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## Study of Surface Morphology and Porosity of Composite Scaffold Nanofiber PVA/CS/HA with Electrospinning Method

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# Abstract

This research aims to compose nanofibers as a scaffold material in bone tissue engineering in terms of surface morphological properties and porosity. HA nanorod was prepared by the precipitation-ultrasonication method, while the PVA/CS/HA nanofiber composites were made by the electrospinning method using a static collector. HA was characterized by using XRD and SEM-EDX, while the PVA/CS/HA nanofiber composites used FTIR and SEM. The results show that HA nanorod has a crystalline size of 10.86 nm, crystallinity level of 52.38%, and Ca/P ratio of 1.70. From the SEM image shows HA nanorod width of 11.6 nm and 97.53 nm in length and some of it still in the form of HA nanoparticles. The diameter and porosity of PVA/CS/HA nanofiber with addition of 0%, 10%, 20% HA were 275, 212, 265 nm and 72.94, 69.49, 70.81%, respectively. **Keywords:** Hydroxyapatite, PVA, chitosan, electrospinning.

### 1. Introduction

For more than a decade, the traditional methods of autograft and allograft bone grafts have been used to treat large-scale bone defects. These method is used of grafting the patient's own bone whereas allograft is a method of grafting a donor bone from another person/body [1]. However, these techniques has disadvantages of high surgical risks, such as bleeding, inflammation, infection, chronic disease, and high cost [2]. Whereas allograft bone grafts are associated with the risk of immunoreaction, infectious transmission, and have a record high failure rate for long term use [3], [4].

Scaffold engineering of bone tissue appears as a solution that can provide support for newly formed bone tissue, as a temporary site for interactive cell traffic and for the formation of the extracellular matrix of bone [5]. Hydroxyapatite (HA) meets the requirements as a scaffold biomaterial with mineral components similar to natural bone and has osteoconductivity, bone tissue binding ability, biocompatibility, affinity for biopolymers, and high osteogenic potential [6], [7]. HA crystals found in bone are generally nanometer-shaped needles (nanorods) with a dimension range of 60 nm long and 5–20 nm wide [8].

Chitosan (CS) was chosen due to its various superior properties in medical and tissue engineering applications, including biodegradability, biocompatibility, non-toxic, hydrophilic, and anti-microbial properties [9], [10]. One of the modifications with synthetic polymer PVA is used to overcome the weakness of the properties of CS which is known to be difficult to form fibers. In addition, PVA also has several superior properties for application in the medical field, including biocompatibility, non-toxicity, non-carcinogenicity, bioadhesive, and good mechanical properties [11].

In this study, HA nanorod synthesis was carried out using the precipitation-ultrasonication method, then it was composited with the PVA/CS polymer matrix, and the process of forming nanofibers using the electrospinning method. The effect of HA addition on surface morphological properties and porosity in PVA/CS/HA nanofiber composites was observed because in nanofiber scaffolds, these parameters have an important role in providing space for cell regeneration as a place to interact and form new tissue [12].

#### 2. Method

#### 2.1. Synthesis of Hydroxyapatite

The synthesis of HA nanorod was carried out by the precipitation-ultrasonication method.  $KH_2PO_4$  solution (0.19 M) was added dropwise to a solution of  $Ca(NO_3)_2$  (0.32 M), stirred using a magnetic stirrer for 60 minutes at 80 °C, rotation speed of 600 rpm, and the reaction pH was maintained at pH 9 using NH<sub>4</sub>OH solution (0.25 M) with the following Equation (1).

$$\begin{bmatrix} 10Ca(NO_{3})_{2(aq)} + 6KH_{2}PO_{4(aq)} \\ + 20NH_{4}OH_{(aq)} \end{bmatrix} \rightarrow \begin{bmatrix} Ca_{10}(PO_{4})_{6}(OH)_{2(s)} + 6KOH_{(aq)} \\ + 20NH_{4}NO_{3(aq)} + 12H_{2}O_{(l)} \end{bmatrix}$$
(1)

The solution obtained was sonicated for 30 minutes at 50 °C with ultrasonicator frequency of 45 kHz. Then the solution was precipitated using centrifugation at 2,200 rpm for 10 minutes. The precipitate was taken and washed with DI water, then dried at 110 °C for 5 hours. The sample is ready to XRD characterization process to determine the purity of the phase, crystal grain size, and crystallinity. SEM EDX to determine the structure and Ca/P ratio of the sample.

