



THE UNIVERSITY *of* EDINBURGH

Edinburgh Research Explorer

Electron capture dissociation and drift tube ion mobility-mass spectrometry coupled with site directed mutations provide insights into the conformational diversity of a metamorphic protein

Citation for published version:

Harvey, SR, Porrini, M, Tyler, RC, MacPhee, CE, Volkman, BF & Barran, PE 2015, 'Electron capture dissociation and drift tube ion mobility-mass spectrometry coupled with site directed mutations provide insights into the conformational diversity of a metamorphic protein' *Physical chemistry chemical physics*, vol. 17, no. 16, pp. 10538-10550. DOI: 10.1039/c4cp05136j

Digital Object Identifier (DOI):

[10.1039/c4cp05136j](https://doi.org/10.1039/c4cp05136j)

Link:

[Link to publication record in Edinburgh Research Explorer](#)

Document Version:

Peer reviewed version

Published In:

Physical chemistry chemical physics

General rights

Copyright for the publications made accessible via the Edinburgh Research Explorer is retained by the author(s) and / or other copyright owners and it is a condition of accessing these publications that users recognise and abide by the legal requirements associated with these rights.

Take down policy

The University of Edinburgh has made every reasonable effort to ensure that Edinburgh Research Explorer content complies with UK legislation. If you believe that the public display of this file breaches copyright please contact openaccess@ed.ac.uk providing details, and we will remove access to the work immediately and investigate your claim.



PCCP

Accepted Manuscript



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

1 **Electron capture dissociation and drift tube ion mobility-mass spectrometry coupled**
2 **with site directed mutations provide insights into the conformational diversity of a**
3 **metamorphic protein**

4 Sophie R. Harvey¹, Massimiliano Porrini², Robert C. Tyler³, Cait E. MacPhee⁴, Brian F.
5 Volkman³ and Perdita E. Barran^{5*}

1 School of Chemistry, University of Edinburgh, West Mains Road, Edinburgh, EH9 3JJ,
UK.

2 Institut Européen de Chimie et Biologie (IECB), U869 ARNA – Inserm Chimie et Biologie
des Membranes et des Nano-objets (CBMN), 33607 Pessac Cedex, France

3 Department of Biochemistry, Medical College of Wisconsin, Milwaukee, WI 53226, USA.

4 School of Physics and Astronomy, University of Edinburgh, West Mains Road, Edinburgh,
EH9 3JJ, UK.

5 School of Chemistry, Manchester Institute of Biotechnology, University of Manchester,
Manchester, M1 7DN, UK.

Corresponding Author

*Perdita E. Barran (perdita.barran@manchester.ac.uk)

Running Title: mass spectrometry delineates a metamorphic protein

1 Abstract

2 Ion mobility mass spectrometry can be combined with data from top down sequencing to
3 discern adopted conformations of proteins in the absence of solvent. This multi-technique
4 approach has particular applicability for conformationally dynamic systems. Previously, we
5 demonstrated the use of drift tube ion mobility-mass spectrometry (DT IM-MS) and electron
6 capture dissociation (ECD) to study the metamorphic protein lymphotactin (Ltn). Ltn exists
7 in equilibrium between distinct monomeric (Ltn10) and dimeric (Ltn40) folds, both of which
8 can be preserved and probed in the gas-phase. Here, we further test this mass spectrometric
9 framework, by examining two site directed mutants of Ltn, designed to stabilise either
10 distinct fold in solution, in addition to a truncated form consisting of a minimum model of
11 structure for Ltn10. The truncated mutant has similar collision cross sections to the wild type
12 (WT) for low charge states, and is resistant to ECD fragmentation. The monomer mutant
13 (CC3) presents in similar conformational families as observed previously for the WT Ltn
14 monomer. As with the WT, the CC3 mutant is resistant to ECD fragmentation at low charge
15 states. The dimer mutant W55D is found here to exist as both a monomer and dimer. As a
16 monomer W55D exhibits similar behaviour to the WT, but as a dimer presents a much larger
17 charge state and collision cross section range than the WT dimer, suggesting a smaller
18 interaction interface. In addition, ECD on the W55D mutant yields greater fragmentation than
19 for the WT, suggesting a less stable β -sheet core. The results highlight the power of MS to
20 provide insight into dynamic proteins, providing further information on each distinct fold of
21 Ltn. In addition we observe differences in the fold stability following single or double point
22 mutations. This approach, therefore, has potential to be a useful tool to screen for the
23 structural effects of mutagenesis, even when sample is limited.

24

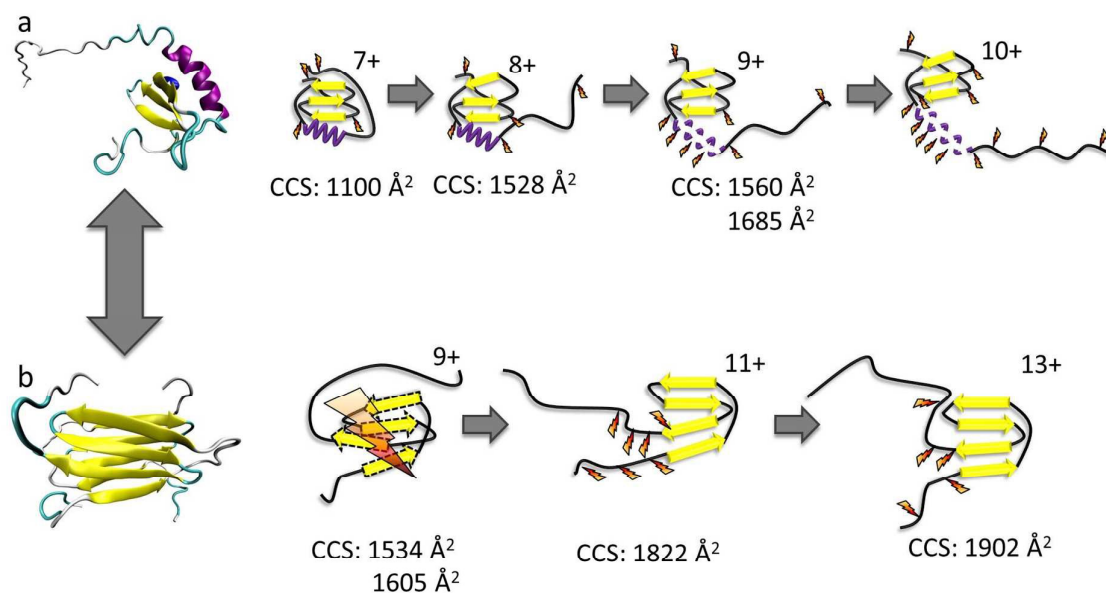
1 Introduction

2 The development of ‘soft’ ionisation techniques such as electrospray (ESI)^{1, 2} and nano-
3 electrospray ionisation (n-ESI)^{3, 4} revolutionised biological mass spectrometry (MS). It is
4 now accepted as a powerful tool in the structural analysis of proteins and protein complexes
5 that can probe solution-phase topologies and even *in vivo* active structures⁵⁻¹⁰. In the early
6 1960s the first reports of combining the technique of ion mobility spectroscopy (IMS) to
7 mass spectrometry were published¹¹⁻¹³. This hybrid technique known as ion mobility-mass
8 spectrometry (IM-MS) provides an extra dimension of information over mass spectrometry
9 alone, separating ions based not only on their mass-to-charge ratio but also on their size and
10 shape, known as their rotationally averaged collision cross section (CCS). IM-MS has been
11 successfully applied to study the conformations adopted by peptides and proteins in the gas-
12 phase and, hence, is gaining importance as a biophysical tool¹⁴⁻¹⁷. Furthermore, the CCS
13 determined from experiment can be compared to theoretical CCS, obtained from coordinates
14 of solved structures (either from NMR or X-ray crystallography) or from molecular
15 modelling, providing confirmatory information on the conformations adopted by the gas-
16 phase protein ions¹⁸⁻²¹.

17 Recently studies which compare the results from IM-MS experiments, on the global fold of a
18 protein, to amino acid level detail obtained from electron capture dissociation (ECD)
19 experiments have been presented²²⁻²⁴. ECD is a fast non-ergodic fragmentation process in
20 which bond dissociation following activation occurs much faster than typical bond vibration
21 and hence ECD is thought not to perturb the higher order structure of proteins enabling
22 protein structural studies to be performed^{25, 26}. ECD has been shown previously to
23 preferentially cleave the backbone of proteins without disruption of weaker non-covalent
24 interactions, such as hydrogen bonding networks present in a protein:peptide complex,²⁷ and

1 has become an influential and sophisticated technique to study protein fold in the gas-phase at
 2 an amino acid level²⁸⁻³¹.

3 Here we use the combination of DT IM-MS and ECD to investigate the two distinct folds of
 4 the chemokine Ltn. Chemokines are small secreted signalling proteins involved in immune
 5 response and are divided into four subclasses depending on the number and positioning of
 6 their conserved cysteine residues³². Lymphotactin is a unique chemokine, containing only a
 7 single disulfide bond, as opposed to the two normally found in chemokines, and is the
 8 defining member of the C subclass of chemokines³³. Another unusual feature of Ltn is that it
 9 is a metamorphic protein³⁴, existing in two distinct conformations in equilibrium (Figure 1) in
 10 a ligand-free state, each of which performs separate functions *in vivo* and are, therefore,
 11 essential for protein function³⁵.



12

13 Figure 1: The conformational equilibrium of lymphotactin; a) monomeric, Ltn10 and b) dimeric, Ltn40. Cartoon
 14 representations of Ltn10 and Ltn40 unfolding pathways as determined through DT IM-MS and ECD, with
 15 yellow arrows representing β -sheet regions and purple spirals representing α -helical regions. For Ltn40 a single
 16 monomeric unit is shown for figure clarity. Favourable sites of ECD fragmentation are shown by orange
 17 'lightening' bolts, for Ltn40 where fragments only map to a monomer unit with no secondary structure, the
 18 extensive fragmentation is depicted by a large lightning bolt and dashed lines represent a lack of defined
 19 structural elements.

