Original article:

DESIGN, SYNTHESIS AND ANTICONVULSANT ACTIVITY OF SOME NEW 5,7-DIBROMOISATIN SEMICARBAZONE DERIVATIVES

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ABSTRACT

A series of 5,7-dibromoisatin semicarbazones have been synthesized in good yield, involving aryl urea and aryl semicarbazide formation. The structures of the synthesized compounds were confirmed on the basis of their spectral data. All the compounds were evaluated for anticonvulsant and CNS depressant activities. Anticonvulsant activity was determined after intraperitoneal (i.p.) administration to mice by maximal electroshock (MES) induced seizure method and minimal motor impairment was determined by rotarod test. A computational study was carried out for prediction of pharmacokinetic properties and making them potentially promising agents for the treatment of epilepsy. Compounds (Z)-1-(5,7-dibromo-2-oxoindolin-3-ylidene)-4-(4-chlorophenyl)semicarbazide (**DH-05**), (Z)-1-(5,7-dibromo-2-oxoindolin-3-ylidene)-4-(3-chloro-4-fluorophenyl)semicarbazide (**DH-11**) and (Z)-1-(5,7-dibromo-1methyl-2-oxoindolin-3-ylidene)-4-(3-chloro-4-fluorophenyl)semicarbazide (**DH-11**) and (Z)-1-(5,7-dibromo-1methyl-2-oxoindolin-3-ylidene)-4-(3-chloro-4-fluorophenyl)semicarbazide (**DH-11**) and (Z)-1-(5,7-dibromo-1methyl-2-oxoindolin-3-ylidene)-4-(3-chloro-4-fluorophenyl)semicarbazide (**DH-12**) exhibited prominent anticonvulsant effect in the series with little or no neurotoxicity and little CNS depressant effect as compared to standard drug.

Keywords: Dibromoisatin, anticonvulsant, maximal electroshock, antidepressant activity

INTRODUCTION

Epilepsy is a disease of complex nature and of different etiology with huge patient load of varying age groups involving both sexes (Verma et al., 2004). On an average, 0.25 million new cases of epilepsy are reported every year. Mainly two kinds of epilepsy, namely grand mal and petit mal are prevalent (Siddiqui et al., 2007). Available drugs are effective in only 60 - 80 % of epileptic patients.

During the past decade, several new drugs have been approved to be used as antiepileptic drugs (Rufinamide, Retigabine, Pregabaline, Remacemide, etc.) (Yogeeswari et al., 2005). Despite the optimal use of available antiepileptic drugs (AEDs), many patients fail to experience seizure control and others do so only at the expense of significant toxic effects. It is estimated that available medication controls seizures in only 50 % of patients or decrease incidence in only 75 % of patients (Agarwal et al., 2004). Anticonvulsant drugs, showing activity with MES (maximal electroshock) test are generally useful in grand mal (Verma et al., 2004).

Literature survey showed that isatin molecule has positive anticonvulsant effects during its initial screening in the MES test (Pandeya et al., 2002; Julian et al., 1952; Popp 1984; Jursic et al., 2002; Hewawasam et al., 2002). Substitution of hydrogen (from –NH in indole nucleus) with methyl group enhanced the lipophilicity of the compounds (Smitha et al., 2008). In recent times, semicarbazones are emerging as novel anticonvulsant drugs. More than 300 compounds have been prepared and tested in animal models. Representative semilike 4-bromobenzaldehvde carbazones. semicarbazone, have shown to have activity comparable with or even exceeding that of phenytoin (Dilantin) in the maximal electroshock (MES)-induced seizure test in mice (Siddiqui et al., 2007). Aryl semicarbazones displaying anticonvulsant activity in the MES screen interact at a specific binding site referred as the hydrogen bonding area and aryl binding site respectively (Dimmock et al., 1995). Terminal amino group of aryl semicarbazones was shown to affect the binding properties at the hydrogen bonding area (Dimmock et al., 1999; Pandeva et al., 1998). However, Pandeva et al. have suggested a new pharmacophore model for semicarbazones displaying anticonvulsant activity (Figure 1). They proposed that terminal amino function of semicarbazones could be substituted with a lipophilic substituted aryl ring (Jain et al., 2010). Proposed pharmacophore model contains four binding sites for interaction with a macromolecular complex in vivo.

- 1. An aryl hydrophobic binding site (A) with halogen substituent preferably at para position
- 2. A hydrogen bonding domain (HBD)
- 3. An electron donor group (D)
- 4. Another hydrophobic-hydrophilic site controlling the pharmacokinetic properties of the anticonvulsant (C)

These new aspects might be useful for designing prototypic molecules with potential anticonvulsant activity (Figure 2). These observations encouraged us to synthesize a series of 5,7-dibromoisatin semicarbazones which were evaluated their anticonvulsant and toxicological profiles.

