



antioxidants

Oxidative Stress in Plants

Edited by

Juan B. Barroso, Mounira Chaki and Juan C. Begara-Morales

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About the Editors

Juan B. Barroso who completed his Ph.D. in 1993 in Biochemistry and Molecular Biology, is now a Full Professor at the University of Jaén, Spain. In 2002, he established an independent research group called Biochemistry and Cell Signaling in Nitric Oxide. He has written more than 160 scientific papers and reviews published in internationally renowned journals, in addition to book chapters, and edited a variety of journal topical issues on plant nitric oxide (NO) metabolism. He also serves as Editorial Board Member of several renowned journals in plant sciences. His pioneering work includes characterization of the generation of NO and its role in plants (1999). Since then, his research group is considered an international reference in the field of NO metabolism in plants, with extensive experience in study of the metabolism of reactive oxygen and nitrogen species (ROS and RNS) in the model plant *Arabidopsis* and plants of agronomic and biotechnological interest under physiological conditions as well as exposed to different biotic and abiotic stress situations. Since NO interacts with molecules such as ROS, RNS, and RSS and with biomolecules like proteins and lipids, the research group's interest is currently focused on the study of NO bioactivity. In fact, in recent years, they have studied the identification and characterization of post-translational modifications of NO-mediated proteins (NO-PTM) and the effect they have on biological activity in plant cells. As a result of these studies, this group has recently pioneered the characterization of electrophilic lipid derivatives resulting from the interaction of nitric oxide with unsaturated fatty acids, called nitrated fatty acids (NO₂-FA), that trigger antioxidant defense mechanisms involved in the maintenance of redox homeostasis in situations of stress.

Mounira Chaki conducted her undergraduate studies at the University of Mohamed First, Morocco. Afterwards, she moved back to the University of Jaén, Spain, where she started her research career. Here, she was awarded her PhD in Molecular and Cellular Biology in 2007 with the highest qualification. During her research career, she focused on the study of the nitro-oxidative stress generated in higher plants in response to different biotic and abiotic stress situations. She is a pioneer in the study of the role of post-translational modifications mediated by nitric oxide, for example, nitration and S-nitrosylation of proteins under stress conditions. She has been trained in recognized research institutions in Spain, France, and Germany, performing cutting-edge research in laboratories led by the most outstanding researchers in the nitric oxide field. She has been awarded with the Marie Curie fellowship, corresponding to a term of two years. She is currently studying the interaction of nitric oxide and derived molecules with lipids and their physiological consequences, which remain unknown in higher plants. This new topic constitutes an important advance in the generation of new knowledge about post-translational modifications mediated by nitrated lipids and their involvement in cellular defense mechanisms. Dr. Chaki has published 44 peer-reviewed papers, most of which are in the top journals in their specific area, in addition to 16 book chapters and 5 licensed patents, as well as contributing to 8 research projects. She has served as an expert for the evaluation of international projects, as well as reviewer of numerous JCR journals. She has also edited some Special Issues in the nitro-oxidative stress field. She teaches courses in the undergraduate and official Master degree in Biotechnology and Biomedicine at the University of Jaén.

Juan C. Begara-Morales completed his Ph.D. in Molecular and Cellular Biology in 2011 with cum laude qualification and received a special doctorate award from the University of Jaén (Spain). His main research line is related to the analysis of nitric oxide (NO) signaling events under physiological and stress conditions in plants. In this field, he has contributed positively to

characterizing the functional modulation of key antioxidant systems by NO-related post-translational modifications, such as S-nitrosylation and tyrosine nitration, in response to abiotic and biotic stresses. He has participated in 8 research projects and he has published 38 scientific papers in international journals and 10 book chapters. Furthermore, he is co-author of 5 invention patents that are commercially licensed and exploited, demonstrating that the connection of his research activity to industry and its potential capacity to generate knowledge that can result in biotechnological applications. He has also edited numerous Special Issues of renowned international journals in the NO and oxidative stress field. Currently, he is a PI of a project whose main goal is to characterize the effect of the cellular oxidation status on the functional modulation of antioxidant systems during the time course of nitro-oxidative stress generated in crops subjected to different adverse environmental conditions.



Oxidative Stress in Plants

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Environmental stresses negatively affect plant growth, development and crop productivity. These adverse conditions alter the metabolism of reactive oxygen and nitrogen species (ROS and RNS, respectively). The high concentrations of these reactive species that exceed the capacity of antioxidant defence enzymes, disturb redox homeostasis, which could trigger damage to macromolecules, such as membrane lipids, proteins and nucleic acids, and ultimately result in nitro-oxidative stress and plant cell death. Significant progress has been made to understand how plants persist in these stressful environments that could be vital to improve plant crop yield. In this special issue “*Oxidative Stress in Plants*”, both original articles and detailed reviews have been published with the aim to provide an up-date view in this research area in higher plants.

In the natural environment, plants are constantly exposed to abiotic stresses, such as extreme temperatures, salt stress, drought and heavy metals that have a huge impact on agriculture worldwide and consequently, lead to massive economic losses. In this sense, three research papers have analysed the effect of abiotic stress on plant growth and development. Dr. Wani’s group [1] studied the role of *Serratia marcescens* BM1 in response to cadmium (Cd) stress in soybean plants by different physiological, biochemical and molecular assays. They found that, in Cd-stressed plants, the *Serratia marcescens* BM1 treatment not only down-regulated Cd levels and oxidative stress markers, but also up-regulated levels of osmolytes, stress-related gene expression and activities of antioxidant enzymes. These authors suggested that inoculation with the *Serratia marcescens* BM1 would promote Cd stress tolerance and phytoremediation potential. The impact of abiotic stress was also reported by Dr. Barroso’s group [2] as they demonstrated the effect of short-term low temperature stress on the metabolism of reactive oxygen and nitrogen species in *Arabidopsis* plants. These authors showed that the low temperature produces nitro-oxidative stress, and reduces cytosolic NADP-malic enzyme activity, which was negatively modulated by the protein tyrosine nitration process. In addition, they proposed that Tyr73 would be a possible residue to be involved in reducing this enzymatic activity. Moreover, Dr. Rivero’s group [3] investigated the response of tomato plants to the effects of calcium and potassium on plant tolerance to combined high-temperature and salinity stress conditions. They showed the positive effect of a rise in calcium and potassium in the nutrient medium on the improvement of oxidative stress produced under these environmental stress injuries. The authors underlined the importance of the correctly administering of nutrient solution and fertilisation to face the damaging effects of adverse conditions in plant cells.

On the other hand, plant cells develop an antioxidant defence mechanism, which includes the non-enzymatic and enzymatic antioxidants for the detoxification of ROS. However, if the ROS production is higher than the ability of the antioxidant systems to scavenge them, it can lead to oxidative stress, and finally to cell death. In this context, Dr. De Maio’s group [4] used citrus plants to investigate the modulation of poly (ADP-ribose) polymerase and antioxidant enzymes, using leaves in different developmental stages, including young, mature and senescent. Their work addressed the physiological, biochemical and molecular changes that occur in plant cells during leaf ageing.

In young leaves, photochemical and glutathione-S-transferase activities increased. However, while the ageing process advanced, the non-enzymatic antioxidant systems reduced and reached the lowest levels in senescent leaves, while poly (ADP-ribose) polymerase activity increased. In the same way, Hasanuzzaman et al. [5] discussed in an extensive review, the available and up-to-date knowledge on the Ascorbate-Glutathione pathway concerning the oxidative stress tolerance, as well as plant defence mechanisms. Furthermore, the review by Laxa et al. [6] provided up-to-date information about the response and function of ROS and RNS, mainly with regard to superoxide radicals, hydrogen peroxide and nitric oxide under drought stress conditions, and their scavenging by the antioxidant defence enzymes in several plant species. To better understand the interaction between chitosan and *Vitis vinifera* L. plants, the original article by Singh et al. [7] analysed the antioxidant potential, the total phenolic content and the expression of ROS detoxification genes in two red grapevine varieties treated by chitosan. They concluded that chitosan induced the phenolic compounds, as well as acted as the organiser for the transfer of polyphenols from the *Vitis vinifera* leaves to the berries.

Another interesting feature of this special issue focuses on investigating the other H₂O₂ targets involved in programmed cell death. Dr. Mano's group [8] studied the mechanism that increased the reactive carbonyl species in the H₂O₂-produced programmed cell death in tobacco Bright Yellow-2 cells. They suggested that H₂O₂ initially inactivates a carbonyl reductase(s), which increases the reactive carbonyl species content, leading to the activation of the caspase-3-like protease of the 20S proteasome. The authors proposed that carbonyl reductase acted as a ROS sensor for inducing programmed cell death.

In plant cells, the ROS metabolism has been widely studied in different compartments, including mitochondria, cytosol, chloroplast, cell wall, plasma membrane, apoplast, glyoxysomes and peroxisomes [9]. The review by Dr. Petřivalský's group [10] provided the present knowledge about the compartment-specific pathways of reactive oxygen species generation and decomposition in plant cells, and the mechanisms that controlling their homeostasis in cell compartments. Likewise, with a particular example at the chloroplastic level, in an in-depth review Miyake [11] summarised the current research concerning the molecular mechanisms of ROS formation and suppression in photosystem I. He established a novel molecular mechanism for the oxidation of the P700 oxidation system in photosystem I and the elimination of ROS formation from the strong relationship between the light and dark reactions of photosynthesis. Furthermore, in an original article, Lewandowska et al. [12] investigated the effect of H₂O₂ on the structure and function of Arabidopsis chloroplastic DJ-1B. They found that AtDJ-1B has double functions, namely holdase and glyoxalase activity, which responded differently to H₂O₂. Glyoxalase activity was reduced by H₂O₂, however the holdase chaperone function did not change. They also analysed the phenotype of T-DNA lines that lacked the protein, and showed that AtDJ-1B was not necessary for plant growth under stress stimuli.

In summary, to better understand the nitro-oxidative stress networks in higher plants (Figure 1), the subjects addressed in this special issue provide an update and new knowledge about ROS and RNS metabolisms in plant responses to adverse environmental stimuli and the modulation of antioxidant systems to control ROS production and accumulation.

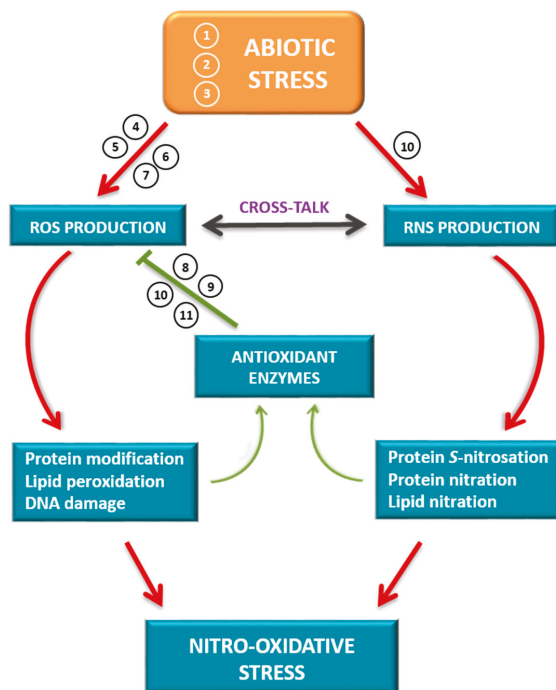


Figure 1. Schematic model of cross-talk between reactive oxygen species (ROS) and reactive nitrogen species (RNS) in plant responses to abiotic stress. Different abiotic stress situations can generate an uncontrolled production of ROS and RNS that oxidatively modify different biomolecules (proteins, lipids and nucleic acids). These modifications can lead to a gain of function of the antioxidant systems to control the production of ROS or generate a situation of cellular damage supported by a process of nitro-oxidative stress. The numbers indicate the relationship of each article in the Special Issue to the subject matter covered. (1) El-Esawi et al., 2020 [1]. (2) Begara-Morales et al., 2019 [2]. (3) García-Martí et al., 2019 [3]. (4) Biswas et al., 2020 [8]. (5) Jankú et al., 2019 [10]. (6) Miyake, 2020 [11]. (7) Lewandowska et al., 2019 [12]. (8) Arena et al., 2019 [4]. (9) Hasanuzzaman et al., 2019 [5]. (10) Laxa et al., 2019 [6]. (11) Singh et al., 2019 [7].

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Article

Serratia marcescens BM1 Enhances Cadmium Stress Tolerance and Phytoremediation Potential of Soybean Through Modulation of Osmolytes, Leaf Gas Exchange, Antioxidant Machinery, and Stress-Responsive Genes Expression

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Abstract: The heavy metal contamination in plant-soil environment has increased manifold recently. In order to reduce the harmful effects of metal stress in plants, the application of beneficial soil microbes is gaining much attention. In the present research, the role of *Serratia marcescens* BM1 in enhancing cadmium (Cd) stress tolerance and phytoremediation potential of soybean plants, was investigated. Exposure of soybean plants to two Cd doses (150 and 300 μ M) significantly reduced plant growth, biomass, gas exchange attributes, nutrients uptake, antioxidant capacity, and the contents of chlorophyll, total phenolics, flavonoids, soluble sugars, and proteins. Additionally, Cd induced the stress levels of Cd, proline, glycine betaine, hydrogen peroxide, malondialdehyde, antioxidant enzymes (i.e., catalase, CAT; ascorbate peroxidase, APX; superoxide dismutase, SOD; peroxidase, POD), and the expression of stress-related genes (i.e., *APX*, *CAT*, *Fe-SOD*, *POD*, *CHI*, *CHS*, *PHD2*, *VSO*, *NR*, and *P5CS*) in soybean leaves. On the other hand, inoculation of Cd-stressed soybean plants with *Serratia marcescens* BM1 significantly enhanced the plant growth, biomass, gas exchange attributes, nutrients uptake, antioxidant capacity, and the contents of chlorophyll, total phenolics, flavonoids, soluble sugars, and proteins. Moreover, *Serratia marcescens* BM1 inoculation reduced the levels of cadmium and oxidative stress markers, but significantly induced the activities of antioxidant enzymes and the levels of osmolytes and stress-related genes expression in Cd-stressed plants. The application of 300 μ M CdCl₂ and *Serratia marcescens* triggered the highest expression levels of stress-related genes. Overall, this study suggests that inoculation of soybean plants with *Serratia marcescens* BM1 promotes phytoremediation potential and Cd stress tolerance by modulating the photosynthetic attributes, osmolytes biosynthesis, antioxidants machinery, and the expression of stress-related genes.

Keywords: *Serratia marcescens* BM1; cadmium; soybean; osmolytes; antioxidants; genes expression

1. Introduction

Over the last few decades, there has been a growing incidence of heavy metal (HM) circulation in plant-soil continuum owing to various natural and anthropogenic activities [1]. HMs, including cadmium (Cd), pose a serious danger to the growth of crop plants, as well as human and animal health [2,3]. Cadmium is a toxic HM with strong mobility in the soil-plant interface and biological toxicity [4]. Cadmium retards vital physio-biochemical activities in plants, which include photosynthesis, biosynthesis of chlorophyll and accessory pigments, and the uptake and assimilation of essential mineral nutrients by triggering the overproduction of reactive oxygen species (ROS), including singlet oxygen ($^1\text{O}_2$), hydrogen peroxide (H_2O_2), superoxide radical ($\text{O}_2\bullet^-$), and the hydroxyl radical ($\text{OH}\bullet$) [5,6]. ROS accumulation causes high toxicity in plant cells [7–16]. Cadmium evokes peroxidation of membrane lipids and modulates the expression of genes of antioxidant defense systems in plants [17]. The presence of Cd inside plants thus causes stress by inducing ionic, osmotic, and oxidative stress [17,18]. Various studies also support the active involvement of antioxidant proteins in combating Cd-induced stress. Gene expression analysis further revealed the active participation of antioxidants genes (*Mn/SOD*, *FeSOD*, *POD*, *CAT*, *APX*, and *GR*) under Cd-induced stress [19,20].

During the course of evolution, plants have evolved intricate strategies to cope with the Cd-induced oxidative stress by activating various signalling networks, which include optimum nutrient homeostasis, enhanced accumulation of osmolytes, ROS detoxification by enhanced activities of antioxidants, and the production of thiol-related compounds [21–24]. Plant stress physiologists are engaged in devising potential sustainable strategies that can unravel mechanisms behind Cd-stress tolerance; however, the field is still emerging. Out of the various strategies that have been adopted to reverse HM-induced stress impacts in plants, the plant growth promoting rhizobacteria (PGPR) interaction is an emerging and effective sustainable way. Various recent studies in the discipline of ecological engineering and management have strongly advocated the use of PGPRs to alleviate HM-induced oxidative stress in plants, like energy crops [25], *Solanum nigrum* [26], *Spartina maritime* [27], and *Lycopersicon esculentum* [20,28,29]. Plant growth promoting rhizobacteria are known to induce tolerance against metal stress via modulating various intrinsic or underlying mechanisms [30]. These mechanisms primarily involve exclusion, extrusion, biotransformation, methylation/demethylation, and accommodation (complex formation) of metals [20,31–34]. Plant growth promoting rhizobacteria also induce the formation of different phytohormones, such as cytokinins, auxins, and gibberellins [35,36]. In addition, the synergistic interaction between microorganisms and plant roots under various abiotic pressures positively regulates the plant performance and edaphic factors [32,37,38]. *Serratia* species are potent PGPRs that are known to induce HM stress tolerance in crop plants [39]. Studies, undertaken so far, have established the role of *Serratia marcescens* in inducing HM stress tolerance in plant species. Cristani et al. [39] studied the possible action of *Serratia marcescens* in lead, Cd, and chromium metal biosorption. Khan et al. [40] analyzed the genome of *Serratia marcescens* RSC-14 and found that this species is an efficient PGPR that can alleviate Cd stress in host plants. By employing proteomic approach, Queiroz et al. [41] indicated that *Serratia marcescens* LG1 can be successfully used for adaptation and tolerance against manganese (Mn). However, the potential of the isolate *Serratia marcescens* BM1 in conferring tolerance against Cd stress in soybean (*Glycine max* L.) has not been studied yet.

Soybean is an economically important food crop worldwide and experiences various environmental stresses, including salinity and HMs, which limit the plant growth and productivity [42–44]. Therefore, enhancing the stress tolerance, productivity and phytoremediation potential of this crop is of utmost importance. Considering the importance of investigating these issues, the present research intends to elucidate the various underlying mechanisms that *Serratia marcescens* BM1 triggers in response to Cd

stress in soybean plants. Growth traits of soybean plants have been investigated. Additionally, various physiological, biochemical and molecular approaches have been assayed.

2. Material and Methods

2.1. Investigation of Cadmium Tolerance of *Serratia marcescens* BM1 Strain

Serratia marcescens BM1 strain used in the present study was isolated from the maize rhizospheric soils of the Egyptian Suez Canal area and proved its capability for indole acetic acid production and inorganic phosphate solubilization [45,46]. In the present research, Cd tolerance of *Serratia marcescens* BM1 strain was investigated in nutrient broth media, containing 0, 150, 300, and 450 μM CdCl_2 (i.e., CdCl_2 was used to provide Cd stress). Bacterial growth was estimated by measuring the optical density (OD = 600 nm) following 36, 48, 72, and 96 h of incubation at 29 °C. This experiment was conducted in four replications.

2.2. Inoculation and Growth of Soybean Plants

Following culturing of *Serratia marcescens* BM1 in nutrient broth media for 4 days at 29 °C, bacterial cells were centrifuged for 4 min at $2000\times g$ and then collected. Resultant pellets were re-suspended in sterilized water, and bacterial culture was then set to 10^8 colony-forming units (CFU) mL^{-1} and used to inoculate soybean plants. Soybean cultivar Giza 35 seeds were obtained from Legumes Institute of Kafrelsheikh in Egypt, and were then sterilized using sodium hypochlorite (8%, *v/v*) for 7 min, washed with sterilized water several times, and left to grow on a wet filter paper for 6 days at 24 °C. The 6-day old plants were then inoculated with *Serratia marcescens* BM1 suspension for 25 min and transferred into hydroponic plastic pots, containing Hoagland plant nutrient solution. Control plants were kept in fresh nutrient broth for 25 min. Experimental treatments were performed as follows: (i) control plants without CdCl_2 and bacterial inoculation (T1); (ii) plants inoculated with *Serratia marcescens* BM1 alone (T2); (iii) plants treated with 150 μM CdCl_2 alone (T3); (iv) plants treated with 150 μM CdCl_2 and *Serratia marcescens* BM1 (T4); (v) plants treated with 300 μM CdCl_2 alone (T5); and, (vi) plants treated with 300 μM CdCl_2 and *Serratia marcescens* BM1 (T6). Soybean plants were irrigated with a Hoagland nutrient solution supplemented with 0, 150, and 300 μM CdCl_2 three times a week. The pots were left in a completely randomized block design in growth chambers of a temperature of 27/19 °C (day/night) and a humidity of 76%. After 7 weeks, plants were collected for the subsequent experimental analyses.

2.3. Morphological Parameters of Plant Root and Shoot

Root and shoot lengths were determined using a measuring tape. Separated roots and shoots were washed with deionized water and weighed to measure their fresh weight. Separated roots and shoots were then oven-dried at 72 °C for 50 h to estimate their dry weights.

2.4. Measurement of Phosphorus, Nitrogen, and Cadmium Uptake

Oven-dried leaf samples were ground into a fine powder and then digested in H_2SO_4 at 190 °C for 5 h. H_2O_2 was then added to the samples and left for 60 min. Digested samples were filtered and then diluted with sterile distilled H_2O . Nitrogen concentration was calculated following Kjeldahl methodology as mentioned by Bremner [47]. Phosphorus concentration was calculated according to the protocol of Murphy and Riley [48]. To determine Cd content, dried leaf samples were grinded to fine powder and then digested in a mixture of HNO_3 : HClO_4 (3:1, *v/v*) at 120 °C. Cadmium was then quantified using an atomic absorption spectrometer (AA6300C, Shimadzu, Kyoto, Japan).

2.5. Measurements of Chlorophyll Content, Leaf Relative Water Content, and Gas-Exchange Attributes

Total leaf chlorophyll content was determined by homogenizing 0.2 g fresh leaf samples in 50 mL of 80% acetone, followed by centrifugation at $14,000\times g$ for 7 min and the absorbance was then spectrophotometrically recorded at 662 and 645 nm as reported by Lichtenthaler [49].

Net photosynthesis rate (P_n), transpiration rate (E), and stomatal conductance (g_s) were measured on expanded leaves of similar developmental stages with a portable gas-exchange system (LI-6400, LI-COR Inc., Lincoln, NE, USA) between 9:30 and 10:30 am, according to the methodology of Holá et al. [50]. Leaf relative water content (RWC) was determined as previously explained by El-Esawi and Alayafi [12].

2.6. Measurement of Total Soluble Sugars, Soluble Protein, Proline and Glycine Betaine Levels

Leaves were ground into a fine powder and homogenized in 100 mM Tris buffer (pH 8.0), followed by centrifugation at $14,000\times g$ for 14 min. Using the protocol of Dey [51], total soluble sugar content was determined by recording the absorbance at 485 nm. The Bradford method [52] was used to estimate the total protein content.

The protocol of Bates et al. [53] was used to estimate proline content. Leaf samples were digested in 5% (*w/w*) sulfosalicylic acid, and centrifuged at $10,000\times g$ for 7.0 min. Supernatant was diluted with sterile distilled water and then mixed with 2% ninhydrin, followed by heating up at 96°C for 30 min, then cooling. Toluene was then added to the mixture, and the absorbance of the upper aqueous phase formed was recorded at 520 nm. Following the protocol of Grieve and Grattan [54], glycine betaine content was estimated by extracting dry leafy samples in hot distilled water at 72°C . To the extract formed, 2 N HCl and potassium tri-iodide solution were added, mixed and cooled on ice for 2 h. Cold 1,2-dichloromethane and distilled water were then added to the mixture where two layers were formed. Organic layer absorbance was recorded at 365 nm.

2.7. Determination of Total Flavonoids and Phenols Contents

The protocol of Zhishen et al. [55] was used to estimate the total flavonoid content by homogenizing oven-dried leaf powder (1.0 g) in distilled water (100 mL), followed by filtration and mixing with a solution composed of distilled H_2O , AlCl_3 , and NaNO_2 . Few drops of NaOH was then added to the mixed solution, which was then diluted with distilled water. The mixture absorbance was recorded at 510 nm. Catechin calibration curve was used.

Total phenolic content was measured by extracting leaf samples (2.0 g) in methanol solution (10 mL, 80%), followed by agitation at 72°C for 18 min. Methanolic extract (2.0 mL) was diluted in 10 mL distilled H_2O comprising 1 N Folin–Ciocalteu reagent (500 μL), and then incubated at 30°C . The mixture absorbance was recorded at 725 nm [56] using gallic acid as a standard.

2.8. Estimation of Hydrogen Peroxide and Malondialdehyde Levels

Following the protocol of Velikova et al. [57], hydrogen peroxide (H_2O_2) content was measured by extracting leaf samples in 0.1% trichloroacetic acid (TCA), followed by centrifugation at $12,000\times g$ for 15 min. Potassium phosphate buffer (10 mM, pH 7.0) and potassium iodide (1 M) were added and well-mixed with the supernatant. The mixture absorbance was then read at 390 nm and H_2O_2 content was estimated following H_2O_2 standard curve. Using the methodology of Heath and Packer [58], malondialdehyde (MDA) level was determined by homogenizing leaf samples in 0.1% TCA followed by centrifugation at $14,000\times g$ for 6 min. Thiobarbituric acid (0.5%) and 20% TCA were added to the supernatant. The mixture was then heated up at 96°C for 25 min, followed by cooling and centrifugation at $9,000\times g$ for 12 min. The supernatant absorbance was recorded at 532 and 660 nm.

2.9. Determination of Leaf Antioxidant Capacity

Leaf antioxidant capacities were estimated by β -carotene-linoleic acid and 2,2'-diphenylpicrylhydrazyl (DPPH) assays [59,60]. Leaf samples (2.0 g) were extracted in methanol solution (10 mL, 80%), and then agitated at 72°C for 18 min. For DPPH assay, 50 μL of methanolic extract (1 mg mL^{-1}) was added to 5 mL of methanolic DPPH solution (0.004%), and then mixed and left in darkness for 25 min. The mixture absorbance was read at 517 nm. For β -carotene-linoleic acid assay, 50 μL of methanolic extract was transferred to β -carotene mixture, and then mixed and incubated for 46 h. The absorbance was read at 470 nm. The results were expressed as IC_{50} in mg mL^{-1} .

2.10. Estimation of Antioxidant Enzymes

Leaf samples were extracted in Tris-HCl (100 mM, pH 7.5) and then mixed with Dithiothreitol (5 mM), PVP-40 (1.5%), MgCl₂ (10 mM), EDTA (1 mM), magnesium acetate (5 mM), phenylmethanesulfonyl fluoride (1 mM), and aproptinin (1 µg mL⁻¹). The mixture was filtered and then centrifuged at 14,000× g for 8 min. The supernatants were utilized for estimating antioxidant enzymes activities. The protocol of Nakano and Asada [61] was used to estimate ascorbate peroxidase (APX) activity by extracting leaf samples in 2 mM AsA, followed by recording the absorbance at 265 nm. Moreover, the methodology of Aebi [62] was used to estimate catalase (CAT) activity, and the absorbance was read at 240 nm. Following the nitroblue tetrazolium photo-reduction method [63], superoxide dismutase (SOD) activity was measured, and the absorbance was read at 540 nm. Using the methodology of Putter and Becker [64], peroxidase (POD) activity was determined by recording the oxidized guaiacol production rate at 436 nm. Enzymes activities were expressed in enzyme unit per milligramme protein (EU mg⁻¹ protein).

2.11. Expression Analysis of Stress-Related Genes

Quantitative real-time PCR (RT-qPCR) assay was applied to investigate the expression levels of 10 stress-related genes, including *APX*, *CAT*, *Fe-SOD*, *POD*, *CHI* (encoding chalcone isomerase in flavonoids biosynthesis pathways; [65]), *CHS* (encoding chalcone synthase in flavonoids biosynthesis pathways), *PHD2* (plant-homeo-domain gene of DNA binding ability and stress tolerance induction; [66]), *VSP* (encoding vegetative storage defense protein; [67]), *NR* (encoding nitrate reductase; [67]) and *P5CS* (encoding pyrroline-5 carboxylate synthetase; [67]). Total RNA was extracted from leaf tissue samples using RNeasy Plant Mini kits (Qiagen, Manchester, UK). Contaminating DNA was then removed and first-strand cDNAs were prepared using Reverse Transcription kits (Qiagen, Manchester, UK). RT-qPCR analysis was conducted as reported in the protocol of QuantiTect SYBR Green PCR kit (Qiagen, Manchester, UK). Reaction volume and PCR amplification conditions were adjusted as mentioned by El-Esawi et al. [43]. Gene-specific primers of Sirhindi et al. [68], Vaishnav et al. [67], El-Esawi et al. [69], and Kim et al. [70] (Table S1) were used for genes amplification. A housekeeping gene, *actin*, was used. The method of Livak and Schmittgen [71] was utilized to estimate relative expression levels.

2.12. Statistical Analysis

The experimental data are means ± standard errors (SEs) ($n = 4$). One-way analysis of variance (ANOVA) and Duncan's multiple range test were carried out using SPSS version 16 (Chicago, IL, USA). Values at $p \leq 0.05$ differ significantly.

3. Results and Discussion

3.1. Cadmium Tolerance of *Serratia marcescens* BM1

Serratia marcescens BM1 strain grew in nutrient broth medium having 0, 150 and 300 µM CdCl₂ at different incubation time intervals with OD = 0.49–0.61, OD = 0.29–0.49 and OD = 0.22–0.35, respectively (Figure 1). However, the bacterium could not grow in the medium containing 450 µM CdCl₂ at all the incubation time intervals (OD = 0). These findings reveal that BM1 was able to grow and tolerate Cd stress up to 300 µM CdCl₂, indicating that this bacterium could be exploited in the phytoremediation and Cd stress tolerance studies.

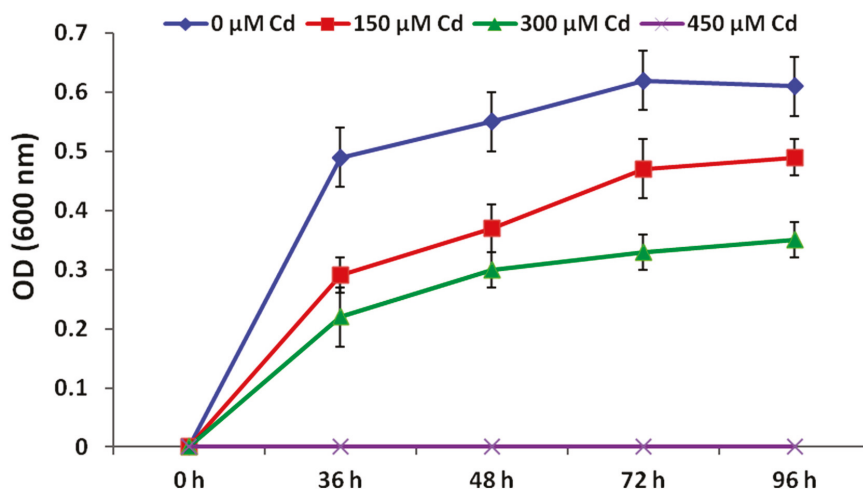


Figure 1. Growth curve of *Serratia marcescens* BM1 in nutrient broth media supplemented with 0, 150, 300, and 450 μM CdCl_2 after 0, 36, 48, 72, and 96 h of incubation at 29 °C.

3.2. *Serratia marcescens* BM1 Enhances Growth and Biomass of Soybean Plants Under Cadmium Stress

Soybean plants exhibited a significant decrease in growth and biomass traits under Cd stress at both doses (150 and 300 μM) with respect to control plants (Table 1). Cadmium stress at T3 (150 μM CdCl_2) and T5 (300 μM CdCl_2) treatments decreased root length by 27.56%, and 47.43% and shoot length by 30.95%, and 50.79%, respectively, with respect to the control plants. Inoculating Cd-stressed soybean with *Serratia marcescens* BM1 enhanced the length of root by 14.20% at T4 (150 μM CdCl_2 + *Serratia marcescens* BM1) and 17.10% at T6 (300 μM CdCl_2 + *Serratia marcescens* BM1) treatments, as compared to T3, and T5, respectively (Table 1). A similar trend also follows in case of shoot length, by which a significant increase of 18.40% at T4 and 27.40% at T6 was noticed with respect to T3, and T5, respectively (Table 1).

Table 1. Growth and biomass of un-inoculated and inoculated soybean plants with *Serratia marcescens* BM1 under Cd stress.

CdCl_2 (μM)	BM1	Root Length (cm)	Root Fresh Weight (g Plant^{-1})	Root Dry Weight (g Plant^{-1})	Shoot Length (cm)	Shoot Fresh Weight (g Plant^{-1})	Shoot Dry Weight (g Plant^{-1})	Root DW/Shoot DW
0	–BM1 (T1)	15.6 \pm 1.11b	1.37 \pm 0.07b	0.15 \pm 0.03b	25.2 \pm 1.4b	1.83 \pm 0.15b	0.37 \pm 0.14b	0.41 \pm 0.05b
	+BM1 (T2)	17.3 \pm 1.25a	1.51 \pm 0.08a	0.18 \pm 0.06a	26.9 \pm 1.6a	1.98 \pm 0.12a	0.42 \pm 0.12a	0.43 \pm 0.04a
150	–BM1 (T3)	11.3 \pm 1.11d	0.98 \pm 0.06d	0.10 \pm 0.05d	17.4 \pm 1.3d	1.24 \pm 0.11d	0.25 \pm 0.15d	0.40 \pm 0.05b
	+BM1 (T4)	12.9 \pm 1.21c	1.18 \pm 0.09c	0.12 \pm 0.04c	20.6 \pm 1.5c	1.41 \pm 0.12c	0.29 \pm 0.11c	0.42 \pm 0.03a
300	–BM1 (T5)	8.2 \pm 1.07f	0.61 \pm 0.08f	0.06 \pm 0.05f	12.4 \pm 1.6f	0.77 \pm 0.15f	0.16 \pm 0.13f	0.37 \pm 0.05d
	+BM1 (T6)	9.6 \pm 1.10e	0.77 \pm 0.06e	0.07 \pm 0.04e	15.8 \pm 1.8e	0.92 \pm 0.12e	0.18 \pm 0.11e	0.39 \pm 0.03c

Values represent the means \pm SE ($n = 4$). Different letters (a–f) next to numbers in the same column indicate significant differences between treatments ($p \leq 0.05$). DW denotes dry weight.

Root and shoot fresh weights showed a significant ($p \leq 0.05$) decrease under Cd stress at T3 and T5 treatments compared to T1 unstressed plants (Table 1). Cadmium stress at T3 and T5 decreases the root fresh weight by 28.46% and 55.47% and shoot fresh weight by 32.24%, and 57.92%, respectively, with respect to unstressed T1 plants. On the other hand, *Serratia marcescens* BM1 application to Cd-stressed soybean plants enhanced the root fresh weight by 20.41% at T4 and 26.23% at T6 treatments with respect to T3, and T5 treatments, respectively. A similar trend was also noticed in shoot fresh weight, in which a significant increase of 13.71% at T4 and 19.48% at T6, was observed with respect to T3, and T5 treatments, respectively (Table 1).

In case of dry weights of root and shoot, Cd induces a significant ($p \leq 0.05$) decrease at both T3 and T5 treatments (Table 1). Cadmium induced significant ($p \leq 0.05$) reductions of 33.33% and 60.00% in root dry weight at T3, and T5 treatments, respectively, with respect to control plants. Cadmium stress also decreased the shoot dry weight by 43.24%, and 59.45% at T3, and T5 treatments, respectively, with respect to control plants. On the other hand, *Serratia marcescens* BM1 application to Cd-stressed plants showed a significant increase in root and shoot dry weight. An increase of 20.00% and 16.66% in root dry weight and 38.10% and 26.66% in shoot dry weight was noticed at T4 and T6 treatments in comparison to plants treated only with Cd at T3, and T5, respectively (Table 1). Furthermore, the root dry weight/shoot dry weight ratio was also higher for plants inoculated with *Serratia marcescens* BM1 in the presence or absence of cadmium. Therefore, *Serratia marcescens* BM1 effect was most profound on root development with a recorded increment in the root/shoot ratio in response to cadmium stress. This could be attributed to the modulation of antioxidant machinery and nutrients uptake in plant cells.

The aforementioned growth effects are consequences of the plant growth-promoting traits of *Serratia marcescens* BM1 strain, which promoted soybean growth and biomass under Cd stress conditions. Similar reports revealed the potential of *Serratia* spp. in enhancing plant growth and biomass under control and stress conditions. *Serratia marcescens* strain SRM significantly improved plant growth and biomass of wheat plants grown in cold temperature [72]. *Serratia marcescens* AL2-16 also improved the growth and biomass of the medicinal plant, *Achyranthes aspera* L. [73]. Furthermore, *Serratia liquefaciens* KM4 improved maize growth and biomass under normal and saline conditions [74]. *Serratia marcescens* CDP-13 promoted plant growth and alleviated salinity stress in wheat [75]. Moreover, Khan et al. [76] revealed that Cd stress significantly reduced plant growth and biomass of *Solanum nigrum* L. plants, whereas *Serratia* sp. RSC-14 inoculation alleviated Cd stress-induced toxic effects by significantly promoting plant growth and biomass.

3.3. *Serratia marcescens* BM1 Modulates the Contents of Cadmium, Nitrogen and Phosphorous in Cadmium-Stressed Soybean Plants

Cadmium stress at T3 (150 μM CdCl₂) and T5 (300 μM CdCl₂) treatments significantly induced the accumulation of Cd over the control plants (Table 2). On the other hand, application of *Serratia marcescens* BM1 to Cd-stressed soybean plants significantly reduced Cd uptake by 21.05% at T4 (150 μM CdCl₂ + *Serratia marcescens* BM1) and 17.86% at T6 (300 μM CdCl₂ + *Serratia marcescens* BM1) treatments with respect to T3, and T5, respectively (Table 2).

Table 2. Minerals uptake, photosynthesis rate (P_n), transpiration rate (E) and stomatal conductance (g_s) in leaves of un-inoculated and inoculated soybean plants with *Serratia marcescens* BM1 under Cd stress.

CdCl ₂ (μM)	BM1	Cd Content (mg g ⁻¹ DW)	N Content (mg g ⁻¹ DW)	P Content (mg g ⁻¹ DW)	P_n ($\mu\text{mol m}^{-2} \text{s}^{-1}$)	E (mmol m ⁻² s ⁻¹)	g_s (mol m ⁻² s ⁻¹)
0	-BM1 (T1)	0.01 ± 0.01e	0.88 ± 0.12b	0.40 ± 0.08b	17.03 ± 1.24b	1.84 ± 0.07b	0.11 ± 0.03b
	+BM1 (T2)	0.01 ± 0.01e	0.93 ± 0.11a	0.44 ± 0.11a	18.92 ± 1.15a	2.03 ± 0.05a	0.13 ± 0.02a
150	-BM1 (T3)	0.19 ± 0.09c	0.72 ± 0.09d	0.29 ± 0.09d	11.12 ± 1.21d	1.49 ± 0.04d	0.06 ± 0.03d
	+BM1 (T4)	0.15 ± 0.05d	0.81 ± 0.07c	0.36 ± 0.10c	13.25 ± 1.23c	1.58 ± 0.07c	0.08 ± 0.02c
300	-BM1 (T5)	0.28 ± 0.11a	0.58 ± 0.08f	0.22 ± 0.11e	8.91 ± 1.05f	1.33 ± 0.07f	0.03 ± 0.01f
	+BM1 (T6)	0.23 ± 0.10b	0.65 ± 0.07e	0.30 ± 0.12d	10.33 ± 1.07e	1.42 ± 0.08e	0.05 ± 0.03e

Values represent the means ± SE ($n = 4$). Different letters (a–f) next to numbers in the same column indicate significant differences between treatments ($p \leq 0.05$).

Under both doses of Cd stress at T3 and T5 treatments, soybean plants exhibited a decrease in N and P contents (Table 2). Cadmium stress at T3 and T5 decreased the N level by 18.18%, and 34.09%, respectively, with respect to unstressed T1 plants. *Serratia marcescens* BM1 application significantly enhanced N content at both Cd treatments. An increase of 12.50% and 12.07% was noticed in N content at T4 and T6 treatments with respect to T3, and T5, respectively (Table 2). Low and high doses of Cd stress also significantly decreased P content at T3 and T5 treatments with respect to unstressed T1 plants. Phosphorous contents showed a significant decline of 27.50% and 45.00% at T3, and T5,

respectively, in relation to control plants. Application of *Serratia marcescens* BM1 to Cd-stressed soybean plants induced phosphorus content by 24.14% and 36.36% at T4 and T6 treatments with respect to T3, and T5, respectively (Table 2). These results indicate that *Serratia marcescens* BM1 was able to alleviate Cd stress-induced oxidative damage by enhancing nutrients uptake in soybean plants. Our results are in agreement with the findings of previous reports which indicated the potential of other *Serratia* strains in mitigating abiotic stress-induced oxidative damage through enhancing nutrients uptake in plants. For example, *Serratia marcescens* strain SRM significantly induced nutrients uptake of wheat plants grown under cold conditions [72]. *Serratia marcescens* TRS-1 also enhanced leaf and root phosphate content of tea plants [77]. Furthermore, *Serratia nematodiphila* LRE07 improved plant growth, biomass and nutrient uptake of *Solanum nigrum* L. plants under Cd stress conditions [78]. *Serratia liquefaciens* KM4 also significantly induced nutrient uptake of maize plants grown under normal and saline stress conditions [74].

3.4. *Serratia marcescens* BM1 Induces Leaf Gas Exchange Attributes, Leaf Relative Water Content and Biosynthesis of Chlorophyll, Sugars, Proteins, Osmolytes, Flavonoids, and Phenolics Under Cadmium Stress

Serratia marcescens BM1 application improved P_n , E and g_s under stress-free and Cd-stressed conditions (Table 2). Soybean exposure to 150 μM CdCl_2 (T3) and 300 μM CdCl_2 (T5) doses significantly decreased P_n , E and g_s values with respect to unstressed T1 plants. *Serratia marcescens* BM1 inoculation nullified the Cd-induced repressions in gas-exchange characteristics and enhanced P_n , E and g_s at T4 (150 μM CdCl_2 + *Serratia marcescens* BM1) and T6 (300 μM CdCl_2 + *Serratia marcescens* BM1) treatments as compared to Cd stressed plants at T3, and T5, respectively (Table 2).

Cadmium stress decreased the chlorophyll content by 21.28% and 30.19% at T3, and T5 treatments, respectively, with respect to unstressed T1 plants (Table 3). *Serratia marcescens* BM1 inoculation significantly ($p \leq 0.05$) induced the biosynthesis of chlorophyll by 5.66% at T4 and 5.67% at T6 treatments as compared to Cd stressed plants at T3, and T5, respectively. Furthermore, Cd stress significantly reduced the leaf relative water content (RWC) with respect to unstressed T1 plants (Table 3). On the other hand, *Serratia marcescens* BM1 inoculation significantly induced RWC in Cd-stressed plants as compared to untreated Cd stressed plants.

Table 3. chlorophyll content, leaf relative water content (RWC), total soluble sugars, protein, proline and glycine betaine (GB) contents in leaves of un-inoculated and inoculated soybean plants with *Serratia marcescens* BM1 under Cd stress.

CdCl ₂ (μM)	BM1	Chlorophyll (mg g ⁻¹ FW)	RWC (%)	Sugars ($\mu\text{g g}^{-1}$ FW)	Proteins (mg g ⁻¹ FW)	Proline ($\mu\text{g g}^{-1}$ FW)	GB ($\mu\text{mol g}^{-1}$ FW)
0	-BM1 (T1)	2.02 ± 0.08b	91 ± 0.53b	2.41 ± 0.14b	1.11 ± 0.14b	20.5 ± 0.29f	2.36 ± 0.04f
	+BM1 (T2)	2.14 ± 0.07a	93 ± 0.48a	2.63 ± 0.16a	1.23 ± 0.13a	35.1 ± 0.24e	3.22 ± 0.07e
150	-BM1 (T3)	1.59 ± 0.09d	55 ± 0.51d	2.17 ± 0.14d	0.88 ± 0.10d	55.2 ± 0.41d	4.17 ± 0.06d
	+BM1 (T4)	1.68 ± 0.08c	68 ± 0.46c	2.29 ± 0.15c	1.05 ± 0.14c	77.3 ± 0.40c	5.13 ± 0.05c
300	-BM1 (T5)	1.41 ± 0.09f	42 ± 0.61e	1.93 ± 0.17e	0.67 ± 0.11f	81.7 ± 0.52b	6.06 ± 0.07b
	+BM1 (T6)	1.49 ± 0.07e	54 ± 0.45d	2.12 ± 0.13d	0.79 ± 0.14e	97.4 ± 0.55a	7.88 ± 0.06a

Values represent the means ± SE ($n = 4$). Different letters (a–f) next to numbers in the same column indicate significant differences between treatments ($p \leq 0.05$). FW denotes the fresh weight of soybean plants.

Low and high doses of Cd stress at T3 (150 μM CdCl_2) and T5 (300 μM CdCl_2) treatments in soybean plants showed a decrease in total soluble sugars and protein contents (Table 3). Both low and high dose of Cd at T3 and T5 caused significant ($p \leq 0.05$) decreases of 9.95%, and 19.91%, respectively, in sugars levels with respect to T1 control plants. A similar trend also follows in these treatments in case of total protein content, whereas a decrease of 20.72% and 39.93% at T3, and T5 was noticed with respect to control plants. On the other hand, *Serratia marcescens* BM1 inoculation significantly enhanced sugars and protein contents at T4 (150 μM CdCl_2 + *Serratia marcescens* BM1) and T6 (300 μM CdCl_2 + *Serratia marcescens* BM1) treatments with respect to T3, and T5, respectively. In case of sugars, a significant increase of 5.53% and 9.84% was noticed at T4 and T6 treatments with respect to T3,

and T5, respectively. Significant increases of 19.32% and 17.91% were also recorded in protein content at T4 and T6 with respect to T3, and T5, respectively (Table 3).

Serratia marcescens BM1 enhanced the biosynthesis of osmolytes, including proline and glycine betaine in control and Cd-stressed soybean plants (Table 3). Cadmium stress at T3 and T5 caused a marked increase in proline content by 2.69, and 3.98 fold, respectively, over control plants. A similar trend was also noticed in glycine betaine, whereas an increase of 1.76 and 2.56 fold at T3 and T5, respectively, was recorded over control plants. Inoculation of plants with *Serratia marcescens* BM1 at both Cd stress conditions had an additive effect on proline and glycine betaine biosynthesis and maximum values of these parameters were recorded at T6 treatments.

Serratia marcescens BM1 inoculation improved the levels of flavonoids and phenolics under control and Cd stress conditions (Table 4). Cadmium stress lowered flavonoid content by 29.95% at T3 (150 μM CdCl_2) and by 59.77% at T5 (300 μM CdCl_2) with respect to unstressed T1 plants. However, inoculating Cd-stressed soybean plants with *Serratia marcescens* BM1 induced flavonoids content by 10.69% and 41.37% at T4 (150 μM CdCl_2 + *Serratia marcescens* BM1) and T6 (300 μM CdCl_2 + *Serratia marcescens* BM1) treatments, with respect to T3, and T5, respectively (Table 4). The phenolic content also exhibited a significant decrease of 23.59% and 41.01% at T3, and T5 treatments, respectively, with respect to T1 control plants. *Serratia marcescens* BM1 application exhibited higher phenolic content of 9.55% and 24.76% at T4 and T6 treatments with respect to T3, and T5, respectively (Table 4).

Table 4. The hydrogen peroxide (H_2O_2) content, lipid peroxidation (MDA) level, total flavonoid content (mg catechin/g extract), total phenolic content (mg gallic acid/g extract) and antioxidant activity (DPPH and β -carotene-linoleic acid) of leaves of un-inoculated and inoculated soybean plants with *Serratia marcescens* BM1 under Cd stress.

CdCl ₂ (μM)	BM1	Total Flavonoids	Total Phenolics	H ₂ O ₂ ($\mu\text{mol g}^{-1}$ FW)	MDA ($\mu\text{mol g}^{-1}$ FW)	DPPH (IC ₅₀ , $\mu\text{g mL}^{-1}$)	β -Carotene Linoleic Acid (IC ₅₀ , $\mu\text{g mL}^{-1}$)
0	-BM1 (T1)	7.21 \pm 0.28b	17.8 \pm 0.24b	0.27 \pm 0.05d	20 \pm 1.71e	0.53 \pm 0.03a	0.51 \pm 0.05a
	+BM1 (T2)	9.11 \pm 0.30a	19.1 \pm 0.23a	0.18 \pm 0.06e	16 \pm 1.66f	0.48 \pm 0.05b	0.47 \pm 0.04b
150	-BM1 (T3)	5.05 \pm 0.22d	13.6 \pm 0.22d	0.72 \pm 0.04b	51 \pm 2.27b	0.46 \pm 0.04c	0.44 \pm 0.05c
	+BM1 (T4)	5.59 \pm 0.27c	14.9 \pm 0.24c	0.61 \pm 0.07c	42 \pm 2.24d	0.42 \pm 0.03d	0.41 \pm 0.06d
300	-BM1 (T5)	2.9 \pm 0.23f	10.5 \pm 0.31e	0.85 \pm 0.04a	61 \pm 2.43a	0.38 \pm 0.05e	0.37 \pm 0.05e
	+BM1 (T6)	4.1 \pm 0.26e	13.1 \pm 0.29d	0.60 \pm 0.05c	47 \pm 2.11c	0.35 \pm 0.04f	0.34 \pm 0.05f

Values represent the means \pm SE ($n = 4$). Different letters (a–f) next to numbers in the same column indicate significant differences between treatments ($p \leq 0.05$).

The results of the above parameters demonstrate that *Serratia marcescens* BM1 also mitigated the negative impacts of Cd stress by modulating leaf gas exchange characteristics, photosynthesis, and osmolytes. These results are in agreement with the findings of previous studies which revealed the key roles of *Serratia* isolates in alleviating abiotic stress effects in plants by regulating photosynthesis, leaf gas exchange parameters, and osmolytes biosynthesis. For example, *Serratia nematodiphila* LRE07 enhanced photosynthetic pigments biosynthesis of *Solanum nigrum* L. plants under Cd stress conditions [78]. Khan et al. [76] revealed that *Serratia* sp. RSC-14 inoculation mitigated the negative impacts of Cd stress by significantly improving chlorophyll biosynthesis. Moreover, *Serratia liquefaciens* KM4 significantly induced leaf gas exchange attributes, photosynthesis process, and osmolytes biosynthesis of maize plants grown under saline stress conditions [74].

3.5. *Serratia marcescens* BM1 Inoculation Reduces the Contents of H_2O_2 and MDA Under Cadmium Stress

Cadmium stress at T3 (150 μM CdCl_2) and T5 (300 μM CdCl_2) treatments significantly increased the production of H_2O_2 over the control plants. On the other hand, the application of *Serratia marcescens* BM1 to Cd-stressed soybean plants significantly reduced H_2O_2 level by 15.27% at T4 (150 μM CdCl_2 + *Serratia marcescens* BM1) and 29.41% at T6 (300 μM CdCl_2 + *Serratia marcescens* BM1) treatments with respect to T3, and T5, respectively. MDA production exhibited significant increases at T3 and T5 treatments with respect to T1 unstressed plants. A significant decrease of 17.64% and 22.95% in

MDA content was noticed in plants inoculated with *Serratia marcescens* BM1 at T4 and T6 treatments with respect to T3, and T5, respectively (Table 4), indicating that *Serratia marcescens* BM1 regulates membrane functions through scavenging ROS accumulation. These results are in agreement with the findings of the previous reports, which indicated the potential of *Serratia* isolates in mitigating the Cd stress impacts by reducing the oxidative stress markers, such as MDA, electrolyte leakage, and hydrogen peroxide. Khan et al. [76] revealed that *Serratia* sp. RSC-14 relived the toxic effects of Cd stress by reducing MDA and electrolytes production. Furthermore, *Serratia liquefaciens* KM4 significantly alleviated the salt stress-induced oxidative damage by reducing MDA and H₂O₂ contents [74].

3.6. *Serratia marcescens* BM1 Inoculation Modulates Antioxidant Capacity and Antioxidative Enzyme Activities Under Cadmium Stress

Cd stress decreased the DPPH values by 13.20% and 28.30% at T3 (150 µM CdCl₂), and T5 (300 µM CdCl₂) treatments, respectively, with respect to control plants (Table 4). Nevertheless, *Serratia marcescens* BM1 inoculation further caused a decrease in DPPH values by 8.69% and 7.89% at T4 (150 µM CdCl₂ + *Serratia marcescens* BM1) and T6 (300 µM CdCl₂ + *Serratia marcescens* BM1) treatments with respect to respective T3, and T5 treatments, respectively. β-carotene-linoleic acid of soybean plants also follows a same decreasing trend when grown under Cd stress conditions at T3 and T5 treatments. Inoculating soybean plants with *Serratia marcescens* BM1 further caused a significant decrease in β-carotene-linoleic acid values by 6.81% and by 8.10% at T4 and T6 with respect to T3, and T5, respectively (Table 4).

To scavenge ROS accumulation and mitigate cellular toxicity, plants activate their defence strategies, including their non-enzymatic and enzymatic antioxidant systems, such as ascorbate, SOD, CAT, APX, and POD. In the present study, the activities of antioxidant enzymes in soybean plants, such as CAT, APX, POD, and SOD were elevated in response to Cd stress at T3 (150 µM CdCl₂) and T5 (300 µM CdCl₂) with respect to T1 control plants (Figure 2). Furthermore, the highest significant increase in CAT, APX, POD, and SOD activities was recorded when soybean Cd-stressed plants were inoculated with *Serratia marcescens* BM1 at T6 (300 µM CdCl₂ + *Serratia marcescens* BM1) treatments (Figure 2). These findings are consistent with the results of the previous reports which showed the key roles of *Serratia* strains in alleviating the abiotic stress-induced oxidative damage through regulating the key antioxidant enzymes. For example, *Serratia nematodiphila* LRE07 alleviated the negative impacts of Cd stress in *Solanum nigrum* L. plants by further inducing the activities of anti-oxidative enzymes [78]. *Serratia liquefaciens* KM4 also significantly alleviated the salt stress-induced oxidative damage by upregulating various antioxidative enzymes, such as CAT, APX, POD, and SOD [74].

3.7. *Serratia marcescens* BM1 Induces the Expression Levels of Stress-Related Genes Under Cadmium Stress

The application of Cd stress in both doses on soybean plants exhibited a dynamic increase in the expression levels of antioxidant genes (*APX*, *CAT*, *Fe-SOD* and *POD*; Figure 3) and stress-induced genes (*CHI*, *CHS*, *PHD2*, *VSP*, and *P5CS*; Figure 4). The results also revealed that the expression levels of these 10 genes were further increased upon inoculating soybean plants with *Serratia marcescens* BM1, whereas, the expression levels reached to the maximum extent at T6 (300µM CdCl₂ + *Serratia marcescens* BM1) treatments in case of *APX*, *CAT*, *Fe-SOD*, *POD*, *PHD2*, *VSP*, and *P5CS* genes. The expression profiling of *CHI*, *CHS*, and *NR* genes were induced to a maximum extent in non-stressed soybean plants, inoculated with *Serratia marcescens* BM1 (Figure 4). These results indicate the key role of *Serratia marcescens* BM1 in eliminating toxic free radicals and conferring Cd stress tolerance through upregulating the abiotic stress-related genes. Similar reports indicated the potential of *Serratia* and other PGPBs in improving abiotic stress tolerance by inducing abiotic stress-related genes [74,79].

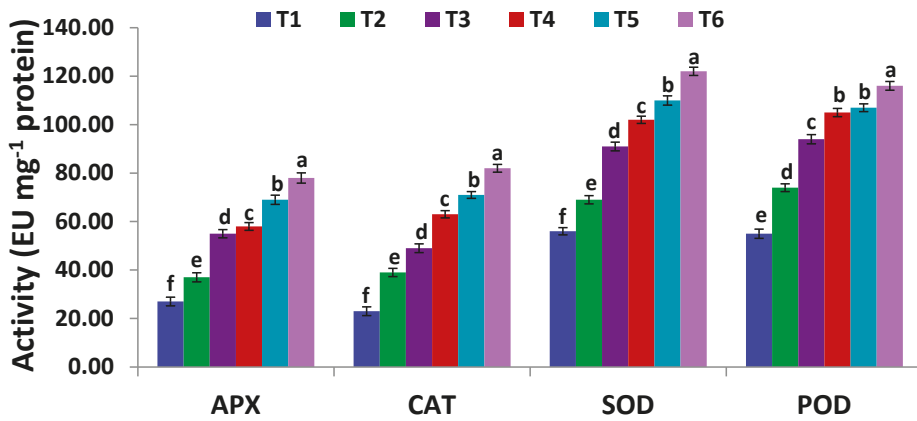


Figure 2. Ascorbate peroxidase (APX), catalase (CAT), superoxide dismutase (SOD), and peroxidase (POD) activities in leaves of un-inoculated and inoculated soybean plants with *Serratia marcescens* BM1 under Cd stress. Data are means ± SE (n = 4). Different letters (a–f) above the chart columns indicate significant differences between treatments (p ≤ 0.05). T1, control with no Cd and BM1 treatment; T2, plant treated with *Serratia marcescens* BM1; T3, plant treated with 150 μM CdCl₂; T4, plant treated with 150 μM CdCl₂ and *Serratia marcescens* BM1; T5, plant treated with 300 μM CdCl₂; T6, plant treated with 300 μM CdCl₂ and *Serratia marcescens* BM1.

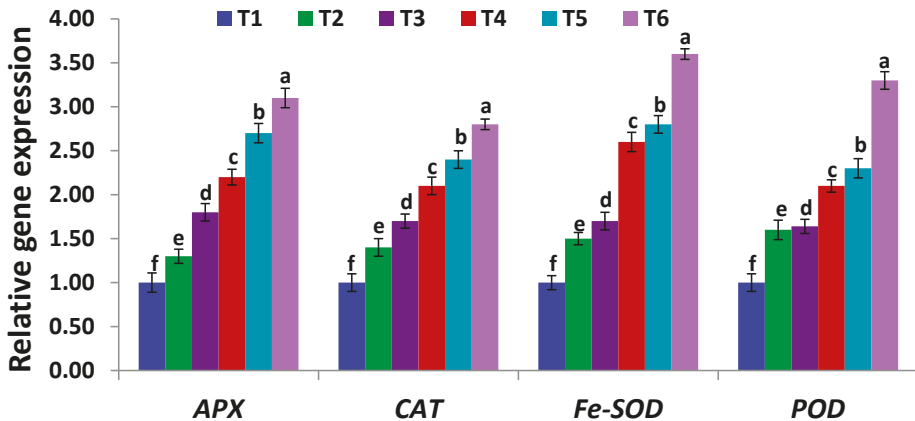


Figure 3. Expression levels of APX, CAT, Fe-SOD, and POD genes in leaves of un-inoculated and inoculated soybean plants with *Serratia marcescens* BM1 under Cd stress. Data are means ± SE (n = 4). Different letters (a–f) above the chart columns indicate significant differences between treatments (p ≤ 0.05). T1, control with no Cd and BM1 treatment; T2, plant treated with *Serratia marcescens* BM1; T3, plant treated with 150 μM CdCl₂; T4, plant treated with 150 μM CdCl₂ and *Serratia marcescens* BM1; T5, plant treated with 300 μM CdCl₂; T6, plant treated with 300 μM CdCl₂ and *Serratia marcescens* BM1.

Overall, cadmium stress-induced oxidative effects have negative impacts on gas exchange attributes and photosynthetic activity, thereby, limiting plant growth and biomass. However, the present study revealed a key role of *Serratia marcescens* BM1 in improving phytoremediation and Cd stress tolerance in soybean plants by modulating the leaf gas exchange attributes, antioxidant machinery, osmolytes, and expression of genes associated with antioxidant defence systems and stress response.

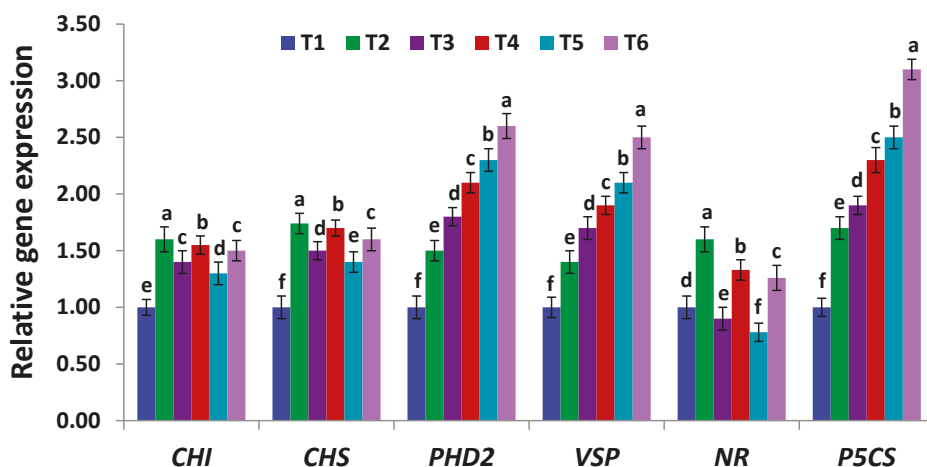


Figure 4. Expression levels of *CHI*, *CHS*, *PHD2*, *VSP*, *NR* and *P5CS* genes in leaves of un-inoculated and inoculated soybean plants with *Serratia marcescens* BM1 under Cd stress. Data are means \pm SE ($n = 4$). Different letters (a–f) above the chart columns indicate significant differences between treatments ($p \leq 0.05$). T1, control with no Cd and BM1 treatment; T2, plant treated with *Serratia marcescens* BM1; T3, plant treated with 150 μ M CdCl₂; T4, plant treated with 150 μ M CdCl₂ and *Serratia marcescens* BM1; T5, plant treated with 300 μ M CdCl₂; T6, plant treated with 300 μ M CdCl₂ and *Serratia marcescens* BM1.

4. Conclusions

In the present study, Cd stress reduced the growth, biomass, photosynthetic attributes, mineral nutrients uptake, and contents of flavonoids, phenolics, soluble sugars, and proteins of soybean plants. Cadmium also increased the contents of cadmium, hydrogen peroxide, and malondialdehyde. Conversely, *Serratia marcescens* BM1 inoculation mitigated the deleterious impacts of cadmium stress by significantly inducing the plant growth, biomass, gas exchange attributes, nutrients uptake, antioxidant capacity, and the contents of chlorophyll, total phenolics, flavonoids, soluble sugars, and proteins in Cd-stressed soybean plants. *Serratia marcescens* BM1 also triggered the activities of antioxidant enzymes and the expression of stress-related genes. Considering these results, it could be concluded that *Serratia marcescens* BM1 might promote Cd stress tolerance through the modulation of photosynthesis, leaf gas exchange traits, antioxidant machinery, osmolyte biosynthesis, and the expression of stress-related genes.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2076-3921/9/1/43/s1>, Table S1: Gene-specific primers sequences used in the present study.

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Article

Short-Term Low Temperature Induces Nitro-Oxidative Stress that Deregulates the NADP-Malic Enzyme Function by Tyrosine Nitration in *Arabidopsis thaliana*

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Abstract: Low temperature (LT) negatively affects plant growth and development via the alteration of the metabolism of reactive oxygen and nitrogen species (ROS and RNS). Among RNS, tyrosine nitration, the addition of an NO₂ group to a tyrosine residue, can modulate reduced nicotinamide-dinucleotide phosphate (NADPH)-generating systems and, therefore, can alter the levels of NADPH, a key cofactor in cellular redox homeostasis. NADPH also acts as an indispensable electron donor within a wide range of enzymatic reactions, biosynthetic pathways, and detoxification processes, which could affect plant viability. To extend our knowledge about the regulation of this key cofactor by this nitric oxide (NO)-related post-translational modification, we analyzed the effect of tyrosine nitration on another NADPH-generating enzyme, the NADP-malic enzyme (NADP-ME), under LT stress. In *Arabidopsis thaliana* seedlings exposed to short-term LT (4 °C for 48 h), a 50% growth reduction accompanied by an increase in the content of superoxide, nitric oxide, and peroxynitrite, in addition to diminished cytosolic NADP-ME activity, were found. In vitro assays confirmed that peroxynitrite inhibits cytosolic NADP-ME2 activity due to tyrosine nitration. The mass spectrometric analysis of nitrated NADP-ME2 enabled us to determine that Tyr-73 was exclusively nitrated to 3-nitrotyrosine by peroxynitrite. The in silico analysis of the *Arabidopsis* NADP-ME2 protein sequence suggests that Tyr73 nitration could disrupt the interactions between the specific amino acids responsible for protein structure stability. In conclusion, the present data show that short-term LT stress affects the metabolism of ROS and RNS, which appears to negatively modulate the activity of cytosolic NADP-ME through the tyrosine nitration process.

Keywords: NADP malic enzyme; low temperature; nitric oxide; tyrosine nitration; peroxynitrite; reactive oxygen species; reactive nitrogen species; nitro-oxidative stress

1. Introduction

Low temperature (LT) is an environmental issue that affects plant physiology and biochemistry at different levels, including antioxidant enzymes, photosynthesis, gene expression, nutrients, and water uptake [1–9]. At the economic level, the impact of LT on plant crops, such as pepper, rice, tomato, maize, and olive, and tropical and subtropical fruits worldwide, is very significant as it influences both their production and quality [1,10,11]. However, model plants, such as *Arabidopsis thaliana*, act as a very useful tool to decipher the molecular mechanism of response to LT stress [12–16].

LT usually induces nitro-oxidative stress, mediated by the overproduction of reactive oxygen (ROS) and nitrogen (RNS) species [1]. Interestingly, an increasing number of reports suggest that certain reduced nicotinamide-dinucleotide phosphate (NADPH)-generating dehydrogenases might be involved in the protection mechanism against nitro-oxidative stresses induced by adverse environmental conditions [17–23]. In plants, several NADPH-generating systems come into play, such as ferredoxin-NADP reductase as a component of photosystem I, and a group of NADP-dehydrogenases that have been found in different subcellular locations. This group of enzymes includes NADP-isocitrate dehydrogenase (NADP-ICDH), glucose-6-phosphate dehydrogenase (G6PDH), 6-phosphogluconate dehydrogenase (6PGDH) and the NADP-malic enzyme (NADP-ME), also called NADP-malate dehydrogenase.

The NADP-malic enzyme, together with the other NADP-dehydrogenases, is a key component of the NADPH-production systems required to maintain the redox balance in cells. It has been identified from bacteria to humans as an enzyme that catalyzes the reversible oxidative decarboxylation of L-malate to pyruvate, CO₂, and NADPH [24–26]. In plants, different isoenzymes have been described in plastids and cytosol. In *Arabidopsis*, cytosolic NADP-ME2 is considered to be responsible for most NADP-ME activity in mature tissues [27–29] and has been linked to a wide range of processes [30], such as lignin biosynthesis, by providing NADPH [31], and to control cytosolic pH by balancing the synthesis and degradation of L-malate [32]. Other roles that have been suggested for NADP-ME include the control of stomatal closure through the degradation of L-malate during the daytime and seed germination [33]. The presence of a cytosolic NADP-ME isoform has been reported in the guard cell complexes of C₃ plant wheat. However, a more profound analysis of the NADP-ME isoforms in plants is still required. These studies will contribute to unraveling the biological role of plastidic and cytosolic isoenzymes in the same tissue, or even different NADP-MEs in the same subcellular location. Four NADP-ME isoforms have been identified in monocot rice (*Oryza sativa*) [34] that, unfortunately, have not yet been characterized at the molecular level. In addition, the transcripts of different NADP-ME isoforms, located in the cytosol and photosynthetic and non-photosynthetic organelles, have been identified in *Flaveria* sp., which exhibit different C₃ and C₄ photosynthetic pathways [35]. Interestingly, NADP-ME has also been proposed to be involved in plant responses to biotic and abiotic stress (reviewed by [30]).

One of the regulatory mechanisms of plant response to stress is protein function modulation via nitric oxide (NO)-related posttranslational modifications (PTMs) [36–38]. Interestingly, different NADPH-generating enzymes have been identified as being the target of these NO-PTMs [39–41], but information on the specific impact of these modifications to their function in the nitro-oxidative stress context is still scarce. Along this line, LT is one of the main abiotic stresses that modulates the metabolism of ROS and RNS, and also affects NADP-ME function [1], which suggests the regulation of this enzyme by NO-PTMs, such as tyrosine nitration, as reported for NADPH-generating systems [41,42]. S-nitrosylation, the attachment of NO to a specific cysteine residue, is an NO-PTMs that has been widely analyzed as a regulatory process during plant response to stress [43]. However, tyrosine nitration also appears to play an important role during plant response to the nitro-oxidative stress generated under environmental insults [44]. This NO-PTM is produced by the addition of a nitro group (-NO₂) to the tyrosine residue aromatic ring which gives rise to 3-nitrotyrosine. This results in significantly reducing local pKa, which can affect the tyrosine function [45]. Different factors have been proposed to regulate this PTM, including protein structure and environmental compartments. Although information on specific denitrase activity in plants that allow this PTM to be

considered key in signaling processes is still lacking [46], these covalent changes may result in effects such as protein function loss and gain or no functional change [42,47–50] and, therefore, impact cellular function. Indeed, different NADPH-generating enzymes have been proposed to be modulated by tyrosine nitration [41,42], but the effect of NO on protein structure [42] has been analyzed only for NADP-ICDH, with NADP-ME2 being one of the least studied enzymes.

In this context, the main goals of this study were to determine whether LT stress could affect the cross-talk between ROS and RNS metabolisms in *Arabidopsis thaliana* seedlings and to study its potential correlation with the group of NADPH-generating dehydrogenases. As NADP-ME was the only NADP-dehydrogenase to be modulated under LT conditions, we focused on this enzyme to study the potential regulatory effect of tyrosine nitration by an in vitro approach.

2. Materials and Methods

2.1. Plant Material and Growth Conditions

Arabidopsis thaliana ecotype Columbia seeds were surface-sterilized and grown on Petri dishes containing Murashige and Skoog medium (Sigma-Aldrich-Fluka, St. Louis, MO, USA) according to [51]. *Arabidopsis* seeds were subsequently grown for 2 weeks in a 16 h light/dark regime at 22 °C/8 h and 18 °C (long-day conditions) with a light intensity of 50 $\mu\text{E m}^{-2} \text{s}^{-1}$. For cold treatment purposes, the Petri dishes containing *Arabidopsis* seedlings were placed inside different containers with ice and transferred to a cold room set at 4 ± 2 °C for 48 h.

2.2. Crude Extracts of Plant Tissues

Arabidopsis thaliana seedlings were frozen and ground in liquid N₂ using a pestle and mortar. Then an extraction buffer (1:3, *w/v*) containing 100 mM Tris-HCl, pH 7.8, 0.02 (*v/v*) Triton X-100, 1 mM EDTA and 10% (*v/v*) glycerol was added. The resulting homogenates were centrifuged at 17,000× *g* for 20 min, and supernatants were subjected to different analyses.

2.3. Detection of Nitric Oxide (NO), Peroxynitrite (ONOO⁻) and Superoxide Radicals (O₂^{•-}) by Confocal Laser Scanning Microscopy

The root tips of *Arabidopsis thaliana* were used to detect different reactive oxygen and nitrogen species by confocal laser scanning Microscopy (CLSM,) according to [52]. Briefly, samples were incubated in the dark for 1h with the following specific fluorescents freshly made in 10 mM Tris-HCl, pH 7.4: 10 μM 4-amino-5-methylamino-2',7'-difluoro fluorescein diacetate (DAF-FM DA, Calbiochem, San Diego, CA, USA) to detect nitric oxide(NO); 10 μM 3-(*p*-aminophenyl) fluorescein (APF, Invitrogen, Carlsbad, CA, USA) to detect peroxynitrite (ONOO⁻) and 10 μM dihydroethidium (DHE, Sigma-Aldrich-Fluka, St. Louis, MO, USA) to detect superoxide radicals (O₂^{•-}). DAF-FM DA and APF were incubated at 25 °C and DHE at 37 °C. Then samples were washed twice in the same buffer for 15 min and mounted on a microscope slide for examination purposes with a CLSM (Leica TCS SP5 II). The following filters and collection modalities were used: 490 nm excitation and 520 nm emission for DHE, 495 nm excitation and 515 nm emission for DAF-FM DA, and 495 nm excitation and 515 nm emission for APF. The images obtained by CLSM from the control and LT-treated *Arabidopsis* roots were processed and analyzed by the Leica confocal software [52].

2.4. Enzyme Activity Assay

Glucose-6-phosphate dehydrogenase (G6PDH; EC 1.1.1.49), NADP-isocitrate dehydrogenase (NADP-ICDH, EC 1.1.1.42) and NADP-malic enzyme (NADP-ME; EC 1.1.1.40) activity were determined at 25 °C according to [17,23,53] by spectrophotometrically recording the reduction in NADP at 340 nm. The reaction mixture (1 mL) contained 50 mM 4-(2-Hydroxyethyl)piperazine-1-ethanesulfonic acid (HEPES), pH 7.6, 2 mM MgCl₂, and 0.8 mM NADP. The reaction was initiated by adding 5 mM glucose-6-phosphate, 10 mM 2R,3S-isocitrate or 1 mM L-malate, respectively.

2.5. Expression and Purification of Cytosolic *Arabidopsis Thaliana* NADP-Dependent Malic Enzyme 2 (NADP-ME2)

c-DNA was obtained from the total *Arabidopsis* leaf RNA by using the first strand c-DNA synthesis kit (Roche, Basel, Switzerland) following the manufacturer's instruction. Then the amplification of the cytosolic NADP-ME2 gene (NM_121205.3) was performed by PCR using Fast Start High-Fidelity polymerase (Roche, Basel, Switzerland) and specific primer sets: 5'-AGAGATATGGGAAGTACTCCGACTGATTACC-3' and 5'-ACAAAACTTTTTAACGGTAGTTTCTGTACACAG-3'. The PCR product (1785 bp) was then cloned into the pALEXb vector (BiomedalSL, Sevilla, Spain) and the expression of the recombinant protein carrying an N-terminal choline-binding domain was performed according to [48]. Briefly, the recombinant protein was induced by adding 1 mM salicylate and 10 mM 3-methyl benzoate. Protein purification was then carried out using a 1-mL LYTRAP column (Biomedal SL, Sevilla, Spain), in which protein was eluted in 1-mL fractions using a discontinuous gradient of choline [48]. Finally, the purity grade of the recombinant protein expression was analyzed by 10% SDS-PAGE.

2.6. Treatment of NADP-ME with SIN-1 (Peroxynitrite Donor)

Molecule SIN-1 (3-morpholiniosyl-nonimine) is considered a peroxynitrite donor that is a protein-nitrating compound [54]. Cytosolic recombinant NADP-ME2 was incubated at 37 °C for 1 h with increasing concentrations (0 to 5 mM) of SIN-1 (Calbiochem, San Diego, CA, USA) freshly made before use. A polyclonal antibody against 3-nitrotyrosine (dilution 1:2500) was employed to corroborate the nitration of NADP-ME2 by the immunoblot analysis.

2.7. Identification of Nitrated Tyrosine in Recombinant Cytosolic NADP-ME2 by Mass Spectrometric Techniques

To identify the potential targets of tyrosine nitration, the purified recombinant cytosolic NADP-ME2 treated with SIN-1 was processed by MS/MS as previously described [42]. The sample was subjected to a reduction with dithiothreitol (DTT), derivatization with iodoacetamide (IAA), and enzymatic digestion with trypsin (37 °C, 8 h). It was subsequently purified to eliminate the interferences from the digestion process. Then the peptide mixture was analyzed in a MALDI-TOF/TOF (matrix-assisted laser desorption ionization-time of flight/time of flight) mass spectrometer (4800, AB Sciex) to evaluate sample quality. To this end, peptide mass fingerprinting (PMF) and Uniprot databases were used to interpret the MALDI-TOF spectra and for identification purposes (release 2011_02), respectively. The resulting sample was subjected to a liquid chromatography–tandem mass spectrometry (LC-MS/MS) analysis, using a Velos-LTQ mass spectrometer equipped with a micro-ESI ion source (Thermo Fisher, San Jose, CA, USA). Briefly, the sample was evaporated and subsequently diluted in water containing methanol (5%) and acid formic (1%). Next a 10-cm long, 150- μ m id Vydac C18 column (Vydac, IL, USA) was used for the chromatographic process, in which separation was performed at 1 μ L min⁻¹ with a 3% to 40% acetonitrile gradient for 30 min (solvent A, 0.1% formic acid; solvent B, acetonitrile with 0.1% formic acid). The Velos-LTQ instrument was operated in the positive ion mode with a 2 kV spray voltage. The scan range of each full MS scan was m/z 400 to 2000. After each MS scan, the targeted MS/MS spectra were collected to identify both the unmodified and nitrated form of the predicted tyrosine-containing peptides. The parent mass list of the targeted scan was selected to ensure maximum coverage of the tyrosine-containing tryptic peptides for ME. To obtain a potential list of targeted m/z values, the protein was subjected to *in silico* digestion using nitrated tyrosine as a dynamic modification. The resulting list of predicted peptides (in both the nitrated and unmodified forms) was filtered to exclude any peptide not containing tyrosine residues. The MS/MS spectra were searched with the Proteome Discoverer software (Thermo Fisher, San Jose, CA, USA) under the following conditions: peptide mass tolerance 2 Da, fragment tolerance 0.8 Da, enzyme set as trypsin, no missed cleavages. The dynamic modifications were cysteine carbamidomethylation (+57 Da), methionine oxidation (+16 Da), and tyrosine nitration (+45). Searches were carried out using a database containing all the peptides listed in Table S1. Identifications were filtered with XCorr > 3, P(pep) < 15%. The MS

spectra of the nitrated tyrosines were manually validated by comparing the spectra obtained for the unmodified peptide and the nitrated peptide.

2.8. Modeling and the Molecular Evolution Analysis

The tertiary structure of the NADP-malic enzyme from *Arabidopsis thaliana* was modeled at the Swiss Model Server [55], the CPHmodels 2.1 server [56], HHpred [57], IntFold [58], I-Tasser [59], Phyre2 [60], and RaptorX [61]. Model quality was evaluated by three-dimensional profiles (Verify 3D) [62], the distribution of atom–atom contacts (Errat method) [63], Procheck [64], Qmean [65], and Qmean-Z score [66]. The coordinates of the quaternary structure were calculated by super positioning the best model on the Protein Data Bank (PDB)entry 1O0S with XtalView [67].

