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Environmental gating of conductance and temperature dependence in oligothiophene single molecule junctions

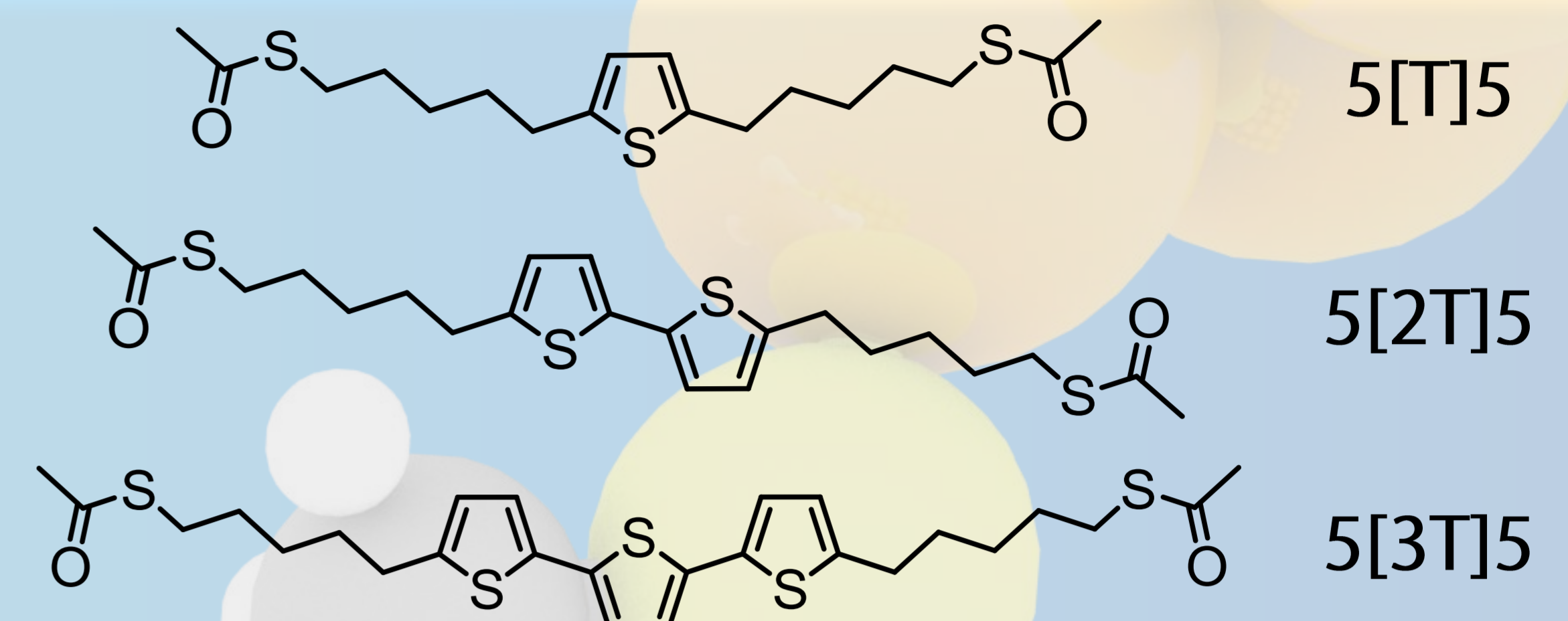
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Introduction

In the last decade, many techniques [1-3] have been developed to fabricate and characterise metal|molecule|metal junctions. Although an isolated molecule within a junction must be affected by supramolecular interactions, this has provoked few detailed studies. We synthesised a series of **oligothiophene** molecular wires and measured their conductance as a function of temperature in different environments (air, dry argon and UHV), and found a remarkably different behaviour. The effect of ambient moisture on the conductance of oligothiophene molecular wires is known in the literature [4], and this study sheds some more light on the mechanism of charge transport through such systems.



Experimental

AIR – CONDUCTANCE INCREASES WITH LENGTH
UHV – CONDUCTANCE DECREASES WITH LENGTH

Molecular conductance has been determined using the non-contact $I(z)$ technique [3], involving the repeated formation of molecular junctions within an STM.

