

# Understanding aerosol-cloud-climate interactions

**Athanasios Nenes**

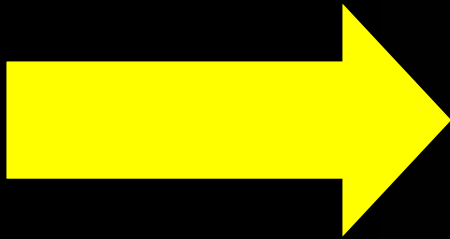
School of Earth & Atmospheric Sciences  
School of Chemical & Biomolecular Engineering  
Georgia Institute of Technology

**CO<sub>2</sub> Forum Seminar**  
Georgia Institute of Technology  
2 April 2009

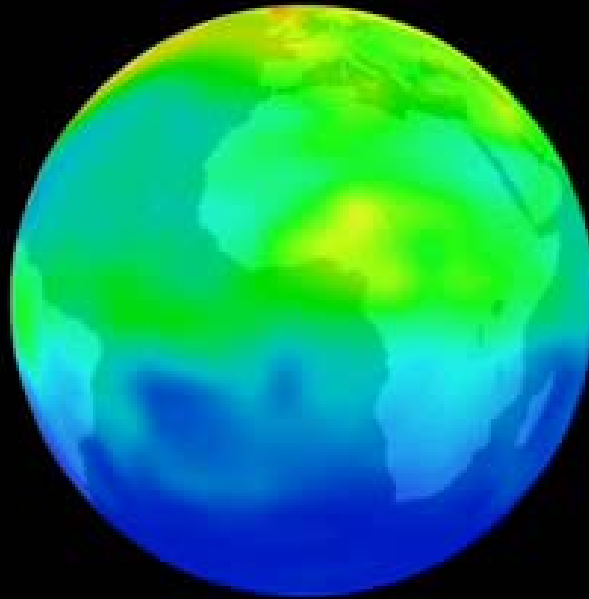
**Acknowledgments: NASA, NSF, NOAA, ONR, CIRPAS  
Seinfeld/Flagan group (Caltech), Adams group (CMU),  
Smith group (NCAR)**

# Earth's Energy Balance

**Sunlight** (this is what we call “visible radiation”)



235 Watts per square meter ( $\text{Wm}^{-2}$ )



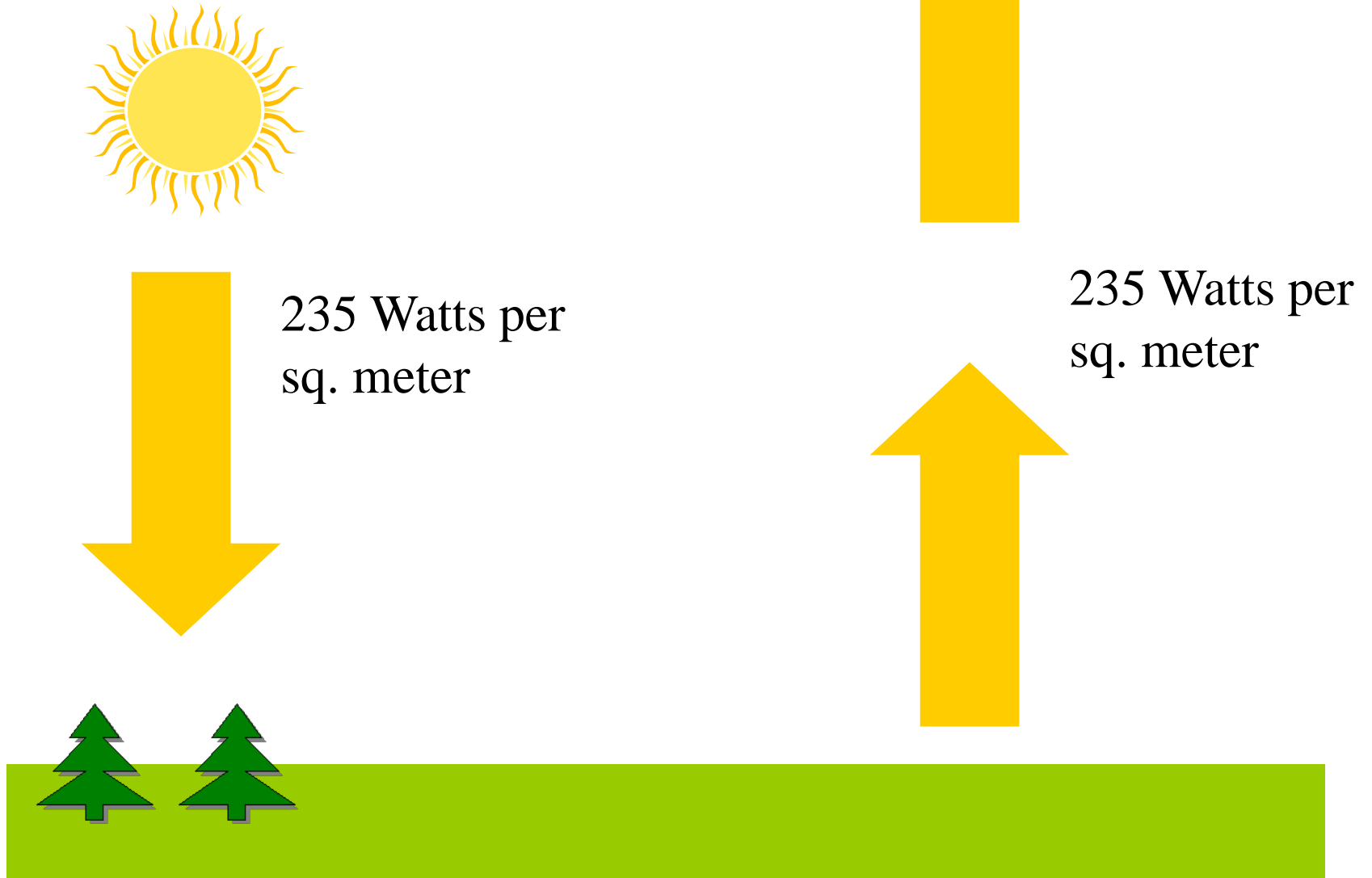
**Heat** (this is what we call “infrared radiation”)



235 Watts per square meter ( $\text{Wm}^{-2}$ )

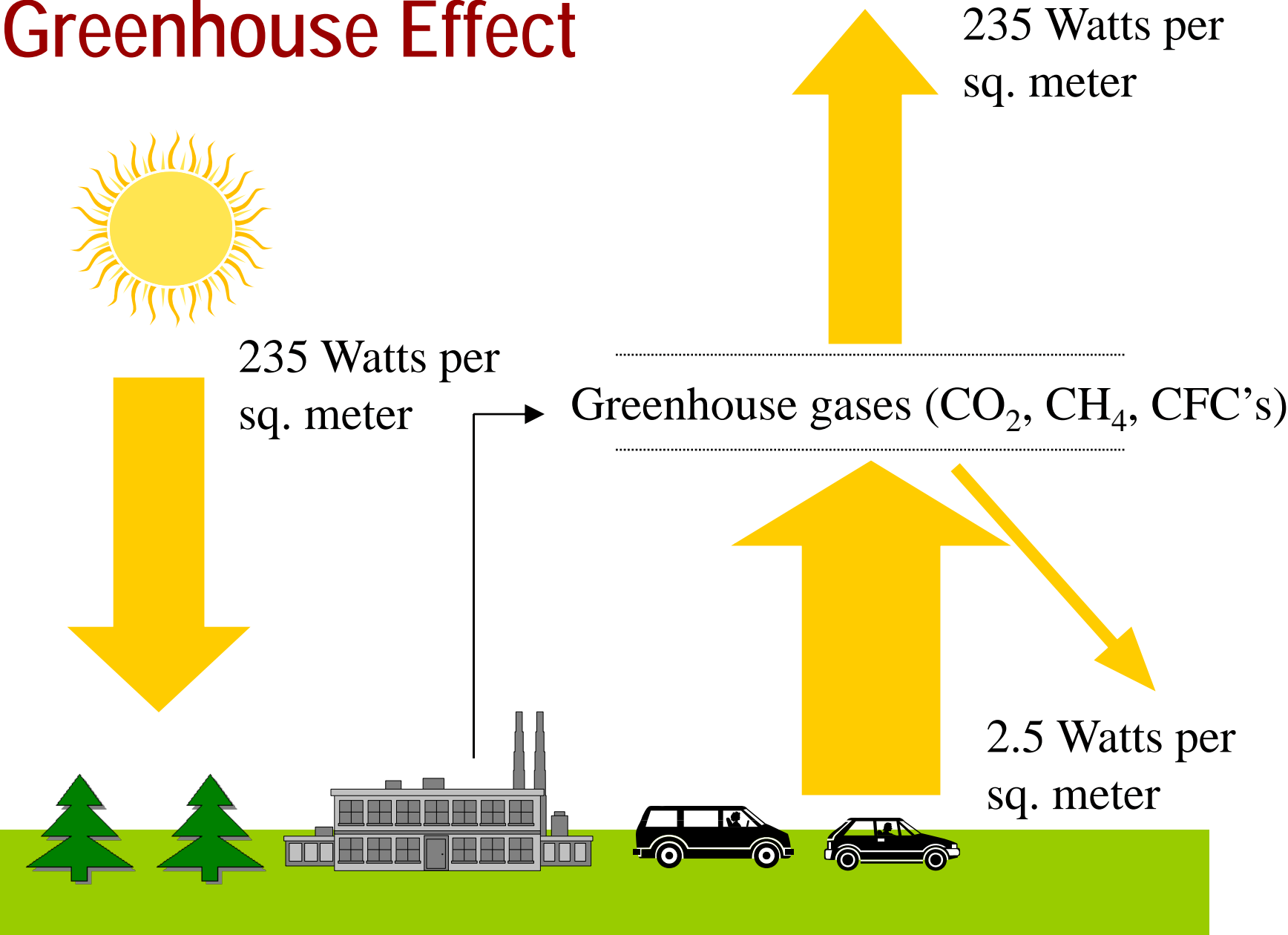
When **energy IN** = **energy OUT**, climate is “in balance” (i.e., steady state)

# Greenhouse Effect

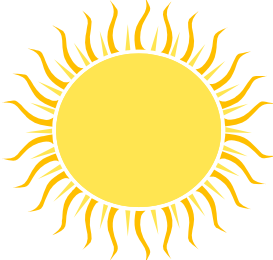


“Natural”, “Preindustrial” state of climate (before late 1700’s)

# Greenhouse Effect



# Greenhouse Effect



235 Watts per  
sq. meter

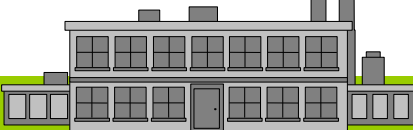
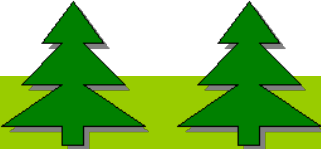


Greenhouse gases (CO<sub>2</sub>, CH<sub>4</sub>, CFC's)

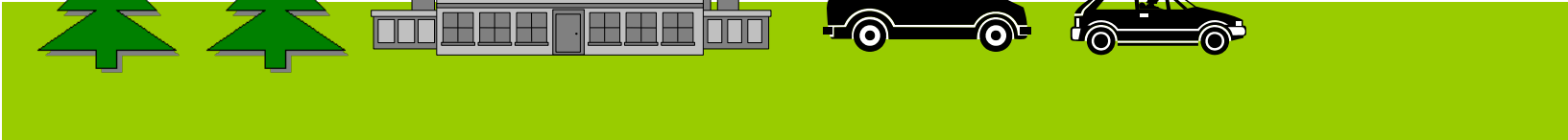


235 Watts per  
sq. meter

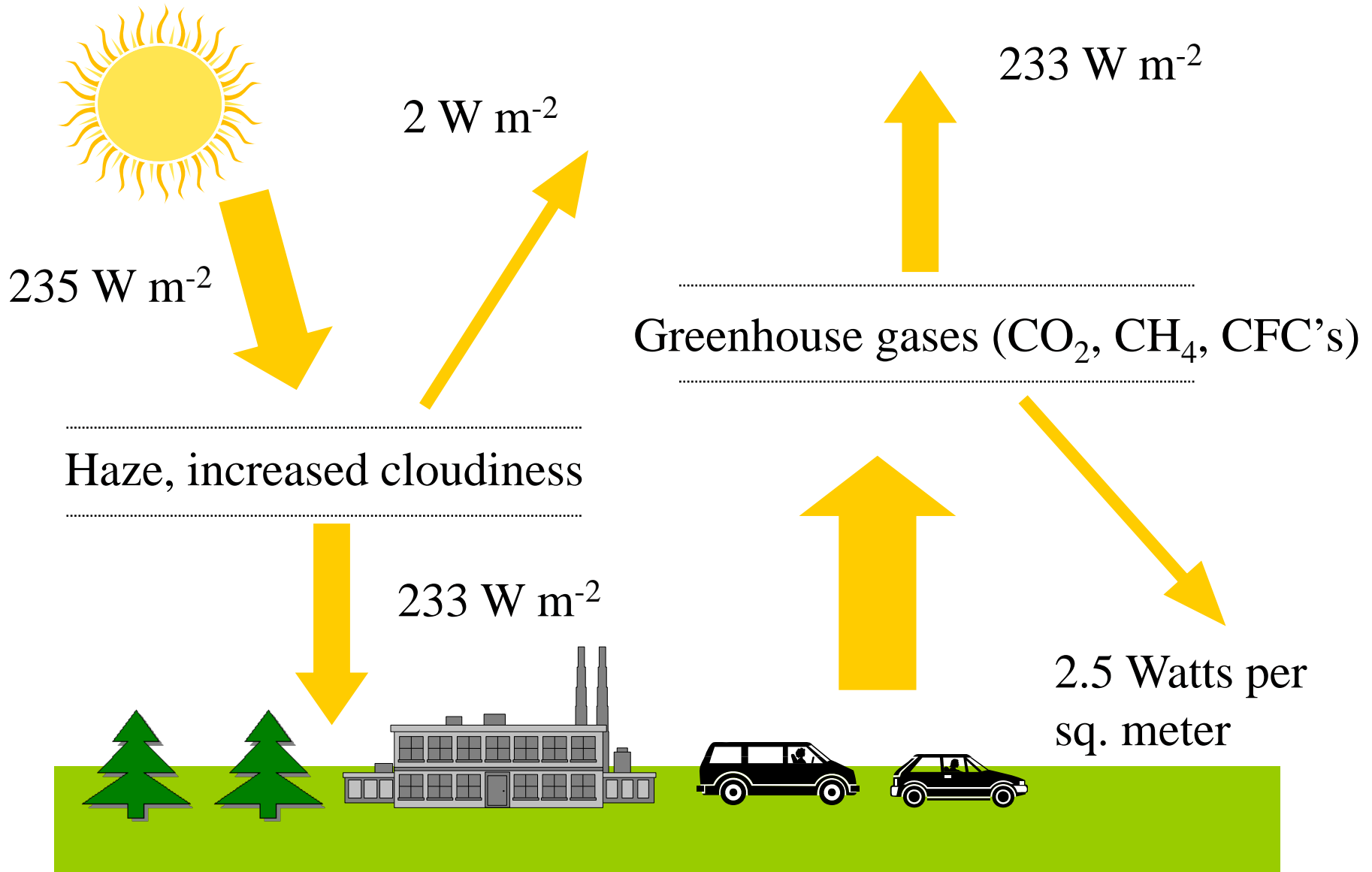
Energy accumulates and surface temperature increases:  
this is "global warming"



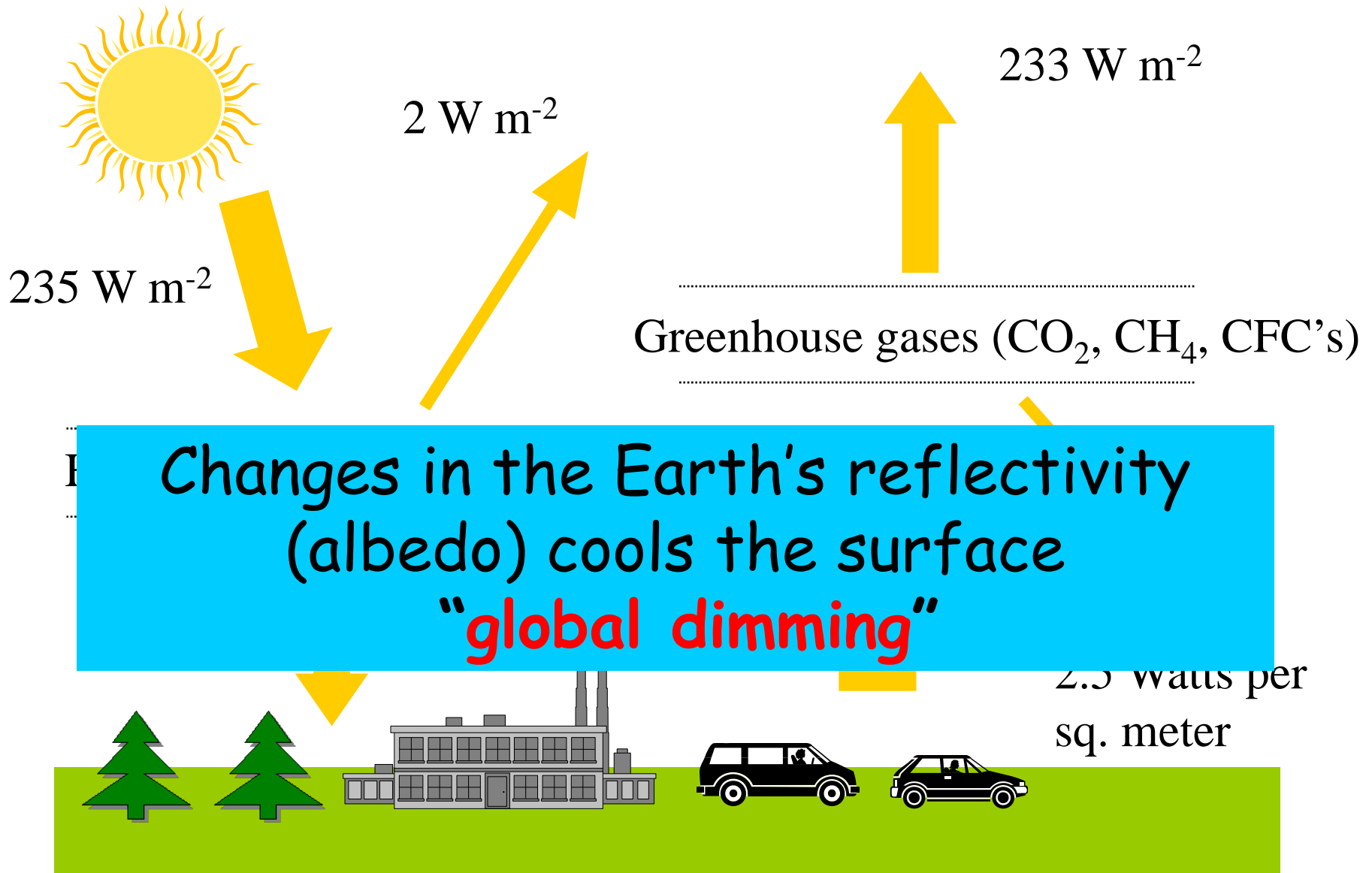
sq. meter



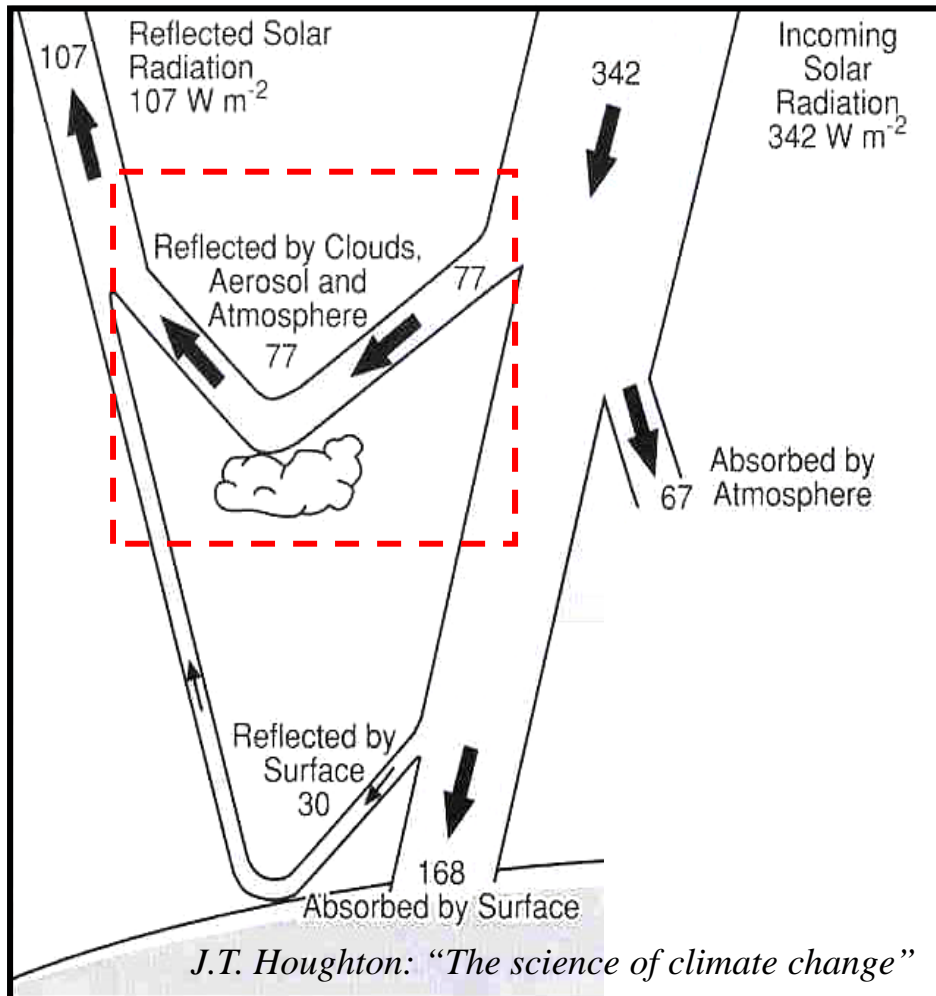
# “Whitehouse” Effect



# "Whitehouse" Effect



# Earth's Albedo: controlling factor of "global dimming"



## *Facts:*

- Clouds account for ~50% of planetary reflectivity (albedo).
- Small changes in clouds yield large changes in global energy balance.
- **A few %** increase in global cloud cover can counteract warming from greenhouse gases.

