



PERFORMANCE OF METAKAOLIN BASED GEOPOLYMER CONCRETE AT ELEVATED TEMPERATURE

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ABSTRACT

Due to the carbon dioxide emission arising from the production of cement, alternative concrete that is environmentally friendly such as metakaolin geopolymer concrete have been developed. However, the performance of metakaolin based geopolymer concrete (MKGC) when exposed to aggressive environment particularly elevated temperature has not been investigated. Therefore, this paper assessed the performance of MKGC exposed to elevated temperatures. MKGC cube specimens of grade 25 were produced using a mix ratio of 1:1.58:3.71. After preparing the specimens, they were placed in an electric oven at a temperature of 60°C for 24 hours. Thereafter, the specimens were stored in the laboratory at ambient temperature for 28 days. The specimens were then exposed to elevated temperatures of 200, 400, 600 and 800°C. After exposure to elevated temperatures, the MKGC specimens were subjected to compressive strength, water absorption and abrasion resistance tests. Results show that at 600 and 800°C, the MKGC lost a compressive strength of 59.69% and 71.71% respectively. Higher water absorption and lower abrasion resistance were also observed.

Keywords: Cement, Compressive Strength, Metakaolin Concrete, Elevated Temperature.

1. INTRODUCTION

Geopolymer are inorganic aluminosilicate materials synthesized at room temperature or slightly above through alkali activation of reactive aluminosilicate [1]. These silicon and aluminum are dissolved in an alkaline activating solution and subsequently polymerizes in to molecular chain which works as calcium silicon hydrate to bind the aggregate materials [2, 3]. The reaction of the materials having aluminosilicate with others having alkalis produce a synthetic alkali silicate material called geopolymer [4]. The geopolymerization process involves dissolution of the geopolymer reactants in a strong alkali solution yielding polymeric Si – O – Al – O bonds in amorphous form [5]. Geopolymer was first introduced by Davidovits in 1978 to describe a family of mineral binders that could cure in alkaline solution as an alternative to conventional Portland cement concrete.

Concrete is the most widely used construction material, and second most consumed material worldwide after water [6]. This is due to numerous advantages it possesses, such as applicability, availability and low cost. However, the manufacturing of cement which is a major binder in concrete leads to global warming. The global warming is caused by the emission of greenhouse gasses such as carbon dioxide (CO₂) in to the atmosphere [7]. Manufacturing of Portland cement is responsible for the emission of up to 1.5 billion tons of CO₂ annually which amounts to 5 – 8% of the total CO₂ emission which poses a threat to the global climate change [2, 8]. If this trend continues, the rate of carbon emission arising from the production of cement may increase up to 10 -15% by 2020 [9]. Therefore, it is imperative to obtain green concrete as an innovative solution to the aforementioned challenge. One of such solution is geopolymer concrete manufacture which is also

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known as zero-cement concrete [8]. Geopolymer concrete has increasingly attracted attention due to its increasingly environmentally beneficial applications as the process of alkali activation will utilize industrial refuse [10]. Geopolymer can be produced by reaction between aluminosilicate materials such as metakaolin, fly ash, granulated blast furnace and alkali activator [2].

Metakaolin is a dehydroxylated form of clay mineral kaolin with coordinated aluminum ion in alumina polyhedron sheet structures [11]. Metakaolin is also a valuable admixture with many excellent advantages including porosity, high specific area, good absorbability and strong coordinated bonds. Metakaolin is produced by heating kaolin at a temperature of 650 – 900°C [12]. During heating process, the structure of kaolinite mineral transforms from crystalline to amorphous which make it a highly reactive pozzolan. This makes metakaolin suitable for the production of metakaolin based geopolymer concrete.

As a newly developed material, metakaolin geopolymer concrete may be exposed to elevated temperatures during fire when it is near electric furnace and reactors [13, 14]. During such exposures the mechanical properties are significantly reduced [15, 16]. Apart from affecting the mechanical properties, Aijaz, *et. al*, [17] reported that an elevated temperature reduces durability properties of concrete structures. Hence, it is important to understand the change in the concrete properties due to extreme temperature exposures [18]. Several researches were conducted on the effect of elevated temperature on the properties of geopolymer concrete. For instance, Mateusz, *et. al* [19] studied the effect of high temperature on mechanical properties of fly ash based geopolymer concrete. Talha [20] and Rivera *et al.* [21] conducted a research on the performance of fly ash based geopolymer concrete exposed to elevated temperature. However, literature on the effect of elevated temperature on the properties of metakaolin based geopolymer concrete is scarce.

Therefore, this paper seeks to study the effect of elevated temperature on compressive strength, water absorption and abrasion resistance of metakaolins based geopolymer concrete.

2. MATERIALS AND METHODS

2.1 Materials

The materials used for the study are:

2.1.1 Metakaolin

Kaolin was obtained from Kankara Local Government, Katsina State, Nigeria. The Kaolin was calcined at 700°C for six hours in order to activate kaolin in to metakaolin. After the metakaolin has cooled down to room temperature, it was sieved through 75µm sieve. The chemical composition of the metakaolin used is presented in Table 1.