The Evolutionary Trace Server was employed to analyze molecular evolution [68] using the model of the tertiary structure as input. Both phylogenetic significance and evolutionary conservation were explored by a BLASTP [69] search the UniProtKB [70] release 2014_10. The prediction of pKa was carried out by PropKa 3.0 [71], and the phosphorylation score was computed on the NetPhos 2.0 Server [72].

2.9. Other Assays

Protein concentration was estimated by the Bio-Rad Protein Assay (Hercules, CA, USA) using bovine serum albumin as a standard. The statistical significance between means was analyzed by a Student's *t*-test.

3. Results

Figure 1A shows the phenotype of the 14-day-old *Arabidopsis thaliana* seedlings grown in MS medium subjected to low temperature (LT) for the last 48 h. These seedlings were smaller in size, and the leaves presented some chlorotic symptoms. The fresh weight of these seedlings also drastically reduced by 50% as a result of LT (Figure 1B), which indicates that the LT conditions caused significant stress to *Arabidopsis thaliana* seedlings.

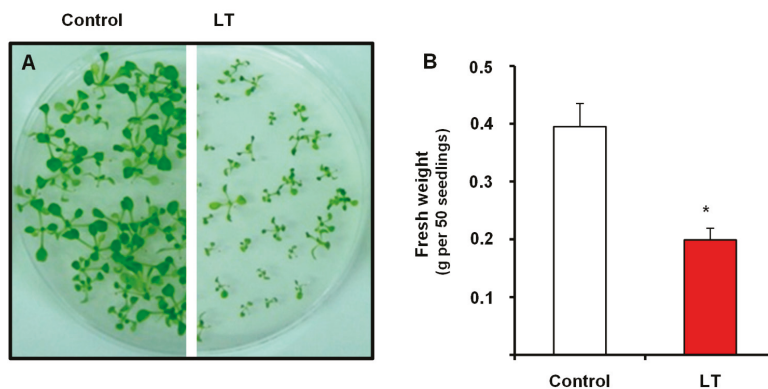


Figure 1. (A) Appearance and (B) fresh weight of the 14-day-old *Arabidopsis* seedlings grown in MS medium subjected to low temperature (LT) for the last 48 h. The results are the mean of at least three different experiments + SEM. For each experiment, 200 seedlings were used. * Differences in relation to the control values were significant at $p < 0.05$.

3.1. NADP-Dehydrogenase Activities under Short-Term LT Stress

It has been previously reported that, as part of the recycling system of NADPH, several NADP-dehydrogenases may be implicated in plant response to some abiotic conditions [23,30,73]. Hence, the activity of the main NADP-dehydrogenases at LT was evaluated. Figure 2A shows how NADP-ME lowered by 25% at LT, whereas a slight, but insignificant increase was recorded for NADP-ICDH (Figure 2B). G6PDH activity showed no change at LT (Figure 2C).

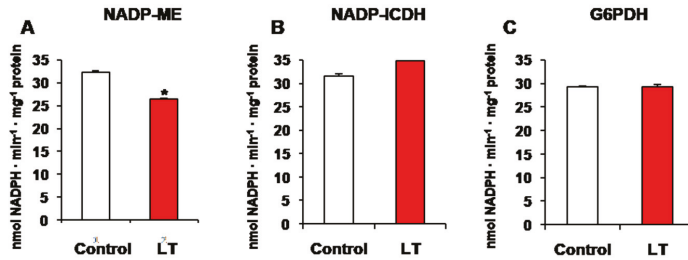


Figure 2. Activity of NADP-dehydrogenase enzymes of the 14-day-old *Arabidopsis* seedlings grown in MS medium subjected to low temperature (LT) for the last 48 h. (A) NADP-malic enzyme (NADP-ME). (B) NADP-isocitrate dehydrogenase (NADP-ICDH). (C) Glucose-6-phosphate dehydrogenase. The results are the mean of at least three different experiments + SEM. * Differences in relation to the control values were significant at $p < 0.05$. NADP: nicotinamide-dinucleotide phosphate.

3.2. Cellular Analysis of Superoxide Radical ($O_2^{\bullet-}$), Nitric Oxide (NO), and Peroxynitrite ($ONOO^-$) in the Roots of the *Arabidopsis* Seedlings Exposed to LT

Figure 3 (panels A–F) illustrates the examination by CLSM of the levels of $O_2^{\bullet-}$, NO, and $ONOO^-$ in the root tips of *Arabidopsis thaliana* seedlings after LT stress for 48 h. $O_2^{\bullet-}$ content was evaluated after incubating *Arabidopsis* seedlings with fluorescent probe DHE. In the roots of the control seedlings, slight green fluorescence related to $O_2^{\bullet-}$ was observed in the tips (Figure 3A). However, increased green fluorescence was detected after exposing seedlings to LT (Figure 3B). Regarding NO production, a significant increase was detected in the roots of the seedlings exposed to LT by using DAF-FM as the fluorescence probe compared to the control plants (Figure 3C,D). APF was used to analyze $ONOO^-$ content, which resulted from the reaction between $O_2^{\bullet-}$ and NO. In the control roots, $ONOO^-$ showed a very slight fluorescent signal (Figure 3E). Nevertheless, this RNS notably increased as a result of LT stress, with homogeneous distribution throughout roots (Figure 3F), similarly to the observed NO distribution.

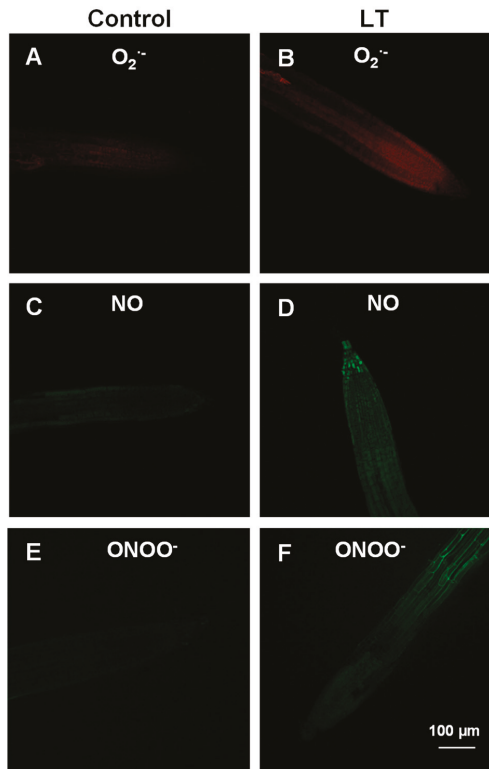


Figure 3. Fluorescence and confocal images showing the in vivo detection of superoxide radical ($O_2^{\bullet-}$) (panels A and B), nitric oxide (NO) (panels C and D), and peroxynitrite ($ONOO^-$) (panels E and F) in *Arabidopsis* root tips after exposing seedlings to low temperature (LT) for 48 h.

3.3. Expression and Purification of Cytosolic NADP-ME2. Effect of Peroxynitrite ($ONOO^-$)

As NADP-ME was the only NADP-dehydrogenase to be modulated under LT stress, and to gain further insight into cytosolic *Arabidopsis thaliana* NADP-ME2, which is the most abundant in this plant [74], the recombinant protein was obtained by cloning the cDNA encoding the enzyme and overexpression in *Escherichia coli*. Figure 4A shows the different fractions obtained after the expression and affinity purification of LYTAG recombinant NADP-ME. The molecular weight of this recombinant construction was 84.3 kDa and resulted from the cytosolic NADP-ME2 protein (63 kDa) containing Ly-tag (21.28 kDa). The fractions with an acceptable purity grade (E5 to E7) showed NADP-EM activity of $9850 \mu\text{mol NADPH min}^{-1} \text{mg}^{-1}$ protein and were used for the subsequent experiments. To analyze the potential action of peroxynitrite generated under LT stress, the recombinant protein was treated with SIN-1 (3-morpholiniosydnonimine) as the peroxynitrite donor. Figure 4B depicts the inhibitory effect of $ONOO^-$ on NADP-ME activity in a dose-dependent manner, which ranged from 33% with 0.1 mM SIN-1 to 56% with 5 mM SIN-1. An immunoblot analysis, run with an antibody against 3-nitrotyrosine, was employed to corroborate the nitration of recombinant NADP-ME as a result of SIN-1 treatment (Figure 4C).

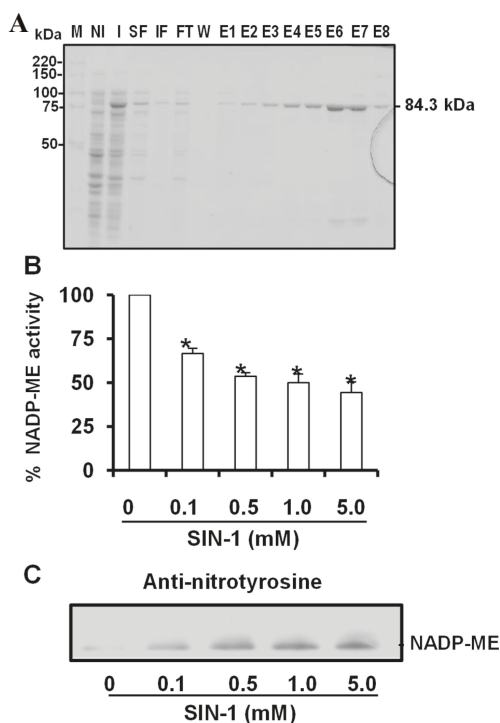


Figure 4. Purification of recombinant cytosolic NADP-ME2 and effect of SIN-1 (3-morpholinosydnonimine (peroxynitrite donor)) on its activity. **(A)** Purity of recombinant NADP-ME after expression and purification was evaluated by SDS-PAGE (10%) and Coomassie blue staining. M, molecular markers; NI, total protein in no-induced culture; I, total protein in induced culture; SF, soluble fraction; IF, insoluble fraction; FT, flow-through; W, wash; E1-E8, elution fractions. **(B)** Effect of SIN-1 on recombinant NADP-ME2 activity. **(C)** Representative immunoblot showing the degree of tyrosine nitration of NADP-ME2 treated with different concentrations of SIN-1 and detected with an antibody against 3-nitrotyrosine (dilution 1:2500). 3 μg of protein was used per line. The specific activity of recombinant NADP-ME was 9850 $\mu\text{mol NADPH min}^{-1} \text{mg}^{-1}$ protein. Data are the means \pm SEM of at least three replicates. * Differences from the control values were significant at $p < 0.05$.

3.4. Spectral Characterization of the Nitrated Recombinant *Arabidopsis* NADP-ME2

There are 25 tyrosine residues in the sequence of the cytosolic NADP-ME2 in *Arabidopsis* plants. Thus to be able to identify which of these residues is(are) a target(s) of this NO-related post-translational modification, the recombinant protein was treated with peroxynitrite, digested with trypsin, and the resulting peptides were analyzed by MALDI-TOF/TOF mass spectrometry. After the LC-MS/MS analysis, a list of scanned and identified peptides was obtained (Table S1), among which only one was identified to contain a nitrated tyrosine. Figure 5 shows the comparison of the nitrated (top) and unmodified (bottom) MS/MS spectra of these identified peptides from NADP-ME2. Nitrated peptide DAHYLTGLLPVILSQDVQER ($Z = 3$) had 21 amino acids and a mass of 2409 Da (2364 Da, plus 45 Da), which is compatible with Tyr-73 nitration (Figure 5).

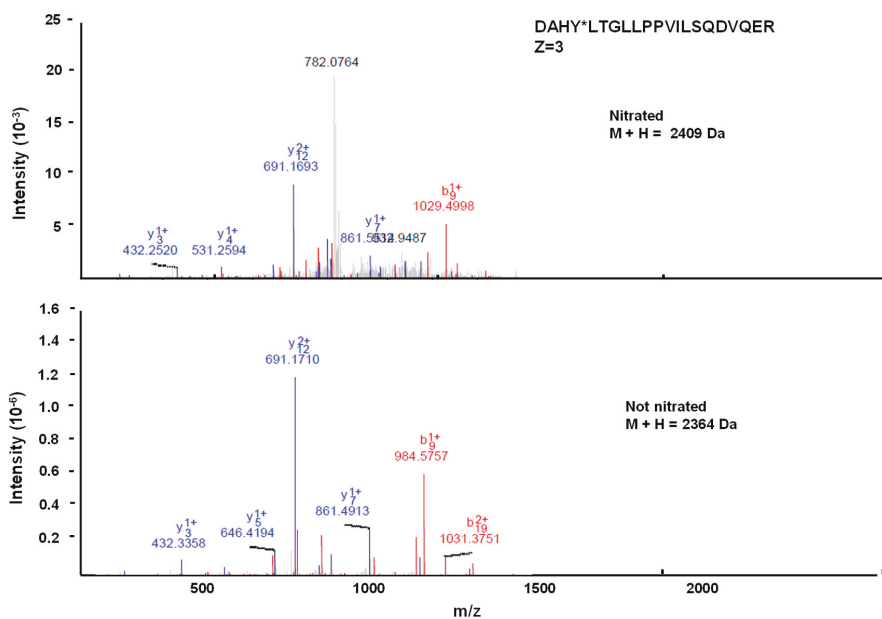


Figure 5. Comparison of the nitrated (top) and unmodified (bottom) MS/MS spectra of the identified peptide (DAHY*LTGLLPPVILSQDVQER) from *Arabidopsis thaliana* NADP-ME2. Peptide fragment ions are indicated by “b” if the charge is retained on the N-terminus (red) and by “y” if the charge is maintained on the C-terminus (blue). The subscript indicates the number of amino acid residues in the considered fragment from either N-terminus or C-terminus. The superscript indicates the charge (1+ or 2+) to the backbone fragmentation.

3.5. Modeling the NADP-ME2 from *Arabidopsis Thaliana*

The tertiary structure of NADP-ME2 from *A. thaliana* was modeled as described in the Materials and Methods section. The quality evaluation revealed that the models computed by servers RaptorX (95% coverage) and IntFold (100% coverage) gave scores, which led us to use IntFold for the final model and to carry out subsequent studies. This model was computed using Protein Data Bank (PDB) entries 2aw5 (human malic enzyme) and 1O0S (*Ascaris suum* malic enzyme complexed with NADH) as a template. These entries share 48% and 46% identity, respectively, with the sequence from *A. thaliana*; the RMSD (root-mean-square deviation) of the model with the templates being 0.62 Å with 2aw5 and 1.73 Å with 1o0s. The coordinates yielded a Qmean value [65] of 0.784, with a Qmean-Z score [66] of 0.2, and an Errat overall quality factor [63] of 78.26, and 89.3% of the residues had an averaged 3D-1D score above 0.2 (Verify3D) [62], while 93.7% of the residues were in the most favored regions in the Ramachandran plot and seven lay in outlier regions.

As NADP-malic enzymes are complex molecules with a double dimer structure that yields a dimer interface and a tetramer interface, the coordinates of the biological assembly of the enzyme from *Ascaris suum* (PDB code 1O0S) were used as a template to identify the interfaces in the *A. thaliana* model (Figure 6). At this point, it is noteworthy that no further refinements were carried out and that the overlapping regions were limited to face monomers from different dimers that did not share any interface.

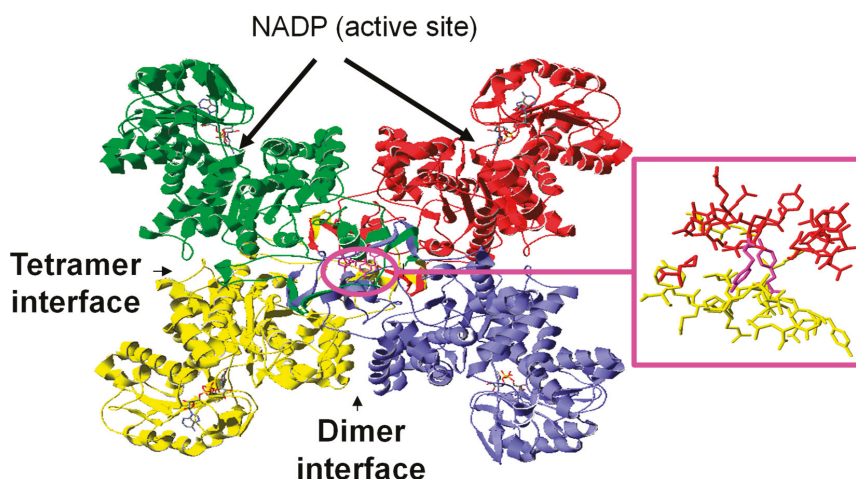


Figure 6. Schematic drawing of the model of the quaternary structure of the malic enzyme from *A. thaliana* showing Y73 (magenta) and details of the residues within 10 Å from Y73, which reveal its interfacial location. The four monomers are depicted in different colors.

3.6. Location of the Nitrated Tyrosine Site in NADP-ME2

The malic enzyme is a dimer of the dimer with close contacts at the dimer interface, whereas the association of the two dimers to yield the tetramer is weaker. In general, the tertiary structure of all malic enzymes is similar. However, small differences may influence catalytic and regulatory mechanisms. Each monomer is divided into four domains (A, B, C, and D) that behave as rigid bodies during the conformational transition from the open to the closed form upon the binding of the divalent cation and the substrate [75].

The evolutionary analysis of the 25 Tyr residues of the NADP-ME2 sequence from *A. thaliana* shows that 15 Tyr residues are preserved in plants, with Tyr-136 being absolutely preserved through evolution in plants, mammals, and bacteria (Table S2). Through these 15 preserved residues, Tyr 158, Tyr242, Tyr457, Tyr 491, and Tyr 592 are predicted as being highly susceptible to phosphorylation (score > 0.75). Among the Tyr residues preserved in plants and predicted as non-phosphorylated, Tyr73 was identified as being nitrated by mass spectrometry. Tyr73 was a particular residue showing the highest estimated pKa (Table S2) which, according to the model, was located on the surface by protruding into the opposite monomer (Figure 6).

4. Discussion

Low temperature (LT) is an important environmental factor that significantly affects plant development and ROS metabolism [76]. *Arabidopsis thaliana*, as a model plant, has been used to analyze the effects of LT on physiology, gene expression, and metabolism at different levels, including photosynthesis, polyamines, and ROS, carbon, and nitrogen metabolisms [77–83]. Previous data have shown that NADPH regeneration by a family of NADP-dehydrogenases can prove essential as a cofactor in cellular redox homeostasis as it is an indispensable electron donor in several enzymatic reactions, biosynthetic pathways, and detoxification processes [30]. The involvement of NADPH in the metabolism of ROS and RNS has also been established as being necessary for not only the functioning of the ascorbate–glutathione cycle, via its participation in the regeneration of reduced glutathione (GSH) by glutathione reductase [84] but also for the activity of the NADPH-dependent thioredoxin system [85]. Consequently, NADPH appears to be important in the functioning of these central cell antioxidants by playing a key role in plant response to oxidative damage.

Similarly, NADPH is also necessary for $O_2^{\bullet-}$ generation by NADPH oxidase (NOX) [86] and is a cofactor required for NO generation by L-arginine-dependent nitric oxide synthase activity [87,88]. In addition, NADPH-generating enzymes appear to be involved in the plant response to different abiotic stresses [17–23]

In this study, by using *Arabidopsis thaliana* as a model, we analyzed the mechanism of response to short-term LT, specifically in relation to the status of some ROS, RNS, and NADP-malic enzyme activity. Additionally, the potential regulatory effect of tyrosine nitration on NADP-ME2 activity was analyzed in particular by an in vitro approach.

4.1. Differential Behaviors of NADP-Dehydrogenases, ROS, and RNS Molecules during Short-Term LT

The ROS metabolism under LT stress has been analyzed in several studies using different plant species [76,89–95]. More recently, occurrence of NO in higher plants and its importance under different stress conditions have become a new area of intense research to understand the role of NO and its related molecules (RNS) in the signaling events leading to plant response to LT stress conditions and their interaction with the ROS metabolism [1,11,96–100].

Cold acclimation in LT situations is the result of different processes, such as cryoprotectant and phospholipid production, protein stabilization, ion homeostasis maintenance, and stress situations, which are usually related to the scavenging of reactive oxygen species (ROS) [11,83]. In fact, LT-induced oxidative stress in *Arabidopsis* has been previously described and is characterized by an increase in lipid oxidation and H_2O_2 content, and by variation in antioxidant enzymes and the reduced functionality of catalase activity, ascorbate peroxidase (APX) and also soluble antioxidants, such as ascorbate [101–103]. The combined effect of less antioxidant capacity and increased ROS is associated with the damage and degradation of photosystems I and II, which finally affect plant growth [81,104]. Our finding, which indicates increased superoxide radical, falls in line closely with all these previously reported data, while our experimental LT conditions (4 °C for 48 h) led to a highly significant reduction in growth (Figure 1) accompanied by oxidative stress.

Similarly, previous data have shown that LT also provokes increased NO content in different plant organs [98,105], particularly in *Arabidopsis thaliana* leaves [96,106]. The increase in NO content observed in the roots of *Arabidopsis* seedlings also falls closely in line with these studies. However, an increase in peroxynitrite ($ONOO^-$) content under LT stress conditions has not yet been reported. This increase would be expected to occur as $ONOO^-$ is quickly produced by a chemical reaction that takes place between superoxide and nitric oxide radicals [107], both of which increase at LT. $ONOO^-$ is the most powerful oxidant that can be produced in nitro-oxidative stress situations, such as salinity and cadmium stress [73,108], which can mediate the tyrosine nitration of proteins and affect the function of ferredoxin-NADP reductase, carbonic anhydrase, NADH-dependent hydroxypyruvate reductase, and different components of the ascorbate–glutathione cycle [37,41,47,48,109,110]. This NO-related posttranslational modification must, therefore, be a mechanism to extend the regulatory and signaling effects of NO under physiological and stress conditions.

Different NADP-dehydrogenases have been shown to be modulated by diverse types of abiotic stresses; e.g., heavy metal [17,111,112], ozone [113], salinity [18,23,114,115], drought [116], and wounding [117]. In pea plants left under natural senescence conditions in which ROS and RNS increased, we also recently identified Tyr392 to be a target of tyrosine nitration that inhibits cytosolic NADP-ICDH [42].

Under our experimental conditions, the analyzed NADP-dehydrogenase activities were differentially affected by LT in *Arabidopsis* seedlings due to the reduced activity caused by NADP-ME activity, while G6PDH and NADP-ICDH activities remained unaffected. These results suggest that NADP-ME could be the NADP-dehydrogenase enzyme that is more sensitive to the oxidation state generated by ROS and RNS production at LT stress. Interestingly, NADP-ME is also the most sensitive to H_2S treatment in sweet pepper fruits [118]. This suggests that NADP-ME could be affected at low oxidation levels during the course of this abiotic stress, whereas the function of other dehydrogenases

(G6PDH and ICDH) could remain in an attempt to maintain the cellular redox state by producing NADPH under nitro-oxidative stress. This could be crucial for plant survival under environmental insults as NADPH acts as a vital electron donor in several reductive and detoxification processes, and in antioxidant processes. Furthermore, NADPH is required in the metabolism of reactive oxygen and nitrogen species (ROS and RNS) [51]. Similar to NADP-dehydrogenase enzymes, NO-PTMs differentially regulate Asc-GSH cycle components during plant response to the nitro-oxidative stress generated by salinity [47,48]. In this case, APX and monodehydroascorbate reductase (MDAR) are inactivated by tyrosine nitration, whereas glutathione reductase (GR) is not affected by this NO-PTM, which suggests a key role for GR to maintain the GSH level needed for the functioning of this antioxidant system. As the Asc-GSH cycle requires NADPH for functioning, this key antioxidant system could be regulated by tyrosine nitration at both levels: directly by regulating the function of its components; indirectly by regulating the enzymes that produce reduced power in the form of NADPH. However, these data differ from those on the behavior of NADP-dehydrogenase activities in other species exposed to LT. For example, pepper plants subjected to LT conditions (8 °C) showed an increase in the oxidative markers, such as lipid oxidation and protein nitration, together with an overall induction of the activity of major NADPH-generating systems (6PGDH, NADP-ICDH, G6PDH, and NADP-ME) after 24-hour treatments. These different results could be a result of the different exposure times to LT stress and, therefore, suggest that the oxidation level reached during this stress could be responsible for the specific modulation of NADPH-generating enzymes.

4.2. NADP-ME Activity Is Dysregulated by Nitration

As stated above, diverse types of environmental conditions have the capacity to modulate different NADP-dehydrogenases. Furthermore, nitro-oxidative stress situations can generate peroxynitrite (ONOO⁻) by a quick chemical reaction between superoxide radical and nitric oxide [107], which can mediate tyrosine nitration and affect the function of several proteins [47,48,109,110]. As NADPH regeneration can be a key cofactor in cellular redox homeostasis, the regulation of NADPH-generating enzymes by NO could play a key role in plant response to these adverse situations as these systems may act as a second line of defense to maintain the functioning of the main antioxidant systems [51]. Indeed, the activity of different NADPH-generating enzymes is regulated by RNS. For instance, ferredoxin-NADP reductase activity is inhibited by tyrosine nitration under high-temperature stress [41], and NADP-dependent isocitrate dehydrogenase activity is also inhibited by Tyr392 nitration, accompanied by increased ROS and RNS under natural senescence conditions [42]. When taking all these data together, tyrosine nitration seems to inhibit NADPH-generating enzymes in these situations. With NADP-EM2, Tyr73 nitration also inhibits the activity of the recombinant protein in a concentration-dependent manner. The inactivation of different NADP-generating enzymes, such as FNR, NADP-ICDH, or NADP-ME2, suggests that this NO-PTM may impair NADPH levels and, therefore, jeopardize the cellular redox state. To determine the functional role of this Tyr73 as a target of tyrosine nitration, we performed an *in silico* analysis. Previous reports indicate that chemical tetranitromethane (TNM) is able to mediate both cysteine oxidation and tyrosine nitration [119]. In line with this, there are also reports that TNM induces the complete inhibition of NADP-ME activity [120]. Loss of enzyme activity has been observed at pH 8.0, but not at pH 6.3, while NADP-ME has almost been 90% deactivated by incubation with an 80-fold molar excess of TNM for 5 min at 30 °C. Substrate L-malate or Mg²⁺ alone offers no protection, while NADP provided considerable protection. As reported for other NADP-dehydrogenases, in the presence of L-malate and Mg²⁺, NADP totally protects enzyme activity. This suggests that tyrosine residue may be located at or near the active site of maize NADP-ME. Spectral analysis of the modified enzyme has indicated that the modification of at least one tyrosine residue per subunit results in the complete loss of enzyme activity. The fluorescence study of unmodified and modified enzymes has postulated that the essential tyrosine residue of maize NADP-ME is possibly involved in L-malate binding. By using knockout *Arabidopsis* mutants of NADP-ME2, it has been recently suggested that this enzyme does not play a key role in the

response mechanism to oxidative stress [121]. In this context, our data indicated that this enzyme was negatively modulated by several RNS produced under nitro-oxidative stress conditions caused by short-term LT. This could partly explain why NADP-ME does not appear to be involved in oxidative stress because it appears to be partially deactivated [121].

Tyrosine nitration is a NO-mediated posttranslational modification (NO-PTM) that, in some cases, modulates the function of a specified antioxidant enzyme [37,122]. For example, for pea cytosolic APX, a key enzyme for regulating H₂O₂, cellular levels under salinity and nitro-oxidative-associated stress conditions, the enzyme is jointly regulated by this NO-PTM. Thus peroxynitrite inhibits APX activity by the nitration of Tyr235, which is located at the bottom of the pocket where the haem group is enclosed [48]. Furthermore, MDAR activity, another component of the Asc-GSH cycle, is also inhibited by Try345 nitration, which could be involved in the cofactor binding site [47]. Interestingly, NADP-ICDH activity is also down-regulated by Tyr392 nitration as a result of the oxidative process generated during the natural senescence of pea plant roots. In this case, the nitration process also appears to be involved in the disruption of the enzyme-cofactor interaction [42]. In addition, peroxisomal hydroxypyruvate reductase, an enzyme involved in the photorespiration process, is down-regulated by tyrosine nitration. In this case, Tyr 198 nitration has been proposed to be responsible for this inhibition [110]. In this context, NADP-ME regulation by tyrosine nitration is another example of the key role played by tyrosine nitration in the regulation of the target protein function. It is noteworthy that this process becomes relevant when it regulates crucial processes in plants, such as antioxidant defense systems. However, the regulation mechanism is not always straightforward and, for NADP-ME, modulation is especially complex [75].

Tyr73 is a particular residue as determined by both location and estimated pKa. Residue Tyr73 is at a contact distance from Glu11 and Asp12 from the same monomer, and Glu78, Asp70, Tyr 587, and Arg588 from the neighbor monomer. The mechanism underlying the effect of Tyr73 nitration on functionality may be the disruption of these interactions as it has been reported for the pigeon malic enzyme that the residues located at the N-end play an important role in subunit interactions and Mn(II) malate binding [123].

5. Conclusions

The present data show that short-term LT stress affects the metabolism of both ROS and RNS in *Arabidopsis* seedlings, which appears to be accompanied by nitro-oxidative stress. Of all the different analyzed NADP-dehydrogenases, cytosolic NADP-ME was negatively modulated under LT conditions. In this context, the in vitro analysis that used recombinant *Arabidopsis* cytosolic NADP-ME2 treated with peroxynitrite enables us to demonstrate that this enzyme is negatively modulated by nitration, which falls in line with the reduction observed in NADP-ME activity under short-term LT stress and also limits NADPH supply. At the molecular level, Tyr73 was identified as the most likely residue to be involved in this negative NADP-ME regulation by this NO-PTM.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2076-3921/8/10/448/s1>, Table S1: List of peptides scanned and peptides identified by LC-MS/MS in the recombinant NADP-ME2 from *Arabidopsis thaliana*, Table S2: Evolutionary conservation of the Tyr residues in ME related sequences found in UniprotKB data base.

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Article

Amelioration of the Oxidative Stress Generated by Simple or Combined Abiotic Stress through the K^+ and Ca^{2+} Supplementation in Tomato Plants

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Abstract: Abiotic stressors such as drought, heat, or salinity are major causes of yield loss worldwide due to the oxidative burst generated under these conditions. Recent studies have revealed that plant response to a combination of different environmental stressors is unique and cannot be deduced from the response developed to each stress when applied individually. Some studies have demonstrated that a different management of some nutrients in the irrigation solution may provide an advantage to the plants against abiotic stressors. Thus, the aim of this study was to investigate if an increase in potassium (K^+) and calcium (Ca^{2+}) concentration in the nutrient solution may have a positive effect on the amelioration of oxidative stress which occurs under the combination of salinity and heat in tomato plants. Our results indicated that plants irrigated with an increase in K^+ and Ca^{2+} concentrations in the irrigation solution from 7mM (K^+) to 9.8 mM and from 4 mM (Ca^{2+}) to 5.6 mM, respectively, induced a recovery of the biomass production compared to the plants treated with salinity or salinity + heat, and subsequently irrigated with the regular Hoagland solution. This was correlated with a better performance of all the photosynthetic parameters, a reduction in the foliar concentration of H_2O_2 and a lower lipid peroxidation rate, and with a better performance of the antioxidant enzymes ascorbate peroxidase (APX), dehydroascorbate reductase (DHAR), glutathione reductase (GR), and NADPH oxidase. Our results showed that these enzymes were differentially regulated at the transcriptional level, showing a higher reactive oxygen species (ROS) detoxification efficiency under salinity and under the combination of salinity and heat, as compared to those plants irrigated with common Hoagland. An increase in K^+ and Ca^{2+} in the irrigation solution also induced a lower Na^+ accumulation in leaves and a higher K^+/Na^+ ratio. Thus, our study highlights the importance of the right management of the plant nutritional status and fertilization in order to counteract the deleterious effects of abiotic stress in plants.

Keywords: abiotic stress combination; heat stress; salinity; photosynthesis; lipid peroxidation; ROS detoxification; antioxidant-related genes; antioxidant enzymes

1. Introduction

Tomatoes (*Solanum lycopersicum*) are the second-most important cultivar in the world economically. Southeastern agriculture in Spain is characterized by its production of high-quality agronomical

products with a high added value that are commercialized in both national and international markets, resulting in large revenues for the Spanish regions dedicated to tomato cultivation. Unfortunately, adverse environmental conditions, such as salinity (coming from both low-quality irrigation water and salinized soils), scarcity of water, and high temperatures, negatively affect plant production and considerably increase economic losses. These environmental conditions are even more detrimental when they act in concert, resulting in great economic losses worldwide. On the other hand, the climate change predictions expected in the next 50–80 years, as published by the International Panel for Climate Change [1], indicate a worsening of the current environmental conditions that, together with the increase in the world's population, make clear the urgent need for obtaining cultivars and the proper management of crops, in order to increase abiotic stress tolerance and improve yields.

Salinity reduces crop production and negatively impacts the rate of leaf expansion, protein synthesis, energy production, lipid metabolism, opening of stomata, and photosynthesis through osmotic stress caused by soil water deficits, ion toxicity, and nutrient imbalances [2]. Similarly, heat stress affects biomass accumulation in crops of agronomic importance, such as tomato, by impairing the correct functioning of photosystem II (PSII) and reducing the chlorophyll content, resulting in a reduction in plant growth [3,4], which in turn negatively affects final fruit production.

In the last few years, our research group has successfully demonstrated that the combination of two or more abiotic stressors generated a specific plant response that could not be elucidated from the responses obtained when these stressors were applied individually [5–7]. This highlights the importance of studying abiotic stressors in combination rather than separately. Individual as well as combined abiotic stressors induce imbalances in the redox cell homeostasis due to the accumulation of reactive oxygen species (ROS), such as hydrogen peroxide (H_2O_2) [8]. Despite their potential to cause damage, it has been shown that, at non-toxic concentrations, H_2O_2 plays a key role in signal transduction mechanisms, which in turn activate defense responses to various stressors [9].

To remove excess ROS and to maintain redox homeostasis, plants have a well-integrated antioxidant defense system that is comprised of antioxidant compounds (i.e., ascorbate, glutathione, β -carotene, etc.) and enzymes, such as superoxide dismutase (SOD), catalase (CAT), and those involved in the ascorbate-glutathione (AsA–GSH) cycle [10]. Under control conditions, these enzymes are tightly regulated for achieving a high efficiency in ROS detoxification. However, abiotic stressors cause inhibition at transcriptional and post-transcriptional levels, as well as post-translational modifications that translate into a decrease in detoxification efficiency, and thus ROS accumulation and deleterious damage to the cells [8].

Plant growth and metabolism are also dependent on plant nutritional status. In this sense, potassium (K^+) and calcium (Ca^{2+}) nutrition plays a fundamental role. Most biochemical and physiological processes, such as protein synthesis, regulation of plant stomata, water use, control of ionic balance, activation of enzymes, and many other processes, are affected by K^+ , positively contributing to the survival of plants exposed to various abiotic stressors [11,12]. In plants subjected to salinity stress, it is common to find a K^+ deficiency [13,14], and a low concentration of K^+ under saline conditions can result in the formation of ROS and cell damage [15]. Also, it has been shown that Ca^{2+} is essential for selective transport by plant cells. An extra Ca^{2+} supply may contribute to cell wall and plasma membrane integrity, and thus may reduce the effect of salinity on plants [16,17]. Moreover, it has also been shown that K^+ nutrition improved with an application of external Ca^{2+} under salinity stress [18]. ROS have been described as playing an important role in the root's response to nutrient deprivation. In this sense, K^+ deprivation induced ROS accumulation in Arabidopsis [12,18,19] and tomato roots [20], with these ROS proposed to participate in the signaling cascade that results in the induction of the genes encoding for the high affinity K^+ transporter HAK5 [12]. The study of salinity and heat stress applied individually has contributed important information to the current knowledge on abiotic stressors, but recent research studies have indicated that a plant's response to a combination of two or more stressors is unique and cannot be elucidated from the study of single stressors [6,7,21–23], so stress combination studies are deemed very necessary. Thus, the aim of the

present study was to investigate the beneficial effects of an increase in K^+ and Ca^{2+} concentration in the irrigation solution on the tolerance of tomato plants to the combination of salinity and high temperature, through the amelioration that these elements can exert on the oxidative stress generated under these conditions.

2. Materials and Methods

2.1. Preliminary Experiments on K^+ and Ca^{2+} Doses in the Nutrient Solution

Before the main set of experiments was performed, a preliminary experiment was conducted in order to confirm the hypothesis that an increase in the concentrations of K^+ and Ca^{2+} in the irrigation solution of tomato plants could exert a positive effect on plant biomass production under an abiotic stress combination, and more specifically, under the combination of salinity and heat, two of the most devastating abiotic stressors on plant productivity that act jointly in the southeast of Spain. For this purpose, tomato seeds (*Solanum lycopersicum* L.) cultivar “Boludo” (kindly offered by Monsanto, Torre Pacheco, Murcia, Spain) were sown in vermiculite in a growth chamber under optimal and controlled conditions of light ($500 \mu\text{mol m}^{-2} \text{s}^{-1}$), photoperiod (16/8 h day/night), humidity (60–65%), and temperature (25 °C) (Chamber A, see Table 1 below). When the plants reached a height of 8–10 cm, they were transferred to 20 L containers (a total of 96 plants, two plants per container) and grown in an aerated hydroponic system containing Hoagland solution (Supplementary Table S1, [24]). The plants were grown under these conditions for 10 days in order to acclimate them to this growing system. After this time, half of the plants (48 plants) were transferred to another growth chamber with the same light, photoperiod, and humidity parameters described previously, but with the environmental (air) temperature set at 35 °C (Chamber B, see Table 1 below). At this time, the different treatments (T1, T2 and T3) were applied as follows ($n = 12$):

Table 1. Nutritional and environmental temperature treatments applied in the preliminary experiment to the tomato plants in order to select the most significant nutritional treatment for further experiments.

Growth Chamber	Treatments	[NaCl]	[K^+]	[Ca^{2+}]
Chamber A (25 °C)	Control	0 mM	7 mM	4 mM
	Salinity	60 mM	7 mM	4 mM
	Salinity + T1	60 mM	8 mM	4.7 mM
	Salinity + T2	60 mM	9.8 mM	5.6 mM
	Salinity + T3	60 mM	11 mM	6.5 mM
Chamber B (35 °C)	Heat	0 mM	7 mM	4 mM
	Salinity + Heat	60 mM	7 mM	4 mM
	Salinity + Heat + T1	60 mM	8 mM	4.7 mM
	Salinity + Heat + T2	60 mM	9.8 mM	5.6 mM
	Salinity + Heat + T3	60 mM	11 mM	6.5 mM

Plants were kept under these conditions for 21 days, after which time the fresh weight and dry weight of whole plants were recorded (Supplementary Figure S1). Based on the results obtained in this preliminary experiment, treatment 2 (T2) (i.e., 9.8 mM K^+ and 5.6 mM Ca^{2+}) was selected for further experiments, as it was the treatment that obtained a better biomass production under the stressors applied.

2.2. Experimental Design, Plant Material and Growth Conditions

Tomato seeds (*Solanum lycopersicum* L.) cv. “Boludo” were sown as described previously. When the plants were 8–10 cm in height. A total of 48 plants were transferred to 120 L tanks (with one plant

per tank) and grown in an aerated hydroponic system containing the control Hoagland solution (as described in Supplementary Table S1) for 10 days in order to adapt them to this system. At this time, the plants were divided between two polycarbonate greenhouses (24 plants per greenhouse) located in an experimental station in Santomera (Murcia, Spain) during the spring growing season, and the different treatments were applied for 30 days. Salinity was applied at the same time that plants were separated into the two different greenhouses, as shown in Table 2. Thus, the experiment consisted of six treatments, as follows:

Table 2. Nutritional and environmental temperature treatments applied to tomato plants under greenhouse conditions for the physiological, biochemical and molecular determinations.

Greenhouse	Treatments	[NaCl]	[K ⁺]	[Ca ²⁺]
Greenhouse A (25 °C)	Control	0 mM	7 mM	4 mM
	Salinity	60 mM	7 mM	4 mM
	Salinity + K/Ca	60 mM	9.8 mM	5.6 mM
Greenhouse B (35 °C)	Heat	0 mM	7 mM	4 mM
	Salinity + Heat	60 mM	7 mM	4 mM
	Salinity + Heat + K/Ca	60 mM	9.8 mM	5.6 mM

Control (25 °C), heat stress (35 °C), salinity stress (25 °C + 60 mM NaCl), salinity combined with increases in K⁺ and Ca²⁺ concentrations (25 °C, 60 mM NaCl, 9.8 mM K⁺ and 5.6 mM Ca²⁺), salinity and heat stress combined (60 mM NaCl + 35 °C), and salinity and heat stress combined with the increases in K⁺ and Ca²⁺ concentration in the nutrient solution (60 mM NaCl, 35 °C, 9.8 mM K⁺ and 5.6 mM Ca²⁺) were set as treatments. Eight tanks per treatment were randomly distributed in both greenhouses (Supplementary photograph 1). The pH of the all the nutrient solutions used in our experiments was maintained between 5.5 and 6.1. Water lost by transpiration was replaced every two days and nutrients were added every week to restore their initial concentrations. To apply the temperature treatments, this parameter was set at a maximum of 25 °C for greenhouse A (control) and 35 °C for greenhouse B. The greenhouse climate control system operated as described in Rodriguez-Ortega et al. [25].

2.3. Plant Sampling

Six plants were used per treatment. Thirty days after starting the treatments, the tomato plants were harvested and divided into leaves, stems, and roots, and their fresh weight (FW) and dry weight (DW) were recorded. Half of the harvested plant material was immediately submerged into liquid N₂ and stored for future procedures. Fully expanded leaves corresponding to the 6th–8th position (middle leaves) were separated and used for the biochemical and molecular measurements. The other half of the plant material was oven-dried for 72 h at 70 °C and weighed to determine the dry weight (DW). Dried plant material was used for ion analysis.

2.4. Ion Analysis and Quantification

Dried plant material was digested with HNO₃:HClO₄ (2:1, v:v), and Na⁺ and K⁺ concentrations were determined by atomic absorption spectrometry (Perkin-Elmer 5500, Waltham, MA, USA).

2.5. Leaf Gas Exchange

Net photosynthesis rate (A), stomatal conductance (gs), and transpiration rate (E) were measured at the end of the experiment in the youngest fully-expanded leaf of each plant (six plants per treatment), using a LI-6400XT photosynthesis system (Li-Cor, Inc., Lincoln, NE, USA). Leaf gas exchange was measured as described by Martinez et al. [6].

2.6. H₂O₂ Quantification

H₂O₂ was extracted as described by [26], with some modifications [23,27,28]. The concentration of H₂O₂ in the extracts was determined as described by Martinez et al. [6]

2.7. Lipid Peroxidation

Malondialdehyde (MDA), as a degradation product of lipid peroxidation, was determined as outlined by Fu and Huang [29] with the modifications listed by Mestre et al. [24], using an MDA extinction coefficient of 155 mM⁻¹ cm⁻¹.

2.8. RNA Extraction and qRT-PCR Experiments

Total RNA was isolated from whole tomato leaves, as described by Martinez et al. [6]. The primer sequences used for the different genes tested as well as the primer sequences of the internal controls are available in Supplementary Table S2. The normalization of transcript expression against housekeeping genes, reaction components, and PCR settings were carried out as described by Martinez et al. [6]. Expression rates as well as log₂ values of the different genes can be found in Supplementary Table S3.

2.9. Enzymatic Activities

All the enzymatic activities described were extracted as described by Martinez et al. [6]. Enzymatic activities were expressed as per mg protein [30].

CAT was assayed as described by Chance and Maehly [31]. The activity was expressed as the change in absorbance at 240 nm of a solution of 12.5 mM H₂O₂ in a 50 mM KH₂PO₄ buffer (pH 7.0) at 30 °C. CAT activity was calculated using an extinction coefficient of 39.4 mM⁻¹·cm⁻¹ [32].

Activities of the Cu/ZnSOD and FeSOD isoenzymes were assayed as described by McCord and Fridovich [33], with some modifications [34]. The reaction mixture and the quantification of Cu/ZnSOD and Fe SOD are fully described in Martinez et al. [6]

Cytosolic APX (cAPX), monodehydroascorbate reductase (MDHAR), and dehydroascorbate reductase (DHAR) activities were assayed as described by Miyake and Asada [35] and the full method can be found in Martinez et al. [6].

GR activity was assayed as described previously by Mestre et al. [24], with the modification of Martinez et al. [6].

NADPH oxidase activity was measured following the method described by Kaundal et al. [36]. The NADPH oxidase-dependent generation of O₂⁻ in plant tissue has been determined by the reduction of the tetrazolium salt 3'-[1-(phenylamino)-carbonyl]-3,4-tetrazolium}bis(4-methoxy-6-nitro)benzenesulfonic acid hydrate (XTT) by O₂⁻. In the presence of O₂⁻, XTT generates a soluble yellow formazan that can be quantified spectrophotometrically at 490 nm. The rate of O₂⁻ generation was calculated using an extinction coefficient of 2.16 × 10⁴ M⁻¹ cm⁻¹ [37].

Different absolute values obtained for each treatment and enzyme (Supplementary Table S4) were normalized against its control and log₂ values were calculated (Supplementary Table S5). Enzymatic activities were represented as a heat map of these log₂ values obtained.

2.10. Statistical Analysis

The data were first tested for homogeneity of variance and normality of distribution, and the Tukey HSD (Honestly-significant-difference) multiple range test was used to determine differences between means ($p \leq 0.05$), with $p < 0.05$ *, $p < 0.01$ **, $p < 0.001$ *** and n.s. as not significant. The relative transcript expression assayed by quantitative PCR (qPCR) was calculated using the 2^{-ΔCt} method. Heat maps for transcript expression and enzymatic activities were created using R.

3. Results and Discussion

In arid and semi-arid regions, the scarcity of good quality water for agriculture necessitates the use of saline water, which has caused scientists to focus on studying the effect of salinity on plants. However, salinity is usually aggravated by the high temperatures that are found in these regions [5]. Our research group has previously shown that the study of the abiotic stressors in combination is essential, as plants showed a specific response to this combination that could not be deduced from the response to the individually-applied stressors alone [5,7]. As shown in Figure 1, the FW and DW of tomato plants were negatively affected by the application of salinity and salinity + heat, with DW reductions of 33% and 38%, respectively, as compared with control plants (Figure 1B). When the irrigation solution was complemented with increases in K^+ and Ca^{2+} , the effects of salinity and the combination of salinity and heat were diminished, obtaining plants with less biomass than control plants, but with reductions of only about 10% (Figure 1B). Thus, plants grown under salinity or under the combination of salinity and heat but with higher concentrations of K^+ and Ca^{2+} in their nutrient solution increased their DW by 33% and 47% respectively, as compared to plants grown under these stressors and irrigated with the normal Hoagland solution. Capula-Rodriguez, et al. [38] highlighted the importance of the use of higher levels of K^+ and Ca^{2+} in the irrigation solution for the mitigation of the effect of the combination of multiple stressors (salinity, alkalinity, and boron) in tomato plants, thus supporting our results. These authors observed that plants grown under the combination of high alkalinity, salinity, and an excess of boron showed improved growth when supplemented with greater concentrations of Ca^{2+} and K^+ , which was related to enhanced phosphorous concentration, maintenance of chlorophyll *a* concentration, and/or the partial restoration of the uptake of other nutrients under these stress conditions.

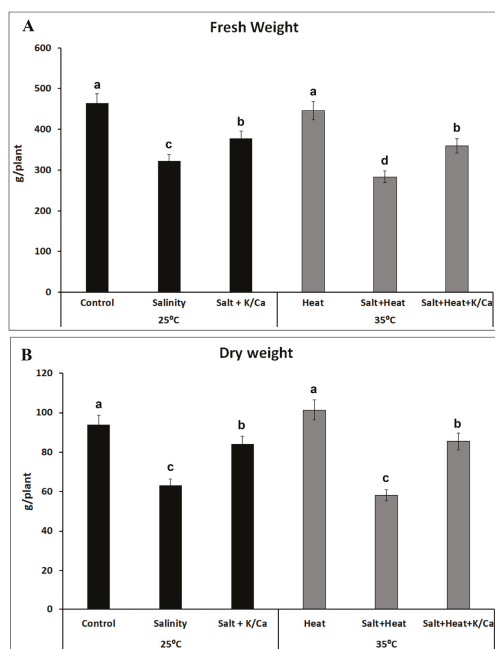


Figure 1. Fresh (A) and dry weight (B) of tomato plants grown under control, salinity (60 mM NaCl) or salinity and higher concentration of K^+ and Ca^{2+} in the irrigation solution at optimal temperature (25 °C) or heat stress (35 °C). Data represent means \pm SE ($n = 6$). Bars with different letters within each panel represent data with significant differences at $p < 0.05$ (Turkey HSD).

In general, salinity problems have been related to an excess of NaCl in the irrigation water, which causes Na⁺ toxicity in the plants, modifies the absorption of K⁺ and other nutrients by the roots, and exerts severe toxic effects on genes and enzymes, causing detrimental alterations to the plant's metabolism [39–41]. Therefore, it is vital for plants to re-establish cellular ionic homeostasis to maintain correct metabolic functioning and growth. Our ionic analysis in tomato leaves detected that heat, and the combination of salinity and heat, reduced the endogenous concentration of Ca²⁺ and K⁺ (Figure 2A,B). Ca²⁺ was reduced by about 18% by heat and the combination of salinity and heat (Figure 2A), whereas K⁺ was reduced by these stressors by about 23% (Figure 2B), as compared to plants grown under optimal conditions (control plants). However, when salinity and heat were applied jointly, and the irrigation solution was supplemented with higher concentrations of K⁺ and Ca²⁺, the endogenous Ca²⁺ concentration obtained for the leaves was similar to control plants. Salinity did not induce significant changes in the leaves' endogenous concentration of Ca²⁺ or K⁺ under any temperature treatment. However, surprisingly, when plants were grown under salinity and irrigated with higher K⁺ and Ca²⁺ concentrations, the endogenous concentration obtained for K⁺ was significantly lower than that obtained under control conditions, with a reduction of 12% (Figure 2B).

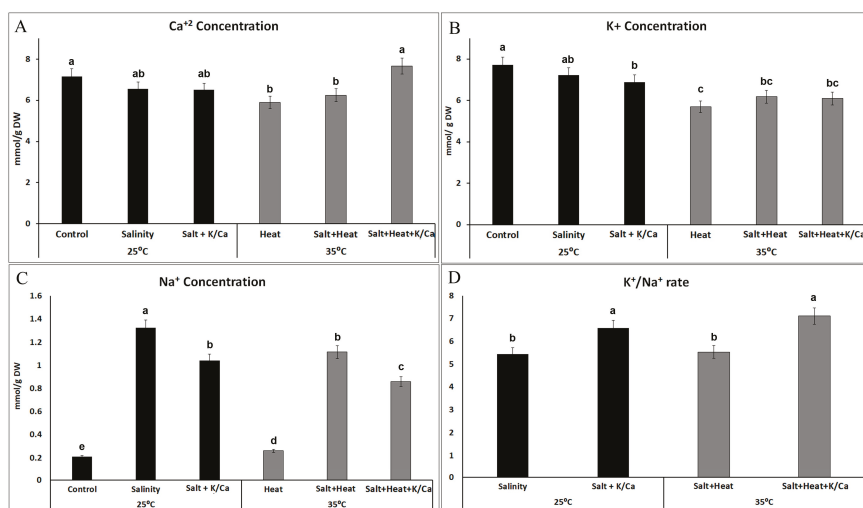


Figure 2. Ca²⁺, K⁺ and Na⁺ concentration (A–C) and K⁺/Na⁺ ratio (D) in tomato leaves grown under control, salinity (60 mM NaCl) or salinity combined with higher K⁺ and Ca²⁺ concentration in the irrigation solution under optimal temperature (25 °C) or under heat stress (35 °C). Values represent means ± SE (*n* = 6). Bars with different letters represent data with significant differences at *p* < 0.05 (Tukey HSD).

The irrigation of plants with a nutrient solution enriched in K⁺ and Ca²⁺ and grown under a combination of stressors did not induce significant changes in the K⁺ concentration as compared to plants irrigated with the normal Hoagland solution under the same environmental conditions (Figure 2B). These data coincided with those previously observed in a study by Martinez et al., where the combination of salinity and heat caused an inhibition of K⁺ uptake. These reductions could be due to the antagonistic effect of K⁺ with Na⁺ [42]. In order to confirm Na⁺ and K⁺ homeostasis in our experiments, the endogenous Na⁺ concentration of leaves was studied (Figure 2C).

As expected, the Na⁺ concentration increased significantly in those treatments where 60 mM NaCl was supplied. The salinity treatment had the highest Na⁺ concentration, and the application of salinity combined with heat induced a significantly lower Na⁺ accumulation in tomato leaves, with a reduction of 16% compared to salinity alone (Figure 2C). Also, the plants that received a nutrient solution

enriched with K^+ and Ca^{2+} and grown under stress conditions showed a lower Na^+ accumulation in the leaves (~22% less) as compared with plants grown under the same conditions but irrigated with the normal Hoagland solution. Thus, our results indicated that endogenous concentrations of Na^+ and K^+ were lower when plants were irrigated with a nutrient solution enriched with K^+ and Ca^{2+} under abiotic stress. The K^+/Na^+ ratio has been extensively related to the salt tolerance in plants [43]. In this sense, our experiments showed that this ratio was higher in those plants that were irrigated with higher concentration of K^+ and Ca^{2+} (Figure 2D) under salinity or under the combination of salinity and heat, which could be related with plants that have an increased tolerance to these stressors. In this light, we found a positive correlation between biomass production and the K^+/Na^+ ratio in plants irrigated with higher K^+ and Ca^{2+} and grown under salinity ($DW_{[Salt+K^+/Ca^{2+}]} - K^+/Na^+ \text{ ratio}_{[Salt+K^+/Ca^{2+}]}$, $r = 0.971^{***}$) or the salinity and heat combination ($DW_{[Salt+Heat+K^+/Ca^{2+}]} - K^+/Na^+ \text{ ratio}_{[Salt+Heat+K^+/Ca^{2+}]}$, $r = 0.811^{***}$). It has been postulated that some compounds (i.e., osmoprotectants and antioxidant compounds, among others) exert a regulatory function, maintaining a cytosolic K^+ concentration by preventing NaCl-induced K^+ leakage from the cell [12,14]. For example, K^+ has been reported to result in the accumulation of osmolytes and augmentation of antioxidant components in the plants exposed to water and salt stress [44]. Thus, our results may indicate that an increase in K^+ concentration in the nutrient solution may have an influence on the concentration of antioxidants and other beneficial compounds in plants, and therefore on the maintenance of the K^+/Na^+ homeostasis in plants. The increase in K^+ and Ca^{2+} concentrations in the irrigation solution improved tomato salinity tolerance under both optimal temperature and heat stress. The most aggressive stress treatment for tomato plants was salinity and heat; however, an increase in K^+ and Ca^{2+} concentration in the irrigation solution induced a higher recovery of these plants under salinity and heat stress.

At the end of the experimental period and before the plants were harvested, photosynthetic parameters, such as CO_2 assimilation rate, stomatal conductance, transpiration rate, and water use efficiency (calculated from the CO_2 assimilated versus the water transpired) were measured (Figure 3). Changes in these parameters are usually good indicators of stress in plants and they are directly correlated with plant growth. CO_2 assimilation rate (Figure 3A), transpiration rate (Figure 3B), and stomatal conductance (Figure 3C) had a similar behavior in our experiments with plants under stress conditions. As compared to control conditions, the salinity and salinity + heat treatments resulted in a significant reduction of those parameters with respect to control plants, with salinity + heat being more aggressive than salinity itself. These data were consistent with the findings by Pinero, et al. [45] and Rivero et al. [7], who reported that both the salinity and salinity + heat treatments led to a reduction in stomatal conductance, reducing the plant's ability to supply CO_2 to the photosynthetic apparatus, thus decreasing the CO_2 assimilation rate under these stress conditions. On the contrary, it has also been observed that heat stress alone can favor a greater transpiration rate, in order to reduce leaf temperature, which may lead to an increase in the CO_2 assimilation rate. The increase described for those parameters was also observed in these experiments, which were therefore in agreement with the results found by Rivero et al. [7]. Interestingly, when plants were grown under salinity or salinity + heat but supplemented with a nutrient solution enriched with K^+ and Ca^{2+} , CO_2 assimilation, transpiration rate, and stomatal conductance partially recovered to values similar or closer to control levels (Figure 3A–C). These photosynthetic parameters are closely related to growth rate in plants, and our results indicate a positive correlation among them, where a reduction in plant growth (Figure 1) was directly related with an inhibition of these photosynthetic parameters, and the recovery of these by an extra supply of the plants with K^+ and Ca^{2+} implied better growth under abiotic stress conditions. Abass et al. [44] and Bohra and Doerffling [46] found similar results in maize plants and rice, respectively, through the addition of K^+ to a saline soil. Both studies concluded that the addition of K^+ significantly alleviated the harmful effects of salinity, by improving plant growth and gas exchange parameters.

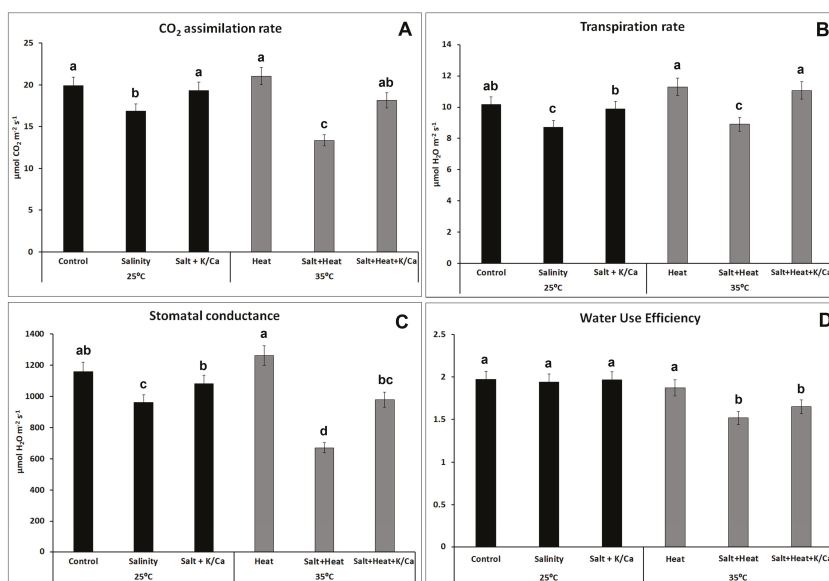


Figure 3. Photosynthetic parameters in tomato leaves grown under control, salinity (60 mM NaCl) or salinity combined with higher K^+ and Ca^{2+} concentration in the irrigation solution at optimal temperature (25 °C) or heat stress (35 °C). Values are means \pm SE ($n = 6$). Bars with different letters within each panel represent data with significant differences at $p < 0.05$ (Tukey HSD).

In addition to K^+ , Ca^{2+} also plays an important role in the plant cell. Among its numerous cell functions, Ca^{2+} contributes to the cell's signaling processes and helps with the detoxification of ROS [18,47,48]. Bhattacharjee [49] observed that the addition of Ca^{2+} helped to protect against heat-induced oxidative damage. It is widely known that salinity and heat stress can enhance oxygen-induced cellular damage due to a high production of ROS [6,22,50,51], so the importance of enhancing the antioxidant defense plant system to cope with these stressors should be highlighted. In recent studies, it has been observed that plants subjected to different combinations of abiotic stressors accumulated large amount of ROS response transcripts, which defined the importance of the antioxidant machinery on the acclimation pathways during combined stressors [50,52]. Martinez et al. [6] showed that tomato plants subjected to a combination of salinity and high temperature had an overproduction of ROS that caused photosynthesis inhibition, breakdown of photosynthetic pigments, and plant growth reduction. However, although the study of the combined effect of two abiotic stressors has begun and there are many publications that demonstrate the importance of studying stressors in combination, there is little information about the effect of the nutrition management and its possible relationship with abiotic stress combination tolerance in plants. In light of this question, a detailed study of the plant's oxidative metabolism was performed in our plants in order to verify a possible role of the extra supply of K^+ and/or Ca^{2+} on the enhancement of the antioxidant defense pathways in plants.

Under these environmental stress conditions, where water supply to the plant is being compromised (i.e., with regards to salinity, heat or both), the internal cellular CO_2 concentration is restricted due to stomatal closure. This, together with the continuous incident sunlight to the plant, results in a transfer of excess electrons to molecular oxygen, generating a superoxide radical ($O_2^- \bullet$), which is immediately detoxified to H_2O_2 by superoxide dismutases (SODs). In addition, the stress condition generates irreversible damage to the cellular membrane's polyunsaturated fatty acids. Thus, two of the most common oxidative stress markers used to measure oxidative damage to the cells are H_2O_2 concentration and membrane lipid peroxidation, measured as the concentration of MDA derivatives. These two

parameters were evaluated in our experiments with tomato plants (Figure 4). Any of the stress conditions used in our experiments resulted in a significant increase in the H_2O_2 concentration with respect to control plants, with the salinity treatment being the one with the highest concentration of this compound, which was found to be almost double the amount found in control plants. On the other hand, when plants were grown either under salinity or the combination of salinity and heat but irrigated with a nutrient solution enriched with K^+ and Ca^{2+} , the concentration of H_2O_2 was significantly lower than that found under the same environmental conditions but irrigated with the normal Hoagland solution. Thus, plants irrigated with higher K^+ and Ca^{2+} and grown under salinity had 20% less H_2O_2 than those grown under salinity but irrigated with the normal nutrient solution. Interestingly, when salinity was combined with heat, the extra supply of K^+ and Ca^{2+} in the nutrient solution resulted in a 30% reduction in H_2O_2 concentration. For the MDA content, the results obtained were similar, with the higher concentration of this compound found in plants under salinity and salinity + heat treatments and irrigated with the normal nutrient solution. Again, the addition of extra K^+ and Ca^{2+} to the nutrient solution resulted in a significant reduction in the level of lipid peroxidation of these plants under these stress conditions. H_2O_2 accumulation and membrane lipid peroxidation leads to oxidative damage to the cells, inhibiting photosynthesis and inducing plant growth inhibition [7,28]. Our results showed that tomato plants grown under these treatments, where H_2O_2 was accumulated to higher concentrations (Figure 4A), showed a higher membrane lipid peroxidation (Figure 4B) and an significant impairment of the photosynthetic parameters (Figure 3), with the concomitant inhibition of plant growth.

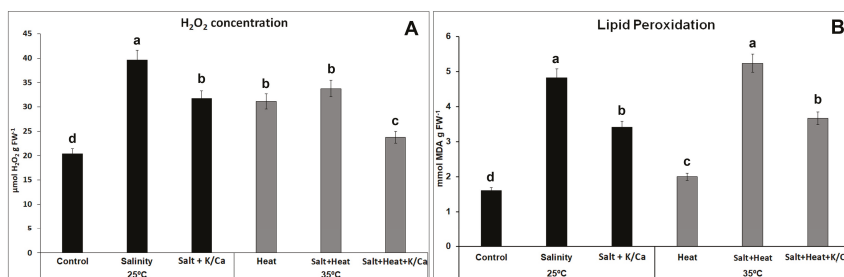


Figure 4. H_2O_2 concentration (A) and Lipid peroxidation rate, measured as MDA concentration (B) in tomato leaves grown under control, salinity (60 mM NaCl) or salinity combined with a higher concentration of K^+ and Ca^{2+} in the irrigation solution at optimal temperature (25 °C) or under heat stress (35 °C). Values represent means \pm SE ($n = 6$). Bars with different letters within each panel represent data with significant differences at $p < 0.05$ (Tukey HSD).

The impairment of the cell antioxidant machinery under abiotic stress conditions can be caused by an overproduction of ROS, an inhibition of the antioxidant enzymes, or both. This inhibition may occur at transcriptional or post-transcriptional levels, or by post-translational modifications of the antioxidant proteins necessary for counteracting the deleterious effects of ROS. In order to assess these impairments, the expression of the main oxidative metabolism-related transcripts and the activity of the main antioxidant enzymes were measured under the different stress treatments used in our study (Figure 4A). After examining both heatmaps (Figure 5A,B) it is clear that tomato plants have a specific antioxidant response under the combination of salinity and heat, and that the response of the genes or enzymes cannot be deduced from the individual responses obtained under salinity or heat applied individually. These results confirm those obtained previously by Rivero et al. [7], Martinez et al. [5], and Martinez et al. [6], and highlight the importance of studying abiotic stressors in combination. The expression levels of the transcripts that code for Fe-SOD and Cu/Zn-SOD were upregulated under all the stress conditions (Figure 5A), and correlated well with the SOD activity found, with the exception of Cu/Zn-SOD activity in plants grown under the combination of salinity + heat and irrigated with the nutrient solution enriched with K^+ and Ca^{2+} , in which case this enzyme was inhibited.

The first cellular line of defense for ROS detoxification is constituted by SODs, which detoxify superoxide radicals into H₂O₂. In this sense, the over-production of H₂O₂ observed under the different stress treatments applied in our experiments was due to an over-expression of Fe-SOD and Cu/Zn-SOD encoding transcripts (Figure 4A), with a concomitant increase in SOD activity. H₂O₂ levels found in the stress treatments of plants supplemented with the K⁺/Ca²⁺ enriched nutrient solution were lower than those of plants grown with the normal nutrient solution under the same stress conditions, which may be due to: (1) a lower production of this compound in these treatments, due to the beneficial effects of the extra supply of K⁺ and Ca²⁺ under stress conditions, or (2) a more effective H₂O₂ detoxification driven by an overexpression of Fe-SOD encoding transcripts (Figure 5A), with a concomitant increase in SOD activity (Figure 5B, Supplementary Table S4) under these conditions. H₂O₂ accumulation also has a deleterious effect in plant cells, which correlated well with the biomass production obtained in our experiments under all the stress treatments used (Figure 1). Therefore, an effective detoxification system for H₂O₂ must work in coordination in order to avoid the cellular accumulation of this compound. In this sense, CAT and the ascorbate–glutathione cycle act together to detoxify H₂O₂ from the cells to H₂O. In our experiments, CAT encoding transcripts were upregulated under salinity, independently of the nutrient solution used. Also, this transcript was upregulated under heat stress, whereas in the salinity + heat treatment it was downregulated. These results correlate well with the CAT activity obtained under these conditions, concluding that the extra supply of K⁺ and Ca²⁺ did not have any effect on the transcription and activity of this enzyme under stress conditions, but the combination of salinity and heat induced the inhibition of its transcription.

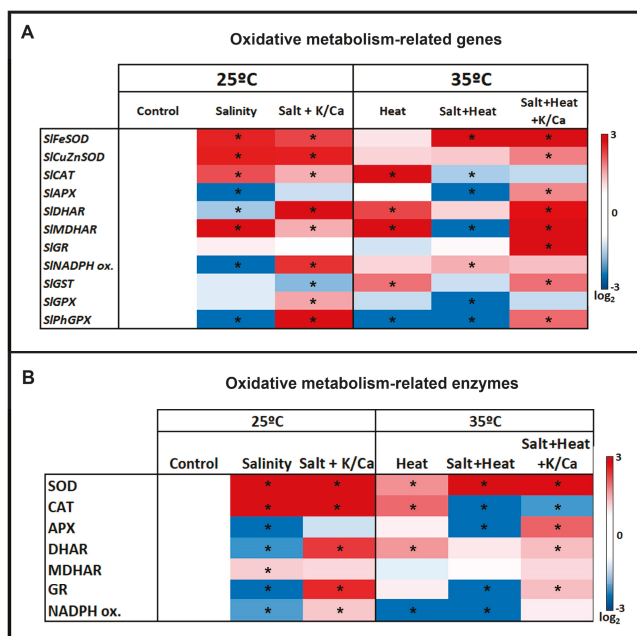


Figure 5. Oxidative metabolism-related gene expression (A) and enzymatic activities (B) in tomato leaves grown under control, salinity (60 mM NaCl) or salinity and higher concentration of K⁺ and Ca²⁺ in the irrigation solution at optimal temperature (25 °C) or heat stress (35 °C). Red color represents a higher relative expression or activity with respect to control plants and blue color represents a lower relative expression or activity. Scale is the log₂ values of the expression after normalization with respect to control plants. Absolute values (gene expression and enzymatic activities) as well as log₂ values can be found in Supporting Information Tables S2–S5, respectively. Asterisks are representative of significant differences with respect to control plants (*p* < 0.05).

The ascorbate–glutathione (AsA–GSH) cycle acts in parallel to CAT in the effective detoxification of H_2O_2 . In our experiments, the transcription levels of the different transcripts that codify for the antioxidant enzymes from the AsA–GSH cycle were differently affected depending on the stress treatment and the irrigation solution used. Thus, salinity stress downregulated the transcripts that codify for APX, DHAR, and NADPH oxidase, with the concomitant inhibition of their related enzymes. Interestingly, the supplementation of plants with a nutrient solution enriched with K^+ and Ca^{2+} reverted the regulation observed for DHAR and NADPH oxidase under salinity, inducing the upregulation of those genes as compared to control plants. However, the activity of DHAR and NADPH oxidase did not show significant differences with respect to control plants. MDHAR transcripts were upregulated and the activity of this enzyme was also higher under salinity as compared to control plants, independently of the irrigation solution used. Heat stress induced an upregulation of all the transcripts belonging to the AsA–GSH pathway, except for *SIAPX* and *SIGR*, which did not show significant differences with respect to control plants. When salinity and heat stress were applied jointly, a general downregulation of the transcripts codifying for the enzymes belonging to the AsA–GSH pathway was observed, with the concomitant inhibition of their enzymatic activities. However, the feeding of the plants with the nutrient solution enriched with K^+ and Ca^{2+} reverted the expression profile observed under the salinity and heat combination for most of the genes, with an upregulation pattern observed in all of them as compared to the control plants, which correlated with the activities measured for these enzymes. The inhibition of one or more of these enzymes in plants grown under salinity or under salinity + heat may explain the high H_2O_2 concentration found under these treatments, where the detoxification of this compound was not as efficient as compared to control plants. This resulted in the accumulation of this compound, oxidative damage, and plant growth inhibition [7,28]. On the other hand, the supplementation of plants with a nutrient solution containing a higher concentration of K^+ and Ca^{2+} seemed to induce the transcription of all the genes involved in the efficient detoxification of H_2O_2 (i.e., *SIAPX*, *SIDHAR*, *SIMDHAR*, *SIGR*, *SINADPH* oxidase), significantly reducing the oxidative damage observed in plants grown under the combination of salinity + heat, and the activation of stress-related genes implied in the signaling mechanisms for ROS detoxification [10]. Thus, K^+ and Ca^{2+} may contribute to the maintenance of the cellular redox homeostasis, protecting photosynthesis and electron transport systems and increasing plant tolerance to abiotic stressors.

GST, PhGPX, and GPX catalyze the GSH-dependent reduction of H_2O_2 and organic peroxides, including lipid peroxides to H_2O or alcohols [44]. In our experiments, the expression of the transcripts *SIGST*, *SIPhGPX*, and *SIGPX* were determined in order to compare them with the level of lipid peroxidation observed under the different treatments and to verify if a supplementation of the plant irrigation solution with K^+ and Ca^{2+} might have any effect on the transcription of these genes. Our results showed that under salinity and salinity + heat, these three genes were downregulated with respect to control plants. However, in plants fed with a nutrient solution enriched with K^+ and Ca^{2+} , an upregulation of the *SIGPX* and *SIPhGPX* (under salinity) and of the *SIGST* and *SIPhGPX* (under salinity + heat) was observed. These results correlated well with the data obtained for lipid peroxidation level in these plants (Figure 4B), where it was observed that K^+ and Ca^{2+} used in higher than recommended concentrations lowered the lipid peroxide level under salinity and salinity + heat as compared with plants irrigated with the normal nutrient solution and grown under the same stress conditions. In this sense, our results may indicate that K^+ and Ca^{2+} could play a role in the transcriptional regulation of the genes that codify for the enzymes involved in the reduction of the H_2O_2 and, consequently, may also play a role in the protection of membrane lipids from peroxidation in plants under abiotic stress conditions.

4. Conclusions

The results obtained in this work highlight the complexity and importance of using the correct supplementation in the plant irrigation solution. We have demonstrated that the plant's tolerance to salinity—and more importantly, to the combination of salinity and heat—can be improved by

the proper enrichment of the nutrient solution with K^+ and Ca^{2+} . These nutrients, used in higher concentrations than recommended, helped to control oxidative damage through the transcriptional regulation of the main enzymes implicated in ROS detoxification and the recovery of membrane lipid peroxides. Consequently, a better performance of the photosynthetic apparatus was maintained, leading to optimal growth rates. Our study highlights the importance of plant nutrition in the signaling processes that control plant tolerance responses to abiotic stress combination. However, more studies are needed in order to fully elucidate the role of K^+ and Ca^{2+} in these signaling mechanisms.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2076-3921/8/4/81/s1>, Table S1: Nutritional composition of the Hoagland solution used for tomato plants growth, Table S2: Primers used for the quantification of the expression levels of the oxidative metabolism-related transcripts by qPCR, Table S3: Relative expression values of the oxidative metabolism-related transcripts. Values were normalized against control samples and log₂ was calculated and shown. Values are means of $n = 3$, Table S4: Absolute activities of the oxidative metabolism-related enzymes. Values obtained were normalized using soluble protein content of each sample and treatment. Values are the means \pm SE ($n = 3$), Table S5: Log₂ of the oxidative metabolism-related enzymes. Values obtained in Table S3 were normalized against control and log₂ was calculated, Figure S1. Fresh weight (FW) of tomato plants obtained at the end of the preliminary experiment (see Materials and Methods section).

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Article

The Ageing Process Affects the Antioxidant Defences and the Poly (ADP-ribose) Activity in *Cistus Incanus* L. Leaves

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Abstract: The ageing process in living organisms is characterised by the accumulation of several deleterious changes occurring in cells and tissues. The increase of reactive oxygen species with the advancement of age is responsible for the oxidative damage to proteins, lipids and DNA, enhancing the risk of diseases. The antioxidant response and the activation of the poly(ADP-ribose)ation process represent the first defences activated by organisms at all life stages to counteract damage to cell structures and genomic material. The regulation of poly(ADP-ribose)ation with age is little known in plants, especially in combination with antioxidant defences modulation. In this study, the relationships between poly (ADP-ribose) polymerase (PARP) activity and enzymatic and non-enzymatic antioxidant pool have been studied together with the photosynthetic apparatus efficiency in the Mediterranean species *Cistus incanus* L., examining leaves at different developmental stages: young, mature and senescent. The photosynthetic performance was evaluated by chlorophyll *a* fluorescence measurement, the total soluble and fat-soluble antioxidant capacity, as well as the activities of enzymes superoxide dismutase (SOD), catalase (CAT), peroxidase (POD) and glutathione-S-transferase (GST), were determined by spectrophotometer, PARP activity was assessed by radioactive labelling. The highest photochemical activity was observed in young leaves, together with the highest GST activity. With the progress of the ageing process, the non-enzymatic antioxidant pool (namely ascorbic acid, α -tocopherol) declined, reaching the lowest value in senescent leaves, whereas PARP activity rose significantly. The overall results indicate that the decline of photosynthetic apparatus efficiency during senescence is due to the reduction of specific defences against oxidative damages, which increase the damages to DNA, as demonstrated by PARP activity rise.

Keywords: *Cistus incanus* L.; photosynthetic apparatus; antioxidants; poly (ADP-ribose) polymerase (PARP) activity

1. Introduction

Ageing is a natural process associated with the time-dependent general decline in the physiological function of an organism. It represents a multifactorial phenomenon, including genetic, physiological and biochemical changes related to the natural process of growth, to genetic defects and to the relationship between genotype and environmental conditions [1–3].

In plants, leaf ageing is a tightly regulated process with a crucial biological purpose: during senescence, metabolic changes and an ordered degradation of structures occur in cells. The decrease of stomatal conductance and photosynthetic rates in the leaves [4] are followed by the degradation of chlorophyll molecules responsible for changes of the leaf colour [5]. By contrast with animals, evident alterations are not observed in the mitochondria and the nucleus that remain intact until the final stages of leaf senescence. Metabolic changes include the hydrolysis of proteins, lipids, nucleic acids and pigments, that are accumulated during the growth phase [6]. At the cellular level, oxidative stress plays an essential role in the ageing process, which seems to be strongly associated with the changes in the prooxidant/antioxidant balance. The oxidative stress occurs when the reactive oxygen species (ROS) generation is transiently or chronically enhanced, and the antioxidant protection system does not counteract the disturbed physiological condition [7]. ROS are generally short-lived highly reactive molecules, derived from the partial reduction of oxygen; they are endogenously generated from healthy cellular metabolism or produced from exogenous sources, including pesticides, ultraviolet (UV) light, metal ions, smoke, ionizing radiation [8–10]. ROS represent a continuous challenge for eukaryotic cells, which may maintain under control their excess to avoid apoptosis, necrosis, autophagy and senescence [11–14]. The cell has particular defence mechanisms in protecting against ROS excess, including the enzymatic and non-enzymatic antioxidants. The main antioxidant enzymes are superoxide dismutase (SOD), catalase (CAT), peroxidase (POD), glutathione peroxidase (GPX), glutathione reductase (GR), and glutathione S transferase (GST) [15]. The non-enzymatic antioxidant defence system includes ascorbic acid (vitamin C), α -tocopherol (vitamin E), glutathione (GSH) and β -carotene [16]. The capacity of cells to counteract the overproduction of free radicals declines with age. The consequences are cumulative damages to important biological macromolecules, like DNA, proteins, and lipids [17,18]. The DNA may be damaged as single-strand breaks (SSBs), double-strand breaks (DSBs), oxidized bases and cross-linking sites [19,20]. The failure in damage repair leads to the cell surviving with altered genetic information. Alternatively, a severe mutational load may cause cell death [21]. In plants such as in animals, several DNA repair mechanisms have evolved to guarantee the integrity of genetic information. The DNA lesions produced by ROS are mainly restored by the base excision repair pathway [22–24]. During the process of base excision repair, two nuclear proteins, the poly(ADP-ribose) polymerase 1 (PARP-1) and the poly(ADP-ribose) polymerase 2 (PARP-2), regulate the accessibility of nicked DNA to other repair enzymes [25]. The poly(ADP-ribose)ylation (PARylation) process represents one of the first cellular responses to oxidative and other types of DNA damages being “sensors” of DNA damaged and involved in the maintenance of genomic stability [26]. Under normal conditions, PARPs have a shallow basal enzymatic activity, which increases dramatically under conditions of cellular stress [27,28]. PARPs activation induces the synthesis of poly(ADP-ribose) (PAR) from nicotinamide adenine dinucleotide (NAD⁺) and the release of nicotinamide as reaction by-product [29,30]. Defects in DNA repair lead to PARP activation and progressive oxidative DNA damage with ageing [31]. A strong correlation has been found between resistance to different stressors, including oxidative stress and longevity in mammalian cells [32,33]. The highest PARylation levels were in long-lived species, in which similar levels of PARP protein were expressed [32]. Thus, poly(ADP-ribose)ylation capacity was correlated with the ageing process and species-specific longevity [26,34]. Besides, during senescence, in eukaryotes, the activity, stability and localization of several proteins are affected by different post-translational modifications, such as phosphorylation, glycosylation, ubiquitylation, methylation and acetylation [4]. Histone acetylation, methylation, phosphorylation, and PARylation are known to influence the remodelling of the chromatin structure, regulating DNA replication and accessibility of the transcriptional machinery to it with indirect control of gene expression. Plant PARylation was first described in the 1970s, and since then nine proteins with PARP signature were characterized [35–38]. Besides performing similar functions to those described for the animal counterparts, plant PARPs [39] are also implicated in response to abiotic and biotic stresses [40–42], in stress tolerance [38,43] and in developmental processes [44]. The role of

poly(ADP-ribose)ylation and its implication in the ageing process has been addressed in several studies concerning animals, but not in plants.