1 Ltn exists in equilibrium between a monomeric, conserved chemokine fold, known as Ltn10
2 and a distinctive dimeric fold, known as Ltn40^{35, 36}. The interconversion between these two
3 folds involves a complete restructuring of the core residues³⁷. We have studied both
4 conformations of wild type (WT) Ltn using gas-phase techniques, which exclude
5 conformational constriction due to buffers or additives and enable the intrinsic stability of
6 each structural element in both Ltn10 and Ltn40 to be assessed. Our earlier work indicated
7 that the Ltn10 fold could be transferred and retained in a solvent-free environment, with the
8 intrinsically disordered (ID) tail being associated with the structural core at the low charge
9 states of the protein and unfolding as the charge state increases²⁴. We hypothesized, and this
10 was supported by ECD data, that the tail unfolding is then followed by unravelling of the α -
11 helix, with little perturbation of the β -sheet region, as summarised in Figure 1a. For Ltn40 a
12 structural collapse is observed for the lowest charge state, leading to a small CCS and
13 extensive backbone dissociation during ECD. Whilst the higher charge states of the dimer
14 show greater stability suggesting the structural core remains stable and unperturbed in these
15 conformations, (Figure 1b).

16 The studies presented here utilise DT IM-MS and ECD to examine the conformational
17 equilibrium of this metamorphic protein further; through the study of specific mutants
18 designed to mimic either the Ltn10 or Ltn40 fold, whilst limiting conformational
19 interconversion. The results provide detailed insight into the unfolding of each distinct Ltn
20 conformation, in conjunction with studying the effect of specific site mutations on both folds
21 and allowing any subsequent increase or decrease in fold stability to be identified.

22 **Results and discussion**

23 **WT 1-72**

1 Initial studies focused on a truncated Ltn mutant, which contains the structural core of the
2 protein but does not contain the last 21 amino acids which form an intrinsically disordered
3 tail in the WT protein. This construct enables us to study a minimum structural model for
4 Ltn10, which contains the two main types of protein secondary structural elements; α -helix
5 and β -sheets.

6 We have established conditions which enable both WT monomer (Ltn10) and WT dimer
7 (Ltn40) to be preserved, which we utilise for all mutants studied here²⁴. The mass spectra of
8 WT 1-72 shows that the protein presents mainly as a monomer (M_{1-72}) and is capable of
9 accepting between four and eight protons (Figure 2a). The majority of the intensity in the WT
10 1-72 mass spectra is due to the $[M_{1-72}+5H]^{5+}$ and $[M_{1-72}+6H]^{6+}$ species, suggesting that these
11 two charge states are particularly conformationally stable. A WT 1-72 dimer (D_{1-72}) is also
12 observed under these conditions but at much lower intensity than the monomer, presenting
13 over the charge state range $[D_{1-72}+8H]^{8+}$ to $[D_{1-72}+12H]^{12+}$ (Figure 2a) it is not discussed any
14 further.

15 Previously, the CCS adopted by WT 1-72 were determined using DT IM-MS²⁴, Figure 2b and
16 Supplementary Information Table S1. In summary, as the charge state increases so does the
17 CCS, this is attributed to more extended solution-phase conformations being more solvent
18 and charge accessible, hence accepting larger numbers of charges, and consequently having a
19 larger CCS. Coulombic repulsion between charged groups can also have an effect in the gas-
20 phase and can cause an increase in CCS with increasing charge. Over the charge state range
21 $[M_{1-72}+4H]^{4+}$ to $[M_{1-72}+6H]^{6+}$ the CCS increases slightly (11%), which suggests that these
22 species are present in a stable, compact, conformationally similar family in which the
23 addition of charge does not significantly influence the conformations adopted. The
24 experimental CCS observed are all smaller by $\sim 15\%$ than the theoretical value (1176 \AA^2)

1 obtained from NMR coordinates of the full length protein with residues 73-93 removed (PDB
2 1J9O). These smaller values indicate a compaction of structure further than that achieved by
3 *in vacuo* minimising the NMR structure; additionally we anticipate a tethering of the loose N-
4 terminus to the protein core.

5 Two distinct conformations are observed for the $[M_{1-72}+7H]^{7+}$ species, the more compact
6 conformation appears similar in CCS to those observed for $[M_{1-72}+4H]^{4+}$ to $[M_{1-72}+6H]^{6+}$
7 suggesting this is probably of a similar, compact, possibly folded structure. The second larger
8 conformation indicates WT 1-72 is more unfolded at this charge state. A further jump in CCS
9 is observed moving to the $[M_{1-72}+8H]^{8+}$ species.

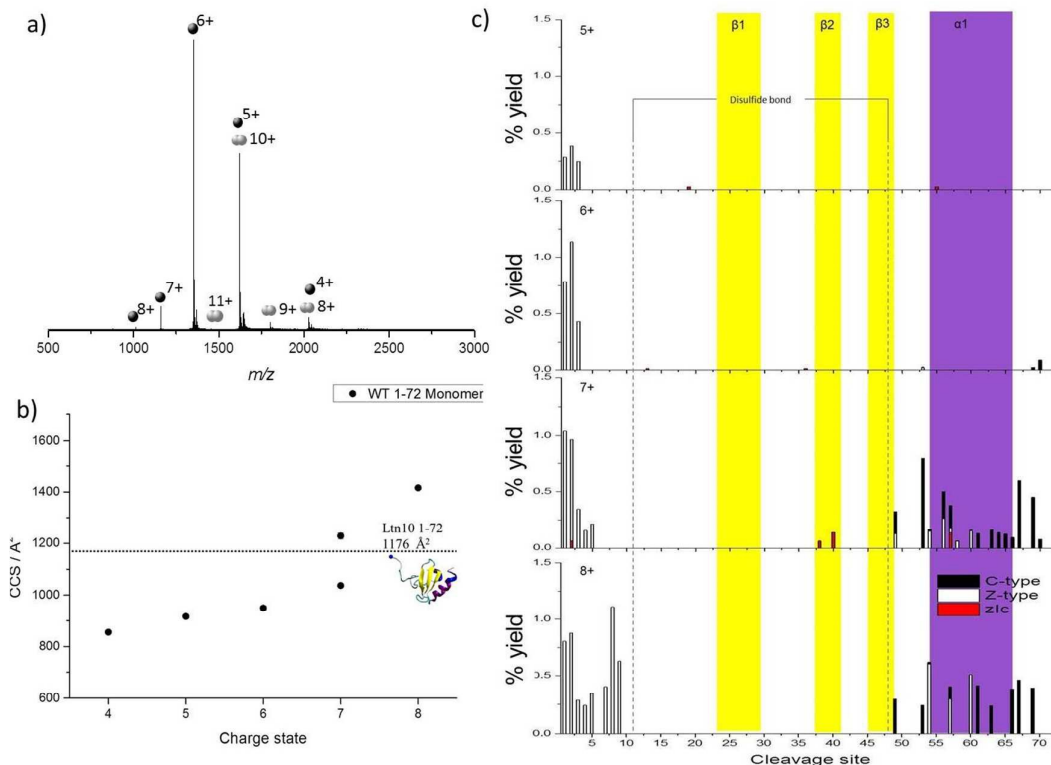
10 The charge states $[M_{1-72}+5H]^{5+}$ to $[M_{1-72}+8H]^{8+}$ were observed at high enough intensity for
11 effective trapping and sufficient fragmentation efficiency for analysis by ECD fragmentation
12 (Supplementary Information Figures S1 and S2). Following ECD experiments, fragmentation
13 maps were plotted in which the percentage yield of all fragment ions observed are plotted
14 against cleavage site, considering c and z-type fragments as well as zIc fragments.
15 Fragmentation yields were calculated considering remaining precursor, charge reduced
16 species and fragment ions. zIc fragments are internal fragments, formed by multiple
17 cleavages and which have a z-type N-terminus and c-type C-terminus²⁴, where “I” refers to
18 the fact that they are internal. Internal fragments have been attributed previously to charge
19 remote fragmentation³⁸ and are thought to be particularly prevalent in sequences containing
20 basic residues, and hence are likely to play an important role in the fragmentation of highly
21 basic proteins such as chemokines. Furthermore, internal fragments were observed previously
22 at significant intensity for WT lymphotactin²⁴.

23 As charge state increases, increased fragmentation is observed attributed to a number of
24 factors; mainly, ECD fragmentation is more efficient at higher charge states³⁹, furthermore,

1 at higher charge states protein ions are likely to be more extended than their lower charged
2 counter parts and hence there are a greater number of sites available for cleavage and
3 dissociation. By mapping the fragmentation observed onto the secondary structural elements
4 of the protein a detailed picture of how the protein restructures and, hence, the stability of
5 each of the different structural elements, can be obtained as a function of charge. All
6 identified c, z and zIc fragments for $[M_{1-72}+5H]^{5+}$ to $[M_{1-72}+8H]^{8+}$ are therefore shown in
7 Figure 2c, as a function of cleavage site onto which the regions involved in formation of each
8 β -strand and the α -helix are mapped.

9 For $[M_{1-72}+5H]^{5+}$ and $[M_{1-72}+6H]^{6+}$, limited fragmentation and subsequent dissociation is
10 observed, resulting in the identification of several c, z and zIc fragments. The low intensity is
11 most likely due to the compact nature of these ions coupled with their low charge states and
12 hence low ECD fragmentation efficiency. As charge state increases both the extent and the
13 intensity of fragmentation observed increases in the N-terminal region suggesting this region
14 is beginning to unfold from the structural core of the protein. In addition at the $[M_{1-72}+7H]^{7+}$
15 and $[M_{1-72}+8H]^{8+}$ charge states we observe an increase in the fragmentation in the α -helical
16 region (highlighted in purple in Figure 2c), for both c and z type fragments in addition to zIc
17 fragments, suggesting this region is beginning to unravel from the structural core of the
18 protein, losing any non-covalent stabilising interactions. This observation is consistent with
19 our DT IM-MS observations in which we see a significant increase in CCS for the $[M_{1-}$
20 $_{72}+7H]^{7+}$ and $[M_{1-72}+8H]^{8+}$ species. For $[M_{1-72}+8H]^{8+}$ the extent of fragmentation occurring in
21 the α -helix decreases in comparison to $[M_{1-72}+7H]^{7+}$, this is attributed to the lower signal
22 intensity of the $[M_{1-72}+8H]^{8+}$ species, leading to fewer ions being trapped and subjected to
23 ECD and highlights the difficulties of performing such ECD studies on multiple charge
24 states. At all charge states very little fragmentation is observed in the β -sheet core
25 (highlighted in yellow in Figure 2c) suggesting this region retains its stabilising interactions

- 1 even after desolvation and ionisation and is more stable than the α -helical region, in part
- 2 attributed to the disulfide bond which would further stabilise this region.



