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CORROSION MONITORING DATABASE FOR KRAFT RECOVERY BOILERS

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ABSTRACT

A database containing corrosion-related information for recovery boilers was developed based on published data critically reviewed and extracted following an extensive literature survey. The database can be used either to aid in material selection for a particular environment or to predict the behavior of a material in a variety of environments. The results of a search may be viewed on-screen, printed, or exported to a spreadsheet for further manipulation or presentation in graphical format.

INTRODUCTION

Fireside corrosion of kraft recovery boilers is a continuing problem in the pulp and paper industry. Excessive corrosion causes water-wall thinning and can lead to tube leakage or rupture. This can then lead to smelt-water explosions that cause equipment damage and injuries to mill personnel. The corrosion mechanism in the lower furnace area is believed to be sulfidation. However, other industries, including petroleum, chemical, and electrical power generation, have experienced sulfidation in the past and have conducted research in this area. Some data on the corrosion behavior of materials used in kraft recovery boilers has been generated at IPST, PPRIC, the Swedish Corrosion Institute, and a few other research groups. [1-4]

A detailed survey of the available literature concerning the oxidation and/or sulfidation of various steels, nickel-base alloys, and component metals relevant to recovery boilers was conducted. The articles collected were critically reviewed and any relevant data were extracted for use in the database. The data included is divided into three main categories: isothermal, cyclic temperature, and cyclic gas with or without cyclic temperature. The database includes experimental results on the effect of smelt on the corrosion rate. A strict procedure was followed to ensure that the numerical values entered into the database were accurate and that all data were entered into the appropriate record.

The database contains data collected from 200 references and is fully searchable. These references contained a large amount of data on many materials, gas environments, and smelts applicable to recovery boilers. The desired temperature range for data to be entered into the database was 300°-600°C, however, several graphs contained data at temperatures slightly above or below this target—these data were included for completeness. The exposure time for each data point is included in the data. Although there are approximately 15,000 data points in the database, not all possible combinations of variables are included.

DATABASE DESIGN

After investigating the capabilities of several commercial database packages, it was decided that the corrosion database would be designed using Microsoft®

Access. As mentioned above, a review of the data presented in the collected articles revealed that the data could be broken down into three main categories: isothermal, cyclic temperature, and cyclic gases with or without temperature cycling. This fundamental division was incorporated into the design of the database. The basic structure of the database is based upon tables. The corrosion data collected was entered into three tables corresponding to the division discussed above.

The relationship among the tables is part of the database design and defines the types of questions the database can answer. A forms-based structure is provided to aid the end-user in asking the appropriate questions, called queries. This form structure guides the user through a series of steps in the query process. Each step asks the user to enter a value(s) or select an option(s). The user may go back to a previous form at any point in the process to make any appropriate changes. When all the forms have been completed a query is run on the chosen corrosion data table.

If no data in the database satisfies all the input parameters, a result that may be of interest to the user, a dialog box appears informing the user that no matches have been found. Alternate input values will need to be entered to locate any available data points. If the query does locate data, a results form appears on-screen listing the first value found. Navigation buttons are present on this form that allow the user to view all the records returned. The total number of records returned is also given. A revised query with smaller parameter ranges may be warranted if this value is too large. The user is given several options on how to proceed at this point.

First, the records found may satisfy the needs of the user without further modification. In this case the records may be printed. Second, the data records may be exported to a spreadsheet for further manipulation and/or to put the data into a graphical form. Third, as noted above, a modified query based on the results may be conducted. This third option can be repeated until the user is satisfied with the results. At this point the records may be printed out as in the first case or the data may be further manipulated as in the second case.

Another feature of the database is its capability to search the references by author and/or title keywords. Once the desired reference is located, the data extracted from the associated article are available for retrieval. These data are treated in the same way as the data returned by conventional queries as discussed above.

