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1	Comparison of endogenous metabolism during long-term anaerobic starvation of
2	nitrite/nitrate cultivated denitrifying phosphorus removal sludges
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14	Abstract: Denitrifying phosphorus removal (DPR) by denitrifying
15	phosphorus-accumulating organisms (DPAOs) is a promising approach for reducing
16	energy and carbon usage. However, influent fluctuations or interruptions frequently
17	expose the DPAOs biomass to starvation conditions, reducing biomass activity and
18	amount, and ultimately degrading the performance of DPR. Therefore, a better
19	understanding of the endogenous metabolism and recovery ability of DPAOs is
20	urgently required. In the present study, anaerobic starvation (12 days) and recovery
21	were investigated in nitrite- and nitrate-cultivated DPAOs at 20 ± 1 °C. The cell decay

22	rates in nitrite-DPAO sludges from the end of the anaerobic and aerobic phase were
23	0.008 day^{-1} and 0.007 day^{-1} , respectively, being 64% and 68% lower than those of
24	nitrate-DPAO sludges. Nitrite-DPAO sludges also recovered more rapidly than
25	nitrate-DPAO sludge after 12 days of starvation. The maintenance energy of
26	nitrite-DPAO sludges from the end of the anaerobic and aerobic phase were
27	approximately 31% and 34% lower, respectively, than those of nitrate-DPAO sludges.
28	Glycogen and polyphosphate (poly-P) sequentially served as the main maintenance
29	energy sources in both nitrite-and nitrate-DPAO sludges. However, the transformation
30	pathway of the intracellular polymers during starvation differed between the sludges.
31	Nitrate-DPAO sludge used extracellular polymeric substances (EPS) (mainly
32	polysaccharides) as an additional maintenance energy source during the first 3 days of
33	starvation. During this phase, EPS appeared to contribute to $19 - 27\%$ of the ATP
34	production in nitrate-DPAOs, but considerably less to the cell maintenance of
35	nitrite-DPAOs. The high resistance of nitrite-DPAOs to starvation might be
36	attributable to frequent short-term starvation and exposure to toxic substances such as
37	nitrite/free nitrous acids in the parent nitrite-fed reactor. The strong resistance of
38	nitrite-DPAO sludge to anaerobic starvation may be exploited in P removal by
39	shortcut denitrification processes.
40	

41 **Keywords:** Starvation; Denitrifying phosphorus-accumulating organisms;

42 Maintenance; Intracellular polymers; Extracellular polymeric substances; Decay

1. Introduction

44	Owing to large fluctuations in the flow and composition of wastewater, the
45	microorganisms responsible for biological wastewater treatment plants are frequently
46	exposed to long-term famine conditions (days and sometimes weeks) (Lu et al., 2007).
47	During sludge storage, by which large influent variations can be adjusted and flexible
48	plant operation can be achieved, microorganisms may experience starvation
49	(Morgenroth et al., 2000). Starvation significantly reduces the amount and activity of
50	active microorganisms, and risks degrading the capacity, efficiency and robustness of
51	wastewater treatment systems (Hao et al., 2010b; Wang et al., 2013b). Starvation is
52	crucially important in enhanced biological phosphorus removal (EBPR) processes,
53	since it alters the levels of intracellular storage compounds in the functional microbes
54	(i.e., polyphosphate-accumulating organisms, PAOs) (Vargas et al., 2013). Indeed,
55	excessive consumption of intracellular polymers (Yilmaz et al., 2007) or excessive
56	decay of both PAOs and intracellular polymers (Miyake and Morgenroth, 2005) has
57	been implicated in EBPR failure.
58	In the absence of external sustenance, starved microorganisms primarily undergo
59	the endogenous processes consisting of cell maintenance and cell decay (Lu et al.,
60	2007; Wang et al., 2012). The impacts of starvation on PAOs and endogenous
61	processes have been extensively investigated (Lopez et al., 2006; Yilmaz et al., 2007;
62	Lu et al., 2007; Hao et al., 2010a; Wang et al., 2012; Vargas, et al., 2013), and
63	effective strategies for maintaining the biomass activity have been accordingly

64	proposed. Lopez et al. (2006) examined the effects of long-term (weeks) anaerobic
65	and aerobic starvation on the composition and activity of PAOs. They concluded that,
66	under aerobic starvation conditions, PAOs are notably attenuated by endogenous
67	processes, whereas no significant PAOs decay occurs under anaerobic starvation. Lu
68	et al. (2007) proposed an intermittent aerobic-anaerobic strategy for the long-term
69	storage of EBPR sludge. In this strategy, the PAOs decay more slowly than in aerobic
70	storage, and glycogen and poly-P are used at a slower rate than in anaerobic and
71	anoxic storage. A similar recovery strategy was recommended by Yilmaz et al. (2007),
72	who found that alternating anoxic/anaerobic and aerobic operation effectively
73	maintains the biomass activity of activated sludge used for biological nitrogen (N) and
74	phosphorus (P) removal, thereby enabling quick activity recovery (i.e., full recovery
75	within 4 days).
76	Unlike PAOs in traditional EBPR processes, the impacts of starvation on
77	denitrifying polyphosphate-accumulating organisms (DPAOs) have been little
78	reported. Denitrifying phosphorus removal (DPR) by DPAOs is a viable and
79	sustainable technology, as N and P can be simultaneously removed with lower carbon
80	source requirements, aeration costs and cell yields (Murnleitner et al., 1997). In
81	particular, since DPAOs can use nitrite as an electron acceptor, DPR is naturally
82	amenable to shortcut nitrification. By replacing nitrate with nitrite, the oxygen cost
83	and carbon consumption of DPR can be reduced by approximately 25% and 40%,
84	respectively (Abeling et al., 1992). Therefore, DPR by nitrite could be used for

85	innovative biological nutrient removal (BNR) systems where energy and carbon
86	savings are a priority, for example, linking nitrite pathways (i.e., partial nitrification +
87	nitrite-based denitrification) to EBPR (Guisasola et al., 2009; Marcelino et al., 2011;
88	Zhou et al., 2011; Tayà et al., 2013). Moreover, as nitrite enriched PAOs need less
89	carbon source (i.e., intracellular PHA) for P-uptake, and eventually they might have
90	higher PHA accumulation which can be used to speed up their anoxic metabolism
91	after the endogenous period.
92	Recently, identifying the inhibitory effects of nitrite and the feasibility of
93	nitrite-based DPR has received increasing attention (Guisasola et al., 2009; Marcelino
94	et al., 2011; Zhou et al., 2011). However, DPAOs metabolism, especially their
95	endogenous metabolism, has not been properly elucidated. To our knowledge, the
96	endogenous characteristics of nitrite metabolism of DPAOs have not been assessed. A
97	better understanding of the mechanism of the impact of starvation on nitrite- and
98	nitrate-DPAOs, and endogenous metabolism of DPAOs may favor the development
99	of strategies for improvement of the robustness and performance of DPR processes,
100	the resuscitation of DPR systems after famine scenarios, and the storage of DPAOs
101	sludge.
102	Increasing evidence shows that extracellular polymeric substances (EPS) can
103	serve as carbon and energy sources for active biomass growth under starvation
104	conditions (Zhang and Bishop, 2003; Wang et al., 2005; Wang et al., 2007; Liu et al.,
105	2007; Flemming et al., 2010). Wang et al. (2005) found that most biodegradable EPS,

106	especially polysaccharides, are located in the core of aerobic granular sludge, and that
107	this fraction of EPS can be depleted after long-term starvation (20 days), as evidenced
108	by the void structure in the core of starved granules. Since most previous starvation
109	investigations of EBPR sludge did not involve EPS, the contribution of EPS to the
110	maintenance of metabolic activity of PAOs/DPAOs remains unclear.
111	The purpose of this study is to identify the differences between the endogenous
112	characteristics of nitrite- and nitrate-DPAO sludges during 12-day anaerobic
113	starvation, and to better understand the endogenous metabolism of nitrite-DPAOs. The
114	transformation of intracellular polymers and post-starvation activity recovery are also
115	compared between nitrite and nitrate-DPAO sludges. We highlight the different
116	transformation pathways of intracellular polymers in nitrite- and nitrate-DPAOs
117	biomass. We also attempt to clarify the role of EPS (especially polysaccharides) in
118	nitrite/nitrate-DPAOs under anaerobic starvation conditions.
119	2. Materials and methods
120	2.1. Set up and long-term operation of parent reactors
121	DPAOs sludge was enriched in two identical laboratory-scale sequencing batch
122	reactors (SBR _{NO3-} and SBR _{NO2-} , using nitrate and nitrite as electron acceptors,
123	respectively) with a working volume of 7.5 L as outlined by Wang et al. (2011). Both
124	SBRs were independently operated in a cyclical anaerobic-anoxic-aerobic pattern
125	with a cycle time of 8 h (15-min filling period, a 120-min anaerobic period, a 210-min
126	anoxic period, a 30-min aerobic period, a 20-min settling period, a 15-min effluent

