



**UNIVERSITI PUTRA MALAYSIA**

**PREPARATION AND CHARACTERIZATION OF  $\text{Bi}_2\text{O}_3\text{-M}_2\text{O}_5$  ( $\text{M} = \text{P}, \text{AS}, \text{V}$ ) OXIDE ION CONDUCTORS**

**LEE SIEW LING.**

**FSAS 2004 25**

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Bi<sub>2</sub>O<sub>3</sub>-M<sub>2</sub>O<sub>5</sub> (M = P, As, V) OXIDE ION CONDUCTORS**

**By**

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**Thesis Submitted to the School of Graduate Studies, Universiti Putra Malaysia,  
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**May 2004**



**for Bon and family with love**



Abstract of thesis presented to the Senate of Universiti Putra Malaysia in fulfilment of  
the requirement of the degree of Doctor of Philosophy

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 $\text{Bi}_2\text{O}_3\text{-M}_2\text{O}_5$  ( $\text{M} = \text{P, As, V}$ ) OXIDE ION CONDUCTORS**

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**Chairman: Professor Lee Chnoong Kheng Ph.D.**

**Faculty: Science and Environmental Studies**

$\text{Bi}_2\text{O}_3\text{-M}_2\text{O}_5$ ,  $\text{M} = \text{P, As, V}$  systems and related materials were prepared by solid state reactions. The phase purity of the materials was determined by X-ray diffraction (XRD). Further characterization using ac impedance spectroscopy and differential thermal analysis (DTA) were carried out on single phase materials. Besides, inductively coupled plasma-atomic emission spectrometry (ICP-AES), density measurement, thermogravimetric analysis (TGA), differential scanning calorimetry (DSC), scanning electron microscopy (SEM), Fourier-transform infrared (FT-IR) spectroscopy and Raman spectroscopy were also performed on selected materials. The crystal system and space group of the single phase materials were determined.

Two narrow solid solution series were formed in  $x\text{Bi}_2\text{O}_3\text{-P}_2\text{O}_5$ :  $5.5 \leq x \leq 6$  and  $7 \leq x \leq 7.25$ . In DTA study, a phase transition was clearly seen in  $\text{Bi}_7\text{PO}_{13}$  and  $\text{Bi}_{29}\text{P}_4\text{O}_{53.5}$  at  $\sim 860^\circ\text{C}$ .

XRD shows that single phase materials were formed in  $x\text{Bi}_2\text{O}_3\text{-As}_2\text{O}_5$  binary system when  $x = 5, 5.5, 5.667, 5.75, 6$  and  $7$ . Among these, materials in the composition range of  $5 \leq x \leq 6.25$  appeared to be solid solutions. Attempts to synthesize materials of composition of  $x\text{Bi}_2\text{O}_3\text{-As}_2\text{O}_5$ ,  $1 \leq x \leq 4$  were unsuccessful.

Single phase materials were formed in  $x\text{Bi}_2\text{O}_3\text{-V}_2\text{O}_5$  binary system,  $5 \leq x \leq 6$  and  $x = 7$ . A phase transition was observed in  $\text{Bi}_{17}\text{V}_3\text{O}_{33}$  and  $\text{Bi}_{23}\text{V}_4\text{O}_{44.5}$  at  $\sim 180^\circ\text{C}$ . However, its origin is unknown.

Materials of composition  $x\text{Bi}_2\text{O}_3\text{-M}_2\text{O}_5$ ,  $5.5 \leq x \leq 6$  ( $\text{M} = \text{P}$ ) and  $5 \leq x \leq 6$  ( $\text{M} = \text{As}, \text{V}$ ) are refined in triclinic symmetry with space group of *P-1*. Meanwhile, monoclinic symmetry was found in materials where  $x = 7, 7.25$  ( $\text{M} = \text{P}$ ) and  $x = 7$  ( $\text{M} = \text{As}, \text{V}$ ). The XRD and IR patterns of both series of  $x\text{Bi}_2\text{O}_3\text{-As}_2\text{O}_5$ ,  $5 \leq x \leq 6.25$  and  $x\text{Bi}_2\text{O}_3\text{-V}_2\text{O}_5$ ,  $5.5 \leq x \leq 6$  solid solutions are very similar since these materials are isostructural. Generally, lattice parameters, volumes and densities of the materials in  $x\text{Bi}_2\text{O}_3\text{-M}_2\text{O}_5$  system,  $\text{M} = \text{P}, \text{As}, \text{V}$  increased with the increase of Bi content.

