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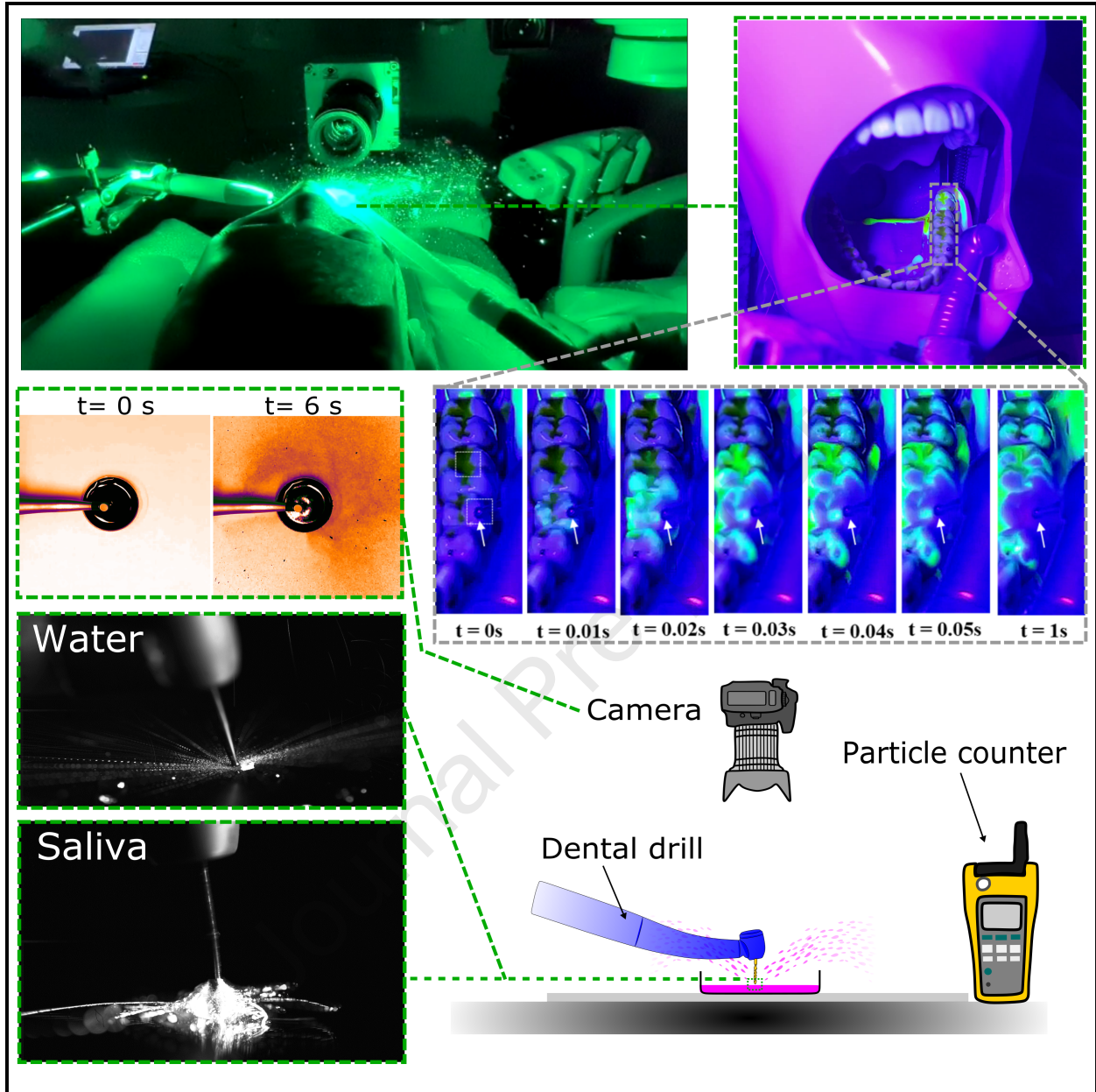
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Removal and Dispersal of Biofluid Films by Powered Medical Devices: Modelling Infectious Agent Spreading in Dentistry

Ian Eames^{1,*}, Francesco D'Aiuto^{2*}, Somayah Shahreza¹, Yousef Javanmardi¹, Ramanarayanan Balachandran¹, Martin Hyde³, Yuan-Ling Ng⁴, Kishor Gulabivala⁴, Sara Watson¹, Hywel Davies¹, Nicolas Szita⁵, Janette Khajeh¹, Jeanie Suvan^{2,*}, Emad Moeendarbary^{1*}

¹Department of Mechanical Engineering, University College London, Torrington Place, London, WC1E 7JE, UK

²Unit of Periodontology, UCL Eastman Dental Institute, University College London, UK

³TSI, 30 Millbank, Westminster, London SW1P 4WP, UK

⁴Unit of Endodontology, UCL Eastman Dental Institute, University College London, UK

⁵Department of Biochemical Engineering, University College London, Bernard Katz Building, Gower Street, London, WC1E 6BT, UK

*corresponding authors

Ian Eames, i.eames@ucl.ac.uk

Francesco D'Aiuto, f.daiuto@ucl.ac.uk

Jeanie Suvan, j.suvan@ucl.ac.uk

Emad Moeendarbary, e.moeendarbary@ucl.ac.uk

Lead contact: Emad Moeendarbary, e.moeendarbary@ucl.ac.uk

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Summary

Medical procedures can disperse infectious agents and spread disease. Particularly, dental procedures may pose a high risk of disease transmission as they use high-powered instruments operating within the oral cavity that may contain infectious microbiota or viruses. Here we assess the ability of powered dental devices in removing the biofluid films and identified mechanical, hydrodynamic, and aerodynamic forces as the main underlying mechanisms of removal and dispersal processes. Our results indicate that potentially infectious agents can be removed and dispersed immediately after dental instrument engagement with the adherent biofluid film while the degree of their dispersal is rapidly depleted due to removal of the source and dilution by the coolant water. We found that droplets, created by high-speed drill interactions typically travel ballistically while aerosol-laden air tends to flow as a current over surfaces. Our mechanistic investigation offers plausible routes for reducing the spread of infection during invasive medical procedures.

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Introduction

Medical procedures using powered instruments span a broad spectrum of specialities including orthopaedics, otorhinolaryngology, ophthalmology, and dentistry, and have the potential to release biofluids into the local vicinity. While biofluids such as blood, saliva, mucous, or tears play various roles, such as nutrient conveyance, aid digestion and lubrication, they also have the potential to transmit viral and bacterial pathogens from one person to another (Xu et al., 2020). Several surgical procedures involve cutting bone or sinewy tissue, which demand a great deal of mechanical energy introduced either electrically or pneumatically. To mitigate tissue damage due to heat generated during cutting, coolant (usually water) is introduced continuously to quench the cutting surfaces. The presence of biofluids, water, air and moving surfaces in the form of instrument tips or blades creates a potential for dispersing infectious agents including splashes, aerosols (WELLS, 1934) and droplets and spreading infection through inhalation or a contact route (Tang et al., 2006).

While power-driven instrument types are common across clinical sciences, generally differing in their size and speed, dentistry represents a unique setting as it deals with the hardest tissues in the human body (i.e., enamel and dentine) requiring the fastest cutting drills and robust cooling mechanisms to prevent thermal damage to the dental pulp. Furthermore, the oral cavity, as the gateway to the body, is an open environment containing multiple biosolids and biofluids that serve as a reservoir for microbiota. The close connection with the respiratory tract and nasal pathway makes the oral environment and its associated biofluids potential reservoirs containing infectious agents that transmit diseases such as *Mycobacterium tuberculosis*, Herpes simplex or SARS-CoV-2.

The potential risk of spreading infection during a dental procedure involving an air-turbine drill and water coolant was recognised as early as the 1960's (STEVENS, 1963). The routes however by which infectious agents are removed and dispersed have not been thoroughly studied (Harrel and Molinari, 2004). The current understanding on this topic, encapsulated in international guidance WHO 2020, PHE 2020 ("COVID-19: infection prevention and control (IPC) - GOV.UK," n.d.), is that all instruments that create an aerosol require specialist protocols to mitigate the risk of spreading disease. The challenge of mitigating risk partly involves characterising what is in the air with some confusion over the definition of an aerosol. Typically an aerosol is characterised by particles whose diameter is less than 5 microns with the criterion based on the potential to be inhaled into the lower respiratory tract (Fennelly, 2020). The range for inhalation could be wider (less than 12 microns in diameter) and indeed a droplet size can shrink by as much as 80% due to evaporation. While, the guidance employs an instrument classification based on their power to generate aerosols, they lack the underpinning fundamental science of how instruments interact with biofluid films and their potential to generate agents carrying infection (Epstein et al., 2020).

Motivated by the lack of systematic investigation on the topic (Kumbargere Nagraj et al., 2020; Volgenant and de Soet, 2018), we studied how biofluid films, that may contain virus or bacteria, are removed and dispersed via dental instruments and procedures. We focus on the dispersal mechanism that is centred around the removal of biofluid films and crucially distinguish between coolant fluid that comes from the dental device and the potentially infectious fluid. Using imaging techniques and dyed fluid films, we analysed the fundamental mechanisms in a laboratory setting and assessed the relevant processes under clinically relevant conditions.

Results

Mechanisms of aerosol and droplet generation

Droplet size has an important consequence for transport processes. It is important to clarify the terminology applied to distinguish between the different droplet size. The usual way to distinguish

97 aerosols is based on the potential for deep inhalation setting a scale of 5 microns in diameter, rather
98 than the physical processes that keep the matter in the air. Large droplets settle quickly and move
99 ballistically while aerosols are distinguished by their long residence time in the air. The distinction
100 between these two groups is imprecise, especially since droplets evaporate and shrink. Based on the
101 resolution of our probing techniques, here we distinguish between aerosols, fine droplets and droplets
102 corresponding approximately to <20 microns, <200 microns and >200 microns in diameter respectively.
103 Application of airborne particle counter (detecting <20 μm particles), deposition (detecting >~50 μm
104 particles), and high-speed imaging (detecting >~200 μm particles) techniques allowed us to probe the
105 particles with sizes in these three categories.