127	discharging period, and a 70-min idle period). During the filling period, each reactor
128	was fed with 5.5 L synthetic wastewater (Section 2.2). KNO ₃ or KNO ₂ solution was
129	carefully added to the reactors during the anoxic phase to prevent nitrate or nitrite
130	accumulation. Specifically, 34 mL KNO $_3$ solution was added to A-SBR _{NO3} -, giving an
131	initial NO ₃ ⁻ -N concentration of 34 mg/L; 11 mL KNO ₂ solution was added into
132	A-SBR _{NO2-} in three pulses (70 min of intervals), giving an initial $NO_2^{-}-N$
133	concentration of 11 mg/L per pulse. Effluent was withdrawn from the port at 30 cm
134	above the bottom, leaving 2.0 L of mixed liquor in the reactor.
135	The temperature was maintained at 20 ± 1 °C, and the rotation speed was
136	controlled at 150 ± 10 rpm during the reaction phases. Nitrogen gas was introduced
137	through the headspace for 5 min to ensure anaerobic condition at the beginning of the
138	anaerobic phase of each cycle. In the post-aerobic phase, the dissolved oxygen (DO)
139	concentration was controlled at $2 - 4$ mg/L. The hydraulic retention time (HRT) was
140	10.9 h. To maintain the concentration of the mixed liquid suspended solids (MLSS) at
141	4000 ± 200 mg/L, 125 mL of mixed liquor was periodically discarded at the end of
142	each aerobic period. The solid retention time (SRT) under these conditions was
143	approximately 20 days.
144	2.2. Synthetic wastewater
145	The synthetic wastewater used in this study contained (per L): 257.1 mg
146	CH ₃ CH ₂ COONa (300 mg chemical oxygen demand (COD)); 32.9 mg KH ₂ PO ₄ (7.5
147	mg P); 55.3 mg K ₂ HPO ₄ · 3H ₂ O (7.5 mg P); 38.2 mg NH ₄ Cl; 85.0 mg MgSO ₄ · 7H ₂ O;

148	10.0 mg CaCl ₂ . Therefore, the influent volatile fatty acid (VFA, i.e., propionate) to P
149	ratio was 6.4 mg C/mg P. PAOs were preferentially selected by adding propionate as
150	the sole carbon source (Oehmen et al., 2005). Trace salt solution (0.3 mL/L) and
151	allylthiourea (4 mg/L) were added as described by Wang et al. (2011). The pH of the
152	synthetic wastewater was maintained at 7.5 ± 0.1 by adding NaHCO ₃ .
153	2.3. Starvation batch experiments
154	Once the SBRs had reached steady-state operation, batch experiments were conducted
155	in four identical sealed reactors, each with a working volume of 3.8 L and an
156	overhead space of 0.2 L. The long-term nitrate or nitrite cultivated sludge in SBR_{NO3-}
157	and $\text{SBR}_{\text{NO2-}}$ was divided into two equal portions at the end of the decanting phase,
158	and was then transferred to one of the four batch reactors. For each starvation test, 2.8
159	L synthetic wastewater was rapidly added to each reactor, maintaining the MLSS
160	level at 4000 \pm 200 mg/L. One of the two reactors incubated with nitrate-DPAOs
161	sludge was operated with a 2-h anaerobic reaction (A-R _{NO3-}), the other was operated
162	with a 2-h anaerobic reaction, a 3.5-h anoxic reaction and a 0.5-h aerobic reaction
163	(O- R_{NO3-}). The corresponding nitrite-DPAOs sludges (operated under the same
164	conditions) from the end of the anaerobic and aerobic phases are called A-R $_{\rm NO2-}$ and
165	$O-R_{NO2-}$, respectively. After one operation cycle, each reactor was sparged with
166	nitrogen gas for 10 min to maintain anaerobic conditions, and was left idle for the
167	next 12 days. During the starvation experiments, each reactor was sparged with
168	nitrogen gas for 10 min per day to remove any H_2S accumulated by the activity of

169	sulfate-reducing bacteria (Morgenroth et al., 2000; Yilmaz et al., 2007). All tests were
170	carried out at 20 \pm 1 °C, and the pH was manually controlled at 7.5 \pm 0.1 by adding
171	0.3 M HCl or 0.3 M NaOH.
172	Liquid- and solid-phase samples were taken from each reactor on days 0, 1.5, 3, 5,
173	8, and 12. The mixed liquor was filtered through a Millipore filter unit (pore size =
174	0.45 μ m), and the liquid portion was retained for analysis of VFA, NH ₄ ⁺ -N, PO ₄ ³⁻ -P,
175	$NO_3^{-}N$ and $NO_2^{-}N$. The solid portion (biomass) was centrifuged, freeze-dried and
176	retained for analysis of intracellular polymers, including poly- β -hydroxybutyrate
177	(PHB), poly-β-hydroxyvalerate (PHV), poly-3-hydroxy-2-methylvalerate (PH2MV),
178	and glycogen. Samples were also taken for MLSS, mixed liquor volatile suspended
179	solids (MLVSS) and EPS measurements.
180	2.4. Recovery batch experiments
181	After 12 days of starvation, the biomass activities of the starved nitrite/nitrate DPAO
182	sludges were assessed in batch tests. The sludges in the four batch reactors were
183	washed three times with 2.8 L propionate-free synthetic wastewater. At the beginning
184	of the recovery test, 2.8 L of propionate-free synthetic wastewater was rapidly added
185	to the reactors containing activated sludge from the end of the anaerobic phase
186	(denoted AS _{ANA.end}) (A-R _{NO3-} and A-R _{NO2-}). The anoxic reaction (3.5 h) was started by
187	adding KNO ₃ /KNO ₂ as described in <i>Section 2.1</i> , followed by a 0.5-h aerobic reaction.
188	The reactors containing activated sludge from the end of the aerobic phase (denoted
189	$AS_{AER.end}$ (O-R _{NO3-} and O-R _{NO2-}) were rapidly supplemented with 2.8 L of synthetic

190	wastewater at the beginning of the recovery test, and then were directly operated
191	through the typical cycle for SBR_{NO3-} and SBR_{NO2-} , respectively. Two recovery cycles
192	were applied to each of these four reactors.
193	2.5. Analytical methods
194	Liquid- and solid-phase analyses of NH4 ⁺ -N, NO3 ⁻ -N, NO2 ⁻ -N, PO4 ³⁻ -P, MLSS and
195	MLVSS were performed by the standard method (APHA, 1998). DO and pH were
196	measured online using oxygen and pH meters (oxi 3310 and pH 3310, WTW
197	Company, Germany), respectively. Glycogen, poly-β-hydroxyalkanoates (PHA), VFA,
198	and EPS were determined by the procedure detailed in the Supplementary Information
199	(SI) (Text S1). PHA content in the sludge sample was defined as the sum of the
200	measured PHB, PHV and PH2MV. EPS was extracted by the formaldehyde-NaOH
201	method, and was calculated as the sum of polysaccharides, proteins and humic
202	substances.
203	The relative PAOs and glycogen-accumulating organisms (GAOs) abundances in
204	both parent SBRs were estimated by 16S rRNA fluorescence in situ hybridization
205	(FISH), as described in Wang et al. (2013b). Candidatus Accumulibacter phosphatis
206	(hereafter referred to as Accumulibacter), Candidatus Competibacter phosphatis
207	(hereafter referred to as Competibacter), Defluvicoccus-related TFO, and
208	Defluvicoccus-related DF were targeted by appropriate oligonucleotide probes (Text
209	S2 and Table S1).

210 **2.6. Determination of cell decay rate**

- 211 The decay rate of DPAOs was estimated from the measured NH_4^+ -N release rate
- 212 based on the activated biomass composition (CH_{2.09}O_{0.54}N_{0.20}P_{0.015}) (Smolders et al.,
- 213 1994), i.e., primarily based on the reduction in the amount of bacteria. The MLVSS is
- assumed as the sum of PHA, glycogen, and active biomass (Smolders et al., 1995).
- 215 Thus, the active biomass is estimated by subtracting the PHA and glycogen content
- 216 from the MLVSS. Accordingly, the decay rate is calculated based on Eq. (1) (Lesouef
- 217 et al., 1992):
- 218 $b = -\ln (X_t/X_0)/t_d$ (1)
- 219 where b is the decay death rate of the PAOs, X_0 and X_t denote the active biomass
- 220 concentration (without glycogen and PHA) before and after starvation respectively,
- 221 and t_d is the duration of the starvation period.

