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5	PURIFICATION OF FREE ARGININE FROM CHICKPEA (CICER ARIETINUM) SEEDS.
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- 28 Abstract.

Chickpea is a grain legume widely consumed in the Mediterranean Region and other parts of the world. Chickpea seeds are rich in proteins but they also contain a substantial amount of free amino acids, especially arginine. Hence chickpea may represent a useful source of free amino acids for nutritional or pharmaceutical purposes. Arginine is receiving great attention in recent years because it is the substrate for the synthesis of nitric oxide, an important signalling molecule involved in numerous physiological and pathological processes in mammals. In this work we describe a simple procedure for the purification of arginine from chickpea seeds using nanofiltration technology and an ion exchange resin, Amberlite IR-120. Arginine was finally purified through precipitation or cristalization yielding preparations with purities of 91 and 100%, respectively. Chickpea may represent an affordable green source of arginine, and a useful alternative to production through fermentation or protein hydrolysis.

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**Key words**: Chickpea, seed flour, free amino acids, arginine.

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**Running title**: Purification of free arginine from chickpea.

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### 1. Introduction.

53 Chickpea (Cicer arietinum L.) is a staple food in large areas of the world, 54 including the Mediterranean Region, the Middle East, India, and South and Central 55 America. It is a grain legume rich in proteins and carbohydrates (Sánchez-Vioque, 56 Clemente, Vioque, Bautista, & Millán, 1999), and also in secondary components such 57 as polyphenols, alkaloids and free amino acids. Free amino acids are among the most 58 abundant secondary compounds in some legume seeds. For example, they are 3.5 59 times more abundant than polyphenols in lentil seeds (Lens culinaris) (unpublished 60 results). Free amino acids may function in the seeds as phytoalexins, storage of 61 nitrogen, and in metabolic signalling (Bell, 2003). Their abundance in legume seeds 62 may be influenced by environmental factors such as temperature, hydric stress, salt 63 stress, availability of nitrogen, and light (Reggiani, Cantu, & Brambilla, 1988; Wallace, 64 Secor, & Schrader, 1984; Roosens, Thu, Iskandar, & Jacobs, 1998; Colling, Stander, & 65 Makunga, 2010).

Free amino acids in legume seeds include both proteic and nonproteic amino acids (Bell, 2003). Chickpea seeds are particularly rich in free arginine (unpublished results). This amino acid is considered semi essential or conditionally essential because, although animals can synthesize it, exclusion of arginine from the diet results in a sub-optimal weight gain (Tapiero, Mathé, Couvreur, & Tew, 2002). In animals, arginine is an intermediate in the urea cycle and is also the substrate for the synthesis of many nitrogen-containing compounds including nitric oxide, ornithine and

73 polyamines (Morris, 2006). Arginine is directly implicated in ATP synthesis, cellular 74 proliferation, vasodilatation, neurotransmission, calcium release, and immunity 75 (Nieves, & Langkamp-Henken, 2002). In particular, the role of arginine in the synthesis 76 of nitric oxide has stimulated research in this amino acid because nitric oxide is 77 involved in many physiological and pathological processes. The therapeutic properties 78 of arginine in the treatment of diseases related with shortage of NO, including 79 nervous, muscular, circulatory, respiratory, digestive, urinary, reproductive, endocrine, 80 and immune systems have been studied as reported by Ignarro (1989). Enteral and 81 parenteral administration of arginine decreases the probabilities of suffering 82 cardiovascular diseases (Flynn, Meininger, Haynes, & Wu, 2002), lower glucose 83 concentration in blood, and improves reproductive, pulmonary, renal, gastrointestinal, 84 hepatic, and immune functions (Tapiero, Mathé, Couvreur, & Tew, 2002)

Although free arginine, as most amino acids, is industrially produced by fermentation using microorganisms (Utagawa, 2004), plants rich in free arginine such as chickpea seeds may represent a very afordable "green alternative". This is especially true for inexpensive, readily available sources of material, such as chickpeas that are damaged during harvesting and processing. These chickpeas, representing about 20 % of chickpea production, are considered as a by-product and are sold at low prices for feeding livestock, (Ulloa, Valencia, & Garcia, 1988).

