



Article

Chemical Profiling and Biological Evaluation of Nepeta baytopii Extracts and Essential Oil: An Endemic Plant from Turkey

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Abstract: Nepeta baytopii is a poorly studied, endemic Nepeta species (Lamiaceae) of Turkey. For the first time, the biological activities (antioxidant, enzyme inhibition, and cytotoxicity properties) of the hexane, ethyl acetate, methanol, water/methanol, and water extracts and essential oil prepared from N. baytopii aerial parts were assessed. Hydro-methanol (41.25 mg gallic acid equivalent (GAE)/g) and water extracts (50.30 mg GAE/g), respectively showed the highest radical scavenging (94.40 and 129.22 mg Trolox equivalent (TE)/g, for 2,2-diphenyl-1-picrylhydrazyl radical and 2,2-azinobis (3-ethylbenzothiazoline-6-sulfonic acid radical scavenging assays) and reducing (229.37 and 129.55 mg TE/g, for ferric-reducing antioxidant power and cupric-reducing antioxidant capacity assays) capacities in vitro. An interestingly high inhibition was observed for ethyl acetate extract against butyrylcholinesterase (10.85 mg galantamine equivalent/g). The methanol extract showed high cytotoxicity (31.7%) against HepG2 cells. Caryophyllene oxide was identified in high concentrations in the essential oil (39.3%). Luteolin and apigenin and their derivatives were identified from the methanol and water extracts. The results obtained from this study highlighted that the abundance of highly bioactive compounds from Nepeta baytopii ensures the multiple biological activities of the tested extracts, and this suggests a potential use in the pharmaceutical and nutraceutical fields, and therefore should be investigated further.

Keywords: *Nepeta*; polyphenols; antioxidant; enzyme inhibition; phytopharmaceutics



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1. Introduction

Nowadays, humanity faces several problems, including both infectious and non-infectious diseases. The prevalence of some non-infectious diseases, such as Alzheimer's disease, diabetes mellitus, or obesity, is globally increasing by the day, and urgent precautions are needed to combat these diseases. Considering the increasing human population,

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synthetic precautions are still the most common for managing this fact. However, most synthetics have exhibited unfavorable side effects on human health, and we have to change them to safe, natural ones. In this sense, phytochemicals are considered in the natural arsenal for humanity [1–3]. Phytochemicals, including phenols, phenolic acids, flavonoids, tannins, and terpenoids, amongst others, are secondary metabolites possessing biological activities [4–6]. For decades now, humankind has been studying the intricate composition of plant extracts to harness their biological activities. Species of the *Lamiaceae* family possess therapeutic activity regarding gastroenterology, dermatology, and gynecology, and the herbs and leaves of *Lamiaceae* species have been used to treat respiratory complications [7]. The *Lamiaceae* family is a family of flowering plants, consisting of 236 genera composed of 6900–7200 species, and *Nepeta* is one of the largest genera of this family [7]. With some 280 species, the *Nepeta* genus is distributed over central and southern Europe, and western, central, and southern Asia, and North Africa [8].

The distinctive diversity and richness of regions of southwestern Asia, including Turkey and Iran, makes it a hotspot of the *Nepeta* genus [9]. In Turkey, 33 *Nepeta* species have been recorded, and 17 of them are endemic [9]. *Nepeta* species have been used in traditional medicine for their antiseptic, antispasmodic, anti-asthmatic, febrifuge, antitussive, and diuretic properties [8]. Additionally, in Turkey, the members of the *Nepeta* genus have been widely used for colds, cancers, coughing, rheumatism, wound healing, obesity, and stomachaches [10–13]. In addition, a recent comprehensive review published by Shara and colleagues [14] presented the in vivo and in vitro studies reporting the acetylcholinesterase inhibitory, anti-atherosclerotic, anticonvulsant and myorelaxant, antidiabetic, anti-leishmanial, anti-malarial, anti-melanogenesis, antioxidant, anthelmintic, hepatoprotective, cytotoxic, immunomodulatory, cardioprotective, anti-microbial species of the *Nepeta* genus. Salehi et al. [15] reviewed the presence of significant compounds in the *Nepeta* genus, including nepetalactone, β -caryophyllene, germacrene-D, 1,8-cineole, and α -pinene.

The biological activities of some Nepeta species endemic to Turkey, namely, Nepeta italica subsp. cadmea, N. nuda subsp. glandulifera, N. meyeri, N. conferta, and N. cadmea have been previously investigated [8,9,16-19]. However, endemic Nepeta baytopii has received little scientific attention. In an earlier study conducted by Dirmenci et al. [20], the morphological description and threatened categories of four Nepeta species were reported. In their study, N. baytopii were described as perennial with a height of 25–70 cm. The plant has longer-spreading hairs and sessile glands. Leaves are ovate-triangular, and the colour of the corolla is lilac. In the paper, the plant is described as an endangered species based on IUCN categories. In another study performed by Kılıc et al. [21], three Nepeta species were investigated for determining the essential oil composition, and one of the species was N. baytopii. In light of the above-mentioned point, this study was designed to provide additional data and novel insights on N. baytopii. In this sense, this work focuses on the evaluation of the hexane, ethyl acetate, methanol, water/methanol, and water extracts and essential oil prepared from *N. baytopii* aerial parts. Biological properties, namely, antioxidant and enzyme-inhibitory properties of all extracts and essential oil, were determined by using in vitro spectrophotometric methods. To determine the cytotoxic effects of the methanol and water extracts, three cell lines (HepG2, B16A5 and S 17) were used. Moreover, methanol and water extracts and essential oils were chemically characterised by using chromatographic methods. It is anticipated that gathered data herein will contribute towards establishing baseline data on this endemic species with significant medicinal potential.