## *Consequence:*

**Understanding cloud formation is required for assessments of anthropogenic climate change.**

*Clouds are VERY dynamic (difficult to simulate).*



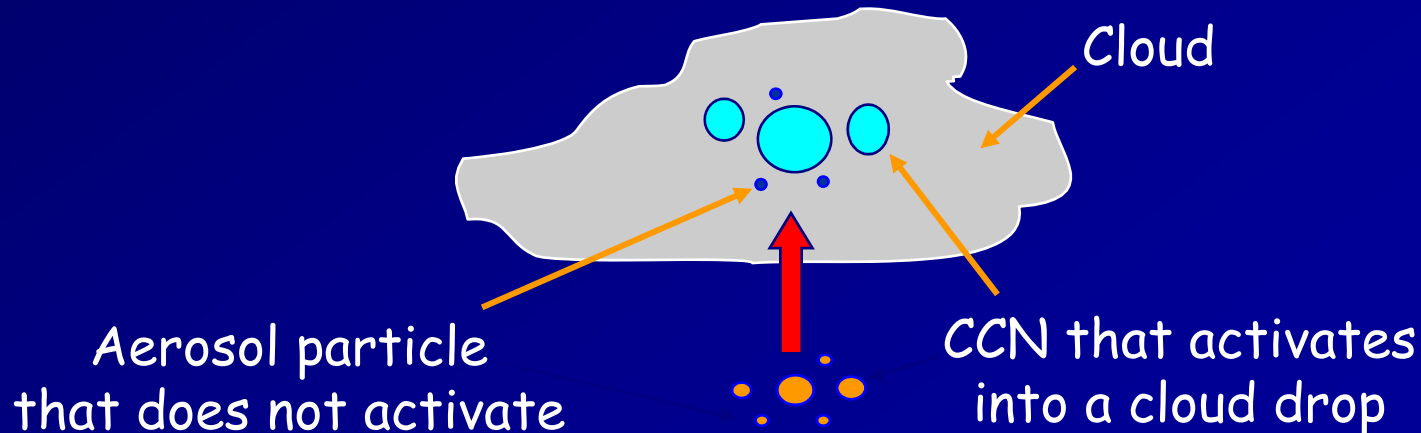
# How do (liquid water) clouds form?

Clouds form in regions of the atmosphere where there is too much water vapor (it is "supersaturated").

This happens when air is cooled (primarily through expansion in updraft regions and radiative cooling).

Cloud droplets nucleate on pre-existing particles found in the atmosphere (aerosols). This process is known as activation.

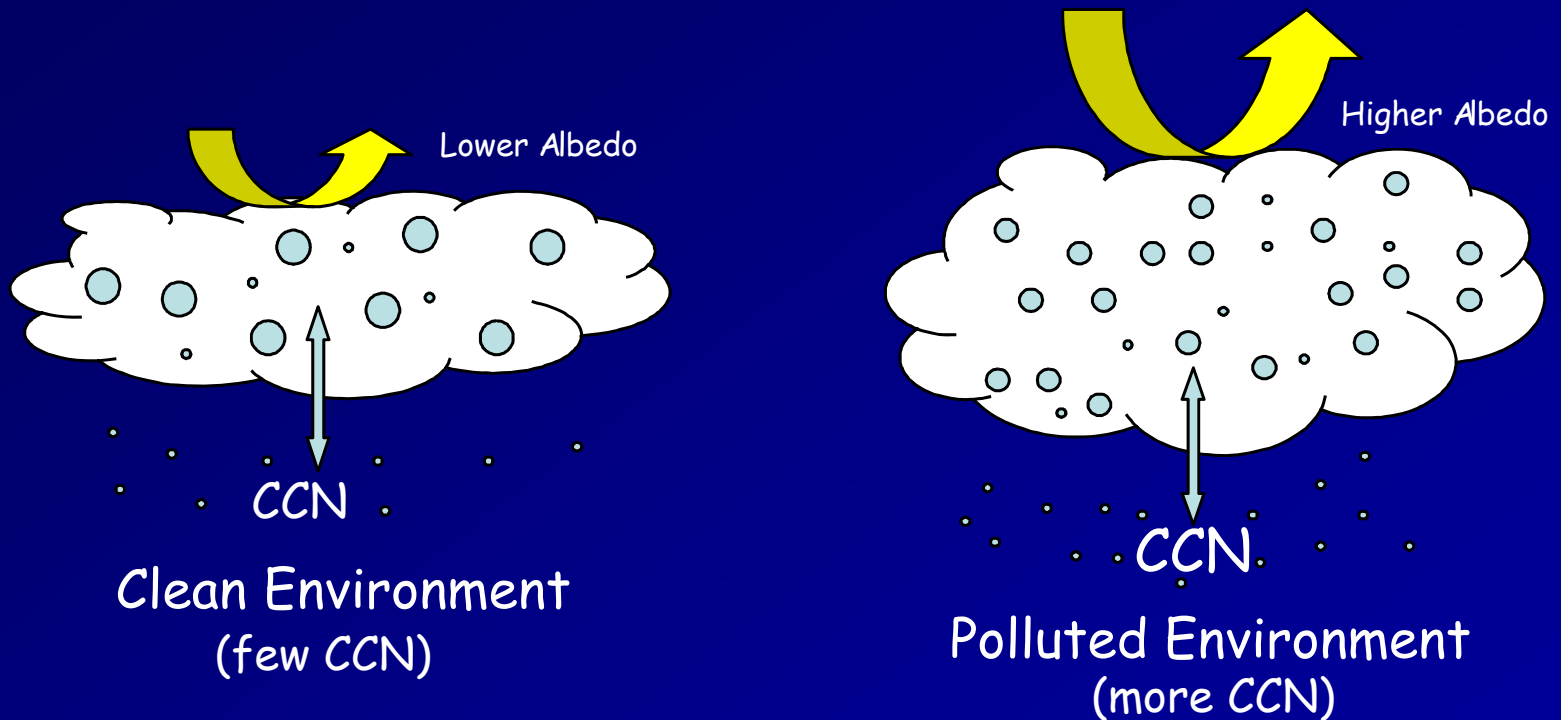
Aerosols that can become droplets are called cloud condensation nuclei (CCN).



# Can humans affect clouds and the hydrological cycle?

Yes! By changing global CCN concentrations (air pollution).

*Result.* Clouds tend to be "whiter", change in precipitate efficiency. This yields a net cooling on climate and is called the "**indirect climatic effect of aerosols**".



Increasing particles tends to cool climate (potentially alot).

Quantitative assessments done with climate models.

# Humans have a large impact on global aerosol

Pollution is a global problem. CCN are emitted together with greenhouse gases.



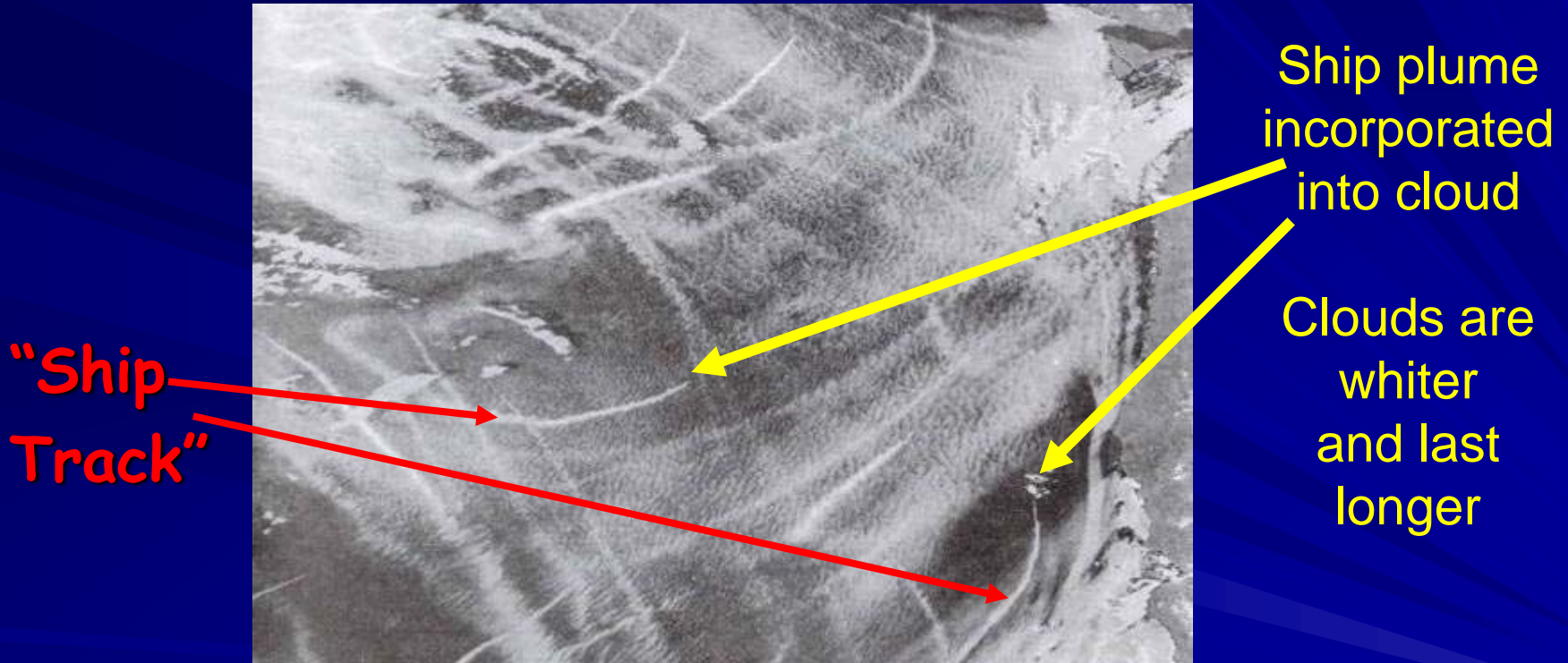
Pollution plumes  
(SE Asia, Summer 2002).



Biomass burning  
(Mediterranean, Summer 2007).

# Observational evidence of indirect effect

"Ship tracks": features of high cloud reflectivity embedded in marine stratus. A result of ship plumes affecting clouds above.



Pollution  $\uparrow$   $\Rightarrow$  Droplet number  $\uparrow$   $\Rightarrow$  Droplet size  $\downarrow$

Droplet size  $\downarrow$   $\Rightarrow$  Cloud reflectivity  $\uparrow$  **AND** Precip  $\downarrow$

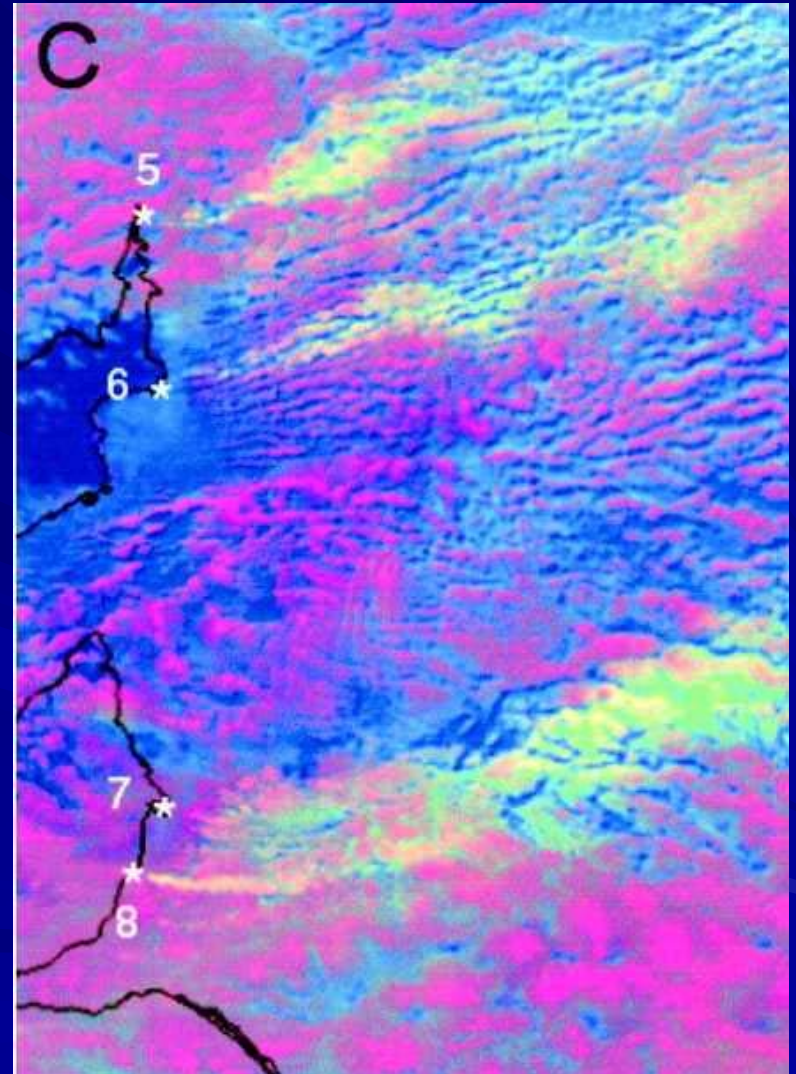
# Observational evidence of indirect effect

Satellite observation of clouds in the Black Sea.

**Red:** Clouds with low reflectivity.

**White:** Clouds that reflect a lot.

**Blue:** Clear sky.



# Observational evidence of indirect effect

Air pollution can affect cloud properties

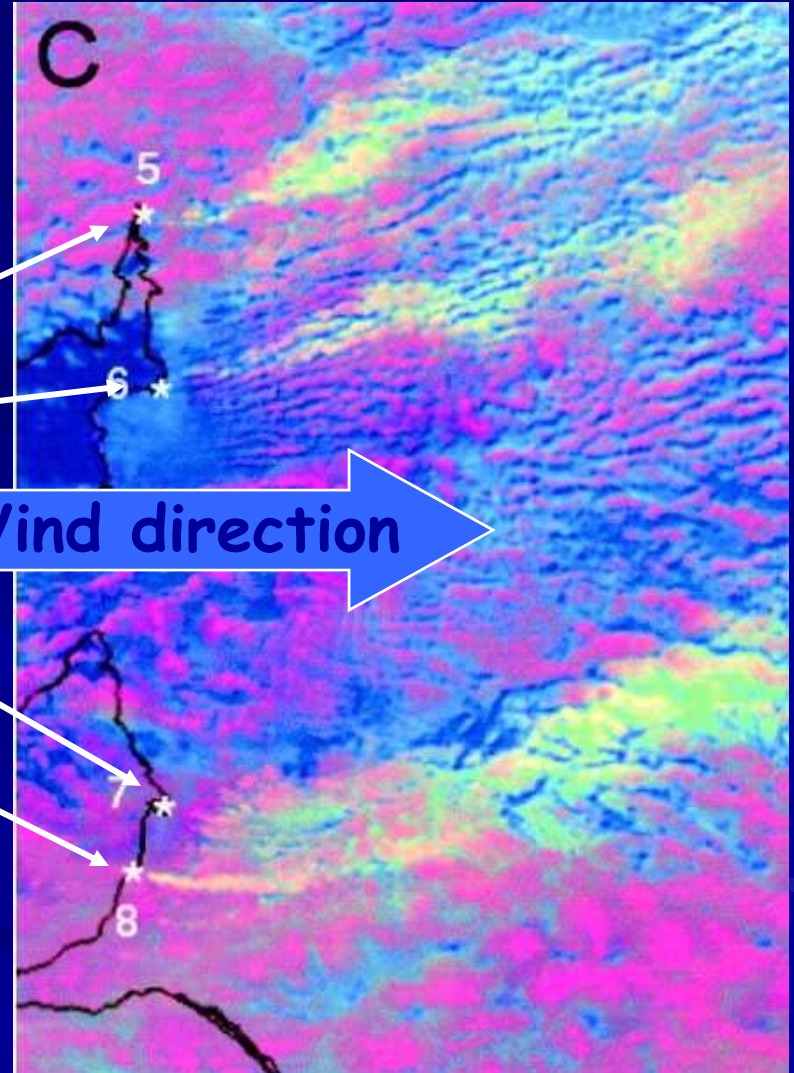
Satellite observation of clouds in the Black Sea.

Power plant

Lead smelter

Port

Oil refineries



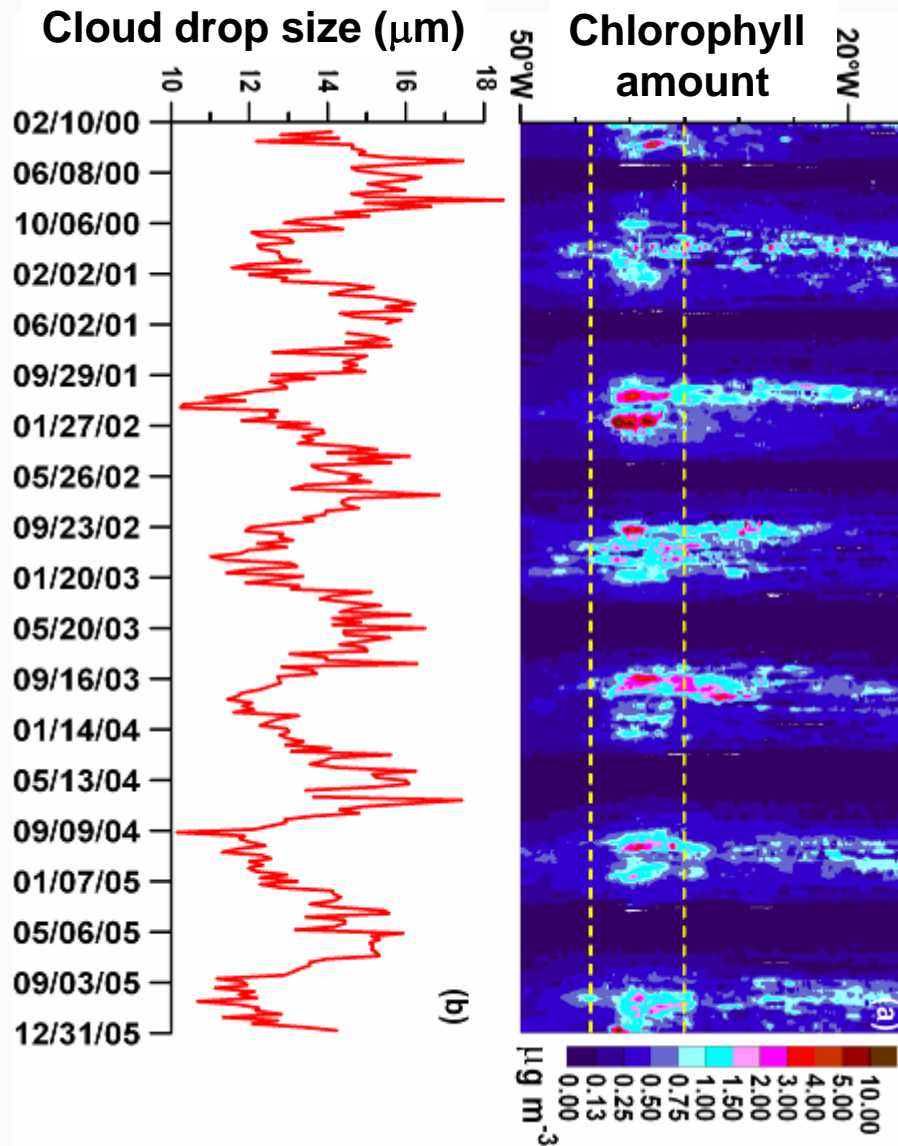
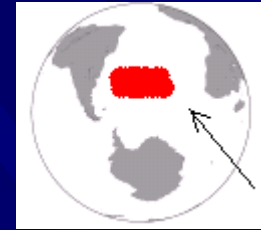
Red: Clouds with low reflectivity.

