Direct Synthesis of Pyridine and Pyrimidine Derivatives

by

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To my parents, Merrill and Dennis Hill, to my sister, Mallory Hill •

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Preface

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Direct Synthesis of Pyridine and Pyrimidine Derivatives

by

Matthew D. Hill

Submitted to the Department of Chemistry on May 9, 2008 in partial fulfillment of the requirements for the Degree of Doctor of Philosophy

ABSTRACT

I. Synthesis of Substituted Pyridine Derivatives via the Ruthenium-Catalyzed Cycloisomerization of 3-Azadienynes

The two-step conversion of various *N*-vinyl and *N*-aryl amides to the corresponding substituted pyridines and quinolines, respectively, is described. The process involves the direct conversion of amides, including sensitive *N*-vinyl amides, to the corresponding trimethylsilyl alkynyl imines followed by a ruthenium-catalyzed protodesilylation and cycloisomerization. A wide range of new alkynyl imines are prepared and readily converted to the corresponding azaheterocycles.

II. Single-Step Synthesis of Pyrimidine Derivatives

The single-step conversion of various *N*-vinyl and *N*-aryl amides to the corresponding pyrimidine and quinazoline derivatives, respectively, is described. The process involves amide activation with 2-chloropyridine and trifluoromethanesulfonic anhydride followed by nitrile addition into the reactive intermediate and cycloisomerization. In situ nitrile generation from primary amides allows for their use as nitrile surrogates. The use of this chemistry with sensitive *N*-vinyl amides and epimerizable substrates in addition to a wide range of functional groups is noteworthy.

III. Direct Synthesis of Pyridine Derivatives

The single-step conversion of various *N*-vinyl and *N*-aryl amides to the corresponding pyridine and quinoline derivatives, respectively, is described. The process involves amide activation with trifluoromethanesulfonic anhydride in the presence of 2-chloropyridine followed by π -nucleophile addition to the activated intermediate and annulation. Compatibility of this chemistry with sensitive *N*-vinyl amides, epimerizable substrates, and a variety of functional groups is noteworthy.

Thesis Supervisor: Assistant Professor Mohammad Movassaghi

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Abbreviations

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Ac	acetyl
atm	atmosphere
Bu	butyl
°C	degree Celsius
^с Нх	cyclohexyl
CH_2Cl_2	dichloromethane
2-ClPyr	2-chloropyridine
cm	centimeter
COD	cyclooctadiene
Ср	cyclopentadiene
d	deuterium
δ	parts per million
DavePhos	2'-(dicyclohexylphosphino)-N.N-dimethylbiphenyl-2-amine
DMAP	4-dimethylaminopyridine
DMF	<i>N.N</i> -dimethylformamide
DMSO	dimethyl sulfoxide
dppf	1.1'-Bis(diphenylphosphino)ferrocene
ee	enantiomeric excess
equiv	equivalent
Et	ethyl
Et ₂ O	diethyl ether
Et ₃ N	triethylamine
EtOAc	ethyl acetate
FT	Fourier transform
g	gram
GC-MS	gas chromatography-mass spectroscopy
h	hour
HRMS	high resolution mass spectroscopy
Hz	Hertz
i	iso
IR	infrared
J	coupling constant
L	liter
LDA	lithium diisopropylamide
LHMDS	lithium hexamethyldisilylamide
Μ	molar
mg	milligram
MHz	megahertz
min	minute
mL	milliliter
mm	millimeter
mmol	millimole
μmol	micromole
mol	mole

.

	n	normal
	N	normal (concentration)
	N_2	dinitrogen
	NMR	nuclear magnetic resonance
	nOe	nuclear Overhauser effect
	р	para
	pH	hydrogen ion concentration
	Ph	phenyl
	Ph ₃ P	triphenylphosphine
	Pr	propyl
	ppm	parts per million
	R	rectus
	$R_{\rm f}$	retention factor
	S	second
	<i>S</i>	secondary
	SPhos	dicyclohexyl(2',6'-dimethoxybiphenyl-2-yl)phosphine
	t	tertiary
÷	TBAF	tetra-n-butylammonium fluoride
	TBS	'butyldimethylsilyl
	TBSOTf	'butyldimethylsilyl trifluoromethanesulfonate
	TBDPS	'butyldiphenylsilyl
	TFA	trifluoroacetic acid
	Tf ₂ O	trifluoromethanesulfonic anhydride
	TfOH	trifluoromethanesulfonic acid
	THF	tetrahydrofuran
	TMS	trimethylsilyl
	VT	variable temperature
	XPhos	dicyclohexyl(2',4',6'-triisopropylbiphenyl-2-yl)phosphine
	Ζ	zusammen

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Chapter I

Synthesis of Substituted Pyridine Derivatives via the Ruthenium-Catalyzed Cycloisomerization of 3-Azadienynes

Introduction and Background

Pyridine derivatives are an important class of azaheterocycle found in many natural products, active pharmaceuticals, and functional materials.¹ Diploclidine² and nakinadine A³ are two examples of recently isolated and structurally diverse natural products containing the pyridine core (Figure 1). Significant pyridine derived pharmaceuticals include atazanavir⁴ (Reyataz[®]) and imatinib mesylate⁵ (Gleevec[®]), and are prescribed for human immunodeficiency virus (HIV) and chronic myelogenous leukemia, respectively (Figure 1). Pyridine derivatives are also incorporated into polymers like polyvinyl pyridine (PVP, Figure 1).⁶ While invention of synthetic methodologies for pyridines has been an important area of chemical research for well over a century, the importance of the pyridine core in both biological and chemical fields continues to inspire development of new syntheses.



Figure 1. Representative compounds containing a pyridine substructure.

Nicotinic acid (also known as vitamin B_3 and niacin) is an important natural building block for pyridine alkaloids (Figure 1). In nature, this important component of coenzymes NAD⁺ and NADP⁺ is synthesized from L-tryptophan by way of the kynurenine pathway in animals or from glyceraldehyde 3-phosphate and L-aspartic acid in many plants.⁷ Both pathways rely on decarboxylation of quinolinic acid as a final step, but are otherwise very different. Nicotine, the addictive substance in tobacco, is formed by incorporation of a pyrrolidine moiety derived from L-ornithine onto the molecular framework of nicotinic acid (Figure 1). Like nicotine, similar alkaloids including anabasine, ricinine, and arecoline all originate from nicotinic acid (Figure 1).⁷

While picoline was isolated in 1846,⁸ Körner's and Dewar's elucidation of the pyridine structure in 1869 and 1871, respectively, marked the beginning of significant chemical research in the field.⁹ Coal tar served as an initial source of pyridine, however recent commercial methods have been developed for its preparation from crotonaldehyde, formaldehyde, and ammonia in the gas phase.^{1b}



Scheme 1. [5+1] Condensation route to substituted pyridines.¹⁰

Historically, many pyridine syntheses rely on condensation of amine and carbonyl compounds. The fragments contributing to the six-atom azaheteroaromatic ring often characterize these and other methods for preparation of the pyridine core. Ammonia has served as the nitrogen source in countless protocols including its [5+1] condensation with 1,5-dicarbonyls (Scheme 1).¹⁰ Like many condensation methods a second oxidation step, often autoxidation, is necessary for aromatization. Ammonia is also frequently used in the [2+2+1+1] Hantzsch pyridine synthesis (Scheme 2).^{11,12} Other pyridine syntheses rely on alkyl or vinyl amines such as the [3+3] example: 1,3-dicarbonyl derivative condensation with a 3-aminoenone (Scheme 3).¹³



Scheme 2. The [2+2+1+1] Hantzsch pyridine synthesis.¹¹



Scheme 3. [3+3] Condensation of a 1,3-dicarbonyl derivative and vinylogous amide.¹³

Other methods of pyridine synthesis have become increasingly important. Boger has developed a [4+2] inverse electron demand hetero-Diels-Alder reaction between enamines and 1,2,4-triazine (Scheme 4).¹⁴ 6π -Electrocyclization approaches, including a recent transition-metal mediated [4+2] example by Ellman (Scheme 5), are also of continuing importance.¹⁵ Over the past several decades, many other transition-metal promoted pyridine syntheses have been reported. A well-developed [2+2+2] approach utilizes two alkyne equivalents and a nitrile species (Scheme 6).¹⁶



Scheme 4. Hetero-Diels-Alder [4+2] approach to pyridine derivatives.¹⁴



Scheme 5. Rhodium mediated [4+2] synthesis of pyridines.¹⁵



Scheme 6. Cobalt mediated [2+2+2] synthesis of substituted pyridines.¹⁶

Recent advancements in cross-coupling chemistry have increased the popularity of azaheterocycle substituent modification and have been described in several reviews.¹⁷ Some of these methods, including the Chichibabin reaction (Scheme 7),¹⁸ rely on the electron deficient character of the pyridine ring. Activated pyridines can be used with numerous transition-metal catalysts to afford a structurally diverse set of pyridine derivatives (Schemes 8 and 9).^{17, 19, 20}



Scheme 7. The Chichibabin reaction.¹⁸



Scheme 8. Iron-catalyzed cross-coupling of activated pyridines.¹⁹



Scheme 9. Suzuki-Miyaura cross-coupling of activated pyridines.²⁰

Due to the importance of pyridines, our group is interested in new methodologies for their synthesis. We envisioned readily available N-vinyl²¹ and N-aryl amides could lead to 3-azadienynes that are capable of forming catalytically generated metal vinylidene intermediates. We believed cycloisomerization of these complexes should give pyridine derivatives in two steps from amide precursors (Scheme 10).



Scheme 10. General strategy for two-step synthesis of pyridine derivatives from readily available substrates.

Metal vinylidene²² complexes (eq 1) can be directly accessed via the transition metal catalyzed isomerization of terminal alkynes.²³ This isomerization is believed to proceed through the initial formation of a metal η^2 -alkyne complex and subsequent formal 1,2-hydride shift of the acetylenic hydrogen to give the thermodynamically favored metal η^1 -vinylidene complex.²³ Metal vinylidenes have been employed for a range of transformations that utilize either a catalytic²⁴ or stoichiometric²⁵ amount of transition metal complexes. Various neutral and cationic transition metal complexes have demonstrated superb activity for providing metal vinylidene intermediates under mild reaction conditions. Experimental and theoretical studies^{23,26} suggest reactivity similar to that of the ketene functional group with an electrophilic C_a -center and nucleophilic C_b -center.²⁷ Furthermore, the direct nucleophilic addition of carbon nucleophiles to metal vinylidene intermediates offers tremendous potential in development of novel carbon-carbon bond forming reactions (eq 1).

Results and Discussion

The metal catalyzed cycloisomerization of dienynes via catalytically generated metal vinylidene intermediates represents a highly effective method for the synthesis of aromatic compounds.²⁸ We sought to explore the use of 3-azadienynes as substrates for a metal–catalyzed cycloisomerization reaction, providing a general approach to a broad range of substituted pyridine derivatives 1 (Scheme 10).²⁹ To take full advantage of the wide range of *N*-vinyl amides available by metal catalyzed *C-N* bond formation,²¹ we required a mild and efficient procedure for the direct conversion of amides 2 to the corresponding 3-azadienynes 3.³⁰ Inspired by recent reports on the electrophilic activation of amides³¹ we developed a single-step process for the conversion of *N*-vinyl/aryl amides 2 to the corresponding alkynyl imines 3. Under our optimum conditions, a cold solution of the *N*-phenyl benzamide (2a, Table 1) in dichloromethane is treated sequentially with 2-chloropyridine (2-ClPyr, 4.0 equiv) and trifluoromethanesulfonic anhydride (Tf₂O, 1.2 equiv), followed by copper trimethylsilylacetylide (2.7 equiv), which affords the desired trimethylsilyl alkynyl imine 3a in 97% yield (Table 2, entry 1, 2.5-g scale).³²

	Tf ₂ O (1.2 equidates base additive (x e $\frac{CH_2Cl_2, -78 \rightarrow 0}{Cu - 2}$ THF, -78 -0 °	v) quiv) °C; 7 equiv)	N SiMe
Entry	Base Additive	x	Yield (%) ^a
1	pyridine	4	<10
2	2,6-lutidine	4	27
3	Et ₃ N	4	<4 ^b
4	Pr ₂ NEt	4	<20
5	2-chloropyridine	4	97
6	2-chloropyridine	3	92
7	2-chloropyridine	2	65

Table 1. Base additive screen for alkynyl imine synthesis.

^a Mass balance is the starting amide. ^b Mixture of products; no recovered SM.

The use of 2-chloropyridine as the base³³ was found to be critical in obtaining the desired alkynyl imines (Table 1).³² Significantly, this single-step and mild procedure provides access to new alkynyl imines, in particular, those derived from *N*-vinyl amides. For comparison, the use of existing methods³⁰ for the synthesis of *N*-2-thienyl and *N*-dihydropyranyl alkynyl imines **3** (Table 2, entries 13 and 15) gave none and <10% yield of the desired product, respectively.

Entry	Amide Substrate (2)	3-Azadienyne (3)	Yield (%) ^a	Product (1)	Yield (%) ^a
		R R R R R R			11
1	2a,R=H R'=H R"=H	3a [^] SiMe ₃	97 ^b	1a	91 ^b
2	R=H R'=H R"=OMe		89		92
3	R = OMe R' = H R'' = H		96		91
4	$R = H$ $R' = CF_3 R'' = H$		73, <10 ^c		89
5	S N Ph	SiMe	81 /3 (s N	75

Table 2. Substrate scope for two-step pyridine synthesis.



^a Isolated yields: all entries are average of two experiments. Optimum conditions used uniformly. ^b Gram-scale experiments. ^c Yield of the corresponding desilylated imine.^{32 d} Kept at -78 °C.^{32 e} 5 mol% of catalyst system used.

Early in our studies we identified the readily available chlorocyclopentadienyl bis(triphenylphosphine) ruthenium complex $(CpRu(PPh_3)_2Cl, 5)^{34}$ as an effective catalyst for cycloisomerization of terminal alkynyl imine 4a to product 1a (Scheme 11).³⁵ While imine 4a could be prepared by protodesilylation of the corresponding trimethylsilyl derivative 3a (Scheme 11), this required an additional step and resulted in decreased stability of the substrate and yield of the cycloisomerization reaction. These considerations prompted the development of a process for the direct use of trimethylsilyl alkynyl imine 3a as substrate. The trimethylsilyl alkynyl imine



Scheme 11. ^a Tf₂O, 2-ClPyr, CH₂Cl₂; Me₃SiC=CCu, THF, $-78 \rightarrow 0$ °C. ^b 5, SPhos, NH₄PF₆, toluene, 105 °C. ^c K₂CO₃, MeOH. ^d 5, toluene, 105 °C.

3a, was used to survey a series of metal complexes, supporting ligands, additives and solvents (Table 3).³² The combination of ruthenium complex **5** (10 mol%), 2-dicyclohexyl-phosphino-2',6'-dimethoxy-1,1'-biphenyl (SPhos³⁶, 10 mol%), and ammonium hexafluorophosphate (1 equiv) in toluene (0.2 M) at 105 °C was identified as the optimal set of conditions, as illustrated by the clean conversion of imine **3a** to quinoline **1a** in 90% yield (Table 2, entry 1, 1.0-g scale).³²



Table 3. Precatalyst and ligand screen.

23	[Ru(COD)Cl ₂] _n	SPhos	10	0
24	CpRu(COD)Cl	-	-	0
25	CpRu(COD)Cl	SPhos	10	<6
26	CpRu(COD)Cl	SPhos	20	<9
27	CpRu(COD)Cl	Ph ₃ P	10	38
28	CpRu(COD)Cl	Ph ₃ P	15	62 [·]
29	CpRu(COD)Cl	Ph ₃ P	20	77
30	CpRu(COD)Cl	(2-furyl) ₃ P	15	36
31	CpRu(COD)Cl	(2-furyl) ₃ P, Ph ₃ P	10, 10	67
32	CpRu(COD)Cl	(2-furyl) ₃ P, SPhos	10, 10	54
33	CpRu(COD)Cl	Ph ₃ P, SPhos	13, 13	88
34	CpRu(COD)Cl	Ph ₃ P, SPhos	15, 15	97
35	CpRu(Ph ₃ P) ₂ Cl	-	-	80
36	CpRu(Ph ₃ P) ₂ Cl	DavePhos	10	30
37	CpRu(Ph ₃ P) ₂ Cl	XPhos	10	74
38	CpRu(Ph ₃ P) ₂ Cl	SPhos	10	95
39	CpRu(Ph ₃ P) ₂ Cl	SPhos	5	56
40	CpRu(Ph ₃ P) ₂ Cl	₽ħ₃₽	10	80
41	CpRu(Ph ₃ P) ₂ Cl	Ph₃P	20	81
42	CpRu(Ph ₃ P) ₂ Cl	(2-furyl) ₃ P	10	76

^a Mass balance is the starting silvlated alkynyl imine **3a**. ^b 2.2 mol % TFA additive. ^c 10 mol % AgOAc additive. ^d Same result at 75 °C. ^e 5 mol % Ru-complex.

Interestingly, neither SPhos nor PPh₃ alone were ideal ligands when used independently with chlorocyclopentadienyl cycloocta-1,5-diene ruthenium complex (CpRuCODCl, 6)³⁷ for cycloisomerization of 3-azadienyne 3a.³² However, the combination of these ligands in conjunction with ruthenium complex 6 provided a catalyst system with equal activity to the optimal system.³² While the exact role of SPhos is unclear at this time,^{38 31}P NMR experiments confirm that PPh₃ out competes SPhos in displacement of COD from 6, providing complex 5 and remaining SPhos, similar to the optimal precatalyst mixture. Also, ¹H NMR monitoring of the cycloisomerization reaction of azadienyne **3a** employing complex **6** and SPhos alone revealed the formation of the inactive CpRu(η ⁶-PhMe)PF₆ complex.³⁹

The optimal reaction conditions proved to be compatible with a variety of C-silyl alkynyl imines (Table 2). In particular, we found even highly sensitive N-vinyl/heterocyclic imines to be excellent substrates (Table 2, entries 9-16), providing a convergent and versatile azaheterocycle synthesis. Importantly, the direct conversion of C-silyl alkynyl imines **3** to the corresponding azaheterocycles **1** with this Ru-catalyst system avoids the isolation of the more sensitive terminal alkynyl imines (i.e., Table 2, entry 4). In only two cases (entries 7 and 16) in situ desilylation was found to be exceedingly slow, prompting the use of the corresponding terminal alkyne derivatives as the substrates for cycloisomerization. In the synthesis of the acid sensitive N-

triisopropylsilylazaindole (entry 16), lowering the catalyst loading (5 mol%) from our standard conditions was beneficial.



Subjecting the alkynyl imine $3a-d_5$ (eq 2) to our standard conditions gave the quinoline $1a-d_5$ (eq 2) with C4-deuterium incorporation (68%).³² The use of terminal alkynyl imine $4a-d_1$ (eq 3, without NH₄PF₆) as substrate provided quinoline $1a-d_1$ (eq 3) with C3-deuterium incorporation (72%). Furthermore, employing ammonium hexafluorophospate- d_4 in the cycloisomerization of alkynyl imine 3a (eq 4) provided the quinoline $1a-d_1$ (eq 4) with C3-deuterium incorporation (68%). The protodesilylated imine 4a (Scheme 11) was not detected as a persistent intermediate by TLC or ¹H NMR monitoring experiments (Table 2, entry 1) and the silyl alkynyl imine 3a was recovered unchanged from the reaction mixture in the absence of Rucomplex 5. Additionally, only a trace amount of the desired desilylated and cycloisomerized product was detected when the ammonium hexafluorophosphate was omitted, returning the starting material as the mass balance.⁴⁰ These observations suggest the direct conversion of the silyl alkynyl imine 3 to the C-silyl metal vinylidene⁴¹ 7 (Scheme 12) followed by protodesilylation and cycloisomerization to give 1.



Scheme 12. Proposed mechanism for synthesis of azaheterocycles 1 from azadienynes 3.

Conclusion

The chemistry described here provides a two-step process for the synthesis of substituted pyridine derivatives from readily available N-vinyl/aryl amides (Scheme 11, steps a and b). Noteworthy features of this chemistry include the single-step conversion of a wide range of readily available amides, including sensitive N-vinyl amides, to the corresponding C-silyl alkynyl imines and their direct Ru-catalyzed protodesilylation and cycloisomerization to the This Ru-catalyzed conversion of C6-trimethylsilyl 3corresponding azaheterocycles. azadienynes to azaheterocycles, not only reduces a three-step sequence^{28c} to a single-step but also does not require the isolation of sensitive and/or inaccessible terminal alkynyl imines as substrates.42

- ⁽⁴⁾ Harrison, T. S., Scott, L. J. Drugs 2005, 65, 2309.
- ⁽⁵⁾ Deininger, M. W. N., Druker, B. J. Pharmacol. Rev. 2003, 55, 401.
- ⁽⁶⁾ Raje, V. P.; Bhat, R. P.; Samant, S. D. Synlett 2006, 2676.

⁽¹⁾ (a) Jones, G. In Comprehensive Heterocyclic Chemistry II; Katritzky, A. R.; Rees, C. W.; Scriven, E. F. V.;

McKillop, A., Eds; Pergamon: Oxford, 1996; Vol. 5; p 167. (b) Henry, G. D. *Tetrahedron* 2004, 60, 6043. (c) Joule, J. A.; Mills, K. In *Heterocyclic Chemistry*, 4th ed.; Blackwell Science Ltd.: Cambridge, MA, 2000; p 63. (d)

Michael, J. P. Nat. Prod. Rep. 2005, 22, 627.

⁽²⁾ Jayasinghe, L.; Jayasooriya, C. P.; Hara, N.; Fujimoto, Y. Tetrahedron Lett. 2003, 44, 8769.

⁽³⁾ Kubota, T.; Nishi, T.; Fukushi, E.; Kawabata, J.; Fromont, J.; Kobayashi, J. Tetrahedron Lett. 2007, 48, 4983.

⁽⁷⁾ Dewick, P. M. In *Medicinal Natural Products A Biosynthetic Approach*, 2nd ed.; John Wiley & Sons Ltd.: West Sussex, England, 2002; p 311.

⁽⁸⁾ Anderson, T. Edinburgh. New Philos. J. 1846, 41, 146.

⁽⁹⁾ Pictet, A. In The Vegetable Alkaloids With Particular Reference to Their Chemical Constitution, 2nd ed.; 1904; p 10. (10) Kelly, T. R.; Lebedev, R. L. J. Org. Lett. 2002, 67, 2197.

⁽¹¹⁾ Hantzsch, A. Liebigs Ann. Chem. 1882, 215, 1.

⁽¹²⁾ For a recent example, see: Guillaume, B.; Charette, A. B. J. Am. Chem. Soc. 2008, 130, 18.

⁽¹³⁾ Baumgarten, P.; Dornow, A. Chem. Ber. 1939, 72, 563.

⁽¹⁴⁾ Boger, D. L.; Panek, J. S.; Meier, M. M. J. Org. Chem. 1982, 47, 895.

⁽¹⁵⁾ Colby, D. A.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2008, 130, 3645.

⁽¹⁶⁾ Chelucci, G.; Falorni, M.; Giacomelli, G. Synthesis 1990, 1121.

^{(17) (}a) Chinchilla, R.; Nájera, C.; Yus, M. Chem. Rev. 2004, 104, 2667. (b) Mongin, F.; Quéguiner, G. Tetrahedron 2001, 57, 4059.

⁽¹⁸⁾ Chichibabin, A. E.; Zeide, O. A. J. Russ. Phys. Chem. Soc. 1914, 46, 1216.

⁽¹⁹⁾ Fürstner, A.; Leitner, A.; Méndez, M.; Krause, H. J. Am. Chem. Soc. 2002, 124, 13856.

⁽²⁰⁾ Billingsley, K.; Buchwald, S. L. J. Am. Chem. Soc. 2007, 129, 3358.

(21) (a) Muci, A. R.; Buchwald, S. L. Top. Curr. Chem. 2002, 219, 131. (b) Hartwig, J. F. In Handbook of Organopalladium Chemistry for Organic Synthesis; Negishi, E., Ed.; Wiley-Interscience: New York, 2002; p. 1051.
(c) Beletskaya, I. P.; Cheprakov, A. V. Coordin. Chem. Rev. 2004, 248, 2337. (d) Dehli, J. R.; Legros, J.; Bolm, C. Chem. Commun. 2005, 973.

⁽²²⁾ For reviews, see: (a) Bruce, M. I.; Swincer, A. G. Adv. Organomet. Chem. 1983, 22, 59. (b) Trost, B. M. Chem. Ber. 1996, 129, 1313. (c) Bruneau, C.; Dixneuf, P. H. Acc. Chem. Res. 1999, 32, 311. (d) McDonald, F. E. Chem. Eur. J. 1999, 5, 3103. (d) Bruneau, C.; Dixneuf, P. H. Angew. Chem. Int. Ed. 2006, 45, 2176. (e) Varela, J. A.; Saá, C. Chem. Eur. J. 2006, 12, 6450.

⁽²³⁾ See reference 22. For mechanistic discussions, see: (a) Wakatsuki, Y.; Koga, N.; Yamazaki, H.; Morokuma, K. J. Am. Chem. Soc. **1994**, 116, 8105. (b) Sheng, Y.; Musaev, D. G.; Reddy, K. S.; McDonald, F. E.; Morokuma, K. J. Am. Chem. Soc. **2002**, 4149. For a review, see: (c) Wakatsuki, Y. J. Organomet. Chem. **2004**, 689, 4092.

⁽²⁴⁾ For representative examples of transition metal catalyzed processes, see reference 22 and: (a) Landon, S. J.;
Shulman, P. M.; Geoffroy, G. L. J. Am. Chem. Soc. 1985, 107, 6739. (b) Trost, B. M.; Dyker, G.; Kulawiec, R. J. J. Am. Chem. Soc. 1990, 112, 7809. (c) McDonald, F. E.; Schultz, C. C. J. Am. Chem. Soc. 1994, 116, 9363. (d)
McDonald, F. E., Schultz, C. C., Chatterjee A. K. Organometallics 1995, 14, 3628. (e) Merlic, C. A.; Pauly, M. E. J. Am. Chem. Soc. 1996, 118, 11319. (f) Maeyama, K.; Iwasawa, N. J. Am. Chem. Soc. 1998, 120, 1928. (g) Manabe,
T.; Yanagi, S.-i.; Ohe, K.; Uemura, S. Organometallics 1998, 17, 2942. (h) Trost, B. M.; Rhee, Y. H. J. Am. Chem. Soc. 1999, 121, 11680. (i) Maeyama, K.; Iwasawa, N. J. Org. Chem. 1999, 64, 1344. (j) McDonald, F. E.; Reddy, K. S.; Díaz, Y. J. Am. Chem. Soc. 2000, 122, 4304. (k) Ohmura, T.; Yamamoto, Y.; Miyaura, N. J. Am. Chem. Soc. 2000, 122, 4304. (k) Ohmura, T.; Yamamoto, Y.; Miyaura, N. J. Am. Chem. Soc. 2000, 122, 4304. (k) Ohmura, T.; Yamamoto, Y.; Miyaura, N. J. Am. Chem. Soc. 2000, 122, 4990.

⁽²⁵⁾ For representative examples of transition metal promoted processes, see reference 22 and: (a) Parlier, A.; Rudler, H. J. Chem. Soc., Chem. Commun. 1986, 514. (b) Liebeskind, L. S.; Chidambaram, R. J. Am. Chem. Soc. 1987, 109, 5025. (c) Barrett, A. G.; Carpenter, N. E. Organometallics 1987, 6, 2249. (d) Wang, Y.; Finn, M. G. J. Am. Chem. Soc. 1995, 117, 8045. (e) McDonald, F. E.; Bowman, J. L. Tetrahedron Lett. 1996, 37, 4675. (f) McDonald, F. E.; Olson, T. C. Tetrahedron Lett. 1997, 38, 7691. (g) McDonald, F. E.; Zhu, H. Y. H. Tetrahedron 1997, 53, 11061.
⁽²⁶⁾ Kostic, N. M.; Fenske, R. F. Organometallics 1982, 1, 974.

(27) (a) Cotton, F. A.; Wilkinson, G. Advanced Inorganic Chemistry, 5th ed.; Wiley-Interscience: New York, 1988; pp. 1122. (b) Birdwhistell, K. R.; Tonker, T. L.; Templeton, J. L. J. Am. Chem. Soc. 1985, 107, 4474.
(28) For related reviews, see: (a) Bruneau, C.; Dixneuf, P. H. Acc. Chem. Res. 1999, 32, 311. (b) Trost, B. M.; Toste,

⁽²⁸⁾ For related reviews, see: (a) Bruneau, C.; Dixneuf, P. H. Acc. Chem. Res. 1999, 32, 311. (b) Trost, B. M.; Toste, F. D.; Pinkerton, A. B. Chem. Rev. 2001, 101, 2067. (c) Nevado, C.; Echavarren, A. M. Synthesis 2005, 167. For related representative reports, see: (d) Trost, B. M.; Dyker, G.; Kulawiec, R. J. J. Am. Chem. Soc. 1990, 112, 7809. (e) Wang, Y.; Finn, M. G. J. Am. Chem. Soc. 1995, 117, 8045. (f) Merlic, C. A.; Pauly, M. E. J. Am. Chem. Soc. 1996, 118, 11319. (g) Ohe, K.; Kojima, M.; Yonehara, K.; Uemura, S. Angew. Chem., Int. Ed. Engl. 1996, 35, 1823. (h) Maeyama, K.; Iwasawa, N. J. Am. Chem. Soc. 1998, 120, 1928.

⁽²⁹⁾ (a) Boger, D. L. J. Heterocycl. Chem. **1998**, 35, 1003. (b) Jayakumar, S.; Ishar, M. P. S.; Mahajan, M. P. Tetrahedron **2002**, 58, 379.

⁽³⁰⁾ For reports on the synthesis of alkynyl imines via a two-step procedure involving imidoyl chlorides, see: (a) Ried, W.; Erle, H.-E. Chem. Ber. 1979, 112, 640. (b) Austin, W. B.; Bilow, N.; Kelleghan, W. J.; Lau, K. S. Y. J. Org. Chem. 1981, 46, 2280. (c) Lin, S.-Y.; Sheng, H.-Y.; Huang, Y.-Z. Synthesis 1991, 235. For related reports, see: (d) Kel'in, A. V.; Sromek, A. W.; Gevorgyan, V. J. Am. Chem. Soc. 2001, 123, 2074. (e) Van den Hoven, B. G.; Alper, H. J. Am. Chem. Soc. 2001, 123, 10214.

⁽³¹⁾ (a) Baraznenok, I. L.; Nenajdenko, V. G.; Balenkova, E. S. *Tetrahedron* **2000**, *56*, 3077. (b) Charette, A. B.; Grenon, M. *Can. J. Chem.* **2001**, *79*, 1694.

⁽³²⁾ See the Experimental Section for details.

⁽³³⁾ Myers, A. G.; Tom, N. J.; Fraley, M. E.; Cohen, S. B.; Madar, D. J. J. Amer. Chem. Soc. 1997, 119, 6072.

⁽³⁴⁾ (a) Gilbert, J. D.; Wilkinson, G. J. Chem. Soc. (A) **1969**, 1749. (b) Blackmore, T.; Bruce, M. I.; Stone, F. G. A. J. Chem. Soc. (A) **1971**, 2376. b) Bruce, M. I.; Windsor, N. J., Aust. J. Chem. **1977**, 30, 1601.

⁽³⁵⁾ (a) For a related report on metal-catalyzed isomerization of *N*-aryl benzamide derived terminal alkynes (i.e., 4a) using W(CO)₅•THF (20-100 mol%) to afford 2-aryl quinolines after treatment with NMO, see ref. 4c. (b) For a report on Ru-catalyzed C2-vinylation of pyridines, see: Murakami, M.; Hori, S. J. Am. Chem. Soc. 2003, 125, 4720.
⁽³⁶⁾ Walker, S. D.; Barder, T. E.; Martinelli, J. R.; Buchwald, S. L. Angew. Chem. Int. Ed. 2004, 43, 1871.

⁽³⁷⁾ Albers, M. O.; Robinson, D. J.; Shaver, A.; Singleton, E. Organometallics 1986, 5, 2199.

⁽³⁸⁾ For related discussions on the steric and electronic effects of phosphines used with complex 6 in Ru-vinylidene formation, see: Trost, B. M.; Rhee, Y. H. J. Am. Chem. Soc. 1999, 121, 11680.

⁽³⁹⁾ McNair, A. M.; Schrenk, J. L.; Mann, K. R. Inorg. Chem. 1984, 23, 2633.

⁽⁴⁰⁾ Incomplete deuterium incorporation at the expected site and leakage to the alternate position (2-7%, eq 1-3) may in part be due to an intermolecular scrambling after cycloisomerization of 8 and prior to release of product 1 (i.e., the intermediate 9 or the metal hydride tautomer in Scheme 12).

 ⁽⁴¹⁾ Schneider, D.; Werner, H. Angew. Chem., Int. Ed. Engl. 1991, 30, 700.
 ⁽⁴²⁾ Movassaghi, M.; Hill, M. D. J. Am. Chem. Soc. 2006, 128, 4592.

