Trace Elements in Ambient Air at Porto Metropolitan Area—Checking for Compliance with New European Union (Eu) Air Quality Standards

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Because of the scientific evidence showing that arsenic (As), cadmium (Cd), and nickel (Ni) are human genotoxic carcinogens, the European Union (EU) recently set target values for metal concentration in ambient air (As: 6 ng/m3, Cd: 5 ng/m3, Ni: 20 ng/m3). The aim of our study was to determine the concentration levels of these trace elements in Porto Metropolitan Area (PMA) in order to assess whether compliance was occurring with these new EU air quality standards. Fine (PM2.5) and inhalable (PM10) air particles were collected from October 2011 to July 2012 at two different (urban and suburban) locations in PMA. Samples were analyzed for trace elements content by inductively coupled plasma-mass spectrometry (ICP-MS). The study focused on determination of differences in trace elements concentration between the two sites, and between PM2.5 and PM10, in order to gather information regarding emission sources. Except for chromium (Cr), the concentration of all trace elements was higher at the urban site. However, results for As, Cd, Ni, and lead (Pb) were well below the EU limit/target values (As: 1.49 ± 0.71 ng/m3; Cd: 1.67 ± 0.92 ng/m3; Ni: 3.43 ± 3.23 ng/m3; Pb: 17.1 ± 10.1 ng/m3) in the worst-case scenario. Arsenic, Cd, Ni, Pb, antimony (Sb), selenium (Se), vanadium (V), and zinc (Zn) were predominantly associated to PM2.5, indicating that anthropogenic sources such as industry and road traffic are the main source of these elements. High enrichment factors (EF > 100) were obtained for As, Cd, Pb, Sb, Se, and Zn, further confirming their anthropogenic origin.

Air pollution is a major environmental health problem (Samet and Krewski, 2007; Costa et al., 2014). According to recent World Health Organization (WHO) estimates, in 2012 approximately 7 million people died (one out of 8 of total global deaths) as a result of air pollution exposure, confirming that air pollution is currently the world's largest single environmental health risk (WHO, 2014). Pollutants of major public health concern include sulfur dioxide (SO2), nitrogen dioxide (NO2), carbon monoxide (CO), ozone (O3), benzene, polycyclic aromatic hydrocarbons (PAH), and particulate matter (PM) (Costa et al., 2014). This has led authorities in the European Union (EU) to develop an extensive body of legislation that establishes health-based standards and objectives for these pollutants in ambient air (European Commission, 2014). Particulate matter (PM) pollution is particularly associated with serious health risks (Chang et al., 2013; Tsai and Yang, 2014). Substantial epidemiological evidence exists indicating that PM is a major health risk factor, with serious consequences on human health, by increasing the incidence of lung cancer, asthma, and cardiovascular diseases (Shrey et al., 2011; Krewski and Rainham, 2007; Craig et al., 2008; Beckerman et al., 2012). European studies on long-term exposure to PM indicate a significant direct association with mortality, particularly from cardiovascular and respiratory diseases (Pelucchi et al., 2009; Vigotti et al., 2007). The WHO has declared that PM poses the greatest air pollution threat globally (WHO, 2013).

Particulate matter (PM) is a mixture of aerosol particles (solid and liquid), with a wide range of sizes and chemical compositions (Ghio et al., 2012). PM is emitted from many anthropogenic sources, including both combustion and noncombustion sources. Important natural sources of PM are sea salt and natural resuspended dust (European Environment Agency, 2014). PM10 ("inhalable particles") refers to particles with a diameter of 10 μ m or less, and a limit value was set at 40 μ g/m3 (on average over a year) for PM10 in ambient air, and entered into force in January 2005. In recent years, a class of smaller particles known as "fine particles" or PM2.5 (particles with a diameter of 2.5 μ m or less) was identified as of particular concern, because these particles penetrate deep into the lung, and several studies showed that PM2.5 is the air pollutant most closely linked with health problems (Tsai and Yang, 2014; Chang et al., 2013, Beckerman et al., 2012). A study from 2009 concluded that in Europe PM2.5 pollution is

associated with more than 492,000 premature deaths, corresponding to a loss of almost 4.9 million years of life (Leeuw and Horálek, 2009). The new EU air quality directive (Directive 2008/50/EC) set a "target" value for PM2.5 at 25 µg/m3, which became a "limit" value in January 2015.

