



UNIVERSITI PUTRA MALAYSIA

**EFFECT OF NANOPARTICULATES ( $Gd_2O_3$ ,  $Eu_2O_3$ ,  $Ce_2O_3$ ) ADDITION  
ON PROPERTIES OF  $(Bi_{1.6} Pb_{0.4})Sr_2 Ca_2 Cu_3 O_{10}$  CERAMICS  
SUPERCONDUCTORS**

**SAEDEH SADAT HOSSEINI RAVANDI**

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SUPERCONDUCTORS**

**By**

**SAEDEH SADAT HOSSEINI RAVANDI**

**Thesis submitted to the senate of University Putra Malaysia  
in fulfillment of the requirements for the degree of Master of Science**

**May 2009**



## **DEDICATIONS**

To Prof. Dr. Halim,  
for his guidance, patience and belief in me...

To my parents, Javad, Mahrokh and my lovely Mohammad,  
for their love, support and understanding...



Abstract of thesis presented to the senate of University Putra Malaysia in fulfillment of requirements for the degree of Master of Science.

**EFFECT OF NANOPARTICULATES ( $\text{Gd}_2\text{O}_3$ ,  $\text{Eu}_2\text{O}_3$ ,  $\text{Ce}_2\text{O}_3$ ) ADDITION ON PROPERTIES OF  $(\text{Bi}_{1.6}\text{Pb}_{0.4})\text{Sr}_2\text{Ca}_2\text{Cu}_3\text{O}_{10}$  CERAMICS SUPERCONDUCTORS**

By

**SAEDEH SADAT HOSSEINI RAVANDI**

**May 2009**

**Chairman : Professor Abdul Halim bin Shaari**

**Faculty : Science**

The effects of Rare Earth oxide (M) nanoparticulates addition in  $(\text{Bi}_{1.6}\text{Pb}_{0.4})\text{Sr}_2\text{Ca}_2\text{Cu}_3\text{O}_{10} + x\text{M}$  with  $\text{M} = \text{Gd}_2\text{O}_3$ ,  $\text{Eu}_2\text{O}_3$ ,  $\text{Ce}_2\text{O}_3$  and  $x = 0.0\text{-}0.05$  system, prepared by the conventional solid state reaction method have been investigated for structural, electrical and magnetic properties. The sizes of nanoparticulates oxide are 150 nm, 4 nm and 50nm respectively for  $\text{Gd}_2\text{O}_3$ ,  $\text{Eu}_2\text{O}_3$  and  $\text{Ce}_2\text{O}_3$ . The samples were calcined at  $800^\circ\text{C}$  and  $830^\circ\text{C}$  for 48 hours followed by sintering at  $850^\circ\text{C}$  for 120 hours. The morphological appearance of the samples with  $\text{Gd}_2\text{O}_3$  addition generally showed platelet-like grains with random orientations, typically of 2223 phase's structure, and by increasing the amount of nanoparticulate the platelet tend to stick together. At high percentage of nanoparticulate addition in  $\text{Gd}_2\text{O}_3$  the microstructure displayed grain size from to  $1.5\text{ }\mu\text{m}$  to  $10\text{ }\mu\text{m}$  and increased

number of pores. The morphologies of fractured surface of pellets of specimen from  $x=0.00$ - $x=0.05$  Ce<sub>2</sub>O<sub>3</sub> nanoparticulate, showed that the flaky nature of the grains gradually vanished and the edges of the grains become softer and the size of the grain changed from 10  $\mu\text{m}$  to 2  $\mu\text{m}$ . At  $x=0.03$  of Eu<sub>2</sub>O<sub>3</sub> the grain size become larger in comparison with other samples.

The volume fraction of the high-T<sub>C</sub> (2223) phase for addition of Gd<sub>2</sub>O<sub>3</sub> decreased from 89% for  $x=0.005$  to 58% for  $x=0.05$  and in the low- T<sub>C</sub> (2212) phase the volume increased from 3.4% to 41.2 % for  $x=0.005$  and  $x=0.05$ , respectively. The *c*-axis parameter decreased from 37.1293 Å to 36.7800 Å at  $x=0.005$  to  $x=0.05$  for Gd<sub>2</sub>O<sub>3</sub> addition on Bi-2223. For addition with Eu<sub>2</sub>O<sub>3</sub>, we can observe some increasing in  $x=0.03$ , from  $x=0.01$ - $x=0.03$  volume fractions increased from  $x=0.01$ - $x=0.03$  according to following percentage of 83.2% to 94.3%, respectively. The lattice parameter in the *c*-axis increased for  $x=0.00$  to  $x=0.03$  in Eu<sub>2</sub>O<sub>3</sub> addition from 37.1264 Å to 37.1432 Å. In addition with Ce<sub>2</sub>O<sub>3</sub> nanoparticulate in pure Bi-2223, with increasing in the amount, the volume fractions reduced from 92.67% for  $x=0.00$  to 57% in  $x=0.05$ , the *c*-axis also decreased from 37.1193 Å to 36.8870 Å for  $x=0.00$ - $0.05$ .