2. Results and Discussion

Folin–Ciocalteu and aluminium chloride assays are rapid, simple, and low-cost procedures which provide an overview of the phenolic and flavonoid contents of plant extracts. These widely used methods underpin detailed phytochemical profiling using cutting-edge technologies. As shown in Table 1, the water extract (50.30 mg GAE/g) possessed the high-

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est concentration of phenolics, followed by the water/methanol extract (41.25 mg GAE/g). The ethyl acetate extract rich in flavonoids contained the value of 27.02 mg RE/g. Hexane extract possessed the lowest phenolic (13.23 mg GAE/g) and flavonoid (7.77 mg RE/g) contents. Additionally, both total amounts of phenolics and flavonoids were significantly affected by the extraction solvents used (p < 0.05). This observation was also confirmed by several authors, who reported that the used solvents affected the level of phenolics and flavonoids of Nepeta extracts [16,22]. Detailed profiles of N. baytopii aerial parts methanol and water extracts were provided in Tables 2 and 3, respectively. UHPLC profiling confirmed the presence of 46 and 43 compounds from methanol and water extracts, respectively. Chromatograms are depicted in Supplemental Material (Figures S1 and S2). Flavones, such as luteolin and apigenin and their derivatives, were identified from both extracts. Fertaric acid, a hydroxycinnamic acid and ester of ferulic acid and tartaric acid, was present in the methanol and water extracts. In general, the detailed phytochemicals of the methanol and water extracts of N. baytopii aerial parts were quite similar, which might be related to the polar nature of the solvents. In addition, the concentration of the different components in the methanol and water extracts might be different, and this was not specified in our data. In accordance with our results, the presence of flavones and hydroxycinnamic acid in the members of the *Nepeta* genus was reported in earlier studies [23–27].

GC-MS was used to determine the composition of *N. baytopii* aerial parts in the essential oil, and the data were presented in Table 3. A total of 10 compounds have been identified from the *N. baytopii* essential oil. Caryophyllene oxide, a sesquiterpenoid oxide common to lemon balm and eucalyptus, was identified in high concentration in *N. baytopii* essential oil (39.3%) [28]. Another sesquiterpene, spathulenol (15.6%), was identified in appreciable amounts from *N. baytopii* essential oil. Kilic and colleagues also reported a lower concentration of caryophyllene oxide in the essential oil of *N. baytopii* aerial part [21]. These differences in the levels of essential oil components could be explained by geographical and climatic differences. Additionally, different compounds (caryophyllene, limonene, nepetalactone and 1,8-cineole, etc.) were identified as main components in the essential oils of some *Nepeta* species [9,29–31].

Table 1. Total bioactive compounds and total antioxidant capacity (by phosphomolybdenum assay) of the tested extracts.

Extracts	TPC (mg GAE/g)	TFC (mg RE/g)
n-Hexane	13.23 ± 0.21 * $^{ m e}$	7.77 ± 0.07 e
Ethyl acetate	$19.57\pm0.24~^{ m d}$	27.02 ± 0.60 a
Methanol	33.81 ± 0.22 c	$23.78\pm0.87^{ m b}$
Water/methanol	41.25 ± 0.18 $^{\mathrm{b}}$	$10.61 \pm 0.54 ^{ m d}$
Water	50.30 ± 0.13 a	13.48 ± 0.18 $^{\mathrm{c}}$
Essential oil	nt	nt

^{*} Values are reported as mean \pm SD. TPC: total phenolic content; TFC: total flavonoid content; GAE: gallic acid equivalent; RE: rutin equivalent; nt: not tested. Different letters indicate significant differences in the tested extracts (p < 0.05).

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Table 2. Chemical composition of the methanol extract.