222 **3. Results**

223 **3.1.** Performance of SBR_{NO2}. and SBR_{NO3}. and relevant microbial populations

224 The SBR_{NO2-} and SBR_{NO3-} steadily operated for 210 days. Typical cycle tests were

225 conducted during steady-state operation (Figure S1) and the DPAOs activities were

estimated from the biochemical reaction rates. The maximum rates of P release and

- 227 uptake, denitrification, and intracellular polymer transformations were higher for
- 228 nitrite DPAOs than those for nitrate-DPAOs (Table 1), indicating that DPAOs
- activities are greater in the nitrite-fed than in the nitrate fed reactors.
- 230 FISH results show that *Accumulibacter*, bound to the PAOMIX probe, were the

231	dominant organisms in both reactors, comprising $66 \pm 1.5\%$ and $75 \pm 1.1\%$ of total
232	biomass in SBR_{NO3-} and SBR_{NO2-} , respectively (Table 1). The ratio of PAOII (unable
233	to use nitrate as an electron acceptor) to total PAOs was 57.6% and 72% in ${\rm SBR}_{\rm NO3-}$
234	and SBR _{NO2} -, respectively. <i>Defluvicoccus</i> -related GAOs were approximately $27 \pm 0.9\%$
235	and 19 \pm 0.6% in SBR _{NO3-} and SBR _{NO2-} , respectively, while <i>Competibacter</i> -related
236	GAOs were nearly undetectable (< 1 %) in both SBRs (Table 1).
237	3.2. Anaerobic starvation in nitrite/nitrate DPAOs biomass
238	3.2.1 Release of NH_4^+ -N and PO $_4^{3-}$ -P, and reduction of MLSS and MLVSS
239	Since the NH_4^+ -N profile reflects the biomass growth condition, it is a useful tool for
240	determining the cell decay of DPAOs (Zeng et al., 2003b; Lu et al., 2007). In all
241	reactors, the NH_4^+ -N concentrations increased gradually during the first 3 days of
242	starvation and rapidly thereafter (Figure 1a), indicating the increasing extent of cell
243	decay. Within 12 days of starvation, cell decay (including cell lysis and the respiration
244	of intracellular materials) released 41.8 mg, 39.5 mg, 18.7 mg and 15.8 mg of
245	NH_4^+ -N/L into the liquid phase of A-R _{NO3-} , O-R _{NO3-} , A-R _{NO2-} and O-R _{NO2-} ,
246	respectively (Figure 1a and Table 2). The ammonia release rates were clearly much
247	higher in nitrate-DPAO sludges than those in nitrite-DPAO sludges (Table 2), whereas
248	no appreciable difference was observed in the NH_4^+ -N release rates between
249	AS _{ANA.end} and AS _{AER.end} (Figure 1a).
250	The starvation period in all reactors was also marked by P release, as

251 intracellular poly-P was degraded to obtain energy for maintenance processes.

252	Specifically, the P release rate was elevated during day $1-5$ in all reactors; thereafter,
253	the P concentration in the bulk remained constant (Figure 1b). Correspondingly, the
254	MLSS concentrations gradually decreased during the first 5 days (Figure 2a). Notably,
255	the MLSS concentration decreased less in nitrite-DPAO sludges (A- R_{NO2-} and O- R_{NO2-})
256	than those in nitrate-DPAO sludges (A- R_{NO3-} and O- R_{NO3-}) (Figure 2a). These findings
257	are related to the lower P release rate in A- R_{NO2-} and O- R_{NO2-} (Table 3). Among these
258	reactors, the MLSS and MLVSS concentrations were most heavily reduced in O-R _{NO3-} ,
259	largely because the storage products (especially glycogen and poly-P) were most
260	depleted in this reactor, accounting for approximately 51% and 33% of the decrease
261	of MLVSS and MLSS, respectively, in O-R _{NO3-} (Table 3). Similarly, Lopez et al.
262	(2006) reported that maintenance processes utilizing organic (PHA and glycogen) and
263	inorganic (poly-P) storage products accounted for about 23% and 29% of the aerobic
264	decrease of MLVSS and MLSS, respectively. The MLVSS, MLSS and MLVSS/MLSS
265	variations in $AS_{ANA.end}$ were comparable to those in $AS_{AER.end}$ from the same parent
266	SBR (Figure 2).

267 3.2.2 Variations in glycogen and PHA contents

In all four reactors, most of the glycogen was consumed within the first 5 days (Figure 3). The glycogen degradation was approximately 21% and 38% lower in the $AS_{ANA.end}$ and $AS_{AER.end}$ sludges, respectively, when compared with those in their nitrate-DPAO sludge counterparts (Figure 3 and Table 3). These findings indicate a relatively lower energy requirement for glycogen hydrolysis in nitrite-DPAOs, and may also correlate

273	with the lower GAO percentage in nitrite-DPAO sludge ($19 \pm 0.6\%$) than in
274	nitrate-DPAO sludge (27 \pm 0.9%) (Table 1). Moreover, the amounts of glycogen
275	degradation were much lower in $AS_{ANA.end}$ (A-R _{NO3-} and A-R _{NO2-}) than those in
276	$AS_{AER.end}$ (O-R _{NO3-} and O-R _{NO2-}). Since glycogen had been partially degraded by
277	anaerobic reactions to supply reducing equivalents before the starvation test, it is
278	likely that less glycogen was available for $AS_{ANA.end}$ as compared to that for $AS_{AER.end}$.
279	These findings agree with our previous observation (Wang et al., 2012) that glycogen
280	degradation rate is approximately 48% lower for $AS_{ANA.end}$ than for $AS_{AER.end}$ after 7
281	days of anaerobic starvation at 15 °C.
282	PHA was synthesized during the 12 starvation days in all four reactors. Most of
283	the PHA was synthesized during the first 5 days (Figure 3), corresponding to the high
284	glycogen degradation. The main PHA components of nitrite-and nitrate-DPAO
285	sludges fed with propionate as the sole carbon source were PH2MV and PHV (Table
286	3). During the 12 days of starvation more PHA was synthesized by $AS_{ANA.end}$ in
287	A-R _{NO2-} (2.79 mmol-C/g-MLVSS) than that by AS _{ANA.end} in A-R _{NO3-} (1.84
288	mmol-C/g-MLVSS) (Table 3), and the PHA synthesis rate in A- R_{NO2} - was almost
289	twice that in A-R _{NO3-} . Similar results were obtained for $AS_{AER.end}$, suggesting that
290	most of the degraded glycogen was converted to PHA in the nitrite-DPAO sludges.
291	3.2.3 Variations in EPS amounts and compositions
292	EPS production is essential for the survival of Accumulibacter in wastewater

treatment systems (Martín et al., 2006). As a candidate carbon and energy source,

294	EPS degradation allows the microorganisms to rapidly adapt to varying influent
295	composition, temperature (Martín et al., 2006) and substrate limitation (Zhang and
296	Bishop, 2003). However, the contribution of EPS to the anaerobic endogenous
297	metabolism of DPAOs has not been previously reported. We also report the first
298	description of EPS variations during anaerobic starvation of DPAO sludges.
299	Figure 4 presents the EPS profiles after 3 and 12 days of starvation. The initial
300	amounts of EPS in A-R _{NO2-} and O-R _{NO2-} sludges were approximately 17.0 and 11.5%
301	lower, respectively, than those in their A- R_{NO3-} and O- R_{NO3-} counterparts (Figure 4d).
302	This discrepancy is attributable to the different polysaccharides content in the EPS of
303	different sludges. In particular, polysaccharides synthesis may be prevented by the
304	presence of free nitrous acid (FNA) in nitrite-DPAOs (Wang et al., 2013a).
305	Specifically, the initial polysaccharides content of EPS was 60.8% and 71.3% higher
306	in A-R _{NO3-} and O-R _{NO3-} than those in A-R _{NO2-} and O-R _{NO2-} , respectively (Figure 4b).
307	Polysaccharides degradation in A- R_{NO3-} and O- R_{NO3-} was almost complete within 3
308	days of starvation (Figure 4b). Similarly, Wang et al. (2007) reported a sharp decrease
309	(approximately 50%) in the polysaccharides content of EPS in highly resistant aerobic
310	granules starved for 4 days. In A- R_{NO2-} and O- R_{NO2-} , polysaccharides degradation
311	throughout the first 3 days was only 3.7 \pm 0.2 mg/g MLVSS and 1.2 \pm 0.0 mg/g
312	MLVSS respectively, accounting for approximately 32% and 12% of the total
313	polysaccharides content, respectively (Figure 4b). At the end of the starvation on day
314	12, the polysaccharides contents remained low in all reactors.