92 The objective of this work was to design an improved purification method to 93 efficiently produce arginine rich extracts that could be used as ingredients for foods, 94 nutritional supplements, and pharmaceuticals.

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- 1. Material and methods.
- 101 1.1. Plant material.

102 Chickpea (*Cicer arietinum*), soy (*Glycine max*), broad bean (*Vicia faba*), lentil 103 (*Lens culinaris*) and common bean (*Phaseolus vulgaris*) seeds were purchased in a local 104 market.

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# 2.2. Extraction of free amino acids.

Seeds were milled using a domestic blender. Flour was suspended in water and taken to pH 4.3 with 1 N HCl for extraction under continuous stirring for 1 hour. Extracts were centrifuged at 15.000 g for 20 min, and pellets were extracted two more times as described above. The three resulting supernatants were pooled and used for the purification of arginine.

111 **2.3.** Purification of arginine.

112 Aliquots of the extracts prepared as described above were concentrated to half 113 of their volume using an Amicon cell filtration unit (Millipore, MA, USA) equipped with 114 a 200 Da nanofiltration membrane (Koch Membrane Systems, MA, USA). The 115 permeate, containing free amino acids and other low molecular weight molecules, was 116 incubated with Amberlite IR-120 (Fluka, MO, USA) to further purify arginine. Amberlite 117 IR-120 is a strong cationic resin that binds arginine at acidic pH. The resin was pre-118 conditioned by treatment with ten volumes of HCl 2 N for two hours, ten volumes of 119 NaOH 2 N for two hours, ten volumes of HCl 2 N for two additional hours, and it was 120 finally washed four times with water. The free amino acid permeate (1 mL) was

121 incubated with 1 mL resin, in 10 mL water taken to pH 2, under continuous stirring at 122 room temperature for 30 min, which led to binding of all free amino acids. The resin 123 was then washed four times for 15 min with ten volumes of pH 2 water. Acidic amino 124 acids were then released by washing with 10 volumes NH<sub>4</sub>OH 0.5 N for 30 min, and 125 finally arginine was recovered by washing three times with 10 volumes NH<sub>4</sub>OH 7.5 N 126 for 30 min. The combined washes were taken to dryness in a speed vacuum and 127 redissolved in the minimum volume of water possible. Solid matter determinations 128 were carried out after drying aliquots at 120° C overnight.

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### 2.4. Precipitation and crystallization.

Precipitation was carried out by addition of excess ethanol, at least four times the volume of extract, and the precipitate was recovered by centrifugation at 15.000 g for 15 min. Crystallization was carried out by addition of the same volume of ethanol and letting the resulting mixture rest for 24 hours at 4° C. Crystals were recovered by centrifugation at 15.000 g for 15 min.

#### 135 2.5. Amino acids analysis.

Amino acids were analyzed by RP-HPLC after derivatization with diethyl ethoxymethylenemanolate, and determined according to the method described by Alaiz, Navarro, Giron, & Vioque (1992), using D, L  $\alpha$ -aminobutyric acid as internal standard and a Novapack C<sub>18</sub> column (300 x 3.9 mm i.d., 4  $\mu$ m, Waters). Electro-sprayionization high-resolution mass spectra were recorded with a micrOTOF-QII High Resolution-of-Flight mass spectrometer (UHR-TOF) with qQ-TOF geometry (Bruker Daltonic, Bremen, Germany).

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2.6. Soluble sugars determination.

Soluble sugars were measured according to Dubois, Gilles, Hamilton, Reber, &
Smith (1956), using a standard curve of glucose.

146 2.7. FPLC gel filtration chromatography.

Samples were analyzed by gel filtration chromatography using a Superdexpeptide column coupled to a FPLC AKTA-purifier system. The eluent used was 0.75 M ammonium bicarbonate at a flow rate of 0.5 mL/min. Elution was monitored at 214 nm and the molecular masses of eluted compounds were determined by comparison with the following molecular weights standards from Pharmacia: blue dextran (2000 kDa), cytochrome C (12500 Da), aprotinin (6512 Da), bacitracin (1450 Da), cytidine (246 Da) and glycine (75 Da).

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### 2. Results and discussion.

