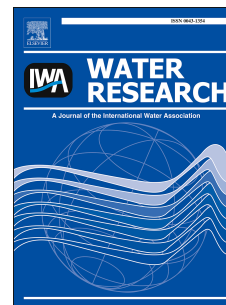


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Effect of Polyhydroxyalkanoates on Dark Fermentative Hydrogen Production from Waste Activated Sludge

Dongbo Wang, Yinguang Chen, Xiaoming Li, Guangming Zeng



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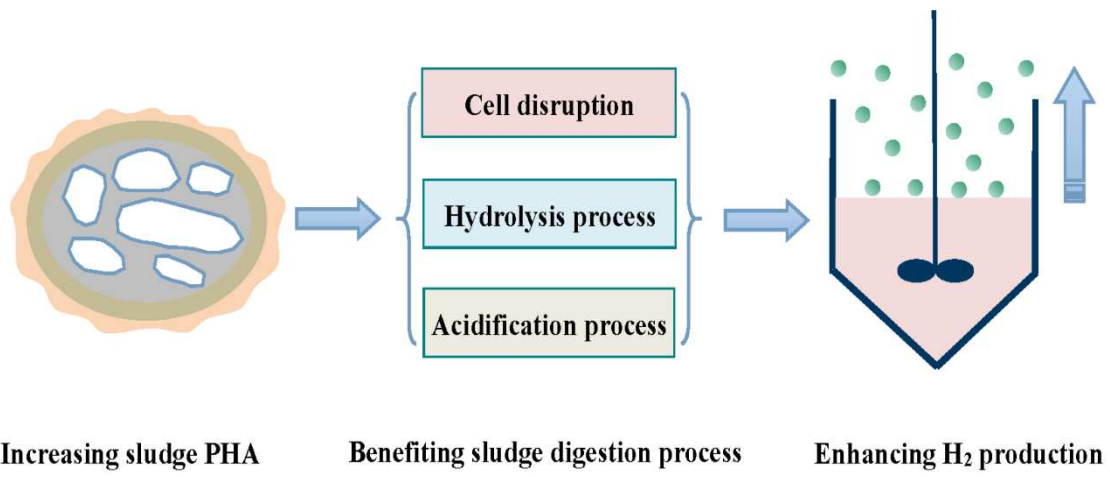
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**Graphical Abstract:**

ACCEPTED MANUSCRIPT

1 **Effect of Polyhydroxyalkanoates on Dark Fermentative Hydrogen**  
2 **Production from Waste Activated Sludge**

3 Dongbo Wang<sup>1,2,3\*</sup>, Yinguang Chen<sup>2\*</sup>, Xiaoming Li<sup>1</sup>, Guangming Zeng<sup>1</sup>

4 <sup>1</sup>College of Environmental Science and Engineering, Hunan University, Changsha 410082, China; Key  
5 Laboratory of Environmental Biology and Pollution Control (Hunan University), Ministry of Education,  
6 Changsha 410082, China

7 <sup>2</sup>State Key Laboratory of Pollution Control and Resources Reuse, School of Environmental Science and  
8 Engineering, Tongji University, 1239 Siping Road, Shanghai 200092, China

9 <sup>3</sup>Advanced Water Management Centre (AWMC), The University of Queensland, QLD 4072, Australia

10 Corresponding author

11 Tel: +86-731-8-8823701. Fax: +86-731-8-8823701

12 E-mail: [w.dongbo@yahoo.com](mailto:w.dongbo@yahoo.com) (Dongbo Wang)

13 Tel: +86-21-65981263. Fax: +86-21-65986313.

14 E-mail: [yg2chen@yahoo.com](mailto:yg2chen@yahoo.com) (Yinguang Chen)

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15 **Abstract:** Polyhydroxyalkanoates (PHA), an intracellular energy and carbon storage polymer, can be  
16 accumulated in activated sludge in substantial quantities under wastewater dynamic treatment (i.e.,  
17 substrate feast-famine) conditions. However, its influence on hydrogen production has never been  
18 investigated before. This study therefore evaluated the influences of PHA level and composition in  
19 waste activated sludge (WAS) on hydrogen production. The results showed that with the increase of  
20 sludge PHA content from 25 to 178 mg per gram volatile suspended solids (VSS) hydrogen production  
21 from WAS alkaline anaerobic fermentation increased from 26.5 to 58.7 mL/g VSS. The composition of  
22 PHA was also found to affect hydrogen production. When the dominant composition shifted from  
23 polyhydroxybutyrate (PHB) to polyhydroxyvalerate (PHV), the amount of generated hydrogen  
24 decreased from 51.2 to 41.1 mL/g VSS even under the same PHA level (around 130 mg/g VSS). The  
25 mechanism studies exhibited that the increased PHA content accelerated both the cell solubilization and  
26 the hydrolysis process of solubilized substrates. Compared with the PHB-dominant sludge, the  
27 increased PHV fraction not only slowed the hydrolysis process but also caused more propionic acid  
28 production, with less theoretical hydrogen generation in this fermentation type. It was also found that  
29 the increased PHA content enhanced the soluble protein conversion of non-PHA biomass. Further  
30 investigations with enzyme analyses showed that both the key hydrolytic enzyme activities and  
31 hydrogen-forming enzyme activities were in the sequence of the PHB-dominant sludge > the  
32 PHV-dominant sludge > the low PHA sludge, which was in accord with the observed order of hydrogen  
33 yield.

34 **Keywords:** hydrogen production from waste activated sludge, dark fermentation, biological nutrient  
35 removal, polyhydroxyalkanoates

## 36 **1. Introduction**

37 The usage of fossil fuel is generally considered as unsustainable due to its diminishing supply and  
38 large contribution to greenhouse gas generation (Lingampalli et al., 2013). Meanwhile, waste activated  
39 sludge (WAS), which is a byproduct of biological wastewater treatment, is inevitably produced in huge  
40 quantities. Therefore, biological production of hydrogen from WAS has attracted much attention (Cai  
41 et al., 2004; Li et al., 2009; Zhao et al., 2010), by which fossil fuel is saved, WAS is reduced and reused,  
42 and the important renewable energy hydrogen is also achieved.

43 In general, the rate of hydrogen production from WAS is low, thus most of previous studies to date  
44 have focused on the enhancement of hydrogen generation efficiency via pretreating sludge (Yang et al.,  
45 2012; Assawamongkholisiri et al., 2013; Kim et al., 2013), controlling operational parameter (Saady,  
46 2013; Gioannis et al., 2013; Zhou et al., 2013), and improving reactor design (Saady, 2013; Jung et al.,  
47 2011). For example, it was found that hydrogen production from WAS could be significantly enhanced  
48 by controlling fermentation pH at constant 10, because this strategy not only improved the hydrolysis  
49 process but also inhibited the activities of hydrogen consuming bacteria of both methanogens and  
50 acetobacteria (Zhao et al., 2010). Besides, it is known that WAS is a nitrogen-rich substrate with low  
51 carbon to nitrogen ratios (around 7/1) whereas the recommended C/N ratio for anaerobic fermentation  
52 system is 20/1 to 30/1 (Kim et al., 2012). Hence, several researches were performed to improve  
53 hydrogen yield through optimizing co-fermentation substrates. It was reported that the bioconversion  
54 of sludge protein and the yield of hydrogen could be largely increased by pertinent addition of  
55 carbohydrate-rich substrates, such as primary sludge, food wastes, and agricultural wastes to WAS  
56 fermentation reactors (Saady, 2013; Zhou et al., 2013; Kim et al., 2012; Chen et al., 2012; Liu et al.,  
57 2013). Despite these important progresses, the enhancement of hydrogen production from WAS by  
58 improving the self-characteristic of sludge has been seldom documented in the literature.

59 Polyhydroxyalkanoates (PHA), an intracellular metabolic intermediate and energy and carbon  
60 storage polymer in wastewater treatment processes, has the ability of rapid and complete degradation  
61 under anaerobic conditions (Reischwitz et al., 1998; Chen and Wang, 2002). PHA can be accumulated  
62 in the external substrate feast stage, but the accumulated PHA is easily consumed in the subsequent  
63 famine stage. As a result, its content in WAS wasted from the traditional wastewater treatment plants  
64 (WWTPs) is usually at low levels (Figure S1, Supporting Information). When using this WAS for  
65 anaerobic fermentation, as mentioned above, the rate of hydrogen production is low. Recently, there  
66 have been increasing evidences showing that WAS with high levels of PHA can be obtained in WWTPs  
67 either by process improvement or by operation optimization. Takabatake et al. (2002) reported that  
68 activated sludge biomass from 4 real WWTPs had the capability to accumulate PHA up to 18.8% of dry  
69 cell weight on average, with the range of 6.0% to 29.5%. Coats et al. (2007) found activated sludge  
70 consortiums capable of synthesizing PHA at 10 to 25% when fed with primary solid fermented liquors.

71 Based on the results, they further proposed a sidestream process for both PHA production and  
72 wastewater treatment. In our recent studies, it was observed that PHA content in WAS withdrawn from  
73 a biological phosphorus removal reactor reached  $116 \pm 5$  mg per gram volatile suspended solids (VSS)  
74 by wasting sludge at 1 h of aeration (Wang et al., 2013).

75 The increase of PHA content in WAS might cause the changes of sludge characteristics, which  
76 further affected the subsequent anaerobic fermentation. To date, however, the influence of PHA on  
77 hydrogen production from WAS has never been reported. Some scientists suggested that the microbial  
78 cells would become more fragile with the increase of intracellular PHA (Budwill et al., 1992; Page and  
79 Cornish, 1993; Lee, 1996). Thus, it is presumed that the increased PHA in WAS might be beneficial to  
80 hydrogen production. If this hypothesis is clearly supported by experimental evidences, a new door  
81 may be opened for both wastewater treatment and hydrogen production from WAS. That is, organic  
82 pollutants in wastewaters are designed to be primarily removed via PHA accumulation, and then the  
83 WAS with high levels of PHA is used for hydrogen production, by which aeration cost in wastewater  
84 treatment process is saved, WAS amount is reduced, and hydrogen yield in WAS anaerobic fermentation  
85 is enhanced.

86 The aim of this paper was to provide a deep understanding of PHA associated with hydrogen  
87 production in dark fermentation. First, the influences of PHA level and composition in WAS on  
88 anaerobic hydrogen production were investigated in batch tests at pH 10. It was reported that alkaline  
89 conditions (especially pH 10) were beneficial to hydrogen production from WAS (Cai et al., 2004; Zhao  
90 et al., 2010), because this method not only enhanced the hydrolysis process but also inhibited the  
91 activities of hydrogen consuming bacteria (Zhao et al., 2010). Then, the reasons for PHA affecting the  
92 yield of hydrogen production were explored from the aspects of the microbial cell disruption, solubilized  
93 substrate hydrolysis, acidification of hydrolyzed products, fermentation type, mass balance, and  
94 activities of key enzymes.

## 95 **2. Materials and methods**

### 96 2.1 The source of sludges with different PHA contents

97 The following activated sludge bioreactors were performed to culture the sludges with different  
98 PHA contents as such characteristic sludges are not available now in real WWTPs. Seed sludge was

99 taken from the secondary sedimentation tank of a municipal WWTP in Shanghai, China, and was  
100 concurrently inoculated into five identical sequencing batch reactors with a working volume of 40 L  
101 each. All reactors were carried out the same and operated with four cycles (6 h per cycle) daily. Each  
102 cycle consisted of a 240 min aerobic period, a 55 min settling period, a 5 min decanting period, and a 60  
103 min idle period. During the aerobic period, air was supplied into all reactors at a flowrate of 20 L/min.  
104 To obtain sludges with different PHA contents, these reactors received 200, 400, 600, 800, and 1000  
105 mg/L of influent chemical oxygen demand (COD) concentrations (acetate was the sole carbon source),  
106 respectively. The concentrations of other nutrients in these synthetic media were the same and were  
107 presented as below (per liter): 0.1 g  $\text{NH}_4\text{Cl}$ , 0.04 g  $\text{KH}_2\text{PO}_4$ , 0.01 g  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , 0.005 g  $\text{CaCl}_2$ , and  
108 0.5mL of a trace element solution. The composition of trace element solution was documented in  
109 previous publication (Wang et al., 2008). After settling period 30 L supernatant was discharged from  
110 each reactor and replaced with 30 L respective medium during the first 6 min of subsequent aerobic  
111 period. About 5.7 L mixture was daily wasted from each reactor at 1.5 h aeration of the second cycle,  
112 thus the sludge retention time was maintained at approximately 7 d. After operation for about 60 d the  
113 five reactors reached stable, and then the wasted sludges were used in the following anaerobic  
114 fermentation tests. These wasted sludges were concentrated at 4 °C for 12 h before use.