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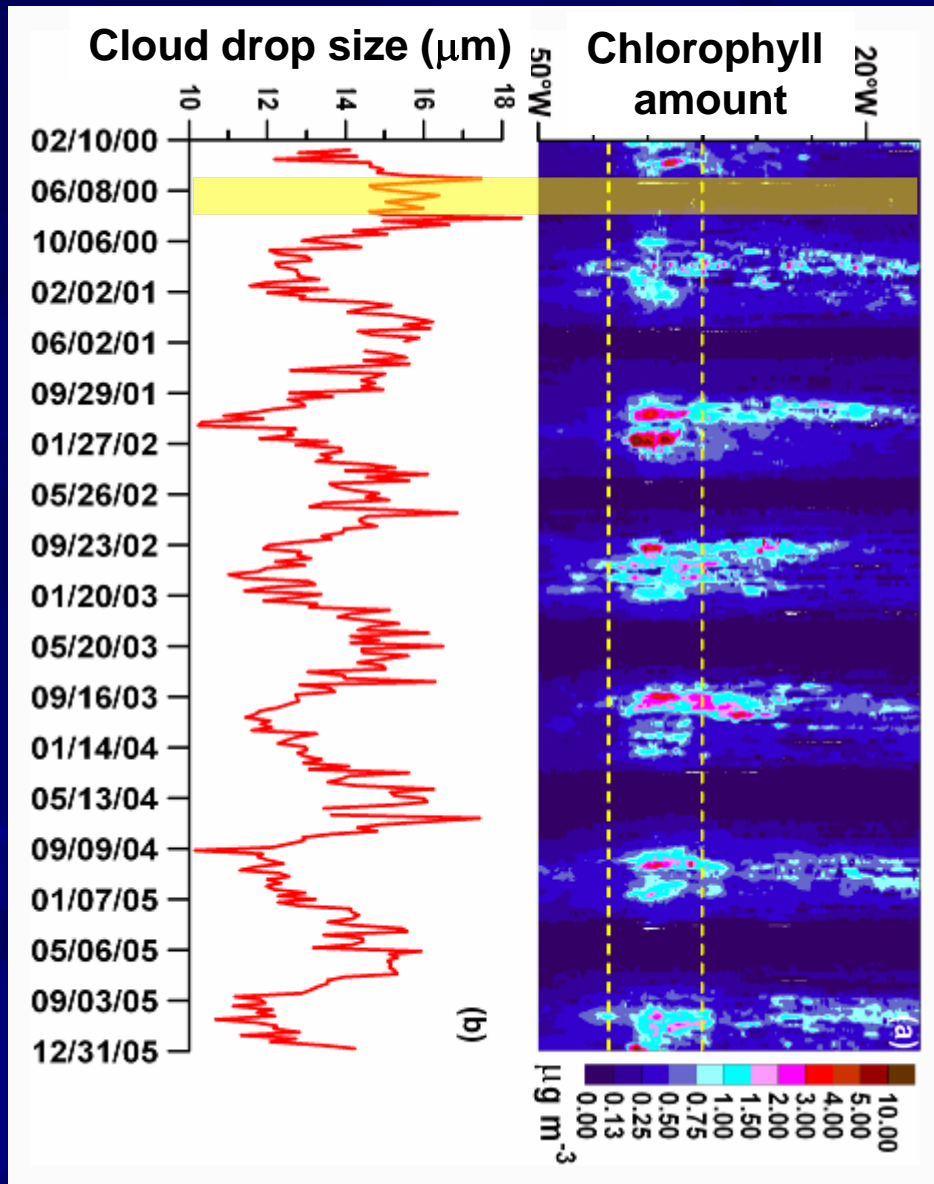
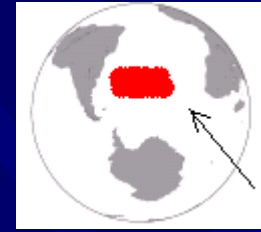
# Phytoplankton affect clouds too...

Location: East of Patagonia (South America)



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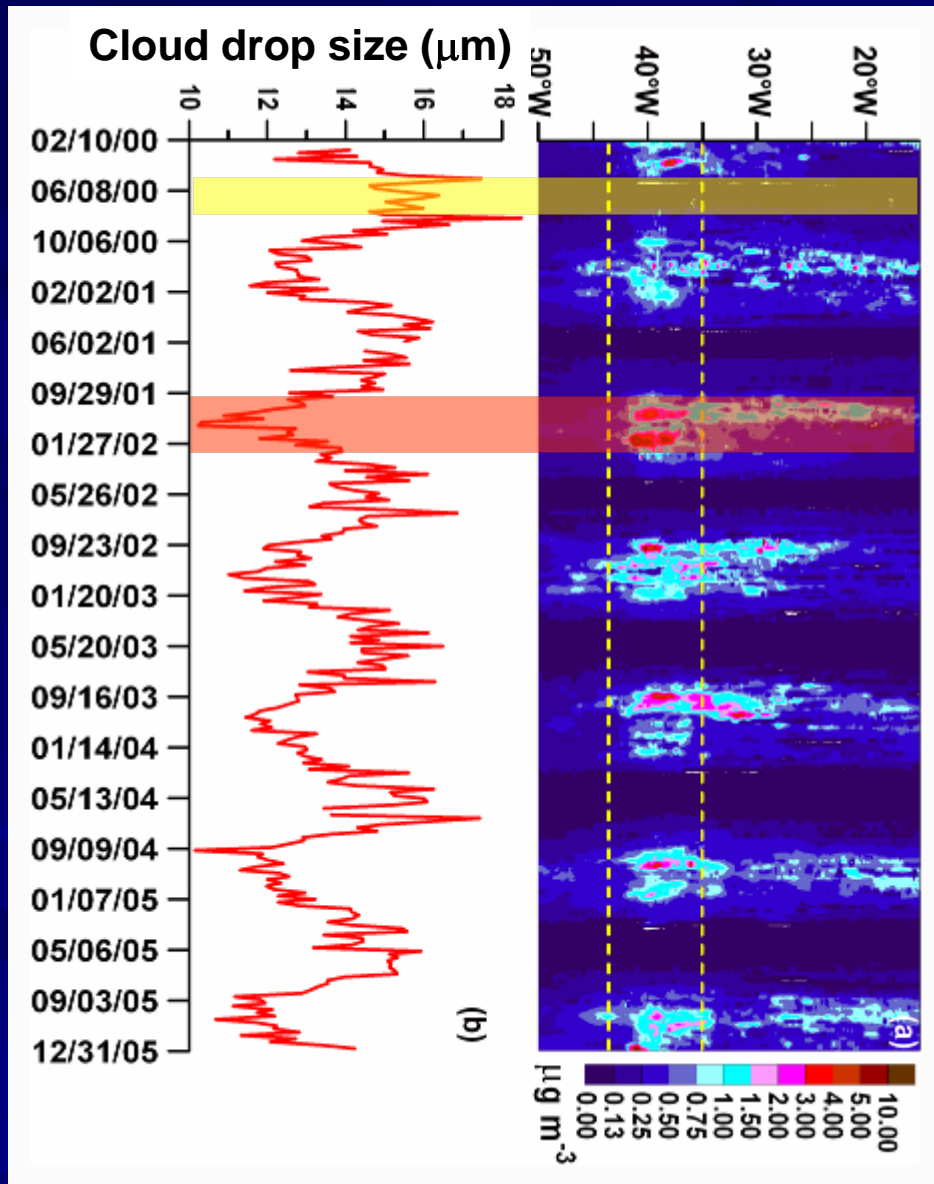
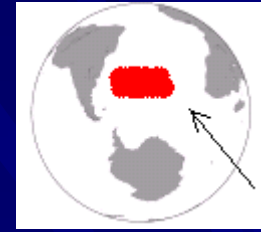


Low chlorophyll period,  
clouds have large drops  
(not very reflective)



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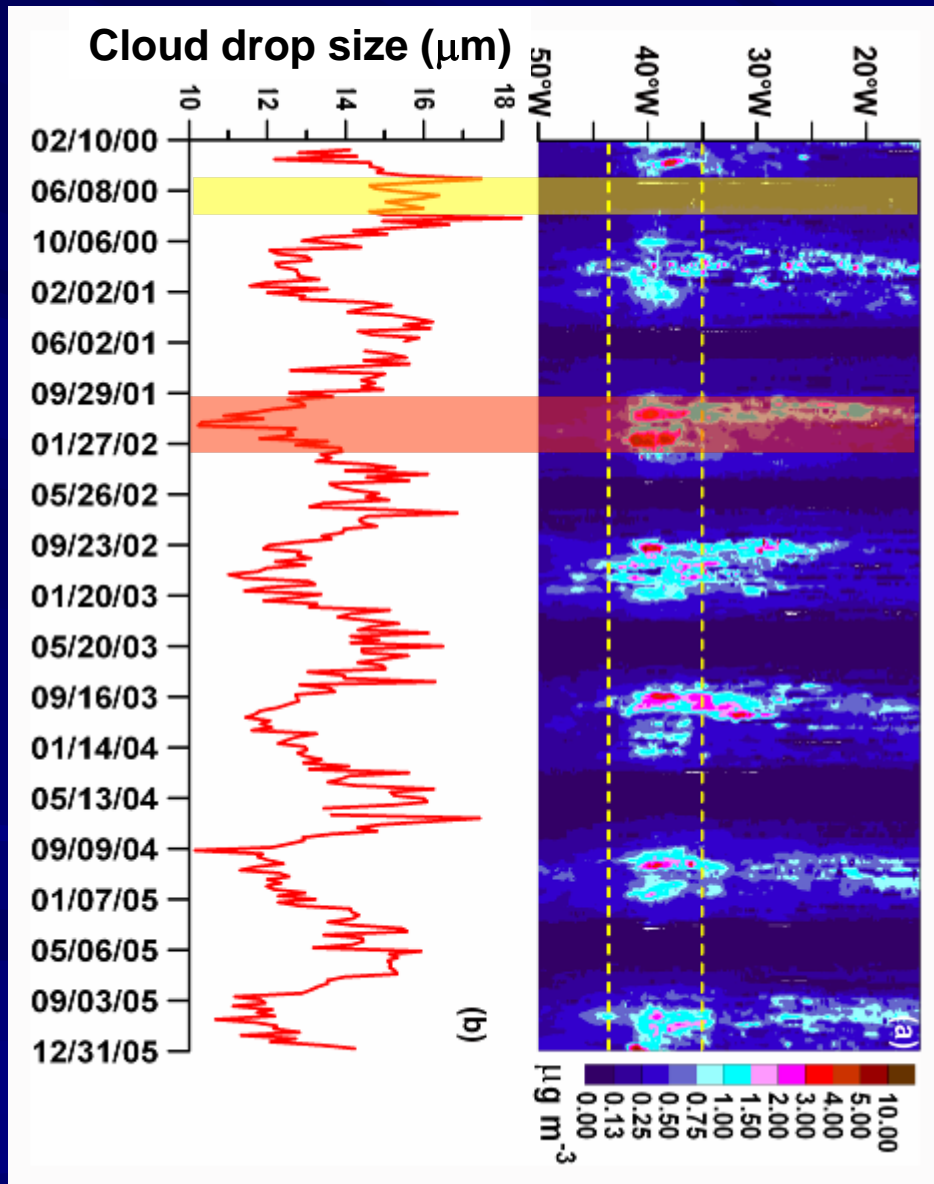
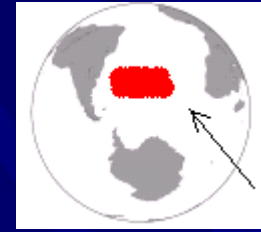


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Location: East of Patagonia (South America)



← Low chlorophyll period,  
clouds have large drops  
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← High Chlorophyll period,  
Clouds have small drops  
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Phytoplankton emissions  
increase particle loads, and  
strongly impact clouds.

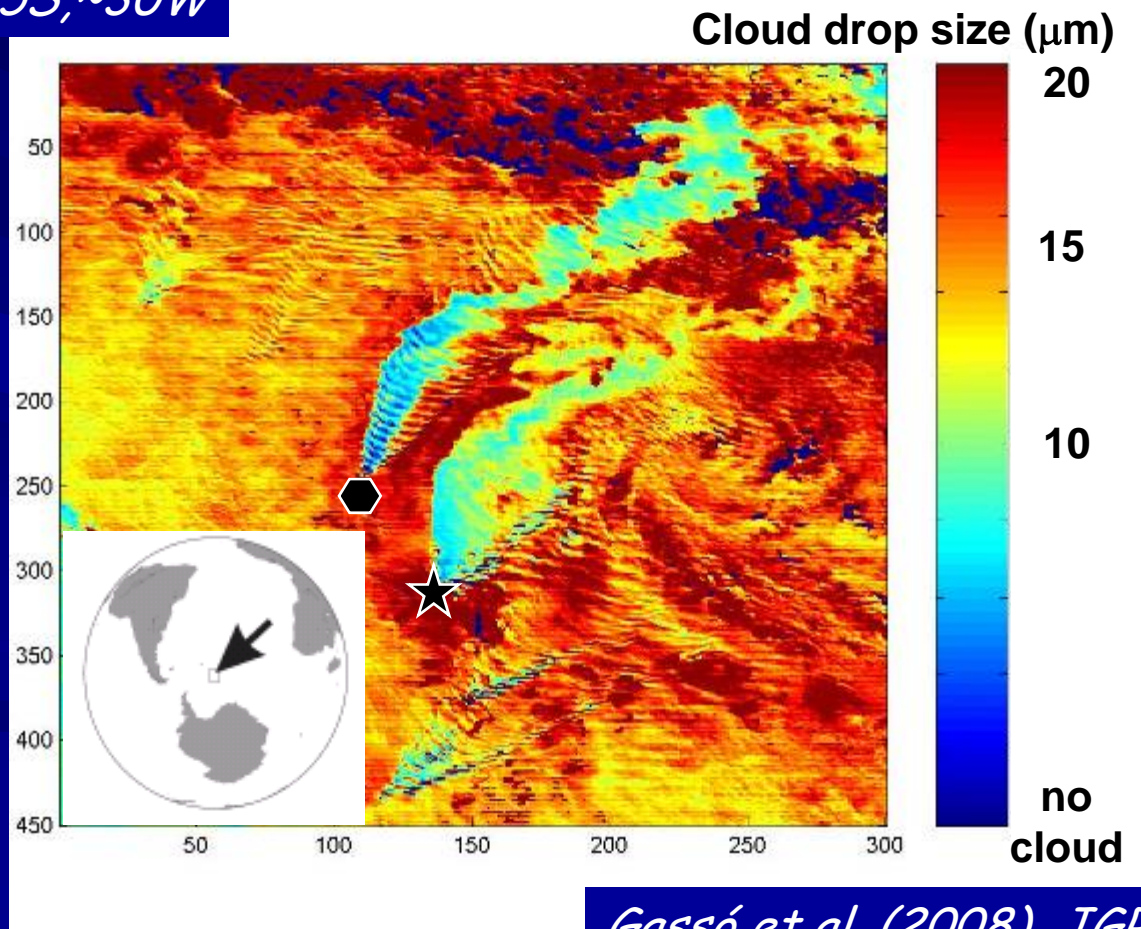
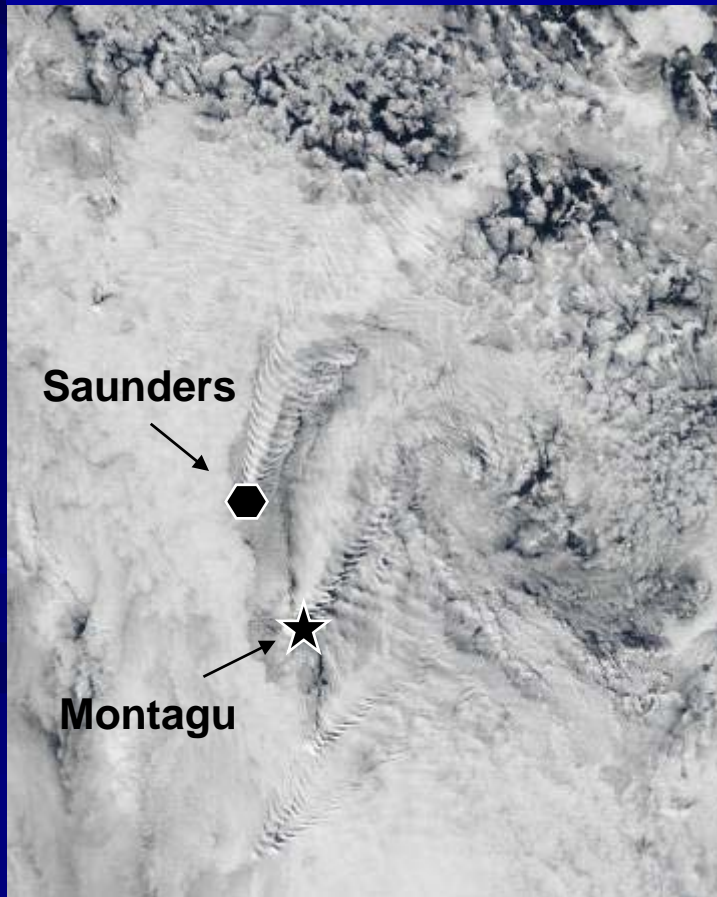
Changes are comparable to  
contrasts between polluted  
and clean environments  
(forcing  $\sim -15 \text{ W m}^{-2}$ ).

*Meskhidze and Nenes, Science, 2006*

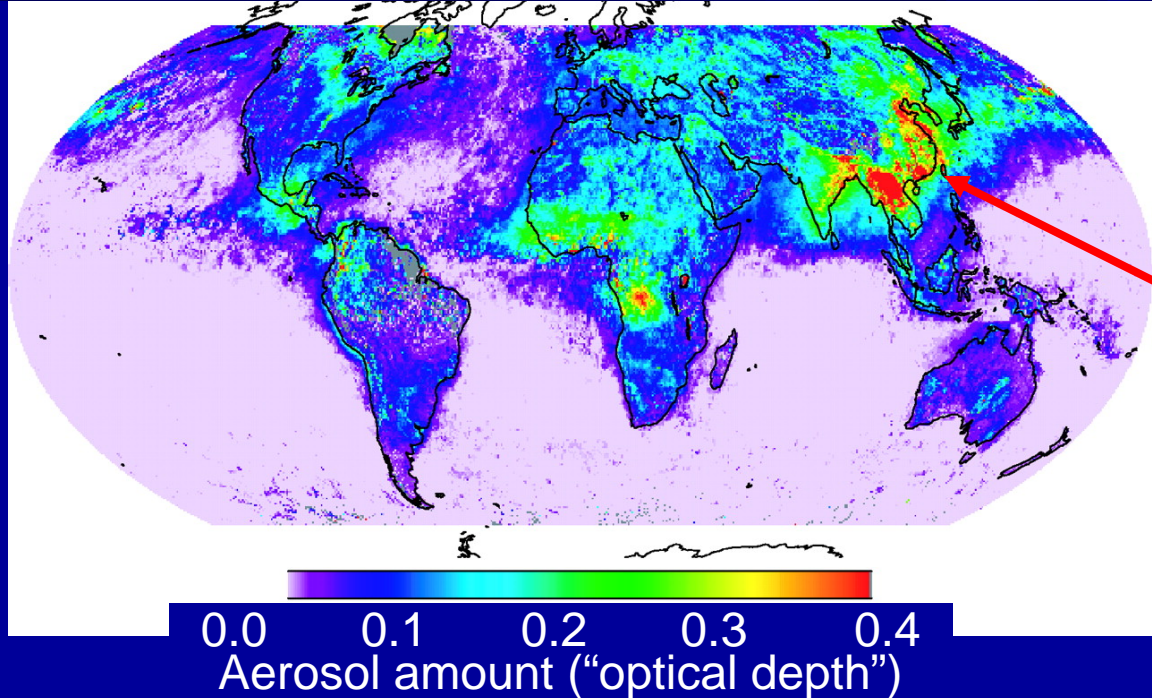
# So do volcanoes (even when "sleeping") ...

Volcanoes continuously emit  $\text{SO}_2$  which becomes sulfate aerosol. The aerosol can substantially increase CCN in volcanic plumes. Cloud in the plume are much more reflective than outside.

Location: Sandwich Islands,  $\sim 55\text{S}, \sim 30\text{W}$



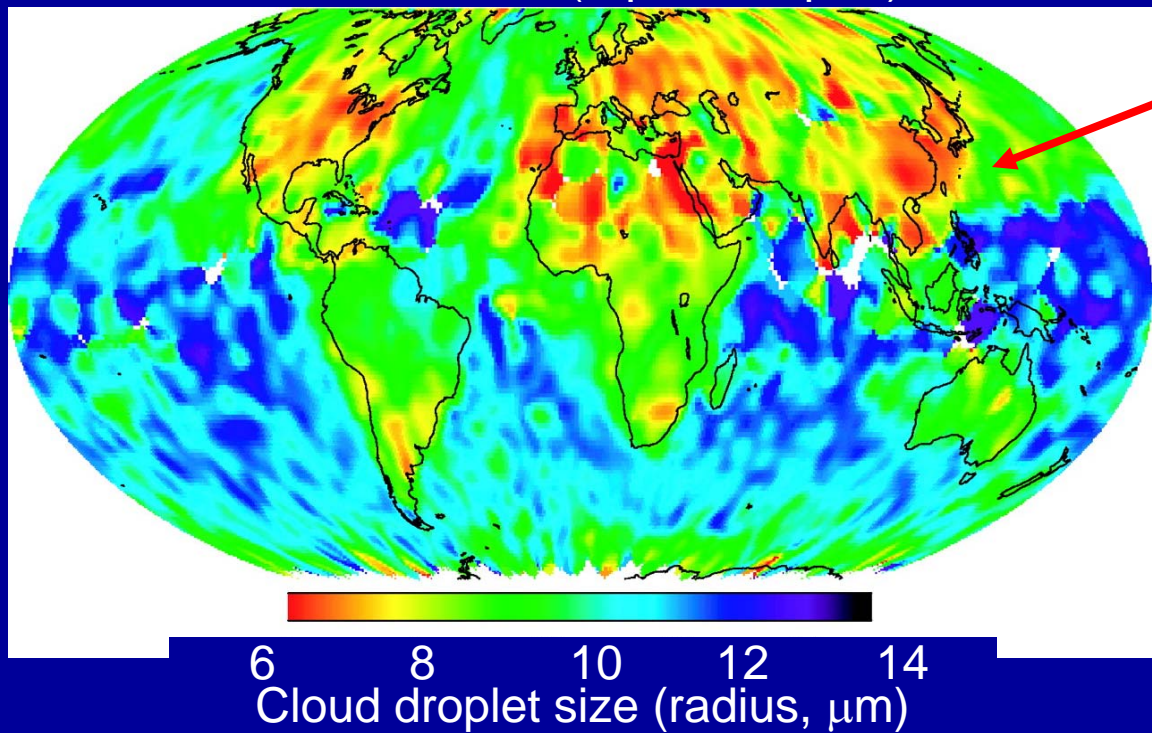
*Gassó et al. (2008), JGR*



**A remote sensing  
global picture...**

A lot of aerosol...

