Laser-Induced Fluorescence of Molybdenocene and Tungstenocene in Low-Temperature Matrices

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The reactive metallocenes, tungstenocene and molybdenocene $(M(\eta^5-C_5H_5)_2; M = Mo, W)$, have been generated by photolysis of the corresponding dihydride complexes, $M(\eta^5-C_5H_5)_2H_2$, in argon and nitrogen matrices at 12 K. The metallocenes have been probed by laser-induced fluorescence with a pulsed tunable laser and by UV/vis absorption spectroscopy. Structured emission is observed from the LMCT excited states (lifetimes < 10 ns). The spectra are complicated by multiple sites/conformers, but emission spectra of a single site/ conformer may be obtained with appropriate selection of matrix and excitation wavelength. Corresponding excitation spectra are measured from the area of selected emission peaks as a function of excitation wavelength. Vibrational progressions are dominated by the ring-metal-ring symmetric stretching mode ($v_4 \approx 300 \text{ cm}^{-1}$). Nevertheless, this mode changes in frequency by $\leq 4 \text{ cm}^{-1}$ ($\leq 1.3\%$) in the LMCT excited state. The bestresolved peaks have a full width at half-maximum of ca. 10 cm⁻¹. Most of the emission is vibrationally fully relaxed, but weak emission peaks arising from v' = 1 states are found for MoCp₂ in N₂ matrices.

Introduction

In spite of the immense number of investigations of the photochemistry of organometallics, spectroscopic information on excited states is very restricted.² The power of laser-induced fluorescence (LIF) as a probe to determine excited-state and ground-state properties of matrix-isolated open-shell metallocenes was demonstrated in the preceding paper through the example of rhenocene.³ We now use the same techniques to examine the d⁴ metallocenes, molybdenocene and tungstenocene (MCp₂; M = Mo, W; Cp = η^5 -C₅H₅). Both WCp₂ and MoCp₂ have been implicated in numerous photochemical reactions in solution.⁴ Here, these two molecules are generated by photolysis of the corresponding dihydride complexes, MCp₂H₂, in argon and nitrogen matrices. They have already been the subject of extensive study by matrix isolation with detection by IR and UV/vis absorption spectroscopy and by magnetic circular dichroism (MCD).⁴⁻⁷ They have a ground electron configuration $e_2{}^3a_1{}^1$, leading to a 3E_2 ground electronic state. In the case of WCp₂, spin-orbit coupling quenches any Jahn-Teller splitting, leading to a well-defined E₂ ($\Omega = 3$) ground spinorbit state. An electronic IR transition to a higher spin-orbit state is prominent. For MoCp₂, the Jahn-Teller effect and spin-orbit coupling are of comparable magnitude, resulting in broader IR bands than for WCp₂ and no IR electronic transition. Both molecules exhibit an intense LMCT absorption band with onset at ca. 400 and 420 nm for WCp₂ and MoCp₂, respectively. The upper state has been demonstrated to be of E_2 symmetry by MCD. The background to the vibrational spectroscopy of metallocenes has been summarized in the preceding paper. We now report vibrationally resolved emission spectra induced by irradiation with a tunable pulsed laser. We deduce excitation spectra from the variation in the area of individual peaks of the dispersed emission spectra as a function of excitation wavelength. Some of the results presented here have been described in preliminary communications.8,9

Experimental Section

The experimental setup for LIF is described in detail in the preceding paper.³ $MoCp_2H_2$ and WCp_2H_2 were synthesized by

standard methods.¹⁰ MoCp₂H₂ was cocondensed with matrix gas for ca. 130 min with the sample held at 310-313 K and the window at 20 K. The matrix was then cooled to 12 K before photolyzing for 60 min ($\lambda > 200$ nm). For WCp₂H₂, the sublimation temperature was 318 K and photolysis time 30 min. In each experiment, it was verified that emission intensity varied linearly with laser energy.

UV/vis and IR absorption spectra of the metallocenes were determined with a matrix apparatus which has been described elsewhere.¹¹ UV/vis spectra were recorded on a Perkin-Elmer Lambda 7G spectrometer (resolution 0.25 nm) and IR spectra on a Mattson/Unicam RS FTIR spectrometer (resolution 1 cm^{-1}).

