

# A Modified Accelerated Chloride Migration Tests for UHPC And UHPFRC With PVA And Steel Fibers

Juliano Provete Vincler, Thomas Sanchez, Vicky Turgeon,

David Conciatori\* and Luca Sorelli

Department of water and civil engineering, Laval University Laval, Canada

\*Corresponding author. E-mail address: david.conciatori@gci.ulaval.ca

## Abstract

Accelerated migration tests which are commonly used to measure chloride diffusion in ordinary cement-based materials cannot be directly applied to composite with very low permeability, such as Ultra High-Performance Fiber Reinforced concretes (UHPFRC). In order to assess the UHPFRC enhancement on the structural durability, there is a critical need to accurately assess the permeability level of the material to chloride ions. The objective of this work is to adapt an existing set-up of accelerated chloride migration test in order to (i) better characterize the resistance of chloride ion penetration in UHPFRC; and (ii) to compare the resistance of chloride ion penetration between UHPC and UHPFRC. The material characterization, the set-up modifications of the existing accelerated migration test, the results are presented. In conclusion, the modification of the test-set-up allowed to accurately measure chloride transport of very low permeability UHPFRC and to shed light on the effect of the fiber reinforcement.

**Keywords:** UHPC, UHPFRC, chloride, diffusion, profiles, migration test

21        **1. INTRODUCTION**

22        Ultra-High-Performance Concrete (UHPC) and Ultra-High-Performance Fiber Reinforced  
23        Concrete (UHPFRC), respectively without and with fibers, were successfully employed for  
24        structural applications thanks to their outstanding compressive strength and tensile toughness  
25        along with its remarkable durability. UHPFRC improved concrete structures in terms of longer  
26        span, material saving, and enhanced lifetime. Decades of research leads to UHPFRC as it is  
27        known today, with its low porosity and high-packing density [1]. The outstanding UHPFRC  
28        durability is based on their extremely low permeability and diffusivity, which is achieved by  
29        maximizing the microstructure packing density [2]–[5]. Thus, external layer of UHPFRC can  
30        be effectively employed to enhance both the strength of damaged concrete structure and their  
31        permeability to external aggressive agents with an optimal gain in terms of the structure lifetime  
32        [6], [7].

33        One of the major durability issues for reinforced concrete structure is unequivocally the  
34        chloride permeability that leads to the corrosion of steel bars [8]–[11]. Very few tests exist on  
35        chloride resistance of UHPFRC in real field conditions and this information is key for  
36        estimating the economic gain on the maintenance cost and the structure lifetime especially in  
37        regions with severe climate conditions [12], [13]. As for example, Thomas et al. monitored the  
38        chloride penetration in a material prism without fibers (UHPC) under real field conditions  
39        which consists of twice-daily tides of wet/dry cycles and winter freeze/thaw cycles [14]. After  
40        a period of about 12-15 years of severe marine exposure, the chloride penetration depth was  
41        limited to 6-10 mm, which was approximately one third of the chloride penetration depth  
42        observed on reference samples of High-Performance Concrete. As for laboratory tests, different  
43        kinds of accelerated tests exist to rapidly assess the chloride ion penetration resistance of

44 concrete [15]–[18]. Depending on the test method, different types of results can be obtained,  
45 such as chloride profile, effective or apparent diffusion coefficient or with a number of  
46 Coulomb to give an indicative rank of the chloride ion permeability of the material. These  
47 results that can be difficult to compare. Many other factors influence the determination of the  
48 chloride ion permeability such as the type of material, the preparation procedure of the  
49 specimen or the testing method. These results are then interpreted to judge of the durability of  
50 the material regarding the chloride ion penetration. However, such tests may not be suitable to  
51 accurately measure the diffusion coefficient of UHPFRC and characterize their durability  
52 regarding the chloride ion permeability. In standard test methods, such as ASTM C1202 [19],  
53 an electrical field applied accelerate the chloride diffusion. That is difficult to measure the  
54 diffusion through samples because of undesirable temperature raise and short-circuit due to  
55 favorable alignment of the fibers. In fact, the recent standard for fabricating and testing  
56 UHPFRC specimen (ASTM C1856 [20]), specify that the ASTM C1202 test method is not  
57 applicable for UHPFRC (including fibers) for the previously mentioned reasons.

58       Despite the difficulty and uncertainty of the ASTM C1202 test method for UHPC with  
59 fibers, some research shows successful tests and concluding results for UHPC with and without  
60 fibers. The results, ranging from 0 to 360 Coulombs, correspond to the negligible chloride ion  
61 permeability of the standard and no significant difference is observable between UHPC matrix  
62 with and without fibers. However, a literature review of the chloride ion permeability of UHPC  
63 in terms of apparent diffusion coefficient, which consider chemical interaction with the  
64 material, shows great scattering of results. The results are not from accelerated test, that is  
65 without electrical current and thus spend much time than accelerated tests. Thomas et al. and  
66 Piérard et al. report respectively an apparent diffusion coefficient of  $1.3 \times 10^{-13} \text{ m}^2/\text{s}$  and

67  $2 \times 10^{-13}$  m<sup>2</sup>/s for UHPC. Tanaka obtain an apparent coefficient of diffusion of  $4 \times 10^{-15}$  m<sup>2</sup>/s for  
 68 UHPFRC exposed for 5 years in an open box girder of a bridge and  $4.7 \times 10^{-16}$  m<sup>2</sup>/s for  
 69 immersed UHPFRC in chloride solution for 2.5 years. Table 1 shows a summary of the results  
 70 from the considered research.

