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Antitumor Agents. 269. Non-Aromatic Ring-A Neo-tanshinlactone Analog, TNO, as a New Class of Potent Antitumor Agents

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Abstract

Tetrahydroneotanshinlactone (TNT) and tetrahydronaphthalene-1-ol (TNO) derivatives were designed, synthesized, and evaluated for cytotoxic activity. The TNO derivatives were found to be a promising novel class of *in vitro* antitumor agents. The cyclohexene ring-A could dramatically affect the antitumor activity and selectivity. Compound **20** showed the highest potency with ED₅₀ values of 0.7 and 1.7 μ M against SK-BR-3 and ZR-75-1 breast cancer cell lines, respectively.

Keywords

Tetrahydronaphthalene-1-ol (TNO) analogs; Tetrahydroneotanshinlactone (TNT) analogs; Cytotoxicity; Breast cancer

Cancer is the second leading cause of death in the United States, accounting for one-quarter of all deaths.¹ Cancer incidence and resultant mortality have increased by approximately 22% since 1990,² and 1,400,000 new cancer cases will be diagnosed in 2009, according to estimates by the American Cancer Society.³ Natural products and natural product-derived compounds have provided more than two-thirds of the clinically used anticancer agents, and thus, are important sources for drug discovery.^{4, 5} Plant-based drug discovery has resulted particularly in the development of new anticancer agents to by-pass multidrug resistance and overcome side effects of current therapeutic medicines.^{6, 7} The above facts prompted us to focus on natural products and their analogs in our anticancer drug program.

Research on tanshinones (isolated from the *Salvia* genus) began in the early 1930s.⁸ Tanshinone I (1) and tanshinone IIA (2) differ structurally in the ring-A system: the former has an aromatic ring, while the latter has a non-aromatic ring (Figure 1). Compounds 1 and 2 have been studied extensively for their antitumor effects, and display different activities and selectivities.⁸ Recent studies indicated that 1 reduced metastasis and tumorigenesis by inhibition of IL-8,⁹ while 2 induced cell differentiation and apoptosis.¹⁰ Neo-tanshinlactone

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(3) (Figure 1), reported by our group previously,¹¹ showed significant and selective *in vitro* anti-breast cancer activity. We further studied how the individual rings in 3 influence the *in vitro* activity, and the results led to the discovery of a novel class of potential anti-breast cancer agents, 2-(furan-2-yl)naphthalen-1-ol derivatives, such as analog 4. However, it remained unclear how ring A affects the activity and selectivity of 3- and 4-analogs. To answer this question, we designed derivatives with two new scaffolds, tetrahydroneotanshinlactone (5, TNT) and tetrahydronaphthalene-1-ol (6, TNO). Like 2, both TNT and TNO derivatives have a non-aromatic ring-A. Different ring sizes, including five- and six-membered rings, were studied and 14 new analogs were designed. This paper reports the synthesis and biological evaluation of 5- and 6-analogs.

As shown in Scheme 1, compound 8 was synthesized by Negishi cross-coupling reaction of compound 7 with 4-methylpent-3-enyl zinc(II) bromide in 96% yield.^{12, 13} Treatment of 8 with AlCl₃ followed by demethylation gave 9 in 84% yield.¹³ Compounds 10 and 11, with five- and six-membered rings, respectively, are commercially available. Compounds 9-11 underwent the previously reported two-step ring closure reactions to afford furochromenones 15–17, which were hydrolyzed by using sodium hydroxide to give ring-opened compounds 18-20. Compound 20, with gem-dimethyl substitution on ring-A, showed significant cytotoxic activity (Table 1), and was chosen for further modification to study the functions of the hydroxyl and carboxylic acid groups (21-28). The selective methylation of the hydroxy group on 20 was achieved by the addition of MeI and 18-crown-6 ether to the crude hydrolysis mixture of 17 without work-up.¹⁴ The resulting carboxylic acid 21 was converted to methyl ester 22 with thionyl chloride and MeOH at room temperature. Meanwhile, the reduction of 20 with lithium aluminum hydride afforded diol 23, which was treated with iodomethane and iodoethane in the presence of Cs2CO3 to generate ethers 24 and 25, respectively. The remaining primary alcohol of 24 was alkylated with iodomethane and iodoethane in the presence of NaH to obtain 26 and 27, respectively. Acetate 28 was obtained by acetylation of 24 with Ac_2O .

