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Research Article

BISPHENOL A DOSE- AND TIME-DEPENDENTLY INDUCES OXIDATIVE STRESS IN RAT LIVER MITOCHONDRIA EX VIVO

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ABSTRACT

Objective: The probable toxic effects of bisphenol A (BPA) on different physiological functions have been reported in animal models. The role of BPA in mitochondrial oxidative stress has not been reported till date. The present study is aimed to elucidate dose- and time-dependent oxidative stress generation by BPA, respectively, in rat liver mitochondria in *ex vivo* model.

Methods: The incubation mixture of BPA-treated groups containing mitochondria, 50 mM potassium phosphate buffer (pH 7.4), and different concentrations of BPA (20–160 μ M/ml) (dissolved in 12% DMSO) in a final volume of 1.0 ml was incubated at 37°C in incubator for different time durations (30 min–2 h). Whereas, the incubation mixture of control group contained DMSO (12%), mitochondria and 50 mM potassium phosphatebuffer (pH 7.4).' will be replaced by 'Whereas, the incubation mixture of control group contained the same constituents except BPA.

Result: We have observed significant decrease in mitochondrial intactness incubated with BPA in dose- and time-dependent manner under bright field and confocal microscopic study compared to control. Further, we have observed a decrease in mitochondrial reduced glutathione (GSH) content and increase in lipid peroxidation and protein carbonylation levels in dose- and time-dependent manner in BPA-exposed mitochondria. We have found a significant increase in the activity of Mn-superoxide dismutase and decrease in the activities of GSH peroxidase, GSH reductase, pyruvate dehydrogenase, and other three enzymes of Kreb's cycle dose and time dependently in BPA-exposed mitochondria. The results indicate that exposure to BPA leads to decrease in intactness of mitochondria and increase in oxidative stress in mitochondria isolated from rat liver in a dose- and time-dependent manner.

Conclusion: It can be concluded that the incubation of mitochondria isolated from rat liver with BPA, caused oxidative stress-mediated damages in mitochondria in both dose- and time-dependent manners.

Keywords: Bisphenol A, Ex vivo model, Intactness of mitochondria, Kreb's cycle enzymes, Mitochondria, Oxidative stress.

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INTRODUCTION

The lifestyle of the people has been rapidly changing due to scientific advancement and proliferation of consumerism in a global manner. It is not only restricted in urban areas but also in the rural areas. In this situation, majority of people in urban and rural areas are leaning toward the consumption of fast foods including the foods packaged in polycarbonate plastic containers or tin containers coated with epoxy resin at the interior are frequently fortified with bisphenol A (BPA). As a result due to weathering in many cases, BPA, the monomer of polycarbonate plastic and epoxy resin, and polymer of BPA are released to the foods, and when the people are exposed to the foods then the humans are also being exposed to the BPA.

BPA is a diphenylmethane derivative (Fig. 1a) which had been commercialized since 1957. BPA is also being used to form polycarbonate plastics through polymerization to make certain plastics and epoxy resins which are used to make a variety of common consumer goods, such as water bottles, sports equipment, compact discs, and digital versatile discs, line water pipes, as coatings on the inside of many food and beverage cans, and in making thermal paper, for example, sales receipts (Fig. 1b) [1]. In 2015, estimation indicated that 4 million tonnes of BPA chemical were produced for manufacturing polycarbonate plastics. The suitability of BPA in some consumer products and food containers raise a question because it shows estrogen-mimicking properties. The authorization of the use of BPA in baby bottles and infant formula packaging has been ended by the US Food and Drug Administration because its uses are not safe. Many bisphenol analogs exhibit endocrine disrupting effects, cytotoxicity, genotoxicity, reproductive toxicity, dioxin-like effects, and neurotoxicity in laboratory studies [2]. BPA causes autism in genetically susceptible children by inducing oxidative stress and mitochondrial dysfunction [3]. The motor functions of duodenal smooth muscle are also inhibited by BPA through induction of oxidative stress. These situations decrease the availability of free Ca²⁺, cause augmentation of the activity of acetylcholinesterase and promotion of the synthesis of nitric oxide (NO) in duodenal visceral smooth muscle [4,5]. Previous studies showed that *in utero* BPA exposure inhibits germ cell nest breakdown in the F1 generation of mice [6]. Earlier studies showed that BPA increased expression of HO-1 gene more than Gadd45 gene at high doses. These genes are responsible for reactive oxygen species (ROS)-dependent liver damages [7].