3

4 **Figure 2: WT 1-72** a) Mass spectra obtained on the DT IM-MS instrument for 100 μ M WT 1-72 in 20 mM
 5 ammonium acetate (AmAc.) b) Average experimental CCS for M_{1-72} obtained from three different day repeats,
 6 error bars are calculated from the standard deviation and generally fall within symbol size. Dashed line
 7 represents theoretical CCS obtained from the clipped NMR structure (PDB 1J9O) removing residues 73-93. c)
 8 Percentage yields calculated for zlc, c and z-type⁴⁰ fragments as a function of cleavage site for M_{1-72} over the
 9 charge states $[M_{1-72}+5H]^{5+}$ to $[M_{1-72}+8H]^{8+}$.

10 These observations are similar to the trends observed for the WT protein in the Ltn10
 11 conformation (Figure 1); one notable difference is that in WT Ltn10 significant unravelling
 12 of the α -helix is not observed until the $[M_{WT}+9H]^{9+}$ species whereas in WT 1-72 this
 13 unravelling is observed from the $[M_{1-72}+7H]^{7+}$ species. At lower charge states the
 14 intrinsically disordered tail is believed to be associated with the structural core of the WT
 15 protein and proposed to uncouple before the α -helix unravels. It is therefore plausible that the
 16 intrinsically disordered tail confers structural stability and protection to the WT Ltn10 fold

1 and the absence of the tail in WT 1-72 causes this construct to unfold at a lower charge state.
2 An additional explanation could be that in the WT 1-72, due to its smaller size, the charges
3 would likely reside closer to each other and therefore Columbic repulsion of charges could
4 induce unfolding at lower charge states than for the WT protein.

5 **Probing the Ltn10 fold through specific site mutations**

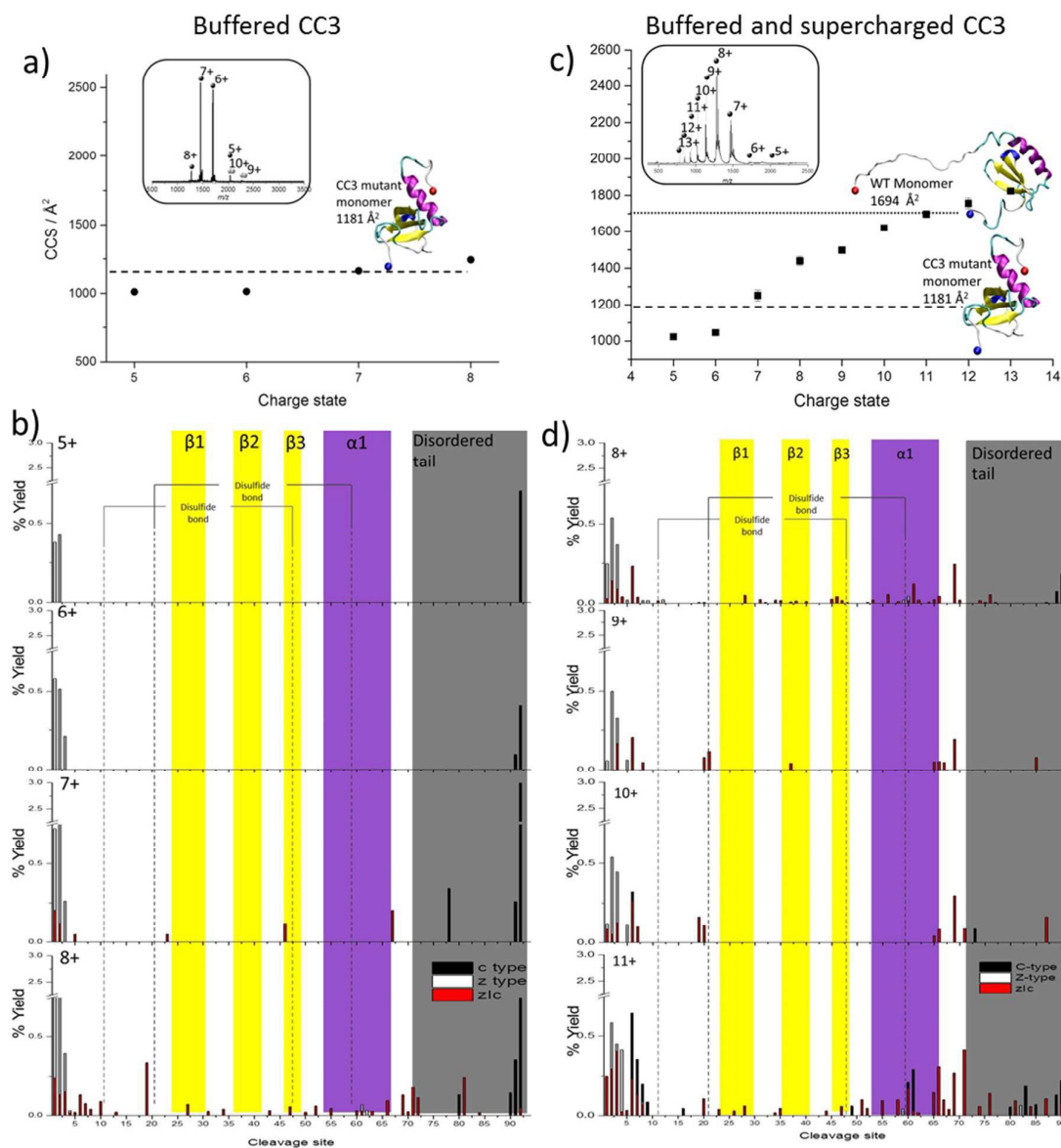
6 In order to further probe the Ltn10 fold, we next considered a mutant, known as CC3,
7 designed to favour the monomeric form of Ltn, Ltn10⁴¹. The CC3 mutant includes an extra
8 disulfide bond based on the third disulfide in the CC chemokine, HCC-2. CC3 was
9 engineered by introducing two extra cysteines in positions 21 and 59, with the aim that the
10 addition of an extra disulfide bond, based on the conserved chemokine fold, would
11 structurally constrain the protein and stop structural rearrangement to form the dimer fold⁴¹.
12 For this mutant very little dimer signal is observed, as expected, (Figure 3a insert and
13 Supplementary information Figure S3). The CC3 monomer presents over a range of charge
14 states from $[M_{CC3+5H}]^{5+}$ to $[M_{CC3+8H}]^{8+}$, with the species which accept either six or seven
15 protons being the most favourable and hence most intense. A small proportion of CC3 dimer
16 is observed but at incredibly low intensity highlighting that for this mutant formation of the
17 dimer is not as favourable as for the WT type protein, hence our analysis will focus on the
18 monomeric CC3.

19 Following DT IM-MS on the CC3 construct we find that it presents over a narrow range of
20 CCSs over the charge states observed, increasing by only 23% from smallest to largest
21 species, (Figure 3a and Supplementary information Table S2). The experimental values
22 obtained are in extremely good agreement with the theoretical value obtained minimising the
23 NMR structure of this mutant (PDB 2HDM). The file used to determine the theoretical CCS,
24 however, does not contain the final 18 amino acids in the intrinsically disordered tail. As

1 experimental values match extremely well with this theoretical value without the full length
2 tail we surmise, that as observed previously for the low charge states of WT Ltn10 (Figure 1),
3 the intrinsically disordered tail is wrapped around the structural core of the protein, but this
4 occurs for all charge states of CC3 studied here under buffered conditions. If the full ID tail
5 were to unfold from the structural core of the protein as a single unfolding event, as for WT
6 Ltn10, a dramatic increase in CCS, between two charge states or two conformations of the
7 same charge state, on the order of 40 % would be expected (Figure 1). The increase in CCS is
8 not as significant for the CC3 mutant suggesting that this restructuring does not occur here.
9 In fact in the CC3 mutant the helix is pinned to the core by the extra disulfide bridge and,
10 therefore, the first unfolding pathway which couples tail and helix unfolding observed for the
11 WT Ltn10 cannot occur for this mutant. Furthermore, this pinning of the α -helix to the
12 structural core and corresponding conformational tightening may also further encourage the
13 association of the tail to the structural core, limiting its unfolding from the core.

14 For ECD experiments the species $[M_{CC3+5H}]^{5+}$ to $[M_{CC3+7H}]^{7+}$ were observed at high
15 enough intensity to be examined (Supplementary Information Figure S4 and S5). Once again
16 by considering c- and z-type fragments in combination with internal fragments a
17 fragmentation map can be built-up to visualise the unfolding of the structurally constrained
18 Ltn10 fold, shown in Figure 3b in which the β -sheet regions and α -helix are again highlighted
19 in yellow and purple, respectively. For all species studied very little fragmentation is
20 observed, this is as expected due to the compact nature of these species and due to the fact
21 that the species are relatively low charged in combination with the two disulfide bonds which
22 can limit fragmentation and dissociation by ECD. Disulfide bonds have been shown to be
23 preferentially cleaved in ECD⁴², which can result in lower backbone fragmentation
24 efficiency. Furthermore in compact proteins which contain numerous non-covalent
25 interactions and those with multiple disulfide bonds, there is the possibility of fragmentation

1 without subsequent dissociation, limiting the observed fragments. Despite this, information
2 about how the CC3 mutant unfolds can still be obtained. At the lowest charge states,
3 $[M_{CC3+5H}]^{5+}$ and $[M_{CC3+6H}]^{6+}$, minor fragmentation is observed at both termini, the higher
4 charged species, $[M_{CC3+7H}]^{7+}$ and $[M_{CC3+8H}]^{8+}$, display slightly more fragmentation sites,
5 consistent with these species being more extended or unfolded and hence capable of
6 accepting a higher number of protons. The more extended nature of these species, would,
7 therefore, provide more available sites for fragmentation and subsequent dissociation.
8 Increased fragmentation is most evident along the N-termini and C-termini suggesting these
9 regions are beginning to decouple from the core, coinciding with an increase in CCS
10 observed for these two species. For both species internal fragments where the intrinsically
11 disordered tail has been lost are observed. The internal fragments further supporting that this
12 protein begins to unfold from both termini, with higher intensity fragmentation being
13 observed as a result of unfolding of the C-terminus consistent with the findings for both the
14 WT and WT 1-72 in which the protein was observed to unfold most extensively from the C-
15 terminus.