The newly synthesized analogs **15–28** were tested for *in vitro* cytotoxic activity against a panel of human tumor cell lines according to previously published methods.¹⁵ Cell lines include: SK-BR-3 (estrogen receptor negative, HER2 over-expressing breast cancer), ZR-75-1 (estrogen receptor positive breast cancer), MDA-MB-231 (estrogen receptor negative breast cancer), A549 (non small cell lung cancer), DU145 (prostate cancer cell line), KB (nasopharyngeal carcinoma), and KB-vin (vincristine-resistant MDR KB subline).

Among the three tetrahydroneotanshinlactone (TNT) derivatives, **15** showed no activity against any tumor cell line tested, which suggested that the five-membered ring A was not favored. Compound **16** was three- to seven-fold more potent than **17** against SK-BR-3, ZR-75-1, A549, and KB-vin cell lines. However, while **16** was less potent compared with **3** against SK-BR-3 and ZR-75-1 breast cancer cell lines, it also showed a broader antitumor spectrum, with greatly enhanced potency against A549 and KB-vin. The results suggested that ring-A could affect the potency and tumor-tissue type selectivity dramatically.

Among tetrahydronaphthalene-1-ol (TNO) derivatives, compounds **18** and **19** displayed only marginal antitumor activity, while **20** showed potent and broad antitumor activity against all tumor cell lines tested (ED_{50} 0.7 µM against SK-BR-3; 1.7 µM against ZR-75-1). Thus, a non-aromatic six-membered ring-A with *gem*-dimethyl substitution was favored for cytotoxic activity, in comparison with unsubstituted five- and six-membered rings. As to the tumor-tissue type selectivity, **20** was significantly active against all tumor cell lines tested, except MDA-MB-231, while **3** and **4** were active against only two of the breast cancer cell lines. In contrast to **20**, compounds **3** and **4** lack the *gem*-dimethyls, and are essentially planar. These results demonstrated that, by changing the molecular conformation and orientation, introduction of a non-aromatic ring-A could greatly influence the antitumor activity against all cell types. In our

prior SAR studies of neo-tanshinlactone (**3**) and the ring-opened analog **4**, the presence of two functional groups from the opened lactone ring-C was critical for antitumor activity, which encouraged us to study comparable derivatives of **20** with ether and ester groups of various sizes. As seen in Table 1, **21–28** showed only moderate to marginal activity against all tumor cell lines tested, but interestingly, still displayed low sensitivity against MDA-MB-231 compared with other tumor cell lines. For example, **25** and **28** showed four-fold higher potency against SK-BR-3 than MDA-MB-231. In summary, the current SAR study indicated that the optimal substituents on the phenyl and furanyl rings are hydroxy and carboxylic acid groups. The preliminary results indicated that the identities of the ring A, hydroxy, and carboxylic acid groups are important to antitumor activity and selectivity. More analogs will be synthesized and evaluated to establish detailed structure-activity relationships (SAR) of this new series of compounds.

In conclusion, tetrahydroneotanshinlactone (TNT) and tetrahydronaphthalene (TNO) derivatives were prepared in order to investigate the effect of the non-aromatic ring-A on *in vitro* antitumor activity. The results indicated that a non-aromatic ring-A could dramatically affect both activity and tumor cell line selectivity, particularly the non-breast cell lines that were studied. Based on this study, a novel class of antitumor agents, TNO derivatives, was discovered and developed. Compound **20** was the most potent analog with an ED₅₀ value of 0.7 μ M against the SK-BR-3 cell line, and showed broader antitumor activity compared with **3** and **4**. Further SAR and mechanism of action studies are ongoing and progress will be reported in due course. In summary, **20** is a promising new lead compound with a novel skeleton for further development toward a new potential clinical trials candidate.

