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# Improved silicon surface passivation of APCVD Al<sub>2</sub>O<sub>3</sub> by rapid thermal annealing

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## Abstract

Short-duration post-deposition thermal treatments at temperatures above those normally used for annealing activation have the potential to further improve the already excellent passivation of crystalline silicon (c-Si) achieved by Al<sub>2</sub>O<sub>3</sub>, but have so far received little attention. In this work we investigate the influence of rapid thermal annealing (RTA) on the surface passivation of c-Si by Al<sub>2</sub>O<sub>3</sub> deposited by atmospheric pressure chemical vapour deposition (APCVD) as a function of RTA peak temperature between 500 and 900°C, and for Al<sub>2</sub>O<sub>3</sub> deposition temperatures between 325 and 440°C. The saturation current density  $J_0$  of undiffused *p*-type surfaces is observed either to increase or decrease following RTA depending on the Al<sub>2</sub>O<sub>3</sub> deposition temperatures provide worse passivation after low temperature heat treatment, but maintain this passivation better at higher RTA temperatures. An exceptionally low  $J_0$  of 7 fA cm<sup>-2</sup>, due to the combination of a very low interface state density  $D_{it}$  and unusually high negative fixed charge density  $Q_{f5}$  is achieved by the use of a short 500– 550°C RTA combined with optimised deposition conditions.

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#### 1. Introduction

Effective surface passivation is essential for the realization of high-efficiency photovoltaic devices based on crystalline silicon. In the last decade  $Al_2O_3$  has emerged to become the gold standard in silicon surface-passivation, providing unprecedented passivation especially of *p*-type surfaces [1]. It is well-established that for  $Al_2O_3$  films deposited at lower temperatures, some post-deposition thermal treatment is generally required to "activate" the surface passivation. This typically takes the form of an extended (10–30 minute) anneal at 400–450°C in N<sub>2</sub> or forming gas. Thermal treatments at higher temperatures have been shown to sometimes degrade, and sometimes improve the surface passivation. Generally, investigations of the effects of higher temperature processing have employed either extended (~10–30 minute) anneals at moderate temperatures (usually <600°C) [2]–[6], intended to activate the passivation, or relatively short (several seconds), high-temperature (~800°C) treatments intended to simulate the contact-"firing" process employed in conventional Si solar cell manufacturing [6]–[11], in which the goal is to demonstrate the stability of the existing passivation.

There appear to be few reported investigations of the effects of very short (~1 s) thermal treatments at moderate temperatures. However, previous work indicates a complex response of  $Al_2O_3$  passivation to thermal processing depending on both temperature and exposure time, as well as  $Al_2O_3$  deposition parameters. In particular, it appears that  $Al_2O_3$  passivation may not always improve or degrade monotonically with increasing annealing time at higher temperatures, but may reach an optimum at some shorter annealing time and subsequently degrade [11]. This suggests that a study of the effects of very short thermal treatments at temperatures above those usually used for "activation" annealing may reveal new pathways for further optimizing  $Al_2O_3$  passivation.

In this work, we make use of rapid thermal annealing (RTA) to investigate the effects of short, well-controlled thermal treatments on  $Al_2O_3$  passivation over a wide temperature range. We investigate these effects for  $Al_2O_3$  films deposited by atmospheric pressure chemical vapour deposition (APCVD), a simple technique well-suited to low-cost, high-throughput manufacturing. Photoconductance carrier lifetime and capacitance–voltage measurements are used to assess the quality of the passivation. We show that such short thermal treatments at moderate temperatures are capable of improving  $Al_2O_3$  passivation beyond the level possible using conventional annealing treatments.

#### 2. Experimental method

Symmetrically passivated Si–Al<sub>2</sub>O<sub>3</sub> lifetime structures were prepared on ~480 µm thick 1  $\Omega$ cm *p*-type (100) floatzone Si wafers. These received a surface damage etch in HF:HNO<sub>3</sub> solution, followed by a standard RCA clean and dilute HF dip immediately prior to Al<sub>2</sub>O<sub>3</sub> deposition. Additional metal–insulator–semiconductor (MIS) structures for capacitance–voltage (C–V) measurements were prepared on single-side-polished 2.8  $\Omega$ cm *p*-type and 1.3  $\Omega$ cm *n*-type (100) float-zone Si wafers. These received only an RCA clean and HF dip prior to deposition. Al<sub>2</sub>O<sub>3</sub> films were deposited by APCVD from triethyldialuminium-tri-(*sec*-butoxide) (TEDA-TSB) and H<sub>2</sub>O (H<sub>2</sub>O:TEDA-TSB molar ratio of ~9:1) at substrate temperatures between 325 and 440°C [12]. The resulting film thickness varied with temperature, but was typically in the range of 10–12 nm. Some samples received a standard activation anneal at 425°C for 30 min in N<sub>2</sub> in a quartz tube furnace following deposition and prior to RTA. Rapid thermal annealing was performed in a UniTemp UTP-1100 oven employing infrared lamps at peak temperatures between 500 and 900°C for 1s in N<sub>2</sub> (40°C/s ramp rate). The sample temperature was monitored during annealing by a thermocouple in contact with the rear side of the sample.

