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Using Stable Isotopes to Determine Dominant Methane Production Pathways of Thaw Ponds in a Subarctic Peatland

By

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B.Sc. Environmental Science: Ecosystems and Sustainability, University of New Hampshire, 2019

THESIS

Submitted to the University of New Hampshire In Partial Fulfillment of The Requirements for the Degree of

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This thesis was examined and approved in partial fulfillment of the requirements for the degree of Master in Earth Science by:

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On November 13, 2020

Approval signatures are on file with the University of New Hampshire Graduate School.

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ABSTRACT

Arctic and subarctic ecosystems are currently warming faster than any other region of the globe, accelerating seasonal permafrost thaw. As thaw progresses, small water bodies can form due to slumping of the peatland surface. These ponds emit methane (CH₄), a strong, radiatively important trace gas, predominantly through ebullition (bubbling). Two different types of methanogenic Archaea present in these systems produce CH₄ through their respective production pathways: acetoclastic and hydrogenotrophic methanogenesis. The acetoclastic pathway forms CH₄ using CH₃COOH, an organic carbon (C) source while hydrogenotrophic methanogenesis uses CO₂, an inorganic C source. Stable isotopes can be used to characterize the relative contribution of these two pathways in overall CH₄ production and to better constrain the global CH₄ budget and improve modeling of future emission scenarios.

We used stable isotopes, carbon-13 (¹³C) of CH₄ and CO₂, deuterium (D) of CH₄, and calculated apparent fractionation factors to determine the relative contribution of acetoclastic versus hydrogenotrophic pathways of methanogenesis in thaw ponds in a subarctic peatland located in the discontinuous permafrost region of northern Sweden. Isotopic analysis was performed on porewater samples (n = 310) and gas captured from ebullition (n = 177). Samples were collected from nine ponds over seven years (2012 to 2019) during the ice-free months (June to September). We tested important physical attributes of the ponds that were related to their formation and CH₄ production pathways. Results indicated that δ^{13} C-CH₄ of ebullition (-86.3‰ to -49.2‰) and porewater (98.2‰ to -42.9‰) and the inferred contribution of hydrogenotrophic vs. acetoclastic methanogenesis differed significantly between certain ponds and pond types. Over the course of this study dissolved and ebullitive δ^{13} C-CH₄ remained relatively constant between years but varied significantly between months. Alternatively, δ D-CH₄ of ebullition (- 397.0‰ to -199.4‰) and porewater (-383.4‰ to -184.8‰) did not differ between sampling years or months.

Pond types that are partially thawed and have a lower daily CH₄ ebullitive flux appear to have a higher relative contribution of hydrogenotrophic methanogenesis while types that are partially or fully thawed and have a higher daily CH₄ ebullitive flux appear to have a higher contribution of acetoclastic methanogenesis relative to other pond types. Differences in CH₄ isotopic composition between pond types indicates that shifts in isotopic emissions could occur as thaw progresses in northern permafrost ecosystems. While we did observe expansion of ponds and landscape slumping at Stordalen Mire over our 7-year study, a strong trend in isotope signal was not observed likely due to the high interannual variability. This unique multi-year study characterized δ^{13} C-CH₄ and δ D-CH₄ values for ebullition and porewater from open-water thaw ponds providing valuable data to constrain the global CH₄ budget and improve modeling of the contribution of these systems to emissions now and in the future.

I. INTRODUCTION

Artic and sub-Arctic ecosystems are currently warming faster than any other region of the planet causing the thaw of permanently frozen soil, or permafrost. Permafrost thaw can manifest as ground slumping and collapse, forming ponds across the landscape that become important sources of methane (CH₄) to the atmosphere as methanogens access newly available C (Burke et al., 2019; Kuhn et al., 2018; Negandhi et al., 2013; Olsen et al., 2011). This production of CH₄, a radiatively important trace gas estimated to be 32 times stronger than carbon dioxide (CO₂) on a 100 year timescale (Holmes et al., 2013), contributes to a positive feedback as CH₄ emissions increase radiative forcing and climate warming that then accelerates permafrost thaw (Schuur et al., 2015; Turetsky et al., 2019).

Methanogenic Archaea are categorized as two different organisms, acetoclasts and hydrogenotrophs that produce CH₄ through unique production pathways: acetoclastic methanogenesis (AM; CH₃COOH \rightarrow CO₂ + CH₄) and hydrogenotrophic methanogenesis (HM; CO₂ + 4H₂ \rightarrow CH₄ + 2H₂O) respectively. Acetoclastic methanogenesis produces CH₄ using acetate (CH₃COOH), an organic C source while HM uses CO₂. The relative contribution of these two pathways in overall CH₄ production can be inferred using stable isotope analysis of carbon-13 (δ^{13} C-CH₄) and deuterium (δ D-CH₃D; Chanton et al., 2006; Chanton et al., 2005; Whiticar, 1999; Whiticar et al., 1986). In addition, the isotopic analysis of dissolved inorganic carbon (δ^{13} C-CO₂) can be interpreted to indicate the relative contribution of CH₄ production pathways via the calculation of apparent fractionation factor (α_C), a measure of the amount of isotopic separation between the source (CO₂) and produced (CH₄) carbon species (Chanton et al., 2005; Whiticar, 1999; Whiticar et al., 1986). Approximate isotopic values indicating contribution of AM are -40 to -60‰ for δ^{13} C and -250 to -400‰ for δ D while contribution of HM is indicated by -60 to -110‰ for δ^{13} C and -170 to -250‰ for δ D (Figure 1; Chanton et al., 2006; Hornibrook et al., 1997; Whiticar et al., 1986).

Determining the relative importance of methanogenesis pathways and constraining the isotopic composition of thaw pond CH₄ emissions is essential for improving calculations of the global CH₄ budget, The global budget of CH₄ is generally determined using either "bottom-up" methods that construct process-based models using field measurements or "top-down" methods that construct inverse models using atmospheric measurements, or a combination of the two. Currently, large discrepancies exist between these approaches with the bottom-up global CH₄ emission calculations almost 30% higher than global CH₄ emissions calculated with top-down methods (Saunois et al., 2020). Field-based isotope measurements are an essential tool to improve agreement between these two methods and provide more accurate estimates of CH₄ emissions from notable sources such as changing permafrost ecosystems (McCalley et al., 2014).

The relative contribution of methanogenic pathways is influenced by climate, hydrologic conditions, and vegetation composition (Berger et al., 2018; Chanton, 2005; Prater et al., 2007; Waldron et al., 1999). These properties change as thaw stage advances in northern peatlands, suggesting that change in the contribution of these two methanogenic pathways and a consequent shift in the emission of δ^{13} C-CH₄ occurs with thaw progression (McCalley et al., 2014). Thaw transitions from dry, elevated palsa plateaus to submerged bog, fen, and pond areas have been documented across the Arctic (Johansson et al., 2006; Johnston et al., 2014; Quinton et al., 2011; Vitt et al., 2000) and associated with differences in isotopic composition of CH₄ diffusive fluxes (McCalley et al., 2014). The presence of different vegetation species and CH₄ transport mechanisms occurring in a system also influence CH₄ fluxes and δ^{13} C-CH₄ signature. The three modes of transport for CH₄ gas flux are plant-mediated, bubbling or ebullition, and diffusion

(Chanton, 2005). Sedge plants common in northern wetlands, including *Carex spp.* and *Eriophorum spp.*, are associated with plant-mediated transport through the process of aerenchyma, acting like a straw to carry gas up through the plant stem and contributing to higher gas efflux (Chanton, 2005; Greenup et al., 2000; Noyce et al., 2014). Methane transported by aerenchyma experiences more fractionation due to oxidation in the rooting zone or rhizosphere. This fractionation enriches the δ^{13} C, making it less negative than the δ^{13} C unaffected by this process (Chanton, 2005). Ebullition is a more rapid process and does not result is isotopic fractionation, making it a reliable sampling method to infer sediment methane dynamics (Chanton & Martens, 1988) and is often the dominant method of transport from aquatic sediments (Bastviken et al., 2011; Walter et al., 2007). The production and presence of sedge plant species, especially *Carex spp.*, drives the relative contribution of acetoclastic versus hydrogenotrophic methanogenesis pathways towards higher contribution of acetoclastic methanogenesis (Bellisario et al., 1999; Prater et al., 2007).

Ebullitive emissions from northern lakes and ponds are a significant source of CH₄ to the atmosphere (Anthony et al., 2020; Burke et al., 2019; Wik et al., 2013) but are not often measured over extended periods of time (Wik et al., 2016). Measurements of the isotopic composition of these ebullitive emissions is even more rare, presenting a gap in scientific understanding of the interannual and seasonal variation of these emissions. Recent data suggests that the relative contribution of CH₄ production pathway does not shift seasonally as indicated by minimal change in δ^{13} C-CH₄ and changes in δ D-CH₄ that could be explained by the influence of evaporation documented on a monthly timescale from ebullitive flux measurements (Jansen et al., 2019; Wik et al., 2020). Consistent relative contribution of CH₄ production pathways across months has also been documented in diffusive flux and dissolved CH₄ measurements (McCalley

et al., 2014). Directional change across consecutive years in isotopic composition of CH_4 ebullitive or diffusive flux from thaw-formed lakes or ponds that would reflect a shift in contribution of methanogenesis pathways has not previously been documented (Negandhi et al., 2013; Townsend-Small et al., 2017). Over a period of three years, $\delta^{13}C$ and δD of dissolved methane ranges were found to narrow after a wetting event in sphagnum and sedge dominated wet areas in northeast Siberia suggesting that relative contribution of methanogenesis pathways could shift in response to changes is climatic conditions in permafrost wetlands (Shingubara et al., 2019). Currently, no studies exist that measure δ^{13} C of CH₄ flux from thaw-formed ponds for period of longer than three years (Blodau et al., 2008; Knoblauch et al., 2015; Nakagawa et al., 2002; Negandhi et al., 2013), making it difficult to determine whether contributions of methanogenesis pathways shifts within ponds in response to increased thaw. Changes in hydrology and vegetation that occur on decadal scales as a result of permafrost thaw (Johansson et al., 2006) could cause shifts in the relative contribution of methanogenesis pathways over a multi-year period in thaw ponds that has not yet been documented by previous short-lived studies.