The critical temperature with increasing amounts of Gd<sub>2</sub>O<sub>3</sub>  $x=0.00$ - $x=0.05$  decreased from 101 K to 96 K, respectively. For samples with the addition of Eu<sub>2</sub>O<sub>3</sub>, T<sub>C</sub> decreased from 101 K to 98 K and started to increase for  $x=0.03$  with T<sub>C</sub> =103.74 K. When Ce<sub>2</sub>O<sub>3</sub> content was further increased, the electrical behavior started to change. The onset

critical temperatures were reduced from 101 K to 93 K for  $x=0.00$ - $x=0.05$ , respectively. Measurements of AC magnetic susceptibility as a function of temperature and AC field amplitudes have been carried out for samples containing  $\text{Gd}_2\text{O}_3$  nanoparticulate at constant frequency  $f=333.3$  Hz. It is shown that with increasing addition content from  $x=0.00$ - $x=0.05$ , the diamagnetic onset temperature decreased from 107 K for pure sample to 97 K for  $\text{Gd}_2\text{O}_3$  in last content. The analysis for comparison is based on the availability of higher percentage of the high- $T_c$  (2223) phase in the sample, the suppression degree of diamagnetic behavior with respect to AC fields; rapid or slow shift of the summit in  $\chi''_{(T)}$  to lower temperature with increasing field amplitude and the sharpness of  $\chi'_{(T)}$  for the intergranular component for the same field amplitude.

Abstrak tesis yang dikemukakan kepada Senat Universiti Putra Malaysia  
Sebagai memenuhi keperluan untuk ijazah Master Sains

**KESAN PENAMBAHAN NANOZARAH ( $\text{Gd}_2\text{O}_3$ ,  $\text{Eu}_2\text{O}_3$ ,  $\text{Ce}_2\text{O}_3$ ) KE ATAS  
SIFAT-SIFAT SUPERKONDUKTOR SERAMIK ( $\text{Bi}_{1.6}\text{Pb}_{0.4}\text{Sr}_2\text{Ca}_2\text{Cu}_3\text{O}_{10}$ )**

**Oleh**

**SAEDEEH SADAT HOSSEINI RAVANDI**

**Mei 2009**

**Pengerusi:** Professor Abdul Halim bin Shaari

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Kesan penambahan nanozarah oksida nadir bumi ( $M$ ) ke dalam sistem  $(\text{Bi}_{1.6}\text{Pb}_{0.4})\text{Sr}_2\text{Ca}_2\text{Cu}_3\text{O}_{10} + xM$  dengan  $M = \text{Gd}_2\text{O}_3$ ,  $\text{Eu}_2\text{O}_3$ ,  $\text{Ce}_2\text{O}_3$  dan  $x = 0.0-0.05$ , yang disediakan dengan kaedah tindakbalas keadaan pepejal yang konvensional telah dikaji struktur, sifat elektrik dan sifat magnetnya. Saiz oksida nanozarah adalah masing-masing 150 nm, 4 nm dan 50 nm untuk  $\text{Gd}_2\text{O}_3$ ,  $\text{Eu}_2\text{O}_3$  dan  $\text{Ce}_2\text{O}_3$ . Sampel-sampel telah dikalsin pada  $800^\circ \text{C}$  dan  $830^\circ \text{C}$  selama 48 jam diikuti dengan persinteran pada suhu  $850^\circ \text{C}$  selama 120 jam. Rupa bentuk morfologi sampel-sampel dengan penambahan  $\text{Gd}_2\text{O}_3$  secara amnya menunjukkan butiran-butiran seperti platlet dengan orientasi yang rawak, khasnya untuk struktur fasa 2223, dan dengan menambahkan amaun nanozarah platlet tersebut cenderung untuk melekat bersama. Pada peratusan penambahan nanozarah  $\text{Gd}_2\text{O}_3$  yang tinggi mikrostruktur menunjukkan saiz butiran dari  $10 \mu\text{m}$  kepada

1.5  $\mu\text{m}$  dan peningkatan bilangan keliangan. Morfologi permukaan retakan pelet bagi spesimen dengan nanozarah  $\text{Ce}_2\text{O}_3$  dari  $x = 0.00 - x = 0.05$ , menunjukkan bahawa keadaan kelupasan pada butiran hilang secara beransur-ansur dan bahagian sisi butiran menjadi semakin licin dan saiz butiran berubah dari 10  $\mu\text{m}$  kepada 2  $\mu\text{m}$ . Bagi  $\text{Eu}_2\text{O}_3$  pada  $x = 0.03$  saiz butiran menjadi lebih besar berbanding dengan sampel-sampel lain.