1

2 **Figure 3: CC3Ltn. DT IM-MS results:** Average experimental CCS for M_{CC3} obtained from three different day
 3 repeats, error bars are calculated from the standard deviation and generally fall within symbol size. Dashed and
 4 dotted lines represents theoretical CCS obtained from NMR structures PDB 2HDM and 1J9O respectively.
 5 Representative structures were obtained from the theoretical calculations by selecting the lowest energy species
 6 among those having a CCS close to the peak position of the distributions calculated. Inserts: Mass spectra
 7 obtained on DT IM-MS instrument of 100 μ M protein a) Buffered CC3 in 20 mM Am. Ac and c) Buffered and
 8 supercharged CC3 in 20 mM Am. Ac + 1% *m*-NBA **ECD analysis:** Percentage yields calculated for c type, z
 9 type and zlc fragments as a function of cleavage site. b) Buffered CC3 over the charge state range $[M_{CC3}+5H]^{5+}$
 10 to $[M_{CC3}+8H]^{8+}$ d) Buffered and supercharged CC3 over the charge state range $[M_{CC3}+8H]^{8+}$ to $[M_{CC3}+11H]^{11+}$.
 11 For figure clarity a y-axis scale break is included for all CC3 fragmentation maps.

12 For $[M_{CC3}+8H]^{8+}$ low intensity fragmentation in the α -helix is also observed suggesting this is
 13 the first secondary structural element to unfold from the structural core a finding which is
 14 consistent with both the WT Ltn10 and the truncated mutant WT 1-72. Less fragmentation in

1 the central β -sheet core is observed than for the WT Ltn10 suggesting that there are still
2 stabilising non covalent interactions here which is consistent with our DT IM-MS
3 experiments in which the $[M_{CC3+8H}]^{8+}$ species is only 23% larger than the most compact
4 species. The lack of fragmentation in this region could also be due to the presence of the two
5 disulfide bonds, which would also act to stabilise this region.

6 For CC3 minimal fragmentation occurs within the β -sheet core, however, there are a number
7 of low intensity internal fragments which are observed within this region for the highest
8 charge state probed under these conditions, namely the $[M_{CC3+8H}]^{8+}$ species. These
9 fragments are of low abundance (19% of the total fragments for this species being due to
10 fragmentation in this region and comprising 6% of the total intensity of fragments). When
11 cleavage in this region occurs it results in z/c fragments generally consisting of more than
12 one β -strand suggesting the fragments at least partially retain some of the stabilising
13 interactions present in the solution fold. This observed stability is most likely due to the two
14 disulfide bonds which would strengthen and stabilise this region, as well as stabilising non-
15 covalent interactions present in this compact, constrained species. Considering the DT IM-
16 MS and ECD data, and the fact that this species is mostly observed at low charge states, it is
17 clear that this mutant is more compact and less prone to restructuring with respect to charge
18 state than the WT protein. The extra disulfide bond in CC3 would confer this stability by
19 essentially pinning the structural core together. To probe CC3 at higher charge states we
20 added a 'supercharging' reagent, *m*-NBA (*meta*-nitrobenzyl alcohol)⁴³.

21 **Further probing the unfolding landscape of Ltn10: Supercharged CC3**

22 Addition of 1% *m*-NBA shifted the highest charge state to $[M_{CC3+13H}]^{13+}$, (Figure 3c insert
23 and Supplementary information Figure S6). In order to probe the conformations of these ions,
24 the CCS for the species $[M_{CC3+5H}]^{5+}$ to $[M_{CC3+13H}]^{13+}$ were then determined (Figure 3c and

1 Supplementary Information Table 3). An increase in CCS of up to 13% at like for like charge
2 states is found, suggesting *m*-NBA promotes more extended structures at all charge states, not
3 only the higher charge states promoted by this reagent. Previous studies have shown that
4 addition of supercharging reagents can alter conformations of protein complexes⁴⁴. From
5 Figure 3c, it is evident that under these conditions we do not see any major increase in the
6 CCS for example the transition between $z = 7$ and $z = 8$ seen in WT Ltn in which CCS
7 increases by ~40%. Instead the CCS increases in smoother steps with respect to charge (from
8 1020 to 1828 Å²), clearly the higher charge states are promoting unfolding, however, the
9 single stage tail-unfolding is not observed here and unfolding is a more continuous process
10 with respect to charge.

11 The experimental CCS obtained for the supercharged CC3 Ltn have been compared both to
12 the theoretical CCS obtained from the minimised NMR coordinates for this mutant (PDB
13 2HDM) and for the WT Ltn10, the structure of which has the intrinsically disordered tail
14 extended out from the structural core of the protein (PDB 1J9O). The theoretical CCS
15 determined from the CC3 mutant compare well with the lowest charge states observed for
16 CC3 under these conditions ($[M_{CC3+5H}]^{5+}$ and $[M_{CC3+6H}]^{6+}$), suggesting that despite the
17 addition of *m*-NBA these species remain folded with the intrinsically disordered tail
18 associated with the structural core. The theoretical CCS obtained from WT Ltn10 (1694 Å²)
19 compares extremely well with the experimental CCS for CC3 Ltn $[M_{CC3+12H}]^{12+}$ and
20 $[M_{CC3+13H}]^{13+}$ species (1777 and 1828 Å²), which suggests the increase in CCS can be
21 attributed to unfolding of the intrinsically disordered tail, however, the continuous increase in
22 CCS suggests this does not occur in a single step as observed for WT Ltn.

23 In order to determine the unfolding pathways accessed through the addition of *m*-NBA the
24 $[M_{CC3+8H}]^{8+}$ to $[M_{CC3+11H}]^{11+}$ charge states were subjected to ECD fragmentation, Figure
25 3d and Supplementary Information Figure S7 and S8. It is interesting to first consider the

1 $[M_{CC3+8H}]^{8+}$ species, which was probed under both buffered and supercharged conditions.
2 Under supercharging conditions more extensive fragmentation is observed along the
3 backbone of the α -helical region, suggesting that this region has become more unstructured
4 and therefore more available for fragmentation under supercharged conditions. This finding
5 is consistent with the observed increase of CCS for this species in the presence of *m*-NBA.

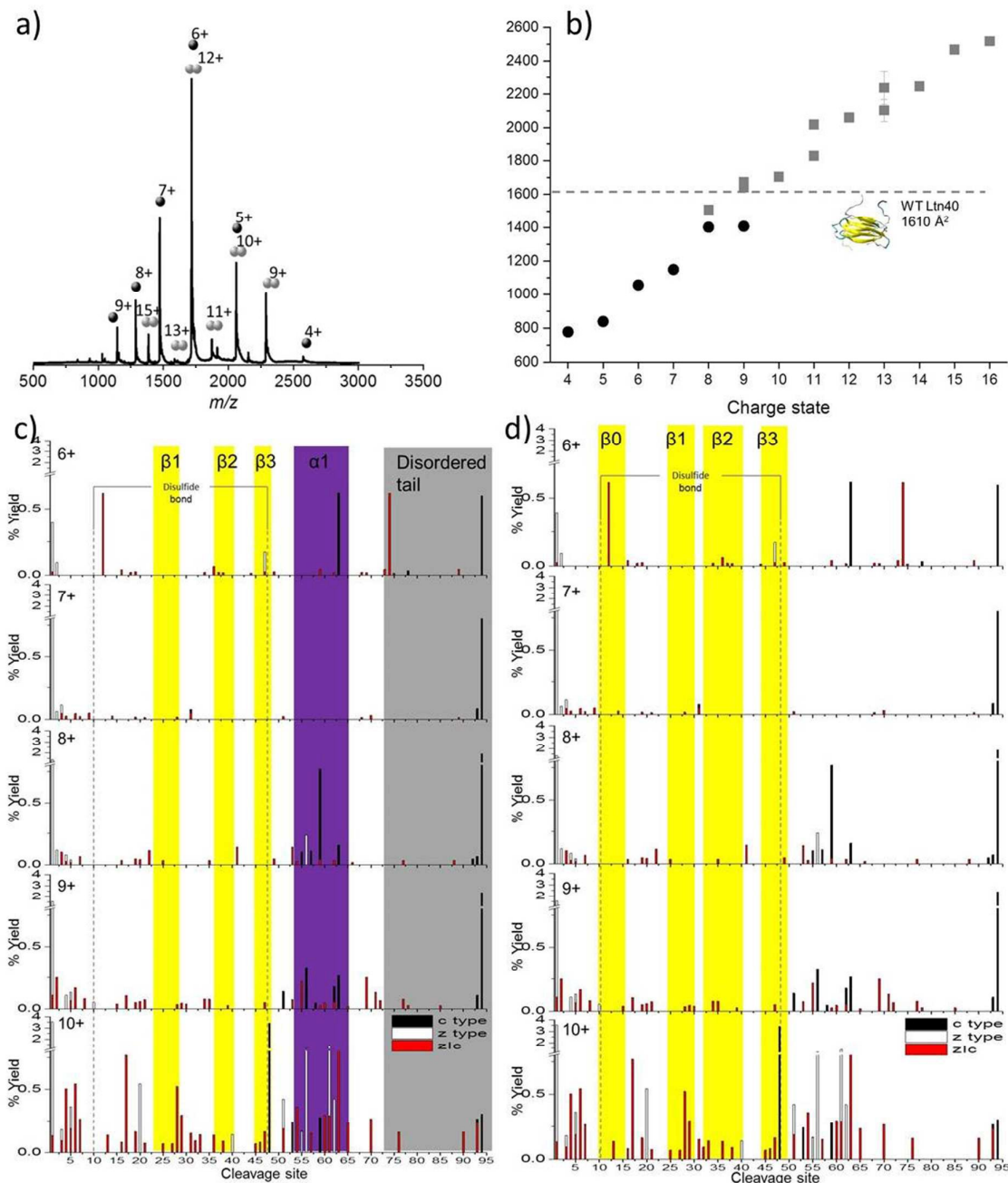
6 As we move to larger CCS, higher charged species, we observe more extensive fragmentation
7 in both termini as well as along the intrinsically disordered tail region, emphasising that these
8 regions are the first to unfold from the core of the protein. At all charge states studied under
9 supercharged conditions zlc fragments are observed which correspond to the complete loss of
10 the intrinsically disordered tail and regions of the N-termini, retaining the full structural core.
11 This observation highlights that under these conditions, for all species of CC3 studied by
12 ECD, the disordered regions are more weakly associated with the structural core of the
13 protein and hence available for fragmentation than under buffered conditions.