Lifetime measurements were performed immediately following  $Al_2O_3$  deposition and after each subsequent heat treatment. The effective excess carrier lifetime  $\tau_{eff}$  was determined as a function of bulk excess carrier density  $\Delta n$  via inductively coupled photoconductance decay measurements using a Sinton Instruments WCT-120 system. Measurements were performed both with short flash illumination using the transient analysis and with quasi-steady-state illumination using the generalised analysis of [13], with the necessary optical constant determined by comparison with transient measurements for samples where these were possible.

Metal-insulator-semiconductor (MIS) samples for C-V measurements were formed following annealing by thermally evaporating Al through a shadow mask to form circular gate contacts ( $\emptyset \approx 700 \ \mu$ m). GaIn eutectic paste was applied manually to form an ohmic rear contact. High-frequency (1 MHz) and quasi-static C-V measurements were performed using an HP 4284A precision LCR meter and HP 4140B picoammeter/DC voltage source. The

interface state density  $D_{it}$  was assessed from the quasi-static capacitance using the method of Berglund [14], while the fixed charge concentration  $Q_f$  was calculated from the measured flatband voltage shift. For full details of the experimental setup and analysis used, see [15].

Some additional  $Al_2O_3$  passivation layers were prepared by atomic layer deposition (ALD) using a Beneq TFS-200 reactor. Both a thermal process using trimethylaluminium (TMA) and  $H_2O$  at 200°C and a plasma-assisted process with TMA and  $O_2$  plasma at 175°C were used.

### 3. $J_{\theta}$ extraction

Surface passivation was assessed in terms of the surface saturation current density  $J_0$ , which has the advantage of being independent of both  $\Delta n$  and the bulk dopant density  $N_b$ , in contrast to the more commonly used effective surface recombination velocity  $S_{eff}$  [16]. In general,  $J_0$  was calculated directly from the measured  $\tau_{eff}(\Delta n)$  at  $\Delta n = 10^{15}$  cm<sup>-3</sup>, according to

$$J_{0} = q n_{i}^{2} \left( N_{b} + \Delta n \right)^{-1} \frac{W}{2} \left( \frac{1}{\tau_{eff}} - \frac{1}{\tau_{int}} \right), \tag{1}$$

where q is the fundamental electron charge,  $n_i = 9.7 \times 10^9 \text{ cm}^{-3}$  is the intrinsic carrier concentration [17], W is the wafer thickness, and  $\tau_{int}$  is the intrinsic bulk lifetime according to [18]. An alternative expression for  $J_0$  may be derived by taking the derivative of (1) with respect to  $\Delta n$  (assuming  $J_0$  is independent of  $\Delta n$ ):

$$J_0 = q n_i^2 \frac{W}{2} \frac{d}{d\Delta n} \left( \frac{1}{\tau_{eff}} - \frac{1}{\tau_{int}} \right).$$
<sup>(2)</sup>

In general, excellent agreement was found between  $J_0$  calculated from (1) and (2) for the examined samples.

Analysis of the data via (1) or (2) is complicated by the fact that contamination of the wafer bulk was also found to occur during the RTA step at higher temperatures. This is apparent from the fact that the resulting  $\tau_{eff}$  exhibits a pronounced dip in low injection, characteristic of a Shockley–Read–Hall (SRH) recombination process. This reduced lifetime remains after stripping the original film and repassivating the surface, and is also observed for samples given an RTA prior to Al<sub>2</sub>O<sub>3</sub> deposition. The lifetime could be recovered to close to its original value by applying a standard phosphorus gettering procedure [19], strongly suggesting the presence of metallic impurities. The magnitude of the lifetime reduction was found to increase with increasing RTA temperature, and was generally negligible at RTA temperatures of 550°C or below.