Mitochondria play an important role in the production of energy by operating tricarboxylic acid cycle and oxidative phosphorylation [8]. Besides, mitochondria profoundly participate in calcium homeostasis of cells involved in motor functions [9]. ROS are produced by singleelectron carriers in the respiratory chain of mitochondria which causes damage to the cell through lipid peroxidation (LPO), protein oxidation, and mitochondrial DNA (mtDNA) mutations. Apoptotic death of cell can be induced by oxidative stress and mitochondria play a central role in this process because it can release cytochrome c into the cytoplasm [10]. There is a possibility of the interaction of BPA in mitochondria-induced cellular apoptotic process. Mitochondria initiate the process of apoptosis by increasing its own oxidative stress and also the oxidative stress in the constituent cell. Therefore, the study of the oxidative stress in BPA-exposed mitochondria will give us an idea about the probable effects of BPA on cellular apoptosis also. Complexes I and III are also thought to be major sites for the production of superoxide and other ROS [11]. In addition, the mitochondrial respiratory chain is sensitive to nitration-mediated damages [12] which can modify the proteins of mitochondria. Protein oxidation and nitration result in altered functions of many metabolic enzymes in the mitochondrial electron-transport chain, including the nicotinamide adenine dinucleotide dehydrogenase, cytochrome c oxidase, and adenosine triphosphate synthase [13]. Hence, our present study has been carried out to test the causal relationship of BPA with hepatic mitochondrial oxidative stress. In our study, BPA produced oxidative stress, *ex vivo*, in mitochondria isolated from rat liver. BPA also showed the induction of oxidative stress, *ex vivo*, in rat liver mitochondria in a dose- and timedependent manner.

METHODS

Chemicals and reagents

All the chemicals used including the solvents were of analytical grade obtained from Sigma Chemicals Co. (USA), Sisco Research Laboratories, Mumbai, India, Merck Limited, Delhi, India.

Animals

Male Charles Foster rats, weighing 120–140 g, were obtained and were fed with standard diet and handled with care in Animal House as per the recommended guidelines of the Kalyani University Animal Ethics Committee.

Animal sacrifice and collection of tissue samples

The animals were sacrificed through cervical dislocation. The liver was removed surgically after carefully opening the abdominal cavity. The collected liver tissues were rinsed well in cold saline and soaked properly with a piece of blotting paper and stored in sterile vials at -20° C for preparing mitochondria.

Preparation of liver mitochondria

Liver mitochondria were isolated according to the procedure of Dutta *et al.* [14] with some modifications. Two grams of tissue was placed in 10 ml of sucrose buffer (0.25[M] sucrose, 0.001[M] EDTA, and 0.05[M] Tris-HCl [pH 7.8]) at 25°C. Then, the tissue was homogenized for 1 min at low speed using a Potter–Elvenjem glass homogenizer (Belco Glass Inc., Vineland, NJ, USA), after which it was centrifuged at 1500 rpm for 10 min at 4°C. The supernatant was then collected and kept in ice. Then, it was centrifuged at 4000 rpm for 5 min at 4°C. The supernatant, thus, obtained was further centrifuged at 14,000 rpm for 20 min at 4°C. The final supernatant obtained was discarded and the pellet was collected and resuspended in sucrose buffer and was stored at -20°C for further assay.