Experimental Section

General Procedures. All reactions were performed in oven-dried or flame-dried round bottomed flasks, modified Schlenk (Kjeldahl shape) flasks, or glass pressure vessels. The flasks were fitted with rubber septa and reactions were conducted under a positive pressure of argon. Stainless steel syringes or cannulae were used to transfer air- and moisture-sensitive liquids. Flash column chromatography was performed as described by Still et al. using silica gel (60-Å pore size, 32-63 µm, standard grade, Sorbent Technologies) or non-activated alumina gel (80-325 mesh, chromatographic grade, EM Science).¹ Analytical thin-layer chromatography was performed using glass plates pre-coated with 0.25 mm 230-400 mesh silica gel or neutral alumina gel impregnated with a fluorescent indicator (254 nm). Thin layer chromatography plates were visualized by exposure to ultraviolet light and/or by exposure to an ethanolic phosphomolybdic acid (PMA), an acidic solution of *p*-anisaldehyde (anis), an aqueous solution of ceric ammonium molybdate (CAM), an aqueous solution of potassium permanganate (KMnO₄) or an ethanolic solution of ninhydrin followed by heating (<1 min) on a hot plate (~250 °C). Organic solutions were concentrated on Büchi R-200 rotary evaporators at ~10 Torr (house vacuum) at 25–35 °C, then at ~0.5 Torr (vacuum pump) unless otherwise indicated.

Commercial reagents and solvents were used as received with the following Materials. Dichloromethane, diethyl ether, tetrahydrofuran, acetonitrile, and toluene were exceptions: purchased from J.T. Baker (CycletainerTM) and were purified by the method of Grubbs et al. under positive argon pressure.² Ammonium hexafluorophosphate was dried at 150 °C under vacuum (~0.5 torr) for 24 h and stored in a glove box under an atmosphere of dinitrogen. The molarity of *n*-butyllithium solutions was determined by titration using diphenylacetic acid as an indicator (average of three determinations).³ Hünig's base and 2-chloropyridine were distilled from calcium hydride and stored sealed under an argon atmosphere. The starting amides were prepared by acylation of the corresponding anilines⁴ or via previously reported copper-catalyzed C-N bond-forming reactions.^{5,6} The ruthenium complex 5 (CpRu(PPh₃)₂Cl) is commercially available and was prepared on large scale according to a literature procedure.⁷ 2-Dicyclohexylphosphino-2',6'-dimethoxy-1,1'-biphenyl (SPhos)⁸ is commercially available and we thank the Buchwald group for providing samples for initial studies. Solutions of Copper (I)

⁽¹⁾ Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923.

⁽²⁾ Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. Organometallics 1996, 15, 1518. ⁽³⁾ Kofron, W. G.; Baclawski, L. M. J. Org. Chem. 1976, 41, 1879.

⁽⁴⁾ For a general procedure, see: DeRuiter, J.; Swearingen, B. E.; Wandrekar, V.; Mayfield, C. A. J. Med. Chem. 1989, 32, 1033.

⁽⁵⁾ For the general procedure used for the synthesis of all N-vinyl amides, see: Jiang, L.; Job, G. E.; Klapars, A.; Buchwald, S. L. Org. Lett. 2003, 5, 3667.

⁽⁶⁾ For related reports, see: (a) Wolfe, J. P.; Wagaw, S.; Marcoux, J.-F.; Buchwald, S. L. Acc. Chem. Res. 1998, 31, 805. (b) Hartwig, J. F. Acc. Chem. Res. 1998, 31, 852. (c) Hartwig, J. F. Angew. Chem., Int. Ed. 1998, 37, 2046. (d) Yang, B. H.; Buchwald, S. L. J. Organomet. Chem. 1999, 576, 125. (e) Muci, A. R.; Buchwald, S. L. Top. Curr. Chem. 2002, 219, 131. (f) Beletskaya, I. P.; Cheprakov, A. V. Coordin. Chem. Rev. 2004, 248, 2337. (g) Dehli, J. R.; Legros, J.; Bolm, C. Chem. Commun. 2005, 973.

⁽⁷⁾ (a) Bruce, M. I.; Hameister, C.; Swincer, A. G.; Wallis, R. C. Inorganic Syntheses 1982, 21, 78. Also, see: (b) Gilbert, J. D.; Wilkinson, G. J. Chem. Soc. (A) 1969, 1749, (c) Blackmore, T.; Bruce, M. I.; Stone, F. G. A. J. Chem. Soc. (A) 1971, 2376, and (d) Bruce, M. I.; Windsor, N. J., Aust. J. Chem. 1977, 30, 1601.

⁽⁸⁾ (a) Walker, S. D.; Barder, T. E.; Martinelli, J. R.; Buchwald, S. L. Angew. Chem. Int. Ed. 2004, 43, 1871. (b) Tomori, H.; Fox, J. M.; Buchwald, S. L. J. Org. Chem. 2000, 65, 5334.

trimethylysilylacetylide were prepared immediately prior to use according to literature procedure.⁹

Instrumentation. Proton nuclear magnetic resonance (¹H NMR) spectra were recorded with a Varian inverse probe 500 INOVA spectrometer. Chemical shifts are recorded in parts per million from internal tetramethylsilane on the δ scale and are referenced from the residual protium in the NMR solvent (CHCl₃: δ 7.27, C₆HD₅: δ 7.16). Data is reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, q = quartet, m = multiplet), coupling constant(s) in Hertz, integration, assignment]. Carbon-13 nuclear magnetic resonance spectra were recorded with a Varian 500 INOVA spectrometer and are referenced from the carbon resonances of the solvent (CDCl₃: δ 77.2, benzene-*d*₆: δ 128.0). Data is reported as follows: chemical shift [multiplicity (s = singlet, d = quartet, m = multiplet), coupling constant(s) in Hertz, assignment]. Infrared data were obtained with a Perkin-Elmer 2000 FTIR and are reported as follows: [frequency of absorption (cm⁻¹), intensity of absorption (s = strong, m = medium, w = weak, br = broad), assignment]. Combustion analysis was performed by Atlantic Microlab, Incorporated. We are grateful to Dr. Li Li for obtaining the mass spectroscopic data at the Department of Chemistry's Instrumentation Facility, Massachusetts Institute of Technology.

⁽⁹⁾ Enda, J.; Kuwajima, I. J. Am. Chem. Soc. 1985, 107, 5495.



<u>N-Phenyl-2-phenyl-4-trimethylsilyl-1-azabut-1-en-3-yne (3a, Table 2, entry 1):</u>

Trifluoromethanesulfonic anhydride (2.51 mL, 15.2 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2a** (2.50 g, 12.7 mmol, 1 equiv) and 2-chloropyridine (4.81 mL, 50.7 mmol, 4.00 equiv) in CH₂Cl₂ (25 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (5.50 g, 34.2 mmol, 2.70 equiv) in THF (60 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes \rightarrow 7% EtOAc in hexanes) to give the alkynyl imine **3a**¹⁰ as a yellow oil (3.42 g, 97%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.21–8.18 (m, 2H, ArH (o -C=N)), 7.51–7.45 (m, 3H, ArH), 7.40–7.36 (m, 2H, ArH), 7.17 (tt, 1H, J = 7.5, 1.1 Hz, ArH), 7.14–7.11 (m, 2H, ArH), 0.14 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	151.7, 150.1, 137.0, 131.4, 128.6, 128.5, 128.3, 125.0, 120.9, 105.4, 97.5, -0.5.
FTIR (neat) cm^{-1} :	3063 (m), 3030 (w), 2960 (m), 1588 (s, C=N), 1565 (s).
HRMS (ESI):	calcd for C ₁₈ H ₂₀ NSi [M+H] ⁺ : 278.1360, found: 278.1365.
Analysis	calcd for C ₁₈ H ₁₉ NSi: C, 77.93; H, 6.90; N, 5.05, found: C, 77.97; H, 6.87; N, 5.10.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.59 (UV, CAM).

⁽¹⁰⁾ For a prior synthesis, see: Ito, Y.; Inouye, M.; Murakami, M. Chem. Lett. 1989, 7, 1261.



2-Phenylquinoline (1a, Table 2, entry 1):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (587 mg, 3.60 mmol, 1.00 equiv), $CpRuCl(PPh_3)_2$ (262 mg, 0.35 mmol, 0.10 equiv) and SPhos (148 mg, 0.35 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3a** (1.00 g, 3.60 mmol, 1 equiv) and toluene (18 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 20-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (15 \rightarrow 50 % EtOAc in hexanes) to afford the quinoline **1a** as a pale yellow solid (668 mg, 90%).¹¹

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.25 (d, 1H, $J = 8.5$ Hz, ArH), 8.21–8.16 (m, 3H, ArH), 7.90 (d, 1H, $J = 8.5$ Hz, ArH), 7.85 (d, 1H, $J = 8.2$ Hz, ArH), 7.75 (ddd, 1H, $J = 8.5$, 7.0, 1.5 Hz, ArH), 7.57–7.52 (m, 3H, ArH), 7.48 (tt, 1H, $J = 7.3$, 1.2 Hz, ArH).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	157.5, 148.4, 139.8, 137.0, 129.9, 129.9, 129.5, 129.0, 127.8, 127.7, 127.3, 126.5, 119.2.
FTIR (neat) cm^{-1} :	3189 (s), 3055 (w), 2091 (s), 1617 (w), 1597 (s), 1491 (m), 1447 (s).
HRMS (EI):	calcd for C ₁₅ H ₁₁ N [M] ⁺ : 205.0886, found: 205.0885.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.51 (UV, CAM).

⁽¹¹⁾ Sangu, K.: Fuchibe, K.: Akiyama, T. Org. Lett. 2004, 6, 353.



N-(4-Methoxyphenyl)-2-phenyl-4-trimethylsilyl-1-azabut-1-en-3-yne (3b, Table 2, entry 2):

Trifluoromethanesulfonic anhydride (174 μ L, 1.06 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2b** (200 mg, 0.88 mmol, 1 equiv) and 2chloropyridine (333 μ L, 3.52 mmol, 4.00 equiv) in CH₂Cl₂ (1.8 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (383 mg, 2.38 mmol, 2.70 equiv) in THF (5.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes→10% EtOAc in hexanes) to afford the alkynyl imine **3b** as a yellow oil (240 mg, 89%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.21–8.15 (m, 2H, ArH (<i>o</i> -C=N)), 7.50–7.42 (m, 3H, ArH (<i>m</i> -C=N, <i>p</i> -C=N)), 7.31–7.25 (m, 2H, ArH), 6.96–6.90 (m, 2H, ArH), 3.83 (s, 3H, CH ₃ O), 0.21 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	157.7, 148.1, 144.2, 137.6, 131.0, 128.5, 128.1, 123.3, 113.7, 104.8, 98.1, 55.6, -0.4.
FTIR (CDCl ₃) cm ^{-1} :	3066 (w), 2962 (m), 2838 (w), 1605 (m, C=N), 1559 (m), 1503 (s), 1251 (s).
HRMS (ESI):	calcd for C ₁₉ H ₂₂ NOSi [M+H] ⁺ : 308.1465, found: 308.1472.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.49 (UV, CAM).



6-Methoxy-2-phenylquinoline (1b, Table 2, entry 2):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (40 mg, 0.24 mmol, 1.0 equiv), CpRuCl(PPh₃)₂ (18 mg, 0.024 mmol, 0.10 equiv) and SPhos (10 mg, 0.024 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3b** (75 mg, 0.24 mmol, 1 equiv) and toluene (1.2 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (20 % EtOAc in hexanes) to afford the quinoline **1b** as a pale yellow solid (52 mg, 91%).¹¹

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.16–8.10 (m, 3H, ArH), 8.08 (d, 1H, $J = 9.2$ Hz, ArH), 7.85 (d, 1H, $J = 8.9$ Hz, ArH), 7.56–7.50 (m, 2H, ArH), 7.48–7.42 (m, 1H, ArH), 7.40 (dd, 1H, $J = 9.2$, 2.7 Hz, ArH (CH ₃ OCCH)), 7.11 (d, 1H, $J = 3.1$ Hz, ArH(CH ₃ OCCH)), 3.99 (s, 3H, CH ₃ O).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	157.8, 155.2, 144.5, 139.9, 135.7, 131.3, 129.1, 129.0, 128.3, 127.5, 122.5, 119.4, 105.1, 55.7.
FTIR (neat) cm ⁻¹ :	3057 (w), 2958 (m), 2836 (w), 1621 (m), 1599 (m), 1493 (s).
HRMS (EI):	calcd for $C_{16}H_{13}NO[M]^+$: 235.0992, found: 235.0984.
TLC (20% EtOAc in hexanes), R _f :	0.39 (UV, CAM).



<u>N-(2-Methoxyphenyl)-2-phenyl-4-trimethylsilyl-1-azabut-1-en-3-yne (3c, Table 2, entry 3):</u>

Trifluoromethanesulfonic anhydride (218 μ L, 1.32 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2c** (250 mg, 1.10 mmol, 1 equiv) and 2chloropyridine (416 μ L, 4.40 mmol, 4.00 equiv) in CH₂Cl₂ (2.2 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (477 mg, 2.97 mmol, 2.70 equiv) in THF (5.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes→10% EtOAc in hexanes) to afford the alkynyl imine **3c** as a yellow oil (326 mg, 97%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.24–8.20 (m, 2H, ArH (o -C=N)), 7.52–7.44 (m, 3H, ArH (p -C=N, m -C=N)), 7.14 (ddd, 1H, J = 8.3, 7.3, 1.8 Hz, ArH), 7.01 (dd, 1H, J = 7.6, 2.1 Hz, ArH), 6.98–6.94 (m, 2H, ArH), 3.85 (s, 3H, OCH ₃), 0.11 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	151.6, 150.5, 141.5, 136.8, 131.3, 128.4, 128.4, 125.7, 120.8, 120.5, 111.5, 104.9, 97.7, 55.9, -0.5.
FTIR (neat) cm ⁻¹ :	3064 (w), 2959 (m), 2900 (w), 2834 (w), 1590 (s), 1567 (s), 1488 (s).
HRMS (ESI):	calcd for C ₁₉ H ₂₂ NOSi [M+H] ⁺ : 308.1465, found: 308.1477.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.49 (UV, KMnO ₄).



8-Methoxy-2-phenylquinoline (1c, Table 2, entry 3):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (53 mg, 0.33 mmol, 1.0 equiv), $CpRuCl(PPh_3)_2$ (24 mg, 0.033 mmol, 0.10 equiv) and SPhos (13 mg, 0.033 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3c** (100 mg, 0.33 mmol, 1 equiv) and toluene (1.6 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (20 % EtOAc in hexanes) to afford the quinoline **1c** as a pale yellow solid (70 mg, 91%).¹²

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.24–8.18 (m, 3H, ArH), 7.93 (d, 1H, $J = 8.5$ Hz, ArH), 7.53 (t, 2H, $J = 7.5$ Hz, ArH), 7.49–7.41 (m, 3H, ArH), 7.10 (dd, 1H, $J = 7.6$, 1.1 Hz, ArH), 4.13 (s, 3H, OCH ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	156.5, 155.7, 140.3, 140.0, 137.0, 129.4, 129.0, 128.5, 127.9, 126.7, 119.7, 119.5, 108.2, 56.3.
FTIR (neat) cm ⁻¹ :	3060 (m), 2934 (m), 2834 (w), 1599 (s), 1556 (s), 1467 (s), 1258 (s).
HRMS (ESI):	calcd for C ₁₆ H ₁₄ NO [M+H] ⁺ : 236.1070, found: 236.1066.
TLC (20% EtOAc in hexanes), R _f .	0.23 (UV, KMnO ₄).

⁽¹²⁾ For a prior synthesis, see: Collin, J.; Jaber, N.; Lannou, M. I. Tetrahedron Lett. 2001, 42, 7405.



<u>N-(3-Trifluoromethylphenyl)-2-phenyl-4-trimethylsilyl-1-azabut-1-en-3-yne (3d, Table 2, entry 4):</u>

Trifluoromethanesulfonic anhydride (251 μ L, 1.52 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide 2d (337 mg, 1.27 mmol, 1 equiv) and 2chloropyridine (481 μ L, 5.08 mmol, 4.00 equiv) in CH₂Cl₂ (2.5 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (552 mg, 3.43 mmol, 2.70 equiv) in THF (7.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes→7.5% EtOAc in hexanes) to afford the alkynyl imine 3d as a pale yellow oil (321 mg, 73%). Desilylation of 3d afforded less than 10% yield of the corresponding terminal alkyne due to rapid nucleophilic addition (i.e., water and methanol) to this terminal alkynyl imine.

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.20–8.17 (m, 2H, ArH (<i>o</i> -C=N)), 7.55–7.47 (m, 4H, ArH), 7.45–7.42 (m, 1H, ArH), 7.40 (s, 1H, ArH), 7.29–7.25 (m, 1H, ArH), 0.06 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	152.3, 151.5, 136.6, 131.8, 131.1 (q, <i>J</i> = 32.0 Hz), 129.2, 128.6, 128.5, 124.6 (q, <i>J</i> = 271.9 Hz), 124.6, 121.5 (q, <i>J</i> = 3.9 Hz), 117.8 (q, <i>J</i> = 3.8 Hz), 106.8, 96.9, -0.7.
FTIR (neat) cm ⁻¹ :	3067 (m), 2962 (m), 1589 (s, C=N), 1567 (s), 1330 (s).
HRMS (ESI):	calcd for $C_{19}H_{19}F_3NSi [M+H]^+$: 346.1233, found: 346.1236.
TLC (20% EtOAc in hexanes) $R_{\rm c}$	0.57 (UV, CAM)



2-Phenyl-7-trifluoromethyl-quinoline (1d, Table 2, entry 4):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (35 mg, 0.22 mmol, 1.0 equiv), $CpRuCl(PPh_3)_2$ (16 mg, 0.022 mmol, 0.10 equiv) and SPhos (9 mg, 0.022 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3d** (75 mg, 0.22 mmol, 1 equiv) and toluene (1.1 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (20 % EtOAc in hexanes) to afford the quinoline 1d as a pale yellow solid (53 mg, 89%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.46 (s, 1H, ArH), 8.30 (d, 1H, $J = 8.2$ Hz, ArH), 8.22–8.18 (m, 2H, ArH(o -C=N)), 8.03 (d, 1H, $J =$ 8.6 Hz, ArH), 7.97 (d, 1H, $J = 8.6$ Hz, ArH), 7.72 (dd, 1H, $J = 8.5$, 1.8 Hz, ArH), 7.59–7.54 (m, 2H, ArH), 7.54–7.50 (m, 1H, ArH).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	158.9, 147.5, 139.1, 136.8, 131.7 (q, <i>J</i> = 32.6 Hz), 130.1, 129.2, 128.9, 127.8 (q, <i>J</i> = 4.6 Hz), 127.5, 125.3, 123.2, 122.1 (q, <i>J</i> = 2.9 Hz), 121.0.
FTIR (neat) cm^{-1} :	3067 (w), 3037 (w), 2917 (w), 1603 (s), 1316 (s).
HRMS (EI):	calcd for $C_{16}H_{10}F_{3}N[M]^{+}$: 273.0760, found: 273.0750.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.46 (UV, CAM).



<u>N-Phenyl-2-(thiophen-2-yl)-4-trimethylsilyl-1-azabut-1-en-3-yne (3e, Table 2, entry 5):</u>

Trifluoromethanesulfonic anhydride (195 μ L, 1.18 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2e** (200 mg, 0.98 mmol, 1 equiv) and 2-chloropyridine (372 μ L, 3.94 mmol, 4.00 equiv) in CH₂Cl₂ (2.0 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (428 mg, 2.66 mmol, 2.70 equiv) in THF (5.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes→7.5% EtOAc in hexanes) to afford the alkynyl imine **3e** as a pale yellow oil (227 mg, 81%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.73 (dd, 1H, $J = 3.7$, 1.2 Hz, SCHCHCHCC=N), 7.48 (dd, 1H, $J = 5.0$, 0.9 Hz, SCHCHCHCC=N), 7.38–7.33 (m, 2H, ArH), 7.20–7.14 (m, 3H, ArH), 7.12 (dd, 1H, $J = 4.6$, 3.7 Hz, SCHCHCHCC=N), 0.16 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	150.5, 144.7, 144.0, 131.6, 130.6, 128.5, 127.7, 125.4, 121.6, 103.7, 96.7, -0.5.
FTIR (neat) cm^{-1} :	3078 (m), 2960 (s), 1563 (s, C=N), 1425 (s), 1252 (s).
HRMS (ESI):	calcd for $C_{16}H_{18}NSSi [M+H]^+$: 284.0924, found: 284.0937.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.70 (UV, CAM).



2-Thiophen-2-ylquinoline (1e, Table 2, entry 5):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (58 mg, 0.35 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (26 mg, 0.035 mmol, 0.10 equiv) and SPhos (15 mg, 0.035 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3e** (100 mg, 0.35 mmol, 1 equiv) and toluene (1.8 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (20 % EtOAc in hexanes) to afford the quinoline **1e** as a pale yellow solid (56 mg, 75%).¹³

¹H NMR (500 MHz, CDCl₃, 20°C):

8.16 (d, 1H, $J = 8.9$ Hz, Ar H), 8.10 (d, 1H, $J = 9.2$
Hz, ArH), 7.82 (d, 1H, $J = 8.6$ Hz, ArH), 7.79 (d,
1H, $J = 7.9$ Hz, ArH (SCHCHCHCC=N)), 7.75 (dd,
1H, <i>J</i> = 3.7, 0.9 Hz, ArH (SCHCHCHCC=N)), 7.71
(ddd, 1H, $J = 8.5$, 7.0, 1.5 Hz, ArH), 7.52–7.47 (m,
2H, ArH), 7.18 (dd, 1H, $J = 4.9$, 3.7 Hz, ArH
(SCHCHCHC=N)).

¹³C NMR (125 MHz, CDCl₃, 20°C):

FTIR (neat) cm^{-1} :

HRMS (ESI):

TLC (20% EtOAc in hexanes), R_f :

128.3, 127.7, 127.3, 126.3, 126.0, 117.8.

152.5, 148.3, 145.6, 136.8, 130.0, 129.4, 128.8,

3079 (w), 3042 (w), 1962 (w), 1617 (m), 1597 (s), 1561 (m), 1502 (s).

calcd for $C_{13}H_{10}NS [M+H]^+$: 212.0528, found: 212.0533.

0.40 (UV, KMnO₄).

⁽¹³⁾ Fürstner, A.; Leitner, A.; Méndez, M.; Krause, H. J. Am. Chem. Soc. 2002, 124, 13856.



<u>N-Phenyl-2-cyclohexyl-4-trimethylsilyl-1-azabut-1-en-3-yne (3f, Table 2, entry 6):</u>

Trifluoromethanesulfonic anhydride (98 μ L, 0.59 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2f** (100 mg, 0.49 mmol, 1 equiv) and 2-chloropyridine (186 μ L, 1.97 mmol, 4.00 equiv) in CH₂Cl₂ (1.0 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (214 mg, 1.33 mmol, 2.70 equiv) in THF (3.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on neutralized silica gel (1% Et₃N/hexanes) to afford the alkynyl imine **3f** as a pale yellow oil (123 mg, 89%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	7.32–7.28 (m, 2H, ArH), 7.12–7.07 (m, 1H, ArH), 6.96–6.93 (m, 2H, ArH), 2.47 (tt, 1H, $J = 23.2$, 6.9 Hz, (CH ₂) ₂ CHC=N), 2.01 (br dd, 2H, $J = 12.6$, 2.0 Hz, ^c C ₆ H ₁₁), 1.86 (dt, 2H, $J = 13.1$, 3.4 Hz, ^c C ₆ H ₁₁), 1.76–1.69 (m, 1H, ^c C ₆ H ₁₁), 1.52 (qd, 2H, $J = 12.5$, 3.4 Hz, ^c C ₆ H ₁₁), 1.36 (qt, 2H, $J = 12.5$, 3.2 Hz, ^c C ₆ H ₁₁), 1.26 (qt, 1H, $J = 12.5$, 3.2 Hz, ^c C ₆ H ₁₁), 0.06 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	159.6, 151.7, 128.4, 124.5, 120.6, 104.0, 98.2, 48.3, 30.3, 26.1, 26.0, -0.5.
FTIR (neat) cm ⁻¹ :	3049 (w), 2933 (s), 2856 (m), 1660 (w), 1594 (s), 1484 (m).
HRMS (ESI):	calcd for C ₁₈ H ₂₆ NSi [M+H] ⁺ : 284.1829, found: 284.1836.
TLC (20% EtOAc in hexanes) R_{c}	$0.67 (UV KMnO_4)$


2-Cyclohexyl-quinoline (1f, Table 2, entry 6):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (40 mg, 0.25 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (18 mg, 0.025 mmol, 0.10 equiv) and SPhos (10 mg, 0.025 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3f** (70 mg, 0.25 mmol, 1 equiv) and toluene (1.2 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (10 % EtOAc in hexanes) to afford the quinoline **1f** as a pale yellow solid (36 mg, 69%).¹⁴

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.09 (d, 1H, $J = 8.5$ Hz, ArH), 8.06 (d, 1H, $J = 8.5$ Hz, ArH), 7.78 (d, 1H, $J = 8.2$ Hz, ArH), 7.69 (ddd, 1H, $J = 8.5$, 7.0, 1.5 Hz, ArH), 7.49 (ddd, 1H, $J =$ 7.9, 7.0, 1.2 Hz, ArH), 7.35 (d, 1H, $J = 8.6$ Hz, ArH), 2.94 (tt, 1H, $J = 12.1$, 3.4 Hz, (CH ₂) ₂ CHCN), 2.08–2.00 (m, 2H, ^c C ₆ H ₁₁), 1.98–1.87 (m, 2H, ^c C ₆ H ₁₁), 1.86–1.76 (m, 1H, ^c C ₆ H ₁₁), 1.64 (qd, 2H, J = 12.5, 3.1 Hz, ^c C ₆ H ₁₁), 1.49 (qt, 2H, $J = 13.1$, 3.4 Hz, ^c C ₆ H ₁₁), 1.35 (qt, 1H, $J = 12.8$, 3.4 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	167.0, 148.0, 136.5, 129.4, 129.1, 127.6, 127.2, 125.8, 119.8, 47.9, 33.0, 26.7, 26.3.
FTIR (neat) cm^{-1} :	3058 (w), 2924 (s), 2850 (m), 1720 (w), 1619 (w), 1601 (m), 1502 (m).
HRMS (EI):	calcd for C ₁₅ H ₁₇ N [M] ⁺ : 211.1356, found: 211.1358.
TLC (20% EtOAc in hexanes), R _f :	0.50 (UV, KMnO ₄).

⁽¹⁴⁾ For a prior synthesis, see: Ishikura, M.; Oda, I.; Kamada, M.; Terashima, M. Synth. Comm. 1987, 17, 959.



<u>N-Phenyl-2-(*tert*-butyl)-4-trimethylsilyl-1-azabut-1-en-3-yne (3g, Table 2, entry 7):</u>

Trifluoromethanesulfonic anhydride (220 μ L, 1.33 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2g** (197 mg, 1.11 mmol, 1 equiv) and 2chloropyridine (420 μ L, 4.44 mmol, 4.00 equiv) in CH₂Cl₂ (2.2 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (482 mg, 3.00 mmol, 2.70 equiv) in THF (5.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes→7.5% EtOAc in hexanes) to afford the alkynyl imine **3g** as a pale yellow oil (236 mg, 83%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.32–7.28 (m, 2H, Ar H), 7.11–7.06 (m, 1H, Ar H), 6.92–6.88 (m, 2H, Ar H), 1.31 (s, 9H, C(C H ₃) ₃), 0.50 (s, 9H, Si(C H ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	162.9, 152.2, 128.5, 124.2, 120.3, 104.7, 97.3, 39.8, 28.0, -0.5.
FTIR (neat) cm ⁻¹ :	3080 (w), 2967 (s), 2868 (m), 1931 (w), 1593 (s), 1477 (s).
HRMS (ESI):	calcd for C ₁₆ H ₂₄ NSi [M+H] ⁺ : 258.1673, found: 258.1675.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.69 (UV, CAM).



(1-tert-Butyl-prop-2-ynylidene)-phenyl-amine (4g, Table 2, entry 7):

Anhydrous potassium carbonate (41 mg, 0.30 mmol, 0.2 equiv) was added to a solution of imine 3g (380 mg, 1.48 mmol, 1 equiv) in methanol (5.0 mL) and stirred at 23 °C. After 25 min., the volatiles were removed under reduced pressure and the residue was purified by flash column chromatography on silica gel (7.5 % EtOAc in hexanes) to afford the alkynyl imine 4g as a yellow solid (259 mg, 95%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	7.32–7.28 (m, 2H, ArH), 7.12 (tt, 1H, $J = 7.3$, 2.3 Hz, ArH), 6.92–6.88 (m, 2H, ArH), 3.14 (s, 1H, C=CH), 1.33 (s, 9H, C(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	161.9, 151.8, 128.7, 124.4, 119.9, 85.4, 76.2, 40.2, 27.8.
FTIR (neat) cm ⁻¹ :	2962 (s), 2925 (s), 1734 (m), 1717 (m), 1684 (s), 1653 (m).
HRMS (EI):	calcd for $C_{13}H_{15}N[M]^+$: 185.1199, found: 185.1193.
TLC (20% EtOAc in hexanes), R _f :	0.66 (UV, KMnO4).



2-tert-Butyl-quinoline (1g, Table 2, entry 7):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (62 mg, 0.38 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (28 mg, 0.038 mmol, 0.10 equiv) and SPhos (16 mg, 0.038 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine 4g (70 mg, 0.38 mmol, 1 equiv) and toluene (1.9 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (7.5 % EtOAc in hexanes) to afford the quinoline 1g as a pale yellow solid (51 mg, 73%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.09 (d, 1H, $J = 8.2$ Hz, ArH), 8.07 (d, 1H, $J = 8.5$ Hz, ArH), 7.78 (dd, 1H, $J = 7.9$, 1.2 Hz, ArH), 7.65 (ddd, 1H, $J = 8.5$, 7.0, 1.5 Hz, ArH), 7.54 (d, 1H, $J = 8.9$ Hz, ArH), 7.48 (ddd, 1H, $J = 8.1$, 6.7, 0.9 Hz, ArH), 1.49 (s, 9H, C(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	147.6, 136.1, 129.6, 129.2, 127.4, 126.6, 125.8, 118.5, 100.0, 38.3, 30.4.
FTIR (neat) cm ⁻¹ :	3061 (w), 2960 (s), 2868 (m), 1619 (m), 1601 (s), 1565 (w), 1503 (s).
HRMS (ESI):	calcd for C ₁₃ H ₁₆ N [M+H] ⁺ : 186.1277, found: 186.1283.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.46 (UV, KMnO ₄)



<u>N-Phenyl-2-(morpholin-4-yl)-4-trimethylsilyl-1-azabut-1-en-3-yne (3h, Table 2, entry 8):</u>

Trifluoromethanesulfonic anhydride (144 μ L, 0.87 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2h** (150 mg, 0.73 mmol, 1 equiv) and 2chloropyridine (275 μ L, 2.91 mmol, 4.00 equiv) in CH₂Cl₂ (1.5 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and Hünig's base (190 μ L, 1.09 mmol, 1.50 equiv) was added via syringe followed by a freshly prepared solution of copper (I) trimethylysilylacetylide (316 mg, 1.96 mmol, 2.70 equiv) in THF (3.0 mL) at 0 °C via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on neutralized silica gel (50% EtOAc in hexanes) to afford the alkynyl imidate **3h** as a burgandy colored oil (166 mg, 80%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	7.27–7.23 (m, 2H, ArH), 7.01 (tt, 1H, $J = 7.3$, 1.1 Hz, ArH), 6.92–6.88 (m, 2H, ArH), 3.79–3.75 (m, 4H, OCH ₂ CH ₂ N), 3.69–3.66 (m, 4H, OCH ₂ CH ₂ N), 0.04 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	151.2, 144.2, 128.5, 122.9, 122.3, 104.2, 93.2, 66.8, 45.9, -0.7.
FTIR (neat) cm ⁻¹ :	2967 (m), 2924 (w), 2861 (m), 1577 (s, C=N), 1420 (m) 1247 (s).
HRMS (ESI):	calcd for C ₁₆ H ₂₃ N ₂ OSi [M+H] ⁺ : 287.1574, found: 287.1580.
TLC (50% EtOAc in hexanes), R _f .	0.70 (UV, KMnO ₄).



