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RESEARCH OBJECTIVES AND SUMMARY OF RESEARCH

Our interests center on the structure and interactions of atoms and molecules, and on their interaction with the radiation field. Our methods involve colliding beam scattering spectroscopy, atomic and molecular beam resonance spectroscopy, highprecision maser techniques and, more recently, optical fluorescence spectroscopy.

1. Hydrogen Maser Studies

We have measured the effect of nuclear mass on the magnetic moment of the bound electron by comparing the electronic g-factor in hydrogen and deuterium. Our experiments achieved a fractional accuracy of 3×10^{-11} , and provide the first accurate verification of the quantum electrodynamic theory of the bound electron in an external field.

2. Production of a New Species of Molecule

We have succeeded in producing a molecular beam of the paramagnetic van der Waals molecule CsHg. Binding is provided through weak polarization forces, roughly 100 times smaller than chemical forces. CsHg represents only one of a large class of molecules which can be created, and which can yield information on atomic interaction mechanisms that cannot otherwise be observed.

3. Spin Exchange Scattering

We have carried out a detailed determination of the potential for an alkali and a molecule with nonzero electron spin. By measuring the spin dependence of the differential scattering cross section, we have been able to elucidate both the long- and short-range behavior of the K-O₂ system. Small-angle scattering results appear to

be well described by elastic scattering theory, a rather unexpected result.

4. Excited State Interactions

During the past year work was started on a program to measure coherence effects in excited atomic systems, and to study interactions between excited atoms. The initial effort is directed toward the production of nonradiating excited dimers.

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A. NUCLEAR MASS CORRECTION TO THE BOUND ELECTRON g-FACTOR

There has been remarkable agreement between precise quantum electrodynamic (QED) calculations of fine and hyperfine structure in simple systems and experimental measurements.¹ Recent theoretical work has extended high-order QED calculations to hydrogenic systems in external fields for which precise experimental evidence is lacking. Grotch and Hegstrom² have computed corrections to the electron g-factor, g_J , in one-electron atoms in an external field to relative orders $a^2 \text{ m/M}$, $a^3 \text{ m/M}$ and $a^2(\text{m/M})^2$, including nuclear mass, and radiative and binding corrections. Comparison of their results with measurements of $G = g_J(H)/g_J(D)$ provides tests of both the nuclear mass-dependent correction terms and the reliability of quantum electrodynamic Zeeman theory. The theoretical result is

$$(G-1)_{\text{th}} = \frac{1}{2} a^2 m_e \left(\frac{1}{M_p} - \frac{1}{M_D} \right) - \left\{ a^2 m_e^2 \left(\frac{1}{M_p^2} - \frac{1}{M_D^2} \right) + 12\pi a^3 m_e \left(\frac{1}{M_p} - \frac{1}{M_D} \right) \right\}$$
$$= (7.248 - 0.026) \times 10^{-9}$$
$$= 7.222 \times 10^{-9}.$$

Previous experimental evidence is inconclusive. Hughes and Robinson³ found $(G-1)_{HR} = 7.2(12) \times 10^{-9}$, in agreement with theory, while Larson, Valberg, and Ramsey⁴ found $(G-1)_{LVR} = 9.4(14) \times 10^{-9}$, in disagreement with theory. These results are sensitive enough to test only the leading theoretical terms, and are slightly discrepant.

Our experimental result is $(G-1)_{exp} = 7.22(3) \times 10^{-9}$, which confirms the leading theoretical term and gives us some confidence in the second and third terms, although the precision is not sufficient to resolve them clearly. Our result confirms the result of Hughes and Robinson, but disagrees with that of Larson, Valberg, and Ramsey.

The major factor contributing to the improved precision of our experiment (3×10^{-11}) over that of the previous experiments $(\sim 1 \times 10^{-9})$ is that we measure the electron transition frequencies in H and D simultaneously in the same spatial region. Since their g-factor ratio is essentially independent of the field, the problem of field stability, which previously limited the accuracy, is secondary.

1. Maser Apparatus

In our experiment we use a double-mode pulsed maser that radiates simultaneously on electron spin transitions H and D in ground state at a magnetic field of 3500 G. The energy levels are shown in Fig. III-1 with the observed transitions indicated by $\pi_{\rm H}$ and $\pi_{\rm D}$.



