## Supplementary material to:

Transformation products of the high-volume production chemicals 1-vinyl-2-pyrrolidinone and 2-piperazin-1-ylethanamine formed by UV photolysis

Benigno José Sieira, Rosario Rodil, Rafael Cela, José Benito Quintana, Rosa Montes
Department of Analytical Chemistry, Institute of Research on Chemical and Biological Analysis (IAQBUS), Universidade de Santiago de Compostela, 15782, Santiago de Compostela, Spain

* Corresponding Authors:
R. Rodil (rosario.rodil@usc.es)
J.B. Quintana (jb.quintana@usc.es)
R. Montes (rosamaria.montes@usc.es)

Phone number: +34 881816035

Table S1. VP and PPE properties and structure.
Table S2. Photochemical kinetic parameters obtained from non-linear fitting to an exponential decay and calculation of quantum yields (UPW: ultrapure water; SW: surface water).

Figure S1. UV photolysis kinetic plots in ultrapure (UPW) and river water (RW) of (A) VP and (B) PPE. Data normalized to time 0 (untreated samples).

Figure S2. UV/Vis absorption spectrum of (a) VP and (b) PPE.
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Figure S4. TPs formation profiles for VP (a) and PPE (b). Signal of TPs normalized to the signal of their parent chemical at time 0.

Figure S5. QTOF product ion spectra of PPE and its TPs. (a) PPE , (b) PPE-87, (c) PPE-115, (d) PPE-126, (e) PPE-146 and (f) PPE-160

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Table S1. VP and PPE properties and structure.

| Compound Name | 1-vinylpyrrolidin-2-one | 2-piperazin-1-ylethanamine |
| :---: | :---: | :---: |
| Acronym | VP | PPE |
| CAS Number | 88-12-0 | 140-31-8 |
| Molecular formula | $\mathrm{C}_{6} \mathrm{H}_{9} \mathrm{NO}$ | $\mathrm{C}_{6} \mathrm{H}_{15} \mathrm{~N}_{3}$ |
| Log D (pH 7.4) ${ }^{(1)}$ | 0.38 | -4.53 |
| $\log P^{(1)}$ | 0.37 | -0.86 |
| Boiling point (760 $\mathrm{mmHg})^{(1)}$ | $217{ }^{\circ} \mathrm{C}$ | 220응 |
| Consumer uses ${ }^{(2)}$ | Washing and cleaning products, cosmetics and personal care products, paints, coatings, and adhesives. Laboratory chemicals. | Adhesives, sealants, coatings, paints. pH regulators and water treatment products. |
| Harmonised classification and labelling (CLPOO) ${ }^{(2)}$ | GHS05 Corrosive, GHS06 Acute toxicity, GHS09 Hazardous to the environment | GHS05 Corrosive, GHS07 Health hazard |

Structure


${ }^{(1)}$ Predicted values, ACDLabs, ${ }^{(2)}$ Data from www.echa.europe.eu

Table S2. Photochemical kinetic parameters obtained from non-linear fitting to an exponential decay and calculation of quantum yields (UPW: ultrapure water; SW: surface water).

|  | $k$ <br> $\left(\mathbf{m i n}^{-1}\right)^{a}$ | $\mathbf{t}_{1 / 2}$ <br> $(\mathbf{m i n})^{\mathrm{a}}$ | $\mathbf{R}^{2}$ | $\varepsilon$ <br> $\left(\mathbf{M}^{-1} \mathbf{c m}^{-1}\right)$ |
| :---: | :---: | :---: | :---: | :---: |
| VP (UPW) | $0.36 \pm 0.03$ | $1.9 \pm 0.2$ | 0.9989 | $4.38 \cdot 10^{3}$ |
| $\left(\mathbf{m o l} /\right.$ einstein) ${ }^{\mathrm{a}}$ |  |  |  |  |

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Figure S1. UV photolysis kinetic plots in ultrapure (UPW) and river water (RW) of (A) VP and (B) PPE. Data normalized to time 0 (untreated samples).



Figure S2: UV/Vis absorption spectrum of (a) VP and (b) PPE.


Figure S3. QTOF product ion spectra of VP and its TPs. (a) VP and (b) VP-86


Figure S3 cont. QTOF product ion spectra of VP and its TPs. (c) VP-116 and (d) VP-227


Figure S4. TPs formation profiles for VP (a) and PPE (b). Signal of TPs normalized to the signal of their parent chemical at time 0.