Incubation of mitochondria with BPA

Initially, the mitochondria were divided into five groups. The incubation mixture of BPA-treated groups containing mitochondria (protein; 1.6 mg/ml), 50 mM potassium phosphate buffer (pH 7.4), and different concentrations of BPA (20-160 µM/ml) (dissolved in 12% DMSO) in a final volume of 1.0 ml was incubated at 37°C in incubator for different time durations (30 min-2 h). On the other hand, the incubation mixture of control group containing mitochondria (protein; 1.6 mg/ml), 50 mM potassium phosphate buffer (pH 7.4), and DMSO (12%) in a final volume of 1.0 ml was incubated at 37°C in incubator for different time durations (30 min-2 h), also. The experimental design of dose- and time-kinetic study has been shown in Fig. 2. Following incubation, (a) the intactness of mitochondria; (b) reactive nitrogen species (RNS) generation; (c) the biomarkers of oxidative stress such as (i) the level of lipid peroxidation (LPO), (ii) the contents of reduced glutathione (GSH), and (iii) the level of protein carbonylation (PCO); and (d) the activities of (i) mitochondrial antioxidant enzymes, (ii) pyruvate dehydrogenase (PDH), and some of the Kreb's cycle enzymes, were determined as described below.







Fig. 2: Experimental design of dose- and time-kinetic study

Determination of mitochondrial intactness by using Janus green B stain

Following incubation, the mitochondria were spread on a slide. After that, a few drops of Janus green B stain were treated with the slide and were left for 5 min for staining in moist chamber. The mitochondria were rinsed once with distilled water so that the complete stain was not gone and a diluted stain remained. Then, the mitochondria were mounted in a drop of distilled water with a coverslip and imaged with bright-field microscope and a confocal system (BD Pathway 855, USA) [14].

Measurement of the levels of RNS in mitochondria

The concentrations of NO, one of the RNS in the incubated mitochondria, were measured spectrophotometrically at 548 nm according to the method of Fiddler, 1977, using Griess reagent [15,16].

Measurement of biomarkers of oxidative stress

LPO level

The levels of LPO in the incubated mitochondria were determined according to the method of Buege *et al.* [17] with some modifications [18]. Briefly, the incubated mitochondria were mixed with thiobarbituric acid-trichloroacetic acid (TBA-TCA) reagent with thorough shaking and heated for 20 min at 80°C. The samples were then cooled to room temperature. The absorbance of the orange chromogen present in the clear supernatant after centrifugation at 2000 rpm for 10 min at room temperature was measured at 532 nm using a UV-VIS spectrophotometer. The unit was used as nmols of TBARS/mg protein.

Reduced GSH content

Reduced glutathione content was estimated by its reaction with DTNB (Ellman's reagent) following the method of Sedlak and Lindsay 1968. Incubated mitochondria were mixed with Tris-HCl buffer (pH 9.0), followed by DTNB for color development. The absorbance was measured at 412 nm to determine the GSH content. The values were expressed as nmole GSH/mg protein [19].

PCO level

The level of PCO was estimated by the method of Levine *et al.*, 1994 [20]. One-fourth of incubated mitochondria were taken in each tube and 0.5 ml dinitrophenylhydrazine in 2M hydrochloric acid was added to the tubes. The tubes were vortexed every 10 min in the dark for 1 h. Proteins were then precipitated with 30% TCA and centrifuged at 10,000 rpm for 10 min. The pellet was washed three times with 1 ml of ethanol: ethyl acetate (1:1, v/v). The final pellet was dissolved in 1 ml of 6 M guanidine-hydrochloride in 20 mM potassium dihydrogen phosphate (pH 2.3). The absorbance was measured spectrophotometrically at 370 nm. The values were expressed as nmol/mg protein.

Determination of the activities of antioxidant enzymes

Manganese superoxide dismutase (MnSOD) activity

MnSOD activity was measured by pyrogallol autoxidation method [21]. To $50 \,\mu$ l of the mitochondrial sample, $430 \,\mu$ l of Tris-HCl buffer ($50 \,m$ M, pH 8.2) and $20 \,\mu$ l pyrogallol (2 mM) were added. An increase in absorbance was recorded at $420 \,n$ m for 3 min in an ultraviolet (UV)-VIS spectrophotometer. The specific activity of the enzyme was expressed as units/mg protein.

Glutathione reductase (GR) activity

GR activity was determined according to the method of Krohne-Ehrich *et al.* [22] with some modifications [23]. Three-milliliter assay mixture contained 50 mM phosphate buffer, 200 mM KCl, and 1 mM EDTA. The reaction was initiated with 1 mM oxidized glutathione (GSSG). The decrease in NADPH absorption was monitored spectrophotometrically at 340 nm. The specific activity of the enzyme was expressed as units/ mg protein.

