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Bender, Philipp; Venero, Diego Alba; Barquín, Luis Fernández ; Costo, Rocío; Hansen, Mikkel Fougt; Frandsen, Cathrine; Fock, Jeppe; Rogers, Sarah; Svedlindh, Peter; Wetterskog, Erik; Johansson, Christer

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Mössbauer, SANS and magnetic characterization of interacting iron oxide nanoparticles (IONPs)



Philipp Bender¹, Diego Alba Venero¹, Luis Fernández Barquín¹, Rocío Costo², Mikkel Fougt Hansen³, Cathrine Frandsen³, Jeppe Fock ³, Sarah Rogers ⁴, Peter Svedlindh ⁵, Erik Wetterskog ⁵, Christer Johansson ⁶

¹ CITIMAC, Universidad de Cantabria, Santander, Spain ² Instituto de Ciencia de Materiales de Madrid, CSIC, Spain ³ Department of Physics/DTU Nanotech, Technical University of Denmark ⁴ ISIS-STFC, Rutherford Appleton Laboratory, United Kingdom ⁵ Faculty of Technology, Uppsala University, Sweden ⁶ Acreo Swedish ICT AB, Göteborg, Sweden

Introduction

Magnetization behavior of ensembles of Iron Oxide Nanoparticles (IONPs) primarily depends on their structural as well as magnetic properties + interparticle interactions. Presence of dipolar interactions has significant goal of the **NanoMag** project. implications in biomedical applications [1].

moment distribution, dipolar interactions) driving those applications is a need for the future and

[EC FP7 NMP project, grant no. 604448]

0.1

0.1

Standardizing the determination of magnetic **Current study:** IONPs with different amounts of parameters (e.g. chemical composition, magnetic dipolar interactions are characterized by Mössbauer, SANS (structural properties) and combined M(H)/ZFC-FC-measurements (magnetic properties) to detect the influence of dipolar interactions on their magnetization behavior.

Samples



100 nm

- IONPs surrounded with rigid silica shell.
- $d_{c} = 11 \, \text{nm}, \, \sigma_{dc} = 0.7 \, \text{nm}$ • Shell prevents agglomeration of IONPs. \rightarrow Shell thickness determined by SANS.

• Comparable *d_c* as S1 but agglomerates. $d_{c} = 10 \, \text{nm}, \, \sigma_{dc} = 0.7 \, \text{nm}$ • On average dipolar interactions should be higher than in S1 due to agglomerates.



Adjusting scattering intensity with core-shell-model [2] enables exact determination of core (d_c) and shell (d_s) size distribution of S1:

 $d_c = 11 \,\mathrm{nm}, \,\sigma_{dc} = 1 \,\mathrm{nm}$ $d_s = 26 \text{ nm}, \sigma_{ds} = 7 \text{ nm}$

Distance between magnetic cores in S1 is large enough so that dipolar interactions should be negligible!

Results: Mössbauer



Results: Magnetic characterization

i) *M(H)* curves of dispersions

S1 powder

with Langevin function, Fit assuming a normal distribution of the core diameters d_c .

q [nm⁻¹]

Fit	results	S1	S2
d _c	[nm]	11.1(1)	9.7(1)



i) Influence of sample preparation

S1: Mössbauer data of dispersion (black), powder (red) and gel (blue) at 250 K show no influence of sample preparation.

ii) Iron oxide composition

Maghemite/magnetite composition analysis using the mean isomer shift

Reduced M_s values can be attributed to spin canting at particle surface [6].

ii) Simulation of ZFC curves including *d_c* distributions [7]

With: $M_s(T) = M_o(1-BT^{3/2})$ [6], $K = K_v + 6K_s/d_c$ [8] K_v: Ansiotropy constant of particle volume, K_s: surface anisotropy constant

S1: Curve can be simulated by introducing a surface anisotropy constant K_s. S2: Curve is shifted probably due to dipolar interactions [9].

iii) Distribution of blocking temperature T_B [7]: $P(T_B) \propto -d(\Delta M)/[TdT]$

[3] corrected for 2nd order Doppler shift gives that main iron oxide is maghemite (at% Fe in magnetite: 25(5) for S1 and O(5) for S2).

iii) Relaxation – size effect / dipolar interactions

S1: Using K=9.05 kJ/m³, V= $\pi/6 \cdot d_c^3$, $\tau_0 = 1$ ns and a measuring time of $\tau \approx$ 5 ns, we estimate a blocking temperature $T_B = KV/[k_B \ln(\tau/\tau_0)] \approx 300K$. Experimentally, we find $T_B \approx 300$ K (data not shown) consistent with the simple estimate. A more accurate analysis is topic for future work

S2: We observe a lower blocking temperature than for S1, possible due to the slightly smaller size or due to dipolar interactions [4,5].

References [1] I. Andreu et al., JMMM 380, 341 (2015) [2] J. Pedersen, Adv. Coll. Interface Sci. 70, 171 (1997) [3] Da Costa et al, J. Phys. Chem. B **118** 11738 (2014) [6] T. Shendruk et al., *Nanotechnology* **18**, 455704 (2007) [4] S. Mørup, E. Tronc, *PRL* **72**, 3278 (1994) [5] S. Mørup, *Europhys. Lett.* **28** 671 (1994)) [7] F. Tournus et al., JMMM 323, 1109 (2011) [8] N. Pérez et al., Nanotechnology 19, 475704 (2008) [9] A. Hillion et al., IEEE Trans. On Mag. 47, 3154 (2011)