

UPCommons

Portal del coneixement obert de la UPC

http://upcommons.upc.edu/e-prints

Aquesta és una còpia de la versió *author's final draft* d'un article publicat a la revista [*Physical Review E*].

URL d'aquest document a UPCommons E-prints: http://hdl.handle.net/2117/90137

Paper publicar¹ / *Published paper*:

Robles-Hernández, B., Sebastián, N. ; Salud, J., Diez-Berart. S., [et. al.] (2016) Molecular dynamics of a binary mixture of twist-bend nematic liquid crystal dimers studied by dielectric spectroscopy. Physical Poview E 93 6 002700 1

by dielectric spectroscopy. Physical Review E, 93. 6. 002700-1-002700-7. Doi: 10.1103/PhysRevE.93.062705

¹ Substituir per la citació bibliogràfica corresponent

2

5

6

8

g

10

11

12

13

14

15

16

17

18

19

20

21

23

Molecular dynamics of a binary mixture of twist-bend nematic liquid crystal dimers studied by dielectric spectroscopy 3

Beatriz Robles-Hernández,¹ Nerea Sebastián,^{1,2} Josep Salud,³ Sergio Diez-Berart,³ David A. Dunmur,⁴

Geoffrey R. Luckhurst,⁵ David O. López,³ and M. Rosario de la Fuente^{1,*}

¹Departamento de Física Aplicada II, Facultad de Ciencia y Tecnología, Universidad del País Vasco, Apartado 644, E-48080 Bilbao, Spain

²Otto-von-Guericke Universitat Magdeburg, Institute for Experimental Physics, ANP, 39106 Magdeburg, Germany

³Grup de Propietas Físiques dels Materials (GRPFM), Departament de Física, E.T.S.E.I.B. Universitat Politècnica de Catalunya,

Diagonal 647, E-08028 Barcelona, Spain

⁴School of Physics and Astronomy, University of Manchester, Manchester M13 9PL, United Kingdom

⁵Chemistry, University of Southampton, Highfield, Southampton SO17 1BJ, United Kingdom

(Received 23 March 2016; published xxxxx)

We report a comprehensive dielectric characterization of a liquid crystalline binary mixture composed of the symmetric mesogenic dimer CB7CB and the nonsymmetric mesogenic dimer FFO9OCB. In addition to the high-temperature nematic phase, such a binary mixture shows a twist-bend nematic phase at room temperature which readily vitrifies on slow cooling. Changes in the conformational distribution of the dimers are reflected in the dielectric permittivity and successfully analyzed by means of an appropriate theoretical model. It is shown that the dielectric spectra of the mixture reflect the different molecular dipole properties of the components, resembling in the present case the characteristic dielectric spectra of nonsymmetric dimers. Comparison of the nematic and twist-bend nematic phases reveals that molecular dynamics are similar despite the difference in the molecular environment.

DOI: 10.1103/PhysRevE.00.002700 22

I. INTRODUCTION

Liquid crystal dimers consisting of two promesogenic 24 groups joined by a flexible methylene chain containing an 25 odd number of carbon atoms have been the focus of intense 26 research activity over the last few years [1-15]. It was found 27 that these compounds, of which 1",7"-bis(4-cyanobiphenyl-28 4'-yl) heptane, CB7CB, is the most studied, show a first 20 order nematic-nematic phase transition. The low-temperature 30 nematic phase is the so-called twist-bend nematic, N_{TB} . In 31 this phase, formed by achiral molecules, the director exhibits 32 periodic twist and bend deformations forming a conical helix 33 having doubly degenerate domains of opposite handedness. 34 Although initially described for dimers of the CBnCB family, 35 the discovery of a low temperature nematic phase boosted 36 research in the area, and a number of materials combining 37 different mesogenic units linked by chains have been identified 38 which exhibit the N_{TB} phase [16–18] and its chiral counterpart 39 [19,20]. It is now recognized that molecular curvature plays 40 crucial role in stabilizing the twist-bend nematic phase. а 41 For liquid crystal dimers the nature of the links between 42 the flexible spacer composed of methylene groups and the 43 mesogenic groups, together with the length of the spacer, play 44 a key role. For example, methylene as opposed to ether links 45 provides a better stabilization of the $N_{\rm TB}$ phase and shorter 46 rather than longer odd spacers also tend to stabilize the phase. 47 The low-energy gaps between the conformers generated by the 48 flexible spacer result in a distribution of energetically favored 49 molecular conformations, promoting a preferred bent shape 50 for molecules having linking chains with an odd number 51 carbon atoms. This also causes a drastic reduction of 52 the bend elastic constant in the conventional nematic phase 53