Comparison of  $\tau_{eff}(\Delta n)$  before and after the application of strong illumination revealed that bulk iron contamination was definitely present, although neither interstitial iron (Fe<sub>i</sub>) nor iron-boron pairs (FeB) are sufficient to explain all of the observed injection dependence, when using the parameters of [20, 21]. It was however found that the post-illumination lifetime data for all samples could be accurately modelled by assuming surface recombination described by a constant  $J_0$  (i.e. low injection conditions at the silicon surface) and two bulk defects: one with the characteristics of interstitial iron (Fe<sub>i</sub>) [20, 21], and one with a  $\sigma_n/\sigma_p$  ratio of 2.5. The resulting effective lifetime is described by the following equation:

$$\tau_{eff}^{-1} = \frac{J_0}{qn_i^2} \frac{2}{W} \left( N_b + \Delta n \right) + \tau_{int}^{-1} + \tau_{SRH,Fe_i}^{-1} + \tau_{SRH,2}^{-1} , \qquad (3)$$

where  $\tau_{SRH,Fe_i}$  and  $\tau_{SRH,2}$  account for bulk SRH recombination via Fe<sub>i</sub> defects and the second unidentified defect, respectively.

Fig. 1 illustrates the process of  $J_0$  extraction via (3). For the samples measured prior to RTA, and for those measured after RTA at the two lowest peak temperatures (500 and 550°C), excellent agreement between the data and Equation (3) was obtained while neglecting the SRH terms, indicating negligible bulk contamination in these samples. Excellent agreement between  $J_0$  calculated by Equations (1) and (2) was also obtained for these samples. For the other samples,  $J_0$  was calculated by adjusting the concentration of the two bulk defect types in order to give good agreement both between Equation (3) and the measured  $\tau_{eff}$ , and also between  $J_0$  calculated from Equations (1) and (2). Fig. 2 shows the extracted bulk defect concentrations as a function of RTA peak temperature, assuming  $\sigma_p = 10^{-15}$  cm<sup>2</sup> for the second defect. Despite some scatter, there is a clear trend of increasing bulk defect concentration with increasing RTA temperature, which is consistent with the diffusion of metallic contaminants. As this is clearly an extrinsic effect not related to the surface passivation, we do not consider it further here.



Fig. 1. (a)  $\tau_{eff}$  and (b)  $J_0$  as a function of  $\Delta n$  for an Al<sub>2</sub>O<sub>3</sub>-passivated sample ( $T_{dep} = 410^{\circ}$ C, no anneal) before and after RTA at 800°C. The various contributions to  $\tau_{eff}$  in Equation (3) are shown, along with  $J_0$  calculated from both Equation (1) ( $J_{0,1}$ ) and Equation (2) ( $J_{0,2}$ ). Prior to RTA,  $\tau_{eff}$  is well-described by Equation (3) without the SRH terms, and  $J_{0,1}$  and  $J_{0,2}$  agree closely without the need for any corrections. After RTA, significant bulk contamination is present, and  $J_{0,1}$  and  $J_{0,2}$  diverge. The bulk SRH components are determined such that good agreement is achieved between  $J_{0,1}$  and  $J_{0,2}$  calculated from the corrected  $\tau_{eff}$ . Dashed lines show the extracted  $J_0$  values in each case.

#### 4. Results and discussion

As shown in Fig. 3,  $J_0$  may either decrease or increase following RTA relative to the as-deposited value, depending on both the deposition and RTA temperatures, but in general increases with the latter. Clear transition points in post-RTA  $J_0$  are apparent around 600°C and above 800°C, which is consistent with earlier reports, and with the out-diffusion of hydrogen- and oxygen-containing species observed at these temperatures during thermal effusion measurements [22], indicating structural modifications of the bulk film and/or interface.

For films deposited at temperatures of 325 and 355°C,  $J_0$  as-deposited is rather high, but improves substantially upon RTA. An optimum RTA temperature exists within the examined range, and this shifts to lower temperatures as deposition temperature increases. For the films deposited at 385°C,  $J_0$  as-deposited is quite variable, because the deposition temperature is in the transition region where substantial changes in interface quality occur [12].  $J_0$  after RTA is generally substantially reduced, except in the case of RTA temperatures greater than 800°C, and displays much less random variation, increasing monotonically with RTA temperature. As the deposition temperature increases further to 410°C, the as-deposited  $J_0$  stabilises at a significantly lower level, while the RTA temperature above which  $J_0$  increases rather than decreases after RTA is reduced to between 600 and 650°C. The inclusion of a standard anneal prior to RTA initially reduces  $J_0$ , but results in somewhat higher post-RTA  $J_0$ . At a deposition temperature of 440°C,  $J_0$  is no longer significantly reduced by annealing, and is significantly increased by RTA at temperatures of 600°C or above.



