African Journal of Biotechnology Vol. 9(8), pp. 1200-1204, 22 February, 2010 Available online at http://www.academicjournals.org/AJB ISSN 1684–5315 © 2010 Academic Journals

Full Length Research Paper

# Effect of processing on iodine content of some selected plants food

Salau, B. A.<sup>1\*</sup>, Ajani E .O<sup>1</sup>, Odufuwa, K. T.<sup>1</sup>, Adegbesan, B. O.<sup>1</sup>, and Soladoye, M. O.<sup>2</sup>

<sup>1</sup>Department of Biochemistry, Faculty of Basic Medical sciences, Olabisi Onabanjo University, Remo Campus, Ikenne, Ogun State, Nigeria.

<sup>2</sup>Department of Plant Science and Zoology, Faculty of Science, Olabisi Onabanjo University, Ago-Iwoye, Ogun State, Nigeria.

Accepted 28 December, 2009

Effect of processing on iodine content of some selected plants food was investigated. Results show significant reduction (p < 0.05) in the iodine content of the processed food compared with the raw forms. The iodine value of 658.60 ± 17.2 µg/100g observed in raw edible portion of *Discorea rotundata* was significantly higher compared with the value of 448.60 ± 2.46, 248.60 ± 2.46 and 300.05 ± 5.66 µg/100g dry matter observed in the boiled yam, pounded yam and yam flour, respectively. The result also indicates that the iodine value of 592.50 ± 8.22 µg/100g dry matter observed in the raw form of cassava was higher when compared with that of garri (366.03 ± 3.82 µg/100g) and cassava flour (216.90 ± 1.03 µg/100g dry matter). Similarly, raw *Zea mays* contain higher iodine (112.24 ± 0.42 µg/100g) compared with boiled maize (79.44 ± 0.64 µg/100g dry matter) and fresh ogi (45.07 ± 1.24 µg/100g dry matter). The result also indicates that raw *Vigina uguicnlata* and *Arachis hypogea* contain higher iodine content (112.22 ± 0.22 and 119.62 ± 0.22 µg/100g dry matter, respectively) when compared with their boiled forms (97.33 ± 1.53 and 83.12 ± 1.35 µg/100g, respectively). Results obtained in this study thus indicate that processing significantly reduces iodine content of food products, hence consideration must be given to different processing methods when accessing lodine intake from different processed foods.

Key words: Processing, iodine content, food products, fermentation, boiling, roasting.

# INTRODUCTION

lodine is an essential trace element of great importance in human nutrition. The element is an integral part of the thyroid hormones (Dunn and Dunn, 2001). Recommended daily allowance of dietary iodine is 180-200  $\mu$ g for adults, > 100  $\mu$ g for children and the daily intake during pregnancy should be at least 230  $\mu$ g iodine (Horst et al., 2005). As iodine is essential for normal brain development (Delange, 2000), it is particularly important that the foetus and young children have adequate intakes (Venturi et al., 2000).

The term iodine deficiency disorder is used to describe the wide range of effects low iodine status can have on health. Iodine deficiency disorders due to iodine decrease or loss during processing or cooking is still a major public health problem in several areas of the world, especially in developing countries (Ozdemir et al., 2009; Wisnu, 2008). One of the serious health effects of iodine deficiency disorders is goiter (Roti and Uberti, 2001; Rasmussen et al., 2002). In very severe iodine deficiency, stunted growth and mental retardation can occur in children (Laurberg, 2004). A number of studies have reported adverse effects on hearing capacity, motor and cognitive function in children associated with moderate and severe iodine deficiency (Venturi et al., 2000; Ruwhof and Drexhage, 2001; Mann and Aitken, 2003).

Food processing is one of the earliest technologies that man has been using. It has the advantage of insuring food supply and in many cases as a necessary step before consumption for a variety of reason, such as increasing stability, improving flavour, decreasing possibility of toxicity and introducing functioning ability.

<sup>\*</sup>Corresponding author: E-mail: immanbisi@yahoo.com. Tel: 08055533192.

