

High-precision mass measurements for reliable nuclear astrophysics calculations

A. Herlert^{*a}, S. Baruah^b, K. Blaum^{cd}, M. Breitenfeldt^b, P. Delahaye^a, M. Dworschak^d,
S. George^{cd}, C. Guénaut^e[†], U. Hager^f, F. Herfurth^d, A. Kellerbauer^{ab}[‡], H.-J. Kluge^{dg},
D. Lunney^e, R. Savreux^d, S. Schwarz^h, L. Schweikhard^b and C. Yazidjian^d

^hNSCL, Michigan State University, East Lansing, MI-48824-1321, USA

High-precision mass measurements as performed at the Penning trap mass spectrometer ISOLTRAP at ISOLDE/CERN are an important contribution to the investigation of nuclear astrophysics processes. Accurate nuclear masses with 0.1 ppm relative mass uncertainty are required for tests of models which are used to predict mass values of nuclides far from the valley of stability. Especially for the determination of the nucleosynthesis pathways, e.g. the rp-process or the r-process, a large number of nuclear masses need to be precisely and accurately known. In addition to a radioactive ion beam facility that delivers a large variety of short-lived nuclides with sufficient yield, a sophisticated experimental setup for precise mass measurements is required. An overview of the results from the mass spectrometer ISOLTRAP is given and the limits and possibilities are described.

International Symposium on Nuclear Astrophysics — Nuclei in the Cosmos — IX June 25-30 2006 CERN, Geneva, Switzerland

*Speaker.

^aPhysics Department, CERN, 1211 Geneva 23, Switzerland

^bInstitut für Physik, Ernst-Moritz-Arndt-Universität, 17487 Greifswald, Germany

^cInstitut für Physik, Johannes Gutenberg-Universität, 55099 Mainz, Germany

^dGSI, Planckstr. 1, 64291 Darmstadt, Germany

^eCSNSM-IN2P3-CNRS, 91405 Orsay-Campus, France

^f Department of Physics, University of Jyväskylä, P.O. Box 35 (YFL), 40014 Jyväskylä, Finland

⁸ Fakultät für Physik und Astronomie, Ruprecht-Karls-Universität, 69120 Heidelberg, Germany

E-mail: alexander.herlert@cern.ch

[†]present address: NSCL, Michigan State University, East Lansing, MI-48824-1321, USA [‡]present address: MPI für Kernphysik, Saupfercheckweg 1, 69117 Heidelberg, Germany

1. Introduction

Nuclear masses play an important role in a large number of physical systems. Especially in the case of astrophysics, nuclear parameters of radionuclides far from the valley of stability are required. For the investigation of the nucleosynthesis pathways, masses with a relative uncertainty of the order of 10^{-7} are needed [1, 2]. However, for a large number of nuclides the present mass uncertainty is above 10 keV, and especially for short-lived nuclides the relative mass uncertainty rapidly rises the farther out the nuclides lie (see Fig. 1). Short half-lives and low production yields makes their investigation an experimental challenge.

Mass models either provide estimates on unknown masses by a global approach, or try to predict the mass values locally [3]. In order to test these mass models, accurate experimental data are required. Various experiments at radioactive ion beam facilities aim at the investigation and precise determination of the masses of short-lived nuclides [4]. The applied techniques range from traditional TOF mass spectrometers to storage rings and Penning trap mass spectrometers [5]. The latter have shown to allow mass measurements for stable ions with a relative uncertainty down to 10^{-11} [6] and for radionuclides of the order of or less than 10^{-8} [7–10].

The Penning trap mass spectrometer ISOLTRAP has already contributed to a number of astrophysical problems by providing accurate mass values. In the case of ⁷²Kr, the identification of a rp-process waiting point was achieved [11] and for ²²Mg a puzzling mass contradiction for the NeNa-cycle in supernovae was resolved [12]. These examples show the need for accurate mass values. In the following, recent investigations on the mass values of neutron-rich nuclides will be presented and the possibilities, challenges, and limitations discussed.



Figure 1: Relative mass uncertainty $\Delta m/m$ of nuclides in color code. Estimated mass values are omitted. The dashed lines indicate the neutron and proton shell closures. Data taken from [13].