Glutathione peroxidase (GPx) activity

GPx activity was measured according to the method of Paglia *et al.* [24], with some modifications as adopted by Dutta *et al.* [25]. The assay system contained 0.05 M phosphate buffer with 2 mM EDTA (pH 7.0), 0.025 mM sodium azide, 0.15 mM GSH, and 0.25 mM NADPH along with suitable amount of mitochondria as the source of the enzyme (final volume of 1 ml). The reaction was started by the addition of 0.36 mM H_2O_2 . The linear decrease of absorbance at 340 nm was recorded using a UV-vis spectrophotometer. The specific activity of the enzyme was expressed as Units/mg protein.

Determination of the activities of PDH and some of the Kreb's cycle enzymes

Pyruvate dehydrogenase (PDH) activity

PDH activity was measured spectrophotometrically according to the method of Chretien *et al.*, [26] by following the reduction of NAD⁺ to NADH at 340 nm using 50 mM phosphate buffer (pH 7.4), 0.5 mM sodium pyruvate as the substrate and 0.5 mM NAD⁺ in addition to the suitable amount of mitochondrial suspension (as the source of enzyme). The specific activity of the enzyme was expressed as units/mg protein.

Isocitrate dehydrogenase (ICDH) activity

The activity of ICDH was measured according to the method of Duncan and Fraenkel [27] by measuring the reduction of NAD⁺ to NADH at 340 nm with the help of a UV-vis spectrophotometer. One millimeter assay mixture contained 50 mM phosphate buffer (pH 7.4), 0.5 mM isocitrate, 0.1 mM MnSO₄, and 0.1 mM NAD⁺ and the suitable amount of incubated mitochondrial suspension as the source of enzyme. The specific activity of the enzyme was expressed as Units/mg protein.

Alpha-Ketoglutarate dehydrogenase (α-KGDH) activity

 α -KGDH activity was measured spectrophotometrically according to the method of Duncan and Fraenkel [27] by measuring the reduction of NAD⁺ (0.35 mM) to NADH at 340 nm using 50 mM phosphate buffer (pH 7.4), 0.1 mM α -ketoglutarate as the substrate and the suitable amount of incubated mitochondrial suspension as the source of enzyme. The specific activity of the enzyme was expressed as Units/mg protein.

Succinate dehydrogenase (SDH) activity

The activity of SDH was measured spectrophotometrically by following the reduction of potassium ferricyanide $[K_3Fe(CN)_6]$ at 420 nm according to the method of Veeger *et al.*, [28] with some modifications [29]. One milliliter assay mixture contained 50 mM phosphate buffer (pH 7.4), 2% (w/v) bovine serum albumin, 4 mM succinate, 2.5 mM potassium ferricyanide $[K_3Fe(CN)_6]$, and the suitable amount of incubated mitochondrial suspension as the source of enzyme. The specific activity of the enzyme was expressed as Units/mg protein.

Estimation of protein

The protein content of the mitochondria from liver tissue was determined by the method of Lowry *et al.* [30].

Statistical evaluation

Each experiment was repeated at least six times. Data are presented as means±standard error. Significance of mean values of different parameters between the incubation mitochondria were analyzed using one-way *post hoc* tests (Tukey's HSD test) of analysis of variances after ascertaining the homogeneity of variances between the incubations. Pair-wise comparisons were done by calculating the least significance. Statistical tests were performed using Microcal Origin version 7.0 for Windows.

RESULTS

Changes in intactness of mitochondria

Fig. 3a-d depicts a significant decrease in the mitochondrial intactness following the incubation of mitochondria with BPA in a dose- and time-dependent manner. It depicts a decrease in the mitochondrial intactness following the incubation of mitochondria with different doses ($20 \,\mu$ M/ml, $40 \,\mu$ M/ml, $80 \,\mu$ M/ml, and $160 \,\mu$ M/ml) of BPA in a time-dependent manner. Fig. 3e reveals the intactness of mitochondria of various groups by the study with bright-field microscope (magnification ×40).