No.	Name	Formula	Rt	[M + H] ⁺	[M – H] [–]	Fragment 1	Fragment 2	Fragment 3	Fragment 4	Fragment 5
1	Quinic acid	$C_7H_{12}O_6$	1.23		191.06	173.04	171.03	127.04	93.03	85.03
2	Pantothenic acid	C ₉ H ₁₇ NO ₅	6.13	220.12		202.11	184.10	174.11	116.03	90.06
3	Caftaric acid (2-O-Caffeoyltartaric acid)	$C_{13}H_{12}O_9$	8.54		311.04	179.03	149.01	135.04	87.01	
4	Neochlorogenic acid (5-O-Caffeoylquinic acid)	$C_{16}H_{18}O_9$	10.11	355.10		163.04	145.03	135.04	117.03	89.04
5	Unidentified iridoid	$C_{16}H_{24}O_{9}$	13.14		405.14	359.14	197.08	179.07	153.05	71.01
6	Salicylic acid-O-hexoside	$C_{13}H_{16}O_{8}$	13.50		299.08	137.02	113.02	93.03	85.03	71.01
7	Mussaenosidic acid or isomer	$C_{16}H_{24}O_{10}$	13.57		375.13	213.08	169.09	151.08	125.06	107.05
8	Kynurenic acid	$C_{10}H_7NO_3$	13.80	190.05		162.06	144.04	116.05	89.04	
9 ¹	Chlorogenic acid (3-O-Caffeoylquinic acid)	$C_{16}H_{18}O_9$	14.84	355.10		163.04	145.03	135.04	117.03	89.04
10	Fertaric acid (2-O-Feruloyltartaric acid)	$C_{14}H_{14}O_9$	14.85		325.06	193.05	178.03	149.06	134.04	87.01
11	Caffeic acid	$C_9H_8O_4$	15.15		179.03	135.04	107.05			
12	Benzofuranecarbaldehyde	$C_9H_6O_2$	15.53	147.04		119.05	91.05	65.04		
13	Cryptochlorogenic acid (4-O-Caffeoylquinic acid)	$C_{16}H_{18}O_9$	16.08	355.10		163.04	145.03	135.04	117.03	89.04
14	5-O-(4-Coumaroyl)quinic acid	$C_{16}H_{18}O_{8}$	17.40		337.09	191.06	173.04	163.04	119.05	93.03
15	4-O-(4-Coumaroyl)quinic acid	$C_{16}H_{18}O_{8}$	18.03		337.09	191.06	173.04	163.04	119.05	93.03
16	Phaselic acid (2-O-Caffeoylmalic acid)	$C_{13}H_{12}O_8$	18.62		295.05	179.03	135.04	133.01	115.00	71.01
17	Loliolide	$C_{11}H_{16}O_3$	20.01	197.12		179.11	161.10	135.12	133.10	107.09
18	Eriodictyol-O-hexoside	$C_{21}H_{22}O_{11}$	20.76		449.11	287.06	151.00	135.04	107.01	83.01
19	7-Deoxyloganic acid isomer	$C_{16}H_{24}O_9$	20.95		359.13	197.08	153.09	135.08	109.06	89.02
20	Luteolin-O-hexosylglucuronide	$C_{27}H_{28}O_{17}$	21.44		623.12	285.04	217.05	199.04	175.04	133.03
21	Luteolin-7-O-glucuronide	$C_{21}H_{18}O_{12}$	22.75		461.07	285.04	217.05	199.04	175.04	133.03
22	Luteolin-7-O-glucoside (Cynaroside)	$C_{21}H_{20}O_{11}$	22.86		447.09	327.05	285.04	284.03	256.04	151.00
23	Apigenin-O-hexosylglucuronide	$C_{27}H_{28}O_{16}$	22.97		607.13	269.05	151.00	113.02		
$24^{\ 1}$	Apigenin-7-O-glucuronide	$C_{21}H_{18}O_{11}$	24.52		445.08	269.05	175.02	151.00	113.02	
25	Rosmarinic acid (Labiatenic acid)	$C_{18}H_{16}O_{8}$	24.71		359.08	197.05	179.03	161.02	135.04	72.99
26^{1}	Eriodictyol (3',4',5,7-Tetrahydroxyflavanone)	$C_{15}H_{12}O_6$	25.40		287.06	151.00	135.04	125.02	107.01	83.01
27	3-O-Methylrosmarinic acid	$C_{19}H_{18}O_{8}$	26.58		373.09	197.05	179.03	175.04	160.02	135.04
28^{1}	Naringenin (4',5,7-Trihydroxyflavanone)	$C_{15}H_{12}O_5$	27.72		271.06	227.07	177.02	151.00	119.05	107.01
29 ¹	Luteolin (3',4',5,7-Tetrahydroxyflavone)	$C_{15}H_{10}O_6$	28.37		285.04	217.05	199.04	175.04	151.00	133.03
30^{1}	Apigenin (4',5,7-Trihydroxyflavone)	$C_{15}H_{10}O_5$	30.23		269.05	227.04	225.06	151.00	149.02	117.03
31	Dimethoxy-trihydroxy(iso)flavone	$C_{17}H_{14}O_7$	30.38		329.07	314.04	313.04	299.02	271.03	
32	Dihydrololiolide	$C_{11}H_{18}O_3$	30.51	199.13		181.12	163.11	135.12	111.04	107.09
33	Undecanedioic acid	$C_{11}H_{20}O_4$	31.30		215.13	197.12	153.13	125.10	57.03	
34	Malyngic acid or isomer	$C_{18}H_{32}O_5$	32.54		327.22	309.21	291.20	229.14	211.13	171.10
35	Nakhsmyrin or isomer	$C_{14}H_{12}O_4$	32.67	245.08		227.07	217.09	203.07	175.04	
36	Nakhsmyrin or isomer	$C_{14}H_{12}O_4$	33.29	245.08		227.07	217.09	203.07	175.04	

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 Table 2. Cont.

No.	Name	Formula	Rt	$[M + H]^+$	$[M-H]^-$	Fragment 1	Fragment 2	Fragment 3	Fragment 4	Fragment 5
37	Dodecanedioic acid	$C_{12}H_{22}O_4$	33.74		229.14	211.13	185.15	167.14		
38	Pinellic acid	$C_{18}H_{34}O_5$	33.83		329.23	311.22	293.21	229.14	211.13	99.08
39	Caffeic acid phenethyl ester	$C_{17}H_{16}O_4$	34.10		283.10	179.03	178.03	161.02	135.04	133.03
40	Salvigenin (5-Hydroxy-4',6,7-trimethoxyflavone)	$C_{18}H_{16}O_6$	35.33	329.10		314.08	313.07	296.07	285.08	268.07
41	Octadecenedioic acid	$C_{18}H_{32}O_4$	37.99		311.22	293.21	235.17	223.17		
42	Stearidonic acid	$C_{18}H_{28}O_2$	40.16		275.20	257.19	231.21	59.01		
43	Hydroxyoctadecatrienoic acid	$C_{18}H_{30}O_3$	40.22		293.21	275.20	235.17	223.13	171.10	59.01
44	Stearidonic acid methyl ester	$C_{19}H_{30}O_2$	42.15	291.23		259.21	241.20	217.20	107.09	93.07
45	Linoleamide	$C_{18}H_{33}NO$	44.44	280.26		263.24	245.23	109.10	95.09	81.07
46	Oleamide	$C_{18}H_{35}NO$	45.68	282.28		265.25	247.24	97.10	83.09	69.07

¹ Confirmed by standards. Fragment: the fragments of compounds reflect a unique pattern in the mass spectrum.