Thaw ponds at Stordalen Mire in Abisko, Sweden have been categorized into four types based on differences in their daily ebullitive CH₄ flux. These types also vary in their physical characteristics and provide a spatial representation of varying thaw stage (Burke et al., 2019)Extensive pre-existing data from Burke et al., (2019) provides a unique opportunity to examine differences in isotopic composition of dissolved and emitted CH₄ that occur in different pond types and draw conclusions about the relationship between pond CH₄ flux, physical characteristics, and relative contribution of methanogenesis pathways.

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The objective of this study is to determine dominant methane production pathways of thaw ponds in a subarctic peatland complex over a multi-year period using stable isotopes (δ^{13} C-CH₄, δ D-CH₄, and δ^{13} C-CO₂). We measured δ^{13} C-CH₄ and δ D-CH₄ of ebullition and δ^{13} C-CH₄, δ D-CH₄, and δ^{13} C-CO₂ of pond sediment porewater from ponds over a period of seven years during the growing season. We hypothesize that (1) δ^{13} C-CH₄ and δ D-CH₄ will vary significantly between individual ponds and pond types established by Burke et al. (2019) indicating a difference in the relative contribution of CH₄ production pathways between pond types with more "bog-like" pond in Types 1-3 having more relative contribution of HM and "fen-like" Type 3 ponds having more relative contribution of AM, and (2) we expect to see no significant difference in isotopic signatures of δ^{13} C-CH₄ and δ D-CH₄ from ebullition and porewater on a monthly timescale or between years over the duration of this study (7 years) due to the lack of seasonal variation previously documented in isotopic composition of dissolved and emitted CH₄ and the decadal timescale of environmental changes in response to permafrost thaw.

II. METHODS

II. 1. SITE DESCRIPTION

Nine thaw ponds were sampled at Stordalen Mire (68°21'N 18°49'E), a peatland located in the discontinuous permafrost region in Abisko, Sweden (Figure 1). This sub-Arctic ecosystem has a mean annual temperature of 0.6°C and a mean annual precipitation of 300mm (Callaghan et al., 2010). Landcover types include permafrost palsas, *Sphagnum spp*. bogs, shallow lakes, fens, and thaw ponds. A large body of literature including decadal measurements of methane emissions and extensive unmanned aerial system imagery documenting vegetation changes at the Stordalen Mire make it an ideal location to study long-term changes (Johansson et al., 2006; Palace et al., 2018; Svensson et al., 1999).

The pond sites for this study were selected based on their boardwalk proximity and accessibility. The ebullitive CH₄ flux from each pond has been previously measured over a four year period (2012 to 2015) by Burke et al., (2019). Thorough descriptions of these ponds and associated flux measurements can be found in Burke et al., (2019) and are summarized here. The seven ponds included in the study were categorized into four types by significant differences in the ebullitive CH₄ flux that coincided with differences in their physical characteristics, including thaw stage. Type 1 ponds were the lowest emitting (median 0.0 mg CH₄ m⁻² day⁻¹), located on intact palsa plateaus, the shallowest ponds in the study, hydrologically isolated, and had a relatively low amount of sedge vegetation and high amounts of *Sphagnum spp*. present. Type 2 ponds emitted the next lowest amount of CH₄ (medians: 4.5 and 3.6 mg CH₄ m⁻² day⁻¹), were surrounded by a combination of intact and collapsing palsas, were deeper than Type 1 ponds, hydrologically isolated, and had increased presence of sedge vegetation relative to Type 1 and contained floating *Sphagnum spp.*. Type 3 ponds emitted the highest amount of CH₄ (medians: 53.4 and 40.9 mg CH₄ m⁻² day⁻¹), were surrounded by intact and collapsing palsas, were some of the deepest ponds in the study, hydrologically isolated, and had some floating Sphagnum spp. with sedge vegetation present along pond edges. Type 4 ponds emitted the second highest amount of CH₄ (medians: 6.4 and 11.7 mg CH₄ m⁻² day⁻¹), were connected to surrounding fullythawed fen areas, not hydrologically isolated, and had relatively high coverage of sedge vegetation and no floating Sphagnum spp. (Burke et al., 2019).

II. 2. EBULLITION SAMPLING

Ebullition (bubble) samples were collected from the eight of the nine study ponds during the growing season (May – September) from 2012-2019; no samples were collected in 2013 and 2015 (Table 1). Every 1-3 days, samples were collected from floating bubble traps to assess



Figure 1. Orthomosaic image of Stordalen Mire in northern Sweden ($68^{\circ}21'N$ 18°49′E) generated from images collected using an unmanned aerial system at an altitude of 70m on July 22, 2019 courtesy of Dr. Michael Palace. Nine ponds included in this study are labeled with corresponding capital letters. Blue polygons enclose approximate pond locations. Pond D is located out of frame to the northwest of the pictured area as indicated by a blue and black arrow. A yellow star (inset figure) indicates the general location of Stordalen Mire within Sweden. The site map and included polygons were rendered using QGIS (v.2.18.23).

isotopic composition and calculate daily ebullitive flux (Figure 2). Bubble traps were deployed using methodology described in detail by Burke et al. (2019) and were designed similarly to bubble traps used by Wik et al. (2013). Gas samples for isotopic composition were collected from bubble traps using 10-ml polypropylene syringes when the accumulated gas volume in a bubble trap was >5 ml. Samples were then transported to the Abisko Scientific Research Station (ANS) and refrigerated at 2°C for up to 24 hours prior to being stored in 15-ml evacuated glass vials with butyl rubber septa.

II. 3. POREWATER SAMPLING

Porewater dissolved gas samples were collected from eight of the nine ponds once monthly during the growing season adjacent to the bubble trap locations in each pond from 2012-2019 with no samples collected in 2013 and 2015 (Table 1). Samples were obtained using

suction with a syringe connected to a 60cm long stainless-steel tube (0.25inch outer diameter) with a perforated end inserted into the peat or sediment to desired depth (Figure 1). Samples (30 ml) were collected at 0, 10, 20, and 40cm depths below the pond bottom. The dissolved gas in the porewater was then equilibrated with 30 ml of room air by shaking the combined air and water for two minutes by



Figure 2. Aerial image of a thaw pond located at Stordalen Mire. (I) CH_4 ebullition samples are collected using a bubble trap floating on the water surface. (II) Dissolved gas porewater samples are collected adjacent to bubble traps to avoid artificial ebullition. A sipper (a) with a perforated end (b) is inserted to depths of 0, 10, 20, and 40cm below the pond bottom to collect 30-ml of water.

hand (McAuliffe, C., 1971). Equilibrated gas samples were then immediately stored in 30-ml evacuated glass vials with butyl rubber septa. Physical limitations such as low daily ebullitive flux and substrates with minimal pore space resulted in uneven sample sizes between individual

ponds (Table 1).

Porewater to be analyzed for δ^{13} C-CO₂ was collected simultaneously with dissolved gas

samples in 2014 and 2019 from eight of the nine ponds at depths of 0, 10, 20, and 40cm below

the pond bottom. 60-ml polypropylene syringes were used to collect and store 25-ml of

porewater that was transported to the ANS to be processed. Samples were stored in 30-ml

evacuated glass vials acidified with 0.5-ml of 21% H₃PO₄ and sealed with a butyl rubber stopper.

The headspace of the stored porewater samples collected in 2014 and 2019 was analyzed

for concentration and isotopic composition of CH4 and CO2 using a continuous-flow Hewlett-

Table 1. Nine thaw ponds at Stordalen Mire were sampled during the growing season over a period of seven years from 2012-2019. Not all ponds were sampled in all years or for both sample types due to field season logistics. Pond type classifications are based on daily ebullitive flux as described in Burke et al. (2019). Sample count is on parentheses next to sample type collected (n=).

Pond	Туре	Sample type collected	Years sampled
Α	1	bubble $(n = 11)$	2014, 2016, 2017, 2018, 2019
		porewater ($n = 56$)	
В	1	bubble $(n = 7)$	2014, 2017, 2018, 2019
		porewater ($n = 38$)	
С	2	bubble ($n = 26$)	2012, 2014, 2016, 2017, 2018, 2019
		porewater $(n = 45)$	
D	2	bubble ($n = 13$)	2014, 2016, 2017, 2018, 2019
		porewater ($n = 31$)	
E	3	bubble ($n = 29$)	2014, 2016, 2017, 2018, 2019
		porewater ($n = 72$)	
F	3	bubble ($n = 43$)	2014, 2016, 2017, 2018, 2019
		porewater ($n = 49$)	
G	4	bubble ($n = 13$)	2012
Н	4	bubble ($n = 35$)	2012, 2016, 2017, 2018, 2019
		porewater $(n = 8)$	
Х	4	porewater $(n = 11)$	2014

Packard 5890 gas chromatograph (Agilent Technologies) at 40°C coupled to a Finnigan MAT Delta isotope ratio mass spectrometer (GC-IRMS) via a Conflo IV interface system (Thermo Scientific) at Florida State University (FSU) as described by McCalley et al. (2014). This instrument was also used for all ebullition and porewater samples from 2012-2016. In summary, the headspace concentrations were converted to porewater concentrations based on their known extraction efficiencies, defined as the proportion of formerly dissolved gas in the headspace. The extraction efficiency for CO_2 relative to dissolved organic carbon (DIC) was based on the CO_2 extraction from dissolved bicarbonate standards (J. Chanton, personal communication).