In recent years, the importance of chemical composition of PM, in particular its trace element content, has been increasingly recognized, in addition to the simple determination of PM mass per cubic meter, as a factor in development of air-pollutant-associated diseases (Bell and Committee, 2012; Franklinet et al., 2008; Ghio et al., 2012). Trace elements, and particularly the so-called "heavy metals," are important ambient air pollutants and are (almost exclusively) associated with PM (i.e., they are present in the solid phase of the air aerosols). Several toxicological studies examined potential mechanisms underlying the toxicity of metal-enriched particulates (Chen and Lippmann, 2009; Ghio et al., 2012; Satsangi et al., 2014; Lippmann et al., 2013). Numerous occupational studies established plausible clinical health endpoints that may be attributed to such exposures (Sarmento et al., 2008; Cakmak et al., 2014). In 1999, the EU Directive 1999/30/EC set a "limit" value for Pb in ambient air at 0.5 µg/m3. More recently, Directive 2004/107/EC, recognizing that "scientific evidence shows that arsenic, cadmium, nickel . . . are human genotoxic carcinogens" and in order "to avoid, prevent or reduce harmful effects . . . on human health and the environment as a whole," also established "target values" for these elements in ambient air. Those target values were set at 6 ng/m3 for As, 5 ng/m3 for Cd, and 20 ng/m3 for Ni (as the total content of these elements and compounds in the PM10 fraction). According to this directive, "Member States shall take all necessary measures not entailing disproportionate costs to ensure that, as from 31 December 2012," the target values are not exceeded.1

Heavy metals such as Pb, Cd, Ni, and As are mainly emitted from anthropogenic (industrial) activities. Because of persistence and potential for long-range atmospheric transport, atmospheric emissions of heavy metals virtually affect even the most remote regions in the globe (WHO, 2007).

Lead (Pb) is a well-known neurotoxin, and exposure to this element has been correlated with developmental and neurobehavioral effects in infants and children and elevated blood pressure in adults (Needleman, 2004). Cadmium (Cd) was categorized as a potential human carcinogen. An increased risk of lung cancer was reported following inhalation exposure in occupational settings.

Kidney and bone are also critical target organs with regard to environmental exposure (WHO, 2007; Ginsberg, 2012). Nickel (Ni) is a known human carcinogen and also produces other health effects such as skin allergies, cardiovascular effects, pulmonary fibrosis, and renal damage (Schaumloffel, 2012; Zhang et al., 2009). Arsenic (As) is actually a metalloid but is commonly included in the list of "heavy metals" based upon its toxicity. Chronic human inhalation exposure to inorganic As is associated with irritation of skin and mucous membranes with increased risk of skin and lung cancer, as well as effects on the brain and nervous system (Sun et al., 2014).

The main aims of this study were to assess whether compliance with the EU limit for Pb was occurring and to determine whether target values recently set for As, Cd, and Ni are in effect. Taking advantage of the multi-elemental capability of inductively coupled plasma–mass spectrometry (ICP-MS), several other trace elements were also measured. A health risk assessment based on the obtained concentrations was performed to assess the current importance of inhalation exposure.

EXPERIMENTAL

Study Areas and Sampling

Two distinct areas of the Porto metropolitan area (PMA), which is the second biggest urban area in Portugal with an area of almost 1600 km² and a total population of more than 1.5 million inhabitants, were selected as sampling sites (SS): (i) SS1, located in an urban area, in the center of Porto, in the confluence of highly traffic streets (41°09_19.6_N, 8°37_13.8_W); and (ii) SS2,

located in a peripheral urban area, in Gandra, a small town about 20 km northeast from Porto (41°10_54.4_N, 8°26_43.6_W). A third sampling site, located in the periphery of Amarante, a small size town, outside of PMA, at 50 km northeast from Porto (41°16_10.8_N, 8 04_10.3_W), was selected as a reference (REF) to estimate background level of trace elements in air (Figure 1).