The aim of the present research was to study the modulation of the antioxidant defences (enzymatic and non-enzymatic antioxidant pool) and poly(ADP-ribose)ylation in response to leaf development to understand the mechanisms at the basis of age-related prooxidant/antioxidant balance and decline of photosynthetic efficiency. To address this, we measured simultaneously oxidative stress markers, namely total soluble and fat-soluble antioxidant capacity, and enzymatic activity of catalase (CAT), superoxide dismutase (SOD), peroxidase (POD) and glutathione-S-transferase (GST) together with measurements of chlorophyll fluorescence emission and PARP activity in young (10-day-old, Y), mature (30 day old; M) and old (45 day old; S) summer leaves of *Cistus incanus* L. In this species the occurrence of a poly (ADP-ribose) polymerase enzyme of 80 kDa, able to synthesize a poly (ADP-ribose) of about 8–10 ADP-ribose unit was already demonstrated [41]. The Mediterranean perennial shrub used in this study is typical of the Mediterranean maquis ecosystems [45] and, from the physiological point of view, is particularly interesting because during the summertime it develops summer leaves especially suitable to cope with drought condition typical of Mediterranean areas [43,46,47]. These leaves exhibit peculiar traits compared to winter ones appearing on the branches in early spring and remaining until the beginning of the winter; during their long lifespan, these leaves experiment and withstand many environmental constraints.

2. Materials and Methods

2.1. Plant Material and Growth Conditions

We conducted the experiment on summer leaves of *Cistus incanus* L. subsp. *incanus*, a seasonally dimorphic species with small xeromorphic leaves (summer habitus) produced by plants from the beginning of May until beginning of June as opposed to large mesomorphic leaves (winter habitus) produced from the end of October until beginning of November [43,45,46].

In early May 2014, 10 healthy plants of *C. incanus*, approximately three years old, were screened in the field (Castelvoturno Nature Reserve, Tyrrhenian coast of southern Italy, Naples) for size and uniformity, excavated from field and immediately transplanted in 15 L pots filled with native soil, transported at Department of Biology of University of Naples Federico II and grown in glasshouse in semi-controlled conditions at a mean temperature of 23 ± 2 °C, 65–70% Relative Humidity (RH), 13 h of photoperiod. In order to avoid any additional stress, plants were watered three times a week with variations depending on the evaporative demand of the environment. At the end of June, after 20 days of acclimation at the glasshouse environmental condition, young (Y—10 days old), mature (M—30 days old) and senescent (S—45 days old) summer leaves were selected for fluorescence emission analyses and determination of enzymatic and non-enzymatic antioxidants and PARP activity.

2.2. Chlorophyll *a* Fluorescence Measurement

Chlorophyll *a* fluorescence measurements were performed on attached young (Y), mature (M) and senescent (S) leaves of *C. incanus* using the Mini-Pam fluorometer (Walz, Germany) equipped with a leaf-clip holder (Leaf-Clip Holder 2030-B, Walz, Germany) which allows the simultaneous recording of the incident Photosynthetic Photon Flux Density (PPFD) on the leaf surface and abaxial leaf temperature. The maximal PSII photochemical efficiency (F_v/F_m) was measured in early morning on 30 min dark-adapted leaves [48]. For the quantum yield of PSII linear electron transport (Φ_{PSII}) determination [49], the fluorescence measurements were conducted at midday under natural environmental conditions of PPFD of $800 \pm (20)$ $\mu\text{mol photons m}^{-2} \text{ s}^{-1}$ and air temperature of $23 (\pm 2)$ °C.

2.3. Isolation of Nuclei and Western Blotting Analysis

Nuclei were isolated as described in Arena et al. [41]. All operations were performed on ice or at 4 °C. Samples of (Y), (M), and (S) leaves (0.1 g) were homogenized at low speed by an Ultra Turrax T8 homogenizer (IKA-WERKE) for 30–40 s. The buffer of homogenization (buffer A) contained 10 mM Tris-HCl pH 8.0, 1 mM ethylenediaminetetraacetic acid (EDTA), 1 mM egtazic acid (EGTA), 1 mM phenylmethylsulfonyl fluoride (PhMeSO₂F), 10 mM MgCl₂, 5 mM 2-mercaptoethanol, 1% Nonidet-P40 (1:4, *w/v*), and protease inhibitor cocktail (5–10 µg/mL). The homogenates were filtered through three layers of cheesecloth and centrifuged at 1500 g for 30 min at 4 °C. This last procedure was repeated four times. Finally, the pellets, which represent the nuclear fractions, were resuspended in a small volume of buffer A without NP 40.

Y, M, and S leaf nuclear fractions (20 µg) were analysed on 10% polyacrylamide gels in the presence of 0.1% sodium dodecyl sulphate (SDS) tension-active according to Arena et al. [41]. Gels were stained in 0.1% Coomassie G in 10% acetic acid and 30% methanol. For immunoblotting, the electrophoresed proteins were transferred onto a polyvinylidene fluoride (PVDF) membrane (Biorad) at 200 V for 1.5 h at 4 °C in the same buffer used for the electrophoretic run. Filter was placed in blocking solution, containing 50 mM Tris-HCl buffer, pH 8.0; 150 mM NaCl; 0.5% (*v/v*) Tween 20 and 3% gelatine for 1.5 h. Subsequently, the filter was incubated for 2 h at room temperature in the same solution integrated with 0, 3% gelatine, in the presence of commercial polyclonal anti-PARP human antibodies (H-250, Santa Cruz), diluted 1:2000, *v/v*. After repeated washings in TBS-Tween 0.5%, the filter was incubated for 1 h, at room temperature, in TBS-Tween 0.5% and 0.3% gelatine, containing anti-rabbit secondary antibodies (Thermo Scientific diluted 1:2000) conjugated with peroxidase. The detection of peroxidase activity was conducted by chemiluminescence, using the kit Super Signal West Dura provided by Pierce. The acquisition and analysis of the images was performed using by Quantity One program in a Chemi-doc apparatus (Bio-Rad).

The same filter was incubated in a stripping buffer containing 62.5 mM Tris-HCl (pH 6.8), 2% SDS, a final concentration of 0.1 M 2-mercaptoethanol, for 30 min at 50–60 °C according to the methods [41]. Afterward the filter was washed in TBST and analysed by immunoblotting with anti-poly(ADPR) (anti-PAR) polyclonal primary antibodies (H000085-05-B01, Alexis, 1:1000) and horseradish peroxidase-conjugated goat anti-mouse secondary antibody (Pierce, 1:2000) as previously described. Poly(ADPR) antibody is able to identify polymers of ADPR at the last of 5 units.

2.4. Poly (ADP-Ribose) Polymerase (PARP) Assay Activity

The assay to measure the ADP-ribosylating activity was carried out for 10 min at 25 °C. The nuclear fractions (20 µg of proteins) were incubated in presence of 0.4 mM [³²P] NAD⁺ (10,000 cpm/nmole) in a 500 mM Tris-HCl buffer, pH 8.0; 50 mM MgCl₂ and 10 mM DTT (reaction mixture) [50]. The reaction was stopped by adding of ice-cold 20% trichloroacetic acid (*w/v*). To isolate the product of the reaction, the mixture was filtered on Millipore filters (HAWPP0001, 0.45 µm) and subjected to various washes using 7% trichloroacetic acid. The radioactivity of insoluble acid material associated with the filter was measured in a liquid phase scintillator (Bechman LS 1701). The enzyme activity is expressed in enzymatic milliunit; 1 mU catalyses the synthesis of 1 nmol ADP-ribose/min, at the optimum of pH and temperature.

2.5. Determination of Total Soluble and Fat-Soluble Antioxidant Capacity

Stock solutions of ascorbic acid and α-tocopherol were prepared in water and in ethanol respectively, just before use. The concentrations were determined by spectrophotometer considering the absorption coefficients from the literature. All samples were frozen under liquid nitrogen and ground with pestle and mortar to a fine powder. After the addition of 1 mL/g of solvent (water or ethanol), the suspensions were homogenized, transferred to polypropylene tubes, and shaken for 1 h at room temperature in the dark. The suspensions were then centrifuged at 10,000 g for 15 min and the

supernatant collected and kept at 4 °C (first extract). The pellets were resuspended, homogenized in another volume of solvent and centrifuged. Finally, the supernatant was joint with the first extract and kept at 4 °C until determinations. Total soluble and fat-soluble antioxidant capacity of Y, M and S leaves were measured according to Prieto et al. [51]. An aliquot of 0.1 mL of supernatant was combined with 1 mL of reagent solution (0.6 M sulfuric acid, 28 mM sodium phosphate, and 4 mM ammonium molybdate). The soluble extracts were incubated at 95 °C, while the fat-soluble extracts were incubated at 37 °C for 90 min in a water bath under a constant shaking. The absorbance was measured at 695 nm against a blank. The blank solution contained 1 mL of reagent solution and the proper volume of the same solvent used for the sample. Water-soluble and fat-soluble antioxidant capacity was calculated using a standard curve of ascorbic acid (extinction coefficient $3.4 \pm 0.1 \times 10^3 \text{ M}^{-1} \text{ cm}^{-1}$) and α -tocopherol (extinction coefficient $4 \pm 0.1 \times 10^3 \text{ M}^{-1} \text{ cm}^{-1}$) respectively and was expressed as equivalents/g of leaves.

2.6. Antioxidant Enzyme Analysis

Catalase (CAT), peroxidase (POD) and superoxide dismutase (SOD) activities were determined in Y, M and S leaves. All operations were conducted in ice-cold. The leaves (1 g) were resuspended and homogenized in 0.1 M phosphate buffer (pH 7.5) containing 0.5 mM EDTA. After centrifugation at 4 °C and 15,000 rpm for 15 min., the supernatants were used to measure the activity of antioxidant enzymes. CAT activity was determined according to Wong and Whitaker [52] and Chance and Maehly [53]. The reaction mixture consisted of 0.1 M phosphate buffer (pH 7.0), 200 mM of H_2O_2 and enzymatic extract. The activity assay was performed by adding of 50–150 μL enzymatic extract to reaction mixture (150 μL H_2O_2 , 2 mL phosphate buffer, and water up to final volume of 3 mL). The decrease of H_2O_2 was monitored at 240 nm and quantified by its molar extinction coefficient ($36 \text{ M}^{-1} \text{ cm}^{-1}$) [54].

POD (EC 1.11.1.7) activity was determined in according to Yuan and Jiang [55] and Chance and Maehly [53]. The reaction mixture contained 0.1 M of phosphate buffer (pH 7.0), 200 mM H_2O_2 , 135mM of guaiacol. SOD assay was determined by adding 500–1000 μL of enzymatic extract to the reaction mixture (1.2 mL of phosphate buffer, 30 μL of H_2O_2 , 200 μL of guaiacol and water in a total volume of 3 mL). Activity was determined by the increase in absorbance at 420 nm caused to guaiacol oxidation ($E = 26.6 \text{ mM}^{-1} \text{ cm}^{-1}$).

SOD (EC 1.15.1.1) activity was assayed according to Sun and Zigman [56] and Khopde et al. [57]. Enzymatic assay (final volume of 3 mL) was conducted by adding a reaction mixture (0.1 M carbonate buffer pH 10.0, 5mM epinephrine pH 2.0) to enzymatic extract. Activity was determined by absorbance increase at 320 nm for 1 min. One unit of SOD activity is expressed as the amount of enzyme required to cause 50% inhibition of epinephrine oxidation under the experimental conditions.

The absorbance increase was recorded at 320 nm for 1 min. The linear portion of the reaction progress curve (product versus time) at 320 nm was calculated ($\Delta \text{Abs}_{320} \text{ min}^{-1}$). The slope was used to determine the rate of epinephrine oxidation, which was calculated in the same conditions but without enzymatic extract. One unit of SOD was defined as the amount of enzyme required to reduce the epinephrine auto-oxidation rate by 50%.

The glutathione S-transferase (GST) activity has been estimated according to Habig et al. [58]. Young (Y), mature (M) and senescent (S) leaf samples (5 g) were homogenized in liquid nitrogen to obtain fine powder and suspended in cold potassium phosphate buffer 0.1 M pH 6.5, EDTA 1 mM. The solution was centrifuged at 4 °C at 7000 g for 30 min; the supernatants were then separated and used for enzyme assays. 25 μL of each sample was added to phosphate-buffered saline (PBS) buffer, containing 1-chloro-2,4-dinitrobenzene (CDNB as substrate for GST) at final concentrations of 1 mM reduced glutathione, 1 mM CDNB and 100 mM potassium phosphate buffer (pH 6.5). The formation of the conjugate complex was monitored by spectrophotometer for 5 min, 25 °C at 340 nm. The enzyme activity was expressed as $\mu\text{M}/\text{min}/\text{g}$ of fresh tissue.

2.7. Statistical Analysis

The software package Sigma-Plot 12.0 (Jandel Scientific, San Rafael, CA, USA) was used for graphical and statistical data processing. Statistically significant differences were checked by one-way analysis of variance (ANOVA) based on a significance level of $p < 0.05$. The normal distribution of data was verified by Shapiro–Wilk and Kolmogorov–Smirnov tests. Percent data were transformed through the arcsine function before statistical analysis. The data are mean \pm standard deviation (SD) ($n = 5$).

3. Results

3.1. Photochemical Apparatus Efficiency

The response of photochemical apparatus in *C. incanus* leaves of different age was assessed by chlorophyll fluorescence emission measurements analysing the indexes: maximal PSII photochemical efficiency (F_v/F_m) and quantum yield of PSII linear electron transport (Φ_{PSII}). As expected, compared to young and mature, senescent leaves showed a strong decrease ($p < 0.05$) of F_v/F_m (Figure 1a) and Φ_{PSII} (Figure 1b). On the contrary, no difference in photochemistry was found between young and mature leaves.

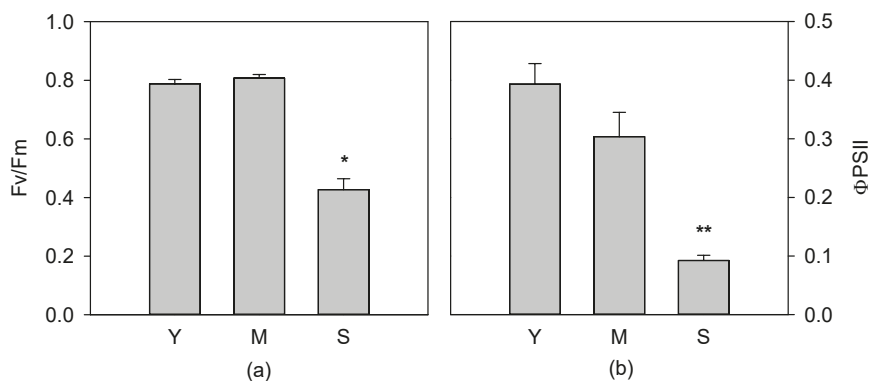


Figure 1. Maximal photochemical efficiency, F_v/F_m (a) and quantum yield of linear electron transport, Φ_{PSII} (b), of young (Y), mature (M) and senescent (S) leaves of *Cistus incanus* L. Data are mean \pm SD ($n = 5$). Results were analysed by one-way analysis of variance (ANOVA) with Holm–Sidak post hoc test. Asterisks represent different levels of significance (** $0.001 < p \leq 0.01$, * $0.01 < p \leq 0.05$) compared to Y leaves.

3.2. Poly(ADP)ribosylation Characterization

No significant difference in PARP activity was found between S and Y leaves; on the contrary, PARP activity was three times lower ($p < 0.01$) in M compared to Y and S leaves (Figure 2).

Moreover, no qualitative and quantitative difference in protein patterns was evidenced (Figure 3a). A single and clear band, corresponding to a protein with molecular weight of about 80 kDa, resulted immunopositive to anti-PARP, which was able to recognize the highly conserved catalytic site of the enzyme (Figure 3b). Western blotting with anti-PAR antibodies used to identify ADPR polymers, consisting of at least five units, showed only one immunopositive signal of about 90 kDa, corresponding to a covalent ADPR protein acceptor (Figure 3c).

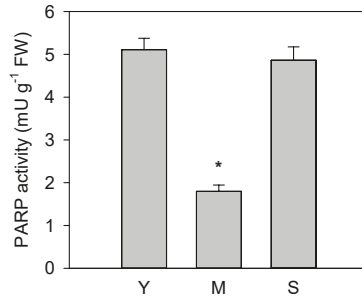


Figure 2. Poly (ADP-ribose) polymerase (PARP) activity measured in young (Y), mature (M) and senescent (S) leaves of *Cistus incanus* L. Data are mean \pm SD ($n = 5$). Results were analysed by one-way ANOVA with Holm-Sidak post hoc test. Asterisks represent different levels of significance (* $0.01 < p \leq 0.05$) compared to Y leaves.

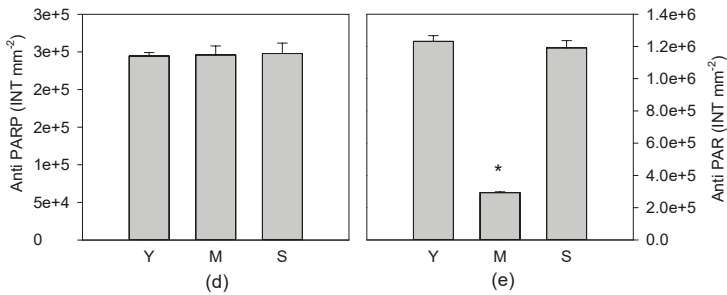
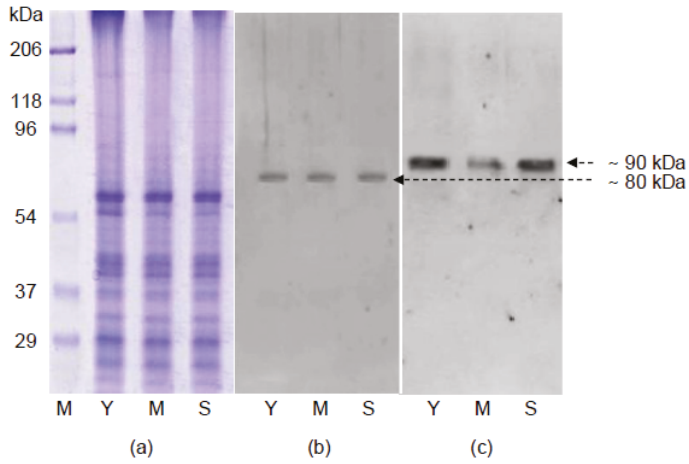


Figure 3. Sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS-PAGE) 10% (a), immunoblotting with anti-PARP (b) and immunoblotting with anti-PAR (c), in nuclear fractions of young (Y), mature (M) and senescent (S) leaves of *Cistus incanus* L.; densitometric analysis of immunopositive bands to anti-PARP (d) and to anti-PAR (e). For densitometric analysis data are mean \pm SD ($n = 3$). Results were analysed by one-way ANOVA with Holm-Sidak post hoc test. Asterisks represent different levels of significance (* $0.01 < p \leq 0.05$) compared to Y leaves.

Densitometric analysis of bands immunopositive to anti-PARP (Figure 3d) and anti-PAR (Figure 3e) was conducted to correlate the intensity of 80kDa proteins, corresponding to their expression, with poly(ADPR) synthesis. Data showed that the intensity of 80kD protein did not change in examined samples, while the intensity of 90kD protein observed in Y and S leaves was significantly higher than that measured in M leaves (Figure 3e). These results suggest that, although the PARP expression was not affected by leaf age, on the contrary the levels of poly(ADPR) produced in the Y and S leaves were significantly higher than those found in M leaves.

3.3. Total Soluble and Fat Soluble Antioxidant Capacity

Total soluble and fat-soluble antioxidant capacity was evaluated to assess if and how the antioxidant activity may be affected by leaf age. The results showed that in Y and M leaves, soluble (Figure 4a) and fat-soluble (Figure 4b) antioxidant capacities did not show any differences, while in S leaves a significant decrease ($p < 0.01$) of both antioxidant capacities occurred.

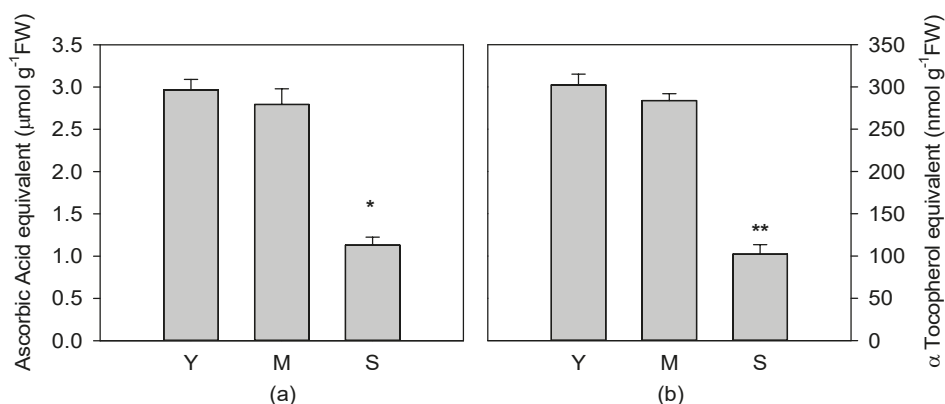


Figure 4. Total soluble antioxidant capacity expressed as ascorbic acid equivalents (a) and fat-soluble antioxidant capacity expressed as α tocopherol equivalents (b), of young (Y), mature (M) and senescent (S) leaves of *Cistus incanus* L. Data are mean \pm SD ($n = 5$). Results were analysed by one-way ANOVA with Holm–Sidak post hoc test. Asterisks represent different levels of significance (** $0.001 < p \leq 0.01$, * $0.01 < p \leq 0.05$) compared to Y leaves.

3.4. Activity of Antioxidant Enzymes

The activities of catalase (CAT), superoxide dismutase (SOD), peroxidase (POD) and glutathione S-transferase (GST) were analysed to examine particular changes in enzymatic activity related to plant aging. The highest ($p < 0.01$) activity for CAT and SOD enzyme was found in S leaves. A significant difference in CAT and SOD activities was also measured in M compared to Y leaves, with higher values ($p < 0.05$) for Y leaves (Figure 5a,b). No significant variation of POD activity was found between Y and M leaves (Figure 5c). GST activity was statistically higher ($p < 0.01$) in young, compared to mature and senescent leaves. GST activity further decreased ($p < 0.001$) in senescent leaves, reaching the lowest value (Figure 5d).

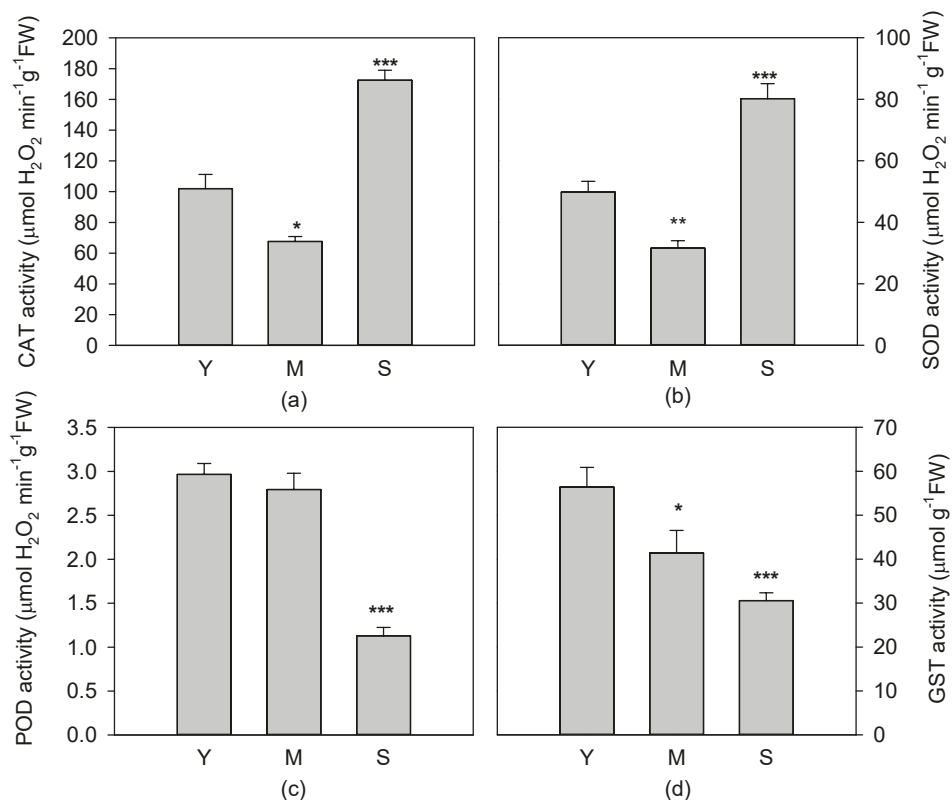


Figure 5. Activity of antioxidant enzymes: catalase, CAT (a), superoxide dismutase, SOD (b), peroxidase, POD (c), and glutathione S-transferase, GST (d) in young (Y), mature (M) and senescent (S) leaves of *Cistus incanus* L. Data are mean \pm SD ($n = 5$). Results were analysed by one-way ANOVA with Holm–Sidak post hoc test. Asterisks represent different levels of significance (** $0 < p \leq 0.001$, ** $0.001 < p \leq 0.01$, * $0.01 < p \leq 0.05$) compared to Y leaves.

4. Discussion

The ageing process has been associated with an increase of reactive oxygen species (ROS), responsible for cumulative damage to biological macromolecules, including lipids, proteins, and nucleic acids [1–4]. In the chloroplast, an impairment of photosynthetic membrane functionality leads to the decrease of photosynthetic activity. In fact, with increasing age, the PSII photochemical activity declines indicating a limited capacity of photosynthetic apparatus in the light-harvesting and conversion at reaction centres. Contextually, the increase of PARP activity at the nuclear level may be considered a rapid response of cells to DNA injuries. It has been demonstrated that poly(ADP-ribose)ylation is one of the most immediate cellular response to genotoxic insults prompted by ionising radiation, alkylating agents, and heavy metals [4,59,60].

Numerous studies suggest a link between poly(ADP-ribose)ylation induced by the DNA damage and ageing/longevity [60,61]; an evident correlation between PARP-1 deficiency and accelerated aging in mammalian cells was also demonstrated [62].

In the present work, we demonstrate for the first time that the PARP activity and consequent poly(ADPR) production are strongly influenced by the leaf aging in plants. In plants, three different nuclear PARPs have been previously characterized [38–40], and in particular in *Cistus incanus* species a PARP of 80 kDa has already identified [41].

The young, mature and senescent leaves of *C. incanus* express only the protein of 80 kDa, recognized by anti-PARP antibodies, specific in the identification of the highly conserved catalytic domain of all three nuclear PARPs. The *Cistus* species used in our experiment express only one of three PARPs documented in plants. We observe a single acceptor of poly(ADPR), corresponding to the same PARP modified with a polymer of about 20 units of ADPR, as indicated by the shift of the molecular weight from 80 kDa to 90 kDa. Therefore, at different leaf growth stages, we do not observe any difference in the length of the synthesized polymer, but only in the amount of poly(ADPR). In particular, the highest levels of poly(ADPR) are found in the Y and S leaves, according to the PARP activity.

In senescent (S) leaves, the increase of PARP activity as well as of CAT and SOD enzymes indicates that all these scavenger activities co-work against the oxidative stress occurring in cells during ageing. The reduction of the total antioxidant capacity together with the decrease of POD and GST activity suggests the inability of cells to counteract definitely ROS production [63]. However, it is likely that in our study, the oxidative damage to DNA does not lead to cell death because PARP enzymes are hyperactivated guaranteeing a continuous repair of the genomic material. It has been demonstrated in plants that, during senescence, cells are subjected to a degradation of cellular structures, and only the nucleolus and mitochondria remain intact until final stages of leaf senescence [64]; therefore, it is reasonable to suppose that mitochondria could be still functioning ensuring an adequate intracellular NAD⁺ supply for PARP activity.

At the functional level, ageing determines a reduction of photosynthetic capacity in senescent leaves (F_v/F_m ratio and Φ_{PSII} decreasing), suggesting a loss of functionality of PSII reaction centres. Although it is likely that the decline of photochemistry may be due to a direct effect of ROS on photosynthetic membranes [5], it cannot be excluded that in senescent leaves the photosynthetic electron transport is diverted in the process other than carbon fixation to provide the reductive power for the scavenger enzymes.

By contrast with S leaves, the high PARP activity found in young (Y) leaves, responsible for the high poly(ADPR) production, seems to be not due to the oxidative DNA damage but rather to other cellular events such as gene expression, transcriptional activity, cell growth, and tissue proliferation [65]. The active growth phase and the high photosynthetic efficiency would guarantee in Y leaves an adequate production of ATP for the synthesis of NAD⁺ for PARP [41]. Moreover, the high total antioxidant capacity, together with the elevated POD and GST activities might assure the apoplastic ROS homeostasis, that control the cell expansion [66]. It is well known that, during the leaf development, a wave of ROS-dependent cell growth sweeps through the leaves [67,68]. Our data suggest that with the progression of ageing, leaves have to carry out a fine-tuning of ROS production and removal adapting to continuous changes of cell metabolism and modulating the antioxidant defences differently [69,70].

5. Conclusions

Our work provides the novel insight that the PARP activity and the consequent poly(ADPR) production are strongly influenced by the leaf aging in plants. Leaf aging is associated to oxidative stress due to the ROS increase, responsible for DNA damage and consequent PARP activation. In the senescent leaves, the hyperactivation of PARP activity guarantees a continuous repair of the genomic material avoiding cell death. Overall, this study contributes to a further understanding of physiological, molecular, and biochemical changes occurring in cells during leaf aging and might act as “trailblazer study” for additional research, such as the study of the impact of environmental stressors (e.g., water shortage) on leaf aging.

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Review

Regulation of Ascorbate-Glutathione Pathway in Mitigating Oxidative Damage in Plants under Abiotic Stress

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Abstract: Reactive oxygen species (ROS) generation is a usual phenomenon in a plant both under a normal and stressed condition. However, under unfavorable or adverse conditions, ROS production exceeds the capacity of the antioxidant defense system. Both non-enzymatic and enzymatic components of the antioxidant defense system either detoxify or scavenge ROS and mitigate their deleterious effects. The Ascorbate-Glutathione (AsA-GSH) pathway, also known as Asada–Halliwell pathway comprises of AsA, GSH, and four enzymes viz. ascorbate peroxidase, monodehydroascorbate reductase, dehydroascorbate reductase, and glutathione reductase, play a vital role in detoxifying ROS. Apart from ROS detoxification, they also interact with other defense systems in plants and protect the plants from various abiotic stress-induced damages. Several plant studies revealed that the upregulation or overexpression of AsA-GSH pathway enzymes and the enhancement of the AsA and GSH levels conferred plants better tolerance to abiotic stresses by reducing the ROS. In this review, we summarize the recent progress of the research on AsA-GSH pathway in terms of oxidative stress tolerance in plants. We also focus on the defense mechanisms as well as molecular interactions.

Keywords: antioxidant defense; free radicals; glyoxalase system; hydrogen peroxide; plant abiotic stress; reactive oxygen species; redox biology; stress signaling

1. Introduction

With the advancement of lifestyle the natural resources are being exploited and the interruption of natural environment is increasing the extremity of various kinds of abiotic stress, including salt stress, drought stress, waterlogging, temperature extremes, including high and low, excess and low light intensity, radiation stress, ozone, metal and metalloid toxicity, and other organic or inorganic pollutants. Environmental extremity narrows ways to increase plant productivity. Ever-increasing population demands newly cultivable areas, including the adverse land areas, even the higher crop production in per unit area.

Any abiotic stress impaired stomatal function, photosystem activity, Calvin cycle, or photosynthetic enzyme activities, as well as altered electron transport chain reactions. Moreover, unfavorable peroxisomal or cytosolic atmosphere led to overwhelm of electron absorption and generate ROS as a common outcome and subsequently causes oxidative damage [1,2]. If the challenges of plant scientists are increasing productivity against the abiotic stresses, their concentrations are moving to the depth for breaking the obstacles at the cellular or organelles levels, where abiotic stresses impose common types of barrier to hinder their function. Reactive oxygen species is an inescapable outcome of aerobic reactions, which are partly reduced or activated by the appearance of oxygen. Reactive oxygen species is a combined name that indicates different, highly active components. Superoxide (O_2^-), hydroxyl (OH^\bullet), and peroxy (ROO^\bullet) are some examples of oxygen radicals. Hydrogen peroxide (H_2O_2), singlet oxygen (1O_2), and ozone (O_3) are the non-radical types of ROS [3]. Reactive oxygen species are important for plants. They have dual role in plants: a small amount of those acts as a signal for inducing abiotic stress responses towards adaptation process, while the excess generation of those causes oxidative damage. However, in severe cases, oxidative damages to membranes (lipid peroxidation), proteins, nucleic acid, including RNA and DNA, and even directs to the oxidative obliteration of the cell [4]. Chloroplast, mitochondrion, membranes of the cell or its ultrastructural organelles, apoplast, and nucleolus are the locations of ROS production. Nonetheless, peroxisome is also considered as a powerful source of ROS since the electron transport chain (ETC) and photochemical reactions are the majority of the processes generating ROS [5–7].

Plants have an antioxidant defense system having non-enzymatic and enzymatic antioxidants in cellular organelles, which scavenges different ROS up to a certain level. If the ROS generation is higher than the scavenging ability of the antioxidant system, then oxidative damage occurs. Antioxidant defense system comprises ascorbate (AsA), glutathione (GSH), carotenoids, tocopherols, flavonoids, etc., which are some commonly known non-enzymatic antioxidants [5]. Ascorbate peroxidase (APX), monodehydroascorbate reductase (MDHAR), dehydroascorbate reductase (DHAR), glutathione reductase (GR), superoxide dismutase (SOD), catalase (CAT), glutathione peroxidase (GPX), glutathione S-transferase (GST), and peroxiredoxin (PRX) are well known enzymatic antioxidant components [8,9]. Among all of these, AsA, GSH, APX, MDHAR, DHAR, and GR comprise the AsA-GSH cycle.

Ascorbate is one of the most powerful substrates for scavenging H_2O_2 . Ascorbate maintains the reduced state of α -tocopherol. Ascorbate is supposed to be concerned in zeaxanthin biosynthesis dissipating excess light energy in the thylakoid membranes of chloroplast and prevents oxidative stress. Ascorbate sustains reduce the state of prosthetic metal ions and maintain the activity of antioxidant enzymes [6]. Glutathione regulates various metabolic functions; it acts as an antioxidant. Glutathione peroxidase and GST utilize GSH as substrate; GPX is responsible for ROS detoxification, whereas GST is liable for xenobiotic detoxification [1]. The glyoxalase system consisting of glyoxalase I (Gly I) and glyoxalase II (Gly II) enzymes detoxifies cytotoxic and oxidative stress creator methylglyoxal (MG), where Gly I uses GSH and after finishing MG detoxification, GSH is recycled [2]. The positive role of AsA-GSH cycle components has been documented in many plants that are affected by abiotic stresses [1,2]. Participation of the GSH/glutathione disulfide (GSSG, the oxidized form of GSH) redox in maintaining a favorable cellular environment and in stress signal and adaptation were discussed in some previous findings. Glutathione participates in signal transduction, the proper pathway, of which remains unrevealed. The presence of AsA and GSH has been reported to improve osmoregulation, plant water status and nutrient status, water use efficiency, photosynthetic performance, and the overall productivity of plants. Exogenous AsA and GSH applications have been reported to enhance the antioxidant defense as well as the overall tolerance of plants against abiotic stresses. Accordingly, the enzymatic antioxidants of AsA-GSH cycle participates in scavenging ROS, whereas AsA and GSH not only directly scavenge a range of ROS but also perform many other functions to maintain a favorable state in cytosol and other cellular organelles to enhance antioxidant capacity and to reduce oxidative stress, which is induced by different abiotic stresses; AsA and GSH also improve the physiological performance of plants. Since the discovery of the AsA-GSH cycle, its most discussed

topics are related to antioxidative protection. However, in this aspect, various other factors should be revealed like physiological factors/processes involved in generating oxidative stress, role of AsA-GSH cycle components in regulating those physiological processes and ultimately the oxidative stress. Considering the multiple vital roles of AsA-GSH cycle in mitigating oxidative stress, this review accommodates presently available and updates of research findings and perspectives.

2. Ascorbate-Glutathione Pathway—An Overview

Ascorbate-Glutathione pathway (also called as Asada–Halliwell pathway) is the major pathway of antioxidant defense, which mainly detoxify the H_2O_2 in a plant cell. Apart from AsA and GSH, its enzymes—APX, MDHAR, DHAR, and GR [6]—have significant roles. Both AsA and GSH are found in the cytosol, nucleus, chloroplast, mitochondria, and peroxisome, where they operate the functions assisted by four enzymes and, therefore, each enzyme has several isoforms that are based on the cellular localization [10]. Both AsA and GSH are present in cellular organelles in a millimolar range, for instance, in *Arabidopsis thaliana*, AsA concentration is the highest (22.8 mM) in the peroxisome, where GSH is highest (14.9 mM) in mitochondria [11,12]. AsA and GSH both have high redox potentials and, therefore, interact with many components and pathways towards the maintenance of a generally reduced state. There are few steps, by which AsA and GSH work coordinately to detoxify H_2O_2 , and at the same time, both AsA and GSH are regenerated. First, the enzyme APX converts H_2O_2 into the water with the help of AsA as an electron donor, which is also converted into monodehydroascorbate (MDHA). This MDHA again regenerates AsA by the activity of MDHAR and a part of this is spontaneously converted into dehydroascorbate (DHA). Later, DHA is reduced to AsA again by using GSH, which results in its oxidation to produce GSSG. Finally, this GSSG regenerates GSH by the activity of GR using NADPH as the electron donor (Figure 1). Both AsA and GSH are strong antioxidants, but the maintenance of their redox state is important in conferring stress tolerance in plants, which largely depends on the activities of the four enzymes that are associated with the AsA-GSH cycle [6,13]. In the next sections, we have described all of the components of the AsA-GSH pathway.

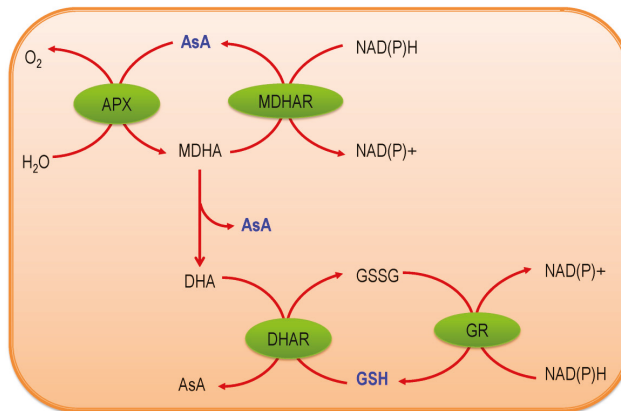


Figure 1. Ascorbate-Glutathione (AsA-GSH) (Ascorbate-Glutathione) pathway [ascorbate, AsA; ascorbate peroxidase, APX; monodehydroascorbate, MDHA; monodehydroascorbate reductase, MDHAR; dehydroascorbate, DHA; dehydroascorbate reductase, DHAR; glutathione, GSH; oxidized glutathione, GSSG; glutathione reductase, GR; Nicotinamide adenine dinucleotide phosphate (reduced form), NAD(P)H; Nicotinamide adenine dinucleotide phosphate (oxidized form), NAD(P)⁺].

3. Components of AsA-GSH Pathway

3.1. Ascorbate

All living organisms either make AsA (also known as Vitamin C) or get it in their foodstuffs. Naturally abundant L-AsA is of a simplest chemical structure and is related to C6 sugars. It is a hexonic acid aldo-1,4-lactone (either L-galacturonic or L-gulonic acid), having an enediol group at C2 and C3 [14]. The enediol group enables L-AsA for donating one or two electrons to form an initial oxidized intermediate (MDHA) and further to an oxidized (DHA) form. The C5 and C6-OH group serves to provide alcoholic nature. They can react with produced acetals, ketals by reacting with aldehydes and ketones, respectively. Having two asymmetric C, L-AsA illustrates a positive optical rotation, which is unaltered by the acidic pH of solution but greatly affected by alkaline pH, which increases over +160° in 2N NaOH solution [14].

In solid-state, L-AsA is stable but oxidizes readily in solution, in particular in the presence of Cu, Fe, or alkali to form DHA. Afterward, two MDHA can undergo spontaneous reaction to rejuvenate one molecule of L-AsA and one molecule of DHA [15].

Ascorbate biosynthesis system is one of the ancient pathways and formed in very primitive life process on this planet. In plant tissue, AsA can be synthesized from several biochemical pathways. D-glucose is the primary substrate for producing AsA, and in this pathway, a set of ten reactions occurred (Figure 2). Ascorbate can be formed via four pathways viz. L-galactose, L-gulose, D-galacturonic acid, and myo-inositol pathway [16–18]. The biosynthesis of AsA is lineal with the cell wall formation. After the initial reactions, the D-galacturonic acid and L-galactose pathways both yielded L-galactono-1,4-lactone. Besides, in L-gulose and myo-inositol pathway, L-gulonic acid is produced, which is further hydrolyzed to form L-gulonono-1,4-lactone, are catalyzed for the synthesizing of AsA in the mitochondria (Figure 2) [19].

In an organism, AsA metabolism comprises of biosynthesis (catabolism) and degradation (anabolism), and the balance between catabolism and anabolism determines the intracellular concentration of AsA. In the previous paragraph, we briefly discussed the biosynthesis of AsA. Hence, we will discuss the degradation and turnover of AsA in this paragraph. In some plants, the AsA turnover rate is relatively very high [20]. On the other hand, AsA is neither stable nor is restricted to oxidoreduction, which changes the equilibrium of AsA and DHA in plant tissue. The AsA pool undergoes turnover in plants. As AsA has prominent responsibility in the redox function metabolism, therefore the recycling of AsA from MDHA (catalyzed by MDHAR using NADPH) and DHA (DHAR using GSH) is the necessity to keep the redox balance as well as the higher total AsA pool [21]; and the functioning of the water-water cycle to optimize photosynthesis [22]. If the oxidized forms are not recovered, they will undergo further degradation to form oxalic or L-tartaric acids (Figure 2) [23]. In a plant cell, AsA act as a multifunctional biosynthetic precursor. While using radioactive ¹⁴C AsA, some studies tried to understand the degradation of AsA and DHA, but the mechanism is still not fully understood (Figure 2) [24]. However, it is well understood that the cleavage between C2 and C3 results in oxalate formation, whereas the cleavage between C3 to C6 produces L-threonate, via L-idonate [25,26]. Furthermore, DHA can be hydrolyzed into 2,3-L-diketogulonate, being further oxidized to unknown intermediate (Figure 2) and catalyzed by ascorbate oxidase (AO). Sometimes, this intermediate produces toxic H₂O₂ non-enzymatically and it may inhibit peroxidase [24,27].

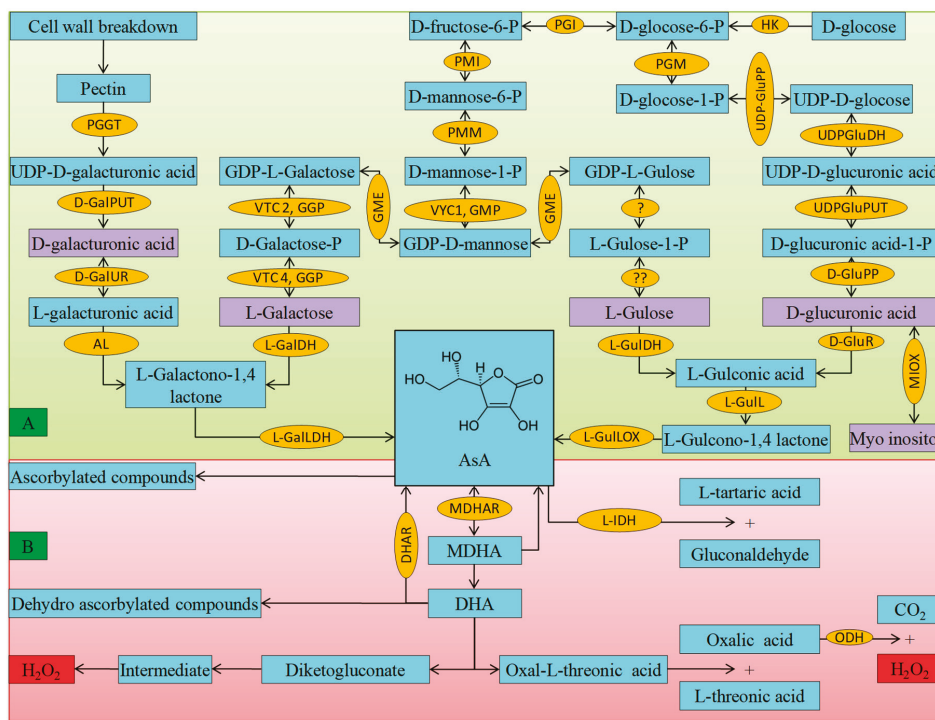


Figure 2. Ascorbate biosynthesis and metabolism is a complex set of reactions, some involving unidentified enzymes; some of the products are reactive and potentially damaging carbonyl compounds, (A) biosynthetic pathway; and, (B) regeneration and degradation pathways in plants. The metabolites in the violate box represent the name of each biosynthetic pathway. The elaborated name of enzymes are as follows (HK: Hexokinase; PGI: glucose-6-phosphate isomerase; PMI: mannose-6-phosphate isomeras; PMM: phosphomannomutase; TC1 or GMP: GDP-D-mannose pyrophosphorylase/mannose-1-phosphate guanylyltransferase; VTC2 or GGP: GDP-D-mannose 3',5'-epimerase, GME: GDP-L-galactose phosphorylase;VTC4 or GPP:L-galactose-1-phosphate phosphatase; GalDH: L-galactose dehydrogenase; L-GalLDH: L-galactono-1,4-lactone dehydrogenase; ?: nucleotide pyrophosphatase or sugar-1-phosphate guanyltransferase; ??: sugar phosphatase; L-GulDH: L-gulose dehydrogenase; L-GulL: L-gulonolactonase; L-GulLOX: L-gulonono-1,4-lactone oxidase; PPGT: polygalacturonate 4-alpha-galacturonosyltransferase; D-GalPUT: D-galacturonate-1-phosphate uridylyltransferase; D-GalUR: D-galacturonate reductase; AL: aldonoalactonase; PGM: phosphoglucomutase; UDPGluPP: UDP-glucose-pyrophosphorylase; UDP-GluDH: UDP-glucose dehydrogenase; UDP-GluPUT: glucuronate-1-phosphate uridylyltransferase; D-GluPP: D-glucurono-1-phosphate phosphatase; MIOX: myo-inositol oxygenase;D-GluR: D-glucuronate reductase; MDHAR: monodehydroascorbate reductase; DHAR: dehydroascorbate reductase;L-IDH: L-Idodonate dehydrogenase).

3.2. Glutathione

Glutathione is an omnipresent low molecular weight tripeptide (γ -L-glutamyl-L-cysteinylglycine; γ -Glu-Cys-Gly), which is a strong antioxidant and an essential metabolite with a multifarious role in plants [28,29]. It was first discovered from yeast cells subsequently in many plants and animal tissue. Later on, in 1936, it was found as the reducing agent present in the plant tissue [30].

Although GSH is composed of glutamine (Glu), cysteine (Cys), and glycine (Gly), three essential amino acids; but some plant may contain homologues of GSH, where Gly is replaced by other amino

acids [31]. In plants, reduced GSH accounts for >98% of total GSH [10]. Generally, cells possess three major reservoirs of GSH cytosol (80–85%), mitochondria (10–15%), and endoplasmic reticulum [32]. The thiol group is specific to γ -glutamyltranspeptidase (GGT) and it allows GSH a higher degree of stability [32,33]. Nevertheless, GSH plays a vital role, including antioxidant defense, xenobiotics detoxification, cell cycle regulation, and apoptosis, reserving cysteine, maintaining redox balance as well as immunity modulation and fibrogenesis [10,29].

In plants, GSH biosynthesis involves two enzymatic steps, which require ATP and the constituent amino acids (Figure 3). In the earliest stratum, γ -glutamylcysteine (γ -EC) is produced by γ -glutamylcysteine synthetase (γ -ECS, EC 6.3.2.2) catalysis and participation from Glu and Cys. In the next stratum, GSH is synthesized from γ -EC and Gly via bonding from the Cys residue of γ -EC with α -amino group of Gly, catalyzed by GSH synthetase (GSH-S, EC 6.3.2.3, also known as GSH synthase). After synthesis in the cytoplasm, GSH is transported to other cellular organelles [34].

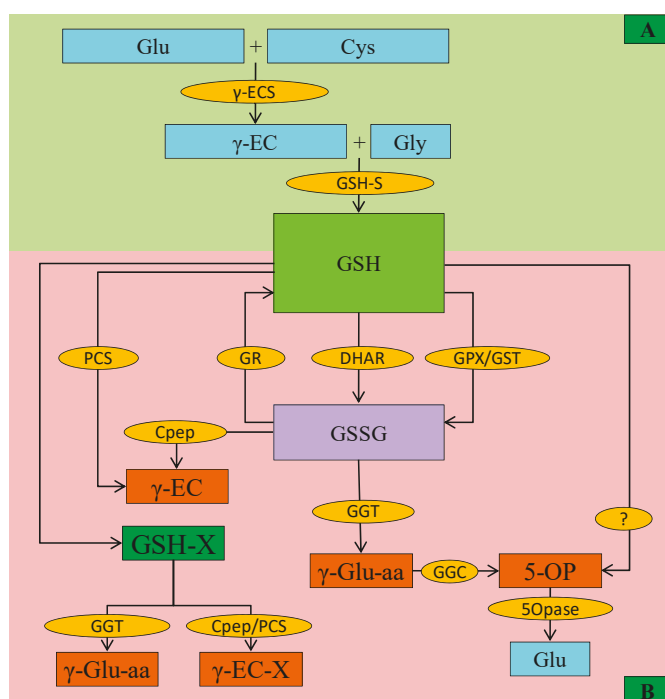


Figure 3. Glutathione biosynthesis, metabolism, and degradation in plants. (A) Biosynthesis the first step occurred in plastid: Glu and Cys form γ -glutamylcysteine (γ -EC) catalyzed by γ -EC synthetase (γ -ECS). The second step occurred in the cytosol or in plastid: γ -EC and Gly bond together to form GSH catalyzed by GSH-S (glutathione synthase). Further, GSH participates in ROS scavenging and is converted into GSH/glutathione disulfide (GSSG) by the enzyme glutathione peroxidase (GPX), glutathione S-transferase (GST), and DHAR. Further GSSG can be recycled to GSH by the activity of glutathione reductase (GR). (B) In the degradation pathway, GSH and S-conjugated compound (GS-X) can be degraded to γ -EC and γ -EC-X by phytochelatin synthase (PCS). While, carboxypeptidase (Cpep) and γ -glutamyl transpeptidase (GGT) both could degrade GS-X to form γ -Glu-aa (aa, amino acid) and γ -EC-X, respectively. Similarly, GSSG is degraded by GGT and Cpep to form γ -Glu-aa and γ -EC, respectively. Further, the produced γ -Glu-aa is converted to 5-oxoproline (5-OP) by γ -glutamyl cyclotransferase (GGC). Besides, GSH is also converted to 5-OP. Although it is thought that this reaction is catalyzed by GGC, still it is unclear. 5-OP is converted to Glu in the next step by the action of 5-oxoprolinase (OPase).

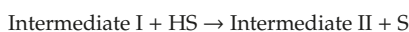
Glutathione is very important for various physiological processes, especially during abiotic stress; it coordinates with AsA turnover and is oxidized to GSSG [35]. Moreover, some other thiol-dependent enzymes, GPX and GST use GSH as co-factor, hence converted to GSSG, which is further reduced back to GSH, with GR catalysis. In higher plants, there are two genes that are reported to encode GRs (*GR1* and *GR2*), where *GR2* is essential for plant development [31,34].

The degradation of GSH is another important phenomenon of GSH metabolism (Figure 3). Up to now, as many as four types of GSH degrading enzymes have been described, which either use GSH or act on GSSG or other GSH-conjugates. Among them, carboxypeptidase activity could degrade GSH itself or GSH-conjugates. Cytosolic PCS is another enzyme that is responsible for the breakdown of GSH-conjugates that are mainly activated during metal/metalloid(s) stress. Another enzyme γ -glutamyl transpeptidase (GGT) acts in GSH transpeptidation or hydrolysis, which is further converted to free Glu by the action of GGC (γ -glutamyl cyclotransferases) and 5-oxoprolinase (5-OPase) (Figure 3) [31,36]. Moreover, another vacuolar GGTs have been reported in *Arabidopsis*, which breaks GSH-conjugates [37]; hence, along with PCS, GGTs is important for metabolizing GSH-conjugates that is formed during secondary metabolites synthesis [37,38].

3.3. Ascorbate Peroxidase

The class I heme-peroxidases; APX (EC 1.11.1.11) occurred in several isoforms in plant cell, viz. cytosolic APX isoform (cAPX), mitochondrial APX isoform (mitAPX), peroxisomal and glyoxysomal APX isoform (mAPX), and chloroplastic APX isoform (chAPX) differed in their substrate specificity, molecular weight, optimal pH ranges between 7 and 8 for maximum activity and stability [39–41]. More importantly, isoforms activity is not stable when AsA is absent. For example, AsA concentrations those are lower than 20 μ M greatly reduced chAPX activity. All of the APX isoforms are heme peroxidase they are inhibited by cyanide and azide. Iron plays a vital role in the APX catalytic site; hence, despite the presence of high AsA concentration, Fe deficiency reduced the activity of cAPX [42]. If the single Cys32 residue near arginine (Arg) 172 residue is altered, APX loses the ability to oxidize AsA to DHA, but it can oxidize other small aromatic molecules. Therefore, the APX properties differ with the guaiacol peroxidases, but they are 33% identical with cytochrome c peroxidase (CCP) [43].

During ROS (H_2O_2) detoxification, APX binds H_2O_2 producing intermediate (I), and heme iron [Fe(V)] is oxidized forming oxyferryl species ($Fe^{4+} = O$). Afterward, APX is regenerated from I, in a two-step reaction with AsA, where the AsA donate an electron and become oxidized. Detailed reactions are shown below (HS = Substrate, S = One electron oxidized form of the substrate).



In plant cell, APX scavenges H_2O_2 , which mainly participates in AsA-GSH cycle catalyzes the reactions produces MDHA (Figure 4) and, subsequently, MDHA yields DHA [6,10,44].

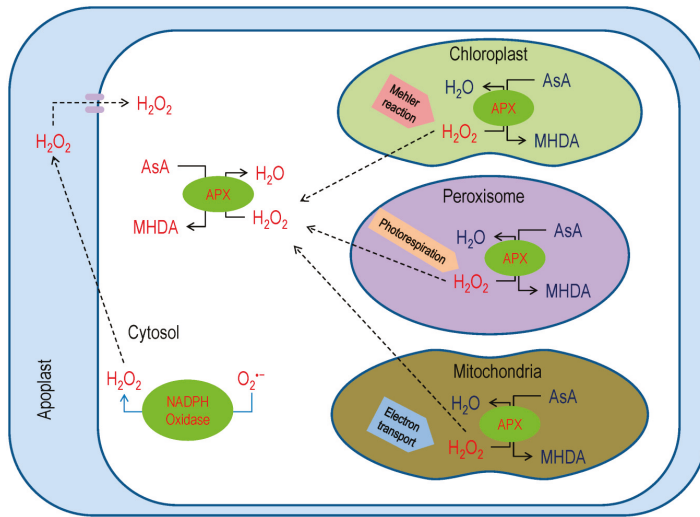


Figure 4. The function of Ascorbate peroxidase (APX) for the abolition of excess reactive oxygen species (ROS) generation in various cellular compartments. Additional details are in the text.

3.4. Monodehydroascorbate Reductase

The MDHAR (EC 1.6.5.4) helps in the revival of AsA [35], having several isoforms that are found in different organelles. Reports suggested that there are three MDHAR genes in tomato, five genes and six isoforms in *Arabidopsis* and rice, and as many as nineteen genes in wheat. In the plant cell, MDHAR activity was detected in different cell organelles, for instance, cytosol, mitochondria, chloroplasts, peroxisomes, and glyoxysomes (Figure 5) [45].

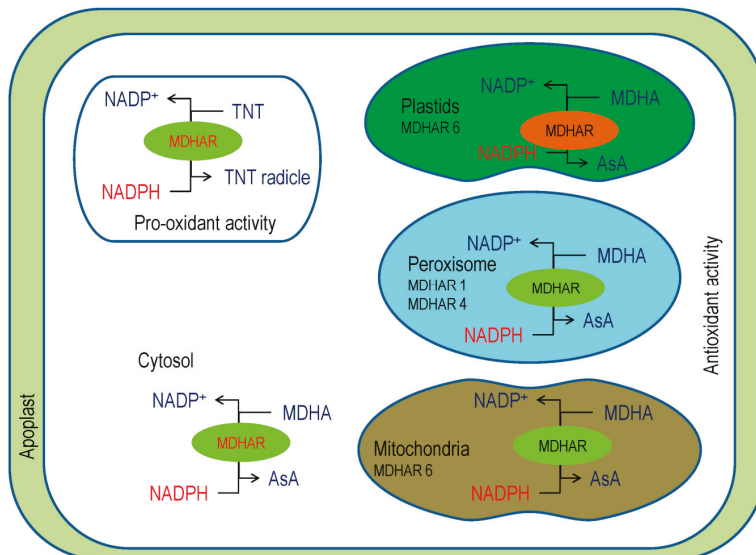


Figure 5. The antioxidant of MDHAR in regenerating AsA to support the removal of reactive oxygen species (ROS) (lower left) contrasts the pro-oxidant role of MDHAR creating 2,4,6-trinitrotoluene (TNT) toxicity.

Although the enzyme is purified from several sources, the detailed structure of this enzyme was published in recent past. Begara-morales et al. [46] coined three-dimensional structure after conducting silico analysis of pea peroxisomal MDHAR. More recently, Park et al. [47] described details MDHAR composition from japonica rice. Those indicated that the structure of MDHAR consists of flavin adenine dinucleotide (FAD) and pyridine binding domain. The structure resembles Fe-S protein reductase [47]. The elucidated structure also indicates that rice MDHAR contains a typical α/β fold, and Arg320 and tyrosine (Tyr) 349 residues are vital for its activity. On the fad-binding domain, the fad is bounded by hydrogen and van der waals bond, where, Gly13, 15 and 297, alanine (Ala) 122 and 319, and threonine (Thr) 123 are involved and highly conserved in the bottom of the crevice. Moreover, Lys53 and Glu178 bridged together, which further bonded with proline (Pro) 49, where both Glu178 and Pro49, are highly conserved, as well as lysine (Lys) 53, Glu178, and FAD bonded each other. Among others, Arg48 compensate FAD phosphate group's negative charge.

In rice, MDHAR α -helices surround β -sheets in the nicotinamide adenine dineucleotide (NAD)-binding domain, where the Tyr174, histidine (His) 315, and phenylalanine (Phe) 348 residues are shifted. Sandwiching between the FAD isoalloxazine ring and Tyr174, as well as steric hindrance moved Tyr174 away. The Phe348 residue shifts outward while His315 comes towards NAD binding site. In addition, the hydrogen bond is formed between Glu178 and nicotinamide ring; Arg202 and ribose and phosphate group, Glu314 and ribose; Glu196 and adenosine ring. Moreover, Glu196 offers rice MDHAR selectivity to NAD; preferring NADH over NADPH [47].

Interestingly, MDHAR can bind substrates other than MDHA, such as isoascorbic acid Evidence showed that phenoxyl radicals, like ferulic acid, quercetin, chlorogenic acid, and coniferyl alcohol, might be reduced by MDHAR [48]. In *Arabidopsis*, MDHAR activity reduces 2,4,6-trinitrotoluene (TNT) and creates its toxicity (Figure 5), but the MDHAR6 mutants are more tolerant, as TNT could not reduce and thus autooxidizes to creates O_2^- [49].

Reports imply that MDHAR response to abiotic stress conditions by reducing MDHA that produces by the excess ROS scavenging, which was observed in many test species (Figure 5) [50,51]. The chlMDHAR is involved in photosynthetic activity during lack of peroxiredoxin [52]. In addition, chlMDAHR activity increased by three- to six-fold during pepper fruit ripening [53].

3.5. Dehydroascorbate Reductase

A major enzyme for GSH assisted DHA recycling is DHAR (EC 1.8.5.1), which is also known by GSH:DHA oxidoreductase or GSH dehydrogenase (AsA) [51,54]. This regeneration process is accomplished at alkaline pH and it is a well known biochemical reaction in plants.

The plant GSH-dependent DHAR is a monomeric enzyme, which is a member of the GSHS-transferase superfamily [55]. *Arabidopsis* possesses three functional DHAR encoding genes, DHAR1 (*At1g19570*), DHAR2 (*At1g75270*), and DHAR3 (*At5g16710*). In recent decades, the attention of researchers towards the DHAR activity in plants for regenerating DHA increased, and a number of investigations were carried out to elucidate the structure and molecular mechanism of DHAR.

The overall three-dimensional structure of DHAR from different plant origin is almost identical, except with some additional short-chain before the α 1-helix. The enzyme has several binding sites. The G site is responsible for binding the GSH in the enzyme. The GSH cysteinyl sulfur bonded Cys20 and occupied disulfide bond. The GSH γ -glutamyl is stabilized, via H-bonds H_2O molecule, and then forms the backbone with serine (Ser) 73 and aspartic acid (Asp) 72. The Phe22 is engaged with the γ -glutamyl group by the van der Waals bond, in addition to hydrogen bonds with Lys59. The Val60 stabilizes the central cysteinyl region. The glyciny group of GSH is loosely bound, forming a salt bridge with Lys47 [54].

The substrate-binding site or DHA binding site or H-site of DHAR enzyme typically exhibits more structural plasticity, but not simultaneously. From the structure of *Pennisetum glaucum* DHAR gene (*PgDHAR1*), it was observed that Lys8 and Asp19 are responsible for DHA binding [56].

The DHAR catalyzing accomplished by the following three reactions (Figure 6):

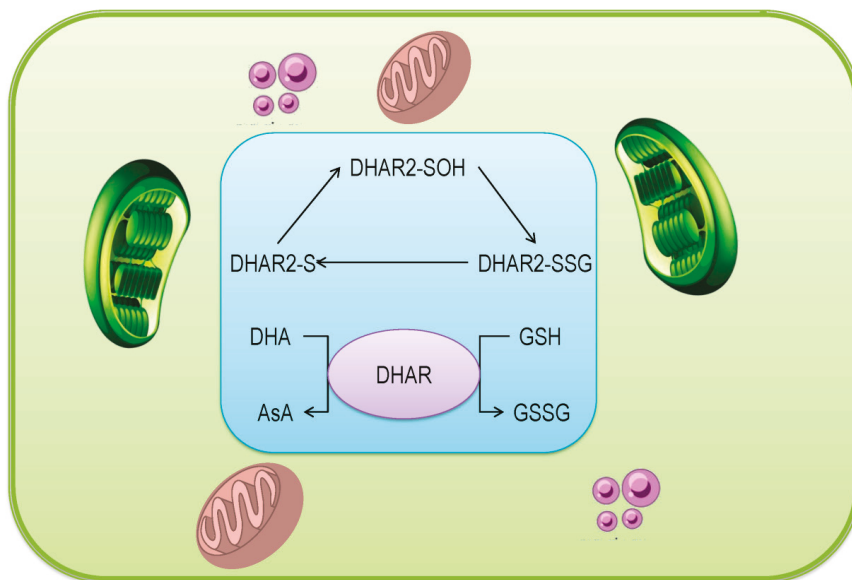


Figure 6. The mechanistic scheme, the ping-pong mechanism for the enzymatic reduction of dehydroascorbate (DHA).

Therefore, the process can be summarized by the following reaction:



As stated earlier, the integral function of DHAR is to reduce DHA to regenerate AsA. During this process, the active site of Cys is oxidized by DHA and further converted to the sulfenic acid. The reaction requires one molecule of H₂O. Knockout mutants of *Arabidopsis* *DHAR1*, *DHAR2*, and *DHAR3*, did not show any significant differences in total AsA content until facing the abiotic stress condition, which confirmed the necessity of DHAR in reducing the DHA during stress [55].

3.6. Glutathione Reductase

The flavoprotein oxidoreductase GR (glutathione reductase, EC 1.8.1.7) and reduced GSSG to GSH, also known by the term GSR or NADP⁺ oxidoreductase, as it employs NADPH for its cellular activity. Although GR is stated as a dimer, the monomeric, heterodimeric, and heterotetrameric forms have also been illustrated [29].

No less than two genes that encoded GR were reported, viz. GR1 and GR2 from higher plants. Where GR1 is cytosolic or peroxisomal and shorter, contrary, GR2 comprises a long N-terminal sequence and mitochondrial or chloroplastic [57]. Up to date, several researchers reported GR isoforms in many plants, for instance, tobacco, spinach, etc. [10]. Although being found in above-stated organelles of the

cell, the chloroplastic isoforms are accounted for 80% of GR [58]. The enzyme possesses a different quaternary structure that is based on the source from which it was purified [29].

Resembles with flavin-containing enzymes, GR exhibits the Rossmann folds, which is very much conserved and serves as the FAD and NADPH binding domains [59]. There is a controversy regarding the number of domain present in GR. Some reports suggested three, while some suggested four. Some researchers suggested an interface domain in GR protein; therefore, the enzyme has four domains, viz FAD-binding domain, NADPH-binding domain, GSSG-binding domain, and an interface domain [60]. Two Arg residues Arg287 and Arg293 are exclusively necessary for the enzymatic activity of GR [61]. Two Cys residue formed a disulfide bridge, which is redox-active and highly conserved. Ser164 replaces Cys residue in higher plants [62].

The enzyme shows high specificity to substrate binding, although the enzyme reduces GSH conjugates as well as mixed GSSG. Although Plant GR can employ NADP^{+} , its affinity towards NADPH is high [29]. The catalytic mechanism of GR accomplishes in two steps. The first step involves NADPH dependent reduction of the flavin moiety, which is further oxidized, meanwhile the disulfide bridge in active site reduced to form an anion–thiolate and release Cys. In the next step, GSSG molecule binds in the active site forming a disulfide bond together with a Cys and histidine (His) separately of the active site. Afterward, one GSH leaves the His, while another followed it and then leaves the Cys residue leaving the disulfide the bridge in the enzyme active site [63].

The brief reaction catalyzing by GR is as follows (Figure 7):

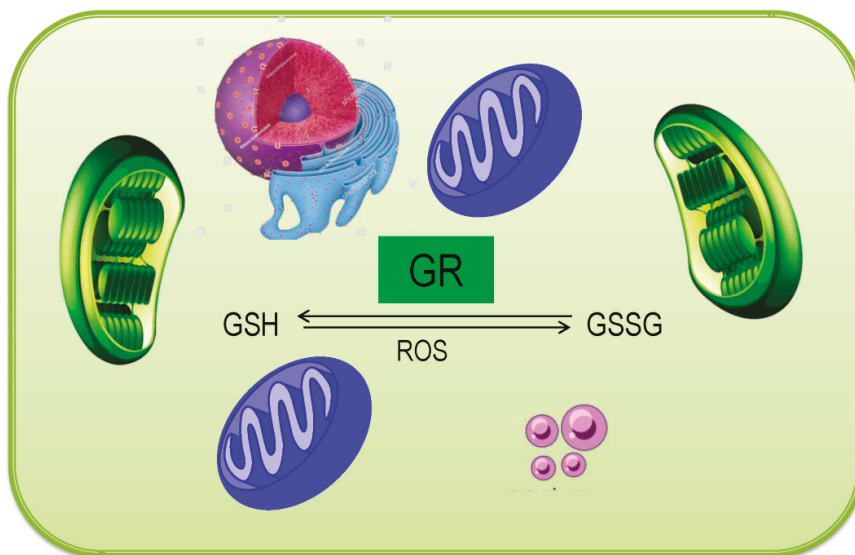


Figure 7. Mechanistic scheme for the enzymatic reduction of glutathione/oxidized glutathione (GSSG) in a plant cell.

During catalysis, pH, and NADPH, and GSSG concentration modulate GR activity. It was reported that low NADPH concentration reduces the GR activity, while below pH 5.5 and over 7.0 is not suitable for GR actions. On the other hand, NADPH-induced GR inhibition was prevented by GSSG [64].

4. Ascorbate and glutathione Redox and its Role in Plant Metabolism

Balanced metabolism is the prerequisite for better productivity in plants, which is always disturbed due to biotic and abiotic stresses. Thus, redox balance is one of the key features of life, by which oxidized products are reduced for further oxidization and energy supply. Moreover, plant cells should counter the oxidation of vital cellular component that occurs continuously due to the presence of 21% atmospheric molecular O₂, which is further complicated due to light-induced overproduction of ROS during photosynthesis. In addition, to keep the electron transport cascades active, simultaneous conversion of electron carriers between reduced and oxidized forms are required. Furthermore, photosynthetic and respiration needs regular electron flux to the electron transport chains from a different site. Therefore, the primary consequence is the generation of O₂⁻ and, subsequently, other ROS, from different enzyme catalysis [10,65]. Although playing a signaling role, over generation of ROS is harmful to cells; thus should be regulated to govern the redox homeostasis [66,67]. For example, AsA, GSH, tocopherols, thioredoxin, glutaredoxin, and peroxiredoxin, and energy metabolism mediators and electron carriers, for example, AsA/DHA, GSH/GSSG, FADH/FAD⁺, NADPH/NADP⁺, and NAD⁺/NADH play vital roles in plant cell for maintaining the redox balance and are termed as redox managers [68]. Among the redox managers, there are significant contributions drawn by AsA and GSH, hence in this section; we will discuss their role to keep redox balance as well as maintaining smooth cellular metabolism.

Reports suggested that, under control condition, the AsA/DHA ratio remains >9. Ascorbate becomes oxidized during ROS scavenging, electron donation to photosystem II (PSII), violaxanthin de-epoxidation, and α -tocopherol reductive quenching [69,70]. While the direct reduction of MDHA by ferredoxin at photosystem I (PSI) and by MDHAR, as well as DHA reduction by GSH dependent DHAR activity, maintains a highly reduced state of AsA pool [2,71]. The biosynthesis and metabolism of AsA are discussed earlier in this article (Section 3.1). In the apoplast and vacuoles, AsA pool is an important redox buffer for ROS detoxification, where AsA recycling is mainly accomplished in the cytosol, and AsA/DHA acts as an oxidative stress sensor [72].

The GSH redox potential depends on the GSH concentration and the ratio of GSH/GSSG. In the GSH pool [GSH/GSSG], if the GSSG remains constant, but total GSH decreases, the equilibrium position dropped and redox balance is disrupted. Thus, proper judgment of the GSH/GSSG could give the idea of the redox ratio [73]. Glutathione serves in a multiplicity of metabolic functions; for instance, it participates in the regeneration of AsA from DHA in the chloroplast by DHAR [74]. Moreover, GSH plays a role in reducing glutaredoxins, functions as a precursor of phytochelatin (PC) synthesis for chelating heavy metal, signal transduction, sulfur metabolism, xenobiotics detoxification, and protects protein thiols against irreversible oxidation with disulfide formation or glutathionylation, which inhibited enzymes, like enolase and 6-phosphogluconolactonase [75]. Plant cells have distinct compartmentation of GSH. Although all other cellular compartments, except vacuole, contain GSH/GSSG redox buffer, the only vacuole is the storehouse of GSH where the GSH-conjugates are degraded [76].

As discussed in earlier (Section 2), both AsA and GSH are connected to the reactions network, the AsA-GSH pathway, and the cellular redox homeostasis depends on the pathway components [76]. In this cycle, AsA performs electron donation for APX that works for H₂O₂ detoxification. Due to its high attraction for H₂O₂, APX is capable of efficient ROS scavenging, even in a low concentration, which gives rise to DHA. The produced DHA is further recycled back, which maintains a high ratio of AsA/DHA. If DHA cannot reduce, it might further be irreversibly hydrolyzed, which decreases the ability of AsA redox pool [77]. In the catalysis process GSSG produced, which is further recycled back, which maintains not only a high GSH/GSSG ratio, but also the balance between GSH and AsA pools [10].

In addition, the redox couples of AsA/DHA and GSH/GSSG can also function in another way for accomplishing redox signaling [78]. As discussed earlier, the AsA/DHA couples create redox balance inside cells. Moreover, AO converts AsA to DHA in the apoplast [79], and it creates a redox

gradient to connect intra- and extra-cellular atmosphere transverse the plasma membrane. Hence, AsA/DHA redox pair functions in apoplastic and cytoplasmic signals [80]. In contrast, the GSH/GSSG couple plays their functions in balancing intracellular redox potential, which in the intracellular redox signaling [81]. In this regard, the GSH distribution in different cellular organelle is very important for understanding cellular redox situation, for which signaling, as well as cellular metabolism, are smoothly going on [12].

5. Overview of Oxidative Stress and Antioxidant Defense in Plants

The production of ROS in living organisms is a usual cellular metabolism, and it is found in a large number in the internal constituents of the cell-like chloroplast, mitochondria, cytosol, peroxisomes, etc. [82–84].

Each plant cell maintains a dynamic balance between ROS and ROS-scavenging antioxidants. Abiotic stress destroys such cellular balance in favor of oxidative reactions by producing a huge amount of ROS [85]. Insufficient energy indulgence in the photosynthetic process during abiotic stresses reduces molecular oxygen and then produces a large amount of ROS, including H_2O_2 , O_2^- , 1O_2 , OH^\bullet , and so on (Figure 8) [10,86]. Reactive oxygen species are extremely reactive molecules and they can damage a large variety of cellular biomolecules, including carbohydrates, nucleic acids, lipids, proteins, etc., and alter their functions [85,87]. In addition, MG, a cytotoxic compound and reactive oxidizer, spontaneously produced in a cell in little amount but under abiotic stresses, its production increased and participated in developing oxidative stress (Figure 8). Similar to ROS, MG production is increased under abiotic stress, which can damage the ultra-structural constituents of cell and cause mutation, and ultimately provokes programmed cell death (PCD) [10].

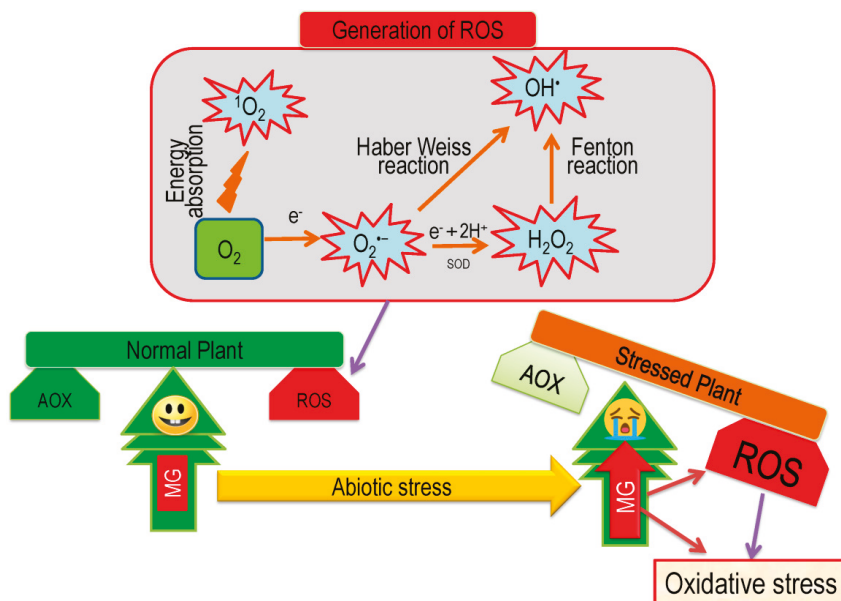


Figure 8. Abiotic stress-induced oxidative stress through the generation of ROS. Additional details are in the text.

Besides causing oxidative stress, ROS and MG play signaling roles for stress tolerance, which controls acclimation and defense responses by modulating some antioxidants and their respective

genes [10,86]. The excess generation of ROS and MG is also able to activate interruption in redox homeostasis, which can give the signal for cellular death or shortening plant life cycle [10,82].

Plant cells have well established antioxidant defense and glyoxalase system for scavenging toxic ROS and MG, respectively. The antioxidant defense system consists of some non-enzymatic components (AsA, GSH, alkaloids, α -tocopherol, non-protein amino acids, and phenolic compounds) and enzymatic components (SOD, CAT, APX, MDHAR, DHAR, GR, GST, and GPX [5,28]). Within the antioxidant defense system, the AsA-GSH pool performs the direct and significant role for minimizing stress effect through scavenging of ROS by using key four enzymes, e.g. APX, MDHAR, DHAR, and GR [28,88,89]. In our previous section (Section 3.3, 3.4, 3.5, and 3.6), we elaborately discussed the function of these four enzymes in ROS detoxification. Usually, in the antioxidant defense system, SOD gives frontline protection against ROS by converting O_2^- to H_2O_2 . Subsequently, CAT and APX scavenge H_2O_2 to H_2O . Glutathione peroxidase and GST also scavenge H_2O_2 to H_2O with the help of GSH (Figure 9) [28].

