

$$\begin{aligned} \chi_{01} &\equiv \chi_{n1} \\ \chi_{n+1,1} &\equiv \chi_{11} \end{aligned}$$

From eqs. (3) it follows that:

$$\begin{aligned} H\Psi_{t1} &= \left(\alpha + 2\beta \cos \frac{2\pi t}{n} \right) \Psi_{t1} + \beta \Psi_{t2} \\ H\Psi_{tp} &= \alpha\Psi_{tp} + \beta(\Psi_{t,p-1} + \Psi_{t,p+1}) \quad p = 2, 3, \dots, m-1 \\ H\Psi_{tm} &= \alpha\Psi_{tm} + \beta\Psi_{t,m-1} \end{aligned} \quad (4)$$

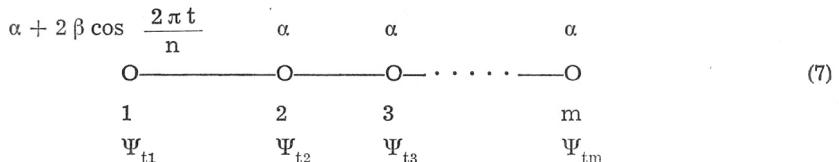
and hence the equation

$$H\Psi_t = E\Psi_t \quad (5)$$

can be written in the form

$$\begin{aligned} c_{t1} \left[\left(\alpha + 2\beta \cos \frac{2\pi t}{n} - E \right) \Psi_{t1} + \beta \Psi_{t2} \right] + \sum_{p=2}^{m-1} c_{tp} \left[(\alpha - E) \Psi_{tp} + \right. \\ \left. \beta (\Psi_{t,p-1} + \Psi_{t,p+1}) \right] + c_{tm} \left[(\alpha - E) \Psi_{tm} + \beta \Psi_{t,m-1} \right] = 0 \end{aligned} \quad (6)$$

Formally, the above relation corresponds to a Hückel problem for a chain of m atoms with a perturbation $\alpha \rightarrow \alpha + 2\beta \cos \frac{2\pi t}{n}$ on the first one.



Here the Ψ_{tp} 's play the role of AO's. The secular equation to be solved is of the form:

$$D_m(w) \equiv \begin{vmatrix} w + 2 \cos \frac{2\pi t}{n} & 1 & 0 & 0 & \dots & 0 \\ 1 & w & 1 & 0 & \dots & 0 \\ 0 & 1 & w & 1 & \dots & 0 \\ \vdots & & & & \ddots & \vdots \\ \vdots & & & & \ddots & \vdots \\ \vdots & & & & \ddots & \vdots \\ 0 & \dots & \dots & \dots & \dots & 1 & w \end{vmatrix} = 0 \quad (8)$$

where

$$w = \frac{\alpha - E}{\beta}$$

A. For $m = 1$ eq. (8) represents $[n]$ annulenes with

$$w = -2 \cos \frac{2\pi t}{n} \quad (9)$$

B. For $m = 2$ it describes $[n]$ radialenes with

$$w = \begin{cases} -\cos \frac{2\pi t}{n} - \sqrt{1 + \cos^2 \frac{2\pi t}{n}} & \text{bonding levels} \\ -\cos \frac{2\pi t}{n} + \sqrt{1 + \cos^2 \frac{2\pi t}{n}} & \text{antibonding levels} \end{cases} \quad (10)$$

Therefore, radialenes cannot possess NBMO's and have always singlet ground states.

The total π -electron energy (in β units) can be approximated by an integral²⁵:

$$\begin{aligned} E_\pi &= 2 \sum_{t=1}^n \left[\cos \frac{2\pi t}{n} + \sqrt{1 + \cos^2 \frac{2\pi t}{n}} \right] = 2 \sum_{t=1}^n \sqrt{1 + \cos^2 \frac{2\pi t}{n}} \\ &\approx 2 \cdot \frac{n}{2\pi} \int_0^{2\pi} \sqrt{1 + \cos^2 t} dt = 2.43201 n \end{aligned} \quad (11)$$

Note that E_π is linearly proportional to the number of carbon atoms. Hence there is no change in stability of the π -electron system when the number of atoms is increased. Radialenes are typical examples of nonaromatic compounds^{26,27}. Steric effects will be of major influence for large n .

Charge densities are given by the formulae:

$$q(\text{inner}) = 1 + \frac{1}{n} \sum_{t=1}^n \frac{\cos \frac{2\pi t}{n}}{\sqrt{1 + \cos^2 \frac{2\pi t}{n}}} \quad (12)$$

$$q(\text{outer}) = 1 - \frac{1}{n} \sum_{t=1}^n \frac{\cos \frac{2\pi t}{n}}{\sqrt{1 + \cos^2 \frac{2\pi t}{n}}}$$

Because

$$\int_0^{2\pi} \frac{\cos t}{\sqrt{1 + \cos^2 t}} dt = 0 \quad (13)$$

the charge density distribution is nearly uniform, deviations being considerable only for small (odd) n.

(4 m + 3) rings show a tendency to become positively charged, while (4 m + + 1) rings have electron affinity. This property seems to have a topological background and it is presently being investigated by us.

C. Generally $D_m(w)$ can be evaluated in the form

$$D_m(w) = P_m(w) + 2 \cos \frac{2\pi t}{n} P_{m-1}(w) \quad (14)$$

where P_m is the characteristic polynomial for the chain of m atoms

$$\begin{aligned} P_0 &= 1 \\ P_1 &= w \\ P_m &= w P_{m-1} - P_{m-2} \end{aligned}$$

Acknowledgement. We would like to thank Dr. D. Cvetković (Beograd) for useful suggestions.

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IZVOD

Teorijski studij radijalena i srodnih molekula

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Prikazano je opće rješenje Hückelovog problema za klasu molekula Q_n^m , kojih su specijalni slučajevi [n] anuleni ($m = 1$) i [n] radijaleni ($m = 2$). Pokazano je također da se opće rješenje može dobiti i promatranjem lanca od m atoma kod kojega se smetnja $(2\beta \cos \frac{2\pi t}{n})$ javlja na prvom atomu.

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Primljeno 10. srpnja 1972.