# Enantioselective Rhodium-Catalyzed Allylation of Cyclic Imines with Potassium Allyltrifluoroborates 

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#### Abstract

This Article presents further examples of the enantioselective rhodium-catalyzed addition of potassium allyltrifluoroborates to cyclic imines. A wide range of substituted allyltrifluoroborates are compatible with this process, and provide protected homoallylic amines with high levels of diastereo- and enantioselection. The reactions display a strong preference for carbon-carbon bond formation at the more substituted terminus of the allyl fragment of the allyltrifluoroborate, regardless of the position of the boron atom. Representative examples of manipulation of the products are also described.


Key words: allyltrifluoroborates, asymmetric catalysis, enantioselectivity, imines, rhodium

## Introduction

The enantioselective rhodium(I)-catalyzed addition of organoboron reagents to $\pi$-electrophiles is now a wellestablished method for the synthesis of enantioenriched chiral compounds. ${ }^{1}$ The stability, functional group tolerance, and usually low toxicity of organoboron compounds, coupled with the ability of a broad range of chiral ligands to impart high levels of enantioselection across several classes of reactions, has resulted in this method being widely applied in synthesis. Thus far, efforts have focused mainly upon nucleophilic arylations ${ }^{1}$ and alkenylations ${ }^{2}$ using the corresponding organoboron reagents, though one example of the use of alkynylboron reagents ${ }^{3}$ has also been described. ${ }^{4,5}$ Although catalytic enantioselective nucleophilic allylations occupy a prominent position in organic synthesis, ${ }^{6,7}$ it was not until our recent investigation ${ }^{8}$ that an enantioselective rhodium-catalyzed allylboration ${ }^{9,10}$ was achieved. Prompted by the general utility of chiral homoallylic amines in synthesis, ${ }^{6 b, 11,12}$ along with the increasing body of work describing the catalytic enantioselective allylborations of imines, ${ }^{11}$ we developed ${ }^{8}$ the enantioselective addition of potassium allyltrifluoroborates to cyclic imines, ${ }^{13}$ catalyzed by a rhodium complex based upon the chiral diene $\mathbf{L} 1^{\text {14,15 }}$ (Scheme 1A). In this study, the reactions using substituted allyltrifluoroborates displayed two notable features. First, clean allylic transposition was observed,

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B. Complementary Stereochemical Outcomes of Crotylations (Ref. 8)


Scheme 1 Previously reported rhodium(I) catalyzed allylation of cyclic imines.
resulting in $\mathrm{C}-\mathrm{C}$ bond formation at the $\gamma$-carbon of the allyltrifluoroborate. Second, high levels of stereochemical transfer were observed. For example, the reactions of aldimine 1 with ( $E$ )- and ( $Z$ )crotyltrifluoroborates 2a and 2b proceeded with different stereochemical outcomes to provide diastereomeric products $\mathbf{3 a}$ and $\mathbf{3 b}$, respectively, with high diastereoand enantioselectivities (Scheme 1B). ${ }^{8}$ A cyclic chairlike transition state TS1 was invoked to rationalize the stereochemical outcomes.

This Article presents further examples of this process, using classes of imines and substituted allyltrifluoroborates additional to those described in our original study. ${ }^{8}$ As well as providing a more thorough understanding of the substrate scope, some of the results using allyltrifluoroborates reveal cases of $\alpha$-selectivity in the reactions, which we have not observed previously. Finally, representative examples of manipulations of the allylation products are presented.

## Further Evaluation of Substrate Scope

In our original investigation, four distinct classes of cyclic imines were shown to undergo enantioselective allylation. These were benzoxathiazine-2,2-dioxides, 1,2,6-thiadiazine-1,1-dioxides, 1,2,5-thiadiazolidine-1,1dioxides, and cyclic sulfamidates. ${ }^{8}$ We have since found that other cyclic $N$-sulfonyl ketimines $\mathbf{4 a - 4 c}$ are also effective substrates (Scheme 2). For example, using the rhodium complex derived from ent-L1 ${ }^{14}$ (the enantiomer of the chiral diene employed in our original study ${ }^{8}$ ), methyl- and $n$-butyl-substituted ketimines underwent allylation with potassium allyltrifluoroborate (2c) under our established conditions ${ }^{8}$ to give benzosultams 5a and

5b, respectively, in moderate yields but with high enantioselectivities. It appears that the allylations of imines containing electron-withdrawing substituents proceed with lower enantioselectivities. For example, a substrate $4 \mathbf{c}$ containing an ethyl ester underwent allylation in only $21 \%$ ee (product $\mathbf{5 c}$ ). In addition, ketimine 6 containing a trifluoromethyl group underwent allylation in $85 \%$ yield but the product 7 was racemic (Scheme 3). In our previous study, benzoxathiazine-2,2dioxides containing electron-withdrawing substituents on the benzene ring were also allylated in lower enantioselectivities under our standard conditions using MeOH (5 equiv) in THF/dioxane, but fortunately the use of $i-\mathrm{PrOH}$ (13 equiv) in toluene/dioxane gave better

## Biographical Sketches



Hamish B. Hepburn was born in 1988 in Dundee, Scotland. He completed a master's degree in chemistry at the University of Edinburgh in 2011, which included a one year placement at AstraZeneca. He is currently conducting


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Hon Wai Lam was born in 1976 in Harrow, England. He received an M.Chem. degree in chemistry from the University of Oxford in 1998. He then moved to the University of Nottingham to carry out his Ph.D. under the direction of Gerald Pattenden. In January 2002, he moved to Harvard

Ph.D. studies in the group of Dr. Hon Wai Lam at the University of Edinburgh, investigating rhodiumcatalyzed enantioselective transformations.
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University as a GSK Postdoctoral Fellow to work with David A. Evans. In October 2003, he joined the School of Chemistry at the University of Edinburgh where he is now a Reader in Organic Chemistry. In October 2013, Hon will take up a new appointment at the University of Nottingham as
is currently undertaking postdoctoral research in the group of Dr. Hon Wai Lam at the University of Edinburgh, focusing on asymmetric catalysis.
the GSK Chair of Sustainable Chemistry. His group's research interests are based around the development of new synthetic methodology, including enantioselective catalysis and $\mathrm{C}-\mathrm{H}$ functionalization chemistry.


5a $57 \%$
96\% ee

5b 64\%
96\% ee


Scheme 2 Enantioselective allylation of imines 4a-4c.


Scheme 3 Allylation of imine 6.
results. ${ }^{8}$ However, application of these modified conditions to the synthesis of products 5 c and 7 did not offer any improvement in enantioselection.
Table 1 presents the reactions of cyclic ketimine $\mathbf{4 a}$ with a range of substituted potassium allyltrifluoroborates. ${ }^{16}$ Consistent with our previous report, ${ }^{8}$ the $\beta$ - $n$-propylsubstituted allyltrifluoroborate 2d was an effective allylating agent, and reacted with $\mathbf{4 a}$ to provide 8 a in high diastereo- and enantioselectivity, though in a modest yield (entry 1). Allyltrifluoroborates not reported in our original investigation were also effective. For example, the $\gamma$-phenyl-substituted allyltrifluoroborate 2 e gave $\mathbf{8 b}$ in $86 \%$ yield as one observable diastereomer in $91 \%$ ee (entry 2 ), while $\beta$-methylallyltrifluoroborate $2 f$ provided $\mathbf{8 c}$ in $56 \%$ yield and $98 \%$ ee (entry 3 ).
Up until this point, all of the reactions that we have studied that involve substituted allyltrifluoroborates have employed reagents in which boron is bonded to a primary carbon atom. In these cases, $\mathrm{C}-\mathrm{C}$ bond formation occurred exclusively at the $\gamma$-carbon of the allyltrifluoroborate. It was therefore of interest to examine the reactions using racemic $\alpha$-methylsubstituted allyltrifluoroborate $\mathbf{2 g}$, where boron is bonded to a secondary carbon atom. On the basis of an experiment using a deuterated potassium allyltrifluoroborate, we speculate that these allylations proceed via the intermediacy of allylrhodium species. ${ }^{8}$ According to this hypothesis, transmetallation of rac-2g with the chiral rhodium complex could therefore, in principle, lead to several interconverting isomeric

Table 1 Enantioselective Allylation of Imine 4a with Substituted Potassium Allyltrifluoroborates. ${ }^{\text {a }}$

${ }^{\text {a }}$ Reactions were conducted using 0.30 mmol of $\mathbf{4 a} .^{\text {b }}$ Isolated yield. Determined by ${ }^{1} \mathrm{H}$ NMR analysis of the unpurified reaction mixtures. ${ }^{\mathrm{d}}$ Determined by chiral HPLC analysis.