REFERENCE	FIBERS	RESULT	TESTING METHODS
Habel et al. [21]	∅ 5.5% steel	88 72	
Thomas et al. [14]	∅	0, 19	
Bonneau et al. [22]	140 kg/m <sup>3</sup> PVA	6, 9	Coulomb ASTM C1202 (6h)
Graybeal [17]	2% steel	18, 18, 26, 39, 76, 360	
Ahlborn et al. [23]	2% steel	15, 75	
Thomas et al. [14]	∅	$1.3 \times 10^{-13}$	ASTM C1556 [24] (63d.)
Piérard et al. [25]	∅	$2 \times 10^{-13}$	NT Build 443 [26] (90d.)
Tanaka [27]	2% steel	$4 \times 10^{-15}$ $4.7 \times 10^{-16}$	Apparent diffusion coefficient (m <sup>2</sup> /s) 5y. exposure in a bridge 2.5y. immersion in chloride solution

71 Table 1. Results of chloride ion penetration resistance from literature

72 While test method ASTM C1202 does not highlight significant difference in result of the  
 73 chloride ion permeability for UHPC or UHPFRC from the different experiments, a difference  
 74 can be observed for diffusion coefficients of UHPC and UHPFRC with natural diffusion tests.  
 75 These results suggest that the ASTM C1202 test method may be insufficient to accurately  
 76 characterize the chloride ion permeability of UHPFRC and the results on the unique UHPC  
 77 matrix may not be representative of the real application of the material with fibers. It can  
 78 therefore lead to strong difference on the lifetime prediction or the optimal design of the steel  
 79 rebar cover in thin UHPFRC elements [28].

80 In order to improve the durability design of UHPFRC, this paper aims to propose an  
81 alternative test method to ASTM C1202 in order to (i) efficiently characterize the resistance of  
82 chloride ion penetration in UHPFRC; and (ii) compare the resistance of chloride ion penetration  
83 between UHPC and UHPFRC with two different types of fibers and volume content.  
84 Additionally, the effect of different preparation procedure of the specimen on the chloride  
85 transport properties was also considered.

## 86 **2. MATERIALS AND METHODS**

### 87 **2.1 Mix designs and sample preparation**

88 **UHPC.** The mix is commercially available under the name of Ductal® and composed of  
89 2195 kg/m<sup>3</sup> of premix, 35 kg/m<sup>3</sup> of superplasticizer type A&F [29] and 125 kg/m<sup>3</sup> of water for  
90 UHPC. The water-to-binder ratio was about 0.17. UHPC mix was batched in a small Hubert  
91 mixer (7 liters) and cast in a polyvinyl chloride cylindrical molds (100 mm diameter and  
92 200 mm height). The sample surface was protected for 24 h with a wet cover before demolding.

93 **UHPFRC-PVA.** The water amount of the UHPC mix with polyvinyl alcohol fibers was  
94 corrected to 150 kg/m<sup>3</sup> to consider the PVA fiber absorption. The water-to-binder ratio is about  
95 0.21 and PVA fibers are 12.0 mm length and 0.2 mm diameter. The fiber volume content was  
96 varied between 1%, 2%, 4%, and 5%. The UHPFRC-PVA samples were prepared with the same  
97 procedure as for UHPC samples.

98 **UHPFRC-STEEL.** The mix UHPC with steel fibers is composed of 125 kg/m<sup>3</sup> of water and  
99 the water-to-binder ratio is about 0.17 [16]. The average geometry is 12.7 mm length and  
100 0.2 mm diameter for steel fiber. Two casting methods were done for this mix respectively  
101 named G1 and G2.

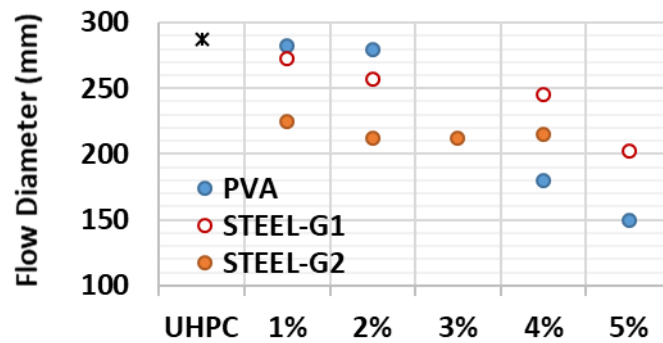
102 The **G1 group** had fiber volumes content varying between 1%, 2%, 4%, and 5%. Two  
103 cylinders for each fiber volume were cast. During placements, the fresh concrete was vibrated,  
104 and the fibers were added before the UHPFRC turned fluid thanks to its thixotropic behavior.  
105 A slump flow test was carried out. After unmolding, the cylinders were thermally cured for one  
106 week according to the following phases: (i) 24 h submersion in lime saturated water at the  
107 laboratory temperature of 20 °C; (ii) 48 h curing at the temperature of 60 °C with relative  
108 humidity of about 90%; and (iii) 96 h submersion in lime saturated water at laboratory  
109 temperature of 20 °C; (iv) curing at 100% relative humidity until the testing day.