Changes in the level of NO of mitochondria

The concentrations of NO in mitochondria in BPA-incubated groups (20 μ M/ml, 40 μ M/ml, 80 μ M/ml, and 160 μ M/ml) were found to be increased significantly (Fig. 3f) when compared to the control group by 3.15%, 4.54%, 8.01%, and 10.79% (in case of 30 min incubation) (*p<0.001); 29.37%, 75.29%, 1.45 folds, and 2.09 folds (in case of 1 h incubation) (*p<0.001); 73.78%, 1.36 folds, 2.94 folds, and 4.16 folds (in case of 1 h 30 min incubation) (*p<0.001); and also 68.35%, 1.42 folds, 3.15 folds, and 4.08 folds (in case of 2 h incubation) (*p<0.001).

Changes in the levels of LPO

The levels of LPO in mitochondria in BPA-incubated groups ($20 \mu M/ml$, $40 \mu M/ml$, $80 \mu M/ml$, and $160 \mu M/ml$) were found to be increased significantly (Fig. 4a) when compared to the control group by 4.77%, 0.24%, 4.30%, and 15.99% (in case of 30 min incubation) (*p<0.001); 30.88%, 49.64%, 76.48%, and 1 fold (in case of 1 h incubation) (*p<0.001); 51.54%, 83.37%, 1.13 folds, and 1.27 folds (in case of 1 h 30 min incubation) (^p<0.001); and also 46.35%, 82.35%, 1.03 folds, and 1.21 folds (in case of 2 h incubation) ([@]p<0.001).

Changes in the contents of reduced GSH

The contents of GSH in mitochondria in BPA-incubated groups ($20 \ \mu$ M/ml, $40 \ \mu$ M/ml, $80 \ \mu$ M/ml, and $160 \ \mu$ M/ml) were found to be decreased significantly (Fig. 4b) when compared to the control group by 2.79%, 7.14%, 8.89%, and 14.74% (in case of 30 min incubation) (*p<0.001); 34.28%, 50.22%, 65.38%, and 70.26% (in case of 1 h incubation) (*p<0.001); 52.52%, 74.19%, 87.55%, and 92.32% (in case of 1 h 30 min incubation) (^p<0.001); and also 54.52%, 81.23%, 87.28% and 92.97% (in case of 2 h incubation) (*p<0.001).

Changes in the levels of PCO

The levels of PCO in mitochondria in BPA-incubated groups ($20 \mu M/ml$, $40 \mu M/ml$, $80 \mu M/ml$, and $160 \mu M/ml$) were found to be increased significantly (Fig. 4c) when compared to the control group by 23.01%,



Fig. 3: Images representing bisphenol A (BPA)-incubated alterations in intactness ([a-d]: Janus green B stained [×40 magnification]; [e] Bright-field microscopy [×40 magnification]) and nitric oxide concentration [f] of mitochondria of control (30 min, 1 h, 1 h 30 min, and 2 h) and BPA (20 μ M/ml, 40 μ M/ml, 80 μ M/ml, and 160 μ M/ml)-incubated groups (30 min, 1 h, 1 h 30 min, and 2 h) (*p≤0.001) (in case of 1 h incubation); (^p≤0.001) (in case of 1 h 30 min incubation); and (*p≤0.001) (in case of 2 h incubation). Results have been expressed as mean±standard error of the mean



Fig. 4: Images representing bisphenol A (BPA)-incubated alterations in the content of reduced glutathione (a), levels of lipid peroxidation (b), and protein carbonylation (c) of mitochondria of control (30 min, 1 h, 1 h 30 min, and 2 h) and BPA (20 μM/ml, 40 μM/ml, 80 μM/ml, and 160 μM/ml)-incubated groups (30 min, 1 h, 1 h 30 min and 2 h) (*p≤0.001) (in case of 1 h incubation); (^p≤0.001) (in case of 1 h 30 min incubation); (®p≤0.001) (in case of 2 h incubation). Results have been expressed as mean±standard error of the mean

63.72%, 80.53%, and 1.03 folds (in case of 30 min incubation) (*p<0.001); 1.13 folds, 3.44 folds, 6.64 folds, and 11.6 folds (in case of 1 h incubation) (*p<0.001); 3.30 folds, 5.65 folds, 7.57 folds, and 13.59 folds (in case of 1 h 30 min incubation) (*p<0.001); and also 3.65 folds, 5.87 folds, 7.42 folds, and 10.6 folds (in case of 2 h incubation) (*p<0.001).

