

PAPER • OPEN ACCESS

## Experimental study on dynamic viscosity of aqueous-based nanofluids with an addition of ethylene glycol

To cite this article: W Safiei *et al* 2020 *IOP Conf. Ser.: Mater. Sci. Eng.* **788** 012094

View the [article online](#) for updates and enhancements.



The banner features a dark blue background with a satellite view of Earth. On the left, there are three circular logos: the top one is 'ECS' in a white circle, the middle one is 'The Electrochemical Society' with a stylized 'ECS' logo, and the bottom one is 'THE KOREAN ELECTROCHEMICAL SOCIETY'. The main text in the center reads 'Joint International Meeting PRIME 2020 October 4-9, 2020' in white and blue. Below this, a blue bar contains the text 'Attendees register at NO COST!' in white. On the right side, there is a large white 'PRIME' logo with a blue arc above it, followed by 'PACIFIC RIM MEETING ON ELECTROCHEMICAL AND SOLID STATE SCIENCE' and '2020' in white. At the bottom right, a blue bar contains the text 'REGISTER NOW' in white with a white arrow pointing right.

## Experimental study on dynamic viscosity of aqueous-based nanofluids with an addition of ethylene glycol

W Safiei<sup>1</sup>, M M Rahman<sup>1,\*</sup>, A H Musfirah<sup>2</sup>, M A Maleque<sup>3</sup> and R Singh<sup>4</sup>

<sup>1</sup> Faculty of Mechanical and Manufacturing Engineering, Universiti Malaysia Pahang, 26600 Pekan, Pahang, Malaysia.

<sup>2</sup> Faculty of Engineering Technology, Universiti Malaysia Pahang, 26300 Gambang, Pahang, Malaysia.

<sup>3</sup> Department of Manufacturing and Materials Engineering, International Islamic University Malaysia, Gombak, Selangor, Malaysia

<sup>4</sup> Department of Production Engineering, Guru Nanak Dev Engineering College, Ludhiana-141006, India

\*Corresponding author: mustafizur@ump.edu.my

**Abstract.** In this study, the effect of adding different nanoparticles in the mixture of deionised water and ethylene glycol on dynamic viscosity is investigated experimentally. In order to prepare for single nanofluids, the dry nanoparticles of SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub> and ZrO<sub>2</sub> were dispersed into 60% volume of deionised water and 40% volume of ethylene glycol as a base fluid using a two-step method. The experiments were performed in the temperature range of 30°C and 70°C and weight fraction ranging between 0.1wt.% and 1wt%. No surfactant used in preparing the nanofluids. The dynamic viscosity data were collected using DV-II+ Pro Brookfield viscometer. The single, dual-hybrid and tri-hybrid aqueous based nanofluids dynamic viscosity results are explicitly presented. From the results, it is exhibited that nanofluid viscosity decreases with increasing liquid temperature and increases with increasing of nanoparticles volume concentration. The viscosity decreases with increasing of deionised water volume percentage in the base fluid. Zirconia single nanofluid at 1wt.% recorded 2.5 times maximum enhancement of viscosity over the base fluid. The results display that single nanofluids have higher dynamic viscosity compared to hybrid nanofluids.

**Keywords.** Nanofluid; Deionized Water; Ethylene Glycol; Dynamic Viscosity; Temperature; Concentration.

### 1. Introduction

Nanofluid is a next-generation liquid which certain volume concentration of nanoparticles dispersed in a base fluid. This is an innovative approach of nano-technology that regards on the engineered colloidal suspension to overcome limited capabilities of thermal-physical properties of conventional liquids. A combination of solid and liquid in transferring the heat is more efficient rather than the liquid itself due to the mass motion of particles circulating in the system. The main idea of this concept introduced by Maxwell during his study in 1873 by adding solid particles into the liquid to improve the physical properties of conventional liquid[1]. However, due to multiple problems such as



clogging and abrasion when micrometres particles were dispersed, there was no real application in any engineering system had been tested since then.