2.1.2 Alkaline Solution

The alkaline solution was prepared by dissolving NaOH crystals in water and mixing the resultant solution of NaOH with NaSiO₃. The solution was then stored at room temperature for 24 hours before use. The concentration of NaOH used is 16 molar concentrations while for Na₂SiO₃ the ratio of Na₂O, SiO₂ and water as provided by the manufacturer are 14.7%, 29.4% and 55.9 % respectively.

2.1.3 Water

The water used for the experiment was clean water fit for drinking which is in compliance with the provisions of [22].

2.1.4 Aggregates

The aggregates used for the experiment were fine and coarse aggregates that were in conformity with the provisions of [23]. The fine aggregate used is sharp river sand with a maximum size of 4.75mm while the coarse aggregate used is crushed gravel with a maximum size of 20mm.

2.2 Mix proportion

Due to the fact that there is no standard method of mix design for geopolymer concrete, a series of trial tests were conducted in order to establish optimum proportion of materials to be used for the production of geopolymer concrete. After trial test, the mix proportions presented in Table 2 were found to be more suitable and were used for the experiment.

Table 1: Chemical Composition of Metakaolin

Oxides (%)	Na ₂ O	MgO	Al ₂ O ₃	SiO ₂	P ₂ O ₅	SO ₃	K ₂ O	CaO	TiO ₂	Mn ₂ O	Fe ₂ O ₃	ZnO	SrO
	0.045	0.496	41.43	55.98	0.250	0.313	0.728	0.056	0.083	0.008	0.608	0.002	0.001

Table 2: Mix proportion of Materials

Sodium silicate	Sodium Hydroxide	Extra water	Metakaolin	Fine aggregate	Coarse aggregate
169kg/m ³	73kg/m ³	3.19kg/m ³	319kg/m ³	506kg/m ³	1182kg/m ³

2.3 Preparation of the Samples

Metakaolin and aggregates were adequately mixed together, after which alkaline solution was added and mixed thoroughly. The fresh metakaolin geopolymer concrete (MKGPC) mix was cast in to a steel mould of 100 x 100 x 100 mm in two layers and each layer was compacted with a poker vibrator. After 24 hours of casting, the concrete cubes were demolded and wrapped with polythene. The specimens were then placed in an electric oven at a temperature of 60°C for 24 hours in order to speed up the polymerization reaction as presented in Plate I. Thereafter, the concrete cubes were removed from electric oven and the polythene was removed and concrete cube specimens stored at room temperature for 28 days as presented in Plate II.

The concrete cube samples were then subjected to elevated temperatures of 200, 400, 600 and 800°C for a period of 2 hours at a heating rate of 10°C/minute as presented in figure 1. After exposure to elevated temperatures the concrete samples were allowed to cool down naturally to room temperature and then subjected to compressive strength, water absorption and abrasion resistance tests. Compressive strength and water absorption tests were carried out in accordance with the provisions of [24] and [25] respectively while abrasion resistance was carried out in accordance with the procedure adopted by [26]

3. RESULTS AND DISCUSSIONS

3.1 Colour Change

Plate I shows color change for geopolymer concrete samples at elevated temperature. It can be seen that at 27 and 200°C, the color of the geopolymer concrete specimens is pink. At 400°C, the color turned dark pink. At 600 and 800°C the color changes to dark grey and whitish grey respectively. The color change may be attributed to the oxidation of iron oxide contained in fine and coarse aggregates. This result is in agreement with the findings of Lee, Choi and Hong [27] & Hager [28] that established that, color change in heated concrete is due to oxidation of iron oxide contained in silicate aggregates.



Plate I: MKGPC wrapped and placed in Oven



Plate II: MKGPC Removed from Oven

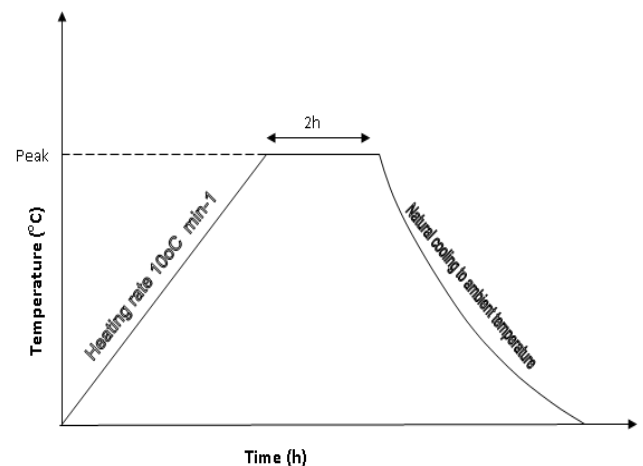


Figure 1: Heating rate for Metakaolin based Geopolymer Concrete Samples

Since color change is due to temperature rise, the color may be used to reveal the exposure temperature of MKGPC.