315	The proteins changes in the four reactors greatly differed from the
316	polysaccharides changes (Figure 4a and b). The proteins content in all reactors
317	slightly increased during the first 3 days of starvation and had decreased to low levels
318	by day 12. It is speculated that, early in the starvation period, cell decay processes
319	released a portion of the intracellular proteins into the extracellular space, where it
320	was captured by EPS. Cellular decay is supported by the increased ammonia
321	concentration in all four reactors during the first 3 days of starvation (Figure 1a). At
322	the end of the starvation period, the proteins contents in all four reactors were heavily
323	reduced, suggesting that proteins may also be hydrolyzed or degraded by the
324	microorganisms as carbon and energy sources. The difference in EPS content
325	variations was negligible between $AS_{ANA.end}$ and $AS_{AER.end}$ from the same parent
326	$(SBR_{NO3-} \text{ or } SBR_{NO2-}).$
327	3.3. Recovery of nitrite- and nitrate-DPAOs activities after 12 days of starvation
328	3.3.1 Recovery of nitrite- and nitrate-DPAOs in AS _{ANA.end}
329	Nitrite denitrification was completed during the first recovery cycle in both A- R_{NO3-}
330	and A-R _{NO2-} . The P uptake efficiency of nitrite-DPAOs was actually slightly increased
331	relative to the pre-starvation efficiency (Figure 5c and d; Figure S1 d). This increase
332	is attributed to the relatively lower "secondary" P release after nitrite depletion by
333	starved microorganism (1.6 mg PO_4^{3-} -P/L vs. 10.8 mg PO_4^{3-} -P/L in a typical
334	pre-starvation cycle). In contrast, the efficiencies of nitrate denitrification and anoxic
335	P uptake in starved nitrate-DPAOs sludge were decreased by 42.9% and 47.2%,

336	respectively, relative to their pre-starvation levels in a typical cycle (Figure 5a and b;
337	Figure S1c and d). The rapid recovery of nitrite-DPAOs activities indicates a higher
338	ability of these organisms to overcome starvation shock. The strong anoxic
339	denitrification and P uptake efficiency during starvation was also partially contributed
340	by PHA synthesis (Figure 3), which provides an electron donor and energy source for
341	the DPAOs. Moreover, the higher amounts of PHA synthesis by nitrite-DPAOs than
342	those by nitrate-DPAOs during the endogenous period (Table 3) may accelerate the
343	anoxic metabolism of nitrite-DPAOs during the recovery period, further benefitting
344	the stable operation of shortcut denitrification P removal systems.
345	In the second recovery batch test of $AS_{ANA.end}$, the concentration of the released P
346	reached 48.0 mg PO_4^{3-} -P/L and 47.3 mg PO_4^{3-} -P/L in A-R _{NO3-} and A-R _{NO2-}
347	respectively, with a respective recovery percentage of 91.7% and 82.3%.
348	3.3.2 Recovery of nitrite- and nitrate-DPAOs in AS _{AER.end}
349	The VFA concentration in O- R_{NO3-} and O- R_{NO2-} (AS _{AER.end}) decreased from its initial
350	5.89 mmol C/L to 5.16 mmol C/L and 4.60 mmol C/L, respectively, at the end of the
351	anaerobic phases in the first recovery batch test (data not shown), representing a
352	decrease of 87.6% and 78.1%, respectively, compared with the amount of VFA
353	assimilated by their parents (SBR $_{NO3-}$ and SBR $_{NO2-}$) in the anaerobic phase. Indeed,
354	during the anaerobic phase of both parent SBRs, the added VFA were completely
355	depleted within 30 minutes (Figure S1).
356	The total amount of released P also sharply declined in $O-R_{NO3-}$ and $O-R_{NO2-}$, by

357	86.9% and 79.4% respectively, relative to their corresponding values in the typical
358	parent SBRs (Table 4). This result may be attributed to rapid degradation of poly-P
359	during starvation, leaving minimal quantities of poly-P in DPAOs cells starved for 12
360	days. During the subsequent anoxic phases, the amount of P uptake in O- R_{NO3-} and
361	O-R _{NO2-} was only 4.1 and 5.7 mg PO_4^{3-} -P/L respectively, indicating a severe
362	deterioration of P removal ability (Table 4). Nevertheless, the effluent P concentration
363	in O-R _{NO2-} remained at 1.3 mg PO ₄ ³⁻ -P/L, suggesting that activity was recovered more
364	rapidly in nitrite-DPAOs sludge than in nitrate-DPAOs sludge (Figure 5b and d). In
365	contrast to the largely diminished P removal, the denitrification efficiencies during the
366	anoxic phase were similar to those observed during typical cycles of the parent SBRs.
367	This result is likely due to the presence of ordinary heterotrophs (OHOs) and
368	denitrifying glycogen-accumulating microorganisms (DGAOs), which might use the
369	residual VFA of the anaerobic phase for anoxic denitrification.
370	4. Discussion
371	The endogenous processes of PAOs can be differentiated into two aspects: (i)
372	maintenance processes linked to utilization of intracellular polymers (mainly poly-P
373	and glycogen) and EPS, which maintain cellular integrity and activity; and (ii) decay
374	processes, which reduce the amount and/or activity of the active biomass (Lopez et al.,
375	2006; Hao et al., 2010b).

376	4.1. Cell maintenance in the nitrite/nitrate-DPAOs sludge
377	4.1.1 Energy and reducing equivalent sources from intracellular polymers
378	While poly-P is consensually regarded as a maintenance source during anaerobic
379	starvation, glycogen is an important energy pool as well as an equivalent reducing
380	source for the maintenance of PAOs (Lopez et al., 2006; Lu et al., 2007; Wang et al.,
381	2012; Vargas et al., 2013). Under anaerobic conditions, glycogen is usually processed
382	for maintenance through a combination of glycolysis and the PHA production
383	pathway (Lopez et al., 2006; Lu et al., 2007). The main reactions involved in
384	glycogen degradation and PHA formation in nitrite- and nitrate-DPAOs were
385	described in details in previous studies (Filipe et al. 2001; Zeng et al. 2003a). The
386	overall reaction (in terms of the molar relationships) is given by Eq. (2):
387	-glycogen + 1/6 PHB + 5/12 PHV + 1/4 PH2MV + 1/2ATP = 0 (2)
388	PHA synthesis in the present starvation test is summarized in Table 3. The ratios
389	among PHB, PHV and PH2MV in the nitrate-DPAOs sludge reactors are
390	well-predicted by Eq. (2). However, the proposed stoichiometry of anaerobic
391	maintenance does not adequately describe the composition of the PHA synthesized by
392	starved nitrite-DPAO sludges. The higher PH2MV content suggests that the
393	investigated nitrite-DPAO sludges adopt different survival metabolic pathways in
394	their endogenous processes. This may correlate with a larger fraction of
395	propionyl-CoA (precursor of PH2MV) produced from acetyl-CoA through the
396	methylmalonyl pathway, as reported by Yagci et al. (2003). Similarly, Oehmen et al.

397	(2006) analyzed the PHA composition in anaerobic test batch culture without
398	propionate addition, and obtained 12% PHB, 41% PHV and 47% PH2MV. They
399	attributed this result to enrichment of Alphaproteobacteria GAOs. The PHA synthesis
400	mechanism in nitrite-DPAOs during starvation is detailed in the SI (Text S3).
401	4.1.2 Maintenance substrate and energy sources from EPS
402	EPS is an important potential energy source that enables DPAOs to manage starvation
403	shock. In all reactors, EPS were almost completely depleted after 12 days of
404	starvation, and minimal non-biodegradable EPS remained in the sludge (Figure 4).
405	Especially, the easily biodegradable components of EPS (i.e., polysaccharides) were
406	rapidly utilized for maintenance during the initial starvation period, followed by
407	utilization of proteins (Figure 4a and b). Indeed, various extracellular enzymes have
408	been detected in activated sludge and biofilms, many of which can potentially degrade
409	EPS components to low-molecular-mass compounds that can then be utilized by
410	microorganisms as carbon and energy sources during starvation periods (Flemming et
411	al., 2010; Zhang and Bishop, 2003; Liu et al., 2007). The main degraders of EPS
412	polysaccharides are hydrolases and lyases (Laue et al., 2006), such as
413	N-acetyl-β-hexosaminidase (Kaplan et al., 2004). Metagenomic analysis of two EBPR
414	sludge communities has revealed the genes encoding enzymes for carbohydrate
415	polymer hydrolysis and subsequent monomer degradation pathways in the dominant
416	flanking species, namely, Flexibacter-like, Xylella-like and Thiothrix-like populations
417	(Martín et al., 2006). Martín et al. (2006) also reported that one of the EPS gene