2-Morpholin-4-yl-quinoline (1h, Table 2, entry 8):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (43 mg, 0.26 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (19 mg, 0.026 mmol, 0.10 equiv) and SPhos (11 mg, 0.026 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3h** (75 mg, 0.26 mmol, 1 equiv) and toluene (1.3 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (20 % EtOAc in hexanes) to afford the quinoline **1h** as a pale yellow solid (35 mg, 62%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	7.95 (d, 1H, $J = 8.7$ Hz, ArH), 7.74 (d, 1H, $J = 7.6$ Hz, ArH), 7.65 (dd, 1H, $J = 7.9$, 1.5 Hz, ArH), 7.56 (ddd, 1H, $J = 8.5$, 7.0, 1.5 Hz, ArH), 7.27 (ddd, 1H, J = 8.0, 6.7, 0.9 Hz, ArH), 6.99 (d, 1H, $J = 9.2$ Hz, ArH), 3.87 (t, 4H, $J = 4.9$, OCH ₂ CH ₂ N), 3.73 (t, 4H, $J = 4.9$ Hz, OCH ₂ CH ₂ N).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	157.6, 147.8, 137.7, 129.8, 127.4, 126.9, 123.4, 122.8, 109.4, 67.0, 45.7.
FTIR (neat) cm ⁻¹ :	3047 (w), 2972 (w), 2917 (w), 2858 (m), 1617 (s), 1605 (m).
HRMS (ESI):	calcd for C ₁₃ H ₁₄ N ₂ O [M] ⁺ : 215.1179, found: 215.1178.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.21 (UV, KMnO ₄).



<u>N-(trans-β-Styryl)-2-phenyl-4-trimethylsilyl-1-azabut-1-en-3-yne (3i, Table 2, entry 9):</u>

Trifluoromethanesulfonic anhydride (133 μ L, 0.81 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2i** (150 mg, 0.67 mmol, 1 equiv) and 2-chloropyridine (254 μ L, 2.69 mmol, 4.00 equiv) in CH₂Cl₂ (1.3 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (292 mg, 1.81 mmol, 2.70 equiv) in THF (5.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes-7.5% EtOAc in hexanes) to afford the alkynyl imine **3i** as a pale yellow solid (159 mg, 78%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.34 (d, 1H, $J = 13.1$ Hz, NCHCHC), 8.18–8.13 (m, 2H, ArH), 7.56 (d, 2H, $J = 7.3$ Hz, ArH), 7.48–7.42 (m, 3H, ArH), 7.39 (t, 2H, $J = 7.6$ Hz, ArH), 7.31 (tt, 1H, $J = 7.3$, 1.2 Hz, ArH), 7.16 (d, 1H, $J = 13.4$ Hz, NCHCHC), 0.39 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	148.0, 139.7, 137.1, 136.7, 133.8, 131.0, 129.0, 128.6, 128.5, 128.0, 127.4, 108.7, 96.8, 0.0.
FTIR (CDCl ₃) cm ^{-1} :	3155 (m), 3062 (m), 3028 (m), 2963 (m), 2902 (m), 1815 (m), 1794 (m), 1643 (w), 1622 (m), 1490 (s).
HRMS (ESI):	calcd for C ₂₀ H ₂₂ NSi [M+H] ⁺ : 304.1516, found: 304.1526.
TLC (20% EtOAc in hexanes), R _f :	0.59 (UV, CAM).



2,5-Diphenyl-pyridine (1i, Table 2, entry 9):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (40 mg, 0.25 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (18 mg, 0.025 mmol, 0.10 equiv) and SPhos (10 mg, 0.025 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3i** (75 mg, 0.25 mmol, 1 equiv) and toluene (1.3 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on neutralized silica gel (20 % EtOAc in hexanes) to afford the pyridine **1i** as a pale yellow solid (44 mg, 77%).¹⁵

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.96 (d, 1H, $J = 2.5$ Hz, ArH), 8.08–8.05 (m, 2H, ArH), 7.98 (dd, 1H, $J = 8.2$, 2.5 Hz, ArH), 7.83 (d, 1H, $J = 8.2$ Hz, ArH), 7.68–7.64 (m, 2H, ArH), 7.54–7.50 (m, 4H, ArH), 7.47–7.42 (m, 2H, ArH).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	156.4, 148.3, 139.2, 137.9, 135.3, 135.1, 129.3, 129.2, 129.0, 128.3, 127.2, 127.0, 120.6.
FTIR (CH ₂ Cl ₂) cm ⁻¹ :	2926 (w), 1735 (s), 1654 (s), 1594 (w), 1472 (s), 1371 (s).
HRMS (ESI):	calcd for C ₁₇ H ₁₄ N [M+H] ⁺ : 232.1121, found: 232.1126.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.43 (UV, CAM).

⁽¹⁵⁾ Berthiol, F.; Kondolff, I.; Doucet, H.; Santelli, M. J. Organomet. Chem. 2004, 689, 2786.



(<u>2-Naphthalen-1-yl-vinyl)-[1-phenyl-3-(trimethyl-silanyl)-prop-2-ynylidene]-amine (3j, Table 2, entry 10):</u>

Trifluoromethanesulfonic anhydride (127 μ L, 0.77 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2j** (175 mg, 0.64 mmol, 1 equiv) and 2chloropyridine (242 μ L, 2.56 mmol, 4.00 equiv) in CH₂Cl₂ (1.3 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (278 mg, 1.73 mmol, 2.70 equiv) in THF (4.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes→7.5% EtOAc in hexanes) to afford the alkynyl imine **3j** as a pale yellow oil (197 mg, 87%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.39 (dd, 1H, $J = 12.8$, 0.6 Hz, NCHCHC), 8.31 (d, 1H, $J = 8.2$ Hz, 8.22–8.18 (m, 2H, ArH), 7.94 (d, 1H, $J = 13.1$ Hz, NCHCHC), 7.89 (d, 1H, $J = 7.9$ Hz, ArH), 7.85 (d, 2H, $J = 7.6$ Hz, ArH), 7.60–7.47 (m, 6H, ArH), 0.38 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	148.8, 142.4, 137.7, 134.6, 134.3, 132.3, 131.6, 131.2, 129.5, 129.4, 129.2, 128.7, 127.1, 126.8, 126.4, 124.8, 124.6, 109.3, 97.5, 0.6.
FTIR (neat) cm ⁻¹ :	3059 (m), 2959 (m), 2141 (w), 1810 (w), 1692 (m), 1644 (w), 1590 (w), 1251 (s).
HRMS (ESI):	calcd for C ₂₄ H ₂₄ NSi [M+H] ⁺ : 354.1673, found: 354.1675.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.77 (UV, CAM).



5-Naphthalen-1-yl-2-phenyl-pyridine (1j, Table 2, entry 10):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (28 mg, 0.17 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (12 mg, 0.017 mmol, 0.10 equiv) and SPhos (7 mg, 0.017 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3j** (60 mg, 0.17 mmol, 1 equiv) and toluene (0.9 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on neutralized silica gel (7.5 % EtOAc in hexanes) to afford the pyridine **1j** as a pale brown solid (44 mg, 88%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.87–8.84 (m, 1H, Ar H), 8.11 (d, 2H, <i>J</i> = 7.02 Hz, Ar H), 7.98–7.87 (m, 5H, Ar H), 7.61–7.46 (m, 7H, Ar H).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	156.4, 150.6, 139.3, 138.4, 136.4, 134.9, 134.0, 131.7, 129.3, 129.0, 128.7, 128.7, 127.6, 127.1, 126.7, 126.3, 125.6, 125.6, 120.1.
FTIR (neat) cm ⁻¹ :	3058 (m), 2930 (w), 1950 (w), 1595 (m), 1548 (m), 1476 (s), 1396 (m).
HRMS (ESI):	calcd for $C_{21}H_{16}N [M+H]^+$: 282.1277, found: 282.1289.
TLC (15% EtOAc in hexanes). Re	$0.46 (UV, KMnO_4)$



[2-(3,4-Dimethoxy-phenyl)-vinyl]-[1-phenyl-3-(trimethyl-silanyl)-prop-2-ynylidene]-amine (3k, Table 2, entry 11):

Trifluoromethanesulfonic anhydride (122 μ L, 0.74 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2k** (175 mg, 0.62 mmol, 1 equiv) and 2-chloropyridine (234 μ L, 2.47 mmol, 4.00 equiv) in CH₂Cl₂ (1.2 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (268 mg, 1.67 mmol, 2.70 equiv) in THF (4.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (15% EtOAc in hexanes) to afford the alkynyl imine **3k** as a yellow oil (216 mg, 96%).

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¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.26 (d, 1H, $J = 13.1$ Hz, NCHCHC), 8.16–8.12 (m, 2H, NCHCHC, ArH), 7.46–7.42 (m, 3H, ArH), 7.14–7.08 (m, 3H, ArH), 6.86 (d, 1H, $J = 8.2$ Hz, ArH), 3.96 (s, 3H, OCH ₃), 3.94 (s, 3H, OCH ₃), 0.38 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	149.6, 149.2, 146.6, 138.2, 137.1, 133.8, 130.7, 129.6, 128.5, 127.8, 121.4, 111.3, 108.8, 108.1, 96.9, 56.1, 55.8, -0.0.
FTIR (neat) cm ⁻¹ :	3057 (w), 2958 (m), 2835 (w), 1599 (s), 1578 (m), 1512 (s), 1267 (s).
HRMS (ESI):	calcd for $C_{22}H_{26}NO_2Si [M+H]^+$: 364.1727, found: 364.1733.
TLC (20% EtOAc in hexanes), R _f :	0.52 (UV, CAM).



5-(3,4-Dimethoxy-phenyl)-2-phenyl-pyridine (1k, Table 2, entry 11):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (45 mg, 0.28 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (20 mg, 0.028 mmol, 0.10 equiv) and SPhos (11 mg, 0.028 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3k** (100 mg, 0.28 mmol, 1 equiv) and toluene (1.4 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on neutralized silica gel (30 % EtOAc in hexanes) to afford the pyridine **1k** as a yellow solid (64 mg, 79%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.92 (d, 1H, $J = 2.1$ Hz, ArH), 8.07–8.03 (m, 2H, ArH), 7.93 (dd, 1H, $J = 8.2$, 2.5 Hz, ArH), 7.81 (d, 1H, $J = 8.2$ Hz, ArH), 7.53–7.49 (m, 2H, ArH), 7.44 (tt, 1H, $J = 7.3$, 1.1 Hz, ArH), 7.22 (dd, 1H, $J = 8.2$, 2.1 Hz, ArH), 7.15 (d, 1H, $J = 1.8$ Hz, ArH), 7.02 (d, 1H, $J = 8.2$ Hz, ArH), 3.99 (s, 3H, OCH ₃), 3.96 (s, 3H, OCB ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	155.7, 149.5, 149.3, 147.9, 139.1, 134.8, 134.8, 130.5, 129.0, 128.9, 126.8, 120.3, 119.5, 111.8, 110.1, 56.1, 56.1.
FTIR (neat) cm ⁻¹ :	3055 (w), 3005 (w), 2966 (m), 2837 (m), 1602 (s), 1590 (s), 1522 (s), 1150 (s), 1022 (s).
HRMS (ESI):	calcd for $C_{19}H_{18}NO_2 [M+H]^+$: 292.1332, found: 292.1337.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.19 (UV, KMnO ₄).



<u>N-(Cyclohexen-1-yl)-2-phenyl-4-trimethylsilyl-1-azabut-1-en-3-yne (31, Table 2, entry 12):</u>

Trifluoromethanesulfonic anhydride (148 μ L, 0.89 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2l** (150 mg, 0.75 mmol, 1 equiv) and 2chloropyridine (282 μ L, 2.98 mmol, 4.00 equiv) in CH₂Cl₂ (1.5 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (323 mg, 2.01 mmol, 2.70 equiv) in THF (3.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes→7.5% EtOAc in hexanes) to afford the alkynyl imine **3l** as a yellow oil (157 mg, 75%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.09–8.05 (m, 2H, ArH), 7.48–7.39 (m, 3H, ArH), 5.21 (t, 1H, $J = 4.0$ Hz, NC=CH), 2.29–2.23 (m, 2H, CH ₂), 2.22–2.16 (m, 2H, CH ₂), 1.82–1.75 (m, 2H, CH ₂), 1.70–1.64 (m, 2H, CH ₂), 0.28 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	148.7, 147.5, 137.2, 130.8, 128.4, 128.0, 110.3, 103.7, 97.6, 27.9, 24.7, 23.1, 22.5, -0.1.
FTIR (neat) cm^{-1} :	3063 (w), 2927 (s), 2857 (m), 1663 (m), 1562 (m), 1448 (m), 1273 (s), 1251 (s).
HRMS (ESI):	calcd for C ₁₈ H ₂₄ NSi [M+H] ⁺ : 282.1673, found: 282.1685.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.61 (UV, KMnO4).



2-Phenyl-5,6,7,8-tetrahydro-quinoline (11, Table 2, entry 12):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (43 mg, 0.27 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (19 mg, 0.027 mmol, 0.10 equiv) and SPhos (11 mg, 0.027 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3l** (75 mg, 0.27 mmol, 1 equiv) and toluene (1.3 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (20 % EtOAc in hexanes) to afford the pyridine **1l** as a pale yellow solid (41 mg, 73%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	7.97–7.94 (m, 2H, Ar H), 7.48–7.36 (m, 5H, Ar H), 3.00 (t, 2H, $J = 6.4$ Hz, C H ₂), 2.82 (t, 2H, $J = 6.4$ Hz, C H ₂), 1.98–1.92 (m, 2H, C H ₂), 1.89–1.83 (m, 2H, C H ₂).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	157.4, 154.9, 140.1, 137.6, 130.9, 128.8, 128.5, 127.0, 118.1, 33.1, 28.8, 23.4, 23.0.
FTIR (neat) cm^{-1} :	3061 (w), 3032 (w), 2935 (s), 2860 (m), 1590 (m), 1566 (m), 1460 (s), 1434 (m), 1253 (m).
HRMS (ESI):	calcd for C ₁₅ H ₁₆ N [M+H] ⁺ : 210.1277, found: 210.1279.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.48 (UV, KMnO ₄).



<u>N-(Thiophen-2-yl)-2-(phenyl)-4-trimethylsilyl-1-azabut-1-en-3-yne (3m, Table 2, entry 13):</u>

Trifluoromethanesulfonic anhydride (195 μ L, 1.18 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2m** (200 mg, 0.98 mmol, 1 equiv) and 2chloropyridine (372 μ L, 3.94 mmol, 4.00 equiv) in CH₂Cl₂ (2.0 mL) at -78 °C. After 10 min., a freshly prepared solution of copper (I) trimethylysilylacetylide (428 mg, 2.66 mmol, 2.70 equiv) in THF (5.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 10 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes->7.5% EtOAc in hexanes) to afford the alkynyl imine **3m** as a yellow oil (181 mg, 65%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.19–8.13 (m, 2H, ArH), 7.48–7.43 (m, 3H, ArH), 7.42 (dd, 1H, J = 3.8, 1.4 Hz, CHS), 7.32 (dd, 1H, J = 5.5, 1.4 Hz, CHCHCHS), 7.07 (dd, 1H, J = 5.5, 3.8 Hz, CHCHS), 0.40 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	152.2, 140.9, 137.7, 130.6, 129.2, 128.5, 127.9, 125.4, 125.3, 112.0, 99.2, -0.6.
FTIR (neat) cm ⁻¹ :	3064 (m), 2960 (s), 2144 (m), 2067 (m), 1537 (m), 1412 (s), 1272 (s), 1253 (s).
HRMS (ESI):	calcd for $C_{16}H_{18}NSSi [M+H]^+: 284.0924$, found: 284.0934.
TLC (20% EtOAc in hexanes), R _f :	0.63 (UV, CAM).



6-Phenyl-thieno[2,3-b]pyridine (1m, Table 2, entry 13):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (35 mg, 0.21 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (15 mg, 0.021 mmol, 0.10 equiv) and SPhos (9 mg, 0.021 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **1m** (60 mg, 0.21 mmol, 1 equiv) and toluene (1.1 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (20 % EtOAc in hexanes) to afford the thienopyridine **1m** as a pale yellow solid (44 mg, 99%).¹⁶

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.15 (d, 1H, <i>J</i> = 8.2 Hz, Ar H), 8.11–8.09 (m, 2H, Ar H), 7.78 (d, 1H, <i>J</i> = 8.6 Hz, Ar H), 7.54–7.49 (m, 3H, Ar H , SC H), 7.44 (tt, 1H, <i>J</i> = 7.3, 1.2 Hz, Ar H), 7.30 (d, 1H, <i>J</i> = 5.8 Hz, SCHC H).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	162.4, 154.7, 139.4, 131.7, 131.3, 129.1, 129.0, 127.4, 127.2, 121.5, 117.0.
FTIR (neat) cm ⁻¹ :	3065 (w), 2919 (w), 2850 (w), 1717 (w), 1572 (s), 1557 (s), 1478 (s), 1453 (s), 1425 (s), 1361 (m), 1332 (s), 1260 (m), 1108 (s).
HRMS (ESI):	calcd for $C_{13}H_{10}NS [M+H]^+$: 212.0534, found: 212.0535.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.45 (UV, KMnO ₄).

⁽¹⁶⁾ Taylor, E. C.; Macor, J. E. J. Org. Chem. 1987, 52, 4280.



[1-Phenyl-3-(trimethyl-silanyl)-prop-2-ynylidene]-thiophen-3-yl-amine (3n, Table 2, entry 14):

Trifluoromethanesulfonic anhydride (294 μ L, 1.77 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2n** (300 mg, 1.48 mmol, 1 equiv) and 2chloropyridine (559 μ L, 5.90 mmol, 4.00 equiv) in CH₂Cl₂ (3.0 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (641 mg, 3.99 mmol, 2.70 equiv) in THF (5.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes→10% EtOAc in hexanes) to afford the alkynyl imine **3n** as a yellow oil (411 mg, 98%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.17 (dd, 2H, $J = 7.6$, 1.8 Hz, ArH), 7.50–7.43 (m, 4H, ArH, CHS)), 7.38 (d, 1H, $J = 5.2$ Hz, CHS), 7.31 (dd, 1H, $J = 5.2$, 3.4 Hz, CHCHS), 0.29 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	149.9, 147.4, 137.6, 131.0, 128.5, 128.1, 124.9, 124.2, 115.6, 106.1, 98.8, -0.4.
FTIR (neat) cm ⁻¹ :	3740 (w), 3108 (m), 3063 (m), 2960 (s), 2899 (m), 2147 (m), 1645 (w), 1586 (m), 1556 (s), 1271 (s).
HRMS (ESI):	calcd for C ₁₆ H ₁₈ NSSi [M+H] ⁺ : 284.0929, found: 284.0933.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.74 (UV, KMnO4).



5-Phenyl-thieno[3,2-b]pyridine (1n, Table 2, entry 14):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (58 mg, 0.35 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (26 mg, 0.035 mmol, 0.10 equiv) and SPhos (15 mg, 0.035 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3n** (100 mg, 0.35 mmol, 1 equiv) and toluene (1.8 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (15 % EtOAc in hexanes) to afford the thienopyridine **1n** as a pale yellow solid (73 mg, 97%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.26 (d, 1H, $J = 8.6$ Hz, ArH), 8.08 (d, 1H, $J = 7.3$ Hz, ArH), 7.78 (d, 1H, $J = 5.5$ Hz, CHS), 7.73 (d, 1H, $J = 8.5$ Hz, ArH), 7.65 (d, 1H, $J = 5.5$ Hz, CHCHS)), 7.52 (t, 2H, $J = 7.0$ Hz, ArH), 7.45 (t, 1H, $J = 7.3$ Hz, ArH).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	156.4, 155.5, 139.8, 131.7, 131.0, 131.0, 129.0, 128.9, 127.4, 125.5, 116.5.
FTIR (neat) cm^{-1} :	3071 (s), 1906 (w), 1564 (s), 1544 (s), 1397 (s), 1280 (s), 1158 (s).
HRMS (ESI):	calcd for C ₁₃ H ₁₀ NS [M+H] ⁺ : 212.0534, found: 212.0534.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.53 (UV, KMnO ₄).



(5,6-Dihydro-4H-pyran-3-yl)-[1-phenyl-3-(trimethyl-silanyl)-prop-2-ynylidene]-amine (30, Table 2, entry 15):

Trifluoromethanesulfonic anhydride (292 μ L, 1.77 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **20** (300 mg, 1.48 mmol, 1 equiv) and 2-chloropyridine (559 μ L, 5.90 mmol, 4.00 equiv) in CH₂Cl₂ (3.0 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (641 mg, 3.99 mmol, 2.70 equiv) in THF (5.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (5% EtOAc in hexanes) to afford the alkynyl imine **30** as a pale yellow oil (416 mg, 99%).

¹ H NMR (500 MHz, C ₆ D ₆ , 20 °C) δ:	8.44–8.40 (m, 2H, ArH), 7.44 (bs, 1H, NC=CHO), 7.24–7.20 (m, 2H, ArH), 7.12–7.08 (m, 1H, ArH), 3.58 (t, 2H, $J = 5.2$ Hz, CH ₂ O), 2.80 (t, 2H, $J = 6.4$ Hz, NCCH ₂), 1.46–1.40 (m, 2H, CH ₂ CH ₂ O), 0.14 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, C ₆ D ₆ , 20 °C) δ:	149.7, 139.9, 138.2, 134.3, 129.9, 127.9, 103.2, 102.1, 100.5, 66.6, 24.8, 22.5, -0.2.
FTIR (neat) cm ⁻¹ :	3063 (w), 2960 (m), 2920 (m), 2850 (w), 1783 (w), 1724 (s), 1675 (w), 1251 (s), 1176 (s).
HRMS (ESI):	calcd for $C_{17}H_{22}NOSi$ [M+H] ⁺ : 284.1465, found: 284.1471.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.67 (UV, KMnO₄).



6-Phenyl-3,4-dihydro-2H-pyrano[3,2-b]pyridine (10, Table 2, entry 15):

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (52 mg, 0.32 mmol, 1.0 equiv), $CpRu(PPh_3)_2Cl$ (23 mg, 0.032 mmol, 0.10 equiv) and SPhos (13 mg, 0.032 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **30** (90 mg, 0.32 mmol, 1 equiv) and toluene (1.6 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 19 h, the reaction vessel was allowed to cool to ambient temperature and the mixture was transferred to a recovery flask with a 10-mL portion of dichloromethane. This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (1 % Et₃N/CH₂Cl₂) to afford the dihydropyranopyridine **10** as a pale yellow solid (47 mg, 70%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	7.92–7.89 (m, 2H, ArH), 7.49–7.47 (m, 1H, ArH), 7.46–7.42 (m, 2H, ArH), 7.35 (tt, 1H, $J = 7.3$, 1.5 Hz, ArH), 7.16 (d, 1H, $J = 8.6$, ArH), 4.24 (t, 2H, $J = 5.5$ Hz, CH ₂ O), 3.05 (t, 2H, $J = 6.6$ Hz, NCCH ₂), 2.19–2.14 (m, 2H, CH ₂ CH ₂ O).
¹³ C NMR (125 MHz, C ₆ D ₆ , 20°C):	151.5, 149.6, 144.0, 140.3, 129.2, 128.5, 127.1, 124.6, 119.6, 66.6, 29.2, 22.8.
FTIR (neat) cm^{-1} :	3061 (w), 3034 (w), 2949 (m), 2874 (m), 1575 (m), 1471 (s), 1458 (s), 1258 (s).
HRMS (EI):	calcd for C ₁₄ H ₁₃ NO [M] ⁺ : 211.0992, found: 211.0987.
TLC (20% EtOAc in hexanes), R _f :	0.40 (UV, KMnO ₄).



<u>N-(N-Triisopropylsilylpyrrol-3-yl)-2-phenyl-4-trimethylsilyl-1-azabut-1-en-3-yne (3p, Table 2, entry 16):</u>

Trifluoromethanesulfonic anhydride (175 μ L, 1.05 mmol, 1.20 equiv) was added via syringe over 1 min to a stirred mixture of amide **2p** (300 mg, 0.88 mmol, 1 equiv) and 2chloropyridine (332 μ L, 3.50 mmol, 4.00 equiv) in CH₂Cl₂ (1.8 mL) at -78 °C. After 5 min., the reaction mixture was warmed to 0 °C. After 20 min., the solution was cooled to -78 °C and a freshly prepared solution of copper (I) trimethylysilylacetylide (380 mg, 2.37 mmol, 2.70 equiv) in THF (5.0 mL) at 0 °C was added via cannula. The reaction mixture was kept at -78 °C for 5 min and then warmed to 0 °C. After 10 min., the crude reaction mixture was filtered through celite (2 cm diam. × 3 cm ht.) and the filtrate was concentrated under reduced pressure. The residue was purified by flash column chromatography on silica gel (100% hexanes→10% EtOAc in hexanes) to afford the alkynyl imine **3p** as a yellow oil (304 mg, 82%).

'Η NMR (500 MHz, C ₆ D ₆ , 20 °C) δ:	8.63–8.60 (m, 2H, ArH), 7.74 (dd, 1H, $J = 3.1$, 1.2 Hz, CHN), 7.50 (dd, 1H, $J = 2.4$, 1.5 Hz, CHN), 7.29–7.24 (m, 2H, ArH), 7.14 (tt, 1H, $J = 7.3$, 1.2 Hz, ArH), 6.73 (dd, 1H, $J = 3.1$, 2.1 Hz, CHCHN), 1.17 (septet, 3H, $J = 7.3$ Hz, Si(CH(CH ₃) ₂) ₃), 0.95 (d, 18, $J = 7.3$ Hz, Si(CH(CH ₃) ₂) ₃), 0.22 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	140.1, 139.2, 138.4, 129.5, 128.3, 127.4, 125.1, 124.4, 105.8, 105.2, 101.0, 18.0, 11.8, -0.2.
FTIR (Neat) cm ⁻¹ :	2948 (m), 2868 (m), 2141 (w), 1547 (w), 1514 (w), 1488 (m), 1252 (s), 1099 (s).
HRMS (ESI):	calcd for $C_{25}H_{39}N_2Si_2 [M+H]^+$: 423.2646, found: 423.2656.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.76 (UV, KMnO ₄).



(1-Phenyl-prop-2-ynylidene)-[1-(triisopropyl-silanyl)-1H-pyrrol-3-yl]-amine (4p, Table 2, entry 16):

Anhydrous potassium carbonate (23 mg, 0.17 mmol, 0.2 equiv) was added to a solution of imine 3p (380 mg, 0.83 mmol, 1 equiv) stirred in methanol (2.8 mL) at 23 °C. After 25 min., the volatiles were removed under reduced pressure and the residue was purified by flash column chromatography on silica gel (7.5 % EtOAc in hexanes) to afford the alkynyl imine 4p as a yellow solid (274 mg, 94%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.16–8.12 (m, 2H, ArH), 7.44–7.36 (m, 3H, ArH), 7.29 (t, 1H, $J = 1.8$ Hz, CHN), 7.18 (dd, 1H, $J = 3.1$, 1.2 Hz, CHCHN), 6.77 (dd, 1H, $J = 2.7$, 2.1 Hz, CHCHN), 3.69 (s, 1H, C=CH), 1.48 (septet, 3H, $J = 7.6$ Hz, Si(CH(CH ₃) ₂) ₃), 1.14 (d, 18H, $J = 7.5$ Hz, Si(CH(CH ₃) ₂) ₃).
¹³ C NMR (125 MHz, C ₆ D ₆ , 20°C):	139.8, 139.7, 139.4, 130.1, 128.8, 128.0, 125.4, 124.6, 107.2, 86.7, 80.4, 18.1, 12.0.
FTIR (neat) cm ⁻¹ :	3296 (m, C≡C–H), 2947 (s), 2868 (s), 2088 (w, C≡C), 1547 (w), 1488 (s), 1098 (s).
HRMS (ESI):	calcd for C ₂₂ H ₃₁ N ₂ Si [M+H]: 351.2251, found: 351.2264.
TLC (30% EtOAc in hexanes), R _f :	0.83 (UV, KMnO4).



<u>5-Phenyl-1-(triisopropyl-silanyl)-1H-pyrrolo[3,2-b]pyridine (1p, Table 2, entry 16):</u>

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate (33 mg, 0.20 mmol, 1.0 equiv), CpRu(PPh₃)₂Cl (7.3 mg, 0.01 mmol, 0.05 equiv)¹⁷ and SPhos (4.1 mg, 0.01 mmol, 0.05 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Imine **3p** (70 mg, 0.20 mmol, 1 equiv) and toluene (1.0 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 90 °C. After 4 h, the reaction vessel was allowed to cool to ambient temperature and was diluted with dichloromethane (3 mL). This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on basic alumina (15 % EtOAc in hexanes) to afford the azaindole 1p as a pale brown solid (46 mg, 65%).

¹H NMR (500 MHz, CDCl₃, 20°C):

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.46–8.42 (m, 2H, ArH), 7.58 (dd, 1H, $J = 8.7, 0.9$ Hz, ArH), 7.53 (d, 1H, $J = 8.5$ Hz, ArH), 7.39–7.34 (m, 2H, ArH), 7.22 (tt, 1H, $J = 7.3, 1.2$ Hz, ArH), 7.14 (d, 1H, $J = 3.4$ Hz, NCHCHC), 7.11 (dd, 1H, $J = 3.4, 0.6$ Hz, NCHCHC), 1.32 (septet, 3H, $J = 7.6$ Hz, Si(CH(CH ₃) ₂) ₃), 0.93 (d, 18H, $J = 7.6$ Hz, Si(CH(CH ₃) ₂) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	151.9, 151.4, 141.8, 135.5, 133.3, 129.2, 128.5, 127.8, 121.3, 114.5, 107.8, 18.3, 13.0.
FTIR (neat) cm^{-1} :	3062 (w), 2948 (m), 2868 (m), 1561 (w), 1510 (w), 1465 (m), 1407 (s), 1138 (m).
HRMS (ESI):	calcd for C ₂₂ H ₃₁ N ₂ Si [M+H]: 351.2251, found: 351.2245.

TLC (20% EtOAc in hexanes), $R_{\rm f}$:

0.58 (UV, KMnO₄).

⁽¹⁷⁾ Use of higher catalyst loadings led to more *N*-desilylation of the product without improvement in yield of 1p.



<u>3-Deutero-2-phenylquinoline (1a-d₁, Equation 4):</u>

An oven-dried pressure vessel containing a magnetic stir bar was charged with ammonium hexafluorophosphate- d_4 (42 mg, 0.25 mmol, 1.0 equiv), CpRu(PPh₃)₂Cl (18 mg, 0.025 mmol, 0.10 equiv) and SPhos (10 mg, 0.025 mmol, 0.10 equiv) under a nitrogen atmosphere in a glovebox and the flask sealed and brought out of the glovebox. Freshly distilled benzene (2 × 3 mL) was added to the pressure vessel and subsequently removed in vacuo. Imine **3a** (70 mg, 0.25 mmol, 1 equiv) and toluene (1.3 mL) were subsequently added via syringe. The flask was flushed with argon, sealed, stirred, and placed in an oil bath at 105 °C. After 21 h, the reaction vessel was allowed to cool to ambient temperature and was diluted with dichloromethane (5 mL). This solution was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (5 % EtOAc in hexanes) to afford the quinoline **1a**- d_1 as a pale yellow solid (46 mg, 89%). ¹H and ¹³C NMR analysis indicates the presence of **1a**-C3-H², **1a**-C4-H², and **1a** in a ratio of 68 : 7 : 25.

¹H NMR (500 MHz, CDCl₃, 20 °C; resonances corresponding to the C3-H² noted by *) δ:8.25

(d, 1H, J = 8.2 Hz, C4-H*), 8.24 (s, 1H, C4-H), 8.21–8.16 (m, 6H, C8-H, C8-H*, C10-H, C10-H*), 7.91 (d, 1H, J = 8.6 Hz, C3-H), 7.85 (dd, 1H, J =8.2, 1.2 Hz, C5-H, C5-H*), 7.74 (ddd, 1H, J = 8.2, 6.7, 1.2 Hz, C7-H, C7-H*), 7.57–7.52 (m, 6H, C6-H*, C6-H, C11-H, C11-H*), 7.48 (tt, 1H, J = 7.6, 1.2 Hz, C12-H, C12-H*).

¹³ C NMR (125 MHz, CDCl ₃ , 20 °C; resor	nances corresponding to the C3-H ² noted by *) δ :157.5
	(C2), 157.4 (C2*), 148.5 (C8'), 148.4 (C8'*), 139.9
	(C9), 139.8 (C9*), 136.9 (C4*), 136.8 (C4), 129.9
	(C7, C7*), 129.8 (C8, C8*), 129.5 (C12, C12*),
	129.0 (C11, C11*), 127.7 (C10, C10*), 127.6 (C5,
	C5*), 127.3 ((4', 4'*), 126.4 (6, 6*), 119.1 (C3),
	119.0 (t, $J = 21$ Hz, C3*).
FTIR (CDCl ₃) cm^{-1} :	3056 (m), 2918 (w), 2248 (w, C–D), 1962 (w), 1616 (m), 1589 (s), 1552 (s), 1589 (s), 1490 (s).
HRMS (ESI):	calcd for C ₁₅ H ₁₁ DN [M+H]: 207.1027, found: 207.1036.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.51 (UV, CAM).

Chapter II

Single-Step Synthesis of Pyrimidine Derivatives

Introduction and Background

Azaheterocycles constitute a very important class of compounds. In particular, pyrimidine derivatives include a large number of natural products, pharmaceuticals, and functional materials (Figure 1).¹ Several examples of pharmaceutically important compounds include trimethoprim,² sulfadiazine,³ Gleevec (imatinib mesilate),⁴ and Xeloda (capecitabine).⁵ Natural and unnatural polymers also contain pyrimidine derivatives.^{1,6} While development of important methodologies for the synthesis of pyrimidines enjoys a rich history, the discovery of new strategies for the convergent synthesis of pyrimidines remains a vibrant area of chemical research.