Results

1. WCp₂ in Nitrogen Matrices. The formation of WCp₂ by photolysis of WCp₂H₂ in a nitrogen matrix is illustrated by the difference IR spectrum in Figure 1. The intense band at 3238 cm⁻¹ is assigned as an electronic transition between spinorbit substates, first seen in argon matrices.^{5,6} The figure also shows the growth of the ring-breathing (1085 cm^{-1}) and C-H deformation (981, 971 cm⁻¹ split) modes of WCp₂ and the loss of W-H modes of the precursor at ca. 1900 $\text{cm}^{-1.12}$ The UV/ vis absorption spectrum of WCp2 shows the structured LMCT transition between 370 and 405 nm. The absorption spectrum before annealing shows three clear progressions in the ringmetal-ring symmetric stretching mode, ν_4' , of ca. 319 cm⁻¹, arising from WCp₂ in three distinct sites/conformers labeled α , β , and γ (Figure 2a; Table I, supplementary material). However, the site structure is simplified into two dominant components, split by ca. 90 cm⁻¹, when the matrix is warmed to 30 K and subsequently recooled (Figure 2b). One component comes close to wavelengths of β site absorption, whereas the other site is quite new (δ). The separation between successive members of each progression, v_4' , is about 314 cm⁻¹ (Table I, supplementary material).

The structure in this spectrum and others in this paper is labeled as follows: upper case letters refer to a progression series, "A" is the series nv_4 (n = 0, 1, 2, ...) and "B" is the combination series due to $v_3 + nv_4$; Greek superscripts symbolize a particular site/conformer; the numerals correspond to a particular overtone of the progression. For example, $A^{\alpha 3}$

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Figure 1. WCp₂H₂ in a nitrogen matrix at 12 K, after 30-min photolysis $\lambda > 200$ nm. The IR difference spectrum shows the formation of WCp₂ (positive absorbances) and depletion of the WCp₂H₂ precursor (negative absorbances).



Figure 2. UV/vis absorption spectra of (a) WCp₂ obtained after deposition of WCp₂H₂ with nitrogen followed by 1-h photolysis, $\lambda > 200$ nm, and (b) WCp₂ after annealing to 30 K and recooling to 12 K. The labels A^{β} and A^{δ} refer to the annealed spectrum and indicate progressions in ν_4 '.

symbolizes the third overtone ($v = 3 \leftarrow 0$) in the v_4 progression of site/conformer α . As for rhenocene, only three vibrations are involved in the vibrational progressions. They are the totally symmetric modes, v_4 , v_3 , and v_2 , corresponding to ring-metalring stretching, C-H deformation and ring-breathing, respectively.¹²

Irradiation into the LMCT band of an unannealed matrix excites an intense emission spectrum (Table II, supplementary material) with a progression in ν_4'' for the three sites/conformers, α , β , and γ .⁹ The (0,0) bands in *emission* for these sites/ conformers are shifted between 25 and 39 cm⁻¹ to lower energy than the corresponding features in the unannealed *absorption* spectrum. The relative intensity of α , β , and γ is dependent on the excitation wavelength but no component may be excited exclusively. The mean value of ν_4'' from all components is 318 cm⁻¹, very close to the value of ν_4'' obtained from the absorption spectrum (see Table II, supplementary material).

The emission spectra show dramatic changes when the matrix is annealed. Irradiation at 392.1 nm, which corresponds to the $A^{\beta}2$ absorption band, induces fluorescence from the A^{β} site only (Figure 3a). A combination progression B ($\nu_3'' + n\nu_4''$) is easily identified, yet there is no clear peak that can be assigned to



Figure 3. (a) Emission spectrum of WCp₂ in a nitrogen matrix at 12 K after annealing, $\lambda_{ex} = 392.1$ nm. (b) Emission spectrum of WCp₂ in an argon matrix at 12 K, $\lambda_{em} = 394.3$ nm.

 ν_2'' . No such well-resolved emission spectrum could be obtained prior to annealing with any excitation wavelength. The spectrum is comparable to that of ReCp₂, although the bands are much broader (fwhm = 73 cm⁻¹; cf. ca. 20 cm⁻¹ for ReCp₂). Figure 3b shows the corresponding spectrum in an argon matrix (see below).

The changes with excitation wavelength that occur in the emission spectra of annealed samples are more subtle than encountered in the ReCp₂ system. Emission from the β -series is detected even if γ - or δ -series absorption bands are irradiated. Emission from the γ -series is also observed on irradiation into the γ -series absorption bands, but emission from the δ -series is very weak, even when the δ -series absorption bands are irradiated directly. These results are suggestive either of energy transfer between sites/conformers or of interconversion between sites.