A complete solid solution series was formed in the  $\text{Bi}_{22}\text{P}_4\text{O}_{43}\text{-Bi}_{22}\text{As}_4\text{O}_{43}$ ,  $\text{Bi}_{22}\text{P}_4\text{O}_{43}\text{-Bi}_{22}\text{V}_4\text{O}_{43}$ ,  $\text{Bi}_{22}\text{As}_4\text{O}_{43}\text{-Bi}_{22}\text{V}_4\text{O}_{43}$ ,  $\text{Bi}_{23}\text{P}_4\text{O}_{44.5}\text{-Bi}_{23}\text{As}_4\text{O}_{44.5}$ ,  $\text{Bi}_{23}\text{P}_4\text{O}_{44.5}\text{-Bi}_{23}\text{V}_4\text{O}_{44.5}$ ,  $\text{Bi}_{23}\text{As}_4\text{O}_{44.5}\text{-Bi}_{23}\text{V}_4\text{O}_{44.5}$ ,  $\text{Bi}_{12}\text{P}_2\text{O}_{23}\text{-Bi}_{12}\text{As}_2\text{O}_{23}$ ,  $\text{Bi}_{12}\text{P}_2\text{O}_{23}\text{-Bi}_{12}\text{V}_2\text{O}_{23}$ ,  $\text{Bi}_{12}\text{As}_2\text{O}_{23}\text{-Bi}_{12}\text{V}_2\text{O}_{23}$  and  $\text{Bi}_7\text{AsO}_{13}\text{-Bi}_7\text{VO}_{13}$  systems. In  $\text{Bi}_7\text{PO}_{13}\text{-Bi}_7\text{AsO}_{13}$  and  $\text{Bi}_7\text{PO}_{13}\text{-Bi}_7\text{VO}_{13}$  systems a two-phase region was seen. All the single phase materials studied above appeared to be oxide-ion conductors. Conductivity increased with increasing vanadium

content, followed by arsenic and phosphorus. Among the materials prepared, the highest conductivity is obtained in  $\text{Bi}_{23}\text{V}_4\text{O}_{44.5}$  with a  $\sigma$  value of  $1.34 \times 10^{-4} \text{ ohm}^{-1} \text{ cm}^{-1}$  at  $300^\circ\text{C}$ . In an attempt to optimize oxide ion conductivity, chemical doping using  $\text{PbO}$ ,  $\text{Sr}(\text{NO}_3)_2$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{Ga}_2\text{O}_3$ ,  $\text{La}_2\text{O}_3$ ,  $\text{Fe}_2\text{O}_3$  etc. was carried out in selected materials, resulting in the formation of limited solid solutions. These materials, however, exhibit conductivity slightly lower than that of the parent materials.

Ball milling process has been carried out in the preparation of  $\text{Bi}_{23}\text{V}_4\text{O}_{44.5}$  and  $\text{Bi}_{14}\text{PO}_8$  in addition to manual grinding prior to firing of the samples. In ball milling process, high-density, fine-grained powders with uniform grain-size distribution were obtained, resulting in an increase in conductivity and dielectric constants.

Sillenite compounds in the  $\text{Bi}_2\text{O}_3\text{-P}_2\text{O}_5$  binary system with Bi:P ratios of 13:1 to 16:1 have been synthesized and found to be solid solutions. Substitution of P by V and As in the material where Bi:P = 14:1 results in partial and complete solid solutions, respectively. Enhancement in conductivity was observed in these solid solutions with V-doped materials exhibiting the highest conductivity. Substitution of P by elements such as  $\text{Pb}^{2+}$ ,  $\text{Sr}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Ga}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Si}^{4+}$ ,  $\text{Ge}^{4+}$ , and  $\text{Ti}^{4+}$  leads to formation of limited solid solutions. Most of these materials have conductivity similar to or slightly higher than that of the parent compound. These materials appeared to be predominantly oxide ion conductors especially at temperatures above  $800^\circ\text{C}$  where  $\gamma \rightarrow \delta'$  polymorphic transformation occurred.

**Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia  
sebagai memenuhi keperluan untuk ijazah Doktor Falsafah**

**PENYEDIAAN DAN PENCIRIAN KONDUKTUR ION OKSIDA  
 $\text{Bi}_2\text{O}_3\text{-M}_2\text{O}_5$  (M = P, As, V)**

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Sistem-sistem  $\text{Bi}_2\text{O}_3\text{-M}_2\text{O}_5$ , M = P, As, V dan bahan-bahan berkaitan telah disediakan melalui tindak balas keadaan pepejal. Ketulenan fasa bahan tersebut dikenalpastikan dengan penggunaan pembelauan sinar-X (XRD). Pencirian lanjutan telah dijalankan ke atas sampel yang berfasa tunggal dengan menggunakan spektroskopi impedans arus ulang-alik dan penganalisa pembezaan terma (DTA). Di samping itu, analisis-analisis termasuk analisis plasma induktif pasangan-pemancaran pengatoman (ICP-AES), penyukatan ketumpatan, analisis termogravimetri (TGA), kalorimetri pembezaan imbasan (DSC), mikroskopi imbasan elektron (SEM), spektroskopi infra merah transformasi Fourier (FT-IR) dan spektroskopi Raman telah dijalankan ke atas sampel-sampel terpilih. Sistem hablur dan kumpulan ruangan bagi sample-sampel yang berfasa tunggal juga dikenalpastikan.