CHEMISTRY

The synthesis of 5,7-dibromoisatin semicarbazones DH-01-DH-14 was achieved as depicted in Scheme 1. Substituted aniline was treated with sodium cyanate in the presence of glacial acetic acid according to the known urea preparation method, to yield aryl urea. Then aryl urea on condensation with hydrazine hydrate in ethanol in the presence of sodium hydroxide gave the aryl semicarbazide. The 5,7dibromoisatin semicarbazones DH-01-DH-14 were prepared by reaction of the appropriate 5,7-dibromoisatin or N-methyl 5,7dibromoisatin with arvl semicarbazide in presence of anhydrous sodium acetate. The synthesized products showed a single spot on TLC and had IR and ¹H-NMR spectra in accordance with their anticipated structures. In general, IR spectra showed the C = Npeak at $1610 - 1598 \text{ cm}^{-1}$ and the NH stretching vibrations at 3424 - 3315 cm⁻¹. The ¹H-NMR spectrum revealed that the CO-NH proton at 12.19 - 11.25 and the aryl NH proton that showed a singlet at 9.99 - 9.72 were D₂O exchangeable. The physical characterization data of the synthesized compounds are given in Table 1.



Figure 1: Suggested pharmacophore model for semicarbazones displaying anticonvulsant activity



Table 1: Physico-chemical parameters data of the newly synthesized derivatives

Code-No	R1	R2	R3	R4	Mol. Formula	Mol. Wt	Rr	% yield	M.P.(°C)	Log P
DH-01	- 20	а.	8 -	H	C9H6N4O2Br2	362	0.31	54	253 - 256	1.36
DH-02	-	÷	-	CH3	C10H8N4O2Br2	376	0.43	42	212 - 214	1.59
DH-03	н	Н	н	H	C15H10N4O2Br2	409	0.44	63	224 - 226	3.25
DH-04	н	н	н	CH3	C16H12N4O2Br2	423	0.60	66	240 - 243	3.49
DH-05	CI	Н	Н	н	C15H9N4O2Br2CI	443.5	0.39	50	177 - 179	3.81
Dh-06	CI	H	Н	CH3	C16H11N4O2Br2Cl	457.5	0.45	44	287 - 289	4.05
DH-07	Br	н	Н	H	C15H9N4O2Br3	488	0.36	63	188 - 190	4.08
DH-08	Br	н	н	CH3	C16H11N4O2Br3	502	0.45	65	273 - 275	4.32
Dh-09	н	н	CI	Н	C15H9N4O2Br2CI	443.5	0.41	57	180 - 182	3.81
DH-10	CI	Н	CI	CH3	C16H11N4O2Br2Cl	257.5	0.48	51	225 - 227	4.05
DH-11	F	CI	н	Н	C15H8N4O2Br2FCI	461.5	0.39	41	156 - 158	3.97
DH-12	F	CI	Н	CH3	C16H10N4O2Br2FCI	475.5	0.47	39	266 - 268	4.21
DH-13	CH3	Н	Н	Н	C16H12N4O2Br2	423	0.55	45	220 - 222	3.74
DH-14	CH3	н	Н	CH3	C17H14N4O2Br2	437	0.47	48	214 - 216	3.98



Scheme 1 for synthesis of 5,7-dibromoisatin semicarbazone derivatives

PHARMACOLOGY

Anticonvulsant screening

The anticonvulsant evaluation of the synthesized compounds was performed using reported procedure (Krall et al., 1978). Male albino mice (CF-1 strain, 18 - 25 g) were used as experimental animal. The synthesized compounds were suspended in polyethylene glycol (PEG- 400). All experimental protocols were carried out with permission from the Institutional Animal Ethics Committee (IAEC). Animals were obtained from the Central Animal House Facility, DR. K. N. Modi Institute of Pharmaceutical Education and Research. Modinagar, Ghaziabad, Uttar Pradesh, India.

Electroshock seizure method

Initial anticonvulsant evaluation of test compounds **DH-01–DH-14** were under taken by maximal electroshock seizure (MES) test following the method of Krall et al. (1978). Maximal seizures were induced by the application of electrical current to the brain via corneal electrodes. For all tests based on MES convulsions, 60 Hz of alternating current (50 mA) was delivered for 0.2 s by corneal electrodes which had been primed with an electrolyte solution containing an anaesthetic agent (0.5 % tetracaine HCl). Abolition of the hind limb tonic extensor spasm was recorded as a measurement of anticonvulsant activity.

Rotarod motor impairment screening

Motor impairment is measured in mice by the rotarod test (Dunham et al., 1957). The mice were trained to stay on an accelerating rotarod (INCO, Ambala, India) that rotated at six revolutions per minute. The rod diameter was 3.2 cm. Motor impairment was indicated by the inability of the animal to maintain equilibrium on the rod for at least 1 min in each of the trials.

CNS depression study

The forced swim method (Porsolt's swim pool test) (Porsolt et al., 1978) was followed to study the CNS depression. Mice were placed in a chamber (diameter 45 cm, height 20 cm) containing water up to a height of 15 cm at 25 ± 2 °C. Two swim sessions were conducted, an initial 15 min pre-test, followed by a 5 min test session 24 h later. The animals were administered an i.p. injection (100 mg/kg) of the test compounds 30 min before the test session. The period of immobility (passive floating without struggling, making only those movements which are necessary to keep its head above the surface of water) during the 5 min test period were measured.

COMPUTATIONAL STUDY

Calculation of physicochemical parameters

A computational study of synthesized compounds (DH-01-DH-14) was performed for prediction of ADME properties. Polar surface area (TPSA) (Ertl et al., 2000), number of rotatable bonds, molecular volume, number of hydrogen donor and acceptor atoms and violations of Lipinski's rule of five (Lipinski et al., 2001) were calculated using Molinspiration online propercalculation toolkit (Molinspiration tv Cheminformatics, 2013). Absorption (%ABS) was calculated by: %ABS=109 - $(0.345 \times \text{TPSA})$ (Zhao et al., 2002). The theoretical toxicity, druglikeness and drugscore properties were calculated in the Osiris Property Explorer (http://www.organicchemistry. org).