Air sampling was performed between October 2011 to March 2012 at SS1 (n = 24 for PM10 and n = 22 for PM2.5), between June and July 2012 at SS2 (n = 17 for PM10 and n = 13 for PM2.5) and between April and May 2013 at REF (n = 6 for PM10) using an ECHO PM (Tecora, Corsico, Italy) air sampler equipped with a PM2.5 and a PM10 inlet at a flow rate of 50 m3/24 h. Periods of 24-h sampling were carried out. The samples were collected using quartz fiber filters (Munktell, Bärenstein, Germany). The filters were conditioned in a borosilicate desiccator (Duran, Mainz, Germany) for 24 h prior to use.

Reagents and Standards

Ultrapure water (resistivity >18.2 M_- cm, at 25 °C) produced with a Milli-Q RG Ultrapure Water Purification System (Millipore, Billerica, MA) and decontaminated plastic labware (by immersion in 10% v/v HNO3 during at least 24 h and thoroughly washed with ultrapure water) were used. Diluted HNO3 solutions were prepared from concentrated (\geq 69% w/w), high-purity HNO3 (TraceSELECT, Fluka, L'Isle d'Abeau Chesnes, France) with ultrapure water. Solutions for ICP-MS analysis were prepared from AccuTrace (AccuStandard, New Haven, CT) solutions: ICP-MS-200.8- CAL1R-1 (multi-element calibration standard), ICP-MS-200.8-IS-1 (internal standards), and ICP-MS-200.8-TUN-1 (tuning solution).

Filter Extraction

Filter extraction was performed according to a standard operating procedure (U.S. Environmental Protection Agency [EPA], 2005).

Briefly, filter samples were sonicated for 3 h with 20 ml 4% HNO₃ (v/v) at 69 °C. Subsequently, samples were filtered through a Whatman 41 filter paper and diluted to 50 ml with ultrapure water. Samples were kept in plastic centrifuge tubes at 4 °C until ICP-MS analysis.

Analysis

Extracts were submitted to trace element analysis by ICP-MS, performed using a VG Elemental (Winsford, UK), PlasmaQuad 3 instrument, equipped with a Meinhard type A pneumatic concentric nebulizer, a quartz water-cooled impact bead spray chamber, a standard quartz tube torch, and nickel sample and skimmer cones. Both the spray chamber and sampling interface were cooled to 10 °C by circulating water. High-purity (99.9999%) argon (Gasin II, Leça da Palmeira, Portugal) was used as the nebulizer and plasma gas. A Gilson (Villier le Bel, France) model Minipuls 3 peristaltic pump was used for sample introduction.

Operating conditions for ICP-MS determinations were as follow: RF forward power 1350 W; nebulizer gas flow rate 0.82 L/min; auxiliary gas flow rate 0.9 L/min; cool gas flow rate 13 L/min; dwell time 10 ms; sweeps per replicate 100; detector, sequential; acquisition mode, peak jumping. The elemental isotopes (*m*/*z* ratios) 9Be, 27Al, 51V, 52Cr, 55Mn, 59Co, 60Ni, 65Cu, 66Zn, 75As, 82Se, 95Mo, 111Cd, 121Sb, 205Tl, 208Pb, 232Th, 238U (as analytical masses) and 45Sc, 89Y, 115In, 159Tb, and 209Bi (as internal standards) were monitored. 83Kr and 77ArCl were also monitored to correct for instrumental isobaric interferences. The instrument was tuned for maximum signal sensitivity and stability using 115In as the target isotope. Limits of detection (LOD) were calculated as the concentration corresponding to 3 times the standard deviation of 10 repeated determinations of the calibration blank signal.

Blank filters were analyzed in the same batch as filter samples, and their metal content was used to correct the metal content of samples. Results were expressed as nanograms per cubic meter, using the calculated volume of air sampled.



FIGURE 1. Location of the sampling sites.

Analytical Quality Control

One spike filter and one blank filter were analyzed in every set of 10 samples to determine the matrix effects and background levels from the filters, respectively. The obtained values are presented in Table 1.

Enrichment Factor (EF)

Enrichment factors (EFs) were calculated to show the degree of enrichment of a given element in PM compared to its relative abundance in crustal material. Typically, this is the first step in attempting to evaluate the importance of crustal and non-crustal sources. The enrichment factor is defined as $EF = (X/AI)_{air}/(X/AI)_{crust}$, where $(X/AI)_{air}$ and $(X/AI)_{crust}$ refer to the ratio of the concentration of element the X to that of AI in the air and in the average crustal material, respectively (Taylor and McLennan, 1985). AI is a common reference element in EF calculations based on its abundance and stability in the crustal surface. The crustal composition for AI was assumed to be 8% (w/w) (Kabata-Pendias, 2011). EFs equal to 1 indicate crustal origin for a given element, whereas EFs > 10 indicate anthropogenic origin.