Some of these processing methods are fermentation, frying and cooking (Raghuuath and Belavady, 1997; Prablivathi and Narasinga, 1979). These processing also introduce chemicals that affect nutritional values adversely. Certain food processing practices often increase the amount of iodine in foods. For example, the addition of potassium iodide to table salt to produce "iodized" salt has dramatically increased the iodine intake of people in developed countries (Clark et al., 2002; Diosady et al., 1997). In addition, iodine-based dough conditioners are commonly used in commercial bread-making, which increases the iodine content of the bread (Connolly et al., 1970). When in elemental form iodine readily sublime and is then rapidly lost to the atmosphere (Laurberg, 2004), its iodate form such as potassium iodate can be reduce to elemental iodine by a variety of reducing agents (Diosady et al., 1997). The absorption and/or utilization of iodine are inhibited by components of certain foods. These food components, called goitrogenic compounds. are found primarily in cruciferous vegetables (for example, cabbage and broccoli), soybean products, cassava root, peanuts, mustard, and millet. Over consumption of these foods may lead to thyroid problems by reducing the amount of available iodine for the manufacture of thyroid hormones (Kontras et al., 2002; Soetan, 2008). It is believed that cooking can inactivate the goitrogenic compounds in these foods, thereby eliminating their negative impact on iodine status (Kontras et al., 2002).

In our previous study, we reported on the iodine content of some tubers, fruit and vegetables locally grown and consume in Ijebu North Local Government Area of Ogun State, Nigeria (Salau et al., 2008). Till date, very few studies have been conducted to assess the losses of iodine during cooking procedures. As with most chemical reactions, an increase in temperature (as encountered in cooking and drying) increases rate of reaction that forms elemental iodine and increase its evaporation rate. In light of the above we set to investigate effect of some local processing methods on iodine content of some selected plants food grown and consumed in Ijebu-North local government of Ogun state, Nigeria.

# MATERIAL AND METHODS

#### Sample collection

Yam, cassava tubers, maize, cowpeas and groundnuts were purchase from two major markets in ljebu-Igbo and Ago-Iwoye in Ijebu-North Local government. Six samples, from each market were purchased randomly. The sample weight varied between 1-5 kg. The six samples for each product were thoroughly mixed together.

#### Sample preparation

#### Yam

The edible portion was prepared by cleaning the dirt with water and

removing the peel with knife. The moisture and iodine content was then determined. The edible portion was further cut into pieces and later divided into six portions for moisture and iodine content determination. The first two (2) portions were mixed together and boiled. The data generated from this was recorded for boiling. The other two (2) portions were first boiled and used for pounded yam. The last two (2) portions were used for yam flour following the method of Song (1992).

#### Cassava

The edible portion was also prepared as described for yam. Two other portions were mixed together and used for cassava flour (lafun). The last two portions were used for cassava meal (garri). The garri was prepared using local method. The edible portion was first grated, dewatered, fermented, sieved and then fried (Song, 1992).

#### Maize

The maize was cleaned and washed in water. 100 g of sample was collected from each portion for moisture and iodine content determination. They were later divided into six portions. Three (3) portions each were mixed together into two parts. One part was boiled and the other part used for ogi following local methods as described by Augustine (1991).

#### Cowpea and groundnuts

Edible portions were prepared by cleaning and washing with water. The portions were later divided into two parts. One part was used as raw sample while the other part was boiled in water. Samples were collected in six replicate for each analysis

#### Sample analysis

#### **Moisture Content**

Thirty (30) g of each sample were taken (6 replicate) into 200 ml crucible, dried in an oven at 105°C for 24 h and moisture content was thereafter determined.

### Ashing

Five (5) g of each dried sample was pulverized using mortar and pestle. Two (2) g of powdered sample was taken and placed in ash crucible. They were then mixed with five (5) g of  $Na_2CO_3$ , 5 ml of 0.5 M NaOH and 10 ml ethanol. The sample was placed in the steam bath at 100°C for about 20 min and later transferred to carbolite furnace for about 15 min at 500°C.

#### **lodine content**

The iodine content was analysed according to Elmsie Caldwells methods as modified by Diosady and Fitzgerald (1983).

#### Statistics

The experimental design was completely randomized. Data were analyzed using the SPSS. Significant difference between the data was determined at p < 0.05 using Duncan multiple range test.