RESULTS AND DISCUSSION

Anticonvulsant and CNS depressant activities

Synthesized 5,7-dibromoisatin semicarbazone derivatives were screened for their anticonvulsant activity by using standard model MES (maximal electroshock seizure test) for their ability to reduce seizure spread (Table 2). Motor impairment screening of the synthesized compounds was also carried out by rotarod test method and CNS depressant effect of the compounds was determined by porsolt's force swim pool method. The anticonvulsant activity was tested after 0.5 and 4.0 h time intervals at dose levels of 30, 100, 300 mg/kg body weight. The CNS depressant activity was studied at a dose level of 100 mg/kg body weight. Phenytoin and Carbamazepine were used as standard drugs.

In this series (DH-01-DH-14), compounds DH-03, DH-06 and DH-08 showed protection against induced seizures at maximum dose (300 mg/kg) after 4.0 h. Compound **DH-05** showed protection at a dose of 100 mg/kg at 4.0 h and compounds DH-11 and DH-12 showed the activity at the dose of 30 mg/kg at 0.5 h. The compounds DH-02, and DH-04 showed protection at a dose of 100 mg/kg at 4.0 h and compound DH-09 showed protection at a dose of 100 mg/kg at 0.5 h. Compounds DH-11 and **DH-12** having R_1 = Flouro, R_2 = Chloro, $R_4 = H$, and $R_1 = Flouro$, $R_2 = Chloro$, R_4 = CH₃ groups respectively were found most active in the series showing activity at 0.5 h at lower dose of 30 mg/kg. In the rotarod motor impairment screening, all compounds except DH-03, DH-05, DH-06 and DH-08, showed motor impairment at the maximum dose of 300 mg/kg.

Table 2: Anticonvulsant and motor impairmentscreening of synthesized compounds (DH-01-DH-14) using maximal electroshock seizure(MES) Model and Rotarod Model

Compound	ME	S ^a	Motor Impt. ^a		
Code	0.5h	4.0h	0.5 h	4.0 h	
DH-01	-	-	-	300	
DH-02	-	100	-	300	
DH-03	-	300	-	-	
DH-04	-	100	-	300	
DH-05		100	-	-	
DH-06		300	-	-	
DH-07	-	-	-	300	
DH-08	-	300	-	-	
DH-09	100	-	-	300	
DH-10	-	-	100	-	
DH-11	-	30	300	-	
DH-12	-	30	300	-	
DH-13	-	100	-	100	
DH-14	-	100	-	100	
Control	-	-	-	-	
Phenytoin	30	30	100	100	
Carbama- zepine	30	100	300	-	

^aDoses of 30, 100, and 300 mg/kg were administered to mice through intraperitoneal route. The figures in the table indicate the minimum dose whereby bioactivity was demonstrated in half or more of mice. The animals were examined 0.5 and 4h after the drug administration. The dash (-) indicates an absence of activity at maximum dose administered (300 mg/kg). Data for Phenytoin and Carbamazepine, used as standard drugs, were obtained referring, Dimmock, J.R., et al 1995 & White, H.S., et al 1995.

Some selected compounds i.e. **DH-05**, **DH-11** and **DH-12** having significant anticonvulsant activity were also tested for their CNS depressant effect. These compounds with respect to control, showed 29.05, 51.24 and 32.95 % increase in immobility time respectively. The standard drug carbamazepine showed 58.63 % increase in the immobility time (Table 3). Percentage CNS depression by **DH-11** was in the vicinity of carbamazepine while **DH-05** and **DH-11** fell far behind.

DH-05, with p-chloro in its structure was found to be the most active compound of this study. It was effective in 4.0 h time interval against induced seizure at lower dose of 100 mg/kg, showing no motor im-

pairment effect in comparison to standard drug Carbamazepine. The activity may be due to the presence of chlorine at para position which is well fitted to receptor site though, it showed reduced CNS depressant effect in comparison to standard drug, carbamazepine.

Using the putative binding site theory proposed by Dimmock et al. (2000a, b) subsequently used by others (Pandeya et al., 2002; Gibson et al., 2009) in postulating the interaction of anticonvulsant compounds at a specific binding site, the molecule observed to interact with the protein receptor as shown in Figure 3.



Figure 3: Proposed binding interactions of the title compounds using putative binding site theory (Dimmock et al., 2000a, b; Pandeya et al., 2002; Gibson et al., 2009).

 Table 3: Data of CNS depressant activity of the selected compounds performed in mice using forced swim test

Com- pounds	Duration of immobility (sec) (mean ± SEM)	% In- creased of immobil- ity
DH-05	68.45 ± 5.28	29.05
DH-11	80.22 ± 2.66	51.24
DH-12	70.52 ± 2.45	32.95
Carbama- zepine	84.14 ± 1.33	58.63
Control	53.04 ± 2.47	-

Tested at 100 mg/kg (i.p. in PEG 400); control animals were administered PEG 400 (i.p.). Each value represents the mean \pm SEM of six mice. The CNS depressant effect was compared with respect to standard drug. **p* < 0.0001. Data was analyzed by unpaired student's *t* test.