110 For **G2 group**, the steel fiber volume contents were considered from 1% to 4%. Four  
111 cylinders for each fiber volume content were cast. No mechanical vibration was applied during  
112 the placement and the fibers were added only after the thixotropic matrix fluidification. After  
113 unmolding, the cylinders were cured according to the following steps: (i) 96h submersion in  
114 lime saturated water at laboratory temperature of 20 °C and (ii) 72 h curing at a temperature of  
115 60 °C with relative humidity of 90%; (iii) curing at 100% relative humidity until the testing  
116 day.

## 117 **2.2 Characterization tests**

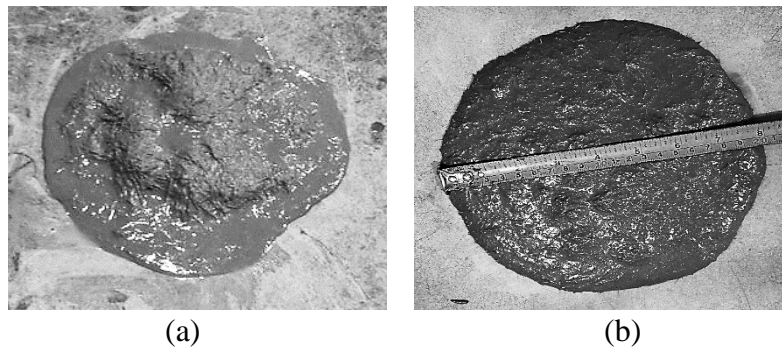
118 **Flow-table test.** The flow-table test was conducted with the apparatus from  
119 ASTM C230 [30]. The test consists of pouring the fresh concrete in the flow mold on the flow  
120 table and measuring the concrete diameter as an average of the largest and the smallest  
121 diameters (D1 and D2). PVA and STEEL-G1 average flow diameter decrease during fiber  
122 volume content increase (Figure 1). Mixes with fiber volume content of 4% and 5%, which  
123 showed fiber segregation, presented the lowest flow diameter [31]. The amount of

124 superplasticizer of STEEL-G2 mixes was increased by 17% to keep a constant flow diameter  
 125 for different fiber volume content. The Figure 2 allows to visually compares the flow diameter  
 126 for the 4%STEEL-G1 and 4%STEEL-G2 mixes. Then, the 4%STEEL-G2 mix is much more  
 127 homogenous than the 4%STEEL-G1 one, that highlight the effects of the different casting  
 128 methods.



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Figure 1. Flow diameter results

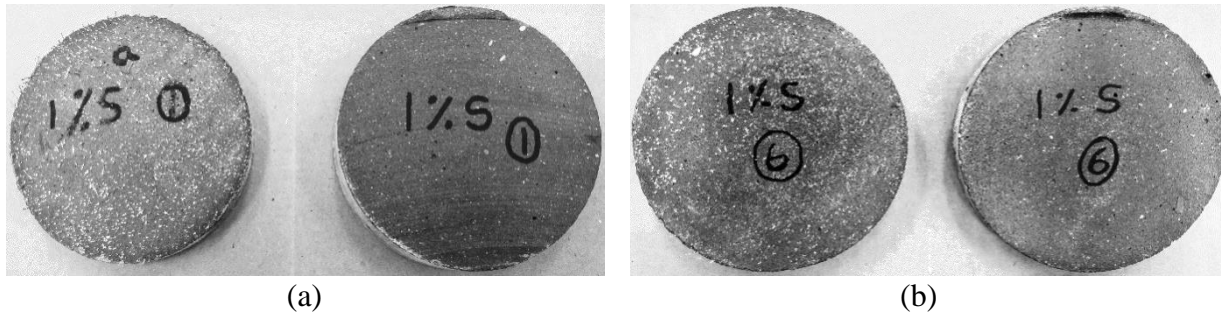


131 Figure 2. Visual comparison of the flow-table test results: (a) 4%STEEL-G1 and (b)  
 132 4%STEEL-G2

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134 **Visual observations.** Test cylinders were cut to eliminate both extremities to prevent milt  
 135 effects. Segregation was observed for UHPFRC-STEEL samples (Figure 3), but not for  
 136 UHPFRC-PVA. This could be explained by the highest density of steel fibers compared to the  
 137 PVA ones. Figure 3b shows that more fibers are present on the top of the STEEL-G2 cylinder

138 compared to the top of the STEEL-G1 cylinder (Figure 3a) meaning that segregation is less  
 139 important for the G2 mixes. The casting procedure of STEEL-G2, which was without vibration,  
 140 could explain this fact.