Changes in the activities of MnSOD

The activities of MnSOD in mitochondria in BPA-incubated groups (20 μ M/ml, 40 μ M/ml, 80 μ M/ml, and 160 μ M/ml) were found to be increased significantly (Fig. 5a) when compared to the control group by 6.76%, 9.18%, 2.09%, and 0.97% (in case of 30 min incubation) (*p≤0.001); 69.41%, 1.78 folds, 3.12 folds, and 5.67 folds (in case of 1 h incubation) (*p≤0.001); 2.44 folds, 4.55 folds, 6.17 folds, and 10.83 folds (in case of 1 h 30 m incubation) (*p≤0.001); and also 2.55 folds, 4.02 folds, 6.60 folds, and 11.29 folds (in case of 2 h incubation) (*p≤0.001).

Changes in the activities of GPx

The activities of GPx in mitochondria in BPA-incubated groups ($20 \ \mu$ M/ml, $40 \ \mu$ M/ml, $80 \ \mu$ M/ml, and $160 \ \mu$ M/ml) were found to be decreased significantly (Fig. 5b) when compared to the control group by 0.97%, 6.42%, 3.57%, and 4.63% (in case of 30 min incubation) (*p<0.001); 33.43%, 51.51%, 71.08%, and 85.48% (in case of 1 h incubation) (*p<0.001); 41.95%, 74.75%, 83.58%, and 92.22% (in case of 1 h 30 min incubation) (^p<0.001); and also 41.25%, 77.02%, 85.39%, and 92.16% (in case of 2 h incubation) (*p<0.001).

Changes in the activities of GR

The activities of GR in mitochondria in BPA-incubated groups (20 μ M/ml, 40 μ M/ml, 80 μ M/ml, and 160 μ M/ml) were found to be decreased significantly (Fig. 5c) when compared to the control group by 4.43%, 11.98%, 14.32%, and 17.45% (in case of 30 min incubation) (*p<0.001); 37.57%, 52.43%, 62.43%, and 79.19% (in case of 1 h incubation) (*p<0.001); 43.25%, 61.00%, 81.50%, and 87.50% (in case of 1 h 30 min incubation) (*p<0.001); and also 45.40%, 65.63%, 78.65%, and 88.02% (in case of 2 h incubation) (*p<0.001).

Changes in the activities of PDH

The activities of PDH in mitochondria in BPA-incubated groups (20 μ M/ml, 40 μ M/ml, 80 μ M/ml, and 160 μ M/ml) were found to be decreased significantly (Fig. 6a) when compared to the control group by 3.95%, 12.66%, 17.29%, and 12.85% (in case of 30 min incubation) (*p<0.001); 27.50%, 47.39%, 61.64%, and 72.37% (in case of 1 h incubation) (*p<0.001); 43.82%, 55.09%, 67.32%, and 80.29% (in case of 1 h 30 min incubation) (^p<0.001); and also 46.68%, 57.73%, 67.67%, and 78.44% (in case of 2 h incubation) (*p<0.001).

Changes in the activities of ICDH

The activities of ICDH in mitochondria in BPA-incubated groups (20 μ M/ml, 40 μ M/ml, 80 μ M/ml, and 160 μ M/ml) were found to be decreased significantly (Fig. 6b) when compared to the control group by 24.80%, 59.06%, 66.34%, and 85.04% (in case of 1 h incubation) (*p<0.001); 55.41%, 74.52%, 83.78%, and 90.15% (in case of 1 h 30 min incubation) (^p<0.001); and also 57.17%, 72.66%, 83.37%, and 90.63% (in case of 2 h incubation) (*p<0.001).

Changes in the activities of α -KGDH

The activities of α -KGDH in mitochondria in BPA-incubated groups (20 μ M/ml, 40 μ M/ml, 80 μ M/ml, and 160 μ M/ml) were found to be decreased significantly (Fig. 6c) when compared to the control group by 1.01%, 8.97%, 0.82%, and 5.12% (in case of 30 min incubation) (*p<0.001); 25.88%, 45.23%, 69.12%, and 80.15% (in case of 1 h incubation) (*p<0.001); 39.39%, 55.18%, 67.79%, and 76.87% (in case of 1 h 30 min incubation) (^p<0.001); and also 27.09%, 49.70%, 65.08%, and 75.18% (in case of 2 h incubation) (*p<0.001).