Element	Spike levels (µg/L)	Found (µg/L)	RSD (%)	Recovery (%)
Be	0	0.07	9.6	98.5
	1	1.05	4.7	
Al	0	5.95	4.6	108.5
	1	7.03	3.2	
V	0	0.05	6.8	97.7
	1	1.03	3.2	
Cr	0	1.39	3.9	103.7
	1	2.43	2.0	
Mn	0	0.35	6.4	100.8
	1	1.35	2.6	
Co	0	0.02	9.1	99.4
	1	1.01	2.6	
Ni	0	0.75	5.6	107.6
	1	1.82	5.8	
Cu	0	1.26	5.6	99.1
	1	2.25	2.3	
Zn	0	2.73	3.9	97.7
	1	3.71	3.4	
As	0	0.02	13.3	95.5
	1	0.97	9.0	
Se	0	0.12	8.9	97.0
	1	1.09	7.8	
Mo	0	0.13	4.8	99.7
	1	1.12	2.2	
Cd	0	0.02	13.5	96.5
	1	0.98	5.1	
Sb	0	0.02	10.7	94.6
	1	0.96	4.7	
TI	0	0.01	10.1	98.1
	1	0.99	6.4	
Pb	0	0.20	5.1	97.4
	1	1.17	4.2	
Th	0	0.12	5.9	99.9
	1	1.12	3.9	
U	0	0.01	13.6	100.3
	1	1.02	4.9	

TABLE 1. Recovery (%) and Precision (% RSD) of the Sample Preparation Procedure

Note. Data are presented as the mean (n = 8).

Human Health Risk Assessment

Human health risk assessment was performed based on the average content of PMbound trace elements determined. Generally, a human health risk assessment involves four recognized steps: 1. Hazard identification: Determine which trace elements are known to produce toxicity. Al, Cr(III), Mn, V, Zn, Se, Sb, and Mo are considered to induce noncarcinogenic effects, while As, Cd, Cr(VI), and Ni induce carcinogenic health effects.

2. Exposure assessment: Estimate the chronic daily intake (CDI) of these trace elements. This was calculated from the following equation:

$$CDI (mg/kg/day) = \frac{\text{total dose (TD, mg/m^3)} \times \frac{\text{inhlation rate (IR, m^3/day)}}{\text{body weight (BW, kg)}}$$
(1)

where TD = $C \times E$, where C is the concentration of the pollutant and E the deposition fraction of particles by size, given by E = $-0.081 + 0.23 \ln(D_p)_2 + 0.23 \operatorname{sqrt}(D_p)$, where D_p is the diameter of particles.

Inhalation rate (IR) is typically assumed to be 20 m₃/d and body weight (BW) to be 70 kg for male adults.

3. Dose-response assessment: Evaluation of the probability of health effects produced by dose of pollutant. Assuming inhalation as the single exposure route, the reference dose (RfD, mg/kg/d) for noncarcinogenic toxic trace elements was calculated from reference concentrations (RfC, mg/m3) provided by the U.S. Environmental Protection Agency (EPA). Similarly, for carcinogenic elements the inhalation slope factor (SF, mg/kg/d) was calculated from inhalation unit risk values (IUR, mg/m3) provided by the U.S. EPA.

4. Risk characterization: Estimation of health risk by combining data from exposure and doseresponse assessment. For noncarcinogenic trace elements, risk is indicated by the hazard quotient (HQ), where HQ = CDI/RfD. This HQ is an estimate of the systemic toxicity posed by a single chemical. For the systemic toxicity due to all chemicals within a single exposure the hazard index (HI) is used, where HI =_ HQ. For carcinogenic trace elements, total carcinogenic risk is estimated in terms of "excess life time cancer risk" (ELCR), where ELCR = CDI \times SF (U.S. Department of Energy [DOE], 1999).

Statistical Data Analysis

Statistical data analysis was performed with SPSS for Windows, version 22 (SPSS, Chicago, IL). The criterion for significance was set as p < .05.