It is harder to record a reliable excitation spectrum for WCp₂ than for ReCp₂ because the laser dyes needed to excite emission in the range 350-400 nm work less efficiently than in the range 450-500 nm. The excitation profile of the $A^{\beta}1$ emission band reproduces the positions of the absorption bands of the β -site well. Despite the selectivity for the β site, the excitation spectrum offers no significant improvement in resolution compared to the annealed absorption spectrum.

The decay kinetics of the excited state show that the emission decays within the duration of the laser pulse. This result allows only an upper limit to be set on the lifetime, τ , of 10 ns.

2. WCp₂ in Argon Matrices. The LMCT absorption band of WCp₂ in an argon matrix at 12 K is slightly blue-shifted in energy compared to that in nitrogen. The vibrational band spacing is ca. 321 cm^{-1} , similar to that determined in nitrogen matrices, but the fine structure $(37-75 \text{ cm}^{-1})$ is much less marked and shows no substantial changes when the matrix is annealed.⁵ Irradiation into the LMCT absorption band leads to intense emission (Figure 3b, Table 3). The emission spectrum consists of a sharp and a broad component, separated by 120 cm⁻¹. The fundamentals ν_4 ", ν_3 ", and ν_2 " are measured from the A, B, and C progressions, respectively (Figure 3b).



Figure 4. Spectra of WCp_2 isolated in an argon matrix at 12 K. (a) Absorption spectrum; (b) excitation spectrum determined from variation of the area of the A2 emission band with excitation wavelength.

TABLE 3: Emission Spectrum for WCp₂ in an Argon Matrix, $\lambda_{ex} = 394.3$ nm (See Figure 3b)

	$\Delta \bar{\nu}, \mathrm{cm}^{-1}$			
$\overline{\nu}$, cm ⁻¹	(from origin)	label	assgt	interval, cm ⁻¹
25 053	0	A0	T ₀	0
24 734	319	A1	$\nu_4^{\prime\prime}$	A0 - 319
24 412	641	A2	$2\nu_4$	A1 - 322
24 262	791	B 0	v3"	T ₀ – 791
24 094	959	A3	$3v_4^{\prime\prime}$	A2 - 318
23 965	1088	C0	$\nu_2^{\prime\prime}$	$T_0 - 1088$
23 942	1111	B1	$\nu_{3}'' + \nu_{4}''$	B0 - 320
23 774	1279	A4	$4\nu_4^{\prime\prime}$	A3 - 320
23 643	1410	C1	$\nu_2'' + \nu_4''$	C0 - 322
23 624	1429	B2	$\nu_3'' + 2\nu_4''$	B 1 - 318
23 473	1580	D0	$2\nu_3^{\prime\prime}$	B0 - 789
23 455	1598	A5	$5\nu_4^{\prime\prime}$	A4 - 319
23 323	1730	C2	$\nu_2'' + 2\nu_4''$	C1 - 320
23 303	1750	B3	$\nu_3'' + 3\nu_4''$	B2 - 321
23 175	1878	E0	?	T ₀ - 1878
23 156	1897	D1	$2\nu_3'' + \nu_4''$	D0 - 317
23 004	2049	C3	$\nu_2'' + 3\nu_4''$	C2 - 319
22 854	2199	E1	?	E0 - 321
22 835	2218	D2	$2\nu_3'' + 2\nu_4''$	D1 - 321
22 686	2367	C4	$\nu_2'' + 4\nu_4''$	C3 - 318
	$\bar{\nu}$, cm ⁻¹ 25 053 24 734 24 412 24 262 24 094 23 965 23 942 23 774 23 624 23 425 23 303 23 175 23 156 23 004 22 854 22 855 22 686	$\begin{array}{c c} & \Delta \bar{\nu}, {\rm cm}^{-1} \\ \hline \nu, {\rm cm}^{-1} & ({\rm from \ origin}) \\ \hline 25 \ 053 & 0 \\ 24 \ 734 & 319 \\ 24 \ 412 & 641 \\ 24 \ 262 & 791 \\ 24 \ 094 & 959 \\ 23 \ 965 & 1088 \\ 23 \ 942 & 1111 \\ 23 \ 774 & 1279 \\ 23 \ 643 & 1410 \\ 23 \ 624 & 1429 \\ 23 \ 473 & 1580 \\ 23 \ 455 & 1598 \\ 23 \ 323 & 1730 \\ 23 \ 303 & 1750 \\ 23 \ 156 & 1897 \\ 23 \ 304 & 2049 \\ 22 \ 854 & 2199 \\ 22 \ 854 & 2199 \\ 22 \ 855 & 2218 \\ 22 \ 686 & 2367 \\ \hline \end{array}$	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