...gives smallest  
cloud droplets

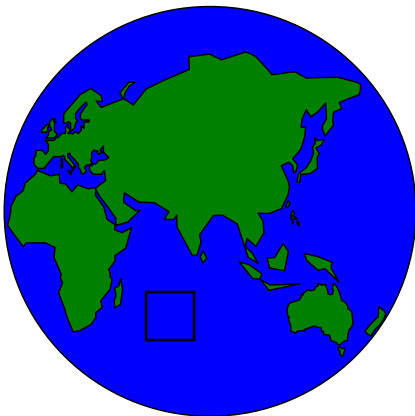


**We see the same  
on all satellite  
platforms...**

Breón et al. (2002)

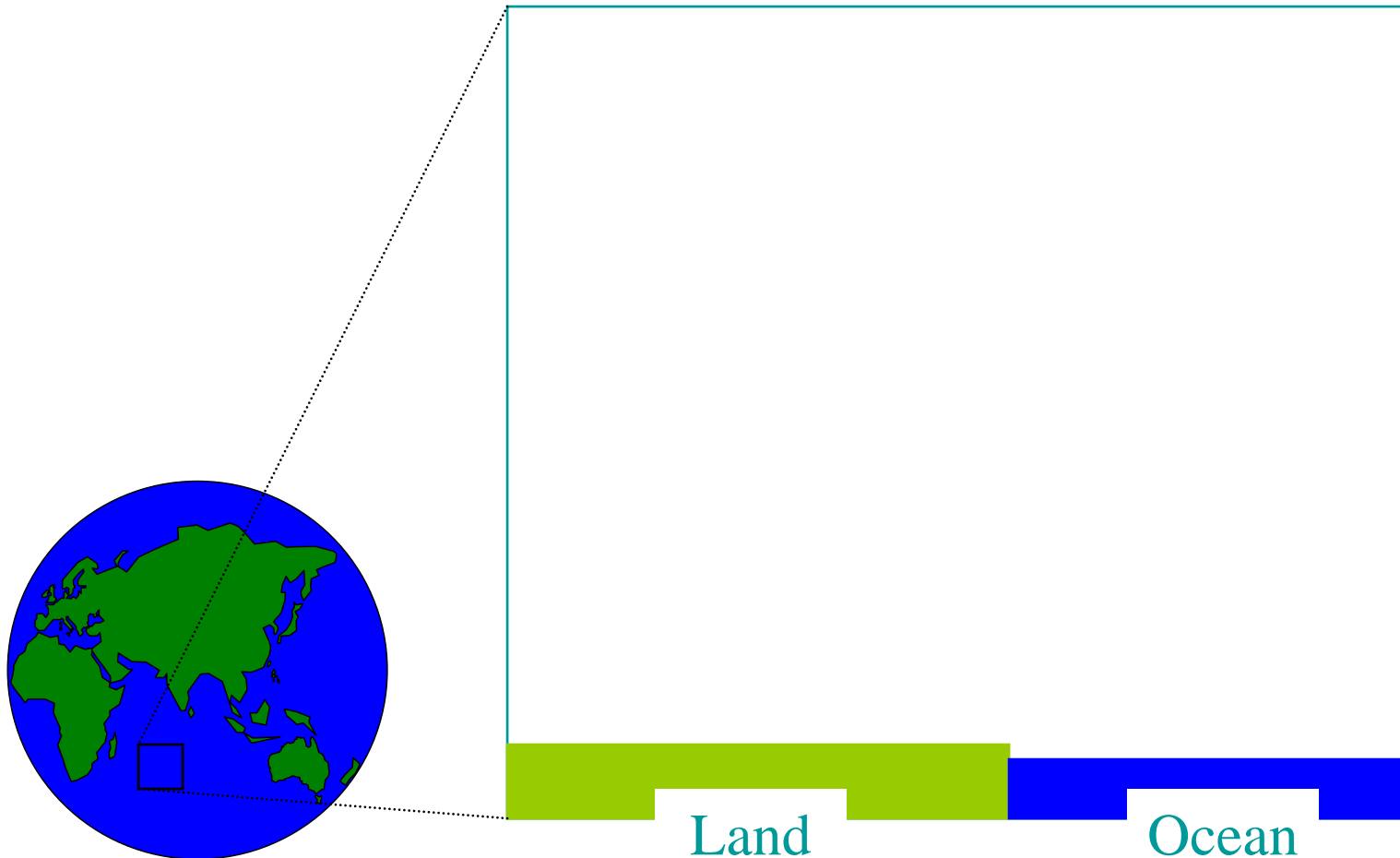
# Global Climate Models: Tools of understanding

- Divide the Earth into small parts (“grid cells”).
- Write differential equations that describe conservation of
  - Energy, Water and other Chemical components
  - Incorporate chemical reactions
  - Incorporate phase changes (for example water-ice-vapor)
- Prescribe initial conditions (temperature everywhere).
- Integrate the equations (numerically) over time.



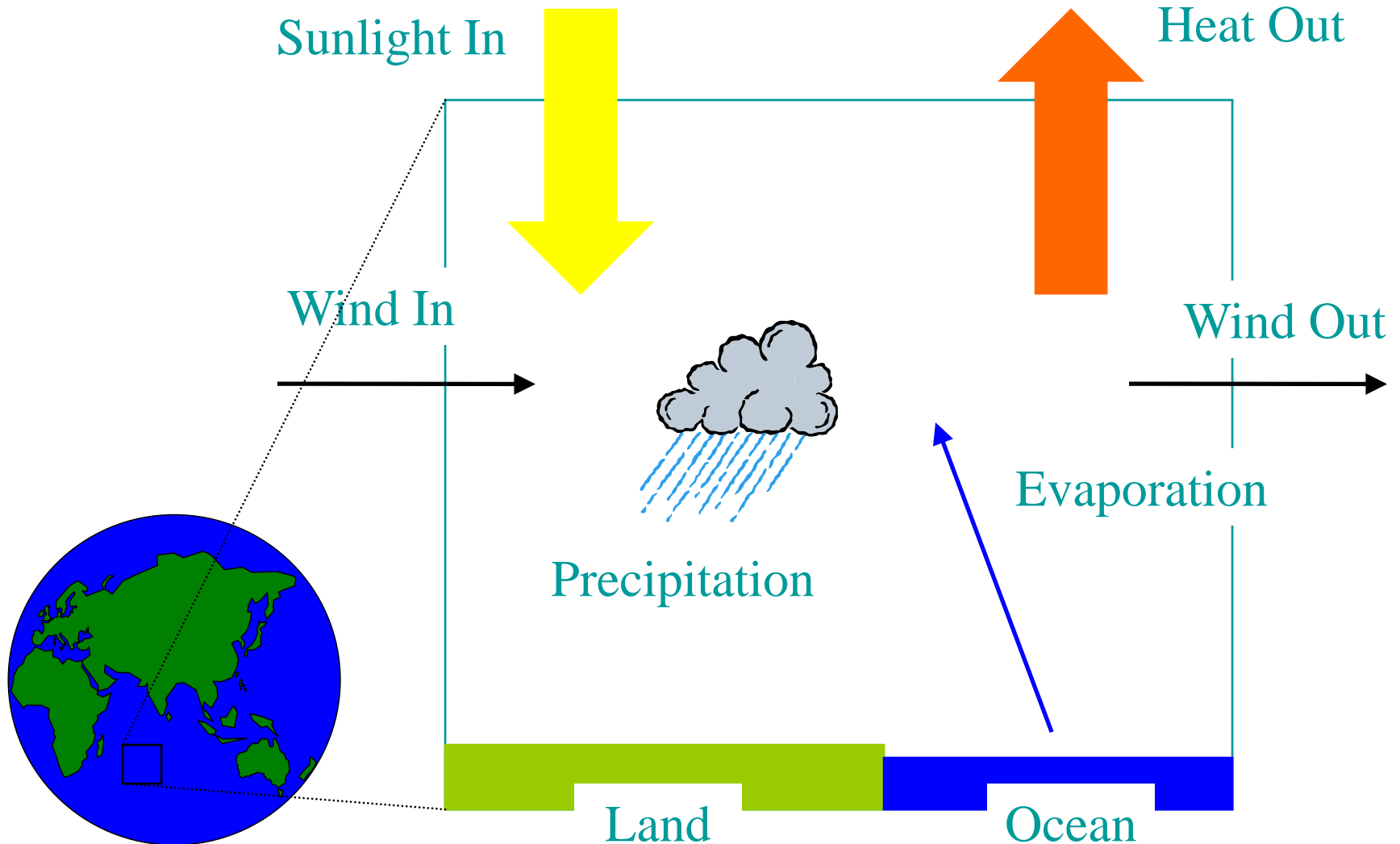
# How Computer Climate Models Work

Example: conservation of energy in the atmosphere



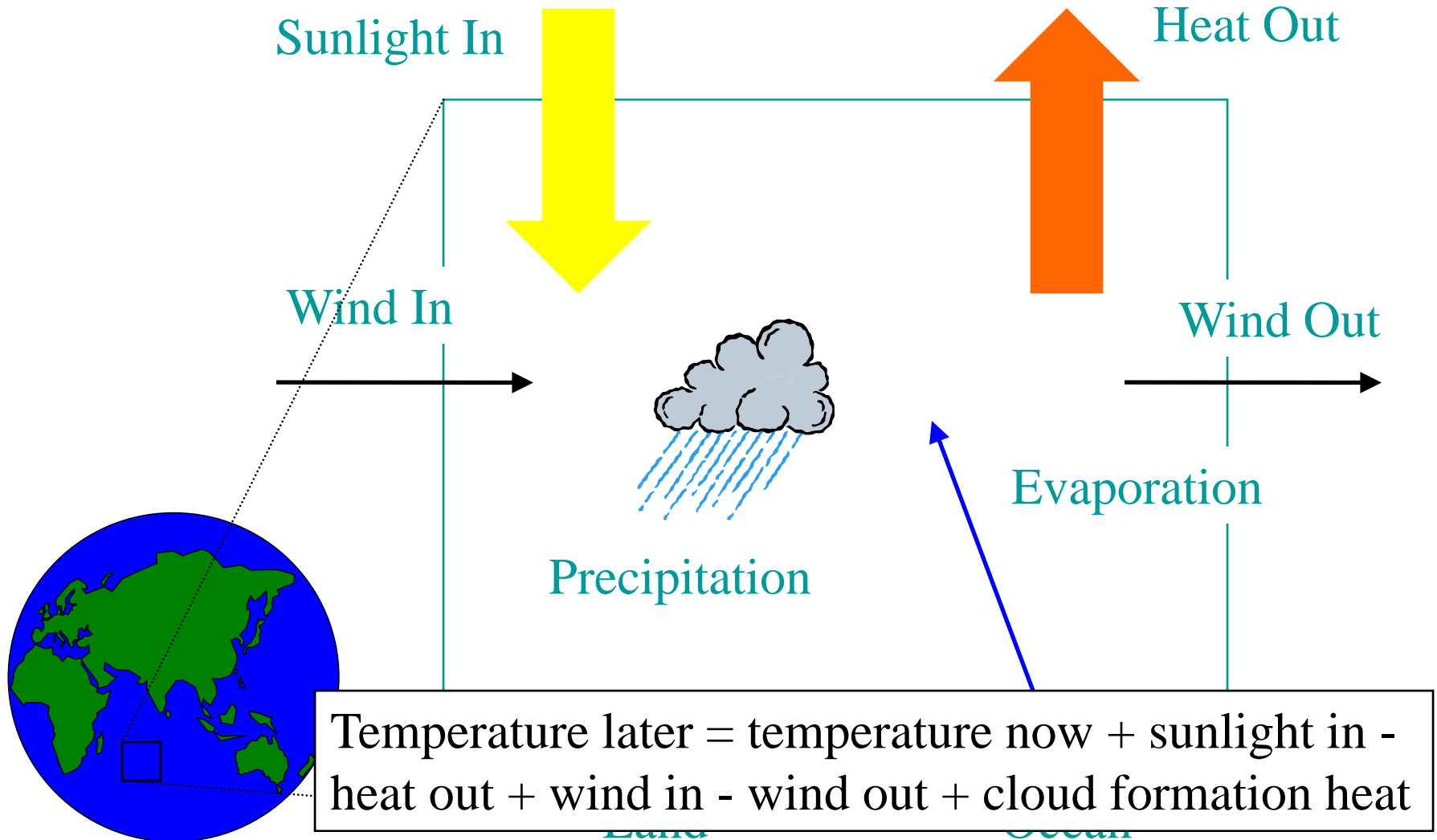
# How Computer Climate Models Work

Example: conservation of energy in the atmosphere



# How Computer Climate Models Work

Example: conservation of energy in the atmosphere





# Anthropogenic Indirect Effect: How do we estimate its global impact?

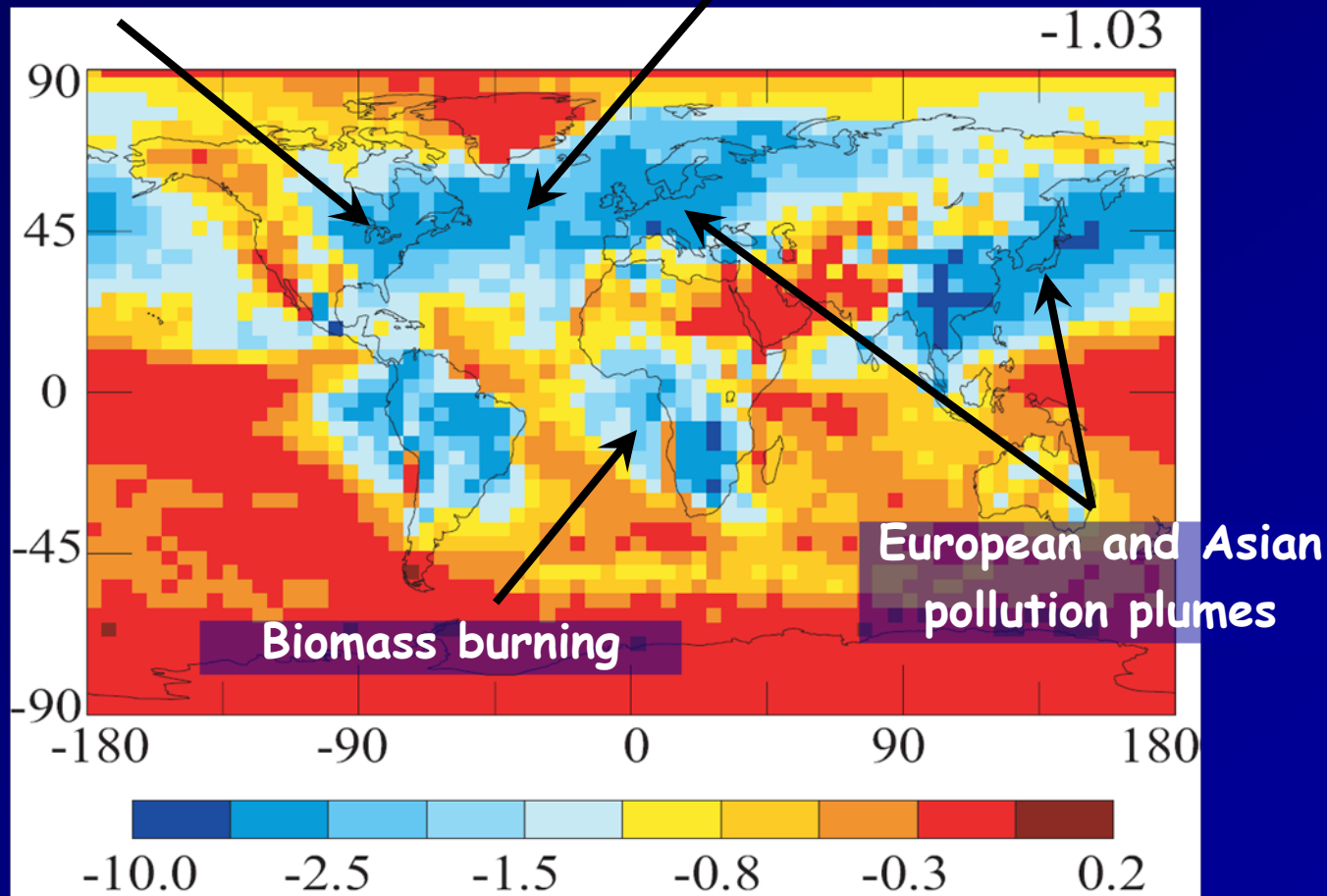
We use a global climate model (GCM)

- simulation with "current day" emissions
- simulation without anthropogenic emissions ("preindustrial" emissions)
- compute the change in energy (radiation) between two simulations ("indirect forcing")
- compare annual average forcing to greenhouse gas warming ( $\sim 2.5 \text{ W m}^{-2}$ )
- Net forcing (greenhouse + indirect) can be used as an index for climate change.