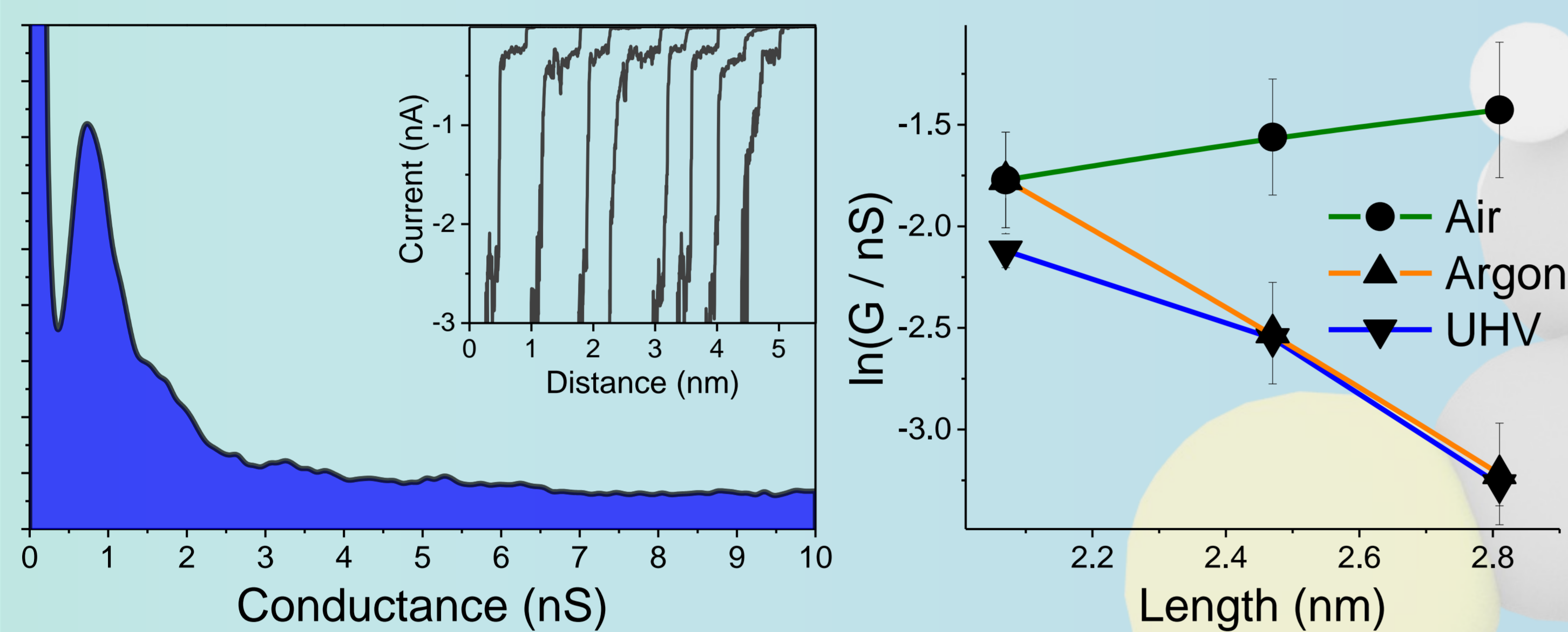


Figure 1: Left: Example of conductance histogram (5[T3]5 in air) and traces (inset). 500 such traces have been used in histogram. Right: Logarithmic plot of conductance in the three environments.

While a temperature dependent conductance is generally taken as sign of incoherent hopping charge transport, it has been previously demonstrated that thermal broadening of transport resonances can explain such behaviour within a **coherent tunnelling mechanism** [5].