141 Figure 3. Bottom (left) and top (right) of the test cylinder for (a) STEEL-G1 samples (b)  
 142 STEEL-G2 samples

143 **Water porosity.** Measures were performed, according to Grandubé [32] and French standard  
 144 [34], on UHPC, UHPFRC with 1% fiber volume content of steel and PVA (Table 2). UHPFRC-  
 145 PVA samples were dried at only 50 °C instead of the 110 °C specified by the standard to avoid  
 146 burning the PVA fibers. Three measurements were performed for each material to ensure the  
 147 repeatability. All tests were done on samples coming from the center parts of the cylinder to  
 148 avoid the effect of the observed segregation. These tests were done in order to confirm that the  
 149 material respect the UHPFRC porosity from the AFGC standard [34].

<b>DRYING TIME</b>	<b>UHPC</b>	<b>UHPFRC 1%PVA</b>	<b>UHPFRC 1%STEEL G1</b>	<b>UHPFRC 1%STEEL G2</b>
<b>6 DAYS</b>	5.6%	4.4%	4.9%	4.4%
<b>2 MONTHS</b>	7.0%	7.6%	7.6%	6.8%

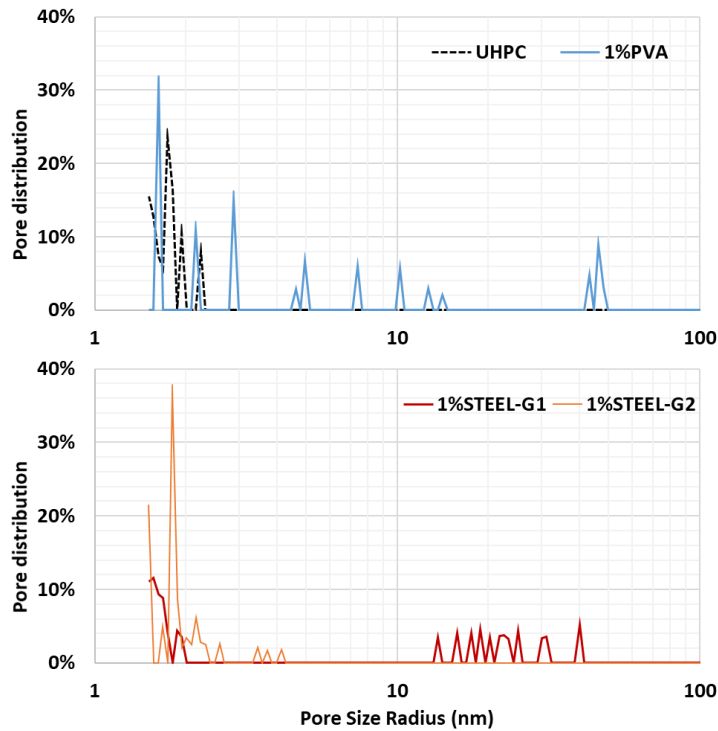
150 Table 2. Water porosity measurements

151 After 6 days of drying and no variation of mass for 24 hours, the materials had a water porosity  
 152 respecting the standard range of UHPC (6 - 9%) and UHPFRC (1.5 - 5%). However, this water  
 153 porosity procedure is not adapted for this kind of concrete which has very low porosity and



154 thus, hard to dry in a short delay (24 hours). After the first validation with AFGC recommended  
155 value, intervals between mass verification was increased to one week and drying lasted two  
156 months. The water porosity results obtained from the extended method revealed less difference  
157 between all mixes than the previous results obtain with the recommended method from  
158 standards. Water porosity for the STEEL-G2 mixes were lower than the STEEL-G1 mixes,  
159 which can be explained by the casting method. Indeed, vibration of G1 mixes may have bonded  
160 smaller air bubbles to form bigger ones which are detectable with water porosity measurements.

161 **Mercury Porosity Intrusion (MIP)** is effective to quantify the pore distribution from 1 nm  
162 up to 100 nm [35]. Analyses were performed for samples of UHPC and UHPFRC with 1%  
163 fibers (Figure 4). All tests were done on samples coming from the center parts of the cylinder  
164 to avoid the effect of the observed segregation. The pore distributions show similar nanopores  
165 (around 2 nm) for all samples. Except for the nanopore of 2 nm, the pore distribution of the  
166 different samples shows some differences. For all UHPFRC samples, capillary pores are present  
167 between 2 and 50 nm, as opposed to the UHPC samples. The presence of fibers in the material  
168 could explain the difference in the pore distribution between the samples with and without  
169 fibers. Not only the presence of fibers but the type of fiber as the pore distribution of the PVA  
170 samples is different that the STEEL samples. A difference can also be noted from the casting  
171 method as STEEL-G1 and STEEL-G2 have different pore distribution between 2 and 50 nm,  
172 where STEEL-G2 has a pore distribution more similar to the UHPC one. The variability of the  
173 porosity from the casting method can be confirmed from the results of the water porosity where  
174 the porosity of the STEEL-G1 and STEEL-G2 are respectively 7.6% and 6.8%. The capillary  
175 pores of 20-30 nm for STEEL-G1, which were also observed in the water porosity measures,  
176 could influence chloride diffusion and accelerate its migration.