# Indirect Forcing calculation ( $W m^{-2}$ )

North America  
pollution plumes

Long-range transport

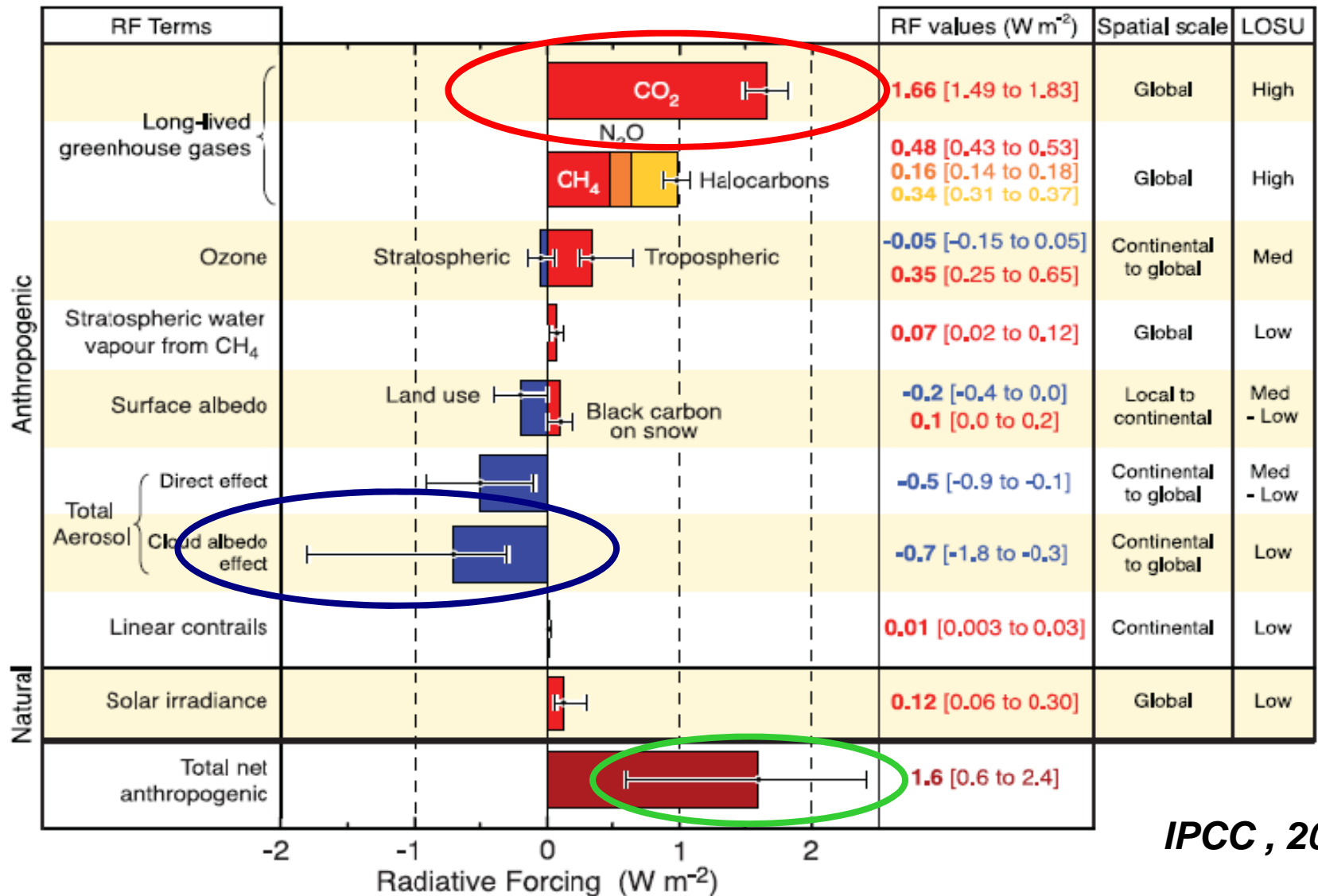


Sotiropoulou et al., 2007

Spatial pattern of IF follows that of aerosol variations

# Components of Human-Induced Climate Change

RADIATIVE FORCING COMPONENTS



©IPCC 2007: WG1-AR4

# The importance of reducing IE uncertainty

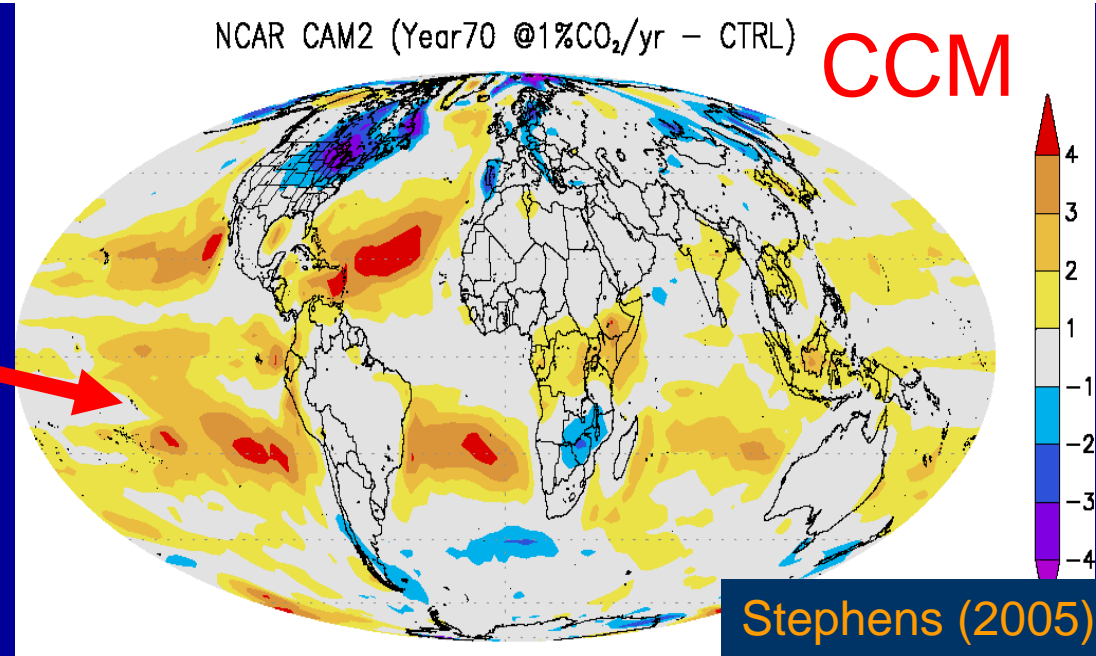
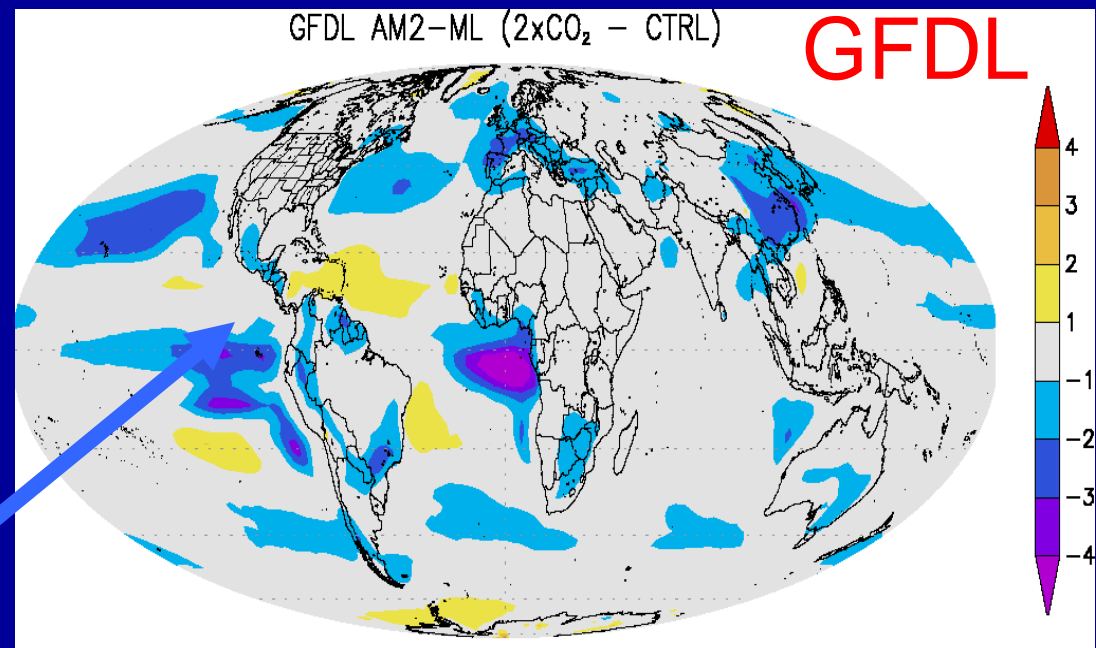
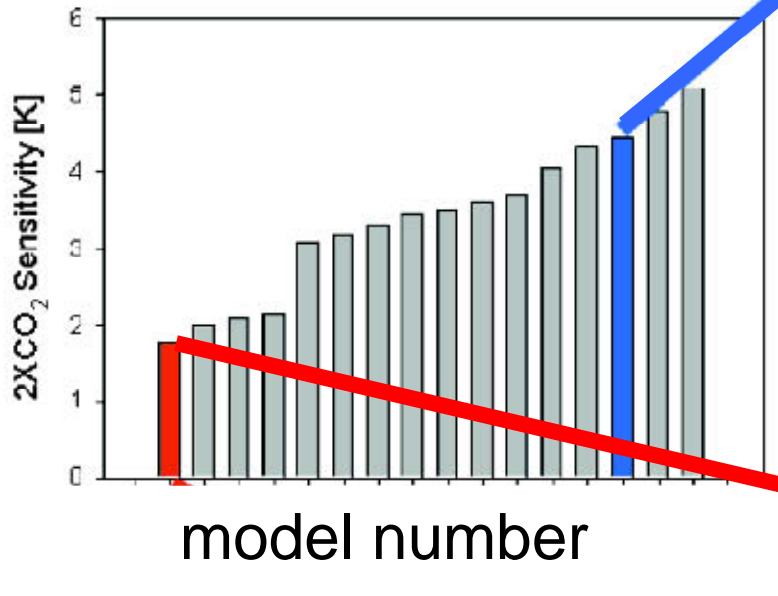
IE diminishes the “punch” of global warming, but we don't know by **how much**.

- If IE is low ( $\sim -0.3 \text{ Wm}^{-2}$ ) then we are “feeling” most of the warming ( $2.3 - 0.3 = 2.0 \text{ Wm}^{-2}$ ) from greenhouse gases.
- If IE is high ( $\sim -2.0 \text{ Wm}^{-2}$ ) then we are “feeling” very little of the warming ( $2.3 - 2.0 = 0.3 \text{ Wm}^{-2}$ ) from greenhouse gases.

The difference in climate sensitivity to  $\text{CO}_2$  is *very large*.  
Improving air quality will lead to accelerated global warming.

The regional impacts of indirect effect can be much larger than global warming on a regional scale.

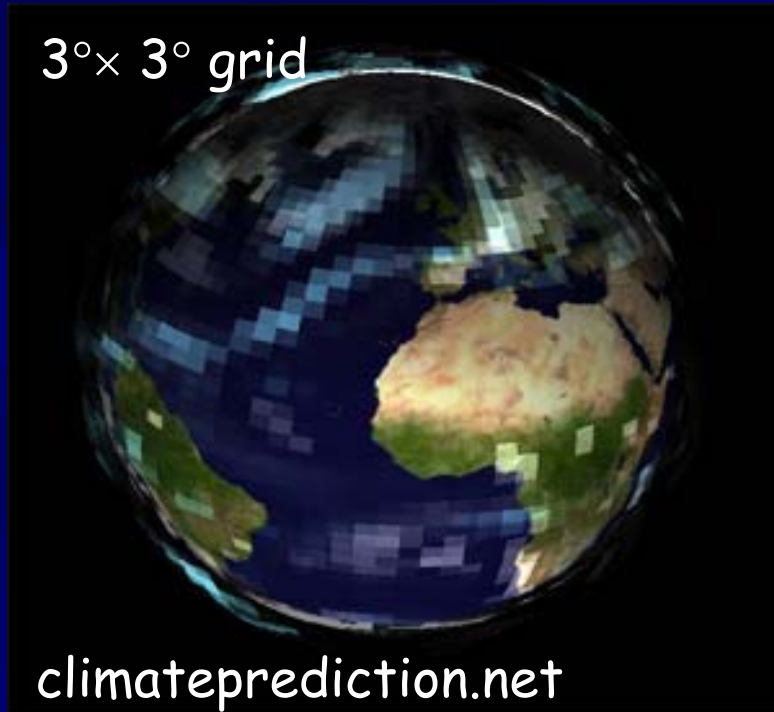
# Treatment of clouds largely determine climate sensitivity...



Stephens (2005)

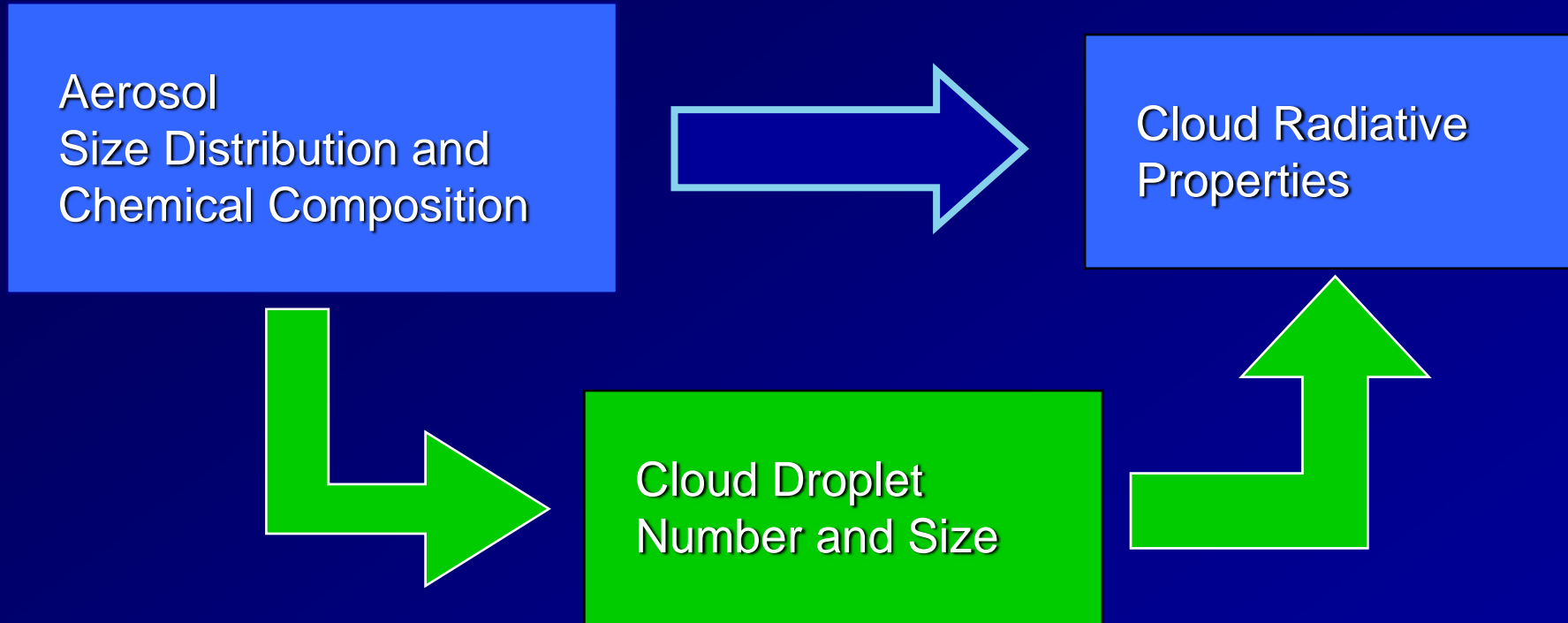
remains the largest source of uncertainty in future climate predictions.

# Problems with GCM assessments of aerosol indirect effect



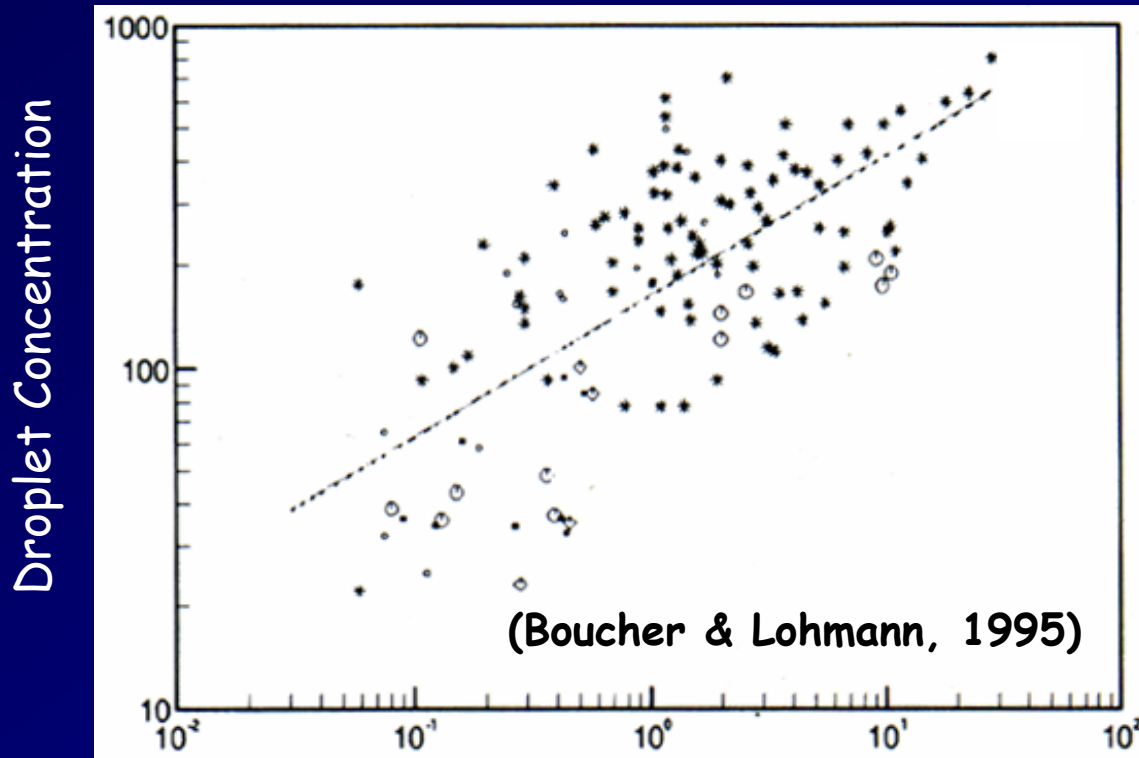
- Cloud formation happens at smaller spatial scales than global climate models can resolve.
- Aerosol-cloud interactions are complex.
- Climate models provide limited information about clouds and aerosols.
- Describing cloud formation explicitly in global models is VERY expensive. These calculations need to be simplified ("parameterized").

# Quantification of the Indirect Effect in Global Climate Models (GCMs)



This problem has historically been reduced to finding the relationship between aerosol number concentration and cloud droplet number concentration. **Empirical relationships** are often used.

# Approach for aerosol- $N_d$ ; empirical



Aerosol sulfate concentration  
(proxy for pollution)

Large variability.

Why?

Unaccounted:

- Meteorology
- Cloud microphysics
- Composition
- etc...

Many studies still utilize this type of approach.

Large predictive uncertainty, without "chances" of improving.



# Current Direction: Use simplified but physically based approaches for cloud processes

## Dynamics

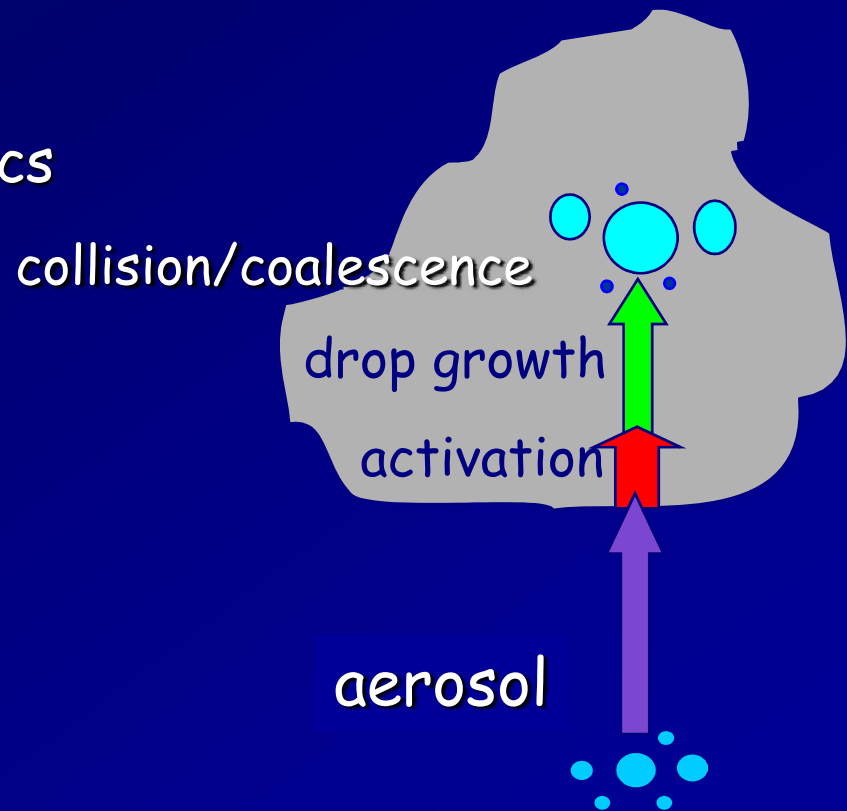
- Updraft Velocity
- Large Scale Thermodynamics

## Particle characteristics

- Size
- Concentration
- Chemical Composition

## Cloud Processes

- Cloud droplet formation
- Drizzle formation
- Rainwater formation
- Chemistry inside cloud droplets

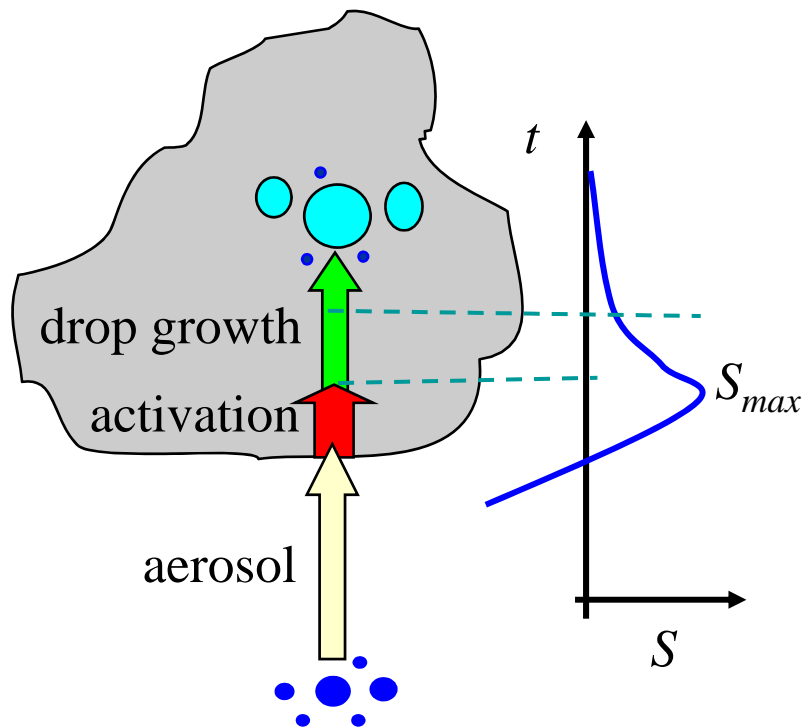


All the links need to be incorporated in global models  
The links need to be **COMPUTATIONALLY** feasible.

# Including explicit physics in GCMs is possible...

**Approach:** use the “simple story of droplet formation”

**Basic ideas:** Solve conservation laws for energy and the water vapor condensing on the aerosol particles contained in the parcel.