The presence of polar groups in the mesogenic units 70 of the dimers can be used to explore the evolution of the 71 conformational population distribution in nematic phases 72 through analysis of the dielectric properties of the material, 73 and so reveal details of the structure in the twist-bend nematic 74 phase. We have successfully employed such an approach to 75 interpret the static and dynamic dielectric behavior of various 76 nematic dimers, ranging from the symmetric dimers CB7CB or 77 CB9CB [1,6] to the highly nonsymmetric CBOnO.Py dimeric 78 family [28,29]. However, until now all dimers showing a 79 stable N_{TB} phase and studied by dielectric spectroscopy have ⁸⁰ had identical terminal groups, and so the main motivation of 81 this paper is to analyze the molecular orientational dynamics 82 in the N_{TB} phase when the dimers are nonsymmetric. We 83 have recently shown that 1''-(2',4-diffuorobiphenyl-4'-yloxy)- ⁸⁴ 9"-(4-cyanobiphenyl-4'-yloxy) nonane (FFO9OCB), an etherlinked nonsymmetric odd dimer, exhibits a monotropic N_{TB} ⁸⁶ phase [30], and that in binary mixtures with CB7CB there 87 is also a monotropic N_{TB} phase, which can be vitrified [31]. ⁸⁸

^[21] above the twist-bend nematic phase. According to the 54 Landau-like theory developed by Dozov [22], the transition 55 to the twist-bend nematic phase occurs when the bend elastic 56 constant vanishes. However, the ultimate physical mechanism 57 causing the director modulation remains controversial. A 58 flexoelectric coupling between bend deformation and electric 59 polarization [23,24] or purely elastic distortions of the director 60 due to molecular curvature [22] are considered to be possible 61 causes behind the formation of such twist-bend nematic 62 deformations. Indeed Monte Carlo simulations of rigid, apolar 63 bent molecules show that this system can still form the 64 twist-bend nematic phase and that the transition is driven by 65 molecular shape alone [25]. The latest theoretical approaches 66 point towards the primary role of the molecular shape through 67 steric interactions for the appearance of the $N-N_{\rm TB}$ transition 68 [26,27].

^{*}Corresponding author: rosario.delafuente@ehu.es

This phase, although monotropic, is stable enough to allow a 89 proper characterization. That is, the addition of a methylene-90 linked dimer (CB7CB) to an ether-linked dimer (FFO9OCB) 91 enhances the stability of the twist-bend nematic phase, as 92 expected from earlier detailed miscibility studies [31-33]. 93 Such behavior makes these mixtures the ideal candidates for 94 the focus of this study. The N_{TB} -N transition is always first 95 order, regardless of the concentration of the two mesogenic 96 compounds in the mixed systems, with a decreasing first-order 97 transition entropy on increasing the FFO9OCB concentration 98 [31]. 99

In this paper we present a complete dielectric study of a 100 mixture with a mole fraction of CB7CB equal to 0.48, having 101 significant proportion of the nonsymmetric component а 102 FFO9OCB, which conveniently shows a broad N_{TB} range that 103 on further cooling becomes a glassy state [N_{TB} glass (-6° C) 104 N_{TB} (77 °C) N (110 °C) I]. Its static dielectric behavior 105 and the frequency-dependent dielectric permittivity will be 106 107 thoroughly examined and discussed in relation to the behavior of pure components CB7CB and FFO9OCB. As will be shown, 108 the mixture presents a dielectric spectrum similar to that of 109 nonsymmetric dimers with three relaxation modes in both 110 nematic phases, which will be analyzed in the framework of 111 the molecular theory proposed by Stocchero *et al.* [34]. 112

113

II. RESULTS AND DISCUSSION

Both pure liquid crystal dimers CB7CB and FFO9OCB 114 were synthesized according to the procedures reported earlier 115 [30,35]. Binary mixtures were prepared in sealed aluminum 116 pans, heated to the isotropic phase, and ultrasonicated. The 117 quality of the mixing was assessed by means of high-resolution 118 calorimetry. More information regarding this preparation 119 procedure, calorimetric studies, and also measurements of 120 splay and bend elastic constants can be found in [31]. 121

Molecular calculations predict a distribution of molecular 122 shapes characterized by two broad peaks with maxima corre-123 sponding to an extended conformer (trans conformer) with an 124 average angle between the mesogenic units of around 120° and 125 hairpin-shaped conformer (cis conformer) with an internal 126 а verage angle around 30°. For high values of the nematic 127 a order parameter the most extended conformers are stabilized 128 the expense of the less extended conformers [1,21,34]. 129 at Figure 1 shows the chemical structures of the component 130 compounds together with a schematic representation of the 131 two main conformers for both dimers. The mean-square 132 dipole moment is given by the averaged vector sum of 133 the constituent dipole moments associated with the nitrile 134 groups for CB7CB (~4 D), and the two C-F bonds and 135 nitrile group for the case of FFO9OCB (~ 2.25 D). As can 136 be appreciated, extended orientationally averaged conformers 137 will contribute very differently to the permittivity component 138 along the director (ε_{\parallel}), with a zero mean-square dipole in 139 the case of CB7CB and a nonzero component in the case 140 of the nonsymmetric FFO9OCB. On the other hand, in both 141 cases the hairpin conformers make a large contribution to the 142 value of $\varepsilon_{||}$. Regardless of the configuration for both dimers 143 there will be a nonzero transverse dipolar contribution to the 144 perpendicular component of the permittivity (ε_{\perp}), which will 145 depend on the average angle between the terminal groups. For 146



FIG. 1. Chemical structures of CB7CB and FFO9OCB. Schematic representation of the two main conformers for both dimers.

the $N_{\rm TB}$ phase "parallel" and "perpendicular" refer to the helix 147 axis.