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Figure 4. Pore size distribution between 1 and 100 nm from MIP measures

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### 2.3 Accelerated migration test method

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**Samples preparation.** The optimal sample thickness chosen was 25 mm for UHPC and UHPFRC-PVA and 50 mm for UHPFRC-STEEL. The samples were saturated in a water solution with 300 mol/m<sup>3</sup> of sodium hydroxide (NaOH) under vacuum pressure for at least 16 hours as prescribed in ASTM C1202.

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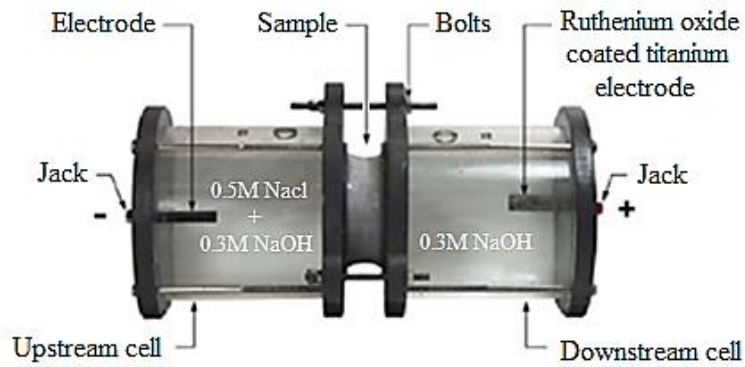
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**Chloride accelerated migration tests** were carried out according to an adapted protocol of ASTM C1202. The test-set-up for migration tests (Figure 5) was adapted regarding the following points: (i) the volume of the solution was adjusted from 0.5 L to 2.7 L to reduce heating issues; (ii) the voltage was increased from 20V to 70V to accelerate the tests; (iii) the sample thickness was reduced in previous work [31]. After preliminary tests, the sample thickness was chosen to be 50 mm for UHPFRC-STEEL and 25 mm for UHPFRC-PVA and

191 UHPC. The electrical potential and the duration of the test were 1400 V/m and 28 days  
 192 respectively. The temperature of the upstream solution, the electrical field passing through the  
 193 sample and the current were regularly measured.



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Figure 5. Chloride ion migration cells [11]

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196 **Grinding operations** were made with a 7.6mm core drill diameter. The concrete powder  
 197 was collected with a small brush to obtain a minimum of 7.0 g of powder per layers of 1 mm  
 198 each. The depth of each layer was measured with a caliper. About 10.0 g of concrete powder  
 199 was collected for each layer.

200 **Chloride titration test.** The total chloride present in the concrete powder was extracted with  
 201 apperal 10% of a nitric acid solution (HNO<sub>3</sub>) with a 69% (± 1%) purity. 5 g of concrete powder  
 202 and 50 ml of acid solution were mixed together. The chloride was extracted through its  
 203 dissolution, for 1 hour, into the acid. The mixture was then filtered using a vacuum and analyzed  
 204 by titration with Mettler Toledo T50 titrator.

205 **Transport model.** The apparent chloride diffusion coefficient  $D_{a,cl}$  obtained from  
 206 accelerated chloride migration tests was calculated using the following equation [36]–[38]:

$$D_{a,cl} = \frac{RTL}{z_{cl}FU} \left( \frac{x_d - \alpha\sqrt{x_d}}{t} \right). \quad (1)$$

207  $R$  represents the ideal gas constant (83,145 J/mol/K);  $T$  the temperature (K);  $L$  the thickness of  
 208 the specimen (m);  $z$  the ion valence ( $z_{Cl} = 1$ );  $F$  the Faraday constant (96,488.46 C/mol),  $U$  is  
 209 the voltage (V);  $x_d$  is the chloride penetration depth (mm);  $t$  the time test (s) and

$$\alpha = 2 \sqrt{\frac{RTL}{zFU}} \operatorname{erf}^{-1} \left( 1 - \frac{2c_d}{c_0} \right), \quad (2)$$

210 where  $c_d$  is the chloride concentration (kg/m<sup>3</sup>) measured at the penetration depth  $x_d$  and  $c_0$  is  
 211 the chloride concentration (kg/m<sup>3</sup>) at the sample surface.

### 212 **3. RESULTS AND DISCUSSION**

213 The migration tests were successfully carried out for all samples of UHPC and UHPFRC-  
 214 PVA. Moreover, only few samples of UHPFRC-STEEL with a fiber volume content lower than  
 215 2-3% have passed the test. Most of UHPFRC-STEEL samples presented corrosion and cracks  
 216 leading to the failure of the accelerated migration test.