Scheme 4 Possible outcomes of reactions using $\alpha$-methyl-substituted allyltrifluoroborate $\mathbf{2 g}$.
allylrhodium species, each of which would provide different products upon reaction with imine $\mathbf{4 a}$ (Scheme 4). As such, it was not clear whether the $\gamma$-selectivity observed previously ${ }^{8}$ would be maintained using rac-2g.
In the event, allylation of $\mathbf{4 a}$ with $\mathrm{rac}-\mathbf{2 g}$ provided a mixture of diastereomeric products $\mathbf{8 d}$ and $\mathbf{1 1}$ with high enantioselectivities, where $\mathrm{C}-\mathrm{C}$ bond formation occurred at the $\alpha$-carbon of $\mathrm{rac}-\mathbf{2 g}$ (Scheme 5). ${ }^{16}$ Other isomeric products such as $\mathbf{8 e}$ were not observed. In addition to


Scheme 5 Reaction of imine $4 a$ with the $\alpha$-methyl-substituted allyltrifluoroborate $\mathbf{2 g}$.
providing further evidence that these allylations proceed via allylrhodium intermediates, this result indicates that the contributions to the observed products are greatest for crotylrhodium species $(E)-\mathbf{9}$, followed by $(Z)-\mathbf{9}$, while the contribution of $\mathbf{1 0}$ is negligible. If the interconversion between the different allylrhodium species is rapid compared with the rates of imine allylation, the product ratio will depend only upon the relative rates of allylation from $(E) \mathbf{- 9}, \mathbf{1 0}$, and $(Z)-\mathbf{9}$, and not upon their equilibrium distribution (Curtin-Hammett-type kinetics ${ }^{17}$ ). However, the high degrees of stereochemical transfer observed in the allylations of imine 1 using $(E)$ - and ( $Z$ )-crotyltrifluoroborates $\mathbf{2 a}$ and 2b (see Scheme 1B) suggests that isomerization between $(E)-9$ and ( $Z$ )-9 is slow compared with the rate of allylation. Therefore, if a similar scenario is operative in the allylation of imine $\mathbf{4 a}$ with $\mathbf{r a c - 2 g}$, it is likely that the ratio of $\mathbf{8 d}$ and $\mathbf{1 1}$ obtained depends significantly on the ratio of $(E)-9$ and $(Z)-9$ formed in the initial transmetallation.
Next, the allylations of aldimine 1 and ketimine 12 with a range of substituted allyltrifluoroborates not reported in our original study ${ }^{8}$ were conducted, and these reactions generally proceeded with high diastereo- and enantioselectivities (Table 2). ${ }^{16}$ In certain cases, the use of $i-\mathrm{PrOH}$ ( 5 equiv) in toluene/dioxane provided higher enantioselectivities compared with our standard conditions of MeOH ( 5 equiv) in THF/dioxane (products 13a, 13b, and 13d). Consistent with the result shown in Table 1, entry 2, 2e was a highly effective allylating agent, and reacted with imines 1 and 12 to give 13a and 13e, respectively, in good yields and high stereoselectivities. Although $\beta$-methyl substituted allyltrifluoroborate $\mathbf{2 f}$ reacted with ketimine $\mathbf{1 2}$ to give 13f in good yield and high enantioselectivity, the ee was lower in the reaction with aldimine $\mathbf{1}$ (13b obtained in $79 \%$ ee). As an example of an allyltrifluoroborate containing substitution at both the $\alpha$ - and $\gamma$-carbons, cyclohexenyltrifluoroborate $\mathbf{2 h}$ was evaluated. This reagent was only moderately effective, as the reaction with aldimine 1 provided 13 c in only $36 \%$ yield, though with good diastereo- and enantioselectivity. ${ }^{16}$ The reaction of $\mathbf{2 h}$ with ketimine $\mathbf{1 2}$ was completely

Table 2 Reaction of Imines $\mathbf{1}$ and $\mathbf{1 2}$ with Substituted Allyltrifluoroborates. ${ }^{\text {a }}$

${ }^{\text {a }}$ Reactions were conducted using 0.30 mmol of $\mathbf{1}$ or $\mathbf{1 2}$. Cited yields are of isolated products. Diastereomeric ratios were determined by ${ }^{1} \mathrm{H}$ NMR analysis of the unpurified reaction mixtures. Enantiomeric excesses were determined by chiral HPLC analysis. ${ }^{b}$ Reaction conducted using $i$-PrOH (5 equiv) in toluene/dioxane instead of MeOH in $\mathrm{THF} /$ dioxane.
unsuccessful, and provided a complex mixture of unidentified products. The $\alpha, \alpha$-dimethyl- substituted allyltrifluoroborate $2 \mathbf{i}$ resulted in $\mathrm{C}-\mathrm{C}$ bond formation at the $\alpha$-carbon exclusively, and provided reverse prenylation products $\mathbf{1 3 d}$ and $\mathbf{1 3 h}$ with good yields and high enantiomeric excesses. Interestingly, our previous study demonstrated that prenyltrifluoroborate $\mathbf{2 j}$, the isomer of $\mathbf{2 i}$, resulted in the formation of the enantiomers of $\mathbf{1 3 d}$ and $\mathbf{1 3 h}$ when $\mathbf{L 1}$ was used (Scheme 6). ${ }^{8}$ The formation of the reverse prenylation products 13d and 13h from both allyltrifluoroborates $\mathbf{2 i}$ ( $\alpha$-selectivity) and $\mathbf{2 j}$ ( $\gamma$-selectivity) suggests that allylation proceeds via allylrhodium species $\mathbf{1 5}$ rather than the isomeric species 14 (Scheme 7).


Ref. 8


1


12

$\left[\mathrm{Rh}(\mathbf{L 1}) \mathrm{Cl}_{2}\right.$ ( $1.5 \mathrm{~mol} \%$ )


2j



Scheme 6 Allylation of imines $\mathbf{1}$ and $\mathbf{1 2}$ with prenyltrifluoroborate $\mathbf{2 j}$.


Scheme 7 Allylrhodium species from allyltrifluoroborates $\mathbf{2 i}$ and $\mathbf{2 j}$.
The reactions of imines $\mathbf{1}$ and $\mathbf{1 2}$ with $\alpha$-methylsubstituted allyltrifluoroborate $\mathbf{2 g}$ proceeded with similar outcomes to the corresponding reaction with imine $\mathbf{4 a}$ (Schemes 8 and 9, compare with Scheme 5). In the case of imine 1, two products ent-3a and ent-3b were produced as an inseparable 2.3:1 mixture in $60 \%$ combined yield, and with enantiomeric excesses of $93 \%$ ee and $97 \%$ ee, respectively (Scheme 8). With imine 12, the two products $\mathbf{1 3 i}$ and $\mathbf{1 3 j}$ were separable, and were isolated in $48 \%$ and $16 \%$ yields, respectively, and in high enantioselectivities (Scheme 9).


Scheme 8 Reaction of imine $\mathbf{1}$ with the $\alpha$-methyl-substituted allyltrifluoroborate $\mathbf{2 g}$.


Scheme 9 Reaction of imine 12 with the $\alpha$-methyl-substituted allyltrifluoroborate $\mathbf{2 g}$.

## Functionalizations of the Allylation Products

To demonstrate the utility of the allylation products, representative transformations were conducted. For example, removal of the sulfonyl group of $\mathbf{1 6}$ (obtained in our original investigation using ligand $\mathbf{L} \mathbf{1}^{8}$ ) was readily accomplished by treatment with $\mathrm{LiAlH}_{4}$; in situ reaction of the resulting amine with $\mathrm{Boc}_{2} \mathrm{O}$ then provided carbamate 17 in $85 \%$ overall yield (Scheme 10).


Scheme 10 Removal of the sulfonyl group of 16.

A hydroboration/oxidation sequence of the alkene of ent16 (prepared as described previously ${ }^{8}$ by the reaction of imine 1 with potassium allyltrifluoroborate, but using chiral diene ent-L1) provided primary alcohol 18, which was transformed into the tricyclic sulfamate 19 by a Mitsunobu cyclization (Scheme 11).


Scheme 11 Conversion of ent-16 into tricyclic sulfamate 19.

Alternatively, treatment of $\mathbf{1 8}$ with $\mathrm{LiAlH}_{4}$ followed by $\mathrm{Boc}_{2} \mathrm{O}$ provided carbamate 20, which was converted into the tetrahydrobenzoxepine 21 by a Mitsunobu cyclization. Amino-substituted tetrahydrobenzoxepines have been shown to exhibit interesting biological


Scheme 12 Conversion of 18 into amino-substituted tetrahydrobenzoxepine 22.
activities; compound 22, for example, is a strong ACAT (acyl coenzyme A, cholesterol $O$-acyltransferase) inhibitor. ${ }^{18}$
Finally, the imine of the allylation product 23 (prepared as described previously ${ }^{8}$ by the reaction of imine $\mathbf{1 2}$ with potassium allyltrifluoroborate, but using chiral diene entL1) underwent a highly diastereoselective reduction upon treatment with DIBAL at $-20^{\circ} \mathrm{C}$ to give 24 in $90 \%$ yield as a single observable diastereomer (Scheme 13). ${ }^{19}$ Heating 24 in dioxane in the presence of ethylene diamine ( 10 equiv) removed the sulfonyl group to provide 1,2-diamine 25 in $88 \%$ yield.