Prediction of ADME properties

A computational study for prediction of ADME properties of synthesized compounds was performed. Topological polar surface area (TPSA), i.e., surface belonging to polar atoms, is a descriptor that was shown to correlate well with passive molecular transport through membranes and, therefore, allows prediction of transport properties of drugs in the intestines and blood-brain barrier crossing (Ertl et al., 2000). The percentage of absorption (%ABS) was calculated using TPSA. From all these parameters, it can be observed that all synthesized compounds exhibited a great %ABS ranging from 74.38 to 82.955 % (Table 4). Furthermore, compounds DH-01, DH-07 and DH-08, DH-12 violated only one and two Lipinski's parameter respectively. None of the other compounds violated Lipinski's parameters, making them potentially promising agents for epilepsy therapy. Although druglikeness of DH-11 and DH-12 was not comparable to standard drug but DH-05 was shown more druglikeness when compared to the standard drug. Drug scores of DH-11 and **DH-12** were low, but for **DH-05**, it lies in the close proximity with phenatoin. Using orisis programe, theoretical toxicity profile (Mutagenic, tumorigenic, irritant and reproductive effective) of the all three compounds **DH-05**, **DH-11** and **DH-12** were analysed and found to be equal to that of the standard drug Phenatoin (Figure 4).

CONCLUSION

In conclusion, the present study indicated that the synthesized compounds have marked anticonvulsant activity as indicated by the protection against maximal electroshock induced seizures in comparison with standard drug (Phenytoin) with little or no neurotoxicity. Further, the study highlights the importance of the structural features of a compound responsible for the anticonvulsant activity. With semicarbazone structure as a template, many modifications can still be carried out and there is ample scope for further exploration which will lead to a systematic structure-activity relationship.



Figure 4: Graphical representation of Druglikeness and drugscore values of compounds DH-05, DH-11, DH-12 and Phenytoin

Com- pound	% ABS	TPSA (A ²)	n- ROTB	MW	MV	n-OHNH donors	n-ON acceptors	Lipinski's violations
Rule	-	-	-	<500	-	<5	<10	<1
DH-1	74.380	100.347	1	361.981	206.293	4	6	1
DH-2	78.125	89.49	1	376.008	223.236	3	6	0
DH-3	79.209	86.351	2	438.079	278.816	3	6	0
DH-4	82.955	75.494	2	452.106	295.758	2	6	0
DH-5	79.209	86.351	2	472.524	292.352	3	6	0
DH-6	82.955	75.494	2	486.551	309.294	2	6	0
DH-7	79.209	86.351	2	516.975	296.701	3	6	1
DH-8	82.955	75.494	2	531.002	313.644	2	6	2
DH-9	79.209	86.351	2	472.524	292.352	3	6	0
DH-10	82.955	75.494	2	486.551	309.294	2	6	0
DH-11	79.209	86.351	2	490.514	297.283	3	6	0
DH-12	82.955	75.494	2	504.541	314.22	2	6	2
DH-13	79.209	86.351	2	452.106	295.377	3	6	0
DH-14	82.955	75.494	2	466.133	312.319	2	6	0

Table 4: Pharmacokinetic parameters important for good oral bioavailability of compounds.

%ABS, percentage of absorption; TPSA, topological polar surface area; n-ROTB, number of rotatable bonds; MW, molecular weight; MV, molecular volume; n-OHNH, number of hydrogen bond donors; n-ON, number of hydrogen bond acceptors

EXPERIMENTAL PROTOCOL

Chemistry

Raw materials for the synthesis were purchased from Merck (India), Spectrochem chemicals (India) and S.D. Fine Chemicals (India). The progress of reaction was monitored by thin layer chromatography, performed on Silica gel-GF coated aluminum plate using iodine vapors and UV light as visualizing agents. Evaporation of solvents was performed at reduced pressure, using a Buchi rotary evaporator. Solid products were purified by recrystallization and column chromatography. Melting points of newly synthesized compounds were determined in open glass capillary with the help of digital melting point apparatus and are uncorrected. FT-IR (KBr) spectra were recorded on a Nicolet 5PC FTIR spectrophotometer (λ -max in cm⁻¹). ¹H NMR spectra was recorded on a Brucker Model-300 NMR Spectrometer in DMSO-d6 using tetramethylsilane (TMS) as the internal reference (chemical shifts in δ ppm). Chemical shifts are reported in ppm downfield from tetramethylsilane (TMS).

Synthesis of 5,7-dibromoisatin

The synthesis of 5,7-Dibromoisatin was based on the method of Lindwall. Isatin (5.0 g, 34 mmol, 1 equiv) was warmed in ethanol (95 %, 100 mL) with stirring until it dissolved. Bromine (16.3 g, 102 mmol, 5.2 mL, 3.0 equiv) was added dropwise to the stirred isatin solution whilst maintaining the temperature of the reaction mixture between 70 to 75 °C. The solution was cooled to room temperature and placed on ice for 30 min. The resulting precipitate was washed with water and cold ethanol and then recrystallized from ethanol to yield bright orange-red crystals of 5,7-dibromoisatin (60 %), m.p. 251–254 °C (Vine et al., 2007, 248-250 °C).