Fig. 2. Bulk concentration of Fe<sub>i</sub> and the second, unidentified defect as a function of RTA peak temperature, as determined during  $J_0$  extraction using Equation (3).  $\sigma_p = 10^{-15}$  cm<sup>2</sup> is assumed for the second defect.



Fig. 3.  $J_0$  as a function of RTA peak temperature for Al<sub>2</sub>O<sub>3</sub> films deposited at various temperatures (indicated in each subfigure) on undiffused 1  $\Omega$ cm *p*-type (100) Si surfaces, with or without an intermediate anneal at 425°C in N<sub>2</sub>.  $J_0$  values are shown for the films as-deposited, after annealing (if applicable), and after RTA.

Fig. 4 shows post-RTA  $J_0$  in the form of a contour map, as a function of both Al<sub>2</sub>O<sub>3</sub> deposition temperature and RTA peak temperature. Fig. 4 shows clearly that the optimum Al<sub>2</sub>O<sub>3</sub> deposition temperature (in terms of minimising  $J_0$ ) depends strongly on the post-deposition thermal processing to be applied. For RTA peak temperatures between 500 and 600°C, the optimum deposition temperature occurs at 410°C. As RTA peak temperature increases, the optimum deposition temperature decreases, falling to ~350°C at an RTA peak temperature of 800°C.



Fig. 4. Contour map of post-RTA  $J_0$  as a function of RTA peak temperature and Al<sub>2</sub>O<sub>3</sub> deposition temperature, derived from the data of Fig. 3. The dotted line shows the deposition temperature corresponding to the minimum value of post-RTA  $J_0$  at a given RTA peak temperature

 $Al_2O_3$  films deposited at higher temperatures display excellent passivation following low-temperature RTA heat treatment, but their passivation quality degrades substantially at higher RTA temperatures. Conversely, films deposited at lower temperatures require moderate heat treatment to reach their minimum  $J_0$  values, which are not as low as those of higher temperature films, but maintain their passivation quality over a broader and higher range of RTA temperatures. This difference in thermal stability may be related to the greater hydroxyl content of the low-temperature films [15], or perhaps more generally to the greater thermal budget required for the structural transitions involved in passivation and depassivation of the surface.

Most interestingly, a minimum  $J_0$  of 7 fA cm<sup>-2</sup>, significantly lower than that obtained by conventional annealing, is observed for Al<sub>2</sub>O<sub>3</sub> films deposited at ~410°C after RTA at 500 or 550°C. This represents an exceptionally good level of surface passivation, and prompts an investigation of the interface properties of these films. C–V measurements of identically prepared MIS structures (Fig. 5) reveal that after RTA,  $D_{it}$  is reduced by a factor of two, from  $11 \times 10^{10}$  eV<sup>-1</sup> cm<sup>-2</sup> to ~5×10<sup>10</sup> eV<sup>-1</sup> cm<sup>-2</sup>. A similar reduction of  $D_{it}$  occurs upon extended (30 minutes) annealing at 425°C in N<sub>2</sub>. However, while the longer, lower temperature anneal results in an increase in negative  $Q_{tot}$ to  $-2.0 \times 10^{12}$  cm<sup>-2</sup> from its as-deposited value of  $-1.2 \times 10^{12}$  cm<sup>-2</sup>, the RTA results in a significantly larger increase, to  $Q_{tot} = -3.3 \times 10^{12}$  cm<sup>-2</sup>. This combination of remarkably low  $D_{it}$  and large  $Q_{tot}$  accounts for the exceptionally low  $J_0$ of surfaces passivated by these films.

In order to show even more clearly the high quality of the passivation layers prepared in this way, Fig. 6 compares  $\tau_{eff}$  vs  $\Delta n$  measured with these films, both as-deposited and after either a standard furnace anneal or an optimised RTA treatment, with that of identical substrates passivated by Al<sub>2</sub>O<sub>3</sub> prepared by standard thermal and plasma-assisted ALD processes. The latter are considered to provide a benchmark for surface passivation quality, where the plasma process is known to yield better results, but is less industrially relevant. Following a standard extended anneal, the APCVD Al<sub>2</sub>O<sub>3</sub> layer performs well, but does not quite reach the level of the thermal ALD film. However, if an optimised RTA is applied in place of the standard anneal,  $\tau_{eff}$  of the APCVD-passivated sample surpasses that of the thermal ALD case. This impressive result shows that CVD processes are capable of forming Al<sub>2</sub>O<sub>3</sub> layers of exceptionally high quality, matching or exceeding that of similar layers prepared by state-of-the-art thermal ALD processes. It also shows the importance of optimising the post-deposition thermal processing to maximise the potential of the deposited layers.