An excitation profile of the A2 band area is shown as Figure 4b (Table 4). Compared to the absorption spectrum (Figure 4a), the excitation profile is much more highly resolved with a clear progression in v_4' . The (0,0) band positions in excitation and emission are coincident. However, the intensity profile of the excitation maxima does not resemble the intensity profile of the absorption and emission spectra: there is no clear v_3' band and the emission intensity decreases monotonically with vibrational quantum number. The A3 emission intensity is very low relative to A2, whereas the A3 absorption intensity is still high.

Expansions of the A1 transition show that the excitation band is narrower (fwhm $\approx 10 \text{ cm}^{-1}$, Figure 5a) than the equivalent band in the emission spectrum (fwhm = 17 cm⁻¹, Figure 5b). The broad component of the emission spectrum is conspicuous in emission but is not seen in excitation, as it has been effectively subtracted out. Both excitation and emission bands are more symmetric than those seen in for ReCp₂.

TABLE 4:Excitation Spectrum of the A2 Emission Bandof WCp2 in Argon Matrices (See Figure 4b)

λ, nm	$\bar{\nu}$, cm ⁻¹	$\Delta \bar{\nu}, \mathrm{cm}^{-1}$ (from origin)	label	assgt	interval, cm ⁻¹
399.15 394.20 389.35	25 053 25 368 25 684	0 315 631	A0 A1 A2	$\begin{array}{c} T_0\\ \nu_4 \\ 2\nu_4 \end{array}$	$0 \\ A0 + 315 \\ A1 + 316$
384.8 384.2	25 988 26 028	935 975	A3 ?	3v4' ?	A2 + 305 ?



Figure 5. (a) Expanded excitation spectrum of WCp₂ in an argon matrix showing the A1 $(1 \leftarrow 0)$ band only. (b) Expanded emission spectrum of WCp₂ in an argon matrix at 12 K, showing just the A1 $(0 \rightarrow 1)$ transition ($\lambda_{ex} = 399.15$ nm).



Figure 6. UV/vis *absorption* spectrum for $MoCp_2$ in a nitrogen matrix at 12 K: (a) before annealing, (b) after annealing to 30 K and recooling.

3. MoCp₂ in Nitrogen Matrices. The LMCT transition of MoCp₂ in a nitrogen matrix has its onset ca. 16 nm to longer wavelength than that of WCp₂. The absorption band is also appreciably broader than that of the tungsten analogue (Figure 6; Table V, supplementary material). Prior to annealing, the absorption spectrum contains two progressions, A^{α} and A^{β} , separated by ca. 140 cm⁻¹ (Figure 6a). The interval between successive members of each progression is about 300 cm⁻¹ and



Figure 7. Emission spectrum of MoCp₂ in a nitrogen matrix at 12 K after annealing to 30 K and recooling ($\lambda_{ex} = 402.0$ nm). The inset shows an expansion of the short-wavelength region including unrelaxed emission.



Figure 8. (a) *Emission* spectrum of $MoCp_2$ in an argon matrix at 12 K. (b) UV/vis absorption spectrum of $MoCp_2$. (c) *Excitation* spectrum determined from the variation of the area of the A2 emission band with excitation wavelength.

corresponds to ν_4' . When the matrix is annealed (Figure 6b) two different series, A^{γ} and A^{δ} , are seen. Each series appears to be superimposed upon a much broader, unstructured band.

Irradiation into the LMCT band produces relatively poorly resolved fluorescence spectra regardless of whether the matrix has been annealed (Figure 7; Table VI, supplementary material). The emission spectrum shows the v_4 " progression and the combination series due to v_3 " + nv_4 ". The onset of the main emission is at 419.5 nm, close to the position of the $A^{\delta 1}$ absorption band. Multiple components are identified when the excitation wavelength is changed, but the large proportion of unstructured emission remains.