Acknowledgments

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- 16. Spectroscopic data. 1-Methyl-7,8-dihydrocyclopenta[*h*]furo[3,2-*c*] chromen-10(6*H*) one (15): ¹H NMR (300 MHz, CDCl₃, ppm): δ 2.19 (p, *J* = 7.5 Hz, 2H, CH₂CH₂CH₂), 2.36 (d, *J* = 1.5 Hz, 3H, CH₃), 3.04 (t, *J* = 7.5 Hz, 2H, CH₂CH₂CH₂), 3.14 (t, *J* = 7.5 Hz, 2H, CH₂CH₂CH₂), 7.19 (d, *J* = 7.8 Hz, 1H, aromatic), 7.36 (q, *J* = 1.2 Hz, 1H, OCH), 7.63 (d, *J* = 8.1 Hz, 1H, aromatic); HRMS

Calcd for C15H13O3 (M+H⁺): 241.0859, found: 241.0858. 1-Methyl-8,9-dihydro-6H-benzo[h]furo [3,2-*c*]chromen-11(7*H*)-one (**16**): ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.80–1.86 (m, 4H, CH₂CH₂CH₂CH₂), 2.35 (d, J = 1.2 Hz, 3H, CH₃), 2.84 (t, J = 5.7 Hz, 2H, CH₂CH₂CH₂CH₂), 2.94 (t, J = 6.0 Hz, 2H, CH₂CH₂CH₂CH₂), 7.01 (d, J = 8.4 Hz, 1H, aromatic), 7.34 (d, J = 0.9 Hz, 1H, OCH), 7.51 (d, J = 8.1 Hz, 1H, aromatic); HRMS Calcd for C₁₆H₁₅O₃ (M+H⁺): 255.1016, found: 255.1012. 1,6,6-Trimethyl-8,9-dihydro-6H-benzo[h]furo[3,2-c]chromen-11(7H)-one (17). 38% yield; mp 101–103 °C; ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.33 (s, 6H, C(CH₃)₂), 1.67–1.71 (m, 2H, CCH₂CH₂CH₂), 1.84–1.88 (m, 2H, CCH₂CH₂CH₂), 2.35 (d, J = 1.2 Hz, 3H, CH₃), 2.97 (t, J = 6.3 Hz, 2H, CCH₂CH₂CH₂), 7.32 (d, J = 8.4 Hz, 1H, aromatic), 7.35 (q, J = 1.2 Hz, 1H, OCH), 7.61 (d, J = 8.7 Hz, 1H, aromatic); HRMS Calcd for C₁₈H₁₉O₃ (M+H⁺): 283.1329, found: 283.1315. 2-(4-Hydroxy-2,3-dihydro-1*H*-inden-5-yl)-4-methylfuran-3-carboxylic acid (18): ¹H NMR (300 MHz, CD₃OD, ppm): δ 2.06 (p, *J* = 7.5 Hz, 2H, CH₂CH₂CH₂), 2.20 (d, *J* = 0.9 Hz, 3H, CH₃), 2.89 (q, *J* = 7.5 Hz, 4H, CH₂CH₂CH₂), 4.94 (s, 1H, OH), 6.83 (d, J = 7.8 Hz, 1H, aromatic), 7.14 (d, J = 7.8 Hz, 1H, aromatic), 7.30 (d, J = 0.9 Hz, 1H, OCH); MS: m/z 257 (M-H⁺). 2-(1-Hydroxy-5,6,7,8tetrahydronaphthalen-2-yl)-4-methylfuran-3-carboxylic acid (19): ¹H NMR (300 MHz, CD₃COCD₃, ppm): δ 1.75–1.77 (m, 4H, CH₂), 2.20 (d, J = 1.2 Hz, 3H, CH₃), 2.69–2.75 (m, 4H, CH₂) 6.67 (d, J = 8.4 Hz, 1H, aromatic), 7.10 (d, J = 8.4 Hz, 1H, aromatic), 7.43 (s, 1H, OCH); HRMS Calcd for C₁₆H₁₅O₄ (M-H⁺): 271.0970, found: 271.0971. 2-(1-Hydroxy-5,5dimethyl-5,6,7,8-tetrahydronaphthalen-2-yl)-4-methylfuran-3-carboxylic acid (20): ¹H NMR (300 MHz, CD₃OD, ppm): δ 1.28 (s, 6H, C(CH₃)₂), 1.62–1.66 (m, 2H, CH₂), 1.78–1.82 (m, 2H, CH₂), 2.21(d, J = 1.5 Hz, 3H, CH₃), 2.70 (t, J = 6.3 Hz, 2H, CH₂), 6.96 (d, J = 8.4 Hz, 1H, aromatic), 7.13 (d, J = 8.4 Hz, 1H, aromatic), 7.33 (d, J = 1.2 Hz, 1H, OCH); HRMS Calcd for C₁₈H₁₉O₄ (M -H⁺): 301.1434, found: 301.1425. 