Table 3. Chemical composition of the infusion.

No.	Name	Formula	Rt	[M + H] ⁺	[M – H] [–]	Fragment 1	Fragment 2	Fragment 3	Fragment 4	Fragment 5
1	Quinic acid	$C_7H_{12}O_6$	1.22		191.06	173.04	171.03	127.04	93.03	85.03
2	Pantothenic acid	$C_9H_{17}NO_5$	6.16	220.12		202.11	184.10	174.11	116.03	90.06
3	Caftaric acid (2-O-Caffeoyltartaric acid)	$C_{13}H_{12}O_9$	8.42		311.04	179.03	149.01	135.04	87.01	
4	Neochlorogenic acid (5-O-Caffeoylquinic acid)	$C_{16}H_{18}O_9$	10.12	355.10		163.04	145.03	135.04	117.03	89.04
5	Unidentified iridoid	$C_{16}H_{24}O_9$	13.10		405.14	359.14	197.08	179.07	153.05	71.01
6	Salicylic acid-O-hexoside	$C_{13}H_{16}O_{8}$	13.48		299.08	137.02	113.02	93.03	85.03	71.01
7	Mussaenosidic acid or isomer	$C_{16}H_{24}O_{10}$	13.54		375.13	213.08	169.09	151.08	125.06	107.05
8	Kynurenic acid	$C_{10}H_7NO_3$	13.76	190.05		162.06	144.04	116.05	89.04	
9 ¹	Chlorogenic acid (3-O-Caffeoylquinic acid)	$C_{16}H_{18}O_9$	14.79	355.10		163.04	145.03	135.04	117.03	89.04
10	Fertaric acid (2-O-Feruloyltartaric acid)	$C_{14}H_{14}O_9$	14.81		325.06	193.05	178.03	149.06	134.04	87.01
11	Benzofuranecarbaldehyde	$C_9H_6O_2$	15.47	147.04		119.05	91.05	65.04		
12	Cryptochlorogenic acid (4-O-Caffeoylquinic acid)	$C_{16}H_{18}O_9$	16.06	355.10		163.04	145.03	135.04	117.03	89.04
13	5-O-(4-Coumaroyl)quinic acid	$C_{16}H_{18}O_{8}$	17.37		337.09	191.06	173.04	163.04	119.05	93.03
14	4-O-(4-Coumaroyl)quinic acid	$C_{16}H_{18}O_{8}$	18.00		337.09	191.06	173.04	163.04	119.05	93.03
15	Loliolide	$C_{11}H_{16}O_3$	19.98	197.12		179.11	161.10	135.12	133.10	107.09
16	Eriodictyol-O-glucuronide	$C_{21}H_{20}O_{12}$	20.69		463.09	287.06	175.02	151.00	135.04	113.02
17	Eriodictyol-O-hexoside	$C_{21}H_{22}O_{11}$	20.75		449.11	287.06	151.00	135.04	107.01	83.01
18	7-Deoxyloganic acid isomer	$C_{16}H_{24}O_9$	20.94		359.13	197.08	153.09	135.08	109.06	89.02
19	Luteolin-O-hexosylglucuronide	$C_{27}H_{28}O_{17}$	21.43		623.12	285.04	217.05	199.04	175.04	133.03

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 Table 3. Cont.