418	cassettes in Accumulibacter encodes UDP-glucose dehydrogenase, which catalyzes
419	the precursor of the glucoronic acid component of EPS. Therefore, the dominant
420	flanking <i>Xylella</i> -like population may be able to degrade the glucoronic acid in
421	Accumulibacter EPS as a carbon and energy source. Additionally, EPS extracted from
422	biofilms were also found to be effectively degraded by their own producing
423	microorganism (Zhang and Bishop, 2003).
424	After 3 days of starvation, at least 90% less polysaccharides were degraded by
425	nitrite-DPAO sludges than by nitrate-DPAO sludges (Figure 4b). This result may be
426	ascribed to the relatively low initial polysaccharides content in the SBR_{NO2} sludge. It
427	should also be mentioned that because the degradation pathway of proteins is more
428	complicated than that of polysaccharides, it is not certain that the decreased proteins
429	(in EPS) contents had been used as substrate to produce energy by cells or only had
430	been hydrolyzed or broken down into its component, i.e., amino acids. Since amino
431	acids in the bulk were not analyzed during the starvation period, the exact amount of
432	EPS proteins being degraded as substrate cannot be accurately determined, and
433	requires further study.
434	4.1.3 Energy production from intracellular polymers and EPS
435	Maintenance energy is critical for the survival of nutrient-limited bacteria. PAOs
436	generally use poly-P and glycogen as maintenance energy sources under anaerobic
437	starvation conditions (Lopez et al., 2006; Lu et al., 2007). However, the main source
438	between them has yet to be clarified. Lu et al. (2007) found that, early in the

439	starvation period, PAOs prefer glycogen to poly-P as an energy source. In contrast,
440	Lopez et al. (2006) found that PAOs sequentially consume poly-P and glycogen,
441	which is supported by lack of any P released during the first day of starvation. No
442	glycogen degradation was observed during an 8-hour anaerobic starvation batch test
443	(Oehmen et al., 2005), supporting the assumption that PAOs use their stored poly-P as
444	the sole source of maintenance energy. Therefore, in the present study, we examined
445	the energy produced by both glycogen and poly-P. Moreover, the polysaccharides in
446	EPS were considered as a potential additional maintenance energy source when
447	calculating the maintenance ATP (Text S4).
448	As shown in Figure 6, glycogen and poly-P were simultaneously utilized during
449	the first 3 days of starvation, similar to our previous observations (Wang et al., 2012).
450	Notably, the polysaccharides in EPS also contributed a sizeable fraction of the
451	maintenance energy, especially in nitrate-DPAO sludge (Figure 6a and b). Hydrolysis
452	of 1 mol poly-P presumably yields 1 mol ATP, while degradation of 1 mol-C glycogen
453	and 1 mol-C polysaccharides should each generate 0.5 mol ATP (Smolders et al.,
454	1994). Maintenance energy production by DPAOs under anaerobic conditions was
455	calculated from the amounts of glycogen consumed and P released (Table 3 and
456	Figure 4).
457	In all reactors, most of the ATP was produced during the first 5 days of starvation.
458	Glycogen was the primary maintenance energy source throughout the first 1.5 days,
459	contributing more than approximately 70% to the total energy production in all four

460	reactors (Figure 6a-d). During day 1.5 – 3, poly-P was increasingly used as an
461	alternative maintenance energy source, and its use dominated (contributed over 50%
462	of the total ATP production) on day 5 in all four reactors (Figure 6a-d). Similar
463	observation was also reported by Lu et al. (2007), who found that glycogen
464	degradation provided the majority of the energy on day 1, after which there was a
465	transition of the primary energy source from glycogen to poly-P. Whereas, Lopez et al
466	(2006) observed a sequential utilization of poly-P and glycogen by PAOs for
467	maintenance energy production under anaerobic conditions. The reasons for this
468	discrepancy have not been presented. Moreover, as these two studies did not present
469	the data of the GAOs percentage, it is not easy to explain this discrepancy from the
470	microbial point of view (e. g., PAOs vs. GAOs). In the present study, we doubt that
471	this result might arise from the relatively high proportion of GAOs in both $\mathrm{SBR}_{\mathrm{NO3}}$.
472	(27 \pm 0.9% of the total biomass) and SBR_{NO2-} (19 \pm 0.6% of the total biomass) (Table
473	1). The major energy source of GAOs is glycogen, while poly-P hydrolysis supplies
474	energy for the dominant PAOs. In the nitrate-DPAOs sludges, polysaccharides
475	provided additional maintenance energy during the first 3 days of starvation, with
476	values of 16.2 and 20.3 ATP mmol/C-mol biomass for $AS_{ANA.end}$ and $AS_{AER.end}$,
477	accounting for 27% and 19% of the total energy production, respectively (Figure 6a
478	and b). After day 5, a marked decline in energy production was observed (Figure
479	6a-d). Such relatively low energy production may not meet the minimal energy
480	demands of DPAOs. The resulting cell lysis is evidenced in a sharp increase in

 NH_4^+ -N concentration (Figure 1).

482	During the 12-day starvation, energy production of nitrite-DPAO sludges in
483	$AS_{ANA.end}$ and $AS_{AER.end}$ were estimated to be 31% and 34% lower, respectively, than
484	those of nitrate-DPAO sludges (Figure 6a-d). The relatively low requirement for
485	maintenance energy of nitrite-DPAOs is further supported by the constant amounts of
486	EPS polysaccharides within the first 3 days (Figure 4). This result may reflect a robust
487	starvation survival response in nitrite-DPAOs. Following each nitrite addition, the
488	parent SBR _{NO2-} was exposed to episodes of famine (no nitrite was available during the
489	first 30 – 70 min of the anoxic period; see Figure S1b). Consequently, nitrite-DPAOs
490	relied on their internal poly-P and/or glycogen reserves for maintenance energy, as
491	evidenced by the irregular P release and glycogen consumption during the anoxic
492	phases of typical cycles (Figure S1d and f). Because of the frequent starvation
493	episodes, and the high number of nitrite/FNA shocks in the parent $\mbox{SBR}_{\rm NO2}$ -, the
494	nitrite-DPAOs were primed to adjust their endogenous mechanisms by slowing down
495	their self-oxidation rate and lowering their maintenance energy, i.e., consuming less
496	glycogen and poly-P, to survive under the imposed starvation conditions. Thus,
497	acclimation of nitrite-DPAOs to the starved conditions may allow them higher
498	adaptability to starvation shock than their nitrate-DPAO counterparts. In addition,
499	$AS_{AER.end}$ required more energy than $AS_{ANA.end}$ (Figure 6a-d), primarily owing to the
500	lower energy available from glycogen and poly-P in AS _{AER.end} .

- 501 **4.2 Decay process of the nitrite/nitrate-DPAO sludges**
- 502 Cell decay refers to any process that reduces the weight (negative growth) and the
- 503 specific activity of biomass. The cell decay rates in A-R_{NO3-}, O-R_{NO3-}, A-R_{NO2-} and
- 504 $O-R_{NO2}$ were estimated to be 0.022, 0.022, 0.008 and 0.007 day⁻¹, respectively (Table
- 505 2). The cell decay rates were much lower in nitrite-DPAO sludges than in
- 506 nitrate-DPAO sludges, and were comparable to the value reported by Lu et al. (2007)
- 507 (0.006 day⁻¹) for PAOs under anaerobic starvation conditions.
- 508 As mentioned in *Section 4.1.3*, frequent starvation episodes in the parent
- 509 nitrite-SBR might evoke stringent starvation responses in the nitrite-DPAOs sludge.
- 510 In other words, nitrite-DPAOs possess a better survival strategy via a relatively low
- 511 maintenance energy requirement to ensure their survival in nutrient-limited systems
- 512 (Salem et al., 2006). The decay rates of DPAOs in AS_{ANA.end} and AS_{AER.end} did not
- 513 appreciably differ in the present study.
- 514 **4.3. Recovery of nitrite/nitrate-DPAOs activities**
- 515 Transition from starvation to full functionality is essential for the survival of bacterial
- 516 systems (Lu et al. 2007). Vargas et al. (2013) found that once wastewater is
- 517 reintroduced, both PAOs and GAOs can recover their initial acetate uptake rates,
- 518 indicating strong survival ability during the starvation period. Yilmaz et al (2007) also
- 519 reported that the P-release and P-uptake in a starved culture were fully recovered
- 520 within 4 days of gradual re-introduction of influent wastewater. In the present study,
- 521 nitrite-DPAOs recovered from 12-day starvation within one day (Figure 5). These