Figure 2: Schematic overview of the ISOLTRAP experiment. The inset shows a cyclotron resonance of 81 Zn⁺, where the solid line is a fit of the known line shape to the data points [14].

2. Experimental setup and procedure

The Penning trap mass spectrometer ISOLTRAP [15, 16] is the prototype experiment for the mass determination on radionuclides at radioactive ion beam facilities using Penning traps. Various additional experiments are meanwhile in operation (CPT [17], JYFLTRAP [18], LEBIT [19], SHIPTRAP [20]) or under construction (TITAN [21], MLLTRAP [22]). A schematic overview of the experimental setup is shown in Fig. 2. The radioactive ions are delivered from one of the two target stations at ISOLDE [23]. The short-lived nuclides are produced by bombarding a fixed target with a 1.4-GeV proton pulse and subsequent spallation, fragmentation, and fission reactions. The target container is heated and the radioactive particles diffuse into an ion source, where ions are formed by either surface ionization, resonant laser ionization, or a plasma. After passing separator magnets, the ions are sent as a 60-keV continuous beam to the ISOLTRAP setup.

By use of a linear radiofrequency (rf) quadrupole structure [24] filled with helium buffer gas, mounted inside a 60-kV high voltage cage, the continuous beam is stopped, accumulated, and bunched as well as cooled for further injection into the first Penning trap. This cylindrically shaped trap also employs helium buffer gas, and with a combination of a quadrupolar rf excitation and gas collisions only the radionuclide ions of interest are mass-selectively centered [25]. Thus unwanted isobaric contaminant ions can be removed with a resolving power of $R = 10^4 - 10^5$, depending on the respective mass difference and the excitation duration [26].



Figure 3: Left: Relative mass uncertainty of published mass values obtained at ISOLTRAP. Right: Maximum yields obtained at ISOLDE during operation at the Synchrotron Cyclotron (SC) and the Proton Synchrotron Booster (PSB) at CERN [27].

The cooled, isobarically pure ion bunch is transferred to the hyperbolically shaped precision Penning trap, where the cyclotron frequency $v_c = qB/(2\pi m)$ of the stored ions with mass m and charge q moving in a magnetic field B is probed with a time-of-flight (TOF) cyclotron resonance detection technique [15, 28]. In order to deduce the mass of the investigated radionuclides, the magnetic field strength B needs to be determined at the same level of precision, by use of a cyclotronfrequency measurement of a well-known stable nuclide ion, either delivered from the ISOLTRAP alkali ion source or the ISOLDE target. By calculating the frequency ratio, the mass ratio of the ions is obtained.

3. Limitations and possibilities for mass measurements at ISOLDE

3.1 Radionuclide half-lives and production yields

In principle, all nuclides delivered by ISOLDE can be investigated with ISOLTRAP. However, due to the preparation steps and the excitation time required for the frequency determination, the half-life is a major constraint. The nuclide with the shortest half-life investigated so far was ⁷⁴Rb with $T_{1/2} \approx 65$ ms [29].

Another important factor is the production yield in the target/ion-source combination. Figure 3 (right) shows an overview of observed maximum production yields at ISOLDE. In addition to the production yield, the efficiency of transfer and stopping at ISOLTRAP has to be taken into account. The transfer from the separator magnets, through the ISOLDE central beam line to the ISOLTRAP setup can be tuned to an efficiency above 80%. The injection and stopping as well as cooling in the RFQ buncher has an efficiency of up to 10% [24]. Although the transfer from the buncher to the final TOF detector is thoroughly tuned before each on-line run, the total efficiency of the ISOLTRAP spectrometer is about 1-5%, i.e. out of 100 ions released on average in the target/ion-source only one to five ions will be detected. This assumes that the half-life of the radionuclides is larger than the duration of the experimental cycle including the excitation time for



Figure 4: Possible pathways of the r-process (solid and dashed lines) as given in [30]. The blue squares denote nuclides for which mass values, obtained with ISOLTRAP, have been published. Red squares indicate nuclides for which data have been recorded that are currently under investigation or being published. Black squares represent stable nuclides.

the frequency determination. Ion loss due to short half-lives in the order of 100 ms and less poses another experimental challenge.