No.	Name	Formula	Rt	[M + H] ⁺	[M – H] [–]	Fragment 1	Fragment 2	Fragment 3	Fragment 4	Fragment 5
20	Luteolin-7-O-glucuronide	C ₂₁ H ₁₈ O ₁₂	22.74		461.07	285.04	217.05	199.04	175.04	133.03
21	Luteolin-7-O-glucoside (Cynaroside)	$C_{21}H_{20}O_{11}$	22.85		447.09	327.05	285.04	284.03	256.04	151.00
22	Apigenin-O-hexosylglucuronide	$C_{27}H_{28}O_{16}$	22.96		607.13	269.05	151.00	113.02		
23^{1}	Apigenin-7-O-glucuronide	$C_{21}H_{18}O_{11}$	24.51		445.08	269.05	175.02	151.00	113.02	
24	Rosmarinic acid (Labiatenic acid)	$C_{18}H_{16}O_{8}$	24.71		359.08	197.05	179.03	161.02	135.04	72.99
25	N-trans-Feruloyltyramine	$C_{18}H_{19}NO_4$	25.12	314.14		194.08	177.05	149.06	145.03	121.07
26^{1}	Eriodictyol (3',4',5,7-Tetrahydroxyflavanone)	$C_{15}H_{12}O_6$	25.40		287.06	151.00	135.04	125.02	107.01	83.01
27	3-O-Methylrosmarinic acid	$C_{19}H_{18}O_8$	26.59		373.09	197.05	179.03	175.04	160.02	135.04
28^{1}	Luteolin $(3',4',5,7$ -Tetrahydroxyflavone)	$C_{15}H_{10}O_6$	28.39		285.04	217.05	199.04	175.04	151.00	133.03
29 ¹	Apigenin (4',5,7-Trihydroxyflavone)	$C_{15}H_{10}O_5$	30.23		269.05	227.04	225.06	151.00	149.02	117.03
30	Dihydrololiolide	$C_{11}H_{18}O_3$	30.52	199.13		181.12	163.11	135.12	111.04	107.09
31	Undecanedioic acid	$C_{11}H_{20}O_4$	31.32		215.13	197.12	153.13	125.10	57.03	
32	Malyngic acid or isomer	$C_{18}H_{32}O_5$	32.55		327.22	309.21	291.20	229.14	211.13	171.10
33	Nakhsmyrin or isomer	$C_{14}H_{12}O_4$	32.69	245.08		227.07	217.09	203.07	175.04	
34	Nakhsmyrin or isomer	$C_{14}H_{12}O_4$	33.30	245.08		227.07	217.09	203.07	175.04	
35	Dodecanedioic acid	$C_{12}H_{22}O_4$	33.76		229.14	211.13	185.15	167.14		
36	Pinellic acid	$C_{18}H_{34}O_5$	33.85		329.23	311.22	293.21	229.14	211.13	99.08
37	Salvigenin (5-Hydroxy-4',6,7-trimethoxyflavone)	$C_{18}H_{16}O_{6}$	35.35	329.10		314.08	313.07	296.07	285.08	268.07
38	Octadecenedioic acid	$C_{18}H_{32}O_4$	38.00		311.22	293.21	235.17	223.17		
39	Stearidonic acid	$C_{18}H_{28}O_2$	40.19		275.20	257.19	231.21	59.01		
40	Hydroxyoctadecatrienoic acid	$C1_8H_{30}O_3$	40.23		293.21	275.20	235.17	223.13	171.10	59.01
41	Stearidonic acid methyl ester	$C_{19}H_{30}O_2$	42.15	291.23		259.21	241.20	217.19	107.09	93.07
42	Linoleamide	C ₁₈ H ₃₃ NO	44.45	280.26		263.24	245.23	109.10	95.09	81.07
43	Oleamide	$C_{18}H_{35}NO$	45.71	282.28		265.25	247.24	97.10	83.09	69.07

¹ Confirmed by standards. Fragment: the fragments of compounds reflect a unique pattern in the mass spectrum.

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The dearth of scientific information regarding the antioxidant capacities of *N. baytopii* has fuelled the need for the comprehensive evaluation of the antioxidant properties of the different extracts and the essential oil of this endemic species. In order to evaluate the antioxidant properties of the extracts and essential oil of N. baytopii aerial part, six bioassays were conducted. These assays included free-radical scavenging (DPPH and ABTS), reducing power (FRAP and CUPRAC), metal chelating, and phosphomolybdenum. These findings are presented in Table 4. Several studies have reported the relationship between high phenolic/flavonoid content and antioxidant activity [32-34]. The total antioxidant capacity of the extracts was assessed using the phosphomolybdenum method. As shown in Table 4, the methanol (2.45 mmol TE/g) and ethyl acetate (2.36 mmol TE/g) extracts were the most active. However, we did not observe any statistical difference among the ethyl acetate and methanol extracts (p > 0.05). Moreover, essential oil N. baytopii aerial parts also showed a better total antioxidant ability than those of water/methanol and nhexane extracts (p < 0.05). The total antioxidant ability could be attributed to the presence of different compounds in the extracts or essential oils. In this sense, as can be seen in Figure 1, we observed a weak correlation between total phenolics and phosphomolybdenum results, but the correlation value was high for total flavonoids. Determining the ability of natural compounds to quench free radicals provides an estimation of their possible scavenging activity in other systems. As shown in Table 4, the water/methanol extract of *N. baytopii* aerial part showed the highest scavenging activity against DPPH (94.40 mg TE/g) and ABTS (129.22 mg TE/g). In the ABTS assay, n-hexane, ethyl acetate, and the essential oil exhibited similar scavenging abilities (p > 0.05). In addition, these extracts and the essential oil did not have any scavenging ability on the DPPH radical. The reducing capacity of the compounds to donate an electron and thus act as reducing agents is commonly assessed using two widely used methods, namely, FRAP (ferric ion) and CUPRAC (cupric ion) assays [35]. In the present study, the water extract of N. baytopii exhibited the highest Fe^{3+} (129.55 mg TE/g)- and Cu²⁺ (229.37 mg TE/g)-reducing potentials. In these reduced power assays, all tested samples exhibited different abilities (p < 0.05). As can be seen in Tables 1 and 5, generally, the free radical scavenging and reduced power results could be correlated with their total phenolic contents (r > 0.8). Pearson's correlation coefficient values are given in Figure 1. Thus, it resulted that phenolic compounds are the main contributors to the antioxidant properties of N. baytopii. Similarly, several researchers have reported a strong correlation between antioxidant properties and the total amounts of phenolics [32,36,37]. Moreover, some authors have argued that the phenolic compounds in the members of the Nepeta genus were main players in the antioxidant assays [30,38,39]. As another mechanism, transition metals are known to participate in Fenton reactions, generating free radicals and exacerbating the oxidative stress status. Therefore, the chelation capacity of *N. baytopii* aerial parts extracts and essential oil were assessed. Results presented herein demonstrated that the water extract and water/methanol extracts possessed a stronger chelating ability as compared with other extracts and essential oils (p < 0.05). The metal-chelating abilities of the tested extracts might be due to the presence of phenolics, and the correlation analysis was confirmed by this fact (r = 0.77).



Figure 1. Pearson's correlation between total bioactive compounds and antioxidant properties and enzyme inhibition effects (p < 0.05).