Scheme 13 Conversion of 23 into 1,2-diamine 25

## Conclusion

In summary, the reactions described herein, using imines and potassium allyltrifluoroborates additional to those described in our original study, ${ }^{8}$ further illustrate the scope of the enantioselective rhodium-catalyzed allylboration of cyclic imines. In particular, these studies highlight the strong preference for $\mathrm{C}-\mathrm{C}$ bond formation to occur at the more highly substituted end of the allyl fragment of the trifluoroborate, regardless of the position of the boron atom (e.g. compare Schemes 5, 8, and 9, along with products $\mathbf{1 3 d}$ and $\mathbf{1 3 h}$ in Table 2 with Schemes 1B and 6). Finally, the utility of the allylation products was demonstrated by representative transformations.
All commercially available reagents were used as received. Anhydrous dioxane was purchased from Sigma-Aldrich and used
without further purification. Anhydrous THF, toluene, and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ were obtained by passage through activated alumina columns using a solvent purification system. Flash column chromatography was carried out using silica gel (Fisher Scientific $60 \AA$ particle size $35-70$ micron). Melting points were recorded on a Gallenkamp melting point apparatus and are uncorrected. IR spectra were recorded on a Shimadzu IRAffinity-1 instrument. ${ }^{1} \mathrm{H}$ NMR spectra were recorded on a Bruker AVA500 ( 500 MHz ) or a Bruker AVA400 ( 400 MHz ) spectrometer. Chemical shifts $(\delta)$ are quoted in parts per million ( ppm ) downfield of tetramethylsilane, using residual protonated solvent as internal standard $\left(\mathrm{CDCl}_{3}\right.$ at $7.27 \mathrm{ppm}, \mathrm{CD}_{3} \mathrm{CN}$ at $1.94 \mathrm{ppm}, \mathrm{CD}_{3} \mathrm{OD}$ at 3.31 ppm ). Abbreviations used in the description of resonances are: $s$ (singlet), d (doublet), t (triplet), q, (quartet), app (apparent), br (broad), m (multiplet). Coupling constants $(J)$ are quoted to the nearest 0.1 Hz . Proton-decoupled ${ }^{13} \mathrm{C}$ NMR spectra were recorded on a Bruker AVA500 (125.8 MHz) spectrometer. Chemical shifts $(\delta)$ are quoted in parts per million (ppm) downfield of tetramethylsilane, using deuterated solvent as internal standard $\left(\mathrm{CDCl}_{3}\right.$ at $77.0 \mathrm{ppm}, \mathrm{CD}_{3} \mathrm{CN}$ at $118.26 \mathrm{ppm}, \mathrm{CD}_{3} \mathrm{OD}$ at 49.0 $\mathrm{ppm})$. Assignments were made using the DEPT sequence with secondary pulses at $90^{\circ}$ and $135^{\circ}$. Proton-decoupled ${ }^{19} \mathrm{~F}$ NMR spectra were recorded on a Bruker AVA400 (376 MHz) spectrometer. Chemical shifts ( $\delta$ ) are quoted in parts per million (ppm) downfield of $\mathrm{CFCl}_{3}$, using residual protonated solvent as internal standard $\left(\mathrm{CFCl}_{3}\right.$ at 376.38 MHz with respect to tetramethylsilane at 400.00 MHz ). High-resolution mass spectra were recorded using electrospray ionization (ESI): or electron impact (EI): techniques on a Finnigan MAT 900 XLT spectrometer. Optical rotations were performed on an Optical Activity POLAAR 20 polarimeter. Chiral HPLC analysis was performed on an Agilent 1100 instrument using $4.6 \times 250 \mathrm{~mm}$ columns. Authentic racemic samples of products for chiral HPLC assay determinations were obtained using $[\mathrm{Rh}(\operatorname{cod}) \mathrm{Cl}]_{2}(2.5$ $\mathrm{mol} \%$ ) as an achiral precatalyst. Chiral diene ent-1 was prepared as described previously. ${ }^{14}$ Imines $1,{ }^{8} \mathbf{4 a},{ }^{20} \mathbf{4 b},{ }^{21} \mathbf{4 c},{ }^{13 \mathrm{c}} \mathbf{6},{ }^{22}$ and $\mathbf{1 2}{ }^{23}$ were prepared according to previously described procedures. Potassium allyltrifluoroborates $\mathbf{2 c} \mathbf{c}^{24}$ and $\mathbf{2 d}{ }^{8}$ were prepared as described previously.

## Preparation of Potassium Allyltrifluoroborates: General Procedure A

Following a slight modification of the procedure of Lennox and Lloyd-Jones, ${ }^{25}$ to a solution of the appropriate allylboronic acid pinacol ester ( 1.0 equiv) in $\mathrm{MeOH}(2 \mathrm{~mL} / \mathrm{mmol})$ and MeCN ( 2 $\mathrm{mL} / \mathrm{mmol}$ ) at room temperature was added a solution of KF (4.0 equiv) in $\mathrm{H}_{2} \mathrm{O}(0.1 \mathrm{~mL} / \mathrm{mmol})$, and the mixture was stirred for 5 min until complete dissolution occurred. To this solution was added a solution of L-(+)-tartaric acid (2.05 equiv) in THF (1.5 $\mathrm{mL} / \mathrm{mmol}$ of allylboronic acid pinacol ester) dropwise, and the resulting mixture was stirred at room temperature for $1 \mathrm{~h} . \mathrm{MeCN}$ ( $5 \mathrm{~mL} / \mathrm{mmol}$ of allylboronic acid pinacol ester) was added and the reaction was stirred for an additional 5 min before being filtered and concentrated in vacuo to leave a mixture of the potassium allyltrifluoroborate and pinacol. This residue was heated under reduced pressure to remove pinacol to leave the potassium allyltrifluoroborate as a white solid.

Potassium (E)-3-phenylprop-2-en-1-yltrifluoroborate (2e) ${ }^{\mathbf{2 6}}$
The title compound was prepared according to General Procedure A from the corresponding allylboronic acid pinacol ester ${ }^{27}$ (488 $\mathrm{mg}, 2.00 \mathrm{mmol}), \mathrm{KF}(0.46 \mathrm{~g}, 8.00 \mathrm{mmol})$, and L-(+)-tartaric acid $(0.62 \mathrm{~g}, 4.13 \mathrm{mmol})$ to give a white solid $(327 \mathrm{mg}, 73 \%)$ that displayed spectroscopic data consistent with those reported previously. ${ }^{26}$

## Potassium 2-methallyltrifluoroborate (2f)

The title compound was prepared according to General Procedure A from the corresponding allylboronic acid pinacol ester ${ }^{28}$ (478
$\mathrm{mg}, 2.63 \mathrm{mmol})$, $\mathrm{KF}(0.61 \mathrm{~g}, 10.5 \mathrm{mmol})$, and $\mathrm{L}-(+)$-tartaric acid $(0.81 \mathrm{~g}, 5.38 \mathrm{mmol})$ to give a white solid ( $248 \mathrm{mg}, 58 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( 400 MHz CD 3 CN ): $\delta=4.34(\mathrm{~s}, 1 \mathrm{H}), 4.32(\mathrm{~s}, 1 \mathrm{H}), 1.69$ (s, 3 H ), 1.14 (br s, 2 H ).
${ }^{13} \mathrm{C}$ NMR ( $100.6 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=116.5,110.3,25.0$, the carbon adjacent to the boron was not observed.
${ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=-152.4$.

## ( $\pm$ )-Potassium $\alpha$-methylallyltrifluoroborate (2g)

The title compound was prepared according to General Procedure A from the corresponding allylboronic acid pinacol ester ${ }^{28}$ ( 910 $\mathrm{mg}, 5.00 \mathrm{mmol}), \mathrm{KF}(1.16 \mathrm{~g}, 20.0 \mathrm{mmol})$, and $\mathrm{L}-(+)$-tartaric acid $(1.54 \mathrm{~g}, 10.3 \mathrm{mmol})$ to give a white solid ( $570 \mathrm{mg}, 70 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=6.04$ (ddd, $J=17.3,10.3,6.9$ $\mathrm{Hz}, 1 \mathrm{H}), 4.66(\mathrm{~d}, J=17.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.60(\mathrm{~d}, J=10.3 \mathrm{~Hz}, 1 \mathrm{H})$, 1.14 (br s, 1 H ), $0.85(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $100.6 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=110.9,107.0,25.1,14.4$.
${ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=-146.5$.

## Potassium 1-cyclohex-2-enyltrifluoroborate (2h) ${ }^{26}$

The title compound was prepared according to General Procedure A from the corresponding allylboronic acid pinacol ester ${ }^{29}$ (416 $\mathrm{mg}, 2.00 \mathrm{mmol}), \mathrm{KF}(0.46 \mathrm{~g}, 8.00 \mathrm{mmol})$ and $\mathrm{L}-(+)$-tartaric acid $(0.62 \mathrm{~g}, 4.13 \mathrm{mmol})$ to give a white solid $(290 \mathrm{mg}, 78 \%)$ that displayed spectroscopic data consistent with those reported previously. ${ }^{26}$

## Potassium $\alpha, \alpha$-dimethylallyltrifluoroborate (2i)

The title compound was prepared according to General Procedure A from the corresponding allylboronic acid pinacol ester ${ }^{28}$ ( 550 $\mathrm{mg}, 2.80 \mathrm{mmol})$, KF ( $0.65 \mathrm{~g}, 11.2 \mathrm{mmol}$ ), and L-(+)-tartaric acid $(0.86 \mathrm{~g}, 5.74 \mathrm{mmol})$ to give a white solid ( $330 \mathrm{mg}, 67 \%$ ).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=6.07-6.00(\mathrm{~m}, 1 \mathrm{H}), 4.61-4.55$ (m, 2 H ), 0.79 ( $\mathrm{s}, 6 \mathrm{H}$ ).
${ }^{13} \mathrm{C}\left(100.6 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}\right): \delta=111.1,105.1,23.7\left(2 \times \mathrm{CH}_{3}\right)$, the quaternary carbon adjacent to the boron was not observed.
${ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{CN}$ ): $\delta=-151.3$.

## Rh-Catalyzed Allylation of Imines: General Procedure B

A vial containing the appropriate cyclic imine ( 0.30 mmol ) and the appropriate potassium allyltrifluoroborate $(0.45 \mathrm{mmol})$ was sealed and flushed with $\mathrm{N}_{2}$ before anhydrous THF ( 3 mL ) was added. To this solution was added a stock solution of the rhodium-chiral diene complex $[\mathrm{Rh}(\text { ent-L1 }) \mathrm{Cl}]_{2}(10.0 \mathrm{mM}$ in anhydrous dioxane, $0.45 \mathrm{~mL}, 0.0045 \mathrm{mmol}=3 \mathrm{~mol} \% \mathrm{Rh})$ followed by $\mathrm{MeOH}(60 \mu \mathrm{~L}$, $1.50 \mathrm{mmol})$, and the resulting mixture was heated to $55^{\circ} \mathrm{C}$ for 15 h. The reaction was cooled to room temperature, filtered through a short plug of silica using EtOAc as eluent, and concentrated in vacuo. Purification of the residue by column chromatography gave the allylated product.