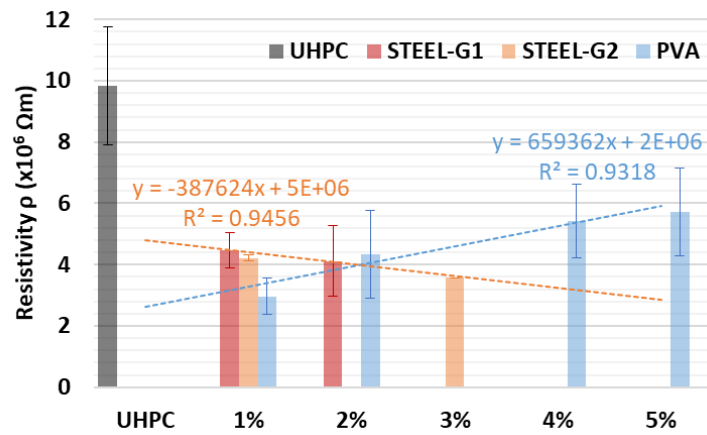
#### 217 **3.1 Verification of the new accelerated migration test procedure**

218 **Resistivity measures.** Considering the applied voltage  $U$  (V), the current  $I$  (A), the thickness  
 219  $L$  (m) and the surface  $S$  (m<sup>2</sup>) of the sample, its electrical resistivity  $\rho$  ( $\Omega \cdot m$ ) can be obtained  
 220 with [39]

$$\rho = \frac{US}{IL}. \quad (3)$$

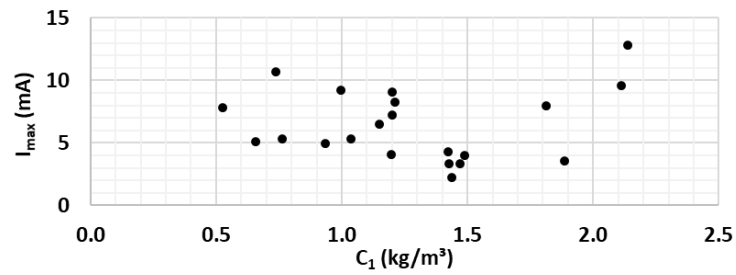
221 The resistivity calculated (Figure 6) are very close to each other despite the different the  
 222 sample thickness used (25 and 50 mm) because the electrical field applied ( $E = U/L$ ) are  
 223 similar. The samples with fibers presented a lower resistivity than the UHPC one. For the  
 224 UHPFRC-PVA and STEEL-G1 samples, these results agree with the water porosity measures

225 (J2.2). The porosity of 1%PVA and 1%STEEL-G1 were greater than the porosity of UHPC  
 226 samples resulting in a greater ionic quantity in pore solutions, and thus a lower resistivity. The  
 227 addition of PVA fibers seems linearly proportional to the resistivity of the samples. Despite a  
 228 lower water porosity for STEEL-G2 compared to the UHPC samples, the STEEL-G2 resistivity  
 229 was lower. The conductivity of the steel fibers as opposed to the PVA fibers could explain this  
 230 phenomenon.



231  
 232 Figure 6. Initial resistivity of the different samples

233 **Maximal current.** The maximal current ( $I_{max}$ ) measured during migration tests were  
 234 between 2 and 13 mA as observed with Figure 7. The current that represent the ionic flux  
 235 passing through the sample remained very low and did not influence the chloride concentration  
 236 measures in the first grinded layer ( $C_1$ ). The applied voltage did not affect the chloride  
 237 concentration in the first layer as well. In very few cases, the electrode in the downstream cell  
 238 presented oxidation which required an immediate cleaning to avoid an increase of the resistance  
 239 and a drop in the electrical potential.



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Figure 7. Maximal current measurements according to the chloride concentration at the first grinded layer

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**Maximal temperature.** The current variation did not significantly increase the temperature during the migration tests (Figure 8). Temperature was in between 20 °C and 30 °C which does not involve Joule effects and is much lower than the limit of acceptability (90 °C) of ASTM C1202. These results are in accordance with the expected results for the proposed modifications of the test. A lower electrical field and an augmentation of the volume of the solution allows to keep the temperature at a stable level. For some samples with high-fiber segregation, extensive fiber corrosion was observed. In those cases, the migration test was stopped when the current exceeded 80 mA corresponding to an increase of temperature up to 55 °C. Those last results are not considered in this work.



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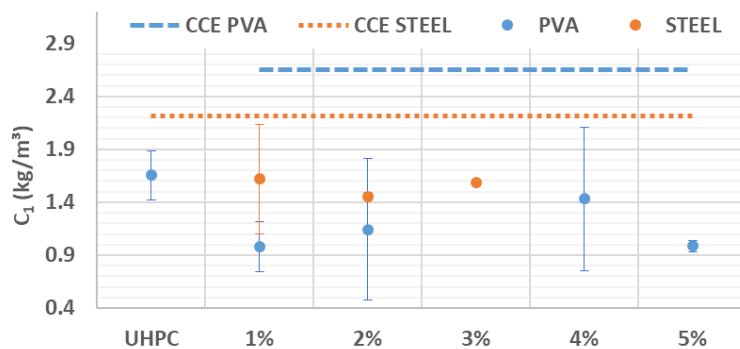
Figure 8. Temperature vs current measurements