156 Many legumes store free amino acids in their seeds. As an example, figure 1 157 shows total free amino acids and free arginine contents in the seeds of five of the most 158 popular grain legumes, which accumulate in particular large amounts of arginine. This 159 is especially the case of chickpea, with arginine representing 53 % of total free amino 160 acids. This is in contrast to total arginine content in chickpea, which represents 10 % of 161 all proteic and non-proteic amino acids. Thus, chickpea could be a good source of 162 arginine provided that an affordable purification protocol could be implemented to 163 purify arginine from the free amino acids pool. Other free amino acids at substantial 164 levels in chickpea seeds are glutamic acid, aspartic acid, leucine and tryptophan (Table 165 1). Extraction of free amino acids was carried out at pH 4.3 in order to minimize 166 protein solubilization, since this pH corresponds to the isolectric point of storage 167 proteins in chickpea seeds (Sanchez-Vioque et al., 1999). Figure 2A shows the FPLC gel

filtration profile of these extracts. Although globulins are insoluble at this pH, other components including albumins, polyphenols and sugars are solubilized in addition to free amino acids. Similarly, during the process of production of chickpea protein concentrates and isolates, the aqueous fraction generated after isolectric precipitation of proteins at pH 4.3 (Sanchez-Vioque et al., 1999) may be used for the production of an arginine rich fraction.

174 Nanofiltration using a 200 Da membrane allows for separation of free amino 175 acids from higher molecular weight components such as proteins, polysaccharides, 176 polyphenols and fibre. Figures 2B and 2C show the FPLC gel filtration profile of the 177 nanofiltration retentate and permeate, respectively.

178 Ion exchange resins are frequently used in the purification of amino acids in 179 industrial settings (Leuchtenberger, Hutmacher, & Drauz, 2005). Strong cationic resins 180 are especially useful in the purification of amino acids with basic R-functional groups 181 such as the guanidinium group in arginine (Utagawa, 2004). These resins have been 182 used in the past for the purification of free amino acids from legume seeds, including 183 homoarginine from Lathyrus sativus (Rao, Ramachandran, & Adiga, 1963) and L. cicera 184 (Bell, 1962), and canavanine from Canavalia ensiformis (Bass, Harper, Rosenthal, 185 Phuket, & Crooks, 1995). Specifically Amberlite IR-120 resin has been used for the 186 determination of arginine in grape juice (Li, Liang, Feng, Liu, & Wang, 2008). Figure 3 shows the total mass, soluble sugars and free amino acids contents in a representative 187 188 experiment of purification arginine from chickpea seeds using Amberlite IR-120 resin. 189 Free amino acids and soluble sugars represented 2.6 % and 27.8 % of total mass in the 190 permeate from nanofiltration, respectively (Figure 3 first group of bars). All free amino 191 acids were bound to the resin after incubation for 30 min, while 58% of total mass and

192 80% of soluble sugars remained in the soluble phase (Figure 3, second group of bars). 193 Washes using pH 2 water allowed for removal of the remaining sugars and unidentified 194 components bound to the resin (Figure 3, third group of bars) without any losses of 195 bound amino acids (Table 3). The resin was also washed using 0.5 N NH<sub>4</sub>OH in order to 196 remove poorly bound acidic amino acids (Figure 3, fourth group of bars). Aspartic and 197 glutamic acid accounted for 85% of the eluted amino acids (Table 2). Finally, arginine 198 was recovered from the resin by washing three times with 7.5 N NH<sub>4</sub>OH (Figure 3, fifth 199 group of bars, Tables 2 and 3).

200 The final washes were pooled and taken to dryness to yield a brownish syrup 201 that was used for precipitation or crystallization in order to further purify arginine. 202 Precipitation yielded a white pellet containing 90.7 % arginine (Table 2). The exact 203 mass of this precipitate was 175.1191 ( $M^+$ ) similar to 175.1190 ( $M^+$ ) of the pure 204 arginine. Crystallization yielded white crystals containing pure arginine (Table 2). The 205 exact mass of this pellet was 175.1190 (M<sup>+</sup>) identical to the theoretic mass expected 206 for pure arginine. While crystallization would be the method of choice to produce 207 arginine of the highest purity, precipitation might be preferred when a higher yield is 208 required. Thus, although the precipitated preparation is only 90.7 % arginine, the yield 209 of precipitation is 38% vs. 22% for crystallization.