2-(1-Methoxy-5,5-dimethyl-5,6,7,8-tetrahydronaphthalen-2yl)-4-methylfuran-3-carboxylic acid (21): ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.30 (s, 6H, $(CH_3)_2$, 1.63–1.67 (m, 2H, CCH₂CH₂CH₂CH₂), 1.77–1.83 (m, 2H, CCH₂CH₂CH₂), 2.36 (d, J = 0.9Hz, 3H, CH₃), 2.76 (t, J = 6.3 Hz, 1H, CCH₂CH₂CH₂), 3.52 (s, 3H, OCH₃), 7.16 (d, J = 8.4 Hz, 1H, aromatic), 7.23 (d, J = 8.4 Hz, 1H, aromatic), 7.29 (d, J = 1.5 Hz, 1H, OCH); MS: m/z 315 (M +H⁺). Methyl 2-(1-methoxy-5,5-dimethyl-5,6,7,8-tetrahydronaphthalen-2-yl)-4-methylfuran-3carboxylate (22): ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.30 (s, 6H, (CH₃)₂), 1.63–1.67 (m, 2H, CCH₂CH₂CH₂), 1.77–1.83 (m, 2H, CCH₂CH₂CH₂), 2.20 (d, J = 1.2 Hz, 3H, CH₃), 2.75 (t, J = 6.3 Hz, 1H, CCH₂CH₂CH₂), 3.46 (s, 3H, OCH₃), 3.72 (s, 3H, COOCH₃), 7.14 (d, J = 8.1 Hz, 1H, aromatic), 7.23 (d, J = 8.1 Hz, 1H, aromatic), 7.27 (d, J = 0.9 Hz, 1H, OCH); MS: m/z 329 (M+H⁺). 2-(3-(Hydroxymethyl)-4-methylfuran-2-yl)-5, 5-dimethyl-5, 6, 7, 8-tetrahydronaphthalene-1-ol(23): ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.30 (s, 6H, (CH₃)₂), 1.63–1.67 (m, 2H, CCH₂CH₂CH₂), 1.80–1.84 (m, 2H, CCH₂CH₂CH₂), 2.11 (d, *J* = 0.9 Hz, 3H, CH₃), 2.71 (t, *J* = 6.3 Hz, 2H, CCH₂CH₂CH₂), 4.58 (s, 1H, CH₂OH), 6.97 (d, J = 8.4 Hz, 1H, aromatic), 7.20 (d, J = 8.4 Hz, 1H, aromatic), 7.28 (d, J = 0.9 Hz, 1H, OCH); MS: m/z 385 (M–H⁺). (2-(1-methoxy-5,5dimethyl-5,6,7,8-tetrahydronaphthalen-2-yl)-4-methylfuran-3-yl)methanol (24): ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.30 (s, 6H, (CH₃)₂), 1.64–1.68 (m, 2H, CCH₂CH₂CH₂), 1.77–1.83 (m, 2H, CCH₂CH₂CH₂), 2.12 (d, *J* = 0.9 Hz, 3H, CH₃), 2.69 (t, *J* = 6.3 Hz, 1H, CH₂OH), 2.77 (t, *J* = 6.3 Hz, 2H, CCH₂CH₂CH₂), 3.46 (s, 3H, OCH₃), 4.41 (d, J = 5.7 Hz, 2H, CH₂OH), 7.16–7.22 (m, 2H, aromatic), 7.27 (d, J=0.9 Hz, 1H, OCH); MS: m/z 323 (M+Na⁺). (2-(1-Ethoxy-5,5-dimethyl-5,6,7,8tetrahydronaphthalen-2-yl)-4-methylfuran-(2-(1-methoxy-5,5-dimethyl-5,6,7,8tetrahydronaphthalen-2-yl)-4-methylfuran-3-yl)methanol (24): ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.30 (s, 6H, (CH₃)₂), 1.64–1.68 (m, 2H, CCH₂CH₂CH₂), 1.77–1.83 (m, 2H, CCH₂CH₂CH₂), 2.12 (d, J = 0.9 Hz, 3H, CH₃), 2.69 (t, J = 6.3 Hz, 1H, CH₂OH), 2.77 (t, J = 6.3 Hz, 2H, CCH₂CH₂CH₂), 3.46 (s, 3H, OCH₃), 4.41 (d, J=5.7 Hz, 2H, CH₂OH), 7.16–7.22 (m, 2H, aromatic), 7.27 (d, J = 0.9 Hz, 1H, OCH); MS: m/z 323 (M+Na⁺). (2-(1-Ethoxy-5,5-dimethyl-5,6,7,8tetrahydronaphthalen-2-yl)-4-methylfuran-3-yl)methanol (25): ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.18 (t, J = 7.2 Hz, 3H, CH₂CH₃), 1.30 (s, 6H, (CH₃)₂), 1.64–1.68 (m, 2H, CCH₂CH₂CH₂), 1.77– 1.83 (m, 2H, CCH₂CH₂CH₂), 2.12 (d, *J* = 0.9 Hz, 3H, CH₃), 2.76 (t, *J* = 6.3 Hz, 1H, CCH₂CH₂CH₂), 2.87 (br, 1H, CH₂OH), 3.58 (q, J = 7.2 Hz, 2H, CH₂CH₃), 4.39 (s, 2H, CH₂OH), 7.18 (s, 2H, aromatic), 7.26 (d, J = 0.3 Hz, 1H, OCH); MS: m/z 313 (M–H⁺). 2-(1-Methoxy-5,5dimethyl-5,6,7,8-tetrahydronaphthalen-2-yl)-3-(methoxymethyl)-4-me thylfuran (26). 44 % yield; ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.30 (s, 6H, (CH₃)₂), 1.64–1.67 (m, 2H, CCH₂CH₂CH₂), 1.77–1.83 (m, 2H, CCH₂CH₂CH₂), 2.10 (d, J = 1.2 Hz, 3H, CH₃), 2.77 (t, J = 6.3 Hz, 1H, CCH₂CH₂CH₂), 3.33 (s, 3H, CH₂OCH₃), 3.49 (s, 3H, OCH₃), 4.32 (s, 2H, CH₂OCH₃),

