Tropical Journal of Pharmaceutical Research December 2017; 16 (12): 2985-2990 ISSN: 1596-5996 (print); 1596-9827 (electronic) © Pharmacotherapy Group, Faculty of Pharmacy, University of Benin, Benin City, 300001 Nigeria.

> Available online at http://www.tjpr.org http://dx.doi.org/10.4314/tjpr.v16i12.24

Original Research Article

Characterization and identification of *in vitro* metabolites of (-)-epicatechin using ultra-high performance liquid chromatography-mass spectrometry

Rui Jun Cai, Xiao Ling Yin*, Jing Liu, Da Xu Qin, Gui Zhen Zhao

Department of Pharmacy, The People's Hospital of Jiuquan, Jiuquan, GanSu 735000, China

*For correspondence: Email: 496014540@qq.com; Tel: +86-937-6982217

Sent for review: 4 June 2017

Revised accepted: 27 November 2017

Abstract

Purpose: To characterize and identify metabolites of (-)-epicatechin in microsomal fraction of rat hepatocytes (MFRHs).

Methods: A single incubation of (-)-epicatechin (1 mL, 50 μ g/mL) in MFRH (0.5 mg/mL) was used for the generation of metabolites. Thereafter, the sample was subjected to protein precipitation prior to analysis with ultra-high performance liquid chromatography coupled to linear ion-trap orbitrap mass spectrometry (UHPLC-LTQ-Orbitap MS).

Results: Nine metabolites of (-)-epicatechin were characterized on the basis of high resolution mass measurement, MS spectra and literature data. Based on their structures, the major metabolic routes of (-)-epicatechin in MFRHs were identified as hydroxylation, dihydroxylation and glycosylation.

Conclusion: This is the first report on metabolites of (-)-epicatechin in MFRHs, and it is helpful in gaining deeper insight into the metabolism of (-)-epicatechin in vivo. The results will also provide guidance in research on the pharmacokinetics of new drugs.

Keywords: (-)-Epicatechin, Metabolites, Hydroxylation, Dihydroxylation, Glycosylation, Rat liver microsomes, Pharmacokinetic studies

This is an Open Access article that uses a funding model which does not charge readers or their institutions for access and distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0) and the Budapest Open Access Initiative (http://www.budapestopenaccessinitiative.org/read), which permit unrestricted use, distribution, and reproduction in any medium, provided the original work is properly credited.

Tropical Journal of Pharmaceutical Research is indexed by Science Citation Index (SciSearch), Scopus, International Pharmaceutical Abstract, Chemical Abstracts, Embase, Index Copernicus, EBSCO, African Index Medicus, JournalSeek, Journal Citation Reports/Science Edition, Directory of Open Access Journals (DOAJ), African Journal Online, Bioline International, Open-J-Gate and Pharmacy Abstracts

INTRODUCTION

(-)-Epicatechin and its isomer (+)-catechin, which belong to the flavan-3-ol family, are ubiquitously distributed in plants [1]. They have been used in many fields, such as food and medical industries due to their anti-oxidative [2], anti-microbial [3,4], and cardio-protective properties [5]. Research on the metabolism of drugs is a very important step in discovery of new drugs. It is also vital for drug development, pharmacokinetics, and clinical pharmaceutics [6]. However, not much is known about the metabolites of (-)-epicatechin [7]. On the other hand, more than 40 metabolites of the isomer of (-)-epicatechin, (+)-catechin have been isolated and characterized through *in vitro* and *in vivo* studies [8-11].

The technique of liquid chromatography coupled with mass spectrometry (LC-MS) is a useful tool for studying the metabolites of drugs in *vitro or in vivo* [12]. However, ultra–high performance liquid

© 2017 The authors. This work is licensed under the Creative Commons Attribution 4.0 International License

chromatography (UHPLC) can provide a higher and faster separation, and less solvent consumption than LC-MS [13], while highresolution mass spectrometry (HRMS) can provide elemental composition by accurate mass measurement. Therefore, UHPLC-HRMS has been widely used for the characterization of drug metabolites among several different LC/MS platforms [14-15].