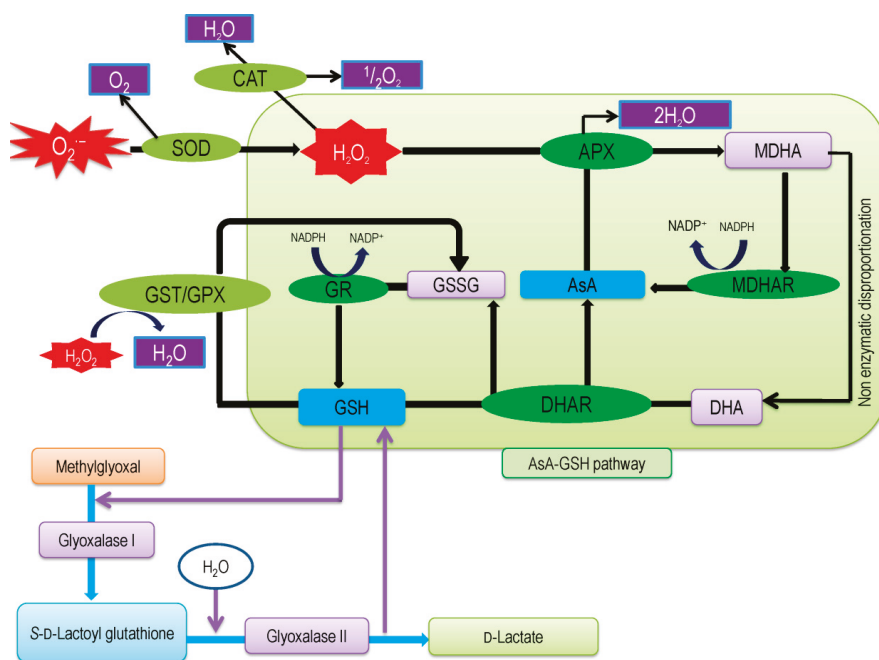


Figure 9. AsA-GSH pathway of the antioxidant defense system and its relation with the glyoxalase system. Additional details are in the text.

Toxic MG is detoxified in the cell by glyoxalase system. Glutathione is not only the major element of AsA-GSH cycle, but it also plays a significant function in the MG detoxification system. Glyoxalase system is composed of two vital enzymes, Gly I and Gly II. In glyoxalase system, MG is detoxified to non-toxic compound in two steps reactions; in the initial step, MG is transformed to S-D-lactoyl-glutathione through the utilization of GSH and in the final step S-D-lactoyl-glutathione transformed in to D-Lactate, where GSH is recycled back [11]. Moreover, GSH contributes to metal chelation. It enhances the amount of PC under stress condition, which makes a complex with metal and drives into the cell vacuole as inert form [90].

6. Role of AsA-GSH in Regulating Oxidative Stress under Abiotic Stresses

Abiotic stress-induced excess ROS causes oxidative stress in plants followed by cellular damage, even death. Hence, the plant itself defends against this higher ROS accumulation by their defense mechanism. Plant significantly activates the AsA-GSH pathway for ROS detoxification. In this section, we will discuss the involvement of AsA-GSH cycle for alleviating oxidative stress upon various abiotic stresses reviewing recently published articles (Tables 1–3).

6.1. Salinity

One of the most devastating abiotic stress factors—salinity by which cultivable land is becoming barren thus reduces total crop production day by day. Oxidative stress is the most dangerous event under salt inundation is imposed by salinity-induced ionic and osmotic stress [10]. Hence, these ionic and osmotic stress both disturb the photosystem, and thus cause excess ROS, such as $^1\text{O}_2$, O_2^- , H_2O_2 , and OH. Salinity-persuaded acute ROS accumulations, then bother cellular redox followed by cellular damage counting membrane dysfunction, DNA damage, collapse the enzymatic action, along with distraction of the antioxidant defense system [91,92]. At this point, the plant synthesizes cellular AsA and GSH, which act as non-enzymatic antioxidants by involving their enzymatic components to detoxify ROS up to tolerable levels (Table 1).

However, the enzymes of AsA-GSH pathway showed their differential responses intolerant and sensitive varieties due to saline toxicity. Among salt-tolerant (Pokkali) and sensitive (BRRI dhan29) rice cultivars. Pokkali responded by enhancing the enzymatic activities of the AsA-GSH cycle, where, lowered APX and higher DHAR activity along with unchanged MDHAR and GR activities were found from BRRI dhan29. Rahman et al. [91,93] reported about the well involvement of AsA-GSH cycle in salt-stressed *O. sativa* where ROS generation was extreme. Here, salt exposed rice enhanced the reduced and oxidized GSH content with a lesser amount of AsA by the higher APX, MDHAR, DHAR, and GR activities against overproduced ROS. *Vigna radiata* was grown under the saline condition [94] and where salt-induced oxidative stress was marked with extreme O_2^- and H_2O_2 overgeneration. Salt-stressed *V. radiata* augmented GSH and GSSG contents along with lowered AsA, whereas up-regulated the activity of all enzymatic antioxidants of AsA-GSH cycle and thus responded with elevated ROS [95]. Salt exposed *Lens culinaris* up-stimulated both MDHAR and DHAR activities, which resulted in a lesser amount of AsA and indicated the overproduced H_2O_2 detoxification [96]. Recently, Singh et al. [97] disclosed the incremental activity of enzymatic antioxidants, including APX, DHAR, and GR, with lower AsA, GSH, and GSSG contents, because of salt-induced higher ROS accumulation in *Solanum lycopersicum*. Similarly, 150 mM salt-treated *S. lycopersicum* also decreased AsA content, which might be used in H_2O_2 detoxification, while better GSH showed its role in lowering H_2O_2 . Ahmad et al. [98] also observed higher APX, and GR activities, while MDHAR and DHAR activities again reduced as well as supported AsA-GSH mediated ROS regulation. Ahanger et al. [99] reported the same response of *S. lycopersicum* upon saline toxicity. Both activities of APX and GR were enhanced in salt-treated *Triticum aestivum* besides elevated H_2O_2 generation and resulted in higher GSH accumulation [100]. The activity of APX, MDHAR, DHAR, and GR enhanced in salt-stressed *S. lycopersicum* to check the excessive H_2O_2 generation, which resulted in lowered AsA and GSH contents [92].

The changes in AsA-GSH pathway were investigated in salt-stressed *Nitraria tangutorum* by applying a varied level of NaCl (100, 200, 300, and 400 mM) [101]. They noticed a gradual enhancement of AsA, DHA, GSH, and GSSG contents by keeping pace with sequential increment of salt-induced H_2O_2 . Here, increased MDHAR and DHAR activities in stressed seedlings also contributed to increasing AsA, and higher DHAR and GR were responsible for better GSH and GSSG contents [92,102]. Talaat et al. corroborated these results with salt-exposed *Phaseolus vulgaris* [103]. Thus, as a part of plant antioxidant defense under salinity, AsA-GSH pathway is very efficient to regulate extra ROS for being tolerant.

6.2. Drought

Drought is another most important abiotic stress, which generates excess ROS accumulation and thus causes variation in the enzymatic activities of AsA-GSH pathway for ROS detoxification. The enzymatic responses of AsA-GSH pathways varied, depending upon plant species, plant age, drought intensity, and duration [10]. Commonly, drought up-regulated the enzymatic antioxidant activities of AsA-GSH pool [10,104]. Plant tolerance to drought stress is categorized based on stress-induced endogenous antioxidants contents along with enzymatic activities (Table 2). *Dendranthema grandiflorum* responded differentially according to their tolerant and sensitive varieties, where tolerant one comparatively displayed better enzyme activity of antioxidants than sensitive [105]. Lou et al. [106] demonstrated how *T. aestivum* responded upon drought exposure. Hence, they noticed that the AsA-GSH cycle responded considerably with excess ROS generation by significant variation of GSH/GSSG and AsA/DHA redox along with the steady increment of H₂O₂. Their team also observed the enzymatic up-stimulation of AsA-GSH pathway to alleviate stress by scavenging excess ROS in *T. aestivum* spike. Thus, *T. aestivum* showed higher participation of AsA with higher APX activity in drought exposure for scavenging extra H₂O₂, as well as higher enzymatic activity to run the AsA-GSH pathway systematically [107].

Table 1. Role of AsA-GSH in regulating oxidative stress under salinity and drought.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Regulation	References
<i>Triticum aestivum</i> L.	100 mM NaCl	GSH content increased by 15%; Stimulated APX and GR activities by 78% and 56%, respectively	Increased H ₂ O ₂ content about 79%	[100]
<i>T. aestivum</i> L. cv. BARI Gom-21	12% PEG for 48 and 72 h	Decreased AsA content at 48 h, but after 72 h, AsA content again enhanced; Increased GSH and GSSG content where GSH/GSSG ratio decreased time-dependently; Enhanced the activities of APX, MDHAR, and GR	Enhanced the H ₂ O ₂ content by 62% and increased O ₂ ⁻ accumulation	[113]
<i>T. aestivum</i> L.	10% PEG	Reduced AsA/DHA and GSH/GSSG redox; Increased enzymatic antioxidants actions of AsA-GSH cycle	Increased H ₂ O ₂ production	[107]
<i>T. aestivum</i> L.	35–40% field capacity (FC) water	Increased GSH/GSSG by 64% while decreased AsA/DHA by 52% respective with a duration of stress; Enhanced APX, MDHAR, DHAR and GR activities	Increased H ₂ O ₂ along with stress duration	[106]
<i>T. aestivum</i> cv. Pradip	150 and 300 mM NaCl	Reduced AsA content upto 52%; Increased reduced and oxidized GSH accumulation by 55% and 18%, respectively with 32% higher GSH/GSSG ratio; Increased APX activity with 29% reduction of GR activity; Slightly increased MDHAR and DHAR activity	Enhanced H ₂ O ₂ generation by 60%	[28]
<i>Oryza sativa</i> L. cv. BRR1 dhan47	150 mM NaCl	Increased GSH accumulation while reduced AsA content by 49% Increased GSH content and lowered the redox status of both AsA/DHA and GSH/GSSG; Upregulated the activity of APX, MDHAR, DHAR, and GR	Increased the production of O ₂ ⁻ with 82% higher H ₂ O ₂ accumulation	[93]

Table 1. Cont.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Regulation	References
<i>O. sativa</i> L. cv. BRRI dhan49	300 mM NaCl	Reduced AsA and GSH accumulation by 51% and 57%, respectively; Decrease GSH/GSSG redox by 87%; Showed lowered APX (27%), MDHAR (24%), DHAR and GR (25%) activities	Increased H ₂ O ₂ content upto 69%	[114]
<i>O. sativa</i> L. cv. BRRI dhan54	300 mM NaCl	Improved AsA content by 51% with higher GSH content; Decreased GSH/GSSG ratio by 53%; Showed higher APX (27%) and DHAR activities while decreased both GR (23%) and MDHAR activities	Accumulated 63% higher H ₂ O ₂ content	[114]
<i>Brassica napus</i> L. cv. BINA sharisha 3	100 and mM NaCl	Reduced the AsA content by 22%; Increased GSH content by 72% and GSSG content by 88%; Unaltered the GSH/GSSG ratio; Amplified APX activity by 32%, decreased DHAR activity by 17%; Slightly increased GR activity	Accumulated higher H ₂ O ₂ content by 76%	[115]
<i>B. napus</i> L. cv. BINA sharisha 3	200 mM NaCl	Reduced the AsA content (40%) along with increased GSH (43%) and GSSG (136%) contents; Decreased the GSH/GSSG ratio (40%); Amplified the APX activity (39%) and reduced the MDHAR (29%) and DHAR (35%) activities; Improved GR activity (18%)	Showed 90% more H ₂ O ₂ content	[115]
<i>B. napus</i> L.	15% PEG	The AsA accumulation remained unaltered and reduced the AsA/DHA ratio; Enhanced GSH content by 19% and GSSG by 67% and decreased GSH/GSSG ratio; Increased APX, MDHAR, DHAR and GR activities	Higher accumulation of H ₂ O ₂ by 55%	[116]
<i>B. campestris</i> L.	15% PEG	Decreased AsA content by 27% with a decrease of AsA/DHA ratio; Increased GSH content by 33% with higher GSSG content by 79% and lowered GSH/GSSG ratio; Decreased DHAR activity	Higher accumulation of H ₂ O ₂ about 109%	[116]
<i>B. juncea</i> L.	15% PEG	Increased the AsA content and did not affect the AsA/DHA ratio; Increased GSH content by 48% and GSSG by 83% and decreased GSH/GSSG ratio; Increased APX, MDHAR, DHAR and GR activities	Accumulation of 37% higher H ₂ O ₂	[116]
<i>B. juncea</i> L. cv. BARI Sharisha 11	10% PEG	Reduced AsA content (14%) while increased both GSH (32%) and GSSG (48%) contents; Enhanced APX activity (24%); Decreased MDHAR and DHAR (33%) activities along with 31% increased GR activity	Acute generation of H ₂ O ₂ (41%)	[117]
<i>B. juncea</i> L. cv. BARI Sharisha 11	20% PEG	Decreased AsA content by 34% while increased the content of GSH by 25% and GSSG by 101%; Up-regulated APX activity by 33%; Decreased activity of MDHAR and DHAR (30%)	Extreme generation of H ₂ O ₂ by 95%	[117]

Table 1. Cont.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Regulation	References
<i>B. napus</i> L. cv. BINA Sarisha 3	10% PEG	Increased AsA (21%), GSH (55%) and GSSG contents while decreased GSH/GSSG ratio Unaltered the activities of APX, and increased the activity of MDHAR, DHAR, and GR (26%)	Elevated the H ₂ O ₂ production	[11]
<i>B. napus</i> L. cv. BINA Sarisha 3	20% PEG	Unaltered AsA content along with higher content of GSH (46%) and GSSG and reduced GSH/GSSG ratio; Reduced the APX and MDHAR activities along with the higher activity of DHAR and GR (23%)	Showed higher H ₂ O ₂ production	[11]
<i>B. napus</i> L. cv. BINA sharisha 3	10% PEG	Increased AsA, GSH (31%) and GSSG (83%) accumulation with lowered GSH/GSSG ratio; Increased APX activity while reduced MDHAR and DHAR activities, but GR activity remained unaltered	Increased H ₂ O ₂ content by 53%	[52]
<i>B. napus</i> L. cv. BINA Sharisha 3	20% PEG	Slightly increased AsA content with 26% and 225% increase of GSH and GSSG content, respectively; Reduced GSH/GSSG ratio; Increased APX activity while decreased the activity of MDHAR, DHAR, and GR (30%)	Increased about 93% H ₂ O ₂ content	[52]
<i>B. rapa</i> L. cv. BARI Sharisha-15	20% PEG	Slightly increased AsA content with 72% and 178% increase of GSH and GSSG content, respectively; Reduced GSH/GSSG ratio by 38%; Increased APX, MDHAR, DHAR, and GR activity	Increased about 131% H ₂ O ₂ content	[104]
<i>Cucumis melo</i> L. cv. Yipintianxia No. 208	50 mM of NaCl:Na ₂ SO ₄ :NaHCO ₃ :Na ₂ CO ₃ (1:9:9:1 M)	Improved AsA, GSSG and DHA contents; Lowered GSH content; Reduced the ratio of AsA/DHA and GSH/GSSG; Stimulated the activity of APX by 96% and DHAR by 38% while reducing the activity of MDHAR and GR by 48% and 34%, respectively	Increased H ₂ O ₂ accumulation	[118]
<i>Solanum lycopersicum</i> L., var. Lakshmi	0.3 and 0.5 g NaCl kg ⁻¹ soil	Reduced AsA and AsA/DHA ratio; Lowered GSH and GSSG accumulation with decreased GSH/GSSG redox; Increased APX activity by 28%, DHAR activity by 28% and GR activity by 14%	Enhanced H ₂ O ₂ and O ₂ ⁻ accumulation	[97]
<i>S. lycopersicum</i> L.cv. Boludo	60 mM NaCl, 30 days	Reduced the activities of APX, DHAR, and GR; Increased MDHAR activity	Higher H ₂ O ₂ generation	[119]
<i>S. lycopersicum</i> L. var. Pusa Ruby	150 mM NaCl	Decreased AsA and GSH content with a higher content of DHA and GSSG; Increased APX, MDHAR, DHAR and GR activities	Higher generation of H ₂ O ₂ and O ₂ ⁻	[92]
<i>S. lycopersicum</i> L. var. Pusa Rohini	150 mM NaCl	Reduced AsA content by 42%; Increased both GSH and GSSG accumulation; Enhanced the activity of APX and GR by 86% and 29%, respectively with reduction of the activity of MDHAR and DHAR by 38% and 32%, respectively	Accumulated about 3 fold higher H ₂ O ₂ content	[99]

Table 1. Cont.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Regulation	References
<i>S. lycopersicon</i> L. cv.K-21	150 mM NaCl	Reduced AsA content by 40% with 50% higher GSH content; Lowered GSSG content by 23% while increased GSH/GSSG ratio by 112%; Increased APX (86%) and GR (92%) activity along with the lowered activity of MDHAR (32%) and DHAR (30%)	Elevated H ₂ O ₂ content about 175%	[98]
<i>Nitraria Tangutorum</i> Bobr.	100,200, 300 and 400 mM NaCl	Increased AsA, DHA, GSH and GSSG accumulation decreased their redox status; Enhanced the activity of APX and GR; Unvaried the activity of DHAR and MDHAR but increased DHAR activity only at 300 mM NaCl	Increased O ₂ ⁻ and H ₂ O ₂ content by 38–98 and 49–102% respectively	[101]
<i>Camellia sinensis</i> (L.) O.Kuntze	300 mM NaCl	Enhanced the AsA and GSH content; Increased APX activity	Elevated H ₂ O ₂ and O ₂ ⁻ content	[120]
<i>Phaseolus vulgaris</i> L. cv. Nebraska	2.5 and 5.0 dS m ⁻¹ prepared from a mixture of NaCl, CaCl ₂ and MgSO ₄	Increased AsA, GSH, DHA and GSSG accumulations; Enhanced AsA/DHA and GSH/GSSG status; Stimulated the enzymatic activity of APX, MDHAR, DHAR and GR activities	Accumulated higher H ₂ O ₂ content	[103]
<i>Vigna radiate</i> L. cv. BINA moog-1	25% PEG	Reduced AsA content along with higher GSH content of 92%; Increased GSSG content by 236% and reduced GSH/GSSG ratio; Amplified the activity of APX (21%) and GR while reduced MDHAR and DHAR activities	Elevated H ₂ O ₂ content by 114% with higher O ₂ ⁻ generation	[111]
<i>V. radiata</i> L.	200 mM NaCl	Reduced AsA content; Increased GSSG and GSH accumulation and lowered GSH/GSSG ratio; Amplified the activity of APX, MDHAR, DHAR, and GR	Increased H ₂ O ₂ content by 80% and O ₂ ⁻ generation by 86%	[95]
<i>V. radiata</i> L. cv. BARI Mung-2	5% PEG	Reduced AsA content where decreased AsA/DHA ratio by 54%; Increased GSSG content; Upregulated the activity of APX and GR (42%) while downregulated the MDHAR (26%) and DHAR activities	Elevated H ₂ O ₂ and O ₂ ⁻ accumulation	[50]
<i>Lens culinaris</i> Medik cv. BARI Lentil-7	20% PEG	Lowered AsA content with higher total GSH content; Unaltered the APX and GR activities while the increased activity of MDHAR and DHAR (64%)	Accumulated higher H ₂ O ₂ content	[96]
<i>L. culinaris</i> Medik cv. BARI Lentil-7	100 mM NaCl	Reduced AsA content by 87% while increased total GSH content by 260%; Improved the activity of APX, MDHAR, DHAR (286%) and GR (162%)	Increased H ₂ O ₂ content by 15%	[96]
<i>Anacardium occidentale</i> L.	21-day water withdrawal	Enhanced total AsA and GSH content; Increased APX activity	Reduced H ₂ O ₂ generation	[112]
<i>Arabidopsis</i>	12-day water withhold	Showed higher GSH and GSSG accumulation; Reduced GSH/GSSG ratio; Increased GR activity	Increased H ₂ O ₂ accumulation rate	[108]
<i>Cajanus cajan</i> L.	Complete water withholding for 3, 6 and 9 days	Decreased GSH/GSSG ratio; Increased the activity of APX, DHAR, and GR	Higher H ₂ O ₂ content	[109]

Table 1. Cont.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Regulation	References
<i>Amaranthus tricolor</i> L.cv. VA13	30% FC	Increased AsA and GSH contents by 286% and 98%, respectively; Improved APX, MDHAR, DHAR, and GR activity by 371%, 379%, 375%, and 375%, respectively	No increment of H ₂ O ₂ content	[110]
<i>A. tricolor</i> L.cv. VA15	30% FC	Increased AsA and GSH contents along with higher redox status of AsA/total AsA and GSH/total GSH; Enhanced the activity of APX, MDHAR, DHAR, and GR by 37%, 45%, 40%, and 2%, respectively	Accumulated higher H ₂ O ₂ content by 137%	[110]
<i>C. sinensis</i> (L.) O. Kuntze	20% PEG	Higher contents of both AsA and GSH; Enhanced the APX activity	Higher accumulation of H ₂ O ₂ and O ₂ ⁻	[120]

Drought-stressed *A. thaliana* enhanced GSH and GSSG content along with the higher GR activity [108]. Hence, *Arabidopsis* showed the GSH dependent H₂O₂ detoxification to attain tolerance. Higher total AsA was accumulated in *Cajanus cajan* upon complete water restriction conditions for up to nine days to defend against excess H₂O₂ toxicity [109]. Hence, drought enhanced the enzymatic activity of APX, DHAR, and GR for decreasing GSH/GSSG, as well as controlling ROS level.

Similarly, tolerant genotype VA13 of *Amaranthus tricolor* showed comparatively better tolerance under drought stress than sensitive one (VA15) by expressing differential responses of the enzymatic and non-enzymatic ROS detoxification pathways [110]. Hence, VA13 expressed remarkable increment in AsA-GSH redox by accelerating the enzymatic antioxidative actions by which increased non-enzymatic antioxidants (AsA and GSH) accumulation, which are vital for ROS detoxification.

Vigna radiata responded differently regarding different drought intensity [111] to control diverse levels of ROS. Moderate drought imposed by 10% polyethylene glycol (PEG) induced comparatively lowered ROS than severe drought (by 20% PEG). Therefore, severe drought-stressed *Brassica* showed a larger use of AsA-GSH pathways against higher H₂O₂ generation than moderate stress. Here, higher stress caused a higher increase of APX activity along with lowest MDHAR and DHAR activity, while GR activity reduced differently than lower stress exposure to rapeseeds seedlings. Additionally, Hasanuzzaman et al. [52] also observed AsA and GSH both antioxidants contents reduced under severe drought condition, but increased under moderate stress. Bhuiyan et al. [104] found increased AsA content in *B. rapa* under drought (20% PEG). They also observed increased APX activity in drought-stressed seedlings, which assisted in efficiently scavenging the H₂O₂. Another two enzymes related to AsA regeneration MDHAR and DHAR also upregulated, as a result the AsA level was increased and strongly maintained its redox balance during oxidative stress situation. Nahar et al. [111] narrated the function of AsA as ROS detoxifier under drought stress where AsA content reduced in *V. radiata* with the increasing of ROS generation. Here, drought-induced higher APX activity enhanced the oxidation of AsA by scavenging H₂O₂ and improved GR activity increased the supply of GSH for involving ROS detoxification. *Anacardium occidentale* also showed the active participation of AsA-GSH cycle by integrative responses of both non-enzymatic and enzymatic antioxidants for drought-induced excess ROS regulation, where the higher accumulation of AsA and GSH, along with APX activity, coordinately reduced the overproduced H₂O₂ [112]. Thus, the AsA-GSH pathways involve in ROS detoxification as well as ROS homeostasis by eliminating excess ROS for keeping them up to the requirement of functioning cell signals.

6.3. Toxic Metals/Metalloids

Due to fast industrialization of the modern world and unrestrained anthropogenic activities, toxic metals/metalloids stresses have become a gargantuan problem for the plant growth and

development [121]. Plants experience toxic metals/metalloids stress try to survive to some extent by using their well established antioxidant defense system. But, the activity and performance of defense system differ with stress concentration, stress duration, plant type, and age of the plant.

The enzymes of AsA-GSH pathway confirmed their differential responses to different toxic metals/metalloids stress (Table 2). Mahmud et al. [122] confirmed that due to Cr stress, the few components of AsA-GSH pathway increased their amount or activity in *B. juncea* L. cv. BARI Sharisha-11. They found five days duration of 0.15 mM and 0.3 mM K_2CrO_4 treatment decreased the content of AsA, but did not change the GSH content. Moreover, activities of APX and GR were enhanced; however, the activities of MDHAR and DHAR were diminished. The higher APX and GR activity might play a function in scavenging excess ROS. A similar upregulation of APX and GR was also recorded in *B. napus* L. cv. BINA sharisha 3 due to Cd treatment [123]. From two separate experiments, they also found Cd stress (0.5 mM and 1.0 mM $CdCl_2$) for 48 h decreased the AsA content, but increased GSH content only under 0.5 mM $CdCl_2$ treatment. Exposure of *Gossypium* to 50 and 100 μM $Pb(NO_3)_2$ for six weeks increased the H_2O_2 content and APX activity [124]. The addition of 150 μM $NiCl_2 \cdot 6H_2O$ in growing media of *B. juncea* L. for one week increased the H_2O_2 content. Moreover, Ni stress decreased the AsA level but augmented the content of GSH and GSSG. Nickel also diminished the function of DHAR and MDHAR, however enhanced APX and GR activity [125]. Similar differential responses of AsA-GSH pathway components were also observed under As [126] and Al [50] toxicity. It can be stated that overproduced ROS plays the signaling role to some extent and inaugurate the higher activity of AsA-GSH enzymes under metals/metalloids toxicity. The upregulation of enzymes plays a significant role in maintaining the redox balance of AsA-GSH pathway under stress condition.

Table 2. Status of AsA-GSH in regulating oxidative stress under metal/metalloid stress.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Regulation	References
<i>Brassica napus</i> L. cv. BINA sharisha 3	Cd (0.5 mM and 1.0 mM $CdCl_2$), 48 h	Reduced AsA content by 20% under 0.5 mM and 32% under 1.0 mM $CdCl_2$ treatment; Increased GSH content only under 0.5 mM $CdCl_2$ stress but enhanced level of GSSG by 34% under 0.5 mM and 65% under 1.0 mM $CdCl_2$ treatment; Increased function of APX by 39% and 43% under 0.5 mM and 1.0 mM $CdCl_2$ treatment but MDHAR and DHAR activity were diminished in dose dependant fashion; GR activity increased by 66% due to 0.5 mM $CdCl_2$ treatment but reduced by 24% due to 1.0 mM $CdCl_2$ treatment	Enhanced H_2O_2 content by 37% under 0.5 mM and 60% under 1.0 mM $CdCl_2$ treatment	[88]
<i>Gossypium</i> spp. (genotype MNH 886)	Pb [50 and 100 μM $Pb(NO_3)_2$], 6 weeks	Increased APX activity	Increased H_2O_2 content	[124]
<i>T. aestivum</i> L. cv. Pradip	As (0.25 and 0.5 mM $Na_2HAsO_4 \cdot 7H_2O$), 72 h	Reduced AsA content by 14% under 0.25 and 34% under 0.5 mM $Na_2HAsO_4 \cdot 7H_2O$ treatment; Increased GSH content by 46% and 34%, GSSG content by 50 and 101% under 0.25 and 0.5 mM $Na_2HAsO_4 \cdot 7H_2O$ stress; Enhanced APX function by 39% and 43% but decreased DHAR function by 33% and 30% under 0.25 and 0.5 mM $Na_2HAsO_4 \cdot 7H_2O$ treatment; Increased GR function by 31% under 0.25 mM	Increased H_2O_2 content by 41% under 0.25 and 95% under 0.5 mM $Na_2HAsO_4 \cdot 7H_2O$ treatment	[127]

Table 2. Cont.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Regulation	References
<i>B. napus</i> L. viz. ZS 758, Zheda619, ZY 50 and Zheda 622	Cr (400 μ M), 15 days	Increased GSH and GSSG content; Increased APX activity	Increased H ₂ O ₂ content	[128]
<i>Oryza sativa</i> L. cv. BRRI dhan29	As (0.5 mM and 1 mM Na ₂ HAsO ₄), 5 days	Decreased AsA content by 33 and 51% and increased DHA content by 27% and 40% under 0.5mM and 1mM Na ₂ HAsO ₄ treatment, respectively; Decreased ratio of AsA/DHA; Enhanced GSH content by 48 and 82% under 0.5mM and 1mM Na ₂ HAsO ₄ treatment, respectively; Enhanced GSSG content whereas lessened GSH/GSSG ratio by 25% under 0.5mM and 41% under 1mM Na ₂ HAsO ₄ treatment; Augmented the function of APX, MDHAR, and GR, however, reduced the activity of DHAR	Increased H ₂ O ₂ content by 65% and 89% under 0.5mM and 1mM Na ₂ HAsO ₄ treatment, respectively	[126]
<i>O. sativa</i> L. cv. Disang (tolerant)	100 μ M AlCl ₃ , 48 h	Increased AsA content in both roots and shoots; Enhanced the GSH content in shoots; Higher activities of APX, MDHAR, DHAR, and GR,	Elevated the generation of H ₂ O ₂ and O ₂ ⁻	[129]
<i>O. sativa</i> L. cv. Joymati (sensitive)	100 μ M AlCl ₃ , 48 h	Higher accumulation of AsA in both roots and shoots; Reduced the GSH content in roots while shoots content was unaltered; Increased APX, MDHAR, DHAR activities; Slightly increased GR activities	Higher accumulation of H ₂ O ₂ and O ₂ ⁻	[129]
<i>V. radiata</i> L. cv. BARI Mung-2	Cd (mild: 1.0 mM CdCl ₂ , severer: 1.5 mM CdCl ₂), 48 h	Declined AsA content by 31% due to mild and 41% due to severe stress; Enhanced DHA level and reduced AsA/DHA ratio; GSH content did not change due to mild stress but enhanced owing to stress severity; GSSG level enhanced, and GSH/GSSG ratio decreased in dose-dependent manner; Increased function of APX but lessened MDHAR and DHAR function due to both level of stress; GR activity increased only due to severe stress	H ₂ O ₂ level and O ₂ ⁻ generation rate was augmented by 73% and 127% due to mild and 69% and 120% due to severe Cd stresses	[130]
<i>V. radiata</i> L. cv. BARI Mung-2	Cd (1.5 mM CdCl ₂), 48 h	AsA content decreased by 27%, and the ratio of AsA/DHA reduced by 80% whereas DHA content increased considerably; Augmented the function of APX and GR however lessened function of MDHAR and DHAR	Increased H ₂ O ₂ level and O ₂ ⁻ generation rate	[131]
<i>O. sativa</i> L. cv. BRRI dhan29	Cd (0.25 mM and 0.5 mM CdCl ₂), 3 days	AsA content and AsA/DHA ratio reduced by 37% and 57% due to 0.25 mM CdCl ₂ and reduced by 51% and 68% due to 0.5 mM CdCl ₂ , respectively; DHA content increased significantly; GSH content enhanced due to 0.25 mM CdCl ₂ stress, but reduced due to 0.5 mM CdCl ₂ stress; GSSG content enhanced by 76% under 0.25 mM and 108% under 0.5 mM CdCl ₂ stress; Reduced ratio of GSH/GSSG in dose dependant manner; Enhanced APX, MDHAR and GR activity	Enhanced H ₂ O ₂ by 46% under 0.25 mM CdCl ₂ and 84% under 0.5 mM CdCl ₂ treatment whereas O ₂ ⁻ generation rate increased in dose dependant manner	[132]

Table 2. Cont.

<i>O. sativa</i> L. cv. BRRI dhan29	Cd (0.3 mM CdCl ₂), 3 days	Lessened level of AsA and AsA/DHA ratio but enhanced DHA level; Enhanced the level of GSH and GSSG however lessened GSH/GSSG ratio; Enhanced the action of APX, MDHAR, and GR whereas declined DHAR function	Overproduced ROS (H ₂ O ₂ and O ₂ ⁻)	[133]
<i>O. sativa</i> L. Zhunliangyou 608	Cd (5 μM Cd(NO ₃) ₂ ·4H ₂ O), 6 days	Reduced AsA content; Increased GSH content; Slightly reduced the APX activity	H ₂ O ₂ content increased by 22.73%	[134]
<i>Abelmoschus esculentus</i> L. Moench	Pb (100 mg L ⁻¹), 21 days	Increased AsA content	Enhanced H ₂ O ₂ content	[135]
<i>B. juncea</i> L. cv. BARI Sharisha-11	Cr (mild: 0.15 mM K ₂ CrO ₄ , severe: 0.3 mM K ₂ CrO ₄), 5 days	AsA content lessened by 19% due to mild and 32% due to severe stress whereas DHA level enhanced by 83% due to mild and 133% due to severe stress as well as AsA/DHA ratio lessened by 47% due to mild and 82% due to severe stress; GSH content did not change considerably but GSSG content enhanced by 42% due to mild and 67% due to severe stress as well as GSH/GSSG ratio lessened by 26% due to mild and 41% due to severe stress; The function of APX enhanced by 21% due to mild and 28% due to severe stress; The activity of MDHAR and DHAR reduced by 25 and 32% under mild and 31 and 50%, under severe stress, respectively; Mild stress increased the activity of GR by 19% while severe stress increased by 16%	H ₂ O ₂ level enhanced by 24% and 46% due to mild and severe stress. Similarly, O ₂ ⁻ generation rate also raised in a dose-dependent manner	[122]
<i>B. campestris</i> L. cv. BARISharisha 9, <i>B. napus</i> L. cv. BARI Sharisha 13 and <i>B. juncea</i> L. cv. BARI Sharisha 16	Cd (mild: 0.25 mM CdCl ₂ , severer: 0.5 mM CdCl ₂), 3 days	Decreased level of AsA, augmented level of DHA as well as decreased AsA/DHA ratio in all studied cultivars; GSH and GSSG level enhanced, but GSH/GSSG ratio lessened in all studied cultivars; APX and GR activities of all species increased significantly under both levels of Cd toxicity	Enhanced H ₂ O ₂ level and O ₂ ⁻ production rate in all tested cultivars in a concentration-dependent fashion	[136]
<i>B. juncea</i> L. BARI Sharisha-11	Cd (mild: 0.5 mM CdCl ₂ , severer: 1.0 mM CdCl ₂), 3 days	Reduced AsA content with higher DHA content and thus decreased AsA/DHA ratio; Increased GSH and GSSG levels as well as declined GSH/GSSG ratio; APX activity increased where GR increased at mild stress but remained unaltered at severe stress; Decreased MDHAR and DHAR activities	Enhanced the H ₂ O ₂ and O ₂ ⁻ level	[137]
<i>V. radiata</i> L. cv. BARI Mung-2	Al (AlCl ₃ , 0.5 mM), 48 and 72 h	Enhanced DHA content but reduced AsA level and AsA/DHA ratio; Increased level of GSH and GSSG but the diminished ratio of GSH/GSSG; Augmented APX activity but decreased MDHAR and DHAR activity	Enhanced H ₂ O ₂ level by 83% and O ₂ ⁻ generation rate by 110%	[50]

Table 2. Cont.

<i>T. aestivum</i> L. cv. Pradip	Pb [mild: 0.5 mM Pb(NO ₃) ₂ , severer: 1.0 mM Pb(NO ₃) ₂], 2 days	AsA decreased in a dose-dependent manner; Mild stress improved the GSH level, but severe stress reduced it; Increased GSSG content; Increased APX activity; Diminished activity of MDHAR and DHAR in a concentration-dependent fashion; Mild stress improved GR activity but severe stress reduced it	Mild stress increased H ₂ O ₂ levels by 41%, but severe stress enhanced it by 95% while O ₂ ⁻ generation rate also increased in a dose-dependent manner	[35]
<i>B. juncea</i> L. cv. BARI Sharisha-11	Cd (mild: 0.5 mM CdCl ₂ , severer: 1.0 mM CdCl ₂), 3 days	AsA content decreased by 24% due to mild and 42% due to severe stress whereas DHA level enhanced by 79% due mild and 200% due to severe stress; Decreased AsA/DHA ratio in dose-dependent manner; GSH and GSSG content enhanced by 19% and 44%, respectively, due to mild stress, while only GSSG content enhanced due to severe stress by 72%; The ratio of GSH/GSSG declined by 17% due to mild and 43% due to severe stress; Enhanced APX by 15% due to mild and 24% due to severe stress; The activity of MDHAR and DHAR reduced by 12% and 14% due to mild stress whereas 17% and 24%, due to severe stress, respectively; The activity of GR enhanced under mild stress by 16% and lessened under severe stress by 9%	Level of H ₂ O ₂ enhanced by 43% due to mild and 54% due to severe stress. Augmented O ₂ ⁻ generation rate in a dose-dependent manner	[138]
<i>B. juncea</i> L. cv. varuna	Ni, (150 µM NiCl ₂ ·6H ₂ O), 1 week	AsA content decreased by 61% whereas GSH and GSSG content increased by 75% and 151%, respectively; Enhanced function of APX by 60% and GR by 72%; DHAR and MDHAR activities were decreased by 62% and 65%, respectively	Increased H ₂ O ₂ by 3.23-fold	[125]
<i>Pisum sativum</i> L. cv. Corne de Bélier	Pb (500 mg PbCl ₂ kg ⁻¹), 28 days	Increased APX and GR activity	Increased H ₂ O ₂ content	[139]
<i>O. sativa</i> L. cv. BRRI dhan54	Ni (0.25 mM and 0.5 mM NiSO ₄ ·7H ₂ O)	Diminished content of AsA and enhanced content of DHA as well as the lessened ratio of AsA/DHA by 73% and 92% under 0.25 mM and 0.5 mM NiSO ₄ ·7H ₂ O stress; GSH and GSSG level enhanced in a dose-dependent manner. However, the GSH/GSSG ratio reduced only under 0.5 mM NiSO ₄ ·7H ₂ O treatment; Increased APX, MDHAR, DHAR and GR activity by 70%, 61%, 19% and 37% under 0.25 mM NiSO ₄ ·7H ₂ O and 114%, 115%, 31% and 104% under 0.5 mM NiSO ₄ ·7H ₂ O treatment, respectively	Increased H ₂ O ₂ content by 28% and 35% due to 0.25 mM and 0.5 mM NiSO ₄ ·7H ₂ O treatment	[68]
<i>Capsicum annuum</i> L. cv. Semerkand	Cd (0.1 mM CdCl ₂), 3 weeks	Enhanced AsA and GSH content	Increased H ₂ O ₂ content	[140]
<i>C. annuum</i> L. cv. Semerkand	Pb (0.1 mM PbCl ₂), 3 weeks	Enhanced AsA and GSH content	Increased H ₂ O ₂ content	[140]
<i>Zea mays</i> L. cv. Run Nong 35 and Wan Dan 13	Cd (50 mg 3CdSO ₄ ·8H ₂ O kg ⁻¹ soil), 6 months	Decreased GSH content	Increased accumulation of H ₂ O ₂	[141]

6.4. Extreme Temperature

Along with the rise in average global temperature, HT stress has been turned into a topic to be concerned about among environmentalists and researchers worldwide. In general, a 5 °C temperature rise above the optimum temperature of growth is considered to be extreme temperature stress or HT stress or heat shock to any plant species [142,143]. Heat stress causes denaturation of protein and membrane lipids, enzyme inactivation, inhibited protein synthesis, and loss of membrane integrity [144], which results from the disruption of cellular homeostasis through the ROS formed in a mass amount under heat stress [143,145]. Focusing on the role of AsA-GSH pathway to scavenge these ROS, different crop species under different levels of extreme or HT stress have been studied (Table 3).

Khanna-Chopra and Chauhan [146] selected a warmer season to induce HT stress to two different cultivars of wheat (*T. aestivum*), which are Hindi62 (heat-tolerant) and PBW343 (heat-sensitive). They sowed the wheat seeds in mid-January and considered it as heat stress environment, while the control plants were sown in mid-November and considered as the non-stress environment. Data were collected at seven days interval up to 35 days after anthesis (DAA), and the results showed a sharp increase in H₂O₂ content up to 14 days, but then declined. Whereas, MDHAR and DHAR enzymes' activity only increased in Hindi62, but APX and GR activities showed a fluctuating pattern of alteration in both cultivars [146]. Another cereal *Z. mays* when experimented similarly with two different cultivars; LM-11 (heat-sensitive) and CML-32 (heat-tolerant), exposed to 40 °C for 72 h, resulted in higher APX and GR activities in CML-32 roots, while a reduction occurred in the shoot. In LM-11, none of the enzyme activity or AsA content was affected [147]. Higher levels of O₂⁻ production rate and H₂O₂ content were observed in *Ficus concinna* seedlings under 48 h of HT (35 °C and 40 °C) stress condition, where AsA and GSH contents were unaffected at 35 °C, while declining AsA at 40 °C temperature [148]. The activity of APX, MDHAR, DHAR, and GR enzymes increased at 35 °C, but then again reduced at 40 °C to the level of control plants [148]. Under similar heat stress condition (40 °C, 48 h), *V. radiata* seedlings resulted in decreased GSH content and MDHAR-DHAR activities, but higher APX-GR activities [50]. Kiwi fruit (*Actinidia deliciosa*) seedlings, when exposed to 45 °C in an incubator for 8 h, resulted in higher AsA content and enhanced activity of all the AsA-GSH cycle enzymes [143]. Tomato seedlings were studied in two different aspects: short-term heat shock (40 °C, 9 h) [149] and long-term heat stress (38/28 °C day/night, seven days) [150]. In both experiments, the enhancement of O₂⁻ generation rate and H₂O₂ content were recorded, but enzyme (APX and GR) activity was only increased at short-term stress condition [149], while the long-term heat exposure reduced all four enzymes activities and GSH content [150]. Similar enzymatic activity was observed in *Nicotiana tabacum* seedlings after seven days of heat (35 °C) stress [151]. From the above discussion, it can be stated that heat stress prevailing for longer duration is less likely to have the capability to modulate AsA-GSH pathway as compared to short-term heat stress.

Table 3. Role of AsA-GSH in regulating oxidative stress under extreme temperature, flooding, and atmospheric pollutant.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Mitigation	References
<i>Actinidia deliciosa</i>	45 °C, 8 h	Increased content of AsA; Higher activity of APX, MDHAR, DHAR, and GR	Increased H ₂ O ₂ content	[143]
<i>Zea mays</i> L. cv. Ludan No. 8	46 °C, 16 h	Decreased GSH, and GSSG content, but interestingly GSH/(GSH + GSSG) ratio increased; Reduced GR activity	-	[154]
<i>Cinnamomum camphora</i>	40 °C, 2 days	Reduced AsA content with higher DHA content; Increased GSH and GSSG content; Enhanced the activities of APX, MDHAR, DHAR, and GR	Higher content of H ₂ O ₂ and O ₂ ⁻	[166]
<i>S. lycopersicum</i> L. cv. Ailsa Craig	40 °C, 9 h	Higher APX and GR activities by 74% and 45%, respectively	H ₂ O ₂ content increased by 49%	[149]

Table 3. Cont.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Mitigation	References
<i>S. lycopersicum</i> L.cv. Boludo	35 °C, 30 days	Increased the APX, DHAR and GR activities; Reduced the MDHAR activity	Increased H ₂ O ₂ content	[119]
<i>Vicia faba</i> L. cv. C5	42 °C, 48 h	Enhanced the AsA, GSH and GSSG content significantly; The enzymatic activity of APX and GR also enhanced	Extreme accumulation of O ₂ ⁻ and H ₂ O ₂	[167]
<i>V. radiata</i> L. cv. BARI Mung-2	40 °C, 48 h	Decreased 64% in AsA/DHA ratio; GSSG pool increased; Higher APX (42%) and GR (50%) activities but declined activities of MDHAR (17%) and DHAR	Higher H ₂ O ₂ content and O ₂ ⁻ production rate	[50]
<i>Z. mays</i> cv. CML-32 and LM-11	40 °C, 72 h	Increased AsA content in both shoot and root of tolerant (CML-32) one, but unaffected in the susceptible (LM-11) one; Both APX and GR activity increased in roots of CML-32 but reduced in the shoot	Higher H ₂ O ₂ accumulation, especially in shoots	[147]
<i>L. esculentum</i> Mill. cv. Puhong 968	38/28 °C day/night, 7 days	AsA+DHA and DHA increased by 220% and 99% respectively; AsA/DHA ratio decreased by 33%; Higher GSSG (25%), but reduced GSH content (23.4%) and GSH/GSSG ratio (39%); APX, MDHAR, DHAR and GR activities declined	Enhanced O ₂ ⁻ generation rate and H ₂ O ₂ content by 129% and 33% respectively	[150]
<i>Nicotiana tabacum</i> cv. BY-2	35 °C, 7 days	Total GSH and AsA contents rose after 7 days heat stress; Increased MDHAR, DHAR and GR activities up to 72 h	The increasing trend of H ₂ O ₂ generation was observed up to 72 h, and then a sharp decline occurred	[151]
<i>Ficus concinna</i> var. <i>subsessilis</i>	35 °C and 40 °C, 48 h	AsA content reduced at 40 °C but GSH content similar to control at both 35 and 40 °C; DHA content enhanced by 49% at 35 °C and by 70% at 40 °C; APX activity increased by 51% and 30% at 35 °C and 40 °C; Activities of MDHAR, DHAR, and GR increased at 35 °C, but GR activity decreased by 34% at 40 °C	At 35 °C, 103% higher H ₂ O ₂ content and 58% higher O ₂ ⁻ production rate and at 40 °C those were 3.3- and 2.2-fold respectively	[148]
<i>T. aestivum</i> cv. Hindi62 and PBW343	Heat stress environment, Late sown (Mid-January)	Higher activities of MDHAR and DHAR was observed in heat-tolerant (Hindi62) one whereas other enzyme activities seemed mostly to decline with time	The content of H ₂ O ₂ was higher up to 14 DAA compared to non-stressed seedlings	[147]
<i>G. hirsutum</i> cv. Siza	Waterlogged pot for 3 days and 6 days	Increased content of AsA by 20% at 3 days and 30% at 6 days of waterlogging; Lower APX, MDHAR and GR activities	Enhanced O ₂ ⁻ generation rate by 22 and 53% and H ₂ O ₂ content by 10 and 39% at 3 and 6 days of waterlogging, respectively	[154]
<i>Sesamum indicum</i> L. cv. BARI Til-4	Waterlogged pot by 2 cm standing water on the soil surface for 2, 4, 6 and 8 days	Reduced AsA content upto 38%; Enhanced GSH and GSSG content significantly; Increased APX and MDHAR activities; Reduced DHAR activity upto 59%; GR activity decreased upto 23%	Increased H ₂ O ₂ content sharply	[168]
<i>Z. mays</i> cv. Huzum-265 and Huzum-55	Root portions waterlogged for 21 h	Reduced AsA content in both cultivars; Increased APX activity in both cultivars	-	[159]

Table 3. Cont.

Plant Species	Stress Levels	Status of AsA-GSH Component(s)	ROS Mitigation	References
<i>Glycine max</i> L.	Waterlogged pot for 14 days	GSH activity declined sharply in roots but shoots unaffected; Reduced GR activity in shoots but roots unaffected	-	[153]
<i>Trifolium repens</i> L. cv. Rivendel and <i>T. pratense</i> L. cv. Raya	2 cm standing water on the soil surface for 14 days and 21 days	Increased contents of both oxidized and reduced AsA observed in both genotypes	Higher H ₂ O ₂ generation in both genotypes	[158]
<i>V. radiata</i> L. cvs. T-44 and Pusa Baisakhi; and <i>V. luteola</i>	Pot filled with water to 1–2 cm height below the soil level, 8 days	Increased activities of both APX and GR in tolerant genotypes but in susceptible one, activities reduced	Reduced contents of O ₂ ⁻ and H ₂ O ₂ in susceptible (Pusa Baisakhi) cultivar	[156]
<i>O. sativa</i> L. MR219-4, MR219-9 and FR13A	Complete submergence for 4, 8 and 12 days	APX activity declined by 88% in FR13A under 4 days of submergence but decreased about 64 and 83% under 8 and 12 days of submergence; GR activity increased in FR13A and MR219-4 cultivars by 10- and 13-fold respectively after 8 days	-	[152]
<i>Allium fistulosum</i> L. cv. Erhan	Waterlogging (5 cm) at substrate surface for 10 days	Lower APX and GR activities	Increased rate of O ₂ ⁻ generation by 240.4% and 289.8% higher H ₂ O ₂ content	[157]
<i>C. cajan</i> L. genotypes ICPL 84,023 and ICP 7035	Soil surface waterlogged (1–2 cm) for 6 days	Reduced APX and GR activities in susceptible genotype, which was higher in tolerant one	Lower accumulation of H ₂ O ₂ and rate of O ₂ ⁻ generation	[155]
<i>S. melongena</i> L. cv. EG117 and EG203	Flooding with a water level of 5 cm, 72 h	Increased AsA content in susceptible EG117 genotype GSH content in both genotypes; Increased APX activity but decreased GR activity	-	[169]
<i>S. lycopersicum</i> cv. ASVEG and L4422	Flooding with a water level of 5 cm, 72 h	Increase in both AsA and GSH contents; Non-significant changes in APX and GR activities	-	[169]
<i>Lolium perenne</i>	Grown in an area with high air pollution	APX and DHAR activities decreased while MDHAR and GR activities increased	A higher concentration of H ₂ O ₂ in pollens	[162]
<i>Populus deltoides</i> × <i>Populus nigra</i> cvs. Carpaccio and Robusta	O ₃ treatment (120 nmol mol ⁻¹ for 13 h), 17 days	No impact on AsA and GSH contents; DHAR activity decreased while GR and MDHAR activity increased	-	[164]
<i>Fragaria x anansa</i>	High dose of carbon monoxide (CO) nitroxide (NO _x) and sulfur dioxide (SO ₂)	The activity of both APX and GR increased upto medium dose but reduced under high dose	H ₂ O ₂ content as well as O ₂ ⁻ generation rate increased	[165]
<i>O. sativa</i> L. cvs. SY63 and WXJ14	Continuous O ₃ exposure for up to 79 days	Both AsA and GSH contents are more likely to decrease; APX, MDHAR, DHAR, and GR activity increased up to 70 days of O ₃ exposure	Both O ₂ ⁻ generation rate and H ₂ O ₂ contents increased	[163]
<i>Prosopis juliflora</i>	Grown in the polluted industrial region	The content of AsA and APX activity increased under polluted environment	-	[161]
<i>Erythrina orientalis</i>	Grown in a polluted industrial area	Increased activities of both APX and GR enzymes recorded	-	[160]
<i>T. aestivum</i> L. cv. BARI Gom-26	Acidic pH (4.5) of growing media	Increased AsA and GSH content; Improved redox balance of GSH/GSSG; Increased activity of APX, MDHAR, DHR, and GR	H ₂ O ₂ contents increased by 209%	[170]

6.5. Flooding

Changes in global climate result in the frequent or unexpected occurrence of heavy rainfall in different regions of the globe, which causes a sudden flood and disrupts the normal ecosystem [2]. Such changes in the ecosystem may cause the extinction of plants species and imbalance in the natural environment [2]. Flooding induced production of ROS and subsequent cellular damage has been authenticated in many studies so far [152–154]. Following are the discussion regarding crop species facing flooding stresses and modulation of their AsA-GSH pathway by flooding stress (Table 3).

Pigeon pea (*C. cajan*) seedlings that are exposed to waterlogged condition for six days revealed that tolerant cultivar could increase APX and GR activities, but a susceptible one cannot [155]. They also observed that, unlike other cases, waterlogging caused a lower accumulation of H₂O₂ and O₂⁻ [155]. In another experiment with *V. radiata*, Sairam et al. [156] showed that waterlogging similarly reduced the H₂O₂ and O₂⁻ production rate in susceptible cultivar, while the tolerant ones remained unaffected. However, both APX and GR enzymes' activity increased in tolerant genotypes, while the susceptible one got reduced [156]. The enhanced production rate of O₂⁻ and H₂O₂ content under flooding stress have been reported in cotton [154], Welsh onion [157], and clover [158] plants. Cotton (*G. hirsutum* cv. Siza) plants after three and six days of flood exposure raised the AsA content, but reduced the activity of APX, MDHAR, and GR [154]. A similar reduction in APX and GR enzymes activities was also recorded in Welsh onion (*Allium fistulosum* L.) after 10 days of waterlogging stress [157]. When *Z. mays* seedlings were waterlogged for 21 h at their root portions, they resulted in reduced AsA content and increased APX activity [159]. On the other hand, under long duration (14 days) flooding stress, *Glycine max* L. plants showed a reduction of GSH activity in roots and GR activity in the shoot, but the GSH in shoot and GR in root were not affected [153]. In case of complete submergence of *O. sativa* L. plants for two, four, or eight days, elevated levels GR enzyme activity was recorded, while APX enzyme activity increased only in tolerant cultivar [152]. Accordingly, the discussion reveals that the impact of flooding stress on AsA-GSH pathway varies depending upon the plant species and duration.

6.6. Atmospheric Pollutants

Atmospheric pollutants are the substances that are assembled in the air to a level or magnitude that is dangerous for living beings. Plants that are grown under different levels of atmospheric pollution have shown their oxidative stress responses and AsA-GSH pathway regulation in different manners (Table 3).

Erythrina orientalis plants were grown in three different locations of Philippines: La Mesa (a non-polluted area); and, Makati and Quezon (highly air-polluted cities). The results revealed that plants grown in the non-polluted area had lower activities of APX and GR as compared to the ones grown in highly polluted areas [160]. A similar increase in APX activity along with higher AsA content was recorded in *Prosopis juliflora* plants grown under polluted industrial region [161]. In a recent experiment, Lucas et al. [162] studied *Lolium perenne* plants that were grown under two different areas of Spain, Madrid, and Ciudad Real, where Madrid was considered to be more polluted than Ciudad Real. The findings indicated that the pollens of *L. perenne* accumulated higher concentration of H₂O₂ and in shoots APX and DHAR activity declined, but the activity of MDHAR and GR increased in the shoot of *L. perenne* plants that were grown in Madrid [162]. When rice seedlings were exposed to continuous O₃ treatment, the results showed a remarkable increase in both O₂⁻ generation rate and H₂O₂ content. In addition, contents of AsA and GSH reduced, while APX, MDHAR, DHAR, and GR activity increased upto 70 days of O₃ exposure in SY63 cultivar and upto 79 days of O₃ exposure in WXJ14 cultivar [163]. Ascorbate and GSH contents were not affected by O₃ exposure in the *Populus* seedlings, but DHAR activity was lower, while the activity of GR and MDHAR was higher after 17 days of O₃ treatment [164]. Young strawberry (*Fragaria x anansa*) seedlings were exposed to three different levels of CO, NO_x, and SO₂, which are as follows: CO @ 133, 267, and 533 ppm, NO_x and SO₂ @ 25, 50, and 199 ppm corresponding to low, medium, and high dose, respectively. As a result of exposure to these atmospheric pollutants, H₂O₂ content as well as O₂⁻ generation rate increased.

However, at low and medium doses of their exposure APX and GR activity increased, while at a high dose that decreased [165]. All sorts of atmospheric pollutants have a remarkable effect on AsA-GSH pathway, but further studies are required to demonstrate that those pollutants completely induced the modification of the AsA-GSH pathway.

6.7. Other Stress

Conklin et al. confirmed the positive role of AsA in protecting plants from ultraviolet (UV) radiation [159], where they found that Vit-C deficient mutant of *A. thaliana* was suffered by stress-induced damages than that of wild type. AsA-deficient mutants also showed sensitivity to O₃ stress due to a lower biosynthesis of AsA [171]. Gao and Zhang [172] reported that vitc1 mutants of *A. thaliana* showed physiological disorders and greater oxidative damages than the wild type, which was due to lower activities of antioxidant enzymes. Mutant plants also showed lower GSH/GSSG and higher DHA/(AsA+DHA) ratio than the wild type. Singh et al. [173] observed a decrease in AsA-GSH cycle enzymes in UV-exposed plants, which in turn affected the plants with oxidative stress. Similar to higher plants, marine macroalga *Ulva fasciata* also showed a positive correlation between enhanced the functions of AsA-GSH cycle and better tolerance of plants to UV radiation [174]. In their study, scavenging of H₂O₂ was regulated by AsA-GSH cycle components, especially APX and GR. Noshi et al. [175] reported that AsA-GSH redox pool provided better protection of *Arabidopsis* from high-light mediated oxidative stress, which was mainly attained due to the higher activities of DHAR. However, both AsA and GSH were found to be responsible for conferring high light (HL) stress [175]. Later, Zheng et al. [176] that susceptibility of *Arabidopsis* mutant was to HL stress were related to the deficiency of AsA and GSH. When AsA deficient *A. thaliana* mutant (vtc2-1) exposed to HL, they generated a high level of H₂O₂ (an oxidative stress marker) than the wild type, which was highly and negatively correlated with the total AsA content. The lack of AsA also resulted in lower chlorophyll (chl) content, chl fluorescence parameters, and PSII photochemistry [176]. Recently, Choudhury et al. [177] studied the metabolomics of *A. thaliana* grown under HL and found that the increased biosynthesis of GSH supports the photochemistry that supports *Arabidopsis* better survival under HL stress.

The pivotal role of the AsA-GSH cycle was also observed in low pH stress also. Bhuyan et al. [170] tested five spring wheat cultivars at different levels of low pH stress. Their observation exhibited that low-pH stress resulted in elevated O₂⁻ and H₂O₂ generation. A decrease in AsA content with increased DHA content was observed, although the APX activity decreased. Increased MDHAR activity was observed, but the ratio of AsA/DHA was not increased. Decreased GSH content and increased GSSG content were found where DHAR and GR activity decreased, resulting in a drop of the GSH/GSSG ratio.

7. Exogenous Use of AsA and GSH in Conferring Abiotic Stress Tolerance

While considering the vital role of both AsA and GSH and their redox researches have been trying to explore the possibilities of using exogenous AsA and GSH in protective plants from abiotic stress. However, the effects are not straightforward due to their species and dose dependency. In the next sections, we provided a summary of the recent results on plant abiotic stress tolerance while applying exogenous AsA and GSH.

7.1. Exogenous AsA

As a non-enzymatic antioxidant, AsA is vital for plant defense mechanism by involving in stress perception and subsequent signaling, and therefore plant responses [178]. Besides its regenerative nature, AsA is also able to donate electrons with and/or without the help of enzymes, and thus significantly detoxifies ROS [179]. Thus, exogenous AsA application is the most prominent for enhancing plant tolerance due to its efficient protection against lipids and proteins oxidation under abiotic stresses [180].

Ascorbate can be exogenously applied as a foliage application, seed treatment, and co-treatment for the alleviation of stress-induced damages [181]. Many researchers reported about the supplemental

AsA-mediated antioxidant defense regulation in various plant species under different stressors, such as salt stress [182], drought [183], extreme temperature [184], ozone [185], and heavy metal stress [186].

Supplemental AsA application effectively lowered the oxidative stress in salt-stressed *Phaseolus vulgaris*, as indicated by the reduction of malondialdehyde (MDA) and ROS accumulation through activating their immune systems related with up-regulation of SOD, CAT and GR activities [187]. The AsA recovered salinity-induced oxidative damage in *Caralluma tuberculata* by lowering the activity of APX, POD, CAT, and GR, which were increased upon saline toxicity [188]. Exogenous AsA-induced plant tolerance, especially on the AsA-GSH pathway, is cultivar dependent [189]. Hence, they used both salt-tolerant *O. sativa* cv. Pokkali and salt-sensitive *O. sativa* cv. Peta to exogenously apply AsA as co-treatment with salinity and found a reduction of H₂O₂ generation in both cultivars. Here, AsA enhanced endogenous AsA and GSH, along with higher SOD, APX, and GR activities in salt-stressed both cultivars in line with lowered ROS and MDA production. However, Pokkali showed more prominent responses of salt tolerance than Peta. Finally, Wang et al. [189] suggested that exogenous AsA differentially increased the salt tolerance mechanism, and thus lessened salt-induced ROS in two rice cultivars. Exogenous AsA enhanced the salinity tolerance of *Z. mays* through protecting oxidative stress with stimulation of plant antioxidant defense [190]. In this study, AsA was used as seed priming against 100 mM NaCl, and AsA restored the salt-induced membrane damage. Hence, external AsA improved the non-enzymatic antioxidants, including Pro, AsA, and GSH accumulation, where SOD and GPX activities increased. Rady and Hemida [190] found lowered CAT activity in AsA treated seedlings under salt stress, which pointed out the AsA-induced decline of H₂O₂ generation.

Plants get relief from drought stress by exogenous AsA application, which was reported by previous researchers [114,191]. Alam et al. [114] studied the AsA induced attenuation of oxidative stress in *B. napus*, *B. campestris*, and *B. juncea* under 15% PEG, indicated by decreasing lipoxygenase (LOX) activity, H₂O₂, and MDA contents. This AsA mediated oxidative stress mitigation was described by AsA caused the strengthening of plant antioxidant defense mechanisms. Hence, exogenous AsA not only responsible for modulating AsA-GSH cycle, but also increased other enzymatic antioxidants activities, such as CAT, GPX, Gly I, and Gly II in all plant species, except GST, which was only increased in *B. napus*. Exogenous AsA mitigated the PEG-induced oxidative stress in *Z. mays* where AsA used as co-treatment, later endogenous AsA content increased, followed by scavenging surplus H₂O₂ generation and a reduction of lipid peroxidation [191]. The higher transcript levels of SOD, CAT, APX, GR, MDHAR, and DHAR were induced in tall fescue by AsA application under PEG-induced water crisis, in respect with the only stressed condition [192]. Subsequently, Xu et al. [192] recommended AsA as a phytoprotectant to improved plants tolerances upon drought stress.

Exogenously applied 50 µM AsA decreased high temperature (HT, 45/35°C)-induced elevated H₂O₂ and MDA contents with lowered electrolyte leakage (EL) in *V. radiata* [184]. Hence, supplemental AsA altered the heat-induced lowered SOD, CAT, APX, and GR activities with increasing endogenous AsA and GSH contents. The AsA also enhanced the antioxidant capacity of tomato to cope with low-temperature stress [193]. The foliar application of AsA decreased the EL and MDA content in *T. aestivum* seedlings when exposed to the combined stress of herbicide and low temperature (−2 °C) [194]. This AsA-induced lowered oxidative damage might be because of ROS scavenging under stress indicated by AsA mediated lowered O₂[−] and H₂O₂, which were attributed by increasing POD, APX, and GR activities.

Seed priming with AsA also increased plant tolerance to metal stress. Hence seed priming with AsA of *A. esculentus* showed the alleviation of Pb-induced oxidative stress that was confirmed by lowered H₂O₂ and MDA contents [135]. This AsA-induced alleviation of oxidative stress supported by exogenous AsA mediated increment of endogenous AsA contents, as well as upregulation of SOD, POD, and CAT activities in Pb-stressed *A. esculentus*. The AsA priming also increased the anthocyanins content in Pb-exposed seedlings, which again enhanced the metal tolerance by checking ROS production. Previously, exogenous foliar application of AsA on rice seedlings increased AsA, and

GSH contents, while enhanced both AsA/DHA and GSH/GSSG redox status, along with higher APX and GR activities under Cd stress [195].

Alamri et al. [196] investigated the potentiality of exogenous AsA to remove the metal-induced oxidative stress in *T. aestivum*. They observed that AsA suppressed the higher content of MDA and H₂O₂ in Pb exposed seedlings by improving the antioxidant enzymes activities, including SOD, CAT, and GR. Thus, AsA-mediated higher activity of enzymatic antioxidants could be responsible for the lowered membrane damage indicated by EL as well as Pb tolerance.

The AsA supplementation also showed its effective role in the mitigation of Cd-induced oxidative stress. Zhang et al. [141] reported that AsA application, as a foliar spray, could become a potent tool to alleviate Cd toxicity in *Z. mays*. They used 0.1, 0.3, and 0.5 mM of AsA in against 3.36 mM Cd contamination, while they observed a remarkable gradual reduction of both H₂O₂ and MDA contents under stressed conditions with increasing AsA levels. Foliar AsA application improved endogenous GSH along with the augmentation of SOD, POD, CAT, and GR activities, which are in line with AsA-induced lessening of oxidative stress in Cd-exposed *Z. mays*.

Thus, exogenous AsA application scavenges ROS in the plant under abiotic stresses, and then protects cell membrane stability. Therefore, reduced MDA content and EL were reported with AsA application as a sign of AsA-induced alleviation of oxidative damages. Accordingly, such exogenous AsA-induced strengthening of plant antioxidant defense, along with lessening oxidative stress, explained the potential of AsA for conferring abiotic stresses.

7.2. Exogenous GSH

At endogenous level, being an active participant of AsA-GSH cycle, GSH scavenges H₂O₂ in enzyme-dependent pathways; GSH is a substrate for GPX; GSH detoxifies lipid hydroperoxides together with GSTs; and, GSH/GSSG induces signals for abiotic stress adaptation [6]. Moreover, several research studies reported that the exogenous application of GSH proved to confirm the additional beneficial effects for enhanced the antioxidant defense system and abiotic stress tolerance.

After exogenous GSH pretreatment, mung bean plants were imposed with HT (42 °C), and beneficial effects were noticed. It enhanced chl and leaf RWC; increased cellular GSH content and GSH/GSSG lowering GSSG content; amplified APX, MDHAR, DHAR, GR, GPX, GST, and CAT activities; exogenous GSH pretreatment upheld the activity of Gly I and Gly II of MG detoxification system. The upregulation of both antioxidant and glyoxalase system ensured the HT tolerance. Meanwhile, GSH supplementation with HT decreased H₂O₂, O₂⁻, MDA level, the activity of LOX, and MG content [95]. Increased temperature (35 °C) in root-zone variably affected physiological processes, growth, and Calvin cycle, which mediated inconsistency in antioxidant components; HT also affected antioxidant enzymes' gene expression of *Cucumis sativus* L. seedlings. HT-induced reduction of GSH content, the ratio of GSH/GSSG, photosynthetic pigments level, photosynthesis, and changes of linked gene expression were evident. HT also augmented soluble protein, proline (Pro), O₂⁻ generation and MDA level, expression of genes, and antioxidant enzymes functioning. The application of supplemental GSH with HT upheld soluble protein, Pro, antioxidant enzymes activity, and its linked gene expression, as well as inhibited O₂⁻ generation and lipid peroxidation, than to HT treatment without GSH [197].

Exogenous GSH improved AsA, and GSH contents, GSH/GSSG, APX, MDHAR, DHAR, and GPX activity of antioxidant system in drought exaggerated mung bean seedlings, which helped to relieve the adverse effect reducing the ROS including H₂O₂ and O₂⁻, both in content and visually in the leaf spots of which were visualized through histochemical detection. Exogenous GSH also decreased the LOX activity, which caused the oxidation of lipid. Exogenous GSH also up-regulated the activity of Gly I and Gly II, therefore, reduce the toxic consequence of MG, and MG-induced oxidative damage [111].

Exogenous GSH (1.0 mM) positively regulated an antioxidant system in wheat plants facing lead (Pb) stress. The imposition of Pb diminished growth, the relative water content of leaf, and chl *a* and *b* content; amplified Pro level, H₂O₂, and O₂⁻ generation, and lipid peroxidation. Glutathione supplementation with Pb stress improved the AsA and GSH contents, GSH/GSSG, activities of MDHAR,

DHAR, GR, SOD, CAT, and GPX, and decreased oxidative damage. The decline of H_2O_2 and O_2^- generation and membrane lipid peroxidation was clear evidence, together with an increased level of Pro and chl, which contributed overall tolerance to Pb toxicity [35]. Pretreatment with 100 μ M GSH with 50 μ M Cd reversed growth reduction and concealed Cd-provoked MDA buildup. In contrast to Cd compelled plants, GSH pretreatment reversed photosynthetic pigment destruction, downregulated Cd accumulation in root and shoot. Exogenous GSH considerably increased the functioning of POD and SOD. In contrast to the Cd affected plants, exogenous GSH pretreated plants extensively reassured decrease in Cu or augmented in Fe levels, which were due to Cd [198]. Exogenous GSH and Cys were applied on lead (100 and 500 $mg\ L^{-1}$) affected *Iris lactea* var. chinensis, growth, accumulation of Pb, and nonprotein thiol (NPT) accumulation pattern were observed. The addition of GSH improved GSH biosynthesis in root and shoot. Endogenous shoot level Cys was recorded for exogenous Cys addition. Exogenous GSH application, together with buthionine sulfoximine (BSO) addition, regulated enzymes involved in GSH biosynthesis. This GSH played an imperative function in Pb accumulation and adaptation to this stress [199]. Exogenous GSH application improved the germination and growth of *Arabidopsis*, tobacco, and pepper under mercury (Hg) stress. Exogenous GSH also conferred Cd, Cu, and Zn stress tolerance. Exogenous GSH downregulated H_2O_2 and O_2^- generation and MDA content, whereas upregulated chl level under Hg. Outstandingly, exogenous GSH reduced Hg accumulation in *Arabidopsis*. GSH showed high binding empathy to Hg, as compared to Cd, Cu, or Zn [200].

Salt-tolerant Pokkali and sensitive cultivar Peta of rice were scrutinized for the role of exogenous GSH on them. Exogenous GSH increased the activity of SOD, APX, and GR, the amount of AsA and GSH, and reversed chloroplasts' H_2O_2 and MDA accumulation in either cultivar affected by salinity (200 mM NaCl). However, tolerance was prominent in cv. Pokkali [163]. The supplementation of GSH inverted the pessimistic properties of salinity stressed (NaCl 100 mM) tomato plants improving the transcript levels and activities of enzymes that are linked to GSH biosynthesis and metabolism. The biosynthesis-related enzymes were gamma-glutamylcysteine synthetase (γ -ECS), glutathione synthetase (GS), whereas, others were GST, GPX, and GR. Exogenous supplemental GSH helped to upregulate the activity of SOD, peroxidase (POD), CAT, APX, MDHAR, DHAR and GR, GSH level, and GSH/GSSG in salt-stressed plants [201]. Externally applied GSH lessened the oxidative damage in different soybean genotypes via reducing H_2O_2 and MDA level, which were produced due to salinity. Glutathione supplementation minimizing the oxidative damage further contributed in yield attributes, and yield performance, which was seeds $plant^{-1}$ and pods $plant^{-1}$, 100-seed weight and yield [202]. Defensive function of supplemental GSH (1 mM GSH) was examined for salt (200 mM NaCl) stressed mung bean. Mung bean plants when imposed with exogenous GSH and NaCl elevated AsA and GSH levels, GSH/GSSG, enhanced APX, MDHAR, DHAR, GR, SOD, CAT, GPX, and GST activities were recorded. Exogenous supplemental GSH also augmented the activity of Gly I and Gly II under salinity. Enhanced antioxidant and glyoxalase system components that resulted from the effect of exogenous GSH application had several beneficial effects. MDA, H_2O_2 , and MG, O_2^- production turned down, and leaf RWC and chl level raise; all of which made mung bean seedlings capable to perform better under saline growing media [95]. Glutathione was exogenously applied on tomato plants affected by salinity. Additional of GSH decreased oxidative stress. The reason behind this was revealed as redistribution of light energy in PSII, higher cellular GSH, GSH/GSSH ratio and activities of SOD, CAT and APX, MDHAR, DHAR, GR, and GRx. Glutathione supplementation revolutionized growth inhibition, Na^+ and Cl^- ions balance, and Na^+/K^+ . Chloroplast, as well as stomatal function related to photosynthetic performance, were documented to improve after the application of GSH with salinity [203].

8. Interaction of Other Pathways with AsA-GSH Pathways in Regulating ROS Metabolism

Beside oxidative stress mitigation, the AsA-GSH cycle also interacts other pathways to reduce ROS and oxidative stress. Therefore, in this section, we will discuss the potentiality of AsA-GSH

pathway components and their interaction with other pathways to modulate the ROS metabolism in plants.

8.1. Interaction of AsA-GSH Cycle with NO Metabolic Pathway

Although AsA-GSH cycle protects cellular components from oxidative damage, its components, especially the proteins (APX, MDHAR, DHAR, and GR), are also vulnerable to the oxidative damage, which can modify their activity, hence breaking down the antioxidant defense. In plants, nitric oxide can be produced from several biochemical pathways, both oxidative and reductive [35]. In the GSH pool, the reduced form of GSH can interact with NO and produce GSNO, which is further catalyzed by the action of GSNO reductase (GSNOR) and release NO and GSSG, and maintains the equilibrium of NO and nitrosothiols, as well as balance the redox state in the cell [204]. Moreover, S-nitrosylation could modify protein interactions, thus tinkering the antioxidant response [205]. Reports suggest that all of the proteins of the AsA-GSH cycle are influenced through S-nitrosylation and/or nitration, which are accomplished from the interaction with NO. Among the AsA-GSH cycle, enzymes APX is the most studied, which is directly influenced by NO metabolism. For example, the inactivation of APX1 is caused due to the oxidation of Cys32 [205]. Contrary, nitrosylation of APX1 active-site Cys32 increases its activity and this post translation modification (PTM) is performed during salinity stress, which increases oxidative stress as well as S-nitrosothiols [206]. Among the other enzymes, MDHAR is negatively modulated by nitration, which causes enzymatic inactivity by altering the position of the cofactor binding site [206], and hence disturb the AsA recycling process. Although information is available on the nitration and activity modulation of DHAR proteins [207,208], but the involvement of Try in this process, as well as the structural alteration impact of the enzyme, is still unclear. Moreover, GR is also targeted for nitration, which is reported to inhibit its activity in a mammalian cell, but the chloroplastic and cytosolic GR of pea is not affected by nitration [206].

8.2. Signaling Role of AsA-GSH Cycle Components and Interaction with Other Pathways

Ascorbate serves as the co-factor for redox enzymes, as well as a precursor for several biosynthetic pathways. In addition, AsA is an important reducing agent for Fe, Cu, and Mn, thus act as a pro-oxidant controlling toxic OH[•] production from the Haber-Weiss and Fenton reaction [209]. Besides, its role as an antioxidant is the most important part of detoxifying ROS. As a pro-oxidant AsA regenerate α -tocopherol. Moreover, AsA also works in the photo-protection that is mediated by the xanthophylls cycle, where violaxanthin de-epoxidase use AsA as a co-factor [210]. Moreover, AsA is employed as the substrate for organic acid (oxalate and tartrate) biosynthesis (Figure 2). Rapid cell expansion is correlated with AO activity, which oxidized AsA [211]. During cell expansion, Pro residues present in the glycoproteins of cell wall undergo hydroxylation where prolyl hydroxylase use AsA as a cofactor [210]. Furthermore, AsA can potentially upregulate cytosolic free Ca²⁺ via anion channels and play a signaling role [212]. More than this breakdown of AsA to DHA by APX or AO creates an electrochemical gradient over the plasma membrane, which also has a signaling role.

Under abiotic stress conditions, GSH triggers adaptation or PCD by intercellular signaling [213]. Glutathionylation of protein Cys residues suggests its redox signaling role, which alters the transcription of proteins [214]. In *Arabidopsis*, stomatal movement induced from methyl jasmonate (MeJA) is regulated by intracellular GSH [215]. In tobacco, both GSH and GSSG application induce Ca²⁺ signaling as well as the expression of a specific gene, which supports the involvement of GSH with signal pathways that connect the Ca²⁺-dependent protein kinase [216]. The protein family peroxiredoxins (Prxs) are also GSH dependent and catalyze the reduction of H₂O₂ [217]. The GSSG can be exchanged with sulfhydryl groups of proteins and produce protein-GSH disulfide conjugates, which has a long half-life and plays a vital role in cellular signaling [218]. Moreover, GSH influences translation, and PTM of proteins, modulation of metabolism, and gene expression [219]. Hence, the mechanistic process of GSH signaling role should be focused on in future studies.

8.3. AsA-GSH Cycle Interaction with Phytohormone Biosynthesis Pathways

Ascorbate regulates phytohormone biosynthesis; hence, modulating plant development [220]. Ascorbate shows activity, where cell developments are affected by hormonal signaling and modulate effective signaling processes [221]. The abscisic acid (ABA) involvement in stagnating growth and metabolism suggests the crucial role of AsA sensing for plant survival [222]. In addition, a number of dioxygenases that are directly related to hormonal biosynthesis require AsA as a cofactor [223]. Moreover, a low AsA induces PR proteins, but do not alter antioxidative enzymes. Thus, AsA acts as a “crosstalking” signal, where ABA acts as an important intermediary signal induces PR1 proteins in many plants. Hence, phytohormone signaling arises the AsA-dependent PR genes regulation [224]. On the other hand, 1-aminocyclopropane-1-carboxylate (ACC) synthase (ACS) and ACC oxidase (ACO) genes encoding ethylene biosynthetic enzymes is induced by GSH. Further, GSH increases serine acetyl transferase (SAT) level and confers Ni toxicity tolerance [225]. In rice, the overexpression of SA metabolism genes gave raise to both SA and GSH content under oxidative stress [226]. Therefore, GSH triggers phytohormones, and vice versa, along with other signaling genes [227].

8.4. Interaction of AsA-GSH Pathway with Glyoxalase Pathway

There is an intimate relationship between AsA-GSH cycle and the glyoxalase pathway through GSH, where it plays a vital role in the detoxification of MG. Methylglyoxal is a respiratory byproduct and produced usually in plants and detoxified by the glyoxalase system. However, MG is overproduced under stress, which causes toxicity [51]. Moreover, MG can disfunctionate antioxidant enzymes [1]. In the MG detoxification process, Gly I (EC 4.4.1.5) and Gly II (EC 3.1.2.6) work simultaneously to detoxify MG (Discussed in Section 6). In this pathway, Gly I uses GSH as a cofactor and conjugates MG with GSH to form S-D-lactoylglutathione (SLG), Gly II, and then produce D-lactate breaking SLG, and regenerate GSH (Figure 9) [1], thus playing important interaction with glyoxalase system.

8.5. Interaction AsA-GSH Pathway with Xenobiotics Detoxification Pathways

Xenobiotic detoxification involves the conjugation of toxic xenobiotics with GSH, which are further transferred to the vacuole by using ATP driven tonoplast transporter. This detoxification enables secondary metabolites biosynthesis as well as storing in the vacuole, such as anthocyanin. Plants are having GSH-dependent enzyme, GST, which detoxify herbicides by conjugating it with GSH. Therefore, the glutathionylated metabolites are imported to vacuolar by ABC (ATP-binding cassette) transporters. However, the GST mainly functions in catalyzing natural products that were observed with xenobiotics and, similar to those, catalyzes alternative GSH-dependent biotransformation reactions and binds and carries phytochemicals between cellular compartments [228].