14 Additionally, in the $[M_{CC3+11H}]^{11+}$ species we see greater fragmentation in the α -helix
15 suggesting this region is becoming more unfolded and beginning to unravel from the
16 structural core of the protein. We do observe fragmentation in the β -sheet region for this
17 species, but at a lower level than that observed for the α -helix, consistent with our previous
18 findings both for the WT Ltn and WT 1-72 constructs which showed that this region remains
19 stable in the Ltn10 fold, due in part to the disulfide bonds. As the protein becomes more
20 unfolded we see greater fragmentation including evidence for electron mediated reduction of
21 disulfide bridges, which suggests they are more accessible. It is important to note that due to
22 the conformational constriction implemented in this mutant by the addition of the second
23 disulfide bridge, we do not see the unfolding pathway involving the complete loss of the α -
24 helix and disordered tail in a single fragment, that is observed for the WT Ltn10, until we
25 reach the most extended species, $[M_{CC3+11H}]^{11+}$. The extra disulfide bridge in CC3

1 successfully structurally constrains the Ltn10 fold and inhibits the unfolding limiting the
2 possibility of unstructuring the core residues needed for interconversion to Ltn40⁴¹.

3 **Probing the Ltn40 fold through specific site mutation**

4 The final mutant studied here, known as W55D Ltn, contains a single point mutation
5 (tryptophan at position 55 is mutated to an asparagine) shown by NMR to shift the
6 equilibrium towards the dimeric structure³⁵. The amino acid replacement is located in the
7 beginning of the α -helix in the Ltn10 conformation and is designed to destabilise this
8 conformation and push the equilibrium towards the dimeric fold (Ltn40), with loss of the
9 tryptophan side chain disrupting the Ltn10 hydrophobic core and destabilising this fold. In
10 our experiments, however, the most intense species is the monomer (M_{W55D}) presenting over
11 a range of charge states from $[M_{W55D}+4H]^{4+}$ to $[M_{W55D}+9H]^{9+}$, Figure 4a. The dimeric species
12 (D_{W55D}) is also observed, at somewhat lower intensity than the monomer, over the charge
13 state range $[D_{W55D}+9H]^{9+}$ to $[D_{W55D}+15H]^{15+}$ wider by three charge states than the WT dimer,
14 Figure 4b. The equilibrium of W55D monomer to dimer could not be significantly shifted
15 towards the dimer through the alteration of pH over the range 2.8 to 9.8 or through increasing
16 the buffer strength, Supplementary Information Figure S9. The fact that the highest intensity
17 peaks observed are due to the monomeric species is discussed in more detail later.



1

2 **Figure 4: W55D ltn.** a) Mass spectra obtained on the DT IM-MS instrument for 100 μM W55D in 20 mM
 3 AmAc. b) Average experimental CCS obtained from three different day repeats, error bars are calculated from
 4 the standard deviation and generally fall within symbol size. Dashed line represents theoretical CCS of Ltn40
 5 obtained from the NMR structure (PDB 2JP1). **Monomeric ECD analysis:** Percentage yields calculated for c
 6 type, z type and internal fragments as a function of cleavage site for M_{W55D} over the charge states $[M_{W55D}+6H]^{6+}$
 7 to $[M_{W55D}+10H]^{10+}$. Fragments are mapped onto c) Ltn10 fold and d) Ltn40 fold. For figure clarity a y-axis
 8 break is included.

9 From DT IM-MS it is clear that the monomeric change in CCS with respect to charge is
 10 different for W55D than for CC3, WT Ltn and WT 1-72 under buffered conditions. We no

1 longer see a stable ensemble of species presenting over a narrow CCS range, which can then
2 be followed by a major increase in CCS^{24} , instead the CCS is observed to increase almost
3 uniformly with charge from 778 to 1407 \AA^2 , Figure 4b and Supplementary Information Table
4 S4.

5 The dimeric form of the W55D Ltn mutant also appears notably different to the WT Ltn and
6 presents over an extremely wide range of CCS, (Figure 4b and Supplementary Information
7 Table 4). In its dimeric form W55D Ltn presents over a CCS range of 1509 to 2519 \AA^2 , with
8 the most extended form observed ($[D_{W55D}+16H]^{16+}$) being 32% larger than the most
9 extended species observed for the WT Ltn40 ($[D_{WT}+13H]^{13+}$). In this case, as no PDB file
10 exists for the mutant dimer, theoretical comparisons were made with the WT dimer (PDB
11 2JP1), however, this file does not contain the intrinsically disordered tail and the theoretical
12 value is therefore based on a structure which is missing 33 amino acids from each chain.
13 From comparison of the experimental and theoretical CCSs we see that D_{W55D} populates a
14 wide range of conformations, both more compact and more extended than this theoretical
15 value. Most conformations are far more extended than the theoretical CCS, which suggests
16 that either the tail is extended out from the structural core and/or the whole structure is more
17 flexible and, therefore, capable of existing in a larger range of conformations, which in turn
18 suggests a smaller interface region is required to stabilise the dimer. It is indeed likely that
19 the wide range of CCS observed is due to a combination of both of these factors. It also
20 appears that less dimer signal is observed for this mutant in comparison to the WT, perhaps
21 suggesting that it is not as stable during ionisation and desolvation and is dissociating into
22 monomer during this process. In order to further probe the flexibility and stability of the
23 W55D mutant ECD analysis was performed both on the monomeric ($[M_{W55D}+6H]^{6+}$ to
24 $[M_{W55D}+10H]^{10+}$) and dimeric ($[D_{W55D}+9H]^{9+}$, $[D_{W55D}+11H]^{11+}$ and $[D_{W55D}+13H]^{13+}$) species
25 of W55D Ltn, Supplementary Information Figures S10 to S12.

1 Discussion will first focus on the monomeric form of W55D. The monomeric species
2 presents us with an interesting challenge in identifying the protein fold; can we use this
3 combination of techniques to determine if the W55D Ltn monomer possess an ECD/IM-MS
4 signature of the conserved chemokine monomer fold, Ltn10, as with monomeric WT 1-72
5 and CC3 Ltn, or are the monomeric species we observe due to dissociation of the dimer and
6 hence exist as a single unit in the dimeric Ltn40 fold?

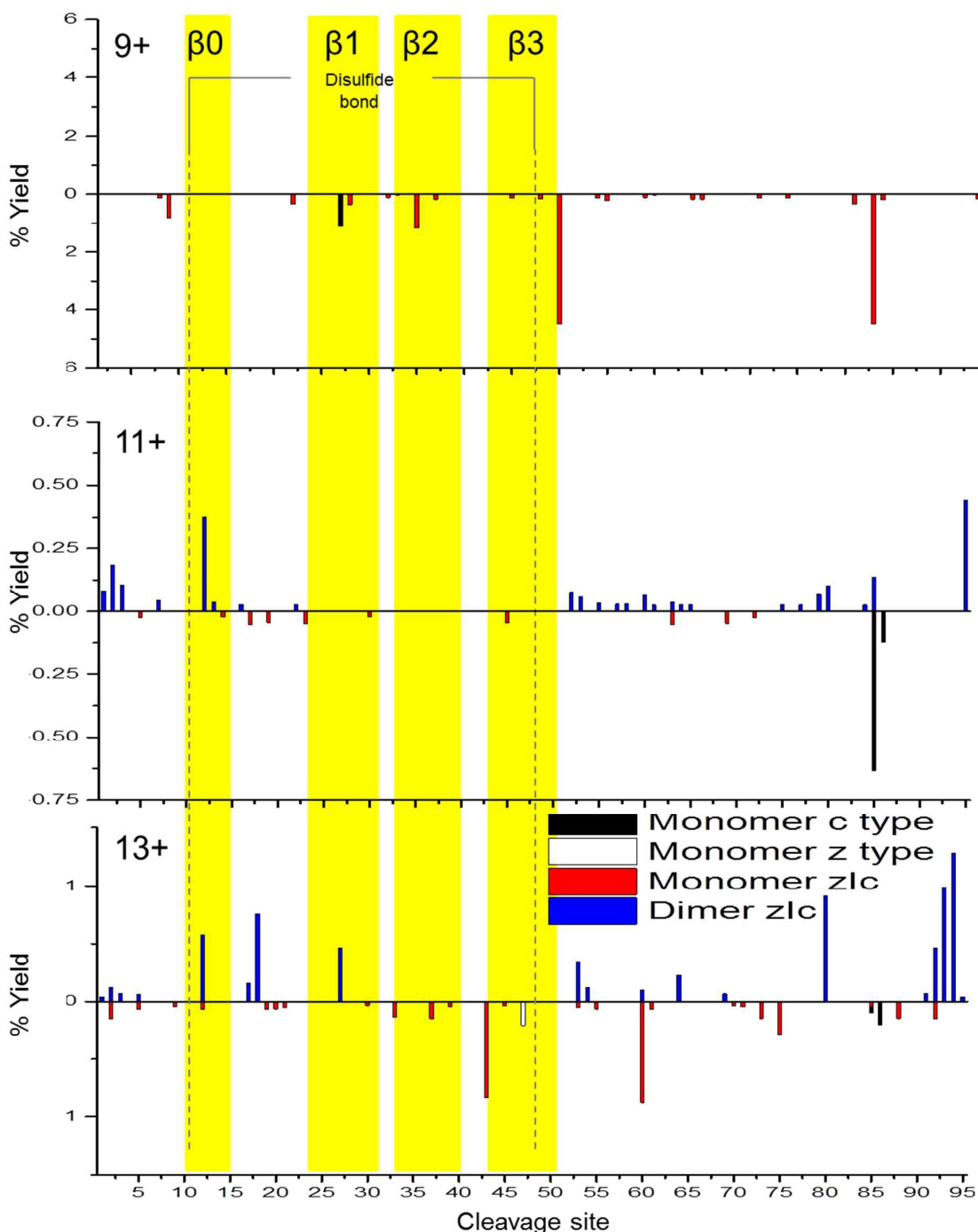
7 In order to determine which, if either, fold the monomeric species of W55D Ltn exists in we
8 have mapped the fragmentation observed for the monomeric W55D over the charge states
9 $[M_{W55D}+6H]^{6+}$ to $[M_{W55D}+10H]^{10+}$ onto both the monomeric fold and a single chain of the
10 dimeric fold, Figure 4c and 4d, respectively. As expected the monomer fragmentation
11 increases as a function of charge, however, the trends observed can enable us to identify the
12 conformation present. As we move to more extended and hence higher charged species
13 extensive fragmentation is observed in the N-terminus suggesting this region is beginning to
14 unfold, as observed for both Ltn10 and Ltn40.