The era of nanofluid begins in 1995 after Choi discovered its great potential[2]. Single nanofluid is referred to a one-type or mono-type of nanoparticle with a small percentage of concentration disperse into the base fluid. While in recent years, dispersing two or more different types of nanoparticle in the base fluid, which can be considered as hybrid nanofluid attracts great attention of scholars globally to further enhance thermal-physical properties of single nanofluid [27]. According to Zhang et al.[3] and Li et al.[4], that hybrid or composite nanoparticles that disseminate in the base fluid exhibited higher and significantly improved heat transfer, cooling effect and antifriction performance than single nanofluid, especially in the metal cutting process. Due to its advantages, particularly on enhancing thermal-physical properties of the base fluid, nanofluid receives greater attention nowadays and promising in the future. Therefore, researchers have focused on the potential of nanofluid on thermal conductivity[5][6], heat transfer[7][8], a viscosity[9][10] and the application of nanofluid in heat exchanger[11], photovoltaic[12] as well as in machining[13].

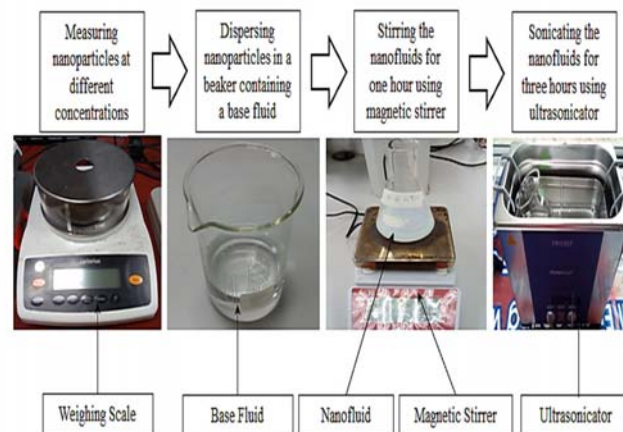
Even though most of the study focus on nanofluid thermal conductivity, however, nanofluid dynamic viscosity deserves the same attention because it has a severe effect on pumping power requirement and the coefficient of heat transfer in engineering systems[10]. Furthermore, the enhancement of nanofluid viscosity over the base fluid is higher than the enhancement of nanofluid thermal conductivity[10]. Garg et al.[14] reported that viscosity of copper nanoparticles enhanced four times over ethylene glycol and about 2 times enhancement of thermal conductivity for the same nanofluid and base fluid. Viscosity can be described as the internal resistance of the liquid to flow. For instance, in a laminar flow, the pressure drop is directly related to liquid viscosity that circulating in a system. Hence, viscosity is one of the essential characteristics, and the behavior of nanofluids viscosity should be explored. Sharma et al.[15] studied various hybrid nanofluids viscosity of different concentrations between 0.5 vol% and 3 vol.%. and viscosity was examined between 25°C and 50°C. The results exhibited Newtonian behavior of hybrid nanofluids and viscosity ratio of hybrid nanofluids decrease with increasing of temperature. The maximum enhancement of viscosity recorded was 52.8% of (80%CeO<sub>2</sub>+20%Cu) hybrid nanofluids. Sundar et al. [16] reported various viscosity results of Fe<sub>3</sub>O<sub>4</sub>-ethylene glycol/water nanofluids with different nanoparticle volume fractions and liquid temperatures. Based on experimental results, it showed that the viscosity of Fe<sub>3</sub>O<sub>4</sub>-ethylene glycol/water nanofluids increased by increasing the nanoparticle volume fraction and decreasing temperature. The maximum enhancement was recorded 2.9 times over the base fluid at a nanoparticle volume fraction of 1%. Bahadir et al.[17] reported that nanofluids have different rheology behavior. Based on the investigation at low temperature which below than 10°C, Alumina behaves non-Newtonian liquid and CNTs behave as a Newtonian liquid especially in high shear rate.

