

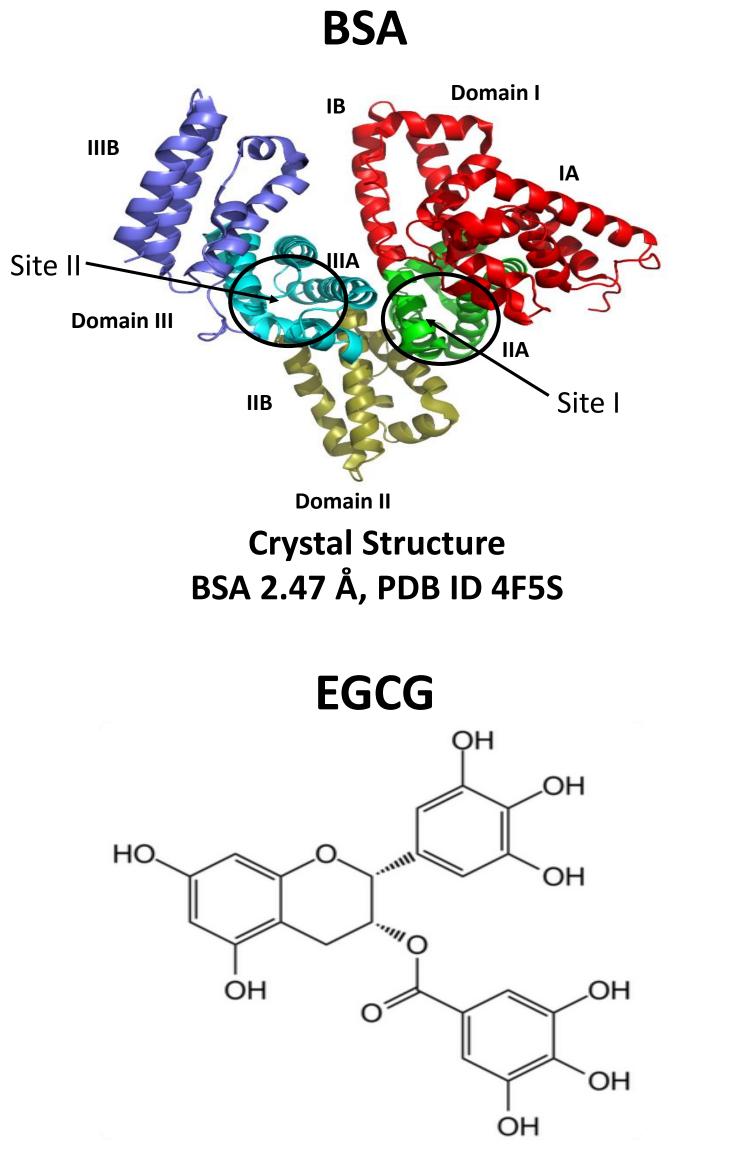


## **Oxidation effects in antiaggregogenic properties of Epigallocatechingallate**

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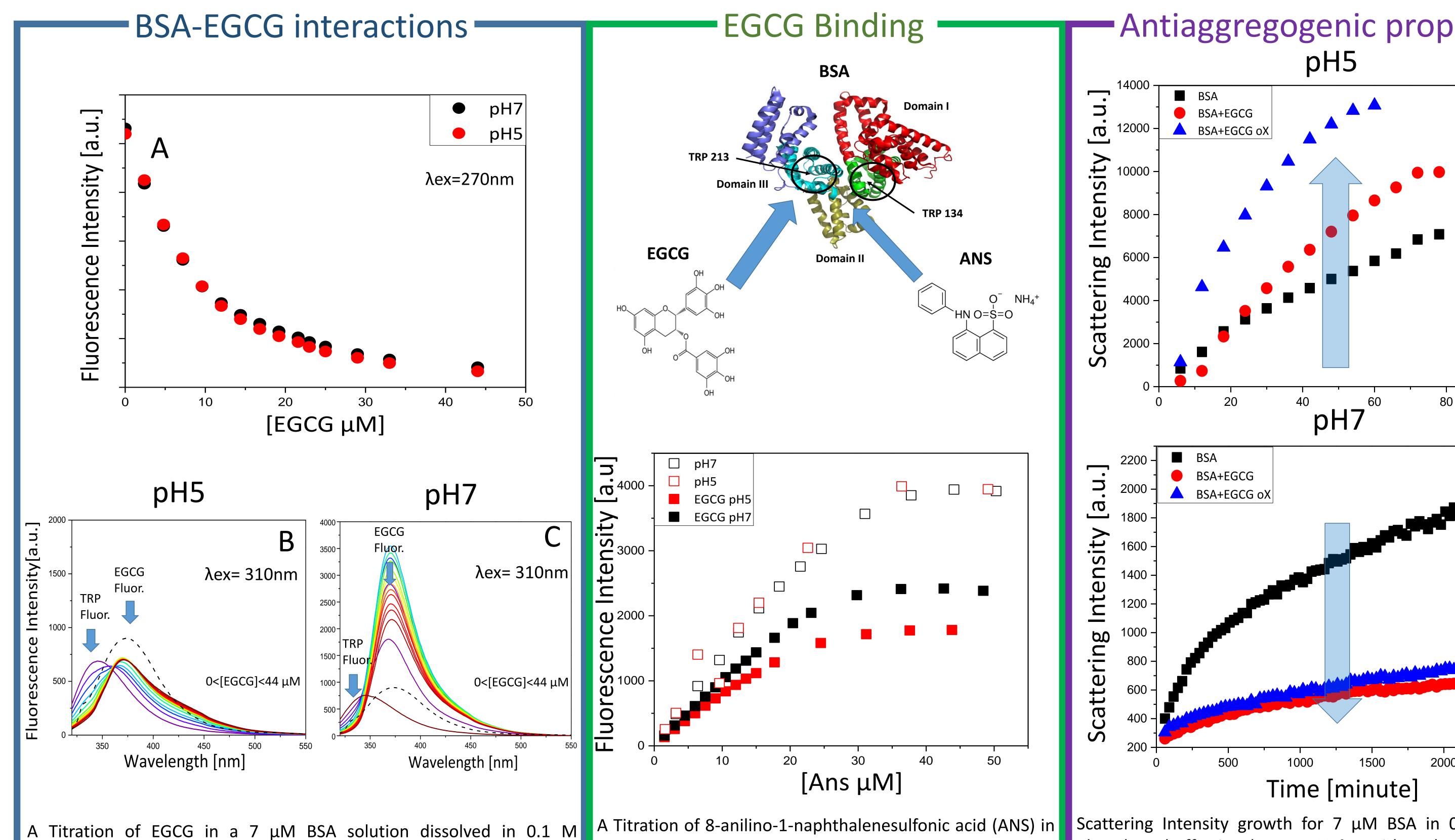
Epigallocatechin-gallate (EGCG), the most abundant flavonoid in green tea, has been extensively studied for its potential in the treatment of amyloid related disorders. This molecule was found to modulate abnormal protein self-assembly, reducing resulting cellular toxicity. EGCG is known to suppress or to slow down the aggregation processes of several proteins, thus supporting the idea that general mechanisms regulate its anti-aggregogenic effects and, interestingly, in the oxidised form it demonstrated an higher efficiency in reducing