## Rh-Catalyzed Allylation of Imines: General Procedure C

A vial containing the appropriate cyclic imine ( 0.30 mmol ) and the appropriate potassium allyltrifluoroborate ( 0.45 mmol ) was sealed and flushed with $\mathrm{N}_{2}$ before anhydrous toluene ( 3 mL ) was added. To this solution was added a stock solution of the rhodium-chiral diene complex $[\mathrm{Rh}(\text { ent-L1) }) \mathrm{Cl}]_{2}(10.0 \mathrm{mM}$ in anhydrous dioxane, $0.45 \mathrm{~mL}, 0.0045 \mathrm{mmol}=3 \mathrm{~mol} \% \mathrm{Rh})$ followed by $i-\mathrm{PrOH}(115$ $\mu \mathrm{L}, 1.50 \mathrm{mmol}$ ), and the resulting mixture was heated to $55^{\circ} \mathrm{C}$ for 15 h . The reaction was cooled to room temperature, filtered through a short plug of silica using EtOAc as eluent, and concentrated in vacuo. Purification of the residue by column chromatography gave the allylated product.

## (S)-3-Methyl-3-(prop-2-en-1-yl)-2,3-dihydro-[1,2]-benzothiazole-1,1-dioxide (5a)

The title compound was prepared according to General Procedure B from imine $\mathbf{4 a}$ ( $54 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) and allyltrifluoroborate 2 c ( $66 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a yellow oil ( $38 \mathrm{mg}, 57 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.32$ ( $30 \% \mathrm{EtOAc} /$ hexane) .
$[\alpha]_{\mathrm{D}}^{20}-54.2\left(c 0.70, \mathrm{CHCl}_{3}\right)$.
IR (neat): $3250(\mathrm{NH}), 1373,1271,1254,1148,1121,1049,916$, $770,760 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.76(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.64$ ( $\mathrm{td}, J=7.7,1.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.53 (td, $J=7.7,1.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.39 (d, $J$ $=7.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.68 (dddd, $J=17.0,10.3,7.9,6.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.24-$ 5.17 (m, 2 H ), 4.65 (br s, 1 H ), 2.71 (dd, $J=14.1,7.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), $2.60(\mathrm{dd}, J=14.1,7.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.64(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=144.6,135.7,133.3,131.7$, 129.2, 123.0, 121.3, 121.2, 62.9, 45.7, 27.8.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{NO}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 224.0740, found: 224.0735 .

HPLC: Chiralcel OD-H column (90:10 hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 230 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=17.8 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=24.0$ $\min ; 96 \%$ ee.

## (S)-3-Butyl-3-(prop-2-en-1-yl)-2,3-dihydro-[1,2]-benzothiazole-1,1-dioxide (5b)

The title compound was prepared according to General Procedure B from imine $\mathbf{4 b}$ ( $67 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) and allyltrifluoroborate $\mathbf{2 c}$ ( $66 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a yellow oil ( $51 \mathrm{mg}, 64 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.45$ ( $30 \% \mathrm{EtOAc} /$ hexane ).
$[\alpha]_{\mathrm{D}}^{20}-30.3$ (c 1.45, $\mathrm{CHCl}_{3}$ ).
IR (neat): 3281 (NH), 1466, 1375, 1271, 1153, 1132, 1034, 930, $766,683 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.76$ (d, $J=7.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.64 ( $\mathrm{td}, J=7.7,1.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.53 (td, $J=7.7,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.35(\mathrm{~d}, J$ $=7.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.63 (dddd, $J=17.1,10.2,7.6,7.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $5.20-$ $5.14(\mathrm{~m}, 2 \mathrm{H}), 4.54(\mathrm{~s}, 1 \mathrm{H}), 2.70-2.60(\mathrm{~m}, 2 \mathrm{H}), 1.93-1.87(\mathrm{~m}, 2$ $\mathrm{H}), 1.45-1.36(\mathrm{~m}, 1 \mathrm{H}), 1.32-1.23(\mathrm{~m}, 2 \mathrm{H}), 1.01-0.97(\mathrm{~m}, 1 \mathrm{H})$, $0.85(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=142.9,136.0,133.2,131.6$, 129.3, 123.3, 121.5, 121.1, 66.3, 45.1, 40.0, 25.7, 22.6, 13.8.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{NO}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 266.1209, found: 266.1205.

HPLC: Chiralpak AD-H column ( $90: 10$ hexane $: i$-PrOH, 0.8 $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=24.7 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=29.2$ $\min ; 96 \%$ ee.

## (S)-Ethyl 5-methyl-1,1-dioxo-3-(prop-2-en-1-yl)-2,3-dihydro-[1,2]-benzothiazole-3-carboxylate (5c)

The title compound was prepared according to General Procedure B from imine $\mathbf{4 c}(76 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) and allyltrifluoroborate 2c ( $66 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a colorless oil ( $49 \mathrm{mg}, 55 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.35$ ( $30 \% \mathrm{EtOAc} /$ hexane ).
$[\alpha]_{\mathrm{D}}^{20}-22.4\left(c 0.90, \mathrm{CHCl}_{3}\right)$.
IR (neat): 3258 (NH), 1722 (C=O), 1261, 1231, 1182, 1144, 1134, $1040,702,660 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.63(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.51$ (s, 1 H ), 7.38 (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.77 (dddd, $J=16.9,10.2,7.8$, $6.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.69$ (s, 1 H ), $5.23-5.17$ (m, 2 H$), ~ 4.36-4.26(\mathrm{~m}, 2 \mathrm{H})$, $2.96(\mathrm{dd}, J=13.9,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.71(\mathrm{dd}, J=13.9,6.4 \mathrm{~Hz}, 1 \mathrm{H})$, $2.49(\mathrm{~s}, 3 \mathrm{H}), 1.34(3 \mathrm{H}, \mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=169.5,144.6,138.0,132.7$, 131.4, 131.0, 125.1, 121.1, 120.7, 68.5, 63.4, 44.6, 21.8, 14.1 .

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{NO}_{4} \mathrm{~S} \quad[\mathrm{M}+\mathrm{H}]^{+}$: 296.0951, found: 296.0951.

HPLC: Chiralcel OD-H column (90:10 hexane: $i$-PrOH, 0.8 $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ minor $)=13.9 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ major $)=18.4$ $\min ; 21 \%$ ee.

## 4-(Prop-2-en-1-yl)-4-(trifluoromethyl)-3,4-dihydro-[1,2,3]-benzoxathiazine-2,2-dione (7)

The title compound was prepared according to General Procedure B from imine $6(75 \mathrm{mg}, 0.30 \mathrm{mmol})$ and allyltrifluoroborate 2 c and was purified by column chromatography (10\% EtOAc/hexane) to give a yellow solid ( $75 \mathrm{mg}, 85 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.30(10 \% \mathrm{EtOAc} /$ hexane $)$.
m.p. $76-77^{\circ} \mathrm{C}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} /\right.$ hexane $)$.

IR (neat): 3285 (NH), 1614, 1489, 1454, 1435, 1377, 1260, 1180, $856,762 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.53(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.48$ (dt, $J=7.5,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.34$ (td, $J=8.0,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.16$ (dd, $J=8.2,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.61-5.53(\mathrm{~m}, 1 \mathrm{H}), 5.41-5.36$ (m, 2 H$), 5.13$ $(\mathrm{s}, 1 \mathrm{H}), 3.05(\mathrm{dd}, J=14.5,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.85(\mathrm{dd}, J=14.5,8.0$ $\mathrm{Hz}, 1 \mathrm{H}$ ).
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=150.8,131.5,128.4,127.0(\mathrm{q}$, $J=2.6 \mathrm{~Hz}), 126.5,124.3(\mathrm{q}, J=286.3 \mathrm{~Hz}), 124.2,120.0,117.9$, $65.8(\mathrm{q}, J=29.1 \mathrm{~Hz}), 41.3$.
${ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=-74.7$.
HRMS (ESI): Exact mass calcd for $\mathrm{C}_{11} \mathrm{H}_{11} \mathrm{~F}_{3} \mathrm{NO}_{3} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 294.0406, found: 294.0408.

To facilitate determination of enantiomeric excess, 7 was converted into the primary alcohol 7 a resulting from a hydroboration-oxidation sequence of the terminal alkene, according to the following procedure:

## 4-(Trifluoromethyl)-3,4-dihydro-[1,2,3]-benzoxathiazine-2,2dione (7a)

To a solution of the alkene $7(66 \mathrm{mg}, 0.23 \mathrm{mmol})$ in THF ( 3 mL ) at $0{ }^{\circ} \mathrm{C}$ was added $9-\mathrm{BBN}(0.5 \mathrm{M}$ in THF, $1.37 \mathrm{~mL}, 0.68 \mathrm{mmol})$ over 2 min . The mixture was warmed to room temperature over 1 h and then stirred for a further 23 h . The reaction was cooled to 0 ${ }^{\circ} \mathrm{C}$ and $3 \mathrm{M} \mathrm{NaOH}(1 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}_{2}\left(30 \mathrm{wt} . \%\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}, 2 \mathrm{~mL}\right)$ were added successively. The resulting mixture was stirred for 1 h at room temperature, diluted with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$, acidified with 2 M HCl , and extracted with EtOAc ( $3 \times 20 \mathrm{~mL}$ ). The combined organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and concentrated in vacuo. Purification of the residue by column chromatography ( $60 \%$ EtOAc/hexane) gave the alcohol $7 \mathrm{a}(61 \mathrm{mg}, 86 \%$ ) as a colorless gum.
$\mathrm{R}_{\mathrm{f}}=0.41$ ( $60 \%$ EtOAc/hexane) .
IR (neat): $3288(\mathrm{OH}$ and NH$), 2924,1454,1375,1177,1159$, $1115,1055,856,762 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 500 \mathrm{MHz}\right): \delta=7.61(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.55-$ $7.52(\mathrm{~m}, 1 \mathrm{H}), 7.41-7.38(\mathrm{~m}, 1 \mathrm{H}), 7.19(\mathrm{dd}, J=8.2,1.2 \mathrm{~Hz}, 1 \mathrm{H})$, 3.60-3.52 (m, 2 H$), 2.45-2.39(\mathrm{~m}, 1 \mathrm{H}), 2.14-2.08(\mathrm{~m}, 1 \mathrm{H}), 1.76-$ $1.68(\mathrm{~m}, 1 \mathrm{H}), 1.38-1.26(\mathrm{~m}, 1 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ): $\delta=152.9,132.5,125.0(\mathrm{q}, J=$ 286.0 Hz ), 127.4, $126.4(\mathrm{q}, J=286.0 \mathrm{~Hz}), 120.7,119.9,68.1(\mathrm{q}, J$ $=28.6 \mathrm{~Hz}$ ), $62.0,33.1,26.6$.
${ }^{19} \mathrm{~F}$ NMR ( $\left.\mathrm{CD}_{3} \mathrm{OD}\right): \delta=-78.0$.
HRMS (ESI): Exact mass calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{~F}_{3} \mathrm{NO}_{4} \mathrm{SNa}[\mathrm{M}+\mathrm{Na}]^{+}$: 334.0331, found: 334.0331 .