106
107 We assessed three common dental devices (dental drill or air-rotor handpiece, ultrasonic scaler, 3-in-1
108 air-water syringe) for their potential to generate aerosols/ droplets by mechanical rotation/vibration of
109 surfaces (bur or ultrasonic tip) or flow of air/water through small orifices (Fig 1a). Air-rotor generates the
110 finest particles because of the fastest bur rotation and the highest air speed. Generated droplets are
111 propelled ballistically, while the created aerosol cloud around the drill is dispersed by the air jet and the
112 coolant spray generates a turbulent aerosol jet flow that slows rapidly with distance due to entrainment
113 (Fig 1b-d). The droplets generated by an ultrasonic scaler appear to be larger and move typically with
114 an average velocity of $\sim 2 \text{ ms}^{-1}$ (Fig 1b-d). The air-rotor handpiece propels droplets at a greater initial
115 velocity compared to ultrasonic scaler (Fig 1b) with droplets reaching velocities of over $\sim 10 \text{ ms}^{-1}$ at
116 proximity to the rotating bur. Close to the devices, the droplets travel in a linear path while far from the
117 device, they move with a parabolic trajectory (Fig 1b).

118
119 The ability of the instrument to convert the coolant water into droplets of different sizes depends on the
120 balance between the surface tension force and the inertial forces created through either of
121 vibrating/rotating surfaces, air and water flow. The different strengths in the mechanical and
122 aero/hydrodynamic forces lead to contrasting droplet sizes with aerosols/fine droplets tending to be
123 generated from the fast-moving surfaces (bur of the air-rotor or vibrating tip of the ultrasonic scaler). Air
124 flow jet (expelled from the air-rotor handpiece) create droplets while very large droplets are formed by
125 the water jet (3-in-1 air-water syringe). We estimated the Weber number (We), which is a measure of
126 the relative strength of inertial to surface tension forces, to characterise the potential of the instrument
127 to generate droplets of different sizes. The ultrasonic scaler generates larger droplets, which splash
128 when impacting surfaces ($We \sim 20$), while the air-rotor creates faster moving small droplets ($We \sim 1.5$),
129 that have a greater tendency to follow the air flow (Fig 1e). Based on the type and strength of the inertial
130 forces, we categorised the ability of the instruments to convert the coolant water into aerosols, fine
131 droplets, and droplets on a diagram in Fig 1f (see Scaling analysis in the method section).

132
133 The inertial forces that involve in the generation of droplets from the coolant water similarly drive the
134 removal of fluid film in the form of splashes, aerosols and droplets of different sizes. Therefore, as
135 examined in the next sections, aerosols and droplets can be generated from the biofluid layer through
136 either mechanical interactions caused by a surface vibrating or rotating while in contact with the layer,
137 aerodynamic interactions caused by the air flowing onto the layer and hydrodynamic interactions caused
138 by the flow of coolant water jet or droplets hitting the layer. To build up a conceptual picture of the
139 removal processes, we designed a series of experiments starting with an idealised interaction with a
140 fluid film and then building up to more complex interactions involving model teeth and mouth of a
141 manikin.

142 143 **Interaction of powered instruments and adherent layers in controlled setting**

144
145 The first set of experiments involved the interaction between a fixed instrument and a fluid film focussing
146 specifically on the air-rotor (Fig 2a), which produced the highest amount of aerosol and fine droplets
147 with high velocity characteristic (Fig 1e). Furthermore, examination of the air-rotor and ultrasonic scaler
148 interacting with a thin fluid layer placed on a circular glass slide clearly suggested the negligible droplet

149 generation by ultrasonic scaler compared to air-rotor particularly when operated at a fix position (SI Figs
150 2 and 3).

151
152 To distinguish between the interconnected removal mechanisms a fixed air-rotor was operated to run
153 the drill, water, and coolant separately and its interaction with a thin layer of biofluid under three
154 operating modes was examined; **mode-1**: the normal operating condition which involves rotation of the
155 bur driven by the air jet while allowing the expulsion of both the air and coolant jet; **mode-2**: rotation of
156 the bur driven by the air jet allowing the expulsion of the air while water expulsion was inhibited; and
157 **mode-3**: rotation of the bur driven by the air jet while both air and water jet expulsion were inhibited.

158 159 **Deposition measurements**

160
161 Under these three modes and using dyed water (dyed either water film or coolant water), first we imaged
162 the droplet deposition during continuous 2 min operation of the instrument engaging with a layer of water
163 covering the bottom of a plastic dish (Fig 2a). Surprisingly, during the continuous operation in mode-1
164 and -2 no droplets and splashes were detected and only a limited number of splashes were observed
165 immediately after operation was stopped in mode-1 (Movies 1-3). Interestingly, in the absence of air and
166 water jets (mode-3), upon start of the bur rotation, the immediate interaction of the bur with the adherent
167 fluid layer generated large splashes (Fig 2b, Movie 4). This was followed by a significant deposition of
168 dyed droplets during the 2 min interaction of the bur with the dyed layer which generated a continuous
169 cloud of fine aerosol deposited with distinct asymmetric patterns involving a complex interaction
170 between the turbulent air flow induced by the bur and the gravity-driven flow induced by the aerosols
171 (Fig 2b, first column). The intensity of the dye pattern increased in time due to the steady flow of aerosol-
172 laden air. The aerosol-laden air appeared to flow over the edge of the dish while the lip of the dish
173 perturbed the low-speed aerosol-laden flow and created a narrow shadow around the dish (Fig 2b).

174
175 Next, we dyed the coolant water instead and investigate the dispersal patterns generated as the result
176 of water jet expulsion and the interaction of the drill with water jet and the clear/ undyed water film layer
177 (Fig 2b, columns 2, 3 and 4, Movies 5-7). When the air-rotor was interacted with the clear layer under
178 mode-1 (using dyed coolant water), significant deposition was observed (Fig 2b, second column) which
179 was less pronounced compared to mode-3 (Fig 2b, first column). The removal of the water layer reduced
180 the amount of deposition (Fig 2b, third column). However, very small amount of deposition (mostly
181 located at the proximity to the lip of the dish) was detected when bur was removed (Fig 2b, last column).

182
183 Finally, investigation of the effects of layer thickness revealed that the intensity and the spread area of
184 the deposition depend on the thickness of the layer when air-rotor was operating in mode-3 (Fig 2c).
185 When the thin layer was engaged with the bur, the continuity of the dyed water layer was affected due
186 to the removal of the water and limited amount of deposition was detectable (Fig 2c). However, when
187 the thickness of the dyed layer increased a continuous reservoir of dyed water was available for removal/
188 droplet generation and therefore the deposition area was expanded due to a high mass flux of droplets
189 and continuous flow of aerosol-laden current (Movies 8-10).

190 191 **Aerosol measurements**

192
193 Our simple photography technique was capable of capturing the dynamics of droplet (with diameters
194 larger than $\sim 50 \mu\text{m}$) deposition and suggested distinctive removal and dispersion mechanisms.
195 However, finer droplets (typically less than $\sim 20 \mu\text{m}$) are known to have a higher degree of retainment
196 within the air and penetration into respiratory system making them more likely source of disease
197 transmission. Therefore, we tested the validity of our findings for significantly finer particles by employing
198 an airborne particle counter to probe the dispersal evolution of 0.3 to $10 \mu\text{m}$ droplets. Consistent with
199 the deposition tests, no aerosolised droplets were detected under mode-2 (expulsion of air without
200 presence of coolant water) as the flow of air pushed the dyed layer away from the bur, preventing the

201 bur engagement with the water layer (Fig 2d). However, presence of aerosols was recorded under
202 mode-1 and mode-3 when the bur was engaged with water jet or sufficiently thick fluid layer generating
203 a significant flow of aerosolised droplets (Fig 2e,f).

204 These observations suggest that under mode (1) the aerosol (in the form of either infectious or non-
205 infectious fine droplets) can be generated via three sources: the bur interacting with the primary layer
206 (Drill interaction), the bur interacting with air/water jet that directly hitting the bur and also the dish
207 (Air+Water+Drill), and the air/water jet direct dispersion or expulsion after hitting the dish (Air+Water).
208 Our examination of the distinct contributions from each source to aerosol production (Fig 2e,f), indicated
209 that Air+Water+Drill generated the highest levels of aerosol with a wide range of sizes (0.3 to 10 μm)
210 that remained within the air beyond 2 mins operation, while the Air+Water produced significantly lower
211 aerosol levels with almost negligible amount for 5 and 10 μm droplets (Fig 2e,f) consistent with
212 deposition experiments (Fig 2b). Interestingly, compared to Air+Water+Drill, Drill condition produced
213 significantly lower amounts of (potentially infectious) aerosol, which diminished rapidly. Furthermore, for
214 each curve we observed an initial peak (occurring few seconds after the start of the operation) with the
215 fastest peak occurring for the Drill, suggesting the significant inertial power of fast rotating bur and effects
216 of pre-engagement and wetting of the bur with the biofluid film. Removal of water layer from the dish
217 had minimal effects on production of small aerosols (0.3 μm , SI Fig 1d) while the amount of large
218 aerosols (5 μm , SI Fig 1e) significantly increased under the presence of water layer which is consistent
219 with the deposition measurement (Fig 2b, second and third columns).
220

221 Simultaneous measurements of aerosol at 0.4 m and 2 m distances (using two probes located on the
222 same height) showed a lag of ~ 20 s in the occurrence of the peak at the further distance (Fig 2h) while
223 the aerosol concentration decayed after ~ 60 s at both sites after the operation stopped (Fig 2h and SI
224 Fig 1f-h). The concentration of aerosol at the further point was dramatically lower for large droplets, with
225 the loading of 5 and 10 μm approached to almost zero.
226

227 **High speed imaging**

228
229 Conducting high speed imaging we also visualised the interaction of rotating bur with the thin fluid film
230 and investigated how the fluid film is removed and aerosolised (Fig 3). Three stages of removal were
231 revealed: the initial rotation of the burr, steady rotation with droplet generation and full removal of the
232 film (Fig 3, Movies 11-16). During the initial rotation of the bur, the water was also rotated by the bur
233 creating initial thin water filaments that fragmented and produced droplet ejecta whose size became
234 progressively finer as the water layer diminished (Fig 3a). The ability of a bur to remove an adherent
235 biofluid film depends on the rheological properties of the biofluid which was assessed by comparing the
236 removal of water with unstimulated saliva collected from a participant (Fig 3b). While similar processes
237 observed for both water and saliva layers, the timescale of the processes was longer for saliva layer.
238 The saliva layer initially rotated around the bur at longer timescale, longer filaments were formed and
239 fine droplet formation via fragmentation was greatly suppressed compared to water film due to increased
240 viscosity (Fig 3b).
241