II. 4. METHANE ISOTOPE SYSTEMATICS

Isotopic compositions of CH₄ and CO₂ were quantified using the standard delta (δ) notation. This reflects the ratio (*R*) of ¹³C to ¹²C or D to H in the sample as a relative difference (denoted as δ^{13} C-CH₄ and δ D-CH₄) from the Vienna Pee Dee Belemnite (VPDB) and Standard Mean Ocean Water (SMOW) international standard materials respectively, expressed in per mil (‰) values. An example for C is as follows:

Eq 1.
$$\delta^{13}C = \frac{R - R_{VPDB}}{R_{VPDB}} = \frac{R}{R_{VPDB}} - 1$$

The apparent fractionation factor $CO_2 \rightarrow CH_4$ (α_C) was calculated as:

Eq 2.
$$\alpha_C = \frac{\delta^{13}C - CO_2 + 1000}{\delta^{13}C - CH_4 + 1000}$$

Apparent fractionation factor can be interpreted to represent differences in the relative importance of CH₄ production mechanisms with higher α_C indicative of HM and lower α_C indicative of AM (Hornibrook et al., 1997; Michael J. Whiticar, 1999; M.J Whiticar et al., 1986). Fractionation factors can be difficult to estimate in field studies because while CO₂ is the precursor for CH₄ produced via the hydrogenotrophic pathway, it is not the immediate precursor for the acetoclastic pathway. However, porewater CO₂ is primarily from respiration of organic matter and is isotopically indistinguishable from organic matter enabling a representative fractionation factor to be determined for AM as well (Conrad, 2005; Corbett et al., 2013; Whiticar, 1999). Therefore, dominant CH₄ production mechanisms can be determined by the apparent fractionation factor of a sample.

II. 5. TUNABLE INFRADER LASER DIRECT ABSORPTION SPECTROMETER (TILDAS) MEASUREMENT AND CALIBRATION

Isotopic compositions of CH₄ were measured using a tunable infrared laser direct absorption spectrometer (TILDAS; Aerodyne Research Inc.) located at the Trace Gas Biogeochemistry Laboratory at the University of New Hampshire (UNH) via direct injections. The TILDAS is a modification of the quantum cascade laser spectrometer instruments described by McCalley et al. (2014) and Santoni et al. (2012) and instead allows for manual injection of the sample into the instrument. The TILDAS uses a room-temperature continuous-wave midinfrared laser whose frequency was tuned to scan rapidly at 900 kHz across ¹²CH₄ and ¹³CH₄ absorption lines in the 7.5-µm region. The laser light enters a multipass sample cell (effective path length about 200m) containing sample air at low pressure (~5kPa) and is detected by a thermoelectrically cooled detector. The associated TDL-Wintel (Aerodyne Research, Inc.) sampling system software was also used for the analysis. The TDL Wintel program averages high-frequency spectra to produce independent ¹²CH₄ and ¹³CH₄ mixing ratios in the sample airstream at 1 second intervals, allowing the ratio of ¹³C-CH₄ to ¹²C-CH₄ or D-CH₄ to H-CH₄ to be denoted in standard δ notation per mil (‰) as described previously (Equation 1). To keep a reasonable concentration regime for infrared spectroscopic analysis a target mixing ratio of 8ppm was used. Therefore, all samples are measured at 8ppm with dilution, or at the original concentration if it is already <8ppm. Samples with CH₄ mixing ratios greater than 8ppm are appropriately diluted with zero air via the sampling program. In some cases, the high concentration of ebullition samples required then to be diluted by hand prior to dilution by the sampling program. All ebullition samples were diluted at a ratio of 1ml sample:9 ml room air, as the sample concentration was not always known ahead of time, prior to injection for analysis. If the target mixing ratio (8ppm) was not able to be reached, samples were further diluted by hand and re-run.

Isometric tanks, standards	δ ¹³ C (‰) ±0.2‰	δD (‰)
B-isol	-54.5	-266
H-isol	-23.9	-171
L-isol	-66.5	-156
T-isol	-38.3	-157

Table 2. Isometric standards used to with an Aerodyne calibration system to calibrate three standard tanks at UNH.

Three working standard calibration tanks were used with the TILDAS at UNH with methane mixing ratios of approximately 2, 5, and 10ppm CH₄. These tanks were calibrated for δ^{13} C-CH₄ and δ D-CH₄ using an Aerodyne calibration system containing four Isometric methane isotope standards, each with 2500ppm CH₄. These four primary tanks were calibrated to the VPDM scale via flask samples that were also analyzed for δ^{13} C by GC-IRMS at FSU, allowing comparable measurements to be made between the two methods. The isotopic methane ratios of each standard are given in Table 2. The δ D are considered preliminary calibrations. The precision for δ D measurement is ~3‰.

The TILDAS and associated sample analysis system also includes an Alicat flow controller (Alicat Scientific) for each of the three standard tanks, an Alicat on an ultra zero air (dilution flow), a computer with an Igor-based program for calibration control and analysis, RS232 communication between the computer and an Alicat control box, and other valves for automated operation. An Igor-based program selects and sends each standard to the instrument at a controlled flow rate along with the dilution flow for a given period. We selected flows to generate CH₄ levels from approximately 12 ppm to <1ppm via a variable flow rate, and thus span the range of CH₄ levels in this study with manual dilutions. A Keeling plot analysis method was used to determine the relationship between the spectroscopic and the standard isotopic ratios. This linear relationship was then applied to the measured spectroscopic isotopic ratios of the UNH calibration tanks. A fourth tank containing 5% methane was also calibrated (Table 3).

UNH standard	δ ¹³ C (‰)	δD (‰)
tanks		
Low cal tank	-49.25	5.82
(~2ppm)		
Mid cal tank	-45.61	-160.22
(~5ppm)		
High cal	-44.19	-143.93
tank(~9ppm)		
5% CH ₄ tank	-37.67	-149.30

Table 3. δ^{13} C (‰) and δ D (‰) of four standard tanks at UNH used with TILDAS sample analysis.

II. 6. STATISTICAL ANALYSES

Univariate statistical analyses were performed using the R Project for Statistical Computing (v. 3.5.3) software and the JMP Pro 15 software. Significance of all statistical analyses were determined at the $\alpha = 0.05$ level. Isotopic signature for δ^{13} C-CH₄ and δ^{13} D-CH₄ data was checked for normality using the Anderson-Darling test for normality. Differences in isotopic composition between ponds, pond types, and sample types were assessed using one-way analysis of variance (ANOVA) and the Tukey-Kramer honestly significant different (HSD) test for normally distributed data. Differences in isotopic composition between sampling months and years were assessed using one-way ANOVA and the Tukey-Kramer HSD test for normally distributed data and the Kruskal-Wallis rank sum test and the Dunn's test were used for nonnormal data. A linear regression analysis was used to compare instrumentation methodologies for determining δ^{13} C.

III. RESULTS

III. 1. ISOTOPIC COMPOSITION OF STABLE ISOTOPES OF CH₄ AND CO₂

Our multi-year dataset of stable isotopes of CH₄ ebullition and porewater from eight subarctic thaw ponds shows variation in isotopic signature among individual ponds and over time on both monthly and interannual timescales. Overall, the δ^{13} C-CH₄ of ebullition and porewater during the growing season from 2012-2019 ranged from -86.3‰ to -49.2‰ and 98.2‰ to -42.9‰, respectively (Figure 3a, 3b). Porewater δ^{13} C-CH₄ data was normally distributed while ebullition data was not (Table 4). The mean and standard deviation of ebullition δ^{13} C-CH₄ was -66.7 ± 8.6‰ (n=177) and -74.9 ± 8.8‰ (n=313) for porewater δ^{13} C-CH₄ (Table 5). When compared between individual ponds using a one-way ANOVA, mean δ^{13} C-CH₄ of ebullition fell into three significantly different groups (Figure 3b). Pond G ebullition, a Type 4 pond that is fully thawed with high relative amounts of sedge vegetation, was the most enriched and was significantly different from all other ponds. Ponds H and F were more depleted than G and were not significantly different from each other. Ponds C and E were the most depleted and were not significantly different from one another or from ponds A and B. Pond D was not significantly different from ponds A, B, H, or F (Tukey HSD, P < 0.001; Figure 3b). Mean δ^{13} C-CH₄ of porewater was also compared using a one-way ANOVA that resulted in three significantly different groups (Figure 3a). However, the ponds did not fall into the same groupings as were



Figure 3. Porewater (a) and ebullition (b) δ^{13} C-CH₄ from 2012-2019 growing seasons grouped by pond. Whiskers represent the 10th and 90th percentiles, and colors indicate different ponds. Black diamonds and horizontal black lines denote means and medians respectively. Lowercase letters along the top of each plot correspond to statistically significant groupings based on a Tukey HSD post-hoc test (*P* < 0.001). Colored points represent individual data points. The number of samples (*n*) in each group is noted along the bottom. A horizontal dotted black line at -60‰ indicates a transition from greater relative contribution of acetoclastic (-40 to -60‰) to hydrogenotrophic (-60 to -110‰) methanogenesis (Whiticar, 1999). Ponds A and B are Type 1, ponds C and D and Type 2, E and F are Type 3, G, H, and I are Type 4.

seen in ebullition data. Ponds F and H, Type 3 and 4 ponds respectively that are fully thawed with high relative amounts of sedge vegetation, were not significantly different from each other



Figure 4. Box plots of porewater (left) and ebullition (right) δ^{13} C-CH₄ from 2012-2019 sampling years grouped by pond Type. Whiskers represent the 10th and 90th percentiles and black dots represent outliers. Colors indicate different pond Types. Black diamonds and horizontal black lines denote means and medians respectively. Uppercase letters along the top correspond to statistically significant groupings based on a Tukey HSD post-hoc test (*p* < 0.05). The number of samples (*n*) in each group is noted along the bottom. A horizontal dotted black line at -60‰ indicates a transition from greater relative contribution of acetoclastic (-40 to -60‰) to hydrogenotrophic (-60 to -110‰) methanogenesis (Whiticar, 1999).

and were the most enriched. Ponds E, A, D, and C were not significantly different from each other and were more depleted in δ^{13} C-CH₄. Pond I was not significantly different from ponds H, F, E, A, D, or C. Pond B was significantly different from all other ponds and had the most depleted mean δ^{13} C-CH₄ of porewater (Tukey HSD, *P* < 0.001; Figure 2). When grouped by pond Type, δ^{13} C-CH₄ of porewater and ebullition decrease and become more enriched from pond Type 1 to pond Type 4 (Figure 4). However, pond Type 4 had the only significantly different mean δ^{13} C-CH₄ of ebullition data and pond Type 1 had the only significantly different mean δ^{13} C-CH₄ of porewater (Figure 4; *P* < 0.05). In general, δ^{13} C-CH₄ of ebullition and porewater (Table 4). However, mean δ^{13} C-CH₄ of ebullition and porewater from the same pond with all years combined was not significantly different, except for pond B (Tukey HSD, *P* < 0.001, Figure 3). This was also true when grouped by pond Type, except for pond Type 4 (Tukey HSD, *P* < 0.001, Figure 4).

