

## 12. Abhijit Pisal - Synthesis of low density and high temperature resistant Y<sub>2</sub>O<sub>3</sub> doped silica aerogels

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### Abstract

Commercialization of aerogels has been slow down due to high cost and manufacturability issues [1]. Therefore, in the present paper, we manufactured silica aerogels at low cost and low risk. In this paper, we report the experimental results on the synthesis of low density and high temperature resistant Y<sub>2</sub>O<sub>3</sub> doped silica aerogels. Silica sols were prepared by keeping the molar ratio of TEOS: ethanol (EtOH): water (0.01 M HCl as acid catalyst): Ammonium Fluoride (0.5 M NH<sub>4</sub>F as base catalyst) was kept constant at 1:15:7:0.6 respectively, while the weight percent of Y<sub>2</sub>O<sub>3</sub> powder was varied from 0.1 to 4%. The aerogels have been produced by two-stage sol-gel process followed by supercritical CO<sub>2</sub> drying. The best quality silica aerogel in terms of low high-temperature thermal conductivity (0.097 W.mK<sup>-1</sup> at 1000 °C), low density (26 mg/cc) and high optical transmission (about 87% in the red region) have been obtained with molar ratio of 1TEOS: 15EtOH:7water:0.6 NH<sub>4</sub>F and 3% of Y<sub>2</sub>O<sub>3</sub> powder. The best quality Y<sub>2</sub>O<sub>3</sub> doped silica aerogel is as shown in figure (Fig 1).



Fig 1. Y<sub>2</sub>O<sub>3</sub> doped silica aerogel

The Y<sub>2</sub>O<sub>3</sub> doped aerogels have been characterized by SEM (Scanning Electron Microscopy), FTIR (Fourier Transform Infra-Red Spectroscopy), Optical transmittance, TG/DTA (Thermogravimetry/Differential Thermal Analysis, Thermal conductivity and BET (Brunauer-Emmett-Teller) analysis. The experimental results on the physical and thermal properties of Y<sub>2</sub>O<sub>3</sub> doped silica aerogels under normal and high temperature have been discussed by taking into account the chemistry and porosity of aerogels.

### References

[1] G. Carlson, D. Lewis, K. McKinley, J. Richardson and T. Tiltson, "Aerogel commercialization : Technology, Markets and costs," J. Non-Cryst. Solids, vol.186, pp. 372–379, Jun. 1995.

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