Pecahan isipadu bagi fasa  $T_C$ -tinggi (2223) bagi penambahan  $\text{Gd}_2\text{O}_3$  berkurang dari 89 % untuk  $x = 0.005$  kepada 58 % untuk  $x = 0.05$  dan dalam fasa  $T_C$ -rendah (2212) isipadu meningkat dari 3.4 % kepada 41.2 % bagi  $x = 0.005$ ,  $x = 0.05$ , masing-masing. Pemalar paksi- $c$  berkurang dari 37.1293 Å kepada 36.7800 Å pada  $x = 0.005$ ,  $x = 0.05$  bagi penambahan  $\text{Gd}_2\text{O}_3$  ke atas Bi-2223. Untuk penambahan dengan  $\text{Eu}_2\text{O}_3$ , dapat diperhatikan sejumlah peningkatan dalam  $x = 0.03$ , dari  $x = 0.00$  kepada  $x = 0.03$  pecahan isipadu meningkat menurut peratusan berikut iaitu 83.2 % kepada 94.3 %, masing-masing. Pemalar kekisi dalam paksi- $c$  meningkat dengan penambahan  $\text{Eu}_2\text{O}_3$   $x = 0.00$  kepada  $x = 0.03$  dari 37.1264 Å kepada 37.1432 Å. Bagi penambahan dengan nanozarah  $\text{Ce}_2\text{O}_3$  ke dalam Bi-2223 tulen, dengan penambahan amauannya, pecahan isipadu berkurang dari 92.6 7% bagi  $x = 0.00$  kepada 57 % bagi  $x = 0.05$ , paksi- $c$  juga berkurang dari 37.1193Å kepada 36.8870Å bagi  $x = 0.00-0.05$ . Suhu genting dengan penambahan amauun  $\text{Gd}_2\text{O}_3$   $x = 0.00 - x = 0.05$  berkurang dari 101 K kepada 96 K, masing-masing. Dengan penambahan  $\text{Eu}_2\text{O}_3$   $T_C$  berkurang dari 101 K kepada 98 K dan mula meningkat bagi  $x = 0.03$  dengan  $T_C = 103.74$  °K. Apabila kandungan  $\text{Ce}_2\text{O}_3$  terus ditingkatkan, sifat elektrik akan mula berubah. Suhu genting permulaan berkurang dari 101 K to 93 K bagi  $x = 0.00 - x = 0.05$ , masing-masing. Pengukuran kerentenan magnet AC sebagai fungsi

suhu dan amplitud medan AC telah dijalankan bagi sampel-sampel yang mengandungi nanozarah  $\text{Gd}_2\text{O}_3$  pada frekuensi malar  $f = 333.3 \text{ Hz}$ . Dapat diperhatikan bahawa dengan peningkatan kandungan bahan penambah dari  $x = 0.00 - x = 0.05$ , suhu permulaan diamagnet berkurangan dari  $107 \text{ K}$  bagi sampel tulen kepada  $97 \text{ K}$  bagi kandungan  $\text{Gd}_2\text{O}_3$  yang akhir. Analisis bagi perbandingan adalah berdasarkan kepada terdapatnya peratus fasa  $T_C$ -tinggi (2223) yang lebih tinggi di dalam sampel. Ketumpatan arus genting yang bergantung pada suhu telah diambil daripada data kerentanan AC bagi beberapa sampel terpilih; pengurangan darjah kelakuan diamagnet terhadap medan AC; laju atau perlahannya anjakan puncak kepada suhu lebih rendah dalam  $\chi''_{(T)}$  dengan peningkatan amplitud medan dan ketajaman  $\chi'_{(T)}$  bagi komponen antara butiran bagi amplitud medan yang sama.