7.17 (dd, J = 8.4 Hz, 2H, aromatic), 7.28 (d, J = 1.2 Hz, 1H, OCH); MS: m/z 315 (M+H⁺). 2-(1-Ethoxy-5,5-dimethyl-5,6,7,8-tetrahydronaphthalen-2-yl)-3-(methoxymethyl)-4-methylfuran (27): ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.19 (t, J = 7.2 Hz, 3H, CH₂CH₃), 1.30 (s, 6H, (CH₃)₂), 1.64–1.67 (m, 2H, CCH₂CH₂CH₂), 1.77–1.83 (m, 2H, CCH₂CH₂CH₂), 2.09 (d, J = 0.9 Hz, 3H, CH₃), 2.77 (t, J = 6.3 Hz, 1H, CCH₂CH₂CH₂), 3.32 (s, 3H, CH₂OCH₃), 3.59 (q, J = 6.9 Hz, 2H, CH₂CH₃), 4.32 (s, 2H, CH₂OCH₃), 7.16 (dd, J = 8.1 Hz, 2H, aromatic), 7.26 (d, J = 0.9 Hz, 1H, OCH); MS: m/z 329 (M+H⁺). (2-(1-Methoxy-5,5-dimethyl-5,6,7,8-tetrahydronaphthalen-2-yl)-4-methylfuran-3-yl)methyl acetate (28): ¹H NMR (300 MHz, CDCl₃, ppm): δ 1.30 (s, 6H, (CH₃)₂), 1.63–1.67 (m, 2H, CCH₂CH₂CH₂), 1.77–1.83 (m, 2H, CCH₂CH₂CH₂), 2.06 (s, 3H, CH₂OCOCH₃), 2.06 (d, J = 0.9 Hz, 3H, CH₃), 2.76 (t, J = 6.3 Hz, 1H, CCH₂CH₂CH₂), 3.48 (s, 3H, OCH₃), 5.01 (s, 2H, CH₂OCOCH₃), 7.13–7.19 (m, 2H, aromatic), 7.30 (d, J = 1.2 Hz, 1H, OCH); MS: m/z 365 (M+Na⁺).