In nature, the pyrimidine ring is synthesized from glutamine, bicarbonate, and aspartate.^{1b} These starting materials are converted to orotate (Figure 1), a ribonucleotide biosynthetic precursor, in four enzymatic reactions. In this sequence, carbamoyl phophate synthetase II transforms glutamine, ATP, and bicarbonate to carbamoyl phosphate. Subsequent condensation of carbamoyl phosphate with aspartate is catalyzed by aspartate transcarbamoylase, affording carbamoyl aspartate. Dihydroorotase promoted dehydration followed by oxidation with dihydroorotate dehydrogenase affords the ribonucleotide precursor, orotate.



Figure 1. Representative compounds containing a pyrimidine substructure.

In 1818, Brugnatelli synthesized the first pyrimidine derivative, alloxan, by nitric acid oxidative degradation of uric acid (Scheme 1).⁷ Another early report, by Frankland and Kolbe in 1848, described the first synthesis of cyanalkine by heating propionitrile with potassium metal (Scheme 1).⁸ Gabriel and Colman first isolated pyrimidine in 1899 by decarboxylation of pyrimidine-4-carboxylic acid.⁹ Since these early reports many important contributions describing a variety of synthetic strategies for preparation of pyrimidine derivatives have been published.

Many of these prevailing strategies rely on condensation of N–C–N fragments, most often amidines or guanidines, with 1,3-dicarbonyl derivatives (Scheme 2).^{1,10} Another versatile approach to pyrimidine synthesis utilizes N–C fragments. Nitriles are a common N–C source and have been used to form pyrimidines in many syntheses. Cyanamide is a particularly useful nitrile derivative in the synthesis of pyrimidines as illustrated in Scheme 3.¹¹

Brugnatelli alloxan synthesis:7

Frankland and Kolbe cyanalkine synthesis:8



Scheme 1. Early reports on the synthesis of pyrimidine derivatives.



Scheme 2. Representative synthesis of a pyrimidine by condensation of a N-C-N fragment and a diketone.¹⁰



Scheme 3. Representative use of cyanamide in condensation with acetoacetone for the synthesis of a pyrimidine.¹¹

With advances in cross-coupling chemistry, substituent modification of existing pyrimidine derivatives has recently gained considerable attention. Several reviews are available that describe advances in this important synthetic approach to pyrimidine derivatives.¹² Many of these procedures rely on inherent reactivity associated with the pyrimidine core (Scheme 4).^{12,13}

Additionally, activated heterocycle cross-coupling has become particularly important with recent advances (Scheme 4).^{12,14,15}

Metalation and electrophilic trapping of pyrimidines.¹³

Iron-catalyzed cross-coupling of activated pyrimidines.14





Scheme 4. Representative derivatization reactions for synthesis of pyrimidine derivatives.

Due to the importance of pyrimidines, our group is interested in new methodologies for their synthesis. Herein is reported a mild, convergent, and single-step procedure for the conversion of readily available *N*-vinyl and *N*-aryl amides¹⁶ to the corresponding substituted pyrimidines and quinazolines, respectively (eq 1).



Results and Discussion

We recently reported a mild procedure for electrophilic activation of sensitive amides en route to pyridine derivatives.^{17,18} We recognized the unique reactivity associated with electrophilic activation of amides using 2-chloropyridine $(2-ClPyr)^{19}$ in combination with trifluoromethanesulfonic anhydride (Tf_2O) .²⁰ The current study concerns trapping of highly activated amide derivatives with weakly nucleophilic nitriles to directly provide the corresponding pyrimidine derivatives (eq 1).

Benzamide 1a and cyclohexanecarbonitrile (2a) were used to identify the optimum reagent combination (Table 1). The use of 2-ClPyr and Tf₂O allowed direct conversion of

benzamide 1a to the corresponding quinazoline 3a (Table 1, entry 7).²¹ Other base additives largely returned the starting amide 1a after aqueous work-up. Superstoichiometric quantitites of 2-chloropyridine were found to have an inhibitory effect (Table 1, entry 8), perhaps by competing with the addition of the weakly nucleophilic nitrile 2a (vide infra). Under optimal conditions, the addition of Tf₂O (1.1 equiv) to a cold solution of amide 1a (1 equiv), nitrile 2a (1.1 equiv), and 2-ClPyr (1.2 equiv) in dichloromethane followed by warming afforded the desired quinazoline 3a in 88-90% isolated yield.

HN	OMe Hx	Tf ₂ O base additive	N	
Ph	0 N 1a 2a	CH₂Cl₂ 78-→45 °C	Ph N CHx 3a	
Entry	Base Additive	Equiv	Isolated Yield (%) ^a	
1	none	0	29	
2	Et ₃ N	1.2	0	
З	ⁱ Pr₂NEt	1.2	14	
4	pyridine	1.2	26	
5	2,6-lutidine	1.2	28	
6	2,4,6-collidine	1.2	19	
7	ethyl nicotinate	1.2	59	
8	3-bromopyridine	ə 1.2	54	
9	2-bromopyridine	ə 1.2	63	
10	2-chloropyridine	€ 1.0	72	
11	2-chloropyridin	ne 1.2	90	
12	2-chloropyridine	ə 3.0	81	

Table 1. Base additive screen.

^a Tf₂O (1.1 equiv), ^cHxCN (2a, 1.1 equiv), 45 °C, 16 h.

We next explored the substrate scope with a variety of secondary amides and nitriles (Table 2). While electron rich *N*-vinyl and *N*-aryl amides proceeded to afford the corresponding pyrimidine derivatives at ambient temperatures (Table 2, condition A), less reactive electron deficient substrates required heating (Table 2, conditions B and C). Electron donating and electron withdrawing substituents were tolerated in *N*-aryl benzamide derivatives (Table 2, entries 1-9). A wide range of nitriles, including electron rich and electron deficient benzonitriles in addition to saturated and unsaturated nitriles (Table 2, entries 10-16) were compatible with this chemistry. A variety of sensitive *N*-vinyl amides (Table 2, entries 14-21) served as

substrates, giving the corresponding pyrimidine derivatives. Significantly, the use of epimerizable substrates (Table 2, entries 20 and 21) provided the corresponding pyrimidine derivatives without loss in optical activity. For the most reactive substrates, the introduction of the nitrile prior to the low temperature activation of the secondary amide is essential for optimum results (Table 2, entry 18).²² In the case of highly reactive amides, excess nitrile was found to increase the yield of the desired pyrimidine product (Table 2, entry 17).

Ent	ry Amide			Nitrile	Conditions	Product	Yield (%) ^b
			ic De	N ^{#+9} Hx		R ^a N CHx	
		H-	H*				
1	Ph	Н	OMe		В		89
2	Ph	Н	Н		В		71
3	Ph	CF₃	Н		C		61
4	4-MeOPh	Н	OMe		Bc		87
5	4-NO ₂ Ph	Н	OMe		С		69
6	'Bu	Н	OMe		С		81
7	۴Hx	Н	OMe		С		73
8	N(CH ₂ CH ₂) ₂ O	Н	н		С		80
9	cyclohex-1-enyl	Н	OMe		В		88 ^d
				N			Me
10		~		R ^a = 4-NO ₂ Ph	С		86
11	<mark>،</mark> ۲	\sim	JMe	R ^a = 4-MeOPh	С		88
12	Ph N	</td <td></td> <td>R^a = 4-(CO₂Et)Ph</td> <td>С</td> <td></td> <td>74</td>		R ^a = 4-(CO ₂ Et)Ph	С		74
13	H			R ^a = (<i>E</i>)-C ₆ H ₄ CH=CH	I C		68
				N ^{FRa}			ı
14	0	<u>_</u> 0_		R ^a = ^c Hx	Α		92
15	U L]	R ^a = ^{<i>t</i>} Bu	Α		94 (88) ^e
16	Ph N	ſ~		Rª = (CH ₂) ₃ C≡CH	Α		77

Table 2.^a Substrate scope for single-step pyrimidine synthesis.



^a Uniform conditions unless otherwise noted. Tf₂O (1.1 equiv), 2-ClPyr (1.2 equiv), nitrile (1.1 equiv), CH₂Cl₂, heating: A = 23 °C, 1h; B = 45 °C, 16h; C = microwave, 140 °C, 20min. ^b Average of two experiments. ^c 18h. ^d 5 equiv of nitrile. ^e Gram-scale reaction. ^f 1h. ^gTBAF (1 equiv) used to desilylate the product. ^b 3 equiv of nitrile. ⁱ E.e. determined by chiral HPLC analysis of a derivative.

The dehydration of primary amides to the corresponding nitriles using Tf₂O and triethylamine has been reported.²³ Under optimum conditions, treatment of a solution of secondary amide **1a** (1 equiv), primary amide **4** (1.1 equiv), and 2-ClPyr (2.6 equiv) with Tf₂O (2.3 equiv) at -78 °C followed by microwave heating for 20 min., directly gave quinazoline **3a** in 74% yield (eq 2).²⁴ The ready availability of primary amides and their use as nitrile surrogates adds to the utility of this chemistry.



As illustrated in Scheme 5, amide activation and addition of 2-ClPyr to a protonated imidoyl triflate is envisioned to give the highly electrophilic 2-chloropyridinium adduct 5. In contrast to pyridine, 2-ClPyr was found not to add to Tf₂O.^{17,25} Monitoring of the reaction in entry 1 of Table 2 by ¹⁹F NMR spectroscopy revealed the presence of trifluoromethanesulfonate $(-79.6 \text{ ppm}, \text{CD}_2\text{Cl}_2)$ throughout the reaction, without involvement of a persistent imidoyl triflate. In situ ¹³C NMR monitoring of the amide activation using $1a^{-13}C=O$ (166.0 ppm, CD₂Cl₂) led to observation of a new broad resonance (149.8 ppm, CD₂Cl₂) prior to addition of the nitrile. React-IR monitoring during activation of amide 1a with Tf₂O in the absence of nitrile 2a revealed the consumption of 2-ClPyr (1580 cm^{-1}) with concomitant appearance of a new absorption band (1600 cm^{-1}). Introduction of the nitrile 2a to this mixture led to loss of this absorption band and simultaneous release of 2-chloropyridinium trifluoromethane-sulfonate (1620 cm^{-1}) and the trifluoromethanesulfonate salt of the desired product **3a** (1575 cm^{-1}) .²⁵ The broad ¹H, ¹³C, and ¹⁹F NMR resonances observed for the activated intermediate prior to addition of the nitrile suggests equilibration of 5 with the corresponding triflate adduct.²⁵ Reversible addition of nitrile²⁶ and expulsion of 2-ClPyr•HOTf to provide the nitrilium ion 6 is expected to occur en route to pyrimidine derivative 3.²⁷ The inhibitory effect of more nucleophilic base additives and excess 2-ClPyr in addition to the benefit of superstoichiometric quantities of nitrile are consistent with the proposed mechanism.



Scheme 5. Proposed mechanism for dehydrative cyclization and formation of pyrimidine derivatives.

Conclusion

We describe a single-step and convergent procedure for the synthesis of pyrimidine derivatives. This chemistry is compatible with a wide range of secondary amides and nitriles, and allows for unique transformations including that in equation 2. This methodology not only alleviates the need for isolation of activated amide derivatives but also does not require the use of stoichiometric Lewis acids.²¹ The compatibility of this chemistry with epimerizable substrates is

noteworthy and offers a valuable addendum to methodology for azaheterocycle synthesis.²⁸ Future work in this area includes synthesis of densely heteroatom-substituted pyrimidine derivatives and more challenging substrates.

- ⁽⁸⁾ Frankland, E.; Kolbe, H. Justis Liebigs Ann. Chem. 1848, 65, 269.
- ⁽⁹⁾ Gabriel, S.; Colman, J. Ber. Dtsch. Chem. Ges. 1899, 32, 1525.
- ⁽¹⁰⁾ Wendelin, W.; Schermanz, K.; Schweiger, K.; Fuchsgruber, A. Monatsh. Chem. 1983, 114, 1371.
- ⁽¹¹⁾ Miller, A. J. Org. Chem. 1984, 49, 4072.

⁽¹²⁾ (a) Chinchilla, R.; Nájera, R.; Yus, M. Chem. Rev. 2004, 104, 2667. (b) Turck, A.; Plé, N.; Mongin, F.; Quéguiner, G. Tetrahedron 2001, 57, 4489.

- (13) Plé, N.; Turck, A.; Heynderickx, A.; Quéguiner, G. Tetrahedron 1998, 54, 9701.
- ⁽¹⁴⁾ Fürstner, A.; Leitner, A.; Méndez, M.; Krause, H. J. Am. Chem. Soc. 2002, 124, 13856.
- ⁽¹⁵⁾ Billingsley, K.; Buchwald, S. L. J. Am. Chem. Soc. 2007, 129, 3358.
- ⁽¹⁶⁾ (a) Muci, A. R.; Buchwald, S. L. Top. Curr. Chem. 2002, 219, 131. (b) Hartwig, J. F. In Handbook of
- Organopalladium Chemistry for Organic Synthesis; Negishi, E., Ed.; Wiley-Interscience; New York, 2002; p. 1051. (c) Beletskaya, I. P.; Cheprakov, A. V. Coordin. Chem. Rev. 2004, 248, 2337. (d) Dehli, J. R.; Legros, J.; Bolm, C.

Chem. Commun. 2005, 973.

⁽¹⁷⁾ Movassaghi, M.; Hill, M. D. J. Am. Chem. Soc. 2006, 128, 4592.

⁽¹⁸⁾ For elegant studies on the activation of amides using Tf_2O and pyridine, see: (a) Charette, A. B.; Grenon, M.

Can. J. Chem. 2001, 79, 1694. (b) Charette, A. B.; Mathieu, S.; Martel, J. Org. Lett. 2005, 7, 5401.

⁽¹⁹⁾ (a) Myers, A. G.; Tom, N. J.; Fraley, M. E.; Cohen, S. B.; Madar, D. J. J. Am. Chem. Soc. 1997, 119, 6072. (b) Garcia, B. A.; Gin, D. Y. J. Am. Chem. Soc. 2000, 122, 4269.

⁽²⁰⁾ Baraznenok, I. L.; Nenajdenko, V. G.; Balenkova, E. S. Tetrahedron 2000, 56, 3077.

⁽²¹⁾ For quinazoline syntheses requiring stoichiometric activation (with TiCl₄, AlCl₃ or SnCl₄) of imidoyl chloride derivatives, see: (a) Zielinski, W.; Kudelko, A. *Monatsh. Chem.* 2000, 131, 895. (b) Kofanov, E. R.; Sosnina, V. V. Danilova, A. S.; Korolev, P. V. *Zh. Prik. Khim.* 1999, 72, 813. (c) Madroñero, R.; Vega, S. *Synthesis* 1987, 628.
⁽²²⁾ Absence of nitrile during activation of the amide substrate in entry 18 of Table 2 gave the desired product in diminished isolated yield (62%).

⁽²³⁾ Bose, D. S.; Jayalakshmi, B. Synthesis 1999, 64.

⁽²⁴⁾ The complete formation of nitrile 2a by treatment of amide 4 with Tf₂O and 2-ClPyr under the reaction conditions was confirmed independently.

⁽²⁵⁾ See Experimental Section for details.

⁽²⁶⁾ For an example of similar reactivity, the Ritter Reaction, see: (a) Ritter, J. J.; Minieri, P. P. J. Am. Chem. Soc. 1948, 70, 4045. (b) Ritter, J. J.; Kalish, J. J. Am. Chem. Soc. 1948, 70, 4048. For a review, see: Krimen, L. I.; Cota, D. J. Org. React. 1969, 17, 213.

(27) The conversion of 6 to 3 may be facilitated by net addition of 2-chloropyridinium triflate across the nitrilium ion.
(28) Movassaghi, M.; Hill, M. D. J. Am. Chem. Soc. 2006, 128, 14254.

⁽¹⁾ (a) Undheim, K.; Benneche, T. In Comprehensive Heterocyclic Chemistry II; Katritzky, A. R.; Rees, C. W.;

Scriven, E. F. V.; McKillop, A. Eds; Pergamon: Oxford, 1996; Vol. 6; p 93. b) Lagoja, I. M. Chem. & Biodiversity 2005, 2, 1. c) Michael, J. P. Nat. Prod. Rep. 2005, 22, 627. d) Joule, J. A.; Mills, K. In Heterocyclic Chemistry, 4th

ed.; Blackwell Science Ltd.: Cambridge, MA, 2000; p 194. e) Erian, A. W. Chem. Rev. 1993, 93, 1991.

⁽²⁾ Joffe, A. M.; Farley, J. D.; Linden, D.; Goldsand, G. Am. J. Med. 1989, 87, 332.

⁽³⁾ Petersen, E.; Schmidt, D. R. Expert Rev. Anti-infective Ther. 2003, 1, 175.

⁽⁴⁾ Nadal, E.; Olavarria, E. Int. J. Clin. Pract. 2004, 58, 511.

⁽⁵⁾ Blum, J. L. Oncologist 2001, 6, 56.

⁽⁶⁾ (a) Köytepe, S.; Pasahan, A.; Ekinci, E.; Seçkin, T. *Eur. Polym. J.* 2005, 41, 121. (b) Gompper, R.; Mair, H-J.; Polborn, K. Synthesis 1997, 696. (c) Kanbara, T.; Kushida, T.; Saito, N.; Kuwajima, I.; Kubota, K.; Yamamoto, T. Chem. Lett. 1992, 583.

⁽⁷⁾ (a) Brugnatelli, G. Giornale di fisica, chimica, et storia Naturale (Pavia) decada seconda 1818, 1, 117; (b) Brugnatelli, G. Ann. Chim. Phys. 1818, 8, 201.

Experimental Section

General Procedures. All previously unnumbered compounds are numbered in order of appearance beginning with "S." All reactions were performed in oven-dried or flame-dried round bottomed flasks, modified Schlenk (Kjeldahl shape) flasks, or glass pressure vessels. The flasks were fitted with rubber septa and reactions were conducted under a positive pressure of argon. Stainless steel syringes or cannulae were used to transfer air- and moisture-sensitive liquids. Flash column chromatography was performed as described by Still et al. using silica gel (60-Å pore size, 32-63 µm, standard grade, Sorbent Technologies) or non-activated alumina gel (80-325 mesh, chromatographic grade, EM Science).¹ Analytical thin-layer chromatography was performed using glass plates pre-coated with 0.25 mm 230-400 mesh silica gel or neutral alumina gel impregnated with a fluorescent indicator (254 nm). Thin layer chromatography plates were visualized by exposure to ultraviolet light and/or by exposure to an ethanolic phosphomolybdic acid (PMA), an acidic solution of *p*-anisaldehyde (anis), an aqueous solution of ceric ammonium molybdate (CAM), an aqueous solution of potassium permanganate (KMnO₄) or an ethanolic solution of ninhydrin followed by heating (<1 min) on a hot plate (~250 °C). Organic solutions were concentrated on Büchi R-200 rotary evaporators at ~10 Torr (house vacuum) at 25–35 °C, then at ~0.5 Torr (vacuum pump) unless otherwise indicated.

Materials. Commercial reagents and solvents were used as received with the following exceptions: Dichloromethane, diethyl ether, tetrahydrofuran, acetonitrile, and toluene were purchased from J.T. Baker (CycletainerTM) and were purified by the method of Grubbs et al. under positive argon pressure.² 2-chloropyridine was distilled from calcium hydride and stored sealed under an argon atmosphere. The starting amides were prepared by acylation of the corresponding anilines³ or via previously reported copper–catalyzed C–N bond–forming reactions.^{4,5}

Instrumentation. All reaction conducted at 140 °C were performed in a CEM Discover Lab Mate microwave reactor. Proton nuclear magnetic resonance (¹H NMR) spectra were recorded with a Varian inverse probe 500 INOVA spectrometer. Chemical shifts are recorded in parts per million from internal tetramethylsilane on the δ scale and are referenced from the residual protium in the NMR solvent (CHCl₃: δ 7.27, C₆HD₅: δ 7.16, CHDCl₂: δ 5.32). Data is reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, q = quartet, m = multiplet), coupling constant(s) in Hertz, integration, assignment]. Carbon-13 nuclear magnetic resonance spectra were recorded with a Varian 500 INOVA spectrometer and are referenced from the carbon resonances of the

⁽¹⁾ Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923.

 ⁽²⁾ Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. Organometallics 1996, 15, 1518.
⁽³⁾ For a general procedure, see: DeRuiter, J.; Swearingen, B. E.; Wandrekar, V.; Mayfield, C. A. J. Med. Chem. 1989, 32, 1033.

⁽⁴⁾ For the general procedure used for the synthesis of all *N*-vinyl amides, see: Jiang, L.; Job, G. E.; Klapars, A.; Buchwald, S. L. Org. Lett. **2003**, *5*, 3667.

⁽⁵⁾ For related reports, see: (a) Wolfe, J. P.; Wagaw, S.; Marcoux, J.-F.; Buchwald, S. L. Acc. Chem. Res. **1998**, 31, 805. (b) Hartwig, J. F. Acc. Chem. Res. **1998**, 31, 852. (c) Muci, A. R.; Buchwald, S. L. Top. Curr. Chem. **2002**, 219, 131. (d) Beletskaya, I. P.; Cheprakov, A. V. Coordin. Chem. Rev. **2004**, 248, 2337. (e) Dehli, J. R.; Legros, J.; Bolm, C. Chem. Commun. **2005**, 973.

solvent (CDCl₃: δ 77.2, benzene-*d*₆: δ 128.0, DMF-*d*₇: δ 163.2, CD₂Cl₂: δ 54.0). Data is reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, q = quartet, m = multiplet), coupling constant(s) in Hertz, assignment]. Fluorine-19 nuclear magnetic resonance spectra were recorded with a Varian 300 INOVA spectrometer and are recorded in parts per million on the δ scale and are referenced from the fluorine resonances of trifluoroacetic acid (CD₂Cl₂: δ -76.6). Data is reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, q = quartet, quint = quintet, m = multiplet), coupling constant(s) in Hertz, assignment]. Infrared data were obtained with a Perkin-Elmer 2000 FTIR and are reported as follows: [frequency of absorption (cm⁻¹), intensity of absorption (s = strong, m = medium, w = weak, br = broad), assignment]. In situ IR reaction monitoring was performed on an ASI ReactIR 1000 spectrometer. Chiral HPLC analysis was performed on an Agilent 1100 Series HPLC with a Chiralpak AD-H column. We thank Dr. Li Li at the Massachusettes Institute of Technology Department of Chemistry instrumentation facility for obtaining mass spectroscopic data.



4-Cyclohexyl-6-methoxy-2-phenyl-quinazoline (3a, Table 2, entry 1):

Trifluoromethanesulfonic anhydride (92 μ L, 0.56 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1a** (115 mg, 0.506 mmol, 1 equiv) and 2-chloropyridine (58 μ L, 0.61 mmol, 1.2 equiv) in dichloromethane (1.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the nitrile **2a** (61 mg, 0.56 mmol, 1.1 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 16 h, the reaction mixture was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 7.5% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product **3a** as a white solid (145 mg, 90%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.67–8.63 (m, 2H, Ar H), 8.02 (d, 1H, $J = 9.1$ Hz, Ar H), 7.56–7.46 (m, 4H, Ar H), 7.37 (d, 1H, $J = 2.6$ Hz, Ar H), 4.00 (s, 3H, OC H ₃), 3.49 (tt, 1H, $J =$ 11.1, 3.3 Hz, ^c C ₆ H ₁₁), 2.09–1.84 (m, 7H, ^c C ₆ H ₁₁), 1.64–1.52 (m, 2H, ^c C ₆ H ₁₁), 1.45 (qt, 1H, $J =$ 12.7, 2.8 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	173.0, 158.6, 157.9, 147.2, 139.0, 131.3, 130.1, 128.6, 128.4, 125.4, 122.6, 102.3, 55.9, 41.7, 32.1, 26.8, 26.4.
FTIR (neat) cm ⁻¹ :	3064 (w), 2931 (m), 2852 (w), 1622 (w), 1567 (w), 1546 (s), 1222 (s).
HRMS (ESI):	calcd for $C_{21}H_{23}N_2O [M+H]^+$: 319.1810, found: 319.1807.
TLC (15% EtOAc in hexanes), $R_{\rm f}$.	0.50 (UV, CAM).


4-Cyclohexyl-2-phenyl-quinazoline (S3b, Table 2, entry 2):

Trifluoromethanesulfonic anhydride (92 μ L, 0.56 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S1b (100 mg, 0.507 mmol, 1 equiv) and 2-chloropyridine (57 μ L, 0.61 mmol, 1.2 equiv) in dichloromethane (1.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the nitrile 2a (61 mg, 0.56 mmol, 1.1 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 16 h, the reaction mixture was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product S3b as a white solid (104 mg, 71%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.72–8.67 (m, 2H, ArH), 8.18 (dd, 1H, $J = 8.3$, 1.3 Hz, ArH), 8.10 (dd, 1H, $J = 8.5$, 1.3 Hz, ArH), 7.86 (ddd, 1H, $J = 8.3$, 6.9, 1.3 Hz, ArH), 7.61–7.48 (m, 4H, ArH), 3.60 (tt, 1H, $J = 11.2$, 3.4 Hz, ^c C ₆ H ₁₁), 2.09–1.84 (m, 7H, ^c C ₆ H ₁₁), 1.62–1.52 (m, 2H, ^c C ₆ H ₁₁), 1.45 (qt, 1H, $J = 12.8$, 3.0 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	174.9, 160.2, 151.2, 138.9, 133.3, 130.5, 129.8, 128.8, 128.7, 126.8, 124.3, 121.9, 41.7, 32.3, 26.8, 26.4.
FTIR (neat) cm ⁻¹ :	3066 (w), 2933 (s), 2852 (s), 1615 (m), 1570 (s), 1546 (s), 1497 (s), 1344 (s), 1027 (m).
HRMS (ESI):	calcd for $C_{20}H_{21}N_2 [M+H]^+$: 289.1705, found: 289.1704.
TLC (15% EtOAc in hexanes), R _f :	0.56 (UV, CAM).



4-Cyclohexyl-2-phenyl-7-trifluoromethyl-quinazoline (S3c, Table 2, entry 3):

Trifluoromethanesulfonic anhydride (82 μ L, 0.50 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S1c (120 mg, 0.450 mmol, 1 equiv) and 2-chloropyridine (51 μ L, 0.54 mmol, 1.2 equiv) in dichloromethane (1.5 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the nitrile 2a (54 mg, 0.50 mmol, 1.1 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product S3c as a white solid (97 mg, 61%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.72–8.69 (m, 2H, ArH), 8.40 (s, 1H, ArH), 8.29 (d, 1H, $J = 8.7$ Hz, ArH), 7.74 (d, 1H, $J = 8.7$ Hz, ArH), 7.59–7.54 (m, 3H, ArH), 3.60 (tt, 1H, $J =$ 11.4, 3.4 Hz, ^c C ₆ H ₁₁), 2.10–1.86 (m, 7H, ^c C ₆ H ₁₁), 1.64–1.53 (m, 2H, ^c C ₆ H ₁₁), 1.45 (qt, 1H, $J =$ 12.7, 3.0 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	175.4, 161.3, 150.6, 138.1, 134.8 (q, $J = 33$ Hz), 131.1, 131.0, 128.9, 128.8, 127.7 (q, $J = 4.3$ Hz), 125.7, 123.8 (q, $J = 271$ Hz), 123.1, 122.3 (q, $J = 4.1$ hz), 42.0, 32.3, 26.7, 26.3.
FTIR (neat) cm ⁻¹ :	2929 (s), 2857 (m), 1575 (s), 1549 (s), 1499 (m), 1344 (s), 1126 (s).
HRMS (ESI):	calcd for C ₂₁ H ₂₀ F ₃ N ₂ [M+H] ⁺ : 357.1579, found: 357.1587.
TLC (20% EtOAc in hexanes), R _f .	0.74 (UV, CAM).



4-Cyclohexyl-6-methoxy-2-(4-methoxy-phenyl)-quinazoline (S3d, Table 2, entry 4):

Trifluoromethanesulfonic anhydride (71 μ L, 0.43 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S1d (100 mg, 0.389 mmol, 1 equiv) and 2-chloropyridine (44 μ L, 0.47 mmol, 1.2 equiv) in dichloromethane (1.3 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the nitrile 2a (47 mg, 0.43 mmol, 1.1 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 18 h, the reaction mixture was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product S3d as a white solid (118 mg, 87%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.62–8.58 (m, 2H, ArH), 7.97 (d, 1H, $J = 9.1$ Hz, ArH), 7.50 (dd, 1H, $J = 9.1$, 2.7 Hz, ArH), 7.35 (d, 1H, $J = 2.7$ Hz, ArH), 7.06–7.03 (m, 2H, ArH), 3.99 (s, 3H, OCH ₃), 3.91 (s, 3H, OCH ₃), 3.46 (tt, 1H, $J = 11.2$, 3.2 Hz, ${}^{c}C_{6}H_{11}$), 2.08–1.84 (m, 7H, ${}^{c}C_{6}H_{11}$), 1.62–1.51 (m, 2H, ${}^{c}C_{6}H_{11}$), 1.45 (qt, 1H, $J = 12.7$, 3.2 Hz, ${}^{c}C_{6}H_{11}$).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	172.9, 161.4, 158.4, 157.6, 147.2, 131.7, 131.0, 129.9, 125.3, 122.2, 113.9, 102.4, 55.8, 55.5, 41.7, 32.0, 26.8, 26.4.
FTIR (neat) cm ⁻¹ :	3001 (w), 2932 (s), 2852 (w), 1623 (m), 1545 (s), 1515 (s), 1250 (s), 1223 (s), 1167(s).
HRMS (ESI):	calcd for $C_{22}H_{25}N_2O_2$ [M+H] ⁺ : 349.1916, found: 349.1913.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.45 (UV, CAM).



<u>4-Cyclohexyl-6-methoxy-2-(4-nitro-phenyl)-quinazoline (S3e, Table 2, entry 5):</u>

Trifluoromethanesulfonic anhydride (80 μ L, 0.49 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S1e (120 mg, 0.440 mmol, 1 equiv) and 2-chloropyridine (50 μ L, 0.53 mmol, 1.2 equiv) in dichloromethane (1.5 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the nitrile 2a (53 mg, 0.49 mmol, 1.1 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product S3e as a yellow solid (111 mg, 69%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.84–8.81 (m, 2H, ArH), 8.38–8.35 (m, 2H, ArH), 8.05 (d, 1H, $J = 9.1$ Hz, ArH), 7.57 (dd, 1H, $J = 9.1$, 2.6 Hz, ArH), 7.39 (d, 1H, $J = 2.7$ Hz, ArH), 4.02 (s, 3H, OCH ₃), 3.51 (tt, 1H, $J = 11.4$, 3.4 Hz, ^c C ₆ H ₁₁), 2.09–1.86 (m, 7H, ^c C ₆ H ₁₁), 1.64–1.53 (m, 2H, ^c C ₆ H ₁₁), 1.46 (qt, 1H, $J = 12.5$, 3.1 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	173.5, 158.8, 156.2, 148.9, 147.0, 144.9, 131.5, 129.1, 126.1, 123.8, 123.1, 102.3, 56.0, 41.8, 32.1, 26.7, 26.3.
FTIR (neat) cm ⁻¹ :	2929 (s), 2851 (m), 1621 (w), 1595 (w), 1544 (m), 1512 (s), 1339 (s), 1256 (m), 1223 (m).
HRMS (ESI):	calcd for C ₂₁ H ₂₁ N ₃ NaO ₃ [M+Na] ⁺ : 386.1481, found: 386.1459.
TLC (30% EtOAc in hexanes), R _f :	0.58 (UV, CAM).



<u>2-tert-Butyl-4-cyclohexyl-6-methoxy-quinazoline (S3f, Table 2, entry 6):</u>

Trifluoromethanesulfonic anhydride (88 μ L, 0.53 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S1f (100 mg, 0.480 mmol, 1 equiv) and 2-chloropyridine (55 μ L, 0.58 mmol, 1.2 equiv) in dichloromethane (1.6 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the nitrile 2a (58 mg, 0.53 mmol, 1.1 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product S3f as a pale yellow solid (124 mg, 86%).

'H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.91 (d, 1H, $J = 9.1$ Hz, ArH), 7.46 (dd, 1H, $J = 9.1$, 2.7 Hz, ArH), 7.32 (d, 1H, $J = 2.7$ Hz, ArH), 3.97 (s, 3H, OCH ₃), 3.40 (tt, 1H, $J = 11.4$, 3.2 Hz, ^c C ₆ H ₁₁), 1.98–1.80 (m, 7H, ^c C ₆ H ₁₁), 1.58–1.50 (m, 2H, ^c C ₆ H ₁₁), 1.49 (s, 9H, C(CH ₃) ₃), 1.40 (qt, 1H, $J = 12.7$, 3.1 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	172.2, 170.9, 157.5, 146.5, 130.9, 124.6, 121.6, 102.1, 55.8, 41.5, 39.6, 32.0, 29.9, 26.7, 26.4.
FTIR (neat) cm ⁻¹ :	2948 (w), 2912 (m), 2848 (m), 1619 (w), 1556 (m), 1497 (m), 1221 (s).
HRMS (ESI):	calcd for $C_{19}H_{27}N_2O[M+H]^+$: 299.2123, found: 299.2121.
TLC (20% EtOAc in hexanes), R _f .	0.60 (UV, CAM).