522	findings well correspond with the lower maintenance energy requirement and decay
523	rate in nitrite-DPAO sludges than those in nitrate-DPAO sludges. In particular, for the
524	nitrite-DPAOs sludge, less (about 40%) PHA consumption for anoxic P uptake as
525	well as high PHA synthesis during the endogenous period than that of the
526	nitrate-DPAOs sludge (Table 3) may speed up their anoxic metabolism during the
527	recovery period, which thus provides an additional benefit to the stable operation of
528	shortcut denitrification P removal systems.
529	Microorganisms in AS _{ANA.end} rapidly recovered their intracellular polymer
530	transformation ability, especially for the nitrite-DPAOs in A-R _{NO2-} , where glycogen
531	synthesis and PHA degradation reached 179% and 99%, respectively, of their
532	pre-starvation levels (Figure 5g and Figure S1f). This observation is consistent with
533	the strong recovery of denitrifying P removal efficiency with the value of 136%
534	(Table 4), indicating a high activity of nitrite-DPAOs. The post-starvation nitrate
535	reduction rate in A-R _{NO3-} was approximately 35% less than the pre-starvation rate.
536	Since no external carbon was added in the anoxic phase during the first recovery
537	batch test, OHOs could not have been involved in nitrite/nitrate reduction. Thus,
538	nitrate denitrification was accomplished by PAOI and/or DGAOs (Zeng et al., 2003b).
539	PAOI can simultaneously reduce nitrate and remove anoxic P, but PAOII cannot use
540	nitrate as an electron acceptor (Flowers et al., 2009). Therefore, the reduced nitrate
541	denitrification rates suggest that starvation shock may inhibit the activity of PAOI
542	and/or DGAOs.

543	The metabolism of intracellular materials in $AS_{AER.end}$ was severely inhibited by
544	long-term starvation, as evidenced by the decrease in P uptake and release. Conversely,
545	both reactors achieved complete denitrification, probably because the abundant
546	propionate (2.89 and 3.22 mmol C/L for O-R _{NO3-} and O-R _{NO2-} , respectively) derived
547	from the preceding anaerobic phase stimulated the denitrification performance of
548	OHOs. However, the low effluent P concentration in $O-R_{NO2}$ confirmed that a fully
549	functional system could be restored by the relatively high fraction of PAOII (Table 1).
550	In summary, the relatively low consumption of intracellular polymers, slow cell
551	decay and rapid recovery of activity in nitrite-DPAO sludges (especially originating
552	from the anaerobic end phase of the DPR process) demonstrate a strong ability to
553	cope with starvation shock. These endogenous process characteristics of nitrite-DPAO
554	sludges, combined with 25% reduction in the oxidation cost and 40% reduction in
555	carbon consumption, may be exploited in efficient shortcut denitrification P removal.
556	5. Conclusions
557	(1) The anaerobic maintenance energy was approximately 30% lower in
558	nitrite-DPAOs sludge than that in nitrate-DPAOs sludge. Glycogen and poly-P
559	sequentially served as the primary maintenance energy sources in starved
560	nitrite/nitrate-DPAOs. The polysaccharides in EPS were rapidly consumed by
561	nitrate-DPAOs sludge during the first 3 days of starvation; conversely, nitrite-DPAOs
562	sludge converted less of these polysaccharides into maintenance energy.
563	(2) The estimated cell decay rates in A- R_{NO3-} , O- R_{NO3-} , A- R_{NO2-} and O- R_{NO2-} were

564	0.022, 0.022, 0.008, and 0.007 day ⁻¹ , respectively. Clearly, the cell decay rates were
565	lower in nitrite-DPAOs than those in nitrate-DPAOs, indicating a better stringent
566	starvation response by nitrite-cultivated DPAOs than by their nitrate-cultivated
567	counterparts.
568	(3) After 12 days of starvation, nitrite-DPAO sludges recovered more rapidly than
569	nitrate-DPAO sludges. The denitrifying P removal efficiencies, as well as the
570	transformation rates of intracellular polymers in the nitrite-DPAO sludges (especially
571	that from the end of the anaerobic phase) were almost identical to their pre-starvation
572	values, indicating a rapid return (within 1 day) to full functionality.
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- 678
- 679

680 Figure captions

- 681 **Fig. 1.** Variations in (a) NH_4^+ -N and (b) PO_4^{3-} -P concentrations in A-R_{NO3-}, O-R_{NO3-},
- $A-R_{NO2}$, and $O-R_{NO2}$ during the 12 days of anaerobic starvation.
- 683 Fig. 2. Variations in (a) MLSS, (b) MLVSS, and (c) MLVSS/MLSS ratio in A-R_{NO3-},
- $O-R_{NO3-}$, $A-R_{NO2-}$, and $O-R_{NO2-}$ during the 12 days of anaerobic starvation.
- **Fig. 3**. Variations in glycogen and PHA during the 12 days of anaerobic starvation: (a),
- 686 A-R_{NO3-}, (b), O-R_{NO3-}, (c), A-R_{NO2-} , and (d), O-R_{NO2-}.
- **Fig. 4.** EPS contents profile during the 12 days of anaerobic starvation in A- R_{NO3-} ,
- 688 O-R_{NO3}-, A-R_{NO2}-, and O-R_{NO2}-: (a) proteins, (b) polysaccharides, and (c) EPS
- 689 (calculated as the sum of proteins, polysaccharides and humics). Data are the averages
- 690 and their standard deviations in triplicate tests.
- **Fig. 5.** Variations in NOx⁻-N, $PO_4^{3^-}$ -P, glycogen and PHA in the recovery batch tests
- after 12 days of anaerobic starvation: (a), (e), $A-R_{NO3}$.; (b), (f), $O-R_{NO3}$.; (c), (g),
- 693 A- R_{NO2} ; and (d), (h), O- R_{NO2} .
- **Fig. 6.** ATP production profile during the 12 days of anaerobic starvation: (a), A-R_{NO3-},
- 695 (b), $O-R_{NO3-}$, (c), $A-R_{NO2-}$, and (d), $O-R_{NO2-}$.
- 696
- 697

propionate supplied a	s carbon sou	urce in a ty	pical cycle						/		
		FIS	SH quantificat	tion		PO4 ³⁻ -P	Gly/	PHA/	PHB/	PHV/	PH2MV/
	$Ac_{Tot}^{a}(\%)$	$Ac_{I}^{b}(\%)$	Ac_{II}^{c} (%)	<i>De</i> ^d (%)	<i>Co^e (%)</i>	/ VFA ^f	VFA ^g	VFA ^g	VFA ^g	VFA ^g	VFA ^g
This study								/			
SBR _{NO3-} (nitrate-based)	66 ± 1.5	28 ± 1.9	38 ± 1.4	27 ± 0.9	<1	0.32	0.44	1.54	0.08	0.66	0.80
SBR _{NO2-} (nitrite-based)	75 ± 1.1	21 ± 1.2	54 ± 1.7	19 ± 0.6	<1	0.36	0.31	1.46	0.11	0.61	0.74
Previous studies						\mathbf{X}					
Oehmen et al. (2005)	63 ± 1.3	-	-	<1	<1	0.42	0.32	1.23	0.04	0.55	0.65
Carvalho et al. (2007)	75	44	31	<1	<1	0.40	0.32	0.97	0	0.40	0.57
Tayà et al. (2013)	85	55	30	-	<10	0.34	0.49	1.47	-	-	-

Table 1. Comparison of the anaerobic carbon transformations, PO_4^{3-} -P release, and biomass compositions (± standard error) with propionate supplied as carbon source in a twicel cycle

^a Ac_{Tot}: Total Accumulibacter. ^b Ac I: Type I Accumulibacter. ^c Ac II: Type II Accumulibacter. ^dDe: Defluvicoccus. ^eCo: Competibacter.

CER

^fUnits PO₄³⁻-P mmol/C mmol. ^gUnits C mmol/C mmol. - No data.

698

699 700

	Starvation	Initial MLVSS	Temperature	Starvation	SRT	NH4 ⁺ -N	Cell decay rate
	condition	(g/L)	(*C)	period (d)	(d)	release (mg/L)	(1/d)
This study							
A-R _{NO3-}			~			41.8 ^a	0.022^{a}
O-R _{NO3-}	anaerohic	25	20 + 2	12	20	39.5 ^b	0.022 ^b
A-R _{NO2-}	unuerobie	2.0		12	20	18.7 ^a	0.008^{a}
O-R _{NO2-}						15.8 ^b	0.007 ^b
Previous studies			Z'				
Lu et al. (2007)	anaerobic	1.6	22 ± 2	8	24	3.2 ^b	0.006 ^b (day 4– 8)
Hao et al. (2010a)	anaerobic	-	22 ± 0.5	7	12	-	0.036 ± 0.003^{b}
Wang et al. (2012)	anaerohic	-2.1	10 15	7	20	1.12 ^a	0.0006 ^a
wally et al. (2012)		2.4	10 - 15	,	20	1.59 ^b	0.0008^{b}
Vargas et al. (2013)	aerobic-anaerobic	2.6	20 ± 2	21	46.8	14 ^b	0.029 ^b

Table 2. Comparison of cell decay of PAOs/DPAOs in this study and previous studies during the 12 days of anaerobic starvation.