We examined temporal trends of δ^{13} C-CH₄ between ponds and found that on a monthly timescale porewater δ^{13} C-CH₄ is significantly more depleted in September than in June (Figure 5a) while ebullition follows an opposing trend with δ^{13} C-CH₄ significantly more enriched late in the summer in August than in June (Figure 5b). Pairwise comparisons using the Dunn's test revealed that Pond F was the only pond with significant difference in δ^{13} C-CH₄ between months for both ebullition and porewater (Table 7, 9, S2, S4). Interannually, porewater δ^{13} C-CH₄ is relatively stable as five out of six sampling years are not significantly different (Figure 5c). The 2016 sampling season is significantly more enriched than other sampling years, except for 2012 (Figure 5c). Ebullition δ^{13} C-CH₄ becomes more depleted from 2012 to 2019. Ebullition δ^{13} C-CH₄ in 2012 is significantly more enriched than data from the 2017 and 2019 sampling seasons but is not significantly different from 2016 and 2018 (Figure 5d). Pairwise comparison using the Dunn's test determined that Ponds E and H had significant differences in ebullition δ^{13} C-CH₄ was significantly different between sampling years within ponds E, F and H (Table 6, S1) and ebullition δ^{13} C-CH₄ was significantly different between sampling months within ponds A, C, D, E, and F (Tables 8, S3). When grouped by pond type, ebullition and porewater δ^{13} C-CH₄ of Type 3 and 4 ponds were generally more enriched across sampling years while Type 1 and 2 ponds were relatively depleted (Figure 6). Both porewater and ebullition monthly mean δ^{13} C-CH₄ was generally more enriched for all pond types in 2016, as seen in the porewater data (Figure 5c, 6).



Figure 5. Porewater (a, c) and ebullition (b, d) δ^{13} C-CH₄ from all ponds grouped by month (a, b) and by year (c, d). Whiskers represent the 10th and 90th percentiles and black dots represent outliers. Black diamonds and horizontal black lines denote means and medians respectively. The number of samples (*n*) in each group is noted along the bottom. A horizontal dotted black line at -60‰ denotes a transition from greater relative contribution of acetoclastic (-40 to - 60‰) to hydrogenotrophic (-60 to -110‰) methanogenesis (Whiticar, 1999). 5a and 5c: Mean δ^{13} C-CH₄ was compared using a one-way ANOVA and a Tukey HSD post-hoc test for data that was normally distributed. Uppercase letters represent significantly different groupings of sampling seasons or months as determined by an all pairs Tukey-Kramer HSD test. Significant p values are marked by *. 5b and 5d: The Kruskal-Wallis rank sum test was used to assess data that did not meet the assumptions for a one-way ANOVA and is plotted as χ^2 and *p*. Uppercase letters indicate pairwise differences between sampling seasons or months as determined by a Dunn's test following the Bonferroni method ($\alpha = 0.05$).



Figure 6. Porewater and ebullition δ^{13} C-CH₄ from 2012-2019 sampling years grouped by pond Type and sample type. Months along the lower x-axis indicate the month of sample collection while years along the upper x-axis indicate year of sample collection. Colors indicate different pond Types while circles and triangles indicate ebullition (bubble) and porewater samples respectively. Points are averages of ≥ 2 data points. A horizontal dotted black line at -60‰ indicates a transition from relative contribution of acetoclastic (-40 to -60‰) to hydrogenotrophic (-60 to -110‰) methanogenesis (Whiticar, 1999). Across and within years δ^{13} C-CH₄ remains relatively constant within pond and sample type.

The δ D-CH₄ of bubbles and porewater ranged from -397‰ to -199‰ and from -383‰ to - 184‰, respectively. Ebullition δ D-CH₄ data was normally distributed while porewater data was not (Table 4). The mean and standard deviation of bubble δ D-CH₄ was -310 ± 44‰ and -315 ± 36‰ for porewater δ D-CH₄ (Table 5). Mean δ D-CH₄ of ebullition was not statistically different

between individual ponds or when grouped by pond Type (Tukey HSD, P = 0.654 and 0.354).

Mean δ D-CH₄ of porewater was also not statistically different between individual ponds or when grouped by pond Type (*Tukey HSD*; *P* =0.867 and 0.958). Mean δ D-CH₄ was also not found to be significantly different between ebullition and porewater samples from the same pond or pond Type (Tukey HSD, *P* = 0.815 and 0.637).

When examining temporal trends in δ D-CH₄ we found neither porewater nor ebullition δ D-CH₄ was significantly different between sampling months (Figure 7a, 7b). Ebullition δ D-CH₄ was also not significantly different between sampling years (Figure 7d), while porewater δ D-CH₄ became more depleted from 2017 to 2019 (Figure 7c). 2017 was significantly more enriched that δ D-CH₄ porewater data from the 2019 sampling season (Figure 7c).

Table 4. Results of the Anderson-Darling test for normality on δ^{13} C-CH₄ and δ D-CH₄ porewater and ebullition data. The hypothesis of normality is rejected for p-values less than or equal to 0.05. Significant p-values indicating a rejection of the hypothesis of normality are noted with *.

Data type	Sample type	\mathbf{A}^2	р
δ ¹³ C-CH ₄	porewater	0.517	0.165
δ^{13} C-CH ₄	ebullition	0.914	0.022*
δD-CH ₄	porewater	1.669	>0.0001*
δD-CH ₄	ebullition	0.341	0.484

We examined isotopic composition of CH₄ (δ D-CH₄ and δ ¹³C-CH₄) in years 2017-2019 and found that ebullition data from pond Types 1 and 2 had more enriched mean δ D-CH₄ and δ ¹³C-CH₄ than porewater from the same pond types (Figure 8). Ebullition data from Type 3 was more enriched in δ ¹³C-CH₄ compared to porewater from the same pond but did the δ D-CH₄ did not differ. Ebullition data from Type 4 had more depleted mean δ D-CH₄ and δ ¹³C-CH₄ than



Figure 7. Box plots of porewater (a, c) and ebullition (b, d) δD -CH₄ from all ponds grouped by month (a, b) and by year (c, d). Whiskers represent the 10th and 90th percentiles and black dots represent outliers. Black diamonds and horizontal black lines denote means and medians respectively. The number of samples (*n*) in each group is noted along the bottom. A horizontal dotted black line at -250‰ denotes a transition from greater relative contribution of hydrogenotrophic (-170to -250‰) to acetoclastic (-250 to -400‰) methanogenesis (Whiticar, 1999). Significant p values are marked by *. 7a and 7c: The Kruskal-Wallis rank sum test was used to assess data that did not meet the assumptions for a one-way ANOVA and is plotted as χ^2 and *p*. Uppercase letters indicate pairwise differences between sampling seasons or months as determined by a Dunn's test following the Bonferroni method ($\alpha = 0.05$). 7b and 7d: Mean δ^{13} C-CH₄ was compared using a one-way ANOVA and a Tukey HSD post-hoc test for data that was normally distributed. Uppercase letters represent significantly different groupings of sampling seasons or months as determined by an all pairs Tukey-Kramer HSD test.

porewater from the same pond type. Mean δ^{13} C-CH₄ of porewater became increasingly more enriched moving from Type 1 to Type 4 however, mean δ D-CH₄ of porewater varied minimally between pond Types. Mean δ D-CH₄ of ebullition was the most enriched in pond Type 2 and the most depleted in pond type 4. Mean δ^{13} C-CH₄ of ebullition was most depleted in Type 1 and most enriched in Type 4 (Figure 8). We compared isotopic composition of dissolved CH₄ in porewater by depth from 2017-2019 and found pond type to be more indicative of isotopic composition than depth (Figure 9). Mean δ^{13} C-CH₄ from all depths of 0 to 40cm was most depleted in Type 1 and most enriched in Type 4. Mean δ D-CH₄ fell within a narrow range of -336 to -298‰ across depths and pond types (Figure 9).

In addition to δ^{13} C-CH₄, the δ^{13} C-CO₂ of pond porewater was measured in 2014 and 2019. Porewater δ^{13} C-CO₂ when years were grouped together ranged from -22.2 to 5.6‰ with a mean and standard deviation of -10.4 ± 6.5‰ (n = 166) and differed significantly between pond Types (Table 5, Figure 10a). Pond Type 3 was the most enriched and varied significantly from pond Types 1, 2 which were more depleted. Pond Type 4 was not significantly different from any other pond Type (Tukey HSD, P < 0.001; Table 5; Figure 10a).