In Fig. 2 we present both components of the static dielectric 149 permittivity for three mixtures (CB7CB mole fractions: 0.21, 150 0.48, and 0.82) together with the corresponding values for the 151 two pure compounds. The measurements were taken using 152 an Agilent Precision LRC meter E4890A and Instec cells of 153 8 μ m thickness with antiparallel planar rubbing. Samples were held on a hot stage (TMSG-600) and temperature controller 155 (TMS-93) from Linkam. The perpendicular component ε_{\perp} 156 was obtained directly using harmonic probe fields of low 157



FIG. 2. Temperature dependence of the static permittivity of the two pure compounds and the three mixtures: (\Box) FFO9OCB; (•) $X_{CB7CB} = 0.21$; (Δ) $X_{CB7CB} = 0.48$; (\diamond) $X_{CB7CB} = 0.82$; (\bigstar) CB7CB. Discontinuous lines correspond to the phase transition N_{TB} -N.

amplitude, $0.5 V_{rms}$, well below the threshold voltage of the 158 Fréedericksz transition. On the other hand, the parallel com-159 ponent ε_{\parallel} was measured by applying voltages well above the 160 transition, which align the director parallel to the electric field. 161 We have chosen 5 kHz as most convenient for the aligning 162 field frequency [6]. As can be observed for all the samples, the 163 behavior at the I-N transition is as expected for materials with 164 positive dielectric anisotropy: ε_{\perp} decreases and ε_{\parallel} increases. 165 On further lowering the temperature, ε_{\perp} is gradually stabilized 166 in a very similar way, reflecting the increase of orientational 167 order and the subtle effect that changes in the conformational 168 distribution can exert on the averaged transverse component 169 of the molecular dipole moment. However, the behavior of 170 the parallel component strongly depends on the symmetric 171 nonsymmetric structure of the dimers. The aforementioned or 172 initial increase in the parallel permittivity on lowering the tem-173 perature at the I-N transition for CB7CB can only be explained 174 by the slight stabilization of hairpin conformers, which carry 175 large longitudinal dipole moment, as opposed to the very а 176 broad distribution of molecular conformers in the isotropic 177 phase [34]. After the initial increase, the parallel permittivity 178 rapidly decreases in the nematic phase on further reducing the 179 temperature. This behavior suggests a significant reduction of 180 the average mean-square molecular dipole moment as the ori-181 entational order of the phase increases. Molecular calculations 182 for CB7CB using a continuous torsional potential [1] for the 183 connecting methylene chain showed that the increase of orien-184 tational order produces a progressive increase of the population 185 of the more extended conformers with zero longitudinal dipole 186 moment at the expense of the hairpin-shaped ones, which is in 187 agreement with the observed temperature dependence of $\varepsilon_{||}$. 188 Such changes in the conformational distribution seem to be 189 common for liquid crystal dimers subject to a nematic potential 190 [34]. On entering the $N_{\rm TB}$ phase, the continuous decrease of 191 both components of the permittivity is accelerated, and at low 192 temperatures there is even a sign reversal in the dielectric 193 anisotropy from positive to negative. 194

For the nonsymmetric FFO9OCB dimer, the behavior is 195 slightly different. The parallel component also decreases as the 196 temperature is reduced but less drastically than for CB7CB; 197 this observation is related to the nonzero value of the average 198 dipole moment of the extended conformers [30]. This case is 199 intermediate between the two extremes: The CBnCB family 200 has two equal terminal groups, and no net longitudinal dipole 201 component for extended conformers, while in the CBOnO.Py 202 family [28,29] only one of the terminal groups has a significant 203 dipole moment. In the latter case, the static permittivity 204 similar to the monomers of the nCB family, showing a is 205 continuous increase as temperature decreases. Interestingly, 206 the static permittivity of the mixtures reveals the adequacy of 207 the model for the interpretation of the dielectric results. ε_{\parallel} 208 behaves in between the tendencies exhibited by both pure 209 compounds, smoothly approaching from one to the other 210 depending on the mole fraction. This is evidence that the 211 contribution of the net longitudinal dipole of the extended 212 nonsymmetric dimer hinders the drastic reduction of $\varepsilon_{||}$ seen 213 for CB7CB and implies a higher value of the dielectric 214 anisotropy. Also noteworthy is the behavior of ε_{\perp} at the N-N_{TB} 215 transition: While there is a decrease in the value for CB7CB, 216 that of the mixtures shows an appreciable increase which can 217

be explained in terms of the growth of the conical angle and ²¹⁸ the contribution of a net longitudinal dipole moment. ²¹⁹