Figure S5: QTOF product ion spectra of PPE and its TPs. (a) PPE , (b) PPE-87 and (c) PPE-115




Figure S5 cont: QTOF product ion spectra of PPE and its TPs. (d) PPE-126, (e) PPE-146 and (f) PPE-160


Figure S6: Extracted ion chromatograms of VP and identified TP


Figure S6 cont: Extracted ion chromatograms of VP identified TP


Fig S7: Extracted ion chromatograms of PPE and identified TP


Fig S7 cont.: Extracted ion chromatograms of PPE identified TP

Table S3. Summary of EPA T.E.S.T. predicted toxicity endpoints. Note: n.a. means that no prediction was possible.

| Compound | Fathead minnow | Daphnia magna | T. pyrifromis | Oral rat |
| :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{LC}_{50} 96 \mathrm{~h} \mathrm{mg} / \mathrm{L}$ | LC $\mathrm{c}_{0} 48 \mathrm{~h} \mathrm{mg} / \mathrm{L}$ | IGC $5048 \mathrm{~h} \mathrm{mg} / \mathrm{L}$ | $\mathrm{LD}_{50} \mathrm{mg} / \mathrm{Kg}$ |
| $V P$ | $1.73 \mathrm{E}+02$ | $1.36 \mathrm{E}+01$ | $1.62 \mathrm{E}+03$ | $1.36 \mathrm{E}+03$ |
| VP-86 | $4.86 \mathrm{E}+02$ | 4.17E+01 | $3.14 \mathrm{E}+03$ | $1.11 \mathrm{E}+03$ |
| VP-116 | $1.39 \mathrm{E}+03$ | $1.15 \mathrm{E}+02$ | $7.42 \mathrm{E}+03$ | $1.16 \mathrm{E}+03$ |
| VP-132 | $2.49 \mathrm{E}+03$ | $3.61 \mathrm{E}+02$ | $8.92 \mathrm{E}+03$ | $8.92 \mathrm{E}+03$ |
| VP-219 | $1.09 \mathrm{E}+02$ | $2.83 \mathrm{E}+00$ | n.a. | $7.37 \mathrm{E}+02$ |
| VP-227 | $1.91 \mathrm{E}+02$ | $4.32 \mathrm{E}+02$ | $1.53 \mathrm{E}+03$ | $4.71 \mathrm{E}+03$ |
| PPE | $2.55 \mathrm{E}+03$ | $6.07 \mathrm{E}+01$ | $1.59 \mathrm{E}+03$ | $1.78 \mathrm{E}+03$ |
| PPE-87 | $1.37 \mathrm{E}+03$ | $1.31 \mathrm{E}+02$ | $2.02 \mathrm{E}+03$ | $2.09 \mathrm{E}+03$ |
| PPE-111 | $9.06 \mathrm{E}+02$ | $4.43 \mathrm{E}+01$ | $4.27 \mathrm{E}+02$ | $1.29 \mathrm{E}+03$ |
| PPE-115 | $1.07 \mathrm{E}+03$ | $2.15 \mathrm{E}+03$ | n.a. | $1.80 \mathrm{E}+03$ |
| PPE-126 | n.a. | n.a. | n.a. | n.a. |
| PPE-144 | $2.91 \mathrm{E}+03$ | $6.96 \mathrm{E}+01$ | $1.30 \mathrm{E}+03$ | $2.75 \mathrm{E}+03$ |
| PPE-146 | $3.38 \mathrm{E}+03$ | $2.11 \mathrm{E}+02$ | $2.07 \mathrm{E}+03$ | $3.37 \mathrm{E}+03$ |
| PPE-160 | $3.81 \mathrm{E}+03$ | $4.10 \mathrm{E}+02$ | $3.88 \mathrm{E}+03$ | $3.23 \mathrm{E}+03$ |

Table S4. Summary of ECOSAR predicted toxicity endpoints. Note: in some cases two values were obtained for the same TP, the $\left(^{*}\right)$ symbol is used when the estimation is based on the "amide" group class, while the other value was obtained when "aliphatic amine" group class was used.