The weak emission bands appearing at *shorter* wavelengths than the assigned electronic origin $A^{\delta}0$ are enlarged in the inset of Figure 7. The longest wavelength of these bands arises from fully relaxed fluorescence: $A^{\beta}0$ ($\lambda_{em} \approx 416.1$ nm). The other features are observed at shorter wavelengths, suggesting that

TABLE 7: Principal Features of the *Emission* Spectrum of MoCp₂ in an Argon Matrix, $\lambda_{ex} = 416.6$ nm (See Figure 8a)

		$\Delta \bar{\nu}$, cm ⁻¹			
λ, nm	\overline{v} , cm ⁻¹	(from origin)	label	assgt	interval, cm ⁻¹
416.6ª	24 004	0	A0	T ₀	0
422.1	23 691	313	A1	$\nu_4^{\prime\prime}$	$T_0 - 313$
427.6	23 386	618	A2	$2\nu_4^{\prime\prime}$	A1 - 305
430.9	23 207	797	B 0	v3"	T ₀ - 797
433.2	23 084	920	A3	$3\nu_4$	A2 - 302
436.6	22 904	1100	B 1	$\nu_{3}'' + \nu_{4}''$	B0 - 303
438.9	22 784	1220	A4	$4\nu_4^{\prime\prime}$	A3 - 300
442.5	22 599	1405	B2	$\nu_3'' + 2\nu_4''$	B1 - 305
444.7	22 487	1517	A5	$5\nu_4^{\prime\prime}$	A4 - 297
446.0	22 422	1582	D0	$2\nu_{3}''$	B0 - 785
448.5	22 297	1707	B 3	$\nu_{3}'' + 3\nu_{4}''$	B2 - 302
450.7	22 188	1816	A6	$6\nu_4''$	A5 — 299
452.1	22 119	1885	D 1	$2\nu_3'' + \nu_4''$	C0 - 303
454.7	21 993	2011	B4	$\nu_3'' + 4\nu_4''$	B3 — 304
457.1	21 877	2127	A7	$7\nu_4^{\prime\prime}$	A6 - 311
458.4	21 815	2189	D2	$2\nu_3'' + 2\nu_4''$	C1 - 304

^{*a*} Position of the (0,0) is taken from the excitation spectrum.

TABLE 8: Excitation Spectrum of MoCp2 in an ArgonMatrix (See Figure 8c)

λ, nm	$\bar{\nu}$, cm ⁻¹	$\Delta \bar{\nu}, \mathrm{cm}^{-1}$ (from origin)	label	assgt	interval, cm ⁻¹
416.6	24 004	0	A0	T ₀	0
411.4	24 307	303	A1	ν_4	$T_0 + 303$
406.2	24 618	614	A2	$2\nu_4$	A1 + 311
404.4	24 728	724	B0	V3'	$T_0 + 724$
401.2	24 925	921	A3	$3\nu_4$	A2 + 307
399.7	25 019	1015	B1	$\nu_3' + \nu_4'$	B0 + 291

they originate from unrelaxed emission. The proposed transitions are given on Figure 7.

Lifetime measurements of the excited state were attempted, but the duration of the laser pulse again exceeds the emission lifetime, allowing only an upper limit to be set at 10 ns.

4. MoCp₂ in Argon Matrices. The absorption spectrum of MoCp₂ in an argon matrix⁵ is similar in appearance to that in a nitrogen matrix, although the site structure is less distinct. The best-resolved emission spectra are obtained when exciting into the (0,0) transition of the absorption spectrum (Figure 8a, Table 7). (Note that this figure does not show the (0,0) emission band since it is close to the laser line.) Progression series, nv_4 " (A), v_3 " + nv_4 " (B), and $2v_3$ " + nv_4 " (C) are observed. Site structure is apparent in all the spectra recorded, and different features appear to be probed selectively according to the laser wavelength. More congested spectra are obtained when shorter wavelength excitation is used. The spectra all show a substantial unstructured, background fluorescence. Shorter wavelength than the (0,0) band, akin to that seen in a nitrogen matrix.

An excitation spectrum may be obtained by recording the area of the A2 emission band as a function of laser wavelength and energy (Figure 8c, Table 8). The excitation profile shows a dramatic improvement in resolution when compared to the absorption spectrum (Figure 8b), allowing values for the excited-state vibrations v_4' and v_3' to be obtained. Notice that the excitation spectrum gives the impression that short-wavelength excitation does not induce emission. In fact, substantial fluorescence is observed, but it is poorly resolved and the area of the A2 band cannot be measured reliably.

The lifetime of the excited state proved shorter than the duration of the laser pulse (i.e., <10 ns).