RESULTS AND DISCUSSION

The results for the set of 18 elements (Be, Al, V, Cr, Mn, Co, Ni, Cu, Zn, As, Se, Mo, Cd, Sb, Tl, Pb, Th, U) determined in PM2.5 and PM10 are presented in Figures 2 and 3, respectively. Values of Be, Tl, Th, and U are not shown because they were always below LOD (Be: 3 ng/L; TI: 4 ng/L; Th: 2 ng/L; U: 1 ng/L).

Trace Elements Concentrations in Urban, Suburban, and Rural Areas

The most abundant elements in all air samples were Al and Zn, while the least abundant were Mo and Co. Overall, significantly higher trace elements content was observed in SS1 compared to SS2 and REF. The only exception was Cr, which showed a higher content in SS2 (Table 2). In particular, the content of Cd, Pb, and Sb in SS1 was severalfold higher than in the other two locations (SS2 and REF): 15.2- and 39.3- fold higher for Cd; 5.8- and 48- fold higher for Sb; and 4.8- and 9.6-fold higher for Pb.

While some trace elements emissions are due to natural processes such as volcanic eruptions, dust storms, rock weathering, forest fires, and sea salt, most are of anthropogenic origin. Elements such as As, V, Co, Mo, Ni, Se, Sb, Cr, and Mn are emitted during coal combustion (Xu et al., 2011), As, Cr, Cu, Mn, and Zn are released into the atmosphere by smelting industries (González-Castanedo et al., 2014), road traffic involves the emission of a wide range of trace elements that includes Pb, Cu, Zn, and Cd (Gunawardena et al., 2012), and Ni and V are tracers for oil combustion (Peltier et al., 2009; Peltier and Lippmann, 2010). Porto city (SS1) is the largest urban center of PMA and the second largest city in Portugal.

Within the Porto urban area there are several industries of particular significance, such as a refinery, a power plant, a steel factory, manufacturing industries, an international airport, a large seaport infrastructure, and a large co-generation urban solid waste incineration facility. These facilities are well-known important sources of air pollution. Thus, the higher content of trace

elements, especially Cd, Sb, and Pb, in SS1 might be attributed to these industrial activities and traffic emissions. It is well established that Se and As are tracers for coal combustion, while Ni and V are tracers for oil combustion. The concentrations of both Se and As, as measured in PM10, were higher than those at the reference site (Table 2). In addition, most of the Se and As was found to be associated with PM2.5 (Table 3). This suggests that coal combustion is a major source of urban and suburban PM2.5. With regard to Ni, concentrations in SS1 and SS2 are essentially similar, while Ni concentrations, as measured in PM10, at the reference site are lower.

In contrast, for V, the concentrations at SS1 are higher than those at SS2 (Figures 2 and 3). The Ni/V ratio is higher when from combustion sources relying on residual oil or bunker oil than on fuel oil for space heating or motor vehicle fuels (Peltier et al., 2009; Peltier and Lippmann, 2010). Data suggest that on-board heavy oil combustion in the seaport at Porto may be a significant source of PM2.5 in Porto.

As expected, Al was the most abundant trace element in PM10 in SS1 since it is the major crustal constituent of airborne soil, and road dust is known to make an important contribution to coarse aerosol (Gugamsetty et al.,



FIGURE 2. Concentration of PM2.5-bound trace elements (ng/m³) in ambient air at SS1 (urban site) and SS2 (suburban site).



FIGURE 3. Concentration of PM₁₀-bound trace elements (ng/m³) in ambient air at SS1 (urban site), SS2 (suburban site), and REF (reference site).

2012).

Trace Elements Associated With Fine (PM2.5) and Inhalable (PM10) Particles

Figures 2 and 3 show the content (mean • $\}$ SD) of the analyzed trace elements in PM2.5 and PM10 particles at the SS1 and SS2 (at REF only PM10 particles were monitored). Interestingly, the most abundant element in PM2.5 was Zn, while in PM10 it was Al. After Al and Zn, the most abundant elements were Cu, Pb, and Mn. Table 3 shows the ratios of the content of each trace element associated to PM10 and PM2.5. A ratio equal to 1 indicates that all trace elements are present in PM2.5. A ratio of 2 infers that the trace element is equally distributed between PM2.5 and "inhalable coarse particles" (i.e., particles with diameters between 2.5 μ m and 10 μ m). Independently from the sampling site, most of the trace elements amounts in air were associated with PM2.5. The most important exceptions were Al and Mn, elements that were predominantly associated with PM10 (PM10/PM2.5 ratio higher than 3).