8.6. AsA-GSH Cycle Interaction with Metal Chelation Process

Maintaining lower metal/metalloid(s) level inside the cell involves metal sequestration by low molecular weight thiols, for instance, metallothioneins (MTs) and phytochelatins (PCs). The two important enzymes involved in this process, glutaredoxin (GRx) and thioredoxin (TRx), are GSH dependent and neutralize H₂O₂ or controls of protein thiols [229]. On the other hand, PCs are another important chelating agent containing that thiol group that are upregulated by different metal/metalloid(s) [32]. The basic component for this PC is GSH. The biosynthesis of PCs is accomplished by the enzyme PC synthase (PCS), which requires GSH as a substrate. The enzyme is crucial for metal detoxification, metal homeostasis, and stress tolerance [137].

9. Genetic Manipulation of AsA-GSH Pathway and Its Role in Abiotic Stress Tolerance

The regulation of AsA and GSH pool plays an important role in mitigating oxidative stress in plants. To attain this, the regulation of the enzymes that are related to the AsA-GSH pathway is vital. There are many plant studies that considered the genetic manipulation of AsA-GSH pathway. These

studied revealed that the overexpression of AsA-GSH pathway enzymes provided the plants better protection against oxidative stress under various environmental adversities (Table 4). Transgenic tobacco plants overexpressing *PcAPX* showed enhanced tolerance to salt and drought [230]. Transgenic plants exhibited a 347% increase in APX activities under drought stress, as compared to control, which resulted in a remarkable decrease in H₂O₂ content (136%) than that of wild type (309%). The ascorbate content was also higher 63%) when compared to wild type (42%). Similar results were also observed in the case of salt stress [230]. Chin et al. [231] found that transgenic *Arabidopsis* overexpressing *OgCytAPX1* scavenged ROS effectively and showed enhanced tolerance to salt and heat. The overexpression of *Malpighia glabraMDHAR* gene resulted in a higher biosynthesis of AsA, which provided tobacco plants tolerance to salt stress [232]. Shin et al. [233] observed that the coexpression of *B. rapaBrMDHAR* and *BrDHAR* genes provided a remarkable improvement of oxidative stress in *A. thaliana*. The overexpression of *BrMDHAR* and *BrDHAR* showed enhanced MDHAR and DHAR activities and higher AsA/DHA ration. These plants also provided better radical scavenging capacity, which resulted in lower H₂O₂ content. Yin et al. [234] found that *Arabidopsis* plants overexpressing the gene *AtGR1* conferred Al stress tolerance by reducing reactive carboxyl species, which was mainly due to higher GSH level and GR activity. The plant that overexpressed *AtGR1* also maintained the activity of H₂O₂-scavenging enzymes. For instance, GPX and APX activities in Al-treated plants were decreased by 21 and 46%, respectively, but the wild-type plants only showed 8 and 30% decreases in such activities [234].

Modulating several NAC genes [NAC domain consists of three different genes; *NAM* (no apical meristem)-*ATAF* (*Arabidopsis* transcription activation factor)-*CUC* (cup-shaped cotyledon)] are also an efficient way to transform the AsA-GSH cycle, consequently enhancing stress tolerance. The overexpression of wheat *TaNAC2* in *Arabidopsis* lines showed tolerance against freezing, salt, and drought stress by modulating the AsA-GSH cycle [235]. Moreover, ectopic expression of *SINAC2* conferred both salt (200 mM) and drought (20% PEG) tolerance up to 10 days in transgenic *Arabidopsis* lines, which is correlated with the lower accumulation of ROS. In addition, the transcriptomic abundance GSH metabolizing genes was also observed in transgenic lines, leading to increased GSH synthesis and lesser oxidative damage [236].

Table 4. Overexpression of genes related to AsA-GSH pathway and their role in ROS scavenging.

Enzymes	Gene	Gene Source	Target Plants	Regulatory Effects	References
APX	<i>OsAPX2</i>	Rice	Alfalfa	Decreased H ₂ O ₂ and MDA contents; Enhanced salt tolerance	[237]
APX	<i>PcAPX</i>	<i>Populus tomentosa</i>	Tobacco	Increased AsA content; NADP ⁺ /NADPH ratio; Decreased lipid peroxidation and H ₂ O ₂ contents; Increased salt and drought stress tolerance	[230]
APX	<i>CaAPX</i>	<i>Camellia azalea</i>	Tobacco	Enhanced MDHAR and DHAR activity; Regulated ROS generation; Enhanced cold and HT tolerances	[238]
APX	<i>ScAPX6</i>	Sugarcane	<i>Nicotiana benthamiana</i>	Hormonal regulation; Lower ROS generation; Enhanced tolerance to Cu stress	[239]
APX	<i>OgCytAPX1</i>	<i>Oncidium</i>	<i>Arabidopsis</i>	Efficient ROS scavenging capacity; Maintained redox homeostasis and increased GPX activities which resulted in lower H ₂ O ₂	[240]
MDHAR	<i>Am-MDAR</i>	<i>Avicennia marina</i>	<i>A. marina</i>	Increase in MDHAR and GR activity; Increased AsA content; Decreased lipid peroxidation; Improved salt-induced oxidative stress	[241]

Table 4. Cont.

Enzymes	Gene	Gene Source	Target Plants	Regulatory Effects	References
MDHAR	<i>MgMDHAR</i>	<i>Malpighia glabra</i>	Tobacco	DHAR activity increased by 1.8–2.1 fold; AsA/DHA increased by 81–84%; Lipid peroxidation decreased by 41–62%	[232]
MDHAR	<i>BrMDHAR</i>	Rapeseed	<i>Arabidopsis</i>	AsA/DHA ratio increased by 7%; Decrease of H ₂ O ₂ content by 55%; Radical scavenging was 16% higher	[233]
DHAR	<i>DHAR</i>	Human	Tobacco	No changes in AsA content; Enhanced tolerance to chilling and salt stress	[242]
DHAR	<i>DHAR1</i>	Rice	<i>Arabidopsis</i>	AsA content increased by 1.4 fold; Enhanced tolerance to salt	[243]
DHAR	<i>DHAR</i>	Human	Tobacco	Increase in AsA content; Increased the activities of SOD and APX; Enhanced salt tolerance	[244]
DHAR	<i>DHAR</i>	<i>Arabidopsis</i>	<i>Arabidopsis</i>	AsA content increased by 2.0–4.25 fold; Enhanced tolerance to HT stress	[245]
DHAR	<i>DHAR</i>	Rice	Tobacco	AsA content increased by 1.6-fold; Improved tolerance to salinity and chilling	[246]
DHAR	<i>BrDHAR</i>	Rapeseed	<i>Arabidopsis</i>	AsA/DHA ratio increased by 11%; H ₂ O ₂ content decreased by 56%; Radical scavenging was 16% higher	[233]
GR	<i>AtGR1</i>	<i>Arabidopsis</i>	<i>Arabidopsis</i>	Enhanced GSH content and GR activity (0.4- to 1.0-fold higher); H ₂ O ₂ content reduced by 26%	[234]

10. Conclusions and Outlook

Ascorbate and GSH have roles in decreasing oxidative stress, and it has been reported in numbers of research findings. Most of the research findings reported about their roles in antioxidant defense system for scavenging ROS. However, exogenous GSH related research on antioxidant defense system needs further confirmation at the genetic and molecular level. Moreover, without the commonly known ROS, like H₂O₂, OH, O₂⁻, etc. some other oxidative stress-inducing agents, like reactive nitrogen species, MG, etc., should be brought under consideration for research. How GSH can affect the generation of other kinds of oxidative stress-inducing damage. Research that is related to exogenous AsA or GSH-induced GST activity concerning xenobiotic detoxification is rare. The regulation of tocopherol by AsA or GSH can be an interesting area of research. For the reduction of metal-induced oxidative stress protection, GSH plays a vital role by producing PCs and inducing vacuolar sequestration. The credible function of AsA-GSH cycle in this area is so far to explicate. The GSH/GSSG redox is a well-reported term when discussing stress-induced oxidative damage and signal transduction process towards adaptation though the process is not well revealed. Interaction between and among the AsA-GSH cycle components and the hormones, other signaling molecules or any other molecules in oxidative stress, redox regulation, or plant adaptation process is not well understood. It is well known that chloroplast, its photosystem, and Calvin cycle activity or photosynthesis process is the maximum contributor of most of the ROS and oxidative stress under any abiotic stress condition. Several research findings reported about the role of AsA and GSH in improving the chl or carotenoid levels. However, very few of them reported regarding the roles of AsA and GSH in regulating stomatal conductance, Calvin cycle, RuBP activity/regeneration, or photosystem efficiency, which directly generates ROS and results in oxidative stress [203]. Some of the research findings show the positive roles of GSH improving/regulating Pro, which is cellular ROS scavenger or cytosol stabilizer to reduce ROS generation. These are the promising area of future research, which not only will alleviate the oxidative stress, but also improve the photosynthetic efficiency of plants for increasing plant production for the constantly growing population of the planet.

Although, in this review article, we focused on abiotic stress-induced oxidative damage and the role of AsA-GSH cycle to mitigate such adversities, biotic stress (fungi, bacteria, virus, nematodes, and parasitic organisms, etc.) might also alter the essential plant processes as well as cellular metabolism. For example, the production of ROS and oxidative stress, disruption of membranes, hampering photosynthesis, changing enzyme activities, cell death, and yield loss might also be attributed to biotic stress, which is in line with abiotic stress. Biotic stress also disrupts signal transduction, as well as transfigures signal pathways that are associated with stress acclimation. Over the past decade, AsA-GSH cycle has also emerged as an important component for the plant biotic stress response. Similar to abiotic stresses, biotic stresses also alters the metabolism and changes in antioxidant activity. Therefore, AsA-GSH cycle also directly impacts the important metabolomic processes, thus providing an important link between metabolism, signal transduction, and acclimation to plants during biotic stress.

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Review

The Role of the Plant Antioxidant System in Drought Tolerance

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Abstract: Water deficiency compromises plant performance and yield in many habitats and in agriculture. In addition to survival of the acute drought stress period which depends on plant-genotype-specific characteristics, stress intensity and duration, also the speed and efficiency of recovery determine plant performance. Drought-induced deregulation of metabolism enhances generation of reactive oxygen species (ROS) and reactive nitrogen species (RNS) which in turn affect the redox regulatory state of the cell. Strong correlative and analytical evidence assigns a major role in drought tolerance to the redox regulatory and antioxidant system. This review compiles current knowledge on the response and function of superoxide, hydrogen peroxide and nitric oxide under drought stress in various species and drought stress regimes. The meta-analysis of reported changes in transcript and protein amounts, and activities of components of the antioxidant and redox network support the tentative conclusion that drought tolerance is more tightly linked to up-regulated ascorbate-dependent antioxidant activity than to the response of the thiol-redox regulatory network. The significance of the antioxidant system in surviving severe phases of dehydration is further supported by the strong antioxidant system usually encountered in resurrection plants.

Keywords: antioxidant; drought; ROS; RNS; stress; acclimation

1. Introduction

During their ontogenesis, plants face a dynamically changing environment defined by abiotic factors (e.g., light/dark, temperature, nutrient and water availability, and toxic compounds such as heavy metals) and biotic interactions (e.g., beneficial and pathogenic microbes, fungi, insects, other herbivores) [1]. Environmental perturbations which significantly disturb metabolism, development and yield, are considered as stress situations and cause stress responses in biological system. Such imposed stress is commonly accompanied by an increase in the production of reactive oxygen species (ROS) and reactive nitrogen species (RNS) that lead to an imbalance between their production and scavenging. Despite their reactive and thus toxic nature, ROS and RNS are also key components of signal transduction pathways that trigger stress responses. Furthermore, ROS and RNS are involved in plant developmental processes [2–4] and plant-microbe interactions [5,6]. However, excessive ROS and RNS production must be counteracted by the antioxidant system to prevent damage development and cell death.

Drought stress severely impacts plant development, growth and fertility. Drought triggers water loss and a decrease in water potential, which concomitantly leads to a reduction in cell turgor (Figure 1). Among the fastest processes induced by drought is the abscisic acid (ABA)-mediated closure of stomata [7]. Prolonged drought stress and increased stress intensity lead to further acclimation reactions. These responses include osmotic adjustment [8,9], decreased shoot-root ratio [10], cell

wall modifications [11,12], reprogramming of metabolism [13], and activation of the antioxidant system [14,15]. Many of these modifications are measurable and are used to characterize the severity of drought stress. Measurable traits are, for example, the stomatal and mesophyll conductance, net photosynthesis, photorespiration, abundance of osmoprotectants, tissue water potential, ABA content and membrane integrity. Drought avoidance includes morphological adaptations, like leaf curling and increased wax deposition on the leaf surface [16] (Figure 1).

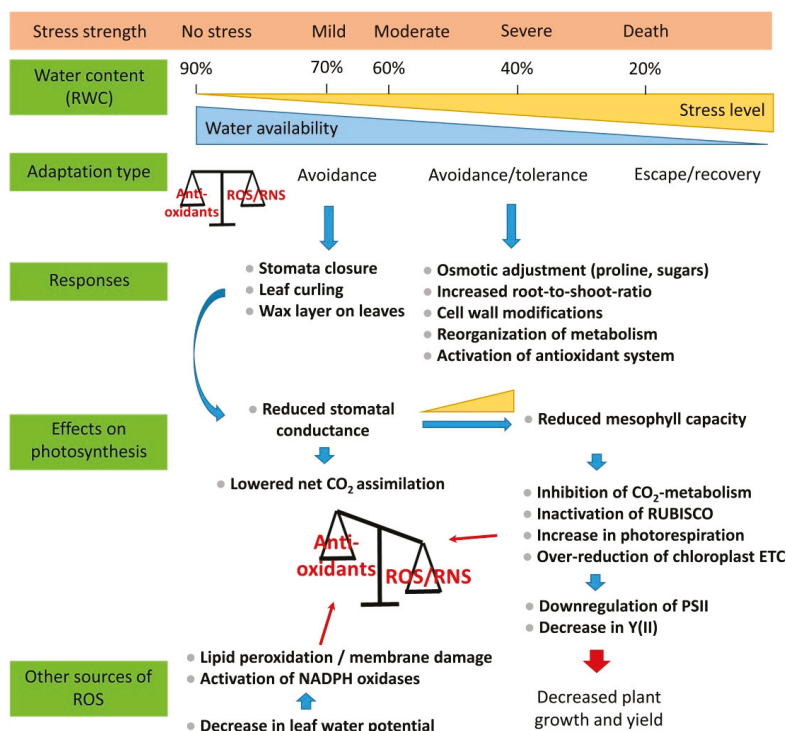


Figure 1. Physiological and biochemical processes triggered by drought.

During evolution, plants developed mechanisms to acclimate to drought or even to withstand dry periods. Extensive research has unraveled the molecular mechanisms of drought and desiccation tolerance. Figure 2 summarizes characteristic features of drought-sensitive, drought-tolerant and desiccation-tolerant plants. Tolerant plants are equipped with higher levels of both osmolytes and non-protein antioxidants, reprogram their metabolism and enhance their antioxidant capacity. Interestingly, sensitive species also activate their antioxidant system. Nevertheless, despite this apparent contradiction, drought tolerance seems to be a function of the antioxidant capacity realized in response to drought. Furthermore, the antioxidant activity not only is important during acute drought stress, but also interferes with recovery from water limitation and resurrection from dehydration.

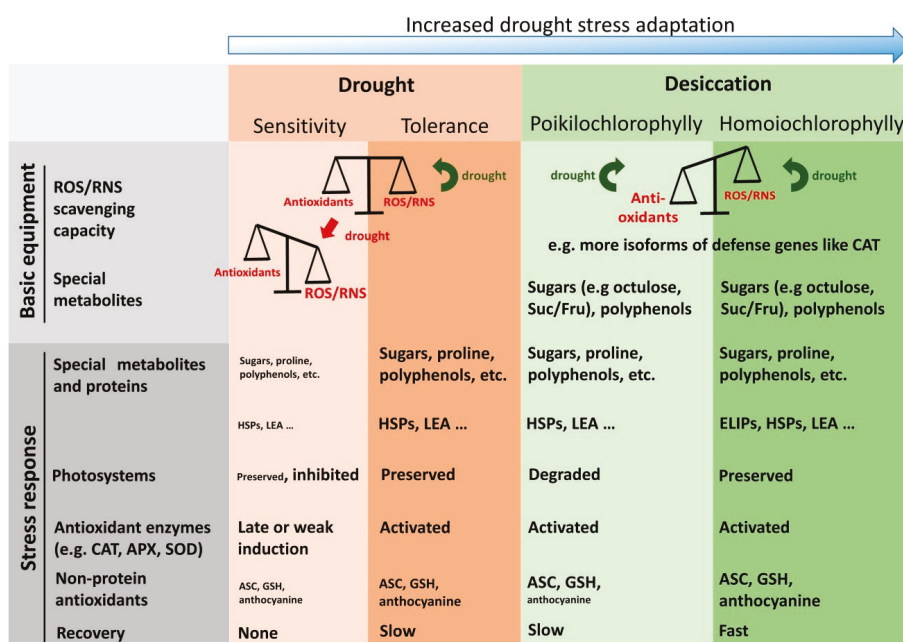


Figure 2. Characteristic features of drought-sensitive, drought-tolerant and desiccation-tolerant plants. The figure summarizes properties related to metabolism, antioxidant defense, and recovery which often are associated with the physiological traits. Red arrow: reactive oxygen species (ROS)/reactive nitrogen species (RNS) gain prevalence; green arrow: status is preserved following drought. Fond size correlates with the strength of stress responses measured. ROS, reactive oxygen species; RNS, reactive nitrogen species; HSP, heat shock protein; LEA, late embryogenesis abundant protein; ELIP, early light-inducible protein; Suc/Fru, sucrose to fructose ratio; CAT, catalase; APX, ascorbate peroxidase; SOD, superoxide dismutase; ASC, ascorbate; GSH, glutathione.

In the beginning of the review we will recall the classification of drought and how drought stress conditions are experimentally induced. This is important information to relate the production of ROS and RNS to the applied stress later in this review. Our review centers on the sites of production and roles of ROS and RNS during dehydration and their detoxification by the antioxidant system. Where possible we will correlate the activation of the antioxidative system to drought tolerance. Furthermore, we will evaluate which antioxidants are involved in drought response in particular. The last section describes the role of the antioxidative system in resurrection plants as an intriguing case of exceptional drought tolerance.

2. Classification and Application of Drought Stress

Drought is classified in mild, moderate and severe stages of stress (Table 1). The transition between the different stages occurs steadily and reflects the progression of drought stress severity both in duration and dehydration strength. Hence, an absolute value of dehydration cannot be assigned to the individual stages of drought stress. The stages are rather categorized in certain ranges. Various units have been used to describe water limitations (Table 1). The overall consensus is that the relative water content (RWC) in mild drought stress ranges between 60–70% compared to the control of ≥90%, in moderate stress between 40–60% and in severe stress between 0–40% (Table 1, Figure 1). Interestingly, these classifications are quite consistent between different species, even though the length of the applied stress to reach these states differs considerably (Table 1). Severe drought

stress conditions can be reached rapidly within a week in soils with low water holding capacity. Mild stress conditions, corresponding to a soil field capacity (SFC) of 70%, are already reached after two days, severe (SFC < 50%) and very severe wilting (SFC < 30%) after five and eight days, respectively, as determined for 25 day-old soybeans grown in a sand-vermiculite mixture [17]. A time period of 1–2 weeks without watering was shown to be the most suitable condition for testing both drought tolerance and recovery of various mesophytic species grown on soil (Table 2). Drought stress can be induced either by withholding water in the case of soil-grown plants or by polyethylene glycol (PEG) in both agar-plates and liquid cultures [18]. The use of PEG-infused agar systems allows generating a defined water potential in the substrate [19]. However, the majority of these systems were only applicable for seedlings for a long time. Recently, Frolov and colleagues [20] established an agar-based polyethylene glycol infusion drought model for six-to-eight-week-old *Arabidopsis* plants. This system is extremely valuable as it allows analyzing the response of adult plants and thus a more appropriate developmental stage in terms of agricultural application.

The occurrence and severity of drought-induced injury varies between different developmental stages of the plant and also depends on duration and strength of the applied stress.

Table 1. Classification of drought stress by different units that describe the water availability for different species at the various stages of drought stress.

Plant Species	Unit	Control	Mild	Moderate	Severe	Very Severe	Length of Stress Application	Reference
<i>Camellia sinensis</i> (Tea)	Soil moisture content [%]	19.5	15.2	10.17	5.54		week(s)	[21]
<i>Arabidopsis thaliana</i>	Water content [g water/g dry soil]	2.2	1.2		0.7		weeks	[22]
<i>Arabidopsis thaliana</i>	Relative soil water content [%]	85–90		45–50	30–35		week(s)	[23]
<i>Biserrula peccinans</i>	Water holding capacity	70–90	40–60		20–40		month	[24]
Common bean	Soil field capacity [%]	90	70		50	30	weeks	[17]
Jujube tree	Relative soil moisture [%]	80	70	60	40		weeks	[25]
Lemon balm and thyme	Relative soil water content [%]	70		40	25		months	[26]
<i>Malus lupulensis</i>	Soil field capacity [%]	75–85		45–55			months	[27]
Poplar	Relative soil water content [%]	70	45		20		month	[28]
Soybean	g_s intervals [mol H ₂ O m ⁻² s ⁻¹]	>0.2	0.1–0.2		<0.1		week	[29]
Tomato	Soil field capacity [%]	100		50			weeks	[30]
<i>Valeriana officinalis</i>	Available water content/relative water content (%)	100/77.3	70/67.2		65.1/50	51.4/30	months	[31]
Wheat	Relative soil water content [%]	80–90	35–43		20–25		week	[32]
Wheat	Relative water content [%]	80–100	60–80		40–60		weeks	[33]
Wheat	Soil field capacity [%]	85		55			months	[34]
<i>Populus deltoides</i>	Water potential [MPa]	–0.1	–0.5		–1.26		week	[35]
Wheat and maize	Water potential [MPa] in the presence of PEG6000		–0.4	–0.8	–1.5		week	[36]

g_s: gram(s).

Table 2. Exemplary experimental design for testing drought tolerance in different plant species.

Plant Species	Drought Stress (Age of Plants, Duration, Re-Watering)	Medium	Reference
<i>Arabidopsis thaliana</i>	2-weeks-old, 13 d no water, re-hydration for 2 d	soil	[37]
<i>Arabidopsis thaliana</i>	2-weeks-old, 5 d no water	MS medium	[38]
<i>Arabidopsis thaliana</i>	2-weeks-old, 12 d no water, re-hydration for 4 days	soil	[39]
Rice	2-weeks-old, 4 d 20% PEG-6000, 1–10 d re-watering	hydroponics	[40]
Rice	40-days-old, 7 d no water, 1–10 d re-watering	soil	[40]
Sugarcane	120-days-old, 10 d no water, re-watering	soil	[41]
Tobacco	14 d without water, 3 d re-watering	soil	[42]
Tomato	8-weeks old, up to 21 d no water	soil	[43]
Wheat	3-leaves stage, 72 h 20% PEG-6000 in 1/2 Hoagland solution (HS), 1 d re-watering with 1/2 HS	hydroponics	[44]

d, day(s); h, hour(s); MS medium, Murashige-Skoog medium; PEG, polyethylene glyco.

3. ROS and RNS Generation during Dehydration and Its Combination with Other Stresses

Stress-induced production of ROS and RNS occurs in different cell compartments [45]. They are used to transmit signals to the nucleus and other compartments to reprogram cell performance including gene expression [46,47]. The underlying mechanisms are known as retrograde and anterograde signaling pathways [1,48]. This paragraph focuses on the sources of ROS and RNS, and their accumulation in response to drought stress.

3.1. ROS during Drought

The first response of plants to drought is the closure of stomata in order to minimize water loss due to transpiration. Because of ongoing photosynthesis in the light, the increased gas diffusion barrier facilitates depletion of the intercellular carbon dioxide (CO₂) concentration. Decreased availability of CO₂ stimulates ribulose-1,5-bisphosphate oxygenation and, thus, photorespiratory hydrogen peroxide (H₂O₂) production in the peroxisomes. This effect has been studied in detail and was frequently summarized, e.g., with respect to drought and H₂O₂ production in wheat and potato as C₃ field crops [49]. Insufficient availability of the electron acceptor CO₂ slows down the oxidation of nicotinamide adenine dinucleotide phosphate (NADPH) in the Calvin–Benson cycle. Lack of NADP⁺ causes a backlog of electrons and over-reduction of the photosynthetic electron transport which in turn increases the reduction rate of oxygen as alternative electron acceptor in the Mehler reaction at photosystem I (PSI) and enhanced release of superoxide anion (O₂•[−]) and hydrogen peroxide (H₂O₂). Hence, chloroplasts are primary targets of excess light and CO₂ starvation in drought. In addition, photorespiration produces NADH in the mitochondrion.

A highly reduced chloroplast NADPH-pool via thioredoxin (TRX) reduction activates the NADPH-dependent malate dehydrogenase and, thereby, the malate valve for export of reducing equivalents to the cytosol and mitochondrion. The disequilibrium between electron supply and consumption in photosynthesis is efficiently transmitted to the respiratory electron transport chain (ETC) in the mitochondrion. Activation of alternative oxidase (AOX) and induction of *aox* gene expression are hallmarks of drought response [50–52]. Even under normal conditions, 1–2% of oxygen is consumed to produce ROS due to an over-reduction at complex I and III in the oxidative phosphorylation [53]. Under drought, the capacities of AOX, plant uncoupling proteins (PUCPs) and ATP-sensitive potassium channels are stimulated to dissipate excess electron flow in ETC [54]. Respiratory functions are inhibited by about two-thirds in drought-stressed plants as reviewed by Atkin and Macherel [55]. These studies included dehydration regimes of various intensities and on different time scales. The authors commented that the missing response in tolerant species might be due to enhanced antioxidant defense. Additionally, ROS are produced at the apoplast. Interestingly, the production of apoplastic ROS is coupled to calcium signaling [56]. Respiratory burst oxidase homolog (RBOH) proteins in the plasma membrane are calcium and phosphorylation-sensitive enzymes generating superoxide anions in the apoplast in response to drought, but also many other stresses [57,58]. Cell wall-associated kinases (WAKs) are members of the receptor-like kinase (RLK) family and participate in the perception of turgor pressure changes during drought probably linking ROS bursts with phosphorylation of RBOHs [59]. Apoplastic ROS also induce lipid peroxidation giving rise to malondialdehyde (MDA) as an indicator for membrane damage especially during drought. After dismutation of superoxide to H₂O₂ in the apoplast, transfer of H₂O₂ from the apoplast to the cytosol may also contribute to the intracellular ROS signature.

Table 3 summarizes changes of ROS and RNS amounts in response to drought stress. Maize growing in soil at 20% water saturation deficit accumulated twice the H₂O₂ amount of well-watered control plants [60]. Likewise, H₂O₂ reached thrice the contents of control rice if exposed to 200 mmol/L mannitol for two days [61,62] and in *Ailanthus altissima* plants that were kept unirrigated for 14 days [63], respectively. Thus, accumulation of ROS under drought is a prototypic case of stress-induced responses.

Table 3. Changes in reactive oxygen species (ROS) and nitric oxide (NO) amounts upon drought or osmotic stress treatment in various plant species. Data originate from green leaf tissue if not indicated otherwise. Increase in percent was chosen due to different detection methods with different units. Effects were estimated from graphs, figures and tables if not directly given in the text or supplements.

ROS/RNS Species	Plant Species	Stress Application	Observed Change in ROS/RNS Concentration (% Relative to Control)	Reference
H ₂ O ₂	<i>Ailanthus altissima</i>	No water for 14 d	+166	[63]
	<i>Arabidopsis thaliana</i>	200 mmol/L mannitol for 6 h	+50	[64]
	<i>Brassica rapus</i>	10% PEG for 2 d	+30	[65]
		20% PEG for 2 d	+65	
	<i>Citrus reticulata</i>	No water for 3 d	+16,6	[66]
		No water for 6 d	+37,5	
	<i>Cleome spinosa</i>	No water for 9 d	+45,5	[67]
		No water for 10 d	+25	
	<i>Crambe abyssinica</i>	50% MWHC for 32 h	+15	[68]
		50% MWHC for 136 h	+84	
	<i>Helianthus annuus</i> cultivars	40% SFC for 21 d	Variable, see literature	[69]
		10% PEG for 5 d	+68	
	<i>Helianthus annuus</i> Aydin	20% PEG for 5 d	+50	[70]
		10% PEG for 5 d	+15	
	<i>Helianthus annuus</i> Musala	20% PEG for 5 d	+30	[71]
		No water for 7 d	+490	
	<i>Medicago sativa</i>	200 mmol/L mannitol for 2 d	+200	[61]
	<i>Oryza sativa</i>	5% PEG for 28 d	+200	[62]
		10% PEG for 28 d	+225	
		15% PEG for 28 d	+300	
		20% PEG for 28 d	+380	
	<i>Oryza sativa</i> callus	−0.5 MPa for 1 d	Age dependent, see literature	[72]
		−2.0 MPa for 1 d		
	<i>Sorghum bicolor</i> M-81E	No water for 7 d	+28.9	[73]
	<i>Sorghum bicolor</i> Roma	No water for 7 d	+54.9	[74]
	<i>Stevia rebaudiana</i>	15% PEG for 30 d	+220	[75]
	<i>Triticum aestivum</i>	50% RWC for 12 d	+40	[76]
		15% PEG for 2 d	+45	
	<i>Triticum aestivum</i> seedlings	15% PEG for 4 d	+200	[77]
		15% PEG for 6 d	+280	
<i>Triticum aestivum</i> (booting)	No water for 52 d	+70	[77]	
<i>Triticum aestivum</i> (filling)	No water for 69 d	+43	[77]	
<i>Zea mays</i> growth zones	20 % less SWC till 3 d after 5th leaf	Doubled across all zones	[60]	
O ₂ • ⁻	<i>Crambe abyssinica</i>	50% MWHC for 32 h	+18	[68]
	<i>Helianthus annuus</i>	10 % PEG for 1 d	−60	[78]
	<i>Oryza sativa</i> roots and leaves	−0.5 MPa for 1 d	Age dependent, see literature	[72]
		−2.0 MPa for 1 d		
<i>Sorghum bicolor</i>	10% PEG for 1 d	−22.5	[78]	
NO	<i>Ailanthus altissima</i>	No water for 14 d	+125	[63]
	<i>Ananas comosus</i>	30% PEG for 25 d	Variable emission, see literature	[79]
	<i>Arabidopsis thaliana</i>	No water for 4 d	+150	[80]
	<i>Citrus aurantium</i>	13% PEG for 12 d	+150	[81]
	<i>Cucumis sativus</i>	Root aeration for 5, 10, 15 h plus rewatering	Variable, see literature	[82]
	<i>Hordeum vulgare</i>	No water for 18 d	Doubled production rate	[83]
	<i>Lotus japonicus</i> roots and leaves	No water for 5 d	+80	[84]
			+33	
	<i>Medicago truncatula</i> roots and leaves	No water for 3, 9, 11 d plus rewatering	Variable, see literature	[85]
	<i>Oryza sativa</i>	200 mmol/L mannitol for 1, 6, 24 h	Variable, see literature	[61]
	<i>Oryza sativa</i>	No water for 9 d	+200	[86]
	<i>Oryza sativa</i> seeds	20% PEG for 1 d	−75	[87]
	<i>Poncirus trifoliolate</i>	No water 6 h	+200	[88]
	<i>Saccharum</i> spp. roots and leaves	−0.4 MPa (PEG) for 1 d	Variable, see literature	[89]

MWHC, maximum water holding capacity; RWC, relative water content; SWC, soil water content; SFC, soil field capacity; d, day(s); h, hour(s).

3.2. ROS, Oxidative Post-Translational Modifications and Redox Signalling

Within proteins, the thiol groups of both cysteine (Cys) and methionine (Met) are the major sites of oxidative post-translational modifications (PTMs) [90]. Thiols are prone to successive oxidation to

sulfenic (R-SOH), sulfinic (R-SO₂H), and sulfonic (R-SO₃H) acids [91]. Cys oxidation and reduction efficiently regulates enzyme activities. A well-established system is the redox system of chloroplasts in which the redox input is provided by ferredoxin (Fd), NADPH and glutathione (GSH), redox signals are transmitted on target proteins by TRX, NADPH-thioredoxin reductase (NTRC) and glutaredoxins (GRX) [92]. Peroxiredoxins (PRX) are thought to sense the redox state of the cell and act in signaling instead of ROS detoxification [92]. Oxidative PTMs and the role of PRX in plant redox signaling are subjects of recent reviews and, thus, are not discussed in detail here [92,93].

3.3. RNS during Drought

Reactive nitrogen species are less diverse than ROS. Nitric oxide (NO) is a gaseous signaling molecule involved in germination, development, hormone regulation, and stress management. While homologues of animal NO synthase are absent from plants [94], the described mechanisms for NO production include (i) nitrate reductase (enzymatic, cytosol/plasma membrane), (ii) xanthine oxidoreductase (enzymatic, peroxisome), (iii) NO-associated proteins (enzymatic, mitochondria/plastids), (iv) nitrite: NO reductase (enzymatic, plasma membrane), (v) electron transport chain (non-enzymatic, mitochondria/chloroplast), and (vi) a poorly understood mechanism using arginine, polyamine or hydroxylamine [95–97]. The bioactive NO concentration is influenced by the nitrogen nutrient supply, the concentration of the storage compound nitrosogluthathione (GSNO), the activity of the GSNO reductase, and turnover mechanisms including the interaction with hemoglobins [98–100].

Osmotic stress, established by exposing rice roots to 200 mmol/L mannitol, increased the NO amount threefold within 24 h in rice leaves [61]. The same increase in NO was observed in rice after withholding irrigation for nine days, while a significant increase was undetected after three days [86]. Since both studies focused on leaves, the large time scale difference is striking and may reflect the time span needed to establish similar stress levels. This interpretation is supported by the fact that an osmotic shock treatment with 210 mmol/L mannitol corresponds to an applied osmotic potential of approximately -1.1 MPa [101], while an equivalent osmotic potential after withholding water was reached only at days 4 and 5 [86]. The data also point to changes in drought sensitivity during development. Most plants respond more sensitive to dehydration in early developmental stages. Therefore, one explanation for the discrepancies between the above mentioned studies might be attributed to differences in the plant growth stages of 16 [61] versus 42 days [86], leaving juvenile leaves more sensitive to drought. In this context, it should be mentioned that the ratio of developing to mature cell in the leaf lamina changes significantly during the early phase of development. Furthermore, the antioxidant response to paraquat was compromised in young *Arabidopsis* leaves [102]. Mature leaves were able to compensate ROS accumulation much more efficiently due to an increase in APX activity. The authors suggested different photoprotective regulatory mechanisms in the two leaf types. Furthermore, it was concluded that the redox-state of plastoquinone A (Q_A) is the determinant of tolerance to paraquat-induced oxidative stress [102]. A similar observation was made in *Fagus sylvatica* L. Here, resistance to paraquat-induced oxidative stress was mediated by an increase in SOD activity in mature leaves [103]. In the tea plant (*Camellia sinensis*), cold-sensitivity of young leaves is correlated with inhibited expression of genes related to cell membranes, carotenoid metabolism, photosynthesis and the antioxidative system [104]. In contrast, transcripts belonging to the gene ontology groups of chloroplasts, cell membranes, redox processes, glutathione metabolism and photosynthesis were increased in mature leaves in response to cold. Hence, the antioxidative system plays an important role in establishing acclimation and hardening to stress.

In tree species like *Ailanthus altissima*, NO amounts increased three-fold after withholding water for 14 days [63]. NO is reported as an important positive regulator for Crassulacean acid metabolism (CAM) in pineapple leaves as described by Freschi et al. [79]. Emission of NO gradually increased from 40 to 140 pmol·h⁻¹·g⁻¹ dry weight upon treatment with 30% PEG 6000 for 5 days. Of PEG, 30% corresponds to a water potential of -1.03 MPa [105] and, thus, is similar to osmotic stress induced by

200 mmol/L mannitol [87]. NO quantification was mostly achieved by using fluorescence probes like diaminofluorescein (DAF) or diaminorhodamine (DAR) derivatives. To overcome drawbacks related to limited specificity, new probes are presently engineered to improve sensitivity and specificity [106]. Nevertheless, cell- and tissue-imaging with DAF-2 diacetate in dehydrating pineapples localized NO in chlorenchyma, trichoma and epithelial cells but did not resolve subcellular compartmentation.

NO also plays a significant role in regulating germination during drought in grasses like wheat and rice [87,107]. Endogenous NO counteracts programmed cell death and vacuolization induced by gibberellic acid. The NO amount in aleurone layers drops by 75% after 24 h of osmotic stress compared to controls (20% PEG-6000). Exogenous application of NO donors alleviates the effect and delays germination. Thus, a synergistic effect of NO is seen with ABA allowing postponing germination until growth conditions improve. Under such conditions, germination is inhibited and resumed only after growth conditions have improved. Expression of rat neuronal NO synthase (nNOS) in plants constitutively increases NO levels twofold in *A. thaliana* [80] and 1.5-fold in *O. sativa* [61]. These nNOS-plants accumulate more biomass and less H₂O₂ after withholding water for 14 d (*A. thaliana*) or upon treating rice with 200 mmol/L mannitol. These results assign a significant role to NO in shaping the acclimation to drought. They also show that the NO effect partly antagonizes the effects of ROS in this process.

In general, information on plant specific endogenous RNS signaling is still scarce. The production of NO occurs in similar subcellular compartments as ROS but our knowledge on its induction, regulation of enzyme activities, and substrates emerges only slowly. Hence, many groups use NO donors to artificially expose plants to RNS. Currently, research focuses on synergistic versus antagonistic effects of RNS and ROS, especially in the field of abiotic stress, and promises a more integrative concept. Experiments on genetic model systems are needed which link the dynamics of specific markers for RNS signaling with proteomic and transcriptomic analyses.

3.4. Nitrosylation by ONOO⁻ and GSNO

Antagonistic and synergistic effects relate to reaction products of RNS and ROS and antioxidants, respectively. Thus, GSNO forms by reaction of NO with reduced glutathione, while peroxynitrite (ONOO⁻) forms at sites of simultaneous formation of O₂•⁻ and NO. GSNO triggers S-nitrosylation, while ONOO⁻ causes tyrosine nitration. Several targets of these reactions are part of the antioxidant defense system like PRX, ascorbate peroxidase (APX), monodehydroascorbate reductase (MDHAR), dehydroascorbate reductase (DHAR) and catalase (CAT) [108,109]. Especially during drought in *Lotus japonicus* NO amounts doubled in roots, but interestingly not in leaves [84]. S-nitrosylation of proteins is promoted in roots. The authors hypothesized that roots are prone to nitrosative stress, and leaves to oxidative stress.

Higher NO concentrations in roots compared to leaves were also reported in sugarcane [89] and bluegrass [110] and support this rule of thumb. One function of NO in roots concerns root patterning as described for pea, tomato, tobacco, and cucumber facing drought conditions [82,111–113]. Such differential effects have also been reported for pollen development and stigma function which respond preferentially to either RNS or ROS, respectively. Apparently, ROS and RNS play unique roles in developmental signaling which should be explored further [114]. Furthermore, GSNO serves as a mobile carrier of NO allowing for long distance signaling. In contrast, ONOO⁻ is highly reactive and characterized by a short half-life of 10 to 20 ms, and thus is discussed as a linker between ROS and RNS signaling [115]. Moreover, specific analyses are needed to clarify the NO-related effects on metabolism and to see whether RNS signaling is exclusively transmitted by ONOO⁻ and GSNO.

3.5. ROS/RNS in Stress Combinations with Drought

Responses to drought are accentuated if dehydration is combined with other abiotic stresses. Exceptions from this rule concern drought combined with ozone and high CO₂. The antagonising effect is traced back to stomata closure triggered by ozone [116] or high CO₂ [117]. Iyer and colleagues [116]

described this phenomenon in *Medicago truncatula*. Here, ROS levels increase in response to drought and ozone by 2-fold and 2.8-fold, respectively, compared to the well-watered condition. However, ROS levels in response to combined drought and ozone stress are indistinguishable from the control (well-watered plants). In contrast, NO levels are elevated only in response to drought by approximately 2-fold, while ozone has no effect. Simultaneous application of the two stresses again did not lead to significant changes. Interestingly, jasmonic acid and salicylic acid synthesis are induced after application of NO-donors in *A. thaliana* which might explain the mitigating effect of ozone in combination with drought [118]. Again, both reports vary in species and treatment, but indicate that RNS signaling is directly involved in stress response and alters the ROS effects.

In the natural environment, dry periods often coincide with high temperature and high light. Malondialdehyde (MDA) is an indicator for lipid peroxidation and oxidative damage and significantly increases in green tissue of citrus cultivars exposed to a combination of drought and heat (10 d, 40 °C). The increase is absent in single stress applications [119]. The stronger effect of a drought/heat combination is also seen in maize. Here, MDA levels increase by 225%, while the single applications elevated MDA levels by only 45% (−0.7 MPa PEG for 8 h) or 92% (2 °C/h increase from 28 to 42 °C for 8 h), respectively [120]. In cotton cultivars, no significant differences in H₂O₂ levels are observed for drought and combined drought/heat stress [121].

Combining heat (42 °C) and drought in succulent purslane for seven days doubles MDA content, while single stress treatments increase the MDA amount only by 20%. Interestingly, O₂•[−] amount raises 2.5-fold under heat and combined stress, but not in plants exposed to drought [122]. Surprisingly, the leaf H₂O₂ level decreases in grapevine upon deprivation from water for four days followed by treatment with heat (1 h, 42 °C) or high light (1 h, 2000 μmol quanta s^{−1}m^{−2}) [123]. None of the double or triple stress treatments including drought alters the H₂O₂ amounts above the levels measured during control treatments. Significant variations between cultivars are only seen in single treatments and a heat/high light treatment.

These examples support the theory by Suzuki and colleagues [1] that the response to a combined stress is unique and cannot be simply extrapolated from the responses to single stresses. For instance, the response to stress combinations on signaling pathways and responses can be synergistic, antagonistic or independent. Antagonistic and, thus, positive interactions are observed for the combination of drought and high CO₂ [124]. However, combined stress often leads to negative interactions, and the consequences are synergistic rather than additive [1]. This is also true for high light and drought [125]. Both, high light and drought realize an over-reduced state of photosynthetic ETC. With respect to high light the over-reduction is caused by an excess of light energy, while the over-reduction following drought is caused by a limited CO₂ availability after stomatal closure and the concomitant inhibition of the Calvin–Benson cycle. Consequently, in both cases ROS and RNS are generated, but the ROS/RNS signatures differ in both cases [126].

The described examples demonstrate the importance to investigate plant responses and signaling pathways in combined stress. However, most laboratory studies on plant stress responses consider one stress at a time, whereas plants in the field usually are exposed to different stresses simultaneously. For example, drought stress is often accompanied by heat and high light intensities [117,127]. Therefore, it has to be kept in mind that any treatment applied under controlled growth chamber conditions fails to reflect field conditions. Ecotypes of the same plant species adopt distinct adaptive responses to acclimate to their local habitats. Such naturally occurring biodiversity in terms of sensitivity vs. tolerance of closely related species, the extreme adaptability of specialists and the special case of crop plant monocultures cannot be treated in this review focusing on ROS and RNS-dependent signaling.

4. Response of the Redox Network under Drought

The activation of the antioxidant system via retrograde signaling is a key process in plant acclimation to oxidative stress. Thus, the upregulation of antioxidant enzymes represents an important marker for drought stress. In the cell, the production and scavenging of ROS and RNS is strictly

controlled and the equilibrium can be perturbed by several biotic and abiotic stresses [128]. Plants have evolved complex redox signaling networks in which ROS and RNS are used as signals to regulate normal and stress-related physiological processes including antioxidant mechanisms to combat the toxic effects of ROS and RNS [129,130]. Plants keep ROS under control by an efficient and versatile scavenging system. The antioxidant defense comprises low molecular weight compounds such as GSH, ascorbate (ASC), α -tocopherol, carotenoids, and enzymes including CAT, SOD, and the thiol peroxidases of the PRX and glutathione peroxidase (GPX) type [131].

Thiol peroxidases are linked to the NADPH-thioredoxin reductase (NTR), ferredoxin-dependent TRX reductase (FTR) and GSH/GRX systems [132,133]. Mechanism of ROS production and their scavenging by high antioxidant capacity has been associated with tolerance of plants to abiotic stresses [128]. Recently, a new function was assigned to thiol peroxidases in redox regulation, namely as TRX oxidases [134]. This mechanism allows for reading out the balance between reductive electron input and oxidative electron drainage and tunes the redox and activity state of target proteins.

4.1. Effect of Drought Stress on the Antioxidant System and Redox Homeostasis

During drought stress, up-regulation of antioxidant systems occurs at both the transcriptional and post-transcriptional level. Table 4 gives examples for quantitative drought responses of antioxidative enzymes and enzymes involved in regeneration of non-protein antioxidants. APX, catalase (CAT) and GPX represent the principal ROS scavengers in plants. Among these three, APX appears to be induced most strongly on post-transcriptional level (Table 4). In contrast to CAT and GPX, APX is also regulated on transcriptional level based on the data summarized in Table 4. Cytosolic, chloroplastic and peroxisomal APX activities are commonly enhanced in all species of the plant kingdom. The activity of cytosolic APX is increased during drought in pea [135]. The *alx8* mutant (altered expression of APX2) of *Arabidopsis* reveals improved drought tolerance [136,137]. Over-expression of peroxisomal or cytosolic APX from poplar in transgenic tobacco increases plant performance under drought [138,139]. CAT is a tetrameric, heme-containing enzyme that catalyzes the dismutation of H_2O_2 into H_2O and O_2 in the peroxisome. CAT2 plays a crucial role when the plant is exposed to a severe drought stress [140]. Compared to APX activation, stimulation of CAT is moderate (Table 4). Even though CAT activation seems predominantly taking place on post-transcriptional level, there are examples for complex regulation of CAT activity under severe drought which involves gene expression, translation and protein turnover [141].

Table 4. Antioxidant enzymes regulated in plants under drought.

Antioxidative Enzyme	Plant Species	Transcriptional Regulation	Post-Transcriptional Regulation	Reference
Ascorbate peroxidase (APX)	Alfalfa			
	<i>Arabidopsis thaliana</i>	APX1 1.66-fold	only severe: 15%	[142]
	<i>Arabidopsis thaliana</i>	APX1 ns	800%	[143]
	<i>Arabidopsis thaliana</i>	APX3 2-fold		[144]
	Bean (tol)		34% (14 d stress)	[145]
	Bean (sens)		ns	[145]
	<i>Carrizo citrange</i>	APX2 5.5-fold	50%	[146]
	<i>Carrizo citrange</i>	cAPX 2-fold	Total ns	[147]
	<i>Cleopatra mandarin</i>	APX2 10-fold	50%	[146]
	<i>Cleopatra mandarin</i>	cAPX 0.5-fold	Total ns	[147]
	<i>Coffea canephora</i> (tol)		219%	[148]
	<i>Coffea canephora</i> (sens)		168%	[148]
	Cotton (tol)		up to 50%	[149]
	Cotton (tol)		60% protein level roots	[150]
	Cotton (sens)		90% protein level roots	[150]
Date Palm	APX-46 4-fold		[151]	
Date Palm	APX-1 4-fold		[151]	
Maize		25%	[152]	
Pea	cAPX1 3-fold (not log2-fold)	cAPX1 50%	[136]	
Poplar (dry climate)		200%	[153]	
Poplar (wet climate)		50%	[153]	
Rice (tol)		Initially 40%, after 5 days: 40%	[154]	
Rice (sens)		80%	[154]	
Tobacco	APX1 299%	300%	[155]	
Tobacco	thyAPX and strAPX ns		[156]	
Wheat	2.29-fold (rel. expression)	35%	[157]	

Table 4. *Cont.*

Antioxidative Enzyme	Plant Species	Transcriptional Regulation	Post-Transcriptional Regulation	Reference
Catalase (CAT)	Alfalfa		100% (moderate)	[142]
	Alfalfa		ns (severe)	[142]
	<i>Arabidopsis thaliana</i>		30%	[158]
	<i>Carrizo citrange</i>	1.5-fold	ns	[147]
	<i>Cleopatra mandarin</i>	1.5-fold	ns	[147]
	<i>Coffea canephora</i> (tol)		109%	[148]
	<i>Coffea canephora</i> (sens)		58%	[148]
	Maize		50%	[152]
	Pea		100%	[136]
	Rice (tol)		Initially 80%, after 5 days: 55%	[154]
	Rice (sens)		96%	[154]
	Bean (tol)		ns	[145]
	Bean (sens)		ns	[145]
Dehydroascorbate reductase (DHAR)	Tobacco	CAT1-2 ns		[156]
	Tobacco	CAT3 2.4-fold (rel. expression)	45%	[155]
	Wheat (tol)		90%	[159]
	Wheat (sens)		80%	[159]
	Cotton		up to 50%	[149]
	Fescue		33%	[160]
	Date Palm	DHAR-25 1.4 fold		[151]
	Date Palm	DHAR-2 1.4-fold		[151]
	Wheat	2.3-fold (rel. expression)	44%	[157]
	Wheat (tol)		65%	[161]
Wheat (sens)		29%	[161]	
Glutathione peroxidase (GPX)	Alfalfa		ns	[142]
	Poplar (dry climate)		160%	[153]
	Poplar (wet climate)		400%	[153]
	Potato		50%	[162]
	Tortula	2.9-fold (rel. expression)	ns	[163]
	Wheat (tol)		ns	[164]
Wheat (sens)		92%	[164]	

Table 4. *Cont.*

Antioxidative Enzyme	Plant Species	Transcriptional Regulation	Post-Transcriptional Regulation	Reference
Glutathione reductase (GR)	<i>Arabidopsis thaliana</i>			[158]
	<i>Carrizo citrange</i>	2-fold	65%	[147]
	<i>Cleopatra mandarin</i>	2-fold	90%	[147]
	Cotton		50%	[149]
	Cowpea (tol)	3.5-fold (rel. expression)	up to 80%	[165]
	Cowpea (sens)	4-fold (rel. expression)	ns	[165]
	Maize		20%	[152]
	Poplar (dry climate)		33%	[153]
	Poplar (wet climate)		180%	[153]
	Bean (tol)		800%	[145]
Bean (sens)		90% (7 d stress)	[145]	
Tobacco		125% (7 d stress)	[145]	
Tobacco			[156]	
Tortula			[155]	
Tortula			[163]	
Wheat			[157]	
Wheat (tol)	1.6-fold (rel. expression)	35%	[164]	
Wheat (sens)		100%	[164]	
Wheat (sens)	2.1-fold (rel. expression)	30%	[164]	
Wheat (tol)		ns	[164]	
Wheat (sens)		36%	[164]	
Glutathione S-transferase (GST)	Tortula		40%	[163]
	Wheat (tol)		113%	[161]
	Wheat (sens)		46%	[161]
	Wheat (tol)		ns	[164]
Wheat (sens)		ns	[164]	
Monodehydroascorbate reductase (MDHAR)	Wheat	2.3-fold (rel. expression)	65%	[157]
	Tobacco	1.6-fold (rel. expression)		[156]
Protein disulphide isomerase (PDI)	Stiff brome	BdPDIL1-1 > 1-fold (rel. expression)		[166]
	Stiff brome	BdPDIL1-2 0.67-fold,		[166]
	Stiff brome	BdPDIL7-2 0.33-fold		[166]
	Stiff brome	BdPDIL2-1 > 1-fold (rel. expression)		[166]
	Stiff brome	BdPDIL3-1, BdPDIL5-1 and BdPDIL8-1 (between 0.33 and 1-fold)		[166]

Table 4. *Cont.*

Antioxidative Enzyme	Plant Species	Transcriptional Regulation	Post-Transcriptional Regulation	Reference
Peroxiredoxin (PRX)	Date Palm	PRXR-18 1.1-fold		[151]
	Date Palm	PRXR-1 1.5-fold		[151]
	Date Palm	PRXR-2 4.3-fold		[151]
Superoxide dismutase (SOD)	Alfalfa		Total SOD ns	[142]
	<i>Arabidopsis thaliana</i>		MnSOD 30%	[142]
	<i>Arabidopsis thaliana</i>	ns	100%	[158]
	Bean (tol)		30% (7 d of stress)	[145]
	Bean (sens)		25% (7 d of stress)	[145]
	<i>Carrizo citrange</i>		ns	[147]
	<i>Cleopatra mandarin</i>	CuZnSOD 2-fold	100%	[147]
	<i>Coffea canephora</i> (tol)	FeSOD 1.5-fold	558%	[148]
	<i>Coffea canephora</i> (sens)		100%	[148]
	Date Palm		up to 450%	[151]
Blue Grass		100% (25 d of stress)	[160]	
Fescue		30% (25 d of stress)	[160]	
Maize	SOD-13 1.2-fold	20%	[152]	
Maize	SOD-11 1.26-fold		[152]	
Pea		100% (chloroplast and cytosol)	[136]	
Poplar (dry climate)		Cu/ZnSOD 5-fold	[153]	
Poplar (wet climate)		cAPX1 6-fold (rel. expression)	[153]	
Rice (tol)			20-50%	[154]
Rice (sens)			60%	[154]
Tobacco		ns	ns	[155]
Wheat (tol)			First 24 h 90%	[159]
Wheat (sens)			First 24 h 80%	[159]

Table 4. *Cont.*

Antioxidative Enzyme	Plant Species	Transcriptional Regulation	Post-Transcriptional Regulation	Reference
Thioredoxin (TRX)	Date Palm	TRX-40 1.1-fold		[151]
	Date Palm	TRX-44 1.3-fold		[151]
	Date Palm	TRX-37 1.3-fold		[151]
	Date Palm	TRX-16 1.3-fold		[151]
	Date Palm	TRX-31 1.3-fold		[151]
	Date Palm	TRX-12 1.1-fold		[151]
	Wheat (tol)		TRX -h 32%	[161]
	Wheat (sens)		TRX -h 41%	[161]

APX, ascorbate peroxidase; CAT, catalase; DHAR, dehydroascorbate reductase; GPX, glutathione peroxidase; GR, glutathione reductase; GST, glutathione-S transferase; MDHAR, monodehydroascorbate reductase; PDI, protein disulfide isomerase; PRX, peroxiredoxin; SOD, superoxide dismutase; TRX, thioredoxin. Black color, up-regulation; red color, down-regulation; ns, not significantly changed.

Besides APX, other components of the ASC-GSH cycle, namely MDHAR, DHAR, glutathione-S-transferase (GST) and glutathione reductase (GR), work synergistically in different cell compartments. MDHAR, DHAR, GST and GR transcripts and activity are predominantly induced under drought stress (Table 4). Among these four enzymes, GR is activated strongest. GR activation can be compared to the one observed for CAT. In general, upregulation of the ASC-GSH metabolism and associated enzymes efficiently scavenge H_2O_2 under drought stress as observed in wheat [167].

Moreover, PRXs are also up-regulated and accumulated in cotton [150], date palm [151] and wheat [161] upon drought (Table 4). This indicates that plants activate compensatory mechanisms to counteract enhanced H_2O_2 production in response to drought stress. In addition to their reductive function in detoxifying H_2O_2 , alkyl hydroperoxide and $ONOO^-$, PRX play a role in redox signaling and transmit information on the cell ROS state to target proteins [134,168].

SODs are a class of metalloenzymes that catalyze the dismutation of two molecules of $O_2^{\bullet-}$ into molecular oxygen and H_2O_2 . The activation of SOD isoforms (Mn-SOD, Fe-SOD, Cu,Zn-SOD) is interpreted as a measure to counteract $O_2^{\bullet-}$ accumulation in diverse cell compartments under drought in e.g., *Arabidopsis* [158], blue grass [160], citrus [147], *Coffea canephora* [148], date palm [151], fescue [160], pea [135], poplar [153], tepary bean [145] and wheat [159]. Apparently, SOD is a critical component of the ROS-scavenging system likely by minimizing the reaction of $O_2^{\bullet-}$ with, e.g., NO to form $ONOO^-$, unsaturated fatty acids for peroxidation or with proteins. In line with this assumption transgenic plants overexpressing Cu,Zn-SOD are more tolerant to drought stress [168].

A set of other important proteins belonging to the TRX superfamily is usually highly activated under drought stress. In general, TRXs are induced under different environmental stresses including dehydration, salinity, heat or cold [169]. Under several stresses, atypical and canonical TRX have the capacity to reduce oxidized antioxidant enzymes in the chloroplast, cytosol and mitochondria [170,171]. TRXs are localized in cytosol, chloroplast, mitochondrion, endoplasmic reticulum and nucleus [132]. Strongly responding oxidoreductases are represented by atypical chloroplastic TRX (CDSP32 and CDSP34), cytosolic or mitochondrial NADPH-TRX reductase (NTRA or B), endoplasmic reticulum-associated protein disulfide isomerase (PDI) and canonical cytosolic TRX (TRX h). NTRA-overexpressing plants exhibit extreme drought tolerance with high survival rates, low water loss and reduced ROS accumulation compared to wildtype and *ntra*-knock out plants [144]. However, TRX transcripts and activity measurements in date palm [151] and wheat [161] also indicate a down-regulation of some TRX members in response to drought stress.

4.2. Distinct Patterns of Antioxidative System Activation in Sensitive and Tolerant Species

As summarized in Figure 2, drought-sensitive species also activate their antioxidative system. The data given in Table 4 confirm this assumption. However, they point out that not only the magnitude of activation might be decisive but also which enzymes are activated. For instance, the activation of the major scavenger APX and CAT is stronger in tolerant species compared to their sensitive counterparts. In contrast, sensitive species activate GPX more than tolerant species. Changes in the activation of the antioxidant system between sensitive and tolerant species are visualized in Figure 3. Obviously, sensitive plants predominantly activate the glutathione-dependent scavenging system, while the ascorbate-dependent system is only induced moderately or are even down-regulated (Figure 3). On the other hand, tolerant species showed a stronger activation of ascorbate-dependent scavenging system compared to the glutathione-dependent system. Moreover, inactivation is only apparent for the TRX-dependent scavenging system in tolerant species. Because drought stress leads to an over-reduction of the electron transport chain, down-regulation of TRX may counteract excessive reduction of target proteins. On the other hand, TRX-dependent reduction of PRX is compromised under this condition. However, PRX can be regenerated by other enzymes like GRX and NTRC [92]. Moreover, drought conditions necessitate a high capacity of detoxifying enzymes such as APX and CAT to suppress ROS accumulation. Furthermore, PRX are involved in redox-signaling [92] which might be their predominant function under drought stress.

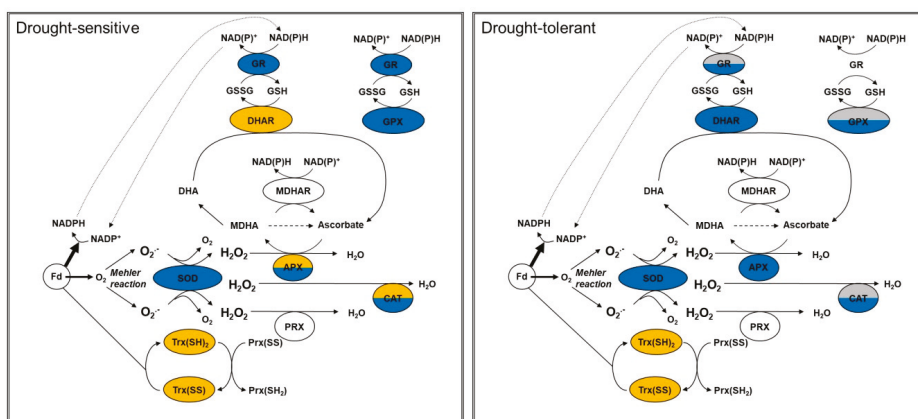


Figure 3. Changes in the activation of the antioxidative system in sensitive and tolerant species. Orange, downregulation, blue, upregulation, grey, no significant changes, no color, no data. APX, ascorbate peroxidase; CAT, catalase; DHAR, dehydroascorbate reductase; Fd, ferredoxin; GPX, glutathione peroxidase; GR, glutathione reductase; MDHAR, monodehydroascorbate reductase; PRX, peroxiredoxin; SOD, superoxide dismutase; TRX, thioredoxin.

There is not much information on drought tolerance and NO signaling. However, a recent study investigated root extracellular and leaf intracellular NO contents in drought-tolerant and -sensitive sugarcane genotypes. Here, drought tolerance was correlated with an increased extracellular NO concentration due to an increased nitrate reductase (NR) activity [89]. Furthermore, the simultaneous decrease in S-nitrosogluthathione reductase (GSNOR) implicates that tolerant plants possess a higher GSNO reservoir. As mentioned before, GSNO is a mobile carrier of NO allowing long distance transport. As observed for roots, likewise, the leaf intracellular NO content was elevated in the tolerant species when compared to the sensitive [89].

When evaluating the role of the ascorbate- and glutathione-dependent pathways in drought tolerance, it must be taken into consideration that the basal levels of the different antioxidants in sensitive and tolerant species were not compared. However, *Arabidopsis* plants lacking the cytosolic APX1 show a collapse in the entire chloroplast-located H₂O₂-scavenging system, which is accompanied with increased H₂O₂ levels and protein oxidation, respectively [172]. In a direct comparison with TRX-dependent peroxidase activity, APX activity was 7-fold and 2-fold higher in leaf extracts and chloroplasts, respectively [173]. Thus, a predominant role of the ascorbate-dependent antioxidative system should be assumed. At this point, a deeper screen through the literature may not be helpful to test the hypothesis since most studies only present data on changes of selected antioxidant enzymes in a few tolerant and sensitive species. Future investigations should explicitly address the hypothesized role of the ascorbate-dependent ROS defense in drought tolerance in tolerant and sensitive genotypes within plant families. If the hypothesis can be confirmed, the ascorbate-dependent scavenging system can be a target for improving plant tolerance towards drought in biotechnological application.

5. The Role of the Antioxidative System in Desiccation Tolerance

Drought stress induces major transcriptional reprogramming in plants via ABA-dependent and ABA-independent pathways regardless whether a plant is sensitive or tolerant to drought. This is also true for resurrection plants. Research has shown that resurrection plants use similar mechanisms and strategies to respond and adapt to drought as sensitive species. However, if processes like perception, signaling and responses are as similar as assumed, which specific features provoke the tolerance to desiccation of vegetative tissues? The major difference to drought-sensitive plants is that the protective machinery of resurrection plants is held in an activated, 'primed' state. To achieve this, the basal

levels of osmolytes like sugars and polyamines, non-enzymatic and enzymatic antioxidants are often increased in desiccation tolerant plants. High levels of sugars like trehalose, sucrose and raffinose prevent protein denaturation, stabilize membranes and act as ROS scavengers [174,175]. In addition, unique sugars such as the C8-sugar octulose also accumulate to up to 90% of the soluble sugars in photosynthetically active leaves [176]. Despite this, Djilianov and colleagues [177] found that the initial Suc/Fru ratio is increased in the desiccation-tolerant plant *H. rhodopensis* compared to the sensitive species *C. eberhardtii*. The differences and similarities between drought sensitivity, and drought and desiccation tolerance are compiled in Figure 3.

Significant evidence indicates that the strong antioxidant status is a prerequisite of desiccation tolerance in resurrection plants. Thus, glutathione is suggested to be an important player in the dehydration response [178]. The non-enzymatic antioxidants ascorbate and glutathione turn more oxidized during dehydration [177,179], while the total glutathione content increases. The increase in GSSG remains elevated during desiccation of the tolerant species *H. rhodopensis*. In addition, activities of antioxidant enzymes like SOD, peroxidase (POD), CAT and GR increase in response to drought in the fern *Selaginella tamariscina* [180]. Resurrection plants are well equipped with genes encoding antioxidant enzymes. For instance, *H. rhodopensis* contains more genes encoding SOD, CAT, MDHAR and GR than the model plant *A. thaliana* [181]. The *H. rhodopensis* genome encodes eight catalase genes and, thus, five more than the *Arabidopsis* genome [181]. Expression of specific *Cat* genes is upregulated following drought/desiccation. The importance of CAT activity during desiccation is shown by an experiment in which leaves were sprayed with the catalase inhibitor 3-aminotriazole (0.1 mmol/L 3-AT). Plants that were treated by 3-AT never recover completely from desiccation and die within a month after the treatment [181]. The increased sensitivity of dehydrating plants to CAT inhibitors is interpreted as indication of enhanced photorespiration due to stomatal closure, lack of intercellular CO₂, enhanced oxygenation of RUBISCO and therefore stimulated release of H₂O₂ by glycolate oxidase in the peroxisome. CAT is needed to detoxify the released H₂O₂ and therefore inhibited CAT disturbs redox and ROS homeostasis under drought.

Wang and colleagues [180] compiled drought/dehydration-responsive proteins from both resurrection and common plants [180]. The comparison of tolerant with sensitive phenotypes highlights the role of the antioxidant system in drought tolerance. For instance, CAT, APX and SOD levels are up-regulated in the drought-tolerant CE704 genotype (maize), while CAT and APX levels decreased in the drought-sensitive genotype 2023 [182]. In wheat, TRX-h and glutathione S-transferase are selectively upregulated in the drought-tolerant genotype Khazar-1 [161].

It should be noted that dehydration tolerance depends on additional features of the plants apart from adjusting metabolism including the antioxidant system. Massive water loss usually causes mechanical disruption in hygrophytic and mesophytic plants, e.g., the rupture of the tonoplast/plasmamembrane/cell wall junctions. Such irreversible mechanical damage is prevented in resurrection plants such as *Craterostigma plantagineum* where the tissue shrinks proportionally to the water loss. Thus, special anatomical properties like leaf curling and structurally flexible vessels are important features of dehydration tolerance [183,184].

6. Conclusion and Perspective

Drought tolerance depends on conditional activation of the acclimation program during initial phases of water loss. This also applies for thallophytic and cormophytic resurrection plants which need a hardening period for full expression of the tolerance trait [183,185]. As pointed out in this review, different drought stress regimes and time points of analysis result in distinct states of the ROS and RNS network and the antioxidant defense system. In the initial phases of dehydration, the activation of the hardening program decisively involves the generation of ROS and RNS which assist in activating the redox regulatory network and appropriate gene expression and protein accumulation. It was out of focus of this review to describe the intimate link between ROS, RNS and hormone signaling like

salicylic acid and abscisic acid [186]. In the end ROS and RNS define a regulatory framework of the cell and contribute to link the stress impact to gene expression and whole plant performance [187].

At present our knowledge on specific subcellular ROS, RNS and redox patterns still falls short of the requirements for understanding the drought acclimation response in its entirety. Cell imaging with roGFP for glutathione redox state [188] and Hyper for H₂O₂ [189] will provide important insight on subcellular responses. In addition, in depth redox proteomics detecting the redox state of also low abundant proteins will provide a global view with subcellular resolution.

There is a need to assess the various PTMs in the proteome simultaneously. This is a challenge for current proteomics which for technical reasons often focuses on single or few PTMs only [190]. As functional readout of ROS and RNS, such approaches will realize the necessary temporal and spatial resolution since ROS and RNS partly antagonize each other. Nevertheless, the presence of both reactive species is necessary for full drought acclimation. Additionally, the reaction of NO with O₂•⁻ generates the highly reactive ONOO⁻ which directly nitrates proteins. Cysteine oxidation and tyrosine nitrations are PTMs that change the activity of its target enzymes. Proteomics may tackle this challenge.

Along with the activation of the antioxidative system, other stress markers often increase during periods of progressive dehydration, e.g., H₂O₂ as indicator of redox imbalance, MDA as lipid oxidation product, glyoxylate linked to photorespiration, glutathione as antioxidant, glutamate and proline as precursor and compatible solute, and zeaxanthin with its role in photoprotection. The consensus of what defines drought tolerance is that many traits are needed to prevent biochemical or physiological impairment during water deficit. Several traits contribute to drought tolerance and include reduced water loss, build-up of osmotic potential, synthesis of compatible solutes, dissipation of excess energy, activation of antioxidant defense and repair systems, generation of sclerenchymatic tissue, strengthening the plasmamembrane-cell wall interaction and other mechanisms of growth adjustment such as differentiation of smaller leaves. The recovery from water depletion is affected by light intensity with often negative interference, i.e., slower recovery at high light.

Taken together, strategies to improve drought tolerance in crops need to target several metabolic pathways at the same time. Certainly, the activation of the antioxidative system following drought is one important goal. Attention should also be drawn to the pathways that are selected to increase drought tolerance. In the first instance, overexpressing of certain enzymes can lead to a beneficial increase in drought tolerance, but may delay germination and development for months and, thus, interfere with the growing season. Thus, biotechnological approaches should take into account the temporal and spatial signaling aspect in drought stress acclimation.

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Abbreviations

ABA	abscisic acid;
AOX	alternative oxidase;
APX	ascorbate peroxidase;
ASC	ascorbate;
CAM	crassulacean acid metabolism,
CAT	catalase;
CO ₂	carbon dioxide;
cys	cysteine;
d	day(s);

DAF	diaminofluorescein;
DAR	diaminorhodamine;
DHAR	dehydroascorbate reductase;
ELIP	early light inducible protein;
Suc/Fru	sucrose to fructose ratio;
ETC	electron transport chain;
FTR	ferredoxin-dependent TRX reductase;
g	grams;
GPX	glutathione peroxidase;
GPX	glutathione peroxidase;
GR	glutathione reductase;
GRX	glutaredoxin;
GSH	glutathione;
GSNO	nitrosoglutathione;
GSNOR	S-nitrosoglutathione reductase;
GST	glutathione-S transferase;
h	hours;
H ₂ O	water;
H ₂ O ₂	hydrogen peroxide;
HS	Hoagland solution;
HSP	heat shock protein;
LEA	late embryogenesis abundant protein;
MDA	malondialdehyde;
MDHAR	monodehydroascorbate reductase;
Met	methionine;
MPa	megapascal;
MWHC	maximum water holding capacity;
NAD ⁺ /NADH	oxidized/reduced nicotinamide adenine dinucleotide;
NADP ⁺ /NADPH	oxidized/reduced nicotinamide adenine dinucleotide phosphate;
NO	nitric oxide;
NR	nitrate reductase;
NTR	NADPH-thioredoxin reductase;
NTRC	NADPH-dependent thioredoxin reductase C;
O ₂	molecular oxygen;
O ₂ ^{•-}	superoxide anion;
ONOO ⁻	peroxynitrite;
PDI	protein disulfide isomerase;
PEG	polyethylene glycol;
PTM	posttranslational modification;
PRX	peroxiredoxin;
PSI	photosystem I;
PUCPs	plant uncoupling proteins;
RBOH	respiratory burst oxidase homolog;
RLK	receptor-like kinase;
RNS	reactive nitrogen species;
ROS	reactive oxygen species;
RWC	relative water content;
SFC	soil field capacity;
SOD	superoxide dismutase;
SWC	soil water content;
TRX	thioredoxin;
WAKs	cell wall-associated kinases

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Article

Chitosan Upregulates the Genes of the ROS Pathway and Enhances the Antioxidant Potential of Grape (*Vitis vinifera* L. 'Touriga Franca' and 'Tinto Cão') Tissues

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Abstract: Chitosan is an environmentally-friendly active molecule that has been explored for numerous agricultural uses. Its use in crop protection is well-known, however, other properties, such as bioactivity, deserve attention. Moreover, the modes of actions of chitosan remain to be elucidated. The present study assessed the levels of total phenolic compounds, the antioxidant potential, and the expression of reactive oxygen species (ROS) scavenging genes in the berries (skins and seeds), leaves, cluster stems, and shoots upon chitosan application on two red grapevine varieties (Touriga Franca and Tinto Cão). The application of chitosan on the whole vine before and after veraison led to the increased levels of polyphenols, anthocyanins, and tannins in Tinto Cão berries, and polyphenols and tannins in Touriga Franca berries, respectively. CUPric Reducing Antioxidant Capacity (CUPRAC) and Ferric Reducing Antioxidant Power (FRAP) assays indicated an increase in the antioxidant potential of berries. With the exception of ascorbate peroxidase (APX), all the ROS pathway genes tested, i.e., iron-superoxide dismutase (Fe-SOD), copper-zinc-superoxide dismutase (Cu/Zn-SOD), catalase (CAT), glutathione reductase (GR), glutaredoxin (Grx), respiratory burst oxidase (Rboh), amine oxidase (AO), peroxidase (POD) and polyphenol oxidase (PPO), were found up-regulated in chitosan-treated berries. Results from the analyses of leaves, stems, and shoots revealed that chitosan not only induced the synthesis of phenolic compounds but also acted as a facilitator for the transfer of polyphenols from the leaves to the berries.

Keywords: chitosan; elicitor; *Vitis vinifera* L.; antioxidant activity; secondary metabolites; anthocyanins; ROS pathway; polyphenols; tannins

1. Introduction

Grapes are non-climacteric berry fruits produced by deciduous and perennial woody vines. For centuries, grapes have had a huge impact on the world's economy due to their use in winemaking and in a vast variety of food products [1,2]. The quality of grape-derived products depends on the

levels of secondary metabolites in the berries, including anthocyanins, tannins, stilbenes, flavonoids, and other phenolic compounds. Although these compounds are known for their antioxidant properties, they are particularly important for the flavor, color, and taste of wines as well as other industrial interests like nutraceuticals and pharmaceutical products [3–5].

For years now, increasing the levels of secondary metabolites in grape berries has been a major subject of interest worldwide. Several cultural practices have been studied to improve the accumulation of secondary metabolites in grape and wine [1,6–12]. It was found that vine ‘cordon training’ and pruning led to the increased levels of tannins and anthocyanins in red wine [9]. The total anthocyanin and the total phenolic contents were also higher in Tannat, Malbec, and Xinomavro grapes obtained from vines subjected to cluster thinning, when compared with “not thinned” controls [6–8]. Koundouras et al. reported that deficit irrigation caused a substantial increase of the level of anthocyanins in the skins of Cabernet Sauvignon grapes [10]. Subsequent studies in the following years confirmed this finding [11,12].

Besides the above-mentioned traditional cultural practices, the use of electrical [1] and chemical [13] elicitors have also been attempted. Elicitor-mediated induction of plant defenses is one of the most effective strategies in enhancing the production of secondary metabolic compounds, and the approach has been tested in various crops [13]. An elicitor is defined as “a compound introduced in small concentration to a living system to promote specific responses” [13]. In plants, several molecules, such as phenylalanine, urea, benzothiadiazole, methyl jasmonate, yeast extract, abscisic acid, and chitosan have been used as elicitors [14–18]. Chitosan, in particular, has received a scrutinized interest giving its efficacy in the crop protection [18–22]. Chitosan is a cheap linear polysaccharide derived from the chitin of fungal cell walls and exoskeleton of arthropods [23,24]. Chitosan is safe, biodegradable, non-allergic, bio-compatible, and the second most abundant renewable carbon source after cellulose [25–28]. These properties have ensured their wide industrial use [23]. Chitosan was introduced for agricultural purposes in the 1980s. The first report of chitosan as an elicitor was in pea (*Pisum sativum* L.) and tomato (*Solanum lycopersicum* L.) [19]. Its application resulted in an increased accumulation of the phytoalexin pisatin in pea, and anti-nutrient proteinase inhibitors in tomato. Chitosan was later used to induce the accumulation of secondary metabolites in many other plant species. For example, it was demonstrated that foliar application of chitosan enhances tomato fruit weight, productivity, and resistance against fungal pathogens [20,21], improves the yield and vegetative growth of strawberries [29], and the quality and self-life of kiwifruit [30].

There is a worldwide trend to explore the properties of chitosan for sustainable agriculture [25,27,28]. There are, however, only sporadic reports on in-field applications of chitosan on grapevine plants. Spraying of chitosan combined with yeast on vines before harvest reduced the decay of berries during cold storage [31]. In vitro cultured grape plantlets supplemented with a chitosan gel demonstrated improved growth and resistance towards the fungal pathogen *Botrytis cinerea* [22]. Vitalini et al. reported increased levels of aromatic compounds in wines obtained from chitosan-treated vines, as compared to those vines which were treated with conventional fungicides [15]. There was also an improvement in the overall sensory acceptance of wines [15]. Although Portu et al. did not find a substantial impact on the phenolic composition of grapes and wines following chitosan treatment [16]. The above-mentioned studies highlight the necessity of conducting more research on the response of vines concerning the application of chitosan.

Moreover, the mode of action of chitosan is still unknown for several aspects of plant protection and productivity. Chitosan is known to induce plant defense mechanisms by indirectly stimulating the synthesis of secondary metabolites [16,23]. The metabolic pathways regulated by chitosan, which ultimately lead to drought resistance in white clover, have been reported recently [26]. In particular, it was found that chitosan-induced drought resistance is associated with the accumulation of endogenous chitosan, and the enhancement of the ascorbate–glutathione and tricarboxylic acid cycle pathways [26]. In the present study, an additional insight in the mode of action of chitosan is provided by analyzing vines in a tissue-specific manner, after field application of chitosan during

and after veraison on *Vitis vinifera* L. ‘Touriga Franca’ and ‘Tinto Cão’. ‘Touriga Franca’ and ‘Tinto Cão’ are the most widely grown varieties of grapevines in the Douro Demarcated Region of Portugal and are very important varieties for Port wine production. Berries’ seeds and skins, stems, leaves, and shoots were analyzed for the total phenolic content (TPC), total anthocyanin content (TAC), total tannin content (TTC), antioxidant potential, and the expression of the reactive oxygen species (ROS) pathway genes.

2. Materials and Methods

2.1. Plant Material, Treatments, and Sample Collection

Vitis vinifera L. ‘Touriga Franca’ and *Vitis vinifera* L. ‘Tinto Cão’ (red varieties), grown in the Quinta de Nossa Senhora de Lourdes vineyard (41°19’ N, 7°44’ W, 500 m above mean sea level), were used in this study. The vineyard is in Vila Real, which is situated in the lower Corgo sub-region of the Douro Demarcated Region of northern Portugal. The vines (420A rootstock) were trained with the Guyot system, spaced 2.3 × 0.9 m, on a grass-covered morainic soil, with 15% of gravel and loamy sand. The trial was set up as a complete randomized block design in three replications (lines), with 12 vines per line (chitosan-treated vines and control vines) in each line. Spraying of crop protectants, weed control, and shoot guiding were the same for all vines.