15 The most extended species and therefore highest charged species ($[M_{W55D}+10H]^{10+}$) displays
16 extensive fragmentation into the core residues of both folds suggesting that this species is
17 significantly unstructured. Therefore, we focus on the intermediate charge states where less
18 Colombically driven unfolding is likely. Beginning with the $[M_{W55D}+8H]^{8+}$ species we
19 observe fragmentation in the region of the residues 52-65. This fragmentation is not
20 consistent with the data from the Ltn10 (see for example Figure 2c), where we see
21 fragmentation consistently throughout the region that is helical in solution. Instead
22 fragmentation appears to branch out from the β -sheet core in the Ltn40 fold, (comparing
23 Figure 4c with Figure 4d). This is consistent with the observations made for the dimeric WT
24 Ltn40 fold in which we see fragmentation branching out from the β -sheet core with higher
25 intensity fragments in the first 9 amino acids after the β 3 strand, and supported by NMR data

1 which shows that these regions that flank the β -sheet are associated with the protein core
2 allowing them to be solved. Additionally, If M_{W55D} were to exist as Ltn10 we would expect to
3 see more evidence for the first unfolding pathway of this fold, which involves unfolding of
4 the α -helix and intrinsically disordered tail, with loss of both regions being observed in a
5 single fragment ion. The fragmentation maps suggest that M_{W55D} is likely to be existing as a
6 single unit of the dimeric Ltn40 fold. This could be a result of dissociation of the dimer upon
7 ionisation and transfer of the protein, suggesting that the dimer interface is not as stable as
8 that of the WT type protein. If this is indeed the case we can surmise that over this time scale
9 in the gas phase monomeric W55D also does not restructure to the Ltn10 conformation.

10 To probe dimeric W55D Ltn conformations, the charge states of the dimer where the net
11 charge is an odd number and, therefore, are not coincident with monomer species were
12 considered, namely $[D_{W55D}+9H]^{9+}$, $[D_{W55D}+11H]^{11+}$ and $[D_{W55D}+13H]^{13+}$. As for the
13 monomer, c-type, z-type and zIc fragment yields were considered, however, in this case
14 fragmentation can occur either from the dimer remaining intact and fragmenting or
15 alternatively it may dissociate to monomer and then fragment, both cases were considered
16 here. As both monomeric units comprising the dimer are identical in sequence these
17 experiments do not distinguish between the two chains and therefore the total fragment
18 percentage yields are mapped onto a single monomeric unit, Figure 5. At the lowest charge
19 state studied here, $[D_{W55D}+9H]^{9+}$, ECD leads to extensive dissociation of the dimer to
20 monomer followed by extensive fragmentation along the backbone, represented by below
21 axis histograms Figure 5. The tendency of the lowest dimer charge state to dissociate to
22 monomer was also observed for the WT protein and attributed to either an encounter complex
23 that lacks a stable β -sheet core or a less stable, perhaps collapsed form of the dimer. This
24 tendency of the dimer to dissociate to monomer upon ECD fragmentation is, however,

- 1 observed for all charge states of the W55D Ltn dimer studied here which also suggests that
- 2 this dimer interface is not as strong in this mutant as compared to the WT Ltn.



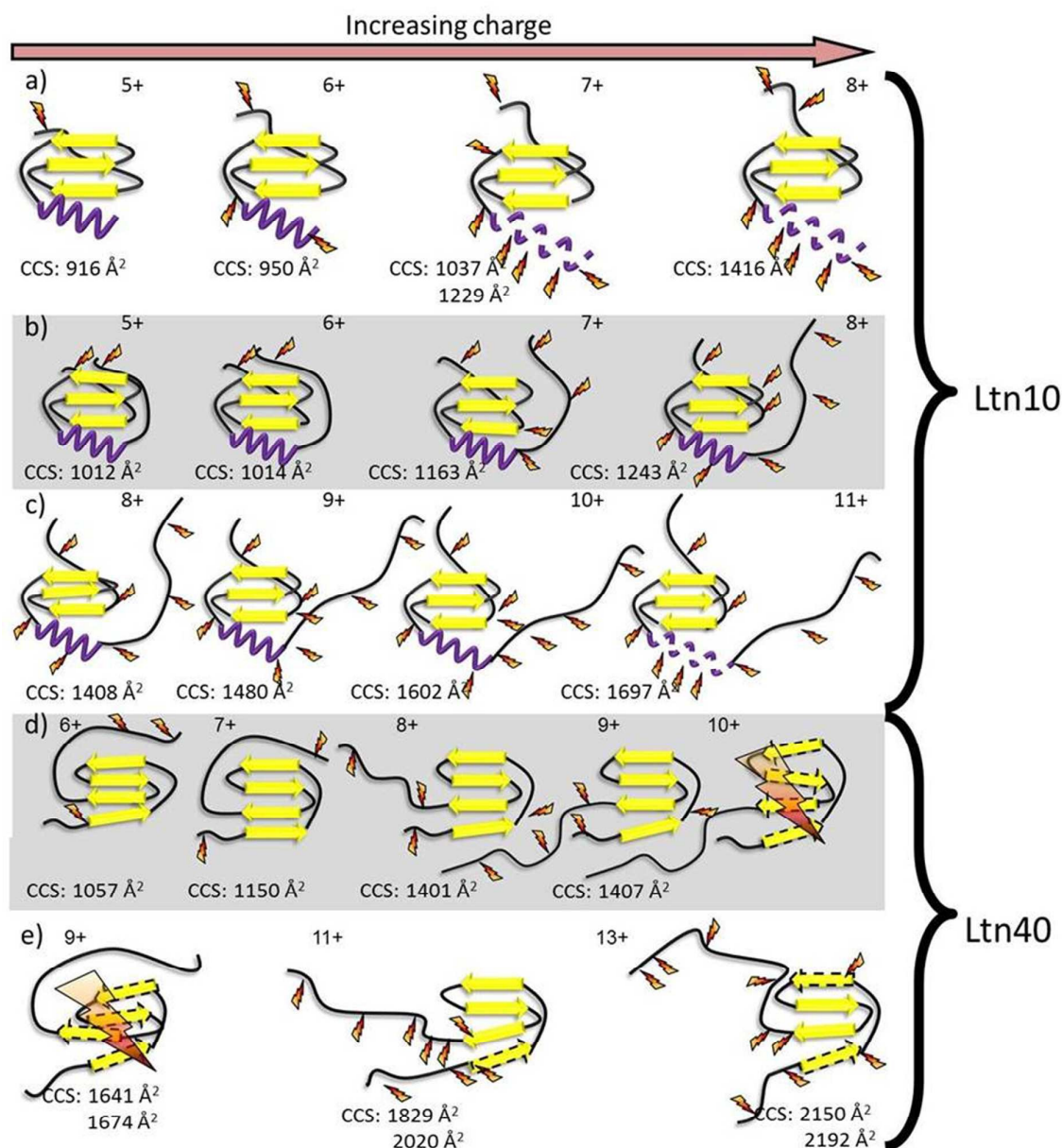
3

4 **Figure 5:** Percentage yields calculated for c type, z type and zIc fragments as a function of cleavage site for
 5 dimeric W55D over the charge states $[D+9H]^{9+}$, $[D+11H]^{11+}$ and $[D+13H]^{13+}$. Above axis histograms represent
 6 fragments formed while retaining the dimer interface. Below axis histograms represent fragments formed
 7 following dimer dissociation to monomer. For figure clarity, the y-axis scale varies with charge state.

1 For the $[D_{W55D}+11H]^{11+}$ the observed dimer fragmentation maps extremely well with the
2 Ltn40 fold, with the majority of the fragmentation occurring at the termini and in the flexible
3 regions just before the β -sheet core. This suggests that some dimeric species which survive
4 the desolvation process are likely to be present in the Ltn40 fold. For $[D_{W55D}+11H]^{11+}$ we
5 begin to see fragmentation within the β_0 strand, which increases in intensity as we go up in
6 charge state to the $[D_{W55D}+13H]^{13+}$ species. The increase in fragmentation along this region
7 suggests it is the least stable of the β -strands and is the first to unfold from the structural core
8 of the protein. This indicates that the Ltn40 conformation unfolds from the N-terminus,
9 which is the converse of the Ltn10 conformation which appears to unfold first from the C-
10 terminus. The ECD fragmentation maps allow us to rank the stability of each of the β -strands
11 from least to most stable ($\beta_0 < \beta_1 < \beta_3 < \beta_2$), which could help in the design of future mutants
12 aimed to mimic this fold but with increased stability and therefore decreased unfolding.