(a) 27°C



(b) 200°C



(c) 400°C



(d) 600°C



(e) 800°C

Plate I: Colour Change and Crack Appearance

3.2 Compressive Strength

Figure 2 presents the compressive strength of MKGC specimens after exposure to elevated temperature. It can be observed that, the residual compressive strength decreases as the temperature rises. At 27°C the MKGPC specimens attained a compressive of 25.8 N/mm². At 200°C, the residual compressive strength decreased to 22.3 N/mm². At 400°C, 600°C, and 800°C the residual compressive strength decreased further to 16.8 N/mm², 10.4 N/mm² and 7.3 N/mm² respectively. This indicates that, the compressive strength of MKGC declines by 13.57%, 34.88%, 59.69% and 71.71% at 200, 400, 600 and 800°C respectively. The reduction in compressive strength could be attributed to the partial decomposition of alkaline materials (sodium silicate and sodium hydroxide) that bind the aggregates together. This is in conformity with the findings of Talha [20].

3.3 Water Absorption

Figure 3 depicts water absorption of MKGC exposed to elevated temperatures. The water absorption increases as the temperature rises. At 27°C, water absorption of MKGC specimen is 3.6%. After exposure to 200°C, water absorption increased to 4.9%. At 400, 600 and 800°C, the percentage of water absorption increased to 5.35%, 6.43% and 7.96% respectively. The higher water absorption observed at elevated temperature could be due to the formation of pores as a result of the partial decomposition of alkaline materials.

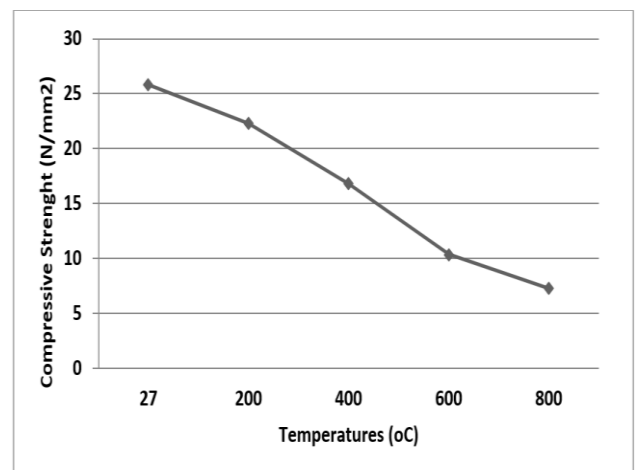


Figure 2: Residual Compressive Strength of Metakaolin Geopolymer Concrete

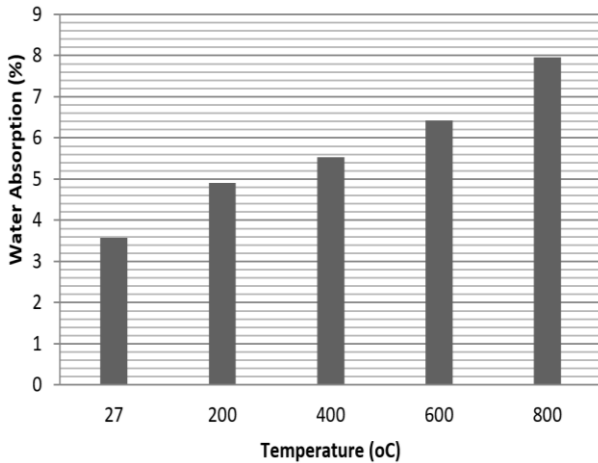


Figure 3: Water Absorption of Metakaolin Geopolymer Concrete Specimens a

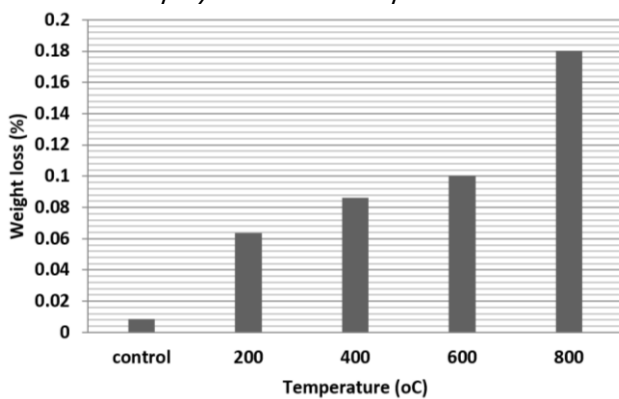


Figure 4: Abrasion Resistance of MKGC Samples Exposed to Elevated Temperature

3.4 Abrasion Resistance

Figure 4 shows the abrasion resistance of MKGC specimens exposed to elevated temperatures expressed as percentage loss of weight. The loss of weight increased as the temperature rises. At 200°C, the MKGC loss a weight of 0.06%. At 400, 600 and 800°C the loss of weight increased to 0.08%, 0.10% and 0.18% respectively.

4. CONCLUSIONS

- i. The color change may be used to reveal the temperature to which metakaolin based geopolymer concrete was exposed to.
- ii. Compressive strength decreases as the temperature rises due to partial decomposition of sodium silicate.
- iii. Water absorption of metakaolin geopolymer concrete increased as the temperature rises while abrasion resistance decreased.

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