The analysis of the frequency-dependent dielectric permittivity for the mixtures in the *N* and N_{TB} phases will give ²²¹ information on the molecular orientational dynamics. In order ²²² to cover the whole frequency range $10^{-2}-1.8 \times 10^9$ Hz and ²²³ properly characterize the dielectric spectra of the materials, ²²⁴ three different analyzers are needed: These are AlphaA ²²⁵ from Novocontrol, HP4192A, and HP4291A. High-frequency ²²⁶ dielectric measurements require the utilization of cells with ²²⁷ untreated metal electrodes, which in our setup consists of a ²²⁸ parallel plate capacitor made of two circular gold-plated brass ²²⁹ electrodes with a 5-mm diameter separated by 50- μ m-thick ²³⁰ silica spacers. ²³¹

For pure materials, the usual alignment achieved 232 spontaneously in these cells with gold electrodes is a random 233 planar alignment; i.e., the director is perpendicular to the 234 probe measuring electric field. For materials having a positive 235 dielectric anisotropy, it is possible to switch in the nematic 236 phase the alignment to homeotropic (director along the field 237 direction) with a suitable bias field. For the mixture studied, 238 although having a positive dielectric anisotropy, the director 239 alignment in the gold-electrode cells was not as described 240 above. Since it is not possible to view the sample in the dielec- 241 tric cell, assessment of the alignment was made by making the 242 appropriate capacitance measurements, and then fitting these 243 to the Havriliak-Negami function [see Eq. (3)]. From the fitting 244 it is possible to determine the extrapolated zero-frequency 245 permittivity, and compare this with values measured in 246 glass cells at low frequency, for which the alignment can 247 be checked visually. Thus, although the dielectric spectrum 248 obtained for the mixture in the gold-electrode cells exhibited 249 some distinctive features of the expected planar alignment, 250 it was not perfect. Similarly, on application of a bias field 251 the final aligned state was not a perfect homeotropic state. 252 We found that by gentle shearing, the degree of homeotropic 253 alignment could be increased, but even the best aligned 254 samples had an extrapolated value for the zero-frequency 255 parallel permittivity about 10% less than that expected from 256 separate low-frequency measurements. Thus we term the 257 state measured as "quasihomeotropic," but for the purposes of 258 analyzing the dynamics probed along the parallel axis of the 259 permittivity, the degree of alignment was satisfactory. 260

The sample was placed at the end of a coaxial line and 261 a modified HP16091A coaxial test fixture was used as the 262 sample holder and then held in a Novocontrol cryostat, which 263 screens the system. Dielectric measurements were performed 264 on cooling with different temperature steps being stabilized 265 to ± 20 mK. However, it should be noted that due to the time 266 required to perform low-frequency scans down to 10^{-2} Hz with 267 the AlphaA analyzer, only a few temperatures were measured 268 in this low-frequency range below $40 \,^{\circ}$ C, and will only be 269 used for a preliminary analysis of the glass transition dynamics 270 where they become relevant. 271

The frequency dependence of the real, $\varepsilon'(\omega)$, and imaginary, ²⁷² $\varepsilon''(\omega)$, parts of the quasihomeotropic dielectric permittivity ²⁷³ components are given in Fig. 3 for temperatures in (b) ²⁷⁴ the nematic, and (c) the twist-bend nematic phases together ²⁷⁵ with (a) a three-dimensional plot of the dielectric losses vs ²⁷⁶ temperature and frequency. Although in the isotropic phase ²⁷⁷



FIG. 3. Mixture with 0.48 mole fraction of CB7CB. (a) Threedimensional plot of the dielectric losses vs temperature and logarithm of the frequency. Frequency dependence of ε' (\Box) and ε'' (\circ) (b) in the *N* phase, and (c) in the *N*_{TB} phase. In (b,c) solid lines are fits to Eq. (3).

there is only one mode, in both nematic phases the dielectric 278 permittivity shows three relaxations, as in other nonsymmetric 279 dimers [11,28,29] and in particular as in FFO9OCB [30]. 280 The interpretation of such previous results required the 281 development of a suitable theoretical model that is worth 282 recalling here. Stocchero et al. [34] proposed a dynamic model 283 based on the time-scale separation between the motion of the 284 mesogenic units and the fast relaxation of the flexible chain. 285 The original model of Maier-Saupe [36] for the dielectric 286 relaxation of a dipolar molecule in a nematic potential was 287 two-state model representing the two orientations of a а 288 dipole along the director axis. In the case of the model of 289 Stocchero et al. the resulting potential presents four deep wells 290 corresponding to the four stable states of the dimer with the 291 terminal groups parallel or antiparallel to the director. Such a 292 potential provides the basis of the kinetic model for nematic 293

dimers which considers that molecular reorientation occurs 294 via a two-step process involving the individual orientational 295 relaxation of the mesogenic units with different rates: $k_i^{T \to C}$ 296 (rate coefficient for the reorientation of the rigid unit i from 297 the *trans* to the *cis* configuration) and $k_i^{C \to T}$ (inverse process). ²⁹⁸ Note that reorientations (end-over-end processes) of the con- 299 nected dipolar groups are considered to be individual, but not 300 independent, because of the correlation imposed by the flexible 301 spacer and accounted for by the conformational distribution 302 through the detailed balance equation $P_T^{eq} k_i^{T \to C} = P_C^{eq} k_i^{C \to T}$ 303 [34], where P_T^{eq} and P_C^{eq} are the equilibrium populations of the 304 trans and cis configurations, respectively. Whole molecular 305 reorientation is excluded, since the energy barrier for the 306 simultaneous rotation of both units is too high and results 307 in a nonsignificant transition rate [34]. As order increases, 308 calculation showed that the four wells become deeper but the 309 barriers are still much smaller for the individual flip-flops than 310 for the whole molecular reorientation. While the dynamics 311 are determined by the energy barriers between the states, the 312 dielectric strength of the relaxations are determined by the 313 relative equilibrium populations of the orientational states. 314