Environmental aerosols toxicity is probably linked not only to increased PM mass, but also to variations in particle size, shape, and chemistry, particularly differences in the concentration of highly toxic elements (Cakmak et al., 2014; Chen and Lippmann, 2009; Satsangi et al., 2014).

	Ratio			
Trace element	SS1/SS2	SS1/REF	SS2/REF	
Al	3.2	4.2	1.3	
As	2.0	8.2	4.1	
Cd	15.2	39.3	2.6	
Co	3.3	5.5	1.7	
Cr	0.7	5.7	8.5	
Cu	5.0	7.0	1.4	
Mn	2.7	4.5	1.6	
Mo	2.8	4.9	1.7	
Ni	1.2	5.8	4.9	
Pb	4.8	9.6	2.0	
Se	2.6	13.1	5.0	
Sb	5.8	48.0	8.3	
V	2.8	5.5	1.9	
Zn	2.5	5.4	2.1	

TABLE 2. Ratio Between the Trace Element Concentration (ng/m^3) in Ambient Air at the Different Sampling Sites, as Measured in PM₁₀

Note. SS1, urban site; SS2, suburban site; REF, reference site.

TABLE 3. Ratio Between the PM_{10} - and $PM_{2.5}$ -Bound Trace Element Concentration (ng/m³) in Ambient Air at SS1 (Urban Site) and SS2 (Suburban Site)

Trace element	SS1 (PM ₁₀ /PM _{2.5})	SS2 (PM ₁₀ /PM _{2.5})
AI	3.3	3.2
As	1.1	1.0
Cd	1.0	1.2
Co	1.5	2.3
Cr	1.3	1.1
Cu	2.1	1.8
Mn	3.2	2.6
Mo	2.0	1.6
Ni	1.7	1.4
Pb	1.0	1.1
Se	1.1	1.3
Sb	1.0	1.1
V	1.5	2.4
Zn	1.3	1.4

Enrichment Factor Analysis

Enrichment factors close to 1 were observed for Al and Mn. Very high EFs (>1000) were obtained for Se, Cd, and Sb in both PM2.5 and PM10 at SS1 (Figure 4a). High EFs (>100) were also observed for As, Mo, Pb, and Zn in both PM2.5 and PM10 at the same sampling site (SS1). At SS2, a similar trend was observed: high EF (>1000) for Sb and Se, followed by As, Cd, Mo, Pb, and Zn (all with EF > 100) in both PM2.5 and PM10 (Figure 4b). These results indicate noncrustal sources for these trace elements in ambient air. As indicated previously, several industrial facilities, including a steel factory, power plant, refinery, and solid waste incineration plant, exist at PMA, and there is sufficient evidence that these facilities are important sources of Cd, Sb, Se, and As (González-Castanedo et al., 2014; Xu et al., 2011). Further, traffic-related

emissions have also been recognized as one of the main sources of trace elements like Pb and Zn (Gunawardena et al., 2012) that also displayed high EF. In contrast, Al, Co, and Mn presented low EF, suggesting that crustal sources such as street dust are the predominant sources of these elements in the ambient air (Gugamsetty et al., 2012).

Human Health Risk Assessment

A human health risk assessment for PMbound trace elements was carried out. The pertinent values of CDI, RfD, HQ, SF, and ELCR for adults are given in Table 4. The mean concentration levels (ng/m3) of different trace elements were used for calculations. The worst-case scenario (SS1) was selected for this risk assessment. As shown in Table 4, for adults, the level of noncarcinogenic risk (HI) was estimated to be 1.06 \times 10-3 for PM2.5 and 4.61×10^{-3} for PM10, and carcinogenic risk (sum of all ELCR) was estimated to be 1.14×10^{-3} -5 for PM2.5 and 5.58 \times 10-5 for PM10. Based upon these findings, it can be concluded that the noncarcinogenic risk (indicated by both HQ and HI) falls well below acceptable levels (<0.2 for HQ and <1 for HI). In contrast, carcinogenic risk (indicated by ELCR), was found to be not negligible (i.e., $>1 \times 10-5$) (U.S. DOE, 1999). Carcinogenic risk was also estimated for SS2 and again observed to be not negligible. However, it needs to be pointed out that Cr was considered to be all in its hexavalent form (Cr_{6+}) in this assessment, which typically is not the case (Wilbur et al., 2012). The carcinogenic risk assessment performed here only considered PM-bound trace elements. Since other compounds such as PAH may also account for carcinogenic risks, a more comprehensive and detailed study is necessary for a complete risk assessment.