Larutan pepejal telah terbentuk dalam sistem  $x\text{Bi}_2\text{O}_3\text{-P}_2\text{O}_5$ :  $5.5 \leq x \leq 6$  dan  $7 \leq x \leq 7.25$ .

Dalam kajian DTA, suatu peralihan fasa didapati dalam  $\text{Bi}_7\text{PO}_{13}$  dan  $\text{Bi}_{29}\text{P}_4\text{O}_{53.5}$  pada  $\sim 860^\circ\text{C}$ .

XRD menunjukkan bahan berfasa tunggal telah terbentuk dalam sistem  $\text{Bi}_2\text{O}_3\text{-As}_2\text{O}_5$  apabila  $x = 5, 5.5, 5.667, 5.75, 6$  and  $7$ . Bahan-bahan dengan komposisi  $5 \leq x \leq 6.25$  wujud sebagai pepejal larutan. Percubaan untuk menyintesiskan komposisi  $x\text{Bi}_2\text{O}_3\text{-As}_2\text{O}_5$ ,  $1 \leq x \leq 4$  telah mengalami kegagalan.

Bahan berfasa tunggal terbentuk dalam sistem  $x\text{Bi}_2\text{O}_3 : \text{V}_2\text{O}_5$ ,  $5 \leq x \leq 6$  dan  $x = 7$ . Suatu peralihan fasa telah didapati dalam  $\text{Bi}_{17}\text{V}_3\text{O}_{33}$  dan  $\text{Bi}_{23}\text{V}_4\text{O}_{44.5}$  pada  $\sim 180^\circ\text{C}$ . Tetapi, sifat asalnya tidak diketahui.

Sistem hablur bagi bahan-bahan berkomposisi  $x\text{Bi}_2\text{O}_3\text{-M}_2\text{O}_5$ ,  $5.5 \leq x \leq 6$  ( $\text{M} = \text{P}$ ) dan  $5 \leq x \leq 6$  ( $\text{M} = \text{As}$ ,  $\text{V}$ ) adalah bersimetri triklinik dengan kumpulan ruangan  $P-1$ . Sementara, sistem hablur bersimetri monoklinik didapati apabila  $x = 7, 7.25$  ( $\text{M} = \text{P}$ ) dan  $x = 7$  ( $\text{M} = \text{As}$ ,  $\text{V}$ ). Pola XRD dan IR bagi kedua-dua siri pepejal larutan  $x\text{Bi}_2\text{O}_3\text{-As}_2\text{O}_5$ ,  $5 \leq x \leq 6.25$  dan  $x\text{Bi}_2\text{O}_3\text{-V}_2\text{O}_5$ ,  $5.5 \leq x \leq 6$  adalah mirip kerana mereka memiliki struktur yang serupa. Secara amnya, parameter kekisi, isipadu dan ketumpatan bahan-bahan dalam sistem  $x\text{Bi}_2\text{O}_3\text{-M}_2\text{O}_5$ ,  $\text{M} = \text{P}, \text{As}, \text{V}$  meningkat dengan penambahan kandungan Bi.