Synthesis of N-methyl 5,7-dibromoisatin

To a suspension of (0.1 mol) of dibromo isatin, in 200 ml of anhydrous methanol, 100 mL of 10 % methanolic potassium hydroxide solution was added in portions with stirring for 30 min. To this mixture, 15 mL of dimethyl sulfate was added and after 1 h, the solution was filtered to remove potassium methyl sulfate. After removal of about 250 mL of solvent under reduced pressure, 40 mL of warm water was added to the obtained residue. On cooling, orange precipitate occurred which was filtered and dried to obtain N-Methyl 5,7-Dibromoisatin (Gupta et al., 2010).

Synthesis of substituted aryl semicarbazides

Substituted aryl ureas were prepared according to the prescribed procedure (Furniss et al., 1998). The substituted phenyl semicarbazides (H, 2-Cl, 4-Cl, 4-Br, and 4-CH₃ etc.) were synthesized according to earlier reported methods.

- Phenyl semicarbazide (Pandeya et al., 1999)
- Yield: 62 % m. p. 245°C. (lit. 247°C);
- N-(2-Chlorophenyl)hydrazinecarboxamide (o-Chlorophenyl semicarbazide) (Pandeya et al., 1999)
- Yield : 65 %, m. p. 176°C (lit. 179°C);
- N-(4-Chlorophenyl)hydrazinecarboxamide (p-Chlorophenyl semicarbazide) (Pandeya et al., 1999)
- Yield : 70 %, m. p. 235°C (lit. 234°C);
- N-(4-Bromophenyl)hydrazinecarboxamide (p-Bromophenyl semicarbazide) (Pandeya et al., 2000)
- Yield: 40 %, m. p. 267°C (lit. 270°C);
- 3-Chloro-4-Flourophenyl semicarbazide (Amir et al., 2010)
- Yield: 60 %, m. p. 94°C (lit. m.p 90-92°C.)
- 4-p-tolylsemicarbazide (p-Methylphenyl semicarbazide) (Wheeler et al., 1924,)
- Yield: 61 %, m. p. decomposing at 272°C (lit. m.p decomposing at 274°C.)

General procedure for synthesis of substituted dibromoisatin semicarbazones

Dibromoisatin semicarbazones **DH-01** to **DH-14** were synthesized according to the procedure specified in literature with some modifications (Amir et al., 2010) as shown in synthetic Scheme 1. To a solution of substituted semicarbazide (0.0025 mole) in 0.5 mL conc. HCl and 12.5 mL water, sodium acetate (0.0025 mole, 0.205g) in 1 mL water was added. About 12.5 mL ethanol was

added to clear the turbidity. This solution mixture was added to an equimolar quantity of the appropriate dibromoisatin or Nmethyl dibromoisatin in alcohol. The solution was stirred for 15 min. precipitation occurred immediately (if immediate precipitation did not occur, reaction mixture would be allowed to stand for 1-2 h), and the solids were filtered, dried and purified by recrystallization from ethanol. The homogeneity of the compounds was checked by TLC using n-Hexane + ethyl acetate (1:1) as mobile phase.

- (Z)-1-(5,7-dibromo-2-oxoindolin-3ylidene)semicarbazide **DH-01**
- IR (KBr): $v_{max} = 3267$ (NH), 1720 (C=O), 1610 (C=N), 1021 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-d₆) δ = 7.33 (brs, 2 H, CONH₂), 7.62-7.85 (m, 2H, ArH), 11.56 (s, 1 H, =N-NH), 11.62 (s, 1 H, NH), ppm. Anal. calcd for C₉H₆Br₂N₄O₂ (359.8858): C 29.88, H 1.86, N 15.62; found C 29.94, H 1.70, N 15.70.
- (Z)-1-(5,7-dibromo-1-methyl-2oxoindolin-3-ylidene)semicarbazide **DH-02**
- IR (KBr): $v_{max} = 3251$ (NH), 1725 (C=O), 1598 (C=N), 1025 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-d₆) $\delta = 2.50$ (s, 3 H, N- CH₃), 7.33 (brs, 2 H, CONH₂), 7.62-7.85 (m, 2H, ArH), 11.56 (s, 1 H, =N-NH), ppm. Anal. calcd for C₁₀H₈Br₂N₄O₂ (373.9014): C 31.90, H 2.10, N 14.86; found C 31.98, H 1.99, N 14.98.
- (Z)-1-(5,7-dibromo-2-oxoindolin-3ylidene)-4-phenylsemicarbazide **DH-03**
- IR (KBr): $v_{max} = 3415$ (NH), 1730 (C=O), 1670 (C=O amide), 1601 (C=N), 1015 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-*d*₆) δ = 7.25-7.53 (m, 5 H, ArH), 7.73-7.84 (m, 2 H, ArH), 9.72 (brs, 1 H, ArNH), 11.65 (brs, 1 H, NH), 12.10 (s, 1 H, CONH) ppm. Anal. calcd for C₁₅H₁₀Br₂N₄O₂ (435.9171): C 41.10, H

2.20, N 12.70; found C 41.20, H 2.05, N 12.75.