Changes in the activities of SDH

The activities of SDH in mitochondria in BPA-incubated groups (20 μ M/ml, 40 μ M/ml, 80 μ M/ml and 160 μ M/ml) were found to be decreased significantly (Fig. 6d) when compared to the control group by 3.51%, 3.36%, 2.60%, and 7.18% (in case of 30 min incubation) (*p<0.001); 16.67%, 59.40%, 74.59%, and 84.03% (in



Fig. 5: Images representing bisphenol A (BPA)-incubated alterations in the activities of Mn-superoxide dismutase (A), glutathione peroxidase (B) and glutathione reductase (C) of mitochondria of control (30 min, 1 h, 1 h 30 min, and 2 h) and BPA (20μ M/ml, 40μ M/ml, 80μ M/ml, and 160μ M/ml)-incubated groups (30 min, 1 h, 1 h 30 min, and 2 h) ("p ≤ 0.001) (in case of 1 h incubation); (^p ≤ 0.001) (in case of 2 h incubation). Results have been expressed as mean±standard error of the mean



Fig. 6: Images representing bisphenol A (BPA)-incubated decreases in the activities of pyruvate dehydrogenase (a), isocitrate dehydrogenase (b), α -ketoglutarate dehydrogenase (c) and succinate dehydrogenase (d) of mitochondria of control (30 min, 1 h, 1 h 30 min, and 2 h) and BPA (20 μ M/ml, 40 μ M/ml, 80 μ M/ml, and 160 μ M/ml)-incubated groups (30 min, 1 h, 1 h 30 min, and 2 h) (*p≤0.001) (in case of 1 h incubation); (^p≤0.001) (in case of 1 h 30 min incubation); (@p≤0.001) (in case of 2 h incubation). Results have been expressed as mean±standard error of the mean

case of 1 h incubation) ($p \le 0.001$); 39.43%, 74.76%, 81.33%, and 87.67% (in case of 1 h 30 min incubation) ($p \le 0.001$); and also 39.47%, 74.43%, 80.50%, and 87.35% (in case of 2 h incubation) ($p \le 0.001$).

DISCUSSION

Mitochondria are the main source of generation of the superoxide anion free radical (O_2) and other ROS [31]. The main mechanisms responsible for mitochondrial ROS production are the oxidative phosphorylation which occurs in the respiratory chain, in particular, its complexes I and III [32], in the inner mitochondrial membrane, and monoamine oxidase in the outer membrane. The noxious action of ROS mainly consists of the peroxidation of lipids, in particular, phospholipids of biological membranes and oxidative damage to proteins and DNA [33]. BPA can leach into food from the epoxy resin lining of cans and from consumer products such as polycarbonate tableware, food storage containers, water bottles, and baby bottles. Additional traces of BPA can leach out of these products when they are heated at high temperatures. Recent studies have also suggested that people may be exposed to BPA by handling cash register receipts which are made up of components containing BPA as one of the constituents.

When the human body is being exposed to BPA (itself contains strong electron donating group (-OH) in its structure) then it will be biotransformed in the liver and gut by glucuronosyltransferases and SULTs, respectively, to BPA glucuronide containing strong electron donating group (-OH) in its structure which can donate electrons to the free oxygen to produce superoxide free anion radical (O_2^{-n}), which is one of the causative factors of oxidative stress in the physiological system. It has been shown that the BPA by decreasing expression of the gene responsible for prevention of oxidative activity can induce production of ROS and subsequent hepatotoxicity as well [34]. NO is a free radical which plays an important role in the pathogenesis of pain, inflammation, neural signal transmission, immune response, control of vasodilatation, and blood pressure [35,36].