HPLC: Chiralpak AD-H column (90:10 hexane: $i$-PrOH, 0.8 $\mathrm{mL} / \mathrm{min} 280 \mathrm{~nm}, 25^{\circ} \mathrm{C}$ ); $\mathrm{t}_{\mathrm{r}}=19.3 \mathrm{~min}, 29.0 \mathrm{~min} ; 0 \%$ ee.

## ( $\boldsymbol{S}$ )-3-Methyl-3-[(S)-1-hex-1-en-3-yl]-2,3-dihydro-[1,2]-benzothiazole-1,1-dioxide (8a)

The title compound was prepared according to General Procedure B from imine $\mathbf{4 a}(54 \mathrm{mg}, 0.30 \mathrm{mmol})$ and allyltrifluoroborate $\mathbf{2 d}$ ( $86 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a yellow oil ( $38 \mathrm{mg}, 48 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.38$ ( $30 \% \mathrm{EtOAc} /$ hexane) .
$[\alpha]_{\mathrm{D}}^{20}-23.3\left(c 0.60, \mathrm{CHCl}_{3}\right)$.
IR (neat): 3219 (NH), 1389, 1377, 1271, 1238, 1153, 1134, 918 , $770,719 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.75(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.62$ ( $\mathrm{td}, J=7.8,1.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.52 (td, $J=7.7,0.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.37 (d, $J$ $=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.59(\mathrm{app} \mathrm{dt}, J=17.0,10.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.09(\mathrm{dd}, J=$ $10.3,1.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.99 (dd, $J=17.1,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.61(\mathrm{~s}, 1 \mathrm{H})$, 2.43-2.37 (m, 1 H$), 1.70-1.64(\mathrm{~m}, 1 \mathrm{H}), 1.63(\mathrm{~s}, 3 \mathrm{H}), 1.46-1.35$ $(\mathrm{m}, 1 \mathrm{H}), 1.30-1.21(\mathrm{~m}, 1 \mathrm{H}), 1.20-1.10(\mathrm{~m}, 1 \mathrm{H}), 0.88(\mathrm{t}, J=7.3$ $\mathrm{Hz}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=143.6,136.8,134.8,132.9$, 129.2, 124.0, 121.4, 119.8, 65.9, 53.7, 30.8, 26.0, 20.6, 13.9.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{NO}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 266.1209, found: 266.1208.

HPLC: Chiralcel OD-H column ( $90: 10$ hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=11.6 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=18.6$ $\min ; 98 \%$ ee.

## (S)-3-Methyl-3-[(R)-1-phenylprop-2-en-1-yl]-2,3-dihydro-[1,2]-benzothiazole-1,1-dione ( $\mathbf{8 b}$ )

The title compound was prepared according to General Procedure B from imine $\mathbf{4 a}(54 \mathrm{mg}, 0.30 \mathrm{mmol})$ and allyltrifluoroborate $\mathbf{2 e}$ ( $100 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $20 \% \mathrm{EtOAc} /$ hexane) to give a gum that solidified on standing to give a white solid ( $77 \mathrm{mg}, 86 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.26(30 \% \mathrm{EtOAc} /$ hexane $)$.
m.p. $78-79{ }^{\circ} \mathrm{C}$ (EtOAc/hexane).
$[\alpha]_{\mathrm{D}}^{20}-88.6\left(c 3.70, \mathrm{CHCl}_{3}\right)$.
IR (neat): $3233(\mathrm{NH}), 1389,1373,1275,1153,1132,930,891$, $756,708 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.77(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.60$ (td, $J=7.6,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.53(\mathrm{td}, J=7.5,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.39-7.27$ (m, 5 H ), 7.17 (d, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.17$ (ddd, $J=17.0,10.2,9.1$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 5.03 (dd, $J=10.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.95 (ddd, $J=17.0,1.4$, $1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.52(\mathrm{~s}, 1 \mathrm{H}), 3.69(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.51(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=143.0,139.1,135.9,135.0$, 133.0, 129.3, 129.3 $(2 \times \mathrm{CH}), 128.6(2 \times \mathrm{CH}), 127.6,123.8,121.4$, 119.2, 66.0, 59.6, 27.1.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{NO}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 300.1053, found: 300.1051.

HPLC: Chiralcel OD-H column ( $90: 10$ hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 230 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=22.4 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=32.8$ $\min ; 91 \%$ ee.

## (S)-3-Methyl-3-(2-methylprop-2-en-1-yl)-2,3-dihydro-[1,2]-benzothiazole-1,1-dione (8c)

The title compound was prepared according to General Procedure B from imine $\mathbf{4 a}(54 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) and allyltrifluoroborate $\mathbf{2 f}$ ( $73 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a yellow gum ( $40 \mathrm{mg}, 56 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.31(30 \% \mathrm{EtOAc} /$ hexane $)$.
$[\alpha]_{\mathrm{D}}^{20}-61.8\left(c 0.55, \mathrm{CHCl}_{3}\right)$.
IR (neat): 3273 (NH), 1369, 1277, 1263, 1163, 1150, 1128, 1057, 893, $758 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.74(\mathrm{t}, J=14.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.64$ (td, $J=7.7,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.53$ (td, $J=7.6,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.43$ (d, $J$ $=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.02-5.01(\mathrm{~m}, 1 \mathrm{H}), 4.84(\mathrm{~s}, 1 \mathrm{H}), 4.75(\mathrm{~s}, 1 \mathrm{H})$, 2.74 (d, $J=13.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.59 (d, $J=13.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), $1.65(\mathrm{~s}, 3$ H), $1.57(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=145.4,140.6,135.4,133.1$, 129.2, 123.2, 121.4, $117.8,62.5,48.9,29.0,23.9$.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{12} \mathrm{H}_{16} \mathrm{NO}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 238.0896, found: 238.0893.

HPLC: Chiralpak AD-H column (90:10 hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=20.6 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=25.6$ $\min ; 98 \%$ ee.
(S)-3-Methyl-3-[(S)-but-3-en-1-yl)-2,3-dihydro-[1,2]-
benzothiazole-1,1-dione ( 8 d ) and ( $S$ )-3-methyl-3-[( $R$ )-but-3-en-1-yl)-2,3-dihydro-[1,2]-benzothiazole-1,1-dione (11)
General Procedure B was followed using imine 4 a ( $54 \mathrm{mg}, 0.30$ mmol ) and allyltrifluoroborate $\mathbf{2 g}$ ( $73 \mathrm{mg}, 0.45 \mathrm{mmol}$ ). Purification by column chromatography ( $10 \%$ EtOAc/hexane) gave the allylated product 8d as white solid ( $47 \mathrm{mg} 66 \%$ ) followed by allylated product $\mathbf{1 1}$ as a white solid ( $17 \mathrm{mg} 24 \%$ ). Recrystallization of $\mathbf{8 d}$ and $\mathbf{1 1}$ from $\mathrm{Et}_{2} \mathrm{O}$ gave colorless crystals, which enabled the stereochemistry of $\mathbf{8 d}$ to be determined by Xray crystallography.
Data for 8d:
$\mathrm{R}_{\mathrm{f}}=0.33(30 \%$ EtOAc/hexane) .
m.p. $72-74{ }^{\circ} \mathrm{C}\left(\mathrm{Et}_{2} \mathrm{O}\right)$.
$[\alpha]_{\mathrm{D}}^{20}-67.8\left(c 1.15, \mathrm{CHCl}_{3}\right)$.
IR (neat): $3250(\mathrm{NH}), 1267,1234,1157,1134,1125,891,766$, $718,586 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.75(\mathrm{~d}, J=7.8,1 \mathrm{H}), 7.62(\mathrm{td}, J$ $=7.8,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.52(\mathrm{td}, J=7.8,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{~d}, J=5.7$ $\mathrm{Hz}, 1 \mathrm{H}), 5.71$ (ddd, $J=16.6,11.0,7.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), $5.08-5.02$ ( $\mathrm{m}, 2$ H), $4.64(\mathrm{~s}, 1 \mathrm{H}), 2.74-2.67(\mathrm{~m}, 1 \mathrm{H}), 1.63(\mathrm{~s}, 3 \mathrm{H}) 1.17(\mathrm{~d}, J=6.9$ $\mathrm{Hz}, 3 \mathrm{H}$ ).
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=143.8,138.1,135.2,133.0$, 129.2, 123.7, 121.4, 117.9, 65.9, 47.0, 26.0, 14.6.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{12} \mathrm{H}_{16} \mathrm{NO}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 238.0896, found: 238.0898.