Esfé et al.[18] investigated viscosity of MWCNTs-TiO<sub>2</sub> hybrid nanofluids in 70% water and 30% EG as a base fluid. At 0.45% and 0.5% volume concentration, the results exhibited that hybrid nanofluids have a close Newtonian behavior, however, at 0.85vol.%, the graph of shear rate over shear stress indicated non-linear correlation which resulting non-Newtonian liquid. Moreover, at a lower concentration, the result indicated that no effect on hybrid nanofluids viscosity upon adding 10% excess of nanoparticles. Kerim et al.[19] studied viscosity dependency on nanoparticle size and concentration ranging from 5wt.% to 20wt.% of various metal oxide nanofluid. The measurements reveal that nanofluid greater than 5wt.% regardless of nanoparticles exhibited non-Newtonian behavior between 10<sup>0</sup> and 10<sup>4</sup> of shear rate. The relative viscosity has a strong dependency on particle size and nanoparticle loading, especially at higher weight concentration. Murshed and Estelle[20] suggested that nanofluid pH value, nanoparticle size and shape should be considered when investigating nanofluid viscosity as the published articles are very limited. Mahbulul et al.[9] also recommended considering nanoparticle size as one of the influencing factors when investigating nanofluid viscosity. Based on published articles and to the best of authors' knowledge, an evaluation of single, dual-hybrid and tri-hybrid of dynamic viscosity and a comprehensive comparison between them is very limited.

Therefore, the objective of this study is to evaluate dynamic viscosity of different nanofluids consists of single, dual-hybrid and tri-hybrid nanoparticles that dispersed in the mixture of deionised water and ethylene glycol. The effect of concentration, composition and temperature of dynamic viscosity of  $\text{Al}_2\text{O}_3$ /water-EG,  $\text{SiO}_2$ - $\text{Al}_2\text{O}_3$ /water-EG and  $\text{Al}_2\text{O}_3$ - $\text{SiO}_2$ - $\text{ZrO}_2$  / water-EG nanofluids are presented.

## 2. Methods and materials

In this study, to prepare for single nanofluids, the dry nanoparticles of  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$  and  $\text{ZrO}_2$  were dispersed into 60% volume of deionised water and 40 % volume of ethylene glycol as a base fluid using a two-step method. Two concentrations were prepared at 0.2wt.% and 1wt.%. The nanoparticle size of  $\text{SiO}_2$ ,  $\text{Al}_2\text{O}_3$ , and  $\text{ZrO}_2$  are 20 nm – 21 nm, 13 nm -18 nm and 17 nm- 25 nm as the information was given by the manufacturer Sigma-Aldrich. Ethylene Glycol in liquid form which has 99.8% purity that contains water percentage less than 0.003% also purchased from Sigma-Aldrich. For dual-hybrid nanofluids,  $\text{SiO}_2$ - $\text{Al}_2\text{O}_3$  nanoparticles with 60:40 composition were dispersed in 60:40 of deionized water and ethylene glycol at 0.1wt.%. In order to evaluate tri-hybrid nanofluid dynamic viscosity, two different tri-hybrid nanofluids were prepared which  $\text{SiO}_2$ - $\text{Al}_2\text{O}_3$ - $\text{ZrO}_2$  and  $\text{Al}_2\text{O}_3$ - $\text{ZnO}$ -MWCNT. The composition of  $\text{Al}_2\text{O}_3$ - $\text{ZnO}$ -MWCNT tri-hybrid nanofluid was 45:45:10 and 80:20 was the base fluid ratio of deionized water and ethylene glycol. While, for  $\text{SiO}_2$ - $\text{Al}_2\text{O}_3$ - $\text{ZrO}_2$  tri-hybrid nanofluid, the hybrid composition was 60:30:10 and the base fluid ratio remain as 60:40. The base fluid was prepared at two different ratios due to evaluate the dependency of dynamic viscosity on the base fluid at a different ratio.



**Figure 1.** Step by step process of various nanofluids preparation at different concentrations.

The use of surfactant in preparing nanofluid is one of the effective methods to obtain homogenize suspension. However, the presence of surfactant in this study must be avoided due to it has an additional effect on the nanofluid dynamic viscosity. This additional factor besides weight concentration, temperature and type of nanofluid which would increase the complexity of the study. Hence, no surfactant was used in preparing the suspensions. The samples were stirred for one hour using magnetic stirrer. Then, the single, dual-hybrid and tri-hybrid nanofluids were being ultrasonicated for 3 hours in ultrasonicator to obtain a homogenised suspension. Figure 1 shows step by step technique in preparing nanofluids from the early process until obtaining a homogenised