## 115 2.2 The source of sludges with different PHA compositions

116 To obtain the sludges with different polyhydroxybutyrate (PHB)/polyhydroxyvalerate (PHV)  
117 fractions but similar total PHA amount, the following activated sludge bioreactors were conducted.  
118 Five identical reactors, as described above, were performed and received the synthetic media with  
119 different ratios of acetate to propionate but the same COD concentration. It is widely accepted that the  
120 composition of wastewater can affect PHB-PHV fraction (Li and Yu, 2011), and the pertinent increase of  
121 propionate concentration in wastewaters will increase the PHV fraction of PHA (Chen et al., 2004).  
122 These reactors received media with 800 mg/L of influent COD concentration, which were prepared with  
123 100% acetate, 85% acetate + 15% propionate, 70% acetate + 30% propionate, 55% acetate + 45%  
124 propionate, and 40% acetate + 60% propionate, respectively. Hereinafter, the sludges withdrawn from  
125 these reactors were defined as sludge-I, sludge-II, sludge-III, sludge-IV, and sludge V, respectively. On  
126 each day, about 5.7 L of sludge-I, sludge-II, sludge-III, sludge-IV, and sludge V mixtures were

127 respectively withdrawn from these reactors at proximately 100, 90, 80, 70, and 65 min of aeration,  
 128 because it was measured via batch tests that these sludges contained similar PHA content at these times.  
 129 All the other operations were the same as those depicted above. It took 54 d before these reactors  
 130 achieved stable characteristic of wasted sludge, and then the wasted sludges began to be used for  
 131 anaerobic fermentation trials. Before use, these wasted sludges were also concentrated at 4 °C for 12 h.

### 132 2.3 The effect of PHA content and composition on hydrogen production

133 The batch tests were performed in ten serum bottles with a working volume of 0.6 L each. Ten  
 134 serum bottles were divided into two groups with five in each. One group (group-I) was used to  
 135 evaluate the effect of PHA content on hydrogen production while the other group (group-II) was  
 136 employed to investigate the PHA composition's influence. Five serum bottles of group-I were  
 137 respectively fed with 300 mL of sludge-200, sludge-400, sludge-600, sludge-800, and sludge-1000, and  
 138 meanwhile the other five serum bottles (group-II) were fed with 300 mL of sludge-I, sludge-II,  
 139 sludge-III, sludge-IV, and sludge-V, respectively. The pH value of sludge mixtures in both group-I and  
 140 group-II was adjusted to 10 by adding 4 M hydrochloric acid (HCl) or 4 M sodium hydroxide (NaOH).  
 141 Oxygen in the bottles was removed from the headspace by nitrogen gas sparging for 30 s. After that,  
 142 all bottles were capped with rubber stoppers, sealed, and placed in an air-bath shaker (120 rpm) at  $37 \pm$   
 143  $1$  °C. During the whole fermentation period (10 d), the pH value in all bottles was controlled to  $10.0 \pm$   
 144  $0.1$  by adding 4 M HCl or 4 M NaOH with an automatic titrator. It should be noted that no extra  
 145 inoculum was added into these fermentation reactors, and therefore WAS was used for both substrate  
 146 and inoculum in this study. The total gas volume was determined via releasing the pressure in the  
 147 bottle using a glass syringe (300 mL) to equilibrate with the room pressure according to the method  
 148 documented in the literature (Owen et al., 1979). As the syringe was always in the bottle, the  
 149 accumulative volume was followed with time. The cumulative volume of hydrogen gas was calculated  
 150 by the following equation described in previous publications (Zhao et al., 2010; Oh et al., 2003).

$$151 \quad V_{H,i} = V_{H,i-1} + C_{H,i} \times V_{G,i} - C_{H,i-1} \times V_{G,i-1} \quad (1)$$

152 Where,  $V_{H,i}$  and  $V_{H,i-1}$  are respectively the cumulative volumes of hydrogen gas in the current (i)  
 153 and previous (i-1) time intervals,  $V_{G,i}$  and  $V_{G,i-1}$  are respectively the total gas volumes in the current and  
 154 previous time intervals, and  $C_{H,i}$  and  $C_{H,i-1}$  are the fractions of hydrogen gas measured by gas



155 chromatography in the current and previous time intervals, respectively.

#### 156 2.4 Long-term semi-continuous reactor operation for the analysis of key enzymes

157 Three typical sludges, which were respectively the low PHA sludge (sludge-200), the  
158 PHB-dominant sludge (sludge-I), and the PHV-dominant sludge (sludge-V), were selected to be fed to  
159 three semi-continuous reactors for the analysis of key enzymes relevant to hydrogen production. The  
160 three semi-continuous reactors were identical with a working volume of 0.6 L each. The three reactors  
161 received 400 mL of sludge-200, sludge-I, and sludge-V, respectively, and the fermentation conditions  
162 were the same as described above. According to the results obtained from the above batch tests the  
163 sludge retention time was maintained at 7, 4.5, and 5.5 d in the sludge-200, sludge-I, and sludge-V  
164 fermentation reactors, respectively. Every day, 57, 89, and 73 mL of fermentation mixtures were  
165 manually withdrawn from the sludge-200, sludge-I, and sludge-V fermentation reactors, respectively.  
166 Then, the same amounts of new sludge-200, sludge-I, and sludge-V were respectively added into these  
167 reactors, which resulted in the VSS loading rate of 1.73 Kg/ (m<sup>3</sup>·d) in the sludge-200 fermentation  
168 reactor, 2.63 Kg/ (m<sup>3</sup>·d) in the sludge-I fermentation reactor, and 2.22 Kg/ (m<sup>3</sup>·d) in the sludge-V  
169 fermentation reactor. After that, all reactors were sparged with nitrogen gas for 30 s to remove oxygen  
170 before they were re-capped and re-sealed. After operation for about 80 days, hydrogen yield reached  
171 stable, and then the assay of key enzyme activities was performed.

#### 172 2.5 Batch fermentation test of the effect of sludge PHA content on cell disruption

173 To eliminate the potential impact of microbial composition on cell disruption, the following batch  
174 fermentation test was conducted. Two fermentation reactors were performed. The sludges fed to the  
175 two fermentation reactors were withdrawn from the same activated sludge bioreactor (i.e., 1000 mg/L of  
176 influent COD fed reactor) but at different aerobic times. One was fed with the sludge wasted at 1.5 h  
177 aeration (i.e., sludge-1000) while the other was fed with the sludge withdrawn at the end of aerobiosis  
178 (this sludge was defined as sludge-1000-I). After concentrating at 4 °C for 12 h, it was measured that  
179 the sludge-1000-I contained 12280 ± 360 mg/L VSS, 14340 ± 340 mg/L total COD, 570 ± 31 mg/g VSS  
180 total protein, 224 ± 15 mg/g VSS total carbohydrate, and 31 ± 5 mg/g VSS PHA. The fermentation  
181 conditions were the same as those described in the section 2.3.

#### 182 2.6 Comparison of protein consumption and hydrogen production among non-PHA sludge,

183 intracellular-PHA sludge, and exogenous-PHA sludge

184 To evaluate the potential effect of PHA on non-PHA biomass during fermentation, we performed  
185 the following batch fermentation experiment. In this batch experiment, three fermentation reactors  
186 were carried out and were respectively fed with 300 mL non-PHA sludge, 300 mL intracellular-PHA  
187 sludge (i.e., sludge-1000), and 224 mL non-PHA sludge + 654 mg exogenous PHA (88% PHB and 12%  
188 PHV). The non-PHA sludge was collected from 1000 mg/L of influent COD fed reactor at 6 h of  
189 aeration, because it was found that PHA content was non-detectable after 6 h of aeration. The  
190 fermentation conditions were also the same as those depicted in the section 2.3. It took about 156, 144,  
191 and 108 h for these reactors to reach the maximal hydrogen production, respectively. At this time,  
192 PHA was non-detectable in all fermentation reactors.

### 193 2.7 Analytical methods

194 Hydrogen fraction in the generated gas was measured via a gastight syringe with 0.2 mL injection  
195 volume and a gas chromatograph (GC112A, China) equipped with a thermal conductivity detector and a  
196 4 mm × 32 m stainless column (Zhao et al., 2010; Xiao et al., 2014). The temperatures of the injection  
197 port, column, and detector were set at 40, 40, and 80 °C, respectively. Nitrogen was used as the carrier  
198 gas at a flowrate of 30 mL/min. The determinations of COD, VSS, and total suspended solids (TSS)  
199 were conducted in accordance with standard methods (APHA, 1998). The measurements of sludge  
200 PHA, protein, carbohydrate, lipid, and short-chain fatty acids (SCFA) were the same as depicted in  
201 previous publications (Wang et al., 2009; Yuan et al., 2006). Carbon, hydrogen, and nitrogen  
202 elemental compositions of fermentation substrates were analyzed by an elemental analyzer (Elemental  
203 Analyzer NA 2500). Microbial extracellular polymeric substances (EPS) containing loosely bound  
204 EPS and tightly bound EPS of activated sludge were measured according to the method documented in  
205 the literature (Mu et al., 2012). Molecular weight (Mw) distribution of the fermentation liquid was  
206 measured via gel-filtration chromatography analyzer (Shimadzu Co., Japan) according to the literature  
207 (Zhao and Chen, 2011). The activities of key hydrolytic enzymes (alpha-glucosidase and protease)  
208 were measured the same as described by Goel et al. (1998). One enzyme unit of alpha-glucosidase was  
209 defined to produce 1 μM of p-nitrophenol in one hour while one enzyme unit of protease was defined to  
210 hydrolyze 1 mg of azocasein per hour (Goel et al., 1998). The measurement of [FeFe] hydrogenase

211 activity was performed according to the method reported in the publications with minor revision (the  
212 debris was centrifuged at 15000 or 20000 g in the publications while it was centrifuged at 12000 g in  
213 this study due to the limit of available centrifuge), and one unit of [FeFe] hydrogenase was defined as  
214 the amount of hydrogenase evolving 1 M hydrogen gas from sodium dithionite reduced methylviologen  
215 per min (Khanna et al., 2011; Bai et al., 2012). Briefly, fermentation mixtures were harvested and  
216 washed for 3 times with 50 mM Tris-HCl (pH 7.5) containing 2 mM dithiothreitol and 1 mM  
217 phenylmethylsulfonyl fluoride. Then, the resuspended cells were sonicated at 20 kHz for 45 min in an  
218 ice bath to break down the cell structure. The debris was centrifuged at 12000g and 4 °C for 30 min,  
219 and the crude extracts in supernatant were obtained for [FeFe] hydrogenase activity measurement. The  
220 analysis was performed in a 5 mL plain tube. A volume of 100 µL crude extracts was added to start the  
221 reaction in the tube containing 50 mM Tris-HCl (pH 7.5), 25 mM sodium dithionite, and 1.5 mM  
222 methylviologen in a final volume of 2 mL. The assay mixtures were bubbled with argon to remove  
223 traces of dissolved oxygen before addition of the crude extracts. The reaction mixture was incubated in  
224 a shaker at 25 °C for 10 min.

## 225 2.8 Statistical analysis

226 All measurements were conducted in triplicate. An analysis of variance was used to evaluate the  
227 significance of results, and  $p < 0.05$  was considered to be statistically significant.

## 228 3. Results and discussion

### 229 3.1 The effect of PHA content and composition in WAS on hydrogen production

230 Table 1 presents the main characteristics of sludges with different PHA contents. It can be seen  
231 from Table 1 that protein, carbohydrate, and PHA are the top three organic compounds in these sludges.  
232 With the increase of PHA content, both protein and carbohydrate contents are decreased. Nevertheless,  
233 the increased PHA content does not result in significant increase of total COD concentration ( $p > 0.05$ ).  
234 In addition, PHB is found to be the dominant fraction of PHA in all sludges, and the percentages of PHB,  
235 PHV, and poly-3-hydroxy-2- methylvalerate (PH2MV) are almost the same among these sludges.

236 Figure 1a shows the time curve of cumulative hydrogen production using sludges with different  
237 PHA contents. It can be seen that the behaviour of hydrogen production at different PHA contents was  
238 similar. The volume of generated hydrogen first increased with the increase of fermentation time and

239 then kept almost constant in the remainder of fermentation period. No hydrogen consumption was  
240 observed in all the fermentation reactors due to the strong alkaline condition controlled in these reactors  
241 (pH 10). It was proven that constant pH 10 could effectively inhibit the activities of methanogens and  
242 acetobacteria (Zhao et al., 2010). All these observations made in Figure 1a were similar to those  
243 reported in the literature at constant pH 10 (Zhao et al., 2010). It can be also found in Figure 1a that  
244 the maximal hydrogen yield was affected by PHA content. With the increase of PHA content,  
245 hydrogen yield increased. For instance, the hydrogen production in the Sludge-200 (with PHA content  
246 of  $25 \pm 3$  mg/g VSS, Table 1) fed reactor increased gradually with fermentation time during the initial  
247 156 h, and no significant increase was found after that time ( $p > 0.05$ ). At time of 156 h the hydrogen  
248 generation was 26.5 mL/g VSS, which was in accord with the datum reported in the literature (Zhao et  
249 al., 2010). Nevertheless, the maximal hydrogen production in the Sludge-1000 (with PHA content of  
250  $178 \pm 11$  mg/g VSS, Table 1) fed reactor was observed at fermentation time of 108 h, and the  
251 corresponding hydrogen production was 58.7 mL/g VSS, which was 2.2-fold higher than that detected in  
252 the Sludge-200 fed reactor. It should be emphasized that the optimal fermentation time in the lowest  
253 PHA sludge (i.e., Sludge-200) was 156 h whereas this value was 108 h in the highest PHA one (i.e.,  
254 Sludge-1000). Clearly, the sludge containing higher PHA produced more hydrogen but required less  
255 fermentation time.