AIR – THERMALLY ACTIVATED CONDUCTANCE
UHV – ACTIVATIONLESS CONDUCTANCE

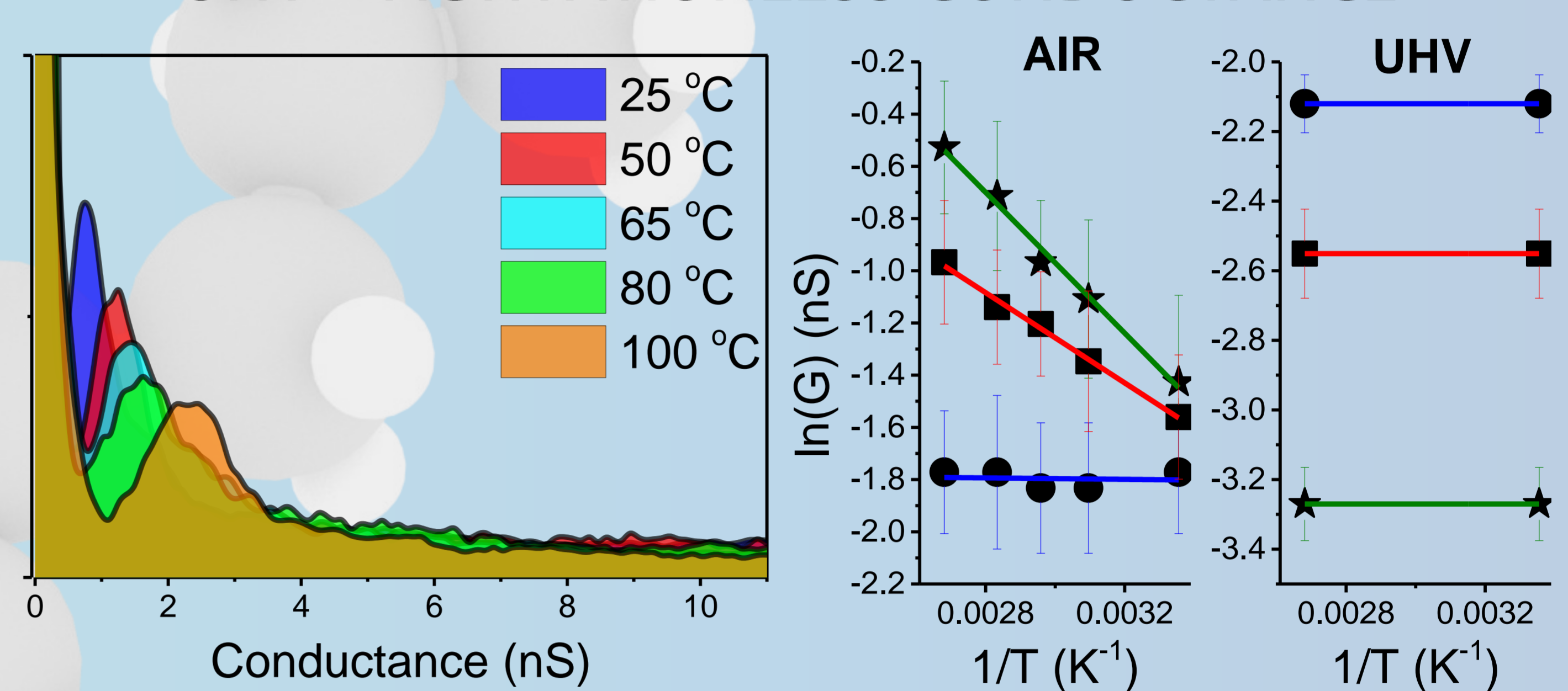


Figure 2: Left: Conductance histograms for 5[T3]5 in air over a range of temperatures. 300-600 current-distance scans have been used to construct each histogram. Right: Arrhenius plot for the three oligomers employed in this study in air and UHV. Activation energies in air are 0.13 eV for 5[T3]5 (green) and 0.08 eV for 5[2T]5 (red). 5[T]5 (blue) showed an activationless mechanism in both environments.

Calculations

NEGF and DFT calculations have been used to understand the observed behaviour: in air, the presence of **ambient moisture** shifts LUMO resonance towards the Fermi level.

As the number of thiophene rings increase, the resonances are moved further, resulting in an **increasing boost** in conductance along the oligothiophene series.

The observed temperature dependence can be explained *via* **thermal broadening** of transport resonances. In vacuum the resonances are far from the Fermi level of the electrodes and there is no thermal activation. In air, 5[T2]5 and 5[T3]5 have resonances shifted close enough to the Fermi level for their thermal broadening to promote a temperature dependent conductance.