DATABASE CAPABILITIES

The capabilities of this database can best be illustrated through examples. The first example will also include several figures illustrating the forms used in building up a complete query. Please note that the references cited after the figure captions for the graphs are the sources of the data in the graphs.

Example 1 – Effect of changing temperature on corrosion rate in a constant gas mixture

This is a fairly simple database query that shows that the records in the database will yield results that are consistent with the known corrosion behavior of materials as the temperature increases. In this case a simple gas environment, H₂S, and two materials (carbon steel and 304 SS) with no

smelt present are chosen as parameters for the query. The first form that appears when starting a query is shown as Figure 1.

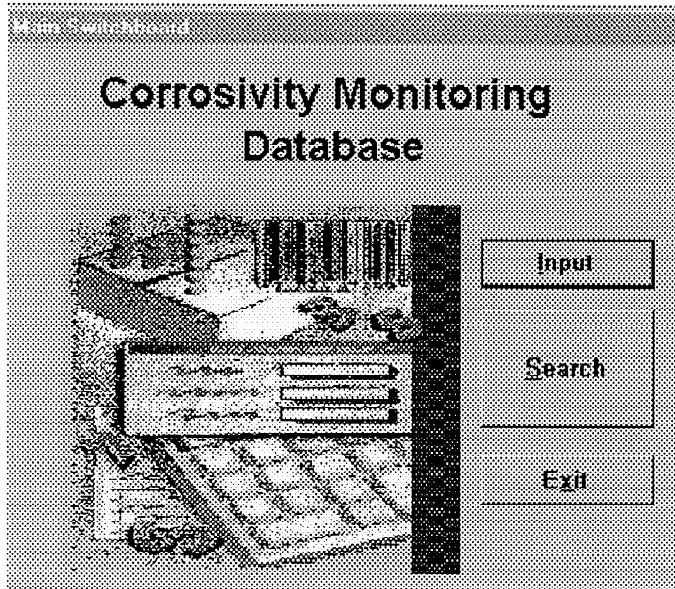


Figure 1 Initial query form

Clicking on the Search button will bring up a second form that allows the user to select a query where all the parameters are required or one in which there is only one parameter needed. See Figure 2.

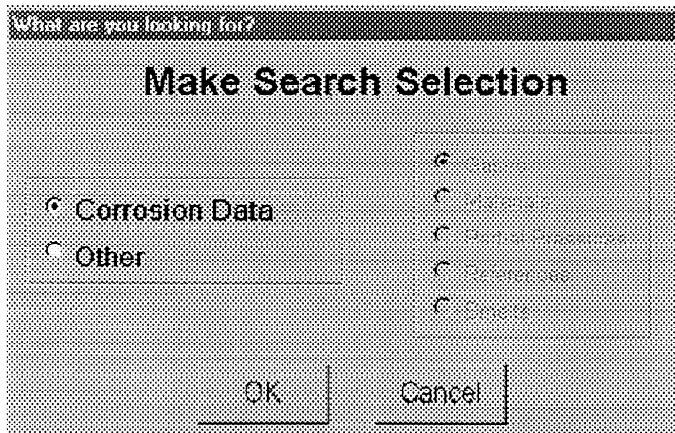


Figure 2 Second query form

In this case “Corrosion Data” is selected and then “OK” is clicked. This will open a third form, Figure 3, that allows the user to specify the environment.

On this form four options need to be chosen. For this example they are isothermal exposure, material by name, no smelt, and input gases. After these are selected, “OK” is clicked.

The form and associated subform buttons based on the selections on the Query selection form are shown in Figure 4. The two subforms associated with this form will pop up when the “Material” button and “Input Gases” button are clicked. The parameters required in the subforms must be entered before the “OK” button on the main form becomes active. The default values for temperature and time encompass the available range.