For symmetric dimers the flipping rates of both units are 315 equal and the resulting dipole moment correlation function 316

$$C_{||}(t) = 4\langle \mu_{||} \rangle^2 P_C^{eq} \exp(-2k^{C \to T} t),$$
(1)

has a monoexponential decay, and so only a single lowfrequency absorption, which, as has been shown for CB7CB and CB9CB [6,11], can be detected at frequencies corresponding to those of the intermediate frequency relaxation observed here and whose strength depends on the conformational distribution. On the other hand, for nonsymmetric dimers, serelaxation rates would be different, and, depending on the shape and dipole moments of each unit, a variety of frequency relaxation profiles is predicted. In the case of FFO9OCB and assuming a relationship between dipole moments of both units given by $\langle \mu_{||} \rangle_{\text{FF}} \approx 1/2 \langle \mu \rangle_{||CN}$, the model leads to the correlation function 328

$$C_{||}(t) = \frac{1}{4} \langle \mu_{||} \rangle_{CN}^{2} \left(P_{T}^{eq} - 3P_{C}^{eq} \right)^{2} \exp(-2k_{B}t)$$

+ 4\langle \mu_{||} \langle_{CN}^{2} P_{C}^{eq} P_{T}^{eq} \exp(-2k_{A}t), (2)

where $k_A = \frac{1}{2}(k_A^{T \to C} + k_A^{C \to T})$ (A stands for the unit con-taining the CN group) and $k_B = 2(P_C^{eq}k_B^{C \to T}) = 2(P_T^{eq}k_B^{T \to C})$ 330 (B stands for the unit containing the FF group). Equation 331 (2) entails the emergence of an additional lower-frequency 332 relaxation mode. Both low-frequency processes are related to 333 the individual flip-flop motions of the units, and dielectric 334 strengths are determined by equilibrium population their of 335 the conformers [30,34]. Experimental results for FFO9OCB 336 clearly confirm [30] such model's prediction. Accordingly, 337 we associate the three dielectric relaxations observed for the 338 mixture to (i) a superposition of the precessional motion 339 of the dipolar groups around the director and the rotation 340 around the long axis at the high-frequency branch of the 341 spectrum (m_2) and (ii) the end-over-end reorientation of the $_{342}$ dipolar groups parallel to the director at low frequencies (m_{1L} 343 and m_{1H} [34]. It is evident that the mixture shows a behavior 344 resembling that of nonsymmetric dimers, with the appearance 345 of a third dielectric mode at low frequencies, that in this case 346 results from the fluorinated group reorientations. 347

The dielectric strengths of m_{1L} and m_{1H} for nonsymmetric 348 dimers are predicted to depend on the conformational pop-349 ulation distribution and on the relative value of the dipole 350 moments of both terminal groups. Although the precise 351 temperature dependence is intricate, the usual trend [11,28,29] 352 that the strength of the lower-frequency mode (m_{1L}) is 353 increases while that of m_{1H} decreases when the temperature is 354 reduced as found for FFO9OCB [30]. Despite the similarity of 355 the dielectric spectra of the mixture with that of FFO9OCB, the 356 contribution of the symmetric CB7CB is clear when comparing 357 the much higher strength of the m_{1H} mode with respect to that 358 pure FFO9OCB, as this mode can be associated with the of 359 reorientation of the nitrile groups of both dimers. In order to 360 describe accurately the behavior of the mixture, the spectra for 361 each temperature have been analyzed by fitting each relaxation 362 mode according to the Havriliak-Negami function through the 363 empirical relationship 364

$$\varepsilon(\omega) = \varepsilon'(\omega) - i\varepsilon''(\omega)$$

= $\sum_{k} \frac{\Delta \varepsilon_{k}}{\left[1 + (i\omega\tau_{k})^{\alpha_{k}}\right]^{\beta_{k}}} - \frac{i\sigma_{dc}}{\omega\varepsilon_{0}} + \varepsilon_{\infty},$ (3)