256 PHA contains several compositions, it is therefore necessary to investigate the effect of PHA  
257 composition on hydrogen production to gain a comprehensive understanding of PHA associated with  
258 hydrogen production. However, since PHB and PHV are the main compositions of PHA in activated  
259 sludge involved in WWTPs with their contents usually above 90%, we only focus on these two  
260 compositions in this study. Table 2 outlined the main characteristics of sludges with different PHA  
261 compositions. From Table 2, it can be found that the main difference among the five sludges was PHA  
262 composition. With the increase of influent propionate ratio, PHV fraction increased. For example,  
263 PHB was the dominant composition of PHA with its percent up to  $79.1 \pm 8.6\%$  of PHA (i.e., sludge-I)  
264 when influent COD was prepared with 100% acetate. When the reactor received 40% acetate + 60%  
265 propionate, PHV was shifted to be the main composition ( $82.8 \pm 7.7\%$  of PHA, sludge-V). Except for  
266 the PHB and PHV fractions, all other characteristics of these sludges were almost the same. Thus these

267 sludges can be employed to evaluate the influence of PHA composition on hydrogen production.

268 Figure 1b illustrates the effect of PHB/PHV fraction on hydrogen production. Although these  
269 sludges contained the same level of total PHA ( $127 \pm 15 \sim 135 \pm 9$  mg/g VSS, Table 2), hydrogen yield  
270 from them were not the same. When the dominant composition of PHA shifted from PHB ( $79.1 \pm$   
271  $8.6\%$  of PHA, Sludge-I) to PHV ( $82.8 \pm 7.7\%$  of PHA, Sludge-V) gradually, the amount of maximal  
272 hydrogen production decreased from 51.2 to 41.1 mL/g VSS. Meanwhile, the optimal fermentation  
273 time increased from 108 to 132 h. It was evident that PHB was more beneficial to hydrogen  
274 production, as compared with PHV. It should also be noted that hydrogen production in the  
275 PHV-dominant sludge (i.e., Sludge-V) fed reactor was still greater than that in the low-PHA sludge (i.e.,  
276 Sludge-200) fed reactor, but the required fermentation time was lower. All the above results showed  
277 that the intracellular polymer PHA could enhance hydrogen production from WAS dark fermentation,  
278 and the different compositions of PHA could cause different effects on hydrogen generation. The  
279 mechanisms of PHA content and composition affecting hydrogen production will be explored in the  
280 following text.

### 281 3.2 Mechanisms of intracellular PHA affecting hydrogen production

282 Besides protein and carbohydrate, as seen in Table 1 and 2, PHA was also one of the primary  
283 organic compounds in these sludges tested in this study, and the changes of its content and composition  
284 were clearly observed to affect hydrogen production (Figure 1). Furthermore, we found that more than  
285 94% of PHA in all tested sludges was degraded during the initial 3 d of fermentation time (Figure S2,  
286 Supporting Information). Thus it was necessary to investigate how hydrogen production was affected  
287 by PHA. During sludge anaerobic digestion, the following four steps are usually included:  
288 solubilization of sludge, hydrolysis of solubilized substrates, acidification of hydrolyzed products, and  
289 methane production (Zhao et al., 2010). Hydrogen is mainly generated in the acidification step.  
290 Since no hydrogen consumption was observed in all reactors, PHA's influence on hydrogen production  
291 was mainly focused on the three former steps.

292 Several researchers showed that PHA as an exogenous plastic material could be completely  
293 decomposed under anaerobic conditions (Reischwitz et al., 1998; Chen and Wang, 2002). However,  
294 compared with the exogenous PHA materials, PHA is an intracellular polymer in this study. This

295 indicates that the microbial cell needs to be disrupted before it can be further degraded. Therefore, its  
296 potential influence on cell disruption was investigated first. In the literature, cell disruption is usually  
297 estimated by the determination of intracellular substrate release (Wang et al., 2013; Tam et al., 2012).  
298 In this study, the variations of soluble protein and carbohydrate were applied to indicate cell breakage,  
299 because the sludges used in this study contained almost the same protein and carbohydrate contents in  
300 EPS (Table 1 and 2). It can be seen from Figure 2a that both the ratios of soluble protein to total  
301 protein and soluble carbohydrate to total carbohydrate at 1 d of fermentation time increased with the  
302 increase of intracellular PHA content. When PHA content increased from  $25 \pm 3$  (Sludge-200) to  $178$   
303  $\pm 11$  (Sludge-1000) mg/g VSS, the ratio of soluble protein (carbohydrate) to total protein (carbohydrate)  
304 increased from  $14.6 \pm 0.9\%$  ( $4.7 \pm 0.4\%$ ) to  $30.5 \pm 2.4\%$  ( $16.2 \pm 1.9\%$ ). PHA has associated proteins in  
305 its biogenesis and degradation, and the degradation of PHA will increase protein solubilization, which  
306 thereby affecting the assessment of cell disruption. Thus, we further determined the VSS reduction  
307 ratio among these reactors at 1 d of fermentation, and the results are shown in Figure 2b. It was  
308 observed that the VSS reduction in the high PHA contained sludge was also greater than that in the low  
309 PHA sludge. Since both the decomposition of PHA and solubilization of other cell compositions will  
310 contribute the VSS reduction, it is necessary to figure out whether the increased VSS reduction in the  
311 high PHA contained sludge is caused by the PHA decomposition. Based on the following equation, it  
312 was further found that except for PHA the solubilization ratio of other cell compositions was 19.9, 21.5,  
313 24.1, 26.8, and 32.4% in the sludge-200, sludge 400, sludge-600, sludge-800, and sludge-1000 reactors,  
314 respectively.

$$\begin{aligned} 315 \quad & \text{Total VSS (g)} \times \text{the measured VSS reduction ratio (\%)} = \text{PHA reduction (g)} + \text{non-PHA cell} \\ 316 \quad & \text{composition (g)} \times \text{the solubilization ratio of non-PHA cell composition (\%)} \end{aligned} \quad (2)$$

317 Where, total VSS is calculated by initial VSS concentration (Table 1)  $\times$  0.3 L, the measured VSS  
318 reduction ratio is shown in Figure 2b, PHA reduction is calculated by initial PHA (according to Table 1)  
319 – remnant PHA (according to Figure 2b and S2), non-PHA cell composition is calculated by total VSS -  
320 initial PHA.

321 It should be noted that the sludges used in above fermentation tests might have different microbial  
322 compositions due to the fact that they were wasted from different activated sludge bioreactors with

323 different influent COD concentrations. Different organisms may have different disruption thresholds.  
324 To eliminate this potential influence, a batch fermentation experiment was carried out using the sludges  
325 withdrawn from the same activated sludge bioreactor but at different aerobic times (Table S1,  
326 Supporting Information). The results showed that the sludge with high PHA level (i.e., sludge-1000)  
327 exhibited higher soluble protein (carbohydrate) to total protein (carbohydrate) ratio, VSS reduction, and  
328 hydrogen production than the low PHA one (i.e., sludge-1000-I).

329 All these results showed that the increased PHA content accelerated cell solubilization, which  
330 thereby caused more soluble substrates for subsequent hydrolysis and acidification stages (Figure S3a,  
331 Supporting Information). Some researchers reported that the microbial cell became more fragile with  
332 the increased intracellular PHA level (Budwill et al., 1992; Page and Cornish, 1993; Lee, 1996), it can be  
333 understood that the increased PHA could accelerate the solubilization of PHA contained biomass.  
334 However, as all fermentation sludges used in this study contained PHA-biomass and non-PHA biomass,  
335 it is unclear whether the increased PHA is beneficial to the cell solubilization of non-PHA biomass  
336 according to the above data, which will be further analyzed below. Figure S3b and S4 (Supporting  
337 Information) present the soluble COD and the ratio of soluble protein (carbohydrate) to total protein  
338 (carbohydrate) in the fermentation systems fed with different PHA compositions at 1 d of fermentation  
339 time. Further analysis found that compared with the PHB dominant sludge (i.e., sludge-I), the  
340 increased PHV fraction did not significantly affect the ratio of soluble protein (carbohydrate) to total  
341 protein (carbohydrate), VSS reduction, and soluble COD concentration ( $p > 0.05$ , Table S2, Supporting  
342 Information), which suggested that PHA composition (i.e., PHB/PHV fraction) caused insignificant  
343 impact on sludge solubilization.

344 It is known that before the solubilized substrates can be directly utilized to produce SCFA and  
345 hydrogen, the solubilized substrates with large Mw require to be hydrolyzed. The hydrolysis rate is  
346 closely relevant to the yield and fermentation time of hydrogen production. Thus we further  
347 investigated the effect of PHA on hydrolysis step. In this study, PHA content was observed to decrease  
348 with fermentation time rapidly (Figure S2, Supporting Information). Although PHA degradation rate  
349 was affected by PHB/PHV fraction, more than 94% of sludge PHA polymer was degraded in the initial  
350 72 h of fermentation time no matter what the original PHA content and composition in sludge were

351 (Figure S2, Supporting Information). Using bovine serum albumin (model protein compound with  
352 average Mw of 67000) and dextran (model polysaccharide compound with average Mw of 23800) as  
353 model substrates of protein and carbohydrate, Zhao et al. (2010) observed that only 54.5% of protein  
354 and 84% of carbohydrate were respectively decomposed under alkaline anaerobic fermentation (pH 10)  
355 even in 84 h of fermentation time. It seems that the anaerobic hydrolysis rate of PHA might be faster  
356 than that of protein and carbohydrate, which are the main compositions of traditional cell.

357 This hypothesis can be strongly supported by the Mw distribution of solubilized substrate shown in  
358 Figure 3. It can be clearly observed that solubilized substrate with low Mw were in the sequence of the  
359 PHB-dominant sludge (i.e., Sludge-I) > the PHV-dominant sludge (i.e., Sludge-V) > the low PHA  
360 sludge (i.e., Sludge-200) at 2 d of fermentation time (Figure 3a). Further investigations revealed that  
361 the percentages of small soluble substrates (Mw < 1000) at this fermentation time showed well positive  
362 correlation with PHA level ( $R^2 = 0.972$ , Figure 3b) but exhibited negative correlation with PHV fraction  
363 ( $R^2 = 0.9124$ , Figure 3c). Considering that the SCFA generation at this fermentation time was found to  
364 be unaffected by both sludge PHA content and composition (Table S3, Supporting Information), it can  
365 be concluded that the increased PHA content benefited the hydrolysis process of solubilized substrates,  
366 and PHB was more beneficial to hydrolysis step, as compared with PHV. Since the sludge with higher  
367 PHA content had lower protein and carbohydrate (Table 1), it can be further inferred that PHA had faster  
368 anaerobic hydrolysis rate than protein and carbohydrate.

369 It is reported that both PHB and PHV can be anaerobically converted to SCFA under anaerobic  
370 conditions through a series of biochemical degradations, and the metabolic pathways are summarized in  
371 Figure S5 (Supporting Information). PHB and PHV are first degraded to their monomers  
372 3-hydroxybutyrate and 3-hydroxyvalerate by depolymerases, respectively. Then, 3-hydroxybutyrate  
373 and 3-hydroxyvalerate are activated by CoA transfer and generated the resultant 3-hydroxybutyryl-CoA  
374 and 3-hydroxyvaleryl-CoA, which are further bio-converted to acetyl-CoA, propionyl-CoA,  
375 butyryl-CoA, and valeryl-CoA. Finally, acetate, propionate, butyrate, and valerate are produced in the  
376 acidification process. As hydrogen is generated in acidification step concurrently, accelerating  
377 hydrolysis process by PHA indicates that more hydrolyzed substrates can be provided for subsequent  
378 hydrogen production. Meanwhile, fermentation time will be also saved if hydrolysis rate is accelerated.



379 It can be therefore understood that the hydrogen production was in the sequence of the PHB-dominant  
380 sludge > the PHV-dominant sludge > the low PHA sludge but the required fermentation time was on the  
381 opposite sequence, because PHB was more beneficial to hydrolysis process than PHV while both PHB  
382 and PHV had faster anaerobic hydrolysis rate than other main cell compositions, such as protein and  
383 carbohydrate.

384 Hydrogen is primarily produced in acidification step, and the fermentation type is also reported to  
385 affect hydrogen generation (Khanal et al., 2004; Li et al., 2009). Thus, we also compared with the total  
386 and individual SCFA production among these reactors. Table 3 illustrates the total and individual  
387 SCFA produced from the three typical reactors at the time of maximal hydrogen production, and the  
388 detailed SCFA information in other reactors is listed in Table S4 (Supporting Information). As seen  
389 from Table 3, the fraction of individual SCFA between the low PHA sludge ( $25 \pm 3$  mg/g VSS, Table 1)  
390 and PHB-dominant sludge ( $132 \pm 11$  mg/g VSS, Table 2) fed reactors was similar. Acetic acid,  
391 propionic acid, and isobutyric acid were the top three SCFA. The total SCFA production in the  
392 PHB-dominant sludge, however, was much higher than that in the low PHA sludge fed reactor ( $375.0 \pm$   
393  $17.1$  vs  $194.3 \pm 11.1$  mg COD/g VSS). With the increase of PHA content the total SCFA yield  
394 increased. Similar results were also observed in other sludges with different PHA levels but similar  
395 PHA composition (Table S4, Supporting Information). Therefore, improvement of acidification of  
396 hydrolyzed products was another reason for the increased PHA sludge showing greater hydrogen  
397 production.