2,4-Dicyclohexyl-6-methoxy-quinazoline (S3g, Table 2, entry 7):

Trifluoromethanesulfonic anhydride (90 μ L, 0.54 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **S1g** (115 mg, 0.493 mmol, 1 equiv) and 2-chloropyridine (56 μ L, 0.59 mmol, 1.2 equiv) in dichloromethane (1.6 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the nitrile **2a** (59 mg, 0.54 mmol, 1.1 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralize silica gel to give the quinazoline product **S3g** as a white solid (119 mg, 74%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.89 (d, 1H, $J = 9.1$ Hz, Ar H), 7.47 (dd, 1H, $J = 9.1$, 2.9 Hz, Ar H), 7.32 (d, 1H, $J = 2.9$ Hz, Ar H), 3.97 (s, 3H, OC H ₃), 3.44 (tt, 1H, $J = 11.5$, 3.2 Hz, ^c C ₆ H ₁₁), 2.95 (tt, 1H, $J = 11.7$, 3.5 Hz, ^c C ₆ H ₁₁), 2.08–1.72 (m, 14 H, ^c C ₆ H ₁₁ , ^c C ₆ H ₁₁), 1.58–1.34 (m, 6H, ^c C ₆ H ₁₁ , ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	172.9, 168.4, 157.4, 146.7, 130.5, 125.0, 122.1, 102.2, 55.8, 47.9, 41.5, 32.2, 32.0, 26.7, 26.6, 26.4, 26.3.
FTIR (neat) cm ⁻¹ :	2927 (s), 2852 (m), 1623 (w), 1556 (m), 1449 (m), 1222 (s).
HRMS (ESI):	calcd for $C_{21}H_{29}N_2O[M+H]^+$: 325.2280, found: 325.2274.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.56 (UV, CAM).



4-Cyclohexyl-2-morpholin-4-yl-quinazoline (S3h, Table 2, entry 8):

Trifluoromethanesulfonic anhydride (97 μ L, 0.59 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **S1h** (110 mg, 0.530 mmol, 1 equiv), nitrile **2a** (64 mg, 0.59 mmol, 1.1 equiv) and 2-chloropyridine (61 μ L, 0.64 mmol, 1.2 equiv) in dichloromethane (1.8 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product **S3h** as a white solid (131 mg, 83%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.92 (d, 1H, $J = 8.3$ Hz, ArH), 7.65–7.58 (m, 2H, ArH), 7.21 (ddd, 1H, $J = 8.2$, 6.4, 1.6 Hz, ArH), 3.97 (t, 4H, $J = 4.7$ Hz, OCH ₂ CH ₂ N), 3.83 (t, 4H, $J = 4.7$ Hz, OCH ₂ CH ₂ N), 3.41 (tt, 1H, $J = 11.4$, 2.9 Hz, ^c C ₆ H ₁₁), 1.98–1.69 (m, 7H, ^c C ₆ H ₁₁), 1.51 (tt, 2H, $J = 9.6$, 4.2 Hz, ^c C ₆ H ₁₁) 1.35 (tt, 1H, $J = 12.8$, 3.4 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	175.8, 158.9, 152.6, 133.3, 126.8, 124.6, 122.3, 118.1, 67.2, 44.7, 41.5, 32.1, 26.7, 26.4.
FTIR (neat) cm ⁻¹ :	3064 (w), 2930 (s), 2852 (s), 1615 (s), 1579 (s), 1554 (s), 1486 (s), 1227 (s).
HRMS (ESI):	calcd for C ₁₈ H ₂₄ N ₃ O [M+H] ⁺ : 298.1919, found: 298.1911.
TLC (20% EtOAc in hexanes), R _f .	0.48 (UV, CAM).



<u>2-Cyclohex-1-enyl-4-cyclohexyl-6-methoxy-quinazoline (S3i, Table 2, entry 9):</u>

Trifluoromethanesulfonic anhydride (79 μ L, 0.48 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S1i (100 mg, 0.432 mmol, 1 equiv), 2-chloropyridine (49 μ L, 0.53 mmol, 1.2 equiv), and nitrile 2a (236 mg, 2.16 mmol, 5.00 equiv) in dichloromethane (1.4 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 16 h, the reaction mixture was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced at to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product S3i as a white solid (124 mg, 89%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.90 (d, 1H, $J = 9.1$ Hz, ArH), 7.47–7.44 (m, 2H, ArH, C=CHCH ₂), 7.32 (d, 1H, $J = 2.7$ Hz, ArH), 3.97 (s, 3H, OCH ₃), 3.41 (tt, 1H, $J = 10.9$, 2.7 Hz, ^c C ₆ H ₁₁), 2.77–2.72 (m, 2H, ^c C ₆ H ₉), 2.39–2.34 (m, 2H, ^c C ₆ H ₉), 2.02–1.70 (m, 11H, ^c C ₆ H ₁₁ , ^c C ₆ H ₉), 1.60–1.49 (m, 2H, ^c C ₆ H ₁₁), 1.41 (qt, 1H, $J = 12.7$, 3.4 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	172.0, 160.1, 157.5, 146.7, 137.4, 132.9, 131.0, 124.8, 122.1, 102.5, 55.8, 41.6, 32.0, 26.8, 26.4, 26.4, 25.6, 23.1, 22.5.
FTIR (neat) cm^{-1} :	2929 (s), 2853 (m), 1621 (w), 1547 (s), 1496 (m), 1226 (s).
HRMS (ESI):	calcd for $C_{21}H_{27}N_2O [M+H]^+$: 323.2123, found: 323.2126.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.67 (UV, CAM).



6-Methoxy-4-(4-nitro-phenyl)-2-phenyl-quinazoline (S3i, Table 2, entry 10):

Trifluoromethanesulfonic anhydride (92 μ L, 0.56 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1a (115 mg, 0.506 mmol, 1 equiv), nitrile S2b (83 mg, 0.56 mmol, 1.1 equiv) and 2-chloropyridine (58 μ L, 0.61 mmol, 1.2 equiv) in dichloromethane (1.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 20% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product S3j as a pale yellow solid (156 mg, 86%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.66–8.62 (m, 2H, ArH), 8.51–8.47 (m, 2H, ArH), 8.13 (d, 1H, $J = 9.3$ Hz, ArH), 8.11–8.07 (m, 2H, ArH), 7.64 (dd, 1H, $J = 9.3$, 2.7 Hz, ArH), 7.57– 7.50 (m, 3H, ArH), 7.24 (d, 1H, $J = 2.7$ Hz, ArH), 3.89 (s, 3H, OCH ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	164.2, 158.8, 158.7, 148.8, 148.6, 144.4, 138.0, 131.3, 131.0, 130.6, 128.8, 128.4, 127.1, 124.1, 122.3, 103.3, 55.9.
FTIR (neat) cm ⁻¹ :	3056 (w), 2971 (w), 1996 (w), 1621 (m), 1543 (s), 1513 (s), 1350 (s), 1222 (m).
HRMS (ESI):	calcd for $C_{21}H_{16}N_3O_3$ [M+H] ⁺ : 358.1192, found: 358.1183.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.36 (UV, CAM).



6-Methoxy-4-(4-methoxy-phenyl)-2-phenyl-quinazoline (S3k, Table 2, entry 11):

Trifluoromethanesulfonic anhydride (92 μ L, 0.56 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1a** (115 mg, 0.506 mmol, 1 equiv), nitrile **S2c** (74 mg, 0.56 mmol, 1.1 equiv) and 2-chloropyridine (58 μ L, 0.61 mmol, 1.2 equiv) in dichloromethane (1.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 20% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product **S3k** as a pale yellow solid (156 mg, 90%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.67–8.64 (m, 2H, ArH), 8.07 (d, 1H, $J = 9.1$ Hz, ArH), 7.94–7.90 (m, 2H, ArH), 7.57–7.45 (m, 5H, ArH), 7.16–7.12 (m, 2H, ArH), 3.95 (s, 3H, OCH ₃), 3.89 (s, 3H, OCH ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	166.1, 161.1, 158.7, 158.1, 148.3, 138.6, 131.6, 130.7, 130.6, 130.2, 128.6, 128.4, 126.1, 122.5, 114.2, 104.6, 55.7, 55.6.
FTIR (neat) cm ⁻¹ :	3006 (w), 2957 (m), 2839 (w), 1608 (s), 1564 (m), 1534 (s), 1499 (s), 1404 (s), 1259 (s), 1221 (m).
HRMS (ESI):	calcd for $C_{22}H_{19}N_2O_2 [M+H]^+$: 343.1447, found: 343.1437.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.33 (UV, CAM).



4-(6-Methoxy-2-phenyl-quinazolin-4-yl)-benzoic acid ethyl ester (S31, Table 2, entry 12):

Trifluoromethanesulfonic anhydride (92 μ L, 0.56 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1a (115 mg, 0.506 mmol, 1 equiv), nitrile S2d (98 mg, 0.56 mmol, 1.1 equiv) and 2-chloropyridine (58 μ L, 0.61 mmol, 1.2 equiv) in dichloromethane (1.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product S3I as a white solid (154 mg, 79%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.67–8.63 (m, 2H, ArH), 8.30 (d, 2H, $J = 8.0$ Hz, ArH), 8.10 (d, 1H, $J = 9.1$ Hz, ArH), 7.98 (d, 2H, $J = 7.9$ Hz, ArH), 7.58 (dd, 1H, $J = 9.1$, 2.7 Hz, ArH), 7.56–7.48 (m, 3H, ArH), 7.31 (d, 1H, $J = 2.7$ Hz, ArH), 4.48 (q, 2H, $J = 7.2$ Hz, CO ₂ CH ₂ CH ₃), 3.87 (s, 3H, OCH ₃), 1.48 (t, 3H, $J = 7.2$ Hz, CO- ₂ CH ₂ CH ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	166.4, 165.7, 158.9, 158.4, 148.5, 142.4, 138.3, 131.7, 131.0, 130.4, 130.0, 130.0, 128.8, 128.5, 126.8, 122.5, 103.9, 61.5, 55.8, 14.6.
FTIR (neat) cm ⁻¹ :	2961 (w), 1717 (s), 1621 (w), 1536 (m), 1407 (m), 1271 (s), 1222 (s).
HRMS (ESI):	calcd for $C_{24}H_{21}N_2O_3$ [M+H] ⁺ : 385.1552, found: 385.1544.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.39 (UV, CAM).



6-Methoxy-2-phenyl-4-styryl-quinazoline (S3m, Table 2, entry 13):

Trifluoromethanesulfonic anhydride (92 μ L, 0.56 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1a** (115 mg, 0.506 mmol, 1 equiv), nitrile **S2e** (72 mg, 0.56 mmol, 1.1 equiv) and 2-chloropyridine (58 μ L, 0.61 mmol, 1.2 equiv) in dichloromethane (1.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 20% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product **S3m** as a white solid (120 mg, 70%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.71–8.68 (m, 2H, ArH), 8.45 (d, 1H, $J = 15.4$ Hz, CH=CH), 8.03 (d, 1H, $J = 9.1$ Hz, ArH), 7.89 (d, 1H, $J = 15.4$ Hz, ArH), 7.80 (d, 2H, $J = 7.4$ Hz, ArH), 7.59–7.55 (m, 3H, ArH), 7.54–7.47 (m, 4H, ArH), 7.45–7.41 (m, 1H, ArH), 4.04 (s, 3H, OCH ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	160.3, 158.5, 158.1, 148.4, 139.1, 138.8, 136.4, 131.0, 130.2, 129.7, 129.1, 128.7, 128.4, 128.2, 126.3, 122.5, 121.3, 101.6, 55.9.
FTIR (neat) cm ⁻¹ :	3059 (w), 2936 (w), 1621 (m), 1560 (m), 1533 (s), 1499 (m), 1408 (s), 1224 (s).
HRMS (ESI):	calcd for $C_{23}H_{19}N_2O[M+H]^+$: 339.1497, found: 339.1492.
TLC (20% EtOAc in hexanes), $R_{\rm f}$.	0.38 (UV, CAM).



4-Cyclohexyl-2-phenyl-7,8-dihydro-6H-pyrano[3,2-d]pyrimidine (S3n, Table 2, entry 14):

Trifluoromethanesulfonic anhydride (63 μ L, 0.38 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S1j (70 mg, 0.34 mmol, 1 equiv), nitrile 2a (41 mg, 0.38 mmol, 1.1 equiv) and 2-chloropyridine (39 μ L, 0.41 mmol, 1.2 equiv) in dichloromethane (1.0 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature. After 1 h, aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyrimidine product S3n as a white solid (94 mg, 93%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.40–8.36 (m, 2H, ArH), 7.47–7.38 (m, 3H, ArH), 4.27 (t, 2H, $J = 5.1$ Hz, CH ₂ CH ₂ CH ₂ O), 3.06 (tt, 1H, $J = 11.9$, 3.5 Hz, ^c C ₆ H ₁₁), 2.98 (t, 2H, $J = 6.6$ Hz, CH ₂ CH ₂ CH ₂ O), 2.18–2.13 (m, 2H, CH ₂ CH- 2CH ₂ O), 1.91–1.69 (m, 7H, ^c C ₆ H ₁₁), 1.44 (qt, 2H, $J = 12.7$, 3.2 Hz, ^c C ₆ H ₁₁), 1.35 (qt, 1H, $J = 12.7$, 3.2 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	161.7, 155.8, 149.6, 146.0, 138.7, 129.3, 128.5, 127.7, 66.7, 38.8, 30.8, 28.1, 26.6, 26.3, 22.2.
FTIR (neat) cm ⁻¹ :	3065 (w), 2930 (s), 2852 (m), 1587 (w), 1565 (s), 1430 (s), 1410 (s), 1208 (m).
HRMS (ESI):	calcd for C ₁₉ H ₂₃ N ₂ O [M+H] ⁺ : 295.1810, found: 295.1796.
TLC (20% EtOAc in hexanes), R_{f} :	0.52 (UV. CAM).



4-tert-Butyl-2-phenyl-7,8-dihydro-6H-pyrano[3,2-d]pyrimidine (S30, Table 2, entry 15):

Trifluoromethanesulfonic anhydride (890 μ L, 5.41 mmol, 1.1 equiv) was added via syringe over 3 min to a stirred mixture of amide **S1j** (1.0 g, 4.9 mmol, 1 equiv), nitrile **S2f** (450 mg, 5.41 mmol, 1.1 equiv) and 2-chloropyridine (560 μ L, 5.90 mmol, 1.2 equiv) in dichloromethane (16 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature. After 3 h, aqueous sodium hydroxide solution (5 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (50 mL) was added to dilute the mixture and the layers were separated. The aqueous layer was extracted with dichloromethane (2 × 50 mL) and the organic fractions were combined, dried over anhydrous sodium sulfate, and were filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 12 × 3 cm) on neutralized silica gel to give the pyrimidine product **S30** as a white solid (1.17 g, 88%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.42–8.37 (m, 2H, ArH), 7.48–7.38 (m, 3H, ArH), 4.27 (t, 2H, $J = 5.1$ Hz, CH ₂ CH ₂ CH ₂ O), 2.98 (t, 2H, J = 6.7 Hz, CH ₂ CH ₂ CH ₂ O), 2.20–2.14 (m, 2H, CH ₂ CH ₂ CH ₂ O), 1.46 (s, 9H, C(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	163.3, 154.8, 150.4, 147.6, 138.6,129.4, 128.5, 127.7, 66.3, 38.2, 28.3, 28.1, 22.1.
FTIR (neat) cm ⁻¹ :	3065 (w), 2955 (m), 2868 (w), 1558 (m), 1429 (m), 1406 (s), 1366 (m), 1353 (m).
HRMS (ESI):	calcd for $C_{17}H_{21}N_2O[M+H]^+$: 269.1654, found: 269.1653.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.67 (UV, CAM).



<u>4-Pent-4-ynyl-2-phenyl-7.8-dihydro-6H-pyrano[3,2-d]pyrimidine (S3p, Table 2, entry 16):</u>

Trifluoromethanesulfonic anhydride (89 μ L, 0.54 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S1j (100 mg, 0.490 mmol, 1 equiv), nitrile S2g (50 mg, 0.54 mmol, 1.1 equiv) and 2-chloropyridine (56 μ L, 0.59 mmol, 1.2 equiv) in dichloromethane (1.6 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature. After 1 h, aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyrimidine product S3p as a colorless oil (107 mg, 78%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.36–8.33 (m, 2H, ArH), 7.48–7.38 (m, 3H, ArH), 4.28 (t, 2H, $J = 5.1$ Hz, CH ₂ CH ₂ CH ₂ O), 2.98 (t, 2H, J = 6.6 Hz, CH ₂ CH ₂ CH ₂ O), 2.92 (t, 2H, $J = 7.2$ Hz, CH ₂ CH ₂ CH ₂ CCH), 2.37 td, 2H, $J = 7.1$, 2.6 Hz, CH ₂ CH ₂ CH ₂ CCH), 2.20–2.14 (m, 2H, CH ₂ CH ₂ CH ₂ CCH), 2.08 (quint, 2H, $J = 7.4$ Hz, CH ₂ CH ₂ CH ₂ CCH), 2.01 (1H, t, $J = 2.7$ Hz, CH ₂ CH ₂ CH ₂ CCH).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	157.2, 155.9, 149.7, 146.9, 138.4, 129.5, 128.6, 127.7, 84.5, 68.8, 66.9, 30.4, 28.1, 26.0, 22.2, 18.5.
FTIR (neat) cm ⁻¹ :	3297 (s), 3066 (m), 2939 (s), 2117 (w), 1587 (s), 1569 (s), 1411 (s), 1202 (m).
HRMS (ESI):	calcd for C ₁₈ H ₁₉ N ₂ O [M+H] ⁺ : 279.1497, found: 279.1496.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.41 (UV, CAM).



<u>4-Cyclohexyl-2,5-diphenyl-pyrimidine (S3q, Table 2, entry 17):</u>

Trifluoromethanesulfonic anhydride (90 μ L, 0.54 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **S1k** (110 mg, 0.493 mmol, 1 equiv), nitrile **2a** (269 mg, 2.47 mmol, 5.00 equiv) and 2-chloropyridine (56 μ L, 0.59 mmol, 1.2 equiv) in dichloromethane (1.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath and warmed to 0 °C, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 1 h, the reaction vessel was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure, and the residue was purified by flash column chromatography (eluent: 5% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyrimidine product **S3q** as a white solid (142 mg, 92%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.59 (s, 1H, Ar H), 8.57–8.54 (m, 2H, Ar H), 7.55– 7.44 (m, 6H, Ar H), 7.38–7.35 (m, 2H, Ar H), 2.90 (tt, 1H, $J = 11.5$, 3.5 Hz, ^c C ₆ H ₁₁), 1.94–1.69 (m, 7H, ^c C ₆ H ₁₁), 1.37 (qt, 1H, $J = 12.8$, 3.2 Hz, ^c C ₆ H ₁₁), 1.30–1.20 (m, 2H, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	171.6, 163.1, 157.4, 138.2, 136.6, 131.7, 130.6, 129.4, 128.9, 128.7, 128.3, 128.1, 42.0, 32.2, 26.3, 26.1.
FTIR (neat) cm ⁻¹ :	3060 (w), 2929 (s), 2853 (m), 1586 (w), 1568 (s), 1525 (s), 1425 (s), 1378 (m).
HRMS (ESI):	calcd for $C_{22}H_{23}N_2 [M+H]^+$: 315.1861, found: 315.1861.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.69 (UV, CAM).



4-Cyclohexyl-2-phenyl-thieno[3,2-d]pyrimidine (S3r, Table 2, entry 18):

Trifluoromethanesulfonic anhydride (72 μ L, 0.43 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S11 (80 mg, 0.39 mmol, 1 equiv), nitrile 2a (47 mg, 0.54 mmol, 1.1 equiv) and 2-chloropyridine (45 μ L, 0.47 mmol, 1.2 equiv) in dichloromethane (1.3 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature. After 1 h, aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyrimidine product S3r as a pale yellow solid (101 mg, 87%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.60–8.58 (m, 2H, ArH), 7.92 (d, 1H, $J = 5.5$ Hz, ArH), 7.60 (d, 1H, $J = 5.5$ Hz, ArH), 7.54–7.48 (m, 3H, ArH), 3.05 (tt, 1H, $J = 11.5$, 3.8 Hz, ^c C ₆ H ₁₁), 2.10–1.81 (m, 7H, ^c C ₆ H ₁₁), 1.56–1.38 (m, 3H, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	168.8, 162.0, 161.4, 138.7, 134.5, 130.3, 128.7, 128.5, 127.4, 125.4, 46.3, 31.3, 26.5, 26.2.
FTIR (neat) cm^{-1} :	2928 (m), 2852 (w), 1535 (s), 1365 (w), 1341 (w).
HRMS (ESI):	calcd for C ₁₈ H ₁₉ N ₂ S [M+H] ⁺ : 295.1269, found: 295.1259.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.52 (UV, CAM).



<u>4-Cyclohexyl-2-phenyl-5*H*-pyrrolo[3,2-*d*]pyrimidine (S3s, Table 2, entry 19):</u>

Trifluoromethanesulfonic anhydride (37 μ L, 0.23 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide S1m (70 mg, 0.20 mmol, 1 equiv), nitrile 2a (46 mg, 0.41 mmol, 2.00 equiv) and 2-chloropyridine (39 μ L, 0.41 mmol, 2.00 equiv) in dichloromethane (0.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath for 5 min. and warmed to 0 °C. The resulting solution was allowed to warm to ambient temperature. After 1 h, triethylamine (1 mL) was introduced to neutralize the trifluoromethanesulfonate salts, followed by TBAF (204 μ L, 1.00 equiv, 1.0 M) to protodesilylate the pyrimidine product. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10-40% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyrimidine product S3s as a pale tan solid (41 mg, 72%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.57–8.52 (m, 2H, ArH), 8.45 (s, 1H, NH), 7.54– 7.47 (m, 3H, ArH), 7.45–7.40 (m, 1H, ArH), 6.79 (dd, 1H, $J = 3.1$, 2.0 Hz, ArH), 3.05 (tt, 1H, $J = 11.4$, 3.5 Hz, ${}^{\circ}C_{6}H_{11}$), 2.09–1.80 (m, 7H, ${}^{\circ}C_{6}H_{11}$), 1.54–1.38 (m, 3H, ${}^{\circ}C_{6}H_{11}$).
¹³ C NMR (125 MHz, DMF- <i>d</i> ₇ , 20 °C) δ:	157.9, 156.9, 151.9, 141.1, 133.6, 130.0, 129.3, 128.6, 125.2, 102.9, 42.4, 32.2, 27.3, 27.0.
FTIR (neat) cm ⁻¹ :	3073 (m), 3019 (m), 2924 (s), 2849 (s), 1996 (w), 1738 (w), 1609 (m), 1543 (s), 1445 (m), 1386 (s).
HRMS (ESI):	calcd for C ₁₈ H ₂₀ N ₃ [M+H] ⁺ : 278.1657, found: 278.1656.
TLC (40% EtOAc in hexanes), $R_{\rm f}$:	0.38 (UV, CAM).



(*R*)-4-[(*tert*-Butyl-dimethyl-silanyloxy)-phenyl-methyl]-2-phenyl-7,8-dihydro-6*H*-pyrano[3,2*d*]pyrimidine (S3t, Table 2, entry 20):

Trifluoromethanesulfonic anhydride (54 µL, 0.33 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide Sli (60 mg, 0.30 mmol, 1 equiv), nitrile S2h⁶ (219 mg, 0.885 mmol, 3.00 equiv) and 2-chloropyridine (34 µL, 0.35 mmol, 1.2 equiv) in dichloromethane (1.0 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath and warmed to 0 °C, and the resulting solution was allowed to warm to ambient temperature for 5 min. before being placed into a preheated oil bath at 45 °C and maintained at that temperature. After 1 h, the reaction vessel was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc in hexanes: SiO_2 : 15 × 1.5 cm) on neutralized silica gel to give the pyrimidine product S3t as a colorless oil (91 mg, 72%, 96% ee). The ee of the product was determined by chiral HPLC analysis of the corresponding desilylated alcohol. The enantiomeric excess of the pyrimidine product was determined to be 95% ee by chiral HPLC analysis [Chiralpak AD-H; 2.5 mL/min; 7% 'PrOH in hexanes; t_R (minor) = 6.74 min., t_R (major) = 9.95 min].

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.44–8.38 (m, 2H, ArH), 7.62–7.57 (m, 2H, ArH), 7.48–7.38 (m, 3H, ArH), 7.35–7.30 (m, 2H, ArH), 7.27–7.24 (m, 1H, ArH), 6.20 (s, 1H, CHOTBS), 4.32–4.21 (m, 2H, CH ₂ CH ₂ CH ₂ O), $3.03-2.92$ (m, 2H, CH ₂ CH ₂ CH ₂ O), $2.20-2.11$ (m, 2H, CH ₂ CH ₂ CH ₂ O), 0.95 (s, 9H, Si(CH ₃) ₂ C(CH ₃) ₃), 0.05 (s, 3H, Si(CH ₃) ₂ C(CH ₃) ₃), 0.00 (s, 3H, Si(CH ₃) ₂ C(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	157.4, 156.0, 151.3, 145.7, 142.6, 138.3, 129.5, 128.5, 128.1, 127.8, 127.3, 127.0, 71.8, 66.8, 28.2, 26.0, 22.0, 18.5, -4.5, -4.7.
FTIR (neat) cm ⁻¹ :	3065 (m), 3032 (m), 2954 (s), 2886 (s), 2856 (s), 1957 (w), 1819 (w), 1586 (m), 1564 (s), 1409 (s), 1252 (s).

⁽⁶⁾ The nitrile **S2h** was prepared by silvlation of the corresponding commercially available cyanohydrin. The optical activity of the starting cyanohydrin was determined to be 96% ee by chiral HPLC analysis.

HRMS (ESI):

calcd for $C_{26}H_{33}N_2O_2Si [M+H]^+: 433.2311$, found: 433.2303.

TLC (20% EtOAc in hexanes), $R_{\rm f}$:

0.67 (UV, CAM).



(S)-N-(5,6-Dihydro-4H-pyran-3-yl)-2-methyl-butyramide (S1n, Table 2, entry 21):

Oxalyl chloride (1.30 g, 10.3 mmol, 1.05 equiv) was added over 1 minute via syringe to a stirred solution of (S)-(+)-2-methylbutyric acid (S7, 1.0 g, 9.8 mmol, 1 equiv) and N,Ndimethylformamide (10 µL) in dichloromethane (33 mL) in an ice-bath at 0 °C. The reaction mixture was removed from the ice-bath after 15 min. and allowed to warm to ambient temperature. After 1.5 h, gas evolution had ceased and dichloromethane saturated with ammonia (33 mL) was added via cannula at ambient temperature. Water (10 mL) was added after 5 min. to remove ammonium salts and the layers were separated. The aqueous layer was extracted with dichloromethane $(2 \times 50 \text{ mL})$, the organic layers were combined and dried over anhydrous sodium sulfate and filtered, and the volatiles were removed under reduced pressure to afford pure primary amide S8 as a white solid (870 mg, 88%). The enantiomeric excess of the amide was determined to be 93% ee by chiral HPLC analysis [Chiralpak AD-H; 1.0 mL/min; 7% 'PrOH in hexanes; t_R (minor) = 14.2 min., t_R (major) = 15.7 min]. The primary amide (850 mg, 8.40 mmol, 1 equiv) was then combined with 5-Bromo-3,4-dihydro-2H-pyran⁷ (1.1 g, 7.0 mmol, 0.83 equiv), copper iodide (160 mg, 0.840 mmol, 0.100 equiv), N,N'-dimethylethylenediamine (148 mg, 1.68 mmol, 0.200 equiv), and potassium carbonate (1.97 g, 14.3 mmol, 1.70 equiv) in toluene (8.4 mL) in a pressure vessel. The resulting reaction mixture was placed in a preheated oil bath at 80 °C and maintained at that temperature. After 16 h, the solution was removed from the bath and allowed to cool to ambient temperature. The crude mixture was diluted with ethyl acetate (30 mL) and filtered through celite; the volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 40% EtOAc in hexanes; SiO₂: 25 × 3 cm) on silica gel to give the amide product S1n as a white solid (950 mg, 74%). The copper catalyzed C-N bond formation occurred without loss of optical activity as confirmed by measuring the enantiomeric excess of the corresponding pyrimidine S3u (see page S24).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	6.93 (s, 1H, C=CH), 6.15 (br s, 1H, NH), 3.95 (t,
	2H, $J = 5.3$ Hz, CH ₂ CH ₂ CH ₂ O), 2.26–2.22 (m, 2H,
	CH ₂ CH ₂ CH ₂ O), 2.16–2.07 (m, 1H, CH(CH ₃)CH ₂
	CH ₃), 1.98–1.92 (m, 2H, CH ₂ CH ₂ CH ₂ O), 1.74–1.64
	(m, 1H, CH(CH ₃)CH ₂ CH ₃), 1.51–1.42 (m, 1H,
	$CH(CH_3)CH_2CH_3$, 1.16 (d, 3H, $J = 6.7$ Hz,
	$CH(CH_3)CH_2CH_3)$, 0.94 (t, 3H, $J = 7.4$ Hz,
	$CH(CH_3)CH_2CH_3).$
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	175.8, 139.5, 113.7, 65.4, 43.3, 27.6, 23.9, 22.1,
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	CH(CH ₃)CH ₂ CH ₃), 0.94 (t, 3H, $J = 7.4$ CH(CH ₃)CH ₂ CH ₃). 175.8, 139.5, 113.7, 65.4, 43.3, 27.6, 23.9, 2 17.7, 12.1.

⁽⁷⁾ Bonner, W. A.; Werth, P. J.; Roth, J. M. J. Org. Chem. 1962, 27, 1575.

FTIR (neat) cm ⁻¹ :	3292 (w), 2968 (m), 2936 (m), 2878 (w), 1727 (s), 1699 (m), 1651 (s), 1510 (m), 1463 (m), 1382 (w), 1165 (m).
HRMS (ESI):	calcd for $C_{10}H_{18}NO_2 [M+H]^+$: 184.1332, found: 184.1337.
TLC (40% EtOAc in hexanes), <i>R</i> _f :	0.34 (UV, CAM).

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(S)-2-sec-Butyl-4-[4-(triisopropyl-silanyloxy)-phenyl]-7,8-dihydro-6H-pyrano[3,2-d]pyrimidine (S3u, Table 2, entry 21):

Trifluoromethanesulfonic anhydride (79 μ L, 0.48 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide Sln (80 mg, 0.44 mmol, 1 equiv), nitrile S2i (361 mg, 1.31 mmol, 3.00 equiv) and 2-chloropyridine (50 μ L, 0.52 mmol, 1.2 equiv) in dichloromethane (1.5 mL) at -78 °C. After 5 min., the reaction mixture was placed in an icewater bath and warmed to 0 °C, and the resulting solution was allowed to warm to ambient temperature for 5 min. before being placed into a preheated oil bath at 45 °C and maintained at that temperature. After 1 h, the reaction vessel was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10->20% EtOAc in hexanes; SiO₂: 15×1.5 cm) on neutralized silica gel to give the pyrimidine product S3u as a colorless oil (115 mg, 60%). The enantiomeric excess of the pyrimidine product was determined to be 93% ee by protodesilylation and chiral HPLC analysis [Chiralpak AD-H; 2.0 mL/min; 3% ⁱPrOH in hexanes; t_R (major) = 9.36 min., t_R (minor) = 11.1 min] of the corresponding alcohol.⁸

¹H NMR (500 MHz, CDCl₃, 20 °C) δ:

8.16–8.12 (m, 2H, ArH), 6.97–6.93 (m, 2H, ArH), 4.28 (t, 2H, J = 5.1 Hz, CH₂CH₂CH₂O), 2.98–2.87 (m, 3H, CH₂CH₂CH₂O, CH(CH₃)CH₂CH₃), 2.20– 2.15 (m, 2H, CH₂CH₂CH₂O), 1.96–1.86 (m, 1H, CH(CH₃)CH₂CH₃), 1.70–1.61 (m, 1H, CH(CH₃) CH₂CH₃), 1.34–1.26 (m, 6H, CH(CH₃)CH₂CH₃, Si(CH(CH₃)₂)₃), 1.14 (d, 18H, J = 7.5 Hz, Si(CH (CH₃)₂)₃), 0.90 (t, 3H, J = 7.4 Hz, CH(CH₃) CH₂CH₃).

¹³C NMR (125 MHz, CDCl₃, 20 °C) δ:

165.5, 157.7, 151.4, 150.9, 145.6, 131.3, 128.9, 119.5, 66.7, 44.2, 29.6, 28.4, 22.1, 20.0, 18.1, 12.9, 12.4.