where ε_{∞} is the extrapolated high-frequency permittivity, $\Delta \varepsilon_k$ 365 is the strength of the corresponding relaxation mode, and $\sigma_{\rm dc}$ is the dc conductivity. The relaxation time τ_k is related 367 to the frequency of maximum loss through the parameters 368 α_k and β_k , which describe the width and the asymmetry of 369 the relaxation spectra, respectively ($\alpha = \beta = 1$ corresponds 370 a simple Debye-like process). The two low-frequency 371 to modes have a near-Debye shape in both mesophases and 372 the high-frequency one is broader and asymmetric (having 373 Cole-Davidson shape with α about 0.8 and β ranging from а 374 0.7 to 0.6). Dielectric strengths for each relaxation are given 375 as a function of temperature in Fig. 4. The strength of the 376 high-frequency mode m_2 in the N phase shows a decrease as 377 the temperature decreases. On the other hand, the strengths of 378 the two low-frequency modes, m_{1L} and m_{1H} , are correlated: 379 while that of m_{1H} decreases, the strength of m_{1L} increases as 380 temperature decreases in the N phase. This is the same trend 381 observed for the corresponding modes for FFO9OCB [30]. 382 The strengths of the three modes exhibit small jumps at the 383 $N-N_{\rm TB}$ phase transition, but a proper quantitative analysis is 384



FIG. 4. Mixture with 0.48 mole fraction of CB7CB. Dielectric strength of the relaxation modes vs temperature: (•) isotropic phase; (Δ) m_2 ; (\Box) m_{1H} ; (\Diamond) m_{1L} .



FIG. 5. Mixture with 0.48 mole fraction of CB7CB. Arrhenius plot of the frequency of the relaxation modes: (•) isotropic phase; (Δ) m_2 ; (\Box) m_{1H} ; (\Diamond) m_{1L} . Solid lines are fits to the VFT law, Eq. (4).

not possible because the achieved alignment is not perfect. In 385 fact we think that in a perfect parallel alignment $\Delta \varepsilon_2$ should 386 be slightly smaller, taking into account the value obtained 387 for FFO9OCB [30], and the strengths of the low-frequency 388 modes slightly higher; the jumps at the transition downwards 389 are provoked by the appearance of the tilt. What is interesting 390 is the general trend: $\Delta \varepsilon_{1H}$ decreases while $\Delta \varepsilon_{1L}$ tends to 391 saturate. It is important to recall that, even if the alignment were 392 perfect, the precise analysis of the temperature dependence 393 would be complicated since it involves both the temperature 394 evolution of the population of the different conformers and 395 the relative value of the dipoles in the terminal mesogenic 396 units [28–30,34]. The frequency dependence of the modes can 397 be determined over a broad temperature range, and as can be 398 observed in Fig. 5 its temperature dependence in the N phase is $_{399}$ similar to that found for FFO9OCB. Remarkably, at the $N-N_{\rm TB}$ 400 phase transition the frequencies remain almost unaltered, 401 compatible with a weakly first order phase transition. The 402 two low-frequency modes follow the Arrhenius law in the 403 N phase with activation energies of 85 kJ mol^{-1} (m_{1L}) and $_{404}$ 70 kJ mol⁻¹ (m_{1H}). At the higher temperatures in the N_{TB} 405 phase, the low-frequency modes also follow the Arrhenius 406 law with activation energies slightly larger, namely, 115 and 407 91 kJ mol⁻¹, respectively. However, in line with the glass 408 transition detected by calorimetry at $-6^{\circ}C$ [31], on reducing 409 the temperature further there is a slowing down of the dynamics 410 as the glassy state is approached, strongly deviating from the 411 Arrhenius behavior. For this temperature range, Fig. 5 shows, 412 superimposed, the corresponding fitting to the VFT (Vogel- 413 Fulcher-Tammann) temperature-dependent relationship, 414

$$f_k = \frac{\omega_k}{2\pi} = f_{k0} \exp\left[\frac{B}{T - T_0}\right],\tag{4}$$

where f_{k0} , B, and T_0 are fitting parameters. The first important ⁴¹⁵ observation is that the m_{1H} mode seems to merge with m_2 , ⁴¹⁶ in the same way as happens with the analogous m_1 and ⁴¹⁷ m_2 for CB7CB on approaching the glassy N_{TB} phase [2]. ⁴¹⁸ However, at least in our frequency-temperature window, m_{1L} ⁴¹⁹ evolves separately. For dielectric relaxation, a glass transition ⁴²⁰ (known as a dynamic glass transition) is obtained when the ⁴²¹ characteristic relaxation frequency reaches 10^{-2} Hz. In Fig. 5 ⁴²²

we can see how this value corresponds to the same temperature 423 for m_2 and m_{1H} (around -10° C) while it is reached at a 424 slightly higher temperature for m_{1L} (around -6° C). This 425 implies that, from a dielectric point of view, two glass 426 transition temperatures involving different molecular motions 427 are present; the lower-temperature glass transition due to the 428 dynamic disorder of the cyanobiphenyl group and that at higher 429 temperatures attributed to the fluorinated group reorientations. 430 In the latter case, the nematic environment at these low 431 temperatures makes the steric interactions of each group 432 much stronger, hindering the reorientations and increasing 433 the required thermal energy. Accordingly, the associated glass 434 transition temperature is higher than for the cyanobiphenyl 435 group [37]. However, the analysis of the dynamics at this 436 double glass transition is beyond the scope of this paper and 437 needs to be confirmed by other experimental techniques and 438 thoroughly examined with more detailed models [2]. 439