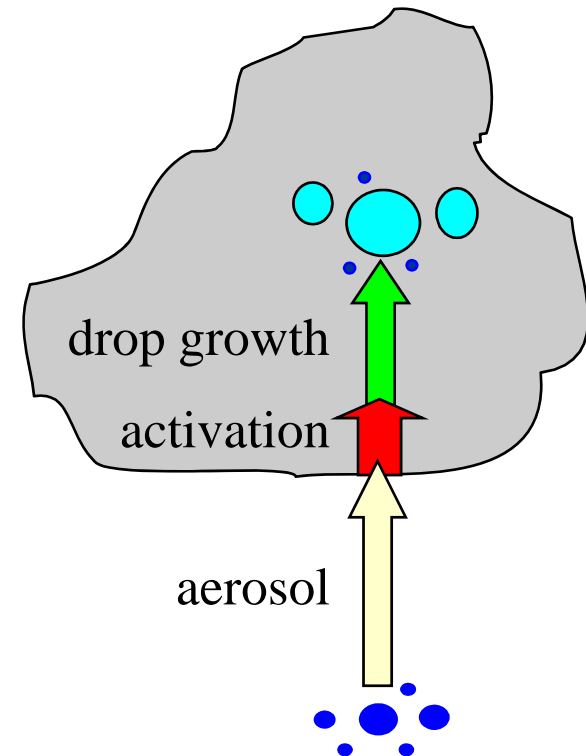
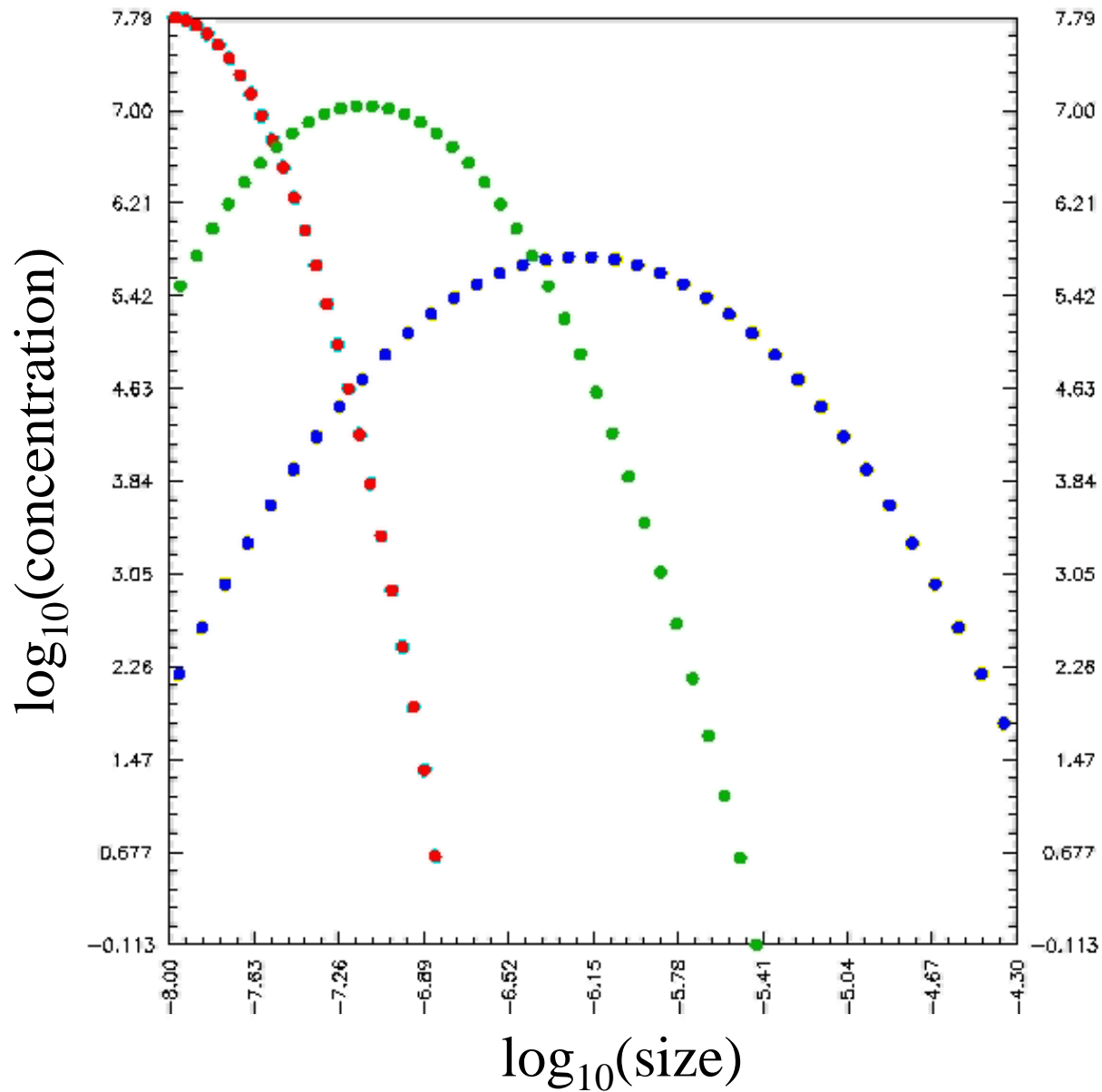


**Steps are:**

- Parcel cools as it rises
- Exceed the dew point at LCL
- Generate supersaturation
- Droplets start activating as  $S$  exceeds their  $S_c$
- Condensation of water becomes intense.
- $S$  reaches a maximum
- No more droplets form

A “classical” nucleation problem

# Cloud droplet formation in updrafts



Including explicit physics in GCMs is possible...

The "good" news:

This theory is well established

The "bad" news:

It is (very, very) SLOW (you have to solve 100's of stiff ODE's).

Fortunately, there is a solution:

"Physically-based" parameterizations.  
They don't solve the "full problem" but only what's important for calculating  $N_d$ :  
 $s_{max}$  and the  $CCN(s_{max})$

# "Mechanistic" Cloud Parameterizations efficiently solve the drop formation problem

Input: P, T, vertical wind, **particle size distribution, composition.**

Output: Cloud properties (droplet number, size distribution).

How: Solve **one algebraic equation** (instead of ODE's).

## Examples:

Abdul-Razzak et al., (1998); Abdul-Razzak et al., (2000);  
Nenes and Seinfeld (2003), Fountoukis and Nenes (2005),  
Ming et al., (2007); Barahona and Nenes (2007)

## Characteristics:

- $10^3$ - $10^6$  times *faster* than numerical parcel models.
- some can treat very complex chemical composition.
- simplified physics, but still you get physically-based links between aerosols and clouds.

# Are parameterizations "good enough"?

Let's apply them for real clouds. In-situ airborne platforms is the ideal laboratory for studying clouds.



CIRPAS Twin Otter



NOAA P3

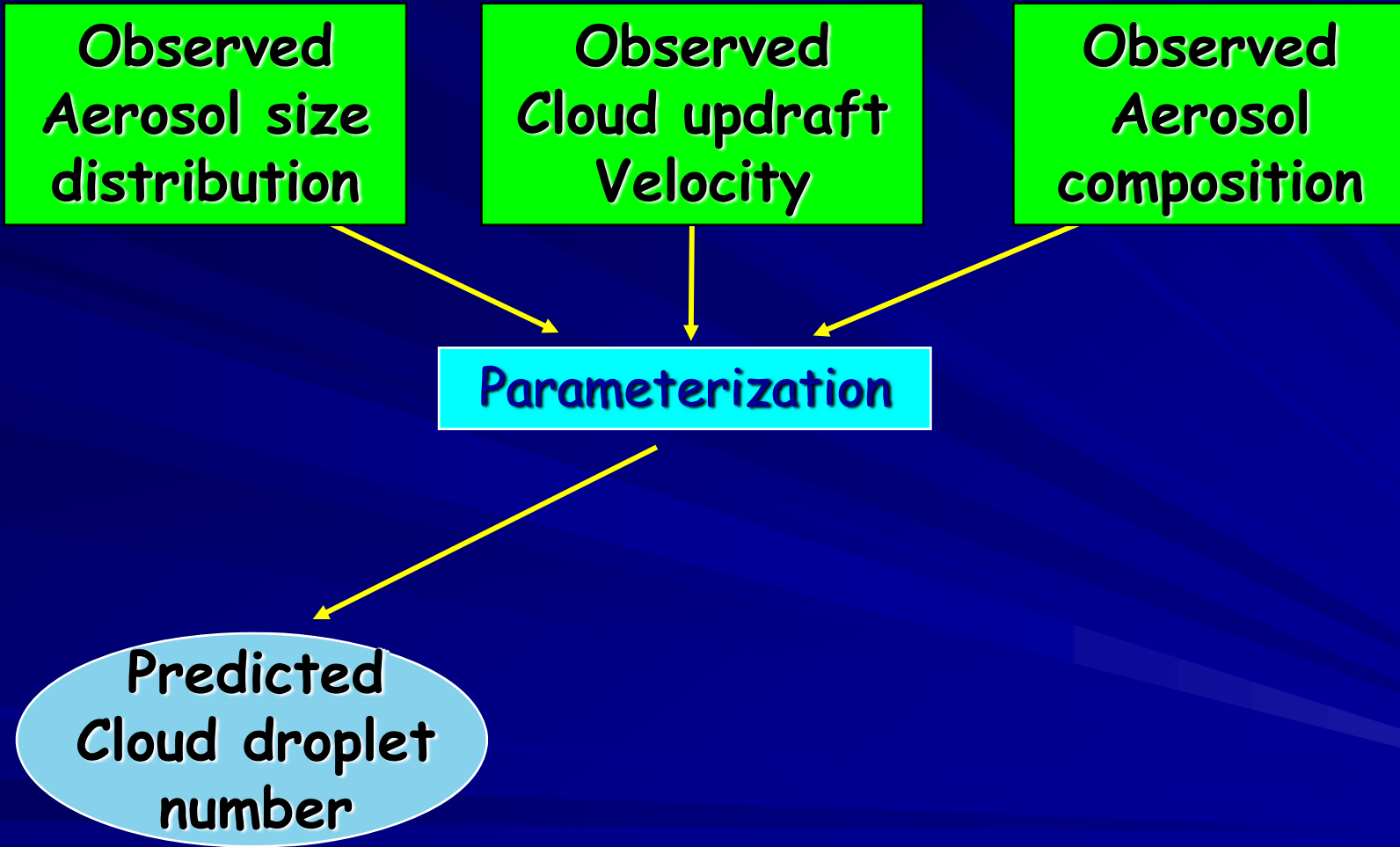
# Parameterization Evaluation CDNC "closure"

Observed  
Aerosol size  
distribution

Observed  
Cloud updraft  
Velocity

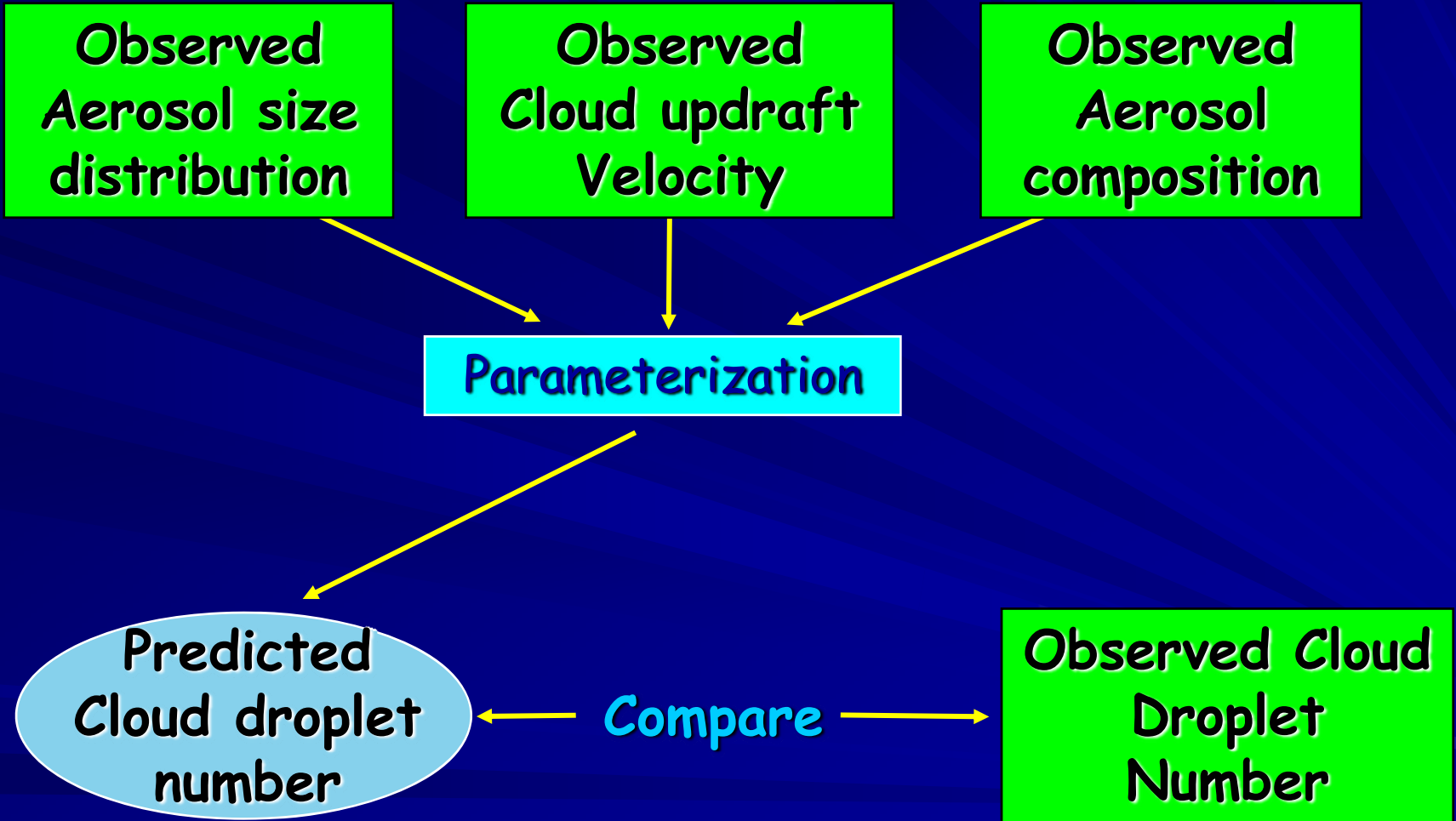
Observed  
Aerosol  
composition

# Parameterization Evaluation CDNC "closure"





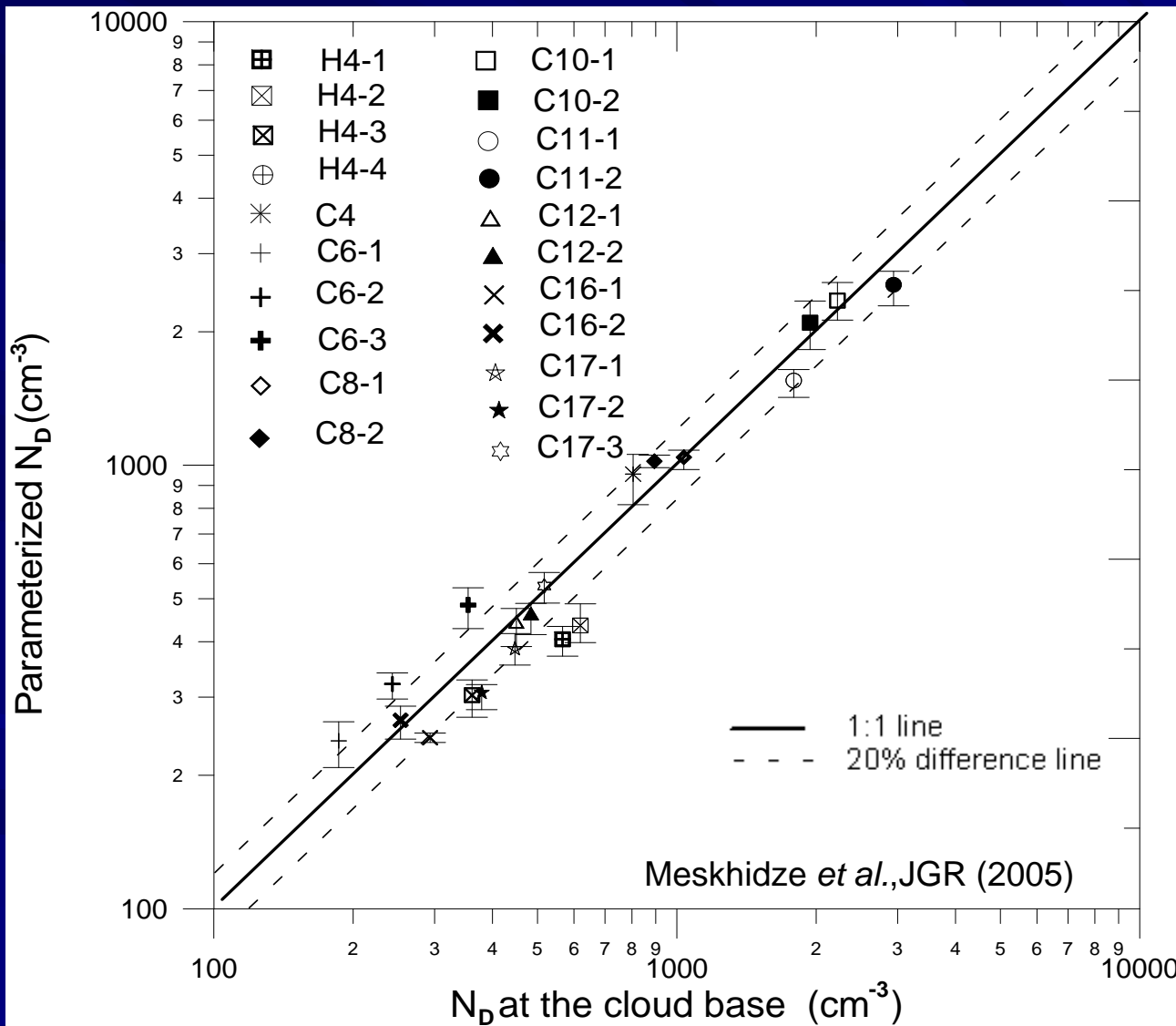
# Parameterization Evaluation CDNC "closure"



# CRYSTAL-FACE (2002) Cumulus clouds



CIRPAS Twin Otter



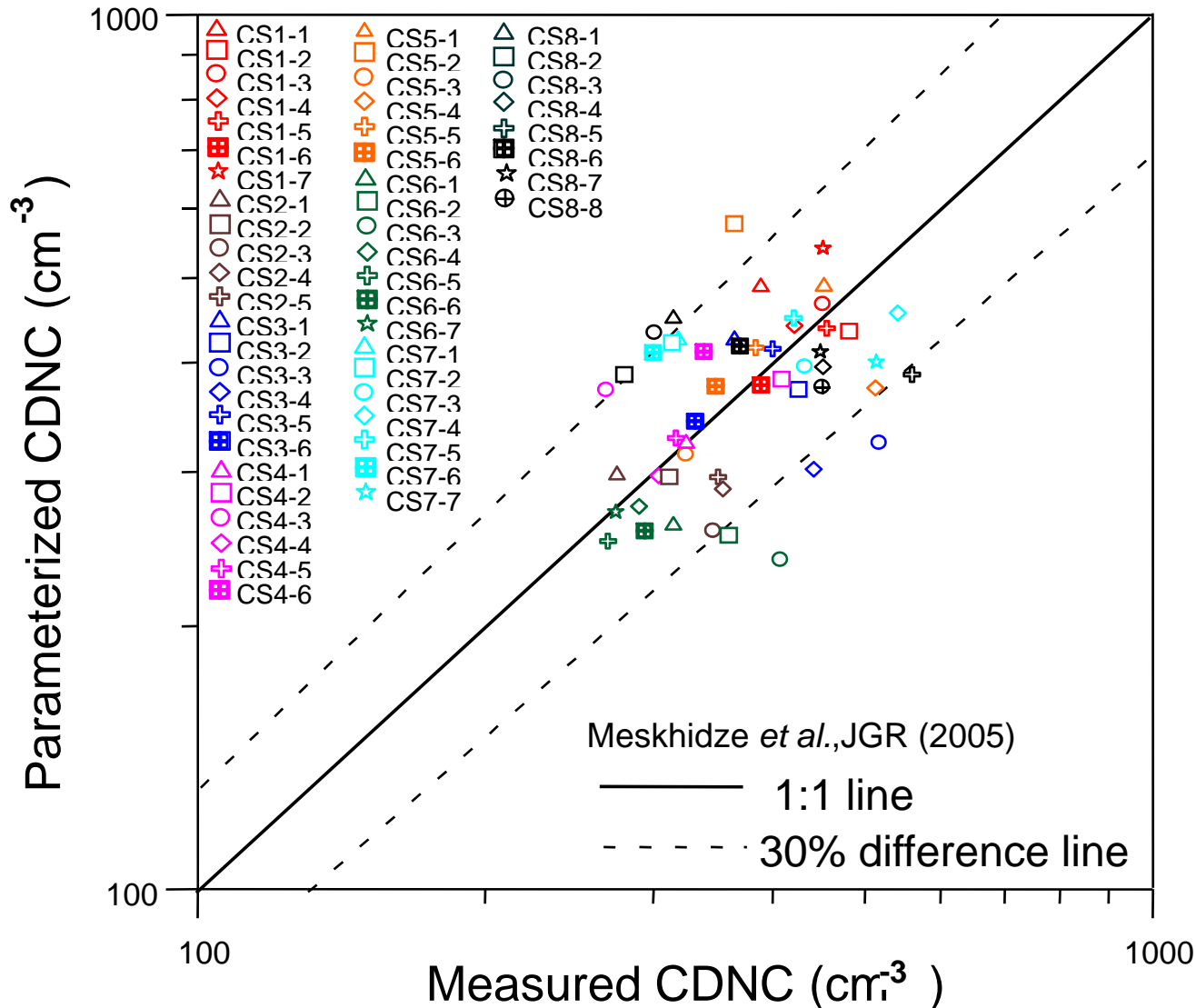
Parameterization  
agrees with  
observed CDNC

Agreement to  
within a few %  
(on average) !