Fig. III-1. Atomic energy levels as a function of magnetic field: (a) hydrogen; (b) deuterium.



Fig. III-2. (a) Double-mode pulsed maser. (b) Maser vacuum assembly.

Design and operating principles for the maser have been described elsewhere.^{5,6} Essentially, the maser produces a signal because of electron free-precession in a magnetic field of 3500 G. Transition frequencies are typically 9.5 GHz, and the decay time is 5 ms, corresponding to a resonance width of 70 Hz (Fig. III-2a).

The atomic beam is produced in a discharge bulb by RF dissociation of H_2 and D_2 . A standard hexapole focusing magnet state-selects the beam, which then enters a collimated storage bulb within a microwave cavity. The storage bulb is coated with DuPont FEP-120 Teflon, which permits the atoms to make approximately 10^4 adiabatic wall collisions before the electron polarization is relaxed. The entire vacuum system is pumped by a Norton-NRC Orb-ion pump, rated at 800 1/s for air at high vacuum (Fig. III-2b).

With the maser assembly resting securely on it, the entire magnet assembly floats on 4 air cushions to minimize vibration of the maser within the magnetic field. The magnet is temperature-regulated within a millidegree at ~34.5°C by two sensitive thermal regulation controls, made up of thermistor bridge sensors, high-gain amplifiers, and proportional heaters. This system maintains field stability, with fractional drift rates typically less than 1×10^{-7} /h.

The magnet has mechanical pole adjustments for parallelism and lateral alignment, and standard Rose shims. With 19 electrical shims, we are able to achieve an rms field homogeneity of $\sim 1 \times 10^{-7}$ over a spherical region 2.5 cm in diameter.

The double-mode transmission cavity is resonant at 9200 MHz and 9650 MHz, which are the H and D transition frequencies, respectively (Fig. III-3). The cavity operates in the TE_{102} and TE_{012} modes, with loaded Q's of ~4000. The modes can be tuned



Fig. III-3. Microwave cavity.

separately and can be simultaneously nearly critically coupled through simple wall irises.

2. Electronics

Because of the required frequency accuracy, all laboratory frequencies are derived in standard ways from a stable Sulzer crystal oscillator which is monitored against the NBS standard from WWVB by a VLF comparator. The oscillator is constantly maintained to an absolute accuracy of 5×10^{-11} ; this results in a final error $(\delta G/G)_{\rm osc} < 3 \times 10^{-12}$.



Fig. III-4. Frequency-generation scheme.

The frequency-generation scheme is shown in Fig. III-4, the pulsed RF switching scheme in Fig. III-5, and the microwave system in Fig. III-6a. One klystron (H) is phase-locked to a reference 9200 MHz, while the other klystron (D) is phase-locked to the first (Fig. III-6b). The atomic transitions are stimulated by "90° pulses," which enter the cavity through an under-coupled port and are generated by mixing pulsed RF power onto the microwave carriers in balanced modulators, with the resonance frequencies corresponding to the first sidebands of the respective carriers. The free-radiating signal power following the pulse is mixed with local-oscillator power from the klystrons in a microwave mixer-preamplifier circuit, thereby heterodyning the signals to intermediate frequencies corresponding (at resonance) to the respective pulsed radiofrequencies, 30.01 for H and 32.01 for D. The signals are then isolated and processed in the identical channels shown in Fig. III-7.



Fig. III-5. (a) Phase-locked oscillator. (b) RF switching networks.



Fig. III-6. (a) Microwave system. (b) Klystron phase-locking system.

Fig. III-7. Signal-detection and signal processing channels.

32 MHz REFERENCE

10... kHz

SCALER



Fig. III-8. Zero-crossing counter.

The frequencies are determined by measuring accurately the period of the residual 10-kHz signals. After filtering and zerocrossing detection, the signals (or "bursts") are counted by circuits (Fig. III-8) that generate "start" and "stop" pulses at the first and (N+1)th zero crossings, thereby defining the period of N cycles. These pulses then start

and stop scalers counting a reference 1-MHz signal. After accumulating a preset number of "bursts," the scaler counts are transferred by a teletype system to paper display and punched paper tape for later computer analysis.

