Investigation of the Temperature Dependent Affinity of SARS-CoV-2 Spike Protein to Gold Nanoparticles Sarah Eckl, Yasmin Khilji-Neal, Dana Moukaled, Hannah Valensi

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BACKGROUND

Based on a previous study on the Beta Amyloid peptide, it was found that the conformation of the peptide (protein) was able to change with external conditions such as pH. We decided to test the conformational changes with pH hopping of the spike protein of SARS-CoV-2. The spike protein is the surface protein of SARS-CoV-2 which initiates an immune response and binds to the receptors on the cell membrane. At a pH of 10, which is a basic condition, the receptor on the spike protein would be facing downwards and not be able to form any gold aggregations. In a pH of 3, which is an acidic condition, we would expect either the 'parallel in opposite direction" or "head to head dimer" which would both form gold aggregation.



Figure 1. Conformational changes of the spike protein between the pH of 3 and 10

Proteins are invisible to the eye and under a microscope so it is essential that they are able to be observed in some sense. We do this by attaching the protein to gold nanoparticles. Proteins will embed a small portion of themselves into the particles and stick out the rest of their form outside of the particle. This allows us to visualize the protein interaction under a transmission electron microscope (TEM). When the proteins are unfolded they will interact with the other proteins attached to other particles and cluster the gold together making a dark clump visible on a TEM image. Nano materials tend to have different properties based on their size so it is essential to test for this variable. We make different samples to test using gold particles ranging from 10nm to 100nm.



Figure 2. The increase in absorbed wavelengths as a function of operation number and nano particle size

prepared and used to observe the shift in the peak position wavelength on the UV-Vis spectrophotometer, we were able to interpret the reversibility of the protein.

RESULTS



at different temperatures and pH levels



different temperatures and pH levels

peak position of adsorption was generally lower at the lower temperature and increased as temperature increases. Another in different places; the aggregates with 100 nm show that peak adsorption occurs within a smaller range of wavelengths.