# CSTRIPE (2003) Coastal Stratus Clouds



CIRPAS Twin Otter

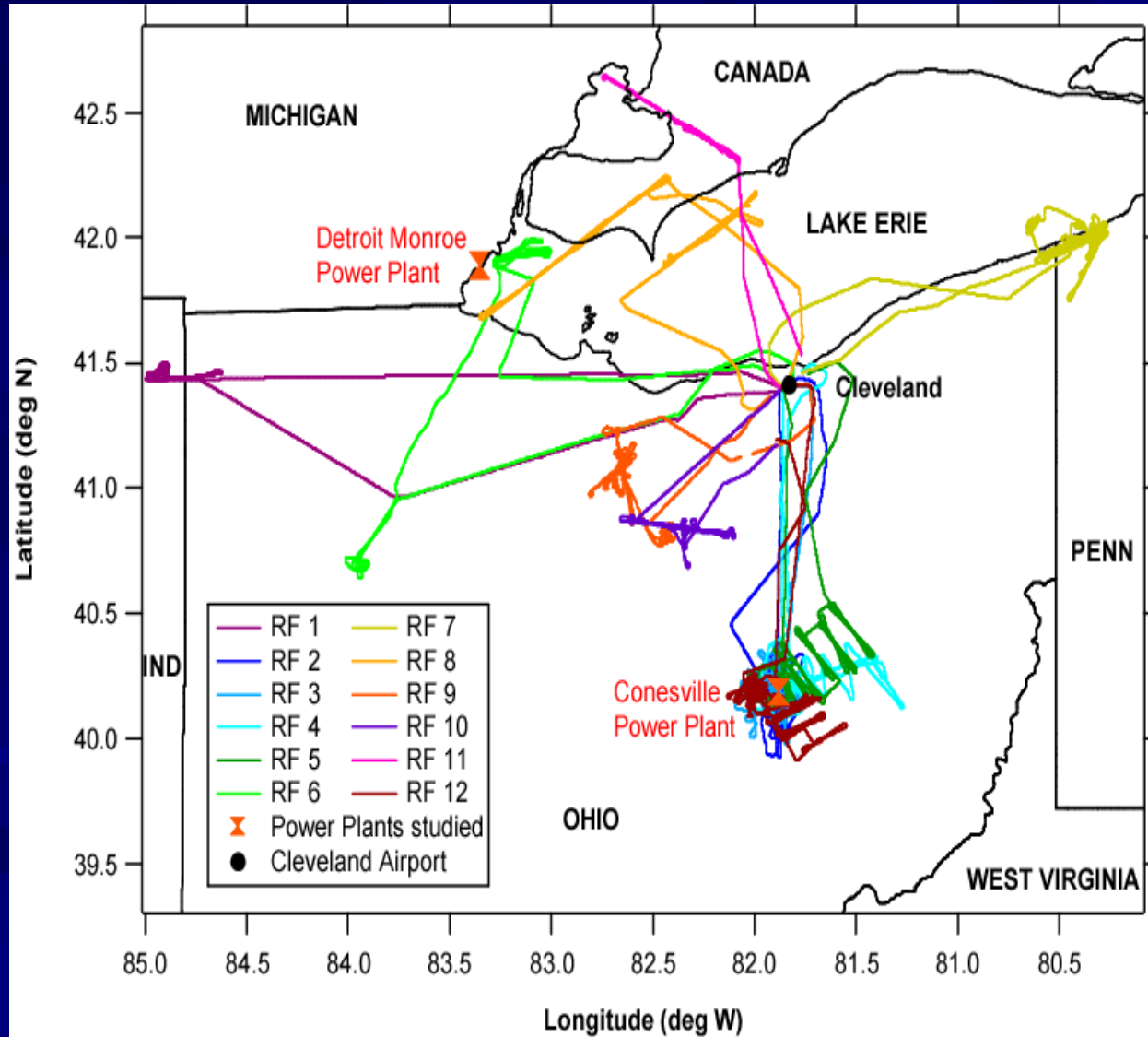


Parameterization  
agrees with  
observed CDNC

Agreement to  
within a few %  
(on average) !

This is a very  
important class  
of clouds.

# CDNC closure during ICARTT (Aug. 2004)



- Cumuliform and Stratiform clouds sampled
- Investigate the effect of power plant plumes on clouds



# Great Agreement

Nd predicted (parameterization), cm-3

- IC03, Aug 06
- ◆ IC05, Aug 09
- ▲ IC06, Aug 10
- ◆ IC09, Aug 16
- × IC10, Aug 17
- ▣ IC11, Aug 18
- ⊠ IC12, Aug 21

Fountoukis *et al.*,  
JGR (2007)

# ... when you know the input

— 1:1 line  
- - - 25% difference line

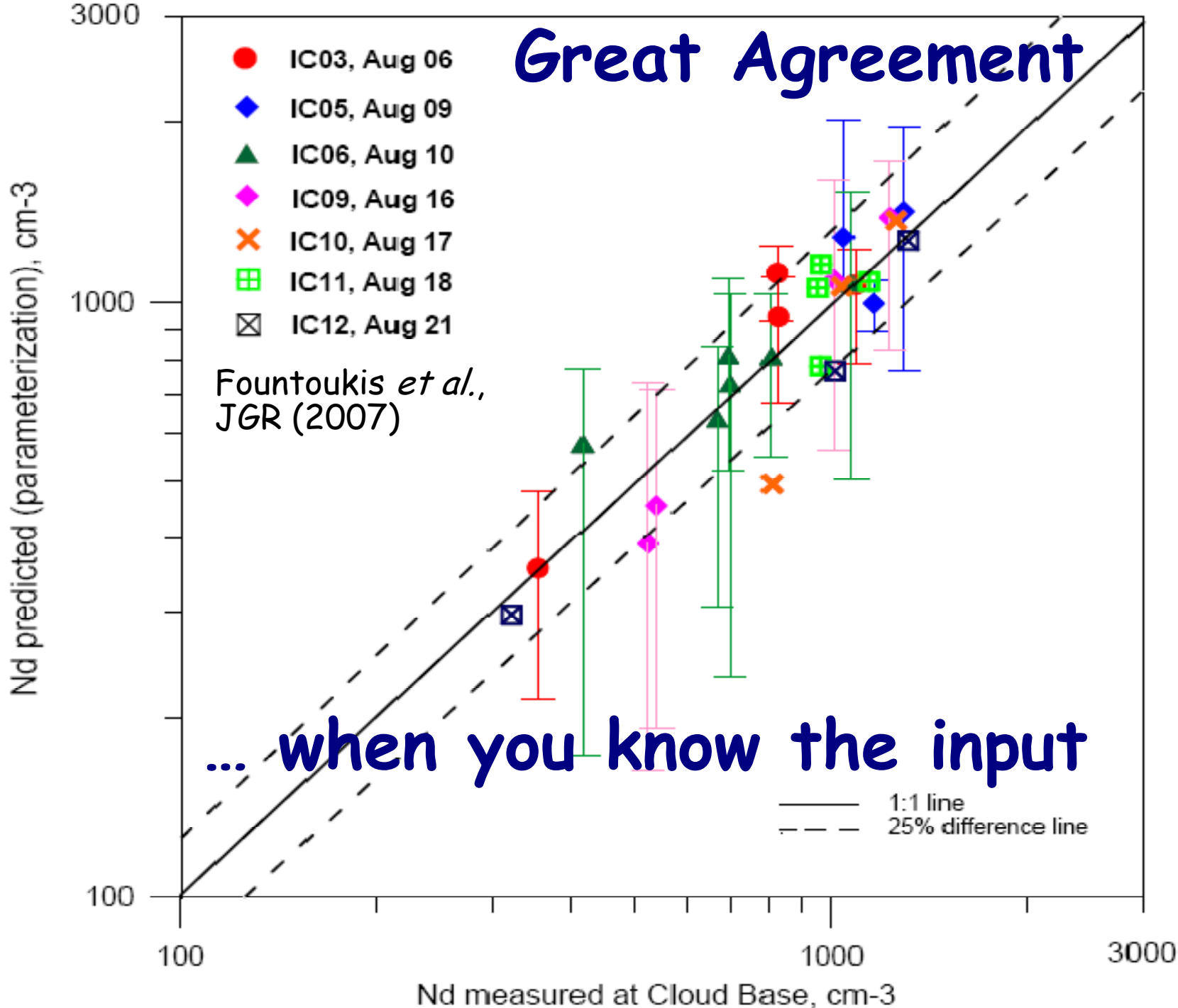
3000

100

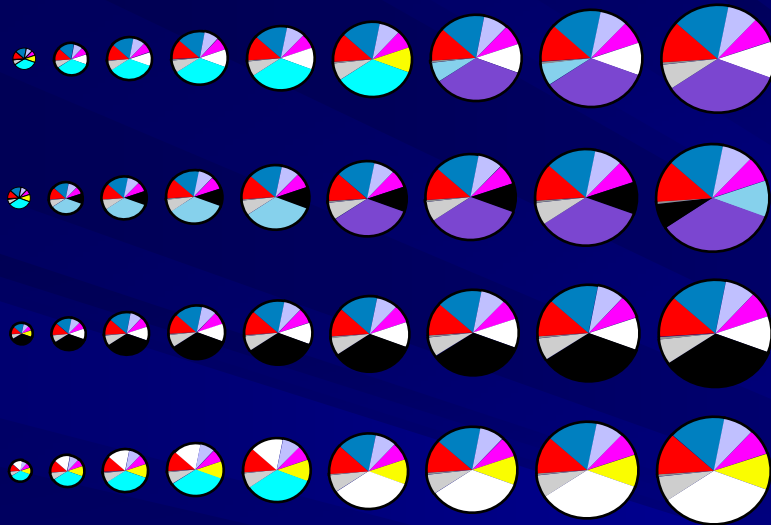
1000

3000

Nd measured at Cloud Base, cm-3



# Problem: Global Aerosol is Vastly Complex



## An integrated "soup" of

- Inorganics, organics (1000's)
- Particles can have uniform composition with size.
- ... or not
- Can considerably vary with space and time (esp. near sources)

CCN activity of particles is a **strong** function ( $\sim d^{-3/2}$ ) of aerosol dry size and (a weaker but **important**) function of chemical composition ( $\sim \text{salt fraction}^{-1/2}$ ).

Predicting CCN concentrations requires knowledge of size distribution and chemical composition.

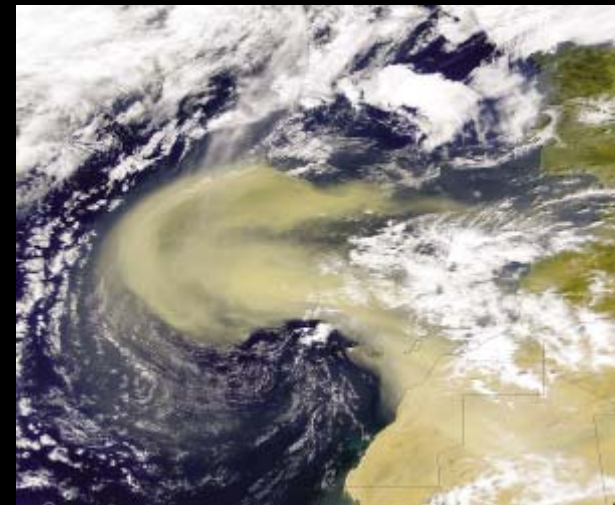
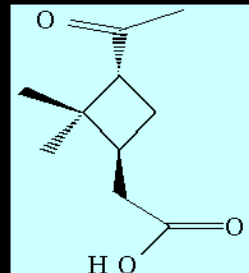
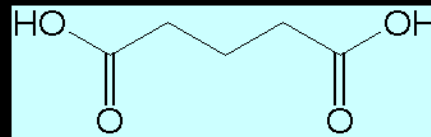
# This aerosol is highly complex and variable

## Some inorganics:

- $(\text{NH}_4)_2\text{SO}_4$ ,  $\text{NH}_4\text{HSO}_4$
- Sulfuric acid
- Seasalt ( $\text{NaCl}$ )
- Crustals ( $\text{Ca}$ ,  $\text{Mg}$ ,  $\text{K}$  salts)
- Nitrates ( $\text{NH}_4\text{NO}_3$ ,  $\text{NaNO}_3$ )
- Chlorides ( $\text{KCl}$ ,  $\text{NH}_4\text{Cl}$ )

## Some organics:

- Glutaric Acid (biomass)
- Pinonic Acid (terpenes)
- Hydroxy Methyl Benzoic Acid (gasoline)
- Palmitic Acid (plant waxes)
- *and hundreds more...*



# The ... headache of organic species:

- They can act as surfactants and facilitate cloud formation.
- They can increase the affinity of particles for water (add solute) and facilitate cloud formation.
- "Waxy" films can form and slow down droplet growth

## Use in-situ data to study the aerosol-CCN link:

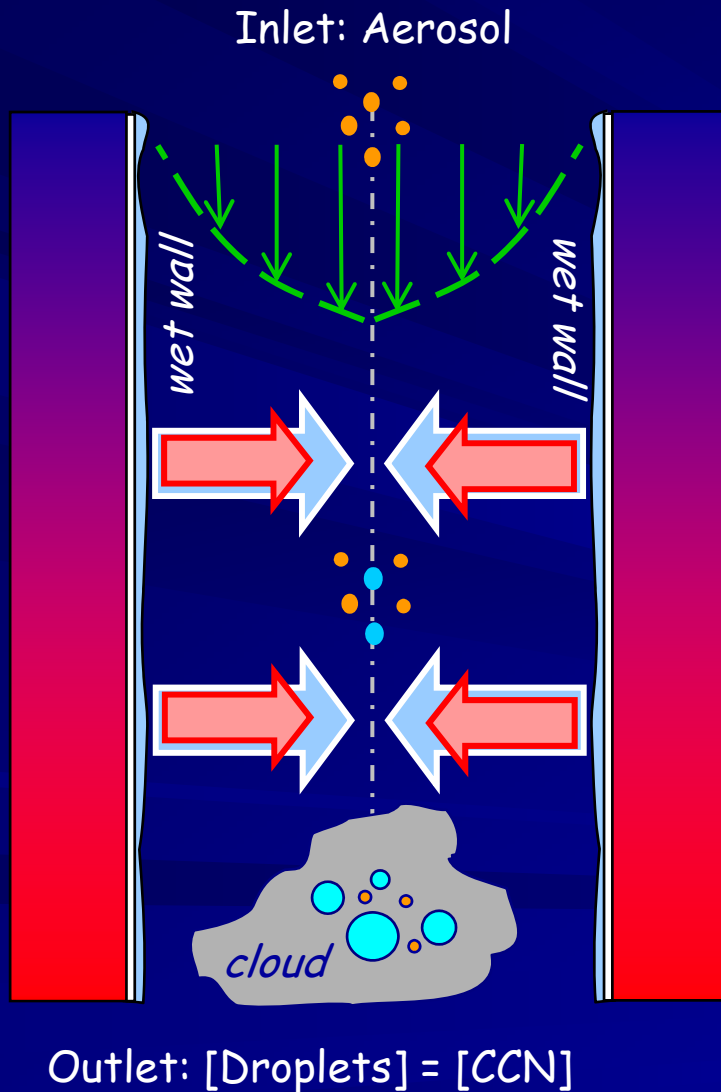
Creative use of CCN activity measurements to "constrain" the importance of organics and mixing state on droplet formation for ambient aerosol.

## Desired information:

- Average molar properties (molar volume, solubility)
- Droplet growth kinetics
- Chemical heterogeneity (mixing state) of aerosol
- Surface tension depression



# Measuring CCN activity of ambient particles: Cloud Chamber



Roberts and Nenes (2005), Patent pending

Metallic cylinder with walls wet.  
Apply T gradient, and flow air.

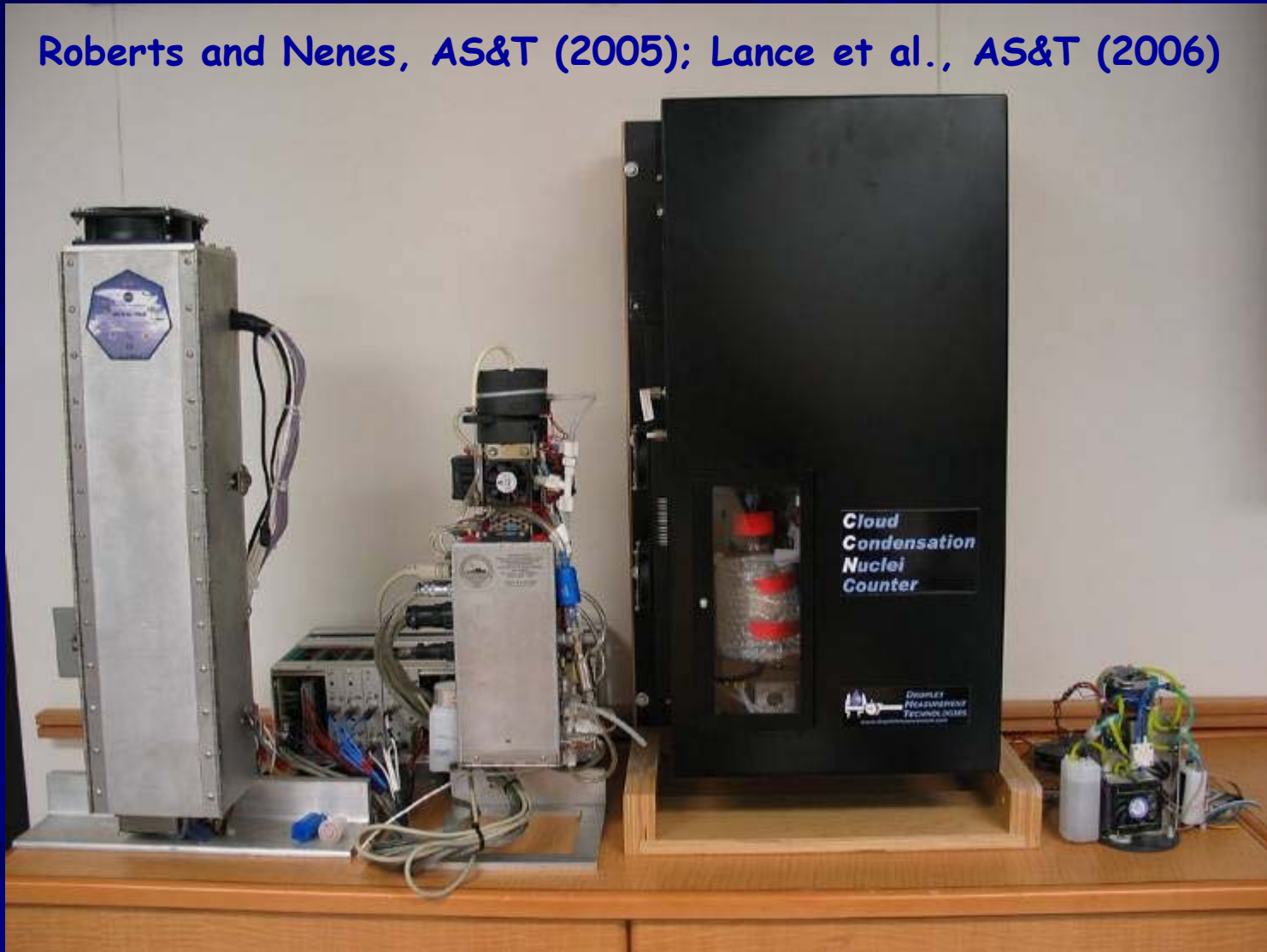
- Wall saturated with  $H_2O$ .
- $H_2O$  diffuses more quickly than heat and arrives at centerline first.
- The flow is supersaturated with water vapor at the centerline.
- Flowing aerosol at center would activate some into droplets.

Count the **concentration** and **size** of droplets that form with a 1 s resolution.