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I certify that an Examination Committee has met on May 29<sup>th</sup>, 2009 to conduct the final examination of Saeedeh Sadat Hosseini Ravandi on her Master of Science thesis entitled "Effect of nanoparticulates (Gd<sub>2</sub>O<sub>3</sub>, Eu<sub>2</sub>O<sub>3</sub>, Ce<sub>2</sub>O<sub>3</sub>) addition on properties of (Bi<sub>1.6</sub> Pb<sub>0.4</sub>)Sr<sub>2</sub> Ca<sub>2</sub> Cu<sub>3</sub> O<sub>10</sub> ceramics superconductor" in accordance with Universiti Pertanian Malaysia(Higher Degree)Regulation 1981. The Committee recommends that the student be awarded the (Name of relevant degree)

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## **DECLARATION**

I declare that the thesis is my original work except for quotations and citation which have been duly acknowledged. I also declare that it has not been previously, and is not concurrently, submitted for any other degree at Universiti Putra Malaysia or at any other institution.

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**SAEDEH SADAT HOSSEINI RAVANDI**

Date:

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## LIST OF SYMBOLS

T	Temperature
$T_C$	Critical temperature
$T_{C(on)}$	Onset critical temperature
$T_{C(R=0)}$	Zero resistance temperature
HTS	High $T_C$ superconductor
BCS theory	Bardeen, Cooper and Schrieffer theory
BSCCO	Bi-Sr-Ca-Cu-O superconducting system
Bi(2212)	Family member in $\text{Bi}_2\text{Sr}_2\text{Ca}_n\text{Cu}_{n+1}\text{O}_{6-2n}$ , n=1
Bi(2223)	Family member in $\text{Bi}_2\text{Sr}_2\text{Ca}_n\text{Cu}_{n+1}\text{O}_{6-2n}$ , n=2
XRD	X-ray diffraction
SEM	Scanning Electron Microscopy
H	Applied magnetic field
$H_e$	External magnetic field
M	Magnetization
$a, b, c$	Lattice parameters
$\chi$	Susceptibility
$\chi'$	Real part of susceptibility
$\chi''$	Imaginary part of susceptibility
AC	Alternating Current
V	Sample volume
$\mathbf{d}_{hkl}$	Reciprocal $d$ vector
$hkl$	Miller indices



## CHAPTER 1

### INTRODUCTION

#### **1.1 What is Superconductivity?**

Superconductors are materials that have no resistance to the flow of electricity, are one of the last great frontiers of scientific discovery. Not only have the limits of superconductivity has not yet been reached, but the theories that explain superconductivity behavior seem to be constantly under review. In 1911 superconductivity was first observed in mercury (Hg) by Dutch physicist Heike Kamerlingh Onnes. When he cooled it to the temperature of liquid helium, 4 K (-452 F, -269 C), its resistance suddenly disappeared. The Kelvin scale represents an "absolute" scale of temperature. Therefore, it was necessary for Onnes to come within four degrees of the coldest temperature that is theoretically attainable to witness the phenomenon of superconductivity. Later, in 1913, he won a Nobel Prize in physics for his research in this area [Kristiana F, et al ; 2004].

In 1913, the element lead (Pb) was discovered to be superconducting at 7.2 K, this record was broken after 17 years with the element niobium (Nb) with  $T_c$  of 9.2 K. later the compounding Nb-Ti alloy was discovered to be superconducting. Among the compounds Nb with other elements,  $Nb_3Ge$  had the highest  $T_c$  of 22.3 K. The discovery of metals and alloys with zero resistance at low temperatures is an interesting phenomena and has wide applications for the future.

The first widely-accepted theoretical understanding of superconductivity was advanced in 1957 by American physicists John Bardeen, Leon Cooper, and John Schrieffer. Their theories of Superconductivity, known as the BCS theory, derived from the first letter of each man's last name - and won them a Nobel prize in 1972. The mathematically-complex BCS theory explained superconductivity at temperatures close to absolute zero for elements and simple alloys. The BCS theory describes superconductivity in low temperature metals such as mercury, based on an attractive interaction between electrons due to their coupling with phonon. However, at higher temperatures and with different superconductor systems, the BCS theory has subsequently become inadequate to fully explain how superconductivity is occurring. [Tinkham, et al; 1996]

Then in 1986, a truly breakthrough discovery was made in the field of superconductivity. Alex Müller and Georg Bednorz, researchers at the IBM Research Laboratory in Rüschlikon, Switzerland, created a brittle ceramic compound that superconductor at the highest temperature then known: 30 K. What made this discovery so remarkable was that ceramics are normally insulators. They don't conduct electricity well at all. So, researchers had not considered them as possible high-temperature superconductor candidates. The Lanthanum, Barium, Copper and Oxygen compound that Müller and Bednorz synthesized, behaved in a not-as-yet-understood way. It was later found that tiny amounts of this material were actually superconducting at 58 K, due to a small amount of lead having been added as a calibration standard - making the discovery even more noteworthy.