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A simple derivation of a general
equilibrium equation, with
two astrophysical applications

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**A simple derivation of a general equilibrium equation,
with two astrophysical applications**

By PER OLOF FRÖMAN

Summary. — If the statistical mechanical formula (1), which has been derived by O. KLEIN, is combined with the well-known thermodynamical equilibrium condition (2) containing the chemical potentials μ_r , one immediately obtains the general equilibrium equation (3), which contains as special cases SAHA's formula (4) and another well-known formula (5) of great astrophysical importance.

Using the powerful concept of a Gibbs' grand canonical ensemble, O. KLEIN has derived the following formula¹:

$$N_{r,s} = \left(\frac{2\pi m_r k T}{h^2} \right)^{\frac{3}{2}} g_{r,s} e^{\frac{\mu_r - E_{r,s}}{k T}}, \quad (1)$$

from which it follows that

$$N_r = \sum_s N_{r,s} = \left(\frac{2\pi m_r k T}{h^2} \right)^{\frac{3}{2}} e^{\frac{\mu_r}{k T}} \sum_s g_{r,s} e^{-\frac{E_{r,s}}{k T}}, \quad (1a)$$

where

$N_{r,s}$ is the mean number of particles (per unit volume) of kind r which are in the internal quantum state s , regardless of the translatory motion of the particle

$g_{r,s}$ is the statistical weight of the internal quantum state s of the particles of kind r

$E_{r,s}$ is the energy of the internal quantum state s of the particles of kind r

μ_r is the chemical potential of the particles of kind r

m_r is the mass of a particle of kind r

T is the absolute temperature

k is Boltzmann's constant

h is Planck's constant.

We now consider a possible reaction

$$\sum_r \nu_r A_r = 0,$$

¹ O. KLEIN, G. BESKOW and L. TREFFENBERG, Arkiv för mat., astr. o. fys., Band 33 B, No. 1 (1946-47); O. KLEIN, Supplemento al volume VI, Serie IX del Nuovo Cimento, N. 2 (1949). For one kind of particle the formula was also given in Professor KLEIN's lectures on statistical mechanics in 1951-52.

between the different kinds of particles. Here, A_r is the symbol for a particle of kind r , and \varkappa_r are coefficients (of which some are positive and some negative). For the above reaction we have the well-known thermodynamic equilibrium condition¹

$$\sum_r \varkappa_r \mu_r = 0. \quad (2)$$

Combining (1) and (2), we immediately obtain the following general equilibrium equation²

$$\prod_r (N_{r,s})^{\varkappa_r} = \prod_r \left\{ \left(\frac{2\pi m_r k T}{h^2} \right)^{\frac{3}{2}} g_{r,s} e^{-\frac{E_{r,s}}{k T}} \right\}^{\varkappa_r} \quad (3)$$

We obtain

$$\prod_r (N_r)^{\varkappa_r} = \prod_r \left\{ \left(\frac{2\pi m_r k T}{h^2} \right)^{\frac{3}{2}} \sum_s g_{r,s} e^{-\frac{E_{r,s}}{k T}} \right\}^{\varkappa_r} \quad (3a)$$

from (1a) and (2), and

$$\frac{N_{r,s}}{N_r} = \frac{g_{r,s} e^{-\frac{E_{r,s}}{k T}}}{\sum_s g_{r,s} e^{-\frac{E_{r,s}}{k T}}} \quad (3b)$$

from (1) and (1a). The formulae (3a) and (3b) together constitute an equilibrium equation which is as general as (3).

We shall now use (3) to obtain two well-known equilibrium equations of considerable astrophysical importance.

I. Let r denote those atoms of a certain kind which have lost r electrons. If (3) is applied to the special reaction in which such an ion loses a further electron, one immediately obtains SAHA's formula in the form

$$N_e \frac{N_{r+1,s'}}{N_{r,s}} = \left(\frac{2\pi m_e k T}{h^2} \right)^{\frac{3}{2}} \cdot \frac{2 g_{r+1,s'}}{g_{r,s}} e^{-\frac{E_{r+1,s'} - E_{r,s}}{k T}} \quad (4)$$

if the electron mass is neglected in comparison with the atomic mass. N_e denotes the mean number of free electrons per unit volume, and m_e denotes the electron mass.

II. Let $N(n_K, n_L, \dots)$ denote the mean number of atoms or ions of a certain kind (per unit volume) with n_K electrons bound in the K -shell, n_L electrons bound in the L -shell, etc. If (3) is applied to the special reaction in which such an atom or ion loses all its electrons, one immediately obtains

$$\frac{N(n_K, n_L, n_M, \dots)}{N(0, 0, 0, \dots)} = \left(\frac{N_e}{2 \left(\frac{2\pi m_e k T}{h^2} \right)^{\frac{3}{2}}} \right)^{n_K + n_L + n_M + \dots} \cdot \left(\frac{2 \cdot 1^2}{n_K} \right) \cdot \left(\frac{2 \cdot 2^2}{n_L} \right) \cdot \left(\frac{2 \cdot 3^2}{n_M} \right) \dots e^{\frac{n_K \varkappa_K + n_L \varkappa_L + n_M \varkappa_M + \dots}{k T}} \quad (5)$$

¹ This equilibrium condition can be derived by the use of Gibbs' grand canonical ensemble. See O. KLEIN, *Supplemento al volume VI, serie IX del Nuovo Cimento*, N. 2 (1949).

² For the sake of simplicity we have suppressed the subscript r on the index s .

if the mass of the electrons is neglected in comparison with the atomic mass. Here $\chi_K, \chi_L, \chi_M, \dots$ denote the (mean) ionization potentials for the K -, L -, M -, ... shells. The formula (5) is very important for calculations of the ionization (and the opacity) of stellar matter.

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