13 **Visualising protein unfolding**

14 By combining the DT IM-MS results with the ECD fragmentation maps for all the mutants
15 studied here we can build up a detailed picture of the unfolding of both Ltn10 and Ltn40,
16 giving insight into fold stability as a function of increased charge, Figure 6.



1

2 **Figure 6:** A cartoon representation of the unfolding of the Ltn10 and Ltn40 folds as shown by DT IM-MS and
 3 ECD, probed through specific mutants, a) WT 1-72 b) buffered CC3 c) buffered and supercharged CC3 d)
 4 monomeric W55D and e) dimeric W55D, where the dimer is represented here by a single monomeric unit.
 5 Yellow arrows represent β -sheet regions and purple spirals represent α -helical regions. Orange 'lightning bolts'
 6 indicate sites where ECD fragmentation is most significant, extensive fragmentation along the backbone is
 7 depicted by a large lightning bolt and dashed lines represent a lack of defined structural elements.

8 Using the methodology presented here we are able to compare the three mutants of this
 9 metamorphic protein. In summary, we find that in the absence of the intrinsically disordered
 10 tail unfolding of the Ltn10 fold still proceeds via the C-terminus in a similar manner as that

1 observed for full length WT Ltn. For WT 1-72, however, we find unravelling of the α -helix
2 occurs at lower charge states than for the full length WT Ltn10, suggesting the intrinsically
3 disordered tail confers some stability or protection to the structural core and in particular the
4 α -helix. We observe that the CC3 Ltn mutant is more structurally constrained than the WT
5 Ltn, as expected for a protein with an additional disulfide bond, and exists in the Ltn10 fold.
6 The positioning of this additional disulfide bond essentially pins the α -helix to the structural
7 core of the protein and blocks the first unfolding pathway of the Ltn10 conformation. For the
8 W55D Ltn our results suggest that the dimer interface of the Ltn40 fold is less stable in this
9 mutant than in the WT protein, leading to dissociation of dimer to monomer both upon
10 ionisation and transfer and as a result of ECD fragmentation. The fragmentation map of this
11 mutant also suggests that for this conformation the first unfolding pathway involves the
12 unfolding of the β 0 strand and N-terminus. The combination of these two techniques and
13 their application to mutagenesis studies can further in the intelligent design of mutants to
14 stabilise protein fold and unfolding by highlighting susceptible regions.

15 **Conclusions**

16 The ability of this combination of techniques to distinguish significant effects of single and
17 double point mutations on protein fold and unfolding highlights the wealth of information
18 that can be obtained in this way. The power of using such techniques becomes evident when
19 you consider the small sample quantities required for such studies, with as little as 25 μ g of
20 protein providing insight. Information can be obtained on the relative stabilities of the
21 mutants in comparison to the WT protein and therefore has the potential to be highly
22 influential in design of site directed mutants.

23 **Experimental Procedures**

24 **Protein expression and purification**

1 All recombinant human Ltn protein used in ECD and DT IM-MS investigations were
2 expressed and purified as previously described.^{41, 45} The lymphotactin mutants CC3 and
3 W55D were prepared by site-directed mutagenesis using complementary primer pairs and the
4 Stratagene QuickChange kit follow the manufacturer's instructions. All expression vectors
5 were verified by DNA sequencing. Purified proteins were frozen, lyophilized and stored
6 at - 20 °C for subsequent study.

7 **DT IM-MS and theoretical calculations**

8 For all ion mobility mass spectrometry experiments samples were ionised and introduced into
9 the instrument using nano-electrospray ionisation. Nano-electrospray tips were made in-
10 house using thin-walled glass capillaries (i.d. 0.5 mm) using a Flaming/Brown micropipette
11 puller (Sutter Instrument Company, Novato, CA, USA). Sample solutions were ionised
12 through a potential applied to a thin platinum wire (0.125 mm Goodfellow) inserted into the
13 glass capillary.

14 MS and DT IM-MS experiments were performed on an in-house modified Q-TOF
15 (Micromass UK Ltd.), adapted in order to carry out separations based on an ion's mobility,
16 and to enable the temperature dependent CCS to be determined. The DT IM-MS instrument
17 was previously modified to include a 5.1 cm long copper drift cell and supplementary ion
18 optics situated post source but before the quadrupole analyser. The instrument and its
19 operation have been described in detail elsewhere⁴⁶ and further details are found in
20 supplementary information.

21 Theoretical CCSs were calculated from NMR structures (PDB identifiers 1J9O, 2JP1 and
22 2HDM for Ltn10, Ltn40 and monomeric CC3 mutant, respectively) using the trajectory
23 method of MOBCAL code⁴⁷. Further details on the procedure are found in Supplementary
24 information.

1 ECD

2 All high resolution mass spectrometry and ECD analysis were carried out on a 12 T Solarix
3 FTICR (Bruker Daltonics). All samples were ionised using nano-electrospray ionisation, n-
4 ESI was performed using a NanoMate (Advion biosciences) running in infusion mode and
5 equipped with a HD_A_0 ESI chip (Advion biosciences). Analysis of WT 1-72 was carried
6 out on 5 μ M WT 1-72 in 100 mM ammonium acetate. Ion accumulation was between 3 and
7 10 seconds for each charge state studied, depending on the intensity of the species, and in all
8 cases ECD fragmentation was summed over 300 scans. For W55D, analysis was performed
9 on 30 μ M W55D in 100 mM ammonium acetate. Ion accumulation was between 0.5 and 8
10 seconds for each charge state studied, depending on the intensity of the species, and in all
11 cases ECD fragmentation was summed over 200 scans. For CC3, analysis was performed on
12 30 μ M CC3 in 100 mM ammonium acetate for 'buffered' analysis and 30 μ M CC3 in 100
13 mM ammonium acetate + 1 % *m*-NBA for 'buffered and supercharged' studies. In both cases
14 ion accumulation was between 0.2 and 10 seconds, depending on the intensity of the species,
15 and ECD fragmentation was summed over 200 scans. For this work it was not possible to
16 collisionally activate the ions post ECD, to reveal cleaved sites on the protein backbone
17 which perhaps have not dissociated due to non-covalent interactions. Future work could
18 employ IRMPD to investigate this.

19 Author Contributions

20 SRH did all of the mass spectrometry experiments and analysis, and prepared the first draft of
21 the manuscript. MP performed the calculations. RT expressed the proteins and with BFV
22 designed the site specific mutants. PEB and CEM helped to analyse the data and PEB wrote
23 the submitted form of this manuscript.

24 Acknowledgments

1 The Schools of Chemistry and Physics at the University of Edinburgh, are thanked for an
2 award of an EPSRC DTA studentship to SRH. Dr. Logan Mackay and Dr. David Clarke are
3 thanked for assistance with FT-ICR MS. Dr. Michel Laguerre (Institut Européen de Chimie et
4 Biologie (IECB), Pessac, France) is thanked for providing access to computational resources.

5 **Supporting Information.**

6 Further experimental details for the drift tube ion mobility mass spectrometry experiments,
7 tables of experimental collision cross sections obtained from DT IM-MS experiments, spectra
8 obtained and further information on the ECD fragmentation experiments can be found in
9 Supporting Information. In addition experimental details and theoretical CCS derivation can
10 also be found in Supporting Information.

- 11 1. Dole, M., Mack, L. L., Hines, R. L., Mobley, R. C., Ferguson, L. D., and Alice, M. B. (1968)
12 Molecular Beams of Macroions, *The Journal of Chemical Physics* 49, 2240-2249.
- 13 2. Yamashita, M., and Fenn, J. B. (1984) Electrospray ion source. Another variation on the free-
14 jet theme, *The Journal of Physical Chemistry* 88, 4451-4459.
- 15 3. Wilm, M. S., and Mann, M. (1994) Electrospray and Taylor-Cone theory, Dole's beam of
16 macromolecules at last?, *International Journal of Mass Spectrometry and Ion Processes* 136,
17 167-180.
- 18 4. Wilm, M., and Mann, M. (1996) Analytical properties of the nanoelectrospray ion source,
19 *Analytical Chemistry* 68, 1-8.
- 20 5. Fenn, J., Mann, M., Meng, C., Wong, S., and Whitehouse, C. (1989) Electrospray ionization
21 for mass spectrometry of large biomolecules, *Science* 246, 64-71.
- 22 6. Hoaglund-Hyzer, C. S., Counterman, A. E., and Clemmer, D. E. (1999) Anhydrous protein ions,
23 *Chem Rev* 99, 3037-3080.
- 24 7. Heck, A. J. R. (2008) Native mass spectrometry: a bridge between interactomics and
25 structural biology, *Nat Meth* 5, 927-933.
- 26 8. Benesch, J. L. P., Ruotolo, B., Simmons, D. A., and Robinson, C. V. (2007) Protein Complexes
27 in the Gas Phase technology for structural genomics and proteomics, *Chemical Reviews* 107,
28 3544-3567.
- 29 9. Morton, V. L., Stockley, P. G., Stonehouse, N. J., and Ashcroft, A. E. (2008) Insights into virus
30 capsid assembly from non-covalent mass spectrometry, *Mass Spectrometry Reviews* 27, 575-
31 595.
- 32 10. Ouyang, Z., Takáts, Z., Blake, T. A., Gologan, B., Guymon, A. J., Wiseman, J. M., Oliver, J. C.,
33 Davisson, V. J., and Cooks, R. G. (2003) Preparing Protein Microarrays by Soft-Landing of
34 Mass-Selected Ions, *Science* 301, 1351-1354.
- 35 11. K. B. McAfee, J., and Edelson, D. (1963) Identification and Mobility of Ions in a Townsend
36 Discharge by Time-resolved Mass Spectrometry, *Proceedings of the Physical Society* 81, 382.
- 37 12. Kebarle, P., and Hogg, A. M. (1965) <Mass Spectrometric study of ions at near atmospheric
38 pressure I The ionic polymerization of ethylene.pdf>, *The Journal of Chemical Physics* 42, 668-
39 674.