III. SUMMARY

440

In our opinion the most interesting aspect of our results 441 is that we have been able to study the dielectric spectra for 442 a nonsymmetric dimer exhibiting the N_{TB} phase. We demon-443 strate that in a mixture containing a nonsymmetric element 444 445 with an adequate molecular dipole the spectra resemble those of the nonsymmetric component by exhibiting three relaxation 446 processes. The spectra can be satisfactorily explained in terms 447 of the theoretical model for dielectric relaxation in liquid 448 crystal dimers proposed by Stocchero et al. [34]. Recently, 449 a study of the dielectric spectra of a six-component mixture, 450 denoted KA(0.2) (three symmetric and three nonsymmetric 451 dimers) in the N and N_{TB} phases was reported, attributing m_{1L} 452

- M. Cestari, S. Diez-Berart, D. A. Dunmur, A. Ferrarini, M. R. de la Fuente, D. J. B. Jackson, D. O. López, G. R. Luckhurst, M. A. Pérez-Jubindo, R. M. Richardson, J. Salud, B. A. Timimi, and H. Zimmermann, Phys. Rev. E 84, 031704 (2011).
- [2] D. O. López, N. Sebastian, M. R. de la Fuente, J. C. Martínez-García, J. Salud, M. A. Pérez-Jubindo, S. Diez-Berart, D. A. Dunmur, and G. R. Luckhurst, J. Chem. Phys. 137, 034502 (2012).
- [3] D. Cheng, J. H. Porada, J. B. Hooper, A. Klittnick, Y. Shen, M. R. Tuchband, E. Korblova, D. Bedrov, D. M. Walba, M. A. Glaser, J. E. MacClennan, and N. A. Clark, Proc. Natl. Acad. Sci. USA 110, 15931 (2013).
- [4] V. Borsch, Y. K. Kim, J. Xiang, M. Gao, A. Jákli, V. P. Panov, J. K. Vij, C. T. Imrie, M. G. Tamba, G. H. Mehl, and O. D. Lavrentovich, Nat. Commun. 4, 2635 (2013).
- [5] V. P. Panov, R. Balachandran, J. K. Vij, M. G. Tamba, A. Kohlmeier, and G. H. Mehl, Appl. Phys. Lett. 101, 234106 (2012).
- [6] B. Robles-Hernández, N. Sebastián, M. R. de la Fuente, D. O. López, S. Diez-Berart, J. Salud, M. B. Ros, D. A. Dunmur, G. R. Luckhurst, and B. A. Timimi, Phys. Rev. E 92, 062505 (2015).
- [7] V. P. Panov, M. Nagaraj, J. K. Vij, Y. P. Panarin, A. Kohlmeier, M. G. Tamba, R. A. Lewis, and G. H. Mehl, Phys. Rev. Lett. 105, 167801 (2010).
- [8] P. A. Henderson and C. T. Imrie, Liq. Cryst. 38, 1407 (2011).

to whole molecule end-over-end rotation of cybotactic groups 453 [38]. Interestingly, although x-ray studies in symmetric dimers 454 such as CB9CB [33] also suggest the presence of domains, our 455 dielectric results of this material [6] show a single relaxation 456 process at frequencies similar to those of m_{1H} as predicted 457 by Stocchero's model. In light of our previous studies and 458 those presented here, we believe that the low-frequency mode 459 described for the KA(0.2) mixture could be attributed to the 460 contribution of the individual reorientation of the mesogenic 461 units of the nonsymmetric compounds constituting the sample. 462 Remarkably, in our mixture we could follow the dielectric 463 modes over a very broad frequency-temperature range both 464 in the N and $N_{\rm TB}$ phases. From a dielectric point of view, 465 both nematic phases exhibit the same molecular active modes, 466 with a similar behavior (only small changes in the strength, 467 related to the director tilt, and in the activation energies) and 468 the molecular environment, nematic or twist-bend nematic, 469 scarcely affects the dielectric properties. However, as the mix- 470 ture shows a glass transition, when reducing the temperature 471 the molecular modes seem to become increasingly collective 472 as already described for CB7CB [2]. 473

ACKNOWLEDGMENTS

474

The authors are grateful for financial support from the 475 MICINN Project No. MAT2015-66208-C3-2-P, and from the 476 Eusko Jaurlaritza-Gobierno Vasco Project No. GI/IT-449-10. 477 The authors also acknowledge the recognition from the 478 Generalitat de Catalunya of GRPFM as Emergent Research 479 Group (2009-SGR-1243). N.S. thanks the Alexander von 480 Humboldt Foundation for support from a Postdoctoral Research Fellowship. 482