Chitosan (molecular weight of 76 kDa and deacetylation degree of 85%) was obtained from Sigma-Aldrich (St. Louis MO) and dissolved in 0.01 M aqueous acetic acid solution to a concentration of 0.1% (w/v). The whole vine, including the fruits, was sprayed with 200 mL of the chitosan solution, using a manual spray lance. Chitosan was applied on the same vines twice, first at the beginning of veraison at the onset of berry ripening, and second at the completion of veraison with all the berries colored red. The meteorological conditions from the beginning of veraison until berry harvesting were as follows: average temperature, humidity, and wind velocity of 23.4 °C, 57.4%, and 6.5 km/h, respectively. Berries and leaves were collected during veraison, after complete veraison, and at complete maturation. Stems and shoots were collected at the complete maturation of the berries, as explained diagrammatically in Figure 1. The same tissues from all 12 vines from a line were pooled to constitute a sample.

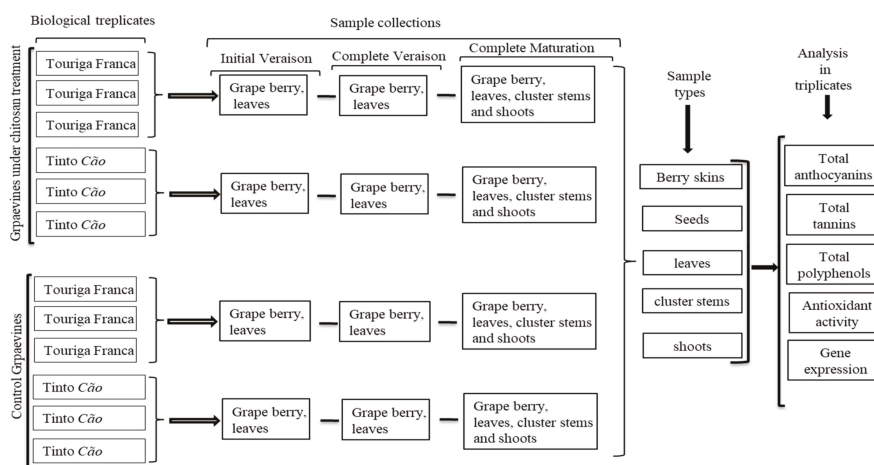


Figure 1. Diagrammatic representation of the complete experimental setup, including sample collection and sample analyses.

2.2. Sample Preparation

The leaves were removed from the vines using a scissor. The seeds were manually separated from the berries' skins. The stems and shoots were chopped in small pieces using a scissor. All processed samples were immediately frozen in liquid nitrogen and stored in $-80\text{ }^{\circ}\text{C}$. Before the analyses, all the samples were freeze-dried and grounded into a fine powder (ca. 0.5 mm size) using a coffee grinder (Q.5321, Qilive, China).

Aqueous ethanol (1.5 mL, 50% *v/v*) was added to 100 mg of powdered sample and the contents were mixed in an agitator for 1 h at room temperature (ca. $25\text{ }^{\circ}\text{C}$). The mixture was then centrifuged at 10,000 rpm for 20 min at $4\text{ }^{\circ}\text{C}$. The supernatant was transferred to a tube and the residue was re-extracted as described above. Both the supernatants were combined, made up to 3 mL with 50% aqueous ethanol, and stored at $-20\text{ }^{\circ}\text{C}$ until further analyses.

2.3. Determination of the Total Phenolic Content (TPC) and the Total Anthocyanin Content (TAC)

The TPC and TAC were estimated using a protocol adapted from two previous publications [5,32]. An Aliquot (200 μL) of the supernatant extract obtained in 2.2 was acidified with 3.8 mL of 1.0 M aqueous HCl. The two solutions were mixed by gentle inversion, followed by incubation at $28\text{ }^{\circ}\text{C}$ for 3 h. The absorbances at 520 nm and 280 nm were recorded using a spectrophotometer (Evolution 201, Thermo Scientific). A520 values were converted into malvidin-3-O-glucoside contents based on a generic extinction coefficient of malvidin-3-O-glucoside [5], and the TAC was expressed as μg of malvidin-3-O-glucoside equivalents per mg dry weight. The TPC was calculated and expressed as μg of epicatechin equivalents per mg dry weight, using a calibration curve for epicatechin.

2.4. Determination of the Total Tannin Content (TTC)

The TTC was determined using the methyl cellulose-mediated precipitation method [33,34]. A mixture was prepared by adding 600 μL of 0.04% methyl cellulose to 200 μL of the supernatant extract obtained in Section 2.2. After incubation for 3 min at room temperature, 400 μL of a saturated solution of ammonium sulphate and 0.8 mL of deionized water were added and the contents mixed by inversion. A control mixture was also prepared, but with deionized water (600 μL) instead of 0.04% methyl cellulose. After incubation for 10 min at room temperature, the mixtures were centrifuged for 5 min at 4,000 rpm ($4\text{ }^{\circ}\text{C}$). The resulting supernatant was read at 280 nm. The TTC was obtained by subtracting the A280 nm extract and the A280 nm control. The tannin value was converted into epicatechin equivalents (μg per mg dry weight) with the aid of a calibration curve.

2.5. Determination of the Antioxidant Activity Using the DPPH, CUPRAC, and FRAP Assays

2.5.1. Sample Preparation

Powdered lyophilized samples (40 mg) were extracted with 1 mL of 70% methanol. After incubation at $70\text{ }^{\circ}\text{C}$ for 30 min, the solution was cooled at room temperature and centrifuged at 11,000 rpm for 20 min in order to obtain the supernatant used for the analyses.

2.5.2. DPPH (2,2-diphenyl-1-picrylhydrazyl) Assay

The DPPH assay used in this study was a modification of previous methods [35–37]. A 4% DPPH solution was prepared in 95% ethanol. The supernatant obtained in 2.5.1 (15 μL) was added to the DPPH solution in a 96-wells microplate to a final volume of 300 μL . The decrease in absorbance at 517 nm (*ABS sample*) was measured with a microplate reader, against a blank prepared with 70% methanol instead of the supernatant (*ABS blank*). The DPPH antioxidant activity was expressed as the percentage of the remaining DPPH in the reaction medium (% AA) using Equation (1).

$$\%AA = \left(ABS\ blank - \left(\frac{ABS\ sample}{ABS\ blank} \right) \right) \times 100 \quad (1)$$

2.5.3. FRAP (Ferric Reducing Antioxidant Power) Assay

The FRAP assay was modified from [38,39]. A portion of an aqueous 10 mM solution of 2,4,6-Tri(2-pyridyl)-s-triazine in 40 mM HCl was mixed with the same volume of 20 mM FeCl₃·6H₂O and 10 times higher volume of acetate buffer of pH 3.6 (3.1 g sodium acetate and 16 mL acetic acid per L). The mixture was incubated at 37 °C for 20 min. A portion (900 µL) of the ferric tripyridyltriazine (Fe³⁺-TPTZ) mixture and 20 µL of supernatant were diluted up to 1000 µL with deionized water and incubated for 10 min, and the absorbance was measured at 593 nm. A standard curve was prepared using fresh FeSO₄ solutions, and the ferric reducing power was expressed as mg FeSO₄ equivalents per mg dry weight.

2.5.4. CUPRAC (CUPric Reducing Antioxidant Capacity) Assay

The CUPRAC assay was carried out as described by [40]. Briefly, 0.2 mL of supernatant was added with 0.5 mL of 2 × 10⁻³ M Cu(II) chloride, 0.5 mL ethanol, 1 mL of 7.5 × 10⁻³ M neocuproine (Nc), 1 mL of 1 M ammonium acetate (NH₄/Ac) buffer pH 7.0, and 0.8 mL ethanol. After 30 min of incubation at room temperature, the absorbance at 450 nm was recorded against a water blank. The standard calibration curve of trolox was constructed as absorbance vs. concentration, and the antioxidant activity expressed as mg trolox equivalents per mg dry weight.

2.6. RNA Extraction, cDNA Synthesis, and qRT-PCR Analysis

Total RNA was extracted from frozen tissues harvested after complete maturation of the vines, using the CTAB (cetyl trimethyl ammonium bromide) buffer as described in [41]. Total RNA was extracted by using the RNeasy Mini Kit (Qiagen, Germany) and a DNase treatment was performed respectively. The integrity of the total RNA was verified by running samples on 1.0% denaturing agarose gels, followed by quantification with a ND-1000 Spectrophotometer (NanoDrop Technologies, Wilmington, DE) at wavelengths of 230 and 280 nm. A total of 1 µg of DNase-treated RNA was used for cDNA synthesis, using the First Strand cDNA Synthesis Kit (Solis Biodyne, Tartu, Estonia). Specific internal primers (Table 1) were designed using Primer3 online tool (<http://primer3.ut.ee/>). Quantitative real-time PCR (qRT-PCR) was performed to estimate the relative expression of 10 selected genes of the ROS pathway. Each reaction contained 5 × HOT FIREPo^l® EvaGreen® qPCR Mix Plus (Solis Biodyne, Tartu, Estonia), 200 nM of each primer, 1 µL 10 × diluted cDNA sample, and nuclease-free water to a final volume of 20 µL. Reactions were carried out on PicoReal-Time Thermal Cycler (PRO 96-1500-512, Thermo Scientific, Ratastie, Finland). The PCR program was: 95 °C for 12 min (initial denaturation), followed by 40 cycles of heating at 95 °C for 15 s (denaturation), 52 °C for 20 s (annealing), and 72 °C for 20 s (extension). No final extension was carried out. The software PicoReal 2.0 was used to analyze the data. The relative mRNA level for each gene was calculated as RQ values (relative quantification by using PicoReal 2.0 tool), using actin as the internal reference gene.

Table 1. Primer sequences for the genes of the reactive oxygen species pathway analyzed in this study.

	Gene	Primer Sequence
<i>Fe-SOD</i>	Superoxide dismutase	F 5' CCTTTGTGAACCTAGGCGAACC 3' R 5' TGGCCGGTTAGCTTGAACCTC 3'
<i>Cu/Zn-SOD</i>	Superoxide dismutase	F 5' AGATTGGCATGTGGTGTGTG 3' R 5' ACTCCCACATTACCCAACAACA 3'
<i>CAT</i>	Catalase	F 5' GGTGTTACACCTTCACTCT 3' R 5' GAGATCCTGAGTAGCATGACTG 3'
<i>APX</i>	Ascorbate peroxidase	F 5' ATCTGGTGGTCATACTCTGG 3' R 5' TCTAGGAGAGCCTTGTCTGA 3'
<i>GR</i>	Glutathione reductase	F 5' AATACTAGGGGAGGGTACA 3' R 5' GTTGCCTTGGATGCAGA 3'

Table 1. Cont.

	Gene	Primer Sequence
<i>Grx</i>	Glutaredoxin	F 5' CTGCTCTCACAGCTAAAAGC 3' R 5' CATCACAGATCACAATCCAC 3'
<i>Rboh</i>	Respiratory burst oxidase	F 5' CTGATTCTCAGTAGGAACCTGGT 3' R 5' GAGGTTTGGTCATGTATAGTGC 3'
<i>AO</i>	Amine oxidase	F 5' GTGAATCCGAACAAGAGAAC 3' R 5' AGACTTATTGTAGGGTGTGACC 3'
<i>POD</i>	Peroxidase	F 5' CCTGATCCAACACTAGATGC 3' R 5' GTCTGACTGAAGAAGTCCAGAG 3'
<i>PPO</i>	Polyphenol oxidase	F 5' CTAGAACTCCAGGTTTCATGC 3' R 5' GCTTCTCGTCGTAGAGTGAT 3'

2.7. Statistical Analyses

Statistical analyses were carried out using JMP 11.0 (SAS Institute Inc, Cary, NC). For each vine tissue, the effects of chitosan application on the TPC, TAC, and TTC during veraison, after veraison and at maturity were assessed using a two-way analysis of variance (ANOVA). The comparison among means was determined according to Turkey's test. When only two means were available, an independent *t*-test was used. Antioxidant activities (DPPH, FRAP, CUPRAC) and gene expression data for each tissue were subjected to an independent *t*-test analysis. All analyses were performed in triplicates (technical replicates) and results are expressed as mean \pm standard deviation of data recorded from three independent samples (biological replicates). Differences between means were considered meaningful at $p < 0.05$ and represented by different letters for ANOVA and asterisks for the *t*-test.

3. Results

3.1. Effect of Chitosan Application on Polyphenols in Grapevine

Grape seeds, skins, and juice contain a substantial amount of polyphenolic compounds which explain the health benefits of grape-based nutraceutical products [23]. In this study, polyphenols accumulated more in the seeds than in the skins of 'Tinto Cão', and more in the skins than in the seeds of 'Touriga Franca' (Table 2). At the first and second harvesting times, the TPC increased in leaves, skins, and seeds collected from vines treated with chitosan, in comparison to samples from control vines: increases of 24.18%, 29.51%, and 8.96% in 'Touriga Franca' during veraison; 31.80%, 22.83%, and 13.31% in 'Touriga Franca' after veraison; 10.53%, 23.01%, and 3.40% in 'Tinto Cão' during veraison; 2.09%, 20.40%, and 18.17% in 'Tinto Cão' after veraison (Table 2). At maturity, the TPC was also higher in the skins of chitosan-treated vines than in those of control vines, with 8.96% and 25.71% increases in 'Touriga Franca' and 'Tinto Cão', respectively. An increase was also recorded in the leaves of 'Touriga Franca' (41.44%), with no change in the leaves of 'Tinto Cão'. At complete maturation, the TFC was higher in the stems from chitosan-treated vines than those of control vines, with increases of 28.73% and 31.75% for 'Touriga Franca' and 'Tinto Cão', respectively. The application of chitosan had no effect in the TPC in the seeds and shoots of the two varieties (Table 2).

Table 2. The total phenolic content.

Variety/Tissue	Harvesting Time/Treatment					
	During Veraison		After Veraison		At Complete Maturation	
	Control	Chitosan	Control	Chitosan	Control	Chitosan
'Touriga Franca'						
berry seeds	109.69 ± 5.04 b	119.27 ± 6.58 a	89.95 ± 5.13 c	101.92 ± 2.47 b	94.67 ± 2.26 c	93.65 ± 5.10 c
berry skins	105.54 ± 2.73 c	136.69 ± 5.40 a	110.61 ± 4.67 c	135.86 ± 5.37 a	112.70 ± 4.23 c	122.80 ± 1.92 ab
cluster stems	NA	NA	NA	NA	NA	NA
leaves	70.28 ± 6.27 c	87.27 ± 1.70 a	63.11 ± 0.56 d	83.18 ± 0.86 a	55.65 ± 1.09 e	78.71 ± 1.48 b
shoots	NA	NA	NA	NA	37.23 ± 2.36 a	34.09 ± 0.64 a
'Tinto Cão'						
berry seeds	192.64 ± 5.34 b	199.19 ± 2.64 a	154.15 ± 1.86 d	182.14 ± 5.54 c	146.32 ± 4.25 e	144.05 ± 1.23 e
berry skins	80.50 ± 3.61 f	99.03 ± 2.14 e	112.24 ± 5.07 d	135.13 ± 0.60 b	129.95 ± 4.17 c	163.36 ± 9.57 a
cluster stems	NA	NA	NA	NA	158.61 ± 5.28 b	208.97 ± 8.37 a
leaves	223.28 ± 10.10 b	246.79 ± 1.92 a	191.77 ± 3.86 d	195.77 ± 3.86 d	204.76 ± 8.89 c	201.16 ± 3.16 cd
shoots	NA	NA	NA	NA	49.59 ± 2.44 a	52.65 ± 1.51 a

(TPC; µg epicatechin equivalents per mg dry weight) in grapevine tissues after chitosan application before and after veraison. All data were obtained from three biological replicates; means ± standard deviations within a row followed by different letters (a–f) are statistically different ($p < 0.05$; ANOVA Turkey's test). NA = not analyzed.

3.2. Effect of Chitosan Application on Anthocyanins in Grapevine

Anthocyanins are a special class of polyphenols involved in resistance against diseases in the leaves and shoots, and responsible for the color of berries and wines [1]. Table 3 shows that anthocyanins accumulated mainly in the skins of the fruits, as compared to the seeds where they were usually not detected. After application of chitosan, the TAC increased in the skins of 'Touriga Franca' and 'Tinto Cão' during veraison by 35.27% and 56.02%, respectively and after veraison by 17.42% and 10.51%, respectively. At complete maturation, a 22.33% increase in TAC was also recorded in the skins of 'Tinto Cão', in comparison with the skins from control vines. On the contrary, the TAC decreased by 10.55% in the skins of 'Touriga Franca' at maturity (Table 3). Before maturation, the TAC also increased in the leaves of chitosan-treated vines, 28.21% in 'Touriga Franca' and 22.37% in 'Tinto Cão' during veraison, and 37.50% and 14.93% after veraison. At complete maturation, the TAC was also higher in 'Touriga Franca' leaves (35.90% increases) treated with chitosan, while no change was recorded for 'Tinto Cão' (Table 3). At complete maturation, more anthocyanins accumulated in the stems of chitosan-treated vines than in those of control vines, with 113.76% and 16.04% increases of TAC in 'Touriga Franca' and 'Tinto Cão', respectively. The TAC of the shoots was not affected by the application of chitosan (Table 3).

Table 3. The total anthocyanin content.

Variety/Tissue	Harvesting time/Treatment					
	During Veraison		After Veraison		At Complete Maturation	
	Control	Chitosan	Control	Chitosan	Control	Chitosan
'Touriga Franca'						
berry seeds	0.00 ± 0.00 a	0.00 ± 0.00 a	0.00 ± 0.00 a	0.00 ± 0.00 a	0.00 ± 0.00 a	0.00 ± 0.00 a
berry skins	16.52 ± 0.44 e	22.35 ± 0.91 cd	20.86 ± 0.87 d	24.49 ± 0.34 ab	26.02 ± 1.02 a	23.28 ± 0.55 bc
cluster stems	NA	NA	NA	NA	NA	NA
leaves	0.39 ± 0.02 b	0.50 ± 0.01 a	0.32 ± 0.02 c	0.44 ± 0.01 ab	0.39 ± 0.01 b	0.53 ± 0.01 a
shoots	NA	NA	NA	NA	1.55 ± 0.11 a	1.12 ± 0.02 a
'Tinto Cão'						
berry seeds	0.00 ± 0.00 a	0.00 ± 0.00 a	0.00 ± 0.00 a	0.00 ± 0.00 a	0.00 ± 0.00 a	0.00 ± 0.00 a
berry skins	5.32 ± 0.19 f	8.30 ± 0.40 e	21.16 ± 1.32 d	23.39 ± 0.25 c	28.74 ± 0.97 b	35.16 ± 1.16 a
cluster stems	NA	NA	NA	NA	2.25 ± 0.01 b	2.68 ± 0.02 a
leaves	0.76 ± 0.06 b	0.93 ± 0.04 a	0.67 ± 0.01 c	0.77 ± 0.03 b	0.78 ± 0.03 b	0.77 ± 0.03 b
shoots	NA	NA	NA	NA	0.14 ± 0.02 a	0.15 ± 0.01 a

(TAC; µg malvidin-3-O-glucoside equivalents per mg dry weight) in grapevine tissues after chitosan application before and after veraison. All data were obtained from three biological replicates; means ± standard deviations within a row followed by different letters (a–f) are statistically different ($p < 0.05$; ANOVA Turkey's test). NA = not analyzed.

3.3. Effect of Chitosan Application on Tannins in Grapevine

Tannins contribute to the astringency of grapes and, in combination with anthocyanins, define the stability of wine color [34]. In this study, tannins tended to accumulate more in the seeds than the skins, although this depended on the growth stage of the vines (Table 4). At all harvesting times, the TTC increased in the skins and seeds collected from vines treated with chitosan, as compared to control vines: 14.18% and 325.15% in 'Touriga Franca', 9.61% and 9.23% in 'Tinto Cão' during veraison; 37.48% and 159.02% in 'Touriga Franca', 52.62% and 21.32% in 'Tinto Cão' after veraison; 20.46% and 32.12% in 'Touriga Franca', 48.46% and 6.01% in 'Tinto Cão' at complete maturation. The effect of chitosan on the TTC in the leaves was varied among varieties. A decrease in the leaves of 'Touriga Franca' during (11.62%) and after (9.73%) veraison and an increase at complete maturation (90.33%); and the exact opposite for 'Tinto Cão' i.e., an increase during (36.14%) and after (23.71%) veraison, and a decrease after complete maturation (11.98%) (Table 4), was observed. A variety effect was also observed at maturity for the stems, with chitosan application leading to a lower TTC in 'Touriga Franca' (35.12 decreases) and a higher TTC in 'Tinto Cão' (37.01% increase), in comparison of the stems of control vines. Treatment of vines with chitosan led to an increased TTC in the shoots of both varieties, with 114.48% and 23.48% increases in 'Touriga Franca' and 'Tinto Cão', respectively (Table 4).

Table 4. The total tannin content.

Variety/Tissue	Harvesting time/Treatment					
	During Veraison		After Veraison		At Complete Maturation	
	Control	Chitosan	Control	Chitosan	Control	Chitosan
'Touriga Franca'						
berry seeds	33.19±4.57 e	141.11 ± 5.95 a	48.23 ± 5.28 d	124.93 ± 14.32 b	95.35 ± 5.32 c	125.98 ± 4.54 b
berry skins	74.24 ± 3.68 b	84.78 ± 4.78 a	59.72 ± 3.78 c	82.09 ± 6.41 a	47.41 ± 3.02 d	57.11 ± 0.76 c
cluster stems	NA	NA	NA	NA	31.63 ± 3.11 a	20.52 ± 2.28 b
leaves	41.75 ± 5.42 a	36.90 ± 1.11 b	35.97 ± 3.45 b	32.47 ± 2.33 c	17.06 ± 1.15 d	32.47 ± 3.37 c
shoots	NA	NA	NA	NA	17.26 ± 1.62 b	37.02 ± 0.66 a
'Tinto Cão'						
berry seeds	129.04 ± 5.34 b	140.95 ± 2.64 a	108.50 ± 1.86 c	131.63 ± 5.54 b	106.81 ± 1.23 c	113.64 ± 4.25 c
berry skins	43.69 ± 3.61 b	47.89 ± 2.14 a	26.47 ± 5.07 d	40.40 ± 0.60 b	32.03 ± 4.17 c	47.55 ± 9.57 a
cluster stems	NA	NA	NA	NA	102.66 ± 5.28 b	140.65 ± 8.37 a
leaves	64.78 ± 10.10 c	88.19 ± 1.92 a	62.67 ± 3.86 c	77.53 ± 3.86 b	89.43 ± 8.89 a	78.72 ± 3.16 b
shoots	NA	NA	NA	NA	30.75 ± 2.44 b	37.97 ± 1.51 a

(TTC; µg epicatechin equivalents per mg dry weight) in grapevine tissues after chitosan application before and after veraison. All data were obtained from three biological replicates; means ± standard deviations within a row followed by different letters (a–e) are statistically different ($p < 0.05$; ANOVA Turkey's test). NA = not analyzed.

3.4. Effect of Chitosan Application on the Antioxidant Potential of Grapevine

In this study, three assays were used for assessing the antioxidant potential of grapevine samples, namely DPPH, FRAP, and CUPRAC [35,38,40]. Data in Figure 2 shows that chitosan application did not affect the antioxidant potential of the seeds of the two varieties studied. Among skins, the antioxidant potential remained unchanged for 'Touriga Franca' (Figure 2A,C,E), while it increased for 'Tinto Cão': 11.54%, 27.06%, and 32.51% for the DPPH (Figure 2B), FRAP (Figure 2D) and CUPRAC (Figure 2F), respectively. With respect to the control, FRAP and CUPRAC increased in the stems of 'Touriga Franca' (5.30% and 6.10%, respectively; Figure 2C,E) and 'Tinto Cão' (25.74% and 31.53%, respectively; Figure 2D,F) following chitosan application, while no change was recorded for the DPPH (Figure 2A,B). Chitosan-treated leaves exhibited more antioxidant activities than control leaves: for 'Touriga Franca', 7.72%, 13.60%, 10.32% increases for the DPPH (Figure 2A), FRAP (Figure 2C) and CUPRAC (Figure 2E), respectively, were recorded. For 'Tinto Cão', 1.23% ($p > 0.05$), 9.70%, 8.70% increases for the DPPH (Figure 2B), FRAP (Figure 2D) and CUPRAC (Figure 2F), respectively, were recorded; however, these increases were non-significant. Chitosan application had no effect on the FRAP and the CUPRAC of the shoots of both varieties (Figure 2C–F). In terms of DPPH, a decreased

activity (8.33%) was recorded in the shoots of chitosan-treated ‘Touriga Franca’ (Figure 2A), while an increased activity (11.80%) was recorded in the shoots of chitosan-treated ‘Tinto Cão’ (Figure 2B).

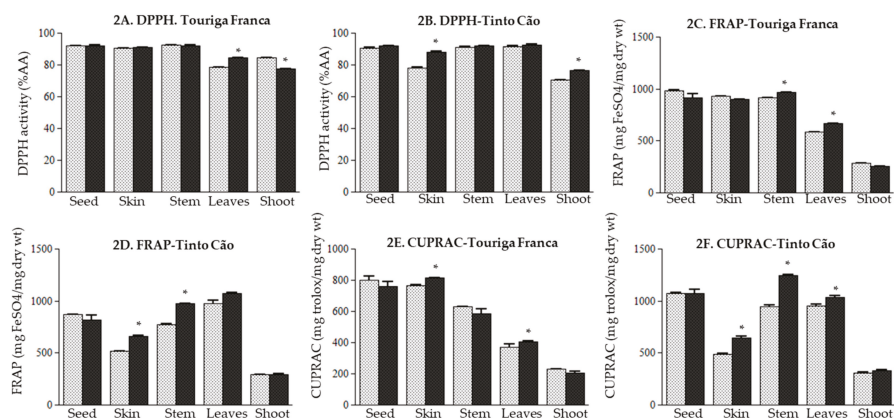


Figure 2. (A,B) 2,2-diphenyl-1-picrylhydrazyl activity (DPPH), (C,D) Ferric Reducing Antioxidant Power (FRAP), (E,F) Copper Reducing Antioxidant Capacity (CUPRAC) in the tissues of grapevine after application of chitosan before and after veraison. Tissues were collected at complete maturation of the berries. Gray column = control samples; black column = chitosan-treated samples. All data were obtained from three biological replicates; standard deviations and the levels of significance ($p < 0.05$; t -test) are represented by error bars and asterisks above the columns, respectively.

3.5. Effect of Chitosan Application of the Expression of Genes of the ROS Producing and Scavenging Pathway

A qRT-PCR was used for assessing the effect of chitosan on 10 genes of the ROS pathway in grapevine tissues collected at complete berry maturation.

With one exception (*Fe-SOD* remained unaffected in the shoots of ‘Tinto Cão’), chitosan upregulated or tended to upregulate the expression of the genes *Cu/Zn-SOD* and *Fe-SOD* in all tissues of ‘Touriga Franca’ and ‘Tinto Cão’ (Figure 3). The most important changes were observed for *Fe-SOD* in the stems of ‘Touriga Franca’ (2.0-fold increase; Figure 3A); and for *Cu/Zn-SOD* in the leaves of ‘Touriga Franca’ (3.3-fold increase), leaves of ‘Tinto Cão’ (3.0-fold increase), stems of ‘Touriga Franca’ (7.4-fold increase), and stems of ‘Tinto Cão’ (7.1-fold increase) (Figure 3C,D).

CAT expression was also upregulated in all the tissues of grapevine after application of chitosan, with 1.3 to 7.6-fold increases recorded. The magnitude of the change was particularly important in the skins (3.2 and 4.3-fold for ‘Touriga Franca’ and ‘Tinto Cão’, respectively) and the shoots (7.6 and 5.0-fold for ‘Touriga Franca’ and ‘Tinto Cão’, respectively) (Figure 4A,B). A varietal effect was observed for *GR* expression, which following chitosan application, was upregulated in the tissues of ‘Tinto Cão’ (with the exception of a non-significant effect in the shoots), and remained unchanged in the tissues of ‘Touriga Franca’ (with the exception of a 1.3-fold increase in the skins) (Figure 4E,F). For both varieties, *Grx* expression was upregulated in the seeds, skins, and leaves, downregulated in the stems, and remained unaffected in the shoots. The highest changes were recorded in the berries, with 5.1-fold and 5.0-fold increases in the skins and seeds of ‘Touriga Franca’ (Figure 4G), and 3.0-fold and 2.7-fold increases in the skins and seeds of ‘Tinto Cão’ (Figure 4H). In contrast to *CAT*, *GR*, and *Grx*, the expression of *APX* was downregulated in all the tissues of both varieties upon chitosan treatment, although a non-significant effect was observed for the shoots (Figure 4C,D). The highest decreases were recorded in the skins and leaves (4.0–4.5-fold changes) and in the stems (5.2–8.1-fold changes).

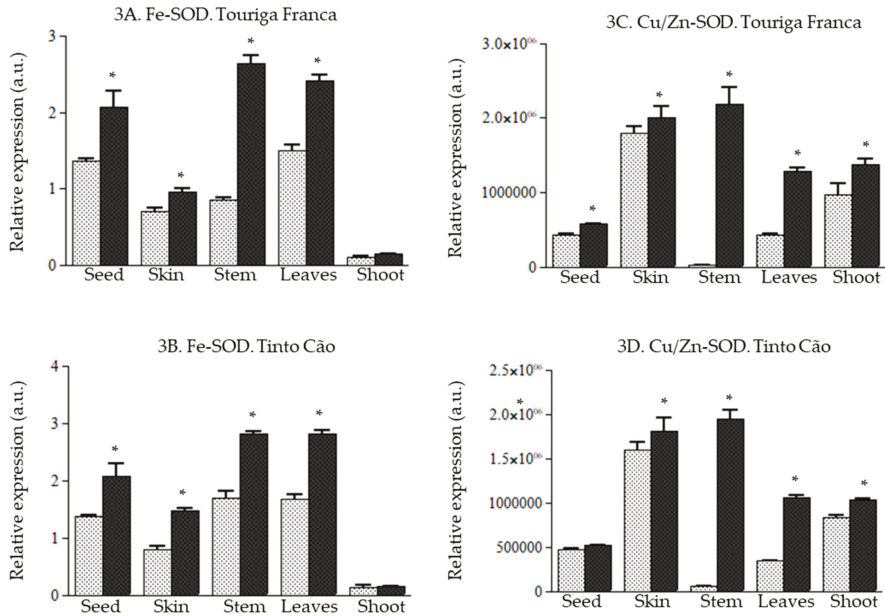


Figure 3. Effect of chitosan on *Fe-SOD* (Fe superoxide dismutase) and *Cu/Zn-SOD* (Cu/Zn superoxide dismutase) genes in the tissues of grapevine (A). relative expression of *Fe-SOD* gene in Touriga Franca, (B). relative expression of *Fe-SOD* gene in Tinto Cão, (C). relative expression of *Cu/Zn-SOD* gene in Touriga Franca, (D). relative expression of *Cu/Zn-SOD* gene in Tinto Cão. Tissues were collected at complete maturation of the berries. Gray column = control samples; black column = chitosan-treated samples. All data were obtained from three biological replicates; standard deviations and the levels of significance ($p < 0.05$; t -test) are represented by error bars and asterisks above the columns, respectively.

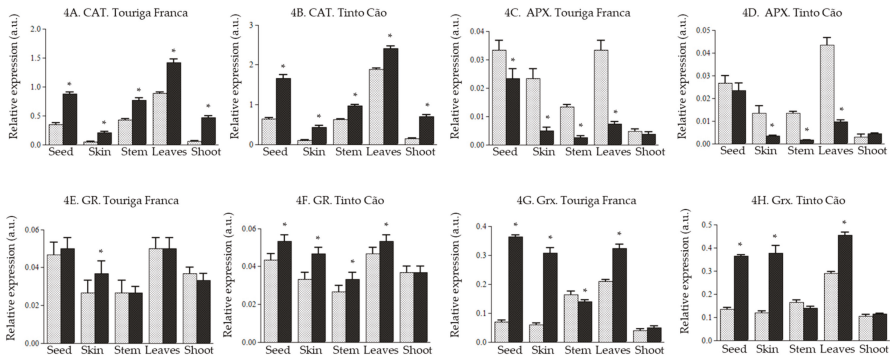


Figure 4. Effect of chitosan on *CAT* (catalase), *APX* (ascorbate peroxidase), *GR* (glutathione reductase), and *Grx* (glutaredoxin) genes in the tissues of grapevine (A). relative expression of *CAT* gene in Touriga Franca, (B). relative expression of *CAT* gene in Tinto Cão, (C). relative expression of *APX* gene in Touriga Franca, (D). relative expression of *APX* gene in Tinto Cão, (E). relative expression of *GR* gene in Touriga Franca, (F). relative expression of *GR* gene in Tinto Cão, (G). relative expression of *Grx* gene in Touriga Franca, (H). relative expression of *Grx* gene in Tinto Cão. Tissues were collected at complete maturation of the berries. Gray column = control samples; black column = chitosan-treated samples. All data were obtained from three biological replicates; standard deviations and the levels of significance ($p < 0.05$; t -test) are represented by error bars and asterisks above the columns, respectively.

Following chitosan application, there was an upregulation of the expression of *AO* (1.4–3.0-fold increases) and *POD* (1.4–31.5-fold increases) genes in all tissues of ‘Touriga Franca’ and ‘Tinto Cão’ (Figure 5C–F). In particular, the expression of *POD* was 31.5 and 11.0 times higher in the leaves of chitosan-treated ‘Touriga Franca’ and ‘Tinto Cão’, respectively, compared to that in the leaves of control vines (Figure 5E,F). The genes *Rboh* (Figure 5A,B) and *PPO* (Figure 5G,H) were also studied upon chitosan application and, with few exceptions of non-significance (*Rboh* in the stems and shoots of both varieties and *POP* in the stems and seeds of ‘Tinto Cão’), all showed an upregulation in the tissues of ‘Touriga Franca’ and ‘Tinto Cão’ i.e., 1.3–1.4-fold increase for *Rboh* and 1.5–5.0-fold increase for *PPO*, with the highest changes noticed in the shoots.

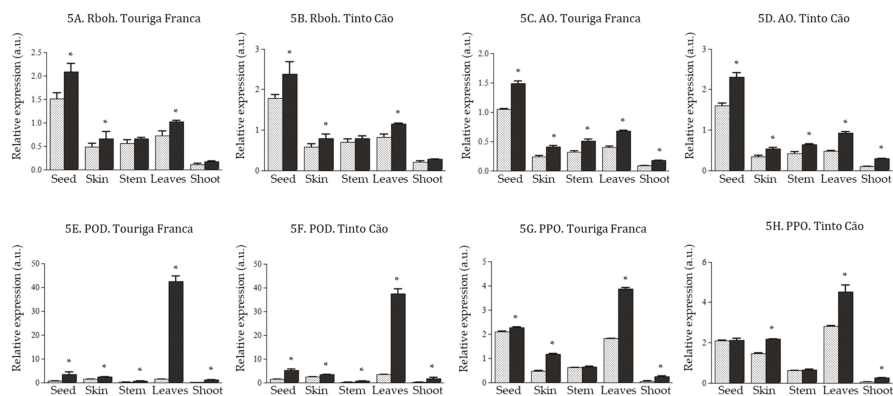


Figure 5. Effect of chitosan on *Rboh* (respiratory burst oxidase), *AO* (amine oxidase), *POD* (peroxidase), and *PPO* (polyphenol oxidase) genes in the tissues of grapevine (A). relative expression of *Rboh* gene in Touriga Franca, (B). relative expression of *Rboh* gene in Tinto Cão, (C). relative expression of *AO* gene in Touriga Franca, (D). relative expression of *AO* gene in Tinto Cão, (E). relative expression of *POD* gene in Touriga Franca, (F). relative expression of *POD* gene in Tinto Cão, (G). relative expression of *PPO* gene in Touriga Franca, (H). relative expression of *PPO* gene in Tinto Cão). Tissues were collected at complete maturation of the berries. Gray column = control samples; black column = chitosan-treated samples. All data were obtained from three biological replicates; standard deviations and the levels of significance ($p < 0.05$; t -test) are represented by error bars and asterisks above the columns, respectively.

4. Discussion

Improved fruit yield and quality is a high priority in vitiviculture, and the use of elicitors holds the potential to help achieve this aim, while decreasing the reliance on pesticides and fertilizers. Previous experiments have instructively illustrated the feasibility of using elicitors to raise the levels of secondary metabolites in grapes. In the previous studies by Portu et al. solutions of methyl jasmonate (10 mM) and chitosan (0.03% in acetic acid 0.01 M) were sprayed over the leaves of *Vitis vinifera* L. ‘Tempranillo’ (200–400 mL per vine) at veraison, and a week later, methyl jasmoate was applied which led to an increased accumulation of anthocyanins and stilbenes in the grape and wine, while chitosan treatment did not have a substantial impact on phenolic compounds [16,42]. There is, however, a report that the synthesis of anthocyanins and stilbenes is promoted by chitosan (50 µg/mL in 0.1% v/v acetic acid) in *Vitis vinifera* L. ‘Barbera’ grape cell suspensions [43]. For *Vitis vinifera* L. ‘Cabernet Sauvignon’, application of chitosan in three weekly intervals from fruit set to harvest had no effect on the TPC of the grapes, and caused a reduction in extraction of phenolics into the wine during vinification [44]. Thus, chitosan efficacy needs to be validated for each variety/clone in consideration with the dosage, timing, and mode of application. In this study, these variables were studied with a view to establishing the most appropriate conditions for ensuring high levels of phenolic compounds in all tissues. Thus,

the whole vine was sprayed with a solution of 0.1% (w/v) chitosan in 0.01 M aqueous acetic acid before and after veraison.

The data in Tables 2 and 3 show that during veraison, after veraison, and at complete maturation, chitosan raised the TPC and the TAC of the leaves and berry's skins of 'Touriga Franca' and 'Tinto Cão', with one exception of slight decrease in TAC in the skins of 'Touriga Franca' at the maturity. The TTC increased in the seeds of chitosan-treated 'Touriga Franca' and 'Tinto Cão' at all harvesting times, while an opposite response was observed for the leaves (Table 4), suggesting a varietal dependency. Chitosan is known to stimulate the activity of phenylalanine ammonia lyase, suggesting an induction of the phenylpropanoid pathway [45,46], which would explain the TPC and TAC in this study. The varietal difference observed for the TTC in the leaves might be explained by a preferential activation of the enzymes responsible for anthocyanin synthesis when chitosan is applied on 'Touriga Franca' leaves, limiting the substrates for the production of tannins [47]. The positive effects of chitosan on the TPC, TAC, and TTC indicate that the treatment could also be regarded as a useful methodology in increasing resistance to some pathogens, as observed with grey mould [45] and powdery mildew infections [18].

Despite increased TPC, TAC and TTC in the berry's skins at maturity following chitosan application, the antioxidant potential measured by DPPH, CUPRAC and FRAP increased in 'Tinto Cão', and remained unchanged in 'Touriga Franca' (Figure 2), which again indicates a varietal dependency. The synthesis and accumulation of phenolic compounds in plants in response to the environmental conditions and cultural practices, are dependent on various factors including the cultivars [2,4,48,49]. This may contribute towards an increased or decreased antioxidant potential [50]. Indeed, chitosan and its formulations are reported to improve and/or reduce the antioxidant potential depending on the crop studied e.g., maize, soybean, jaborandi, Thymus and tobacco [51–54]. Overall, there was only a loose correlation between TPC/TAC/TTC and the DPPH. On the other hand, a good correlation was found between TPC/TAC/TTC and FRAP/CUPRAC in the stems and the leaves (data not shown).

ROS, such as O_2^- (superoxide), are free radicals derived from oxygen that is continuously produced during normal aerobic respiration in plants or in response to injury and can be unfavorable to plant fitness [55–59]. Several endogenous antioxidants are produced by plants to slow or stop the production of ROS. SOD (including *Cu/Zn-SOD* and *Fe-SOD*) plays a major role in scavenging ROS in several plants [60–63], including grapevine [64,65]. The present study demonstrated an upregulation of the SOD gene upon chitosan application. Overall, there was an upregulation of genes of the ROS pathway (*Fe-SOD*, *Cu/Zn-SOD*, *CAT*, *GR*, *Rboh*, *AO*, *PPD*, and *PPO*) in all the tissues of both varieties, with the exception of *APX* in all tissues and *Grx* in the stems of both varieties (Figures 3–5). Notably, *POD* increased following application of chitosan, 51.5 and 11.0-fold in the leaves of 'Touriga Franca' and 'Tinto Cão', respectively (Figure 5E), suggesting a crucial role for *POD* in the ROS pathway. Recently, it was reported that maize plants are able to cope with drought, following application of chitosan because of the potentiation of the activities of the antioxidant enzymes SOD, *CAT*, *APX*, *GR* and guaiacol peroxidase [46]. This shows that chitosan may induce defense responses in grapevine by activating the ROS pathway, as suggested in some earlier studies [66–69]. Exogenous chitosan reportedly interacts with chitin synthase, and chitin deacetylase in plants, to produce chitosan oligomers that are responsible for signal perception in cells, leading to the activation of various genes, including genes of the ROS pathway [50,70,71]. Elicitor molecules have been used to stimulate a broad range of responses including defense mechanism [72] in field conditions as well as in vitro cultures for elicitation of secondary metabolites [73–76]. However, it was reported earlier that acetic acid alone itself may exert some effect [77]. This aspect will be included in the next study as an extra control sample.

A tissue-specific analysis of the response of grapevine to chitosan in this study shows that chitosan might also act by influencing the transport of secondary metabolites in the vine. At maturity, the TPC, and the TAC remained unchanged in the shoots (Tables 2 and 3), suggesting that shoots are not implicated in the transport of phenolic acids and anthocyanins to the berries. Instead, chitosan seems to stimulate the transfer of leaf anthocyanins to the berries via the stems. At maturity, that transfer seems complete in the case of 'Tinto Cão', since the TAC and TPC increased in the stems and skins,

but remained unchanged in the leaves; and incomplete in the case of ‘Touriga Franca’ since a high TAC and TPC is still measured in the leaves. The response of the TTC in the leaves depended on the variety and the harvesting time (Table 4), suggesting that leaf tannins are not transported to the berries. Instead, shoot tannins seem to be transferred to the berries. Finally, the data in Tables 2–4 and Figure 2 shows that there is a prospect for adding values to grapevine stems and leaves via the application of chitosan. In general, the levels of phenolics and the antioxidant potential were constantly high in the stems and leaves of both varieties at all harvesting times.

5. Conclusions

Present study steps ahead towards a better understanding of chitosan–*Vitis vinifera* L. interaction and elicitation mechanism at the molecular level and suggests its application for value addition in viticulture practices.

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Correction

Correction: Biswas, M.S. et al. Inactivation of Carbonyl-Detoxifying Enzymes by H₂O₂ Is a Trigger to Increase Carbonyl Load for Initiating Programmed Cell Death in Plants. *Antioxidants* 2020, 9, 141

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The author wishes to make the following correction to this paper [1]. The H₂O₂ concentration of one experimental condition was mistyped in the index in Figure 7A,B. The correct Figure 7 is as follows:

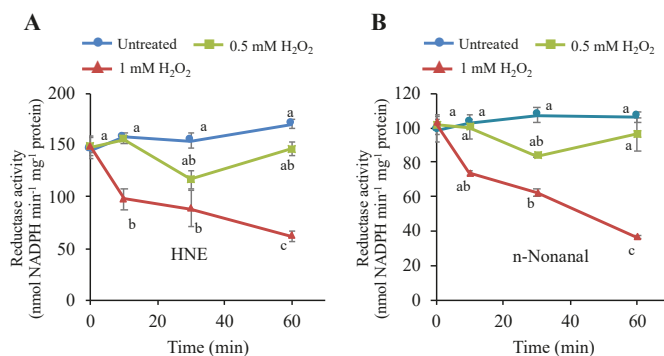


Figure 7. Effects of H₂O₂ on the NADPH-dependent HNE-reducing and *n*-nonanal-reducing activities in tobacco BY-2 cells. Four-d cultured cells were treated with H₂O₂ at 0.5 mM and 1 mM, as in Figure 1. Then, cells were harvested at the indicated time point, and proteins were extracted as in the Materials and Methods section. The reductase activities for (A) HNE and (B) *n*-nonanal were determined as in the Materials and Methods section. Each point represents the mean of three independent experiments and the error bars of the SEM. Different letters represent significantly different values ($p < 0.05$ on Tukey test).

These changes have no material impact on the discussion and conclusions of the paper. The authors would like to apologize for any inconvenience caused to the readers by these changes.

Conflicts of Interest: The authors declare no conflicts of interest.

Reference

1. Biswas, M.S.; Terada, R.; Mano, J. Inactivation of Carbonyl-Detoxifying Enzymes by H₂O₂ Is a Trigger to Increase Carbonyl Load for Initiating Programmed Cell Death in Plants. *Antioxidants* **2020**, *9*, 141. [[CrossRef](#)] [[PubMed](#)]



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Review

On the Origin and Fate of Reactive Oxygen Species in Plant Cell Compartments

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Abstract: Reactive oxygen species (ROS) have been recognized as important signaling compounds of major importance in a number of developmental and physiological processes in plants. The existence of cellular compartments enables efficient redox compartmentalization and ensures proper functioning of ROS-dependent signaling pathways. Similar to other organisms, the production of individual ROS in plant cells is highly localized and regulated by compartment-specific enzyme pathways on transcriptional and post-translational level. ROS metabolism and signaling in specific compartments are greatly affected by their chemical interactions with other reactive radical species, ROS scavengers and antioxidant enzymes. A dysregulation of the redox status, as a consequence of induced ROS generation or decreased capacity of their removal, occurs in plants exposed to diverse stress conditions. During stress condition, strong induction of ROS-generating systems or attenuated ROS scavenging can lead to oxidative or nitrosative stress conditions, associated with potential damaging modifications of cell biomolecules. Here, we present an overview of compartment-specific pathways of ROS production and degradation and mechanisms of ROS homeostasis control within plant cell compartments.

Keywords: cell wall; chloroplasts; cytoplasmic membrane; cytosol; glyoxysomes; mitochondria; peroxisomes; plant cell; reactive oxygen species

1. Introduction

Subcellular compartmentalization in eukaryotic cells forms the basis for highly selective separation of biochemical reactions and metabolic pathways, and delimit their potential mutual interferences [1]. Current knowledge indicates that within each compartment of eukaryotic cells, specific redox characteristics have evolved [2,3]. Unique redox aspects of different cell compartments are integrated at the molecular level by a group of small diffusible and reactive molecules, reactive oxygen species (ROS), which provide communication between intracellular compartments [4]. ROS are produced in all forms of aerobic life by divergent reactions including a partial reduction of the molecular oxygen [5,6]. Previously, ROS were described as toxic by-products of the aerobic metabolism and their increased levels were connected with multiple forms of cellular damage mediated by oxidative modifications of cell biomolecules. Nowadays, the accumulated evidence suggest that ROS function as key redox signaling and effector molecules in vital biological processes such as cell growth, differentiation, proliferation and responses to a wide spectrum of external stimuli [7,8]. In plant cells, signaling networks of ROS are closely connected to central physiological processes of energy generation and consumption such as respiration, photosynthesis and photorespiration and in plant responses to abiotic and biotic stress conditions [9–11].

In general, signaling functions of ROS are related to tightly regulated site- and time-specific modulations of ROS levels, whereas uncontrolled ROS accumulation as a consequence of either induced

production or defective catabolism, or their combination, is associated with damaging oxidative effect leading to cellular damages or even to the cell death [12]. ROS can modify proteins structures and functions through oxidative reactions namely with protein thiol groups or iron-containing clusters. On the other hand, toxic effects of accumulated ROS are also exploited as chemical defense within immune repertoires of diverse organisms against invading pathogens, such as the well-described ROS burst leading to the plant cell death within the plant hypersensitive response to biotrophic pathogens [13].

Individual ROS show a high variability in their chemical properties, reactivity and involvement in redox signaling pathways [14]. As a general rule, radical forms of ROS are more reactive compared to non-radical ROS. Moreover, ROS reactivity is significantly affected by presence of free Fe^{2+} ions involved in formation of highly reactive hydroxyl radical in a pathway termed the Fenton reaction, which results in production of hydroxyl radicals as the most reactive ROS form with a single unpaired electron, capable to react with virtually all biomolecules. High reactivity of hydroxyl radicals leads to subsequent cellular damages, including changes in protein structures, lipid peroxidation and membrane destruction [15]. However, recent findings support the suggested role of the hydroxyl radical as being more than a destructive agent, as its oxidative properties can facilitate seed germination, growth, stomatal closure, reproduction, immune responses, plant cell death and adaptations to stress conditions [16].

Singlet oxygen ($^1\text{O}_2$) is formed in photosynthetic membranes in reactions between triplet-state chlorophyll molecules and the molecular oxygen [17]. Observed half-life of $^1\text{O}_2$ is approximately 3 μs and it has been shown that a minor portion of singlet oxygen is able to diffuse to a distance of several nanometers, where it can react with diverse biomolecules and potentially mediate its signaling functions [18,19]. Enzyme activity of lipoxygenase represents an additional source of $^1\text{O}_2$ in plants [20]. Highly reactive $^1\text{O}_2$ can be efficiently removed by quenching with low-molecular lipophilic compounds like β -carotene and tocopherol, or by a scavenging action of protein D1 in photosystem II [21].

Superoxide anion radical (O_2^-) is a highly reactive nucleophilic species with a half-life about 1 μs , which often serves as an initiator of reaction cascades generating other ROS, namely hydrogen peroxide. Superoxide occurs with a similar half-life as singlet oxygen, but strongly differs in its target reaction partners. Polyunsaturated fatty acid within thylakoid membranes are considered the main reaction partners of $^1\text{O}_2$, whereas superoxide preferentially reacts with other radical compound including nitric oxide and components of protein hem-containing and non-hem iron centers [22].

Hydrogen peroxide (H_2O_2) is the least reactive non-radical ROS. Its signaling pathways are mediated mainly through interactions with proteins containing redox-sensitive moieties, such as metal centers or cysteine thiols, whose oxidation controls protein biological activity [23,24]. H_2O_2 can diffuse across lipid membranes through the aquaporin channels, which is a crucial feature for H_2O_2 intracellular signaling [25]. In plants, it was demonstrated that H_2O_2 transport could also be mediated by aquaporin homologues from the plasma membrane intrinsic factor (PIP) family and by the tonoplast intrinsic protein (TIP) [26]. Excess H_2O_2 is known to trigger chloroplast and peroxisome autophagy and programmed cell death in plants [27].

2. Subcellular Localization and Functions of ROS Production in Plant Cells

On a quantitative scale, in contrast to animals where mitochondria play a predominant role in the ROS biogenesis in most cell types, plant cells generate ROS in a highly variable manner by multiple pathways depending on the plant tissue, developmental stage and external conditions [9,28]. ROS metabolism has been extensively studied in all plant cell compartments including cell wall, apoplast, plasma membrane, cytosol, mitochondria, chloroplasts, peroxisomes and glyoxysomes [29]. More stable ROS species like hydrogen peroxide can even diffuse, cross cell membranes and transport their signaling effects into other compartments (Figure 1). Moreover, ROS cooperate and cross-talk with signaling pathways of plant hormones such as abscisic, jasmonic and salicylic acid or ethylene [30,31].

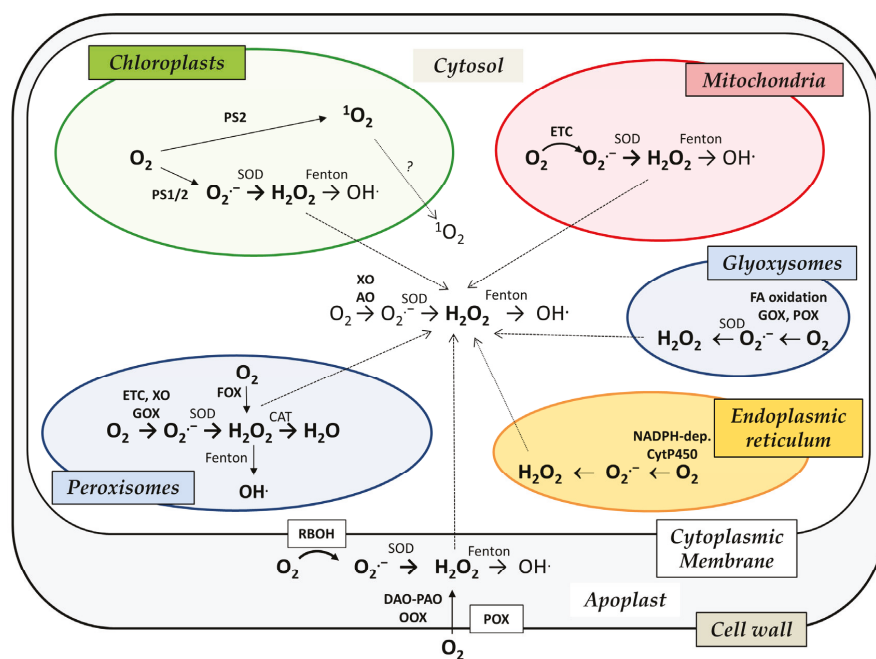


Figure 1. Schematic overview of reactive oxygen species (ROS) sources in plant cell compartments. AO, aldehyde oxidase; CAT, catalase; DAO, diamine oxidase; GOX, glycolate oxidase; ETC, electron transport chain; FOX, flavin oxidases; OOX, oxalate oxidase; PAO, polyamine oxidase; POX, peroxidases; PS, photosystem; RBOH, NADPH oxidase; SOD, superoxide dismutase; XO, xanthine oxidase/dehydrogenase.

2.1. Cell Wall, Apoplast and Cytoplasmic Membrane

The cell wall, apoplast and cytoplasmic membrane form crucial boundary compartments of plant cells securing their integrity vital for intracellular processes and communication with external environment including interactions with microbial symbionts or pathogens [32,33]. An important portion of ROS produced under physiological conditions and in plant responses to stress stimuli is formed within these compartments, which represent the major site of ROS production in plant-pathogen interactions [34]. Intensive ROS production termed as the oxidative burst is activated upon the recognition of pathogens or other biotic stress stimuli by immune receptors localized in plasma membrane, which triggers signaling pathways leading to transient localized increase of ROS [35]. In plants, receptor-like kinases (RLKs) represent key elements of the communication between the external environment and the cellular interior. Apoplastic ROS production occurs often following activation of RLK signaling in a wide array of cellular processes [36]. Thus, intricate interconnections exist between RLKs, extracellular ROS generation and ROS signal transduction and perception within the plant ROS sensing machinery. ROS can further diffuse as H_2O_2 inside the plant cells via membrane aquaporin channels [25]. Furthermore, apoplastic ROS are key players in physiological processes such as cell growth, which is particularly dependent on ROS involvement in increasing extensibility of the cell wall during root hair and pollen tube elongation, as well as in leaf growth [37,38]. It is known that hydroxyl radical promotes cell elongation by loosening the cell wall by oxidative cleavage of pectins and xyloglucans. On the other hand, H_2O_2 can initiate polysaccharide cross-linking and protein disulfide bond formation which restricts elongation growth. Several apoplastic and plasma-membrane proteins are involved in the regulation of apoplastic balance between hydroxyl radical and H_2O_2 , which

regulates cell expansion by direct interactions with cell wall components and also ROS interactions with intracellular transcriptional networks and cytoskeleton [37].

Plant-specific class III peroxidases, as members of a large multigene family of peroxidases (POXs, EC 1.11.1.7) are localized in the plant cell wall and comprise a significant source of apoplastic ROS [39]. Peroxidases contain eight conserved cysteine residues, one or more glycosylation sites, and a signal peptide which divert them into cell wall, apoplast or vacuole [40]. Peroxidases show a broad substrate specificity and can operate either in ROS-consuming or ROS-generating mode through reaction mechanisms based on peroxidation or hydroxylation cycle, respectively [41–44]. In ROS-consuming peroxidase mode, one oxygen atom from H_2O_2 is transferred to the hem group of the ferric enzyme in the ground state and the first intermediate is formed. Then an electron from a reducing substrate generates the second intermediate and the cycle is finished by regeneration of ferric enzyme [41]. H_2O_2 reduction catalyzed by peroxidases occurs with simultaneous oxidation of electron donors such as phenolic compounds [45]. Yet, the presence of certain reductants is not the only condition for peroxidases functions; the second one is the apoplast alkalization. In ROS-producing mode, peroxidases produce ROS due to the oxidase cycle forming the third intermediate. Plant class III peroxidases are responsible for production of apoplastic ROS and involved in stress signaling after pathogen attack, wounding and under abiotic stress stimuli [46]. Beside their function in signaling pathways, POXs are also involved in polymerization of suberin and lignin, important compounds of passive plant defense barriers [42] in a cross-talk with signaling pathways of jasmonic acid [47].

The apoplastic oxidative burst occurs as a result of ROS-producing activities of several groups of enzymes including cell wall peroxidases, plasma membrane NADPH oxidases, amine oxidases, lipoxygenases, oxalate oxidases and quinone reductases [48,49]. Lower apoplastic pH compared to the cytosolic pH has a crucial control effect on redox properties of protein cysteine thiols and overall redox conditions. Another important feature of the apoplastic space is decreased antioxidant capacity as a consequence of low abundance of low-molecular weight antioxidants such as glutathione and ascorbate. Decreased antioxidant capacity in the apoplast is critical for activation of ROS signaling pathways. Thus, the apoplastic oxidative burst results in enhanced apoplastic oxidation state, while cytoplasm stays reduced. Redox gradients formed across the cellular membrane allow for differential regulation of proteins on cell surface such as receptors important for stress signal perception [50]. Oxalate oxidase and amine oxidases belong to ROS sources found in the apoplast. Oxalate oxidase (EC 1.2.3.4) catalyzes oxidation of oxalate to H_2O_2 and CO_2 , and its gene expression is strongly induced by increased levels of salts and salicylic acid [51]. Copper-containing amine oxidases (CuAOs, EC 1.4.3.6) together with polyamine oxidases (PAOs, EC 1.5.3.3) are abundant plant proteins involved in the catabolism of polyamines, important plant growth regulators [52,53]. CuAOs are homodimer proteins contain copper ion and 2,4,5-trihydroxyphenylalanine quinone cofactor in both subunits. PAOs are monomeric and require flavin adenine nucleotide for their enzymatic activity. Enzymes of these classes are responsible for di- and polyamine catabolism, catalyzing oxidative deamination of their substrates, producing appropriate aldehydes, ammonia and H_2O_2 [54]. Produced apoplastic H_2O_2 is then used in modifications of the cell wall under physiological or stress conditions [55] or can function as a signal molecule, e.g., in the opening of Ca^{2+} channels [56,57].

Germins, a group of proteins with oxalate oxidases activity, represents another contribution to apoplastic ROS levels. Germins are apoplastic proteins with an oligomeric structure appearing at the onset of germination in plants, and are involved in plant development and defense responses to pathogen attack [58]. Oxalate oxidases contain manganese in their structure and are responsible for oxidation of oxalate. Moreover, plants with higher oxalate oxidase activity were found more resistant to plant pathogens [59]. Apart from germins, the germin protein family includes germin-like proteins (GLPs) which lack the oxalate oxidase activity but were shown to possess superoxide dismutase (SOD) activity [60].

NADPH oxidases (termed NOXs in animals) are family of proteins catalyzing production of superoxide in cytoplasmic membrane, termed as respiratory burst oxidase homologues (Rboh) in

plants [61,62]. Rbohs are considered the major ROS sources within the oxidative burst orchestrated in plant responses to pathogen attacks [63]. Within molecular events following the recognition of microbe- and damage-associated molecular patterns (MAMPs and DAMPs, respectively), the ROS is often observed as a biphasic ROS increase [64]. Rbohs were believed to specifically localize in plasma membrane [62]; however, recently, NtRbohD was observed localized in lipid rafts, membrane micro-domains in tobacco cells that can be coupled to other membrane components. It seems that different sub-membrane distribution of Rbohs is critical for their functions [65,66].

Protein structures of plant NADPH oxidases contain six conserved transmembrane helices, where the 3rd and 5th helix bind two hem groups through four histidine residues. The hem groups are essential for transfer of electrons across the membrane to the molecular oxygen as the acceptor in the extracellular space [67]. N-terminal Ca^{2+} binding motifs and cytosolic FAD- and NADPH-binding domains in C-terminal regions are another typical structural feature of plant Rbohs. Plant Rbohs share a high similarity in their amino acid sequences of regions spanning membrane C-terminal cytosolic regions, whereas N-terminal cytosolic regions of NADPH oxidases show higher variability. NADPH oxidases require Ca^{2+} binding and phosphorylation to become fully activated; however, the exact activation mechanism has not been yet deciphered. These two activation events were suggested to occur synergistically [68], whereas other studies indicated the phosphorylation occurs as the first step followed by Ca^{2+} binding [69]. Both Rbohs activity and localization are controlled through phosphorylation by protein kinases from several families, including calcium-dependent and calmodulin-like domain protein kinases (CDPKs) [66,70]. The product of NADPH oxidase catalyzed reaction, superoxide anion radical is a membrane impermeable species due to its negative charge; however, under decreased pH conditions the superoxide is protonated and can pass through the membranes. In this way, the pH status can affect compartmentalization of created product: in plants, where the physiological range is found around pH 5, approximately 16% of superoxide produced by Rbohs is transformed to a membrane permeable hydroperoxyl form [62].

Superoxide dismutases (SODs) constitute the first line of ROS catabolism with a crucial function in all compartments where superoxide radicals are produced [51]. Specifically in the apoplast, Cu-Zn SOD were reported to have a physiological role in the lignification process in the vascular tissue of spinach hypocotyls [71]. It is noteworthy that concentrations of the key redox regulators glutathione (GSH) in the apoplast are orders of magnitude lower compared to the cytosol, where GSH is found in the millimolar range [72]. Furthermore, most apoplastic ascorbic acid (AA) in physiological conditions is oxidized, while over 90% of total leaf tissue AA is found in the reduced form [73], and oxidized AA derivatives, such as oxalic acid, were suggested to function as signaling molecules [74]. On the other hand, AA has been suggested to be the most important antioxidant for the detoxification of ROS in the apoplast in stress conditions [75]. Therefore, it seems that the low abundance of low molecular antioxidants in the apoplast might extend the half-life of H_2O_2 , and thus enable H_2O_2 diffusion and propagation of ROS signals within and from the apoplast compartment.

2.2. Cytosol

The cytosol has not been regarded as an important compartment in plant ROS production. Cytoplasmic NADPH is a central component of redox maintenance in the cytosol as it controls the thiol/disulfide status and participates in the production of reducing substrates for antioxidant enzymes involved in ROS catabolism. On the other hand, NADPH also supplies electrons to plasma membrane NADPH oxidases as ROS producing enzymes [76]. Xanthine oxidase/dehydrogenase (XO; EC 1.2.1.37) and aldehyde oxidase (AO; EC 1.2.3.1) are molybdenum- and flavin-containing enzymes, which play important roles in plant purine metabolism and phytohormone biosynthesis, respectively. Both XO and AO contribute to increased ROS production in multiple animal pathologies and were suggested as potential sources of ROS also in plants. Interestingly, tomato and *Arabidopsis thaliana* XO has been shown to produce superoxide but not H_2O_2 in vitro, whereas its animal counterpart can produce both superoxide and H_2O_2 [77]. Superoxide-generating activity was lost in *A. thaliana* Atxdh1 T-DNA

insertion mutant and RNA interference lines, providing molecular evidence that plant XO generates superoxide. Unlike XO, AO produces only H_2O_2 , which was surprisingly found not sensitive to inhibition by diphenyl iodonium, a known inhibitor of flavin-containing enzymes. Application of abscisic acid or water-stressed induced ROS production and AO and XO upregulation both in leaves and roots, indicating that plant AO and XO might be significant sources for ROS accumulation under stress condition [77].

Recently, AO isoforms AO2 and AO3 and their superoxide-producing activity were observed in *Nicotiana benthamiana* plants inoculated with the tomato bushy stunt virus, suggesting involvement of plant AO in defense mechanisms against viral infection [78]. It seems that another specific AO isoforms AsAO4 has a critical function in delaying senescence in siliques by catalyzing aldehyde detoxification under stress condition [79]. AAO4 enzyme oxidizes an array of aromatic and aliphatic aldehydes and generates superoxide and H_2O_2 in an aldehyde-dependent manner. Interestingly, XDH1 can play spatially specific dual and opposing roles in ROS metabolism in *A. thaliana* defense responses to powdery mildew [80]. In leaf epidermal cells, XDH1 functions as an oxidase, together with the NADPH oxidases RbohD and RbohF, to generate superoxide. Superoxide dismutation to H_2O_2 and subsequent accumulation of H_2O_2 in the fungal haustorial structures within infected epidermal cells contributes to constrain the haustorium, thereby contributing to powdery mildew resistance. In contrast, leaf XDH1 functions in the xanthine dehydrogenase mode producing uric acid in local and systemic tissues to scavenge H_2O_2 from stressed chloroplasts, thus contributing to plant protection from biotic stress-induced oxidative damage.

Besides its role in ROS production, the cytosol is supposed to play a key role in redox signal integration. ROS signaling from the apoplast and organelles needs to pass through the cytoplasm to achieve its effects and modulate gene expression in the cell nucleus [81]. The signal can be transmitted directly from H_2O_2 or via redox sensitive proteins to the MAPK cascade and to redox-dependent transcription factors such as heat stress transcription factors [82]. A detailed discussion of ROS signaling mechanisms in plant cells is not within the scope of this paper; nevertheless, multiple aspects of plant ROS signaling have been recently covered elsewhere [9,11,12,29,36,61,83].

2.3. Mitochondria

Mitochondrial respiration, encompassed by a directed electron flow from reduced organic substrates to the molecular oxygen through components of respiratory chain in the inner mitochondrial membrane, is inheritable associated with ROS generation [84]. Therefore, superoxide production occurs during normal operation of the respiratory chain, but its rate is highly increased in conditions of decelerated respiratory rates, e.g., by respiratory chain inhibition or limited ADP availability, resulting in a highly reduced state of mitochondrial electron transport chain. Superoxide disproportionation to H_2O_2 and O_2 is strongly accelerated by superoxide dismutase present in mitochondrial matrix [85].

In animals, respiratory complexes I and III are considered main sources of mitochondrial ROS, namely of superoxide anion radical, formed by a partial reduction of the molecular oxygen by highly reactive redox intermediates [86]. In plants, succinate-dependent H_2O_2 production in mitochondria was reported to be faster than pyruvate/malate-dependent H_2O_2 production, indicating a larger role for Complex II compared to Complex I [87]. Moreover, the ubiquinone pool might serve as another site of ROS production in plant mitochondria [88]. Using a *A. thaliana* mutant in the Complex II subunit succinate dehydrogenase 1-1, Complex II was shown to contribute to localized mitochondrial ROS production that regulates plant stress and defense responses [89].

ROS originating from plant mitochondria are known to impact both mitochondrial respiratory and cellular functions, including a variety of signaling cascades in the cell, comprising retrograde signaling, plant hormone action, programmed cell death and defense against pathogens [90]. Observed mitochondrial steady-state levels of H_2O_2 are two orders of magnitude lower compared to peroxisomes and chloroplasts. However, in specific plant tissue or developmental stages, major production and accumulation of mitochondrial ROS can occur. During seed imbibition and hydration, high rates of

mitochondrial respiration contribute as the major source of superoxide, which need to be properly balanced by the antioxidant machinery to avoid oxidative damage of biomolecules potentially resulting in defective seed germination [91]. Superoxide production in plant mitochondria can be minimized by several pathways that enable bypassing the electron transport chain. Proton leak across the membrane is facilitated by uncoupling proteins, whereas alternative oxidase (AOX) bypasses proton pumping on Complex III and IV [92]. Interestingly, overexpression of the *A. thaliana* uncoupling protein 1 in tobacco reduced ROS generation, induced the antioxidant system and increased resistance to diverse abiotic stress conditions [93,94].

2.4. Chloroplasts

Within photosynthetically active cells, chloroplasts act as another site of the major ROS production, connected with multiple redox reactions and electron transport chains localized in thylakoid membranes. Singlet oxygen ($^1\text{O}_2$) is a unique ROS species produced constitutively in plant leaves in light. Chlorophylls pigments in the antenna system and in the reaction center of photosystem II are primary sources of $^1\text{O}_2$ in plants, which is formed in a reaction between the molecular oxygen and chlorophyll molecules in the triplet state [17,95,96]. The highly reactive nature of $^1\text{O}_2$ is characterized by its half-life of about 3 μs in water solutions, but about 100 μs in hydrophobic environment of lipid membranes. A small part of singlet oxygen is able to diffuse to distances of several nanometers and it has been proposed that $^1\text{O}_2$ can be involved in the signaling of programmed cell death or light acclimation processes [18]. Production of $^1\text{O}_2$ is enhanced under light stress conditions, and lipoxygenase activity can provide an additional source of $^1\text{O}_2$ [19]. Singlet oxygen can be detoxified namely by quenching by small lipophilic antioxidants like tocopherol or β -carotene. Carotenoids are capable of effective $^1\text{O}_2$ scavenging resulting in formation of excited triplet states which return efficiently to the ground state through a heat release mechanism. Among carotenoids occurring in plant cells, β -carotene is functionally the most important [97]. Plastoquinone and D1 protein of photosystem II are known as other potential $^1\text{O}_2$ scavengers [21]. Recently, a role of $^1\text{O}_2$ produced by the chloroplast lipoxygenase was reported in leaf wounding using *A. thaliana* *lox2* mutants where $^1\text{O}_2$ produced in wounded plants is involved in lipid or protein oxidation which can act as signaling mechanisms [20].

Chloroplast superoxide anion radical is known to occur both in photosystem I and II. In photosystem I, superoxide is generated by the Mehler reaction during low NADP^+ concentrations and in presence of iron-sulfur proteins, reduced thioredoxin and ferredoxin [98]. Electron leakage to molecular oxygen on the acceptor side of photosystem II produces superoxide anion radical which after dismutation to H_2O_2 is reduced to hydroxyl radical by the non-hem iron. H_2O_2 formation by incomplete water oxidation occurs at the donor side of photosystem II, where it can be reduced to hydroxyl radical by manganese [99]. Another sources of chloroplastic ROS generation represent the oxygenase activity of ribulose-1,5-bisphosphate carboxylase/oxygenase (RuBisCO, EC 4.1.1.39) producing phosphoglycerate and phosphoglycolate that are metabolized by ongoing reactions in peroxisomes where H_2O_2 is formed [100].

Chloroplastic H_2O_2 is metabolized within glutathione-ascorbate cycle, called Foyer-Asada-Halliwell pathway, important cycle within the antioxidative mechanisms of plant cells [101–103]. Ascorbate peroxidase (APX, EC 1.11.1.11) scavenges H_2O_2 using ascorbate as an electron donor to form monodehydroascorbate (MDHA) as a first step of the pathway. Four APX isoforms are found within plant cells: thylakoid tAPX, chloroplast stromatal soluble sAPX, cytosol cAPX and glyoxysome membrane form mAPX. Then monodehydroascorbate reductase (MDHAR, EC 1.6.5.4) transforms MDHA to dehydroascorbate (DHA), which utilizes NADPH produced during photosynthesis to regenerate ascorbate. MDHAR also catalyzes the elimination of H_2O_2 from mitochondria and peroxisomes [104]. Ascorbate regeneration from DHA is mediated by dehydroascorbate reductase (DHAR, EC 1.8.5.1). Oxidized thiol groups of DHAR are then returned to a reduced state by reaction with reduced glutathione, which is in turn regenerated from the oxidized glutathione by glutathione reductase (GR, EC 1.6.4.2), enzyme with a critical role in redox pathways and essential for maintaining

reduced pool of GSH [105]. Its localization is predominantly chloroplastic; however, minor GR activity is also found in cytosol and mitochondria [106].

2.5. Peroxisomes and Glyoxysomes

Plant peroxisomes accommodate multiple key metabolic pathways such as β -oxidation of fatty acids, photorespiration, ureides metabolism and xenobiotic detoxifications. Considerable amounts of ROS are produced in peroxisomal reaction pathways, peroxisomes are therefore included among major intracellular sources of plant ROS [107]. A short electron transport chain comprising cytochrome b and NADH-dependent ferric-chelate is located in the peroxisomal membrane (EC 1.16.1.7). Superoxide is formed on cytochrome b within the peroxisomal respiration and then released into the cytosol. Xanthine oxidase (XO, EC 1.17.3.2) contributes the peroxisomal ROS pool with superoxide formed during oxidation of xanthine or hypoxanthine to uric acid [108]. During photorespiration, H_2O_2 is produced in chloroplasts as a result of RuBisCO oxygenase activity followed by reaction of glycolate oxidase in peroxisomes [100]. Cu, Zn-SOD catalyzes dismutation of superoxide produced by matrix XO as another source of peroxisomal H_2O_2 , which can be also produced by activities of several flavin oxidases. Furthermore, enzymes participating in ROS catabolism such as monodehydroascorbate reductase mediating superoxide detoxification are also present in peroxisomes [109].

Catalases (CAT, EC 1.11.1.6) are tetrameric hemoproteins catalyzing H_2O_2 decomposition to water and oxygen. CAT are primarily active in peroxisomes and glyoxysomes, the sites of high H_2O_2 generation and turnover [101]. According to their catalytic mechanisms, CAT enzymes can be categorized into two groups: monofunctional with dismutating activity and bifunctional with dismutating/peroxidatic activities. By the current enzyme nomenclature, plant CAT isoforms comprise class I present in leaves and participating in H_2O_2 scavenging during photorespiration, class II are in vascular bundles, and CAT class III as key enzymes of H_2O_2 removal in the glyoxylate cycle [110].

Glyoxysomes were considered for long as a rather specialized organelle among plant microbodies derived from a single ancestral peroxisome, are important for specific plant aerobic pathways, especially glyoxylate cycle and β -oxidation of fatty acids [111]. However, recent studies accentuate unifying features of how these dynamic organelles contribute to energy metabolism, development and responses to environmental challenges. On the functional level, the major part of peroxisomal and glyoxysomal proteins are related to fatty acid oxidation. Analysis of triglyceride metabolism in *A. thaliana* seedlings showed that only two enzymes, isocitrate lyase and malate synthase, distinguish glyoxysomes from other peroxisomes [112]. Importantly, glyoxysomes as well as all peroxisome-like organelles share a number of characteristic enzymes related to ROS metabolism [107,113,114]. Nevertheless, glyoxysomes represent essential compartments of plant cells during seed germination [111,115]. Glyoxysomes are involved in mobilization of storage lipids in germinating seeds and represent an important source of both superoxide and H_2O_2 resulting from the β -oxidation of fatty acids and the activity of enzymes such as glycolate and urate oxidases [116,117]. Similarly to peroxisomes, glyoxysomes also contain catalase and H_2O_2 -forming peroxidase.

2.6. Endoplasmic Reticulum

The endoplasmic reticulum (ER) is a dynamic organelle that fulfills many cellular functions including calcium storage, protein production and lipid metabolism. Metabolic pathways of protein biosynthesis, folding and post-translational modifications such as glycosylation, disulfide bond formation and chaperone-mediated protein folding processes are harbored in ER [118]. ER is also an essential cell compartment where oxidation, hydroxylation and deamination of cellular components or xenobiotics occur [119]. ROS generated in ER and moving the cytoplasm are partly produced in the electron transport chain located in membranes associated with ER; moreover, H_2O_2 is produced in ER for proper oxidative protein folding [120]. Stress-triggered H_2O_2 production can signal outside ER to induce components of antioxidant defense and balance the redox status of the cell [121,122]. In *A. thaliana*, ROS produced within the ER stress response pathway are involved the water stress-induced programmed cell

death [123]. Superoxide radical is a by-product of oxidation and hydroxylation processes involving cytochrome P450 and cytochrome P540 reductase in a presence of reduced NADPH [124].

3. Conclusions

In recent years, we have witnessed a substantial progress in our understanding of ROS-dependent redox signaling in plants involved in plant growth and developmental and responses to abiotic and biotic stress stimuli. It has become increasingly evident that cellular ROS signaling pathways are clearly confined in a spatio-temporal manner, similarly to that observed for other second messengers, and that ROS redox signaling is tightly connected to cellular compartmentalization, which allows organelle-specific signaling responses. Roles of low molecular antioxidants and ROS-catabolizing enzymes as important regulators of ROS levels in cellular compartments have been uncovered.

Within the plant ROS landscape, a simultaneous operation of both highly conserved and highly specific mechanisms of ROS production and degradation has been recognized in different plant cell organelles. It has become evident that light-driven reactions in thylakoid membranes can be considered the major ROS source in photosynthetically active tissues, whereas mitochondrial ROS production is of major importance in the dark or in non-green tissues. Furthermore, ROS production in the apoplastic space was uncovered to play a crucial role in plant interactions with external environment, including microbial organisms or root growth and development. However, some well-accepted concepts are continuously questioned by advances in the plant ROS field; e.g., singlet oxygen, generally thought to be produced in photosynthetic center, was reported to be involved in osmotic stress-induced cell death in *A. thaliana* roots, suggesting a novel light-independent mechanism of its generation [125].

Recent advances in the development and application of redox-sensitive probes enable specific measurements of individual ROS in different cellular compartments of intact plant cells. The application of these probes permits to analyze simultaneously spatio-temporal ROS modulations and functional associations between ROS metabolism and signaling and organelle functions. It is noteworthy that the current knowledge on the quantitative aspects of ROS metabolism in plant cells *in vivo* is still quite limited. This is mainly associated with current protocols used to detect ROS which are in many case not suitable for accurate and reliable ROS quantification [9,126]. Recent development of quantitative ROS transcription-based bioreporters seems a promising new strategy to achieve quantitative cellular mapping of ROS changes in plant responses to stress stimuli [127].

Moreover, actual advances of omics technologies provide new strategies to study in detail the site-specific functions of different components of ROS metabolism in plant cell signaling. With advances from single-cell to single-organelle transcriptomics, proteomics and metabolomics, specific roles of individual metabolites and enzymes in the regulation of ROS metabolism can be achieved. The recently introduced “ramanomics” approach might provide an efficient tool of non-invasive quantitative profiling of cellular compartments and monitoring of molecular interactions and transformations in live cells and their subcellular structures. Advancing field of the high-resolution microscopy combined with mentioned genetically-encoded redox sensors proteins is expected to provide new insights into the compartment-specific landscapes of short-lived reactive oxygen species in plant cell compartments, and to further elucidate their role in plant physiology and stress responses.