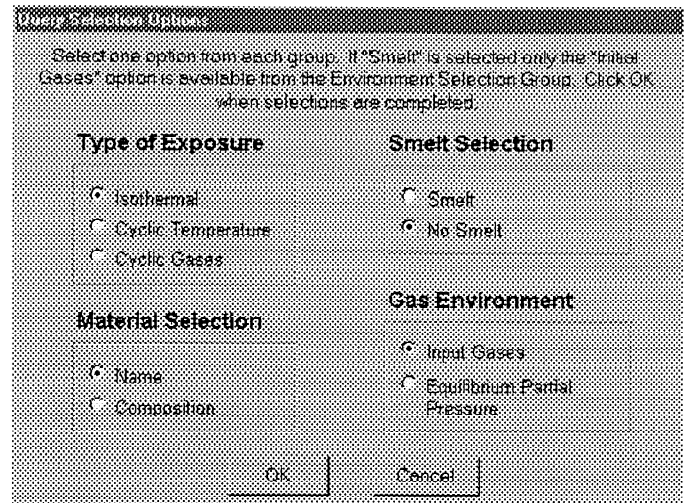


Figure 3 Third query form

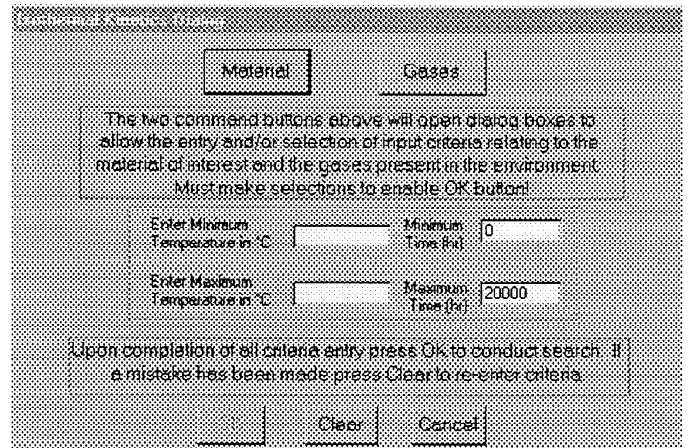


Figure 4 Fourth query form

On the Materials subform, the option needed for this query example was to allow a particular material to be chosen from a category of materials. Selecting this option updates the subform to show the available categories. At this point a decision has to be made—select “all materials” because both carbon steel and a stainless steel are needed or run two separate queries.

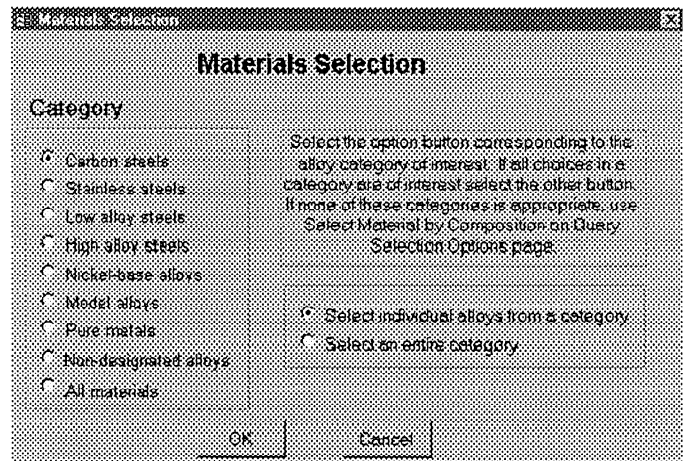


Figure 5 Materials subform

“All materials” was selected and then “OK” clicked. A pop-up form then appears that allows the appropriate material selections to be made. The subform seen after updating and the pop-up form are shown in Figures 5 and 6.

Material	Al	C	Cr	Cu	Fe	Mn	Mg
C1018	0.05	0.18	0.05	0.02	98.85	0.77	0.0
304	0	0.08	19	0	68.6	2	0
SA-210	0	0.21	0	0	98.9	0.65	0
SA-192	0	0.12	0	0	99.1	0.45	0
304L	0	0.03	19	0	67.9	2	0

Choose a single material by clicking on the desired row. Use Ctrl+Click, Shift+Click, or Ctrl+Shift+Click to make multiple selections that are non-adjacent ranges, or non-adjacent ranges, respectively.