242 **Interaction of powered instruments and adherent layers assessed in simulated clinical setting**

243
244 To gain a more realistic insight relevant to the clinical situation, we next investigated the interaction
245 between an air-rotor and teeth using either a set of adult teeth model or a manikin. Considering real
246 tooth geometry meant that the bur and air-water flows interact with uneven/ rough surfaces and the
247 concave section of the crown (a potential saliva reservoir) while the oral cavity has significant impact on
248 containment of the splashes and the aerosol flow.
249

250 In the first series of tests (Fig 4a), teeth (44-47 ISO 3950) were coated with simulated saliva mixed with
251 fluorescein dye, and an air-rotor handpiece was held by hand near the teeth with the bur contacting the
252 occlusal surface of tooth 46 (ISO 3950). Upon turning on the air-rotor (within first 300 ms), the presence

253 of cooling water immediately diluted the fluorescein coating leading to the flow of dye mixture over and
254 away from the teeth (Movie 17). The convex cusps of the crown created a pool of diluted dyed water
255 pooling at the occlusal pit. Consistent with the tests in Fig 2b, engagement of the drill with the pooled
256 dyed water led to generation and ejection of a small number of fine dyed droplets detected after
257 ~1000ms at the distances up to ~20 cm away from the tooth (Fig 4a).

258
259 Next, the dispersion of the coolant water was examined by mixing the coolant water with the red food
260 dye. The dyed coolant water initially coated the bur and the crown of the tooth before any observable
261 droplet dispersion (Movie 18). Within 100 ms, we observed droplet deposition generated through
262 mechanical and aero/hydrodynamic processes. The drill's mechanical interactions (via its shank similar
263 to Fig 1a) with coolant aerosol-jet potentially generates aerosols with the smallest droplet size (Fig 1f)
264 while the bur engagement with the coolant pool at the occlusal pit produces fine droplets (as observed
265 in Fig 2c). The aerosol-laden coolant jet, inertially impacted the tooth surface and finer components
266 dispersed in the air and flow as current along the surface (Movies 17,18).

267
268 Finally, using a manikin we explored the influence of drill orientation/movement and the mouth geometry
269 in directing and confining droplet splatter. Simulant-saliva was mixed with fluorescein powder and
270 applied to cover the teeth (34-38 ISO 3950) and the bur was engaged with teeth 35 on the buccal cusp.
271 As soon as the air-rotor started, the simulant-saliva layer was rapidly removed from the teeth (in less
272 than 1s) by the water jet before the bur engaged with the tooth (Fig 4c). While the water jet diluted the
273 dyed simulant-saliva (as detected through increase in the fluorescent light intensity, Fig 4d), the
274 movement of the drill along the buccal side and its engagement with the teeth surface led to a splatter
275 outside the mouth (Fig 4e). Droplets (mostly generated aerodynamically by the coolant jet and
276 mechanically by the bur) were propelled through the air with a tendency to be entrained into the vortex
277 created by flow separation at the side of the manikin's mouth, leading to the deposition of a mixture of
278 dyed and clear large droplets on the manikin face (Fig 4e). Thorough scrutinisation of all areas around
279 the manikin indicated a discrete number of (<5) fluorescent splashes at distances up to 1 m from the
280 head towards the foot. No dyed splashes were observed in the 4 settling plates placed within 50 cm of
281 the manikin head while further examination of these plates with a fluorescent microscope revealed a
282 small number of fine spots (Fig 4f). Owing to the absence of a propelled air component, the removal
283 pattern changed dramatically with the ultrasonic scaler as the coolant water from the agitator simply
284 flushed the simulant-saliva layer from the teeth and splashed around the mouth only when operated
285 continuously over several teeth.

286
287 Finally, we assessed the degree of the suspension of fine droplets (up to 2.5 microns) in the air (Fig 4g)
288 when either air-rotor or ultrasound scaler was operated in the manikin mouth by a dentist running a 3min
289 routine dental procedure. We detected a rise of $120 \mu\text{gm}^{-3}$ of PM2.5 (average value $50 \mu\text{gm}^{-3}$) during
290 the air-rotor operation and this level dropped dramatically when drilling ceased (10 minutes after the
291 procedure was stopped, the mass loading dropped to $5 \mu\text{gm}^{-3}$). During the ultrasonic scaler operation
292 at a *fixed position* and directed into the mouth, the air sampler did not detect a significant change
293 showing values below $3 \mu\text{gm}^{-3}$ of PM2.5 (Fig 4g). However, when the ultrasonic scaler was swept
294 around the mouth, and in some cases impinged on the air sampler (Fig 4f), the PM2.5 reached a
295 maximum of $100 \mu\text{gm}^{-3}$ (average of $30 \mu\text{gm}^{-3}$). As soon as the procedure ceased, the measured particle
296 loading in the air dropped rapidly back to the levels prior to the procedure starting and at a much faster
297 rate than observed with the dental drill.

298 299 Discussion

300
301 The primary mode of disease transmission in the clinical setting is fluidic (Bourouiba, 2021), either
302 through fine aerosols entering the air that can be inhaled, or through aerosols, fine droplets and splashes
303 that settle on surfaces and are transferred via contact (Peng et al., 2020; Tang et al., 2006). Infectious
304 agents spread by medical devices affix those generated by normal pathways (Fig 5a) including

305 breathing, speaking, coughing and sneezing (Abkarian et al., 2020). The powered medical procedures
306 and especially dental operations (as they use high powered instruments) involve complex interactions
307 between fast-moving surfaces, air and water jets that make the assessment of the risk of the spread of
308 infectious materials from patient to medical practitioners due to medical procedures challenging (Fig
309 5a).

310
311 Numerous studies have analysed the potential risk of disease transmission by aerosol and splatter
312 associated with dental procedures. Splatter tests with dyed water coolant show spray deposition over
313 large distances (Harrel et al., 1998) with a range of droplet sizes deposited within 2 m and aerosols
314 potentially dispersed further (Allison et al., 2020). CFU microbiological assays have provided
315 overwhelming evidence that the use of air rotors and drills enhances the spread of bacteria from the
316 mouth compared to when transported away from the patient through breathing, speaking, or coughing
317 (Rautemaa et al., 2006). Many previous studies do not distinguish between the clean splashes/ droplets/
318 aerosols (ejected directly from the devices (Sergis et al., 2020) or indirectly through interaction of ejected
319 clean flow with other surfaces) and those that contain the infectious biomaterials (mostly generated from
320 removal of infectious biofluid films). More recent studies have attempted to measure the distribution of
321 aerosolised simulant-saliva laden with a virus or bacteria (such as *Streptococcus mutans*) that were
322 continuously introduced into a phantom head mouth, while powered devices were operated on the teeth
323 (Ionescu et al., 2020; Vernon et al., 2021). Vernon et al (Vernon et al., 2021) analysed the influence of
324 mitigation strategies (such as rubber dam and aspiration) on aerosol loading and CFU on settling plates,
325 and contrasted the spread from high-speed air rotors with lower speed electric drills. While these
326 quantitative measurements highlighted the importance of drill speed, air and availability of saliva on the
327 dispersal process (Holliday et al., 2021) the exact mechanisms by which infectious agents are removed
328 and subsequently dispersed were not thoroughly analysed.

329
330 The airborne spread of infection is fluidic in nature and relies on how infectious materials that are mostly
331 embedded within adherent biofluid films are removed, enter the air and are being dispersed. Therefore,
332 assessing the risk of disease transmission from powered medical instruments first requires a thorough
333 understanding of the removal and dispersal processes that are involved during instrument interaction
334 with adherent biofluid films. Indeed, the small size of bacteria and especially viruses compared to a large
335 body of fluid means that they move with the fluid and as such, tracking the fluid can provide an
336 appropriate proxy for following transport of infectious agents. Consequently, analysis of the removal and
337 dispersal of the adherent biofluid films using dye techniques or airborne particle counters (as in our
338 study) can provide a valid approximation to evaluate the spatiotemporal spread of infectious agents.

339
340 Our laboratory tests identified three independent mechanisms for removing biofluid films: mechanical
341 due to vibrating/rotating surfaces, aerodynamic caused by air flow and hydrodynamic caused by water
342 flow or droplet impact. Our bright-field visualisations of the deposition and aerosol measurements (Figs
343 2 and 4), support the view that the aerosol cloud generated during dental procedures mostly flow as a
344 current (with an estimated velocity of ~ 0.08 m/s) and continuously settles, as it moves along the surface
345 and is not dispersed randomly unless there exists turbulence in the air (generated externally for example
346 by ventilation systems or the movement of people).

347
348 Our data confirmed that the operation of the air-rotor has the higher ability to potentially remove and
349 disperse the infectious agents (Fig 1, SI Figs 2, 3). As summarised in Fig 5b, three different mechanisms
350 appear during the operation of an air-rotor with the high-speed rotation of the bur and its interaction with
351 film having the greatest potential for film removal and the subsequent dispersal. Depending on the
352 geometrical constraints and operational orientation, the three mechanisms may engage with each other
353 in an additive or subtractive fashion to remove and disperse potentially infectious biofluid film. For
354 instance, when the bur interacts with a flat surface (Fig 2), the air flow may act to deplete the fluid film
355 or reduce the thickness of the fluid film near at the bur which decreases the amount of potentially
356 infectious biofluid to be exposed to the drill and turn to fine aerosols through mechanical interactions.

357 However, when the bur is located in an occlusal pit (Fig 4), the air flow may enhance the removal process
358 and splash generation by exerting high shear forces on the film and particularly at the cusp of the teeth
359 edge (Fig 5c). Furthermore, while we designed the experimental configurations to enhance the removal
360 of the biofluid film coating the teeth, only a small fraction of the adherent layer was observed to be
361 removed and deposited over a short distance (Fig 5c).