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Review

Molecular Mechanism of Oxidation of P700 and Suppression of ROS Production in Photosystem I in Response to Electron-Sink Limitations in C3 Plants

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Abstract: Photosynthesis fixes CO₂ and converts it to sugar, using chemical-energy compounds of both NADPH and ATP, which are produced in the photosynthetic electron transport system. The photosynthetic electron transport system absorbs photon energy to drive electron flow from Photosystem II (PSII) to Photosystem I (PSI). That is, both PSII and PSI are full of electrons. O₂ is easily reduced to a superoxide radical (O₂^{•−}) at the reducing side, i.e., the acceptor side, of PSI, which is the main production site of reactive oxygen species (ROS) in photosynthetic organisms. ROS-dependent inactivation of PSI *in vivo* has been reported, where the electrons are accumulated at the acceptor side of PSI by artificial treatments: exposure to low temperature and repetitive short-pulse (rSP) illumination treatment, and the accumulated electrons flow to O₂, producing ROS. Recently, my group found that the redox state of the reaction center of chlorophyll P700 in PSI regulates the production of ROS: P700 oxidation suppresses the production of O₂^{•−} and prevents PSI inactivation. This is why P700 in PSI is oxidized upon the exposure of photosynthesis organisms to higher light intensity and/or low CO₂ conditions, where photosynthesis efficiency decreases. In this study, I introduce a new molecular mechanism for the oxidation of P700 in PSI and suppression of ROS production from the robust relationship between the light and dark reactions of photosynthesis. The accumulated protons in the luminal space of the thylakoid membrane and the accumulated electrons in the plastoquinone (PQ) pool drive the rate-determining step of the P700 photo-oxidation reduction cycle in PSI from the photo-excited P700 oxidation to the reduction of the oxidized P700, thereby enhancing P700 oxidation.

Keywords: P700; P700 oxidation system; photorespiration; photosynthesis; Photosystem I (PSI); reactive oxygen species (ROS); reduction-induced suppression of electron flow (RISE); repetitive short-pulse (SP) illumination (rSP illumination treatment)

1. Introduction

Numerous researchers have shown that, in photosynthetic organisms, oxidative damage due to enhanced production of reactive oxygen species (ROS) occurs when environmental stress (e.g., extreme low/high temperatures, high salinity, and oligotrophic inorganic components) decreases the photosynthetic efficiency [1–21]. Much of this research has focused on the Mehler reaction, whereby a superoxide radical (O₂^{•−}), one of the most important ROS species, is formed via a one-electron reduction of O₂ in Photosystem I (PSI) of the chloroplast thylakoid membranes. In addition to PSI, ROS production in Photosystem II (PSII) has long been studied, and numerous studies have been published regarding its relationships to oxidative damage in PSII across oxygenic photosynthetic organisms [22–29]. This review focuses on our recent findings regarding the molecular mechanisms of the production and suppression of ROS in PSI.

We try to answer the following questions in this review: “Is the excess accumulation of electrons in PSI truly harmful to photosynthetic organisms?” and “Does ROS production occur *in vivo*?” Driever and Baker [30,31] and Ruuska et al. [30,31] reported that the *in vivo* activity of the Mehler reaction is too small compared to the electron flux in photosynthesis. Additionally, considering the photosynthetic capacity of many photosynthetic organisms, the photosynthetic rate is saturated at a light intensity level corresponding to approximately 25% of natural sunlight or lower [2,3,31–37]. This indicates that photosynthesis regularly proceeds under conditions of an excess photon supply, which can cause the photosynthetic electron transport system to be full of electrons. To address this issue, we asked “How do photosynthetic organisms manage to escape from the accumulation of electrons in PSI?” We then developed a method for suppressing photosynthesis and imitating the accumulation of electrons in PSI. Using this approach, we succeeded in showing how ROS are produced and suppressed in PSI [38]. In what follows, the robust and universal characteristics of the suppression of ROS production in photosynthetic organisms are discussed.

2. The Accumulation of Electrons in the PSI Acceptor Side Promotes ROS Production and Oxidative Damage

In the photosynthetic electron transport system, the reaction center chlorophyll P700 in PSI functions in the photo-oxidation reduction cycle (Figure 1). Ground state P700 is photoexcited to produce the excited P700 ($P700^*$), which then donates an electron to the primary electron carrier (A_0) to produce the oxidized P700 ($P700^+$). Subsequently, $P700^+$ accepts an electron from the reduced plastocyanin (PC) to regenerate P700. The oxidized PC in turn is reduced by the electrons coming from PSII. The electron in the reduced A_0 flows to ferredoxin (Fd) through the electron carriers: A_1 , F_x , and F_A/F_B [39]. If a light pulse (light intensity: 20,000 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$; pulse duration: 400 ms) is applied to C3-sunflower intact leaves in the dark under atmospheric conditions, all P700 molecules are oxidized to $P700^+$ (Figure 2A). $P700^+$ then rapidly decreases during the light pulse, which is due to the occupation of $P700^*$ in the photo-oxidation reduction cycle of P700, indicating the accumulation of electrons at the acceptor side of PSI. If the light pulse is applied every 10 s on C3-sunflower intact leaves (i.e., repetitive short-pulse (rSP) illumination treatment), PSI is deactivated (Figure 3) [38]. Conversely, PSII is not deactivated at all. PSI deactivation is all suppressed when atmospheric O_2 partial pressure drops to 2 kPa. The atmospheric rSP illumination treatment then produces a superoxide radical (O_2^-) in PSI [32,40–43]. These results suggest that the electrons accumulated on the PSI acceptor side during one light pulse are used for O_2 reduction, which triggers the O_2^- production reflected as $P700^*$ accumulation. O_2^- accumulation occurring over time with rSP illumination treatment indicates progressive PSI oxidative damage to the photosynthetic organism. PSI inactivation due to rSP illumination treatment represents a case of photosynthetically induced CO_2 fixation inactivation [9,35,38,44]. Unlike PSII, PSI requires approximately one week for functional recovery following inactivation [7,9,35,36,45,46]. Therefore, PSI inactivation can be fatal to plant growth [7,35,36,42]. Exposure of plants to low temperatures also inactivates PSI and the recovery of PSI function can take several weeks.

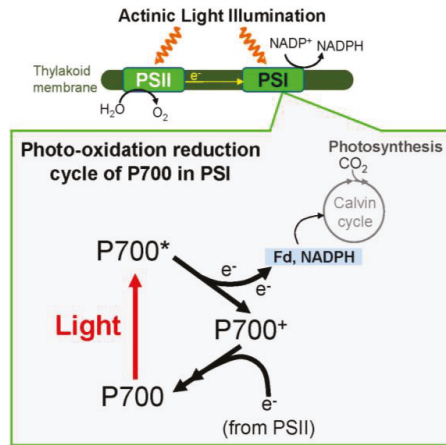


Figure 1. The photo-oxidation reduction cycle of P700 in Photosystem I (PSI). The reaction center chlorophyll P700 in PSI absorbs the light energy, and the P700 is excited to P700*. The P700* donates electrons to the electron carrier, A_0 , and concomitantly produces the oxidized P700, P700+. The P700+ is reduced by electrons from Photosystem II (PSII) through plastoquinone (PQ), the cytochrome (Cyt) *b6/f*-complex, and plastocyanin (PC). The reduced PC directly donates electrons to P700. There are then P700 turnovers in the photo-oxidation reduction cycle of P700 in PSI. The electron on A_0 flows to NADP+ to produce NADPH through the electron carriers A_1 , F_x , F_A/F_B , and Fd. In the photosynthetic linear electron flow, electrons extracted from H₂O oxidation in PSII flow to NADP+ for the production of NADPH.

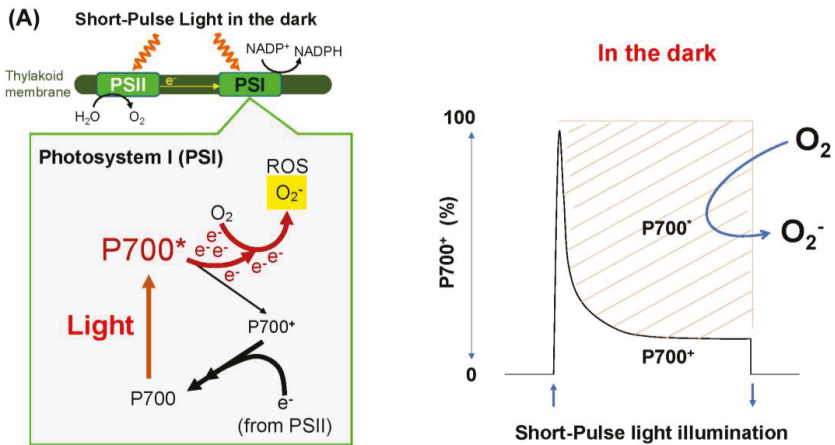


Figure 2. Cont.

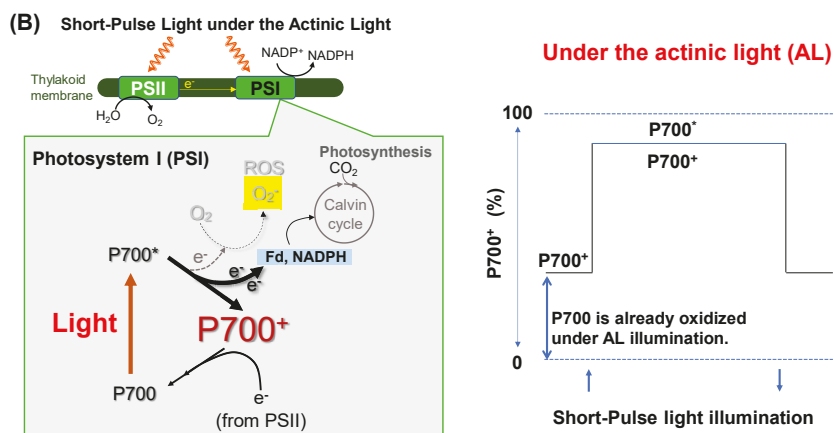


Figure 2. The kinetics of oxidized P700 during short-pulse (SP) illumination in the dark and under actinic light. (A) An SP illumination ($20,000 \mu\text{mol photons m}^{-2} \text{s}^{-1}$, 400 ms) to intact leaves in the dark rapidly oxidizes the reaction center chlorophyll P700 in PSI from the fully reduced state to the maximally oxidized state. Afterwards, oxidized P700 (P700^+) is reduced by electrons from PSII, with the accumulation of the excited state of P700 (P700^*), as shown by the increase in shaded area. P700^* can donate an electron to O_2 , thereby producing a superoxide radical (O_2^-) through the one-electron reduction by the electron carriers, A_0/A_1 , F_x , and $\text{F}_\text{A}/\text{F}_\text{B}$, localized at the acceptor side of PSI. (B) In actinic light (AL) illumination, before SP illumination, P700 is already oxidized to P700^+ . SP illumination further oxidizes P700 to its maximum, which is lower than the oxidation obtained in the dark, because P700^* accumulates under the steady-state AL illumination. P700^+ produced by SP illumination is not reduced, because P700 in PSI experiences turnover with the rate-determining step of P700^+ reduction in the P700 photo-oxidation reduction cycle. The Y-axis shows the oxidation ratio of P700 from 0% to 100%, and the X-axis shows the time for SP illumination: upward arrow: SP illumination starts; downward arrow: SP illumination stops.

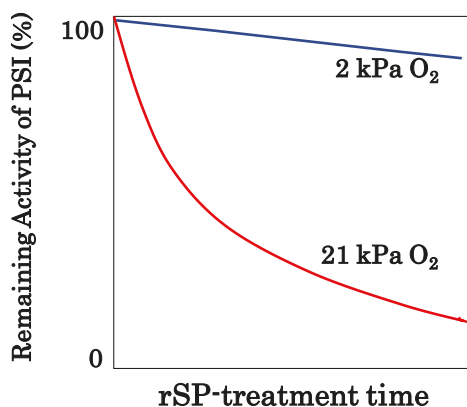


Figure 3. Repetitive short-pulse (rSP) illumination treatment inactivates PSI. In angiosperms, rSP illumination treatment of the intact leaves inactivates PSI activity [38]. The inactivation is suppressed under low O_2 conditions (2 kPa O_2). Short-pulse illumination reduces O_2 to a superoxide radical, O_2^- , and the accumulated O_2^- from rSP illumination treatment oxidatively attacks PSI components. Red line: the PSI activity gradually decreases as rSP illumination treatment proceeds under atmospheric O_2 (21 kPa) condition; blue line: PSI inactivation by rSP illumination treatment is suppressed under low O_2 (2 kPa) condition.

3. ROS Production Is Suppressed by Oxidation of P700

As described above, ROS-dependent oxidative damage in PSI was shown *in vivo*. Next, a new question arose: essentially, how do plants survive under natural sunlight? The answer was quickly obtained. PSI inactivation by rSP illumination treatment was suppressed under illumination by actinic light (AL) [38]. Furthermore, the extent of PSI inactivation decreased with increasing AL intensity. In other words, the light-induced inactivation of PSI was prevented by light itself. The effect of AL illumination was that P700 was oxidized to P700⁺ (Figure 2B). Increasing AL intensity increased P700⁺ in the P700 photo-oxidation reduction cycle. Afterwards, P700⁺ accumulation prevented P700* accumulation during SP illumination (Figure 2B). In other words, high levels of P700⁺ were maintained, even during light pulses. Thus, the role of AL illumination from the perspective of the P700 photo-oxidation reduction cycle was made clear: both P700* accumulation in the dark and P700⁺ accumulation under AL illumination during an SP illumination, indicate the rate-determining step (RdS) transition in the P700 photo-oxidation reduction cycle. In the dark, P700* oxidation is rate-limited, whereas under AL illumination, P700⁺ reduction is rate-limited. AL illumination suppresses rSP-illumination-treatment-induced PSI inactivation through the rate-determining step transition mechanism. That is, AL illumination suppresses the accumulation of P700*, which can donate electrons to O₂ to produce O₂⁻ in PSI. Subsequently, we found the robust effect of AL illumination, which was that P700 was oxidized and suppressed PSI inactivation by rSP illumination treatment [42].

Numerous studies have reported the oxidation of P700 in PSI under high intensity AL or low CO₂ conditions [33,34,47–54]. Accumulated P700⁺ during the P700 photo-oxidation reduction cycle absorbs excess photon energy and dissipates it as heat [55–57]. Concomitantly, P700⁺ accumulation inhibits O₂ reduction, producing O₂⁻ by suppressing P700* accumulation [38]. This is the molecular mechanism that reduces the rate of the Mehler reaction in intact leaves under AL illumination. If P700 is not subsequently oxidized, ROS production will increase, due to the promotion of the Mehler reaction, and PSI will stop functioning [38,42,58].

4. Molecular Mechanism of Oxidation of P700 in Photosystem I under AL Illumination

In photosynthesis, a light reaction functions with a dark reaction in a tightly coupled state. For example, the light reaction proceeds in the thylakoid membranes of chloroplasts in C₃-plants, where photosynthetic linear electron flow produces NADPH and ATP. The dark reaction proceeds in both photosynthesis and photorespiration, where both NADPH and ATP are used for the regeneration of ribulose 1,5-bisphosphate (RuBP) (one of the substrates for RuBP carboxylase/oxygenase which is consumed in the dark reaction). In photorespiration, reduced Fd is used in chloroplasts. Electron flow, which produces both NADPH and the reduced Fd, and proton flow, which produces ATP in the light reaction, are coupled, with both e⁻ flow and H⁺ flow required for the consumption of NADPH, the reduced Fd, and ATP in the dark reaction. This tight coupling of the light reaction with the dark reaction can cause a dangerous situation in which ROS are produced in the photosynthetic electron transport system. The limitation of the dark reaction is the limitation of the light reaction, which can lead to the accumulation of electrons in the photosynthetic electron transport system. For example, in the photo-oxidation reduction cycle of P700 in PSI, electrons would accumulate at the acceptor side of PSI, where O₂ can be reduced to O₂⁻, a ROS. As described above, unless P700 in PSI is oxidized, PSI suffers from oxidative damage.

We show below how the light reaction tightly couples with the dark reaction in photosynthesis. We later show how P700 is oxidized in response to the limitation of the dark reaction, for example, in drought, low temperature, and high light situations.

4.1. Robust Relationships Support Tight Coupling between the Light and Dark Reactions in Photosynthesis

Much evidence has accumulated that clarifies a robust tight coupling between the light and dark reactions in C₃ photosynthesis (Figure 4). For instance, Genty et al. [59] reported a positive linear

relationship with an origin of zero between the photosynthetic linear electron flow rate (J_f) in the light reaction and the photosynthetic electron consumption rate (J_g) based on NADPH consumption—in the dark reaction [59]. Consistently, a positive linear relationship with an origin of zero was also found between the photosynthetic linear electron flow rate (J_f) in the light reaction and the electron consumption rate (J_g)—based on NADPH consumption and reduced Fd—in the dark reaction, where photosynthesis and photorespiration both occur (Equation (1)) [30,31,60]. These results indicate that photosynthetic linear electron flow is the sole driver of both photosynthesis and photorespiration. In other words,

$$J_f = J_g \quad (1)$$

Recently, a positive linear relationship with an origin of zero between the reduced Fd oxidation rate (v_{Fd}) and the photosynthetic linear electron flow rate (J_f) was found (Equation (2) [61]). These results indicate that electron flow via Fd is driven by both photosynthesis and photorespiration. In other words,

$$v_{Fd} = k_{fd} \times Fd^- = J_g \quad (2)$$

The constant k_{fd} is the apparent rate constant of the reduced Fd oxidation rate, while Fd^- is the amount of reduced ferredoxin.

A positive linear relationship with an origin of zero was reported between J_f and the pmf production rate ($vH^+ = gH^+ \times pmf$ (gH^+ , proton conductance) (Equation (3)) [62–65]). These results indicate that photosynthetic linear electron flow provides the formation of pmf. In other words,

$$k_{H^+} \times J_f = vH^+ \quad (3)$$

The rate constant k_{H^+} reflects lumen H^+ accumulation efficiency, which depends on H_2O oxidation in PSII and on the Q-cycle in the Cyt b_6/f complex. Furthermore, a positive linear relationship with an origin of zero was found between vH^+ and the proton usage rate (JgH^+), due to the regeneration of ATP required for both photosynthesis and photorespiration (Equation (4)) [66]. These results indicate that vH^+ also reflects the pmf usage rate during photosynthesis and photorespiration. In other words,

$$vH^+ = JgH^+ \quad (4)$$

Furthermore, a positive linear relationship with an origin of zero was found between JgH^+ and J_g ($=J_f$) (Equation (5)) [66]. These results indicate that JgH^+ was determined by the electron consumption rate (J_g), based on the consumption of both reduced Fd and NADPH. In other words,

$$JgH^+ = k_{jg} \times J_g = k_{H^+} \times J_f \quad (5)$$

Equation (5) indicates that there is a tight coupling between the light and dark reactions during photosynthesis and photorespiration. Constant k_{jg} is the coefficient that links e^- flow with H^+ flow during photosynthesis and photorespiration. These findings indicate that the operation of the light reaction is driven by the dark reaction and vice versa, and provide a concrete explanation for the observed decrease in J_f under environmental stress, when the photosynthetic rate decreases due to a dark reaction rate limitation (i.e., when JgH^+ and J_g decrease).

Tight coupling between the light and dark reactions, as shown in Equation (5), can cause the accumulation of electrons at the acceptor side of PSI and the accumulation of protons in the luminal space of thylakoid membranes under stress conditions, e.g., drought, low temperatures, and high intensity of light, which decrease the photosynthesis efficiency [2,3,8,67–69]. For example, under constant light intensity, decreases in both J_g and JgH^+ decrease J_f . These situations cause a reduction in PQ and the non-photochemical quenching of chlorophyll fluorescence, both of which reflect the accumulations of electrons and protons in the photosynthetic electron transport system [32,34,42,58,70–72].

However, the electrons do not accumulate at the acceptor side of PSI; P700 is oxidized under stress conditions [32,34,49,50,53,72]. From the model of the photo-oxidation reduction cycle of P700, the reduction of P700⁺ must be the rate-determining step to accumulate the oxidized P700 in PSI. The suppression of electron fluxes in both light and dark reactions does not limit the reduction of P700⁺ in the cycle. That is, the electron flux to the oxidized P700 should be suppressed such that P700 is oxidized.

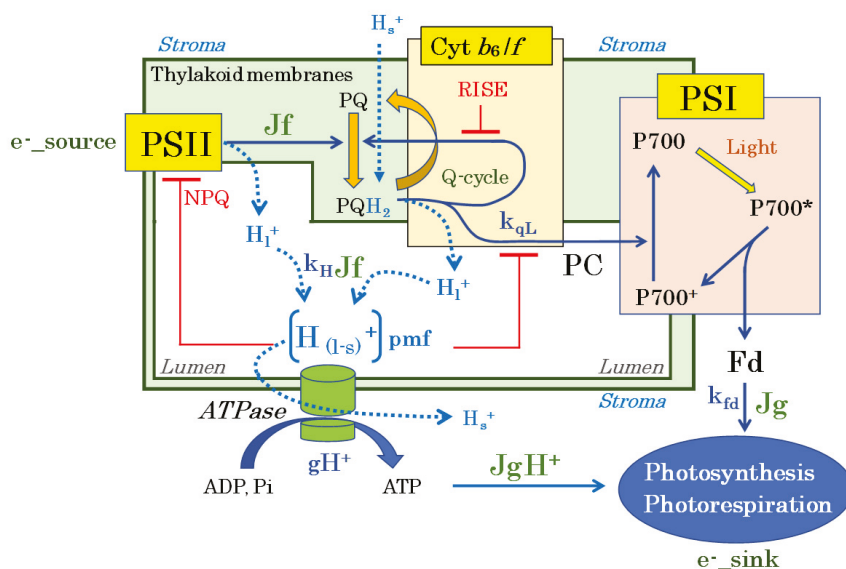


Figure 4. Tight coupling between the light and the dark reactions in photosynthesis of C3 angiosperms. Blue arrows: electron flow; blue dotted arrows: H⁺ flow; H_s⁺: stroma H⁺; H_l⁺: lumen H⁺; [H(1-s)⁺]: difference in H⁺ concentration (ΔpH) between the stroma and the lumen, i.e., the proton motive force (pmf). Photosystem II (PSII): electron source. P700 functions in the photo-oxidation reduction cycle to catalyze the electron flow from plastocyanin (PC) to Fd in Photosystem I (PSI). P700 is photoexcited to P700*, which donates electrons to Fd through electron carriers (A₀, A₁, F_A/F_B, and F_X) to produce P700⁺. In turn, P700⁺ is reduced by electrons from PSII through plastoquinone (PQ), the Cyt b₆/f complex, and PC. The Q-cycle in the Cyt b₆/f complex transfers electrons from reduced PQ along two routes: first, PC; second, oxidized PQ. Enhanced reduction of PQ suppresses the reduction of oxidized PQ in the Cyt b₆/f complex, which slows down the activity of the Q-cycle, and the electron flux from the Cyt b₆/f complex to P700⁺ in PSI. H⁺ conductance (gH⁺) is an apparent rate constant that depends on the concentrations of ADP, Pi, ATPase, and the catalytic constant (k_(ATPase)). NPQ, non-photochemical quenching of chlorophyll fluorescence, is activated by the acidification of the luminal space of the thylakoid membrane, which is reflected by the increase in pmf. NPQ suppresses the quantum yield of PSII, which is reflected by the electron flux in photosynthetic linear electron flow, J_f; the pmf also suppresses the oxidation activity of reduced PQ by the Cyt b₆/f complex. Tight coupling of the light reactions (the production of electron and proton flows) to the dark reaction (the consumption of electrons and protons in both photosynthesis and photorespiration) yields a robust relationship whereby electron flux (J_f) is equal to the electron flux in electron sinks (photosynthesis and photorespiration) expressed as J_g. The pmf is the driving force to synthesize ATP by ATP synthase using ADP and Pi, and the pmf-dissipation rate (vH⁺) is expressed as gH⁺ × pmf. As a robust model, J_f = vF_d = J_g [61]; vF_d is the oxidation rate of reduced Fd; k_{Jf} × J_f = gH⁺ × pmf [64,65]. vH⁺ = gH⁺ × pmf = J_gH⁺ [66]. J_g and J_gH⁺ mutually determine each other. Meanwhile, pH decreases in the luminal space of the thylakoid membrane during the photo-oxidation of water in PSII and the Q-cycle in the Cyt b₆/f complex. The decrease in pH (i.e., increased pmf) acts as the driving force for ATP production by ATPase.

4.2. How Do Protons and Electrons Accumulate in Both the Lumenal Space of Thylakoid Membranes and the PQ Pool in Response to the Suppression of the Dark Reaction?

Generally, it is proposed that the electron flux from PQ to P700 in PSI is regulated by the oxidation activity of the reduced PQ (PQH₂) by the cytochrome (Cyt) *b₆/f*-complex [48,67,70,71,73–81]. The PQH₂ oxidation activity by the Cyt *b₆/f*-complex decreases with a lowering pH. Therefore, the accumulation of protons in the lumenal space of thylakoid membranes suppresses the electron flow to the oxidized P700 in PSI, which is known as photosynthesis control [67,73,74]. Furthermore, the overreduction of PQ inhibits the Q-cycle in the Cyt *b₆/f*-complex, and contributes to the oxidation of P700 in PSI, which is known as the reduction-induced suppression of electron flow (RISE) [70,71]. We propose that the RISE is also involved in photosynthesis control. To clarify the molecular mechanism of oxidation of P700 in PSI upon exposure to the limitation of photosynthesis, we must understand how protons and electrons accumulate in both the lumenal space and the PQ pool in the photosynthetic electron transport system in response to the activities of both photosynthesis and photorespiration.

The induction mechanism of proton gradient (Δ pH) formation across the thylakoid membrane, the accumulation of protons in the lumenal space of thylakoid membranes, is shown as below: Δ pH formation is observed as a pmf increase [65,82,83]. In general, Δ pH mainly occupies the pmf at low photosynthetic efficiency, e.g., high light and low CO₂ conditions. The velocity of change in pmf [$d(\text{pmf})/dt$] is determined by the difference between the pmf generation rate and the pmf decay rate. The pmf generation rate depends on the photosynthetic linear electron flow rate (Equation (3)), and the pmf decay rate depends on the ATP production rate, which is driven by both photosynthesis and photorespiration (Equation (4)): thus, the velocity of change in pmf is as shown (Equations (6) and (7)).

$$d(\text{pmf})/dt = k_{H^+} \times Jf - vH^+ \quad (6)$$

$$= k_{H^+} \times Jf - gH^+ \times (\text{pmf}) \quad (7)$$

The rate constant k_{H^+} reflects the H⁺ accumulation in the lumenal space of thylakoid membranes, which is driven by photosynthetic linear electron flow. At the steady state, $d(\text{pmf})/dt = 0$, and the pmf production rate is $k_{H^+} \times Jf$. This depends on H₂O oxidation in PSII and on Q-cycle rotation in the Cyt *b₆/f* complex. Furthermore, the pmf decay rate, vH^+ , is expressed as $gH^+ \times (\text{pmf})$. The gH^+ , H⁺ conductance, is a rate constant that reflects the apparent rate constant of pmf decay. The vH^+ also reflects the pmf dissipation rate, and vH^+ can be replaced with JgH^+ as follows (Equation (8)):

$$d(\text{pmf})/dt = k_{H^+} \times Jf - JgH^+ \quad (8)$$

We reveal that, in a steady state where $d(\text{pmf})/dt = 0$, based on Equations (7) and (8), Equation (9) was obtained.

$$k_{H^+} \times Jf = vH^+ (= gH^+ \times \text{pmf}) = JgH^+ \quad (9)$$

Equation (9) shows that the photosynthetic linear electron flow activity in the light reaction links photosynthesis and photorespiration activity in the dark reaction through pmf formation and dissipation. The dependence of pmf on Jf , gH^+ , and JgH^+ is shown as follows (Equations (10) and (11)).

$$\text{pmf} = (k_{H^+} \times Jf)/gH^+ \quad (10)$$

$$= JgH^+/gH^+ \quad (11)$$

Based on this model, we will discuss the molecular mechanism of proton accumulation in the lumenal space of thylakoid membranes in response to the activities of photosynthesis and photorespiration. For example, if the extent of the decrease in the dark reaction rate, shown as the decrease in both Jg and JgH^+ , was smaller than that in gH^+ , then pmf increased. This is how protons accumulate in the lumenal space of thylakoid membranes.

We consider the accumulation of electrons in the photosynthetic electron transport system, reflected as the redox states of both PQ and Fd. The reduced state of PQ is evaluated by a parameter of chlorophyll fluorescence, $1 - qL$ [72,84]. The qL shows the oxidation state of PQ in thylakoid membranes. A higher value of $1 - qL$ shows the high accumulation of electrons in the PQ pool. The oxidation rate of the reduced PQ can be expressed as $kqL \times (1 - qL)$. Additionally, the oxidation rate of the reduced Fd (Fd^-) is expressed as $k_{Fd} \times Fd^-$. Both kqL and k_{Fd} are the apparent rate constants. In the light reaction, the electron fluxes for the oxidations of both the reduced PQ and the reduced Fd are equal to J_f , as shown in the equation (12).

$$J_f = kqL \times (1 - qL) = k_{Fd} \times Fd^- = J_g \quad (12)$$

The kqL is given a positive value and is expected to decrease by two factors: first, kqL decreases with a decrease in pH in the luminal space of thylakoid membranes (photosynthesis control) [67,85,86]; second, kqL decreases by the RISE [70,71]. Generally, a decrease in J_g increases the value of $1 - qL$ [34,72]. These facts indicate, in response to the decrease in electron sink activity (both photosynthesis and photorespiration), that the extent of the decrease in kqL is larger than that in J_g . That is, from Equation (12), we can clearly understand why $1 - qL$ increases. This is how electrons accumulate in the PQ pool of thylakoid membranes.

On the other hand, a decrease in J_g does not induce the reduction of Fd [61]. For example, lowering the intercellular partial pressure of CO_2 decreases the photosynthesis rate, and J_g decreases; however, the reduced level of Fd either does not change or decrease. These facts show that the extent of the decrease in k_{Fd} was almost the same as that in J_g . The k_{Fd} depends on both the activity of Fd^-NADP^+ oxidoreductase (FNR) and $NADP^+$ regeneration efficiency. The decrease in the photosynthesis rate lowers the $NADP^+$ regeneration rate, which decreases k_{Fd} .

4.3. How Do the Accumulated Protons in the Luminal Space and Electrons in the PQ Pool Oxidize P700 in PSI?

The decrease in $J_f (=J_g)$ shows the negative linear relationship with $P700^+$ (Figure 5) [42,87]. The decrease in photosynthesis efficiency induces the oxidation of P700 in PSI. As described above, the suppression of photosynthesis enhances both the pmf and the RISE, which would cause the reduction of $P700^+$ in the photo-oxidation reduction cycle of P700 in PSI to be the rate-determining step. The decrease in J_g lowers J_f in the tight coupling of the light reaction with the dark reaction. Here, we have to pay an attention to the magnitude of the decrease in J_f . The extent of the decreases in gH^+ and kqL are larger than that in J_g , so the pmf and $(1 - qL)$ increase. We also must consider that the magnitude of J_f would be downregulated by both the pmf and the RISE, such that the reduction of $P700^+$ should be the rate-determining step in the photo-oxidation reduction cycle of P700 in PSI. J_f must be lower than the theoretical value of J_g obtained by the suppression of photosynthesis. For example, if J_f decreases to the same extent as the decrease in J_g in the response to the decrease in C_i , the photo-excited P700 ($P700^*$) would accumulate, because the efficiency of photo-excitation of P700 to $P700^*$ does not change at a constant intensity of AL. This deduction contradicts the experimental facts; that is, that suppressed photosynthesis induces the oxidation of P700.

The RISE was evidenced to have the potential to stop the electron flow in the light reaction and to oxidize P700 in PSI of cyanobacteria [70,71]. Flavodiiron protein (FLV) catalyzes the reduction of O_2 to H_2O using NADPH in the cyanobacteria [88,89]. The electron flux in the FLV reaction is so high that it alternates the electron flux in photosynthesis under CO_2 -deficient conditions [88,90]. The cyanobacteria deficient in FLV show suppressed photosynthesis, with the accumulation of electrons in the PQ pool [70]. Furthermore, illumination by SP light reduces PQ almost transiently with the suppression of O_2 evolution under steady-state conditions [70]. FLV can oxidize PQ even though photosynthesis can function at a higher activity. FLV keeps PQ in an oxidized state so as not to induce the RISE for the full activity of photosynthesis, if enough CO_2 is supplied to cyanobacteria.

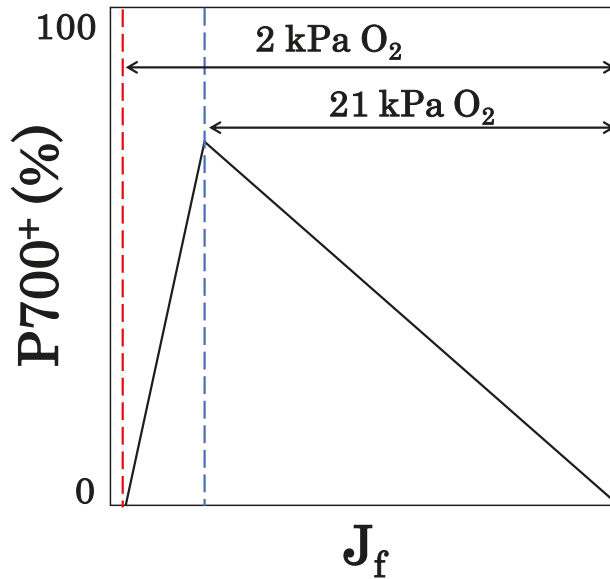


Figure 5. Dependence of oxidized P700 ($P700^+$) induction on electron flux in photosynthetic linear electron flow in intact leaves. Generally, P700 is oxidized with a decreasing electron flux of photosynthetic linear electron flow (J_f) by lowering CO_2 partial pressure. At 21 kPa O_2 , when both photorespiration and photosynthetic CO_2 fixation are functioning, $P700^+$ shows the maximum value at the CO_2 compensation point, while J_f shows the minimum value (blue dotted line). The horizontal arrow at 21 kPa O_2 shows the range over which J_f is driven by both photosynthetic CO_2 fixation and photorespiration. At 2 kPa O_2 , when no detectable photorespiration occurs and photosynthetic CO_2 fixation functions normally, $P700^+$ starts to decrease to the minimum at the CO_2 compensation point after reaching its maximum value, while J_f reaches its minimum value (red dotted line). The horizontal arrow at 2 kPa O_2 shows the range over which J_f is driven by photosynthetic CO_2 fixation.

Furthermore, we found that inductions of the pmf and RISE oxidize P700 in PSI of the C3-plant wheat leaves [42,87]. A decrease in the intercellular partial pressure of CO_2 lowers the photosynthetic rate and increases the photorespiration rate. However, the photorespiration rate cannot reach the value theoretically derived from Rubisco kinetics at lower CO_2 [37,91]. The induction of the pmf and RISE lowers the activity of the light reaction, which causes the activity of the dark reaction, the photorespiration rate, to be suppressed, even though the potential activity of the dark reaction exceeds the suppressed activity of the light reaction. These results show that the oxidation of P700 requires the suppression of electron flux in the light reaction much more so than in the dark reaction; thus, the suppressed activity in the dark reaction is lower than the theoretical potential activity. As a result, the reduction of $P700^+$ becomes the rate-determining step in the P700 photo-oxidation reduction cycle. This is the essence of the molecular mechanism of P700 oxidation.

On the other hand, the induction of the pmf and RISE does not always oxidize P700 in response to the limitation of photosynthesis. Unless the electron flux of the light reaction dips below the potential electron flux of the dark reaction, which is, theoretically, expected from the limited activity of photosynthesis, the limitation of the PSI acceptor side is so strong that P700 is not oxidized (Figure 5). When photosynthesis is severely limited, where both photosynthesis and photorespiration functions are suppressed, P700 cannot be oxidized because the oxidation of $P700^*$ becomes the rate-determining step in the P700 photo-oxidation reduction cycle.

4.4. Suppression of the Production of ROS in PSI by the P700 Oxidation System

We summarize the molecular mechanism of P700 oxidation in PSI of chloroplasts (Figure 6). Both PSI and PSII absorb photon energy to excite the reaction center chlorophylls P680 (PSII) and P700 (PSI). Similarly to P700 in PSI, P680 undergoes a photo-oxidation reduction cycle. The ground state of P680 is photo-excited to P680*. The P680* donates electrons to PQ through electron carriers in PSII. Additionally, then, the electrons start to flow from the reduced PQ to PSI, through the Cyt *b₆/f*-complex, and plastocyanin (PC). During the electron flow, H⁺ from both H₂O oxidation in PSII and the reduced PQ (PQH₂, plastoquinol) oxidation of the Cyt *b₆/f*-complex accumulate in the luminal space of thylakoid membranes. The accumulated protons are the pmf to drive ATP synthase to produce ATP in the stroma. On the other hand, NADPH is produced by ferredoxin-NADP oxidoreductase (FNR) at the acceptor side of PSI. ATP, NADPH, and the reduced Fd are used for the regeneration of RuBP in both photosynthesis and photorespiration. The regeneration rates of NADP⁺, the oxidized Fd, and ADP are determined by the activities of both photosynthesis and photorespiration.

As shown in Equations (5), (9), and (12), the light reaction tightly couples with the dark reaction. The suppression of photosynthesis decreases Jg or JgH⁺. Afterwards, the regeneration rates of ADP, Pi, and NADP⁺ decrease. That is, gH⁺ decreases to increase the pmf, and kqL decreases to increase (1 – qL), which induces the RISE. Both the pmf and the RISE lower kqL, as reflected in the suppressed activity of the reduced PQ oxidation by the Cyt *b₆/f*-complex. NPQ induced by the pmf also downregulates the quantum yield of PSII. The suppressed electron flow from PSII to the oxidized P700 also causes the reduction of P700⁺ in the photo-oxidation reduction cycle to be the rate-determining step, which leads to the oxidation of P700.

We compared the rate-determining step in the photo-oxidation reduction cycle of P700 in PSI in response to the limitation of electron sink activities and the photosynthesis and photorespiration activities at high light conditions (Figure 6). In Case (I), we set the electron sink to a large size. The black arrows show the electron flow in the photosynthetic electron transport system from the redox reaction of PQ to the electron sink, including both photosynthesis and photorespiration, through the Cyt *b₆/f*-complex, PC, and P700 in PSI and Fd. The width of the black arrows reflects the magnitude of the apparent rate constant in each elementary reaction. The rate-determining step of the P700 photo-oxidation reduction cycle in PSI is the reduction of P700⁺, leading toward the oxidation of P700⁺. Both the pmf and RISE contribute to the suppression of the Q-cycle activity of the Cyt *b₆/f*-complex, which causes the reduction of P700⁺ to be the rate-determining step.

In Case (II), the electron sink is set to a medium size. Environmental stress, e.g., drought and/or low temperatures, can suppress photosynthetic activities. That is, both the pmf and RISE increase in response to the decrease in both gH⁺ and kqL, as determined by Equations (11) and (12). The apparent rate constant for the reduction of P700⁺ further decreases, compared to Case (I). The magnitude of the apparent rate constant for P700⁺ reduction is lower than the apparent constant for P700* oxidation. The increase in the suppression of the Q-cycle in the Cyt *b₆/f*-complex enhances the rate-determining step, the reduction of P700⁺, in the P700 photo-oxidation reduction cycle of PSI.

In Case (III), the electron sink is smaller than the medium size in Case (II). The activities of both photosynthesis and photorespiration are further suppressed, compared to Case (II). However, the rate-determining step leads to the elementary reaction of the oxidation of the photo-excited P700, P700*, and P700* then accumulates, although both the pmf and RISE are kept to the same or a slightly greater extent or as in Case (II). That is, the suppression of the apparent rate constant of the oxidation of P700* is larger than that of the oxidation activity of the reduced PQ in the Cyt *b₆/f*-complex in response to the limitation of the electron sink. Electrons accumulate at the acceptor side of PSI, and the electrons flow to O₂ to produce O₂⁻, the production of ROS. The transition of the rate-determining step from the reduction of P700⁺ to the oxidation of P700* is a dangerous condition that induces oxidative damage to PSI.

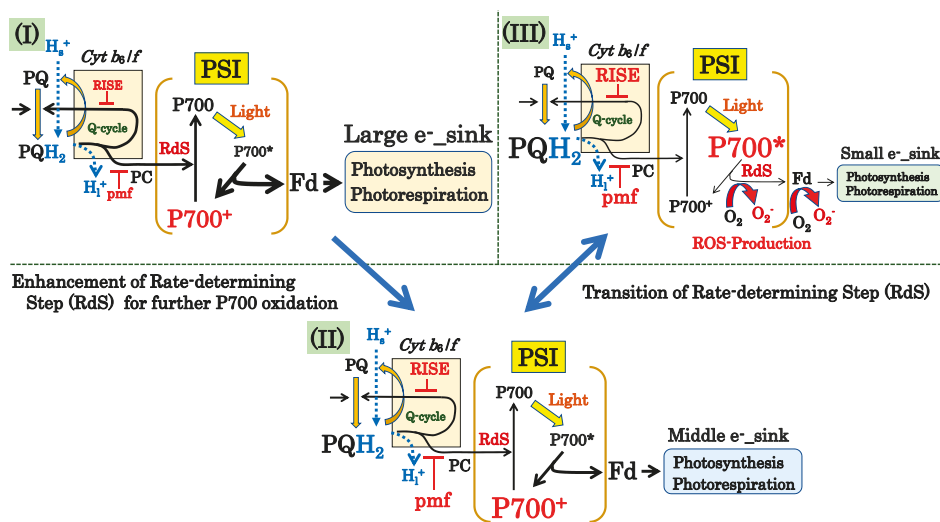


Figure 6. The P700 oxidation system and ROS production mechanism. The photosynthetic linear electron flow in the light reaction is tightly coupled to the dark reaction, photosynthesis and photorespiration (electron sinks). The P700 oxidation system enhances the rate-determining step of P700⁺ reduction to stimulate the oxidation of P700 in PSI, in response to the decrease in electron sinks. Once exposed to a further decrease in electron sinks, P700⁺ reduction cannot remain the rate-determining step, and the rate-determining step transitions from P700⁺ reduction to the oxidation of P700*. Three distinct scenarios of the rate-determining step can take place under saturated light intensity: In Case (I) with the large electron sink, both the pmf and RISE are induced by the two electron sinks: photosynthesis and photorespiration. Afterwards, the oxidation activity of the reduced PQ by the Cyt *b₆/f*-complex is suppressed to decrease the reduction rate of P700⁺. As a result, the photo-oxidation reduction cycle of P700 in PSI turns over with the rate-determining step of P700⁺ reduction. In Case (II) with the medium-sized electron sink, PQ is further reduced, and the pmf is further accumulated, compared to Case (I), by the decreased gH⁺ and k_{qL}. The electron flux from the reduced PQ to the oxidized P700 is then further suppressed. The oxidation of P700 becomes the rate-determining step in the photo-oxidation reduction cycle of P700 in PSI. In Case (III) with the small electron sink, the further induced pmf and RISE cannot decrease the reduction rate more than the suppression of the oxidation of P700* due to the smaller electron sink, compared to Case (II). Afterwards, P700* accumulates. As a result, the photo-oxidation reduction cycle of P700 in PSI turns over with the rate-determining step of P700* oxidation. From Case (II) to Case (III), the rate-determining step transitions from the reduction of P700⁺ to the oxidation of P700* in the photo-oxidation reduction cycle of P700 in PSI. The accumulated P700* increases the chance of the reduction of O₂ to O₂⁻, the production of ROS at the acceptor side of PSI. The reduced electron carriers in PSI—A₀, A₁, F_x, F_A/F_B, and the reduced Fd—can react with O₂ to produce O₂⁻ [2,39–41,92,93].

4.5. Relationship between Photorespiration- and Flavodiiron Protein (FLV)-Dependent Electron Flows and Their Contribution to P700 Oxidation in PSI

Following the model of the P700 oxidation system, photosynthetic linear electron flow contributes to the induction of either the pmf or RISE, which suppresses the electron flow from the Cyt *b₆/f*-complex to the oxidized P700 through PC. Afterwards, P700 is oxidized, as shown in Figures 5 and 6. From cyanobacteria to gymnosperms, in the evolution of green lineage oxygenic photosynthetic organisms, flavodiiron protein (FLV) drives O₂-dependent electron flow in the photosynthetic electron transport system, in addition to the photorespiration pathway [88–90,94–96]. FLV reduces O₂ to H₂O using NADPH as the electron donor [88]. Different from photorespiration, FLV-dependent electron flow does

not consume ATP; i.e., it only induces the pmf. Therefore, FLV-dependent electron flow has a higher capacity to oxidize P700. In fact, in the cyanobacteria *Synechocystis* sp. PCC6803 and *Synechococcus* sp. PCC7002, the deletion of *flv* genes caused the photoinhibition of PSI, which was enhanced by the reduction of P700 in PSI [58]. Furthermore, the liverwort *Marchantia polymorpha*, which was deficient in *flv* genes, suffered from the photoinhibition of PSI due to the reduction of P700 [58]. Angiosperms, however, do not have FLV genes in their genomes [87,97]. Among land plants, the electron flux of photorespiration-dependent photosynthetic linear electron flow was compared to FLV-dependent electron flux at the CO₂ compensation point, where photosynthetic activity was suppressed to zero. The maximum activities of both FLV-dependent and photorespiration-dependent photosynthetic linear electron flow were then evaluated among land plants [98].

Photorespiration activities at the CO₂ compensation point increased with the evolution of photosynthetic organisms from liverworts to C3 angiosperms to ferns and gymnosperms (Figure 7A). C3-C4 and C4 angiosperms show lower activities of photorespiration, compared to C3 angiosperms. C3 angiosperms show the maximum activities of photorespiration [98]. In C3 angiosperms, photosynthetic linear electron flow was mainly driven by photorespiration, which was the main electron sink in photosynthesis, at the CO₂ compensation point (Figure 7B). An electron sink other than photorespiration was found in liverworts, ferns, and angiosperms. FLV could be an electron sink at the CO₂ compensation point. C3-C4 and C4 angiosperms also had an electron sink other than photorespiration and FLV-dependent photosynthetic linear electron flows, but its mechanism is unknown.

What do C3-angiosperms achieve by discarding FLV genes in their genomes? Different from photorespiration, FLV-dependent electron flow does not consume ATP; it produces it, as described above. FLV-dependent electron flow induces the pmf, which can be a self-regulating electron flow and suppress photosynthesis, because an enhanced pmf suppresses the activity of the Cyt *b₆/f*-complex, as shown in Figures 5 and 6. In liverworts, ferns, and gymnosperms, FLV-dependent photosynthetic linear electron flow functions immediately after the actinic light illumination starts, and its flux is suppressed at the steady state of photosynthesis [98]. This indicates that FLV-dependent photosynthetic linear electron flow is generally not essential to photosynthesis organisms, and that there is a molecular mechanism to reduce FLV-dependent photosynthetic linear electron flow under a steady state of photosynthesis. On the other hand, photorespiration drives both the light reaction and the dark reaction, and inevitably functions with photosynthesis. This situation implies that a tight coupling of the light reaction with the dark reaction driven by both photosynthesis and photorespiration can regulate both the pmf and RISE in response to their activities, because the dark reaction induces and dissipates the pmf and RISE simultaneously. This could be a sensitive trigger for the regulation of P700 oxidation. If FLV-dependent photosynthetic linear electron flow participates in the photosynthetic linear electron flow driven by both photosynthesis and photorespiration, the automatic regulation of photosynthetic linear electron flow will collapse. Therefore, C3 angiosperms do not require FLV-dependent photosynthetic linear electron flow, and discard FLV genes in their genomes. These conclusions explain why angiosperms can have a high activity of photosynthesis, if photosynthetic organisms having FLV genes in their genomes have lower photosynthetic activity.

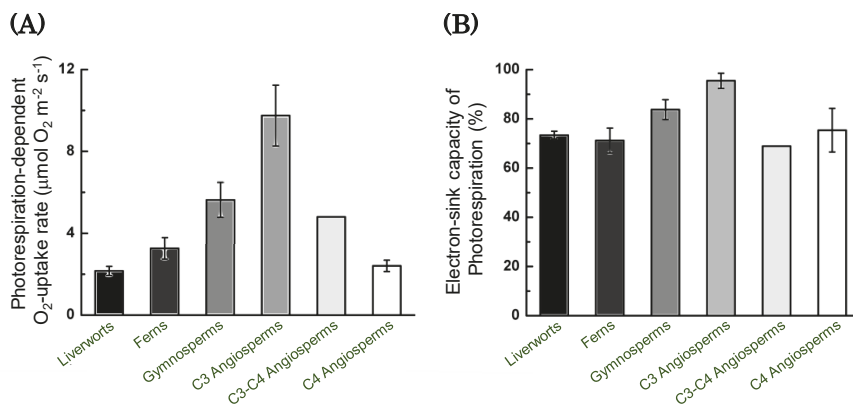


Figure 7. Evolution of tight coupling between the light and the dark reactions in photosynthesis. Upon spreading on the surface of the earth, oxygenic photosynthetic organisms inevitably metabolized 2-phosphoglycolate to regenerate 3-phosphoglycerate via the photorespiration pathway, because ribulose 1,5-bisphosphate (RuBP) carboxylase/oxygenase (Rubisco) catalyzes both RuBP carboxylation and oxygenation under atmospheric conditions. Consequently, photorespiration inevitably accompanies photosynthetic CO₂ fixation. In C3 angiosperms, both photosynthetic CO₂ fixation and photorespiration have become the main electron sinks in the photosynthetic process. Hanawa et al. [98] found that photorespiration activity increased with the evolution of oxygenic photosynthetic organisms. Photorespiration activity is evaluated as a photorespiration-dependent O₂-uptake at the CO₂ compensation point, at which point its activity peaks [66]. (A) Photorespiration activity increases with the evolution of oxygenic photosynthetic organisms from liverworts to C3 angiosperms through ferns and gymnosperms. (B) The electron sink capacity of photorespiration reaches its maximum in C3 angiosperms. The activities of flavodiiron (FLV)-dependent photosynthetic linear electron flow and other alternative photosynthetic linear electron flow (AEF) decrease upon the evolution of oxygenic photosynthetic organisms from liverworts to C3 angiosperms through ferns and gymnosperms. On the other hand, C3-C4 and C4 plants show lower photorespiration activity, and other electron sinks function at the CO₂ compensation point. Data are from Hanawa et al. [98].

5. Conclusions

Oxygenic photosynthetic organisms have a robust, common molecular mechanism to suppress the production of reactive oxygen species (ROS), the P700 oxidation system, because the ROS-dependent oxidative inactivation of PSI is lethal to these organisms. When cyanobacteria first appeared on the Earth about 2–3 billion years ago, they already had a P700 oxidation system [58]. Photosynthetic organisms regulate the activity of plastoquinol oxidation by the cytochrome (Cyt) *b₆/f*-complex using two strategies: pmf formation and the reduction-induced suppression of electron flow (RISE), both of which are used to suppress the electron flux from the Cyt *b₆/f*-complex to the reaction center chlorophyll P700 in PSI. The induction of both the pmf and RISE is sensitive to changes in the activities of both photosynthesis and photorespiration in land plants, which drive photosynthetic linear electron flow. The magnitudes of both the pmf and RISE are determined by the activity of photosynthetic linear electron flow. Upon exposure to environmental stress, e.g., low temperature, and/or drought, suppressed activities of both photosynthesis and photorespiration cause both the pmf and RISE to oxidize P700 in the manner described in the text for the suppression of ROS production in PSI.

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Article

Bifunctional Chloroplastic DJ-1B from *Arabidopsis thaliana* is an Oxidation-Robust Holdase and a Glyoxalase Sensitive to H₂O₂

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Abstract: Members of the DJ-1 protein family are multifunctional enzymes whose loss increases the susceptibility of the cell to oxidative stress. However, little is known about the function of the plant DJ-1 homologs. Therefore, we analyzed the effect of oxidation on the structure and function of chloroplastic AtDJ-1B and studied the phenotype of T-DNA lines lacking the protein. In vitro oxidation of AtDJ-1B with H₂O₂ lowers its glyoxalase activity, but has no effect on its holdase chaperone function. Remarkably, upon oxidation, the thermostability of AtDJ-1B increases with no significant alteration of the overall secondary structure. Moreover, we found that *AtDJ-1B* transcript levels are invariable, and loss of AtDJ-1B does not affect plant viability, growth and stress response. All in all, two discrete functions of AtDJ-1B respond differently to H₂O₂, and AtDJ-1B is not essential for plant development under stress.

Keywords: chaperone; glyoxalase; holdase; redox

1. Introduction

α -dicarbonyls, such as glyoxal (GO) and methylglyoxal (MG), are toxic compounds produced during glycolysis, metal-catalyzed glucose auto-oxidation, and lipid peroxidation. When they react with proteins, they form advanced glycation end-products (AGEs), which have been implicated in the progression of diseases such as diabetes, atherosclerosis, and neurological disorders [1–3]. Plants also accumulate sugar-derived MG as a byproduct of the Calvin cycle [4]. GO and MG are being removed by a glutathione (GSH)-dependent two-enzyme system, consisting of glyoxalase I (GLYI) and glyoxalase II (GLYII) [5]. A third glyoxalase, initially termed GLYIII [6], is a GSH-independent enzyme. This GLYIII protein is a member of the DJ-1/ThiJ/PfpI superfamily, and it is conserved across all kingdoms. In humans and plants, this enzyme is termed DJ-1, and it is known to exhibit a variety of cellular functions. Human DJ-1 protects rat neurons against oxidative stress [7], has chaperone [8], glyoxalase [9], and deglycase activity [10], and forms a scaffold for RNA-associated proteins [11].

DJ-1 has a conserved cysteine (Cys106 in human DJ-1), which is essential for its glyoxalase activity [12], and is easily oxidized to sulfinic and sulfonic acid by H₂O₂ [13,14].

Humans have only one isoform of homodimeric DJ-1, while plants have six isoforms encoded by tandem repeats and resulting in DJ-1 pseudodimers with varying subcellular localization [15]. In plants, the role of these DJ-1 isoforms in protecting against oxidative stress is poorly understood. Among Arabidopsis DJ-1 proteins, the experimental evidence is limited to cytosolic AtDJ-1A and chloroplastic AtDJ1-C. AtDJ-1A protects the plant from high light by acting as an antioxidant enzyme and a copper chaperone for superoxide dismutase (SOD) [16]. However, the SOD chaperone activity of DJ-1 proteins has been disputed [17]. Another DJ-1 homolog, AtDJ-1C, was found to be essential for plant viability and chloroplast development [18]. Previously, we identified AtDJ-1A to be sulfenylated *in vivo* [19], and more recently we also identified AtDJ-1B as a redox-sensitive sulfenylated protein in the chloroplasts [20].

In this study, we focus on AtDJ-1B, and evaluate the effect of H₂O₂ on the activity of this enzyme. We found that oxidation increases the thermal stability of AtDJ-1B, but inactivates its glyoxalase activity. We further demonstrated that AtDJ-1B is a holdase and that oxidation of AtDJ-1B has no effect on its chaperone activity. Finally, the analysis of two independent SALK T-DNA lines showed for the first time that the loss of AtDJ-1B does not affect the viability, growth, or stress response of Arabidopsis plants.

2. Materials and Methods

A summarized table of buffers referred to in this section can be found in Supplementary Information.

2.1. Cloning and Purification of Recombinant AtDJ-1B

An N-terminal Tobacco Etch Virus (TEV)-cleavage site was introduced into a codon-optimized Arabidopsis AtDJ-1B open reading frame by polymerase chain reaction (PCR), before subcloning into a pDEST15 expression vector with an N-terminal glutathione-S-transferase (GST)-tag (Gateway technology, Thermo Fisher Scientific, Waltham, MA, USA). This expression vector was transformed into SHuffle[®] T7 competent *E. coli* (New England Biolabs, Ipswich, MA, USA) and plated on agar to obtain a single colony, which was grown overnight at 30 °C in Luria-Bertani Broth (LB) supplemented with 100 µg/mL ampicillin. 1 L Terrific Broth (TB, with 100 µg/mL ampicillin) cultures were inoculated with this overnight pre-culture and grown at 30 °C until they reached the exponential growth phase, then cooled to 20 °C, supplemented with 0.4 mM isopropyl-β-D-1-thiogalactopyranoside and further grown overnight at 20 °C.

Cells were pelleted and resuspended in Lysis Buffer, lysed in a cell cracker at 20 kilopounds per square inch, and centrifuged at 40,000 × *g* for 30 min, 4 °C. The supernatant was passed through a 0.45-µm filter and loaded onto Glutathione (GSH) Sepharose High Performance resin (GE Healthcare Europe, Diegem, Belgium) equilibrated with Binding Buffer A. After 1 h of incubation, unbound material was removed by washing the resin with 10 column volumes Binding Buffer, and the GST-tag was cleaved by incubating the resin with an 1 mg of TEV protease per estimated 10 mg of total bound protein overnight at 4 °C. Non-bound material (containing cleaved AtDJ-1B) was collected and incubated for 1 h at 4 °C with Ni²⁺-Sepharose 6 Fast Flow resin (GE Healthcare Europe, Diegem, Belgium) equilibrated with Binding Buffer B, for the purpose of capturing the His-tagged TEV protease. Non-binding protein was collected and concentrated by Vivaspin 20 (Sartorius, Göttingen, Germany) centrifugal filtration and injected onto a size-exclusion (SE) Superdex75 16/60 column (GE Healthcare Europe, Diegem, Belgium) equilibrated with SE Buffer. Protein eluates containing AtDJ-1B were collected and concentrated by centrifugal filtration, and protein concentration determined by Bradford assay [21]. Protein was then flash-frozen in liquid nitrogen and stored at −80 °C.

2.2. Circular Dichroism (CD)

Purified AtDJ-1B was buffer-exchanged into 50 mM 4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid (HEPES) pH 7.3, 500 mM NaCl using Bio-Spin[®] 6 columns (Bio-Rad Laboratories N.V., Temse, Belgium), the protein concentration measured by Bradford assay, and treated with either 5 mM tris(2-carboxyethyl)phosphine (TCEP), or with several H₂O₂ molar ratios (2:1, 10:1; 100:1). Treatments with TCEP or H₂O₂ were carried out at 25 °C for 1 h, after which samples were buffer-exchanged into degassed, filtered 20 mM sodium phosphate pH 7.4, 100 mM KF on Bio-Spin[®] 6 (Bio-Rad Laboratories N.V., Temse, Belgium). These samples were centrifuged for 10 min at 20,000 × *g*, and the protein concentration was determined with the Bradford assay. Samples were vacuum-degassed for 10 min and CD spectra were collected at 20 °C using a J-715 spectropolarimeter (JASCO Inc., Easton, MD, USA) with a 0.1 mm cuvette from 250–185 nm using 1.0 nm sampling and averaged over 6 technical replicates. The spectrum of the buffer alone was subtracted from the resulting data, and the data converted to mean residue ellipticity. Spectra were deconvoluted by CONTIN ridge-regression analysis [22] within the DICHROWEB server [23,24], using a reference data set optimized for 190–240 nm.

2.3. Thermal Unfolding

The thermal stability of AtDJ-1B was assessed by monitoring the change in the absorbance of circularly polarized light at 222 nm as a function of temperature. Immediately following spectra acquisition by CD as described above, the same samples were subjected to a temperature gradient from 15–85 °C at a ramp of 1 °C min^{−1}, and the absorption at 222 nm measured at 0.2 °C intervals. The linear slopes of the initial and final baselines of the sigmoidal unfolding curves were fitted with equation 1. For equation 1, *A* is ellipticity in mdeg (θ), *m* and *k* are the pre-unfolding intercept and slope respectively, *T* is temperature, *n* and *l* are the post-unfolding intercept and slope respectively.

$$(A - (m + k * T)) / ((n + l * T) - (m + k * T)) \quad (1)$$

The resulting data of [U]/[F] was then plotted against T (Kelvin), and fitted with a Boltzmann Sigmoidal equation (Equation 2) in GraphPad Prism (version 7.0), where *y* is the data for [U]/[F], *slope* is the steepness of the curve, *x* is temperature in Kelvin, and *T_m* is the parameter for melting temperature (*T_M*).

$$y = y_{\min} + (y_{\max} - y_{\min}) / (1 + \exp\left(\frac{T_m - x}{\text{slope}}\right)) \quad (2)$$

2.4. Glyoxalase Assay

The glyoxalase activity of AtDJ-1B was estimated by monitoring the degradation of its substrate, glyoxal, in function of time. The amount of glyoxal in the sample was measured by derivatizing it with 1,2-diaminobenzene, as described for methylglyoxal [25].

A reaction mix containing 7.62 mM glyoxal and 571 nM AtDJ-1B in Assay Buffer A was incubated for 2 h at 30 °C on a thermoblock with mixing. Every 20 min an aliquot of the reaction mix was taken, 2-fold diluted in Assay Buffer A, and 1,2-diaminobenzene (Sigma-Aldrich, Overijse, Belgium) and HClO₄ were added to give a final concentration of 186 nM AtDJ-1B, 575 μM 1,2-diaminobenzene (Sigma-Aldrich, Overijse, Belgium) and 0.5 M HClO₄. After 5 min mixing, the absorbance of the derivatized sample at 340 nm was measured. For each time point measurements from control standard solutions of 0, 2, 4, 6, 8, 10 mM of glyoxal derivatized in the same manner as the samples containing AtDJ-1B were taken for generation of a standard curve. The assays were performed for three experimental replicates with averaging of two technical replicates.

Purified AtDJ-1B was either reduced with 5 mM TCEP or oxidized by treatment with varying molar excess (2-fold, 4-fold, 6-fold, 8-fold, 10-fold, or 100-fold) of H₂O₂, or with 5 mM diamide (Sigma-Aldrich, Overijse, Belgium). After 1 h of incubation at 25 °C the oxidizing or reducing agents

were removed by buffer exchange to Assay Buffer A using Bio-Spin® columns (Bio-Rad Laboratories N.V., Temse, Belgium).

2.5. Chaperone Assay

AtDJ-1B protein sample was buffer exchanged to Assay Buffer B and either reduced with 5 mM TCEP or oxidized by treatment with either 2-fold or 10-fold molar excess H₂O₂ to AtDJ-1B of 2:1 and 10:1 for 1 h at 25 °C. The samples were again buffer-exchanged into Assay Buffer B.

The thermal unfolding of citrate synthase (CS; from porcine heart; Sigma-Aldrich, Overijse, Belgium) was induced at 44 °C in Assay Buffer B with 0.24 μM citrate synthase (20 μg mL⁻¹) and either 3-fold molar excess of Hsp90 (Sigma-Aldrich, Overijse, Belgium) as positive control, 5-fold molar excess of lysozyme (from chicken egg white, (Sigma-Aldrich, Overijse, Belgium)) as negative control, or 5-fold or 20-fold molar excess of AtDJ-1B protein. Aliquots (17 μL) were taken every 20 min and added to a cuvette containing the CS Assay Buffer to a final volume of 500 μL. Citrate synthase (CS) activity was followed by monitoring the rate of decrease of acetyl-CoA at 233 nm.

2.6. Oxidation of AtDJ-1B for Mass Spectrometric Analysis

AtDJ-1B was reduced with 50 mM TCEP for 30 min at room temperature and excess of TCEP was removed by Bio-Spin® (Bio-Rad Laboratories N.V., Temse, Belgium) to a buffer containing 50 mM HEPES pH 7.3, 0.5 M NaCl. 73 μg of protein (25 μM) was mixed with 125-fold molar excess of dimedone prior to the addition of 10-fold molar excess of H₂O₂, and incubated at 37 °C for 1 h. H₂O₂ and dimedone were removed by Bio-Spin™ and the protein concentration determined. Iodoacetamide (IAM) was added at 400-fold molar excess (i.e., final concentration 8 mM) and the sample incubated for 1 h at room temperature in the dark.

2.7. Determination of Cysteine Oxidation States by Mass Spectrometry

For the identification of modified residues in AtDJ-1B, 10 μg of desalted proteins were denatured by methanol/chloroform precipitation and digested O/N with trypsin or chymotrypsin at 30 °C in 50 mM NH₄HCO₃ (pH 8.0).

After dissolving in eluent C (0.1% (v/v) trifluoroacetic acid in 2% (v/v) acetonitrile (ACN)), peptides were directly loaded onto reversed-phase pre-column (Acclaim PepMap 100, Thermo Fisher Scientific, Waltham, MA, USA) and eluted in backflush mode. Peptides were separated using a reversed-phase analytical column (Acclaim PepMap RSLC, 0.075 × 250 mm, Thermo Scientific), equilibrated in eluent A (0.1% (v/v) hydrofluoric acid (FA) with 4% eluent B (0.1% (v/v) FA in 80% (v/v) ACN, with a linear gradient of 4%–27.5% eluent B (0.1% hydrofluoric acid in 98% acetonitrile) for 100 min, 27.5%–40% eluent B for 10 min, 40%–95% eluent B for 1 min and holding at 95% for the last 10 min at a constant flow rate of 300 nL/min on an EASY-nLC 1000 ultra performance liquid chromatography (UPLC) system (Thermo Fisher Scientific, Waltham, MA, USA). Orbitrap Fusion Lumos tribrid mass spectrometer (Thermo Fisher Scientific, Waltham, MA, USA) was used for the analysis of the resulting peptides, which were then subjected to NanoSpray Ionization (NSI) source followed by tandem mass spectrometry (MS/MS) in Fusion Lumos coupled online to the UPLC. Orbitrap at a resolution of 120,000 was used for intact peptide detection. Peptides were selected for MS/MS using HCD (higher energy collisional dissociation) setting as 30; Orbitrap at a resolution of 30,000 was used for ion fragment detection. The top 20 precursor ions above a threshold ion count of 5.0³ in the MS survey scan with 20.0 s dynamic exclusion were subjected to a data-dependent procedure that alternated between one MS scan followed by 20 MS/MS scans. The electrospray voltage applied was 2.1 kV. MS1 and MS2 spectra were obtained with an AGC target of 4E5 ions and a maximum injection time of 50ms and an AGC target of 5E4 ions and a maximum injection time of 100 ms, respectively. The *m/z* scan range for MS scans was 350 to 1500. MS/MS data processing was performed using Sequest HT search engine within Proteome Discoverer 2.2 against a homemade protein database containing the recombinant AtDJ-1A and AtDJ-1B sequences. Trypsin or chymotrypsin was specified

as cleavage enzyme allowing up to 2 missed cleavages, 5 modifications per peptide, and up to 7 charges. Mass error was set to 10 ppm for precursor ions and 0.2 Da for fragment ions. Oxidation on Met, sulfenic-dimedone, sulfenic or sulfonic on Cys were considered as variable modifications. False discovery rate (FDR) was assessed using a fixed value PSM validator and thresholds for protein, peptide and modification site were specified at 1%. Site of covalent modification were manually validated. The mixed disulfides were identified by the use of the DBond software (Hanyang University, South Korea) [26].

2.8. Gene Expression Levels

The transcript level changes upon various treatments were obtained from several RNA Sequencing (RNA-Seq) experimental datasets, including: 3-h high light stress on *cat2-2* plants [27], treatment of Col-0 plants with 50 μM Antimycin A [28], Restricted Gas and Continuous Light (RGCL) treatment [29] of *cat2-2* or Col-0 plants for 24 h [30], 24 h of methyl viologen treatment of Col-0 plants (He et al., submitted) and *Pseudomonas syringae* infection of Col-0 plants (Stael et al., personal communication). Data visualization was performed using Heatmapper [31], genes were hierarchically clustered using Euclidean distance with average linkage.

2.9. Plant Material

All mutants used in this study are SALK T-DNA insertion lines in Col-0 background: *dj1a* (SALK_049637), *dj1b-4* (SALK_046449), *dj1b-9* (SALK_093414). All lines were genotyped prior to further analysis and homozygous plants were selected for phenotyping and confirmation of gene expression levels by RT-qPCR.

2.10. RT-qPCR

To confirm that T-DNA lines are true knockout lines, their RNA was extracted from three-week-old plants rosettes using TRIzol solubilization and extraction [32] followed by RNeasy Plant Mini Kit (Qiagen, Venlo, The Netherlands). First strand cDNA synthesis was performed using iScript cDNA Synthesis Kit (Bio-Rad Laboratories N.V., Temse, Belgium) using 1 μg of total RNA used as input material. The 5-fold diluted cDNA and gene-specific primers (Supplementary Table) were used for RT-qPCR performed by iCycler iQ (Bio-Rad Laboratories N.V., Temse, Belgium), with SYBR Green I Master Kit according to manufacturer's instructions. Data was analyzed by qBASE+ (Biogazelle, Zwijnaarde, Belgium), using *ELONGATION FACTOR 1 α* (*EF-1 α*) and *POLYUBIQUITIN 5* (*UBI5*) as reference genes. For each data point three biological and three technical replicates were used.

2.11. Growth Conditions and Plant Stress Assays

To assess the effect of high light, plants were grown in soil for 21 days in a controlled chamber (100 $\mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ light intensity, 16 h/8 h light/dark, 21 °C, 50% relative humidity). Three-week-old plants were transferred to high light (600 $\mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$) for 72 h. The bright-light pictures of the plants, as well as measurements of photosystem II (PSII) maximum efficiency (F_v'/F_m') using an Imaging-PAM-Series chlorophyll fluorescence system (HeinzWalz, Effeltrich, Germany) were taken every 24 h.

To assess mutants' response to various stress conditions in vitro, plants were surface-sterilized by fumigation, vernalized for 3–4 days at 4 °C and grown under controlled conditions (16 h/8 h light/dark, 100 $\mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ light intensity, 21 °C, 70% relative humidity) for 21 days. The medium was half-strength Murashige-Skoog medium (1/2 MS, 1% w/v sucrose, 0.8% w/v agar) containing one of the following additives: 25 mM mannitol, 50 mM NaCl, 25 nM methyl viologen or 1 μM 3-aminotriazole (3-AT). Plants were also grown on control plates, i.e., 1/2 MS with no additives. Plant rosette measurements were performed using ImageJ software [33] from 9 until 19 days after germination.

The Restricted Gas exchange and Continuous Light (RGCL) treatment was used to trigger photorespiratory stress in plants growing on 1/2 MS agar plates [34]. 21 days after vernalization, plates (either 1/2 MS or 1/2 MS + 3-AT) were sealed with multiple layers of Parafilm[®] M (Bemis

Company Inc., Oshkosh, WI, USA) in order to restrict gas exchange and transferred to continuous light ($100 \mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ light intensity, 21°C , 70% relative humidity) for 10 days. The bright-light pictures of the plants, as well as measurements of photosystem II (PSII) maximum efficiency (F_v'/F_m') using an Imaging-PAM-Series chlorophyll fluorescence system (HeinzWalz, Effeltrich, Germany) were taken every 2–3 days.

3. Results

3.1. AtDJ-1B Contains Multiple Oxidant-Sensitive Cysteines

The conserved catalytic cysteine of human DJ-1 is recognized as being highly prone to oxidation, with the most commonly observed oxidation state of this cysteine being sulfonic acid [35]. Arabidopsis AtDJ-1B is sulfenylated in planta [20], but nothing was known about its sulfenylation sites. AtDJ-1B is a pseudodimer containing two conserved DJ-1/PfpI domains in which there are eight cysteines, six within the N-terminal subunit, and two in the C-terminal subunit (Figure 1A). Of these eight cysteines, two correspond to the conserved catalytic cysteines of homodimeric human DJ-1 (Cys109 and Cys314), and seven are predicted to be solvent-exposed based on a homology model of AtDJ-1B (Cys129 being the sole buried residue). Hence, we sought to characterize the extent of cysteine oxidation in recombinantly expressed AtDJ-1B after exposure to H_2O_2 . Mass spectrometry (MS) analysis of AtDJ-1B treated for 1 h with a 10-molar excess of H_2O_2 showed, as expected, sulfenylation of the conserved active-site cysteines (Cys109 and Cys314) (Figure 1B). Sulfenylation also occurred at Cys110, and to a lesser extent at Cys129 and Cys339 (Figure 1B).

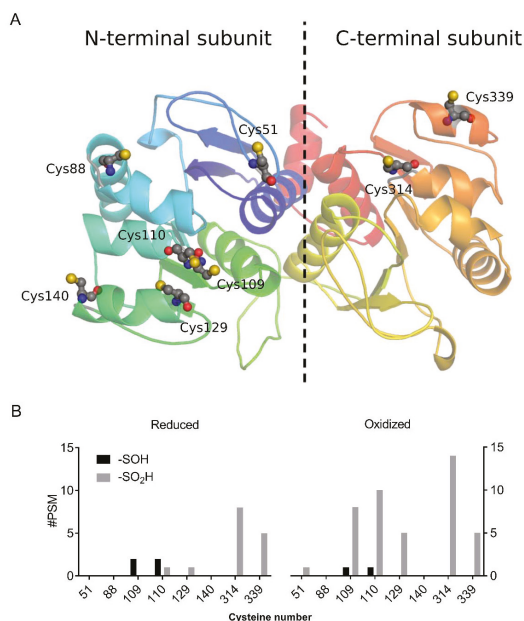


Figure 1. AtDJ-1B cysteines are differentially oxidized upon H_2O_2 treatment. (A) Homology model of AtDJ-1B with annotated cysteines in ball-and-stick representation. A dotted line approximately marks the 2-fold pseudosymmetry axis arising from the pseudo-dimeric AtDJ-1B monomer. Model generated using the I-TASSER structural prediction server, wherein the crystal structure of AtDJ-1D (PDB ID, 3UK7) was applied as a threading template. (B) Number of peptides (#PSM) detected by mass spectrometry that contained sulfenylated (-SOH) and sulfinylated (-SO₂H) cysteines, either in a sample reduced by 5 mM TCEP or oxidized by 10-fold molar excess of H_2O_2 for 1 h.

3.2. Oxidized AtDJ-1B Becomes More Thermostable

To assess the effect of the redox state of AtDJ-1B on its structural stability, we used circular dichroism (CD). We compared the relative secondary structure content of reduced and oxidized AtDJ-1B using increasing molar ratios of H_2O_2 (Figure 2A). The secondary structure of AtDJ-1B was not significantly altered upon oxidation, and both reduced and oxidized AtDJ-1B are characterized by 36–37% α -helical and 15–17% β -strand content, in agreement with previous findings, and comparable to what was observed for human DJ-1 [36].

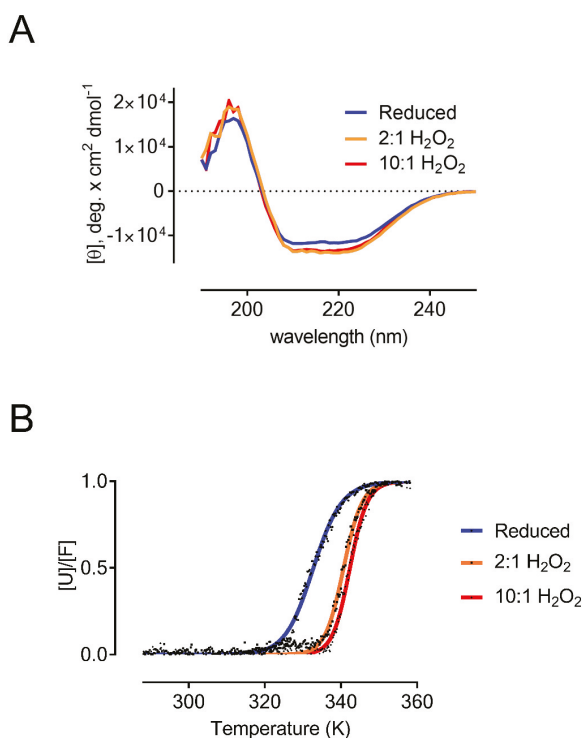


Figure 2. Oxidation does not influence the secondary structure of AtDJ-1B but increases its thermostability. **(A)** Comparative CD spectra of AtDJ-1B reduced with TCEP (blue trace), and DJ-1B oxidized by treatment with either a 2-fold (orange) or 10-fold (red) molar excess of H_2O_2 . A slight spectral shift can be observed for the H_2O_2 -treated protein relative to the reduced protein, though this spectral difference does not ultimately relate to any significant change in secondary structure. **(B)** Thermal unfolding of AtDJ-1B fitted to the Gibbs-Helmholtz equation. The change in ellipticity at 222 nm was followed by CD as a function of temperature for reduced AtDJ-1B (5 mM TCEP), and oxidized AtDJ-1B (either 2-fold, or 10-fold molar excess of H_2O_2). Data were converted to fraction of unfolded protein ($[U]/[F]$) and baseline-subtracted as described in Materials & Methods.

To evaluate the effect of oxidation on the thermostability of AtDJ-1B, we determined the melting temperature (T_M) of AtDJ-1B by following the change in ellipticity at 222 nm in function of temperature (Figure 2B). After fitting the data with a Boltzmann Sigmoidal equation, we obtained a T_M of 59.7 °C for reduced AtDJ-1B. For oxidized AtDJ-1B, the T_M increased to 67.6 °C and 69.3 °C following treatment with 2-fold and 10-fold molar excess of H_2O_2 , respectively. This demonstrates that the oxidized form of AtDJ-1B is significantly more thermostable than the reduced form.

3.3. Oxidation Inactivates the Glyoxalase Activity of AtDJ-1B

One of the demonstrated functions of DJ-1 proteins is as a GSH-independent glyoxalase enzyme [9,37,38]. To measure the glyoxalase activity of AtDJ-1B, we monitored the consumption of glyoxal in function of time through derivatization with 1,2-diaminobenzene (Figure 3). Then, under near-steady-state conditions (8 mM glyoxal) the effect of increasing molar ratios of H₂O₂ on the glyoxalase activity of AtDJ-1B was evaluated. Pre-treatment of AtDJ-1B with a 2-fold molar excess H₂O₂ resulted in a decrease of the relative activity to ~64% of the activity of the reduced enzyme, and a further decrease to ~16% after treatment with 10-fold molar excess H₂O₂. This dose-dependent oxidative inactivation of AtDJ-1B was further confirmed by complete inactivation of glyoxalase activity following treatment with a 100-fold molar excess of H₂O₂. Oxidation of AtDJ-1B by treatment with 5 mM diamide (disulfide bond formation) also elicited a decrease to ~17% of full activity.