The aim of this investigation was to isolate and characterize (-)-epicatechin metabolites from MFRHs.

EXPERIMENTAL

Reagents

Authentic (-)-epicatechin standard (purity > 98.0 %) was product of Chengdu Biopurify Phytochemicals Co, Ltd (Sichuan, China). Acetonitrile (HPLC grade) was purchased from Fisher Scientific (Fisher, Fair Lawn, USA). Reduced nicotinamide adenine dinucleotide phosphate (NADPH) was supplied by Zhong Sheng Rui Tai Biotech (Beijing, China), while a Milli-Q system (Millipore, MA, USA) was used for preparing distilled water. Rat liver microsomes were product of BD Biosciences (Bedford, MA, USA). Magnesium chloride and Tris-HCL buffer used in this experiment (pH 7.4) were of analytical grade.

Incubation of microsomes

Metabolic transformation was carried out in vitro in a final volume of 1 mL by incubating (-)epicatechin with RLMs in a shaking water bath at 37 °C. (-)-Epicatechin (50 µg/mL) was preincubated for 5 min in 0.1 mol/L Tris-HCl buffer (pH 7.4) containing 5mM MgCl₂ and 0.5 g/mL MFRHs. Next, NADPH (1 mM) was added to initiate the reaction. The reaction was allowed to proceed for I h. and then stopped by introduction of ice-cold acetonitrile (1 mL) to the reaction mixture. The mixture was vortexed and clarified by centrifugation at 4 °C for 10 min at 15,000 rpm. The metabolites were identified by injection of 5 µL of the supernatant into UHPLC-LTQ-Orbitrap MS. A solution prepared in a similar manner but lacking (-)-epicatechin served as blank. All analyses were done in triplicate.

UHPLC-LTQ-Orbitrap analysis conditions

All UHPLC-LTQ-Orbitrap analyses were performed with LTQ/Orbitrap XL hybrid mass spectrometer (Thermo Electron, Germany) equipped with An Accela UHPLC system (Thermo Fisher Scientific) comprising an auto sampler, a de-gasser component and a quaternary pump, via an electrospray ionization source (ESI) (Thermo Electron, Bremen, Germany). Column chromatography was carried out at room temperature using an Acquity™ UPLC BEH C18 column (2.1 \times 50 mm, 1.7 μ m) with a mobile phase of water (solvent A) and acetonitrile (solvent B) through gradient elution (0.2mL per/min) as follows: 0.2 mL/min: 0 - 2 min, maintained at 5 % B; 2 - 3 min, increased from 5 to 10 % B; 3 - 15 min, increased from 10 to 15 % B; 15 - 17 min, increased from 15 to 80 % B; 17 – 23 min, maintained at 80 % B; 23 – 24 min, decreased from 80 to 5 % B; 24 - 28 min, maintained at 5 % B.

Negative ion mode was used in the MS/MS operations, and mass analysis was carried out at a resolution of 30,000 in the range of 100 to 800 m/z. The capillary voltage, source voltage, capillary temperature, tube lens sheath gas flow and auxiliary gas flow rate were 35 V, 3.0 kV, 350 °C, 110 V, 30 and 10 arbitrary units, respectively.

Data processing

Data acquisition and processing were performed with Thermo X caliber 2.1 workstation (Thermo Fisher Scientific), and the results were compared with that from blank MFRH samples obtained under identical conditions.

RESULTS

Metabolic routes of (-)-epicatechin

In order to identify the metabolites of (-)epicatechin, the first step of this work was to study the MS² fragmentation pattern of (-)epicatechin. The parent ion [M-H]⁻ was at *m/z* 289.0704 (-0.9 ppm, $C_{15}H_{13}O_6$) and the MS² spectrum yielded fragment ions at m/z 245.0811 (1.1 ppm, $C_{14}H_{13}O_4$), *m/z* 205.0499 (1.8 ppm, $C_{11}H_9O_4$) and *m/z* 179.0346 (4.0 ppm, $C_9H_7O_4$) by loss of CO₂ moiety (44 Da), $C_4H_4O_2$ (84 Da), and $C_6H_6O_2$ (110 Da) (Figure 1). These product ions aided the identification of the metabolites of (-)-epicatechin.