362
363 There is a growing number of practical methods to reduce the potential of infectious agents spread in
364 dentistry, including reducing the availability of biofluids through the use of dams (Fine et al., 1993),
365 application of suction devices to remove aerosol-laden close to the point of generation (Vernon et al.,
366 2021). Hassandarvish et al (Hassandarvish et al., 2020) have shown that mouthwash can reduce viral
367 load in biofluids in a laboratory setting. Previous measurements reported the viscosity of the human
368 saliva to be at least twice the water viscosity (CE et al., 2000). Consequently, our experiments on the
369 human saliva (Fig 3b) indicate that increasing the viscosity of a biofluid (by replacing water with human
370 saliva) suppresses removal mechanisms, which are especially important during the start of drilling.
371 Other groups have suggested changing the rheological properties of the coolant water to reduce
372 aerosolisation (Plog et al., 2020). Therefore, manipulating the rheological properties of the fluids
373 (biofluids and coolants) involved during powered medical procedure are among other possible ways to
374 suppress the aerosol generation.

375
376 In summary, our work provided a mechanistic view of the general processes of biofluid film removal and
377 dispersal by powered medical devices, specifically in the context of dentistry. This provides an important
378 steppingstone to understand and propose mitigation strategies to reduce the risk of the spread of
379 airborne infection.

380 381 **Limitations of study**

382
383 Our study focuses on modelling infectious agent spreading employing the dye technique and thus no
384 specific microbiota or virus was used in our study. However, one limitation of such techniques is their
385 inability to predict the levels of infectivity of the dispersed biofluid precisely. Indeed, infectious agents
386 embedded within the fluid body may get inactivated by heat or mechanical forces generated in drilling
387 or desiccation following droplet evaporation.

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400 401 **Author contributions**

402 IE and EM designed and performed the majority of the experiments. RB, MH and HD helped with laser
403 sheet imaging. FD, YLN, KG, JK and JS helped with the clinical setting experiments. SS, SW and NS
404 helped with laboratory setting experiments. EM analysed the data with helps from IE and YJ. IE and EM
405 wrote the manuscript with contributions from FD, YJ and JS. All authors discussed the results and
406 commented on the manuscript.

407 408 **Declaration of interests**

409 The authors declare no competing interests.

Journal Pre-proof

410 **Figure titles and legends**

411
412 **Figure 1 Droplet generation capacity of different dental devices.**

413 **(a)** High speed photography (5000 frames per s) of aerosol and droplets showing an instantaneous view
414 (left images) and a maximum projection (right images) of 100 image sequences (corresponding to 20
415 ms) highlighting the trajectory of the dispersed phase. The panels (from top to bottom) correspond to
416 air-rotor, air-rotor (without burr), low-speed drill with external 3-in-1 coolant jet and back-exhaust,
417 ultrasonic scaler, low-speed drill and 3-in-1. The red, green and blue arrows show regions that aerosols
418 (<20 microns), fine droplets (20-200 microns) and droplets (>200 microns) were generated, respectively.
419 **(b)** Characterization of the spray dynamics for air-rotor (left panel) and ultrasonic scaler (right panel).
420 The velocity contours were estimated by tracking individual droplets using PTV. Scale =1 cm. **(c,d)** The
421 distribution of droplet diameter and speed for the air-rotor and ultrasonic scaler. **(e)** The size and velocity
422 of individual particles were combined to estimate the distribution of the Weber number. **(f)** Regime
423 diagram showing the characterisation of different dental instruments according to their potential to
424 generate aerosols, fine droplets and droplets expressed through the movement of a liquid jet (II_1) or by
425 mechanical agitation (II_2). The inset images are taken from the regions specified by red, green and blue
426 squares in (a).

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430 **Figure 2 Removal and dispersion of the adherent biofluid film in laboratory setting.**

431 **(a)** Schematic of the laboratory setup used to analyse the mechanisms of layer removal with photos
432 taken from the above and the air sampler located 40 cm away from the dish. **(b)** Top view images
433 showing the temporal evolution of dye deposition (from either dyed coolant water or dyed fluid layer)
434 due to air-rotor handpiece operating under different modes. Scale=10 cm. **(c)** Effects of fluid layer
435 thickness on the temporal evolution of dye deposition due to a drill engaging with the layer. Scale=10
436 cm. Arrows in (b) and (c) point to the regions that small amount of deposition was detected. **(d-g)** Air
437 particle count (sampled continuously over 5 min at 40 cm distance from the drill by a particle counter
438 that measured cumulative particle count every 5s in 0.2 liters of air) as a function of time and under a
439 variety of drill-air-water configurations. To estimate the baseline, the particle counter ran for 0.5 min
440 prior to the operation of the drill. Then while particle counter was continuously running, the air-rotor
441 handpiece was operated for 2 min. After the drill operation was stopped, the particle counter was kept
442 running for additional 2.5 min. **(d)** Measurements of air particle counts (0.3 μ m and 5 μ m inset) when air-
443 rotor was operated on a thick layer of water film under three operating condition of air only (mode-2 but
444 without the drill), drill with air only (mode-2) and drill only (mode-3). **(e,f)** Measurements of air particle
445 counts (0.3 μ m in (e) and 5 μ m in (f)) when air-rotor was operated on a layer of water film under three
446 operating condition of drill only (mode-3), air with water only (mode-1 but without drill) and drill with air
447 and water (mode-1). The insets are the zoom of the first 1.5 min. **(g)** Influence of the layer thickness on
448 0.3 μ m particle count (5 μ m inset) in mode-3 (drill rotation with inhibited expulsion of air and water). **(h)**
449 Simultaneous measurements of particle count (0.3 μ m) at 0.4m and 2.0m from drill only (see SI Fig 1f
450 for 5 μ m particles and SI Figs 1g-h for individual unaveraged curves). The curves in (d and g) are the
451 smoothed data from individual measurements representing the trend. The curves in (e), (f) and (h) are
452 the smoothed, averaged of data from three independent experiments with shades indicating the
453 standard error.

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456 **Figure 3: The dynamics of biofluid film removal and influence of fluid properties.**

457 **(a,b)** High-speed images (7000 frames per s) capturing the dynamics of bur rotating at ~20000 rad/s
458 and engaging with water or unstimulated saliva (from a human participant) droplet/film. In **(a,b)**, the top
459 images are the instantaneous single snapshots, the middle images were created by overlaying single
460 snapshots over a period of time. The bottom images were created by colour coding single snapshots
461 and overlaying on top of one another. **(a)** The removal of a droplet of water (~ 2 mm in diameter) collated

462 in three sequences: acceleration of bur (0-0.055s), steady full-speed rotation of bur (0.085-0.13s),
 463 leading to full removal of droplet/film (0.14-0.195s). Scale=5 mm. **(b)** The interaction of a thin film of
 464 saliva (~500 μm thickness) with bur is shown at different stages (0-0.22s, 0.300-0.52s and 0.86-1.08s)
 465 after the start of the drill rotation at $t=0$. Scale=5 mm.
 466

467

468 **Figure 4: Assessment of biofluid film removal and dispersal in clinically relevant conditions.**

469 **(a)** The removal of a fluorescently dyed simulated saliva layer by an air-rotor handpiece operating on
 470 model teeth. The teeth (45-47 ISO 3950) were coated with dye, the drill engaged with tooth 46 and the
 471 coolant water was undyed. Using fluorescent lamp, the dyed droplet splatters were detected up to 10 cm
 472 away from the model teeth. The blue arrow points to the regions of undyed water deposition and the yellow
 473 arrows indicate the regions that dyed flow or small deposition could be detected. Scale=1 cm. **(b)** The
 474 dispersal of coolant water (dyed with red food colouring) from air-rotor handpiece operating on model teeth.
 475 Arrows point to the regions that small amount of deposition could be detected. **(c-g)** Measurements
 476 conducted in a simulated clinical setting with a dentist who performed procedures on a phantom head
 477 (located on a dental chair). The air-rotor handpiece pointed at the buccal cusp of the occlusal surface of
 478 teeth 35, while teeth 34-38 were coated with fluorescently dyed simulant saliva layer. **(c)** Image
 479 sequences show the dilution of the simulant saliva loaded with fluorescein dye. The arrows point to the
 480 tip of the bur. Scale= 1 cm. **(d)** Intensity profiles at two locations on the teeth (areas located close to the
 481 tip and ~ 1 cm away from the tip) rapidly decayed due to dilution by the coolant water. **(e)** The local
 482 splatter pattern imaged on the surface of the manikin's face (located ~20 cm away from drilling point)
 483 after a 3 min continuous drilling procedure. **(f)** A small number of fluorescent particles (around ~200 μm
 484 diameter) were detected in areas up to 0.5 m away from the drilling point. Fluorescent imaging was used
 485 to scan the tracer Petri dishes distributed up to 2m away from the head. Scale=1 mm. **(g)** Total mass
 486 loading in the air for sub 2.5 micron particles during the operation of either air-rotor handpiece in fixed
 487 position or the ultrasonic scaler operating in fixed/ static or moving conditions. The inset is the zoom of
 488 the dotted area.
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494 **Figure 5: Summary of the potential risks involved in transmission of infectious agents and the**
 495 **critical mechanisms for the removal and dispersal by powered instruments in dentistry.**

496 **(a)** Collage showing the production of aerosols and droplets by powered mechanical devices in the
 497 dental setting and their link through contact and airborne transmission routes. **(b)** Insert images show
 498 the influence of either drill, drill/air and drill/air/water on the removing dyed simulant saliva. Schematic
 499 shows the removal of adherent layers through three mechanisms: mechanical (moving or vibrating
 500 surface), aerodynamic (due to the air movement) and hydrodynamic (impact of droplets or movement
 501 of water). Color contours and arrows show the qualitative comparison between the levels of the shear
 502 stresses generated by different mechanisms. **(c)** The interaction between an air-rotor and teeth covered
 503 with dyed simulant saliva under conditions of drill, drill+air and drill+air+water. These panel were used
 504 as insets in (b).
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508 **Movies 1-10:** Top view videos (related to Figure 2) showing the interaction of air-rotor with a water layer.
509 **Mode-1:** The normal operating condition which involves rotation of the bur driven by the air jet while
510 allowing the expulsion of both the air and coolant jet.

511 **Mode-2:** Rotation of the bur driven by the air jet allowing the expulsion of the air while water expulsion
512 was inhibited.

513 **Mode-3:** Rotation of the bur driven by the air jet while both air and water jet expulsion were inhibited.