Table 5. Mean ebullition and porewater δ^{13} C-CH₄, δ D-CH₄, δ^{13} C-DIC, and fractionation factor (α) from four pond Type classifications. Mean δ^{13} C-CH₄ contains data from all sampling years. Mean δ D-CH₄ contains data from 2017-2019 sampling seasons. Mean δ^{13} C-DIC and α are from porewater samples collected during the 2014 and 2019 sampling seasons. from 2014 and 2019.

Pond Type	Ebullition δ ¹³ C-CH ₄ (‰)	Porewater δ^{13} C-CH ₄ (‰)	Ebullition δD-CH4 (‰)	Porewater δD-CH ₄ (‰)	Porewater δ^{13} C-CO ₂ (‰)	Porewater $\alpha_{\rm C}$
1	-71.37 ± 7.18	-79.92 ± 8.58	-294 ± 41	-283 ± 27	-10.67 ± 7.16	1.073 ± 0.0057
2	-71.29 ± 7.29	-76.52 ± 6.79	-315 ± 40	-317 ± 66	-13.34 ± 5.22	1.069 ± 0.0060
3	-67.78 ± 7.60	-71.01 ± 8.16	-315 ± 34	-317 ± 39	-7.11 ± 5.28	1.066 ± 0.0084
4	-59.74 ± 7.25	-69.28 ± 4.78	-313 ± 35	-309 ± 54	-9.33 ± 5.11	1.067 ± 0.0072

The apparent fractionation factor ($\alpha_{\rm C}$) calculated from porewater δ^{13} C-CO₂ and δ^{13} C-CH₄ ranged from 1.046 to 1.083 with a mean and standard deviation of 1.07 ± 0.01 (n = 162; Table 5,

Figure 10b). Mean $\alpha_{\rm C}$ of porewater varied significantly between pond Type with years grouped together when compared using a one-way ANOVA. Pond Type 1 had the highest mean α and was significantly different from pond Types 3 and 4. Pond Types 3 and 4 had lower mean α and were not significantly different from each other. Pond Type 2 was not significantly different from pond Types 1, 3, or 4 (Figure 10b; Table 1; *P* < 0.001). Mean α did not differ between sampling years of 2014 and 2019, except for pond Type 4 which had a lower mean α in 2019 than in 2014 (Figure 11). Pond Type 1 porewater in 2019 had the most depleted mean δ^{13} C-CH₄ while pond Type 4 was the most enriched. Porewater from pond Type 1 in 2019 has the highest mean α value while pond Type 4 in 2019 has the lowest. Mean α of pond Types 3 and 4 both decrease between 2014 and 2019 while pond Type 2 does not (Figure 11).

III. 2. METHOD COMPARISON

Porewater δ^{13} C-CH₄ samples were analyzed using either a QCLS or a GC-IRMS. A linear regression analysis determined a positive correlation between δ^{13} C-CH₄ from the two sample methods. The linear model y = 0.84x – 13.9 fit 80% of the samples (*P* < 2.2e-16, *R*² = 0.80; Figure 12). The QCL and GC-IRMS methods yielded similar results for data from the 2019 growing season, allowing data from sampling years and methods to be combined to make interannual and conclusions.



Figure 8. Average porewater and ebullition δ D-CH₄ and δ^{13} C-CH₄ from 2017-2019 sampling years grouped by pond type and sample type. Different pond types are indicated in distinctive colors, and circles and triangles represent ebullition (bubble) and porewater samples respectively. Light gray lines denote standard deviation around the average values found for each pond type. Gray arrows indicate greater contributions of the denoted methanogenesis pathways or oxidation (Chanton et al., 2005).



Figure 9. Average porewater δ D-CH₄ and δ^{13} C-CH₄ from 2017-2019 sampling years grouped by pond type and depth. Different pond types are indicated in distinctive colors, and shapes denote different sampling depths. Standard deviation is depicted with light gray around the average values found in the pond type. Pond type is a stronger indicate of isotopic composition of δ^{13} C -CH₄ than depth as pond types fall into distinct groups but depth increments do not. Gray arrows indicate greater contributions of the denoted methanogenesis pathways (Chanton et al., (2005).



Figure 10. Porewater (a) δ^{13} C-CO₂ and (b) fractionation factor (α) from 2014 and 2019 sampling years grouped by pond type. Whiskers represent the 10th and 90th percentiles and black dots represent outliers. Colors indicate different pond types. Black diamonds and horizontal black lines denote means and medians respectively. Uppercase letters along the top correspond to statistically significant groupings based on a Tukey HSD post-hoc test (p < 0.001). The number of samples (n) in each group is noted along the bottom.



Figure 11. Average porewater δ C-CO₂ and δ^{13} C-CH₄ from 2014 and 2019 sampling years grouped by pond type. Standard deviation is depicted with light gray around the average values found for each pond type. Pond types are indicated in distinctive colors, and shapes denote different sampling years. Black lines mark slope of differing fractionation factor (α_c) at 0.01 intervals.



Figure 12. Porewater δ^{13} C-CH₄ from the 2019 sampling season analyzed using both a QCLS and a **GC-IRMS**. Circles represent individual data points. A black diagonal line represents the 1:1 line. The linear regression model with had an equation of y = 0.84x -13.9, *P-value* < 2.2*e*-16, and an R^2 value of 0.80 determining that 80% of the data fit the linear regression model, determining that data analyzed with the two methods is comparable.

Table 6. Comparison of ebullition δ^{13} C-CH₄ by year for each pond using the Kruskal-Wallis rank sum test. *P* values less than or equal to 0.05 indicate significant differences between sampling years within a pond and are marked with a *. See table S1 for corresponding pairwise comparisons.

2		
χ^2	DF	р
3.1313	2	0.209
NA	NA	NA
4.5691	4	0.334
3.5027	2	0.173
9.6903	3	0.0214*
9.0192	3	0.0290*
NA	NA	NA
17.9995	4	0.0012*
NA	NA	NA
	x ² 3.1313 NA 4.5691 3.5027 9.6903 9.0192 NA 17.9995 NA	χ² DF 3.1313 2 NA NA 4.5691 4 3.5027 2 9.6903 3 9.0192 3 NA NA 17.9995 4 NA NA

Ebullition δ^{13} C-CH₄ by year

Table 7. Comparison of ebullition δ^{13} C-CH₄ by month for each pond using the Kruskal-Wallis rank sum test. Comparison with 1 degree of freedom (DF) were conducted using the Wilcoxon rank sum test. *P* values less than or equal to 0.05 indicate significant differences between sampling months within a pond and are marked with a *. See table S2 for corresponding pairwise comparisons.

			nonun
Site	χ^2	DF	р
Pond A	3.4545	2	0.1778
Pond B	3.1429	2	0.2077
Pond C	0.0043	2	0.9979
Pond D	3.0945	2	0.2128
Pond E	5.3891	2	0.0676
Pond F	6.0929	2	0.0475*
Pond G	0.7143	1	0.398
Pond H	5.0238	2	0.0811
Pond I	NA	NA	NA

Ebullition $\delta^{13}\text{C-CH}_4$ by month

Table 8. Comparison of porewater δ^{13} C-CH₄ by year for each pond using a one-way ANOVA. *P* values less than or equal to 0.05 indicate significant differences between sampling years within a pond and are marked with a *. See table S3 for corresponding pairwise comparisons.

Site	DF	р
Pond A	4	< 0.0001*
Pond B	3	0.3554
Pond C	4	0.0294*
Pond D	4	0.0199*
Pond E	4	< 0.0001*
Pond F	4	< 0.0001*
Pond G	NA	NA
Pond H	2	0.571
Pond I	NA	NA

Porewater δ^{13} C-CH₄ by year

Table 9. Comparison of porewater δ^{13} C-CH₄ by month for each pond using a one-way ANOVA. When degrees of freed (DF) was equal to one a pooled t-test was used. *P* values less than or equal to 0.05 indicate significant differences between sampling months within a pond and are marked with a *. See table S4 for corresponding pairwise comparisons.

Porewater δ^{13} C-CH₄ by month

Site	DF	р
Pond A	3	0.0916
Pond B	3	0.183
Pond C	3	0.0899
Pond D	3	0.382
Pond E	3	0.2778
Pond F	3	0.0256*
Pond G	NA	NA
Pond H	2	0.7216
Pond I	1	0.7241

Table 10. Comparison of porewater δD -CH₄ by month for each pond using the Kruskal-Wallis rank sum test. Comparison with 1 degree of freedom (DF) were conducted using the Wilcoxon rank sum test. *P* values less than or equal to 0.05 indicate significant differences between sampling months within a pond and are marked with a *.

Site	χ^2	DF	р
Pond A	0.9001	2	0.6376
Pond B	0.8549	2	0.6522
Pond C	1.4433	2	0.486
Pond D	0.4558	2	0.7962
Pond E	2.9655	2	0.227
Pond F	5.4652	2	0.0651
Pond G	NA	NA	NA
Pond H	0.2	1	0.6547
Pond I	NA	NA	NA

Porewater **\delta**D-CH₄ by month

Table 11. Comparison of porewater δ D-CH₄ by year for each pond using the Kruskal-Wallis rank sum test. Comparison with 1 degree of freedom (DF) were conducted using the Wilcoxon rank sum test. *P* values less than or equal to 0.05 indicate significant differences between sampling months within a pond and are marked with a *.