- [9] C. S. P. Tripathi, P. Losada-Pérez, C. Glorieux, A. Kohlmeier, M. G. Tamba, G. H. Mehl, and J. Leys, Phys. Rev. E 84, 041707 (2011).
- [10] M. Sepelj, A. Lesac, S. Baumeister, S. Diele, H. L. Nguyen, and D. W. Bruce, J. Mater. Chem. **17**, 1154 (2007).
- [11] D. A. Dunmur, G. R. Luckhurst, M. R. de la Fuente, S. Diez, and M. A. Peréz-Jubindo, J. Chem. Phys. 115, 8681 (2001).
- [12] C. Meyer, G. R. Luckhurst, and I. Dozov, Phys. Rev. Lett. 111, 067801 (2013).
- [13] C. Meyer, G. R. Luckhurst, and I. Dozov, J. Mater. Chem. C 3, 318 (2015).
- [14] D. A. Dunmur, M. R. de la Fuente, M. A. Perez-Jubindo, and S. Diez, Liq. Cryst. 37, 723 (2010).
- [15] D. A. Dunmur, M. Cestari, S. Diez-Berart, A. Ferrarini, M. R. de la Fuente, D. J. B. Jackson, D. O. López, G. R. Luckhurst, M. A. Pérez-Jubindo, R. M. Richardson, J. Salud, B. A. Timimi, and H. Zimmermann, *Presented at the 23rd International Liquid Crystal Conference, Kraków, Poland, July 11–16, 2010*, Oral 60 (unpublished).
- [16] R. J. Mandle, E. J. Davis, C. C. A. Voll, C. T. Archbold, J. W. Goodby, and S. J. Cowling, Liq. Cryst. 42, 688 (2015).
- [17] R. J. Mandle, E. J. Davis, C. T. Archbold, C. C. A. Voll, J. L. Andrews, S. J. Cowling, and J. W. Goodby, Chemistry 21, 8158 (2015).
- [18] Z. Ahmed, C. Welch, and G. H. Mehl, RSC Adv. 5, 93513 (2015).

- [19] E. Gorecka, N. Vaupotič, A. Zep, D. Pociecha, J. Yoshioka, J. Yamamoto, and H. Takezoe, Angew. Chem., Int. Ed. 54, 10155 (2015).
- [20] A. Zep, S. Aya, K. Aihara, K. Ema, D. Pociecha, K. Madrak, P. Bernatowicz, and H. Takezoe, J. Mater. Chem. C 1, 46 (2013).
- [21] M. Cestari, E. Frezza, A. Ferrarini, and G. R. Luckhurst, J. Mater. Chem. 21, 12303 (2011).
- [22] I. Dozov, Europhys. Lett. 56, 247 (2001).
- [23] R. B. Meyer, in *Molecular Fluids*, edited by R. Balian and G. Weill (Gordon and Breach, New York, 1976).
- [24] S. M. Shamid, S. Dhakal, and J. V. Selinger, Phys. Rev. E 87, 052503 (2013).
- [25] R. Memmer, Liq. Cryst. 29, 483 (2002).
- [26] C. Greco and A. Ferrarini, Phys. Rev. Lett. 115, 147801 (2015).
- [27] N. Vaupotič, S. Curk, M. A. Osipov, M. Čepič, H. Takezoe, and E. Gorecka, Phys. Rev. E 93, 022704 (2016).
- [28] N. Sebastián, M. R. de la Fuente, D. O. López, M. A. Pérez-Jubindo, J. Salud, S. Diez-Berart, and M. B. Ros, J. Phys. Chem. B 115, 9766 (2011).
- [29] N. Sebastián, M. R. de la Fuente, D. O. López, M. A. Pérez-Jubindo, J. Salud, and M. B. Ros, J. Phys. Chem. B 117, 14486 (2013).

- [30] N. Sebastián, D. O. López, B. Robles-Hernández, M. R. de la Fuente, J. Salud, M. A. Pérez-Jubindo, D. A. Dunmur, G. R. Luckhurst, and D. J. B. Jackson, Phys. Chem. Chem. Phys. 16, 21391 (2014).
- [31] D. O. López, B. Robles-Hernández, J. Salud, M. R. de la Fuente, N. Sebastián, S. Diez-Berart, X. Jaen, D. A. Dunmur, and G. R. Luckhurst, Phys. Chem. Chem. Phys. 18, 4394 (2016).
- [32] K. Adlem, M. Čopič, G. R. Luckhurst, A. Martelj, O. Parri, R. M. Richardson, B. D. Snow, B. A. Timimi, R. P. Tuffin, and D. Wilkies, Phys. Rev. E 88, 022503 (2013).
- [33] E. Ramou, Z. Ahmed, C. Welch, P. K. Karahaliou, and G. H. Mehl, Soft Matter 12, 888 (2016).
- [34] M. Stocchero, A. Ferrarini, G. J. Moro, D. A. Dunmur, and G. R. Luckhurst, J. Chem. Phys. 121, 8079 (2004).
- [35] P. J. Barnes, A. G. Douglass, S. K. Heeks, and G. R. Luckhurst, Liq. Cryst. 13, 603 (1993).
- [36] G. Meier and A. Saupe, Mol. Cryst. 1, 515 (1966).
- [37] S. Diez-Berart, D. O. López, J. Salud, J. A. Diego, J. Sellarés, B. Robles-Hernández, M. R. de la Fuente, and M. B. Ros, Materials 8, 3334 (2015).
- [38] R. R. Ribeiro de Almeida, C. Zhang, O. Parri, S. N. Sprunt, and A. Jákli, Liq. Cryst. 41, 1661 (2014).