514

515 Investigating effects of air flow:

516 **Movie 1:** Operation in mode-2 but with removed bur on dyed layer.

517 **Movie 2:** Operation in mode-2 on dyed layer.

518 **Movie 3:** Operation in full condition (mode-1) on dyed layer.

519

520 Investigating droplet splashes from water/air jet:

521 **Movie 4:** Operation of drill only (mode-3) on dyed film (~3mm thickness).

522 **Movie 5:** Operation in full condition (mode-1) with the flow of dyed water on undyed layer (~3mm
523 thickness).

524 **Movie 6:** Operation in full condition (mode-1) with the flow of dyed water on dish with no water layer.

525 **Movie 7:** Operation in mode-1 but with removed bur and flow of dyed water on undyed layer (~3mm
526 thickness).

527

528 Investigating effects of layer thickness:

529 **Movie 8:** Operation of drill only (mode-3) on thin dyed water layer (less than ~0.5mm thickness).

530 **Movie 9:** Operation of drill only (mode-3) on medium dyed water layer (~1mm thickness).

531 **Movie 10:** Operation of drill only (mode-3) on thick dyed water layer (~4mm thickness).

532

533

534 **Movies 11-16:** Videos (taken with high-speed camera) of the bur engaging with water (movies 11-13,
535 Fig 3a) or unstimulated saliva collected from human participant (movies 14-16, Fig 3b) droplet/ film.

536

537 **Movie 17:** Video of the interaction of the air-rotor (operating in mode-1) with model teeth coated with
538 fluorescent dye. The video was taken using fluorescent lamp.

539 **Movie 18:** Video of the interaction of the air-rotor (operating in mode-1 and the coolant water dyed by
540 high concentration of food dye) with model teeth. The video was taken using a bright field light.

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560 **STAR Methods**

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562 **Resource availability**

563 **Lead contact**

564 Further information and requests for resources and data should be directed to the lead contact,
565 Professor Emad Moeendarbary (e.moeendarbary@ucl.ac.uk).

566

567 **Materials availability**

568 This study did not generate new unique reagents.

569

570 **Data and Code Availability**

571 Additional Supplemental Items are available from Mendeley Data at
572 <https://dx.doi.org/10.17632/v9px86xh8w.1> or

573 [https://data.mendeley.com/v1/datasets/v9px86xh8w/draft?a=553d231d-9f66-46fa-b0d0-](https://data.mendeley.com/v1/datasets/v9px86xh8w/draft?a=553d231d-9f66-46fa-b0d0-d50c3695d716)
574 [d50c3695d716](https://data.mendeley.com/v1/datasets/v9px86xh8w/draft?a=553d231d-9f66-46fa-b0d0-d50c3695d716)

575

576 **Experimental Model and Subject Details**

577 No experimental model or human subject was used in this study.

578

579

580 **Method details**

581

582 The experimental tests were performed at the Royal National ENT and Eastman Dental Hospitals,
583 UCLH. Two identical dental suites were used for the tests, with each serviced by clean air entry of 10
584 air-changes per hour (ACH) giving a potential air replenishment time of 6 minutes. Assessments in a
585 simulated clinical setting were conducted on a phantom head with upper and lower dental arch
586 containing 32 teeth. For the laboratory tests (performed in the UCL Environmental Fluid Mechanics
587 Laboratory, Roberts Building), the instruments were analysed in isolation on a surface adjacent to a
588 dental chair. The instruments used in this study are listed in SI Table 1.

589

590 **Brightfield imaging**

591 For the analysis of the sprays in Fig 1a, the instruments were held in position over a sink, illuminated
592 with strong diffuse lighting and recorded using a high-speed CMOS camera colour camera (Phantom
593 VEO710, Photron Inc, US). The images were recorded at a rate of 5000-7500 frames per second.

594

595 For imaging dynamics of biofluid film removal (Fig 3), approximately 100-200 μ l of either water or the
596 unstimulated saliva

597 collected from the human participant was laid on a flat metallic surface using small pipette tip. The
598 interaction of the drill with the water or saliva layer was investigated using high speed camera (Phantom
599 VEO710, Photron Inc, US) with the imaging speed of 5000-7500 frames per second. The procedure of
600 unstimulated saliva collection involves resisting the swallowing of the participant's saliva and spitting
601 into a small test tube every 20s for 2 min to collect approximately 2ml saliva.

602

603 **Laser sheet imaging**

604 Spray dynamics in Fig 1b was captured using laser illuminated Mie scattering technique with a laser
605 vertical plane bisecting the dental instrument cross-section. This technique leads to capturing a much
606 lower aerosol/droplet density in the image and is capable of measuring the velocity of individual droplets.
607 High speed imaging of the spray was carried out using a 1000 mW, 515.3nm continuous diode laser
608 (Genesis MX514-1000 SLM OPS Laser-Diode System) and a high-speed CMOS camera. The camera
609 was fitted with a 100 mm lens producing an imaging window size varying from 60 x 100 mm to 30 x 50
610 mm. The images were captured at a frame rate of 5000-7500 Hz. The region of interest was set to
611 observe the near field spray characteristics. An ultrasonic scaler, high speed air-driven drill and 3-in-1

612 air-water syringe were used to generate sprays. TSI Insight 4G software was utilised to capture the
613 images.

614

615 **Fluorescein imaging**

616 The very small size of the viruses and low diffusivity of bacteria within a body of fluid mean that a dye
617 model is an appropriate tool for tracking the infectious biofluid transport. Furthermore, the dye technique
618 affords high spatial resolution in terms of tracking which enables the mechanistic view of the dispersal
619 processes to be unpicked.

620

621 To examine the mechanism of saliva removal, a series of tests were performed using a phantom head.
622 A simulant saliva (Biotene Oral Balance) was well-mixed with sodium fluorescein salt (0.625 mg/ml in
623 simulant-saliva), applied over the teeth and illuminated using a UV lamp. Two types of fluorescein tests
624 were applied. In the first, dyed simulant-saliva was applied to the teeth (34-38 ISO 3950) of a manikin
625 by a dentist and an NSK air-rotor applied to tooth 35 with the drill in contact with the buccal cusp of the
626 occlusal surface. The drill was applied at about 30 degrees from the horizontal. In the second, dyed
627 simulant-saliva was applied to teeth 45-47 of a full-teeth model and an air-rotor applied to the occlusal
628 surface of teeth 46 by an experimental professor. The air-rotor was aligned 10 degrees from the vertical
629 plane.

630

631 The fluorescein salt concentration was initially extremely high that it absorbed light when applied to the
632 teeth (appeared as dark green), but strongly fluoresced during dilution and in the presence of UV light.
633 To capture the potential for simulant-saliva removal, a series of petri dishes were placed at distances
634 20, 40 and 80 cm from the head (in the chest direction) and 40 cm above the head. All the dishes were
635 in the same plane. Prior to each test, the lids of the dishes were removed and replaced 10 minutes after
636 the test started. Each test consisted of 200s continuous operation of the instruments. Photographs and
637 videos of the experiments were recorded and analysed after the tests. During these tests, air was
638 monitored for aerosol concentration PM2.5 and PM10 using a Temtop M2000 (Elitech) which was
639 placed next to the phantom head. During the tests, 4 people present in the room to control the various
640 components of the experiments.

641

642 **Deposition tests**

643 Splatter tests involved the interaction between powered instrument and a layer of fluid (water in Fig 2 or
644 simulant saliva in Figs 4 and 5). The instrument was held in position and in contact with the centre of a
645 60 mm petri dish, placed onto a A0 sheet of white craft paper and a camera affixed above tests. Either
646 the coolant water or the water layer was dyed using rhodamine dye (Merck Life Science, UK) and
647 illuminated by a diffuse light source. Continuous videos were taken using a visualiser (IPEVO V4K Ultra
648 HD) installed on top of the instrument.

649

650 **Air sampling**

651 During the fluorescein imaging tests within the hospital, air was sampled for a period of 10 minutes using
652 a Temtop M2000 (Elitech). The device was placed adjacent to the phantom head and at the same level
653 as the settling plates. The PM2.5, PM10 and particle count levels were recorded during the tests and
654 for periods after the tests. Ventilation brought filtered air into the room and the 10 ACH for the room
655 meant that the air born particle load was low; therefore, there is no need for the baseline subtraction
656 and the raw data is plotted in Fig 4g.

657

658 During the splatter deposition tests in the laboratory, air was sampled using two Fluke 985 (Fluke, US)
659 airborne particle counters that were placed flat on a workbench (pointing towards the petri dish) with a
660 distance of 0.4 m and 2.0 m away from the dish. The ventilation system delivered unfiltered air to the
661 laboratory (Roberts Building, UCL) and the additional components due to the local sources of aerosols
662 was eliminated by subtracting the background concentration (SI Fig 1).

663

664 **Quantification, reproducibility and image analysis**

665 Experiments were repeated at least three times and plots and images are representative of at least
 666 three independent tests. For experiments in Figs 2e,f,h, the curves from three independent experiments
 667 were smoothed then averaged and the standard error was calculated as indicated by the shades.
 668 Individual curves related to effects of distance is presented in SI Fig. S1 g, h, which show the lag time
 669 due to distance more clearly. Quantification and plotting were performed in MATLAB (Mathworks) or
 670 Origin (OriginLab). Commercially available software, Imaris (BitPlane, South Windsor, CT, USA), was
 671 used to analyse the images. After optimising image volume rendering, spot object tools were employed
 672 to automatically segment and track the droplets. An autoregressive algorithm with a maximum inter-
 673 frame distance of 150 μm and a gap size of 3 μm was used to calculate the position of spots over time.
 674 The droplet sizes was analysed using the 'Analyse Particle' plugin in ImageJ (National Institutes of
 675 Health, USA); this technique was capable of identifying droplet size above 200 microns. The data was
 676 plotted using MATLAB.