398 From Table 3, it can be also observed that even under the same total PHA level (Table 2) both  
399 individual SCFA fraction and total SCFA production varied significantly ( $p < 0.05$ ) when PHA  
400 composition changed. The dominant PHA monomer shifted from PHB (sludge-I, Table 2) to PHV  
401 (sludge-V, Table 2), the top three SCFA were propionic acid, acetate acid, and n-valeric acid instead of  
402 acetic acid, propionic acid, and isobutyric acid. It was reported that the fermentation type also affected  
403 hydrogen production, and the higher the acetic or the lower the propionic were generated, the greater  
404 hydrogen was produced (Khanal et al., 2004; Li et al., 2009). When 1 mol monomeric  
405 3-hydroxybutyrate (or 3-hydroxyvalerate) is fermented, 1 mol NADH and 1 mol  $H^+$  will be generated in  
406 the step of 3-hydroxybutyryl-CoA (or 3-hydroxyvaleryl-CoA) degradation, which can be further formed

407 1 mol H<sub>2</sub> (Janssen and Schink, 1993; de María and Domínguez, 2010). That is, if 1 mol-C (or 1 g) of  
408 PHB and 1 mol-C (or 1 g) of PHV are respectively fermented, the theoretical hydrogen production from  
409 PHB will be higher than that from PHV (1 mol-C: 0.25 vs 0.2 mol H<sub>2</sub>; 1g: 11.6 vs 10 mmol H<sub>2</sub>).  
410 Moreover, the PHV-dominant sludge produced less total SCFA than the PHB-dominant sludge. Similar  
411 observation was made in other sludges with different PHB/PHV fractions (Table S4, Supporting  
412 Information). Thus, it can be understood that the PHV-dominant sludge produced less hydrogen than  
413 the PHB-dominant sludge. From the metabolic pathways for anaerobically converting PHB and PHV  
414 to SCFA presented in Figure S5 (Supporting Information), we can see that acetic acid and butyric acid  
415 are the resultant of PHB fermentation whereas acetic acid, propionic acid, and valeric acid are the  
416 resultant of PHV fermentation, which might be the reason for more propionic acid generated in the  
417 PHV-dominant sludge fed reactor.

418 According to the above analysis, it is clear that biomass containing higher PHA is easier to be  
419 disrupted, which is thereby beneficial to dark fermentative hydrogen production. After the disruption  
420 of PHA contained biomass, PHA will be released to the fermentation system. Thus, one might want to  
421 know whether the released PHA affects the fermentation of non-PHA biomass. To evaluate this  
422 potential impact, one batch fermentation test using non-PHA sludge, intracellular-PHA sludge, and  
423 non-PHA sludge + exogenous PHA (with equivalent PHA content to intracellular-PHA sludge) was  
424 carried out, and the results are shown in Table 4. It can be seen that the protein consumption ratio,  
425 SCFA production, and hydrogen production in the exogenous-PHA sludge reactor were much higher  
426 than those in the non-PHA sludge reactor. Since the exogenous-PHA reactor was fed with the same  
427 sludge (i.e., non-PHA sludge) as the non-PHA sludge reactor, and the added exogenous PHA did not  
428 contain any protein, the increased protein consumption rate in the exogenous-PHA reactor indicated that  
429 the presence of PHA benefited the fermentation of non-PHA biomass.

430 To obtain a deeper understanding, COD mass balance among these fermentation systems was  
431 assessed, and the results are displayed in Figure 4. It was found that the COD ratios of both SCFA  
432 and hydrogen in the exogenous-PHA reactor were higher than those in the non-PHA reactor, which were  
433 consistent with the lower VSS and soluble protein remained in the fermentation system. It should be  
434 highlighted that the average COD ratio of soluble protein in the exogenous-PHA reactor was only 61.1%

435 of that in the non-PHA reactor (8.0% vs 13.1% of total COD). Although the average COD ratio of  
436 VSS remained in the former was also lower as compared with that in the latter (49.3% vs 63.0%), VSS  
437 reduction from the aspect of sludge solubilization was approximately the same in the two fermentation  
438 systems. In the non-PHA reactor, all VSS reduction was ascribed to sludge solubilization, and the  
439 average ratio of VSS reduction was 37.0% ( $1 - 63.0\% = 37.0\%$ ). In the exogenous-PHA reactor,  
440 however, both the exogenous PHA decomposition and sludge solubilization contributed to VSS  
441 reduction. Considering that the exogenous PHA, which was completely decomposed during  
442 fermentation, accounted for about 22.3% (as COD) of total VSS  $[(0.576 \text{ g PHB} \times 1.67 + 0.078 \text{ g PHV}$   
443  $\times 1.92) / (0.576 \text{ g PHB} \times 1.67 + 0.078 \text{ g PHV} \times 1.92 + 12.16 \text{ g VSS} / \text{L} \times 0.224 \text{ L} \times 1.42) = 0.223]$ , the  
444 VSS reduction come from sludge degradation should be 36.6%  $\{[(1 - 49.3\%) - 22.3\%] / (1 - 22.3\%) =$   
445  $0.366\}$ . These results indicated that the presence of PHA could not enhance non-PHA sludge  
446 solubilization but promote the soluble protein conversion. From Table 4, it was observed that the C/N  
447 ratio of exogenous-PHA system was 9.39, which was higher than that of non-PHA system (7.08). It  
448 was reported that pertinent increase of C/N ratio benefited the conversion of protein and the production  
449 of SCFA and hydrogen (Chen et al., 2012; Feng et al., 2009). Therefore, the increased C/N ratio may  
450 be the main reason for the presence of PHA enhancing soluble protein conversion and hydrogen  
451 production.

452 It should be also noted that the intracellular-PHA sludge reactor showed higher hydrogen yield,  
453 SCFA production, and VSS reduction as compared with the exogenous-PHA reactor, though they had the  
454 similar PHA content and C/N ratio (Table 4 and Figure 4). As demonstrated above, intracellular PHA  
455 could accelerate sludge solubilization whereas exogenous PHA could not enhance it, which may be the  
456 main reason for the increased VSS reduction and hydrogen and SCFA production in the  
457 intracellular-PHA sludge reactor. Another possibility might be that exogenous PHA would tend to be  
458 crystallized, which would be more resistant to enzymatic attack. Moreover, due to the limit of our  
459 available exogenous PHA, the compositions of exogenous PHA (88.0% PHB + 12.0% PHV) and  
460 intracellular PHA (78.3% PHB + 20.2% PHV + 1.5% PH<sub>2</sub>MV) were not completely the same, which  
461 might also affect hydrogen yield and SCFA production. The results of COD mass balance analysis also  
462 showed that with the increase of sludge PHA content, the COD ratios of produced hydrogen and SCFA

463 increased (Figure S6a, Supporting Information). When the dominant composition of PHA shifted from  
464 PHB to PHV, both ratios decreased (Figure S6b, Supporting Information).

465 The production of hydrogen during sludge anaerobic fermentation at pH 10 is primarily related to  
466 sludge hydrolysis and acidification because both methanogens and acetobacteria are inhibited (Zhao et  
467 al., 2010). Determination of enzyme activities is an alternative method to evaluate microbial activities  
468 (Nybroe et al., 1992). Thus, the activities of key hydrolytic and hydrogen-forming enzymes were  
469 finally measured to reflect the microbial activities of key hydrolytic and hydrogen-producing microbes.  
470 Protease and  $\alpha$ -glucosidase are the key enzymes for protein and carbohydrate hydrolysis, respectively,  
471 while [FeFe] hydrogenase is the key enzyme in the biochemical metabolism for the production of  
472 molecular hydrogen (Goel et al., 1998; Khanna et al., 2011; Bai et al., 2012). As seen in Figure 5, the  
473 Sludge-I (i.e., PHB-dominant sludge) fed reactor had the highest activities of both key hydrolytic  
474 enzymes and hydrogen-forming enzyme while the Sludge-200 (i.e., low PHA sludge) fed reactor had the  
475 lowest ones. The activities of protease,  $\alpha$ -glucosidase, and [FeFe] hydrogenase in the Sludge-V (i.e.,  
476 PHV dominant sludge) fed reactor were lower than those in the Sludge-I (i.e., PHB-dominant sludge)  
477 fed reactor but much higher than those in the Sludge-200 (i.e., low PHA sludge) fed reactor. All these  
478 observations were consistent with the detected order of hydrogen production.

### 479 3.3 Implications for wastewater and WAS treatments

480 This study reveals for the first time that enhanced hydrogen production can be achieved from  
481 increased PHA sludge. The findings obtained in this work have important implications to sludge  
482 fermentation systems for hydrogen production. Although numerous studies have been performed in the  
483 field, the strategy to enhance hydrogen generation from the aspect of improving the self-characteristic of  
484 sludge has never been reported. Thus, the findings of this study can provide a new solution to enhance  
485 hydrogen generation. More importantly, a new door may be opened for both wastewater treatment and  
486 hydrogen production from WAS based on the findings achieved in this work. In the conventional  
487 WWTPs, wastewaters are first influent into the anaerobic phase, where carbon sources (e.g., acetate)  
488 will be taken up and further converted to intracellular PHA. In the subsequent aerobic phase, the  
489 accumulated PHA will enter into the tricarboxylic acid cycle and be oxidized to provide carbon and  
490 energy (oxygen is consumed and CO<sub>2</sub> is formed) for cell growth and nutrient removal (Figure S1,

491 Supporting Information). Therefore, if the wasted WAS contained higher PHA by either process  
492 improvement or operation optimization as mentioned in the “Introduction” section, less PHA will be  
493 oxidized in the bioreactors of WWTPs. As a result, less oxygen is required (i.e., aeration is saving),  
494 less CO<sub>2</sub> is formed, and less cell growth is also occurred. Furthermore, the increase of sludge PHA  
495 level can accelerate cell solubilization and solubilized substrate hydrolysis processes, as demonstrated in  
496 this work, which thereby enhances the subsequent hydrogen production from WAS.

497 Several strategies have been verified to promote hydrogen production from WAS, such as ozone or  
498 ultrasound pretreatment (Yang et al., 2012), heat pretreatment (Assawamongkholsiri et al., 2013), acid or  
499 alkaline pretreatment (Cai et al., 2004; Assawamongkholsiri et al., 2013), and co-digestion of WAS and  
500 other biosolids (Zhou et al., 2013; Kim et al., 2012; Chen et al., 2013; Liu et al., 2013). However, these  
501 strategies require either consumption of energy, addition of chemicals, or transportation of other  
502 substrates. In comparison, the PHA based technology developed on the basis of the findings in this  
503 work does not have these limitations, because this strategy enhances hydrogen production by improving  
504 the self-characteristic of sludge in wastewater treatment step. It should be noted that this PHA  
505 accumulation based method can be integrated with other strategies (e.g., alkaline condition applied in  
506 this study). Thus, hydrogen production may be further enhanced if we combine the PHA accumulation  
507 based method with other strategies, such as heat, ozone or ultrasound pretreatment and co-fermentation  
508 substrate optimization, which remains to be investigated in future.

#### 509 **4. Conclusions**

510 This study evaluated the influences of intracellular PHA level and composition on hydrogen  
511 production from anaerobic WAS fermentation. The results showed that the sludge containing higher  
512 PHA not only promoted hydrogen yield but also shortened fermentation time. Compared with PHV,  
513 PHB was more beneficial to hydrogen production. The increased PHA content accelerated cell  
514 solubilization and solubilized substrate hydrolysis processes and enhanced soluble protein conversion.  
515 Compared with the PHB-dominant sludge, the increased PHV fraction not only slowed the hydrolysis  
516 process but also caused more propionic acid production, with less theoretical hydrogen generation in this  
517 fermentation type. Enzyme analyses further showed that both the key hydrolytic enzyme activities and  
518 hydrogen-forming enzyme activities were in the sequence of the PHB-dominant sludge > the

519 PHV-dominant sludge > the low PHA sludge, which was consistent with the observed order of hydrogen  
520 production.

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### 526 **Supporting Information**

527 This file contains Tables S1 –S4, and Figures S1 – S6.

### 528 **References**

529 APHA (American Public Health Association), 1998. Standard Methods for the Examination of Water  
530 and Wastewater, 20th ed. Washington, DC, USA.

531 Assawamongkholisiri, T., Reungsang, A., Pattra, S., 2013. Effect of acid, heat and combined acid-heat  
532 pretreatments of anaerobic sludge on hydrogen production by anaerobic mixed cultures. Int. J.  
533 Hydrogen Energ. 38, 6146-6153.

534 Bai, L., Wu, X., Jiang, L., Liu, J., Long, M., 2012. Hydrogen production by over-expression of  
535 hydrogenase subunit in oxygen-tolerant *Klebsiella oxytoca* HP1. Int. J. Hydrogen Energ. 37,  
536 13227-13233.

537 Budwill, K., Fedorak, P.M., Page, W.J., 1992. Methanogenic degradation of poly(3-hydroxyalkanoates).  
538 Appl. Environ. Microb. 58, 1398-1401.

539 Cai, M.L., Liu, J.X., Wei, Y.S., 2004. Enhanced biohydrogen production from sewage sludge with  
540 alkaline pretreatment. Environ. Sci. Technol. 38, 3195-3202.

541 Chen, L.J., Wang, M., 2002. Production and evaluation of biodegradable composites based on  
542 PHB–PHV copolymer. Biomaterials 23, 2631-2639.

543 Chen, Y., Xiao, N., Zhao, Y., Mu, H., 2012. Enhancement of hydrogen production during waste  
544 activated sludge anaerobic fermentation by carbohydrate substrate addition and pH control. Bioresour.  
545 Technol. 114, 349-356.