254 **3.2 Accelerated migration tests results**

255 **Chloride concentration at the surface sample.** With the chloride concentration in the  
256 upstream cell of  $C_0 = 500 \text{ mol/m}^3$ , the maximal chloride mass concentration that could penetrate  
257 the concrete pore solution, noted  $CCE_{max}$  ( $\text{kg/m}^3$ ), is:

$$CCE_{max} = C_0 \cdot M_{Cl} \cdot V_{w/c} \quad (4)$$

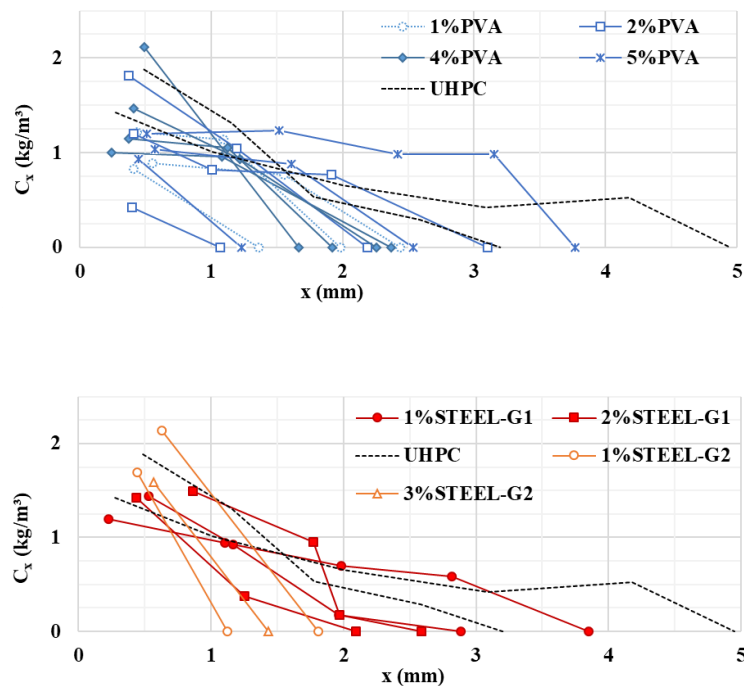
258 where  $M_{Cl}$  (35,453 g/mol) is the chloride molar mass and  $V_{w/c}$  ( $\text{m}^3/\text{m}^3_{\text{concrete}}$ ) is the volume of  
259 water necessary to cast one cubic meter of concrete.  $CCE_{max}$  does not consider hydration  
260 reactions during the concrete cure. This calculated upper bound allows to verify the consistency  
261 of the results and a possible sample cracking during the migration test. Indeed, all samples with  
262 chloride mass concentration higher than the maximal value  $CCE_{max}$  have presented signs of  
263 cracking. Chloride mass concentrations measured for the first grinded layer (Figure 9) were  
264 lower than the maximal value ( $CCE_{max}$ ). The samples which results are presented in this work  
265 did not show any sign of cracking.



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267 Figure 9. Chloride mass concentration at the first grinded layer

268 **Chloride concentration profiles** presented in a first approach gross results of the chloride  
269 diffusion in the different materials. The maximum depth reach by chloride is about 5 mm for  
270 the UHPC samples (Figure 10). As for UHPFRC, the maximum depth reach is 3.8mm (STEEL-

271 G1). The two different kinds of fiber (STEEL and PVA) show similar chloride profile except  
 272 for STEEL-G1 which is slightly higher. STEEL-G2 samples have a higher chloride  
 273 concentration in the first layer but the penetration depths are lower than the STEEL-G1 samples.  
 274 The highest chloride concentrations in the first layer grinded are 1.9 kg/m<sup>3</sup> for the UHPC and  
 275 2.1 kg/m<sup>3</sup> for the UHPFRC.



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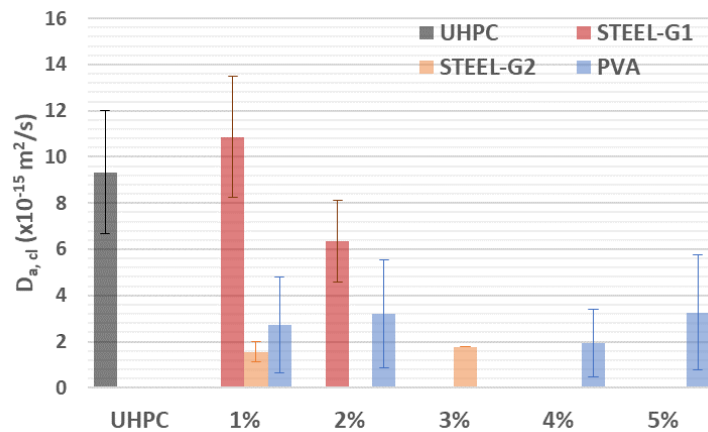
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278 Figure 10. Chloride profile for UHPC, UHPFRC-PVA and UHPFRC-STEEL migration  
 279 tests

280 **Apparent chloride diffusion coefficient  $D_{a,Cl}$  according to the fiber volume.** The average  
 281 diffusion coefficient was calculated using equation (1) for UHPC, UHPFRC-PVA and  
 282 UHPFRC-STEEL with maximum and minimum values (Figure 11). The quantity of fibers in  
 283 the sample does not seem to significantly affect the measured diffusion according to the range  
 284 of the results. Then, median apparent diffusion coefficients and their standard deviations were  
 285 calculated for the four types of materials regardless the fiber volume (Figure 12). That allows