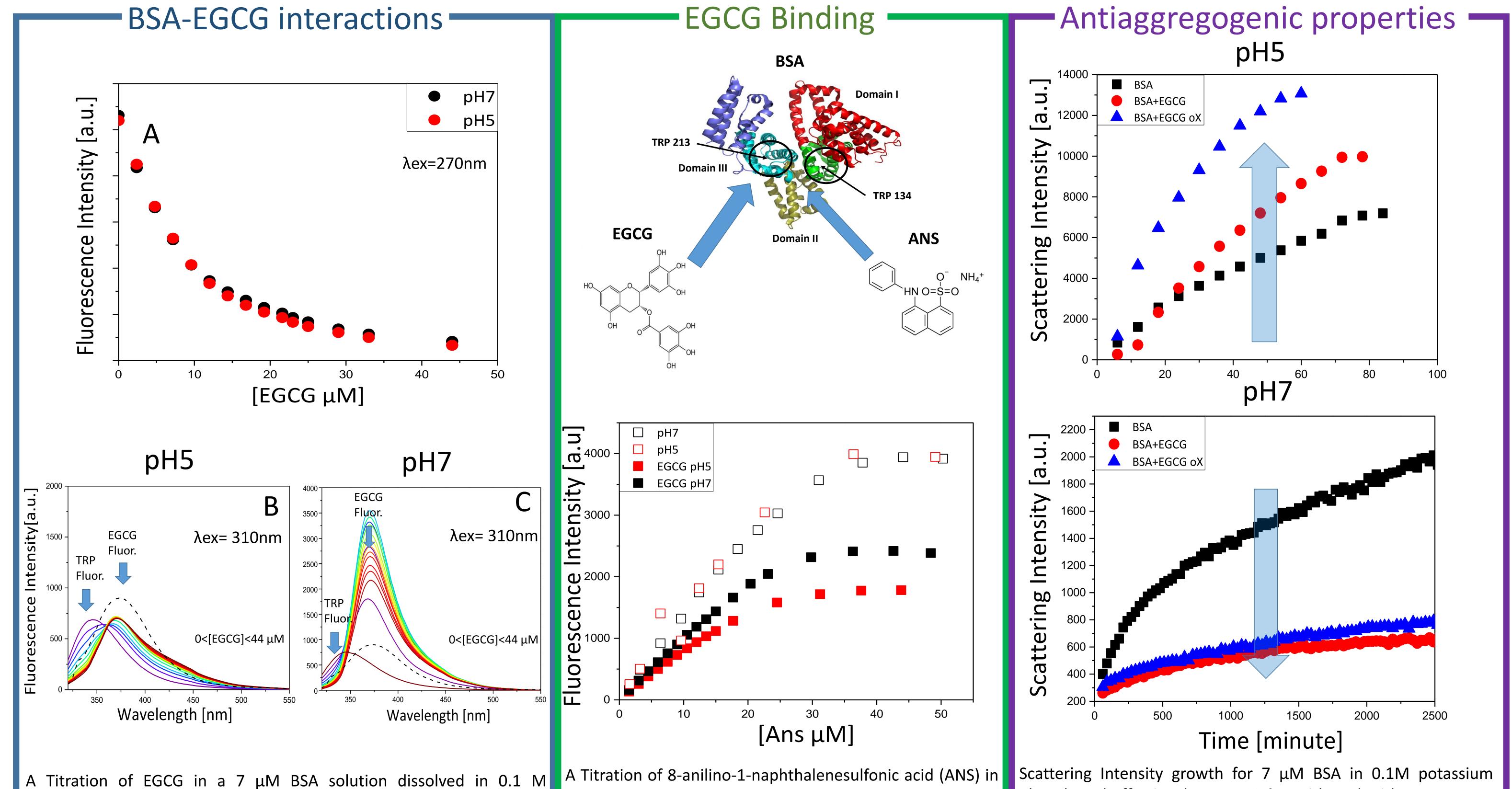


protein aggregation with respect to intact molecule.

We here investigate the effects of intact and oxidized EGCG the thermal aggregation pathway of Bovine Serum Albumin (BSA), a well-known model protein whose aggregation processes are known in details.

By means of different spectroscopic methods, we evaluate similarities and differences of the two molecules during protein aggregation. Different solution conditions are investigated, close and away from the isoelectric point of the protein, with the aim of eliciting the role of electrostatics in the observed EGCG-Protein interaction and in the supramolecular assembly which are dramatically dependent on solution conditions.





potassium phosphate buffer @270nm and @310nm.

Quenching of the two tryptophans is EGCG dose-dependent and it is the same for both solutions panel A. Increasing ANS and EGCG. Fluorescence intensity in the triggers the aggregation. At neutral pH, the insertion of fluorescence intensity reported in panel C indicate a better presence of EGCG is very different in EGCG or EGCG oX reduces protein aggregation. This interaction between the protein and the small-molecule at particular in the presence of EGCG the intensity is process is mediated by electrostatics which plays an pH7. In particular EGCG interact with the hydrophobic site II of BSA near the Tryptophan 213.

in the presence and absence of 24  $\mu$ M EGCG @380nm.

There is a competitive binding process between At acidic pH, the insertion of the EGCG or EGCG ox much lower, this indicates a binding preference of important role in the process. EGCG over ANS.

a 7 μM BSA solution in 0.1 M potassium phosphate buffer 🚺 phosphate buffer incubate at 60 °C with and without EGCG or EGCG oxidized [8.5 μM].

## Conclusions

BSA-EGCG interactions were different as a function of different solution, better binding interaction at pH7 was reported on the blue panel. Green panel shows a competitive binding between ANS and EGCG as a function of pH value. These data confirm our hypothesis and shows a better ANS intensity in the presence of EGCG at pH7. Finally violet panel shows how the electrostatics can mediate the antiaggregogenic properties of EGCG with and without oxidation.