546 Chen, Y., Randall, A.A., McCue, T., 2004. The efficiency of enhanced biological phosphorus removal

- 547 from real wastewater affected by different ratios of acetic to propionic acid. *Water Res.* 38, 27-36.
- 548 Coats, E.R., Loge, F.J., Wolcott, M.P., Englund, K., McDonald, A.G., 2007. Synthesis of  
549 polyhydroxyalkanoates in municipal wastewater treatment. *Water Environ. Res.* 79, 2396-2403.
- 550 de María, C.G., Domínguez, K.B.H., 2010. Simultaneous kinetic determination of 3-hydroxybutyrate  
551 and 3-hydroxyvalerate in biopolymer degradation processes. *Talanta* 80, 1436-1440.
- 552 Feng, L., Chen, Y., Zheng, X., 2009. Enhancement of waste activated sludge protein conversion and  
553 volatile fatty acids accumulation during waste activated sludge anaerobic fermentation by  
554 carbohydrate substrate addition: The effect of pH. *Environ. Sci. Technol.* 43, 4373-4380.
- 555 Goel, R., Mino, T., Satoh, H., Matsuo, T., 1998. Enzyme activities under anaerobic and aerobic  
556 condition in activated sludge sequencing batch reactor. *Water Res.* 32, 2081-2088.
- 557 Gioannis, G.D., Muntoni, A., Poletini, A., Pomi, R., 2013. A review of dark fermentative hydrogen  
558 production from biodegradable municipal waste fractions. *Waste Manage.* 33, 1345-1361.
- 559 Janssen, P.H., Schink, B., 1993. Pathway of anaerobic poly- $\beta$ -hydroxybutyrate degradation by  
560 *Ilyobacter delafieldii*. *Biodegradation* 3, 179-185.
- 561 Jung, K.W., Kim, D.H., Kim, S.H., Shin, H.S., 2011. Bioreactor design for continuous dark fermentative  
562 hydrogen production. *Bioresour. Technol.* 102, 8612-8620.
- 563 Khanal, S.K., Chen, W.H., Li, L., Sung, S., 2004. Biological hydrogen production: effects of pH and  
564 intermediate products. *Int. J. Hydrogen Energy* 29, 1123-1131.
- 565 Khanna, N., Dasgupta, C.N., Mishra, P., Das, D., 2011. Homologous overexpression of [FeFe]  
566 hydrogenase in *Enterobacter cloacae* IIT-BT 08 to enhance hydrogen gas production from cheese  
567 whey. *Int. J. Hydrogen Energy* 36, 15573-15582.
- 568 Kim, S., Choi, K., Kim, J., Chung, J., 2013. Biological hydrogen production by anaerobic digestion of  
569 food waste and sewage sludge treated using various pretreatment technologies. *Biodegradation* 24,  
570 753-764.
- 571 Kim, M., Yang, Y., Morikawa-sakura, M.S., Wang, Q., Lee, M.V., Lee, D., Feng, C., Zhou, Y., Zhang,  
572 Z., 2012. Hydrogen production by anaerobic co-digestion of rice straw and sewage sludge. *Int. J.*  
573 *Hydrogen Energy* 37, 3142-3149.
- 574 Lee, S.Y., 1996. Bacterial Polyhydroxyalkanoates. *Biotechnol. Bioeng.* 49, 1-14.

- 575 Li, J.Z., Zheng, G.C., He, J.G., Chang, S., Qin, Z., 2009. Hydrogen producing capability of anaerobic  
576 activated sludge in three types of fermentations in a continuous stirred-tank reactor. *Biotechnol. Adv.*  
577 *27*, 573-577.
- 578 Li, W., Yu, H., 2011. From wastewater to bioenergy and biochemicals via two-stage bioconversion  
579 processes: A future paradigm. *Biotechnol. Adv.* *29*, 972-982.
- 580 Lingampalli, S.R., Gautam, U.K., Rao, C.N.R., 2013. Highly efficient photocatalytic hydrogen  
581 generation by solutionprocessed ZnO/Pt/CdS, ZnO/Pt/Cd<sub>1-x</sub>Zn<sub>x</sub>S and ZnO/Pt/CdS<sub>1-x</sub>Sex hybrid  
582 nanostructures. *Energy Environ. Sci.* *6*, 3589-3594.
- 583 Liu, X., Li, R., Ji, M., Han, L., 2013. Hydrogen and methane production by co-digestion of waste  
584 activated sludge and food waste in the two-stage fermentation process: Substrate conversion and  
585 energy yield. *Bioresour. Technol.* *146*, 317-323.
- 586 Mu, H., Zheng, X., Chen, Y., Chen, H., Liu, K., 2012. Response of anaerobic granular sludge to a shock  
587 load of zinc oxide nanoparticles during biological wastewater treatment. *Environ. Sci. Technol.* *46*,  
588 5997-6003.
- 589 Nybroe, O., Jorgensen, P.E., Henze, M., 1992. Enzyme activities in waste water and activated sludge.  
590 *Water Res.* *26*, 579-584.
- 591 Oh, S.E., Van Ginkel, S., Logan, B.E., 2003. The relative effectiveness of pH control and heat treatment  
592 for enhancing biohydrogen gas production. *Environ. Sci. Technol.* *37*, 5186-5190.
- 593 Owen, W.F., Stuckey, D.C., Healy, J.B., Young, L.Y., McCarty, P.L., 1979. Bioassay for monitoring  
594 biochemical methane potential and anaerobic toxicity. *Water Res.* *13*, 485-492.
- 595 Page, W.J., Cornish, A., 1993. Growth of *Azotobacter vinelandii* UWD in fish peptone medium and  
596 simplified extraction of poly- $\beta$ -hydroxybutyrate. *Appl. Environ. Microb.* *59*, 4236-4244.
- 597 Reischwitz, A., Stoppok, E., Buchholz, K., 1998. Anaerobic degradation of poly-3-hydroxybutyrate and  
598 poly-3-hydroxybutyrate-co-3-hydroxyvalerate. *Biodegradation* *8*, 313-319.
- 599 Saady, N.M.C., 2013. Homoacetogenesis during hydrogen production by mixed cultures dark  
600 fermentation: Unresolved challenge. *Int. J. Hydrogen Energ.* *38*, 13172-13191.
- 601 Takabatake, H., Satoh, H., Mino, T., Matsuo, T., 2002. PHA (polyhydroxyalkanoate) production  
602 potential of activated sludge treating wastewater. *Water Sci. Technol.* *45*, 119-126.



- 603 Tam, Y.J., Allaudin, Z.N., Lila, M.A.M., Bahaman, A.R., Tan, J.S., Rezaei, M.A., 2012. Enhanced cell  
604 disruption strategy in the release of recombinant hepatitis B surface antigen from *Pichia pastoris* using  
605 response surface methodology. *BMC Biotechnol.* 12, 70.
- 606 Wang, D., Chen, Y., Zheng, X., Li, X., Feng, L., 2013. Short-chain fatty acid production from different  
607 biological phosphorus removal sludges: The influences of PHA and gram-staining bacteria. *Environ.*  
608 *Sci. Technol.* 47, 2688-2695.
- 609 Wang, D., Li, X., Yang, Q., Zeng, G., Liao, D., Zhang, J., 2009. Biological phosphorus removal in  
610 sequencing batch reactor with single-stage oxic process. *Bioresour. Technol.* 2008, 5466-5473.
- 611 Wang, D., Li, X., Yang, Q., Zheng, W., Liu, Z., Liu, Y., Cao, J., Yue, X., Shen, T., Zeng, G., Deng, J.,  
612 2009. The probable metabolic relation between phosphate uptake and energy storages formations  
613 under single-stage oxic condition. *Bioresour. Technol.* 100, 4005-4011.
- 614 Xiao, N., Chen, Y., Chen, A., Feng, L., 2014. Enhanced bio-hydrogen production from protein  
615 wastewater by altering protein structure and amino acids acidification type. *Sci. Rep.* 4, 3992.
- 616 Yang, S.S., Guo, W.Q., Cao, G.L., Zheng, H.S., Ren, N.Q., 2012. Simultaneous waste activated sludge  
617 disintegration and biological hydrogen production using an ozone/ultrasound pretreatment. *Bioresour.*  
618 *Technol.* 124, 347-354.
- 619 Yuan, H., Chen, Y., Zhang, H., Jiang, S., Zhou, Q., Gu, G., 2006. Improved bioproduction of  
620 short-chain fatty acids (SCFAs) from excess sludge under alkaline conditions. *Environ. Sci. Technol.*  
621 40, 2025-2029.
- 622 Zhao, Y., Chen, Y., 2011. Nano-TiO<sub>2</sub> enhanced photofermentative hydrogen produced from the dark  
623 fermentation liquid of waste activated sludge. *Environ. Sci. Technol.* 45, 8589-8595.
- 624 Zhao, Y., Chen, Y., Zhang, D., Zhu, X., 2010. Waste activated sludge fermentation for hydrogen  
625 production enhanced by anaerobic process improvement and acetobacteria inhibition: the role of  
626 fermentation pH. *Environ. Sci. Technol.* 44, 3317-3323.
- 627 Zhou, P., Elbeshbishy, E., Nakhla, G., 2013. Optimization of biological hydrogen production for  
628 anaerobic co-digestion of food waste and wastewater biosolids. *Bioresour. Technol.* 130, 710-718.

**Table 1. The main characteristics of the sludges withdrawn from the five reactors fed with different influent COD concentrations <sup>a</sup>**

Parameter	Sludge-200	Sludge-400	Sludge-600	Sludge-800	Sludge-1000
<b>TSS</b>	13595 ± 306	13608 ± 280	13549 ± 393	13835 ± 328	13710 ± 375
VSS	12120 ± 285	12128 ± 271	12174 ± 332	12310 ± 410	12243 ± 296
Total COD	14290 ± 240	14310 ± 310	14220 ± 420	14550 ± 370	14370 ± 390
Total protein <sup>b</sup>	575 ± 29	556 ± 23	542 ± 20	517 ± 27	491 ± 18
Total carbohydrate <sup>c</sup>	231 ± 14	202 ± 11	181 ± 9	165 ± 13	142 ± 7
PHA <sup>d</sup>	25 ± 3	87 ± 7	107 ± 8	146 ± 12	178 ± 11
Lipid and oil	5.8 ± 0.4	5.5 ± 0.8	5.7 ± 0.9	5.3 ± 0.8	5.2 ± 0.5

<sup>a</sup> Results are the averages and their standard deviations of triplicate measurements. Sludge-X represents the sludge withdrawn from the reactor which is fed with X mg/L COD. The unit for total suspended solids, VSS, and total COD is mg/L while the remainder is expressed in mg/g VSS.

<sup>b</sup> EPS protein content in sludge-200, sludge-400, sludge-600, sludge-800, and sludge-1000 was respectively 56 ± 4, 52 ± 5, 49 ± 3, 55 ± 7, and 51 ± 5 mg/g VSS.

<sup>c</sup> EPS carbohydrate content in sludge-200, sludge-400, sludge-600, sludge-800, and sludge-1000 was respectively 26 ± 2, 28 ± 3, 24 ± 3, 25 ± 5 and 23 ± 2 mg/g VSS.

<sup>d</sup> The percentages of PHB, PHV, and PH2MV are 77.3%, 19.5%, and 3.2% in the sludge-200, and 79.5%, 18.6%, and 1.9% in the sludge-400, and 77.6%, 18.9%, and 3.5% in the sludge-600, and 80.1%, 17.5%, and 2.4% in the sludge-800, and 78.3%, 20.2%, and 1.5% in the sludge-1000, respectively.

**Table 2. The main characteristics of the sludges wasted from the five reactors fed with the same COD concentration but different acetate to propionate ratios <sup>a</sup>**

Parameter	Sludge-I	Sludge-II	Sludge-III	Sludge-IV	Sludge-V
TSS	13340 ± 410	13125 ± 376	13037 ± 423	13295 ± 357	13453 ± 442
VSS	11810 ± 265	11776 ± 293	11485 ± 316	11995 ± 325	12160 ± 384
Total COD	14130 ± 370	13680 ± 340	13550 ± 310	14160 ± 330	14190 ± 350
Total protein <sup>b</sup>	505 ± 32	510 ± 27	507 ± 23	501 ± 30	505 ± 26
Total carbohydrate <sup>c</sup>	169 ± 15	162 ± 13	172 ± 15	166 ± 11	164 ± 12
PHA	132 ± 11	135 ± 9	130 ± 14	127 ± 15	129 ± 10
PHB fraction	79.1 ± 8.6	52.9 ± 4.7	31.5 ± 5.9	20.3 ± 5.2	13.7 ± 1.6
PHV fraction	18.2 ± 1.5	41.5 ± 3.8	62.3 ± 5.4	76.1 ± 7.9	82.8 ± 7.7
Lipid and oil	5.1 ± 0.5	5.3 ± 0.7	4.8 ± 0.7	5.2 ± 0.4	4.3 ± 0.9

<sup>a</sup> Results are the averages and their standard deviations of triplicate measurements. The unit for total suspended solids, VSS, and total COD is mg/L; PHB and PHV fractions are expressed in % of total PHA; the remainder is expressed in mg/g VSS.

<sup>b</sup> EPS protein content in sludge-I, sludge-II, sludge-III, sludge-IV, and sludge-V was respectively 63 ± 6, 58 ± 4, 59 ± 6, 62 ± 7, and 65 ± 8 mg/g VSS.

<sup>c</sup> EPS carbohydrate content in sludge-I, sludge-II, sludge-III, sludge-IV, and sludge-V was respectively 31 ± 4, 33 ± 5, 29 ± 2, 35 ± 6, and 30 ± 2 mg/g VSS.