Siri larutan pepejal yang lengkap terbentuk dalam sistem-sistem  $\text{Bi}_{22}\text{P}_4\text{O}_{43}$ - $\text{Bi}_{22}\text{As}_4\text{O}_{43}$ ,  $\text{Bi}_{22}\text{P}_4\text{O}_{43}$ - $\text{Bi}_{22}\text{V}_4\text{O}_{43}$ ,  $\text{Bi}_{22}\text{As}_4\text{O}_{43}$ - $\text{Bi}_{22}\text{V}_4\text{O}_{43}$ ,  $\text{Bi}_{23}\text{P}_4\text{O}_{44.5}$ - $\text{Bi}_{23}\text{As}_4\text{O}_{44.5}$ ,  $\text{Bi}_{23}\text{P}_4\text{O}_{44.5}$ - $\text{Bi}_{23}\text{V}_4\text{O}_{44.5}$ ,  $\text{Bi}_{23}\text{As}_4\text{O}_{44.5}$ - $\text{Bi}_{23}\text{V}_4\text{O}_{44.5}$ ,  $\text{Bi}_{12}\text{P}_2\text{O}_{23}$ - $\text{Bi}_{12}\text{As}_2\text{O}_{23}$ ,  $\text{Bi}_{12}\text{P}_2\text{O}_{23}$ - $\text{Bi}_{12}\text{V}_2\text{O}_{23}$ ,  $\text{Bi}_{12}\text{As}_2\text{O}_{23}$ - $\text{Bi}_{12}\text{V}_2\text{O}_{23}$  dan  $\text{Bi}_7\text{AsO}_{13}$ - $\text{Bi}_7\text{VO}_{13}$ . Dalam sistem-sistem  $\text{Bi}_7\text{PO}_{13}$ - $\text{Bi}_7\text{AsO}_{13}$  dan  $\text{Bi}_7\text{PO}_{13}$ - $\text{Bi}_7\text{VO}_{13}$ , dua fasa yang berlainan telah dikesan. Bahan-bahan berfasa tunggal yang dikaji merupakan konduktor ion oksida. Kekonduksian meningkat dengan penambahan kandungan vanadium, diikuti oleh arsenik dan fosforus. Di antara bahan-bahan yang disediakan,  $\text{Bi}_{23}\text{V}_4\text{O}_{44.5}$  telah menunjukkan kekonduksian yang paling tinggi dengan  $\sigma = 1.34 \times 10^{-4} \text{ ohm}^{-1} \text{ cm}^{-1}$  pada  $300^\circ\text{C}$ . Dalam usaha untuk meningkatkan kekonduksian, pendopan kimia dengan menggunakan  $\text{PbO}$ ,  $\text{Sr}(\text{NO}_3)_2$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{Ga}_2\text{O}_3$ ,  $\text{La}_2\text{O}_3$ ,  $\text{Fe}_2\text{O}_3$  dan sebagainya telah dilakukan pada bahan-bahan terpilih. Pepejal larutan yang terhad telah terhasil. Walau bagaimanapun, kekonduksian bahan-bahan terdop tersebut adalah kurang daripada bahan-bahan induk.

Selain daripada tindak balas keadaan pepejal, proses pengisaran bebola juga dijalankan dalam penyediaan  $\text{Bi}_{23}\text{V}_4\text{O}_{44.5}$  dan  $\text{Bi}_{14}\text{PO}_8$ . Dalam proses pengisaran bebola, bahan yang berketumpatan tinggi, berserbuk halus dan bertaburan seragam telah dihasilkan, dan seterusnya membawa kepada peningkatan dalam kekonduksian dan pemalar dielektrik.

Kompoun sillenit dalam sistem  $\text{Bi}_2\text{O}_3$ - $\text{P}_2\text{O}_5$  dengan nisbah Bi:P berjulat antara 13:1 dan 16:1 telah disintesiskan dan larutan pepejal telah diperolehi dalam bahan-bahan tersebut. Penggantian P oleh V dalam  $\text{Bi}_{14}\text{PO}_8$  menghasilkan larutan pepejal yang

terhad, sementara larutan pepejal yang lengap telah diperolehi apabila P dalam  $\text{Bi}_{14}\text{PO}_8$  digantikan oleh As. Peningkatan dalam kekonduksian telah diperhatikan dalam larutan pepejal yang didop dengan V. Pembentukan larutan pepejal yang terhad telah diperolehi apabila P digantikan oleh  $\text{Pb}^{2+}$ ,  $\text{Sr}^{2+}$ ,  $\text{Al}^{3+}$ ,  $\text{Ga}^{3+}$ ,  $\text{Fe}^{3+}$ ,  $\text{Si}^{4+}$ ,  $\text{Ge}^{4+}$ , dan  $\text{Ti}^{4+}$ . Kekonduksian bahan-bahan tersebut adalah lebih kurang sama dengan kompoun induk. Bahan-bahan tersebut merupakan konduktur ion oksida terutamanya pada suhu lebih daripada  $800^\circ\text{C}$ , di mana peralihan fasa  $\gamma \rightarrow \delta'$  berlaku.

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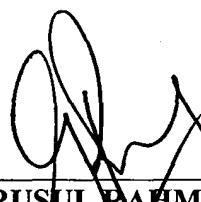
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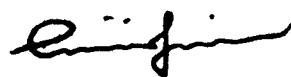
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## **DECLARATION**

I hereby declare that the thesis is based on my original work except for the quotations and citation which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted for any other degree at UPM or other institutions.



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Date: 29/7/2004



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