3. Error Analysis

The systematic errors affecting this experiment have been analyzed thoroughly,⁷⁻⁹ and will be discussed fully as part of F. G. Walther's thesis research. The major sources of error are the inhomogeneity shift,⁷ spin-exchange pulling,⁸ the second-order Doppler shift, wall shifts⁹ and instrumental pulling in the detection system. The cavity can be tuned sufficiently well to virtually eliminate cavity pulling, and many other sources of error, which are always present, are small enough to be neglected. We remove the inhomogeneity shift by using several different collimators and extrapolating to zero collimator length. Spin-exchange pulling has been extensively studied, with corrections being made from computer integrations of the complete rate equations. Wall shifts occur because of a change in hyperfine frequency during wall collisions. These effects are well-known and lead to a shift in G.

Instrumental effects are caused primarily by mixer distortion (phase shift vs amplitude) and finite delay times in the counters, although bandpass pulling and leakage power mechanisms are also present. Instrumental pulling, however, can be removed by successively interchanging the intermediate frequencies of H and D and by using different sideband combinations of the IF carriers. All data are taken by using 4 different sideband combinations; this provides two redundant means of removing instrumental pulling. Therefore we not only have data corrected for instrumental effects but also a test of the consistency of the instrumental corrections.

In Table III-1 important corrections to the raw data are listed, with typical values given for both the total magnitude of the correction and the estimated uncertainty associated with the correction.

Although three separate sets of data, after extrapolation to remove the inhomogeneity effect, agree within error bars of approximately 2×10^{-11} , we have some evidence that other effects (presumably related to contaminated surfaces) can cause shifts of several parts in 10^{-11} . Although the presence of these effects can be

Source of Error	Magnitude of Correction	Estimated Uncertainty
Instrumental Effects	5×10^{-11}	0.5×10^{-11}
Shimming and Inhomogeneity	2	0.5
Second-Order Doppler	1	_
Wall Shift	1	0.3
Cavity Pulling	< 1	<0.3

Table III-1. Corrections to raw data.

independently determined and apparently did not affect our final data, we have allowed an additional 2×10^{-11} uncertainty for these effects to be added quadratically to our apparent error bar, which results in a final error bar of $\sim 3 \times 10^{-11}$. Since some subjective judgment is included in this estimate, it should be viewed as our best estimate of a 60% confidence interval.

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References

- 1. S. J. Brodsky, Proc. International Conference on Precision Measurement and Fundamental Constants, Gaithersberg, Maryland, 1970.
- 2. H. Grotch and R. A. Hegstrom, Phys. Rev. A 4, 59 (1971).
- 3. H. G. Robinson and W. M. Hughes, Proc. International Conference on Precision Measurement and Fundamental Constants, Gaithersberg, Maryland, 1970.
- D. J. Larson, P. A. Valberg, and N. F. Ramsey, Phys. Rev. Letters <u>23</u>, 1369 (1969).
- 5. D. Kleppner, H. M. Goldenberg, and N. F. Ramsey, Phys. Rev. 126, 603 (1962).
- 6. P. F. Winkler, D. Kleppner, T. Myint, and F. G. Walther (to appear in Phys. Rev. A).
- 7. D. Brenner, Phys. Rev. <u>185</u>, 26 (1969).
- 8. P. L. Bender, Phys. Rev. 132, 2154 (1963).
- 9. P. W. Zitzewitz, E. E. Uzgiris, and N. F. Ramsey, Rev. Sci. Instr. 41, 81 (1970).

B. MOLECULAR-BEAM STUDIES OF VAN DER WAALS MOLECULES: THE CsHg SYSTEM

We have produced and detected, for the first time, a free paramagnetic van der Waals molecule. The molecule CsHg has been produced and detected in a molecular beam. The theory of the magnetic-resonance spectrum of the $^{2}\Sigma$ molecule has been developed,

and we are searching for resonance transitions at low magnetic field. Our goal is to obtain information on atomic interactions and the structure of the weakly bound mole-cules, especially about the spin-rotation coupling constant. The formation mechanism of these molecules in an atomic beam has been described previously.¹



Fig. III-9. Schematic diagram of the apparatus.