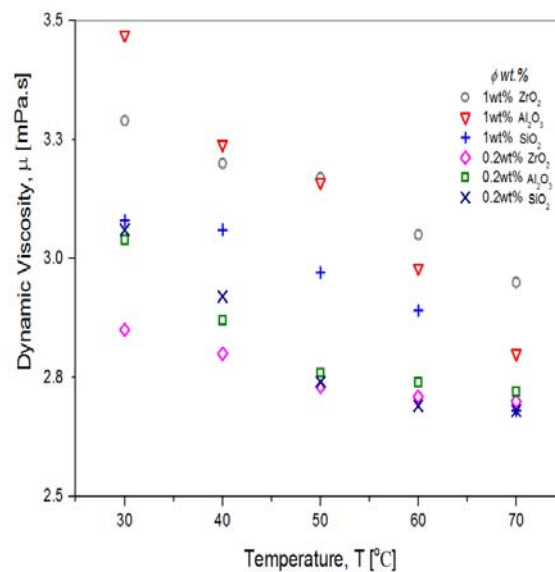
suspension. Dynamic viscosity of these nanofluids was measured using DV-II+ Pro Brookfield viscometer, as shown in figure 2, with the temperature range within 30°C and 70°C.



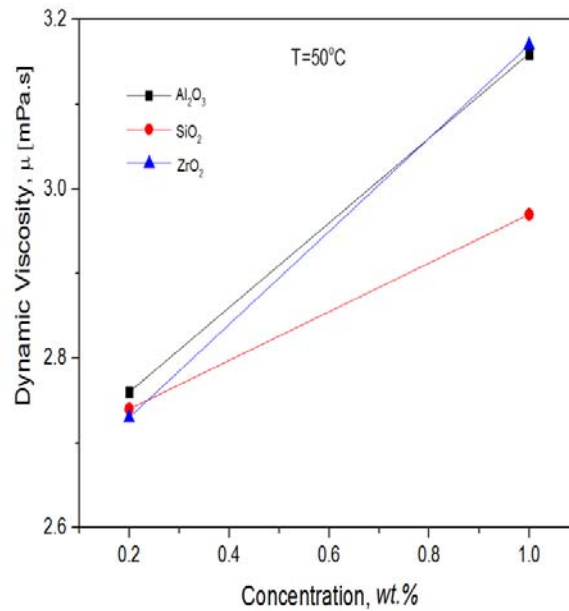
**Figure 2.** DV-II+ Pro Brookfield viscometer.

### 3. Results and discussion

Figure 3 shows the experimental results of dynamic viscosity of  $\text{Al}_2\text{O}_3$ ,  $\text{SiO}_2$  and  $\text{ZrO}_2$  single nanofluids over temperature and concentration. Absolute dynamic viscosity of aqueous based nanofluids reduces with increasing of liquid temperature. As the temperature gradually rising, the bonding between molecules in nanofluids become weaker, resulting in low viscosity as presented in the results. The low viscosity of nanofluid at higher temperature is also associated with the increase of gases viscosity due to higher intensity of molecules movement leaving the nanofluid[21]. Dynamic viscosity increases remarkably with increasing nanoparticles weight concentration, as shown in figure 3(b). Similar findings also reported by Masoud et al. [22] Kole et al. [23] and Sundar et al. [24].



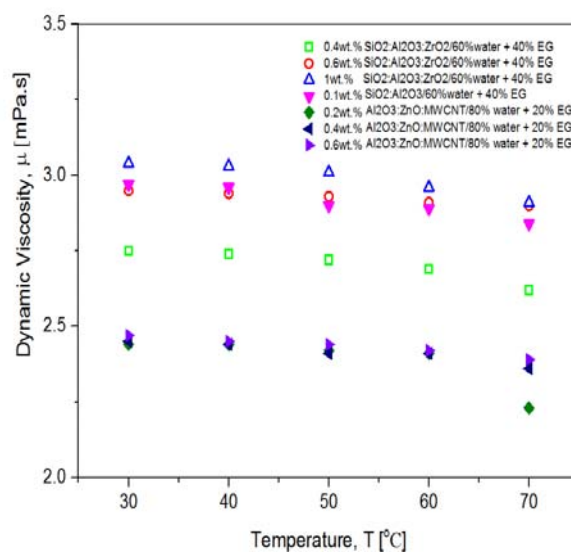
(a)