### 2.2. Synthesis of PVA/CS/HA Nanofiber Composite

Dissolve 0.52 grams of PVA powder in 4 ml of distilled water using a magnetic stirrer at a speed of 700 rpm and a temperature of 110 °C for 3 hours. Dissolving 0.1 gram of chitosan in 5 ml of 2% acetic acid using a magnetic stirrer at a speed of 700 rpm and room temperature for 3 hours. The homogeneous dissolved PVA was added with chitosan and stirrer at a speed of 700 rpm and a temperature of 110 °C for 2 hours until it was homogeneous. Adding 10 and 20% HA powder to PVA/CS solution of 0.054 and 0.108 grams respectively and stirrer at 700 rpm and 110 °C for 2 hours. The solution was sonicated for 30 minutes and then electrospinning with a needle to collector distance of 16.5 cm, an electrospinning voltage of 12 kV, and a flowrate of 80 ml/minute for 90 minutes. Make initial observations using an optical microscope to see the formation of nanofibers. The PVA/CS/HA composites were then characterization by FTIR and SEM.

#### 3. Result and Discussion

#### 3.1. Analysis XRD of Hydroxyapatite

Figure 1 shows the XRD HA diffraction pattern of fitting results using Rietica software with a comparison database derived from Crystallography Open Database (COD) number 9013627 to determine the lattice parameters, crystal structure, and size of HA nanoparticles. Where the model has a lattice value a = b = 9.4081 Å and c = 6.8887 Å with a value of  $a = b \neq c$ ;  $\alpha = \beta = 90^{\circ}$ ;  $\gamma = 120^{\circ}$  which indicates that hydroxyapatite has a hexagonal crystal structure. The average value of the crystal size resulting from the refinement of the HA sample was 10.86 nm. Whereas in a study conducted by Mehta *et al.* [16] using the same method produced an average crystal size of 36.51 nm and found that giving temperature treatment by sintering could increase the crystal size of hydroxyapatite nanorod.



Figure 1. Refinement hydroxyapatite XRD pattern.

The degree of crystallinity from the XRD test results in this study is calculated using Equation 2.

$$\chi_c = \frac{\text{Area Fraction of Crystal}}{\text{Area Fraction of Crystal + Area Fraction of Amorf}} \times 100\%$$
(2)

From the calculation using this formula, the degree of crystallinity of the HA synthesized results is 52.38%. This value fulfills the crystallinity level that must be met in the body, which is around 40–60%. ISO standardized specifications (ISO 13779-1:2000) state that to make HA coatings that have sufficient mechanical strength, at least they must have a degree of crystallinity of more than 45%. This shows that the results of the synthesis of this hydroxyapatite sample have met the minimum crystallinity level of the given ISO standard. In previous studies proved that the degree of crystallinity of hydroxyapatite can be increased with increasing reaction temperature [13]–[16].

#### 3.2. Analysis SEM-EDX of Hydroxyapatite

Preparation of hydroxyapatite nanorod using the precipitation-ultrasonication method in this study produced hydroxyapatite in the form of partially agglomerated nanoparticles (see Figure 2). The length and width dimensions of the measurement results using ImageJ software obtained an average nanorod length of 97.53 nm and a width of 11.67 nm.

The results of the elemental analysis from the SEM-EDX characterization in Figure 3 show that the Ca/P value in the hydroxyapatite sample in this study through the calculation of the elemental mass ratio is 1.70. This result is close to the mass ratio of Ca/P hydroxyapatite in natural bone which is 1.67 [17] which gives it superior mechanical properties [18], [19].



Figure 2. The results of SEM characterization of HA samples with magnification (a) 100,000 times; (b) 25,000 times.



Figure 3. SEM EDX hydroxyapatite results.



Figure 4. FTIR spectrum of PVA/CS (grey line), PVA/CS/HA 10% (red line), and PVA/CS/HA 20% (blue line).