⁽⁸⁾ The use of the (S)-N-(4-methoxyphenyl)-2-methylbutyramide variant of amide S1n as the substrate with the same nitrile (S2i, 1.1 equiv) under standard conditions (B or C, see text) provided the corresponding quinazoline in 58% yield (condition C) but with complete racemization (0%ee). This is likely due to the compounded effect of the low reactivity of the amide and the nitrile in addition to the likely slower rate of cyclization of intermediate 6 (Scheme 5) leading to quinazolines.

FTIR (neat) cm^{-1} :

HRMS (EI):

2961 (s), 2868 (s), 1605 (s), 1558 (m), 1541 (w), 1508 (s), 1463 (m), 1270 (s).

calcd for $C_{26}H_{40}N_2O_2Si [M]^+$: 440.2859, found: 440.2865.

TLC (20% EtOAc in hexanes), $R_{\rm f}$:

0.37 (UV, CAM)

Direct conversion of secondary amide 1a and primary amide 4 to quinazoline 3a.



<u>4-Cyclohexyl-6-methoxy-2-phenyl-quinazoline (eq 2):</u>

Trifluoromethanesulfonic anhydride (193 μ L, 1.17 mmol, 2.30 equiv) was added via syringe over 1 min to a stirred mixture of amide **1a** (115 mg, 0.506 mmol, 1 equiv), cyclohexanecarboxamide **4** (71 mg, 0.56 mmol, 1.1 equiv) and 2-chloropyridine (125 μ L, 1.32 mmol, 2.60 equiv) in dichloromethane (1.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath for 5 min. and warmed to 0 °C. The resulting solution was warmed to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product **3a** as a white solid (119 mg, 74%).

See 4-Cyclohexyl-6-methoxy-2-phenyl-quinazoline (3a, Table 2, entry 1) experimental page for spectroscopic data.

In situ IR analysis of the conversion of amide 1a and nitrile 2a to quinazoline 3a:

All reactions were performed in a reaction vessel under an atmosphere of argon with the React-IR probe submerged completely in the reaction mixture.



<u>4-Cyclohexyl-6-methoxy-2-phenyl-quinazoline (3a):</u>

In situ IR monitoring of the addition of trifluoromethanesulfonic anhydride (96 μ L, 0.58 mmol, 1.1 equiv) via syringe over 1 min to a mixture of amide **1a** (120 mg, 0.528 mmol, 1 equiv) and 2-chloropyridine (60 μ L, 0.63 mmol, 1.2 equiv) in dichloromethane (2.7 mL) at 0 °C revealed consumption of both amide **1a** and 2-chloropyridine with concomitant appearance of a persistent band at 1600 cm⁻¹ corresponding to the activated compound. After 5 min., nitrile **2a** (63 mg, 0.58 mmol, 1.1 equiv) was added via syringe, and the resulting solution was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 6 h, the reaction vessel was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 20% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product **3a** as a white solid (106 mg, 63%).⁹

⁽⁹⁾ Dines, T. J.; MacGregor, L. D.; Rochester, C. H. Spectrochimica Acta Part A 2003, 59, 3205.



Control IR Experiments:

Assignment of the 2-chloropyridine and the 2-chloropyridinium triflate characteristic stretches: In situ IR monitoring of the addition of trifluoromethanesulfonic acid (47 µL, 0.53 mmol, 1) 1 equiv) via syringe to a solution of 2-chloropyridine (50 μ L, 0.53 mmol, 1 equiv, 1580 cm⁻¹) in CH_2Cl_2 (2.2 mL) at 0 °C resulted in the expected 2-chloropyridinium triflate salt (1620 cm⁻¹).

Assignment of the 4-Cyclohexyl-6-methoxy-2-phenyl-quinazolin-1-ium triflate (3a•HOTf) characteristic stretch:

In situ IR monitoring of the addition of trifluoromethanesulfonic acid (78 µL, 0.88 mmol, 2) 2.0 equiv) via syringe to a solution of quinazoline 3a (140 mg, 0.29 mmol, 1 equiv, 1550 cm⁻¹) and 2-chloropyridine (50 µL, 0.53 mmol, 1 equiv, 1580 cm⁻¹) in CH₂Cl₂ (2.2 mL) at 0 °C resulted in the expected mixture containing 2-chloropyridinium triflate salt (1620 cm⁻¹) and the 4-cyclohexyl-6-methoxy-2-phenyl-quinazolin-1-ium triflate salt (3a•HOTf, 1575 cm⁻¹).

The same characteristic resonance for 3a•HOTf was observed in the absence of 2-3) chloropyridine. In situ IR monitoring of the addition of trifluoromethanesulfonic acid (43 µL, 0.49 mmol, 1 equiv) to a solution of quinazoline 3a (140 mg, 0.29 mmol, 1 equiv, 1550 cm⁻¹) in CH₂Cl₂ (2.2 mL) at 0 °C resulted in the expected 4-cyclohexyl-6-methoxy-2-phenyl-quinazolin-1-ium triflate salt ($3a \cdot HOTf$, 1575 cm⁻¹).

Assignment of the characteristic stretch at 1600 cm⁻¹ to the activated intermediate in the presence of 2-chloropyridine:

Trifluoromethanesulfonic anhydride (96 µL, 0.58 mmol, 1.1 equiv) was added via 4) syringe over 1 min to a solution of amide 1a (120 mg, 0.528 mmol, 1 equiv) in dichloromethane (2.7 mL) at 0 °C. After 5 min., 2-chloropyridine (60 µL, 0.63 mmol, 1.2 equiv) was added via syringe resulting in the appearance of the characteristic stretch at 1600 cm^{-1} . The nitrile **2a** (63 mg, 0.58 mmol, 1.1 equiv) was immediately added via syringe, and the resulting solution was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 3 h, the reaction mixture was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 20% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product **3a** as a white solid (58 mg, 35%). This non-ideal procedure (order of reagent addition and time) was used to verify the involvement of 2-chloropyridine in the formation of the activated intermediate resulting in the observed stretch at 1600 cm⁻¹.

5) Additionally, in situ IR monitoring of the addition of trifluoromethanesulfonic acid (73 μ L, 0.44 mmol, 1 equiv) via syringe to 2-chloropyridine (42 μ L, 0.44 mmol, 1 equiv, 1580 cm⁻¹) in CH₂Cl₂ (2.2 mL) at ambient temperature resulted in no observable change. The band at 1580 cm⁻¹ related to 2-chloropyridine persisted with out loss in intensity. After 4.5 h, water (70 μ L, 4.4 mmol, 10 equiv) was added via syringe and as expected the 2-chloropyridine stretch (1580 cm⁻¹) disappeared completely with a concomitant appearance of a band consistent with 2-chloropyridinium triflate salt (1620 cm⁻¹).

¹H and ¹⁹F NMR monitoring of the conversion of amide 1a and nitrile 2a to quinazoline 3a:



4-Cyclohexyl-6-methoxy-2-phenyl-quinazoline (3a•HOTf):

Trifluoromethanesulfonic anhydride (80 µL, 0.48 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1a** (100 mg, 0.440 mmol, 1 equiv) and 2-chloropyridine (50 µL, 0.53 mmol, 1.2 equiv) in CD₂Cl₂ (1.5 mL) at -78 °C. After 5 min the reaction vessel was placed in an ice-water bath and warmed to 0 °C. ¹H (500 MHz) and ¹⁹F (282 MHz) NMR analysis revealed broad resonances. Additionally, complete consumption of 2-chloropyridine and the starting amide **1a** was confirmed. ¹⁹F NMR was informative and revealed a broad peak corresponding to a triflate ion at δ -79.6 along with remaining trifluoromethanesulfonic anhydride (δ -72.4, ~16%). The nitrile **2a** (53 mg, 0.48 mmol, 1.1 equiv) was added via syringe. After 5 min., ¹H and ¹⁹F NMR analysis revealed a small set of resonances consistent with the protonated quinazoline product **3a**•HOTf along with dominant resonances corresponding to those observed during activation of the amide as described above. Again, the ¹⁹F NMR was informative and revealed predominantly a broad resonance corresponding to a triflate ion at δ -79.6 along with unreacted trifluoromethanesulfonic anhydride (13%) at δ -72.4. While a trace amount of the product was observed, the best conditions for the synthesis of **3a** involve heating to 140 °C.



<u>4-Cyclohexyl-6-methoxy-2-phenyl-quinazoline (3a):</u>

Trifluoromethanesulfonic anhydride (80 μ L, 0.48 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1a** (100 mg, 0.440 mmol, 1 equiv), nitrile **2a** (53 mg, 0.48 mmol, 1.1 equiv) and 2-chloropyridine (50 μ L, 0.53 mmol, 1.2 equiv) in CD₂Cl₂ (1.5 mL) at -78 °C. After 5 min., the reaction vessel was placed in an ice-water bath for 5 min. and warmed to 0 °C; the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and a sample was subject to ¹H (500 MHz) and ¹⁹F NMR (282 MHz) analysis. Complete conversion to the desired product was observed by this crude ¹H NMR analysis. The observed resonances corresponded to 2-chloropyridium trifluoromethane-sulfonate, protonated quinazoline **3a**•HOTf, and the remaining nitrile **2a**. ¹⁹F NMR analysis of the crude reaction mixture revealed only a broad resonance corresponding to triflate anion at δ -79.6 and weak resonance at δ -72.4 for the trace amount of remaining trifluoromethanesulfonic anhydride. Aqueous sodium hydroxide solution (1 mL, 1N)

was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinazoline product **3a** as a white solid (127 mg, 91%).

Control ¹H and ¹⁹F NMR Experiments:

Assignment of the 2-chloropyridinium triflate resonances:

1) Addition of trifluoromethanesulfonic acid (1 equiv) via syringe to 2-chloropyridine (1 equiv) in CD₂Cl₂ (700 μ L) at 23 °C followed by ¹H and ¹⁹F NMR analysis revealed the formation of the expected 2-chloropyridinium triflate. ¹H NMR (500 MHz) δ : 15.8 (br-s, 1H), 8.76 (br-m, 1H), 8.54 (m, 1H), 8.02–7.98 (m, 2H). ¹⁹F NMR (282 MHz) δ : –79.3.

2) Consistent with the IR experiments described above, the addition of trifluoromethanesulfonic anhydride (32 µL, 0.19 mmol, 1.1 equiv) via syringe to a solution of 2chloropyridine (17 µL, 0.18 mmol, 1 equiv) in CD₂Cl₂ (600 µL) at 23 °C under an atmosphere of argon followed by ¹H and ¹⁹F NMR analysis revealed no change after 24 h. Importantly, ¹⁹F NMR (282 MHz) analysis only reveals a persistent resonance at δ -72.4 (Tf₂O). After 24 hours, water (50 μ L) was added to this sample to give the expected 2-chloropyridinium triflate (δ -79.5).

Assignment of the 4-Cyclohexyl-6-methoxy-2-phenyl-quinazolin-1-ium triflate (3a•HOTf) resonances:

3) Addition of trifluoromethanesulfonic acid (56 µL, 0.63 mmol, 2.0 equiv) via syringe to a solution of 2-chloropyridine (36 µL, 0.38 mmol, 1.2 equiv) and quinazoline **3a** (100 mg, 0.310 mmol, 1 equiv) in CD₂Cl₂ (1 mL) at 23 °C followed by ¹H and ¹⁹F NMR analysis revealed the formation of the expected 2-chloropyridinium triflate and the 4-cyclohexyl-6-methoxy-2-phenyl-quinazolin-1-ium triflate (**3a**•HOTf). ¹H NMR (500 MHz) δ : ¹H NMR (500 MHz) δ :14.9 (br-s, 2.6H), 9.05 (br-s, 2.2H), 8.72 (dd, 1.1H, *J* = 5.5, 1.8 Hz), 8.70–8.63 (m, 3.0H), 8.28 (ddd, 1.0H, *J* = 8.2, 7.7, 1.9 Hz), 7.92 (dd, 0.9H, *J* = 9.3, 2.6 Hz), 7.82–7.75 (m, 2.8H), 7.75–7.70 (m, 2.0H), 7.61 (d, 1.1H, *J* = 2.7 Hz), 4.10 (s, 3.0H), 3.74 (tt, 1.1H, *J* = 11.4, 3.4 Hz), 2.14–1.90 (m, 7.2H), 1.70–1.59 (m, 2.1H), 1.48 (qt, 1.1H, *J* = 13.0, 3.5 Hz). ¹⁹F NMR (282 MHz) δ : –79.6.

4) Addition of trifluoromethanesulfonic acid (56 μ L, 0.63 mmol, 2.0 equiv) via syringe to a solution of quinazoline **3a** (100 mg, 0.310 mmol, 1 equiv) in CD₂Cl₂ (1 mL) at 23 °C followed by ¹⁹F NMR analysis of the mixture suggests *mono*-protonation to give the quinazolinium triflate **3a**•HOTf. ¹⁹F NMR (282 MHz) δ : -78.0, -79.1.

¹³C NMR monitoring of the conversion of amide 1a-¹³C and nitrile 2a to quinazoline 3a-¹³C:



4-Cyclohexyl-6-methoxy-2-phenyl-quinazoline-¹³C (3a-¹³C):

2-Chloropyridine (50 µL, 0.53 mmol, 1.2 equiv) was added via syringe to a solution of amide¹⁰ 1a-¹³C (¹³C=O, 100 mg, 0.439 mmol, 1 equiv) in CD₂Cl₂ (1.1 mL) at ambient temperature in an NMR tube under an atmosphere of argon. A sharp resonance corresponding to the carbonyl of amide $1a^{-13}C$ ($\delta 166.0$) was observed. Trifluoromethanesulfonic anhydride (80 µL, 0.48 mmol, 1.1 equiv) was added via syringe at 23 °C and the ¹³C NMR spectrum of the resulting mixture was immediately recorded. The starting amide was completely consumed and a new and persistent broad resonance was detected (δ 149.8). Addition of nitrile 2a (53 mg, 0.48 mmol, 1.1 equiv) at 23 °C and immediate ¹³C NMR monitoring led to observation of two new sharp resonance at $\delta 166.9$ and $\delta 96.9$ along with the remaining broad resonance at $\delta 149.8$ (~0.9:0.4,1.0, respectively). After 2 h heating of the mixture at 45 °C, the sample was cooled to 23 °C and the ¹³C NMR analysis of the reaction mixture revealed a dominant new resonance at δ 155.1 attributed to the desired product guinazoline 3a-¹³C•HOTf and disappearance of the broad resonance at δ 149.8. The resonances at δ 166.9 and δ 96.9 were weak (~5%) but remained The assignment of the resonance at $\delta 155.1$ to $3a^{-13}C$ •HOTf was confirmed detectable. independently by protonation of a sample of product $3a^{-13}C$ with TfOH (1 equiv) in CD₂Cl₂. The NMR tube was placed in a 45 °C oil bath and maintained at that temperature for an additional 14 h. At this time, the only dominant 13C resonance was that attributed to the desired product quinazoline $3a^{-13}C$ -HOTf (δ 155.1). An aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL103, was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15×1.5 cm) on neutralized silica gel to give the isotopically enriched quinazoline product $3a^{-13}C$ as a white solid (79 mg, 56%).

⁽¹⁰⁾ Amide $1a^{-13}C$ was readily prepared from the commercially available benzoic acid-*carboxy*- ^{13}C (99% atom % ^{13}C , Ph¹³CO₂H).

Chapter III

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Direct Synthesis of Pyridine Derivatives

Introduction and Background

As discussed in Chapter I, the pyridine substructure is one of the most prevalent heterocycles found in natural products, pharmaceuticals, and functional materials.¹ Many powerful methodologies for the synthesis of these heterocycles rely on condensation of amines and carbonyl compounds or cycloaddition reactions.²³ Cross-coupling chemistry also allows introduction of substituents to activated heterocycles.⁴ Our two-step pyridine synthesis⁵ (Chapter I) requires a ruthenium catalyst and can provide di- or tri-substituted pyridines. Herein we report a mild and convergent, transition-metal free, single-step procedure for the conversion of readily available *N*-vinyl and *N*-aryl amides⁶ to the corresponding substituted pyridines and quinolines, respectively (eq 1).



Results and Discussion

In earlier chapters of this thesis we reported a two-step synthesis of pyridines and a single-step synthesis of pyrimidines from readily available amides.⁵ These methodologies were made possible in part due to the recognition of the unique electrophilic activation^{5b} of amides with trifluoromethanesulfonic anhydride $(Tf_2O)^7$ in the presence of 2-chloropyridine (2-ClPyr) as the base additive.⁸ A variety of amides were employed in our pyrimidine synthesis using nitriles as σ -nucleophiles.^{5b} The current study focuses on the direct condensation of amides 1 with a wide range of π -nucleophiles (2 or 3) to provide the corresponding pyridine derivatives 4 (eq 1) in a single step.

We began our studies by investigating the use of alkoxy and silyloxy acetylenes in direct condensation with amides upon activation with Tf₂O and 2-ClPyr (eq 2).^{5b} Under optimum reaction conditions, these electron-rich π -nucleophiles provided the desired pyridine and quinoline derivatives in one step from the corresponding *N*-vinyl and *N*-aryl amides (Table 1, 4a-e). Similarly, the use of ynamide 2d and ynamine 2e readily provided the 4-amino pyridine derivatives in a single step (Table 1, 4f-1). While phenyl acetylene was not sufficiently



Table 1. Single-Step Synthesis of Pyridine Derivatives by Condensation of π -Nucleophiles with Amides.^a

^a Average of two experiments. Uniform conditions unless otherwise noted: Tf_2O (1.1 equiv), 2-CIPyr (1.2 equiv), nucleophile (2 or 3), CH_2CI_2 , heating: A = 23 °C, 1 h; B = 45 °C, 1 h; C = 140 °C, 20 min. ^b nucleophile (2.0 equiv). ^c 2-CIPyr (2.0 equiv). ^d only 10 min at 23 °C. ^e 2-CIPyr (5.0 equiv). ^f only 1 min heating, nucleophile (3.0 equiv). ^g nucleophile (1.1 equiv). ^h heated for 1 h. ⁱ 45% yield using **3a** with condition A. * Reactions performed by my colleague, Omar K. Ahmad.

nucleophilic, the more electron rich derivatives 2f and 2g served as π -nucleophiles in this pyridine synthesis (Table 1, 4m-o). Importantly, both electron-rich and electron-deficient *N*-aryl amides can be condensed with π -nucleophiles 2a-g (Table 1, compare 4i and 4j).



Based on mechanistic findings in our pyrimidine synthesis,^{5b} we propose this single-step pyridine synthesis proceeds by π -nucleophilic addition of acetylenes **2a-g** to an activated electrophile **5**⁹ followed by expulsion of 2-ClPyr•HOTf and annulation of the highly reactive intermediate **6** (eq 2). The condensation of the terminal alkyne **2f** with an *N*-(4-nitrophenyl) amide gave the desired quinoline **4o** in low yield (Table 1, 42% yield) along with 32% yield of cyclohex-1-yl-3-(4-methoxyphenyl)-propynone, the hydrolysis product of the corresponding alkynyl imine.¹⁰ This observation suggests competitive deprotonation of intermediate **6** (R^e=H) when cyclization to heterocycle **4** is slow.



We next examined the direct condensation of enol ethers with N-vinyl and N-aryl amides (eq 3). While ethyl vinyl ether (3a) could be used as a π -nucleophile when heating is not required (Table 1, 4p and 4u), we found triphenylsilyl vinyl ether (3b) to provide superior results in more challenging cases (Table 1, 4v-x). The use of excess nucleophile can be beneficial and provides an improved yield of the product (Table 1, 4u). Importantly, the use of silyl ether 3b in place of ethyl vinyl ether 3a eliminates the competitive addition of EtOH, generated in conversion of 7 to 4 (eq 3), to the activated intermediate 5. Both acyclic and cyclic trimethylsilyl enol ethers can be used in direct condensation with amides (Table 1, 4q-t). However, when desilylation competes with cyclization of oxonium ion 7 (eq 3), the use of more robust silyl enol ether derivatives is preferred. Condensation of amide 1a with enol ether 3e at 23 °C predominantly gave the vinylogous amide 8 (eq 4, 78%, 8:4y, >99:1, reaction performed by my colleague, Omar K. Ahmad) while heating the reaction mixture at 140 °C for 2 h¹¹ provided the desired quinoline (eq 4, 53%, 4y:8, >99:1, reaction performed by my colleague, Omar K. Ahmad).



Consistent with cyclization of intermediate 7 (eq 3), exposure of amide 8 to the standard reaction conditions provided <10% yield of 4y. Whereas the use of triisopropylsilyl ether derivatives was not optimal due to slow cyclization, the use of *tert*-butyldimethylsilyl ethers and microwave irradiation extends this chemistry to less reactive amide substrates (Table 1, 4y-cc). The use of enol ethers as the π -nucleophile in conjunction with electron deficient *N*-aryl amides (Table 1, compare 4y-aa) in this azaheterocycle synthesis is less efficient as compared to the use of acetylenic derivatives as the nucleophile (vide supra). Additionally, it should be noted that formamides do not give the corresponding pyridines with alkynyl or alkenyl π -nucleophiles due to rapid isocyanide formation.



The example shown in equation 5 highlights the greater efficiency of this chemistry when nucleophilic acetylenes are employed in place of enol derivatives. Activation of amide 1m under standard conditions and the use of silyl enol ether 3b provided the intramolecular annulation product 9 rather than the expected quinoline product. However, activation of amide 1m under identical conditions and the use of nucleophile 2d provided the desired quinoline derivative 4dd
without detectable formation of phenanthridine 9 (eq 5). The synthesis of pyridine 4ee from the corresponding N-vinyl amide 1n without loss of optical activity (eq 6) is noteworthy and is consistent with our prior observations.^{5b}



Conclusion

Herein we describe a single-step and convergent procedure for the synthesis of pyridine derivatives. This chemistry is compatible with a wide range of N-vinyl/aryl amides and π nucleophiles. This methodology alleviates the need for isolation of activated amide derivatives and provides rapid access to highly substituted pyridines with predictable control of substituent introduction. The versatility of this chemistry offers a valuable addendum to methodology for azaheterocycle synthesis.¹² Future work in this area will focus on intramolecular trapping of electrophilic activated amide intermediates with σ - and π -nucleophiles to generate polycyclic azaheterocycles.

- ⁽¹⁾ (a) Jones, G. In Comprehensive Heterocyclic Chemistry II; Katritzky, A. R.; Rees, C. W.; Scriven, E. F. V.; McKillop, A., Eds; Pergamon: Oxford, 1996; Vol. 5; p 167. (b) Henry, G. D. Tetrahedron 2004, 60, 6043. (c) Michael, J. P. Nat. Prod. Rep. 2005, 22, 627.
- ⁽²⁾ (a) Boger, D. L. J. Heterocycl. Chem. 1998, 35, 1003. (b) Jayakumar, S.; Ishar, M. P. S.; Mahajan, M. P. Tetrahedron 2002, 58, 379. (c) Zeni, G.; Larock, R. C. Chem. Rev. 2004, 104, 2285. (d) Varela, J. A.; Saá, C. Chem. Rev. 2004, 104, 3787.
- ⁽³⁾ (a) Varela, J. A.; Castedo, L.; Saá, C. J. Org. Chem. 2003, 68, 8595. (b) Sangu, K.; Fuchibe, K.; Akiyama, T. Org. Lett. 2004, 6, 353. (c) Zhang, X.; Campo, M. A.; Yao, T.; Larock, R. C. Org. Lett. 2005, 7, 763. (d) McCormick, M. M.; Duong, H. A.; Zuo, G.; Louie, J. J. Am. Chem. Soc. 2005, 127, 5030 and references therein.
- ⁽⁴⁾ Chinchilla, R.; Nájera, C.; Yus, M. Chem. Rev. 2004, 104, 2667. ⁽⁵⁾ (a) Movassaghi, M.; Hill, M. D. J. Am. Chem. Soc. 2006, 128, 4592. (b) Movassaghi, M.; Hill, M. D. J. Am.

- ⁽⁶⁾ (a) Muci, A. R.; Buchwald, S. L. Top. Curr. Chem. 2002, 219, 131. (b) Hartwig, J. F. In Handbook of Organopalladium Chemistry for Organic Synthesis; Negishi, E., Ed.; Wiley-Interscience: New York, 2002; p. 1051. (c) Beletskaya, I. P.; Cheprakov, A. V. Coordin. Chem. Rev. 2004, 248, 2337. (d) Dehli, J. R.; Legros, J.; Bolm, C. Chem. Commun. 2005, 973.
- ⁽⁷⁾ (a) For elegant prior studies on amide activation, see: Charette, A. B.; Grenon, M. Can. J. Chem. 2001, 79, 1694. (b) Review: Baraznenok, I. L.; Nenajdenko, V. G.; Balenkova, E. S. *Tetrahedron* **2000**, *56*, 3077. ⁽⁸⁾ Myers, A. G.; Tom, N. J.; Fraley, M. E.; Cohen, S. B.; Madar, D. J. J. Am. Chem. Soc. **1997**, *119*, 6072.
- ⁽⁹⁾ For discussion of the structure and chemistry of 5, see ref. 6b.

⁽¹⁰⁾ See the Experimental Section for details.

- ⁽¹¹⁾ Shorter reaction times gave a mixture of amide 8 and product 4y.
- ⁽¹²⁾ Movassaghi, M.; Hill, M. D.; Ohmad, O. K. J. Am. Chem. Soc. 2007, 129, 10096.

Chem. Soc. 2006, 128, 14254.

Experimental Section

General Procedures. All reactions were performed in oven-dried or flame-dried roundbottomed flasks, modified Schlenk (Kjeldahl shape) flasks, or glass pressure vessels. The flasks were fitted with rubber septa and reactions were conducted under a positive pressure of argon. Stainless steel syringes or cannulae were used to transfer air- and moisture-sensitive liquids. Flash column chromatography was performed as described by Still et al. using silica gel (60-Å pore size, 32–63 µm, standard grade, Sorbent Technologies) or non-activated alumina gel (80– 325 mesh, chromatographic grade, EM Science).¹ Analytical thin–layer chromatography was performed using glass plates pre-coated with 0.25 mm 230–400 mesh silica gel or neutral alumina gel impregnated with a fluorescent indicator (254 nm). Thin layer chromatography plates were visualized by exposure to ultraviolet light and/or by exposure to an ethanolic phosphomolybdic acid (PMA), an acidic solution of *p*-anisaldehyde (anis), an aqueous solution of ceric ammonium molybdate (CAM), an aqueous solution of potassium permanganate (KMnO₄) or an ethanolic solution of ninhydrin followed by heating (<1 min) on a hot plate (~250 °C). Organic solutions were concentrated on Büchi R-200 rotary evaporators at ~10 Torr (house vacuum) at 25–35 °C, then at ~0.5 Torr (vacuum pump) unless otherwise indicated.

Materials. Commercial reagents and solvents were used as received with the following exceptions: Dichloromethane, diethyl ether, tetrahydrofuran, acetonitrile, and toluene were purchased from J.T. Baker (CycletainerTM) and were purified by the method of Grubbs et al. under positive argon pressure.² 2-chloropyridine was distilled from calcium hydride and stored sealed under an argon atmosphere. The starting amides were prepared by acylation of the corresponding anilines³ or via previously reported copper–catalyzed C–N bond–forming reactions.^{4,5} Ethoxy acetylene (**2a**) was purchased from Aldrich as a solution in hexanes and purified by kugelrohr distillation before use (% wt. in hexanes determined by ¹H NMR analysis, ~47% wt.). Silyloxy acetylenes **2b** and **2c** were prepared according to Sun, J.; Kozmin., S. A. *Angew. Chem. Int. Ed.* **2006**, *45*, 4991. Ynamide **2d** was prepared according to Buissonneaud, D.; Cintrat, J.-C. *Tetrahedron Lett.* **2006**, *47*, 3139. Silyl enol ether **3b** was prepared according to Schaumann, E.; Tries, F. *Synthesis* **2002**, 191.

Instrumentation. All reaction conducted at 140 °C were performed in a CEM Discover Lab Mate microwave reactor. Proton nuclear magnetic resonance (¹H NMR) spectra were recorded with a Varian inverse probe 500 INOVA spectrometer or a Bruker 400 AVANCE spectrometer. Chemical shifts are recorded in parts per million from internal tetramethylsilane on the δ scale and are referenced from the residual protium in the NMR solvent (CHCl₃: δ 7.27, C₆HD₅: δ 7.16).

⁽¹⁾ Still, W. C.; Kahn, M.; Mitra, A. J. Org. Chem. 1978, 43, 2923.

 ⁽²⁾ Pangborn, A. B.; Giardello, M. A.; Grubbs, R. H.; Rosen, R. K.; Timmers, F. J. Organometallics 1996, 15, 1518.
⁽³⁾ For a general procedure, see: DeRuiter, J.; Swearingen, B. E.; Wandrekar, V.; Mayfield, C. A. J. Med. Chem. 1989, 32, 1033.

⁽⁴⁾ For the general procedure used for the synthesis of all *N*-vinyl amides, see: Jiang, L.; Job, G. E.; Klapars, A.; Buchwald, S. L. Org. Lett. 2003, 5, 3667.

⁽⁵⁾ For related reports, see: (a) Wolfe, J. P.; Wagaw, S.; Marcoux, J.-F.; Buchwald, S. L. Acc. Chem. Res. **1998**, 31, 805. (b) Hartwig, J. F. Acc. Chem. Res. **1998**, 31, 852. (c) Muci, A. R.; Buchwald, S. L. Top. Curr. Chem. **2002**, 219, 131. (d) Beletskaya, I. P.; Cheprakov, A. V. Coordin. Chem. Rev. **2004**, 248, 2337. (e) Dehli, J. R.; Legros, J.; Bolm, C. Chem. Commun. **2005**, 973.

Data is reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, q = quartet, m = multiplet), integration, coupling constant(s) in Hertz, assignment]. Carbon-13 nuclear magnetic resonance spectra were recorded with a Varian 500 INOVA spectrometer or a Bruker 400 AVANCE spectrometer and are recorded in parts per million from internal tetramethylsilane on the δ scale and are referenced from the carbon resonances of the solvent (CDCl₃: δ 77.2, benzened₆: δ 128.0). Data is reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, q = quartet, m = multiplet), coupling constant(s) in Hertz, assignment]. Infrared data were obtained with a Perkin-Elmer 2000 FTIR and are reported as follows: [frequency of absorption (cm⁻¹), intensity of absorption (s = strong, m = medium, w = weak, br = broad), assignment]. Chiral HPLC analysis was performed on an Agilent 1100 Series HPLC with a Whelk-O1 (*S*,*S*) column. We thank Dr. Li Li at the Massachusetts Institute of Technology Department of Chemistry instrumentation facility for obtaining mass spectroscopic data.



6-Cyclohexyl-4-ethoxy-2,3-dimethylpyridine (4a, Table 1):

Trifluoromethanesulfonic anhydride (50 μ L, 0.30 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1b** (50 mg, 0.28 mmol, 1 equiv) and 2chloropyridine (52 μ L, 0.55 mmol, 2.0 equiv) in dichloromethane (900 μ L) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ethyl ethynyl ether (**2a**, 82 mg, 0.55 mmol, 2.0 equiv, 47% wt. in hexane) was added via syringe. The resulting solution was allowed to warm to ambient temperature. After 1 h, triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 20% EtOAc/1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine derivative **4a** as a pale yellow solid (55 mg, 85%).

'H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	6.48 (s, 1H, ArH), 4.06 (q, 2H, $J = 7.0$ Hz, OCH ₂ CH ₃), 2.66–2.58 (m, 1H, ^c C ₆ H ₁₁), 2.46 (s, 3H, CH ₃), 2.10 (s, 3H, CH ₃), 1.98–1.93 (m, 2H, ^c C ₆ H ₁₁), 1.86–1.80 (m, 2H, ^c C ₆ H ₁₁), 1.77–1.71 (m, 1H, ^c C ₆ H ₁₁), 1.48–1.38 (m, 7H, ^c C ₆ H ₁₁ , OCH ₂ CH ₃), 1.32–1.24 (m, 1H, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	164.7, 163.4, 156.5, 117.2, 101.0, 63.5, 47.0, 33.5, 26.8, 26.3, 22.8, 14.8, 10.9.
FTIR (neat) cm ⁻¹ :	3231 (w), 3064 (w), 2955 (s), 2865 (s), 1726 (w), 1634 (m), 1594 (w), 1535 (s), 1498 (s), 1475 (s), 1240 (m).
HRMS (ESI):	calcd for $C_{15}H_{24}NO [M+H]^+$: 234.1852, found: 234.1844.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.27 (UV, KMnO ₄).