In conclusion, although the presence of free amino acids in seed legumes has been known for a long time the contents and potential interest of free amino acids in pulses such as chickpea have not received much attention. We show that chickpea is a potential source of free amino acids, especially arginine, that can be easily purified, as a green alternative to production by fermentation.

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175.

Table 1. Total and free amino acid composition in chickpea seeds. Results, expressed as g / 100 g amino acids, are the average  $\pm$  sd of two determinations.

	Total amino acids	Free amino acids
Asp	$13.5^{a} \pm 0.1$	5.7 ± 0.5
Glu	$18.7^{b} \pm 0.1$	20.3 ± 1.6
Asn		3.9 ± 0.5
Ser	$5.9 \pm 0.1$	$0.9 \pm 0.2$
Gln		$0.0 \pm 0.0$
His	$2.3 \pm 0.1$	$0.5 \pm 0.0$
Gly	$4.3 \pm 0.1$	$1.6 \pm 0.0$
Thr	$4.1 \pm 0.0$	$0.7 \pm 0.0$
Arg	$10.4 \pm 0.1$	53.5 ± 5.0
Ala	$4.6 \pm 0.0$	$0.9 \pm 0.0$
Pro	$1.8 \pm 0.1$	$0.0 \pm 0.0$
Tyr	$2.3 \pm 0.0$	$0.5 \pm 0.0$
Val	$4.1 \pm 0.0$	$0.5 \pm 0.0$
Met	$0.8 \pm 0.0$	$0.5 \pm 0.0$
Cys	$1.1 \pm 0.1$	$0.0 \pm 0.0$
lle	$3.6 \pm 0.1$	$0.2 \pm 0.0$
Trp	$0.8 \pm 0.0$	
Leu	$8.0 \pm 0.1$	$7.3^{c} \pm 0.7$
Phe	$6.0 \pm 0.1$	$1.4 \pm 0.0$
Lys	7.2 ± 0.1	$1.8 \pm 0.2$

293	<sup>a</sup> Asp + Asn. <sup>b</sup> Glu + Gln. <sup>c</sup> Leu + Trp.
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297	Table 2. Amino acid composition (g / 100 g amino acids) of chickpea permeate
298	and washes of the Amberlite resin using 0.5 N NH <sub>4</sub> and concentrated NH <sub>4</sub> (7.5 N).
299	Results are the average ± sd of two determinations.