Site	χ^2	DF	р
Pond A	1.5103	2	0.4699
Pond B	4.5048	2	0.1051
Pond C	1.7888	2	0.4088
Pond D	1.0911	2	0.5795
Pond E	1.2511	2	0.535
Pond F	2.0595	2	0.3571
Pond G	NA	NA	NA
Pond H	2.4	1	0.1213
Pond I	NA	NA	NA

Porewater δD -CH₄ by year

Table 12. Comparison of ebullition δ D-CH₄ by month for each pond using a one-way ANOVA. When degrees of freedom (DF) was equal to 1 a pooled t-test was used. *P* values less than or equal to 0.05 indicate significant differences between sampling years within a pond and are marked with a *.

		v
Site	DF	р
Pond A	1	0.8838
Pond B	NA	NA
Pond C	NA	NA
Pond D	1	0.8862
Pond E	2	0.2004
Pond F	2	0.3092
Pond G	NA	NA
Pond H	1	0.1223
Pond I	NA	NA

Ebullition δD-CH₄ by month

Table 13. Comparison of ebullition δ D-CH₄ by year for each pond using a one-way ANOVA. When degrees of freedom (DF) was equal to 1 a pooled t-test was used. *P* values less than or equal to 0.05 indicate significant differences between sampling years within a pond and are marked with a *.

Ebullition δD-CH₄ by year

Site	DF	р
Pond A	1	0.9606
Pond B	NA	NA
Pond C	NA	NA
Pond D	NA	NA
Pond E	2	0.5046
Pond F	2	0.0486*
Pond G	NA	NA
Pond H	2	0.5113
Pond I	NA	NA

IV. DISCUSSION

IV. 1. CONTRIBUTION OF METHANOGENESIS PATHWAYS VARY BY POND TYPE

The isotopic composition of ebullitive and dissolved CH₄ from nine thaw ponds over a seven-year period indicate that relative contribution of methanogenesis pathway differs between ponds and pond types. Ponds A and B, categorized as Type 1, and ponds C and D, categorized as Type 2 have relatively low δ^{13} C-CH₄ (Table 5, Figure 3, 4)and high α_C (Table 5, Figure 11) suggesting that HM contributes more to production than AM in Type 1 and 2 ponds. Methane produced by relatively higher contribution of HM has been observed to be depleted in the range of -60 to -110‰ for δ^{13} C-CH₄ and enriched in δ D resulting in values ranging from -170 to -250‰ (Chanton et al., 2005; Whiticar, 1999; Whiticar et al., 1986). However, δD-CH₄ from Type 1 ebullition (-219 ± 45 to $-337\pm47\%$) and porewater (-190 ± 18 to $-383\pm26\%$) falls within a range generally observed with higher relative contribution of AM. The same is also true for Type $2 \delta D$ -CH₄ of ebullition (-249±15 to -324±31‰) and porewater (-184±29 to -365±25‰; Table 5). Methane produced by higher relative contribution of AM, has been observed to enrich in δ^{13} C and deplete in δD , ranging from -40 to -60‰ and -250 to -400‰, respectively (Chanton et al., 2005; M.J Whiticar et al., 1986). The mean δ D-CH₄ of pond Type 1 and Type 2 are similar to values found by Jansen et al. (2019) in post-glacial lake sediment gas pockets (-308±43‰) and to ebullition measurements (- $310.6 \pm 21.3\%$) from the same lakes by Wik et al. (2020) during the summer months from three lakes located at Stordalen Mire suggesting that these depleted δD values are likely driven by δD -H₂O of source H₂O, which has been documented to heavily influence δD-CH₄, rather than by production effects (Chanton et al., 2006; Waldron et al., 1999). HM having higher relative contribution to production in Types 1 and 2 is further supported by

their respective mean α_C (Table 1). Larger α_c values are observed when HM contributes more to production relative to AM while lower α_c indicates higher relative contribution of AM (Chanton et al., 2005; Michael J. Whiticar, 1999; M.J Whiticar et al., 1986). Fractionation factor (α_C) decreases from Type 1 to Type 4 indicating a transition between pond types from higher relative contribution of HM to higher contribution of AM (Figure 10b, Figure 11). Type 1 ponds are hydrologically isolated, the shallowest ponds in this study, and located on palsa plateaus with a low amount of sedge vegetation relative to the other pond types. Type 2 ponds have more sedge vegetation than Type 1 but less than Types 3 and 4, are hydrologically isolated with the exception of Pond D, and are bordered by collapsing palsas (Burke et al., 2019). The presence of hydrogenotrophic methanogenesis in Types 1 and 2 implies the presence of decomposed organic matter and low redox potential (Eh) (Bouchard et al., 2015; Walter et al., 2008). The presence of intact palsas containing decomposed organic matter in the form of peat around Types 1 and 2 supports that these ponds would have a higher relative contribution of hydrogenotrophic methanogenesis than ponds with less access to this form of organic matter. Hydrogenotrophic methanogenesis contributes more to production in pond Types 1 and 2 but will have a low influence on ecosystem scale isofluxes as Types 1 and 2 have a relatively low associated daily ebullitive fluxes (Burke et al., 2019)

While still within ranges observed with higher relative contribution of HM, Types 1 and 2 have ebullition signatures that are enriched in both δ^{13} C-CH₄ and δ D-CH₄ in comparison to the associated porewater. This is likely a result of CH₄ oxidation driving enrichment of residual CH₄ in both δ^{13} C and δ D (Chanton et al., 2005). Oxidation has been documented to influence isotopic composition of CH₄ from ebullition of nearby post-glacial lakes at Stordalen Mire (Wik et al., 2020). Methane oxidation in the water column can shift δ^{13} C-CH₄ up to +20‰ between sediment

bubble CH₄ and dissolved CH₄ (Jansen et al., 2019) however, the shallow nature of the ponds in this study (\leq 85cm) make it unlikely that oxidation is able to happen in the water column and may instead be occurring in the sediment before bubbles are released to the water surface (Burke et al., 2019; Chanton, 2005; Wik et al., 2020).

Pond Type 3, consisting of ponds E and F, appears to be transitional reflecting a relatively even mix of HM and AM contributing to production. The δ^{13} C-CH₄ values for Type 3 were significantly more depleted in δ^{13} C-CH₄ than Types 1 and 2 but still within the range of values indicative of higher relative contribution of HM. There is likely a mixture of HM and AM contributing to production in Type 3 driving the enrichment of δ^{13} C-CH₄ (Chanton et al., 2005; Whiticar et al., 1986).

The respective ranges of δ^{13} C-CH₄ and δ D-CH₄ associated with different production pathways have overlap and can vary based on the isotopic composition of parent organic material, kinetic isotope fractionation, and mixture of the two production pathways resulting in a "transitional" isotopic composition between hypothesized ranges (Whiticar, 1999). Additionally, some question remains on the applicability of these ranges to freshwater systems (Wik et al., 2020) being originally based on isotopic data from marine environments (Whiticar et al., 1986). Apparent fractionation factor (α_{C}) can be useful to reconcile these overlaps and distinguish production pathways as fractionation from production remains somewhat constant despite variances in source δ^{13} C and δ D (Whiticar, 1999). Type 3 has a significantly lower mean α_{C} (1.046 to 1.077) than Type 1 with sample years combined, further supporting a shift towards higher relative contribution of AM in Type 3 ponds compared to Types 1 and 2 (Figure 10b; Chanton et al., 2005; Whiticar, 1999; Whiticar et al., 1986). Between 2014 and 2019 the mean α_{C} of Type 3 decreases (Figure 11) reflecting a possible shift over time towards more contribution

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of AM that is not seen in δ^{13} C-CH₄ alone (Figure 6). As the highest emitting pond type (Burke et al., 2019), shifts in the relative contribution of methanogenesis pathway and the associated isotopic composition of Type 3 emissions over time has the greatest potential relative to other pond types to influence isotopic composition at the ecosystem scale.

In Type 4, consisting of ponds G, H, and X, isotopic analysis indicates that AM and HM both contribute to production but AM has a higher relative contribution than is seen in Types 1-3. Mean δ^{13} C-CH₄ of Type 4 is significantly more enriched in ebullition (-49.2 to -76.8±0.50‰) than in porewater (-59.5±0.51 to -75.9±0.14‰), though this is likely driven by inconsistent sampling between individual ponds and less a reflection of oxidation effects that could explain enriched δ^{13} C between the sediment and emitted bubbles within a particular pond (Table 1; Chanton et al., 2005). Regardless of significant differences in mean δ^{13} C-CH₄ of individual ponds and between porewater and ebullition, the overall mean δ^{13} C-CH₄ of Type 4 ponds was the most enriched of all pond types, significantly so for ebullition data (Table 2; Figure 2). Enriched mean δ^{13} C-CH₄ of ebullition indicates higher contribution AM relative to other pond Types and mean δ^{13} C-CH₄ of porewater was also enriched in the transitional range of δ^{13} C values where contribution of both pathways is common, as seen is Type 3 (Chanton et al., 2005; Whiticar, 1999; Whiticar et al., 1986). Type 4 δ D-CH₄ of ebullition (-199±42 to -397±127‰) and porewater $(-228\pm17 \text{ to } -339\pm27\%)$ is depleted within ranges hypothesized for higher contribution of AM (Chanton et al., 2006; Whiticar, 1999; Whiticar et al., 1986). The pairing of enriched δ^{13} C-CH₄ and depleted δ^{13} C-CH₄ suggest AM has a large relative contribution to production in Type 4 ponds and is further supported by fractionation factor (α_c) (Chanton et al., 2005). The relatively low mean α_c of Type 4 supports a larger role of AM compared to Types 1-3 (Figure 10; Chanton et al., 2005; Whiticar, 1999; Whiticar et al., 1986). The relatively high

contribution of AM coincides with relatively high CH₄ daily ebullitive flux from Type 4 ponds and an increased presence of *Carex spp.* and *Eriophorum spp.* sedge vegetation compared to other pond Types (Burke et al., 2019). Type 4 ponds are hydrologically connected to the surrounding mire complex with input from adjacent fen areas, potentially explaining the shift in relative contribution of production pathway as increased hydrologic conductivity provides an input of labile C substrates favorable for AM (Hornibrook et al., 1997).