Discussion

The d^4 metallocenes, WCp₂ and MoCp₂, exhibit intense structured emission from their LMCT excited states. The LIF

TABLE 9: Progression Frequencies $(\bar{\nu}, \text{ cm}^{-1})$ of Metallocenes in the Ground and Excited States with the Estimated Standard Error (95% Probability), Where Available, in Parentheses

and and and and and
v_4 v_3 v_3 v_2 v_2
40 (4) 818 745 1097 1060
2 (3) 826 754 1100 1063
6 790 1088
6 (12) 797
3 (24) 789
7 791 724
02 (12) 791
1104
79 772 737 1001 953

^a Prior to annealing. ^b After annealing. ^c From ref 14. ^d From ref 13.

spectra have many features in common with the spectra of ReCp₂. The emission occurs from the same state as that probed by the laser, with (0,0) bands close to coincidence in emission, absorption, and excitation. The vibrational progression in the ring-metal-ring symmetric stretching mode, nv_4 , dominates the spectrum, with weaker progressions from $v_3 + nv_4$ and sometimes, $v_2 + nv_4$ or $2v_3 + nv_4$. The spectra of WCp₂ and MoCp₂ are complicated by the presence of multiple trapping sites and/or conformers. By judicious choice of conditions, it is possible to obtain emission and excitation spectra dominated by a single site/conformer. Some of the excitation spectra are much sharper than the corresponding absorption spectra.

There are also several features of the spectra which differ significantly from those of ReCp₂, some of which are described in more detail below: (i) the emission lifetimes are all too short to measure with the current apparatus (<10 ns, compared with 72 ns for ReCp₂); (ii) the differences between ground- and excited-state values of ν_4 are very small ($\leq 4 \text{ cm}^{-1}$); (iii) the narrowest emission and excitation bands of MoCp₂ and WCp₂ (in argon matrices) are close to symmetrical, unlike the skew shape observed for ReCp₂; (iv) the intensity profiles of the emission and excitation spectra, especially of WCp₂ in Ar, are not mirror images; (v) the spectra of MoCp₂ show evidence for unrelaxed emission.

1. Vibrational Fine Structure. The totally symmetric vibrational frequencies determined from the LIF spectra are listed in Table 9 together with values for other metallocenes. Although the spectra of MoCp₂ and WCp₂ are dominated by progressions in ν_4 , the differences between ground- and excitedstate values of ν_4 are so small ($\leq 4 \text{ cm}^{-1}$) as to be insignificant when errors are taken into account. It proved possible to determine v_3'' in all emission spectra, but the corresponding value of ν_3' was determined only for MoCp₂ in argon matrices. In that case, a reduction of 67 cm^{-1} (8.4%) was found. Only one value of ν_2'' was measured and none of ν_2' . Since ν_4 is the progression-forming mode, there must be a change in the metalring bond length in the excited state. The lack of a corresponding change in v_4 indicates that v_4 does not correlate simply with bond length and that there must also be changes in coupling between vibrations in the excited state.

The symmetric band shape with fwhm of $10-20 \text{ cm}^{-1}$ found in the sharpest spectra of WCp₂ in solid argon implies that there is no coupling either to matrix phonons or torsional overtones as postulated to explain the skew shape of rhenocene bands.

It is valuable to compare the changes in the frequencies between ground and excited states for the range of metallocenes (Table 9). The transition involved in the cases of ReCp₂, MoCp₂, and WCp₂ transfers an electron from a ligand π -orbital to a metal-ligand bonding orbital. For ReCp₂ this results in an increase in ν_4 , but for the d⁴ metallocenes, there is essentially no change. For ReCp₂ there is also a substantial fall in ν_3 and ν_2 . The data for the d⁴ metallocenes are incomplete, only allowing us to state that there is a similar reduction in ν_3 for MoCp₂. The transition involved in emission from ZnCp also has LMCT character,¹³ though this time the electron is transferred to a metal p-orbital. The changes in frequency are remarkably similar to those found for ReCp₂ in spite of the drastic difference in electron configuration and the reduced value of ν_4 . In contrast, the emission from crystalline RuCp₂ arises from a spin-forbidden ligand-field band,¹⁴ so the electron is transferred to an orbital which is strongly metal-ligand antibonding. As a result, the value of ν_4 falls by 54 cm⁻¹ (16%). The relation between the changes in the frequency and the changes in geometry will be developed in a subsequent paper via Franck-Condon analysis.