## 3.3. Analysis FTIR of PVA/CS/HA Nanofiber Composite

The FTIR test results in Figure 4, sample A1 (presented by grey line) shows the presence of PVA with the detection of the CH bending, CO stretching, and OH stretching functional groups at wave numbers around 1,435; 1,095; and 3,315 cm<sup>-1</sup>, respectively. The presence of chitosan with the appearance of the amide (NH) and amino (CH) functional groups at wave numbers 3,315 and 2,941 cm<sup>-1</sup>. From these results show the increasing HA concentration causes the hydroxyl (OH) functional group in sample A2 (presented by red line) shift to the left (bigger) than sample A1 from 3,315 to 3,331 cm<sup>-1</sup>. This suggests that a stronger hydrogen bond is formed due to the increase in OH from the addition of hydroxyapatite [20]. In the A3 (presented by blue line) sample the hydroxyl group shifted back to a smaller wave number at 3,315 cm<sup>-1</sup> which reflects the agglomeration of HA. The agglomeration that occurs will reduce the interaction between the surface of the HA nanoparticles and the polymer, so that the active OH groups on the surface of the two materials will have less opportunity to interact with each other [21].

## 3.4. Analysis SEM of PVA/CS/HA Nanofiber Composite

Figure 5(a) shows that the surface morphology of the PVA/CS matrix produces fibers that are quite homogeneous, continuous, and without beads. This shows that the PVA/CS composite has dissolved well and the electrospinning parameters (needle to collector distance, voltage, and flowrate) are appropriate. Figure 5(b) shows the addition of 10% HA to produce a homogeneous, continuous fiber, and there are few beads due to the agglomerated HA. Whereas in Figure 5(c) the addition of 20% HA shows less homogeneous fibers, almost broken, and there are many beads with micron-sized agglomerations that can cause the scaffold to underperform in cell adhesion and proliferation [22].



Figure 5. SEM characterization results at magnification of 10,000 times nanofiber (a) PVA/CS, (b) PVA/CS/HA 10%, and (c) PVA/CS/HA 20%.



Figure 6. Nanofiber diameter distribution of (a) PVA/CS, (b) PVA/CS/HA 10%, and (c) PVA/CS/HA 20%.

	Porosity (%)	Fiber Diameter	Fiber Average
		Distribution (nm)	Diameter (nm)
PVA/CS	72.94	204 - 457	$(274.94 \pm 1.90)$
PVA/CS/HA 10%	69.49	118 - 340	$(211.77 \pm 3.60)$
PVA/CS/HA 20%	70.81	160 - 456	$(264.99 \pm 4.71)$

Table 1. Porosity and diameter distribution of PVA/CS/HA nanofibers.

The fiber diameter distribution is shown in Figure 6 that the level of homogeneity of the fiber diameter size decreases with the addition of HA. The average diameter size of PVA/CS/HA fibers decreased in the nanofiber composites without HA to the addition of 10% HA. The smaller size of the fiber diameter is due to the increased viscosity of the solution with the addition of HA [23]. Meanwhile, the addition of 20% HA indicates that the fiber diameter increases again due to a lot of HA agglomeration which causes the HA interaction with the PVA/CS matrix to be disturbed so that when the fiber formation becomes unstable which results in the formation of beads and the fiber diameter size is not homogeneous. Natural bone tissue itself has organic matrix fibers that range in diameter from 100 to 450 nm [24].

Table 1 shows the results of the porosity and fiber diameter which shows a unidirectional relationship. This happens because the simple porosity calculation using OriginPro software depends on the SEM test results' surface area and fiber thickness. The PVA/CS/HA 10% composite, which shows a smaller fiber diameter, causes a smaller surface area and a larger pile of fibers. Thicker than the addition of HA 0 and 20% resulted in less porosity. The greater the porosity the greater the size of the fiber diameter, and vice versa, the smaller the porosity, the smaller the fiber diameter size [23]. The tendency for fibers to accumulate will result in poor porosity and reduced pore size [25].

#### 4. Conclusion

The synthesized hydroxyapatite has a hexagonal crystal structure, a crystal size of 10.86 nm and a crystallinity level of 52.38% with a Ca/P ratio close to the ideal value of HA (1.67), which is 1.70. HA nanorod has been successfully synthesized with dimensions of 97.53 nm in length and 11.67 nm in width with part of it in the form of HA nanoparticles. The addition of HA concentrations in the PVA/CS/HA nanofiber composites can reduce the size of the fiber diameter thereby increasing the surface area of the fiber. In the other hand the addition of highest HA concentration affected in reduced fiber homogeneity, nearly cut fibers, HA agglomeration, and the formation of beads. The porosity of the PVA/CS/HA nanofiber composites has a direct relationship with the average fiber diameter and is inversely related to the thickness of the nanofiber.

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