Note that some ions cannot be delivered at all, since their elemental properties prevent the diffusion out of, or ionization at the target. For example ions of refractory elements like vanadium are at the moment not available at ISOLDE. Nevertheless, new ISOLDE beams might become available in the future, if those elements are addressed by use of ion chemistry, i.e. by the production of specific molecules in the target that can easily diffuse out of the target (for example fluorides, oxides, or sulfides).

3.2 Isobaric and isomeric contaminants

Depending on the target material and the ion source in use, various isobaric and isomeric contaminating ions may be present in the delivered radioactive ion beam. Therefore, target development is an important issue at ISOLDE. For example, a new quartz transfer-line target was designed and built [31], which allows to freeze out alkali contaminants like rubidium ions in the transfer from the target container to the ion source. This kind of target was recently used to produce a very clean beam of neutron-rich zinc isotopes for mass measurements at ISOLTRAP.

Thus it was possible to measure the cyclotron frequencies of $^{71-81}Zn^+$, for which the expected contaminants, especially rubidium isotopes, were reduced to such small amounts that they could be handled and removed in the ISOLTRAP preparation trap. With a yield of 100–1000 ions per second for $^{81}Zn^+$, it was possible to observe on the average one ion in each experimental cycle, enough to obtain several cyclotron resonances for $^{81}Zn^+$ as shown in the inset of Fig. 2.

Another way to suppress or remove isobaric contaminants from the radioactive ion beam is the application of ion-molecule chemistry. Element-specific reactions can be used to produce a

TT.	1	• •		,	C	1. 11	1	. 1	•	1 1
HI	γh_	nracision	macc	monsuromonts	tor	roliable	nucloar	· actronhy	C10C 0	alculations
111)	$< n^{-}$	precision	mass	measurements	101	renuore	nacieur	usitophy	sics ci	accatations

element	mass numbers	references
Ne	18,19,23,24	[32]
Na	21,22	[12]
Mg	22	[12]
Ar	32-34,42-45	[7, 33, 34]
Cr	56,57	[35]
Cu	68,68m,70,70m,70n	[36, 37]
Kr	72-78,80,82,84,86,88-95	[11, 30, 34, 38]
Rb	74-81,82m,83,84,86-89,90m,91-94	[34, 39, 40]
Sr	76-83,86-88,91-95	[39-42]
Sn	124,129,130,130m,131,132	[42]
Xe	114–123,126,136	[43, 44]
Cs	117-122,122m,123-132,134-142	[45, 46]
Ba	123–128,131,138–144	[46, 47]
Ce	132–134	[48]
Pr	133–137	[47]
Nd	130,132,134–138	[47, 49]
Pm	136–139,140m,141,143	[47]
Sm	136–141,141m,142,143	[47]
Eu	139,141,142m,143–149,151,153	[47 - 49]
Dy	148,149,154	[47 - 49]
Но	150	[47]
Tm	165	[48]
Yb	158–164	[48]
Hg	179–188,190–197,185m,187m,189m,191m,193m,197m,200	[50]
Pb	187,187m,196,198	[50, 51]
Bi	197	[50]
Ро	198	[50]
At	203	[50]
Fr	209–212,221,222,230	[52, 53]
Ra	226,229–232	[46, 52, 53]

Table 1: Unstable and stable nuclides measured with ISOLTRAP.

clean molecular sideband. As an example, the investigation of neutron-rich tin nuclides, $^{128-134}$ Sn, became possible when the radionuclides of interest were shifted out of the mass range with Cs contamination, by adding an enriched sample of 34 S to the target in order to form tin-sulfide molecules (SnS).

A further challenge are isomeric states with half-lives of more than 10 ms that are also produced in the target and may contaminate the radionuclide beam. While the isomeric states can usually not be resolved and isolated by the separator magnets of ISOLDE and even by the preparation trap of ISOLTRAP, the precision Penning trap allows isomeric cleaning with a resolving power

element	mass numbers	year
Ne	17	2004
Al	26,27	2006
Κ	35–38,43–46	2004
Ca	38,39	2006
Mn	56–57,58m,59–63	2003,2006
Fe	61–63	2006
Ni	57,60,64–69	2003
Cu	65–67,69,73,74,76	2003
Zn	71m,72–81	2005
Ga	63–65,69–78	2003
Se	70–73	2002
Br	72–75	2002
Ag	98–101,103,112,114,116,118m,120,121	2002,2006
Cd	114,120,124,126	2005,2006
Sn	127,128m,133,134	2004
Cs	145,147	2002
Ba	130	2003
Tl	181,183,186m,187,196	2002
Pb	197,208	2002
Bi	190–196,215,216	2002
Fr	203,205,229	2002
Ra	214	2002