**Table 3. Comparison of individual and total SCFA production from the low PHA sludge (Sludge-200), PHB-dominant sludge (Sludge-I), and PHV-dominant sludge (Sludge-V) at the time of maximal hydrogen production <sup>a</sup>**

		Low PHA sludge	PHB-dominant sludge	PHV-dominant sludge
acetic	concentration	114.2 ± 8.2	211.0 ± 12.0	109.3 ± 1.9
	fraction	58.7 ± 2.2	56.3 ± 1.5	33.6 ± 0.6
propionic	concentration	32.3 ± 2.9	59.3 ± 2.6	122.1 ± 5.9
	fraction	16.6 ± 1.0	15.8 ± 0.3	37.5 ± 1.2
n-butyric	concentration	7.5 ± 2.0	16.0 ± 0.3	5.3 ± 0.5
	fraction	3.8 ± 1.0	4.3 ± 0.2	1.6 ± 0.2
isobutyric	concentration	20.3 ± 9.0	46.1 ± 5.7	25.6 ± 3.6
	fraction	10.5 ± 4.6	12.3 ± 1.6	7.9 ± 1.1
n-valeric	concentration	13.2 ± 1.1	29.4 ± 4.5	42.7 ± 3.7
	fraction	6.8 ± 0.4	7.8 ± 1.0	13.1 ± 1.0
isovaleric	concentration	6.8 ± 0.5	13.2 ± 2.3	20.3 ± 1.1
	fraction	3.5 ± 0.4	3.5 ± 0.6	6.2 ± 0.3
total SCFA	concentration	194.3 ± 11.1	375.0 ± 17.1	325.3 ± 7.4

<sup>a</sup> The data are the averages and their standard deviations of triplicate measurements. The unit for concentration is mg COD/g VSS while fraction is expressed as % of total SCFA.

**Table 4. Comparison of protein consumption and hydrogen production among non-PHA sludge, exogenous-PHA sludge, and intracellular-PHA sludge at the time of maximal hydrogen production<sup>a</sup>**

	Protein consumption ratio (%)	SCFA production (mg COD/ g VSS)	H <sub>2</sub> production (mL/ g VSS)
Non-PHA sludge <sup>b</sup>	15.8 ± 1.1	189.7 ± 13.2	26.1 ± 1.5
Exogenous-PHA sludge <sup>c</sup>	23.6 ± 1.5	326.5 ± 18.5	47.2 ± 1.9
Intracellular-PHA sludge	27.9 ± 2.3	387.4 ± 21.9	59.3 ± 2.8

<sup>a</sup> Results are the averages and their standard deviations of triplicate measurements. The calculated total COD in the non-PHA sludge, exogenous-PHA sludge, and intracellular-PHA sludge reactor was respectively 4.28, 4.31, and 4.31 g based on the characteristics of these sludges, while the measured C/N ratios of these sludges were 7.08, 9.39, and 9.37, respectively.

<sup>b</sup> After 12 h concentration, it was detected that the non-PHA sludge contained 12160 ± 390 mg/L VSS, 14250 ± 370 mg/L total COD, 586 ± 37 mg/g VSS total protein, and 239 ± 18 mg/g VSS total carbohydrate.

<sup>c</sup> The calculation of VSS included both the biomass and exogenous PHA.

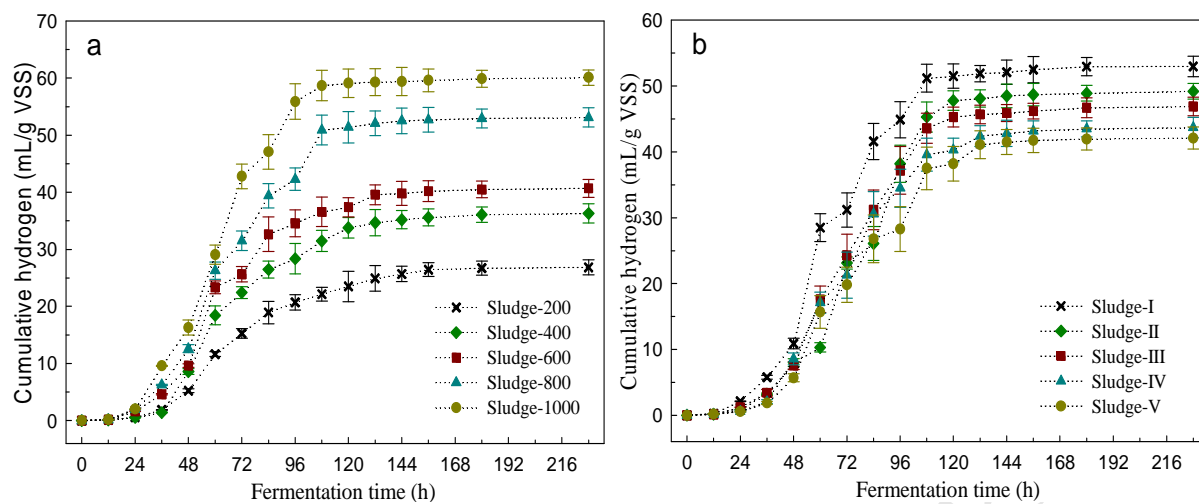


Figure 1. Effect of sludge PHA content (a) and composition (b) on hydrogen production during sludge dark fermentation. Error bars represent standard deviations of triplicate tests.

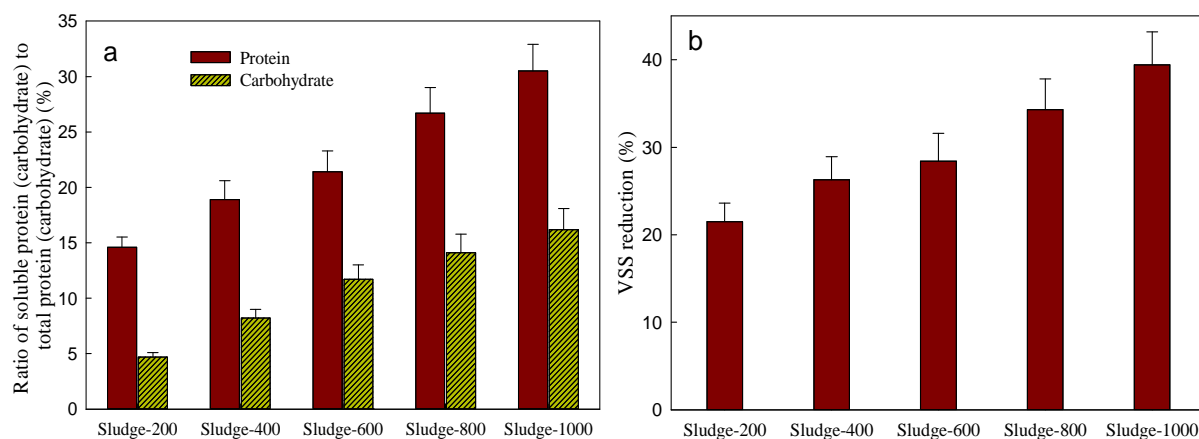


Figure 2. Effect of sludge PHA content on soluble protein and carbohydrate release ratios (a) and VSS reduction (b) at 1 d of fermentation time. Results are the averages and their standard deviations of triplicate measurements.

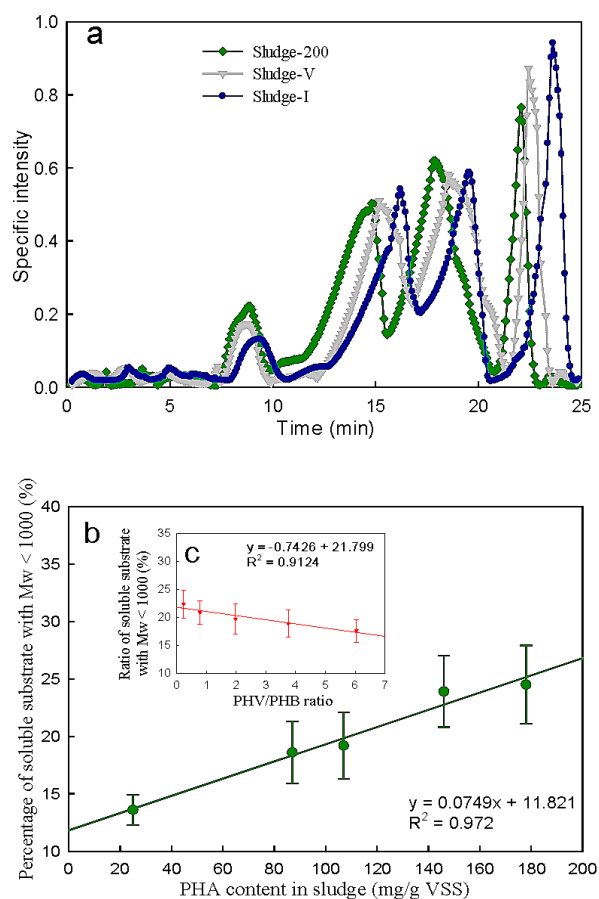


Figure 3. Mw distribution of soluble substrate in the low PHA sludge (Sludge-200), PHV-dominant sludge (Sludge-V), and PHB-dominant sludge (Sludge-I) fed reactors (a), and the correlation between PHA content (b) and composition (c) and percentage of soluble substrate with Mw < 1000 at 2 d of fermentation time. The data are the averages and their standard deviations of triplicate measurements.



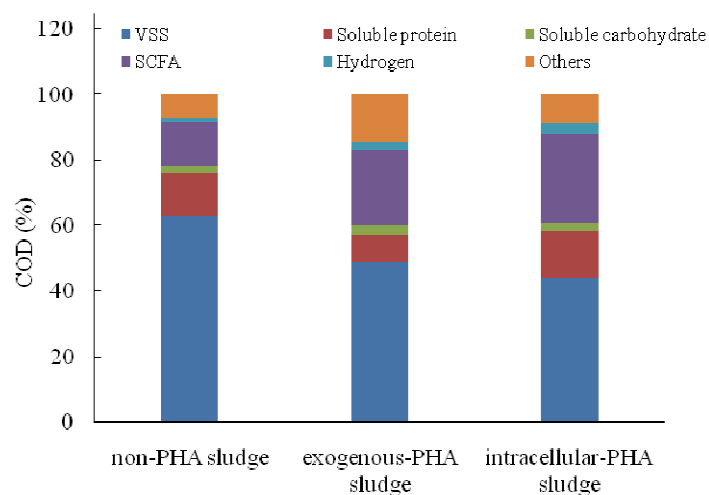


Figure 4. COD mass balance analysis of the non-PHA sludge, exogenous-PHA sludge, and intracellular-PHA sludge fermentation reactors at the time of maximal hydrogen production. The data reported are the averages of triplicate measurements. The COD conversion coefficients are 1.42 g COD/g VSS, 1.67 g COD/ g PHB, 1.92 g COD/g PHV, 2.11 g COD/g PH2MV, 8 g COD/ g H<sub>2</sub>, 1.5 g COD/g protein, 1.06 g COD/ g carbohydrate, 1.07 g COD/g acetic, 1.51 g COD/g propionic, 1.82 g COD/g butyric, and 2.04 g COD/g valeric.

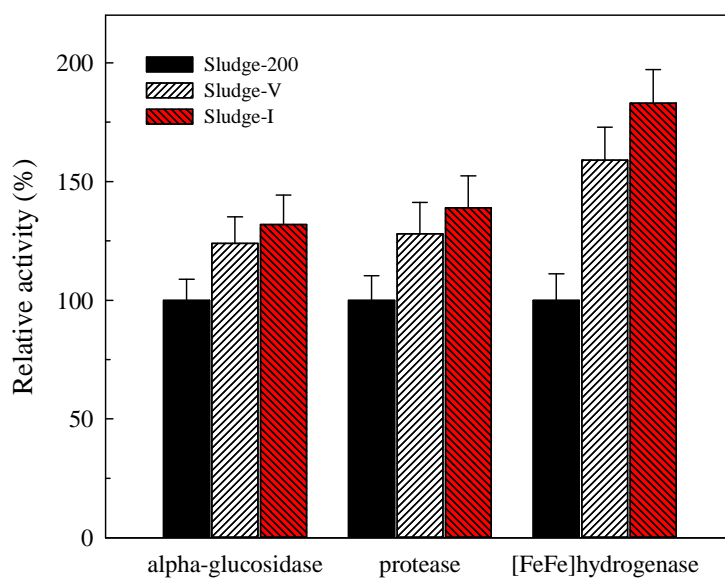


Figure 5. Comparison of the relative activities of key hydrolytic and acid-forming enzymes between the three semi-continuous fermentation reactors after stable operation. Error bars represent standard deviations of triplicate measurements.

**Highlights:**

- The intracellular PHA affected hydrogen production from anaerobic WAS fermentation
- The increase of intracellular PHA benefited hydrogen production
- Compared with PHV, PHB was more beneficial to hydrogen production
- The increased PHA accelerated solubilization and hydrolysis processes
- The increased PHV slowed hydrolysis process and increased propionic acid yield

## Supporting Information

### **Effect of Polyhydroxyalkanoates on Dark Fermentative Hydrogen Production from Waste Activated Sludge**

Dongbo Wang<sup>1,2,3\*</sup>, Yinguang Chen<sup>2\*</sup>, Xiaoming Li<sup>1</sup>, Guangming Zeng<sup>1</sup>

<sup>1</sup>College of Environmental Science and Engineering, Hunan University, Changsha 410082, China; Key Laboratory of Environmental Biology and Pollution Control (Hunan University), Ministry of Education, Changsha 410082, China

<sup>2</sup>State Key Laboratory of Pollution Control and Resources Reuse, School of Environmental Science and Engineering, Tongji University, 1239 Siping Road, Shanghai 200092, China

<sup>3</sup>Advanced Water Management Centre (AWMC), The University of Queensland, QLD 4072, Australia

Corresponding author

Tel: +86-731-8-8823701. Fax: +86-731-8-8823701

E-mail: [w.dongbo@yahoo.com](mailto:w.dongbo@yahoo.com) (Dongbo Wang)

Tel: +86-21-65981263. Fax: +86-21-65986313.