677

678 **Scaling analysis**

679 The disruption of a water/air interface through mechanical agitation via a bur or vibrating tip or flow
 680 through a nozzle leads to droplets. The potential for generating an aerosol (<50 μm), fine droplets (50
 681 to 200 μm) or droplets (>200 μm) depends on the magnitude of the forces that act on the water films
 682 and jets and this potential was assessed prior to the experimental study. For a water jet issuing from a
 683 hole with diameter H , at a speed U_j moving through air, the inertial force of the fluid is ρU_j^2 . The potential
 684 for generating large droplets can be assessed by a characteristic measure based on comparing the
 685 surface tension (σ) that stabilises a droplet, and inertial forces, that destabilise the droplet: $\Pi_1 = \frac{\sigma}{\rho U_j^2 H}$,

686 where ρ is the density of fluid This measure is the inverse of the Weber number. When the flow is slow
 687 and inertial forces are weak, Π_1 is large and millimetric droplets are created. When the flow is fast and
 688 inertial forces are large compared to surface tension force, Π_1 is small and an aerosol will be generated.
 689 For moving surfaces with angular velocity ω and length scale δ , a centrifugal acceleration on an
 690 adherent water films scales as $\omega^2 \delta$ and a nominal centrifugal force $\rho \omega^2 \delta^2$ giving a second
 691 dimensionless measure $= \frac{\sigma}{\rho \omega^2 \delta^3}$. The equivalent measure for an instrument vibrating with a frequency
 692 Ω and displacement of the surface Δ is $\Pi_2 = \frac{\sigma}{\rho \Omega^2 \Delta^3}$. The typical surface displacement measurement of
 693 30 μm (Lea et al., 2002). The measures Π_1 and Π_2 form a discriminatory measure and a comparative
 694 measure between different instruments of their potential to generate aerosols and droplets.

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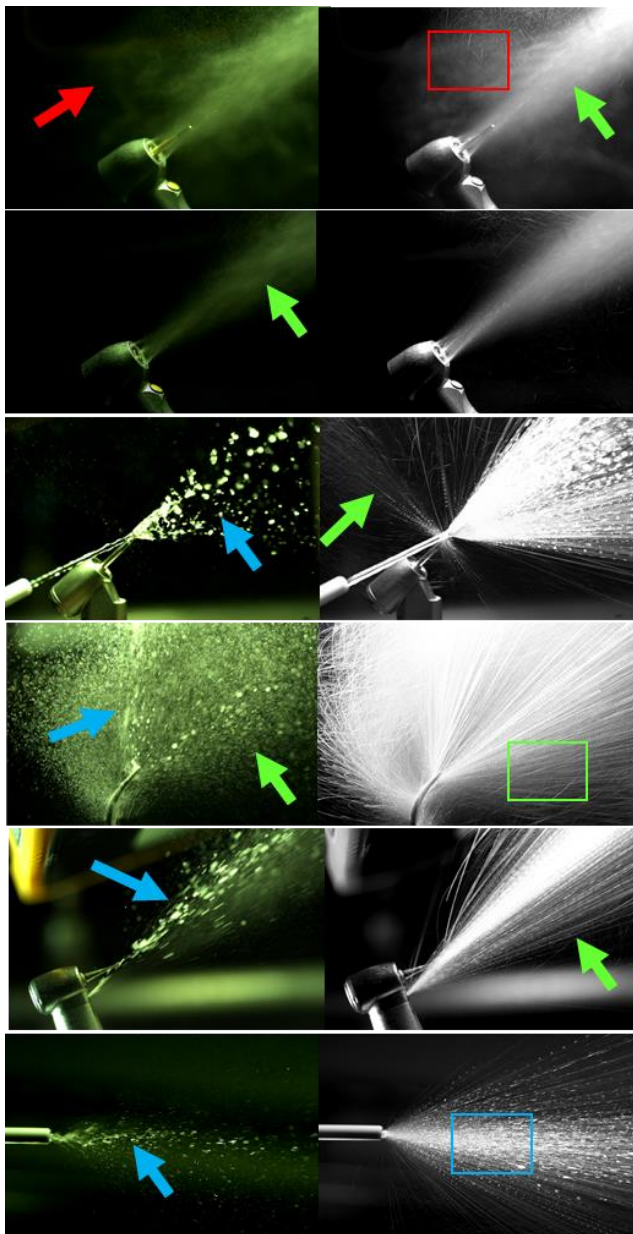
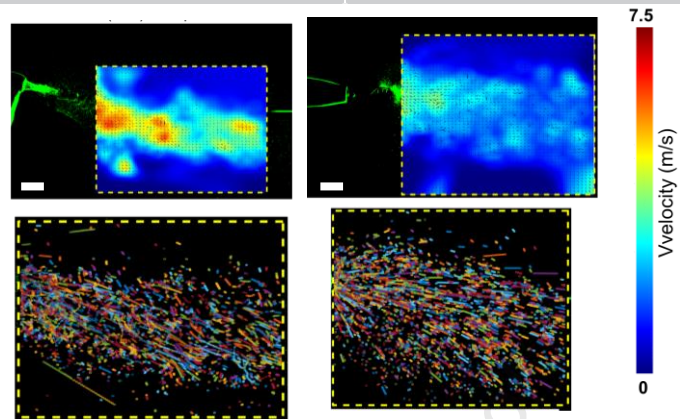
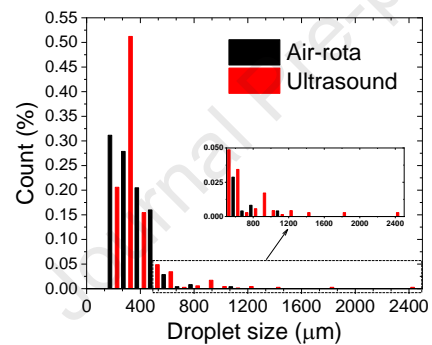
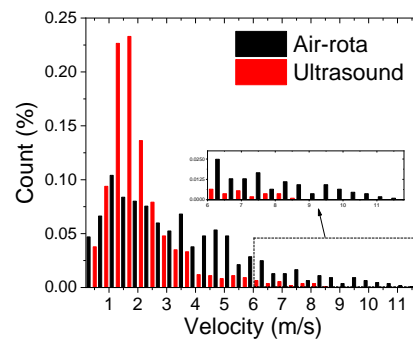
696 **Quantification and statistical analysis**

697 Figures represent averaged or representative results of multiple independent experiments or
 698 simulations. The method section provides details concerning the number of independent experiments.
 699 Analyses were performed using data analysis toolbox in Microsoft Excel or Origin.