Yam and its Products	Moisture content (g %)	lodine content (µg/100g dry matter)	% difference in iodine content
Fresh yam (edible portion)	68.80 ± 0.20	658.65 ± 17.26 <sup>a</sup>	Not determined
Roasted yam	24.60 ± 0.09	592.48 ± 5.82 <sup>b</sup>	-9.94
Boiled yam	74.91 ± 2.31	$448.60 \pm 2.46^{\circ}$	-31.89
Yam flour	12.00 ± 1.68	$307.40 \pm 3.66^{d}$	-53.33
Pounded yam	64.53 ± 1.82	248.60 ± 2.16 <sup>e</sup>	-62.26

 Table 1. lodine content of yam and its products.

Values are expressed as mean of 6 determinations  $\pm$  SEM.

Values in the same column with thee same superscript are not significantly different from each others.

Table 2. lodine and Moisture content of cassava (Maintial esculena) and its products.

Cassava and its product	Moisture content (g %)	lodine content (μg/100g dry matter)	% difference in iodine content
Fresh cassava tuber (edible portion)	58.25 ± 1.18	$592.50 \pm 8.22^{a}$	Not determined
Cassava flour (lafun)	14.15 ± 0.22	216.90 ± 1.23 <sup>b</sup>	-63.39
Cassava meal (garri)	18.32 ± 0.38	$366.03 \pm 3.82^{\circ}$	-38.22

Values are expressed as mean of 6 determinations ± SEM.

Values in the same column with the same superscript are not significantly different from each others.

Table 3. lodine and iodine content of maize (Zea maize) and its product.

Maize and its products	Moisture content (g %)	lodine content (μg/100g)	% difference in iodine content
Dry maize (edible portion)	0.05±0.88	112.24±1.42 <sup>a</sup>	Not determined
Boiled maize	51.01±1.23	79.44±2.64 <sup>b</sup>	-29.22
Ogi	66.04±1.21	45.01±1.23 <sup>°</sup>	-59.90

Values are expressed as mean of 6 determinations ± SEM.

Values in the same column with the same superscript are not significantly different from each others.

# RESULT

Shown in Table 1 is the result of the moisture and iodine content of yam and yam products. Iodine content of all the processed products were significantly (p < 0.05) lowered than the raw food. The least reduction (9.94%) was observed in roasted yam, while the highest reduction (62.26%) was observed in pounded yam. The result of moisture and iodine content of cassava and its products is shown in Table 2. The two processed products were significantly lowered (p < 0.05) in iodine content when compared with the raw edible portion. The least reduction in iodine content was observed in garri (38.22%) while lafun has the highest reduction (63.39%)

Table 3 is the result of the moisture and iodine contents of maize and its products. Boiling of maize and processing to ogi was observed to significantly reduce the iodine content when compared with the raw edible portion. Highest loss of iodine (59.90%) was observed in ogi whereas as a percentage iodine reduction of 29.22% was observed in the boiled maize. The result of the moisture and iodine content of cowpea and groundnut (raw edible portions and their products) are shown in Table 4. The result indicates that boiling significantly reduced iodine content by 29.95% in cowpea and 18.63% in groundnut. No significant difference was observed between the iodine content of raw groundnut and cowpea whereas the iodine content of boiled groundnut was significantly higher than that of the boiled cowpea.

No significant correlation was observed between the iodine content and moisture content of all the products.

# DISCUSSION

The iodine content of foods is generally reflective of background levels as well as processing technology and manufacturing practices. For example, the high iodine content of milk and dairy products has been attributed to the use of iodine-containing supplements in feed for dairy cattle, iodophor-based medications, teat dips and udder washes as well as iodophors used as sanitizing agents in

Raw processed form	Moisture content (g %)	lodine content (μg/100g)	% difference in iodine content
Raw cowpea (edible portion)	11.15 ± 1.33	112.22 ±1.25 <sup>ª</sup>	Not determined
Boiled cowpea	55.46 ± 2.92	83.12± 1.35 <sup>b</sup>	-29.93
Raw groundnut edible portion	21.12 ± 1.30	119.62±3.11 <sup>a</sup>	Not determined
Boiled groundnut	61.21 ± 2.48	97.33±1.54 <sup>°</sup>	-18.63

**Table 4.** lodine and Moisture content iodine content of cowpea (*Vigina uguiculata*), groundnut (*Arachis hypogea*) and their products.