HPLC: Chiralcel OD-H column (90:10 hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=15.0 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=22.5$ $\min ; 98 \%$ ee.
Data for 11:
$\mathrm{R}_{\mathrm{f}}=0.29(30 \% \mathrm{EtOAc} /$ hexane $)$.
m.p. $134-136{ }^{\circ} \mathrm{C}\left(\mathrm{Et}_{2} \mathrm{O}\right)$.
$[\alpha]_{\mathrm{D}}^{20}-65.2\left(c 0.50, \mathrm{CHCl}_{3}\right)$.
IR (neat): 3283 (NH), 1366, 1273, 1234, 1177, 1152, 1134, 934, $764,590 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.77(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.66$ ( td, $J=7.8,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.55(\mathrm{td}, J=7.8,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.37(\mathrm{~d}, J$ $=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.85(\mathrm{ddd}, J=17.1,10.3,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.24(\mathrm{dd}, J$ $=10.3,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.22-5.17(\mathrm{~m}, 1 \mathrm{H}), 4.33(\mathrm{~s}, 1 \mathrm{H}), 2.67-2.61$ $(\mathrm{m}, 1 \mathrm{H}), 1.62(\mathrm{~s}, 3 \mathrm{H}), 0.87(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=143.7,138.2,135.2,133.3$, 129.3, 123.2, 121.5, 118.1, 66.0, 47.3, 27.7, 15.0.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{12} \mathrm{H}_{16} \mathrm{NO}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 238.0896, found: 238.0897.

HPLC: Chiralcel OD-H column ( $90: 10$ hexane: $i$-PrOH, 0.8 $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=13.6 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=22.2$ $\min ; 98 \%$ ee.

## ( $\boldsymbol{S}$ )-4-[( $R$ )-1-phenylprop-2-en-1-yl]-3,4-dihydro-[1,2,3]-

 benzoxathiazine-2,2-dioxide (13a)The title compound was prepared according to General Procedure C from imine $\mathbf{1}(55 \mathrm{mg}, 0.30 \mathrm{mmol})$ and allyltrifluoroborate $2 \mathbf{e}$ ( $101 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a yellow oil ( $81 \mathrm{mg}, 90 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.50(30 \% \mathrm{EtOAc} /$ hexane $)$.
$[\alpha]_{\mathrm{D}}^{20}-85.0\left(c 0.80, \mathrm{CHCl}_{3}\right)$.
IR (neat): 3273 (NH), 1414, 1364, 1321, 1165, 1103, 874, 822, $756,696 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.42-7.37(\mathrm{~m}, 2 \mathrm{H}), 7.35-7.31$ $(\mathrm{m}, 4 \mathrm{H}), 7.12(\mathrm{td}, J=7.5,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.08-7.05(\mathrm{~m}, 2 \mathrm{H}), 6.04$ (ddd, $J=17.4,10.4,7.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.34 (dt, $J=10.4,1.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), $5.16-5.12(\mathrm{~m}, 1 \mathrm{H}), 5.11(\mathrm{dt}, J=17.3,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.75(\mathrm{~d}, J=$ $7.4 \mathrm{~Hz}, 1 \mathrm{H}) 4.24$ (dd, $J=13.3,6.6 \mathrm{~Hz}, 1 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=151.5,138.6,134.8,129.7$, $129.0(2 \times \mathrm{CH}), 128.6(2 \times \mathrm{CH}), 127.7,126.9,125.3,121.5,121.1$, 119.2, 60.2, 52.5 .

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{NO}_{3} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 302.0845, found: 302.0846.

HPLC: Chiralpak AD-H column ( $90: 10$ hexane: $i-\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=13.8 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=15.8$ $\min ; 95 \%$ ee.

## (S)-4-(2-Methylprop-2-en-1-yl)-3,4-dihydro-[1,2,3]-benzoxathiazine-2,2-dioxide (13b)

The title compound was prepared according to General Procedure C from imine $\mathbf{1}(55 \mathrm{mg}, 0.30 \mathrm{mmol})$ and allyltrifluoroborate $\mathbf{2 f}$ ( 73 $\mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a colorless oil ( $47 \mathrm{mg} 65 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.56(30 \%$ EtOAc/hexane).
$[\alpha]_{\mathrm{D}}^{20}-45.2\left(c 0.85, \mathrm{CHCl}_{3}\right)$.
IR (neat): $3273(\mathrm{NH}), 1406,1360,1188,1165,1103,891,816$, $758,675 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.35-7.30(\mathrm{~m}, 1 \mathrm{H}), 7.28-7.26$ (m, 1 H$), 7.24-7.20(\mathrm{~m}, 1 \mathrm{H}), 7.04(\mathrm{dd}, J=8.2,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.01$ (d, $J=0.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.96 (app td, $J=8.6,4.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.91 (s, 1 H), 4.63 (d, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.88 (dd, $J=14.5,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.69$ (ddd, $J=14.5,9.2,0.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 1.76 (s, 3 H ).
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=151.0,139.9,129.5,126.3$, 125.4, 122.4, 119.1, 116.2, 54.4, 42.6, 22.0.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{11} \mathrm{H}_{13} \mathrm{NO}_{3} \mathrm{~S}[\mathrm{M}+\mathrm{Na}]+$ : 262.0508, found: 262.0509 .

HPLC: Chiralpak AD-H column ( $95: 5$ hexane: $i$-PrOH, 0.8 $\left.\mathrm{mL} / \mathrm{min}, 280 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=20.2 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=23.3$ $\min ; 79 \%$ ee.

## (S)-4-[(R)-Cyclohex-2-en-1-yl]-3,4-dihydro-[1,2,3]-benzoxathiazine-2,2-dioxide (13c)

The title compound was prepared according to General Procedure B from imine $\mathbf{1}(55 \mathrm{mg}, 0.30 \mathrm{mmol})$ and allyltrifluoroborate $\mathbf{2 h}$ ( $85 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a $10: 1$ inseparable mixture of diastereomers as a white solid ( $29 \mathrm{mg}, 36 \%$ ). Recrystallization of a small sample of $\mathbf{1 3 c}$ from $\mathrm{Et}_{2} \mathrm{O}$ gave colorless crystals that were suitable for X-ray crystallography.
$\mathrm{R}_{\mathrm{f}}=0.54(30 \% \mathrm{EtOAc} /$ hexane $)$.
m.p. $128-130^{\circ} \mathrm{C}\left(\mathrm{Et}_{2} \mathrm{O}\right)$.
$[\alpha]_{\mathrm{D}}^{20}-38.4\left(c 1.25, \mathrm{CHCl}_{3}\right)$.
IR (neat): $3273(\mathrm{NH}), 1408,1369,1184,1171,1161,878,826$, $766,719 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.35-7.31(\mathrm{~m}, 1 \mathrm{H}), 7.25-7.22$ (m, 2 H ), 7.06-7.04 (m, 1 H$), ~ 6.04-5.99(\mathrm{~m}, 1 \mathrm{H}), 5.62-5.58(\mathrm{~m}, 1$ H), 4.90 (dd, $J=6.4,5.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.54(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.13-$ $3.06(\mathrm{~m}, 1 \mathrm{H}), 2.08-2.00(\mathrm{~m}, 2 \mathrm{H}), 1.82-1.75(\mathrm{~m}, 1 \mathrm{H}), 1.59-1.49$ (m, 2 H ), 1.27-1.18 (m, 1 H ).
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=151.6,132.3,129.3,126.6$, 126.3, 125.5, 121.4, 119.2, 60.1, 39.0, 24.9, 22.2, 21.4.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{NO}_{3} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 266.0845, found: 266.0846.

HPLC: Chiralcel OD-H column (95:5 hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 254 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=14.2 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=24.2$ min; $93 \%$ ee.

## (S)-(2-Methylbut-3-en-2-yl)-3,4-dihydro-[1,2,3]-benzoxathiazine-2,2-dioxide (13d) ${ }^{8}$

The title compound was prepared according to General Procedure C from imine $\mathbf{1}(55 \mathrm{mg}, 0.30 \mathrm{mmol})$ and allyltrifluoroborate $\mathbf{2 i} \mathbf{i}$ ( 79 $\mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $20 \% \mathrm{EtOAc} /$ hexane) to give a yellow oil ( $54 \mathrm{mg}, 71 \%$ ) that displayed spectroscopic data consistent with those reported previously. ${ }^{8}$
$\mathrm{R}_{\mathrm{f}}=0.54$ ( $30 \% \mathrm{EtOAc} /$ hexane ).
$[\alpha]_{\mathrm{D}}^{20}-60.0\left(c 0.50, \mathrm{CHCl}_{3}\right)$.
HPLC: Chiralpak AD-H column (95:5 hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=13.9 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=25.1$ $\min ; 90 \%$ ee.

## (S)-3-Methyl-4-phenyl-3[(R)-1-phenylprop-2-en-1-yl]-2,3-dihydro-[1,2,5]-thiadiazole-1,1-dioxide (13e)

The title compound was prepared according to General Procedure B from imine $\mathbf{1 2}(62 \mathrm{mg}, 0.30 \mathrm{mmol})$ and allyltrifluoroborate $\mathbf{2 e}$ ( $101 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a colorless gum ( $84 \mathrm{mg}, 86 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.26$ ( $30 \% \mathrm{EtOAc} /$ hexane ).
$[\alpha]_{\mathrm{D}}^{20}+52.7\left(c 2.20, \mathrm{CHCl}_{3}\right)$.
IR (neat): 3242 (NH), 1558, 1315, 1175, 1146, 995, 820, 710, 689, $652 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.12-8.08(\mathrm{~m}, 2 \mathrm{H}), 7.70-7.65$ (m, 1 H), 7.59-7.54 (m, 2 H), 7.42-7.38 (m, 4 H), 7.35-7.31 (m, 1 H), 6.24 (ddd, $J=17.0,10.1,9.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 5.03 (dd, $J=10.2,1.2$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 4.97 (d, $J=17.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), 4.48 (s, 1 H ), 3.88 (d, $J=9.3$ $\mathrm{Hz}, 1 \mathrm{H}), 1.67$ (s, 3 H ).
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=181.0,138.1,134.4,133.7$, $130.2(2 \times \mathrm{CH}), 129.4(2 \times \mathrm{CH}), 129.3,129.1(2 \times \mathrm{CH}), 128.9(2 \times$ CH), 128.0, 119.5, 75.9, 57.2, 26.1.
HRMS (ESI): Exact mass calcd for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 327.1162, found: 327.1160.