Figure III-9 is a schematic diagram of the apparatus which was formerly used for spin-exchange scattering experiments, but has been demoted to a molecular-beam resonance machine by adding a C magnet and an RF coil. An oven with a 0.0018 in. converging nozzle, similar to one used by Gordon,² is used to produce a supersonic jet. The oven is heated by 3 independent heaters in order to prevent clogging of the nozzle. A liquid-nitrogen-cooled housing with a water-cooled radiation shield is used to pump the large amount of Cs and Hg streaming from the oven. The skimmer has been eliminated in order to prevent skimmer-jet interaction. Instead, a 0.08 in. slit is situated approximately 1 in. downstream. The oven is loaded with equal amounts (by molar weight) of Cs and Hg and is operated typically at 750°K. A beam of Mach number 14 can be obtained with considerable enhancement of the association fraction of Cs dimers and CsHg molecules over that in the source.

The beam contains Cs, Hg, Cs₂ ($^{1}\Sigma$, $^{3}\Sigma$), Hg₂, CsHg, and other polymers of Cs and Hg. As the beam passes through the Stern-Gerlach magnet, the species are deflected



Fig. III-10. Deflection profile of the beam.

according to their weight and magnetic moment. (Unlike a thermal beam, all species of the jet beam travel at the same velocity.) Figure III-10 shows a profile of the intensity vs the source position with the detector fixed at positions M. A 2×10^{5} /s maximum flux of CsHg is detected, approximately a factor of 10^{3} lower than the Cs flux.

The approximate Hamiltonian for the CsHg molecule in low magnetic field is

$$\mathbf{H} = \mathbf{a} \vec{\mathbf{I}} \cdot \vec{\mathbf{S}} + \mathbf{Y} \vec{\mathbf{N}} \cdot \vec{\mathbf{F}} - \mathbf{g}_{\mathbf{N}}^{\mu} \mathbf{N} \vec{\mathbf{N}} \cdot \vec{\mathbf{H}} - \mathbf{g}_{\mathbf{F}}^{\mu} \mathbf{B} \vec{\mathbf{F}} \cdot \vec{\mathbf{H}},$$

where a and γ are the hyperfine and spin-rotation coupling constants, μ_N is the nuclear magneton, and μ_B is the Bohr magneton. N is the rotational quantum number, and g_N and g_F are the nuclear and electronic g-factors. Figure III-11 is an energy-level diagram in low field; the possible transitions are indicated by arrows. The number of molecules in each quantum state is

$$n = \frac{n_0}{(2I+1)(2N+1)} p(n),$$

where n_0 is the initial flux, and p(n) is the probability of finding the molecule at a particular N. At a beam temperature of ~15°K, n/n_0 is ~8×10⁻⁵. Since g_N is different for each state, the fraction of the beam contributing to a given transition is very small. A factor of eight increase in intensity can be obtained by using the state N = 4. At lower fields J = N + F is a good quantum number. For a $\Delta J = 0$, $\Delta m_J = \pm 1$

molecular-beam resonance transition (solid arrow)

$$\Delta v = \frac{g_{J} \mu_{o}^{H}}{h},$$

where

$$g_J = g_F \frac{J(J+1) + F(F+1) - N(N+1)}{2J(J+1)}$$
.

The state selector selects molecules with F = 4, and for N = 4. The fact that g_J is independent of J and is equal to $g_F/2$ allows simultaneous transitions in m_J states for



Fig. III-11. Energy-level diagram and vector coupling model of CsHg. Solid arrow: $\Delta J = 0$, $\Delta m_j = 1$ transition. Dotted arrow: $\Delta J = 1$, $\Delta m_j = 0$ low-field transition.

J = 8 to J = 1. The spin-rotational constant γ can be deduced from the quadratic effect by observing the breakdown of $\vec{N} \cdot \vec{F}$ coupling as the magnetic field is increased.

An alternative way of observing the resonance is to use $\Delta J = 1$, $\Delta m_{J} = 0$

transitions (dotted arrow) at very low field so that for a transition $J = N - 4 \rightarrow J = N - 3$, we have

 $[g(J+1)-g(J)] \mu_0 Hm_1/h < \Delta_{\nu},$

where Δv is the resonance linewidth. If this condition is met, the transitions caused by all m_T states overlap. This method gives a direct measurement of γ .

Our initial search has been hindered by a large background signal caused by the inability of the magnet to provide adequate separation of all the species. Large detectors and slit widths have aggravated this problem. Work is in progress to eliminate the background by narrowing the slit and detector. The new resonance machine described in Section III-C should eliminate these problems.