Values are expressed as mean of 6 determinations ± SEM.

Values in the same column with the same superscript are not significantly different from each others.

dairy processing establishments (Fischer and Giroux, 1987; Park et al., 1981; Pennington et al., 1986). The elevated iodine levels found in grain and cereal products are related to endogenous iodine in ingredients but, in addition, likely reflect the use of iodine-containing food additives, such as iodate dough conditioners (Varo et al., 1982; Katamine et al., 1986).

In the present study, the major processing methods used were boiling, frying, drying and fermentation. The study indicates that fermentation leads to the greatest loss in iodine content as observed in yam (53.3%), cassava flour (63.39%) and ogi (59.90%). The reduction could be attributed to leaking of iodine into the soaking water and also possibly by evaporation especially during drying of yam and cassava flour. This is in agreement with previous study which reported reduction in certain nutrients during fermentation and drying (Binita and Khetarpaul, 1997; Varo et al., 1982; Nelson and Philip, 1985).

Few older studies with rather inconsistent results exist on the influence of household preparations and processing on the iodine content of foods (Manthey, 1989; Varo et al., 1982; Montag and Grote, 1981). Manthey et al. (1989) found an increase of iodine after cooking and frying of saithe portions of fish, whereas Montag and Grote (1981) reported a considerable decrease in the iodine content after household preparation of plaice and haddock. Our study indicates that cooking reduced iodine content of roasted yam by 10.04%, whereas boiling reduced the iodine content of yam by 31.89%, maize by 29.22%, cowpea by 29.93% and groundnut by 18.63%. Our result indicates a percentage reduction in iodine content that varies from 18.63 to 31.89% as a result of roasting and boiling. The difference observed may be due to differences in cooking time and the nature of plant material. In a previous study, it was reported that roasting of meatballs for 15 min resulted in a 5% loss of iodine. Extension of thermal requirement to 30 min resulted in an increase, almost threefold, of iodine loss (Ozdemir et al., 2009). The same study also reported that the decrease of iodate content and conversion to other iodine species is affected by acidity, moisture content, heating during cooking process and is also influenced by the type of cooking spices and raw materials used.

The observed difference in iodine content between the boiled and pounded yam could be attributed to the release of iodine during pounding.

# Conclusion

The result of this study indicates that local processing method reduced iodine content of the selected plant food grown and consumed in Ijebu-north Local Government Area of Ogun State, Nigeria. Though root and tuber contain high level of iodine, the final content at consumption stage is very low. Thus, improved method of processing of the local plant food available in this area in order to ensure that the dietary requirement of iodine is met by people leaving in this area is met may be advocated.

#### REFERENCES

- Augustine EN (1991). Utilization and processing of maize IITA research guide 35: p. 13.
- Binita R, Khetarpaul N (1997). Probiotic fermentation: Effect on antinutrients and digestibility of starch and protein of indigenous developed food mixture. J. Nutr. Health, pp. 139-147.
- Clark C, Wu T, Liu G, Li P (2002). Iodized salt for iodine deficiency: A systematic review. Endocrinol. Metals Clin. North Am.3: 681-698.
- Connolly RJ, Vidor GI, Stewart JC (1970). Increase in thyrotoxicosis in endemic goiter area after iodation of bread. Lancet, 1: 500-502
- Delange F (2000). The role of iodine in brain development. Proc. Nutr. Soc. 59: 75-79.
- Diosady II, Albert JO, Venkatesh MG, Shone TG (1997). Stability of iodine in iodized salt used for correction of iodine deficiency disorders. Food nutria bulletin. 18: p. 41.
- Diosady LL, Fitzgerald D (1983). A sensitive kinetic assay of determination of iodine in cereals. Food Biochem. 72: 248-254.
- Dunn JT, Dunn AD (2001). Update on intrathyroidal iodine metabolism. Thyroid, 11: 407-414.
- Fischer PWF, Giroux A (1987). lodine content of a representative Canadian diet. J. Can. Diet. Assoc. 48: 24-27.
- Katamine S, Mamiya Y, Sekimoto K, Hoshino N, Totsuka K, Naruse U, Watabe A, Sugiyama R, Suzuki M (1986). Iodine content of various meals currently consumed by urban Japanese. J. Nutr. Sci. Vitaminol. (Tokyo), 32: 487-495.
- Kontras DA, Matovinovic J, Vought R (2002). The ecology of iodine In: Stanbury; J. B Hetzel BS (eds). Endemic goiter and endemic cretinism. New York John Wiley publisher pp.185-195.
- Laurberg P (2004). Victories and Challenges in Optimizing Iodine Intake. Thyroid, 14: 589.
- Mann J, Aitken E (2003). The re-emergence of iodine deficiency in New Zealand? N. Z. Med. J. 351: 1161-1170.