E-mail: [yg2chen@yahoo.com](mailto:yg2chen@yahoo.com) (Yinguang Chen)

## TABLES

**Table S1. Comparison of the soluble protein (carbohydrate)/total protein (carbohydrate) and VSS reduction and hydrogen production in the sludge-1000 and sludge-1000-I fed reactors <sup>a</sup>**

	Fermentation time (d)	Sludge-1000	Sludge-1000-I
Soluble protein/total protein (%)	1	30.2 ± 2.3	15.7 ± 0.9
Soluble carbohydrate/total carbohydrate (%)	1	15.9 ± 1.7	5.6 ± 0.5
VSS reduction (%)	1	41.5 ± 3.2	23.9 ± 2.5
Hydrogen production (mL/g VSS)	4.5	60.1 ± 2.7	26.4 ± 1.2

<sup>a</sup> The data are the averages and their standard deviations of triplicate measurements.

**Table S2. The statistical analysis results of different PHA compositions affecting soluble COD, soluble protein/total protein, soluble carbohydrate/total carbohydrate, and VSS reduction (Compared with Sludge-I).**

		Soluble COD	Soluble protein /total protein	Soluble carbohydrate /total carbohydrate	VSS reduction
Sludge-II	F <sub>observed</sub>	0.36	0.39	0.08	0.37
	F <sub>significant</sub>	7.71	7.71	7.71	7.71
	P <sub>(0.05)</sub>	0.58	0.57	0.79	0.57
Sludge-III	F <sub>observed</sub>	0.56	0.01	0.04	0.01
	F <sub>significant</sub>	7.71	7.71	7.71	7.71
	P <sub>(0.05)</sub>	0.49	0.91	0.85	0.91
Sludge-IV	F <sub>observed</sub>	0.19	0.17	0.10	0.32
	F <sub>significant</sub>	7.71	7.71	7.71	7.71
	P <sub>(0.05)</sub>	0.68	0.70	0.77	0.60
Sludge-V	F <sub>observed</sub>	0.03	0.04	0.23	0.03
	F <sub>significant</sub>	7.71	7.71	7.71	7.71
	P <sub>(0.05)</sub>	0.87	0.84	0.65	0.86

**Table S3. The total SCFA production at 2 d of fermentation time in all reactors <sup>a</sup>**

	Total SCFA (mg COD/g VSS)
Sludge-150	89.5 ± 6.5
Sludge-250	91.1 ± 8.2
Sludge-350	91.9 ± 9.3
Sludge-450	92.5 ± 8.9
Sludge-550	90.9 ± 9.8
Sludge-I	93.4 ± 9.6
Sludge-II	92.8 ± 8.4
Sludge-III	90.6 ± 7.6
Sludge-IV	91.8 ± 6.5
Sludge-V	93.1 ± 8.4

<sup>a</sup>The data are the averages and their standard deviations of triplicate measurements.

**Table S4. The individual and total SCFA production in the Sludge-400, Sludge-600, Sludge-800, Sludge-1000, Sludge-II, Sludge-III, and Sludge-IV fed reactors at the time of maximal hydrogen production <sup>a</sup>**

		Sludge-400	Sludge-600	Sludge-800	Sludge-1000	Sludge-II	Sludge-III	Sludge-IV
acetic	concentration	148.4 ± 11.6	165.2 ± 13.1	213.4 ± 10.3	215.8 ± 14.5	165.3 ± 10.4	145.7 ± 12.3	119.6 ± 11.9
	fraction	55.8 ± 4.2	55.1 ± 4.4	57.5 ± 3.1	56.4 ± 3.9	46.3 ± 3.2	42.5 ± 3.6	36.3 ± 3.6
propionic	concentration	42.6 ± 2.1	50.3 ± 1.9	58.6 ± 1.7	62.3 ± 2.9	93.6 ± 3.2	99.4 ± 3.9	112.3 ± 4.8
	fraction	16.0 ± 0.8	16.8 ± 0.6	15.8 ± 0.5	16.3 ± 0.7	26.2 ± 0.9	29.0 ± 1.1	34.1 ± 1.5
n-butyric	concentration	13.2 ± 0.5	15.6 ± 0.9	15.0 ± 0.4	18.7 ± 0.4	10.6 ± 0.8	9.1 ± 1.1	8.2 ± 0.7
	fraction	5.0 ± 0.2	5.2 ± 0.3	4.0 ± 0.2	4.9 ± 0.4	3.0 ± 0.2	2.7 ± 0.3	2.5 ± 0.2
isobutyric	concentration	29.8 ± 5.1	33.6 ± 4.8	47.0 ± 4.5	43.4 ± 6.5	35.8 ± 2.9	33.2 ± 3.6	28.9 ± 2.4
	fraction	11.2 ± 1.7	11.2 ± 1.6	12.7 ± 1.3	11.3 ± 1.9	10.0 ± 0.8	9.7 ± 1.0	8.8 ± 0.7
n-valeric	concentration	21.6 ± 4.3	22.5 ± 4.5	25.8 ± 4.2	26.8 ± 5.9	32.6 ± 3.5	35.7 ± 3.1	40.6 ± 3.2
	fraction	8.1 ± 1.5	7.5 ± 1.4	6.9 ± 1.1	7.0 ± 1.4	9.1 ± 1.0	10.4 ± 0.9	12.3 ± 1.0
isovaleric	concentration	10.1 ± 1.8	12.5 ± 2.3	11.1 ± 1.9	15.6 ± 2.1	18.9 ± 2.1	19.8 ± 1.9	20.1 ± 1.9
	fraction	3.8 ± 0.7	4.1 ± 0.8	3.1 ± 0.5	4.1 ± 0.6	5.3 ± 0.6	5.8 ± 0.6	6.1 ± 0.6
total SCFA	concentration	265.7 ± 15.8	299.7 ± 17.2	370.9 ± 12.4	382.8 ± 23.4	356.8 ± 14.3	342.9 ± 13.5	329.7 ± 13.9

<sup>a</sup>The data are the averages and their standard deviations of triplicate measurements. The unit for concentration is mg COD/g VSS while fraction is expressed as % of total SCFA.

## FIGURES

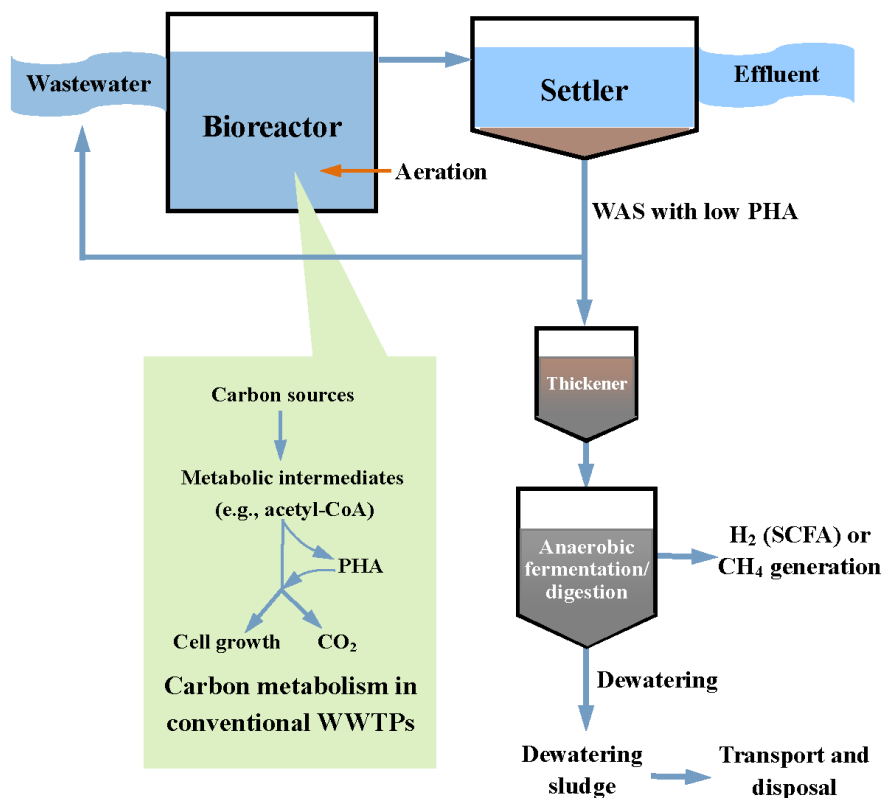


Figure S1. Scheme of the conventional processes for wastewater and WAS treatments in WWTPs.



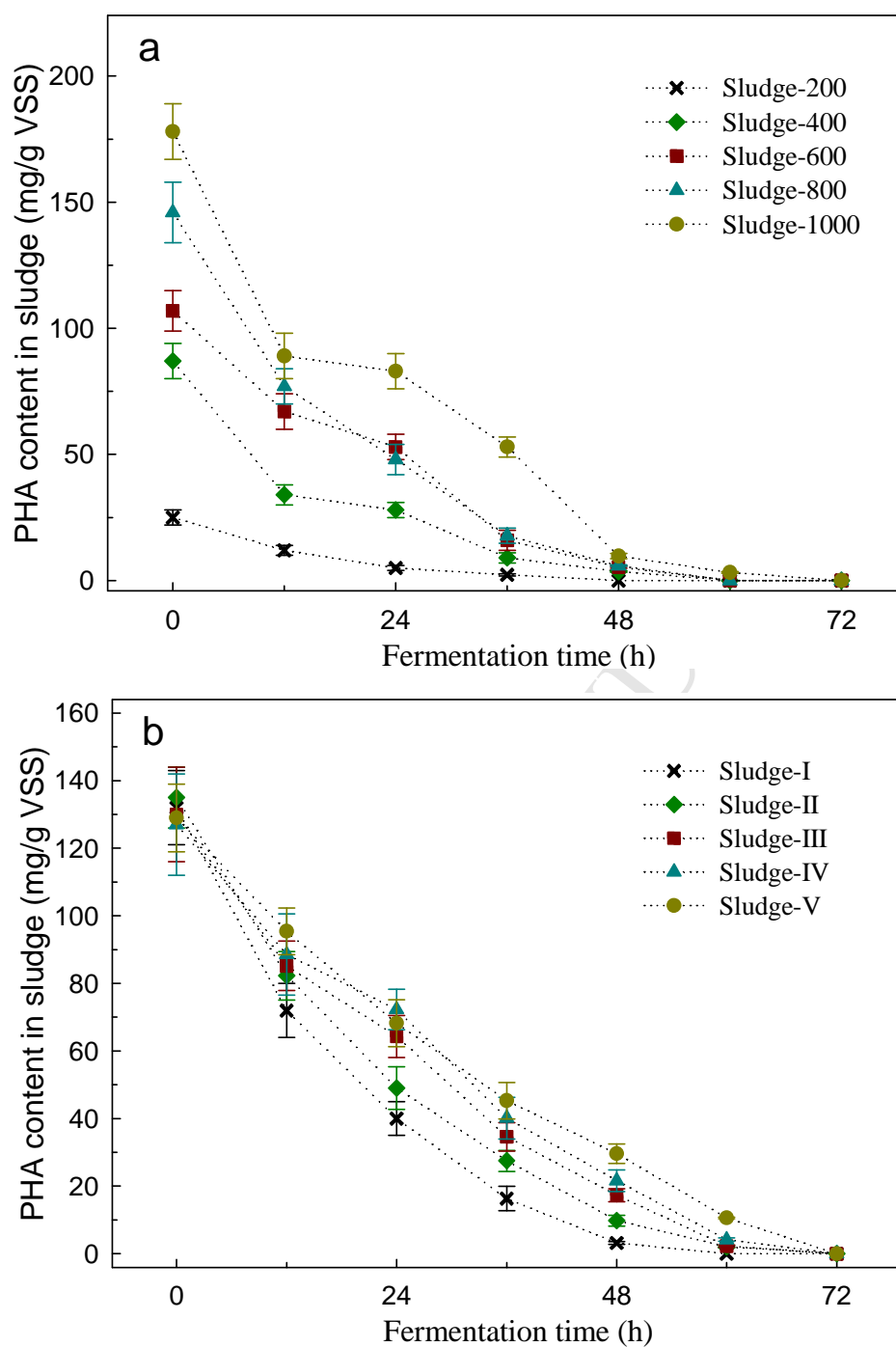


Figure S2. The Curves of PHA variation with fermentation time in reactors fed with different PHA content sludges (a) and PHA constitute sludges (b). Error bars represent standard deviations of triplicate measurements.