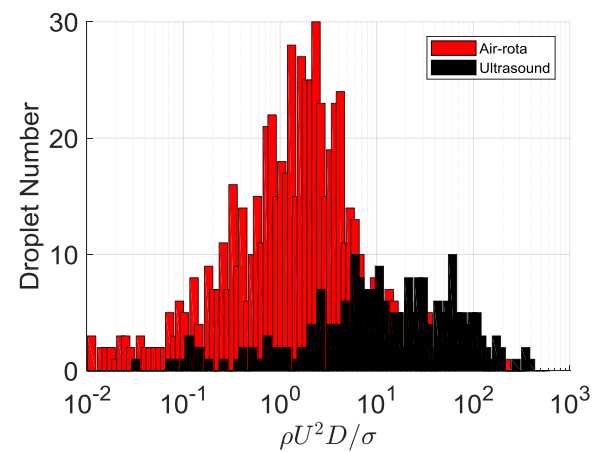
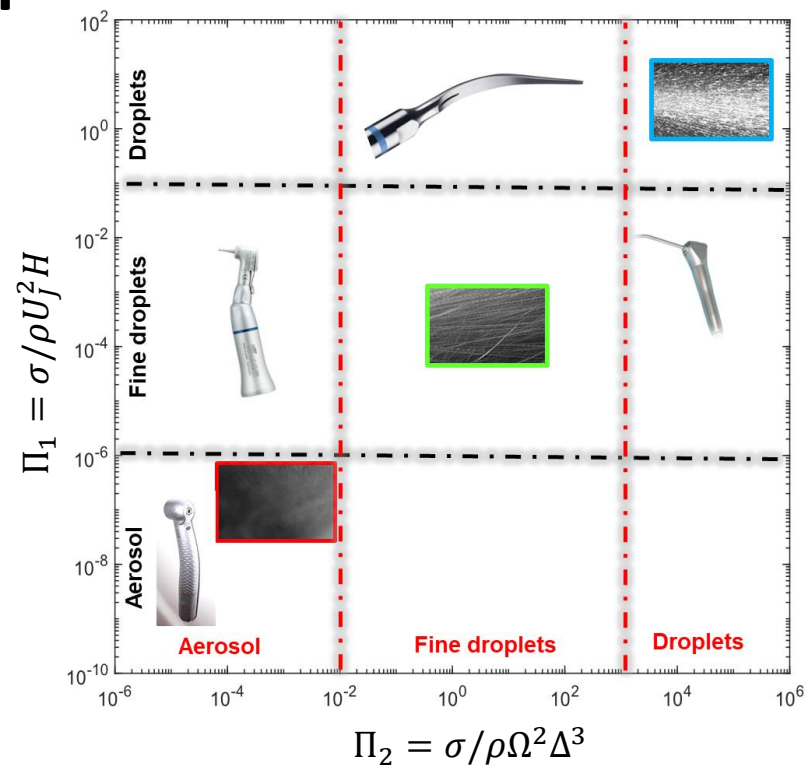
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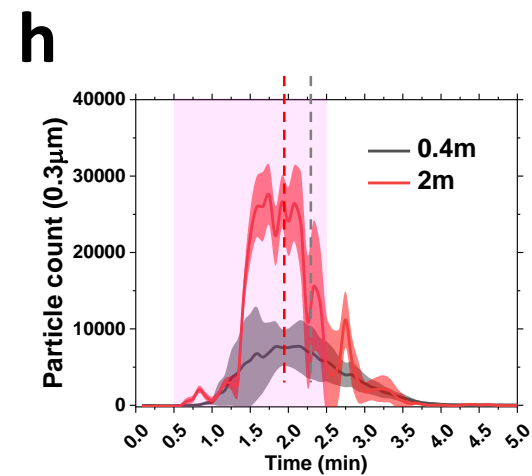
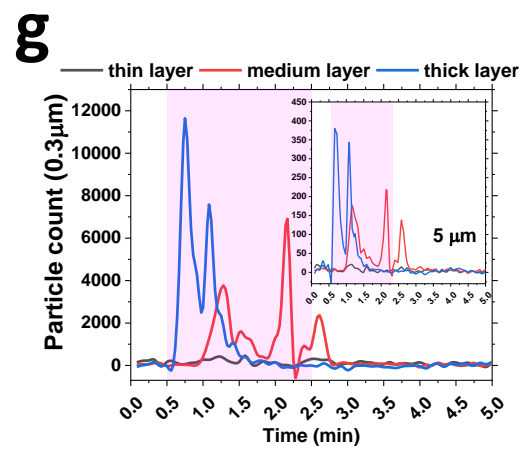
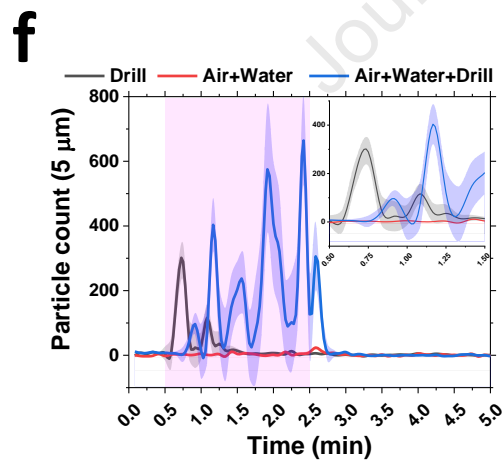
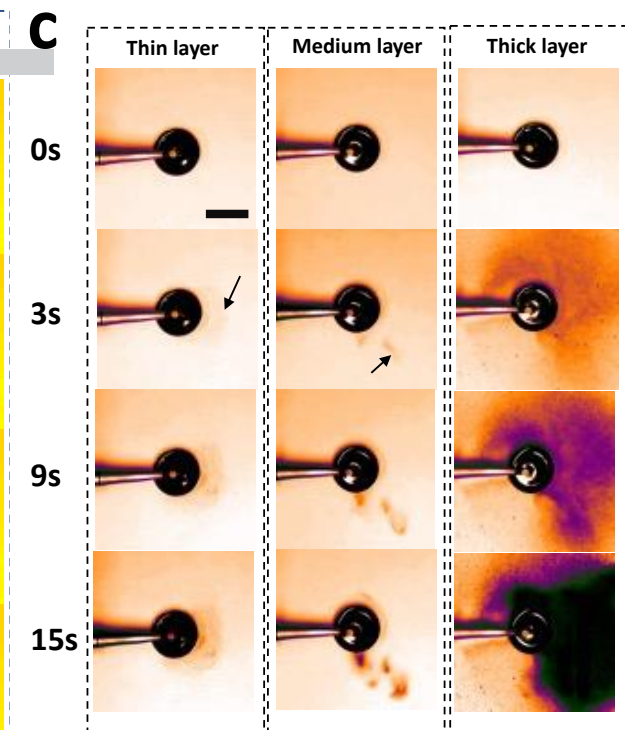
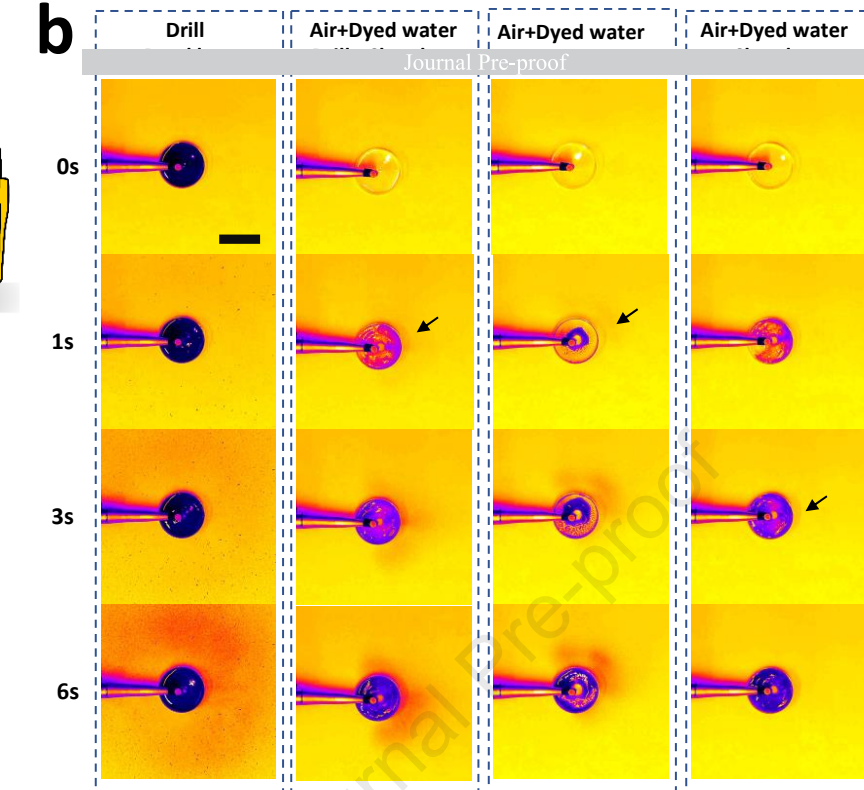
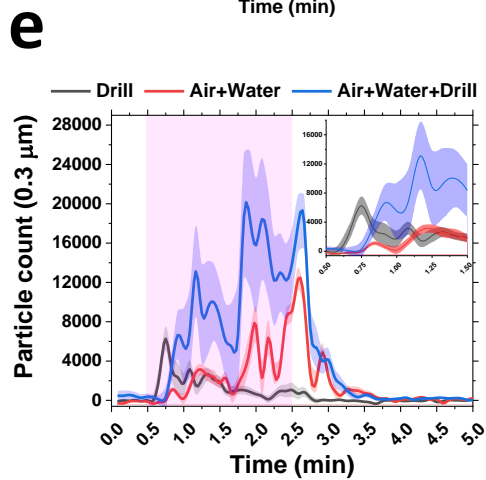
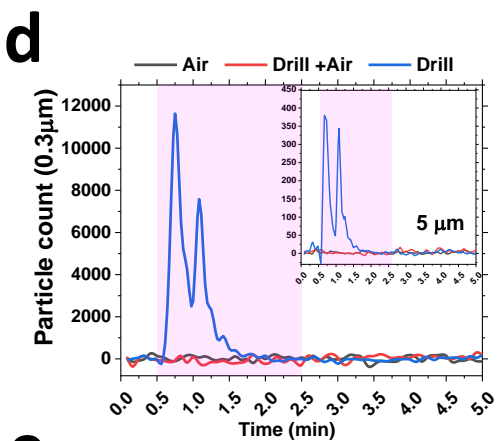
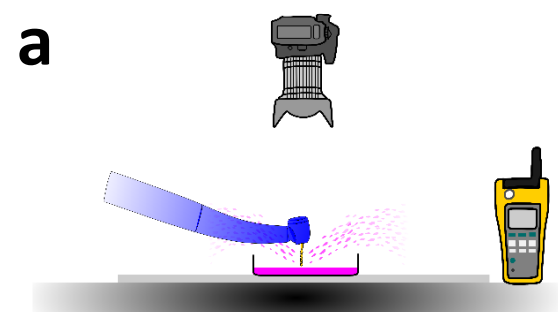
701 **References**