^a All results represent the sludge from the anaerobic end phases.

^b All results represent the sludge from the aerobic end phases.

$\frac{1}{100} = 12 \text{ dups of underose starvation.}$					
Parameters	Nitrate-sludge		Nitrite-sludge		
i didilecers	Nitrate-sludge Nitrite-sludge A-R _{NO3} . O-R _{NO3} . A-R _{NO2} . O-1 770 1650 670 910 357 793 360 347 87.2 150.8 45.3 107 3.03 5.56 2.41 3.4 0.61 1.11 0.48 0.6 0.23 (0.12) 0.58 (0.19) 0.19 (0.07) 0.5 1.07 (0.58) 1.42 (0.46) 1.16 (0.42) 1.4 0.54 (0.29) 1.06 (0.35) 1.44 (0.52) 1.6 1.84 3.06 2.79 3.6 0.30 ^a 0.58 ^a 0.76 ^b 1.0 36.6 \pm 0.4 46.4 \pm 0.5 11.7 \pm 0.8 10 144.0 \pm 2.3 129.0 \pm 13.3 133.1 \pm 9.3 144	O-R _{NO2-}			
MLSS reduction (mg/L)	770	1650	670	910	
MLVSS reduction (mg/L)	357	793	360	347	
PO_4^{3-} -P release (mg/L)	87.2	150.8	45.3	102.7	
Glycogen degradation	2.02	5 56	2.41	2.14	
(mmol-C/g-MLVSS)	3.03	5.50	2.41	3.44	
Glycogen degradation rate ^a	0.61	1 11	0.49	0.00	
(mmol-C/g-MLVSS/d)	0.61	1.11	0.48	0.69	
PHB synthesis	0.22 (0.12)	0.59 (0.10)	0.10 (0.07)	0.54(0.15)	
(mmol-C/g-MLVSS)	0.25 (0.12)	0.38 (0.19)	0.19 (0.07)	0.34 (0.13)	
PHV synthesis	1.07 (0.59)	1 42 (0 46)	1.16 (0.42)	1 40 (0 20)	
(mmol-C/g-MLVSS)	1.07 (0.38)	1.42 (0.46)	1.16 (0.42)	1.40 (0.39)	
PH2MV synthesis	0.54 (0.20)	1.06 (0.25)	1 44 (0 52)	1 66 (0 16)	
(mmol-C/g-MLVSS)	0.34 (0.29)	1.00 (0.33)	1.44 (0.32)	1.00 (0.40)	
PHA synthesis	1 0/	2.06	2 70	2 61	
(mmol-C/g-MLVSS)	1.04	3.00	2.19	5.01	
PHA synthesis rate	0.20a	0.58 ^a	0.76 ^b	1.02 ^b	
(mmol-C/g-MLVSS/d)	0.50	0.38	0.70	1.02	
Polysaccharides consumption	26.6 + 0.4	464 05	117.09	10.1 ± 0.4	
(mg/g-MLVSS)	30.0 ± 0.4	40.4 ± 0.3	11.7 ± 0.8	10.1 ± 0.4	
Proteins consumption	144.0 ± 2.2	120.0 ± 12.2	122.1 ± 0.2	146.8 ± 1.5	
(mg/g-MLVSS)	144.0 ± 2.3	127.0 ± 13.3	133.1 ± 7.3	140.0 ± 1.3	
Total EPS consumption	196.0 ± 0.1	200.1 ± 9.0	162.4 ± 0.3	1776 + 72	
	170.0 ± 0.1	200.1 ± 9.0	102.4 ± 0.3	$1/1.0 \pm 1.2$	

Table 3. Transformations of materials in the liquid- and solid-phase in A-R_{NO3-}, O-R_{NO3-}, A-R_{NO2-} and O-R_{NO2-} during the 12 days of anaerobic starvation.

^a All results obtained during the first 5 days.

(mg/g-MLVSS)

^b All results obtained during the first 3 days.

Data in the brackets represent the fractions of the components in PHA.

typical cycles and recovery batch tests.								
Items	Amount of PO_4^{3-} -P release ^a	Amount of PO ₄ ³⁻ -P uptake ^a	Amount of NO _x -N reduction ^b	P/N ratio				
Nitrate-sl	udge							
SBR _{NO3-}	52.33	36.69	33.37	1.10				
A-R _{NO3-}	47.98 (91.7) *	19.37 (52.8)	19.36 (58.0)	1.00 (63.7)				

33.68 (100.9)

33

30 (90.9)

33 (100.0)

0.12 (7.6)

1.20 (69.0)

0.17 (9.8)

0.80

4.09 (11.1)

35.97 (136.0)

5.73 (21.7)

26.44

Table 4. Comparison of denitrifying P-removal efficiency during anaerobic/anoxic phases in typical cycles and recovery batch tests.

706	^a Units mg PO ₄ ³⁻ -P/L.

O-R_{NO3-}

SBR_{NO2-}

A-R_{NO2-}

O-R_{NO2-}

Nitrite-sludge

707 ^b Units mg NO_x^--N/L .

708 * Results obtained from the second recovery batch tests.

709 Data in brackets represent the percentage recovery relative to values obtained in typical cycles of

710 the corresponding parent SBRs (%).

6.87 (13.1)

47.30 (82.3) *

11.82 (20.6)

57.48

711

712

713

714













Fig. 5



746 747	Supplementary Information
748	Title: Comparison of endogenous metabolism during long-term anaerobic
749	starvation of nitrite/nitrate cultivated denitrifying phosphorus removal sludges
750	
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760	
761	Text S1. The detailed descriptions of analysis of VFA, PHA, glycogen, and EPS
762	a) VFA
763	VFA (propionate) was measured using an Agilent 6890N gas chromatograph (GC)
764	equipped with a 30 m×0.53 mm×1 μm (length×ID×film) DB-WAXetr column and a
765	flame ionization detector (FID) at 220 °C.
766	b) PHA

767	PHA, including poly- β -hydroxybutyrate (PHB), poly- β -hydroxyvalerate (PHV),
768	poly-3-hydroxy-2-methylvalerate (PH2MV), was measured according to a method
769	described by Oehmen et al. (2005). Briefly, approximately 20 mg freeze-dried
770	samples of biomass were put into screw-topped glass tubes, and 2 mL of chloroform,
771	2 mL of an acidified methanol solution (10% H_2SO_4) and 0.1mL benzoic acid
772	methanol solution (2 g of benzoic acid dissolved in 100mL methanol used as an
773	internal standard) were subsequently added. The samples were then digested for 4h at
774	100 °C. After cooling, 1 mL of distilled water was added and mixed vigorously with
775	each sample. Thereafter, 1 h of settling time was allowed to achieve phase separation.
776	When the phases were separated, approximately 1 mL of the bottom organic layer was
777	transferred to the GC vials for analysis. 3 μ L of the chloroform phase was analyzed
778	with a gas chromatograph (Thermo Focus GC). The chromatograph was operated with
779	a HP-5 column (30 m length×0.32 mm ID×0.25 μ m film), a split injection ratio of
780	1:15 and helium as the carrier gas (1.5 mL/min). A flame ionisation detection (FID)
781	unit was operated at 250 °C with an injection port temperature of 230 °C. The oven
782	temperature was set to 80 °C for 4 min, increased at 8 °C /min to120 °C, and then to
783	220 °C at 30 °C /min and held for 2 min.
784	c) Glycogen
785	Glycogen was determined by the anthrone method (Frølund et al., 1996). A 5-ml

volume of 0.6 M HCl was added to weighed (approximately 10 mg), freeze-dried

787 biomass in screw-topped glass tubes, and then heated at 100 °C for 3 h. After cooling