Fig. 5. a) Typical C–V characteristics and resulting  $D_{it}$  distribution for Al<sub>2</sub>O<sub>3</sub> deposited on *p*- and *n*-type substrates at ~410°C, as-deposited and after RTA at 500°C. b) Midgap  $D_{it}$  vs  $Q_f$  determined from multiple such measurements at different points on the same samples, both as-deposited, and after either annealing at 425°C in N<sub>2</sub>, or RTA at 500°C. Inset shows the measured (solid) and set (dashed) RTA temperature profile.



Fig. 6.  $\tau_{eff}$  vs  $\Delta n$  for 480 µm thick *p*-type 1  $\Omega$ cm (100) FZ-Si wafers passivated by APCVD Al<sub>2</sub>O<sub>3</sub> films deposited at 410°C, both as-deposited and after either a standard 30 min anneal at 425°C in N<sub>2</sub>, or an RTA at 500°C. Data are also shown for the same wafers passivated by Al<sub>2</sub>O<sub>3</sub> prepared by thermal ALD (200°C, 20 nm, annealed for 30 min at 425°C in N<sub>2</sub>) and plasma-assisted ALD (175°C, 20 nm, annealed for 15 min at 425°C in S<sup>5</sup> H<sub>2</sub> in Ar).

The degradation of  $J_0$  observed at higher RTA temperatures appears to conflict with previous results on the firing stability of the Al<sub>2</sub>O<sub>3</sub> passivation. In previous work [11], we observed passivation to improve following a conventional fast-firing step in a belt furnace with peak substrate temperature over 800°C. Conversely, a much shorter RTA step at a similar temperature was found to result in a significant deterioration of passivation quality for the same films and substrates. The reason for this apparent discrepancy may be related to the different heating methods employed by the two processes, which in the case of RTA involves high-intensity illumination of the substrate by infrared lamps, which at high intensity can have significant emission in the UV [23]. Such illumination

has previously been linked to surface passivation degradation observed for  $SiO_2$ -passivated surfaces during RTA [23]. It is therefore possible that the observed degradation is related to the presence of this illumination rather than being a purely thermal process. Further work is necessary to clarify this question.

It is possible that the illumination also plays a role in the passivation improvement observed at lower temperatures. It is known that strong illumination with higher-energy photons may induce an increase in the  $Al_2O_3$  fixed charge  $Q_f$  [24], which would be consistent with the strong increase observed in Fig. 5. If this were the case it would raise interesting questions regarding other previous results, since a number of previous works have employed similar RTA systems to perform their activation anneals.

## 5. Conclusions

Systematic investigation of the effects of post-deposition heat treatment by RTA on Al<sub>2</sub>O<sub>3</sub> surface passivation reveals a number of trends.  $J_0$  post-RTA may either increase or decrease depending on Al<sub>2</sub>O<sub>3</sub> deposition temperature and RTA peak temperature, but generally increases with the latter. The optimum deposition temperature depends on the post-deposition thermal processing to be applied. Films deposited at lower temperatures provide worse passivation after low temperature heat treatment, but maintain this passivation better at higher RTA temperatures. Al<sub>2</sub>O<sub>3</sub> passivation exposed to RTA at high temperatures exhibits a degradation that is not observed for similar films exposed to a conventional fast-firing process at the same temperature, pointing to the possible influence of nonthermal effects in the degradation process. On the other hand, RTA at more moderate temperatures can result in passivation improvements beyond those possible with a standard extended anneal at 425°C. An exceptionally low  $J_0$ of 7 fA/cm<sup>2</sup>, due to a combination of very low  $D_{it}$  and unusually high  $Q_f$ , is achieved by the use of a short 500– 550°C RTA combined with optimised deposition conditions. It is shown by direct comparison that this passivation is as good as or better than that of Al<sub>2</sub>O<sub>3</sub> prepared by a standard thermal ALD process, demonstrating the outstanding passivation potential of Al<sub>2</sub>O<sub>3</sub> prepared by relatively simple CVD processes, as well as the potential gains of optimising the post-deposition thermal treatment.

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