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References

- F. Y. Chu, D. E. Pritchard, and D. Kleppner, "Molecular Beam Studies of van der Waals Molecules," Quarterly Progress Report No. 100, Research Laboratory of Electronics, M.I.T., January 15, 1971, pp. 20-23.
- 2. R. J. Gordon, Ph.D. Thesis, Harvard University, 1969 (unpublished).

C. MOLECULAR-BEAM STUDIES OF VAN DER WAALS MOLECULES: NEW APPARATUS

In order to provide a more powerful and flexible facility for the study of van der Waals molecules, a new molecular-beam apparatus has been designed and is being built. Its basic structure is that of the traditional molecular-beam magnetic resonance machine: a source chamber containing the beam oven; a separately pumped velocity selector chamber; a magnet chamber with its own pump, containing two deflecting magnets ("A" and "B") and a homogeneous ("C") magnet with RF coil. A separate chamber containing a hot-wire detector is attached downstream of the magnet chamber.

The new machine will be superior to the present modified scattering apparatus in several respects. The high-temperature jet-beam oven will be provided with a gas inlet, so that, for example, alkali-rare gas van der Waals molecules can be studied. The increased pumping speed of the new apparatus, approximately 4 times that of the old, permits use of noncondensable gases.

The oven will have a separately heated skimmer, at an adjustable distance from the oven nozzle, so that beam conditions can be optimized. A water-cooled baffle will match the skimmer, minimizing scattering of condensable atoms into the beam.

The deflecting magnets are twice as long (10 in.) as in the old apparatus, and are designed to produce a field of approximately 22,000 G, and a gradient of \sim 40,000 G/cm. The large deflection produced by these magnets will help eliminate the background of unwanted atoms.

The C magnet in the new apparatus will be longer and more homogeneous than that in the present machine, with a design homogeneity of better than 1 part in 10^4 . In order to narrow the resonance line further, all of the magnets will be operated from regulated power supplies, with the C magnet field-regulated by a nuclear magnetic resonance probe.

F. Y. Chu, E. Mattison, J. Apt,D. E. Pritchard, D. Kleppner

D. SPIN-EXCHANGE SCATTERING

We have made measurements of the spin-dependent differential cross section in scattering of K from O_2 at thermal energies, and have obtained from this the probability of spin exchange during a collision, $P_{ex}(\theta)$, and the spin summed differential cross section, $\sigma_{sum}(\theta)$, both as functions of energy and angle. The data were then analyzed with the aim of determining the intermolecular forces.

For elastic scattering, the interaction is described by two potentials, depending on the spin state of the KO_2 "molecule," doublet or quartet. The long-distance van der Waals attraction is expected to be spin-independent; and at small intermolecular distances both states are repulsive. In the intermediate range of ~8-15 a₀, electron cloud overlap is expected to cause repulsion in the quartet state and increased attraction in the doublet state, so that the average of the two potentials still follows the van der Waals potential.¹

In the following equations, σ_{sum} is a weighted average of the scattering from the two potentials and P_{ex} depends on the interference between the two scattering amplitudes

$$\begin{split} \sigma_{\mathrm{sum}}(\theta) &= \frac{1}{6} \left[4 \sigma_4(\theta) + 2 \sigma_2(\theta) \right] \\ \mathrm{P}_{\mathrm{ex}}(\theta) &= \frac{4}{27} \; \frac{\left| \mathrm{f}_4(\theta) - \mathrm{f}_2(\theta) \right|^2}{\sigma_{\mathrm{sum}}(\theta)}, \end{split}$$

where f_2 , f_4 , σ_2 , σ_4 are the scattering amplitudes and cross sections for doublet and quartet scattering.