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Table 4. Chemi	ical profile o	f the tested	l essential oil.
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No	Compounds	RRI ^a	(%)
1	Dihyroedulan I	1530	6.1
2	β-Bourbonene	1531	3.7
3	Linalool	1548	4.1
4	cis-p-mentha-2,8-dien-1-ol	1678	5.7
5	Verbonene	1732	2.1
6	(E)-β-Damascenone	1841	0.8
7	Caryophyllene oxide	2017	39.3
8	Hexahydrofarnesyl acetone	2134	2.7
9	Spathulenol	2147	15.6
10	n-Hexadecanoic acid	2912	11.0
	Total identified (%)		91.1

a Relative retention indices are calculated against n-alkanes.

Table 5. Antioxidant properties of the tested extracts.

Extracts	PBD (mmol TE/g)	DPPH (mg TE/g)	ABTS (mg TE/g)	CUPRAC (mg TE/g)	FRAP (mg TE/g)	MCA (mg EDTAE/g)
n-Hexane	$1.28\pm0.15^{\text{ c}}$	na	$12.04\pm0.84~^{\rm d}$	$44.08\pm0.35~^{\mathrm{e}}$	$22.38\pm0.66^{\text{ e}}$	$0.34\pm0.02^{\text{ e}}$
Ethyl acetate	2.36 ± 0.20 a	na	13.33 ± 0.58 d	75.55 ± 0.60 d	29.24 ± 0.27 d	$22.15 \pm 2.16^{\ b}$
Methanol	2.45 ± 0.15 a	90.88 ± 0.37 ^c	$92.43 \pm 1.30^{\ b}$	$165.54\pm1.87^{\text{ c}}$	$88.78\pm1.36~^{\rm c}$	$15.61\pm0.54^{\text{ c}}$
Water/methanol	1.87 ± 0.06 b	94.40 ± 0.09 a	129.22 \pm 0.78 $^{\mathrm{a}}$	221.71 ± 2.59 b	124.78 ± 1.40 b	$26.88\pm2.10~^{a}$
Water	2.09 ± 0.07 a,b	$93.16 \pm 0.20^{\text{ b}}$	86.56 ± 2.54 ^c	$229.37 \pm 1.38~^{a}$	$129.55 \pm 1.23~^{\mathrm{a}}$	$27.14\pm0.58~^{\rm a}$
Essential oil	2.22 ± 0.15 a,b	na	12.10 ± 0.53 d	$21.42\pm0.11~^{\mathrm{f}}$	$10.95\pm0.13~^{\rm f}$	$6.52\pm0.07~^{ m d}$

EDTAE: EDTA equivalents; PBD: phosphomolybdenum; TE: trolox equivalent; na: non active. Different letters indicate significant differences in the tested extracts (p < 0.05).

The inhibitory ability of *N. baytopii* aerial parts extracts and essential oil were tested against enzymes linked to a critical role in the development of diabetes mellitus type II, Alzheimer's disease, and skin hyperpigmentation problems. Diabetes mellitus type II and Alzheimer's disease have escalated to epidemic proportions, and the need for complementary therapeutic agents to effectively manage these debilitating conditions are of paramount importance. From Table 5, the ethyl acetate extract of *N. baytopii* aerial parts exhibited the highest activity against AChE (4.57 mg GALAE/g) and BChE (10.85 mg GALAE/g). In AChE inhibition, n-hexane and methanol extracts displayed similar actions (p > 0.05). The high galantamine equivalent value recorded on BChE supported appreciably high inhibitory action in comparison to other Lamiaceae species [40–42]. The inhibition of BChE has been advocated in the later stage of Alzheimer's disease. During the progression of the disease, the BChE level increases, exacerbating the conditions of the patient [43]. Herein, the ability of *N. baytopii* aerial parts extracts and essential oil to inhibit α -amylase and α -glucosidase was also evaluated. These enzymes play critical roles in hyperglycaemia, the hallmark of diabetes mellitus. In diabetes mellitus type II management, the inhibition of enzymes responsible for the hydrolysis of polysaccharides to glucose monosaccharide, which can be absorbed in the intestinal system. Herein, methanol extract showed the highest (8.15 mmol ACAE/g) activity against α-glucosidase. Interestingly, the hexane and ethyl acetate extract were also good inhibitors of α -glucosidase. α -Glucosidase situated at the brush border of the small intestine catalyses the hydrolysis of disaccharides into glucose. Therefore, the inhibition of α -glucosidase reduces glucose formation, glycaemic peaks, and hyperglycaemia. Apart from these debilitating maladies, the ability of N. baytopii aerial parts extracts and essential oil to inhibit tyrosinase was also evaluated. Tyrosinase is the key enzyme targeted in skin hyperpigmentation treatment. In fact, the inhibition of tyrosinase reduces the production of the brown pigment melanin. Herein, the methanol and water/methanol extracts possessed the highest tyrosinase inhibition values (p > 0.05). The search for natural compounds possessing tyrosinase inhibitory characteristics is of particuPlants 2021, 10, 1176 9 of 15

lar interest in the dermato-cosmetic industry, and this has been fuelled by the interest of the general public for naturally derived products. Observed enzyme inhibitory properties of N. baytopii extracts might be explained by their chemical components. Some components such as apigenin, naringenin, luteolin, and chlorogenic acid have been reported as significant enzymes inhibitors [44–54], and thus, N. baytopii could be considered as a promising source of natural enzyme inhibitors. Interestingly, several researchers reported on the enzyme inhibition abilities of some Nepeta species. For example, Sarikurkcu et al. [55] reported the inhibitory properties of *N. nuda* subsp. *glandulifera* and *N. cadmea* on cholinesterases, amylase, glucosidase, and tyrosinase. When compared with our results, the Nepeta species exhibited lower enzyme inhibition properties than N. baytopii. Furthermore, different research groups reported the enzyme-inhibitory effects of several Nepeta essential oils. For example, N. nuda and N. cadmea essential oils exhibited moderate inhibitory effects on some enzymes, and the main compounds were geijerene and nepetalactone in these essential oils, respectively [9]. As a structure-ability approach, essential oils have a complex nature, and thus, observed enzyme inhibitory abilities could be caused by different factors, including the main compounds and interactions of these components.