| Compound | Acute toxicity |  |  | Chronic toxicity |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Fish 96h LC 50 | Daphnid 48 h LC50 | $\begin{gathered} \text { Green Algae } 96 \mathrm{~h} \\ \mathrm{EC}_{50} \end{gathered}$ | Fish ChV | Daphnid ChV | Green Algae ChV |
| $V P^{*}$ | $7.23 \mathrm{E}+02$ | $9.16 \mathrm{E}+02$ | 4.63E+01 | $4.38 \mathrm{E}+00$ | 7.96E+01 | $1.36 \mathrm{E}+01$ |
| VP-86 | $1.53 \mathrm{E}+03$ | $2.11 \mathrm{E}+03$ | $7.98 \mathrm{E}+01$ | 7.22E+00 | $1.49 \mathrm{E}+02$ | $1.83 \mathrm{E}+01$ |
| VP-116 | $1.97 \mathrm{E}+04$ | $3.29 \mathrm{E}+04$ | $6.50 \mathrm{E}+02$ | $5.36 \mathrm{E}+01$ | $1.46 \mathrm{E}+03$ | $8.57 \mathrm{E}+01$ |
| VP-132 | $3.40 \mathrm{E}+04$ | $5.88 \mathrm{E}+04$ | $1.03 \mathrm{E}+03$ | $8.37 \mathrm{E}+01$ | $2.40 \mathrm{E}+03$ | $1.23 \mathrm{E}+02$ |
| VP-219 | $8.50 \mathrm{E}+01$ | $9.60 \mathrm{E}+00$ | $8.82 \mathrm{E}+00$ | $5.84 \mathrm{E}+00$ | $7.48 \mathrm{E}-01$ | $2.82 \mathrm{E}+00$ |
| VP-227 | $4.08 \mathrm{E}+04$ | $6.85 \mathrm{E}+04$ | $1.33 \mathrm{E}+03$ | $1.10 \mathrm{E}+02$ | $3.01 \mathrm{E}+03$ | $1.73 \mathrm{E}+02$ |
| PPE | $5.48 \mathrm{E}+03$ | 4.31E+02 | $8.09 \mathrm{E}+02$ | $1.13 \mathrm{E}+03$ | $2.37 \mathrm{E}+01$ | $1.99 \mathrm{E}+02$ |
| PPE-87 | $1.14 \mathrm{E}+03$ | $9.83 \mathrm{E}+01$ | $1.54 \mathrm{E}+02$ | $1.79 \mathrm{E}+02$ | $5.89 \mathrm{E}+00$ | $4.05 \mathrm{E}+01$ |
| PPE-111 | $2.63 \mathrm{E}+02$ | $2.58 \mathrm{E}+01$ | $3.13 \mathrm{E}+01$ | $2.76 \mathrm{E}+01$ | $1.76 \mathrm{E}+00$ | $9.03 \mathrm{E}+00$ |
| PPE-115 | $4.79 \mathrm{E}+03$ | $3.77 \mathrm{E}+02$ | $7.06 \mathrm{E}+02$ | $9.84 \mathrm{E}+02$ | $2.07 \mathrm{E}+01$ | $1.74 \mathrm{E}+02$ |
| PPE-115* | $1.91 \mathrm{E}+04$ | $3.19 \mathrm{E}+04$ | $6.34 \mathrm{E}+02$ | $5.24 \mathrm{E}+01$ | $1.43 \mathrm{E}+03$ | 8.40E+01 |
| PPE-126 | $1.05 \mathrm{E}+03$ | $9.39 \mathrm{E}+01$ | $1.38 \mathrm{E}+02$ | $1.49 \mathrm{E}+02$ | $5.82 \mathrm{E}+00$ | $3.71 \mathrm{E}+01$ |
| PPE-144 | $1.54 \mathrm{E}+04$ | $1.13 \mathrm{E}+03$ | $2.44 \mathrm{E}+03$ | $3.94 \mathrm{E}+03$ | $5.78 \mathrm{E}+01$ | 5.70E+02 |
| PPE-144* | $7.34 \mathrm{E}+04$ | $1.35 \mathrm{E}+05$ | $1.94 \mathrm{E}+03$ | $1.53 \mathrm{E}+02$ | $4.78 \mathrm{E}+03$ | $1.95 \mathrm{E}+02$ |
| PPE-146 | $6.31 \mathrm{E}+04$ | $4.16 \mathrm{E}+03$ | $1.11 \mathrm{E}+04$ | $2.24 \mathrm{E}+04$ | $1.92 \mathrm{E}+02$ | $2.40 \mathrm{E}+03$ |
| PPE-160 | $1.75 \mathrm{E}+05$ | $1.07 \mathrm{E}+04$ | $3.31 \mathrm{E}+04$ | $7.75 \mathrm{E}+04$ | $4.62 \mathrm{E}+02$ | $6.79 \mathrm{E}+03$ |
| PPE-160* | $1.30 \mathrm{E}+06$ | $2.99 \mathrm{E}+06$ | $1.95 \mathrm{E}+04$ | $1.37 \mathrm{E}+03$ | $6.04 \mathrm{E}+04$ | $9.96 \mathrm{E}+02$ |


[^0]:    ${ }^{\text {a }}$ Mean $\pm 95 \%$ confidence interval
    ${ }^{\mathrm{b}}$ Value from (Wols et al., 2012)