IV. 2. THAW PROGRESSION SHIFTS METHANOGENESIS PATHWAY

The four pond types in this study are representative of different thaw stages and their differences in dominant methanogenic pathway suggest a shift from higher relative contribution of HM to more contribution from AM as thaw stage progresses. Type 1 ponds are located on a palsa plateau while Type 2 ponds are located along edges of collapsed palsa, suggesting a chronological order between pond types as permafrost thaw and active layer deepening cause progression from Type 1 to Type 2 (Burke et al., 2019; Olsen et al., 2011). The mean δ^{13} C-CH₄ of ebullition and porewater samples from pond Types 1 and 2 are comparable to the mean signature of diffusive CH₄ flux (-79.6 \pm 0.6‰) documented from a partly thawed *Sphagnum spp*. ombrotrophic bog at Stordalen Mire (McCalley et al., 2014). Type 3 ponds are surrounded by collapsing and remnant palsas and are some of the deepest ponds in this study with significantly higher daily ebullitive CH₄ flux than all other pond types (Burke et al., 2019). The transitional isotopic composition of Type 3 may be indicative of a transitional thaw stage as well from partly or fully thaved bog-like environment to a fully thaved fen-like environment. Type 4 ponds have open water and are hydrologically connected and experience input from adjacent fen areas (Burke et al., 2019). The mean δ^{13} C-CH₄ of ebullition and porewater from Type 4 ponds is comparable to the mean signature of diffusive CH₄ flux (-66.3 \pm 1.6‰) from a fully thawed

Eriophiorum spp. fen at Stordalen Mire (McCalley et al., 2014). The fully thawed fen had an increased abundance of acetoclastic genus *Methanosaeta* of methanogens (McCalley et al., 2014). Apparent fractionation factor ($\alpha_{\rm C}$) was determined by McCalley et al. (2014) to be lower in the *Eriophorum spp*. dominated fen (1.046 ± 0.001) than in the Sphagnum bog (1.053 ± 0.002). Mean $\alpha_{\rm C}$ follows a comparable trend in this study between pond types, increasing from 1.067 ± 0.0072 in Type 4 to 1.073 ± 0.0057 in Type 1 (Table 1). Permafrost thaw at Stordalen Mire has progressed as permafrost palsa sites transition to ombrotrophic bogs and finally to minerotrophic fens (Johansson et al., 2006; McCalley et al., 2014). Shifts in isotopic composition of ebullitive CH₄ emissions from Types 1 to 4 are comparable to shifts in isotopic composition of diffusive CH₄ emissions as thaw progresses between bogs and fens suggesting that differences between pond types are chronological, with Type 1 being the least thawed and Type 4 being the most thawed, and similar to progressions seen along palsa to fen thaw gradients.

The thaw ponds in this study had dissolved and emitted δ^{13} C-CH₄ and δ D-CH₄ values similar to values reported in other thaw pond studies and ebullition measurements from postglacial lakes also located at Stordalen Mire. Ebullition δ D-CH₄ values from Types 1-4 were more enriched in δ D than data reported from runnel and polygonal ponds in Nunavut, Canada (median -372.9‰; Negandhi et al., 2013) and from alasses in Yakutsk, eastern Siberia (mean -363±20‰; Nakagawa et al., 2002), but were comparable to values from ebullition measurements of neighboring post-glacial lakes located at Stordalen Mire (mean -310.6±21.3‰; Wik et al., 2020). Ebullition δ^{13} C-CH₄ from Types 1, 2, and 3 in this study were also similar to ebullition δ^{13} C-CH₄ values measured from these same lakes (mean -67.8±4.6‰; Wik et al., 2020), while Type 4 ebullition was similar to values measured from other thaw ponds in Canada (median -58.8‰; Negandhi et al., 2013) and eastern Siberia (mean -61.1±4.4‰; Nakagawa et al., 2002). Dissolved δ^{13} C-CH₄ from pond types 1-4 was similar to data measured by Knoblauch et al. (2015) from polygonal ponds in northeast Siberia. Thaw ponds from varying locations have different emitted isotopic composition of CH₄, highlighting the need to better constrain the isotopic emissions from these features to inform the global CH₄ budget.

As increased temperatures cause an increase in permafrost thaw and active layer deepening, thaw pond formation will continue, and existing early stage ponds may transition to later pond types (Type 4). The difference in relative contribution of CH₄ production pathways between pond types suggests that this transition will be accompanied by a general shift from higher relative contribution of HM towards AM dominance in ponds. Northern wetlands have been associated with a narrow range of δ^{13} C-CH₄ values (-71±1‰) by Fisher et al (2017). The wide range of δ^{13} C-CH₄ emitted from the ponds in this study counters this and suggests that this range could be an inaccurate representation as larger amounts of emissions come from pond types with relatively enriched δ^{13} C-CH₄. Top-down models of high latitude CH₄ emissions currently do not account for a shift in δ^{13} C-CH₄ as thaw progresses and need to incorporate transitioning landscape features including the pond types examined in this study (Deng et al., 2014; McCalley et al., 2014).

IV. 3. TEMPORAL VARIATION IN ISOTOPIC COMPOSITION OF CH4

Our multi-year dataset of isotopic composition of emitted and dissolved CH₄ remains relatively constant across years with limited shifts in δ^{13} C-CH₄ and δ D-CH₄, supporting our hypothesis that we would not see changes in contribution of methanogenesis pathway within the timeframe of this study (Figure 5c, 5d, 7c, 7d). Despite general stability in contribution of production pathways, we do see that ebullition δ^{13} C-CH₄ and porewater δ D-CH₄ become significantly more depleted over the duration respective sampling periods (Figure 5d, 7c). There is a lack of literature studying dissolved and emitted CH₄ from thaw-formed ponds for extended periods of time (>3 years; Blodau et al., 2008; Knoblauch et al., 2015; Nakagawa et al., 2002; Negandhi et al., 2013) so it is difficult to determine what could be driving the depletion of stable isotopes of CH₄. Based on previous work conducted in bog and fen areas if thaw is driving a transition between pond types towards conditions more similar to Types 3 and 4 we could expect to see a directional shift towards more enriched $\delta^{13}C$ as is seen in transitions from relatively depleted δ^{13} C to relatively enriched along bog to fen thaw gradients (McCalley et al., 2014). Progression of thaw to more fen-like Type 4 ponds could also explain the depletion in δD -CH₄ from 2017 to 2019 (Figure 7c). However, it is unlikely that this shift is happening over a three or even seven year time frame given that permafrost thaw and subsequent vegetation shifts occur on a decadal scale at Stordalen Mire (Johansson et al., 2006). Alternatively, the depletion of ebullition δ^{13} C-CH₄ between years could indicate a sampling bias, specifically in the ebullition δ^{13} C-CH₄ data which has the largest number of samples in 2019 relative to all other years (Table 1; Figure 5d). More enriched δ^{13} C-CH₄ in sampling years prior to 2019 could be due to a larger number of samples collected from Types 3 and 4 with more enriched signatures. If the ponds in this study continue to be monitored to examine long-term variation in isotopic composition of CH₄ ebullition is will be critical to maintain even sample distribution across pond types.

Non-directional significant differences are seen in δ^{13} C of dissolved CH₄ that may be explained by climatic differences between sampling years (Figure 5c). In permafrost wetlands, increases in precipitation can cause narrowing of the range of values for δ^{13} C and δ D of dissolved CH₄ (Shingubara et al., 2019). Differences in precipitation could be driving the significant enrichment of porewater δ^{13} C-CH₄ in 2016 relative to other years.

On a monthly timescale, there were no significant differences in δD -CH₄ from ebullition or porewater samples between months during the time period from 2017 to 2019 (Figure 7a, 7b), further supporting our hypothesis that there would be no difference in isotopic composition of CH₄ between sampling months. There were however significant differences in δ^{13} C-CH₄ between months (Figure 5a, 5b). The depletion of porewater δ^{13} C-CH₄ from June to August may be due to the low sample count in September rather than a reflection of CH₄ production mechanisms, as δ^{13} C-CH₄ is not significantly different from June to August when sample counts are more evenly distributed (Figure 5a). Ebullition δ^{13} C-CH₄ becomes significantly more enriched from June to August, contrary to previous findings that δ^{13} C-CH₄ of ebullition does not change significantly on a seasonal timescale (Jansen et al., 2019; Wik et al., 2020). When porewater and ebullition δ^{13} C-CH₄ was grouped by pond it became evident that significant differences between months are driven by Pond F only (Table 7, 9, S2, S4) and are not representative of all ponds in this study. Pond F is Type 3 which we believe to be a transitional pond type. The shifts in monthly isotopic composition could be driven by active layer deepening around this pond as temperatures increase over the duration of the growing season, causing an influx of decomposed organic matter from the surrounding collapsing permafrost. This could be determined by collecting further data on the quality of carbon available in Pond F and other ponds in the study during the growing season. Individual conditions that could be driving shifts in δ^{13} C-CH₄ within this particular pond over the duration of the sampling season need to be further investigated.

V. SUMMARY AND CONCLUDING STATEMENTS

We measured the isotopic composition of δ^{13} C-CH₄, δ D-CH₄, and δ^{13} C-CO₂ from porewater and ebullitive flux of nine thaw ponds categorized into four types (Burke et al., 2019) over a seven year period and determined that relative contribution of hydrogenotrophic and acetoclastic methanogenesis pathways. Mean δ^{13} C-CH₄ of each pond type fell within hypothesized ranges for higher relative contribution of HM versus AM, but δD -CH₄ was not clearly associated with production effects with the exception of Type 4. This may be a result of depleted δD -H₂O occurring in northern terrestrial ecosystems, making production effects difficult to interpret from δ D-CH₄. The apparent fractionation factor was key to identifying differences in dominant production pathway between pond types because it helps to further identify CH₄ production effects. Comparison to bog and fen sites at Stordalen Mire indicate that Type 1, 2 ponds are most comparable to bogs in their isotopic composition of CH_4 and have higher relative contributions of HM, Type 3 is also comparable to bog sites but with values that suggest a more even mix of contribution of HM and AM relative to Types 1 and 2. The Type 4 δ^{13} C-CH₄ and δ D-CH₄ isotopic signatures indicate higher relative contribution of AM than in the other pond types. The shift in isotopic composition of CH₄ moving from pond type 1 to 4 is similar to thaw progressions from bog to fen documented at Stordalen Mire, suggesting that pond types are representative of thaw stages. Increased warming and permafrost thaw in northern permafrost peatlands has the potential to shift the relative contribution of methanogenesis pathways in these systems from higher contribution of HM to AM, and the overall δ^{13} C-CH₄ signatures of landscapes. However, this transition between pond types has not been documented within the timeframe of this study. This work highlights the need for accurate determination of the isotopic composition of thaw pond systems to constrain the global CH₄ budget that does not currently account for the relatively enriched δ^{13} C-CH₄ from northern wetlands.