Several species of the *Lamiaceae* family have been studied for the development of novel chemotherapeutic agents [56–58]. Herein, the methanol and water extracts were tested with HepG2, human hepatocarcinoma cells (Table 6). Hepatocellular carcinoma, the most common liver malignancy, is a leading cause of cancer-related death worldwide [58]. In this work, the methanol extract of *N. baytopii* aerial parts showed high cytotoxicity (31.7%) against HepG2 while water was non-cytotoxic. Melanoma is a type of skin cancer occurring in melanocytes, which are dendritic-like cells producing melanin pigment [59]. We observed that the water extract (70.2%) showed higher cytotoxicity against mouse melanoma cell (B16 4A5). We also determined the cytotoxic effect of *N. baytopii* aerial parts against non-tumoral murine bone marrow stromal, and the results are presented in Table 7. The methanol extract (34.8%) was more cytotoxic than the water extract (61.5%). In the literature, in accordance with the presented results, several *Nepeta* species, such as *N. curvidens* [60], *N. curviflora* [61], and *N. nuda* [22] exhibited remarkable cytotoxic effects on several cell lines.

Table 6. Cellular viability (%) of *Nepeta baytopii* extracts on HepG2, B16 4A5 and S17 cell lines applied at the concentration of $100 \,\mu\text{g/mL}$.

Sample/Cell line	HepG2	B16 4A5	S17
DMSO 0.5%	101 ± 7	88.2 ± 2.1	79.3 ± 4.9
Methanol	31.7 ± 0.5	76.7 ± 2.3	34.8 ± 0.9
Water	108.6 ± 12.6	70.2 ± 3.1	61.5 ± 5.6

Table 7. Enzyme inhibitory effects of the tested extracts.

Extracts	AChE (mg GALAE/g))	BChE (mg GALAE/g)	α-Amylase (mmol ACAE/g)	α -Glucosidase (mmol ACAE/g)	Tyrosinase (mg KAE/g)
n-Hexane	$3.97\pm0.32^{\ \mathrm{b}}$	$6.93\pm1.14^{\text{ b}}$	0.66 ± 0.01 b	$7.87 \pm 0.02^{\ \mathrm{b}}$	$77.84 \pm 1.83^{\text{ b}}$
Ethyl acetate	4.57 ± 0.06 a	10.85 ± 0.73 a	0.84 ± 0.02 a	7.76 ± 0.01 b	78.60 ± 1.58 b
Methanol	$3.65 \pm 0.11^{\ \mathrm{b}}$	$2.98\pm0.46^{\text{ c}}$	0.67 ± 0.02 b	8.15 ± 0.08 a	96.06 ± 0.70 a
Water/methanol	$2.68\pm0.07^{\text{ c}}$	na	0.50 ± 0.01 ^c	0.61 ± 0.04 $^{ m e}$	95.31 ± 1.77 a
Water	na	na	0.10 ± 0.01 e	$1.06 \pm 0.09 ^{\mathrm{d}}$	6.15 ± 1.02 ^d
Essential oil	na	na	0.24 ± 0.01 d	1.64 ± 0.01 ^c	$21.41\pm3.57^{\text{ c}}$

GALAE: galantamine equivalent; KAE: kojic acid equivalent; ACAE: acarbose equivalent; na: non-active. Different letters indicate significant differences in the tested extracts (p < 0.05).

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3. Materials and Methods

3.1. Plant Material

The aerial parts of *Nepeta baytopii* were collected in July 2019 (Genç village, Bingöl, Turkey, 38°43′00″ N, 40°34′09″ E, 1055 m). The plant material was authenticated by one of the authors (R.P). Voucher specimens (GP-1082) were deposited in the Bingöl University, Faculty of Agriculture, Bingöl, Turkey. Twenty-five plants were randomly collected in the same population, and they were dried in a dark condition for 10 days.

3.2. Extraction

The aerial parts of the plant materials were grounded, and then 10 g were separately extracted with hexane, ethyl acetate, methanol, and methanol/water (80%) in maceration technique (for 24 h, room temperature). The extracts were evaporated to dryness and stored at $4\,^{\circ}\text{C}$ until analysis. Regarding water extracts, we used traditional infusion techniques, and 5 g plant materials were kept with 100 mL of boiled water for 15 min. Then, the water extracts were lyophilised. The extracts procedure were performed in triplicate and the obtained extracts were stored at $4\,^{\circ}\text{C}$ until analysis.

3.3. UHPLC-MS Analysis

Chromatographic separation was accomplished with a Dionex Ultimate 3000RS UH-PLC instrument, equipped with a Thermo Accucore C_{18} (100 mm \times 2.1 mm i. d., 2.6 μ m) analytical column for the separation of compounds. Water (A) and methanol (B) containing 0.1% formic acid were employed as mobile phases, respectively. The total run time was 70 min for the elution profile. Mass spectrum analysis was carried out using a Thermo Q-Exactive Orbitrap mass spectrometer (Thermo Scientific, Waltham, MA, USA) equipped with an electrospray ionisation probe interface in positive and negative-ion mode. All detailed analytical conditions have been published [62].