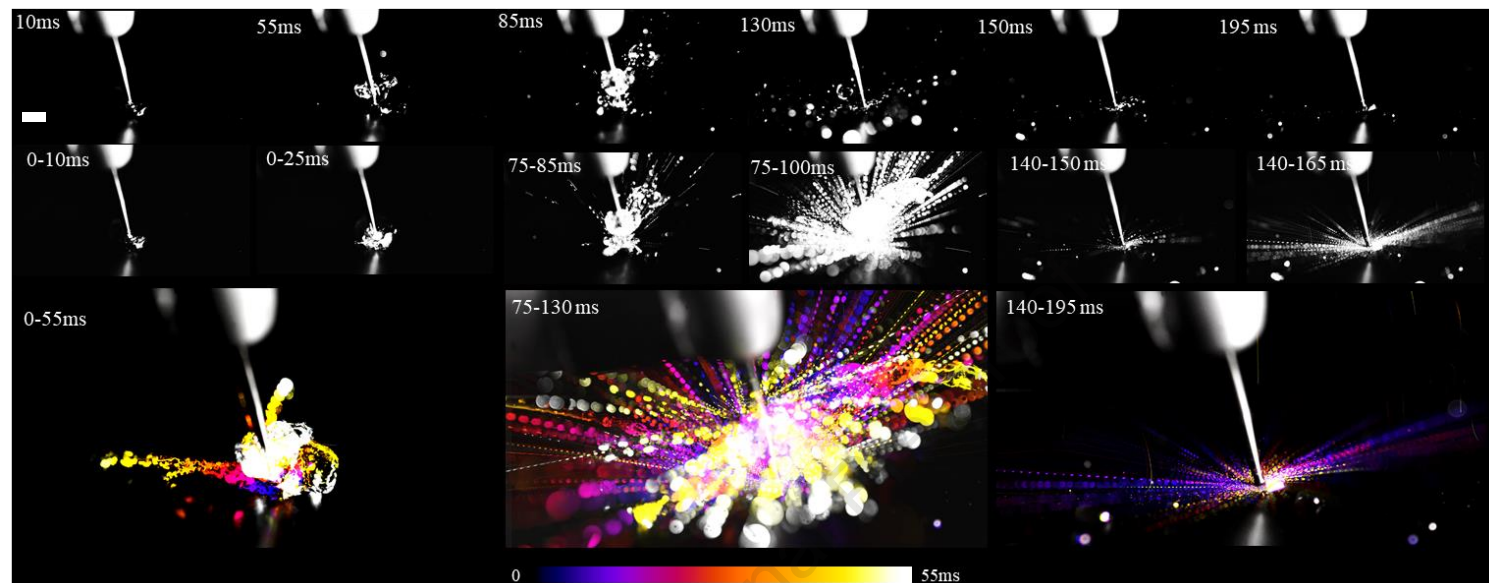
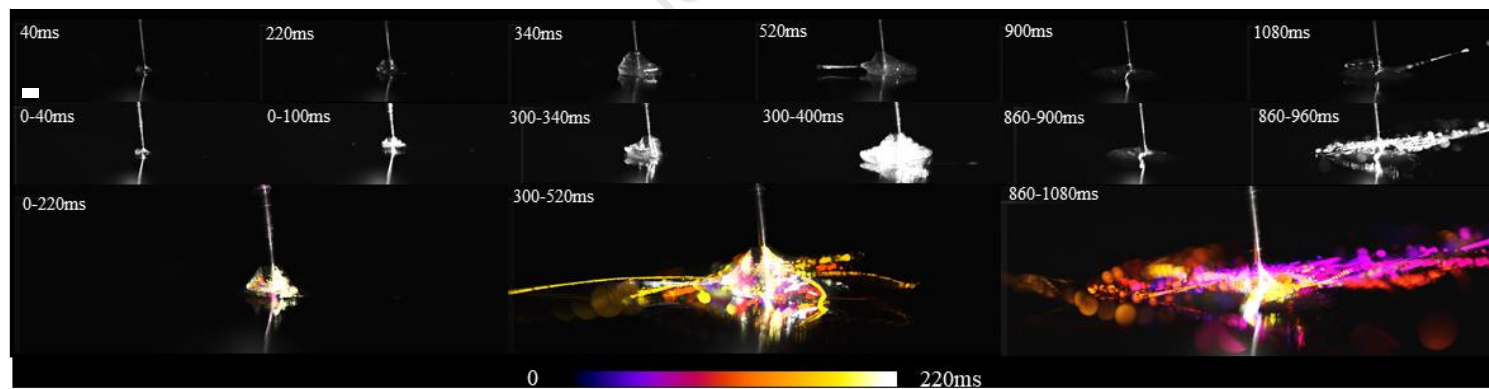
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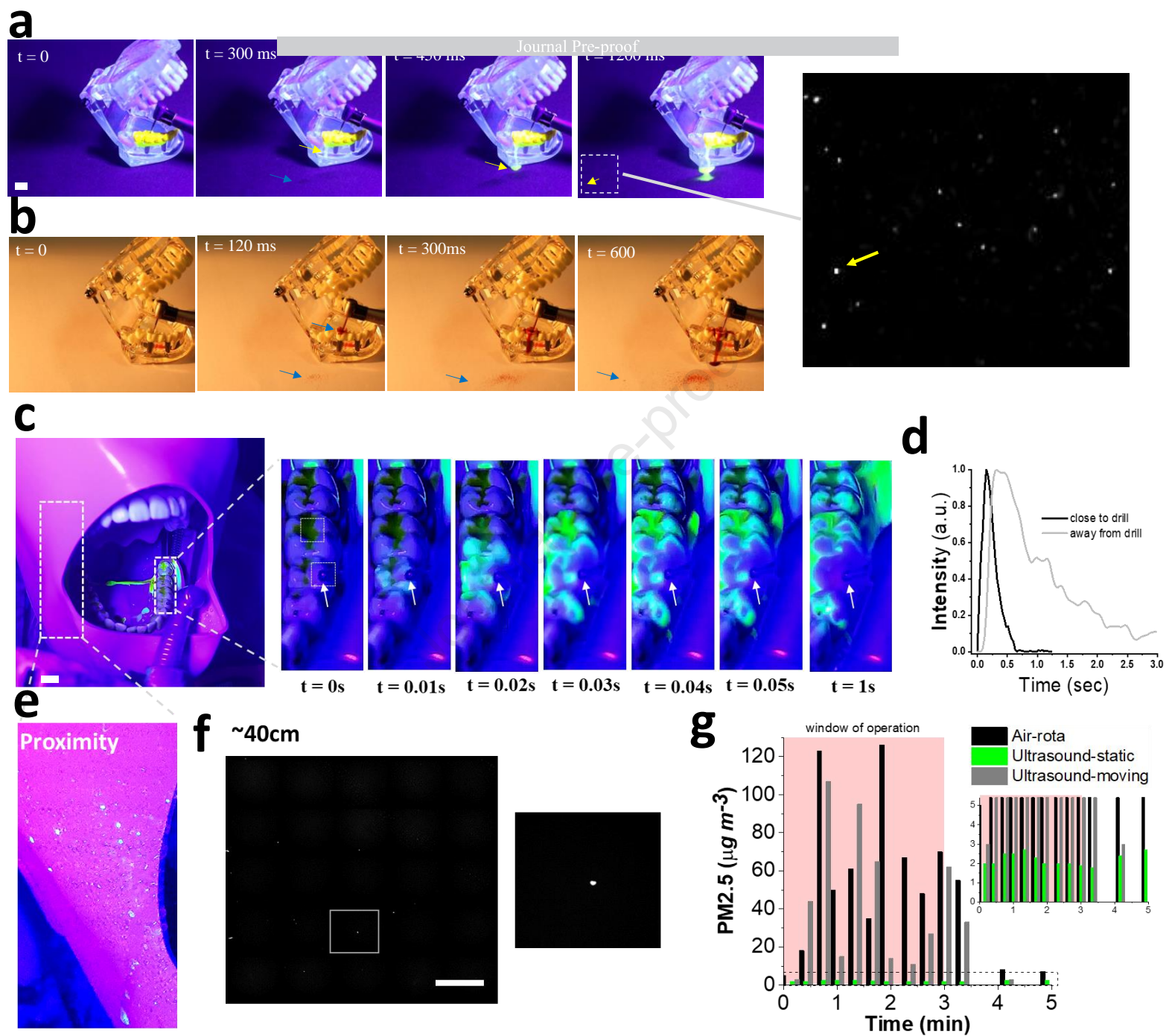
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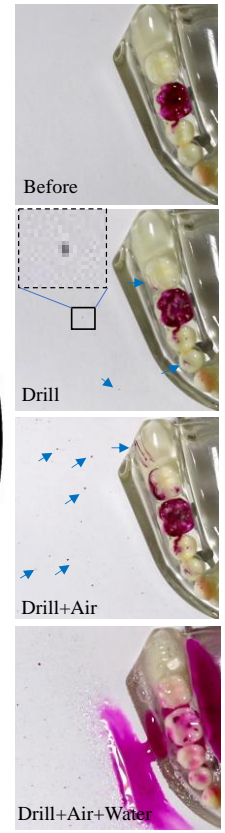
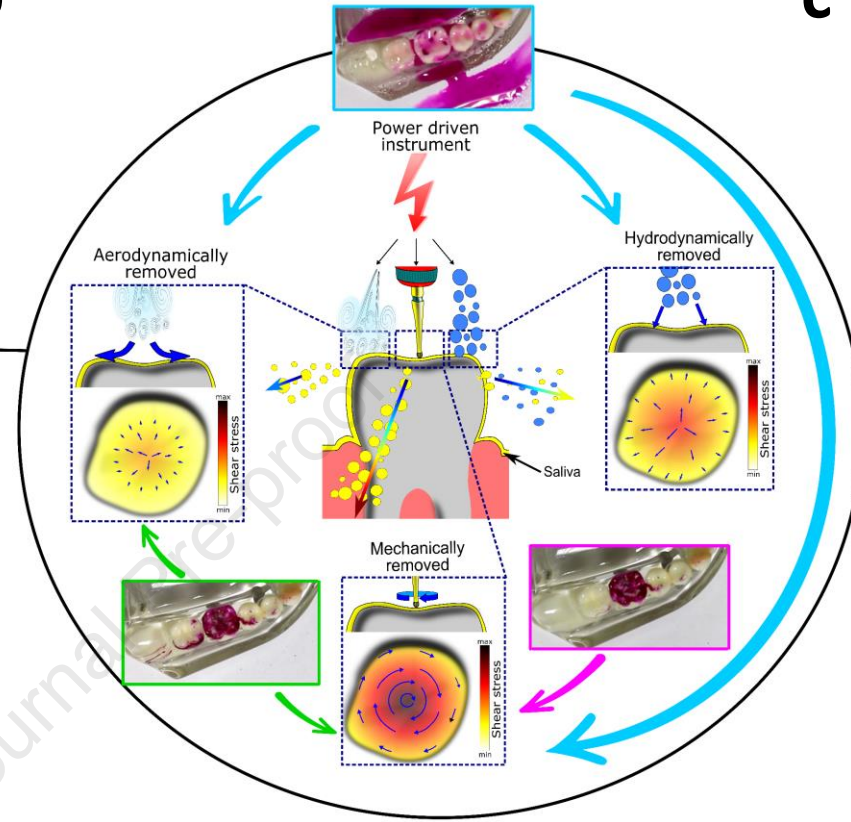
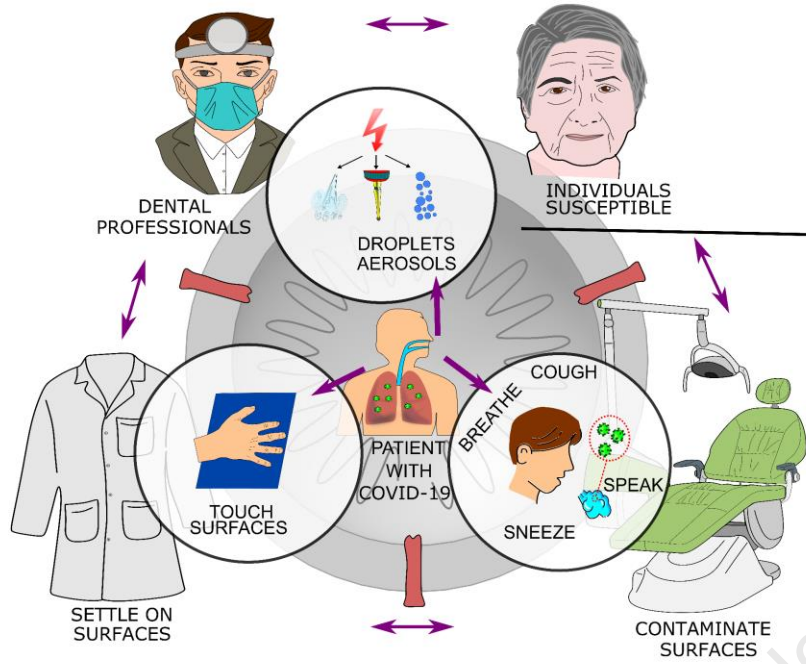
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- Mechanical, hydrodynamic, and aerodynamic forces drive removal/ dispersal processes
- The air-rotor has the highest ability to remove and disperse infectious agents
- The aerosol cloud flows as a current and continuously settles
- Manipulating rheological properties of the fluids can suppress aerosol generation

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