OK Cancel

Figure 6 Materials pop-up

Clicking “OK” in the pop-up form returns the user to the main form. Clicking on the “Input Gases” button opens the associated subform shown in Figure 7.

Change maximum percentage default value >> 0% 100%

Minimum O ₂ %	<input type="text" value="0"/>	Minimum CO ₂ %	<input type="text" value="0"/>	Minimum CO%	<input type="text" value="0"/>
Maximum O ₂ %	<input type="text" value="100"/>	Maximum CO ₂ %	<input type="text" value="100"/>	Maximum CO%	<input type="text" value="100"/>
Minimum H ₂ S%	<input type="text" value="0"/>	Minimum CH ₃ SH%	<input type="text" value="0"/>	Minimum SO ₂ %	<input type="text" value="0"/>
Maximum H ₂ S%	<input type="text" value="100"/>	Maximum CH ₃ SH%	<input type="text" value="100"/>	Maximum SO ₂ %	<input type="text" value="100"/>
Minimum SO ₂ %	<input type="text" value="0"/>	Minimum H ₂ O ₂ %	<input type="text" value="0"/>	Minimum H ₂ %	<input type="text" value="0"/>
Maximum SO ₂ %	<input type="text" value="100"/>	Maximum H ₂ O ₂ %	<input type="text" value="100"/>	Maximum H ₂ %	<input type="text" value="100"/>
Minimum H ₂ %	<input type="text" value="0"/>	Minimum Cl ₂ %	<input type="text" value="0"/>	Minimum HCl%	<input type="text" value="0"/>
Maximum H ₂ %	<input type="text" value="100"/>	Maximum Cl ₂ %	<input type="text" value="100"/>	Maximum HCl%	<input type="text" value="100"/>

Enter the range, in percent, of the gases present in the environment.

OK Cancel

Figure 7 Input Gases subform

This subform allows the user to set the range of gases present. Inert carrier gases are not included, therefore the total percentage does not have to add to 100%. After the desired gas ranges, in this case 100% H₂S, has been entered, clicking on “OK” returns the user to the main form.

The “OK” button is now active. Values are now entered for the temperature and exposure time. Clicking on the “OK” button runs the query based on all the parameter values that have been entered.

After a short time either a message telling the user that no matches were found in the data or a form showing the first record that contains data appears. If the message box appears, the user can modify the input parameters and rerun the query. If matching records are located, they appear in a form similar to that shown in Figure 8. The total number of records retrieved and navigation buttons are located at the bottom of the form.

For this example the data were exported to a spreadsheet for sorting and to present a summary of the data as a graph.

Preview Print New Search Close

Minimum Corrosion Rate: mpy Maximum Corrosion Rate: mpy

There are some records which may not have corrosion rate data

Material: carbon steel

Temperature: °C Pressure: atm Exposure Time: hours

Input Gases:

Calculated Equilibrium Partial Pressures:

Weight Gain: mg/cm² Weight Loss: mg/cm²

Corrosion Rate: mpy

Rate Constant:

Date:

Title:

Author:

Journal/Book:

Page: year:

Records: of (1/1) of 74

Figure 8 Results form

As can be seen in Figure 9, the corrosion rates of both materials increase as the temperature increases, and the corrosion rate of carbon steel is greater than that of stainless steel. One could choose different materials to see their relative performance in this environment or one could do a query with a different gas environment with the same materials very easily.