3.4. Essential Oil Components' Analyses

The dried plant materials (100 g) were subjected to hydro-distillation using a Clevenger-type apparatus for 6 h. EO distillates, once yielded, were dried over anhydrous magnesium sulphate, filtered and then stored in dark bottles at -4 °C until further analysis. The yield was calculated as 0.52% (v/w).

The essential oil was analysed by gas chromatography-flame ionisation detector (GC-FID) and gas chromatography-mass spectrophotometry (GC-MS) techniques [63,64]. GC-MS analysis was conducted by an Agilent 5975 GC-MSD system coupled to an Agilent 7890A GC (Agilent Technologies Inc., Santa Clara, CA, USA). An HP-Innowax FSC column (60 m \times 0.25 mm, 0.25 μ m film thickness) was used with helium (purity 99.99%) as a carrier gas (1.2 mL/min). Other analytical details were reported in our previous papers [63,64]. The identification of components was based on a retention index (RI) determined by coinjection with reference to a homologous series of n-alkanes (C8–C30), under the same experimental conditions. Further identifications were achieved by comparing their mass spectra with those from NIST 05 and Wiley Eighth version, as well as by comparison of their RIs with literature values.

3.5. Total Phenolic and Flavonoid Content

Spectrophotometric methods were used to determine total phenolic and flavonoid content, as conducted previously. Standard equivalents (gallic acid equivalent (GAE) for phenolic and rutin equivalent (RE) for flavonoid) were used to assess the bioactive contents in the plant extracts [65,66].

3.6. Determination of Antioxidant and Enzyme-Inhibitory Effects

Antioxidant protocols included reducing power (cupric-reducing antioxidant capacity (CUPRAC) and ferric-reducing power (FRAP)), metal chelating, phosphomolybdenum (PBD), and free-radical scavenging (2,2-diphenyl-1-picrylhydrazyl (DPPH) and 3-

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ethylbenzothiazoline-6-sulphonic acid (ABTS)) activities. Experimental details were as described previously by [67]. Inhibitory effects of the extracts were tested against different enzymes (tyrosinase, α -amylase, α -glucosidase, and cholinesterases). Trolox and ethylene-diaminetetraacetic acid (EDTA) for antioxidant, galantamine for cholinesterases, kojic acid for tyrosinase, and acarbose for α -amylase and α -glucosidase were used to express antioxidant and enzyme-inhibitory results.

3.7. Cell Culture

The human hepatocarcinoma HepG2 cells and murine bone marrow stromal S17 cells were kindly provided by the Centre for Molecular and Structural Biomedicine of Biomedical and Molecular BME, University of Algarve, Portugal), while mouse melanoma B16 4A5 cells were purchased from Sigma-Aldrich (Taufkirchen, Germany). All cell lines were cultured in Dulbecco's Modified Eagle medium (DMEM), supplemented with foetal bovine serum (10%), L-glutamine (2 mM, 1%), and penicillin (50 U/mL)/streptomycin (50 μ g/mL) (1%), and kept under a humidified atmosphere at 37 °C and 5% CO₂.

3.8. Determination of Cellular Viability and Selectivity

Cells were plated in 96 well plates at 5×10^3 cells/well (HepG2 and S17) and 2×10^3 cells/well (B16 4A5). After a 24 h incubation period, cells were treated with the samples at the concentration of 100 μ g/mL for 72 h. Cells incubated with DMSO at 0.5% (the highest DMSO concentration used in the test wells) were used as the control. The cellular viability was determined by the MTT 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) test, as described formerly [68]. The percentage of viable cells was calculated relative to the control (DMSO, 0.5%).

3.9. Statistical Analysis

All quantitative analyses were performed in triplicate (n = 3), and data were expressed as means \pm S.D. Significant differences in the tested samples were determined by an ANOVA (Tukey test), with a probability value of 5%. Pearson's correlation was estimated to identify the relationship between the total amounts of phenolics and flavonoids, and the biological activities (antioxidant and enzyme-inhibitory effects). R software (Version 3.6.2) was used for the statistical analysis.

4. Conclusions

For the first time, the biological activities and phytochemical profiles of the aerial parts of *N. baytopii*, endemic from Turkey, were evaluated. Extraction of the aerial parts was performed using solvents of different polarity. Furthermore, the essential oil of the plant was prepared by hydro-distillation. The water/methanol and water extracts possessed appreciable amounts of phenolic compounds and showed the highest antioxidant capacities in vitro. Phytochemical profiling revealed the presence of flavones, such as luteolin and apigenin and their derivatives in both the water and methanol extracts. The ethyl acetate extract showed pronounced inhibitory properties against butyrylcholinesterase, highlighting the possibility for a new, efficient Alzheimer's disease therapeutic agent. The high cytotoxicity of *N. baytopii* aerial parts methanol extract against HepG2 suggests further future investigations in this area. The data presented here showed that the endemic *N. baytopii* possessed many interesting biological activities and is certainly encouraging for a future application in the pharmaceutical and nutraceutical fields, although further tests are necessary.

Supplementary Materials: The following are available online at https://www.mdpi.com/article/10 .3390/plants10061176/s1, Figure S1. Total ion chromatograms of methanol extract in positive ion mode (a) and negative ion mode (b). Figure S2. Total ion chromatograms of water extract in positive ion mode (a) and negative ion mode (b).

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