Table 2: Unstable and stable nuclides for which data have been recorded with ISOLTRAP (year of on-line run given) and data analysis is ongoing or data will be published soon.

of the order of $R \approx 10^7$. Thus, either pure isomeric states or the ground state can be prepared as was demonstrated for ⁶⁸Cu [36]. Note that if an ion source with resonant laser ionization (RILIS) is applied, narrow-band laser excitation can state-selectively populate single isomeric states or significantly suppress the unwanted ones, as demonstrated, e.g., in the case of ⁷⁰Cu [37].

4. Recent investigations

The masses of over 300 nuclides have already been measured with ISOLTRAP. An overview is given in tables 1 and 2. For published frequency-ratio values, the corresponding relative mass uncertainties are shown in Fig. 3 (left). While in the beginning the systematic uncertainties were not known and therefore a conservative estimation of the lower limit of the relative mass uncertainty of 1×10^{-7} was applied, the systematic investigation of the setup via the mass measurement of carbon clusters was a major improvement [54]. Carbon clusters consisting of a multiple of ¹²C atoms are the ideal reference masses, because the atomic mass unit is defined as 1/12 of the mass of ¹²C [55]. A systematic study showed the limit of the accuracy of ISOLTRAP at $\Delta m/m = 8 \times 10^{-9}$ [56]. This is added as a systematic uncertainty to the statistical uncertainty from the frequency determination.

Recent results regularly reach relative mass uncertainties of the order of 1×10^{-8} , some of those are shown in Fig. 3 (left). The experimental data of a large number of nuclides is presently being analyzed or the results are close to publication. The respective nuclides are given in table 2. These include neutron-rich zinc and tin radionuclides with the nuclides ⁸⁰Zn and ¹³²Sn, which are relevant for the study of the r-process, as illustrated in Fig. 4. In both cases the studied nuclides lie within the possible path of the r-process, and large deviations from the literature values [13] have been found. The influence on astrophysical calculations is currently investigated.

Two previous results for mass measurements at ISOLTRAP have already contributed to the study of the nucleosynthesis in stellar explosions and the rp-process. In the case of ²²Mg, the former literature mass value was in conflict with results from ²¹Na(p, γ)²²Mg reaction rate measurements [57]. Indeed, the previous mass excess value -397.0(1.3) keV [58] was found to significantly deviate from the new value -399.92(27) keV [12] and a ²¹Na(p, γ)²²Mg resonance energy of 209.7(1.2) keV into the 5714-keV 2⁺ state of ²²Mg was deduced. This independent determination of the resonance energy gives further input to the comparison of the observed γ radiation from oxygen-neon nova explosions with the yield of ²²Ne as expected from nova models, where ²²Ne is produced in the so called NeNa-cycle [57].

The second example is the mass determination of ⁷²Kr. This neutron-deficient nuclide, which lies close to the proton drip line, is one of the three major waiting points of the rp-process. When reaching one of these nuclides, the proton capture sequence stops, since the (γ, p) photodisintegration rate is equal to the (p, γ) proton capture rate and a slow β^+ decay must occur before further protons can be added. With the ISOLTRAP mass measurement the mass excess value was improved from -54110(270) keV to -53941(8) keV [11]. With an uncertainty of less than 10 keV, the new mass value allowed to contrain the effective lifetime in x-ray bursts, where at temperatures of 1-1.5 GK ⁷²Kr remains a strong waiting point with at least 80% of its β^+ decay half-life, thus delaying the burst duration.

5. Technical developments at ISOLTRAP

Technical improvements of the ISOLTRAP mass spectrometer include a new ion detector [59]. The existing micro-channel-plate (MCP) detector was replaced by a channeltron detector with a conversion dynode, where secondary electrons are monitored with an efficiency close to 100%. With this setup an increase of the total efficiency by a factor of about 3 was obtained and nuclides with a lower production yield can be studied.