FIGURE 4. Enrichment factors (EF) of trace elements associated to (a) PM2.5 and (b) PM10.

	CDI	RfD		SF	
Trace element	(mg/kg/day)	(mg/kg/day)	HQ	(mg/kg/day)	ELCR
(a) PM _{2.5}					
Noncarcinogenic tr	ace elements				
Al	43.3×10^{-7}	$1.00 \times 10^{\circ}$	0.43×10^{-5}	-	_
Cr	1.71×10^{-7}	3.00×10^{-3}	5.71×10^{-5}	_	-
Mn	4.48×10^{-7}	1.40×10^{-1}	0.32×10^{-5}	-	-
Mo	0.67×10^{-7}	5.00×10^{-3}	1.34×10^{-5}	_	-
Se	3.96×10^{-7}	5.00×10^{-3}	7.93×10^{-5}	-	-
Sb	3.35×10^{-7}	4.00×10^{-4}	83.7×10^{-5}	_	-
V	3.27×10^{-7}	9.00×10^{-3}	3.63×10^{-5}	-	-
Zn	83.2×10^{-7}	3.00×10^{-1}	2.77×10^{-5}	-	-
			$\Sigma = 1.06 \times 10^{-3}$	_	-
Carcinogenic trace	elements				
As	1.84×10^{-7}	-	_	15.1	2.77×10^{-6}
Cd	2.26×10^{-7}	_	-	6.3	1.42×10^{-6}
Cr	1.71×10^{-7}	_	_	41	7.06×10^{-6}
Ni	2.70×10^{-7}	-	-	0.84	0.23×10^{-6}
					$\Sigma = 1.14 \times 10^{-5}$
(b) PM ₁₀					
Noncarcinogenic tra	ace elements				
AI	56.5×10^{-6}	1.00 × 10°	0.57×10^{-4}	-	-
Cr	0.90×10^{-6}	3.00×10^{-3}	2.99×10^{-4}	-	-
Mn	5.68×10^{-6}	1.40×10^{-1}	0.41×10^{-4}	-	-
Mo	0.52×10^{-6}	5.00×10^{-3}	1.04×10^{-4}	-	-
Se	1.65×10^{-6}	5.00×10^{-3}	3.30×10^{-4}	-	-
Sb	1.37×10^{-6}	4.00×10^{-4}	34.2×10^{-4}	-	-
V	1.95×10^{-6}	9.00×10^{-3}	2.16×10^{-4}	-	-
Zn	44.0×10^{-6}	3.00×10^{-1}	1.47×10^{-4}	-	-
			$\Sigma = 4.61 \times 10^{-3}$	-	-
Carcinogenic trace	elements				
As	7.92×10^{-7}	_	_	15.1	1.20×10^{-5}
Cd	8.82×10^{-7}	_	_	6.3	0.56×10^{-5}
Cr	8.96×10^{-6}	_	_	41	3.67×10^{-5}
Ni	1.83×10^{-6}	_	-	0.84	0.15×10^{-5}
					$\Sigma = 5.58 \times 10^{-5}$

CONCLUSIONS

PMA ambient air quality was studied regarding several trace elements concentrations as measured bound to both PM2.5 and PM10.

Taking as reference EU standards, the air quality in the studied area may be regarded as good. The concentrations of almost all elements with the exception were Cr decreased from the urban to the suburban sampling site. Arsenic, Cd, Ni, Pb, Sb, Se, V, and Zn were mostly associated with PM2.5, suggesting that anthropogenic sources such as industry and road traffic are the main sources of these elements, which was further confirmed by high enrichment factors (EF > 100) obtained for As, Cd, Pb, Sb, Se, and Zn. A health risk assessment showed a negligible noncarcinogenic risk (HQ and HI well below 0.2 and 1, respectively), but a nonnegligible carcinogenic risk (>1 \times 10-5) if all Cr in the ambient air is present as Cr(VI).

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