788	to room temperature, samples were sheared by a vortex mixer (XW-80A,
789	Shanghai Qingpu Huxi Instrument Co., Ltd, China) for 1 min, and were transferred to
790	10-mL tubes, followed by centrifugation (CT15RT, Techcomp, China) for 5 min at
791	10 000 g. About 1 mL supernatant was added to 4 mL of anthrone-H ₂ SO ₄ reagent (0.2%
792	anthrone (w/v) in 80% (v/v) H_2SO_4) in 10-mL colorimetric tubes. All tubes were
793	placed in a water bath at 100 °C for 10 min. After cooling at 4 °C for 5 min in cold
794	water, samples were measured by a UV/VIS spectrophotometer (UV765, Shanghai
795	Precision & Scientific Instrument Co., Ltd, China) at 625 nm. Glucose was used as
796	the standard.
797	d) EPS
798	A heat extraction method (Li et al., 2007) was applied to extract the EPS (including
799	loosely bound EPS (LB-EPS) and tightly bound EPS (TB-EPS)) in the sludge. 30 mL
800	sludge suspension was first dewatered by centrifugation (CT15RT, Techcomp, China)
801	in a 50-mL tube at 4 000 g for 5 min. The supernatant was recovered for water quality
802	analysis. The sludge pellet in the tube was re-suspended into 15 mL of 0.05% NaCl
803	solution. The sludge mixture was then diluted with the NaCl solution (pre-heating to
804	70 °C) to its original volume of 30 mL. Immediately, the sludge suspension was
805	sheared by a vortex mixer (XW-80A, Shanghai Qingpu Huxi Instrument Factory,
806	China) for 1 min, followed by centrifugation at 4 000 g for 10 min. The organic
807	matter in the supernatant was readily extractable EPS, and was regarded as the
808	LB-EPS of the biomass. For the extraction of the TB-EPS, the sludge pellet left in the

809	centrifuge tube was re-suspended in 0.05% NaCl solution to its original volume of 30
810	mL. The sludge suspension was heated to 60 °C in a water bath for 30 min. After
811	cooling to room temperature, the sludge mixture was centrifuged at 4 000 g for 15
812	min. The supernatant collected was regarded as the TB-EPS of the sludge.
813	Both the LB-EPS and TB-EPS extractions were analyzed for proteins (PN),
814	polysaccharides (PS) and humic-like substances (HS). The PN and HS were analyzed
815	by a UV/VIS spectrophotometer (UV765, Shanghai Precision & Scientific Instrument
816	Co., Ltd, China) following the modified Lowry method (Frølund et al., 1995) using
817	bovine serum albumin (Sigma) and humic acid (Fluka) as the standards, respectively.
818	The PS content was determined by the anthrone method described by Frølund et al.
819	(1996) using glucose as the standard.
820	Ctrin ,

821

Text S2. Fluorescent in situ hybridization (FISH) 822 Fluorescence in situ hybridization (FISH) was performed as described by Amann 823 (1995) using Cy5-labelled EUBMIX probes, which are specific for most Bacteria 824 (Daims et al., 1999). The Cy3-labelled PAOMIX probe (PAO462, PAO651 and PAO846) were used to target Candidatus Accumulibacter phosphatis (Crocetti et al., 825 2000), a known PAO; the Cy3-labelled GAOMIX probe (GAOQ431, GAOQ989 and 826 GB_G2), DFMIX probe (TFO_DF 218 and TFO_DF 618, and DF 988 and DF 1020) 827 were used to target Candidatus Competibacter phosphatis, Defluvicoccus-related TFO 828 829 (Cluster I) in Alphaproteobacteria, and Defluvicoccus-related DF (Cluster II) in Alphaproteobacteria (Crocetti et al., 2002; Kong et al., 2002; Wong et al., 2004) to 830 831 determine the microbial population distributions of the PAOs and GAOs (Table S1). 832 All oligonucleotide probes were commercially synthesized, and all hybridization 833 buffer contained 35% (v/v) formamide. Cy5-labeled Acc-I-444 was used to target PAOI Accumulibacter, while FAM-labeled Acc-II-444 was used to target PAOIIA, 834 835 IIC and IID Accumulibacter (Flowers et al. 2009). 836 FISH preparations were visualized using a Zeiss LSM 510 Meta confocal 837 laser-scanning microscope (CLSM) using a Plan-Apochromat 63×oil (NA1.4) 838 objective. Thirty images were taken from each sample for quantification. The 839 percentage of Accumulibacter, Competibacter and Defluvicoccus in the entire 840 bacterial population (EUBMIX) was determined via FISH image analysis using image analyzing software (Image-Pro Plus, V6.0, Media Cybernetics). The standard error of 841

842 the mean (SE_{mean}) was calculated as the standard deviation divided by the square root

843 of the number of images.

Supplementary Table S1.	Oligonucleotide pro	bes used for fluoresce	nce in situ h	vbridization (FISH)
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Probe	Probe sequence (5'-3')	Specificity	FA (%)	Reference
EUB 338	GCTGCCTCCCGTAGGAGT	Most Bacteria	35	Daims et al., 1999
EUB 338-II	GCAGCCACCCGTAGGTGT	Planctomycetales and other	35	Daims et al., 1999
		Bacteria		
EUB 338-III	GCTGCCACCCG AGGAGT	Verrucomicrobiales and other	35	Daims et al., 1999
		Bacteria		
PAO462	CCGTCATCTACWCAGGGTATTAAC	Most Accumulibacter spp.	35	Crocetti et al., 2000
PAO651	CCC TCT GCC AAA CTC CAG	Most Accumulibacter spp.	35	Crocetti et al., 2000
PAO846	GTTAGCTACGGCACTAAAAGG	Most Accumulibacter spp.	35	Crocetti et al., 2000
Acc-I-444	CCCAAGCAATTTCTTCCCC	Most Accumulibacter I	35	Flowers et al., 2009
Acc-II-444	CCCGTGCAATTTCTTCCCC	Most Accumulibacter IIA, IIC	35	Flowers et al., 2009
		and IID		
ALF969	TGGTAAGGTTCTGCGCGT	Most Alphaproteobacteria	35	Crocetti et al., 2002
GAOQ431	TCCCCGCCTAAAGGGCTT	Some Competibacter spp.	35	Crocetti et al., 2002
GAOQ989	TTCCCCGGATGTCAAGGC	Some Competibacter spp.	35	Crocetti et al., 2002
GB_G2	TTCCCCAGATGTCAAGGC	Some Competibacter spp.	35	Kong et al., 2002
TFO_DF218	GAAGCCTTTGCCCCTCAG	Some Defluvicoccus vanus	35	Wong et al., 2004
TFO_DF618	GCCTCACTTGTCTAACCG	Some Defluvicoccus vanus	35	Wong et al., 2004
DF988	GATACGACGCCCATGTCAAGGG	Some Defluvicoccus vanus spp.	35	Meyer et al., 2006
DF1020	CCGGCCGAACCGACTCCC	Some Defluvicoccus vanus spp.	35	Meyer et al., 2006

844

AC'

845 Text S3. The detailed explanation for the PHA synthesis mechanism during

846 anaerobic starvation

847	The PHA composition depends greatly on whether PAOs selectively or randomly
848	condense activated acetyl-CoA and propionyl-CoA to form PHA. Oehmen et al. (2006)
849	proposed that the stoichiometry of PAOs fed with propionate were closely correlated
850	with the model based on selective condensation of activated acetyl-CoA and
851	propionyl-CoA, while GAOs tend to randomly condense activated acetyl-CoA and
852	propionyl-CoA in PHA formation. Lemos et al. (2003) concluded that the notable
853	difference in the components of PHA in various studies is probably due to different
854	populations and/or metabolisms. Because there were more GAOs (27 \pm 0.5%) in
855	nitrate-DPAOs, a better prediction was obtained by Eq. (2) (Section 4.1.1), which was
856	based on the random condensation pattern. Conversely, the relatively high PH2MV
857	content means a larger amount of activated propionyl-CoA production and/or a more
858	preferentially selective binding together of activated propionyl-CoA molecules. The
859	exact underlying reason for this requires further investigation.
860	

- 861 Text S4. The maintenance energy calculation of poly-P, glycogen and
- 862 polysaccharides in the EPS
- 863 The maintenance energy production by DPAOs was calculated from the amounts of
- 864 glycogen consumption and poly-P hydrolysis, based on the assumption of hydrolysis
- of 1 mol poly-P yielding 1 mol ATP, and degradation of 1 mol-C glycogen producing
- 866 0.5 mol ATP (Smolders et al., 1994).

- The content of polysaccharides in EPS was also measured using the anthrone method (Frølund et al., 1996), the same as that of glycogen, which is determined by total carbohydrates using glucose as the standard. 1 mol-C polysaccharides in EPS is deduced to produce 0.5 mol ATP, according to the existed conclusion that degradation of 1 mol-C glycogen produces 0.5 mol ATP (Smolders et al., 1994).
- 872

Figure captions

- **Figure S1** Variations in nutrients and carbon profiles during a typical cycle in
- SBR_{NO3-} and SBR_{NO2-}

879 Fig. S1



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Highlights (85 characters)

► Anaerobic starvation (12d) and recovery of nitrite-DPAO was studied for the first time

► EPS polysaccharides were an additional maintenance energy source for DPAO' survival

Maintenance energy and cell decay were lower for nitrite- than nitrate-DPAO sludge

▶ Nitrite-DPAO had better stringent response to the starvation than nitrate-DPAO

► Nitrite-DPAO sludge had faster starvation recovery than nitrate-DPAO sludge