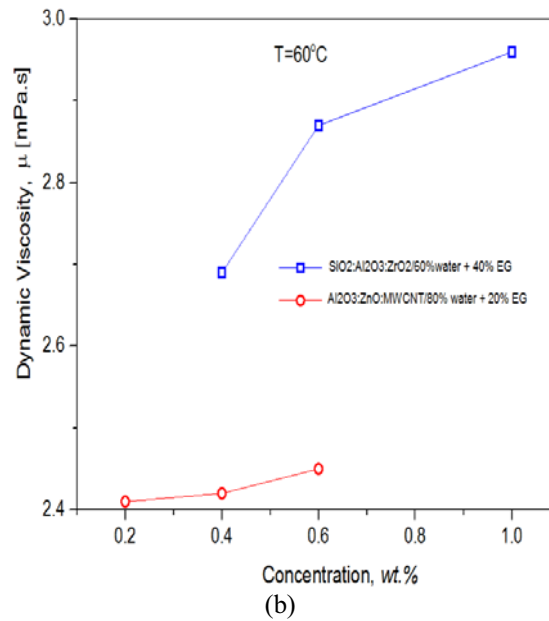
(b)

**Figure 3.** Dynamic viscosity of single nanofluids over (a) Temperature (b) Concentration.

Figure 4 reveals the experimental results of hybrid nanofluids dynamic viscosity over temperature and concentration. The results indicate that viscosity reduces remarkably when the volume percentage of deionised water increasing. Abdul Hamid et al.[25] also reported similar findings that the value of viscosity increased with increasing of ethylene glycol volume fraction ranging from 40% to 60%. Viscosity increases with increasing of weight concentration, indicating similar findings between single and hybrid nanofluids.

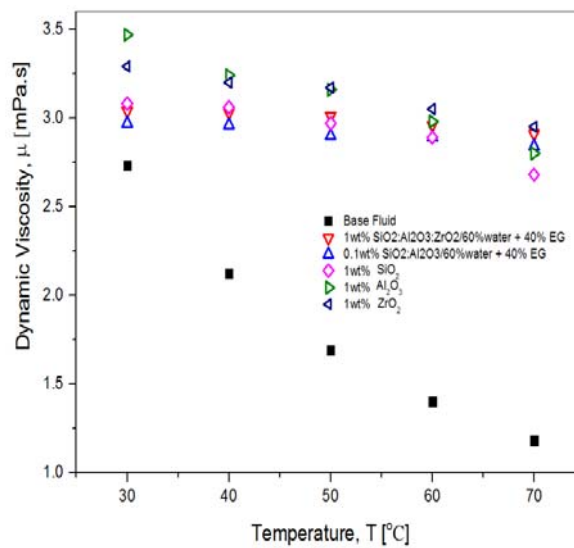


(a)

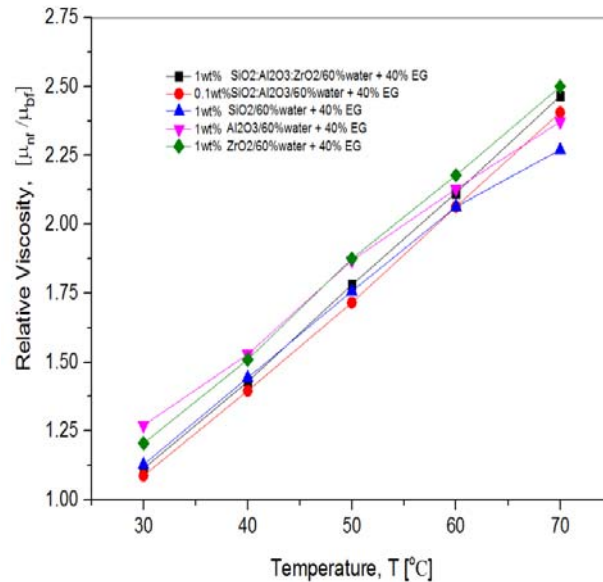


**Figure 4.** Dynamic viscosity of hybrid nanofluids over (a) Temperature (b) Concentration.

Figure 5 depicts a comparison of single and hybrid nanofluid viscosity over the base fluid. The presence of nanoparticles in the nanofluid has enhanced rheological properties with slightly minimal degradation over temperature compared to the base fluid. Dynamic viscosity of 1wt.% ZrO<sub>2</sub> nanofluid recorded the maximum enhancement, which was 2.5 times over the base fluid at 70°C, as shown in figure 6. The relative viscosity of nanofluids has linear proportional, which increases with increasing temperature. According to Esfe et al. [26], the viscosity ratio also enhances with increasing particle concentration and nanoparticle’s diameter.