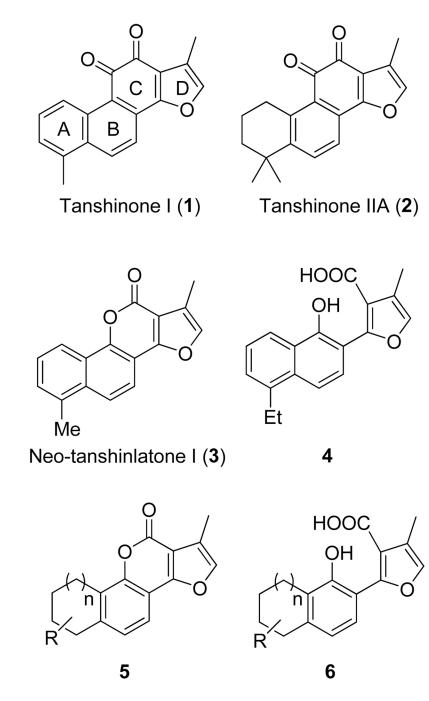
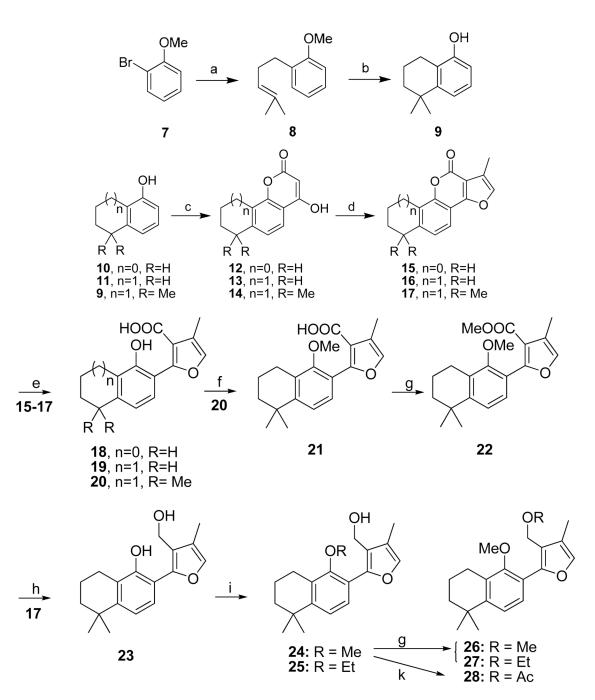


Figure 1.

Structures of tanshinone I (1), tanshinone IIA (2), neo-tanshinlactone (3), analog 4, and two newly designed scaffolds 5-6

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Scheme 1.

Reactions and conditions: (a) 4-methylpent-3-enylzinc(II) bromide, $Pd(Cl_2)(dppf)$, THF, reflux, 1h; (b) (i) AlCl₃, DCM, 0 °C, 15 min; (ii) BBr₃, CH_2Cl_2 ; (c) malonic acid, PPA, 75 ° C, 3 h; (d) chloroacetone, HOAc/NH₄OAc, toluene/EtOH, reflux, 24 h; (e) 5% NaOH (aq), reflux; (f) NaOH, 18-crown-6, MeI, CH₃CN, 90 °C; (g) SOCl₂, MeOH, rt; (h) LiAlH₄, THF; (i) MeI or EtI, Cs₂CO₃, acetone, 50 °C; (g) MeI or EtI, NaH, THF, rt; (k) Ac₂O, Et₃N, DMAP, CH₂Cl₂.

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Compd	SK-BR-3	ZR-75-1	MDA-MB-231	A549	DU145	KB	KB-vin
6	0.8	1.1	37.9	54.2	58.3	>37	>37
4	3.4	1.0	>37	35.8	29.4	30.7	23.6
15	>60	>37	>37	>37	>37	>37	>37
16	5.1	8.3	38.6	10.2	32.7	31.5	10.2
17	35.8	32.3	>37	28.0	31.9	30.5	34.4
18	30.2	30.2	44.6	41.5	38.4	31.4	30.2
19	32.4	21.7	73.5	57.0	40.4	50.0	29.4
20	0.7	1.7	24.7	3.3	3.0	4.7	4.7
21	25.5	41.4	>61	53.5	34.1	43.3	47.8
22	27.1	34.5	61.0	61.0	33.5	53.4	52.1
23	21.3	25.5	39.5	24.5	22.7	22.0	19.9
24	23.7	15.0	28.3	23.3	21.7	25.3	21.7
25	6.6	14.3	39.8	22.9	17.5	16.9	15.3
26	18.5	14.3	47.8	19.4	17.5	11.5	20.1
27	19.8	19.8	42.7	22.9	17.7	21.6	16.5
28	14.0	15.5	55.6	28.7	16.7	25.4	24.9