Identified metabolites

The results from high resolution extracted ion chromatography (HREIC) are shown in Figure 2. Nine metabolites of (-)-epicatechin were tentatively characterized on the bases of high resolution mass measurements, MS spectra, and literature data in negative ion mode. The UPLC-MS data are summarized in Table 1. Metabolite M0 was confirmed as (-)-epicatechin through comparison of the retention time, high resolution mass measurements, and MS^2 spectra with authentic references.

Metabolites M3, M6, M8 and M9 eluted at 7.59, 12.53, 14.53, and 16.87 min, respectively, and possessed un-protonated molecular ion [M-H]⁻ at m/z 305.0652 (-1.3 ppm, C₁₅H₁₃O₇), m/z 305.0655 (0 ppm, C₁₅H₁₃O₇), m/z 305.0654 (-0.7 ppm, C₁₅H₁₃O₇), and m/z 305.0651 (-1.7 ppm, C₁₅H₁₃O₇). These ions were by 16 Da higher than that of (-)-epicatechin, implying that they were derived from (-)-epicatechin by hydroxylation. The diagnostic product ions at m/z 179.0342 (1.8 ppm, C₉H₇O₄), m/z 179.0342 (1.8 ppm, C₉H₇O₄), were observed

in their MS² spectra, which confirmed that they were metabolites of (-)-epicatechin. Therefore, they were tentatively characterized as hydroxylated products of (-)-epicatechin.

The compounds M1 and M2 were eluted at 6.54 and 7.38 min with the same deprotonated molecules at m/z 331.0806 (-1.9 ppm, C₁₇H₁₅O₇), which was a +36 Da (2O)-shift from that of (-)epicatechin. The MS2 spectra of M1 and M2 showed the major fragment ions at m/z 313, m/z287, and m/z 269, due to loss of H₂O, CO₂, and H₂O+ CO₂, respectively from the parent drug which had similar fragmentation pattern with (-)epicatechin. Based on previous analyses, they were identified as dihydroxylated products of (-)epicatechin.



Figure 2: Results of HREIC analysis of (-)-epicatechin metabolites in MFRHs: *m/z* 305.0655, 331.0812, 451.1235

Peak	t _R	Theoretical mass <i>m/z</i>	Experimental mass <i>m/z</i>	Error (ppm)	Formula [M- H] ⁻	MS/MS fragment	Transformation
1	6.54	331.0812	331.0806	-1.9	C ₁₇ H ₁₅ O ₇	MS ² [331]: 313.0700 (100), 287.0910 (49), 269.0794 (32), 205.0495 (22)	Dihydroxylation
2	7.38	331.0812	331.0806	-1.9	C ₁₇ H ₁₅ O ₇	MS ² [331]: 313.0699 (100), 287.0909 (49), 269.0798 (38), 205.0498 (16)	Dihydroxylation
3	7.59	305.0655	305.0652	-1.3	C ₁₅ H ₁₃ O ₇	MS ² [305]: 179.0342 (100), 221.0444 (76), 219.0651 (74)	Hydroxylation
4	8.97	451.1235	451.1230	-1.2	$C_{21}H_{23}O_{11}$	MS ² [451]: 313.0700 (100), 355.0803 (63), 289.0701 (8)	Glycosylation
5	9.82	451.1235	451.1229	-1.4	C ₂₁ H ₂₃ O ₁₁	MS ² [451]: 313.0911 (100), 289.0701 (66), 245.0805 (37)	Glycosylation
0	12.2 9	289.0707	289.0704	-0.9	$C_{15}H_{13}O_6$	MS ² [209]: 245.0500 (100), 205.0810 (40), 203.0709 (19),179.0345 (15), 231.0291 (9), 271.0603 (6)	(-)-epicatechin
6	12.5 3	305.0655	305.0655	0.0	$C_{15}H_{13}O_7$	MS ² [305]: 179.0337 (100), 219.0653 (75), 221.0446 (70)	Hydroxylation
7	13.2 0	451.1235	451.1228	-1.6	$C_{21}H_{23}O_{11}$	MS ² [451]: 289.0703 (100), 245.0806 (58)	Glycosylation
8	14.5 3	305.0655	305.0654	-0.7	$C_{15}H_{13}O_7$	MS ² [305]: 179.0342 (100), 219.0652 (76), 221.0445 (72)	Hydroxylation
9	16.8 7	305.0655	305.0651	-1.7	C ₁₅ H ₁₃ O ₇	MS ² [305]: 179.0342 (100), 219.0650 (79), 221.0443 (74)	Hydroxylation