# Development phases of cloud chamber

Roberts and Nenes, AS&T (2005); Lance et al., AS&T (2006)

scale = 1 m



1<sup>st</sup> version  
April 2002

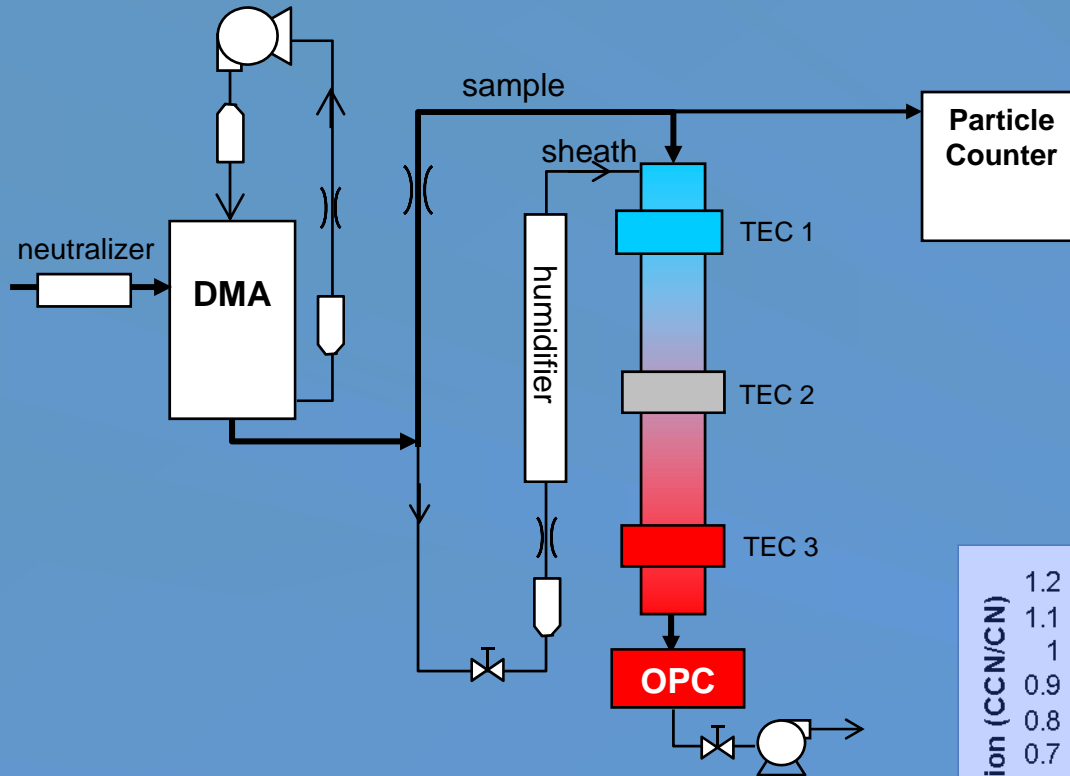
2<sup>nd</sup> version  
January 2003

Commercial ver.  
July 2004

mini version  
August 2006

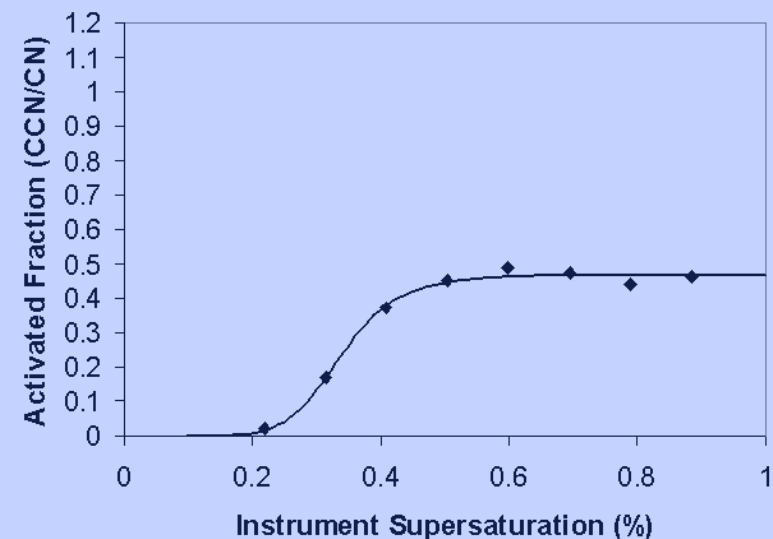
# Key source of information: Size-resolved CCN activity measurements

Measure CCN activity of aerosol with known diameter

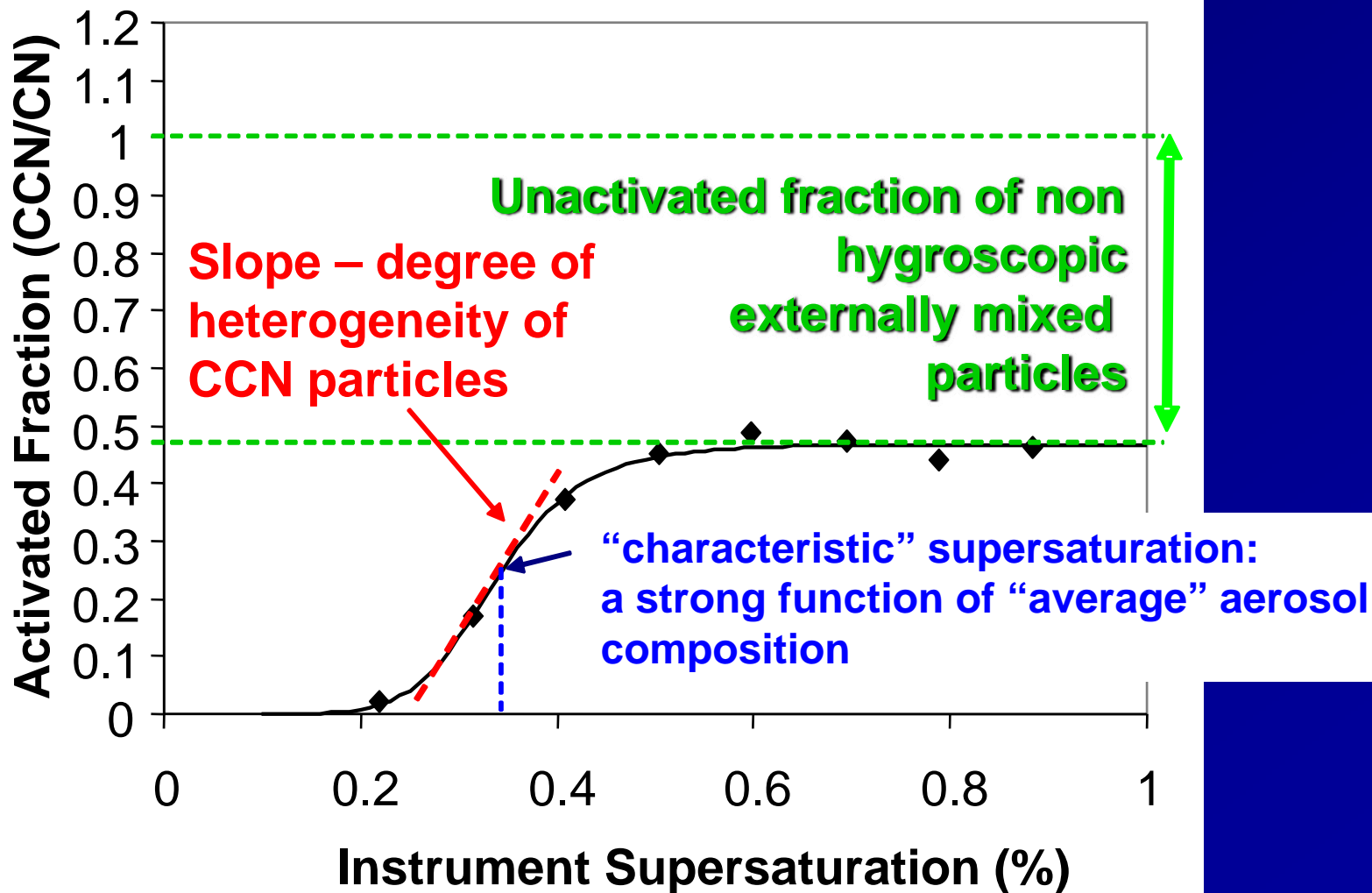


$$\text{Activated Fraction} = \frac{\text{CCN}}{\text{CN}}$$

Results: "activation curves"

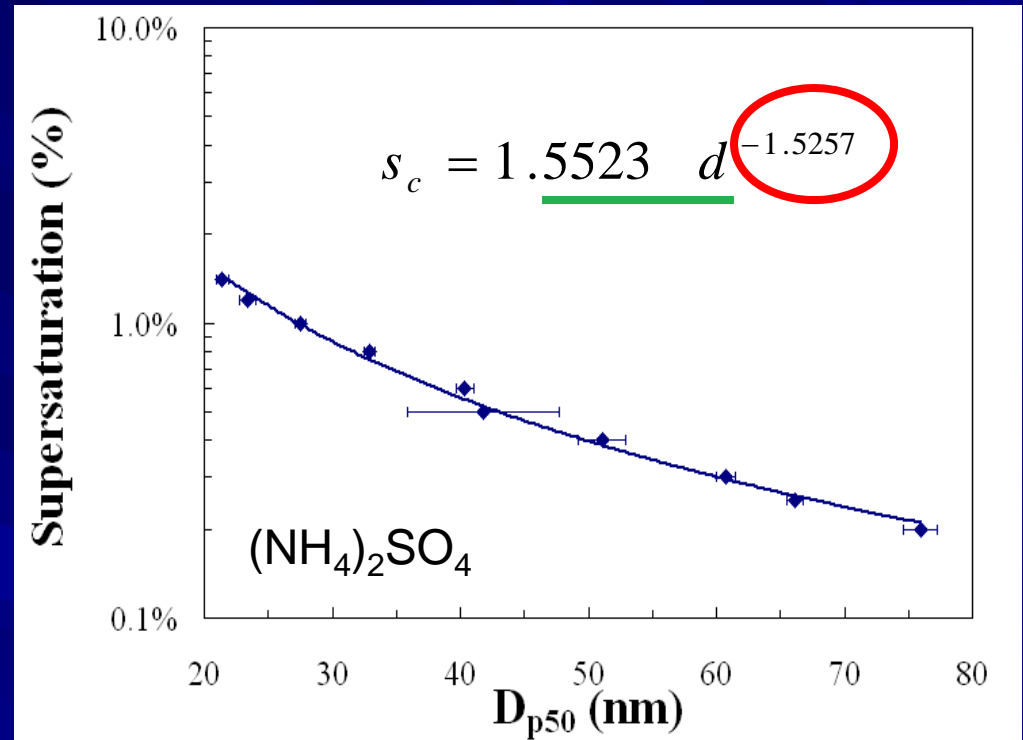


# What the Activation Curves tell us



# Inferring Average Molecular Weight from CCN measurements

- Plot characteristic supersaturation as a function of dry particle size.
- Fit the measurements to a power law expression.
- Relate fitted coefficients to aerosol properties (e.g. molecular weight, solubility) by using Köhler theory:



$$S_c = \left( \frac{256M_w\sigma}{27RT\rho_w} \right)^{1/2} \left[ \sum_i \left( \frac{\rho_w}{M_w} \right) \left( \frac{M_i}{\rho_i} \right) \frac{1}{\varepsilon_i\nu_i} \right]^{1/2} d^{-3/2} = \omega d^{-3/2}$$

... and then solve for average molecular weight of organics.

$$\frac{M_{\text{organic}}}{\rho_{\text{organic}}} = \frac{\varepsilon_{\text{organic}} \nu_{\text{organic}}}{\frac{256}{27} \left( \frac{M_w}{\rho_w} \right)^2 \left( \frac{1}{RT} \right)^3 \sigma^3 \omega^{-2} - \sum_{i \neq \text{organic}} \frac{\rho_i}{M_i} \varepsilon_i \nu_i}$$

Average  
Molecular weight  
of organics

Constants

Data from CCN  
measurements

From ion  
chromatography

A simple and comprehensive way to quantify the contribution of organics to cloud formation.

# Major findings on soluble organics

Many “aged” soluble organics from a wide variety of sources have a  $200\text{--}250\text{ g mol}^{-1}$ . For example:

- Aged Mexico City aerosol from MILAGRO.
- Organic Aerosol from biogenic VOC emissions
  - α-pinene and monoterpene oxidation (Engelhart et al., ACPD).
  - Ozonolysis of Alkenes (Asa-Awuku et al., ACPD)
  - Oleic Acid oxidation (Shilling et al., 2007)
- In-situ samples collected in California (2003, 2005, 2007), New Hampshire (2004), Cleveland (2004), Houston (2006), Greece (2007), Finland (2007), Arctic (2008).

What varies mostly is not the “thermodynamic” properties of the complex organic “soup” but the fraction of soluble material (we can simulate this)...

**Complexity sometimes simplifies things for us.**

# Current efforts: Ice (cirrus) clouds

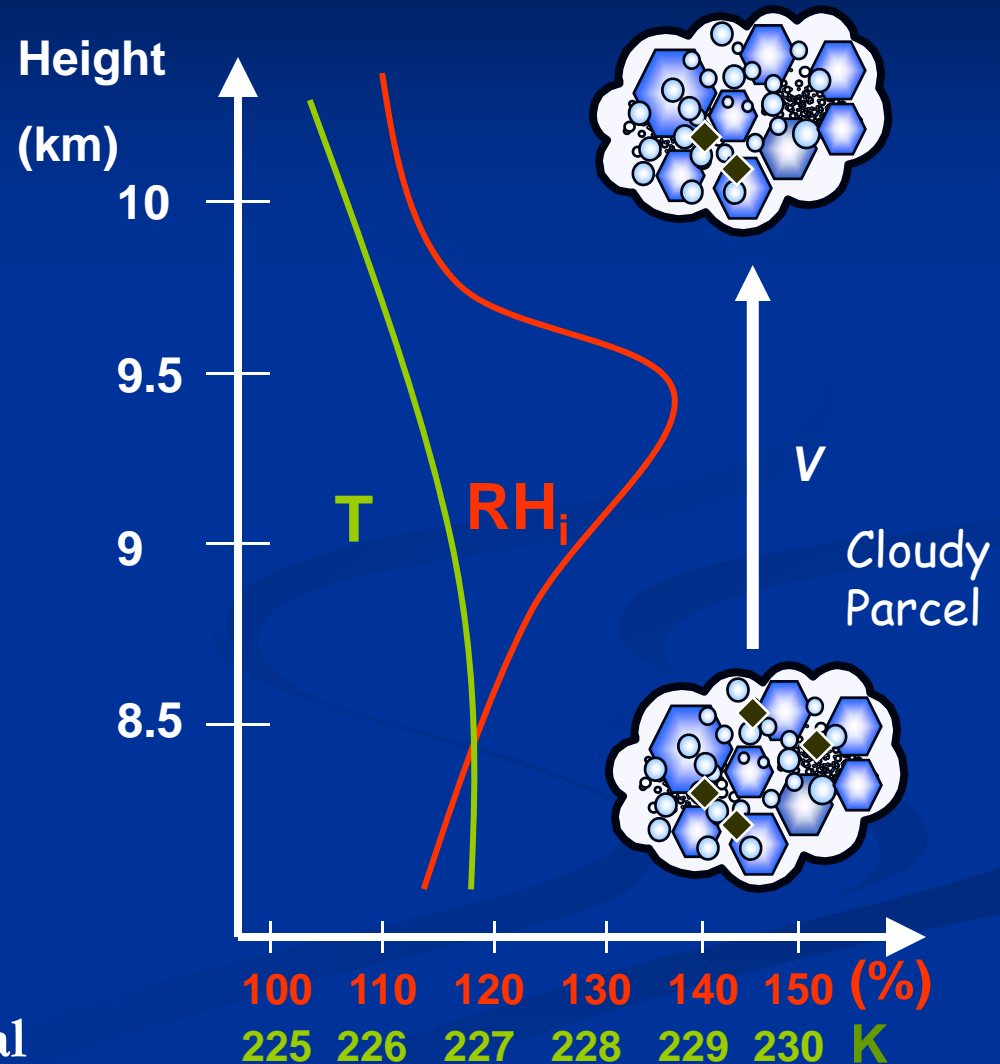
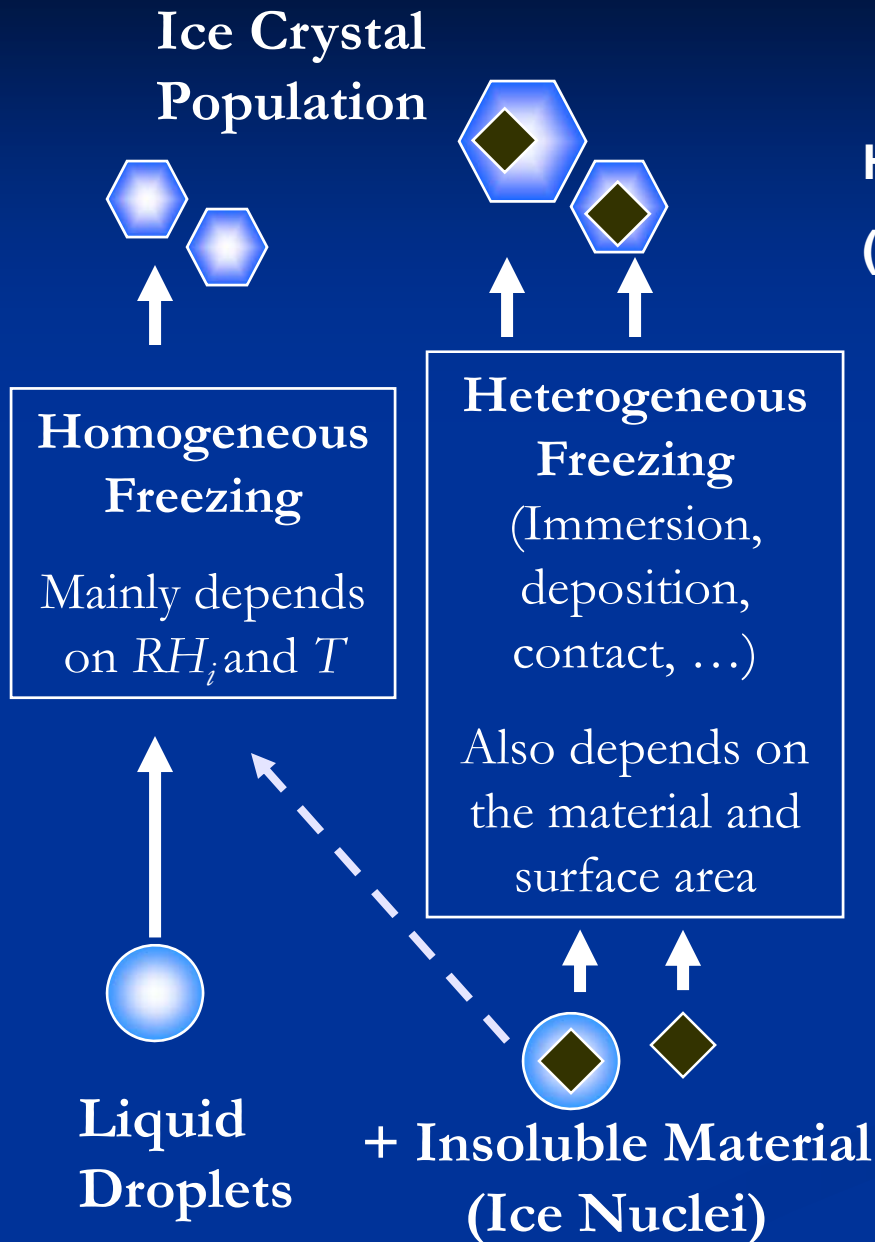


- Cirrus clouds, composed of ice crystals, are important for:
  - Radiative transfer: they tend to warm climate
  - Stratospheric moisture and circulation
  - Regulation of the ocean temperature
  - Heterogeneous chemistry

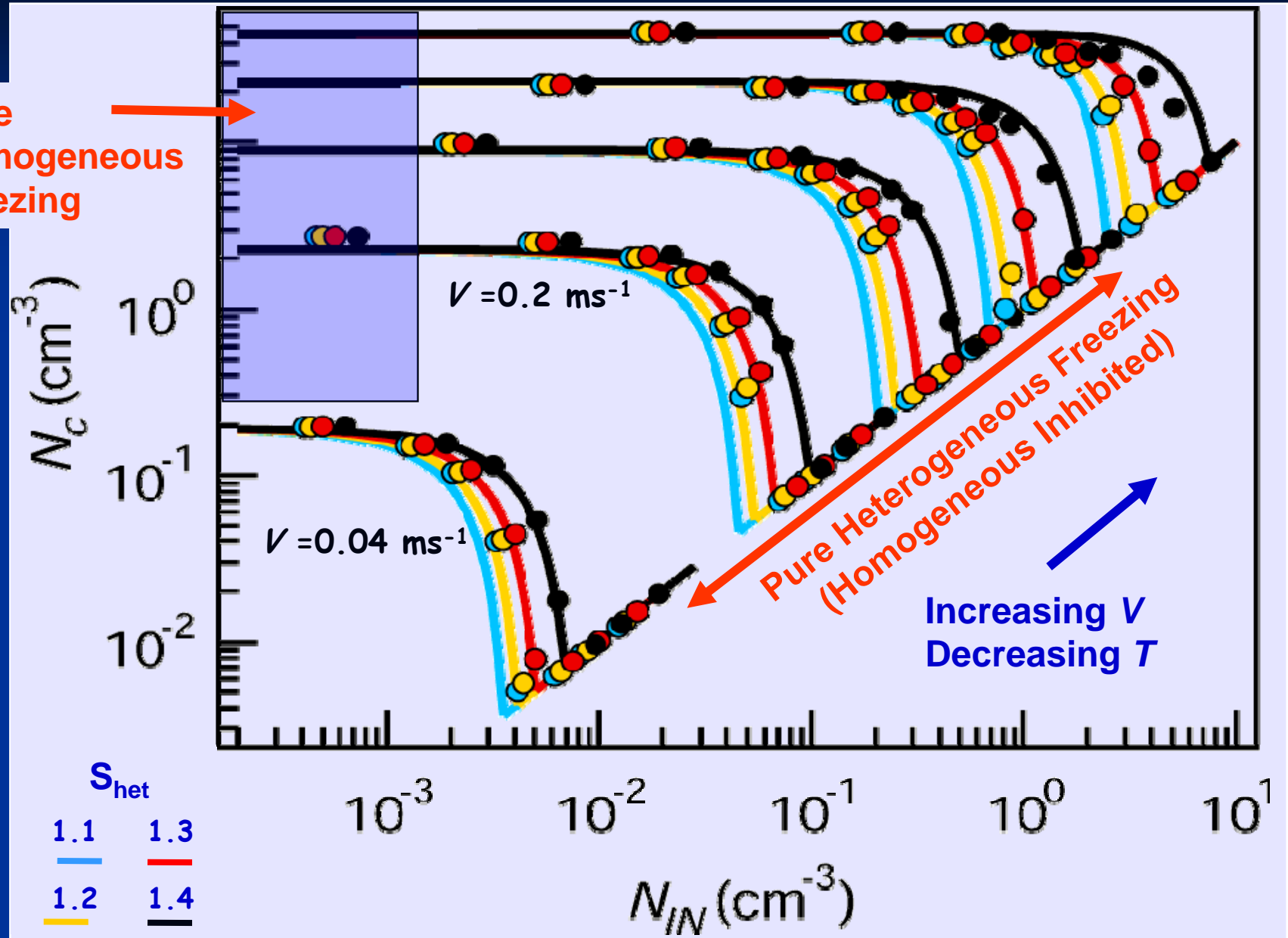
Cirrus may be affected by aircraft emissions, transport of dust and pollution. **Aerosol effects on cirrus (and climate) are highly unknown (worse than for liquid clouds)!!**



# Cirrus (Ice) Cloud Formation



# Parameterizing Ice Formation Possible



# Summary

- The impacts of aerosol on clouds appreciably diminish warming from greenhouse gases.
- The sensitivity of climate to  $CO_2$  increases are very uncertain, ranging from modest to very large.
- A large amount of uncertainty arises from the treatment of clouds and aerosol-cloud interactions in climate models.
- The indirect effect can regionally be **dominant (like in Georgia)**. These impacts will only become larger because of the development of Asian Nations.

# Summary

- Climate models are beginning to include physically-based descriptions of aerosol-cloud interactions.
- Observations provide the “constraints” and “tests” for evaluation of the improved physics developed for climate models.
- A lot of work to do... but we are really seeing the improvements. This could can only be accomplished through the coordinated effort of the scientific community and the support from the funding agencies.