**Figure 5.** Comparison of the base fluid and nanofluids viscosity over temperature



**Figure 6.** The relative viscosity of different nanofluids over temperature.

#### 4. Conclusions

In this study, it can be concluded that nanofluid viscosity decreases with increasing temperature and deionised water volume percentage in the base fluid. However, nanofluid viscosity increases with increasing of weight concentration. These findings observed at both types of single and hybrid nanofluid. Zirconia single nanofluid at 1wt.% recorded 2.5 times maximum enhancement of viscosity over the base fluid. The results display that single nanofluids have higher dynamic viscosity compared to hybrid nanofluids. Nevertheless, this is an excellent finding for hybrid nanofluids because pumping and pressure losses could be minimized when the nanofluids applied in engineering systems. For future works, the dynamic viscosity of nanofluids with higher concentration and the presence of surfactant in the nanofluid can be explored.

#### Acknowledgement

The authors gratefully acknowledge the support of Universiti Malaysia Pahang for providing research grants RDU1703135 and UMP LEAP-3 flagship project number RDU172203.

#### References

- [1] J. C. Maxwell, 1873 "A treatise on electricity and magnetism," *Oxford: Clarendon Press.* 360–366.
- [2] S. U. S. Choi and J. A. Eastman, 1995 "Enhancing thermal conductivity of fluids with nanoparticles," *ASME International Mechanical Engineering Congress and Exposition.* **66** 99–105
- [3] X. Zhang et al., 2016 "Performances of Al<sub>2</sub>O<sub>3</sub>/SiC hybrid nanofluids in minimum-quantity lubrication grinding," *International Journal of Advanced Manufacturing Technolog.* **86** 3427–3441
- [4] Y. Zhang, C. Li, D. Jia, D. Zhang, and X. Zhang, 2015 "Experimental evaluation of the lubrication performance of MoSinf<sub>2</sub>/inf/CNT nanofluid for minimal quantity lubrication in Ni-based alloy grinding," *International Journal of Machine Tools and Manufacture.* **99** 19–33