Table 1: Fragment ions from (-)-epicatechin metabolism in MFRHs identified by UHPLC-LTQ-Orbitrap MS

Compounds M4, M5, and M7 appeared after 8.97, 9.82 and 13.20 min with un-protonated [M-H]⁻ at m/z 451.1230 (-1.2 ppm, C₂₁H₂₃O₁₁), m/z 451.1229 (-1.4 ppm, C₂₁H₂₃O₁₁), and m/z 451.1228 (-1.6 ppm, C₂₁H₂₃O₁₁), 162 Da (C₆H₁₀O₅) higher when compared with (-)-epicatechin. Fragments at m/z 289.0701 (-2.0 ppm, C₁₅H₁₃O₆), m/z 289.0701 (-2.0 ppm, C₁₅H₁₃O₆), and m/z 289.0703 (-1.3 ppm, C₁₅H₁₃O₆) by loss glucosyl (C₆H₁₀O₅, 121 Da) moiety relative to the precursor ion at m/z 451 in their MS² spectra indicated that a glucosyl moiety was present. Thus, they were presumed to be glucosyl products of (-)-epicatechin.

DISCUSSION

The results obtained from preliminary trials with various mobile phase systems in this study showed that good chromatographic peak shapes could be achieved by using mobile system without formic acid, while much lower column pressure was afforded by the inclusion of acetonitrile in the mobile phase. Therefore, the mixture of acetonitrile and water was chosen as the mobile phase solvent system in this study. A gradient elution pattern was adopted.

(-)-Epicatechin is characterized by the presence of many hydroxyl groups on the flavanol skeleton, which make it suitable for detection by ESI in negative mode [15]. To the best of our knowledge, many studies have so far focused mainly on the metabolites of (+)-catechin, while not much has been done on the elucidation of the metabolites of (-)-epicatechin. For example, 40 and 58 metabolites of (+)-catechin were found in in vitro and in vivo investigations, respectively [8-11]. Among these, 7 metabolites were identified in MFRHs [16]. It is worth mentioning that (-)-epicatechin has a metabolic pathway similar to that of (+)-catechin in MFRHs. Therefore, it is scientifically reasonable to assume that the in vivo metabolisms of (-)epicatechin and (+)-catechin will follow a similar route. This will be helpful in the identification of metabolic products of (-)-epicatechin in vivo.

CONCLUSION

The *in vitro* metabolites of (-)-epicatechin have been successfully identified based on UHPLC-LTQ-Orbitrap analysis. Based on the 9 metabolites identified and their MS data, the major metabolic pathways of (-)-epicatechin in RLMs are hydroxylation, dihydroxylation and glycosylation. This is the first report on the metabolites of (-)-epicatechin in RLMs and is considered a useful guide for understanding its metabolism *in vivo*. In addition, the methodology used in this study offers a new perspective for studying the metabolism and pharmacokinetics of new drugs.

DECLARATIONS

Acknowledgement

The authors gratefully acknowledge Dr di Jia for supporting this project.

Conflict of interest

The authors declare that no conflict of interest is associated with this study.

Contribution of authors

We declare that this work was done by the authors named in this article and all liabilities pertaining to claims relating to the content of this article will be borne by the authors. Xiao Ling Yin planned and designed the research; Rui Jun Cai and Jing Liu performed the experiments, Ruijun Cai wrote the manuscript, while Da Xu Qin and Gui Zhen Zhao analyzed the data.