HPLC: Chiralpak AD-H column ( $90: 10$ hexane: $i-\mathrm{PrOH}, 0.8$ $\mathrm{mL} / \mathrm{min}, 254 \mathrm{~nm}, 25{ }^{\circ} \mathrm{C}$ ); $\mathrm{t}_{\mathrm{r}}$ (major) $=23.9 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $) 25.2$ $\min ; 95 \%$ ee.

## (S)-3-Methyl-3-(2-methylprop-2-en-1-yl)-4-phenyl-2,3-dihydro-[1,2,5]-thiadiazole-1,1-dioxide (13f)

The title compound was prepared according to General Procedure B from imine 12 ( $62 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) and allyltrifluoroborate ( 73 $\mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a yellow oil ( $71 \mathrm{mg}, 90 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.26$ ( $30 \% \mathrm{EtOAc} /$ hexane ).
$[\alpha]_{\mathrm{D}}^{20}-25.0\left(c 0.80, \mathrm{CHCl}_{3}\right)$.
IR (neat): 3300 (NH), 1553, 1294, 1175, 1144, 907, 824, 783, 692, $654 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=8.07(\mathrm{dd}, J=8.5,1.2 \mathrm{~Hz}, 2 \mathrm{H}$ ), 7.67-7.62 (m, 1 H), 7.57-7.52 (m, 2 H), 5.08-5.04 (m, 1 H ), 4.87
(s, 1 H ), 4.74 (s, 1 H ), 2.98 (d, $J=14.5 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.73 (dd, $J=$ $14.5,0.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.84(\mathrm{~s}, 3 \mathrm{H}), 1.67(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR ( $125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=182.3,139.5,133.8,130.2(2$ $\times \mathrm{CH}), 129.2(2 \times \mathrm{CH}), 128.9,118.6,71.6,46.7,27.5,23.4$.
HRMS (ESI): Exact mass calcd for $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 265.1005, found: 265.1000.

HPLC: Chiralcel OD-H column (90:10 hexane:i-PrOH, 0.8 $\left.\mathrm{mL} / \mathrm{min}, 254 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=16.1 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=19.2$ $\min ; 97 \%$ ee.

## ( $\boldsymbol{S}$ )-3-Methyl-3-(2-methylbut-3-en-2-yl)-4-phenyl-2,3-dihydro-

 [1,2,5]-thiadiazole-1,1-dioxide ( $\mathbf{1 3} \mathbf{H})^{8}$The title compound was prepared according to General Procedure B from imine 12 ( $62 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) and allyltrifluoroborate $\mathbf{2 i}$ ( $79 \mathrm{mg}, 0.45 \mathrm{mmol}$ ) and was purified by column chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to give a yellow solid ( $72 \mathrm{mg}, 87 \%$ ) that displayed spectroscopic data consistent with those reported previously. ${ }^{8}$
$\mathrm{R}_{\mathrm{f}}=0.25(30 \% \mathrm{EtOAc} /$ hexane $)$.
$[\alpha]_{\mathrm{D}}^{20}+40.7\left(c 1.40, \mathrm{CHCl}_{3}\right)$.
HPLC: Chiralpak AD-H column ( $90: 10$ hexane: $i-\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=22.1 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=29.2$ $\mathrm{min} ; 91 \%$ ee.
( $S$ )-4-[ $(S)$-But-3-en-2-yl]-3,4-dihydro-[1,2,3]-benzoxathiazine-2,2-dioxide (ent-3a) ${ }^{8}$ and ( $S$ )-4-[(R)-but-3-en-2-yl]-3,4-dihydro-[1,2,3]-benzoxathiazine-2,2-dioxide (ent-3b) ${ }^{8}$
General Procedure B was followed using imine $1(55 \mathrm{mg}, 0.30$ mmol ) and allyltrifluoroborate $\mathbf{2 g}(73 \mathrm{mg}, 0.45 \mathrm{mmol})$. Purification by column chromatography ( $20 \%$ EtOAc/hexane) gave a $2: 1$ inseparable mixture of the allylated products ent-3a and ent-3b as a yellow oil ( $43 \mathrm{mg}, 60 \%$ ) that displayed spectroscopic data consistent with those reported previously. ${ }^{8}$
$\mathrm{R}_{\mathrm{f}}=0.52(30 \% \mathrm{EtOAc} /$ hexane $)$.
Data for ent-3a:
HPLC: Chiralpak AD-H column (98:2 hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 225 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=18.5 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=38.6$ $\min ; 93 \%$ ee.
Data for ent-3b:
HPLC: Chiralpak AD-H column ( $98: 2$ hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 225 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=21.7 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=43.0$ $\min ; 97 \%$ ee.

## 3(S)-3-[(S)-But-3-en-2-yl]-3-methyl-4-phenyl-2,3-dihydro-

 [1,2,5]-thiadiazole 1,1-dioxide (13i) ${ }^{8}$ and $3(S)$-3- $(R)$-but-3-en-2-yl]-3-methyl-4-phenyl-2,3-dihydro-[1,2,5]-thiadiazole-1,1dioxide ( $\mathbf{1 3 j})^{8}$General Procedure B was followed using imine 12 ( $62 \mathrm{mg}, 0.30$ mmol ) and allyltrifluoroborate $\mathbf{2 g}$ ( $73 \mathrm{mg}, 0.45 \mathrm{mmol}$ ). Purification by column chromatography ( $10 \%$ EtOAc/hexane) gave the allylated product $\mathbf{1 3 j}$ as a yellow oil ( $13 \mathrm{mg}, 16 \%$ ) followed by the allylated product $\mathbf{1 3 i}$ as a yellow oil ( $38 \mathrm{mg}, 48 \%$ ) that displayed spectroscopic data consistent with those reported previously. ${ }^{8}$ Both $\mathbf{1 3 i}$ and $\mathbf{1 3 j}$ were contaminated with small quantities ( $<10 \%$ ) of each other.
Data for 13i:
$\mathrm{R}_{\mathrm{f}}=0.22$ ( $30 \% \mathrm{EtOAc} /$ hexane ).
$[\alpha]_{\mathrm{D}}^{20}+23.0\left(c 1.00, \mathrm{CHCl}_{3}\right)$.
HPLC: Chiralcel OD-H column (90:10 hexane: $i-\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 254 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=13.0 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=22.2$ $\min ; 97 \%$ ee.
Data for $\mathbf{1 3 j}$ :
$\mathrm{R}_{\mathrm{f}}=0.30$ ( $30 \% \mathrm{EtOAc} /$ hexane ).
$[\alpha]_{\mathrm{D}}^{20}+16.6\left(c 0.30, \mathrm{CHCl}_{3}\right)$.
HPLC: Chiralcel OD-H column (90:10 hexane: $i$-PrOH, 0.8 $\mathrm{mL} / \mathrm{min}, 254 \mathrm{~nm}, 25^{\circ} \mathrm{C}$ ); $\mathrm{t}_{\mathrm{r}}$ (major) $=11.0 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=17.3$ $\mathrm{min} ; 91 \%$ ee.

## tert-Butyl $N$ - [(R)-1-(2-hydroxyphenyl)but-3-en-1-yl]carbamate

 (17)To a solution of allylation product $\mathbf{1 6}^{8}$ in THF ( 1 mL ) at room temperature was added $\mathrm{LiAlH}_{4}(1.0 \mathrm{M}$ in THF, $0.37 \mathrm{~mL}, 0.37$ mmol ) over 1 min at room temperature. The mixture was heated at $60{ }^{\circ} \mathrm{C}$ for 15 h , allowed to cool to temperature, and then cooled with an ice bath. The reaction was quenched carefully with EtOAc $(1 \mathrm{~mL})$, followed by the addition of $\mathrm{EtOH}(1 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}$ (2 $\mathrm{mL})$. To the resulting turbid mixture was added $\mathrm{Boc}_{2} \mathrm{O}(81 \mathrm{mg}$, 0.37 mmol ) in one portion and the resulting mixture was stirred at room temperature for 1 h . The reaction was diluted with EtOAc $(20 \mathrm{~mL})$ and acidified with 2 M HCl until the aqueous layer became clear. The aqueous layer was separated and extracted with EtOAc $(2 \times 20 \mathrm{~mL})$. The combined organic layers were dried ( MgSO 4 ), filtered, and concentrated in vacuo. Purification of the residue by column chromatography ( $6: 1$ hexane: $\mathrm{Et}_{2} \mathrm{O} \rightarrow 2: 1$ hexane: $\mathrm{Et}_{2} \mathrm{O}$ ) gave the carbamate 17 as a colorless oil ( 28 mg , 85\%).
$\mathrm{R}_{\mathrm{f}}=0.54(30 \% \mathrm{EtOAc} /$ hexane $)$.
$[\alpha]_{\mathrm{D}}^{20}+45.2\left(c 1.15, \mathrm{CHCl}_{3}\right)$.
IR (neat): $3310(\mathrm{OH}), 2925,1680,1502,1456,1367,1170,1043$, 918, 860, $750 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=8.50(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.16-7.07(\mathrm{~m}, 2$ H), 6.88-6.80 (m, 2 H$), 5.75(\mathrm{ddt}, J=17.1,10.2,6.9 \mathrm{~Hz}, 1 \mathrm{H})$, 5.27 (br s, 1 H ), $5.13(\mathrm{dd}, J=17.2,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.09(\mathrm{~d}, J=10.4$ $\mathrm{Hz}, 1 \mathrm{H}), 4.89(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 2.62(\mathrm{t}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.47(\mathrm{~s}, 9 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (125.8 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=157.0,154.7,134.6,128.5$, $127.9,126.6,119.8,117.8,117.0,80.7,48.9,38.6,28.4\left(3 \times \mathrm{CH}_{3}\right)$.
HRMS (EI): Exact mass calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{3} \mathrm{~N}[\mathrm{M}+\mathrm{H}]^{+}$: 264.1594, found: 264.1599.
HPLC: Chiralpak AS-H column (98:2 hexane: $i-\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 230 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}$ (major) $=23.1 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=30.6$ $\min ; 93 \%$ ee.