Furthermore, the stability of the mass measurements is improved by stabilizing the magnetic field. It was observed that the fluctuation of the cyclotron frequency is correlated to the temperature variations in the warm bore of the superconducting magnet [60] and the pressure in the helium recovery line. Therefore, new stabilization systems have been installed which aim at stabilizing the center temperature to better than ± 10 mK and the helium pressure to ± 50 mbar.

Finally, a new technique to obtain radionuclides that are not produced at ISOLDE has been tested: the production of radionuclides by in-trap decay [61]. To this end short-lived nuclides are stored in the buffer-gas-filled preparation trap and after a sufficiently long storage period (depending on the half-life), enough daughter nuclide ions from the decay have been collected in the Penning trap for further measurements. After the usual cooling and centering procedure, they are

transferred to the precision trap for mass determination. This technique has been successfully applied to the first ISOLTRAP mass measurements of $^{61-63}$ Fe isotopes, which are not available from ISOLDE. The results are currently under investigation.

6. Conclusion and outlook

The ISOLTRAP mass spectrometer at ISOLDE delivers accurate mass values of short-lived radionuclides. The relative mass uncertainty is currently limited to $\delta m/m = 8 \times 10^{-9}$. With the possibility to reach nuclides with half-lives well below 100 ms and production yields down to 100 ions per second, a large number of radionuclides can be investigated at ISOLDE. Together with other Penning trap mass spectrometers at radioactive ion beam facilities around the world, more and more reliable mass values will become available for astrophysical calculations and tests of mass models.

Besides target development at ISOLDE, which aims at higher yields, more accessible elements, and less contamination, the experimental setup of ISOLTRAP is constantly subject to technical development. A new ion detector for a better overall efficiency, the stabilization of the magnetic field, and new reference ion sources are recent improvements of the setup. Further development is directed at a higher accuracy in the mass determination by use of new rf excitation schemes and ions of higher charge states.

Acknowledgments

This work was supported by the German Ministry for Education and Research (BMBF) under contract 06GF151, by the European Commission under contracts HPMT-CT-2000-00197 (Marie Curie Fellowship), HPRI-CT-2001-50034 (NIPNET), and RII3-CT-2004-506065 (TRAPSPEC), and by the Helmholtz association of national research centres (HGF) under contract VH-NG-037. We also thank the ISOLDE Collaboration as well as the ISOLDE technical group for their assistance.

References

- [1] H. Schatz et al., Phys. Rep. 294 (1998) 167.
- [2] H. Schatz, Int. J. Mass Spectrom. 251 (2006) 293.
- [3] D. Lunney, J.M. Pearson, and C. Thibault, Rev. Mod. Phys. 75 (2003) 1021.
- [4] K. Blaum, Phys. Rep. 425 (2006) 1.
- [5] Ultra-accurate mass determination and related topics, Special issue of Int. J. Mass Spectrom., Vol. 251 (2-3), L. Schweikhard and G. Bollen (Eds.), Elsevier, Amsterdam, 2006.
- [6] S. Rainville, J.K. Thompson, and D.E. Pritchard, Science 303 (2004) 334.
- [7] K. Blaum et al., Phys. Rev. Lett. 91 (2003) 260801.
- [8] U. Hager et al., Phys. Rev. Lett. 96 (2006) 042504.
- [9] G. Bollen et al., Phys. Rev. Lett. 96 (2006) 152501.