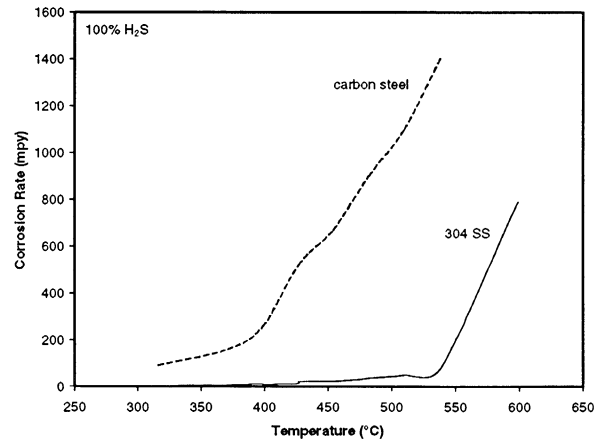


Figure 9 Corrosion rate versus temperature for carbon steel and 304 stainless steel in 100% H₂S [5-7]

Example 2 – Effect of changing S₂ pressure at constant temperature

This is a slightly more complicated database query. While it is no harder to fill out the query forms than in the previous example, a little more thought is required to decide how to limit the number of records retrieved. Once a search strategy has been adopted, everything proceeds as described in the previous section. Figure 10 shows the results for carbon steel at different temperatures. The main result that may be drawn from this plot is that while the corrosion rate of carbon steel is affected by the S₂ partial pressure, large changes are required. However, it is very clear from the figure that a change in temperature has much greater influence on the corrosion rate than does a change in S₂ partial pressure. Figure 11 shows the results of the same query in a narrower S₂ partial pressure range where this effect is emphasized.

Figure 11 shows the results of the same query in a narrower S_2 partial pressure range where this effect is emphasized.

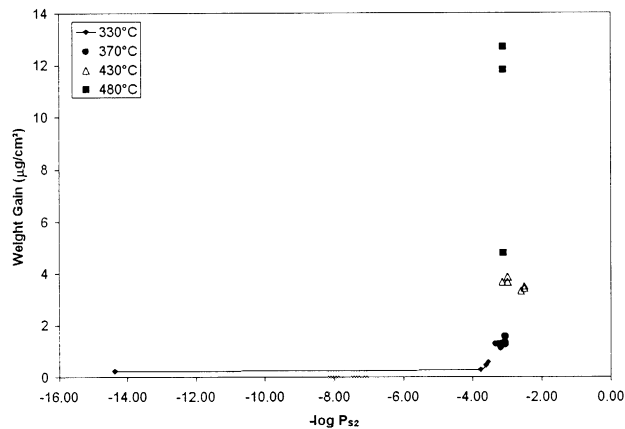


Figure 10 Weight gain versus log of S_2 partial pressure for carbon steel at various temperatures [4, 8]

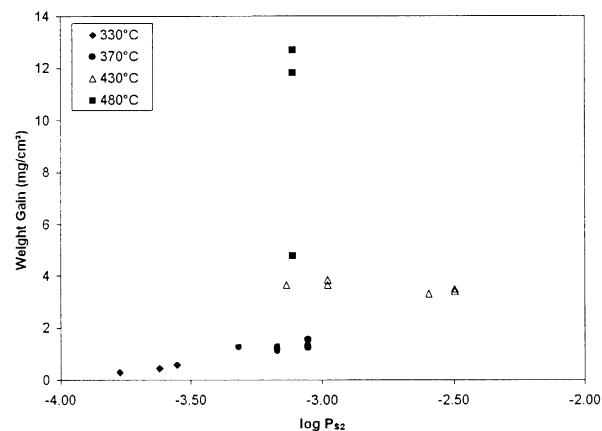


Figure 11 Weight gain versus log of S_2 partial pressure for carbon steel at various temperatures [4, 8]

CONCLUSIONS

A forms-based, user-friendly corrosion database for use by recovery boiler operators has been developed. This database contains information extracted from the published literature on the corrosion of many materials in environments of interest for recovery boiler operation. The database can be used as an aid in material selection for a particular environment or may be used to predict the behavior of a material in a variety of environments. The results of any query can be viewed on-screen and may be printed. The results of the queries can be exported to other databases or plotted in a graphical form defined by the user.

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