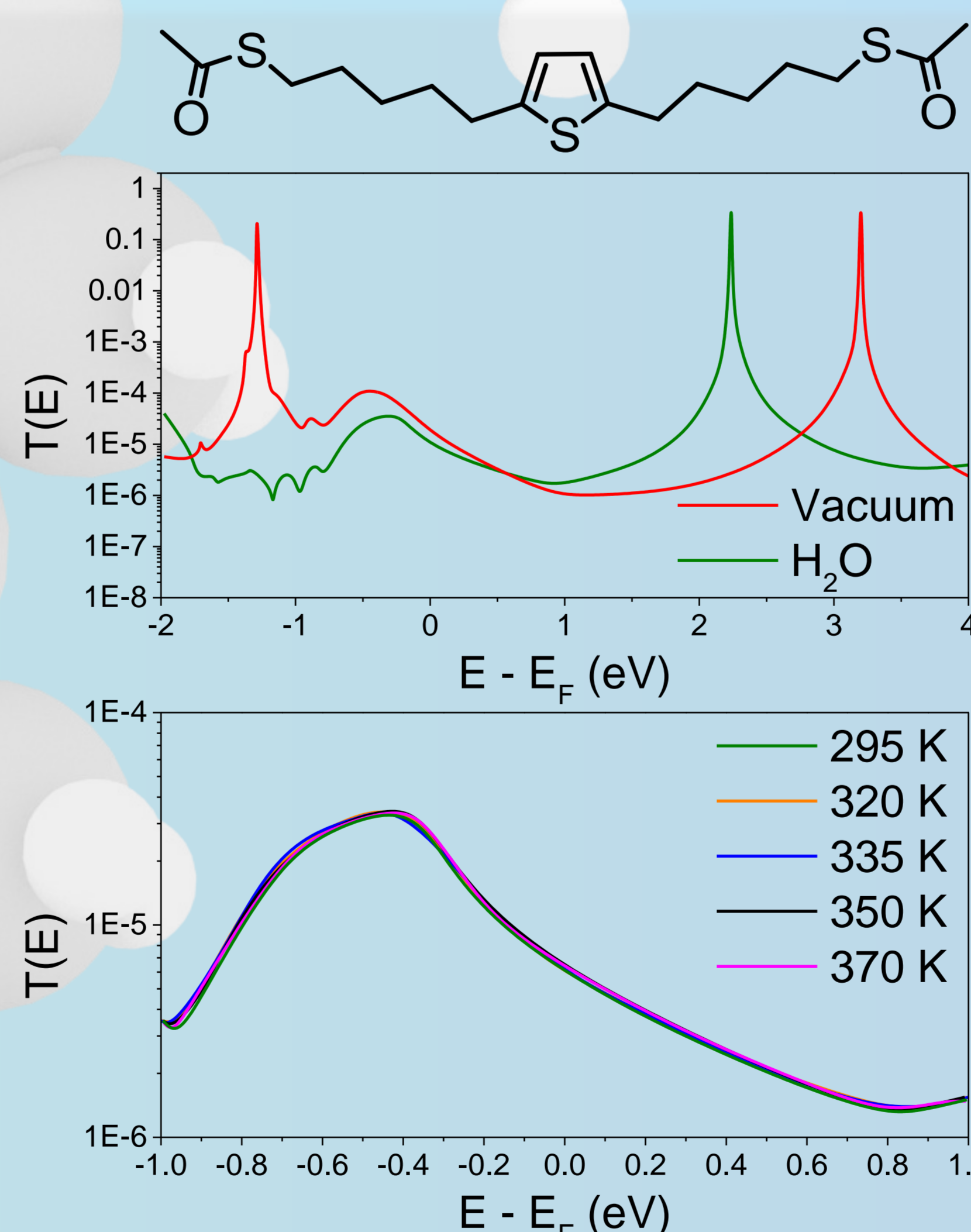


Figure 3: Top: Calculated transmission curves for 5[T]5 in vacuum and in the presence of water. Bottom: Close-up of calculated transmission curves at different temperatures near the Fermi level of the electrodes.