The cross sections were calculated from the usual partial wave expansion by using a 7-parameter model for the derivative of the phase shift with respect to the angular momentum. For the derivative of the phase shift we chose a model that is closely related to the deflection function. This model can easily be varied to produce a desired change in the cross section. In the semiclassical approximation,

$$\chi(b) = 2 \left. \frac{\partial \eta}{\partial \ell} \right|_{\ell = kb},$$

where $\chi(b)$ is the classical angle of deflection as a function of the impact parameter b, $\eta(\ell)$ is the phase shift of the ℓ^{th} partial wave, and k is the wave number, k = $(2mE)^{1/2}/\hbar$. The potential can be calculated from the deflection function, although we have not carried out that calculation.²

For small-angle scattering, σ_{sum} is dominated by σ_4 so that we need only consider



Fig. III-12. σ_{sum} vs E0 and b vs E0. Solid line in σ_{sum} plot is calculated from a given deflection function. Crosses are data points. Vertical scale of σ_{sum} is arbitrary and chosen so that the data points approximately coincide with the calculated cross section.

the quartet scattering in describing the features of σ_{sum} . The position of the broad peak in σ_{sum} corresponds roughly to the maximum angle of classical scattering caused by the attractive branch of the deflection function, the rainbow angle (Fig. III-12). The oscillations are caused by diffraction effects, and their period near a particular position gives a direct measure of the distance between the impact parameters corresponding to that angle of scattering.

Figure III-12 shows the experimental and calculated sum cross sections and the best fit to the deflection function. These selected data exhibit the clearest structure, but other data taken at different energies and out to larger angles can be reproduced equally well by using the fact that $E_{\chi}(E, b)$ is independent of the energy, E, to find $\chi(E, b)$ at the different energies. (The deflection for a given impact parameter can be shown to be inversely proportional to the energy for small angles of deflection.³) The experimental and calculated probability of exchange are given in Fig. III-13. It should be noted that the rapid oscillatory structure which was present in the calculation is averaged away in the data at the larger angles chiefly because these data were averaged over a range of velocities.

The requirement that the calculations reproduce σ_{sum} within experimental error fixes the rainbow angle within 10%, the impact parameter, b_g , at which the deflection function crosses $\chi = 0$ within 2%, and the phase at b_{σ} (or, equivalently, the integral of



Fig. III-13. P_{ex} vs E0. Solid line is calculated. Crosses are data points.

the deflection function from b_g to infinity) within 10%. But σ_{sum} is relatively insensitive to the shape of the deflection function near its minimum and to the long- and short-range interactions for $b > 15 a_0$ or $b < 10 a_0$.

To fit P_{ex} , another constraint must be put on the calculated deflection function. For small scattering angles, the requirement¹ that the average of the doublet and quartet potentials be the van der Waals potential implies that the same is true for the deflection functions. The value of $P_{ex}(\theta)$ with its oscillations removed by averaging determines the difference between the deflection functions and therefore the difference between each one and the van der Waals potential for angles less than the rainbow.

With this information about the shape of the deflection function near its minimum, and the information from σ_{sum} , the calculated deflection function should be accurate within 20% between 10 a and 15 a. At the minimum it should be accurate to 10% and outside this range to 50%.

There are two important results of this work. The first is the determination of the quartet and doublet deflection functions for the KO_2 system; these may be inverted to give the long-range behavior of the intermolecular potentials for KO_2 . This is, as far as we know, the first detailed determination of the potential between an alkali and a molecule with nonzero electron spin. The second result is that the small-angle scattering in this system can be described fully by simple elastic scattering. This is somewhat surprising, since we know⁵ that the wide-angle scattering is dominated by complex formation: $K + O_2 \rightarrow KO_2^* \rightarrow K + O_2$, where KO_2^* indicates an autodissociating state of the KO₂ molecule.

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References

- 1. D. E. Pritchard and F. Y. Chu, Phys. Rev. A 2, 1932 (1970).
- 2. G. M. Carter and D. E. Pritchard (submitted to Phys. Rev.).
- 3. W. H. Miller and H. Kruger, Phys. Rev. Letters, <u>28A</u>, 165 (1968).
- 4. F. T. Smith, R. P. Marchi, and K. G. Dedrick, Phys. Rev. 150, 79 (1966).
- 5. D. O. Ham, Ph.D. Thesis, Department of Chemistry, M.I.T., February 1968 (unpublished).

E. NONRADIATING EXCITED COMPLEXES

The long-range interaction between two atoms is generally the second-order van der Waals interaction $-C/r^6$, but the interaction between identical atoms, one of which is excited, is first-order, D/r^3 . This well-known effect, resonant dispersion interaction, plays an important role in the long-range potential of excited dimers. These excited complexes possess interesting radiative properties; essentially, they provide the simplest example of a two-body coherent system. Therefore the complexes are an

ideal subject for use in investigating the properties of simple super-radiant systems.