- 1 13. Hogg, A. M., and Kebarle, P. (1965) <Mass Spectrometric study of ions at near atmospheric
2 pressure II Ammonium Ions produced by the alpha radiolysis of ammonia and their solvation
3 in the gas phase by ammonia and water molecules.pdf>, *The Journal of Chemical Physics* 43,
4 449-456.
- 5 14. Shelimov, K. B., Clemmer, D. E., Hudgins, R. R., and Jarrold, M. F. (1997) Protein Structure in
6 Vacuo: Gas-Phase Conformations of BPTI and Cytochrome *c*, *Journal of the American*
7 *Chemical Society* 119, 2240-2248.
- 8 15. Clemmer, D. E., Hudgins, R. R., and Jarrold, M. F. (1995) Naked Protein Conformations:
9 Cytochrome *c* in the Gas Phase, *Journal of the American Chemical Society* 117, 10141-10142.
- 10 16. Wyttenbach, T., Von Helden, G., and Bowers, M. T. (1996) Gas-Phase Conformation of
11 Biological Molecules Bradykinin, *Journal of the American Chemical Society* 118, 8355-8364.
- 12 17. Hudgins, R. R., Woenckhaus, J., and Jarrold, M. F. (1997) High resolution ion mobility
13 measurements for gas phase proteins: correlation between solution phase and gas phase
14 conformations, *International Journal of Mass Spectrometry and Ion Processes* 165-166, 497-
15 507.
- 16 18. Harvey, S. R., Porrini, M., Stachl, C., MacMillan, D., Zinzalla, G., and Barran, P. E. (2012)
17 Small-Molecule Inhibition of c-MYC:MAX Leucine Zipper Formation Is Revealed by Ion
18 Mobility Mass Spectrometry, *Journal of the American Chemical Society* 134, 19384-19392.
- 19 19. Bernstein, S. L., Dupuis, N. F., Lazo, N. D., Wyttenbach, T., Condron, M. M., Bitan, G., Teplow,
20 D. B., Shea, J.-E., Ruotolo, B. T., Robinson, C. V., and Bowers, M. T. (2009) Amyloid- β protein
21 oligomerization and the importance of tetramers and dodecamers in the aetiology of
22 Alzheimer's disease, *Nature Chemistry* 1, 326-331.
- 23 20. Bleiholder, C., Dupuis, N. F., Wyttenbach, T., and Bowers, M. T. (2011) Ion mobility-mass
24 spectrometry reveals a conformational conversion from random assembly to β -sheet in
25 amyloid fibril formation, *Nat Chem* 3, 172-177.
- 26 21. Pierson, N. A., Valentine, S. J., and Clemmer, D. E. (2010) <Evidence for a Quasi-Equilibrium
27 Distribution of States for Bradykinin Ions in the Gas Phase.pdf>, *Journal Of physical*
28 *Chemistry B* 114, 7777-7783.
- 29 22. Skinner, O., McLafferty, F., and Breuker, K. (2012) How Ubiquitin Unfolds after Transfer into
30 the Gas Phase, *J. Am. Soc. Mass Spectrom.* 23, 1011-1014.
- 31 23. Kalapothakis, J. M. D., Berezovskaya, Y., Zampronio, C. G., Faull, P. A., Barran, P. E., and
32 Cooper, H. J. (2014) Unusual ECD fragmentation attributed to gas-phase helix formation in a
33 conformationally dynamic peptide, *Chemical Communications*.
- 34 24. Harvey, S. R., Porrini, M., Konijnenberg, A., Clarke, D. J., Tyler, R. C., Langridge-Smith, P. R.,
35 MacPhee, C. E., Volkman, B. F., and Barran, P. E. (2014) Dissecting the Dynamic
36 Conformations of the Metamorphic Protein Lymphotactin, *The Journal of Physical Chemistry*
37 *B*.
- 38 25. Zubarev, R. A., Kelleher, N. L., and McLafferty, F. W. (1998) Electron Capture Dissociation of
39 Multiply Charged Protein Cations. A Nonergodic Process, *Journal of the American Chemical*
40 *Society* 120, 3265-3266.
- 41 26. Breuker, K., Oh, H. B., Lin, C., Carpenter, B. K., and McLafferty, F. W. (2004) Nonergodic and
42 conformational control of the electron capture dissociation of protein cations, *Proceedings*
43 *of the National Academy of Sciences of the United States of America* 101, 14011-14016.
- 44 27. Haselmann, K. F., Jorgensen, T. J., Budnik, B. A., Jensen, F., and Zubarev, R. A. (2002) Electron
45 capture dissociation of weakly bound polypeptide polycationic complexes, *Rapid Commun*
46 *Mass Spectrom* 16, 2260-2265.
- 47 28. Horn, D. M., Breuker, K., Frank, A. J., and McLafferty, F. W. (2001) Kinetic Intermediates in
48 the Folding of Gaseous Protein Ions Characterized by Electron Capture Dissociation Mass
49 Spectrometry, *Journal of the American Chemical Society* 123, 9792-9799.

- 1 29. Breuker, K., Oh, H. B., Horn, D. M., Cerda, B. A., and McLafferty, F. W. (2002) Detailed
2 unfolding and folding of gaseous ubiquitin ions characterized by electron capture
3 dissociation, *Journal of the American Chemical Society* 124, 6407-6420.
- 4 30. Oh, H., Breuker, K., Sze, S. K., Ge, Y., Carpenter, B. K., and McLafferty, F. W. (2002) Secondary
5 and tertiary structures of gaseous protein ions characterized by electron capture dissociation
6 mass spectrometry and photofragment spectroscopy, *Proceedings of the National Academy
7 of Sciences of the United States of America* 99, 15863-15868.
- 8 31. Breuker, K., Brüschweiler, S., and Tollinger, M. (2011) Electrostatic Stabilization of a Native
9 Protein Structure in the Gas Phase, *Angewandte Chemie International Edition* 50, 873-877.
- 10 32. Zlotnik, A., and Yoshie, O. (2000) Chemokines: a new classification system and their role in
11 immunity, *Immunity* 12, 121-127.
- 12 33. Kelner, G., Kennedy, J., Bacon, K., Kleyensteuber, S., Largaespada, D., Jenkins, N., Copeland,
13 N., Bazan, J., Moore, K., Schall, T., and et, a. (1994) Lymphotactin: a cytokine that represents
14 a new class of chemokine, *Science* 266, 1395-1399.
- 15 34. Murzin, A. G. (2008) Metamorphic Proteins, *Science* 320, 1725-1726.
- 16 35. Tuinstra, R. L., Peterson, F. C., Kutlesa, S., Elgin, E. S., Kron, M. A., and Volkman, B. F. (2008)
17 Interconversion between two unrelated protein folds in the lymphotactin native state,
18 *Proceedings of the National Academy of Sciences* 105, 5057-5062.
- 19 36. Kuloglu, E. S., McCaslin, D. R., Kitabwalla, M., Pauza, C. D., Markley, J. L., and Volkman, B. F.
20 (2001) Monomeric solution structure of the prototypical 'C' chemokine lymphotactin,
21 *Biochemistry* 40, 12486-12496.
- 22 37. Tyler, R. C., Murray, N. J., Peterson, F. C., and Volkman, B. F. (2011) Native-State
23 Interconversion of a Metamorphic Protein Requires Global Unfolding, *Biochemistry* 50, 7077-
24 7079.
- 25 38. Li, X., Lin, C., Han, L., Costello, C. E., and O'Connor, P. B. (2010) Charge remote fragmentation
26 in electron capture and electron transfer dissociation, *Journal of the American Society for
27 Mass Spectrometry* 21, 646-656.
- 28 39. Iavarone, A. T., Paech, K., and Williams, E. R. (2004) Effects of charge state and cationizing
29 agent on the electron capture dissociation of a peptide, *Analytical Chemistry* 76, 2231-2238.
- 30 40. Roepstorff, P., and Fohlman, J. (1984) Proposal for a common nomenclature for sequence
31 ions in mass spectra of peptides, *Biomedical Mass Spectrometry* 11, 601.
- 32 41. Tuinstra, R. L., Peterson, F. C., Elgin, E. S., Pelzek, A. J., and Volkman, B. F. (2007) An
33 Engineered Second Disulfide Bond Restricts Lymphotactin/XCL1 to a Chemokine-like
34 Conformation with XCR1 Agonist Activity†, *Biochemistry* 46, 2564-2573.
- 35 42. Zubarev, R. A., Kruger, N. A., Fridriksson, E. K., Lewis, M. A., Horn, D. M., Carpenter, B. K., and
36 McLafferty, F. W. (1999) Electron capture dissociation of gaseous multiply-charged proteins
37 is favored at disulfide bonds and other sites of high hydrogen atom affinity, *Journal of the
38 American Chemical Society* 121, 2857-2862.
- 39 43. Iavarone, A. T., and Williams, E. R. (2002) Supercharging in electrospray ionization: effects on
40 signal and charge, *International Journal of Mass Spectrometry* 219, 63-72.
- 41 44. Sterling, H., Kintzer, A., Feld, G., Cassou, C., Krantz, B., and Williams, E. (2012) Supercharging
42 Protein Complexes from Aqueous Solution Disrupts their Native Conformations, *J. Am. Soc.
43 Mass Spectrom.* 23, 191-200.
- 44 45. Peterson, F. C., Elgin, E. S., Nelson, T. J., Zhang, F., Hoeger, T. J., Linhardt, R. J., and Volkman,
45 B. F. (2004) Identification and Characterization of a Glycosaminoglycan Recognition Element
46 of the C Chemokine Lymphotactin, *Journal of Biological Chemistry* 279, 12598-12604.
- 47 46. McCullough, B. J., Kalapothakis, J., Eastwood, H., Kemper, P., MacMillan, D., Taylor, K., Dorin,
48 J., and Barran, P. E. (2008) Development of an Ion Mobility Quadrupole Time of Flight Mass
49 Spectrometer, *Analytical Chemistry* 80, 6336-6344.

- 1 47. Mesleh, M. F., Hunter, J. M., Shvartsburg, A. A., Schatz, G. C., and Jarrold, M. F. (1996)
2 Structural Information from Ion Mobility Measurements Effects of the Long-Range, *Journal*
3 *of physical Chemistry A* 100, 16082-16086.

4