GSH provides the first line of defense against ROS, as it can scavenge free radicals and reduce H_2O_2 BPA can produce various quinol and semiquinone intermediates which can react with GSH to produce GSH conjugates which are responsible for the generation of oxidative stress. LPO levels are assayed to determine the extent of oxidative damages involving mainly membrane lipids. Free radicals produced due to oxidative stress can easily react with lipids of the cell membrane and thereby initiating a chain reaction leading to the production of lipid peroxides [37]. On the other hand, PCO levels are estimated to study the protein oxidations owing to their early formation and higher stability.

The cellular antioxidant system against oxidative stress comprises the tripeptide GSH and few enzymes regulating its metabolism. GPx uses reduced GSH to remove peroxides produced due to oxidative stress. On the other hand, GR reduces the oxidized GSH back to GSH using NADPH [38].

PDH has been demonstrated to be sensitive which induce oxygen free radicals, which might be a cause of decrease of the enzyme activity [39]. ICDH is a major NADPH producer in the mitochondrial Kreb's cycle pathway and thus plays a key role in cellular defense against oxidative stress-induced damage. During marked elevation in ROS generation, DNAfragmentation and a decrease in the activity of ICDH were observed [40].

A study by Chitra *et al.*, 2003, on epididymal sperm of rats due to BPAinduced oxidative stress had shown that graded dose of BPA elicits the depletion of antioxidant defense system and responsible for oxidative stress generation in epididymal sperm of rats which corresponds to our results on graded incremental dose of BPA induces oxidative



Fig. 7: Scheme showing the probable mechanism of generation of oxidative stress by BPA in hepatic mitochondria

stress in *ex vivo* liver mitochondrial model [41]. A research study by Moon *et al.*, 2012, had shown that BPA doses below the NOAEL (noobserved-adverse-effect level) can induce mitochondrial dysfunction in the liver which is associated with an increase in oxidative stress and inflammation. Our experimental results had shown that graded micromole (μ M) concentrations of BPA induced similar effects in mitochondrial *ex vivo* model [42]. GSH depletion could lead to an oxidative stress condition which is being supported by the presence of increased levels of MDA concentration in the GSH-depleted mitochondrial system. This statement has also been supported by the present study and also by an earlier study [43].

In our study, the intactness of mitochondrial structure, concentrations of NO, levels of LPO and PCO, contents of GSH and activities of various antioxidant enzymes (Mn-SOD, GPx, GR), and also activities of PDH and various Kreb's cycle enzymes (ICDH, α-KGDH, SDH) were found to be altered significantly in all the BPA-incubated groups $(20 \,\mu\text{M/m})$, $40 \,\mu\text{M/m}$, $80 \,\mu\text{M/m}$, and $160 \,\mu\text{M/m}$) as compared to the control group at all the time points studied. The probable mechanism of generation of oxidative stress by BPA is that the BPA can react with reduced GSH and produce GSH conjugate which is responsible for decreasing content of reduced GSH and this reaction may be responsible for alteration of GSH/GSSG ratio. Another mechanism of generation of oxidative stress by BPA is that BPA can generate higher concentration of electron may be by altering the respiratory chain reactions which can produce higher amount of superoxide free radicals. This free radical itself and also by producing hydroxyl radical causes oxidative stress. As a result of BPA-induced oxidative stress, the enzymes of tricarboxylic acid cycle and mitochondrial electron transport chain might be altered. Consequently, ATP synthesis by the mitochondrial respiratory chain of the BPA-exposed mitochondria might be inhibited; and, thus, ATP-driven all the cellular biochemical processes might be affected. Moreover, the alterations in the levels and activities of various parameters showed a linear trend in progression with time. This study has the uniqueness in

the sense that it into account both dose- and time-dependent pattern of BPA in oxidative stress generation.

From the findings of the present study, it can be concluded that the incubation of mitochondria isolated from rat liver with BPA, caused oxidative stress-mediated damages in mitochondria in both dose- and time-dependent manners. This study is formulating an *ex vivo* model for studying xenobiotics-induced mitochondrial oxidative stress. The probable mechanism(s) of generation of oxidative stress in mitochondria by BPA has been elucidated in Fig. 7.

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CONFLICT OF INTEREST

The authors declare that there are no conflicts of interest.

AUTHORS' CONTRIBUTIONS

MD: Designing of problem, experimental study, manuscript preparation.

GP: Designing of problem, manuscript correction.

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