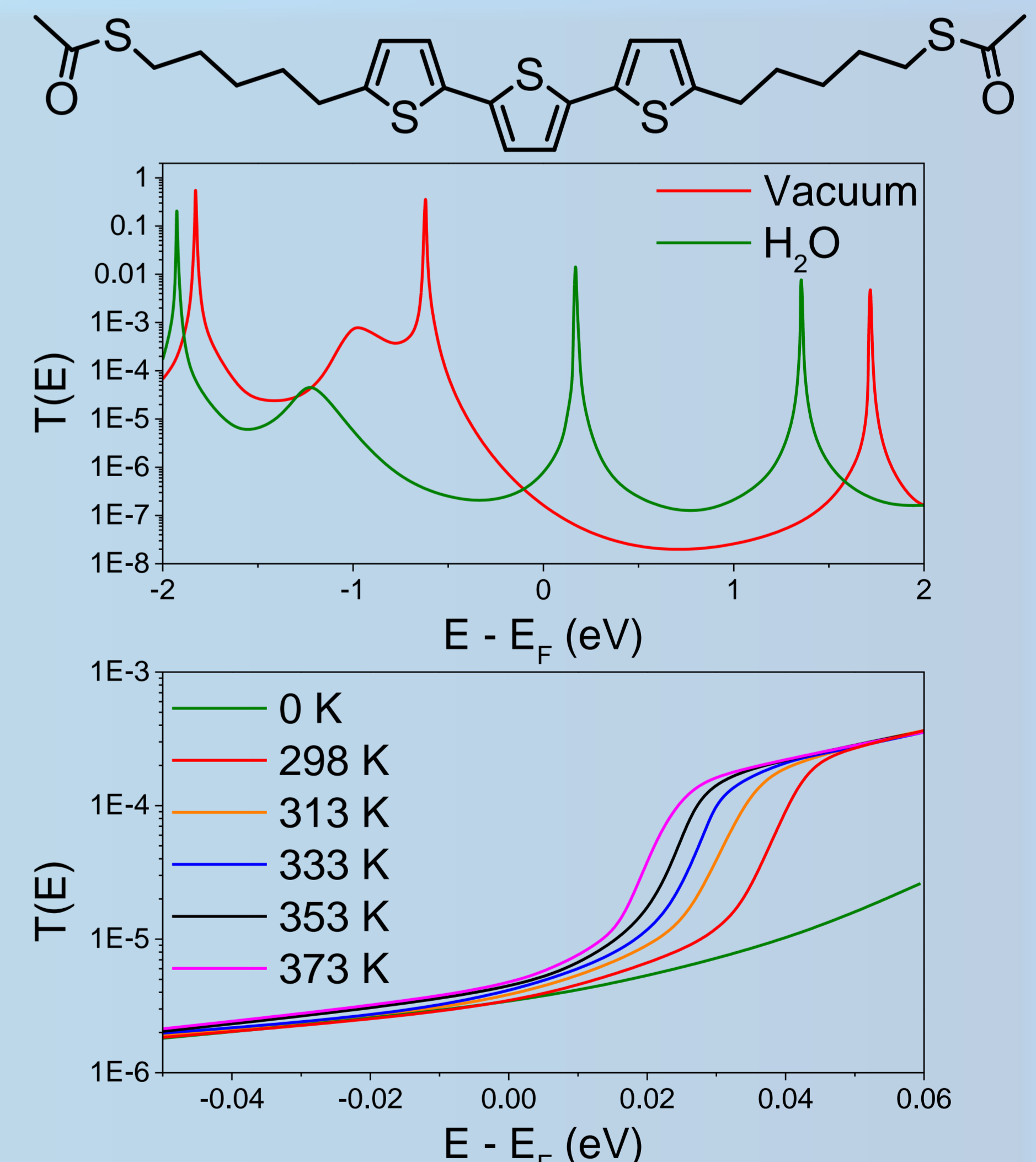


Figure 4: Top: Calculated transmission curves for 5[T3]5 in vacuum and in the presence of water. Bottom: Close-up of calculated transmission curves at different temperatures near the Fermi level of the electrodes.

Conclusions

We synthesised and measured the conductance of a series of thiol-capped α - ω dialkyl oligothiophenes in different environments and over a range of temperatures. We found that the longer oligomers have **increased conductance** and **thermal activation** upon coordination with water (as ambient moisture). NEGF and DFT calculations explained this behaviour within a coherent tunnelling theory: water coordination shifts transport resonances closer to the Fermi level of the electrodes, and the temperature dependence of conductance can be ascribed to their thermal broadening.

References:

- [1] Reed, M. A., Zhou, C., Muller, C. J., Burgin, T. P. & Tour, M. J. *Science* **278**, 252-254 (1997). [2] Xu, B. & Tao, N. J. *Science* **301**, 1221-1223 (2003). [3] Haiss, W., van Zalinge, H., Higgins, S. J. et al. *J. Am. Chem. Soc.* **125**, 15294-15295 (2003). [4] Leary, E., Hobenreich, H., Higgins, S. J. et al. *Phys. Rev. Lett.* **102**, 086801 (2009). [5] Sedghi, G., Garcia-Suarez, V. M., Esdaile, L. J. et al. *Nature nanotechnol.* **6**, 517-523 (2011).