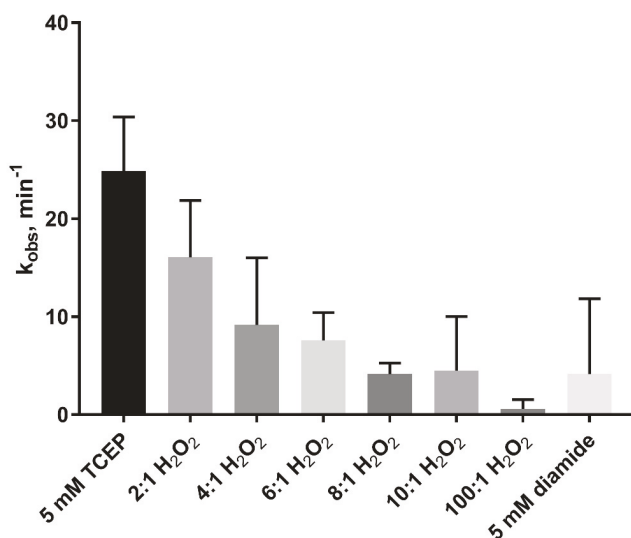


Figure 3. Glyoxalase activity of AtDJ-1B is lost after oxidation. Reduced and oxidized AtDJ-1B were prepared by pretreatment with either TCEP, varying molar ratios of H₂O₂: protein, or diamide. The reductant/oxidant was then removed prior to the activity assay. Displayed are observed rate constant (k_{obs}) values averaged from experimental triplicates with \pm standard deviation (SD) indicated.

3.4. Oxidation does not Affect the Chaperone Activity of AtDJ-1B

Hsp31, a close homolog of AtDJ-1B in *Saccharomyces cerevisiae*, has previously been demonstrated to function as a holdase and confer stress resistance [39–41]. To explore whether AtDJ-1B is capable of similar functionality, we assessed the ability of AtDJ-1B to act as a holdase towards citrate synthase (CS). Here we found that AtDJ-1B protects CS against thermal inactivation at 44 °C, with 5-fold molar excess of AtDJ-1B preserving 35% of CS activity after 40 min heat treatment compared to only 10% preservation when using a 5-fold molar excess lysozyme as a negative control (Figure 4A). Although AtDJ-1B offered less protection than the positive control of Hsp90 (preserving 68% of activity at a 3-fold molar excess), the extent to which AtDJ-1B protected CS against inactivation was proportional to the relative concentration of AtDJ-1B, with 20-fold molar excess resulting in greater protection than 5-fold excess (Figure 4A). Considering a putative role of AtDJ-1B in the oxidative stress response, we evaluated the impact of oxidation on the ability of AtDJ-1B to act as a holdase. Remarkably, no significant difference was observed between the chaperone effectiveness of reduced (TCEP-treated) and oxidized (H₂O₂-treated) AtDJ-1B, with all samples showing a similar level of protection against thermal inactivation of CS (Figure 4B).

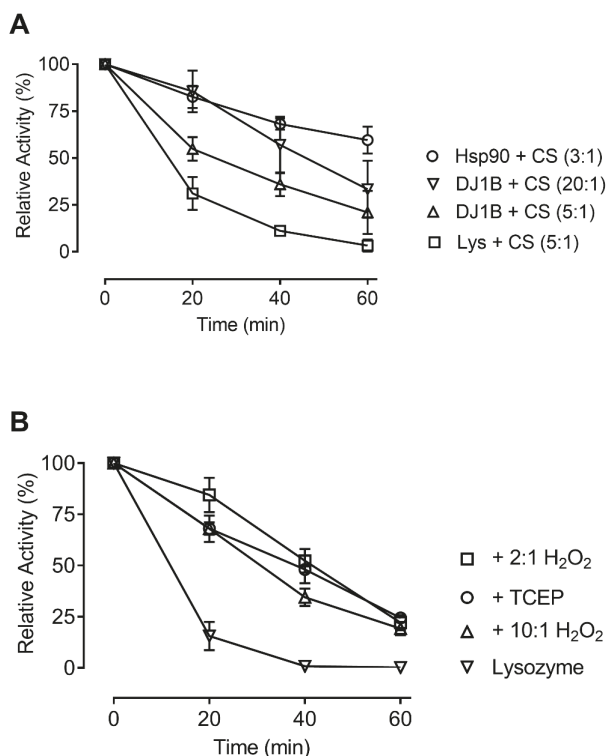


Figure 4. Suppression of thermal inactivation of citrate synthase (CS) by AtDJ-1B is independent of the oxidation state of AtDJ-1B. (A) At 44 °C, CS was incubated with 5-fold molar excess of lysozyme, 3-fold molar excess of Hsp90, 20- or 5-fold molar excess of DJ-1B at 44 °C, and activity measurements taken every 20 min for 1 h. (B) CS was incubated with 5-fold molar excess of lysozyme, or 20-fold molar excess of DJ-1B, either reduced with 5 mM TCEP, or oxidized with 10:1 or 2:1 molar excess of H₂O₂ at 44 °C, and activity measurements taken every 20 min for 1 h. Given is relative activity as a percentage of the full enzymatic activity of CS, just prior to heat-block incubation. Each data point represents average \pm SD for triplicates.

3.5. DJ-1B-Deficient Plants are Phenotypically Identical to Wildtype

Having established the possible activities of DJ-1B and their redox regulation, we sought to determine the phenotype of Arabidopsis T-DNA *dj1b* insertion lines. Two SALK T-DNA *dj1b* lines (further referred to as *dj1b-4* and *dj1b-9*, see Materials and Methods) and one SALK T-DNA *dj1a* line were genotyped and confirmed to be homozygous. The *dj1b-4* and *dj1b-9* mutants contain the T-DNA insert in the 4th exon and the 4th intron, respectively, while in the *dj1a* line the insert is located in the promoter region (Supplementary Figure S1). The lack of gene transcripts was validated by reverse transcription quantitative PCR (RT-qPCR) (see Supplementary data).

When grown in soil in a growth chamber (16 h of light, 8 h of dark, 100 $\mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$ light intensity), the *dj1b* and *dj1a* mutants were phenotypically identical to wildtype Col-0 plants (Figure 5). Since *dj1a* mutants were previously reported to show increased susceptibility to high light stress [16], we sought to confirm these findings and checked whether plants lacking AtDJ-1B show the same phenotype. After 3 days of high light stress treatment, both wildtype and mutant plants had no leaf lesions and their photosystem II efficiency, inferred from Fv'/Fm' ratios, remained unchanged (Figure 6).

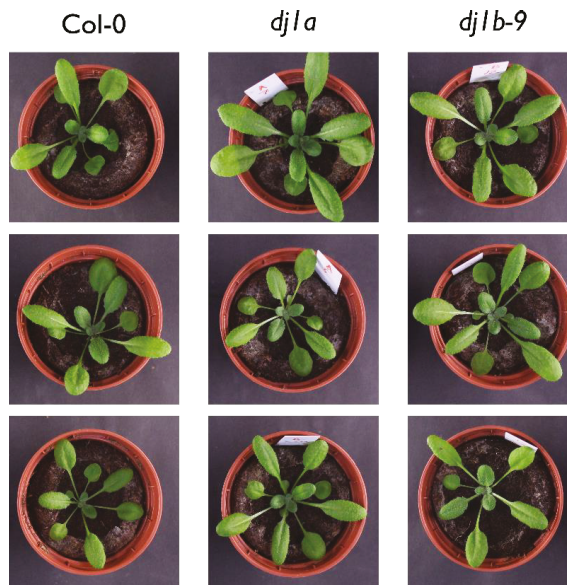


Figure 5. 3-week old plants lacking AtDJ-1A or AtDJ-1B are phenotypically identical to wildtype Col-0 plants. Col-0, *dj1a* and *dj1b-9* plants were grown for 3 weeks in a long-day light regimen (16 h/8 h light/dark, 21 °C for 3 weeks). Representative bright-light images were taken 21 days after sowing.

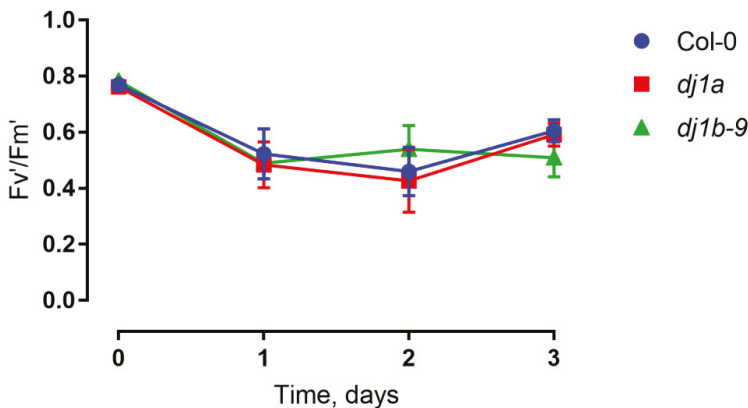


Figure 6. Photosystem II efficiency of mutants subjected to high light treatment was the same as that of wild type (WT). Three-week-old plants were exposed to high light intensities ($600 \mu\text{mol}\cdot\text{m}^{-2}\cdot\text{s}^{-1}$) for 72 h and the F_v'/F_m' ratios were measured. Each data point represents average from biological triplicates \pm SD.

To further characterize the phenotypes of *dj1a* and *dj1b* mutants, we grew them on $1/2$ Murashige-Skoog (MS) agar medium for 3 weeks and assessed their rosette sizes when plants were subjected to different stress agents in the medium: NaCl, mannitol, methyl viologen, 3-amino-1,2,4-triazole. In all cases, mutants grown on control $1/2$ MS medium or subjected to stress were characterized by the same rosette size as wildtype (Figure 7). Moreover, when plants were subjected to photorespiratory stress by a Restricted Gas exchange and Continuous Light (RGCL) treatment [34] for up to 10 days, the F_v'/F_m' ratio of the mutants was the same as for wildtype (Figure 8). Taken together,

these results indicate that the loss of DJ-1A or DJ-1B does not result in altered growth rates and photosynthetic capacities when exposed to the described abiotic stress conditions.

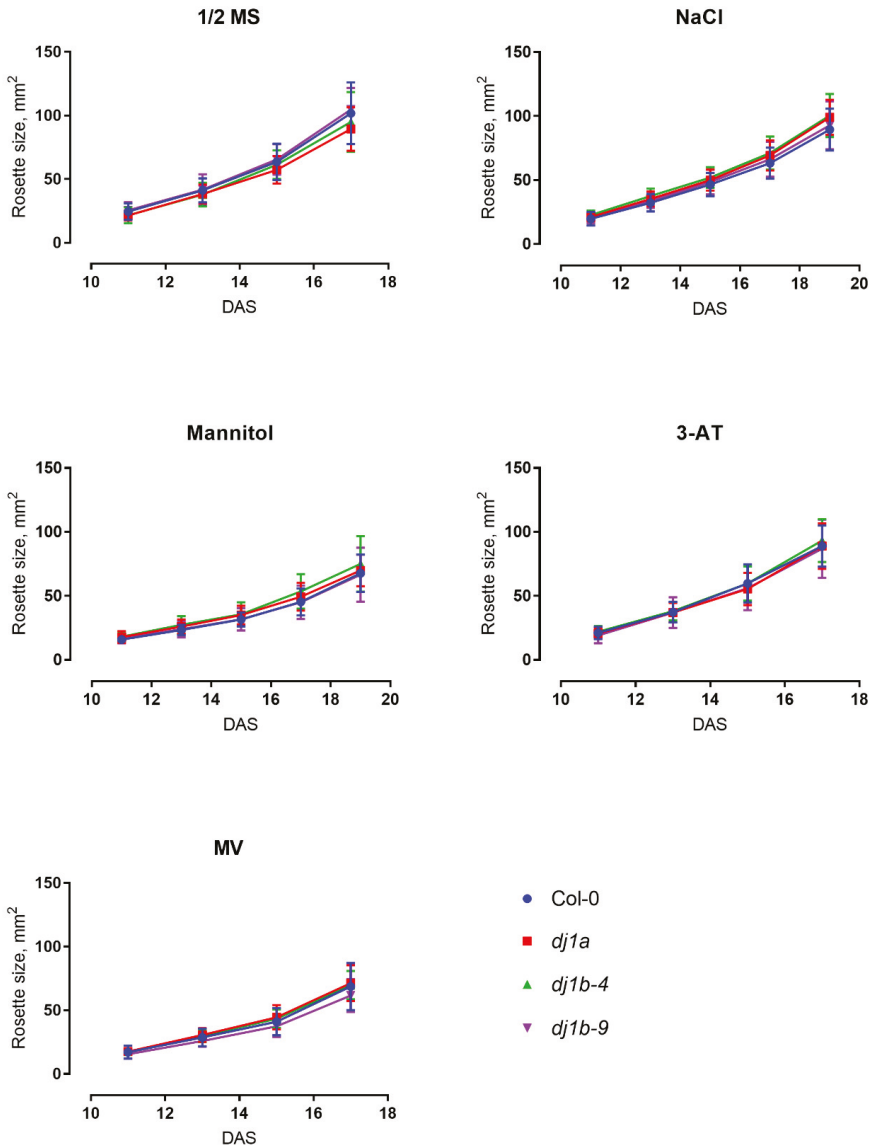


Figure 7. T-DNA insertion lines lacking AtDJ-1A or AtDJ-1B have no growth defects when subjected to various stresses. The plants were grown on 1/2 MS medium with stress-triggering additives: NaCl, mannitol, 3-amino-1,2,4-triazole (3-AT) or methyl viologen (MV) and their rosette sizes were analysed from bright-light images using ImageJ Software. Data points represent biological replicates ± SD.

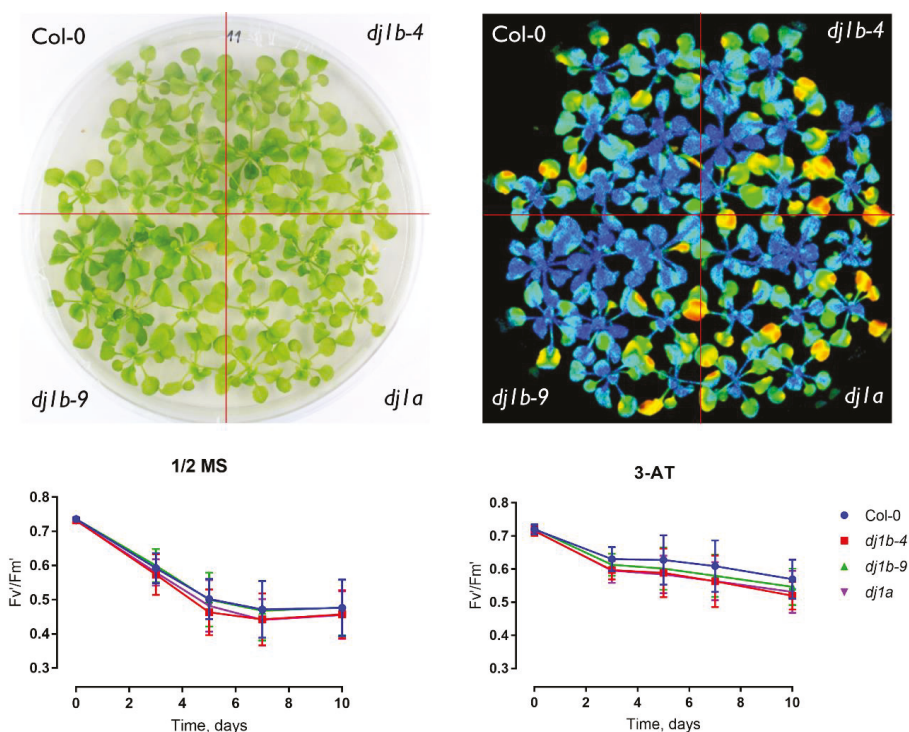


Figure 8. Plants deficient in AtDJ-1A or AtDJ-1B show no photorespiratory phenotype. (Upper panel) T-DNA *dj1a* and *dj1b* insertion lines do not differ from WT, as shown on a representative bright-light image (left) and color-coded F_v'/F_m' image (right) after 7 days of Restricted Gas exchange and Continuous Light (RGCL) treatment. (Lower panel) F_v'/F_m' decrease during the RGCL treatment. Data points represent averages from biological replicates \pm SD.

3.6. Transcript Levels of DJ-1B are Stress-Independent

Since various oxidative stress treatments trigger transcriptome-wide changes in Arabidopsis mRNA levels [42], we analyzed several RNA-Seq datasets, aiming to determine whether mRNA levels of all Arabidopsis DJ-1 homologs are also susceptible to such treatments (Figure 9). We concentrated on treatments triggering oxidative stress, such as RGCL [34], high light [27], methyl viologen treatment, and *Pseudomonas syringae* infection. In addition to wildtype Col-0, we also analyzed the transcriptome of *cat2-2* mutants, which are commonly used as stress-inducible systems to study oxidative stress in vivo [43].

The RNA-Seq dataset analysis confirmed the earlier finding that AtDJ-1A is upregulated under stress [16]. The same trend was observed for AtDJ-1E, whose mRNA levels were also upregulated in the majority of the stress conditions. Remarkably, for AtDJ-1F, which shares 76% protein sequence identity with AtDJ-1E, the mRNA levels are significantly decreased upon stress. The mRNA levels of the other three AtDJ-1 homologs are less susceptible to stress treatments; in particular AtDJ-1B transcription levels are not affected. This result is in line with previous studies of plant DJ-1 homologs showing different expression patterns and transcriptional responses to stress [44].

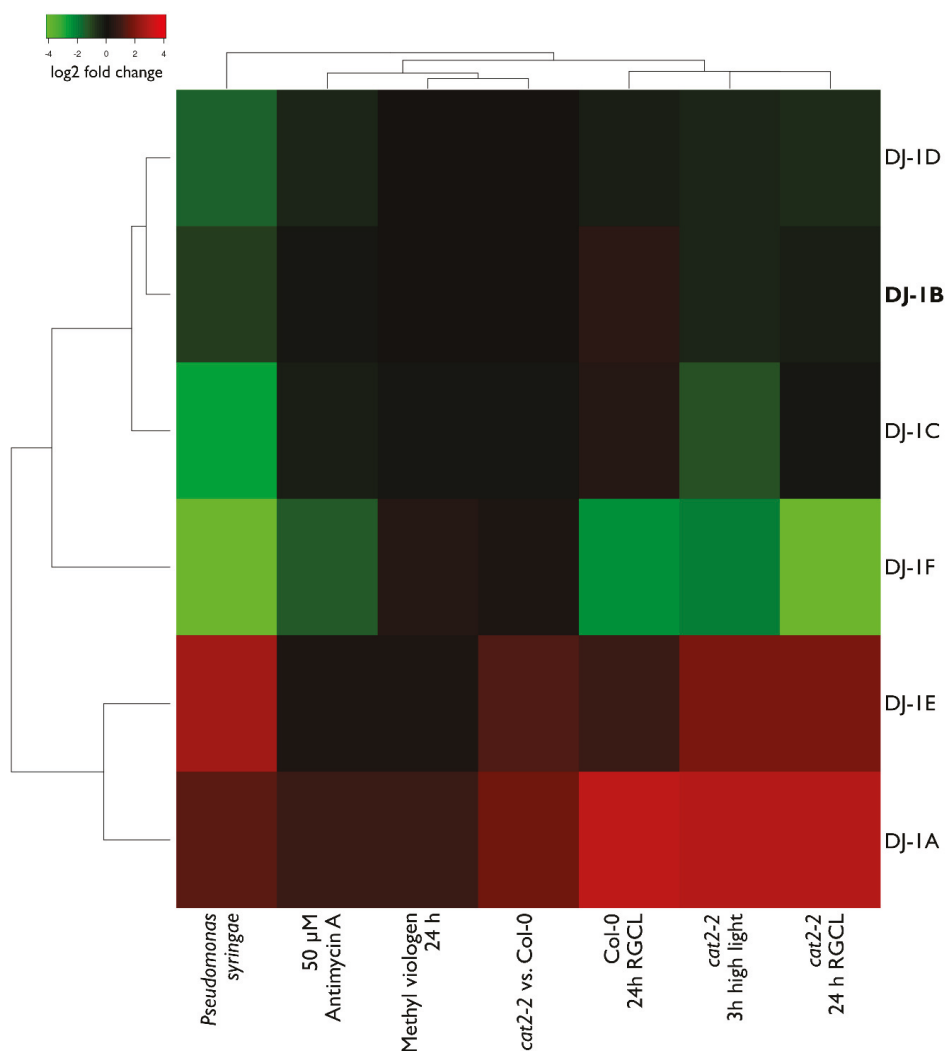


Figure 9. Arabidopsis DJ-1 gene expression levels respond differently after various stress treatments. Arabidopsis plants were either treated by a stress agent (pathogen *Pseudomonas syringae*, methyl viologen, high light, RGCL—Restricted Gas Continuous Light—treatment [29], or two lines (Col-0 and *cat2-2*) were compared to each other. The RNASeq experiments from which the data was obtained, as well as the detailed descriptions of the treatments are referred to in Materials and Methods. Colors represent log2 fold change.

4. Discussion

We report reduced AtDJ-1B as a glyoxalase with specific activity of $600 \text{ nmol} \cdot \text{min}^{-1} \cdot \text{mg protein}^{-1}$ (Supplementary Table). For non-reduced AtDJ-1B, a specific activity of $310 \text{ nmol} \cdot \text{min}^{-1} \cdot \text{mg protein}^{-1}$ was reported [12]. The glyoxalase activity of AtDJ-1B is ~ 35 times lower than for AtDJ-1D when glyoxal is used as substrate (Supplementary Table) [12]. This lower glyoxalase activity could be due to the absence of a conserved histidine, which has been suggested to facilitate proton transfer and glyoxal stereospecificity [45]. In plants under biotic and abiotic stress, and after H_2O_2 treatment,

the intracellular concentration of glyoxals has been shown to increase from 100 to 2000 $\mu\text{mol/g}$ tissue [46,47]. When methylglyoxal is photoreduced by photosystem I during photosynthesis, it donates electrons to O_2 , producing superoxide ($\text{O}_2^{\bullet-}$) [48]. Thus, by removing superoxide-generating glyoxals, enhanced glyoxalase activity contributes to cellular protection during oxidative stress. However, we clearly showed inactivation of the glyoxalase activity of AtDJ-1B by H_2O_2 , the predominant oxidant under oxidative stress conditions (Figure 4). To our knowledge, this is the first report of a DJ-1 protein losing its glyoxalase activity upon oxidation. Moreover, as the GLYI/II enzyme system catalyzes glyoxal/methylglyoxal detoxification with far greater catalytic efficiency than even AtDJ-1D [49], we propose that in plants the majority of glyoxals is detoxified by GLYI/II enzymes, rather than by H_2O_2 -sensitive DJ-1 proteins.

AtDJ-1C has been reported with no glyoxalase activity [12], and its knockout results in non-viable plants [18]. Therefore, we sought to explore possible alternative functions for AtDJ-1B, which shares its chloroplastic localization with AtDJ-1C. We found that AtDJ-1B acts as an oxidation-robust holdase, though a molar excess of AtDJ-1B is needed to protect CS from thermal inactivation. Since porcine CS used for the holdase activity assay is not its natural client protein, it still remains unclear whether the holdase function of AtDJ-1B would be its real physiological function. Furthermore, it is known that dedicated chaperones suppress aggregation of client proteins at stoichiometric concentrations [50], whereas AtDJ-1B showed holdase activity at 5 to 20-fold molar excess. Similarly to human DJ-1 [8], AtDJ-1B could have specific client proteins in Arabidopsis that still have to be identified.

Furthermore, human DJ-1 has been demonstrated to be a redox-dependent holdase, with the oxidized form of DJ-1 able to suppress α -synuclein aggregation significantly better than the reduced form, offering 20 to 60% more protection [8]. On the other hand, the yeast homolog Hsp31 is a redox-independent holdase and this chaperone activity is required for protecting yeast from cytotoxic stress [39]. We also found the holdase activity of AtDJ-1B to be redox-independent. Remarkably, we observed that the thermostability of AtDJ-1B increases upon oxidation, which indicates that the oxidized form of AtDJ-1B is more likely than reduced AtDJ-1B to retain its chaperone activity under sustained heat stress. Other DJ-1 family members have also been shown to be highly thermostable, with a T_M values ranging from 64 to 77 $^\circ\text{C}$ [36,51,52]. Here, we determined a comparatively lower T_M for AtDJ-1B of 59.7 $^\circ\text{C}$, and observe an increase in T_M of approximately 8–10 $^\circ\text{C}$ for oxidized AtDJ-1B. Similar increase in thermostability has also been observed for human DJ-1 and drosophila DJ-1 β , with respective T_M increases of 13.3 and 11.5 $^\circ\text{C}$ upon oxidation in the presence of a 7-fold molar excess of H_2O_2 [51].

Finally, it is worthy of mention that both human and prokaryotic DJ-1 homologs also have deglycase activity [53–55], and it was proposed to be one of the core physiological functional roles of DJ-1 proteins [56]. As glyoxalase function is required for deglycase activity, when glyoxalase activity is lost upon oxidation, the deglycase function will also be abolished. In both its role as a deglycase and as a holdase, it is reasonable to assume that AtDJ-1B has preferential target proteins. Further understanding of the in vivo role of AtDJ-1B will require the identification of its interaction partners in Arabidopsis.

In contrast to AtDJ-1C, whose loss results in infertile plants with numerous developmental defects [18], AtDJ-1B loss does not influence plant viability. We also found that *dj1b* plants show wild-type resistance to stress, at least under the conditions tested. Moreover, while it was previously reported that AtDJ-1A deletion leads to increased susceptibility to high light [16], the phenotype of the *dj1a* SALK T-DNA line described here is identical to wildtype, which directly contradicts earlier findings [16].

The Arabidopsis DJ-1 family originates from one ancestor gene and diverged into six genes through both whole-genome and tandem duplications [57]. The same mechanism is responsible for higher duplication rates in plants in general, when compared to other eukaryotes [58,59]. After duplication, proteins evolve new functions; one of the copies might acquire a novel function, or if the ancestral protein is multifunctional, the duplicates divide the original function, for instance by

differential expression patterns [59]. The analysis of RNASeq data presented in this paper shows that Arabidopsis DJ-1 proteins differ greatly in their expression patterns. These differences will be better understood after the analysis of their cis-regulatory elements in promoter sequences, like described for other plant species [44,60,61]. Moreover, since the mRNA levels are not representative for the protein levels, especially during dynamic phases such as cellular differentiation and stress response [62], it would be very informative to quantify all DJ-1 protein levels, both in stress and control conditions. Generating multiple mutant lines and biochemically characterizing all six Arabidopsis DJ-1 proteins will explain to what degree their functions overlap and whether the findings about the role of AtDJ-1B presented here are applicable to the other homologs as well.

5. Conclusions

In summary, we showed for the first time that Arabidopsis DJ-1B is a bifunctional protein, having both glyoxalase and holdase activity. Importantly, these two functions are differently regulated by H₂O₂; while the glyoxalase activity is lost upon oxidation, the holdase activity is not affected by H₂O₂. We also, for the first time, report the *dj1b* plant phenotype and prove that AtDJ-1B is not necessary for viability, development, or stress resistance of Arabidopsis plants, which might be due to redundant functions of all DJ-1 homologs. To explain why Arabidopsis has retained six DJ-1 homologs, the physiological relevance of the bifunctionality of AtDJ-1B still need to be confirmed in vivo and compared to those of the other AtDJ-1 proteins.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2076-3921/8/1/8/s1>, Figure S1: *AtDJ-1A* and *AtDJ-1B* gene models, Figure S2: *AtDJ-1B* and *DJ-1A* transcript levels (left and right, respectively) in WT and KO T-DNA lines, Table S1: Component summary of buffering solutions used, Table S2: AtDJ-1B specific glyoxalase activities and corresponding observed reaction rates determined during glyoxalase assay, Table S3: Arabidopsis thaliana DJ-1 specific activities [12], Table S4: Primers used for the study, Table S5: Log₂ fold change values of DJ-1 mRNA expression levels, as visualised on Figure 9.

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Article

Inactivation of Carbonyl-Detoxifying Enzymes by H₂O₂ Is a Trigger to Increase Carbonyl Load for Initiating Programmed Cell Death in Plants

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Abstract: H₂O₂-induced programmed cell death (PCD) of tobacco Bright Yellow-2 (BY-2) cells is mediated by reactive carbonyl species (RCS), degradation products of lipid peroxides, which activate caspase-3-like protease (C3LP). Here, we investigated the mechanism of RCS accumulation in the H₂O₂-induced PCD of BY-2 cells. The following biochemical changes were observed in 10-min response to a lethal dose (1.0 mM) of H₂O₂, but they did not occur in a sublethal dose (0.5 mM) of H₂O₂. (1) The C3LP activity was increased twofold. (2) The intracellular levels of RCS, i.e., 4-hydroxy-(*E*)-hexenal and 4-hydroxy-(*E*)-nonenal (HNE), were increased 1.2–1.5-fold. (3) The activity of a reduced nicotinamide adenine dinucleotide phosphate (NADPH)-dependent carbonyl reductase, scavenging HNE, and *n*-hexanal was decreased. Specifically, these are the earliest events leading to PCD. The proteasome inhibitor MG132 suppressed the H₂O₂-induced PCD, indicating that the C3LP activity of the β 1 subunit of the 20S proteasome was responsible for PCD. The addition of H₂O₂ to cell-free protein extract inactivated the carbonyl reductase. Taken together, these results suggest a PCD-triggering mechanism in which H₂O₂ first inactivates a carbonyl reductase(s), allowing RCS levels to rise, and eventually leads to the activation of the C3LP activity of 20S proteasome. The carbonyl reductase thus acts as an ROS sensor for triggering PCD.

Keywords: lipid peroxide; oxidative stress; oxylipin; reactive carbonyl species (RCS); reactive electrophile species (RES); redox signal; ROS

1. Introduction

Reactive oxygen species (ROS), such as superoxide radical and hydrogen peroxide (H₂O₂), are metabolic outcomes of various cellular processes in plants. Versatile biological actions of ROS, collectively termed as oxidative signals, largely depends on the ROS concentration [1]. Recent evidence shows that ROS are involved in diverse signaling functions in plants, from cell cycle regulation to whole plant senescence [2]. The transition from cellular proliferation to cell elongation, which is necessary for plant growth in the early stages of cell differentiation, is regulated by ROS signals [3]. The levels of superoxide radical and H₂O₂ are high in the elongation and meristematic zone of Arabidopsis root [4]. ROS also play an important role in seed germination; superoxide and H₂O₂ are increased in the radicle of Arabidopsis during germination [5]. Thus, ROS signals are necessary in various stages of plant growth and development.

Programmed cell death (PCD) is an essential phenomenon in plants for maintaining their growth and development. In many cases, PCD is started by the rise of ROS levels, which is stimulated by various endogenous or exogenous stimuli. Aleurone layer of cereal grains and the lateral root cap cells are eliminated as a result of PCD with a rise of H₂O₂ level [6,7]. Under water deficit condition rice developing anthers started PCD due to enhancement of H₂O₂ level and down-regulation of antioxidant transcripts [8]. In response to salinity and sorbitol stress, tobacco BY-2 cells increase the level of superoxide radical, which triggers PCD [9]. High temperature causes a rise in the H₂O₂ level in tobacco BY-2 cells before initiation of PCD [10]. H₂O₂ is also involved in the hypersensitive response (HR)-like cell death in tobacco leaves [11]. However, the mechanism of ROS action to initiate PCD was largely unclear.

Recently, we found that, in the mechanism by which ROS initiates PCD, the lipid peroxide-derived reactive carbonyl species (RCS) play critical roles in conveying ROS signals [12]. RCS is a group name of the α,β -unsaturated aldehydes and ketones, such as acrolein and 4-hydroxy-(*E*)-2-nonenal (HNE) [13,14]. RCS are generated from lipid peroxides (LOOH) and act as agents to mediate ROS signal to target proteins in the heat shock-responsive gene regulation [15], senescence [16], abscisic acid (ABA) signaling for stomatal closure [17,18], and auxin signaling for lateral root formation [19] in plants. We reported that tobacco BY-2 cells exposed to H₂O₂ generated acrolein and HNE, and a chemical carbonyl scavenger prevented the initiation of PCD, without affecting the increase in the intracellular ROS level [12].

We further revealed that the targets of RCS were two caspase-like proteases [20]. Caspases are cysteine-containing aspartate-specific proteases that are involved in the initiation of PCD in animals. Plants do not have proteins homologous to animal caspases but have caspase-like proteases that show the same substrate specificity for the artificial peptides to caspase-1 and caspase-3. Vacuolar processing enzyme (VPE), an ortholog of asparaginyl endopeptidase (legumain), shows the caspase-1-like protease (C1LP) activity and is responsible for processing vacuolar proteins [21]. The functions of C1LP in plants and caspase-1 in animals to determine cell fate are comparable [22]. Activation of C1LP is associated with the aluminum-induced PCD in the root tip of a peanut [23]. The caspase-3-like protease (C3LP) is also involved in the initiation of PCD in cadmium-induced PCD [24] and hypersensitive response-induced PCD [25] in *Arabidopsis thaliana*. High temperature treatment to tobacco BY-2 cells increases the H₂O₂ level, followed by the activation of C3LP before PCD occurs [10]. In our previous study [20], we found that H₂O₂ addition to BY-2 cells activated both C3LP and C1LP before PCD symptoms become apparent. The addition of acrolein and HNE to the cell-free extract rapidly activated both C3LP and C1LP, but the addition of H₂O₂ to the extract failed to activate the proteases. These results indicated that, in BY-2 cells, the RCS accumulating due to the H₂O₂ stimulus activated C3LP and C1LP and started PCD. Thus, the activation of C3LP and C1LP was ascribed to RCS that was generated by the oxidative stimulus.

We tentatively concluded that C3LP was more relevant than C1LP to the initiation of PCD because, on addition of lethal dose of acrolein, the activation of the former was more rapid and larger than that of the latter [20]. However, the contribution of C1LP to the H₂O₂-induced PCD remains open. It is also unclear which protein is responsible for the RCS-stimulated activity of C3LP because two distinct proteins have been known to show C3LP activity: the β 1 subunit of the 20S proteasome (PBA1) [25,26] and cathepsin B [27].

Another critical question is about the type of RCS that activated C3LP. In our previous study, we analyzed carbonyls in the cells 2 h after H₂O₂ addition and found that at least eight types of oxylipin carbonyls were increased [12]. The two potent RCS acrolein and HNE, found as the increasing carbonyls, showed the ability to activate C3LP and C1LP *in vitro* [20]. On the other hand, C3LP was activated within 30 min after H₂O₂ addition. It is therefore expected that certain types of RCS should be increased prior to the C3LP activation on the addition of H₂O₂ to BY-2 cells.

In this study, attempting to elucidate the detailed mechanism of the H₂O₂-induced PCD of BY-2 cells, we monitored early biochemical events in response to oxidative stimulus. We confirmed that

C3LP is critical for the PCD, and found its activity is attributable to PBA1. In response to H₂O₂, BY-2 cells accumulated HNE and 4-hydroxy-(*E*)-hexenal (HHE) in 10 min. Interestingly, at the same time, an NADPH-dependent carbonyl reductase(s), showing HNE-reducing activity, was largely inactivated. The same activity in cell-free extract was inactivated by the addition of H₂O₂, indicating that H₂O₂ attenuated the carbonyl reductase activity to enhance the RCS level. We propose a new mechanism of ROS-induced PCD, where the carbonyl reductase(s) acts as an ROS sensor for initiating the PCD process.

2. Materials and Methods

2.1. Culture of Cells

Cell suspension of tobacco BY-2 (*Nicotiana tabacum* L. cv. Bright Yellow-2) (RIKEN BioResource Research Center, Tsukuba, Japan) were cultured in Murashige and Skoog medium (MP Bio, Tokyo, Japan) supplemented with sucrose (30 g L⁻¹), myo-inositol (100 mg L⁻¹), KH₂PO₄ (200 mg L⁻¹), thiamine HCl (0.5 mg L⁻¹), and 2,4-dichlorophenoxyacetic acid (0.2 mg L⁻¹) (all these chemicals were purchased from FUJIFILM Wako Pure Chemical, Osaka, Japan) at pH 5.6. Cells were propagated with continuous rotation at 120 rpm in darkness at 25 °C. Every seven days, 0.5 mL dense cells were sub-cultured to a fresh medium (50 mL). The exponential growth phase cells at the fourth day were selected for experiment [28]. H₂O₂ was added to the culture medium (H₂O₂ treatment). After an incubation, the cells were harvested and washed with distilled water for analyses.

2.2. Cell Viability Assay

Cell viability was determined by fluorescein diacetate (FDA) staining. Harvested cells were incubated in FDA solution (1 µg mL⁻¹) for 5 min, washed twice with distilled water, and then observed under a fluorescence microscope (Leica AF6000, Wetzlar, Germany). Viable cells cleave FDA to form fluorescein, which fluoresce with excitation at 485 nm and emission at 515 nm. Dead cells do not fluoresce.

2.3. Determination of C1LP and C3LP Activities

Protein extract from BY-2 cells was prepared as described previously [20,21]. The harvested cells were frozen with liquid nitrogen and ground to powder with a mortar and pestle. Then, the ground powder of cells was transferred to a centrifuge tube and immediately added the protease extraction medium containing 50 mM sodium acetate, pH 5.5, 50 mM NaCl, 1 mM EDTA, 1 mM phenylmethylsulfonyl fluoride (PMSF), and 0.1 mM E64-d (a thiol protease inhibitor). After centrifugation at 14,000× *g* for 30 min at 4 °C, supernatant was collected as the protein extract.

The tetrapeptide fluorogenic substrates Ac-YVAD-AMC for C1LP and Ac-DEVD-AMC for C3LP and the respective inhibitors (Ac-YVAD-CHO and Ac-DEVD-CHO) were purchased from Peptide Institute (Osaka, Japan). For measuring a protease activity, two tubes were prepared. One had a reaction mixture (0.2 mL) containing the protease sample (50 µL of protein extract or 50 µg of protein) in 20 mM Na-acetate, pH 5.5, 0.1 M dithiothreitol, 0.1 mM EDTA, and 1 mM PMSF with an inhibitor at 0.1 mM, and the other had the same reaction mixture without the inhibitor. They were preincubated at 37 °C for 1 h. Then, the substrate peptide was added to each tube at 0.1 mM. After an incubation at 37 °C for 1 h, the AMC released from the substrate was determined with a fluorescence microplate reader (Twincler LB970, Berthold Japan, Tokyo, Japan) (excitation 380 nm; emission 445 nm). The specific activity of proteases was calculated from the difference between the absence and the presence of an inhibitor. A standard curve was developed using a series of AMC (Peptide Institute) solutions in the 0 nM to 200 nM range.

In vitro activation of C1LP and C3LP activity was done by the addition of either acrolein or H₂O₂ to protein extract. After incubation, the protein extract was allowed passage through a PD MiniTrap G-25 column (GE Healthcare, Tokyo, Japan), equilibrated with protease extraction medium, to remove small molecules. A 0.2 mL of reaction mixture with the eluted protein extract was prepared similarly as described above.

2.4. Analysis of Glutathione

Total glutathione (GSH + oxidized form (GSSG)) was determined as described previously [20,29] with a minor modification. Briefly, harvested cells (about 0.4 g) were immediately ground in liquid nitrogen with mortar and pestle and suspended in two volumes of cold 5% sulphosalicylic acid in a 1.5 mL centrifuge tube. For glutathione analysis, the supernatant was collected after centrifugation at $20,000\times g$ for 15 min at 4 °C. The cell extract (a 0.1 mL aliquot) was neutralized by mixing with 0.9 mL of 0.1 M HEPES-KOH, pH 7.4, containing 5 mM EDTA. The mixture was divided into two fractions for determining total GSH (GSH + oxidized form (GSSG)) and GSSG. A 1.0 mL reaction mixture was prepared with 0.3 mL of neutralized cell extract containing 0.1 M HEPES-KOH, pH 7.4, 5 mM EDTA, 10 mM 5,5'-dithiobis-2-nitrobenzoic acid (DTNB), 0.5 unit glutathione reductase. The reaction was started with the addition of 0.2 mM NADPH, and the rate of absorbance increase at 412 nm was recorded for 1 min. The GSSG content was determined after masking GSH by 20 μ L of 2-vinylpyridine to the neutralized cell extract. The emulsified residual 2-vinylpyridine was removed through a brief centrifuge; then, GSSG was determined, as described above, for total GSH. The GSH content was estimated after the subtraction of GSSG to the total GSH. A GSH standard curve was prepared with authentic GSH (Wako, Japan) in the range of 0–0.1 mM.

2.5. Analysis of Ascorbate

Asc and dehydroascorbic acid (DHA) were determined as described previously [20,30] with some modifications. The cells were extracted with the same protocol of GSH analysis. The total ascorbate (Asc + DHA) content was estimated in a 1 mL reaction mixture containing 0.1 mL aliquot of cell extract, 0.2 mL of 150 mM phosphate, pH 7.4, containing 5 mM EDTA, and 0.05 mL of 10 mM dithiothreitol (DTT). To remove excess DTT, 0.05 mL of 0.5% *N*-ethylmaleimide was added after 10-min incubation at room temperature. To determine Asc, 0.1 mL of water was added in the reaction mixture instead of DTT and *N*-ethylmaleimide. Then, the reaction mixture was incubated at 40 °C for 40 min with 0.2 mL of 10% trichloroacetic acid, 0.2 mL of 44% ortho-phosphoric acid, 0.2 mL of 4% α,α -dipyridyl in 70% ethanol, and 0.3% (*w/v*) FeCl_3 for color development. The absorbance at 525 nm was measured, and the concentration of DHA was calculated from the difference of total ascorbate (Asc + DHA) and Asc. A standard curve was prepared with Asc in the 0–250 μ M range.

2.6. Detection and Quantification of Carbonyls Using HPLC

The fresh cells (0.4 g) were filtered and washed twice with distilled water, then immediately transferred into a screw-capped glass tube containing 2.5 mL of acetonitrile containing internal standard (25 nmol of 2-ethylhexanal) and 0.005% (*w/v*) butylhydroxytoluene to prevent autoxidation of unsaturated organic compounds. Then, the tube was closed and incubated in a water bath at 60 °C for 30 min. Avoiding cell debris, only clear solution was collected by Pasteur pipette to another glass tube, and the carbonyls were derivatized with 2,4-dinitrophenylhydrazine (0.5 mM) and formic acid (0.5 M) for 60 min at 25 °C. Then, the formic acid in the solution neutralized with 0.45 g of NaHCO_3 in 2.5 mL of saturated NaCl. After thorough mixing and subsequent centrifugation, the upper layer was collected, dried, and dissolved in a smaller volume of acetonitrile. The hydrazone-derivatized carbonyls were determined on a reverse phase (C30) HPLC [12,17,31,32].

2.7. Determination of Enzymatic Activity

Proteins were extracted as follows. The cells harvested and washed on a nylon mesh were frozen with liquid nitrogen and ground to powder with a mortar and pestle. About 0.4 g powdered cells were transferred to the reductase extraction medium containing 50 mM KH_2PO_4 , pH 7.4, 1 mM EDTA, and 1 \times Protease Inhibitor Cocktail (Sigma-Aldrich Japan, Tokyo, Japan) and extensively mixed. After centrifugation with $10,000\times g$ for 10 min at 4 °C, 0.5 mL cell extract was allowed passage of the PD MiniTrap G-25 column equilibrated with reductase extraction medium. Protein content was

determined with the Protein Assay CBB solution (Nacalai Tesque, Kyoto, Japan) with bovine serum albumin as the standard.

The aldehyde reducing activity by the reductases of the protein extract was measured using 50 mM Mes-NaOH, pH 6.0, containing 0.1 mM NADPH. Various species of aldehydes (HNE, *n*-hexanal, acetaldehyde, butyraldehyde, (Z)-3-Hexenal) were used as the substrate of reductases and the rate of oxidation of NADPH was measured at 340 nm using a Shimadzu (Kyoto, Japan) MPS-2000 spectrophotometer.

3. Results

3.1. Threshold Level of Hydrogen Peroxide for Inducing PCD

Because addition of H₂O₂ to BY-2 cells caused a rapid increase (within 10 min) in C3LP and C1LP [20], it is expected that certain types RCS are generated in the cells. On the other hand, addition of H₂O₂ is generally oxidative stimulus, and it will cause lipid peroxidation, leading to the formation of various RCS. We attempted to distinguish the RCS that are responsible for the triggering of PCD from those generated as a consequence of the progress of oxidative stress. For this purpose, we have set up two H₂O₂ treatment conditions, i.e., lethal and sublethal. BY-2 cells were treated with H₂O₂ at various concentrations, and it was found that 1 mM H₂O₂ induced about 75% cell death in 5-h incubation, while 0.5 mM H₂O₂ did not increase cell death as compared with untreated control (Figure 1). Specifically, 1 mM H₂O₂ was a lethal dose to induce cell death and 0.5 mM was sublethal. The biochemical events that triggered PCD thus should have occurred earlier than 5 h only in the cells treated with 1 mM H₂O₂ but should not in those treated with 0.5 mM H₂O₂.

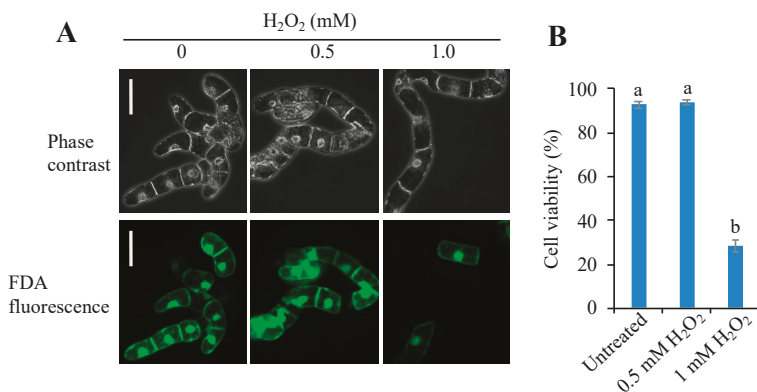


Figure 1. Effects of H₂O₂ concentration on the viability of Bright Yellow-2 (BY-2) cells. A 0.5 mL aliquot of cells from 7-d culture were sub-cultured in 50 mL of fresh culture medium and, after 4 d, the culture medium was supplemented with H₂O₂ to 0.5 mM or 1 mM. After 5 h incubation cells were harvested and stained with fluorescein diacetate (FDA). Living cells and dead cells were counted as described in the Materials and Methods section. Cells forming a single layer under microscopy were chosen for evaluation. (A) Typical phase-contrast (top) and FDA-fluorescence (bottom) images of the same field. Bar = 50 μ m. (B) Percentage of cell viability. The total number of cells was counted under phase contrast observation, and the FDA-stained cells were counted under fluorescence observation. A total of 200 cells were observed in each treatment. Means \pm SE of three independent experiments are shown. Differences among treatments were analyzed by Tukey's test: $p < 0.05$. Different letters among the treatments represent statistically significant differences.

3.2. Caspase-Like Proteases in Tobacco BY-2 Cells Are Activated Only When H₂O₂ Induces PCD

The addition of H₂O₂ to BY-2 cells at a lethal dose activated C3LP and C1LP within 30 min [20]. To compare the effects of sublethal and lethal doses of H₂O₂, we treated BY-2 cells with H₂O₂ at 0.5 and 1 mM and determined the caspase activities (Figure 2). Untreated cells exhibited constitutive levels of C3LP and C1LP activity. H₂O₂ at the lethal dose (1 mM) increased C3LP significantly in 10 min, and after 30 min, the activity reached 2.5-fold. A sublethal dose of H₂O₂ (0.5 mM) did not increase the C3LP activity (Figure 2A). By the lethal dose of H₂O₂, the C1LP activity was also increased to a significantly higher level in 60 min, while in 10 min, the increase was apparently smaller (Figure 2B). The sublethal dose of H₂O₂ caused a very slight increase in the C1LP activity. Thus, early activation of C3LP and C1LP was consistently associated with the later occurrence of PCD.

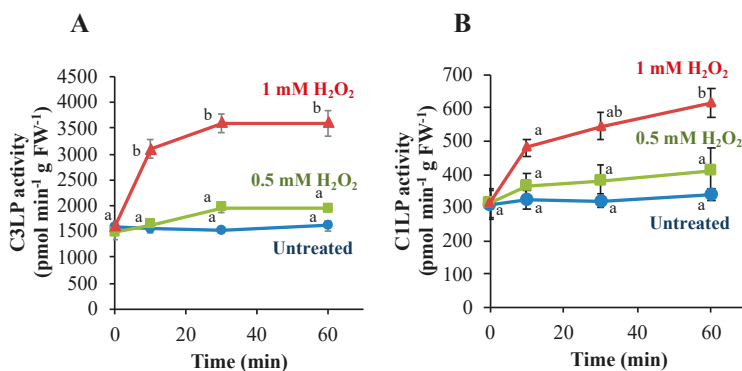


Figure 2. Activation of caspase-3-like protease (C3LP) (A) and caspase-1-like protease (C1LP) (B) in BY-2 cells after exposure to H₂O₂. Four-d cultured cells were treated with H₂O₂ to 0.5 mM or 1 mM, as in Figure 1. Proteins were extracted from the cells as described in the Material and Methods section. The activity of C3LP (A) and C1LP (B) was measured with the respective substrates Ac-DEVD-AMC and Ac-YVAD-AMC, as in the Materials and Methods section. The data are the mean \pm SE. Different letters represent significantly different values ($p < 0.05$ on Tukey test).

3.3. Activation of C3LP, Rather Than C1LP, Triggers PCD

The activation of C3LP in response to a lethal dose of H₂O₂ was more rapid and to a larger extent than C1LP (Figure 2). To evaluate the contribution of each protease to PCD initiation, we treated the cells with the inhibitors specific to C3LP or C1LP and examined whether one of them or both prevents H₂O₂-induced PCD or not (Figure 3). The H₂O₂ treatment caused PCD in 75% cells in 5 h. The addition of C3LP inhibitor efficiently suppressed H₂O₂-induced cell death; only 25% cells were dead. On the other side, C1LP inhibitor could not prevent H₂O₂-induced cell death effectively; 70% cells were dead. We also tested whether these caspase inhibitors prevent the PCD induced by 0.2 mM acrolein. The addition of acrolein at 0.2 mM induced 90% cell death after 5-h incubation (Figure S1). Addition of the C3LP inhibitor prior to the acrolein treatment suppressed the cell death to 30%, as compared with 80% death for the acrolein-treatment without protease inhibitors. On the other side, the C1LP inhibitor did not efficiently prevent acrolein-induced cell death. These results suggest that the activation of C3LP is critical for the PCD initiated by oxidative stimuli.

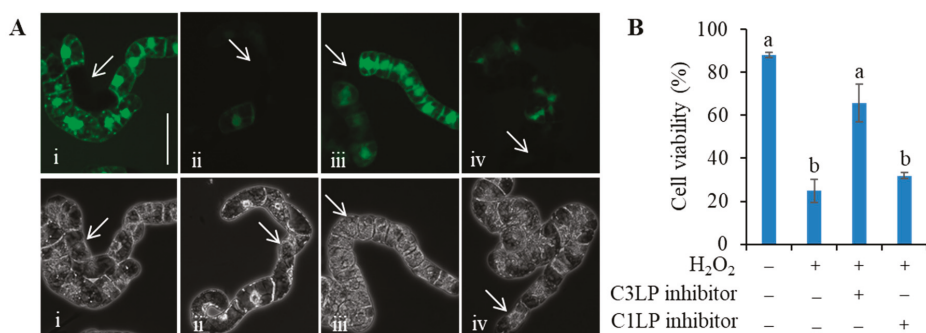


Figure 3. Effects of caspase inhibitors on the H₂O₂-induced cell death. Four-d cultured BY-2 cells were treated, as in Figure 1, with 1 mM H₂O₂ in the presence or absence of an inhibitor. The cells were harvested after 5-h incubation, and FDA-stained cells were counted as living cells. (A) Typical fluorescence images (top) and phase contrast images (bottom) of the same field: (i) untreated cells as control, (ii) 1 mM H₂O₂, (iii) 1 mM H₂O₂ + 20 μM C3LP inhibitor, and (iv) 1 mM H₂O₂ + 20 μM C1LP inhibitor (top). White arrows indicate dead cells. Bar = 50 μm. (B) Percentage of living cells. A total of 400 cells were observed in each treatment. Mean of 3 runs ± SEM. Different letters represent significantly different values (*p* < 0.05 on Tukey test).

3.4. The C3LP Activity of the 20S Proteasome Is Responsible for the H₂O₂-Initiated PCD

Two distinct proteins are known to show the C3LP activity, i.e., PBA1, a subunit of the 20S proteasome [25,26,33] and cathepsin B [27]. Vacca et al. [34] measured ROS accumulation and C3LP in the heat shocked tobacco BY-2 cells in absence and presence of the proteasome inhibitor MG132. They found that MG132 prevented PCD and specifically decreased ROS level and C3LP activity. To clarify which protein is really involved in the H₂O₂-induced PCD, we tested the effect of MG132. BY-2 cells were pre-incubated with MG132 at 0.1 mM for 30 min, and then H₂O₂ was added at 1 mM to induce PCD. H₂O₂ at 1 mM in 5-h incubation induced about 75% cell death, and proteasome inhibitor MG132 significantly reduced cell death to 40% (Figure 4). Thus, the C3LP activity responsible for PCD was attributed to the β1 subunit of the 20S proteasome.

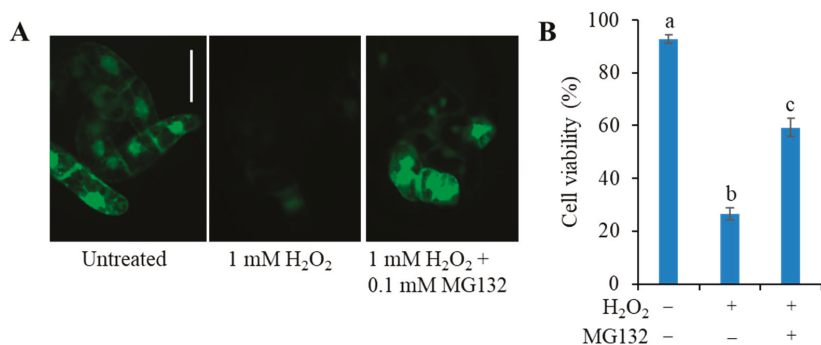


Figure 4. Effect of MG132 on the cell survival in H₂O₂-treated tobacco BY-2 cells. Four-d cultured cells were treated with either 1 mM H₂O₂ or 1 mM H₂O₂ plus 0.1 mM MG132. MG132 was added 30 min before addition of H₂O₂. After 5-h incubation, cell death was determined with FDA staining, as in Figure 1. (A) Typical photographs are shown: untreated cells as control (left), treated with H₂O₂ at 1 mM (middle), and H₂O₂ at 0.2 mM plus 0.1 mM MG132 (right). Bar = 50 μm. (B) Cell viability (%). All values are mean ± SE, and the data represent three independent experiments. Different letters represent significantly different values (*p* < 0.05 on Tukey test).

3.5. Limited Types of Oxylipin Carbonyls Are Increased in Very Early Stages of PCD

We presume that oxylipin carbonyls generated in the H₂O₂-stimulated cells activate C3LP and C1LP and thereby initiate PCD [20]. Because C3LP is activated rapidly on H₂O₂ addition (Figure 2), the oxylipin carbonyls to activate it should also be rapidly increased.

BY-2 cells were treated with the lethal (1 mM) and sublethal (0.5 mM) dose of H₂O₂, and after 10-min incubation, oxylipin carbonyls were extracted and quantified on HPLC (Figure S2). We found that the lethal dose of H₂O₂ increased several types of RCS and other carbonyls as compared with their basal levels in the untreated samples (Figure 5). Among the RCS, HNE increased significantly. HHE also showed an increasing trend although the difference was insignificant. The non-RCS carbonyls acetaldehyde, butyraldehyde, (Z)-3-hexenal, and *n*-hexanal were also increased significantly (Figure 5). In the cells treated with the sublethal dose of H₂O₂ (0.5 mM), HNE, HHE, acetaldehyde, and *n*-hexanal were not changed, but butyraldehyde and (Z)-3-hexenal were increased twofold compared with the untreated cells (Figure 5). When the lethal and sublethal conditions are compared, only limited types of oxylipin carbonyls, i.e., HNE (a RCS), acetaldehyde, and *n*-hexanal (non-RCS carbonyls), were increased. It is expected that these carbonyls are relevant to the C3LP activation and initiation of PCD.

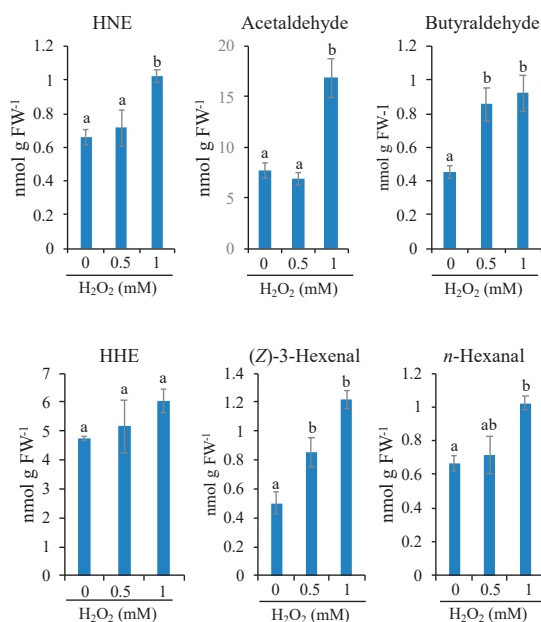


Figure 5. Effects of H₂O₂ on the carbonyl levels in BY-2 cells. To 4-d-cultured cells, H₂O₂ was added to the indicated concentration. After a 10-min incubation, cells were harvested, and the intracellular contents of carbonyl levels were determined as in the Materials and Methods section. Results of 4-hydroxy-(*E*)-2-nonenal (HNE), acetaldehyde, butyraldehyde, 4-hydroxy-(*E*)-hexenal (HHE), 3-(*Z*)-hexenal and *n*-hexanal are shown. Each point represents the mean of three independent experiments and the error bars the SEM. Different letters represent significantly different values ($p < 0.05$ on Tukey test).

HNE and *n*-hexanal have been known to induce PCD in tobacco BY-2 cells [12], and HNE can directly activate C3LP [20]. Here, we tested the ability of acetaldehyde for the activation of C3LP (Figure S3). Four-d cultured cells were treated with various concentrations of acetaldehyde, from 0.5 mM to 10 mM, and then measured for C3LP activity. Acetaldehyde up to 3 mM did not increase C3LP, but at 10 mM, it significantly increased the C3LP activity compared with that in the untreated

cells. Thus, the three oxylipin carbonyls that were increased rapidly on the addition of lethal dose of H_2O_2 were capable of activating C3LP.

3.6. Effects of H_2O_2 on the Intracellular Glutathione and Ascorbate Levels

A possible cause of the RCS increases is the depletion of glutathione (GSH) in the cells because the thiol moiety of GSH provides a major defense against the various electrophiles, including RCS [35,36]. Consumption of ascorbate by H_2O_2 also may facilitate the loss of GSH pool and can be an indirect cause of RCS increase. To measure early changes in the GSH and ascorbate status, tobacco BY-2 cells were treated with a lethal and sublethal doses of H_2O_2 and sampled at different time points (Figure 6). Lethal level of H_2O_2 (1 mM) decreased total glutathione pool (GSH + GSSG) by 40% in 30 min. In 60-min incubation, this lowered level of GSH recovered by about 10% (Figure 6A). The GSH reduction ratio was not dropped by H_2O_2 at 1 mM in 30 min (Figure 6B). The sublethal level of H_2O_2 (0.5 mM) lowered the total glutathione pool by about 35% in 10 min, but afterwards, it recovered (Figure 6A). In contrast, 0.2 mM acrolein, a lethal level, consumed 90% glutathione in 10 min, as observed previously [20]. The lethal level of H_2O_2 , although, induced 68% cell death, but only 10% total GSH pool was decreased in 10 min, and in 60-min incubation, the total GSH pool was not decreased, rather showing an increasing trend. The result suggests that glutathione pool size and GSH reduction ratio are not directly correlated to the initiation of PCD in H_2O_2 -stimulated cells.

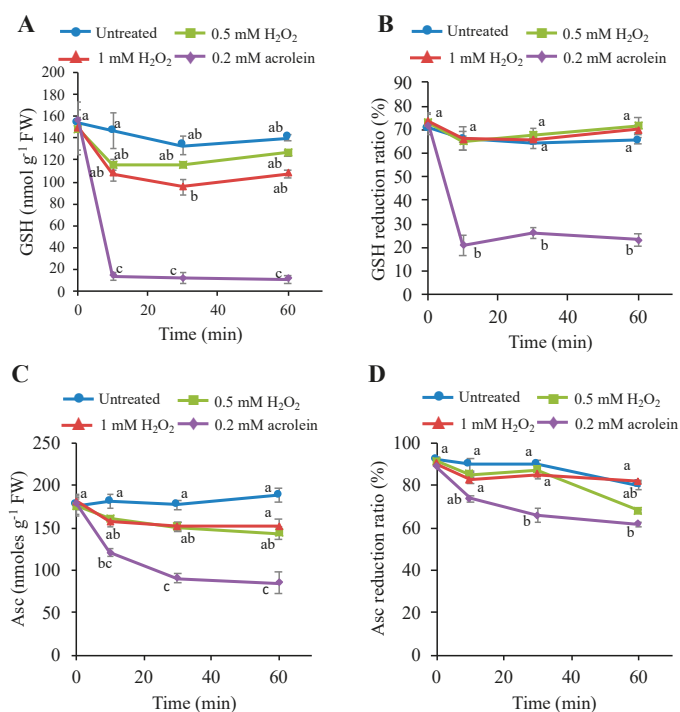


Figure 6. Changes in the contents and reduction ratios of ascorbate (Asc) and glutathione (GSH) in tobacco BY-2 cells treated with H_2O_2 or acrolein. H_2O_2 and acrolein at the indicated concentration was added to 4-d cultured cells, and then Asc and GSH contents were measured as described in the Materials and Methods section. (A) Reduced Asc, (B) Asc reduction ratio (%), (C) contents of reduced GSH, and (D) GSH reduction ratio (%). Each point represents the mean of three independent experiments and the error bars of the SEM. Different letters represent significant different data ($p < 0.05$ on Tukey test).

The changes of Asc content and Asc reduction ratio (Asc/Asc + dehydroascorbate (DHA)) was also measured after treatment with H₂O₂ and acrolein. The lethal level of H₂O₂ (1 mM) decreased an insignificant level of Asc ca. 12% in 10 min, and the sublethal level of H₂O₂ (0.5 mM) did not show any adverse effects on Asc level (Figure 6C). The Asc reduction ratio was also slightly decreased and recovered afterwards (Figure 6D). In contrast, the lethal level of acrolein, i.e., 0.2 mM consumed 70% Asc content in 10 min, as observed previously [20]. These results suggest that the lethal dose of H₂O₂ induced two-third percentage cells death without affecting cellular redox status. Thus, the mechanism of carbonyls accumulation is critical for cell viability.

3.7. H₂O₂ Stimulus Regulates a Carbonyl-Scavenging Capacity in Tobacco BY-2 Cells

In BY-2 cells stimulated with H₂O₂, only limited types of carbonyls were increased early (Figure 5). This was rather unexpected from our previous observation that, in oxidative stressed plant cells, many types of oxylipin carbonyl, including various RCS, are produced [12,31]. This contradiction lead us to conceive the presence of a regulation mechanism of carbonyl increase. We conceived a possibility that H₂O₂ attenuated certain types of carbonyl-scavenging enzyme and eventually increased the levels of limited types of carbonyls, especially HNE. Plant cells have NAD(P)H-dependent HNE-scavenging reductases, including 2-alkenal reductase (AER) [37,38] and aldo-keto reductase [13,14,39]. We treated BY-2 cells with lethal and sublethal doses of H₂O₂, and then extracted proteins and determined the NADPH-dependent HNE reductase activity (Figure 7A). It was found that the H₂O₂ treatment at a lethal dose (1 mM) decreased the reductase activity by 35% in 10 min and by 60% in 60 min. Sublethal H₂O₂ (0.5 mM) treatment caused a slight but insignificant decrease in the reductase activity. Thus, both the accumulation of HNE and the suppression of the HNE-reducing enzyme activity were observed only in the cells undergoing PCD. This result suggested that the inactivation of HNE-reducing enzyme by the lethal level of H₂O₂ was a cause of the enhancement of the intracellular HNE level, which eventually leads to PCD.

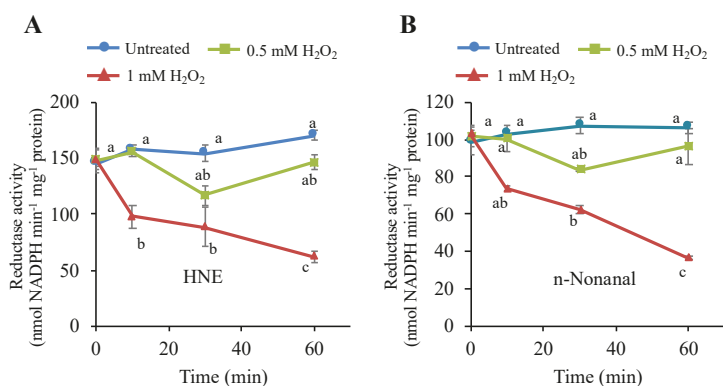


Figure 7. Effects of H₂O₂ on the NADPH-dependent HNE-reducing and *n*-nonanal-reducing activities in tobacco BY-2 cells. Four-d cultured cells were treated with H₂O₂ at 0.5 mM and 1 mM, as in Figure 1. Then, cells were harvested at the indicated time point, and proteins were extracted as in the Materials and Methods section. The reductase activities for (A) HNE and (B) *n*-nonanal were determined as in the Materials and Methods section. Each point represents the mean of three independent experiments and the error bars of the SEM. Different letters represent significantly different values ($p < 0.05$ on Tukey test).

To characterize the inactivated reductase, we also measured the NADPH-dependent activity for the saturated aldehyde *n*-nonanal. The *n*-nonanal-reducing activity responded to H₂O₂ treatment similar to the HNE-reducing activity; it was quickly decreased in the lethal level of H₂O₂, and in the

sublethal level, it was decreased slightly and recovered afterward (Figure 7B). These results suggest that the H_2O_2 -sensitive reductase had substrate specificity for both HNE and *n*-nonanal, i.e., it was a carbonyl reductase. Alternatively, there might be two different reductases for HNE and *n*-nonanal, and both showed similar sensitivity to H_2O_2 .

3.8. H_2O_2 Directly Inactivates the Carbonyl Reductase

The addition of H_2O_2 to living cells will cause a broad range of biochemical changes, and the inactivation of the carbonyl reductase described above (Figure 7) could be a consequence of sequential reactions involving multiple proteins as observed in signal transduction. To verify the role of H_2O_2 in the inactivation of the carbonyl reductase, we prepared protein extract from untreated BY-2 cells and examined the effects of H_2O_2 addition on the NADPH-dependent HNE-reducing activity in it (Figure 8). In this experiment, we added 0.5 mM hydroxylamine to eliminate the catalase activity so that the added H_2O_2 was not immediately scavenged. It was found that H_2O_2 caused a rapid decrease in the HNE-reducing activity. In 5 min at 0.2 mM H_2O_2 , the carbonyl reductase activity was lost by about half. Because the inactivation of HNE-reducing activity was observed in the cell-free protein extract, where intricate interactions among cellular components are very unlikely, we concluded that H_2O_2 directly acted on the carbonyl reductase and inactivated it.

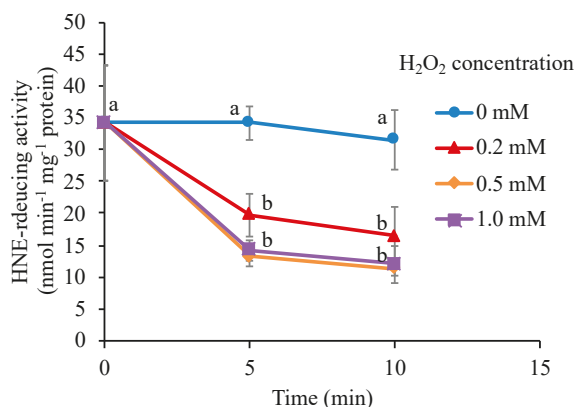


Figure 8. Effect of H_2O_2 on the NADPH-dependent HNE-reducing activity in protein extract from BY-2 cells. Proteins were extracted from untreated cells. H_2O_2 was added at the indicated concentration to the protein extract in the presence of 0.5 mM NH_2OH . The HNE-reducing activity was determined at the indicated time after addition of H_2O_2 , as in the Materials and Methods section. Different letters among the treatments represent significantly different values ($p < 0.05$ on Tukey test).

Because H_2O_2 treatment of the cells caused inactivation of both HNE-reducing and *n*-nonanal-reducing activity, as shown in Figure 7, the H_2O_2 -sensitive carbonyl reductase may show broad substrate specificity for various aldehydes. In the protein extract, we determined carbonyl reductase activity for five different aldehydes, i.e., HNE, *n*-hexanal, acetaldehyde, butyraldehyde, and (Z)-3-hexenal, and examined the effect of H_2O_2 treatment on distinct activities (Figure 9). After H_2O_2 treatment, the carbonyl reductase activity for all these substrates became significantly lower than those in the untreated control. These results suggest that the H_2O_2 -sensitive NADPH-dependent carbonyl reductase possesses a broad substrate specificity.

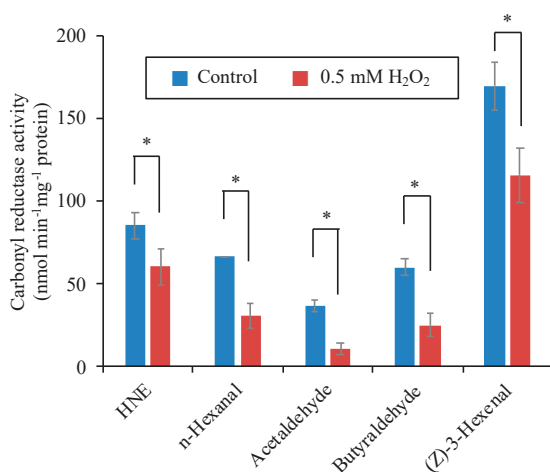


Figure 9. Effects of H₂O₂ on the NADPH-dependent carbonyl reductase activity for different substrates. Proteins were extracted from the untreated tobacco BY-2 cells. H₂O₂ at 0.5 mM was added to the cell extracts, and after 10-min incubation, the enzyme activity for the indicated substrate were determined as described in the Materials and Methods section. Each point represents the mean of three independent experiments and the error bars of the SEM. Asterisks indicate a statistical difference ($p < 0.05$, Student's *t*-test).

4. Discussion

4.1. H₂O₂-Induced Inactivation of Carbonyl-Detoxifying Enzyme(s) Enhances the HNE Level and Triggers PCD

We recently reported that, in response to oxidative stimulus, plant cells generated RCS and that the generated RCS triggered PCD via activation of caspase-like proteases [20]. In this study, we investigated the biochemical mechanism of RCS accumulation at the early stage of oxidative stimulus. We found a larger fraction (70%) of tobacco BY-2 cells were dying in 5 h after the treatment with 1 mM H₂O₂. In exploring the early events caused by the lethal dose of H₂O₂, we found a 2.5-fold increase of C3LP activity even in 10-min incubation, while the increase in the C1LP activity was insignificant. Suppression of C3LP activity by the specific C3LP inhibitor protected cell death, but the C1LP inhibitor did not. We found a 10-min exposure of the cells to a lethal level of H₂O₂ increased only limited species of carbonyls, including highly reactive HNE and HHE. Importantly, the addition of H₂O₂ at lethal level decreased a carbonyl reductase activity in 10-min incubation. These findings suggest that the inactivation of the carbonyl reductase by H₂O₂ is the very initial step of the ROS-mediated PCD in BY-2 cells.

As compared with the condition in living cells, where the HNE reductase was not inactivated by 0.5 mM H₂O₂ (Figure 7), in a cell-free solution, it was rather labile; H₂O₂ at 0.2 mM was high enough to affect the reductase activity (Figure 8). This difference is accounted for by the fact that living cells are equipped with the potent H₂O₂-scavenging system, including catalase and peroxidases. The addition of 0.5 mM H₂O₂ was not harmful to BY-2 cells because catalase and peroxidases with ascorbate scavenged H₂O₂ and thereby protected the carbonyl reductase. An excessive amount of H₂O₂ (1 mM<) could escape from the scavenging system, and the intracellular level of H₂O₂ would go over the threshold, to inactivate the carbonyl reductase. The inactivation profile of the carbonyl reductase in Figure 8, obtained in the presence of a strong inhibitor of heme-type enzymes, likely represented the intrinsic H₂O₂ sensitivity of the enzyme. Attenuation of the carbonyl reductase indirectly activates C3LP for the initiation of PCD. Therefore, the carbonyl reductase acts as a sensor of H₂O₂.

4.2. Consumption of Antioxidants Is Not a Major Cause of PCD

In our previous work, when PCD in BY-2 cells was started by the addition of acrolein, we observed a rapid drop of glutathione pool and significant decrease in the ascorbate pool, along with the activation of caspase-like proteases [20]. This posed a possibility that the loss of these antioxidants by RCS could also be a critical factor in causing PCD. In this work, we analyzed changes in the cellular redox status in response to H₂O₂ at different levels. We found that the effects of H₂O₂ of the lethal and sublethal doses were not different on the pool size and reduction level of both glutathione and ascorbate (Figure 6). Addition of H₂O₂ lowered the glutathione pool size slightly, but it was recovered in 60 min. Importantly, even though glutathione level was restored in 1 mM H₂O₂, PCD was not stopped (Figure 1). These results demonstrate that the main cause of PCD in H₂O₂-stimulated cells was not the redox status changes and supports the central role of RCS as an executor of PCD.

4.3. Candidates of the H₂O₂-Sensitive Carbonyl Reductase

The protein of the H₂O₂-sensing carbonyl reductase in BY-2 cells has not been identified, but several characters of it have been revealed in this study: (i) Its activity depends on NADPH. (ii) It recognizes both HNE and *n*-nonanal as a substrate, i.e., it reduces the carbonyl moiety. The enzyme that has these properties is aldo/keto reductase (AKR). In plants and animals, and in the microorganism, 16 families and above 190 members of reductases present in the AKR superfamily to catabolize RCS. The identified AKR in higher plants classified in four families depending on their functions; detoxify reactive carbonyls, osmoprotectants, secondary metabolism, and membrane transport. In addition to the four families, AKR have three subfamilies; 4A, 4B, and 4C [14,40,41]. There are only limited number of isozymes thus far characterized of the substrate specificity [40,41] and, to our knowledge, there are no reports of their sensitivity to H₂O₂.

In cell-free extract broad range of substrates from RCS, such as HNE, to less reactive carbonyls, such as *n*-hexanal, acetaldehyde, and butyraldehyde, (*Z*)-3-Hexenal showed H₂O₂ at 0.5 mM significantly reduced aldehyde reducing activity (Figure 7). If all these activities are ascribed to a single species of protein, the carbonyl reductase has a broad substrate specificity. Alternatively, it is possible that several different isozymes of carbonyl reductase participated, and they all were sensitive to H₂O₂. Purification of the carbonyl reductase will clarify this question.

4.4. Mechanism of the Inactivation of the Carbonyl Reductase by H₂O₂

ROS can oxidize various amino acid residues on a protein. Cys residue is especially sensitive to the oxidation and indeed such oxidative modification generally changes the normal activity or function of the protein [42]. For example, in response to H₂O₂, Cys199 moiety of OxyR, a six cysteine residues protein, oxidized to sulfenic acid and destabilized its activity. Cys199 modification accelerates to form a disulfide bond with Cys208, which enhances conformational changes of global OxyR and becomes susceptible to oxidation. A site-directed mutant C199S OxyR lost oxidative response and did not activate in the presence of H₂O₂ [43]. Peroxiredoxin (Prx) are ubiquitous enzymes appear in all types of organisms and metabolize peroxide to maintain cellular redox balance. The enzymatic activity of Prxs largely depends on the cysteine residues. The Prx family consists of three cysteine subfamilies, viz. 2-Cys, atypical 2-Cys, and 1-Cys. In human cell, 90% of the H₂O₂ generated in mitochondria reacts with Prx [44]. Thus, it is expected that the thiol moiety of the carbonyl reductase to be sensitive to oxidative signal. H₂O₂ sensors in the AKR family in plants have not been characterized yet. To elucidate the inactivation mechanism, we are trying to purify the H₂O₂-sensing carbonyl reductase.

5. Conclusions

In response to oxidative stimulus by H₂O₂, plant cells significantly increased C3LP activity and accumulate limited species of carbonyls, including RCS. The addition of H₂O₂ did not notably disturb redox balance of the cells but inactivated NADPH-dependent carbonyl reductase(s) in the cells and

cell-free extract. Eventually, the carbonyls load increased and cells underwent PCD. Our results show a new mechanism of ROS sensing in plants: ROS is sensed by the carbonyl reductase(s), and its oxidative inactivation eventually increases carbonyls load and initiates the program of cell death.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2076-3921/9/2/141/s1>: Figure S1. Acrolein-induced cell death was suppressed by the C3LP inhibitor. Four-d cultured BY-2 cells were treated with 0.2 mM acrolein or 0.2 mM acrolein + 20 μ M of each inhibitor (C1LP or C3LP) and harvested after 5 h. (A) Typical fluorescence photographs (top row) and phase contrast microscopy of the same images (bottom): (i) untreated cells as control, (ii) 0.2 mM acrolein, (iii) 0.2 mM acrolein + 20 μ M C3LP inhibitor, (iv) 0.2 mM acrolein + 20 μ M C1LP inhibitor (top). White arrows indicate dead cells. (B) FDA-staining cells were counted for living cells and expressed as percentage. The data are the mean \pm SE. Different letters represent significant different data ($p < 0.05$ on Tukey test). Figure S2: Typical chromatograms of DNP-carbonyls in BY-2 cells. The cells were treated with water as control (A) and H₂O₂ (B). Four-d-cultured cells treated with H₂O₂ at 1 mM and incubation for 10 min. The identified aldehydes are labelled at the top of each peak. Figure S3. Activation of C3LP in BY-2 cells after exposure to acetaldehyde. Four-d cultured cells were treated with various concentration of acetaldehyde as indicated. Protease inhibitors were added to cell extracts to a final concentration of 0.1 mM and caspase assay was performed as described in Materials and Methods. Statistical differences compared to controls are indicated by asterisks. ($p < 0.05$, Student's *t*-test).

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