<u>2-Cyclohexyl-4-ethoxy-5,7-dimethoxyquinoline (4b, Table 1):</u>

Trifluoromethanesulfonic anhydride (83 μ L, 0.50 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1c (120 mg, 0.456 mmol, 1 equiv) and 2chloropyridine (52 μ L, 0.55 mmol, 1.2 equiv) in dichloromethane (1.5 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, ethyl ethynyl ether (2a, 128 mg, 0.91 mmol, 2.0 equiv, 50% wt. in hexane) was added via syringe. The resulting solution was allowed to warm to ambient temperature. After 1 h, triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 30% EtOAc and 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline derivative 4b as a white solid (110 mg, 75%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	6.97 (br s, 1H, ArH), 6.51 (s, 1H, ArH), 6.44 (d, 1H, $J = 2.3$ Hz, ArH), 4.22 (q, 2H, $J = 7.0$ Hz, OCH ₂ CH ₃), 3.93 (s, 3H, OCH ₃), 3.91 (s, 3H, OCH ₃), 2.82–2.72 (m, 1H, ^c C ₆ H ₁₁), 2.04–1.99 (m, 2H, ^c C ₆ H ₁₁), 1.90–1.85 (m, 2H, ^c C ₆ H ₁₁), 1.80–1.74 (m, 1H, ^c C ₁ H ₁₁), 1.90–1.85 (m, 2H, ^c C ₆ H ₁₁), 1.80–1.74
	(iii, iii), C_6H_{11}), $1.00-1.51$ (iii, $5H_1$, C_6H_{11}), OCH ₂ CH ₃), $1.50-1.41$ (m, 2H, ${}^{\circ}C_6H_{11}$), 1.32 (tt, 1H, $J = 12.7$, 3.5 Hz, ${}^{\circ}C_6H_{11}$).
¹³ C NMR (125 MHz, DMF- <i>d</i> ₇ , 20 °C) δ:	168.6, 164.0, 161.0, 158.2, 152.8, 107.6, 100.2, 98.2, 98.2, 64.4, 56.3, 55.7, 47.9, 33.1, 26.7, 26.3, 14.7.
FTIR (neat) cm^{-1} :	2929 (m), 2851 (w), 1725 (w), 1615 (s), 1591 (s), 1451 (m), 1404 (m), 1382 (m), 1206 (m), 1155 (s).
HRMS (ESI):	calcd for $C_{19}H_{26}NO_3 [M+H]^+$: 316.1907, found: 316.1904.
TLC (40% EtOAc in hexanes), R _f :	0.54 (UV, KMnO ₄).



nOe data:



7-Butyl-6-phenyl-8-(triisopropylsilyloxy)-3,4-dihydro-2H-pyrano[3,2-b]pyridine (4c, Table 1):

Trifluoromethanesulfonic anhydride (72 μ L, 0.43 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1d (80 mg, 0.39 mmol, 1 equiv) and 2chloropyridine (75 μ L, 0.79 mmol, 2.0 equiv) in dichloromethane (1.3 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ynol 2b⁶ (200 mg, 0.788 mmol, 2.00 equiv) was added via syringe. The resulting solution was allowed to warm to ambient temperature. After 10 min., triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 20% EtOAc and 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine derivative 4c as a clear oil (110 mg, 63%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.40–7.30 (m, 5H, ArH), 4.18 (t, 2H, $J = 5.1$ Hz, CH ₂ CH ₂ CH ₂ O), 2.95 (t, 2H, $J = 6.6$ Hz, CH ₂ CH ₂ CH ₂ O), 2.56–2.50 (m, 2H, CH ₂ CH ₂ CH ₂ CH ₃), 2.15–2.09 (m, 2H, CH ₂ CH ₂ CH ₂ O), 1.45– 1.38 (m, 2H, CH ₂ CH ₂ CH ₂ CH ₃), 1.35–1.27 (m, 3H, Si(CH(CH ₃) ₂) ₃), 1.25–1.15 (m, 2H, CH ₂ CH ₂ CH ₂ CH ₃), 1.11 (d, 18H, $J = 7.4$ Hz, Si(CH (CH ₃) ₂) ₃), 0.76 (t, 3H, $J = 7.3$ Hz, CH ₂ CH ₂ CH ₂ CH ₃).
¹³ C NMR (100 MHz, CDCl ₃ , 20 °C) δ:	152.4, 149.9, 141.6, 141.0, 140.5, 129.1, 128.1, 127.3, 125.9, 65.8, 32.3, 28.3, 27.2, 23.0, 22.7, 18.3, 14.6, 13.9.
FTIR (neat) cm ⁻¹ :	2981 (w), 2927 (s), 2852 (m), 1731 (w), 1588 (s), 1575 (s), 1468 (m), 1328 (m), 1217 (m), 1123 (s).
HRMS (ESI):	calcd for $C_{18}H_{22}NO_2 [M-Si(^{i}Pr)_3+2H]^+: 284.1645,$ found: 284.1644.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.21 (UV, KMnO ₄).

⁽⁶⁾ For preparation of 2b and 2c, see Sun, J.; Kozmin., S. A. Angew. Chem. Int. Ed. 2006, 45, 4991.



<u>2-sec-Butyl-3-butyl-4-(triisopropylsilyloxy)-5,6,7,8-tetrahydroquinoline (4d, Table 1):</u>

Trifluoromethanesulfonic anhydride (80 μ L, 0.49 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1e** (80 mg, 0.44 mmol, 1 equiv) and 2-chloropyridine (84 μ L, 0.88 mmol, 2.0 equiv) in dichloromethane (1.5 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ynol **2b** (123 mg, 0.485 mmol, 1.10 equiv) was added via syringe. The resulting solution was allowed to warm to ambient temperature. After 10 min., triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 3% EtOAc and 0.5% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine derivative **4d** as a clear oil (134 mg, 73%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	2.92–2.79 (m, 3H), 2.70–2.50 (m, 4H), 1.85–1.70 (m, 5H), 1.62–1.53 (m, 1H), 1.43–1.30 (m, 7H), 1.21 (d, 3H, $J = 6.8$ Hz, CH ₃ CH ₂ CHCH ₃), 1.10 (d, 18H, $J = 7.5$ Hz, Si(CH(CH ₃) ₂) ₃), 0.95–0.92 (m, 3H), 0.79 (t, 3H, $J = 7.4$ Hz, CH ₃ CH ₂ CHCH ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	162.3, 159.8, 155.3, 123.7, 120.4, 38.0, 33.5, 33.0, 30.1, 26.4, 24.4, 23.4, 23.3, 22.9, 21.0, 18.2, 14.6, 14.3, 12.7.
FTIR (neat) cm ⁻¹ :	3257 (w), 3090 (w), 2958 (s), 2933 (s), 2869 (s), 1631 (w), 1612 (w), 1502 (s), 1464 (m), 1427 (m), 1210 (w).
HRMS (ESI):	calcd for C ₂₆ H ₄₈ NOSi [M+H] ⁺ : 418.3500, found: 418.3510.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.76 (UV, KMnO ₄).

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2-Cyclohexyl-5,7-dimethoxy-3-phenyl-4-(triisopropylsilyloxy)quinoline (4e, Table 1):

Trifluoromethanesulfonic anhydride (69 μ L, 0.42 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1c (100 mg, 0.380 mmol, 1 equiv) and 2chloropyridine (43 μ L, 0.46 mmol, 1.2 equiv) in dichloromethane (1.3 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ynol 2c^e (115 mg, 0.418 mmol, 1.10 equiv) was added via syringe. The resulting solution was allowed to warm to ambient temperature. After 1 h, triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 4% EtOAc and 0.5% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline derivative 4e as a clear oil (154 mg, 78%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.41–7.30 (m, 5H, ArH), 6.97 (d, 1H, $J = 2.4$ Hz, ArH), 6.41 (d, 1H, $J = 2.4$ Hz, ArH), 3.95 (s, 3H, OCH ₃), 3.82 (s, 3H, OCH ₃), 2.58 (tt, 1H, $J = 11.1$, 3.5 Hz, ^c C ₆ H ₁₁), 1.75–1.55 (m, 7H, ^c C ₆ H ₁₁), 1.28 (qt, 1H, $J = 13.0$, 3.4 Hz, ^c C ₆ H ₁₁), 1.12–1.00 (m, 2H, ^c C ₆ H ₁₁), 0.83 (d, 18H, $J = 7.2$ Hz, Si(CH(CH ₃) ₂) ₃), 0.75–0.67 (m, 3H, Si(CH(CH ₃) ₂) ₃).
¹³ C NMR (100 MHz, CDCl ₃ , 20 °C) δ:	166.7, 160.2, 157.9, 157.5, 152.6, 137.3, 132.8, 128.2, 127.2, 123.9, 110.4, 100.3, 97.5, 55.6, 55.0, 42.8, 32.3, 26.6, 26.1, 18.1, 14.2.
FTIR (neat) cm ⁻¹ :	3057 (w), 2942 (s), 2865 (s), 1619 (s), 1567 (s), 1465 (m), 1451 (m), 1375 (s), 1358 (s), 1250 (m), 1207 (s).
HRMS (ESI):	calcd for C ₃₂ H ₄₆ NO ₃ Si [M+H] ⁺ : 520.3241, found: 520.3245.
TLC (10% EtOAc in hexanes), $R_{\rm f}$:	0.32 (UV, KMnO ₄).



<u>3-(2-sec-Butyl-3-phenyl-5,6,7,8-tetrahydroquinolin-4-yl)oxazolidin-2-one (4g, Table 1):</u>

Trifluoromethanesulfonic anhydride (80 μ L, 0.49 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1e** (80 mg, 0.44 mmol, 1 equiv) and 2chloropyridine (84 μ L, 0.88 mmol, 2.0 equiv) in dichloromethane (1.5 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ynamide **2d**Error! Bookmark not defined. (91 mg, 0.49 mmol, 1.1 equiv) was added as a solid in one portion and the reaction flask was rapidly purged and sealed under an argon atmosphere. After 1 h, triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 30% EtOAc and 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine derivative **4g** as a white solid (126 mg, 81%).

¹H NMR (500 MHz, CDCl₃, 20 °C, minor atropisomer noted by *) δ:7.48–7.37 (m, 3H, ArH; 3H, ArH*), 7.33–7.28 (m, 1H, ArH; 1H, ArH*), 7.20–7.16 (m, 1H, ArH; 1H, ArH*), 4.28–4.21 (m, 1H, NCH₂CH₂O; 1H, NCH₂CH₂O*), 3.86–3.80 (m, 1H, NCH₂CH₂O; 1H, NCH₂CH₂O*), 3.59–3.48 (m, 1H, NCH₂CH₂O; 1H, NCH₂CH₂O*), 3.14–3.03 (m, 1U, NCH CH O: 1U, NCH CH O*), 3.02–3.90 (m)

1H, NCH₂CH₂O; 1H, NCH₂CH₂O*), 3.03–2.80 (m, 3H, $CH_2CH_2CH_2CH_2$; 3H, $CH_2CH_2CH_2CH_2^*$), 2.69-2.60 CH₃CH₂CHCH₃; (m, 1H, 1H, $CH_3CH_2CHCH_3^*),$ 2.59-2.51 (m, 1H. CH₂CH₂CH₂CH₂; 1H, CH₂CH₂CH₂CH₂*), 1.97-1.62 (m, 5H, $CH_2CH_2CH_2$, $CH_3CH_2CHCH_3$; 5H, CH₂CH₂CH₂CH₂*, CH₃CH₂CHCH₃*), 1.60-1.37 CH₃CH₂CHCH₃; (m, 1H, 1H, $CH_3CH_2CHCH_3^*$), 1.23 (d, 3H, J = 6.7 Hz, $CH_3CH_2CHCH_3$), 0.99 (d, 3H, J = 6.8 Hz, $CH_3CH_2CHCH_3^*$), 0.82 (t, 3H, J = 7.4 Hz, $CH_3CH_2CHCH_3^*$), 0.58 (t, 3H, J = 7.3Hz, $CH_3CH_2CHCH_3$).

¹³C NMR (125 MHz, CDCl₃, 20 °C) δ: 162.4, 162.2, 158.5, 158.5, 156.5, 156.5, 141.0, 140.9, 136.5, 136.4, 132.8, 132.7, 130.1, 129.9, 129.1, 129.1, 129.0, 129.0, 128.2, 128.1, 127.9, 127.7, 62.9, 46.1, 45.9, 39.7, 38.8, 38.8, 33.1, 29.9, 29.1, 24.4, 22.9, 22.9, 22.5, 20.9, 20.9, 12.7, 12.3.

FTIR (neat) cm^{-1} :

HRMS (ESI):

2959 (m), 2933 (m), 1753 (s), 1603 (w), 1575 (w), 1481 (w), 1440 (m), 1409 (m), 1230 (m), 1181 (w).

calcd for $C_{22}H_{27}N_2O_2$ [M+H]⁺: 351.2067, found: 351.2057.

TLC (30% EtOAc in hexanes), $R_{\rm f}$:

0.29 (UV, KMnO₄).



3-(2-Morpholino-3-phenylquinolin-4-yl)oxazolidin-2-one (4h, Table 1):

Trifluoromethanesulfonic anhydride (40 μ L, 0.24 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of urea **1f** (46 mg, 0.22 mmol, 1 equiv) and 2-chloropyridine (25 μ L, 0.27 mmol, 1.2 equiv) in dichloromethane (750 μ L) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ynamide **2d** (83 mg, 0.44 mmol, 1.1 equiv) was added as a solid in one portion and the reaction flask was rapidly purged and sealed under an argon atmosphere. The resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before triethylamine (300 μ L) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 40% EtOAc and 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline derivative **4h** as a white solid (58 mg, 70%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.93–7.90 (m, 1H, ArH), 7.73–7.69 (m, 1H, ArH), 7.66 (qd, 1H, $J = 6.9$, 1.5 Hz, ArH), 7.63–7.58 (m, 1H, ArH), 7.55–7.47 (m, 3H, ArH), 7.45–7.41 (m, 2H, ArH), 4.46–4.40 (m, 1H, NCH ₂ CH ₂ O), 4.08– 4.02 (m, 1H, NCH ₂ CH ₂ O), 3.64–3.52 (m, 5H, NCH ₂ CH ₂ O, N(CH ₂ CH ₂ O), 3.22–3.13 (m, 4H, N(CH ₂ CH ₂) ₂ O), 3.04–2.99 (m, 1H, NCH ₂ CH ₂ O).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	159.6, 157.8, 147.8, 140.4, 135.6, 130.3, 129.6, 129.5, 128.8, 128.7, 128.6, 128.3, 128.2, 125.3, 122.5, 122.2, 66.7, 63.2, 49.7, 46.1.
FTIR (neat) cm ⁻¹ :	3490 (w), 3060 (m), 2965 (m), 2893 (m), 2849 (m), 2248 (w), 1755 (s), 1587 (s), 1491 (s), 1409 (s), 1238 (s), 1117 (s).
HRMS (ESI):	calcd for $C_{22}H_{22}N_3O_3$ [M+H] ⁺ : 376.1656, found: 376.1668.
TLC (40% EtOAc in hexanes), $R_{\rm f}$:	0.28 (UV, KMnO ₄).



<u>3-(2-Cyclohexyl-5,7-dimethoxy-3-phenylquinolin-4-yl)oxazolidin-2-one (4i, Table 1):</u>

Trifluoromethanesulfonic anhydride (69 μ L, 0.42 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1c (100 mg, 0.380 mmol, 1 equiv) and 2-chloropyridine (43 μ L, 0.46 mmol, 1.2 equiv) in dichloromethane (1.3 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ynamide 2d (78 mg, 0.42 mmol, 1.1 equiv) was added as a solid in one portion, and the reaction flask was rapidly purged and sealed under an argon atmosphere. The resulting solution was allowed to warm to ambient temperature. After 1 h, triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 40% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline derivative 4i as a white solid (140 mg, 85%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.49–7.40 (m, 4H, ArH), 7.23–7.19 (m, 1H, ArH), 7.11 (d, 1H, $J = 2.3$ Hz, ArH), 6.57 (d, 1H, $J = 2.3$ Hz, ArH), 4.33–4.28 (m, 1H, NCH ₂ CH ₂ O), 3.97 (s, 3H, OCH ₃), 3.96–3.89 (m, 4H, OCH ₃ , NCH ₂ CH ₂ O), 3.76–3.70 (m, 1H, NCH ₂ CH ₂ O), 3.24–3.18 (m, 1H, NCH ₂ CH ₂ O), 2.67–2.60 (m, 1H, $^{\circ}C_{6}H_{11}$), 1.91–1.74 (m, 3H, $^{\circ}C_{6}H_{11}$), 1.63–1.60 (m, 2H, $^{\circ}C_{6}H_{11}$), 1.35–1.10 (m, 4H, $^{\circ}C_{6}H_{11}$), 1.09–0.99
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	(m, 1H, ${}^{c}C_{6}H_{11}$). 166.4, 160.9, 157.8, 156.0, 152.3, 138.5, 136.3, 133.4, 130.3, 128.8, 128.7, 127.9, 127.8, 111.4, 101.2, 99.9, 62.9, 56.6, 55.8, 47.8, 43.2, 32.5, 32.2. 26.5, 26.5, 26.0.
FTIR (neat) cm ⁻¹ :	2924 (m), 2850 (w), 1745 (s), 1617 (s), 1573 (s), 1479 (w), 1446 (w), 1423 (m), 1407 (m), 1249 (m).
HRMS (ESI):	calcd for $C_{26}H_{29}N_2O_4$ [M+H] ⁺ : 433.2122, found: 433.2107.
TLC (40% EtOAc in hexanes), R _f :	0.31 (UV, KMnO ₄).



4-(5-Phenyl-6-(trimethylsilyl)thieno[3,2-b]pyridin-7-yl)morpholine (4l, Table 1):

Trifluoromethanesulfonic anhydride (89 μ L, 0.54 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1h** (100 mg, 0.492 mmol, 1 equiv) and 2-chloropyridine (233 μ L, 2.45 mmol, 5.00 equiv) in dichloromethane (1.6 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ynamine **2e** (270 mg, 1.48 mmol, 3.00 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 1 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 20% EtOAc was 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine derivatives **41** as a pale yellow oil (138 mg, 76%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.74–7.72 (m, 1H, ArH), 7.62–7.60 (m, 1H, ArH), 7.50–7.47 (m, 2H, ArH), 7.45–7.40 (m, 3H, ArH), 3.98–3.95 (m, 4H, N(CH ₂ CH ₂) ₂ O), 3.42 (br s, 4H, N(CH ₂ CH ₂) ₂ O), 0.05 (s, 9H, Si(CH ₃) ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	181.1, 166.3, 161.8, 158.7, 144.6, 130.6, 129.7, 128.3, 128.3, 125.9, 125.8, 67.0, 51.9, 2.6.
FTIR (neat) cm ⁻¹ :	2955 (m), 2894 (m), 2856 (m), 1572 (w), 1515 (s), 1471 (s), 1445 (m), 1337 (s), 1259 (s), 1113 (s).
HRMS (ESI):	calcd for C ₂₀ H ₂₅ N ₂ OSSi [M+H] ⁺ : 369.1451, found: 369.1452.
TLC (30% EtOAc in hexanes), $R_{\rm f}$:	0.44 (UV, KMnO ₄).



7-(4-Methoxyphenyl)-5-phenylthieno[3,2-b]pyridine (4m, Table 1):

Trifluoromethanesulfonic anhydride (80 μ L, 0.49 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1h (90 mg, 0.44 mmol, 1 equiv) and 2chloropyridine (50 μ L, 0.53 mmol, 1.2 equiv) in dichloromethane (1.5 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the acetylene 2f (117 mg, 0.886 mmol, 2.00 equiv) was added via syringe. The resulting solution was allowed to warm to ambient temperature. After 1 h, triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc and 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine derivative 4m as a pale yellow oil (90 mg, 64%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.13–8.09 (m, 2H, ArH), 7.82–7.77 (m, 3H, ArH), 7.73 (s, 1H, ArH), 7.70 (d, 1H, $J = 5.5$ Hz, ArH), 7.55–7.50 (m, 2H, ArH), 7.46 (tt, 1H, $J = 7.4$, 1.3 Hz, ArH), 7.12–7.09 (m, 2H, ArH), 3.92 (s, 3H, OCH ₃).
¹³ C NMR (100 MHz, CDCl ₃ , 20 °C) δ:	160.6, 157.3, 156.5, 144.7, 140.0, 131.2, 130.9, 130.5, 129.5, 129.0, 129.0, 127.5, 126.1, 115.7, 114.7, 55.6.
FTIR (neat) cm ⁻¹ :	3061 (w), 2931 (w), 2836 (w), 1996 (w), 1608 (m), 1565 (m), 1513 (s), 1357 (m), 1294 (m), 1257 (s), 1031 (m).
HRMS (ESI):	calcd for $C_{20}H_{16}NOS [M+H]^+$: 318.0947, found: 318.0944.
TLC (20% EtOAc in hexanes). Rf:	0.37 (U V, KMnO₄).



nOe data:

15% nOe



5,7-dimethoxy-2-phenylquinoline (4p, Table 1):

Trifluoromethanesulfonic anhydride (55 μ L, 0.33 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1c (80 mg, 0.30 mmol, 1 equiv) and 2-chloropyridine (35 μ L, 0.37 mmol, 1.2 equiv) in dichloromethane (1.0 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the vinyl ether 3a (24 mg, 0.33 mmol, 1.1 equiv) was added via syringe. The resulting solution was allowed to warm to ambient temperature. After 1 h, aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline product **4p** as a pale yellow solid (81 mg, 99%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.36 (d, 1H, $J = 8.5$ Hz, ArH), 7.16 (d, 1H, $J = 8.7$ Hz, ArH), 7.00 (d, 1H, $J = 2.1$ Hz, ArH), 6.47 (d, 1H, $J = 2.3$ Hz, ArH), 3.97 (s, 3H, OCH ₃), 3.95 (s, 3H, OCH ₃), 2.87 (tt, 1H, $J = 12.0$, 3.5 Hz, ^c C ₆ H ₁₁), 2.06–2.0 (m, 2H, ^c C ₆ H ₁₁), 1.94–1.86 (m, 2H, ^c C ₆ H ₁₁), 1.82–1.76 (m, 1H, ^c C ₆ H ₁₁), 1.62 (qd, 2H, J = 12.6, 3.0 Hz, ^c C ₆ H ₁₁), 1.48 (qt, 2H, $J = 12.6$, 3.2 Hz, ^c C ₆ H ₁₁), 1.35 (qt, 1H, $J = 12.8$, 3.5 Hz, ^c C ₆ H ₁₁).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	167.8, 156.2, 135.3, 131.2, 127.9, 116.5, 115.4, 99.7, 97.5, 55.9, 55.8, 47.8, 33.1, 26.8, 26.3.
FTIR (neat) cm ⁻¹ :	2929 (s), 2852 (m), 1643 (m), 1624 (s), 1608 (s), 1580 (s), 1511 (w), 1452 (m), 1397 (m), 1205 (s), 1151 (s).
HRMS (ESI):	calcd for C ₁₇ H ₂₂ NO ₂ [M+H] ⁺ : 272.1645, found: 272.1644.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.28 (UV, KMnO ₄).



2-Cyclohexyl-5,7-dimethoxy-4-methylquinoline (4q, Table 1):

Trifluoromethanesulfonic anhydride (76 μ L, 0.46 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1c (110 mg, 0.418 mmol, 1 equiv) and 2-chloropyridine (48 μ L, 0.50 mmol, 1.2 equiv) in dichloromethane (1.4 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the enol ether 3c (109 mg, 0.836 mmol, 2.00 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 1 h, aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 3% EtOAc and 1% Et₃N \rightarrow 10% EtOAc and 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline product 4q as a white solid (70 mg, 59%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	6.98 (s, 1H, ArH), 6.88 (s, 1H, ArH), 6.44 (s, 1H, ArH), 3.92 (s, 3H, OCH ₃), 3.89 (s, 3H, OCH ₃), 2.83–2.74 (m, 4H, CH ₃ , $^{\circ}C_{6}H_{11}$), 2.02–1.96 (m, 2H, $^{\circ}C_{6}H_{11}$), 1.91–1.84 (m, 2H, $^{\circ}C_{6}H_{11}$), 1.80–1.74 (m, 1H, $^{\circ}C_{6}H_{11}$), 1.59 (qd, 2H, $J = 12.7$, 3.3 Hz, $^{\circ}C_{6}H_{11}$), 1.51–1.40 (m, 2H, $^{\circ}C_{6}H_{11}$), 1.32 (qt, 1H, $J = 12.5$, 3.4 Hz, $^{\circ}C_{6}H_{11}$).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	166.8, 160.4, 158.7, 151.4, 145.9, 119.6, 115.4, 100.5, 97.9, 55.6, 55.6, 47.5, 33.0, 26.7, 26.3, 24.6.
FTIR (neat) cm^{-1} :	2998 (w), 2927 (s), 2851 (m), 1692 (w), 1619 (s), 1593 (s), 1452 (m), 1405 (m), 1252 (m), 1155 (m).
HRMS (ESI):	calcd for C ₁₈ H ₂₄ NO ₂ [M+H] ⁺ : 286.1802, found: 286.1804.
TLC (20% EtOAc in hexanes), R _f ::	0.46 (UV, KMnO ₄).



4-Cyclohexyl-7.9-dimethoxy-2.3-dihydro-1*H*-cyclopenta[c]quinoline (4s, Table 1):

Trifluoromethanesulfonic anhydride (76 μ L, 0.46 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1c (110 mg, 0.418 mmol, 1 equiv) and 2-chloropyridine (48 μ L, 0.51 mmol, 1.2 equiv) in dichloromethane (1.4 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the enol ether 3e (131 mg, 0.836 mmol, 2.00 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 1 h, the reaction mixture was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc and 0.5% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline product **4s** as a white solid (69 mg, 53%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.00 (s, 1H, Ar H), 6.41 (s, 1H, Ar H), 3.92 (s, 3H, OC H ₃), 3.89 (s, 3H, OC H ₃), 3.52–3.44 (m, 2H, C H ₂ CH ₂ CH ₂), 3.04–2.96 (m, 2H, CH ₂ CH ₂ CH ₂), 2.88–2.80 (m, 1H, $^{c}C_{6}H_{11}$), 2.18–2.10 (m, 2H, CH ₂ C H ₂ CH ₂), 1.94–1.70 (m, 7H, $^{c}C_{6}H_{11}$), 1.48–1.36 (m, 3H, $^{c}C_{6}H_{11}$).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	163.3, 160.0, 157.4, 150.2, 149.9, 133.0, 113.9, 100.1, 97.5, 55.7, 55.6, 44.6, 35.4, 31.4, 30.5, 26.9, 26.2, 24.5.
FTIR (neat) cm ⁻¹ :	2998 (w), 2928 (s), 2850 (m), 1621 (s), 1583 (s), 1508 (w), 1450 (m), 1414 (w), 1360 (m), 1251 (m), 1205 (s).
HRMS (ESI):	calcd for C ₂₀ H ₂₆ NO ₂ [M+H] ⁺ : 312.1958, found: 312.1961.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.59 (UV, KMnO ₄).



6-Cyclohexyl-1,3-dimethoxy-7,8,9,10-tetrahydrophenanthridine (4t, Table 1):

Trifluoromethanesulfonic anhydride (76 μ L, 0.46 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1c (110 mg, 0.418 mmol, 1 equiv) and 2-chloropyridine (79 μ L, 0.84 mmol, 2.0 equiv) in dichloromethane (1.4 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the enol ether 3f (142 mg, 0.836 mmol, 2.00 equiv) was added via syringe, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a preheated oil bath at 45 °C and maintained at that temperature. After 1 h, the reaction mixture was allowed to cool to ambient temperature and aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc and 0.5% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline product 4t as a white solid (83 mg, 61%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	6.96 (d, 1H, $J = 2.5$ Hz, ArH), 6.43 (d, 1H, $J = 2.5$ Hz, ArH), 3.93 (s, 3H, OCH ₃), 3.87 (s, 3H, OCH ₃), 3.42–3.48 (m, 2H, CH ₂ CH ₂ CH ₂ CH ₂), 3.00–2.93 (m, 1H, $^{c}C_{6}H_{11}$), 2.87–2.82 (m, 2H, CH ₂ CH ₂ CH ₂ CH ₂ CH ₂), 1.92–1.73 (m, 11H, $^{c}C_{6}H_{11}$, CH ₂ CH ₂ CH ₂ CH ₂ CH ₂), 1.48–1.37 (m, 3H, $^{c}C_{6}H_{11}$).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	165.6, 159.2, 158.6, 149.5, 143.6, 125.6, 114.9, 101.0, 98.2, 55.6, 55.6, 41.8, 32.1, 30.5, 27.1, 26.7, 26.4, 23.1, 22.6.
FTIR (neat) cm ⁻¹ :	2926 (s), 2851 (m), 1617 (s), 1577 (m), 1449 (m), 1406 (w), 1307 (w), 1245 (m), 1206 (s), 1157 (s).
HRMS (ESI):	calcd for $C_{21}H_{28}NO_2 [M+H]^+$: 326.2115, found: 326.2105.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.59 (UV, KMnO ₄).



2-Phenylquinoline (4u, Table 1):

Trifluoromethanesulfonic anhydride (92 μ L, 0.56 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1j** (100 mg, 0.507 mmol, 1 equiv) and 2-chloropyridine (58 μ L, 0.61 mmol, 1.2 equiv) in dichloromethane (1.7 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the vinyl ether **3a** (73 mg, 1.0 mmol, 2.0 equiv) was added via syringe. The resulting solution was allowed to warm to ambient temperature. After 1 h, aqueous sodium hydroxide solution (1 mL, 1N) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline product **4u** as a white solid (77 mg, 74%).⁷

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	8.25 (d, 1H, $J = 8.5$ Hz, ArH), 8.21–8.16 (m, 3H, ArH), 7.90 (d, 1H, $J = 8.5$ Hz, ArH), 7.85 (d, 1H, $J = 8.2$ Hz, ArH), 7.75 (ddd, 1H, $J = 8.5$, 7.0, 1.5 Hz, ArH), 7.57–7.52 (m, 3H, ArH), 7.48 (tt, 1H, $J = 7.3$, 1.2 Hz, ArH).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	157.5, 148.4, 139.8, 137.0, 129.9, 129.9, 129.5, 129.0, 127.8, 127.7, 127.3, 126.5, 119.2.
FTIR (neat) cm ⁻¹ :	3189 (s), 3055 (w), 2091 (s), 1617 (w), 1597 (s), 1491 (m), 1447 (s).
HRMS (EI):	calcd for $C_{15}H_{11}N[M]^+$: 205.0886, found: 205.0885.
Analysis	calcd for C ₁₅ H ₁₁ N: C, 87.77; H, 5.40; N, 6.82, found: C, 87.55; H, 5.37; N, 6.84.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.51 (UV, CAM).

⁽⁷⁾ For a two-step synthesis of **4u**, see Movassaghi, M.; Hill, M. D. J. Am. Chem. Soc. **2006**, 128, 4592.



2-Phenyl-5.6.7.8-tetrahydroquinoline (4v, Table 1):

Trifluoromethanesulfonic anhydride (57 μ L, 0.34 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 1k (63 mg, 0.31 mmol, 1 equiv) and 2-chloropyridine (59 μ L, 0.63 mmol, 2.0 equiv) in dichloromethane (1.1 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the enol ether 3b⁸ (189 mg, 0.626 mmol, 2.00 equiv) was added, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 3% EtOAc and 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine derivative 4v as a pale yellow solid (46 mg, 71%).⁷

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	7.97–7.94 (m, 2H, ArH), 7.48–7.36 (m, 5H, ArH), 3.00 (t, 2H, $J = 6.4$ Hz, CH ₂), 2.82 (t, 2H, $J = 6.4$ Hz, CH ₂), 1.98–1.91 (m, 2H, CH ₂), 1.89–1.83 (m, 2H, CH ₂).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	157.4, 154.9, 140.1, 137.6, 130.9, 128.8, 128.5, 127.0, 118.1, 33.1, 28.8, 23.4, 23.0.
FTIR (neat):	3061 (w), 3032 (w), 2935 (s), 2860 (m), 1590 (m), 1566 (m), 1460 (s), 1434 (m), 1253 (m).
HRMS (ESI):	calcd for C ₁₅ H ₁₆ N [M+H] ⁺ : 210.1277, found: 210.1279.
TLC (20% EtOAc in hexanes), Rr.	0.48 (UV, KMnO₄).

⁽⁸⁾ For preparation of **3b** see Schaumann, E.; Tries, F. Synthesis **2002**, 191.



5-Phenylthieno[3,2-b]pyridine (4w, Table 1):

Trifluoromethanesulfonic anhydride (63 μ L, 0.38 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1h** (70 mg, 0.34 mmol, 1 equiv) and 2-chloropyridine (39 μ L, 0.41 mmol, 1.2 equiv) in dichloromethane (1.2 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the enol ether **3b** (208 mg, 0.688 mmol, 2.00 equiv) was added, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 10% EtOAc and 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine product **4w**⁷ as a pale yellow solid (55 mg, 76%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.26 (d, 1H, $J = 8.6$ Hz, ArH), 8.08 (d, 2H, $J = 7.3$ Hz, ArH), 7.78 (d, 1H, $J = 5.5$ Hz, CHS), 7.73 (d, 1H, $J = 8.5$ Hz, ArH), 7.65 (d, 1H, $J = 5.5$ Hz, CHCHS)), 7.52 (t, 2H, $J = 7.0$ Hz, ArH), 7.45 (t, 1H, $J = 7.3$ Hz, ArH).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	156.4, 155.5, 139.8, 131.7, 131.0, 131.0, 129.0, 128.9, 127.4, 125.5, 116.5.
FTIR (neat):	3071 (s), 1906 (w), 1564 (s), 1544 (s), 1397 (s), 1280 (s), 1158 (s).
HRMS (ESI):	calcd for $C_{13}H_{10}NS [M+H]^+: 212.0528$, found: 212.0534.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.53 (UV, KMnO ₄).