- [10] G. Savard et al., Phys. Rev. Lett. 95 (2005) 102501.
- [11] D. Rodríguez et al., Phys. Rev. Lett. 93 (2004) 161104.
- [12] M. Mukherjee et al., Phys. Rev. Lett. 93 (2004) 150801.
- [13] G. Audi, A.H. Wapstra, and C. Thibault, Nucl. Phys. A 729 (2003) 337.
- [14] M. König et al., Int. J. Mass Spectrom. Ion Process. 142 (1995) 95.
- [15] G. Bollen et al., Nucl. Instrum. Meth. A 368 (1996) 675.
- [16] F. Herfurth et al., J. Phys. B: At. Mol. Opt. Phys. 36 (2003) 931.
- [17] G. Savard et al., Int. J. Mass Spectrom. 251 (2006) 252.
- [18] A. Jokinen et al., Int. J. Mass Spectrom. 251 (2006) 204.
- [19] R. Ringle et al., Int. J. Mass Spectrom. 251 (2006) 300.
- [20] S. Rahaman et al., Int. J. Mass Spectrom. 251 (2006) 146.
- [21] J. Dilling et al., Int. J. Mass Spectrom. 251 (2006) 198.
- [22] J. Szerypo et al., Nucl. Instrum. Meth. B 204 (2003) 512.
- [23] E. Kugler, Hyperfine Interact. 129 (2000) 23.
- [24] F. Herfurth et al., Nucl. Instrum. Meth. A 469 (2001) 254.
- [25] G. Savard at al., Phys. Lett. A 158 (1991) 247.
- [26] H. Raimbault-Hartmann et al., Nucl. Instrum. Meth. B 126 (1997) 378.
- [27] M. Turrión, priv. communication.
- [28] G. Gräff, H. Kalinowsky, and J. Traut, Z. Phys. A 297 (1980) 35.
- [29] A. Kellerbauer et al., Phys. Rev. Lett. 93 (2004) 072502.
- [30] P. Delahaye et al., Phys. Rev. C 74 (2006) 034331.
- [31] E. Bouquerel et al., submitted to Eur. Phys. J. A
- [32] K. Blaum et al., Nucl. Phys. A 746 (2004) 305c.
- [33] F. Herfurth et al., Phys. Rev. Lett. 87 (2001) 142501.
- [34] F. Herfurth et al., Eur. Phys. J. A 15 (2002) 17.
- [35] C. Guénaut et al., J. Phys. G: Nucl. Part. Phys. 31 (2005) S1765.
- [36] K. Blaum et al., Europhys. Lett. 67 (2004) 586.
- [37] J. Van Roosbroeck et al., Phys. Rev. Lett. 92 (2004) 112501.
- [38] D. Rodríguez et al., Nucl. Phys. A 769 (2006) 1.
- [39] H. Raimbault-Hartmann et al., Nucl. Phys. A 706 (2002) 3.
- [40] T. Otto et al., Nucl. Phys. A 567 (1994) 281.
- [41] F. Herfurth et al., Nucl. Phys. A 746(2004) 487c.
- [42] G. Sikler et al., Nucl. Phys. A 763 (2005) 45.
- [43] J. Dilling et al., Eur. Phys. J. A 22 (2004) 163.

- [44] A. Herlert et al., Int. J. Mass Spectrom. 251 (2006) 131.
- [45] H. Stolzenberg et al., Phys. Rev. Lett. 65 (1990) 3104.
- [46] F. Ames et al., Nucl. Phys. A 651 (1999) 3.
- [47] D. Beck et al., Nucl. Phys. A 626 (1997) 343c.
- [48] G. Bollen et al., Hyperfine Interact. 132 (2001) 215.
- [49] D. Beck et al., Eur. Phys. J. A 8 (2002) 307.
- [50] S. Schwarz et al., Nucl. Phys. A 693 (2001) 533.
- [51] C. Weber et al., Phys. Lett. A 347 (2005) 81.
- [52] G. Bollen et al., J. Mod. Optics 39 (1992) 257.
- [53] F. Herfurth et al., *Eur. Phys. J. A* **25**, s01 (2005) 17.
- [54] K. Blaum et al., Eur. Phys. J. A 15 (2002) 245.
- [55] K. Blaum et al., Anal. Bioanal. Chem. 377 (2003) 1133.
- [56] A. Kellerbauer et al., Eur. Phys. J. D 22 (2003) 53.
- [57] S. Bishop et al., Phys. Rev. Lett. 90 (2003) 162501.
- [58] G. Audi, O. Bersillon, J. Blachot, and A.H. Wapstra, Nucl. Phys. A 729 (2003) 3.
- [59] C. Yazidjian et al., submitted to Hyperfine Interact.
- [60] K. Blaum et al., J. Phys. G: Nucl. Part. Phys. 31 (2005) S1775.
- [61] A. Herlert et al., New J. Phys. 7 (2005) 44.