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	Permeate	ermeate 0.5 N NH.OH		7 5 NH.OH		Dracinitad	Crystalized
	Termedie	wash	first wash	second wash	third wash	Arg	Arg
						8	9
Asp	5.29 ± 0.16	27.51 ± 0.09	4.19 ± 0.02	3.23 ± 0.03	$1.66 \pm 0.00$	2.91 ± 0.02	n.d.
Glu	29.40 ± 0.53	57.31 ± 0.09	31.01 ± 0.20	23.69 ± 0.20	$15.50 \pm 0.10$	6.37 ± 0.08	n.d.
Asn	3.45 ± 0.08	$4.01 \pm 0.03$	$3.64 \pm 0.07$	$2.61 \pm 0.10$	$1.72 \pm 0.02$	n.d.	n.d.
Ser	1.36 ± 0.07	1.75 ± 0.00	$1.69 \pm 0.01$	$1.14 \pm 0.10$	$0.71 \pm 0.01$	n.d.	n.d.
Gln	0.26 ± 0.04	0.22 ± 0.00	$0.14 \pm 0.00$	$0.12 \pm 0.00$	$0.08 \pm 0.01$	n.d.	n.d.
His	0.43 ± 0.05	0.38 ± 0.03	$0.15 \pm 0.04$	$0.09 \pm 0.01$	$0.06 \pm 0.00$	n.d.	n.d.
Gly	1.89 ± 0.04	$1.24 \pm 0.02$	$2.11 \pm 0.00$	$1.51 \pm 0.00$	$0.93 \pm 0.01$	n.d.	n.d.
Thr	1.92 ± 0.15	$2.43 \pm 0.01$	$1.69 \pm 0.01$	$1.27 \pm 0.01$	$0.76 \pm 0.00$	n.d.	n.d.
Arg	30.05 ± 0.89	n.d.	25.37 ± 0.28	38.89 ± 0.26	55.99 ± 0.37	90.73 ± 0.07	102.87 ± 0.66
Ala	3.35 ± 0.19	1.23 ± 0.02	$2.86 \pm 0.10$	$1.74 \pm 0.08$	0.94 ± 0.13	n.d.	n.d.
Pro	n.d.*	$1.60 \pm 0.13$	$2.17 \pm 0.14$	$1.42 \pm 0.07$	$0.32 \pm 0.06$	n.d.	n.d.
Tyr	1.77 ± 0.04	$0.29 \pm 0.01$	$1.94 \pm 0.02$	$1.81 \pm 0.00$	$1.58 \pm 0.02$	n.d.	n.d.
Val	0.95 ± 0.12	0.74 ± 0.06	$1.94 \pm 0.23$	$1.24 \pm 0.15$	$0.71 \pm 0.24$	n.d.	n.d.
Met	0.05 ± 0.07	n.d.	$0.47 \pm 0.04$	$0.29 \pm 0.01$	n.d.	n.d.	n.d.
Cys	0.04 ± 0.05	n.d.	$0.25 \pm 0.01$	$0.14 \pm 0.00$	n.d.	n.d.	n.d.
lle	0.92 ± 0.02	$0.20 \pm 0.00$	$1.11 \pm 0.01$	$0.86 \pm 0.01$	$0.54 \pm 0.02$	n.d.	n.d.
Leu + Trp	14.69 ± 0.29	0.78 ± 0.00	14.69 ± 0.13	15.97 ± 0.08	$15.90 \pm 0.07$	n.d.	n.d.
Phe	2.68 ± 0.07	$0.31 \pm 0.01$	$3.53 \pm 0.03$	$3.19 \pm 0.00$	$2.19 \pm 0.12$	n.d.	n.d.
Lys	0.71 ± 0.04	n.d.	$1.05 \pm 0.01$	$0.78 \pm 0.00$	$0.42 \pm 0.00$	n.d.	n.d.
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\*n.d.: not detected.

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Table 3. Detailed total mass, soluble sugars, and amino acid contents, of a representative experiment, in the Amberlite washes using pH 2 water, 0.5 N NH<sub>4</sub>OH and 7.5 N NH<sub>4</sub>OH. Results are the average  $\pm$  sd of three determinations.

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311 Total mass Sugars Amino acids (mg) (mg) (mg) tr1 First pH 2 water wash 141.9 ± 4.7 22.9 ± 2.0 Second pH 2 water wash 40.8 ± 2.9  $18.4 \pm 1.7$ tr Third pH 2 water wash  $11.0 \pm 1.0$  $7.2 \pm 0.3$ tr Fourth pH 2 water wash  $5.2 \pm 0.0$  $3.1 \pm 1.4$ tr Total pH 2 water washes 198.9 51.6 tr 0.5 N NH<sub>4</sub>OH wash  $7.2 \pm 0.0$ tr  $1.7 \pm 0.1$ First 7.5 N NH<sub>4</sub>OH wash 68.0 ± 0.0  $29.9 \pm 0.6$ tr Second 7.5 N NH₄OH wash 33.0 ± 5.1 tr  $6.9 \pm 0.2$ nd<sup>2</sup> Third 7.5 N NH<sub>4</sub>OH wash tr  $2.5 \pm 0.1$ Total 7.5 N NH₄OH washes 39.3 101 tr

Figure legends. Figure 1. Free amino acids (full bars) and arginine (open bars) contents (g / 100 g seed flour) in five commercial grain legumes. Results are the average ± sd of three determinations. Figure 2. FPLC gel filtration profile of A) chickpea extracts, B) nanofiltration retentate, C) nanofiltration permeate. Dashed line in Figure 1A represents the cut-off molecular weigth (200 Da) of the nanofiltration membrane. Figure 3. Balance of total mass (open bars), soluble sugars (grey bars) and free amino acids (black bars) in a representative experiment of arginine purification using Amberlite IR-120 resin. Results are the average  $\pm$  sd of three determinations. 











FIGURE 3