REFERENCES

- Porter LJ. In The Flavonoids, Advances in Research since 1986; Harborne, J. B., Ed.; Chapman & Hall: Landon, 1994; pp 23–56.
- 2. Abd El-Aziz TA, Mohamed RH, Pasha HF and Abdel-Aziz HR. Catechin protects against oxidative stress and inflammatory-mediated cardiotoxicity in adriamycintreated rats. Clin Exp Med 2012; 12: 233–240.
- Dai CM, Wang JB, Kong WJ, Cheng P, Xiao XH. Investigation of anti-microbial activity of catechin on Escherichia coli growth by microcalorimetry. Environ Toxicol Phar 2010; 30(3): 284-288.
- Rawdkuen S, Suthiluk P, Kamhangwong D, Benjakul S. Antimicrobial activity of some potential active compounds against food spoilage microorganisms. Afr J Biotechnol 2012; 11: 13914–13921.
- Nagao T, Hase T, Tokimitsu I. A green tea extract high in catechins reduces body fat and cardiovascular risks in humans. Obesity 2007; 15(6): 1473-1483.
- LaDu BN, Mandel HG, Way EL. (eds.). Fundamentals of Drug Metabolism and Drug Disposition, Baltimore: Williams & Wilkins, 1971. pp. 569-572.
- Harada M, Kan Y, Naoki H, Fukui Y, Kageyama N, Nakai M, Miki W and Kiso Y. Identification of the major antioxidative metabolites in biological fluids of the rat with ingested (+)-catechin and (-)-epicatechin. Biosci Biotechnol Biochem 1999; 63: 973-977.
- 8. Liang J, Xu F, Zhang YZ, Zang XY, Wang D, Shang MY, Wang X, Chui DH, Cai SQ. The profiling and identification of the metabolites of (+)-catechin and study

on their distribution in rats by HPLC-DAD-ESI-IT-TOF-MSn technique. Biomed Chromatogr 2014; 28(3): 401-411.

- Das S, Lamm AS, Rosazza JPN. Biotransformation of (+)-Catechin to Novel B-Ring Fission Lactones. Org Process Res Dev 2011; 15: 231-235.
- Alberto MR, Gomez-Cordoves C, Manca de Nadra MC. Metabolism of gallic acid and catechin by Lactobacillus hilgardii from wine. J Agric Food Chem 2004; 52: 6465-6469.
- Das NP. Flavonoid metabolism. Absorption and metabolism of (+)-catechin in man. Biochem Pharmacol 1971; 20: 3435-3445.
- Kachlicki P, Einhorn J, Muth D, Kerhoas L, Stobiecki M. Evaluation of glycosylation and malonylation patterns in flavonoid glycosides during LC/MS/MS metabolite profiling. J Mass Spectrom 2008; 43(5): 572-586.

- 13. Nováková L, Matysová L, Solich P. Advantages of application of UPLC in pharmaceutical analysis. Talanta 2006; 68(3): 908-918.
- Cai W, Zhang JY, Liu SY, Lu JQ, Zhang HG. Metabolic profiles of 11,13α-dihydroixerin Z in rats using high performance liquid chromatography-LTQ-Orbitao mass spectrometry. Anal Method 2016; 8(4); 854-861
- 15. Cai RJ, Wei J, Jing FL, Zhao GZ, Zhang Y. Identification of metabolites of gardenin A in rat liver microsomes using ultra-high performance liquid chromatography coupled with linear ion-trap Orbitrap mass spectrometry. Trop J Pharm Res 2017: 16(2): 421-427.
- Liu GQ, Dong J, Wang H, Wan LR, Hashi Y, Chen SZ. ESI fragmentation studies of four tea catechins. Chem J Chin Univ 2009; 30(8): 1566-1570.
- Sun JH, Li XR. Identification of metabolites of (+)catechin in rat liver microsomes based on UPLC-LTQ-Orbitap and multiple mass defect filter method. Chin Hosp Pharm J 2016; 36(15): 1264-1267.