- [5] M. I. Pryazhnikov, A. V. Minakov, V. Y. Rudyak, and D. V. Guzei, 2017 “Thermal conductivity measurements of nanofluids,” *International Journal of Heat and Mass Transfer*. **104** 1275–1282
- [6] J. A. Eastman, S. U. S. Choi, S. Li, W. Yu, and L. J. Thompson, 2001 “Anomalous increase in effective thermal conductivities of ethylene glycol-based nanofluids containing copper nanoparticles,” *Applied Physics Letters*. **78** 718–720
- [7] B. Farajollahi, S. G. Etemad, and M. Hojjat, 2010 “Heat transfer of nanofluids in a shell and tube heat exchanger,” *International Journal of Heat and Mass Transfer*. **53** 12–17
- [8] W. Duangthongsuk and S. Wongwises, 2009 “Heat transfer enhancement and pressure drop characteristics of TiO<sub>2</sub>-water nanofluid in a double-tube counter flow heat exchanger,” *International Journal of Heat and Mass Transfer*. **52** 2059–2067
- [9] I. M. Mahbubul, R. Saidur, and M. A. Amalina, 2019 “Latest developments on the viscosity of nanofluids,” *International Journal of Heat and Mass Transfer*. **55** 874–885
- [10] R. Prasher, D. Song, J. Wang, and P. Phelan, 2006 “Measurements of nanofluid viscosity and its implications for thermal applications,” *Applied Physics Letters*. **89** 13
- [11] J. Albadr, S. Tayal, and M. Alasadi, 2013 “Heat transfer through heat exchanger using Al<sub>2</sub>O<sub>3</sub> nanofluid at different concentrations,” *Case Studies in Thermal Engineering*. **1** 38–44
- [12] S. Soltani, A. Kasaeian, H. Sarrafha, and D. Wen, 2017 “An experimental investigation of a hybrid photovoltaic/thermoelectric system with nanofluid application,” *Solar Energy*. **155** 1033–1043
- [13] N. A. C. Sidik, S. Samion, J. Ghaderian, and M. N. A. W. M. Yazid, 2017 “Recent progress on the application of nanofluids in minimum quantity lubrication machining: A review,” *International Journal of Heat and Mass Transfer*. **108** 79–89
- [14] J. Garg et al., 2008 “Enhanced thermal conductivity and viscosity of copper nanoparticles in ethylene glycol nanofluid,” *Journal of Applied Physics*. **103** 7
- [15] S. Sharma, A. K. Tiwari, S. Tiwari, and R. Prakash, 2018 “Viscosity of hybrid nanofluids: Measurement and comparison,” *Journal of Mechanical Engineering and Sciences*. **12** 3614–3623
- [16] L. Syam Sundar, E. Venkata Ramana, M. K. Singh, and A. C. M. De Sousa, 2012 “Viscosity of low volume concentrations of magnetic Fe<sub>3</sub>O<sub>4</sub> nanoparticles dispersed in ethylene glycol and water mixture,” *Chemical Physics Letters*. **554** 236–242
- [17] B. Aladag, S. Halelfadl, N. Doner, T. Maré, S. Duret, and P. Estellé, 2012 “Experimental investigations of the viscosity of nanofluids at low temperatures,” *Applied Energy*. **97** 876–880
- [18] M. Hemmat Esfe, H. Rahimi Raki, M. R. Sarmasti Emami, and M. Afrand, 2019 “Viscosity and rheological properties of antifreeze based nanofluid containing hybrid nano-powders of MWCNTs and TiO<sub>2</sub> under different temperature conditions,” *Powder Technology*. **342** 808–816
- [19] K. Yapici, O. Osturk, and Y. Uludag, 2018 “Dependency of nanofluid rheology on particle size and concentration of various metal oxide nanoparticles,” *Brazilian Journal of Chemical Engineering*. **35** 575–586
- [20] S. M. S. Murshed and P. Estellé, 2017 “A state of the art review on viscosity of nanofluids,” *Renewable and Sustainable Energy Reviews*. **76** 1134–1152
- [21] B. Ruhani, D. Toghraie, M. Hekmatifar, and M. Hadian, 2019 “Statistical investigation for developing a new model for rheological behavior of ZnO–Ag (50%–50%)/Water hybrid Newtonian nanofluid using experimental data,” *Physica A: Statistical Mechanics and its Applications*. **525** 741–751
- [22] M. Afrand, K. Nazari Najafabadi, and M. Akbari, 2016 “Effects of temperature and solid volume fraction on viscosity of SiO<sub>2</sub>-MWCNTs/SAE40 hybrid nanofluid as a coolant and lubricant in heat engines,” *Applied Thermal Engineering*. **102** 45–54
- [23] M. Kole and T. K. Dey, 2010 “Viscosity of alumina nanoparticles dispersed in car engine

- coolant,” *Experimental Thermal and Fluid Science*. **34** 677–683
- [24] L. Syam Sundar, E. Venkata Ramana, M. K. Singh, and A. C. M. Sousa, 2014 “Thermal conductivity and viscosity of stabilized ethylene glycol and water mixture Al<sub>2</sub>O<sub>3</sub> nanofluids for heat transfer applications: An experimental study,” *International Communications in Heat and Mass Transfer*. **56** 86–95
- [25] K. A. Hamid, W. H. Azmi, R. Mamat, N. A. Usri, and G. Najafi, 2015 “Investigation of Al<sub>2</sub>O<sub>3</sub> Nanofluid Viscosity for Different Water/EG Mixture Based,” in *Energy Procedia*. **79** 354–359
- [26] M. Hemmat Esfe, S. Saedodin, S. Wongwises, and D. Toghraie, 2015 “An experimental study on the effect of diameter on thermal conductivity and dynamic viscosity of Fe/water nanofluids,” *Journal of Thermal Analysis and Calorimetry*. **119** 1817–1824
- [27] Zawawi, N. N. M., Azmi, W. H., Redhwan, A. A. M., Sharif, M. Z., & Samykano, M., 2018. Experimental investigation on thermo-physical properties of metal oxide composite nanolubricants. *International Journal of Refrigeration*. **89** 11-21