- Nelson M, Phillips DIW (1985). Seasonal variations in dietary iodine intake and thyrotoxicosis. Hum. Nutr. 39: 213-216.
- Ozdemir Y, Olcay S, Sefik K (2009). Estimation of iodine loss from iodised salt during different storage and cooking conditions. 3rd International Congress on Food and Nutrition 22-25 April 2009 Antalya, Turkey.
- Park YK, Harland BF, Vanderveen JE, Shank FR, Prosky L (1981). Estimation of dietary iodine intake of Americans in recent years. J. Am. Diet. Assoc. 79: 17-23.
- Pennington JAT, Young BE, Wilson DB, Johnson RD, Vanderveen JE (1986). Mineral content of foods and total diets: The selected minerals in foods survey 1982 to 1984. J. Am. Diet. Assoc. 86: 876-891.
- Prablivathi T, Narasinga BS (1979) effect of domestic preparation of cereals and legumes on ionizable iron. J. Sci. Food Agric. 30: 597-602.
- Raghuuath M, Belarady B (1997) Riboflavin and total vitamin B6 content of in down pulses. Varietals difference and the effect of cooking. J. Plant Food, 3: 205-210.
- Rasmussen LB, Ovesen L, Bulow I (2002). Relations between various measures of iodine intake and thyroid volume, thyroid nodularity, and serum thyroglobulin. Am. J. Clin. Nutr. 76: 1069-1076.
- Roti E, Uberti ED (2001). Iodine excess and hyperthyroidism. Thyroid, 11: 493-500.
- Ruwhof C, Drexhage HA (2001). lodine and thyroid autoimmune disease in animal models. Thyroid, 11: 427-436.
- Salau BA, Ketiku AO, Ajani EO, Ezinma EN, Idowu GO, Soladoye MO (2008). Iodine contents of some selected root tubers, cereals and legumes consumed in Nigeria. Afr. J. Biotechnol. 7: 4328-4330.

- Song KH (1992). Traditional processing and utilization of cassava in Africa IITA research guide 41.
- Soetan KO (2008). Pharmacological and other beneficial effects of antinutritional factors in plants. A Review. Afr. J. Biotechnol. 7:4713-4721.
- Varo P, Saari E, Paaso A, Koivistoinen P (1982). Iodine in Finnish foods. Int. J. Vitamin Nutr. Res. 52: 80-89.
- Venturi S, Donati FM, Venturi A (2000). Role of iodine in evolution and carcinogenesis of thyroid, breast and stomach. Adv. Clin. Pathol. 4: 11-17. PMID.
- Wisnu C (2008). Determination of iodine species content in iodized salt and foodstuff during cooking. Int. Food Res. J. 15: 325-330.
- Manthey M (1989). Gehalte an Natrium, Kalium, Jod und Fluorid in Fischer-Zeugnissenl. Deut. Lebenin-Rundsch. 85:318-321.
- Montag A, Grote B (1981). Untersuchungen zur Jod-Brom-Relation in Lebensmitteln. Z. Lebensmittel-Untersuch. Forsch. 172: 123-128.
- Horst K, Serden B, Sunna Z, Peter Q (2005). Chnages of iodine content in fish during household preparation. Deut. Lebensm. Rundsch. 101: 10-16.