- (Z)-1-(5,7-dibromo-1-methyl-2oxoindolin-3-ylidene)-4phenylsemicarbazide **DH-04**
- IR (KBr): $v_{max} = 3410$ (NH), 1732 (C=O), 1675 (C=O amide), 1601 (C=N), 1012 (C-Br) cm⁻¹.
- ¹H NMR (300 MH_Z, DMSO-*d*₆) δ = 2.58 (s, 3 H, N- CH₃), 7.23-7.63 (m, 5 H, ArH), 7.72-7.81 (m, 2 H, ArH), 9.78 (brs, 1 H, ArNH), 12.10 (s, 1 H, CONH) ppm. Anal. calcd for C₁₆H₁₂Br₂N₄O₂ (452.1001): C 42.45, H 2.60, N 12.38; found C 42.48, H 2.55, N 12.42.
- (Z)-1-(5,7-dibromo-2-oxoindolin-3ylidene)-4-(4chlorophenyl)semicarbazide **DH-05**
- IR (KBr): $v_{max} = 3424$ (NH), 1738 (C=O), 1657 (C=O amide), 1607 (C=N), 1018 (C-Br) cm⁻¹.
- ¹H NMR (300 MH_Z, DMSO-*d*₆) δ = 7.15 (d, *J* = 8.1 Hz, 2 H, ArH), 7.75 (d, *J* = 8.1 Hz, 2 H, ArH), 7.76-7.78 (m, 2 H, ArH), 9.76 (brs, 1 H, ArNH), 11.62 (brs, 1 H, NH), 12.03 (s, 1 H, CONH) ppm. Anal. calcd for C₁₅H₉Br₂ClN₄O₂ (469.8781): C 38.16, H 1.90, N 11.80; found C 31.20, H 1.88, N 11.84.
- (Z)-1-(5,7-dibromo-1-methyl-2oxoindolin-3-ylidene)-4-(4chlorophenyl)semicarbazide **DH-06**
- IR (KBr): $v_{max} = 3343$ (NH), 1730 (C=O), 1657 (C=O amide), 1610 (C=N), 1012 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-*d*₆) $\delta = 2.51$ (s, 3 H, N- CH₃), 7.40 (d, *J* = 6.9 Hz, 2 H, ArH), 7.80 (d, *J* = 14.1 Hz, 2 H, ArH), 8.06-8.40 (m, 2 H, ArH), 9.75 (brs, 1 H, ArNH), 11.25 (s, 1 H, CONH) ppm. Anal. calcd for C₁₆H₁₁Br₂ClN₄O₂ (483.8937): C 39.45, H 2.18, N 11.50; found C 39.55, H 2.30, N 11.48.

- (Z)-1-(5,7-dibromo-2-oxoindolin-3ylidene)-4-(4-bromophenyl) semicarbazide **DH-07**
- IR (KBr): $v_{max} = 3324$ (NH), 1740 (C=O), 1648 (C=O amide), 1601 (C=N), 1019 (C-Br) cm⁻¹.
- ¹H NMR (300 MH_Z, DMSO-*d*₆) δ = 7.25 (d, *J* = 8.3 Hz, 2 H, ArH), 7.78 (d, *J* = 8.5 Hz, 2 H, ArH), 7.66-7.79 (m, 2 H, ArH), 9.72 (brs, 1 H, ArNH), 11.70 (brs, 1 H, NH), 12.10 (s, 1 H, CONH) ppm. Anal. calcd for C₁₅H₉Br₃N₄O₂ (513.8276): C 34.80, H 1.70, N 10.80; found C 34.88, H 1.80, N 10.85.
- (Z)-1-(5,7-dibromo-1-methyl-2oxoindolin-3-ylidene)-4-(4bromophenyl)semicarbazide **DH-08**
- IR (KBr): $v_{max} = 3340$ (NH), 1730 (C=O), 1652 (C=O amide), 1609 (C=N), 1017 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-*d*₆) $\delta = 2.56$ (s, 3 H, N- CH₃), 7.50 (d, J = 6.7 Hz, 2 H, ArH), 7.86 (d, J = 14.5 Hz, 2 H, ArH), 8.09-8.42 (m, 2 H, ArH), 9.72 (brs, 1 H, ArNH), 11.65 (s, 1 H, CONH) ppm. Anal. calcd for C₁₆H₁₁Br₃N₄O₂ (527.8432): C 36.15, H 2.10, N 10.55; found C 36.20, H 2.05, N 10.60.
- (Z)-1-(5,7-dibromo-2-oxoindolin-3ylidene)-4-(2chlorophenyl)semicarbazide **DH-09**
- IR (KBr): $v_{max} = 3438$ (NH), 1739 (C=O), 1611 (C=N), 1022 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-*d*₆) $\delta = 7.51-7.63$ (m, 4 H, ArH), 7.87 (m, 2H, ArH), 9.96 (s, 1 H, ArNH), 11.63 (brs, 1H, NH), 12.09 (s, 1 H, CONH). ppm. Anal. calcd for C₁₅H₉Br₂ClN₄O₂ (469.8781): C 38.10, H 1.90, N 11.80; found C 38.15, H 1.88, N 11.84.
- (Z)-1-(5,7-dibromo-1-methyl-2oxoindolin-3-ylidene)-4-(2chlorophenyl)semicarbazide **DH-10**
- IR (KBr): $v_{max} = 3416$ (NH), 1729 (C=O), 1601 (C=N), 1021 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-d₆) $\delta = 2.48$ (s, 3 H,