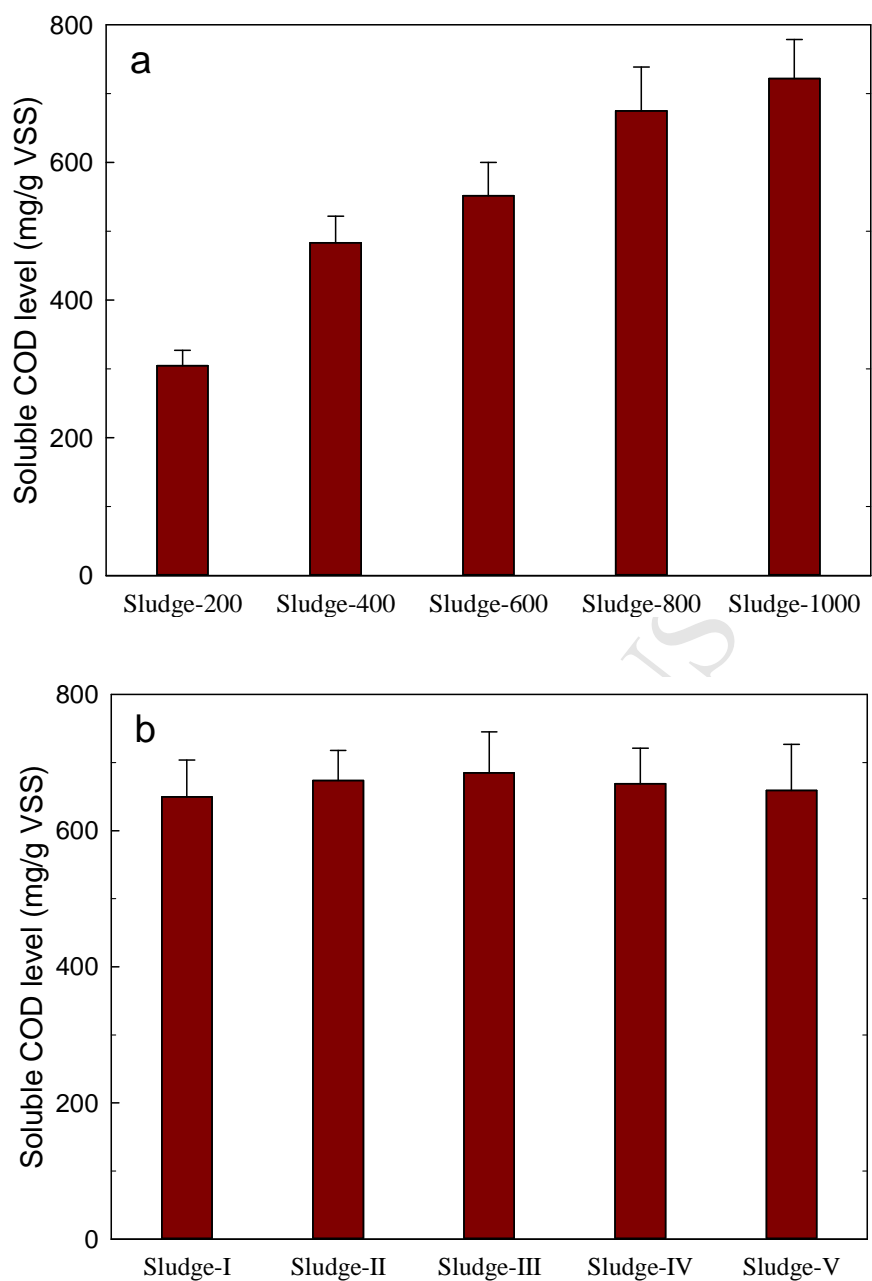


Figure S3. Effect of sludge PHA content (a) and constitute (b) on soluble COD concentration at 1 d of fermentation time. Results are the averages and their standard deviations of triplicate measurements.

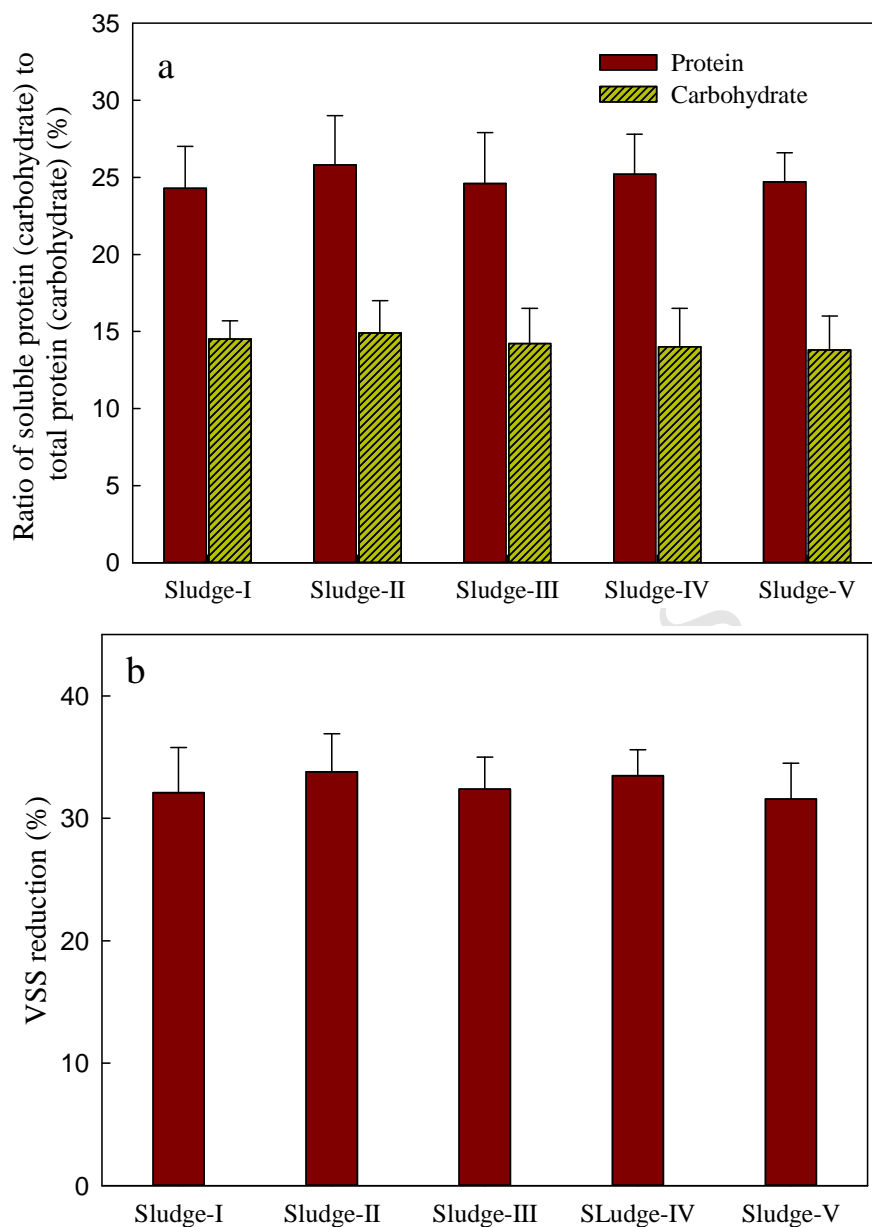


Figure S4. Effect of sludge PHA constitute on soluble protein and carbohydrate release ratios (a) and VSS reduction (b) at 1 d of fermentation time. Results are the averages and their standard deviations of triplicate measurements.

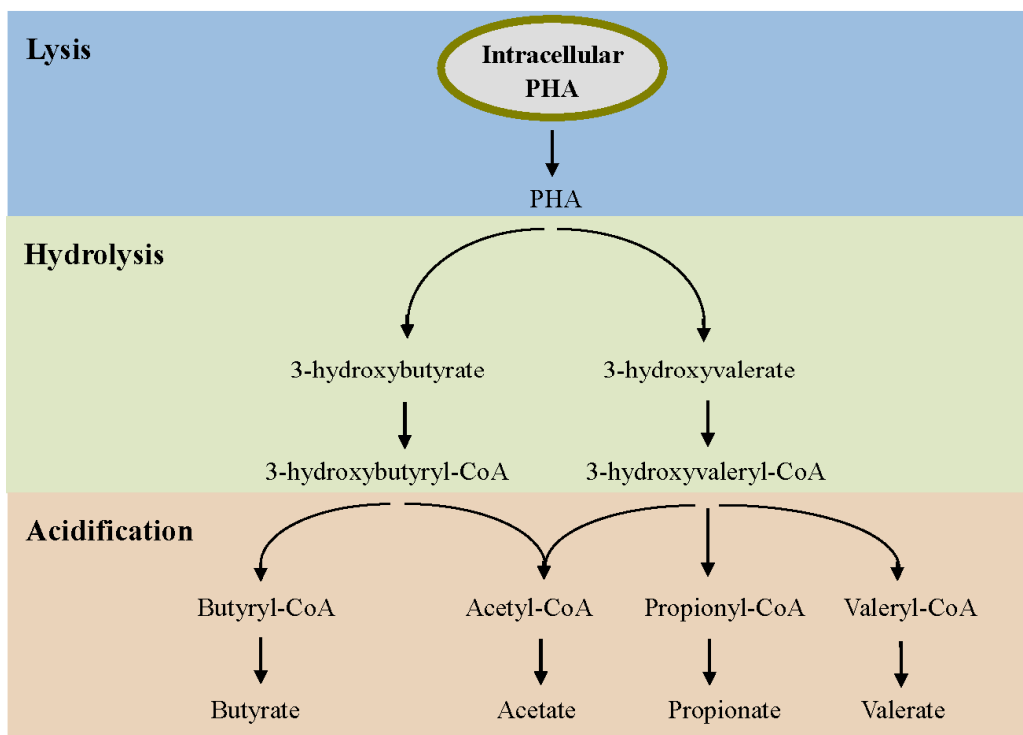


Figure S5. The proposed degradation pathway of PHA to SCFA (Adapted according to the literature (Reischwitz et al., 1998; Janssen and Schink, 1993)).

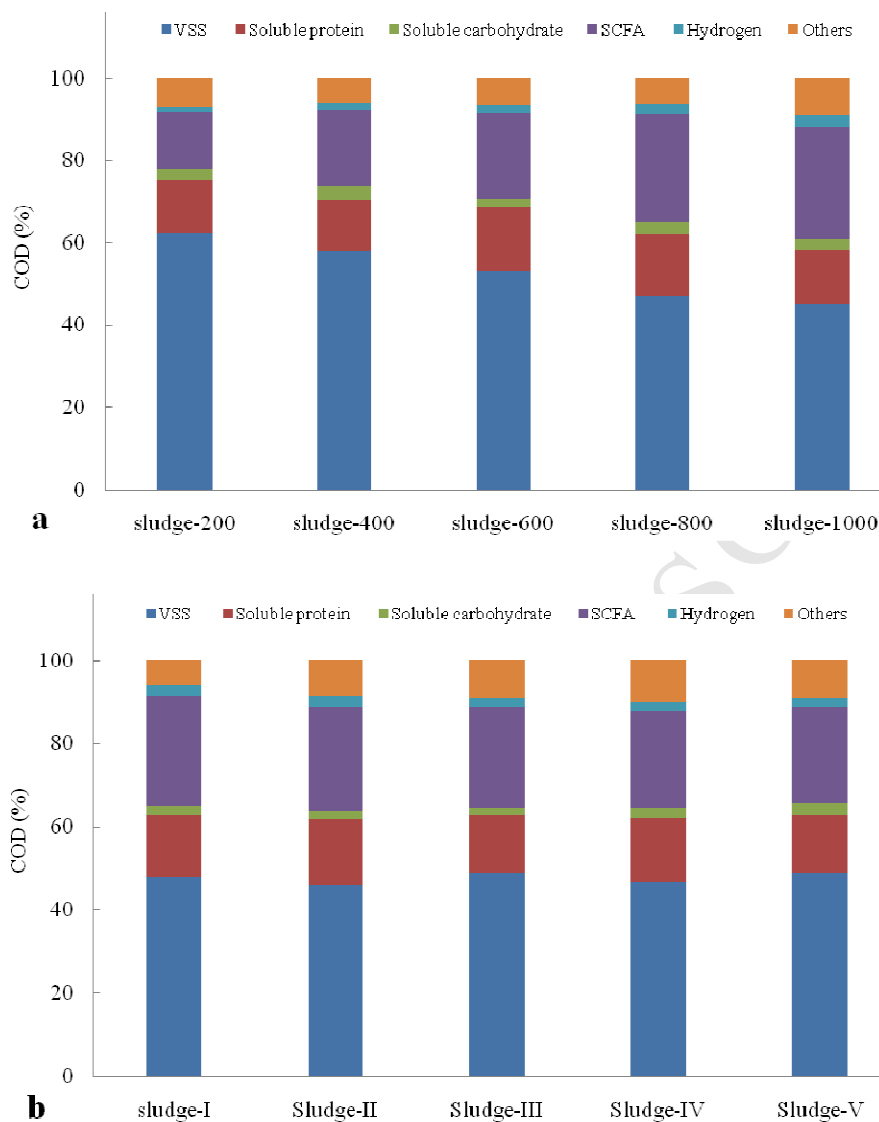


Figure S6. COD balance analysis of the fermentation systems fed with different PHA content sludges (a) and different PHA composition sludges (b) at the time of maximal hydrogen production. The data reported are the averages of triplicate measurements.

**REFERENCES**

- Janssen, P.H., Schink, B., 1993. Pathway of anaerobic poly- $\beta$ -hydroxybutyrate degradation by *Ilyobacter delafieldii*. *Biodegradation* 3, 179-185.
- Reischwitz, A., Stoppok, E., Buchholz, K., 1998. Anaerobic degradation of poly-3-hydroxybutyrate and poly-3-hydroxybutyrate-co-3-hydroxyvalerate. *Biodegradation* 8, 313-319.

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