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APPENDIX

APPENDIX A. TEMPORAL PAIRWISE COMPARISONS OF EBULLITION AND POREWATER $\delta^{13}C\text{-}CH_4$ and $\delta D\text{-}CH_4$ BY POND

Table A1. Pairwise comparisons across sampling years within ponds for ebullition δ^{13} C-CH₄ data using the Dunn's test following the Bonferroni method ($\alpha = 0.05$). Significant p-values indicated by *. Ponds with NA values had insufficient data for the comparison.

	Pairwise comparison of ebullition δ^{13} C-CH ₄ by year							
Pond A		р		Pond E		р		
2019	2017	1	•	2018	2017	0.0451*		
2019	2016	0.5458	•	2019	2017	1		
2017	2016	0.4068	•	2018	2016	1		
Pond B		р	•	2019	2016	1		
		NA	•	2019	2018	0.3287		
Pond C		р	•	2017	2016	0.1373		
2017	2016	1	•	Pond F		р		
2017	2012	0.981	•	2018	2017	0.5604		
2019	2018	1	•	2019	2017	1		
2019	2016	1	•	2019	2018	1		
2019	2012	1	•	2018	2016	1		
2016	2012	1	•	2019	2016	0.2218		
2018	2016	1		2017	2016	0.0257		
2018	2012	1	•	Pond G		р		
2019	2017	0.8148	•			NA		
2018	2017	0.7376	•	Pond H		р		
Pond D		р	•	2019	2018	1		
2017	2016	0.343	•	2016	2012	1		
2019	2016	0.6541	•	2017	2012	1		
2019	2017	1	•	2017	2016	1		
				2019	2017	1		
				2019	2012	0.0109*		
				2018	2017	0.8831		
			•	2019	2016	0.0469		
			•	2018	2012	0.1291		
				2018	2016	0.1079		
-								

Table A2. Pairwise comparisons across sampling months within ponds for ebullition δ^{13} C-CH₄ data using the Dunn's test following the Bonferroni method ($\alpha = 0.05$). Significant p-values indicated by *. Ponds with NA values had insufficient data for the comparison.

	Pairwise comparison of ebullition δ^{13} C-CH ₄ by month						
Pond A		р		Pond E		р	
August	June	0.7735	•	August	June	0.0662	
August	July	0.3007	•	July	June	0.3603	
July	June	1		August	July	1	
Pond B		р		Pond F		р	
August	June	0.4247		August	July	0.0425*	
July	June	1		August	June	0.6997	
August	July	1		July	June	1	
Pond C		р	•	Pond G		р	
July	June	1		August	July	0.4469	
August	June	1		Pond H		р	
August	July	1		August	June	0.0819	
Pond D		р		July	June	0.3727	
July	June	0.6686		August	July	0.8072	
August	June	1		Pond I		р	
August	July	0.3686				NA	

Table A3. Pairwise comparisons across sampling years within ponds for porewater δ^{13} C-CH₄ data using the Tukey-Kramer HSD test. Significant p-values indicated by *. Ponds with NA values had insufficient data for the comparison.

Pairwise comparison of porewater δ^{13} C-CH ₄ by year								
Pond A		р		Pond E		р		
2016	2018	<.0001*		2016	2017	<.0001*		
2014	2018	<.0001*		2016	2018	<.0001*		
2016	2017	0.0013*		2014	2017	0.0074*		
2016	2019	0.0038*		2016	2019	0.0006*		
2019	2018	0.0154*		2014	2018	0.0314*		
2017	2018	0.1614		2014	2019	0.4171		
2014	2017	0.0868		2019	2017	0.1229		
2016	2014	0.1221		2016	2014	0.5523		
2014	2019	0.2625		2019	2018	0.4235		
2019	2017	0.9544		2018	2017	0.9771		

Pond R		n		Pond F		n
	0017	P			2010	<u></u>
2014	2017	0.4873	_	2016	2018	<.0001*
2014	2019	0.5175		2016	2014	<.0001*
2018	2017	0.6482	_	2016	2017	0.0009*
2018	2019	0.6765		2019	2018	0.0107*
2014	2018	0.9987	_	2019	2014	0.0288*
2019	2017	0.9999		2016	2019	0.0674
Pond C		р		2017	2018	0.2926
2016	2014	0.0181*		2019	2017	0.4973
2016	2018	0.2074		2017	2014	0.5111
2017	2014	0.2247		2014	2018	0.9957
2016	2019	0.4487		Pond G		р
2019	2014	0.5378				NA
2016	2017	0.6937		Pond H		р
2017	2018	0.8592		2018	2019	0.5435
2018	2014	0.8238		2018	2016	0.7924
2019	2018	0.9885		2016	2019	0.8116
2017	2019	0.9892		Pond I		р
Pond D		р	_			NA
2017	2018	0.0376*				
2014	2018	0.0862				
2016	2018	0.0856				
2017	2019	0.1722				
2014	2019	0.3814				
2016	2019	0.3861				
2019	2018	0.7583				
2017	2016	0.9501				
2017	2014	0.9673				
2014	2016	1				

	Pairwis	e compari
Pond A		р
June	August	0.1088
June	July	0.129
September	August	0.6246
September	July	0.7791
June	September	0.8991
July	August	0.9449
Pond B		р
September	July	0.1514
September	June	0.4076
September	August	0.5581
August	July	0.6966
June	July	0.8459
August	June	0.9904
Pond C		р
June	September	0.1495
June	August	0.181
June	July	0.1867
July	September	0.6587
August	September	0.707
July	August	0.999
Pond D		р
June	July	0.3528
August	July	0.5517
June	September	0.9217
August	September	0.965
September	July	0.987
June	August	0.9939

Table A4. Pairwise comparisons across sampling months within ponds for porewater δ^{13} C-CH₄ data using the Tukey-Kramer HSD test. Significant p-values indicated by *. Ponds with NA values had insufficient data for the comparison.

APPENDIX B. 2019 DAILY BUBBLE FLUX BY POND

I. EBULLITON SAMPLING AND ANALYSIS

Ebullition (bubble) samples were collected from the seven study ponds during the growing season (June - August) in 2019; No samples were collected in 2013 and 2015. Every 1-3 days samples were collected from floating bubble traps to assess isotopic composition and calculate daily ebullitive flux (Figure 2). Bubble traps were deployed using methodology described in detail by Burke et al. (2019) and were designed similarly to bubble traps used by Wik et al. (2013). Ebullitive flux samples were processed as described by Burke et al. (2019). In summary, the accumulated gas was store in 10-ml polypropylene syringes and refrigerated at 2°C for up to 48hours prior to analysis with a gas chromatograph-flame ionization detector (GC-FID) at the Abisko Scientific Research Station (ANS).

II. FLUX RESULTS BY POND

Table A5. Sample count, range and average daily bubble flux for seven ponds in this study from the 2019 sampling season. Daily bubble flux varied widely between the ponds surveyed but were highest from Type 3 ponds E and F, consistent with previous findings from Burke et al. (2019).

Pond	Pond type	Minimum	Median flux	Maximum	Mean flux	Sample
	from Burke	flux	(mg CH ₄ m ⁻	flux	$(mg CH_4 m^{-2}d^{-1})$	count
	et al.	(mg CH ₄ m ⁻	$^{2}d^{-1}$)	$(mg CH_4 m^{-2}d^{-1})$		
	(2019)	$^{2}d^{-1}$)		1)		
А	1	0.0	0.1	56.6	7.6 ± 18.2	22
В	1	0.1	0.1	2.7	0.4 ± 0.6	30
С	2	0.0	0.7	463.9	28.6 ± 95.8	24
D	2	0.0	0.3	6.8	0.7 ± 1.5	19
Е	3	0.0	41.0	432.5	69.1 ± 81.9	81
F	3	0.3	79.6	2001.7	131.5 ± 234.0	88
Н	4	0.0	14.1	96.7	27.6 ± 27.4	40



Figure A1. Daily bubble flux (mg CH₄ m⁻²d⁻¹) from seven ponds included in this study during the 2019 sampling season. The y-axis was plotted between 0 and 350 mg CH₄ m⁻²d⁻¹ to better display the distribution of the data. Outliers greater than 350 mg CH₄ m⁻²d⁻¹ are omitted from this figure. Whiskers represent the 10th and 90th percentiles and black dots represent outliers. Black diamonds and horizontal black lines across boxes denote means and medians respectively. The number of samples (*n*) in each group is noted along the bottom. Daily bubble flux data was not normally distributed (*p* < 0.0001, Anderson-Darling test for normality) and was assessed using the Kruskal-Wallis rank sum test, results of which are noted as χ^2 and *p*. Lowercase letters indicate pairwise differences between ponds as determined by a Dunn's test following the Bonferroni method ($\alpha = 0.05$). Daily bubble flux varied widely but was significantly different between individual ponds. The pond type associated with each pond by Burke et al. (2019) is noted along the top of the figure.