- N- CH₃), 7.31-7.53 (m, 4 H, ArH), 7.76 (m, 2H, ArH), 9.92 (s, 1 H, ArNH), 12.19 (s, 1 H, CONH.) ppm. Anal. calcd for $C_{16}H_{11}Br_2CIN_4O_2$ (483.8937): C 39.45, H 2.20, N 11.45; found C 39.48, H 2.18, N 11.42.
- (Z)-1-(5,7-dibromo-2-oxoindolin-3ylidene)-4-(3-chloro-4fluorophenyl)semicarbazide **DH-11**
- IR (KBr): $v_{max} = 3235$ (NH), 1728 (C=O), 1607 (C=N), 1021 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-*d*₆) $\delta = 2.51$ (s, 3 H, N- CH₃), 7.27-7.45 (m, 3 H, ArH), 7.86 (m, 2 H, ArH), 9.99 (brs, 1 H, ArNH), 12.06 (s, 1 H, CONH) ppm. Anal. calcd for C₁₅H₈Br₂ClFN₄O₂ (487.8687): C 36.70, H 1.60, N 11.40; found C 36.80, H 1.55, N 11.45.
- (Z)-1-(5,7-dibromo-1-methyl-2oxoindolin-3-ylidene)-4-(3-chloro-4fluorophenyl)semicarbazide **DH-12**
- IR (KBr): $v_{max} = 3233$ (NH), 1727 (C=O), 1604 (C=N), 1019 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-*d*₆) $\delta = 2.49$ (s, 3 H, N- CH₃), 7.25-7.46 (m, 3 H, ArH), 7.82 (m, 2 H, ArH), 9.95 (brs, 1 H, ArNH), 12.11 (s, 1 H, CONH) ppm. Anal. calcd for C₁₆H₁₀Br₂ClFN₄O₂ (501.8843): C 38.10, H 1.98, N 11.02; found C 38.15, H 1.92, N 11.08.
- (Z)-1-(5,7-dibromo-2-oxoindolin-3ylidene)-4-p-tolylsemicarbazide **DH-13**
- IR (KBr): $v_{max} = 3336$ (NH), 1734 (C=O), 1654 (C=O amide), 1598 (C=N), 1013 (C-Br) cm⁻¹.
- ¹H NMR (300 MH_Z, DMSO-*d*₆) δ = 2.74 (s, 3 H, ArCH₃), 7.23 (d, *J* = 7.4 Hz, 2 H, ArH), 7.65 (d, *J* = 6.4 Hz, 2 H, ArH), 7.73-7.75 (m, 2 H, ArH), 9.85 (brs, 1 H, ArNH), 11.58 (brs, 1 H, NH), 12.14 (s, 1 H, CONH) ppm. Anal. calcd for C₁₆H₁₂Br₂N₄O₂ (449.9327): C 42.45, H 2.60, N 12.30; found C 42.48, H 2.58, N 12.35.

- (Z)-1-(5,7-dibromo-1-methyl-2oxoindolin-3-ylidene)-4-ptolylsemicarbazide **DH-14**
- IR (KBr): $v_{max} = 3315$ (NH), 1737 (C=O), 1656 (C=O amide), 1602 (C=N), 1021 (C-Br) cm⁻¹. ¹H NMR (300 MH_Z, DMSO-*d*₆) $\delta = 2.53$ (s, 3 H, N- CH₃), 2.70 (s, 3 H, ArCH₃), 7.45 (d, J = 7.1Hz, 2 H, ArH), 7.68 (d, J = 6.1 Hz, 2 H, ArH), 7.72-7.76 (m, 2 H, ArH), 9.72 (brs, 1 H, ArNH), 12.08 (s, 1 H, CONH) ppm. Anal. calcd for C₁₇H₁₄Br₂N₄O₂ (463.9484): C 43.75, H 3.00, N 12.00; found C 43.80, H 2.98, N 12.21.

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REFERENCES

Agarwal N, Mishra P. Synthesis of 4-aryl substituted semicarbazones of some terpines as novel anticonvulsants. J Pharm Pharmaceut Sci 2004;7:260-4.

Amir M, Ahsan MJ, Ali I. Synthesis of N^{1} -(3-chloro-4-flourophenyl) N^{4} -substituted semicarbazones as novel anticonvulsant agents. Ind J Chem 2010;49:1509-14.

Dimmock JR, Pandeya SN, Quail JW, Pugazhenthi U, Allen TM, Kao GY et al. Evaluation of the semicarbazones, thiosemicarbazones and biscarbohydrazones of some aryl alicyclic ketones for anticonvulsant and other biological properties. Eur J Med Chem 1995;30:303-14. Dimmock JR, Sample HA, John S, Beazely MA, Abrams GD. Isolation and identification of the major urinary metabolite of 4-(4-fluorophenoxy) benzaldehyde semicarbazone after oral dosing to rats. Pharmazie 1999;54:260-2.

Dimmock JR, Vashishtha SC, Stables JP. Anticonvulsant properties of various acetylhydrazones, oxamoylhydrazones and semicarbazones derived from aromatic and unsaturated carbonyl compounds. Eur J Med Chem 2000a;35:241-8.

Dimmock JR, Vashishtha SC, Stables JP. Ureylene anticonvulsants and related compounds. Pharmazie 2000b;55:490-4.