2. Unrelaxed Emission. The emission spectra of $MoCp_2$ in nitrogen matrices show weak features to high energy of the electronic origin with frequencies which fit with emission from v' = 1 levels. This observation is consistent with a very short emission lifetime and a comparable relaxation time. The broadening of emission spectra when higher vibrational quanta are probed may be another manifestation of unrelaxed emission, since it would give rise to multiple overlapping bands. It may also contribute to the broadening of the high-energy features of the excitation spectra (e.g., Figure 4) so the excitation profile does not mirror the emission profile. For comparison, unrelaxed emission is much more important in the gas phase, as has been observed for ZnCp.¹³

Conclusions

The d⁴ metallocenes MoCp₂ and WCp₂ isolated in N₂ and Ar matrices exhibit well-resolved fluorescence spectra when excited with a pulsed tunable laser. Both absorption and emission involve the same fully allowed LMCT excited state with lifetime <10 ns. The spectra are complicated by the presence of multiple sites/conformers, but emission spectra of a single site can be obtained with suitable choice of matrix and excitation wavelength. Similarly, single-site excitation spectra can be obtained with suitable choice of emission maximum. The spectra are dominated by progression series in the symmetric ring-metal-ring stretching mode, nv_4 , with additional progressions in $v_3 + nv_4$, $v_2 + nv_4$, and $2v_3 + nv_4$. Nevertheless, the changes in v_4 between ground and excited states are $\leq 4 \text{ cm}^{-1}$ ($\leq 1.3\%$), implying that the vibrational frequencies do not correlate well with the geometric coordinate.

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Supplementary Material Available: Tables I, II, V, VI: UV/vis absorption and emission spectra for WCp_2 and $MoCp_2$ in nitrogen matrices (5 pages). Ordering information appears on any current masthead page.

References and Notes

(1) Present address: Department of Chemistry, St. Patrick's College, Maynooth, Ireland.

(2) Photoprocesses in Transition Metal Complexes, Biosystems and other Molecules. Experiment and Theory; Kochanski, E., Ed.; NATO ASI Series C; Kluwer Academic Publishers: Dordrecht, 1992; Vol. 376. Turner, J. J.; Johnson, F. P. A.; Westwell, J. R. Coord. Chem. Rev. 1993, 125, 101.

(3) Hill, J. N.; Perutz, R. N.; Rooney, A. D. Preceding paper in this issue.

(4) Grebenik, P.; Grinter, R.; Perutz, R. N. Chem. Soc. Rev. 1988, 17, 453.

(5) Chetwynd-Talbot, J.; Grebenik, P.; Perutz, R. N. Inorg. Chem. 1982, 21, 3647.

- (6) Cox, P. A.; Grebenik, P.; Perutz, R. N.; Robinson, M. D.; Grinter,
 R.; Stern, D. R. *Inorg. Chem.* 1983, 22, 3614.
 (7) Graham, R. G.; Grinter, R.; Perutz, R. N. J. Am. Chem. Soc. 1988,
- 110, 7036.
- (8) Bell, S. E. J.; Hill, J. N.; McCamley, A.; Perutz, R. N. J. Phys. Chem. 1990, 94, 3876.
- (9) Perutz, R. N.; Hill, J. N.; McCamley, A. Coord. Chem. Rev. 1991, 111, 111.

(10) Green, M. L. H.; Knowles, P. J. J. Chem. Soc., Perkin Trans. 1 1973, 989.

- (11) Haddleton, D. M.; McCamley, A.; Perutz, R. N. J. Am. Chem. Soc. 1988, 110, 1810.
- (12) Aleksanyan, V. T. In Vibrational Spectra and Structure; Durig, J. R., Ed.; Elsevier: Amsterdam, 1982; Vol. 11.
 (13) Robles, E. S. J.; Ellis, A. M.; Miller, T. A. J. Phys. Chem. 1992,
- 96, 3247.

(14) Riesen, H.; Krausz, E.; Luginbühl, W.; Biner, M.; Güdel, H. U.; Ludi, A. J. Chem. Phys. **1992**, 96, 4131. Wrighton, M. S.; Pdungsap, L.; Morse, D. L. J. Phys. Chem. **1975**, 79, 66. Hollingsworth, G. J.; Kim Shin, K.-S.; Zink, J. I. Inorg. Chem. **1990**, 29, 2501.