5-(Thiophen-2-yl)thieno[3,2-b]pyridine (4x, Table 1):

Trifluoromethanesulfonic anhydride (69 μ L, 0.42 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide 11 (80 mg, 0.38 mmol, 1 equiv) and 2-chloropyridine (72 μ L, 0.76 mmol, 2.0 equiv) in dichloromethane (1.3 mL) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the enol ether 3b (231 mg, 0.764 mmol, 2.00 equiv) was added, and the resulting solution was allowed to warm to ambient temperature for 5 min. before the reaction vessel was placed into a microwave reactor and heated to 140 °C. After 20 min., the reaction vessel was removed from the microwave reactor and allowed to cool to ambient temperature before triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 5% EtOAc and 0.5% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine derivative 4x as a pale yellow solid (61 mg, 74%).

¹ H NMR (500 MHz, CDCl ₃ , 20°C):	8.18 (d, 1H, $J = 8.5$ Hz, ArH), 7.76 (d, 1H, $J = 5.5$ Hz, ArH), 7.68–7.65 (m, 2H, ArH), 7.59 (dd, 1H, $J = 5.5$, 0.6 Hz, ArH), 7.42 (dd, 1H, $J = 5.1$, 1.1 Hz, ArH), 7.14 (dd, 1H, $J = 5.0$, 3.7 Hz, ArH).
¹³ C NMR (125 MHz, CDCl ₃ , 20°C):	156.1, 150.6, 145.3, 131.5, 131.1, 131.0, 128.2, 127.6, 125.3, 125.1, 115.1.
FTIR (neat):	3098 (w), 2733 (w), 2453 (w), 1996 (w), 1564 (s), 1548 (s), 1437 (m), 1395 (s), 1159 (m).
HRMS (ESI):	calcd for $C_{11}H_8NS_2 [M+H]^+$: 218.0093, found: 218.0096.
TLC (20% EtOAc in hexanes), $R_{\rm f}$:	0.50 (UV, KMnO4).



3-(3.8-Diphenyl-2-(thiophen-2-yl)quinolin-4-yl)oxazolidin-2-one (4dd, equation 5):

Trifluoromethanesulfonic anhydride (52 μ L, 0.32 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1m** (80 mg, 0.29 mmol, 1 equiv) and 2-chloropyridine (33 μ L, 0.34 mmol, 1.2 equiv) in dichloromethane (950 μ L) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ynamide **2d** (59 mg, 0.32 mmol, 1.1 equiv) was added as a solid in one portion and the reaction flask was rapidly purged and sealed under an argon atmosphere. The resulting solution was allowed to warm to ambient temperature. After 1 h, triethylamine (500 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 40% EtOAc/1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the quinoline derivative **4dd** as a pale yellow solid (73 mg, 57%).

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.87–7.81 (m, 4H, ArH), 7.69–7.59 (m, 3H, ArH), 7.57–7.52 (m, 3H, ArH), 7.50–7.44(m, 2H, ArH), 7.28–7.26 (m, 2H, ArH), 6.74 (dd, 1H, $J = 5.1$, 3.9 Hz, ArH), 6.28 (dd, 1H, $J = 3.9$, 1.1 Hz, ArH), 4.51–4.45 (m, 1H, NCH ₂ CH ₂ O), 4.07–4.01 (m, 1H, NCH ₂ CH ₂ O), 3.90–3.84 (m, 1H, NCH ₂ CH ₂ O), 3.41–3.35 (m, 1H, NCH ₂ CH ₂ O).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	157.3, 151.4, 146.1, 145.9, 141.3, 140.8, 138.9, 135.9, 132.2, 131.5, 131.2, 129.9, 129.9, 129.6, 129.2, 129.1, 128.9, 127.9, 127.8, 127.6, 124.7, 121.9, 63.2, 47.2.
FTIR (neat) cm ⁻¹ :	3060 (w), 2916 (w), 2249 (w), 1753 (s), 1601 (w), 1579 (m), 1479 (s), 1421 (s), 1231 (m).
HRMS (EI):	calcd for $C_{28}H_{21}N_2O_2S [M+H]^+$: 449.1318, found: 449.1317.
TLC (50% EtOAc in hexanes), $R_{\rm f}$:	0.43 (UV, KMnO ₄)

Characterization of the product 9 (see text):

¹H NMR (500 MHz, CDCl₃, 20 °C) δ:

8.71 (d, 1H, J = 8.3 Hz, ArH), 8.59 (dd, 2H, J = 7.9, 4.6 Hz, ArH), 8.22 (d, 1H, J = 8.2 Hz, ArH), 7.89 (t, 1H, J = 7.5 Hz, ArH), 7.78–7. 65 (m, 4H, ArH), 7.58 (dd, 1H, J = 5.1, 1.0 Hz, ArH), 7.25 (dd, 1H, J = 5.2, 3.6 Hz, ArH).

¹³C NMR (125 MHz, CDCl₃, 20 °C) δ:

FTIR (neat) cm^{-1} :

HRMS (EI):

TLC (20% EtOAc in hexanes), R_{f} .

154.2, 143.9, 142.7, 133.8, 130.8, 130.4, 129.5, 129.1, 128.3, 128.1, 127.6, 127.6, 127.2, 124.9, 123.7, 122.5, 122.1.

3070 (m), 1956 (w), 1812 (w), 1734 (w), 1610 (m), 1577 (m), 1562 (s), 1519 (m), 1484 (s), 1458 (s), 1430 (s).

calcd for $C_{17}H_{12}NS [M+H]^+$: 262.0685, found: 262.0683.

0.51 (UV, KMnO₄).



(S)-1-(2-sec-butyl-3-phenyl-5,6,7,8-tetrahydroquinolin-4-yl)azetidin-2-one (4ee, equation 6):

Trifluoromethanesulfonic anhydride (15 μ L, 0.091 mmol, 1.1 equiv) was added via syringe over 1 min to a stirred mixture of amide **1n**⁹ (15 mg, 0.083 mmol, 1 equiv) and 2-chloropyridine (16 μ L, 0.17 mmol, 2.0 equiv) in dichloromethane (280 μ L) at -78 °C. After 5 min., the reaction mixture was placed in an ice-water bath and warmed to 0 °C, the ynamide **2h** (16 mg, 0.091 mmol, 1.1 equiv) was added as a solid in one portion and the reaction flask was rapidly purged and sealed under an argon atmosphere. After 1 h, triethylamine (100 μ L) was introduced to neutralize the trifluoromethanesulfonate salts. Dichloromethane (5 mL) was added to dilute the mixture and the layers were separated. The organic layer was washed with brine (2 mL), was dried over anhydrous sodium sulfate, and was filtered. The volatiles were removed under reduced pressure and the residue was purified by flash column chromatography (eluent: 20% EtOAc and 1% Et₃N in hexanes; SiO₂: 15 × 1.5 cm) on neutralized silica gel to give the pyridine derivative **4ee** as a pale yellow oil (19 mg, 69%). The enantiomeric excess of pyridine **4ee** was determined to be 94% ee by chiral HPLC analysis [Whelk-O1 (*S,S*); 0.5 mL/min; 1% 'PrOH in hexanes containing 0.2% Et₃N; *t*_R (minor) = 55.0 min., *t*_R (major) = 61.2 min].

¹ H NMR (500 MHz, CDCl ₃ , 20 °C) δ:	7.46–7.41 (m, 2H, ArH), 7.41–7.36 (m, 1H, ArH), 7.23–7.18 (m, 2H, ArH), 2.95 (t, 2H, $J = 6.3$ Hz, NCH ₂ CH ₂ CO), 2.85–2.62 (m, 7H, NCH ₂ CH ₂ CO, CH ₂ CH ₂ CH ₂ CH ₂ , CH ₃ CH ₂ CHCH ₃), 1.96–1.73 (m, 5H, CH ₂ CH ₂ CH ₂ CH ₂ CH ₂ , CH ₃ CH ₂ CHCH ₃), 1.50–1.40 (m, 1H, CH ₃ CH ₂ CHCH ₃), 1.11 (d, 3H, $J = 6.8$ Hz, CH ₃ CH ₂ CHCH ₃), 0.67 (t, 3H, $J = 7.5$ Hz, CH ₃ CH ₂ CHCH ₃).
¹³ C NMR (125 MHz, CDCl ₃ , 20 °C) δ:	166.0, 161.6, 158.1, 140.9, 136.9, 130.9, 129.9, 129.7, 128.6, 128.6, 127.7, 126.6, 41.0, 38.6, 36.5, 33.2, 29.6, 25.3, 22.9, 22.7, 21.0, 12.5.
FTIR (neat) cm ⁻¹ :	2959 (m), 2933 (m), 2870 (w), 1757 (s), 1602 (w), 1570 (w), 1547 (m), 1432 (m), 1405 (m), 1379 (m), 1192 (m).
HRMS (ESI):	calcd for $C_{22}H_{27}N_2O [M+H]^+$: 335.2118, found: 335.2115.

⁽⁹⁾ The enantiomeric excess of amide **1n** was determined to be 94% ee by chiral HPLC analysis [Whelk-O1 (*S,S*); 1.0 mL/min; 3% ^{*i*} PrOH in hexanes; t_R (major) = 28.6 min., t_R (minor) = 32.5 min].

Appendix A

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Spectra for Chapter I.

Solvent: CDC13 Ambient temperature INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 90.0 degrees Acq. time 3.200 sec Width 10000.0 Hz 16 repetitions OBSERVE H1, 500.2312691 MHz DATA PROCESSING FT size 131072 Total time 1 minute



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-136-



Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 576 repetitions 576 repetitions OBSERVE C13, 125.7832286 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 16 minutes



-137-



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-138-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7537710 MHz DATA PROCESSING FT size 65536 Total time 0 min, 52 sec



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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7537722 MHz DATA PROCESSING FT size 65536 Total time 1 min, 24 sec





Solvent: CDC13 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 136 repetitions OBSERVE C13, 125.7832337 MHz DECOUPLE H1, 500.2332753 MHz Power 44 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz Line broadening 0.3 Hz FT size 131072 Total time 3 minutes



-146-

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-147-

Pulse Sequence: s2pul

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Solvent: CDC13 Ambient temperature File: mh-I-200A INOVA-500 "zippy" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7537722 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec



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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec



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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec



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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature File: mh-I-192column3 INOVA-500 "zippy" PULSE SEQUENCE

PULSE SEQUENCE Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7537721 MHz DATA PROCESSING FT size 65536 Total time 0 min, 52 sec



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-159-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7537725 MHz DATA PROCESSING FT size 65536 Total time 0 min, 52 sec



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-162-

Solvent: CDC13 Ambient temperature INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 90.0 degrees Acq. time 3.200 sec Width 10000.0 Hz 16 repetitions OBSERVE H1, 500.2312700 MHz DATA PROCESSING FT size 131072 Total time 1 minute



-163-



Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 256 repetitions OBSERVE C13, 125.7832280 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALT2-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 7 minutes





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-165-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446543 MHz DATA PROCESSING FT size 65536 Total time 1 min, 24 sec



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Solvent: CDC13 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 120 repetitions OBSERVE C13, 125.7832337 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 3 minutes





-167-



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-168-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7537719 MHz DATA PROCESSING FT size 65536 Total time 1 min, 24 sec



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-170-



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-171-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446540 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec



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-173-



-174-

Solvent: CDC13 Ambient temperature INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 90.0 degrees Acq. time 3.200 sec Width 100000.0 Hz 16 repetitions OBSERVE H1, 500.2312702 MHz DATA PROCESSING FT size 131072 Total time 1 minute

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ب 8.97 Solvent: CDC13 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 160 repetitions OBSERVE C13, 125.7832268 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALT2-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 4 minutes







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-177-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec



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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446552 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec



-181-



Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.763 sec Pulse 65.4 degrees Acq. time 1.000 sec Width 37735.8 Hz With 37/35.8 Hz 3112 repetitions OBSERVE C13, 125.7832263 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated 77.486 .978 26 DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 91 minutes Me Me'' Мe 1g 30.357 ÷ 136.055 29 2 38.340 100.003 147.613 T 1 1 1 111 T 1.1 20 80 0 ppm 100 60 40 200 180 160 140 120

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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature File: mh-II-110fr12-14 INOVA-500 "z1ppy" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7537708 MHz DATA PROCESSING FT size 65536 Total time 1 min, 24 sec

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Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.763 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 8 repetitions OBSERVE C13, 125.7832383 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz 67.009 Line broadening 0.3 Hz 45.682 FT size 131072 Total time 1 minute 1h 29. 26 2 137.703 109.410 77.482 .230 __76.974 157.647 147.842 ويتبا استنقاه ويتعوا فتشاوينا لينا HILLING COLD 20 0 200 180 160 120 100 80 60 40 140

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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7537722 MHz DATA PROCESSING FT size 65536 Total time 1 min, 24 sec

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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446540 MHz DATA PROCESSING FT size 65536 Total time 1 min, 24 sec









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-194-



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-195-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446540 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec





Solvent: CDC13 Ambient temperature User: 1-14-87 File: mh-III-194carbon INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.763 sec Pulse 65.4 degrees Acq. time 1.736 sec Acq. time 1.756 Sec Width 37735.8 Hz 8 repetitions OBSERVE C13, 125.7831641 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on .413 1.15 653 29 29 29 29 29 29 WALTZ-16 modulated DATA PROCESSING 808 Ν Line broadening 0.3 Hz FT size 131072 Total time 1 minute `SiMe₃ 3j 5 -132.279 131.615 34 34 L34 -78.063 7.807 77.555 109.286 97.495 148.776 • with the ship him where whether and as the this should be an interiment in the second and the she have been been all with the second he is in a sid the loss of a filler one was bold on the light of the side of t LANDER MALINE MARKINE The second se ding kelala kelalan kerangkan kelalak kerangkan berangkan kelalak kerangkan kelalak kerangkan serika kerangkan s di ba er heites lö i de en le fontacion de la dicula 20 -20 ppm 80 60 40 200 160 140 120 100 240 220 180

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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446540 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec

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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec











-203-



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-204-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446540 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec



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-208-





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Solvent: CDC13 Ambient temperature INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 2.000 sec Pulse 90.0 degrees Acq. time 3.200 sec Width 10000.0 Hz 16 repetitions OBSERVE H1, 500.2312696 MHz DATA PROCESSING FT size 131072 Total time 1 minute



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Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky"

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INOVA-500 "rocky" PULSE SEQUENCE Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 1192 repetitions OBSERVE C13, 125.7832268 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB Continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 34 minutes 482 5 28.502 .555 9 -127.935 -125.426 -130.567 -129.148 'SiMe₃ 3m 137.698 99.188 -140.907 12.046 52.204 فعار بطحوران أورابنا خامر والوال الا والوازه وأر ومتابا ومعالية والمعالية الله ومرارك ومعارك المركز والمركز والم في التلأمات ويشارك الفريسة المتعادية الأوأ llan h. C. an ang terun sa sa kapang pala pang kala na samanan na kapatan kapan kapan kapan kapan sa ka kapan a at lead to be a second of the eri per have en her beste fan de selfe en de ser bewerkel kinde ser ser ser begre de ser her her her de ser al b 11 200 180 160 120 20 140 100 80 60 40 0 ppm



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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 90.0 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec





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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature File: mh-III-113 INOVA-500 "Zippy" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 90.0 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446549 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec





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Pulse Sequence: s2pul

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Solvent: CDC13 Ambient temperature Mercury-300 "mrhat"

Relax. delay 5.000 sec Pulse 34.1 degrees Acq. time 1.995 sec Width 4506.5 Hz 6 repetitions OBSERVE H1, 300.0986361 MHz DATA PROCESSING FT size 32768 Total time 2 min, 13 sec

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ې 9.10 Solvent: Benzene Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Pulse 65.4 degrees Acq. time 1.000 sec Width 37735.8 Hz 744 repetitions OBSERVE C13, 125.7831721 MHz DECOUPLE H1, 500.2333204 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 12 minutes

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-228-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446540 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec

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Pulse Sequence: s2pul Solvent: Benzene Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446837 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec

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Solvent: CDC13 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 40 repetitions OBSERVE C13, 125.7832309 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALT2-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 1 minute



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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec



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-237-

Pulse Sequence: s2pul

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Solvent: Benzene Ambient temperature INOVA-500 "bullwinkle"

PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions OBSERVE H1, 499.7446834 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec





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Pulse Sequence: s2pul

Solvent: CDC13 Ambient temperature File: mh-III-295 INOVA-500 "zippy"

PULSE SEQUENCE Relax. delay 1.000 sec Pulse 85.8 degrees Acq. time 3.277 sec Width 9998.8 Hz 16 repetitions DBSERVE H1, 499.7446549 MHZ DATA PROCESSING FT size 65536 Total time 1 min, 8 sec





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-242-



-243-

Appendix B

Spectra for Chapter II.

Pulse Sequence: s2pul Solvent: CDCl3 Ambient temperature File: mh-IV-244 INOVA-500 "zippy" PULSE SEQUENCE Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 4 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec



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Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.500 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz .OMe 40 repetitions 40 repetitions OBSERVE C13, 125.7832274 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING N 644 32.069 26.791 128. Line broadening 0.3 Hz FT size 131072 Total time 1 minutes 3a 128.420 6.974 482 30 26.374 -131.253 -130.104 125.440 -246-55.881 102.324 41.727 931 122.574 57 158.558 138.985 .045 147.174 m : **出出这些正式的问题。但是它是我的过去。**他们 and the second state of th hlid beat had all the folio in tail 80 60 160 140 120 100 20 200 180 40 0 ppm



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Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Puise 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 9 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec



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-254-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Puise 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 7 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

-255-





Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.763 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz Willin 37,35.8 m2 40 repetitions OBSERVE C13, 125.7832297 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 1 minutes







c:\pel_data\spectra\mhiv250.sp - mh-IV-250

-257-

Pulse Sequence: s2pul

Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 5 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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-260-

Pulse Sequence: s2pul

Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 5 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.500 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 72 repetitions OBSERVE Cl3, 125.7832274 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 2 minutes





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solvent: CDC13 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.763 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 88 repetitions OBSERVE C13, 125.7832286 MHz DECOUPLE H1, 500.2332753 MHZ POWEr 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 3 minutes





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c:\pel_data\spectra\mhiv273.sp - mh-IV-273

-266-

Pulse Sequence: s2pul

Solvent: CDC13 Ambient temperature File: mh-V-65clean INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 7 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.500 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz Width 3//35.8 Hz 152 repetitions OBSERVE C13, 125.7832297 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING 237 -26.681 553 137 32-Line broadening 0.3 Hz FT size 131072 Total time 5 minutes S3h 126.804 124.593 122.332 26.361 41.462 -268-133.295 .230 152.634 118.056 5.755 158.864 T.L. -r---Т T T T 200 180 160 140 120 100 80 60 40 20 0 ppm



c:\pel_data\spectra\mhv65.sp - mh-V-65

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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 7 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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Pulse Sequence: s2pul Solvent: CDCl3 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 9 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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Pulse Sequence: s2pul

Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 10 repetitions OBSERVE H1, 499.7446521 MHZ DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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Pulse Sequence: s2pul

Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 7 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec





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Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" INDVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.500 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 120 repetitions OBSERVE C13, 125.7832309 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB Continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz Line broadening 0.3 Hz FT size 131072 Total time 4 minutes

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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 7 repetitions OBSERVE H1, 499.7446521 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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Puise Sequence: s2pul Solvent: CDC13 Ambient temperature User: 1-14-87 File: mh-V-52Carbon INOVA-500 "zippy" INOVA-500 "zippy" PULSE SEQUENCE Relax. delay 0.500 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 256 repetitions OBSERVE C13, 125.7832309 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 9 min, 36 sec

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c:\pel_data\spectra\mhiv285.001 - mh-IV-285

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c:\pel_data\spectra\mhv12.001 - mhv12

-303-

Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 11 repetitions OBSERVE H1, 499.7446521 MHZ DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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-312-

Data File C:\HPCHEM\2\DATA\MHV169.D



-313-



-314-

Solvent: CDC13 Ambient temperature User: 1-14-87 INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.500 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 8 repetitions OBSERVE C13, 125.7832337 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 1 minute

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Data File C:\HPCHEM\2\DATA\MHV171.D

Racemis; Chiralcel OD 99% hex 1% IPA 0.5ml/min





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Appendix C

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Spectra for Chapter III.












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Pulse Sequences explif Solventi CDC13 Ambient temperature File: mi-VI-212 INOVA-500 "zippy" PULSE SEQUENCE Relax, delay b.(00 edu Pulse 00.0 deu Acq. time 3.001 sec Width 10504.2 Hz 7 repetitions OBSERVE H1, 499.7417206 MHz DATA PROCESSING FT size 262144 Total time 2 min, 8 sec Ph OSiⁱPr₃ ⁿḃu 4c -327-

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Pulse Sequence: s2pul

Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 6 repetitions OBSERVE H1, 499.7417206 MHz DATA PROCESSING FT size 262144 Total time 2 min, 8 sec



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Solvent: CDCls Ambient temperature User: 1-14-87 INDVA-SDD "rocky" INDVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.750 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 320 repetitions OBSERVE C13, 125.7832274 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 FT size 131072 Total time 13 minutes

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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature File: mh-VI-55fr7-18 INOVA-500 "Zippy" PULSE SEQUENCE Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 8 repetitions OBSERVE H1, 499.7417206 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec



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Pulse Sequence: s2pul

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Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec. Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 10 repetitions OBSERVE H1, 499.7417206 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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Pulse Sequence: s2pul Solvent: CDQ13 Temp. 22.0 0 / 205.1 K File: mh-VI-A3 INUVA-B00 "z1µµy" Pulse SEQUENCE Relax. delay 5.000 sec Pulse 88.0 degrees Acq. time 3.001 sec Width 10004.2 Hz 16 repetitions OBSERVE H1, 404.7417200 MHz DATA PRODESING FT size 202144 Total time 2 min, 8 sec







-343-





Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 5 repetitions OBSERVE H1, 499.7417206 MHz DATA PROCESSING FT size 131072 Total time 2 min, 8 sec

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								·			•	NUC1 P1 PL1 SF01	CHANNEL fl ======= 13C 8.75 usec -3.00 dB 100.6228298 MHz
												CPDPRG2 NUC2 PCPD2 PL2 PL12 PL13 SF02	CHANNEL f2 ====== waltz16 1H 90,00 usec -1,00 dB 14,52 dB 18,00 dB 400,1316005 MHz
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Solvent: CDC135 Ambient temperature User: 1-14-87 INOVA-500 "rocky" INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.200 sec Pulse 69.0 degrees Acq. time 1.736 sec Width 37735.8 Hz 816 repetitions OBSERVE C13, 125.7832268 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 26 minutes ____26.786 306 33.112 OMe L C OMe 6 .812 917 °Hx' ŝ ŝ 4p .486 9 -354-97.508 .673 .843 115.401 187 99. .185 " 47. 167.786 131 135.336 .944 127 with a light to be a start of the 1. Hardel public (Herbitan)) Erne and Million (Herbit ull the line of th all detailed Hubie Hanneld 17.17 20 ppm 60 n 80 40 180 120 100 200 140 160





STANDARD CARBON PARAMETERS

Pulse Sequence: s2pul Solvent: CDC13 Amblent temperature User: 1-14-87 INOVA-500 "bullwinkle"

Relaw. delay 0.050 eec Pulse 36.7 degrees Acq. time 2.000 sec Width 31397.2 Hz 192 repetitions OBSERVE C13, 125.0001306 MHz DECOUPLE H1, 400.7442104 MHz Power 34 dH continuously bu WALTZ-18 wodulated DATA PROCESSINO Line broadening 1.0 Hz FT size 131072 Total time 571 hr, 54 min, 55 sec





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Pulse Seguences expul

Solvent: 00019 Ambient temperature INOVA-500 "builtwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 11 repetitions OBSERVE H1, 499.7417206 MHz DATA PROCESSING FT size 262144 Total time 2 min, 8 sec



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Pulse Sequence: s2pul

Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle"

Relax. delay 5.000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 6 repetitions OBSERVE H1, 499.7417206 MHz DATA PROCESSING FT size 262144 Total time 2 min, N mac



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Pulse Sequence: s2pul Solvent: CDC13 Ambient temperature INOVA-500 "bullwinkle" PULSE SEQUENCE Pulse 85.8 degrees Acq. time 3.277 sec Width 9988.8 Hz 16 repetitions OBSERVE H1, 499.7537710 MHz DATA PROCESSING FT size 65536 Total time 0 min, 52 sec

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Solvent: CDC13 Temp. 20.0 C / 293.1 K User: 1-14-87 INGVA-500 "rocky" INDVA-SUN "FOCKY" PULSE SEQUENCE Relax. delay 0.703 wec Pulse 65.4 degroes Acq. time 1.730 wec Width 37735.8 Hz 100 repatitions OBSERVE, 013, 120,783232 MHz DEOUPLE H1, BD0.2332753 MHz Power 44 HB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 FT size 131072 Total time 4 minutes Ph 4u -366and direction distributions أتتلز بدأكل يسترياه والاتلام علايا STATISTICS AND A state of the environmental and the linn ; Kille H Aring Bit & mass (1/16) 7777 -1--1--1--1--1 20 0 ppm 60 80 40 200 180 140 120 100 160



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Pulse Sequence: s2pul

Solvent: CDC13 Ambient temperature File: mh-III-113 INOVA-500 "zippy"

PULSE SEQUENCE Relax. delay 1.000 sec Pulse 90.0 degrees Acq. time 3.277 sec Width 9998.6 Hz 16 repetitions OBSERVE H1, 499.7446549 MHz DATA PROCESSING FT size 65536 Total time 1 min, 8 sec



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Solvent: CDCl3 Ambient temperature User: 1-14-87 INOVA-500 "rocky" INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.763 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 16 repetitions OBSERVE C13, 125.7832458 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated OATA PRODENTING Line trondenting 0.3 Hz FT size 131072 Total time 1 minute <u>128.978</u> 128.910 127.385 .385 0 Ph 125.541 130.988 4w 116.514 131.029 -372-139.786 ___131.651 7.482 974 230 60 40 20 0 ppm 120 100 200 140 80 180 160



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Solvent: CDC13 Ambient temperature User: 1-14~87 INOVA-500 "rocky" INOVA-500 "rocky" PULSE SEQUENCE Relax. delay 0.763 sec Pulse 65.4 degrees Acq. time 1.736 sec Width 37735.8 Hz 56 repetitions OBSERVE C13, 125.7832332 MHz DECOUPLE H1, 500.2332753 MHz Power 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Line broadening 0.3 Hz FT size 131072 Total time 2 minutes



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Pulse Sequence: s2pul Solvent: CDC13 Temp. 22.0 C / 295.1 K File: mh-VII-12 INOVA-500 "zippy" PULSE SEQUENCE Relax. delay 5,000 sec Pulse 89.0 degrees Acq. time 3.001 sec Width 10504.2 Hz 8 repetitions OBSERVE H1, 499.7417206 MHz DATA PROCESSING FT size 262144 Total time 2 min, 8 sec

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Solvent: CDC13 Ambient temperature User: 1-14-87 INOVA-500 "Focky" PULSE SEQUENCE Relax. delay 0.763 sec Pulse 69.0 degrees Acq. time 1.736 sec Width 37735.8 Hz 192 repetitions OBSERVE C13, 125.7832337 MHz DECOUPLE H1, 500.2332753 MHz DECOUPLE H1, 500.2332753 MHz Dewer 37 dB continuously on WALTZ-16 modulated DATA PROCESSING Resol. enhancement -0.0 Hz FT size 131072 Total time 8 minutem

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Matthew Dennis Hill

Curriculum Vitae

Education	
	Massachusetts Institute of Technology, Cambridge, MA. Ph D. Organic Chemistry (expected 2008)
	Thesis title: "Direct Synthesis of Pyridine and Pyrimidine Derivatives."
	Ohio University, Athens, OH. B.S. Biochemistry; Molecular Biology; B.S.C. Legal Communication
Research Experience	
2003-present	Graduate Research Associate, Massachusetts Institute of Technology Professor Mohammad Moyassaghi, Advisor.
	• Development of efficient methodology for azaheterocycle synthesis.
2003	Undergraduate Researcher, Massachusetts Institute of Technology Professor Alice Ting Advisor
	 Investigation toward incorporation of biotin analogues into biotin transferase.
2001-2003	Undergraduate Researcher, Ohio University
	 Professor Mark McMills, Advisor. Development of new synthetic methodologies for Phorbol analog synthesis
	Development of new synthetic methodologies for Thoroof analog synthesis.
2002	NSF REU Undergraduate Researcher, Columbia University Professor Koji Nakanishi, Advisor
	• Investigation of causes for age-related macular degeneration.
Teaching Experience	
2003-2007	Three semesters teaching assistantship (MIT): two organic chemistry courses (head TA for Professors M. Movassaghi/S. Buchwald) and one laboratory course.
2001-2003	Five quarters peer mentorship (Ohio University): three organic chemistry courses (Professors M. McMills/J. Butcher).
Academic Honors and	d Awards
2007	MIT Wyeth Scholar, Amgen Summer Fellowship (MIT), Morse Travel Grant (MIT)
2003	Rhodes Scholarship State Finalist (Tennessee), Upper Ohio Valley of the American Chemical Society La Vallee Award.
2002	Barry M. Goldwater Scholarship, NSF REU Fellowship at Columbia University, Jeanette Grasselli-Brown Research Fellowship (Ohio University), Omicron Delta Kappa Torch Scholarship (Ohio University), Lela Ewers Science Scholarship (Ohio University), Chemistry Scholarship (Ohio University), Three-time recipient of the Deans Scholarship (Ohio University), Two-time recipient of the Hiram Roy Wilson Scholarship (Ohio University), Two-time recipient of the Jesse Day Undergraduate Chemistry Award (Ohio University).

2001	Lubrizol Foundation Chemistry Scholarship (Ohio University), Sandra Lou McKay Memorial Scholarship (Ohio University), Paul C. and Beth K. Stocker Scholarship (Ohio University), Upper Ohio Valley Section of the American Chemical Society Sophomore Award.
2000	Ohio University Foundation Scholarship, CRC Handbook Award for General Chemistry (Ohio University), Provost Scholarship (Ohio University).

Professional Activities

2007-present	American Chemical Society: member of the Organic Division
2006-present	MIT Chemistry Outreach affiliate (program to stimulate interest in chemistry).
2001-2003	Alpha Chi Sigma: vice president (professional chemistry organization).

Publications

"Observations On the Use of Microwave Irradiation in Azaheterocycle Synthesis," Hill, M. D.; Movassaghi, M. Tetrahedron Lett. 2008, in press.

"New Strategies for Synthesis of Pyrimidine Derivatives," Hill, M. D.; Movassaghi, M. Chem. Eur. J. 2008, in press.

"Synthesis of Substituted Pyrimidines and Quinazolines," Hill, M. D.; Movassaghi, M. Synthesis 2008, 823-827.

"Single-Step Synthesis of Alkynyl Imines from N-Vinyl and N-Aryl Amides. Synthesis of N-Phenyl-2-phenyl-4-trimethylsilyl-1-azabut-1-en-3-yne," Movassaghi, M.; Hill, M. D. Org. Synth. 2008, 85, 88–95.

"Direct Synthesis of Pyridine Derivatives," Movassaghi, M.; Hill, M. D.; Ahmad, O. K. J. Am. Chem. Soc. 2007, 129, 10096–10097.

"Single-Step Synthesis of Pyrimidines by Direct Condensation of Amides and Nitriles," Movassaghi, M.; Hill, M. D. *Nat. Protoc.* **2007**, *2*, 2018–2023.

"Synthesis of Substituted Pyridines and Quinolines," Hill, M. D.; Movassaghi, M. Synthesis 2007, 1115–1119.

"Single-step Synthesis of Pyrimidine Derivatives," Movassaghi, M.; Hill, M. D. J. Am. Chem. Soc. 2006, 128, 14254–14255.

"Synthesis of Substituted Pyridine Derivatives via the Ruthenium-Catalyzed Cycloisomerization of 3-Azadienynes," Movassaghi, M.; Hill, M. D. J. Am. Chem. Soc. 2006, 128, 4592–4593.

Presentations

"New Methodologies for the Synthesis of Azaheterocycles."

- Oral presentation: American Chemical Society National Meeting (Boston, MA, August 2007).
- Poster: Gordon Research Conference: Heterocyclic Compounds (Newport, RI, June2007).
- Oral presentation: Graduate Research Symposium: MIT (Cambridge, MA, January 2007).

"Spectroscopic Determination of Changes in Vesicle Permeability Due to A2E." Oral presentation: REU Summer Seminar: Columbia University (New York, NY, August 2002).