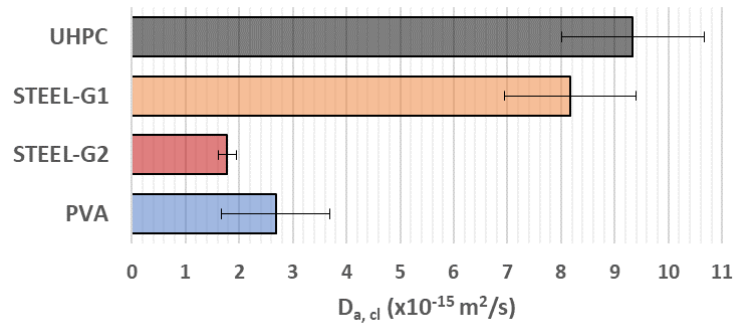


286 to have enough samples for each material to realize a statistical analysis and leads to three  
 287 following conclusions. (i) Considering the water porosity obtain for the different material  
 288 (Table 2), UHPC has one of the lowest porosities. The expected diffusion coefficient for UHPC  
 289 should then be inferior or equal to UHPFRC ones, however, the results presented revealed that  
 290 chloride migration through UHPC is around five times higher than in STEEL-G2 and PVA.  
 291 That clearly demonstrates the influence of fibers on the chloride diffusion  $D_{a,Cl}$  and the  
 292 importance to consider fibers in durability analyses. Nevertheless, that is not possible to observe  
 293 a difference between PVA and steel fibers in regard of the uncertainties. (ii) Pore size  
 294 distribution results with MIP also revealed additional capillarity pores for STEEL-G1 samples  
 295 (Figure 4) that could accelerate the chloride diffusion and explained a higher  $D_{a,Cl}$  than STEEL-  
 296 G2 ones. The casting method for steel fiber samples hence influence chloride diffusion.



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298 Figure 11. Chloride apparent diffusion coefficient calculated for the different samples



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Figure 12. Median apparent diffusion coefficient calculated and its standard deviation according to the type of material

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Furthermore, all the diffusion coefficients obtained are in the range of  $10^{-14}$  to  $10^{-15} \text{ m}^2/\text{s}$  which is very low compared to diffusion coefficients for ordinary concrete, about  $10^{-11}$  to  $10^{-12} \text{ m}^2/\text{s}$  with an usual w/c ratio (0.35-0.60) [37]. As stated from Table 1, the results from different studies do not highlight significant difference for material with and without fibers according to the ASTM C1202 standard. However, other studies (Table 1) reveals some differences with and without fiber samples by natural diffusion tests. The results from this work confirm the assumption that the standard procedure from ASTM C1202 is insufficient to characterize the material with fiber (UHPCFRC) since the presence of fiber does influence the durability of the material. Moreover, the diffusion coefficient of UHPCFRC obtain from this work are in accordance with the results from Tanaka [27] for a 5-year chloride exposure. Therefore, this work clearly demonstrates that the proposed procedure test yields accurate diffusion coefficients in a shorter time: 28-days accelerated migration against 5-year chloride exposure.

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#### 4. Conclusions

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The aim of this work was to propose an alternative test method to ASTM C1202 in order to (i) better characterize the resistance of chloride ion penetration in UHPCFRC; and (ii) compare

318 the resistance of chloride ion penetration between UHPC and UHPFRC with two different types  
319 of fibers and volume content. The effect of different preparation methods for samples with steel  
320 fibers was also considered. The proposed test modifications for the accelerated migration test  
321 consisted of (i) increasing the solution volume to reduce heating issues and (ii) the electrical  
322 field applied to accelerate the time test; (iii) reducing the sample thickness. The resistivity of  
323 the different samples is similar despite variation in their length.

324 All samples had a chloride mass concentration lower than the maximal value expected  
325 ( $CCE_{max}$ ) except for those presenting cracks where the chloride concentration was higher.

326 The procedure for the water porosity was also adapted to obtain more accurate results for  
327 UHPC and UHPFRC samples because of their very low porosity.

328 Based on the presented results, the following conclusions can be drawn:

- 329 • The diffusion results clearly demonstrate that fibers slow down the chloride diffusion in  
330 UHPFRC samples.
- 331 • The casting method for UHPFRC with steel fibers (G1 and G2) has also an influence on  
332 the chloride diffusion.
- 333 • The results from this work are in accordance with the diffusion coefficient obtained with  
334 natural diffusion tests from the literature, however, they were obtained in a shorter time  
335 which confirms the accuracy of the proposed accelerated test method.

336 Further adaptation on the test set-up is so needed to avoid the issue of corrosion and cracks  
337 of steel samples with high fibers volume content. Due to segregation, the steel fiber  
338 concentration can be higher in some samples. That leads to a rise of temperature and current  
339 during testing. An investigation for a more appropriate method is necessary to obtain an

340 effective diffusion coefficient and distinguish the influence of the material geometry, the fibers  
341 influence and the chloride chemical adsorption.

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