The origin of resonant dispersion energy can be seen by the following simple argument. Consider two identical L-C circuits with resonant frequency $\omega_0 = (LC)^{-1/2}$. The energy of each oscillator is $(n+1/2)\hbar\omega_0$, where n is the photon occupation number. If the oscillators are in the ground state, the energy is

$$\mathbf{E}_{\mathrm{O}} = \frac{1}{2}\,\hbar\boldsymbol{\omega}_{\mathrm{O}} + \frac{1}{2}\,\hbar\boldsymbol{\omega}_{\mathrm{O}} = \hbar\boldsymbol{\omega}_{\mathrm{O}}.$$

Now let the oscillators interact by electric coupling of the fringe fields of the capacitor. If the coupling coefficient is κ , then the resonant mode is split into two modes having frequencies

$$\omega'_{0} = \omega_{0}\sqrt{1 \pm \kappa}$$

The energy is

$$\mathbf{E}_{\mathbf{O}}^{\prime} = \frac{1}{2} \, \hbar \boldsymbol{\omega}_{\mathbf{O}} \Big\{ \sqrt{1+\kappa} + \sqrt{1-\kappa} \Big\}.$$

For $\kappa \ll 1$, the change in energy because of interaction is

$$\Delta \mathbf{E}_{_{\mathrm{O}}} = \mathbf{E}_{_{\mathrm{O}}}' - \mathbf{E}_{_{\mathrm{O}}} = -\frac{1}{4} \hbar \omega_{_{\mathrm{O}}} \kappa^2.$$

The capacitors couple like two dipoles, and $\kappa \approx 1/R^3$, where R is the separation of the circuits. Hence $\Delta E_o = -C/R^6$. This is the lumped oscillation analog to the van der Waals attraction between atoms.

Next, assume that one of the oscillators is excited with one photon. Then

$$\mathbf{E}_{1}^{\prime} = \frac{3}{2} \, \, \tilde{\hbar} \omega_{0}^{\prime} \sqrt{1 \pm \kappa} + \frac{1}{2} \, \, \tilde{\hbar} \omega_{0}^{\prime} \sqrt{1 \mp \kappa}.$$

The interaction energy for $\kappa \ll 1$ is

$$\mathbf{E}_{1}' = \frac{1}{2} \, \tilde{\hbar} \boldsymbol{\omega}_{O} \left\{ 4 \pm \kappa - \kappa^{2} \right\}.$$

The change of energy, to lowest order of κ , is

$$\Delta E_1 = \pm \frac{1}{2} \hbar \omega_0 \kappa = \pm D/R^3,$$

where D is a constant. The sign of the interaction is positive if the system is in a symmetric mode, negative if the mode is antisymmetric.

If the photon is localized on one oscillator, the system has mixed symmetry and the

first-order interaction is zero. A system prepared in a state of pure symmetry, however, has a large, first-order, interaction.

The lifetime of an excited molecule bound in the symmetric state by first-order dispersion forces exhibits the shortening associated with a super-radiant system; in this case the lifetime is one half the free-atom lifetime. The antisymmetric state, however, behaves quite differently; it exhibits what might be called "super nonradiance." Essentially, the radiating dipoles of the two atoms are out of phase, and the total dipole is zero.

The radiation theory for two identical atoms and one photon has been attacked theoretically by Stephen¹ and others,²⁻⁵ but until now these systems have not been investigated experimentally.

We are attempting to produce nonradiant excited complexes of two sodium atoms. The complexes will be formed by exciting the forbidden transition, and observed by the delayed fluorescence rate. The initial work is being carried out in a gas cell, although the use of high-intensity atomic beams would have some advantages. We have constructed a flashlamp pumped dye laser and much of the photon-counting equipment.

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References

- 1. M. J. Stephen, J. Chem. Phys. 40, 699 (1964).
- 2. E. A. Power, J. Chem. Phys. 46, 4297 (1967).
- 3. P. R. Fontana and D. Hearn, Phys. Rev. Letters 19, 481 (1967).
- 4. R. H. Lehmberg, Phys. Rev. A 2, 889 (1970).
- 5. V. Ernst and P. Stehle, Phys. Rev. <u>176</u>, 1456 (1968).