Dunham MS, Miya TA. A note on a simple apparatus for detecting neurological deficit in rats and mice. J Am Pharm Assoc Sci 1957;46:208.

Ertl P, Rohde B, Selzer P. Fast calculation of molecular polar surface area as a sum of fragment-based contributions and its application to the prediction of drug transport properties. J Med Chem 2000;43:3714-7.

Furniss BS, Hannaford AJ, Smith PWG, Tatchell AR. Vogel's textbook of practical organic chemistry. Harlow: Longman, 1998.

Gibson A, Harkless J, Alexander M, Scott KR. Enaminones 10. Molecular modeling aspects of the 5-methylcyclohexenone derivatives. Bioorg Med Chem 2009;17:5342-6.

Gupta L, Sunduru N, Verma A, Srivastava S, Gupta S, Goyal N et al. Synthesis and biological evaluation of new [1,2,4]triazi-no[5,6-b] indol-3-ylthio-1,3,5-triazines and [1,2,4]triazino[5,6-b]indol-3-ylthio-pyrimidines against Leishmania donovani. Eur J Med Chem 2010;45:2359-65.

Hewawasam P, Gribkoff VK, Pendri Y, Dworetzky SI, Meanwell NA, Martinez E et al. The synthesis and characterization of BMS-204352 (MaxiPostTM) and related 3fluorooxindoles as openers of maxi-K potassium channels. Bioorg Med Chem Lett 2002,12:1023–6.

Jain J, Kumar Y, Stables J, Sinha R. Menthone semicarbazides and thiosemicarbazides as anticonvulsant agents. Med Chem 2010;6:44-50.

Julian PL, Meyer EW, Printy HC. The chemistry of indoles. In: Elderfield C (ed): Heterocyclic compounds (pp 201-31). New York: Wiley, 1952.

Jursic BS, Stevens ED. Preparation of dibarbiturates of oxindole by condensation of isatin and barbituric acid derivatives. Tetrahedron Lett 2002;43:5681–4.

Krall RL, Penry JK, White BG, Kupferberg HJ, Swinyard EA. Antiepileptic drug development: II. Anticonvulsant drug screening. Epilepsia 1978;19:409-28.

Lipinski CA, Lombardo L, Dominy BW, Feeney PJ. Experimental and computational approaches to estimate solubility and permeability in drug discovery and development settings. Adv Drug Deliv Rev 2001; 46:3-26.

Molinspiration Cheminformatics, Bratislava, Slovak Republic, Available from: http://www.molinspiration.com/services/pr operties.html, (accessed 16.02.13).

Pandeya SN, Mishra V, Singh PN, Rupainwar DC. Anticonvulsant activity of thioureido derivatives of acetophenone semicarbazone. Pharmacol Res 1998;37:17-22. Pandeya SN, Ponnilavarasan I, Pandey A, Lakhan R, Stables JP. Evaluation of pnitrophenyl substituted semicarbazones for anticonvulsant properties. Pharmazie 1999; 54:923–5.

Pandeya SN, Yogeeswari P, Stables JP. Synthesis and anticonvulsant activity of 4bromophenyl substituted aryl semicarbazones. Eur J Med Chem 2000;35:879–86.

Pandeya SN, Raja AS, Stables JP. Synthesis of isatin semicarbazones as novel anticonvulsants – role of hydrogen bonding. J Pharm Pharm Sci 2002;5:266-71.

Popp FD. Potential anticonvulsants. IX. Some isatin hydrazones and related compounds. J Heterocyc Chem 1984;21:1641– 5.

Porsolt RD, Anton G, Blanet N, Jalfer M. Behavioral despair in rats: a new model sensitive to antidepressant treatments. Eur J Pharmacol 1978;47:379-86.

Siddiqui N, Pandeya SN, Khan SA, Stables J, Rana A, Alam M et al. Synthesis and anticonvulsant activity of sulfonamide derivatives-hydrophobic domain. Bioorg Med Chem Lett 2007;17:255–9.

Smitha S, Pandeya SN, Stables JP, Ganapathy S. Anticonvulsant and sedative-hypnotic activities of N-acetyl/methyl isatin derivatives. Sci Pharm 2008;76: 621–36. Verma M, Pandeya SN, Singh KN, Stables JP. Anticonvulsant activity of schiff bases of Isatin derivatives. Acta Pharm 2004;54: 49–56.

Vine KL, Locke JM, Ranson M, Benkendorff K, Pyne SG, Bremner JB. In vitro cytotoxicity evaluation of some substituted isatin derivatives. Bioorg Med Chem 2007; 15:931–8.

Wheeler AS, Bost RW. 4-para-tolylsemicarbazide and certain derivatives. J Am Chem Soc 1924;46:2813–6.

White HS, Woodhead JH, Franklin MR, Mattson RH, Meldrum BS. Antiepileptic drugs (4th ed., p 99) New York: Raven, 1995.

Yogeeswari P, Sriram D, Veena V, Kavya R, Rakhra K, Ragavendran JV et al. Synthesis of aryl semicarbazones as potential anticonvulsant agents. Biomed Pharmacother 2005;59:51–5.

Zhao YH, Abraham MH, Le J, Hersey A, Luscombe CN, Beck G et al. Rate-limited steps of human oral absorption and QSAR studies. Pharm Res 2002;19:1446-57.