## (S)-4-(3-Hydroxypropyl)-3,4-dihydro-[1,2,3]-benzoxathiazine-2,2-dioxide (18)

To a solution of the alkene $\mathbf{1 6}$ (prepared as described previously ${ }^{8}$ by the reaction of imine 1 with potassium allyltrifluoroborate, but using chiral diene ent-L1) ( $225 \mathrm{mg}, 1.00 \mathrm{mmol}$ ) in THF ( 3 mL ) at $0^{\circ} \mathrm{C}$ was added $9-\mathrm{BBN}(0.5 \mathrm{M}$ in THF, $6.0 \mathrm{~mL}, 3.0 \mathrm{mmol})$ over 2 $\min$. The mixture was warmed to room temperature over 1 h and then stirred for a further 23 h . The reaction was cooled to $0^{\circ} \mathrm{C}$ and $3 \mathrm{M} \mathrm{NaOH}(3 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}_{2}\left(30 \mathrm{wt} . \%\right.$ in $\left.\mathrm{H}_{2} \mathrm{O}, 6 \mathrm{~mL}\right)$ were added successively. The resulting mixture was stirred for 1 h at room temperature, diluted with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$, acidified with 2 M HCl , and extracted with $\mathrm{EtOAc}(3 \times 40 \mathrm{~mL})$. The combined organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and concentrated in vacuo. Purification of the residue by column chromatography ( $80 \%$ EtOAc/hexane) gave the alcohol 18 as a white solid ( 182 mg , 75\%).
m.p. $112-113^{\circ} \mathrm{C}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$.
$\mathrm{R}_{\mathrm{f}}=0.36$ ( $80 \% \mathrm{EtOAc} /$ hexane $)$.
$[\alpha]_{\mathrm{D}}^{20}-36.7\left(c 0.49, \mathrm{CHCl}_{3}\right)$.
IR (neat): $3255(\mathrm{OH}), 2880,1485,1452,1425,1371,1175,1107$, $883,760 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): \delta=7.40(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.36$ (ddd, $J=8.3,4.5,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{td}, J=1.6,1.2 \mathrm{~Hz}, 1 \mathrm{H})$, $7.02(\mathrm{dd}, J=8.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.70(\mathrm{dd}, J=10.9,3.8 \mathrm{~Hz}, 1 \mathrm{H})$,
3.70-3.61 (m, 2 H$), 2.26-2.18(\mathrm{~m}, 1 \mathrm{H}), 2.03-1.95(\mathrm{~m}, 1 \mathrm{H}), 1.90-$ $1.82(\mathrm{~m}, 1 \mathrm{H}), 1.76-1.67(\mathrm{~m}, 1 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (125.8 MHz, $\left.\mathrm{CD}_{3} \mathrm{OD}\right): \delta=152.8,130.2,127.9,126.1$, 125.0, 119.3, 62.3, 57.9, 31.3, 29.6.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{NO}_{4} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 244.0638, found: 244.0640.

## (S)-8-Oxa-7 $\lambda^{6}$-thia-6-azatricyclo[7.4.0.0 $\left.{ }^{2,6}\right]$ trideca-1(9),10,12-

 triene-7,7-dioxide (19)To a solution of the alcohol $\mathbf{1 8}(61 \mathrm{mg}, 0.25 \mathrm{mmol})$ and $\mathrm{PPh}_{3}(85$ $\mathrm{mg}, 0.33 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was added a solution of DEAD ( $53 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$. The mixture was allowed to warm to room temperature over 1 h and then stirred for an additional 9 h . The reaction was quenched with $\mathrm{EtOH}(1 \mathrm{~mL})$ and concentrated in vacuo. Purification of the residue by column chromatography ( $60 \% \mathrm{EtOAc} /$ hexane) gave the product ( 56 mg , $>95 \%$ ) as a white solid.
m.p. $85-86^{\circ} \mathrm{C}$ (EtOAc/hexane).
$\mathrm{R}_{\mathrm{f}}=0.57(60 \% \mathrm{EtOAc} /$ hexane $)$.
$[\alpha]_{\mathrm{D}}^{20}-125.0\left(c 0.40, \mathrm{CHCl}_{3}\right)$.
IR (neat): $2982,1485,1450,1392,1206,1175,1103,1005,856$, $758 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=7.30(\mathrm{dddd}, J=8.1,7.3,1.7,0.8$ $\mathrm{Hz}, 1 \mathrm{H}), 7.21(\mathrm{td}, J=7.5,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{dt}, J=7.5,1.0 \mathrm{~Hz}$, $1 \mathrm{H}), 7.01(\mathrm{dd}, J=8.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.20(\mathrm{dd}, J=7.4,2.5 \mathrm{~Hz}, 1$ H), 3.61-3.56 (m, 1 H ), 3.51 (ddd, $J=10.1,8.6,5.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.59$ (ddd, $J=16.5,12.8,7.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.30-2.24(\mathrm{~m}, 1 \mathrm{H}), 2.07-1.99$ (m, 1 H ), 1.92-1.83 (m, 1 H ).
${ }^{13} \mathrm{C}$ NMR $\left(125.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=151.0,129.1,126.6,125.6$, 122.5, 118.8, 62.7, 49.7, 34.0, 23.4.

HRMS (EI): Exact mass calcd for $\mathrm{C}_{10} \mathrm{H}_{12} \mathrm{NO}_{3} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}$: 226.0532, found: 226.0529.
HPLC: Chiralpak AD-H column (95:5 hexane: $i$-PrOH, 0.8 $\left.\mathrm{mL} / \mathrm{min}, 280 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ minor $)=18.6 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ major $)=20.2$ $\min ; 92 \%$ ee.

## tert-Butyl- $N$-[(S)-4-hydroxy-1-(2-hydroxyphenyl)butyl] carbamate (20)

To a solution of the cyclic sulfamate $18(100 \mathrm{mg}, 0.41 \mathrm{mmol})$ in THF $(2 \mathrm{~mL})$ at room temperature was added $\mathrm{LiAlH}_{4}(2.0 \mathrm{M}$ in THF, $0.62 \mathrm{~mL}, 1.24 \mathrm{mmol}$ ) dropwise over 4 min . The mixture was heated at $60{ }^{\circ} \mathrm{C}$ for 2 h , allowed to cool to room temperature, and then cooled with an ice bath. The reaction was quenched carefully with EtOAc $(2 \mathrm{~mL})$, followed by the addition of $\mathrm{EtOH}(2 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(2 \mathrm{~mL})$. To the resulting turbid mixture was added $\mathrm{Boc}_{2} \mathrm{O}$ ( $268 \mathrm{mg}, 1.24 \mathrm{mmol}$ ) in one portion and the resulting mixture was stirred at room temperature for 1 h . The reaction was diluted with EtOAc ( 40 mL ) and acidified with 2 M HCl until the aqueous layer became clear. The aqueous layer was separated and extracted with EtOAc $(2 \times 40 \mathrm{~mL})$. The combined organic layers were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and concentrated in vacuo. Purification of the residue by column chromatography ( $80 \% \mathrm{EtOAc} /$ hexane ) gave the carbamate 20 ( $87 \mathrm{mg}, 75 \%$ ) as a colorless gum.
$\mathrm{R}_{\mathrm{f}}=0.42(80 \% \mathrm{EtOAc} /$ hexane $)$.
$[\alpha]_{\mathrm{D}}^{20}-33.6\left(c 0.24, \mathrm{CHCl}_{3}\right)$.
IR (neat): $3305(\mathrm{OH}$ and NH), 2980, $1680(\mathrm{C}=\mathrm{O}), 1502,1456$, $1367,1292,1253,1165,752,742 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=8.28(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.18(\mathrm{dd}, J=$ $12.1,4.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.93(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.90(\mathrm{dt}, J=7.5,1.1$ $\mathrm{Hz}, 1 \mathrm{H}), 5.13(\mathrm{~s}, 1 \mathrm{H}), 4.86(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.72(\mathrm{dt}, J=6.2$, $0.9 \mathrm{~Hz}, 2 \mathrm{H}), 2.06-1.95(\mathrm{~m}, 2 \mathrm{H}), 1.73-1.58(\mathrm{~m}, 2 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (125.8 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=157.3,154.9,128.8,128.4$, $126.3,120.3,117.8,80.8,62.3,49.0,30.6,29.5,28.3\left(3 \times \mathrm{CH}_{3}\right)$.

HRMS (ESI): Exact mass calcd for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{NO}_{4}[\mathrm{M}+\mathrm{H}]^{+}: 282.1700$, found: 282.1696 .
tert-Butyl $\quad N$-[(S)-2,3,4,5-tetrahydro-1-benzoxepin-5-yl] carbamate (21)
To a solution of the alcohol $20(68 \mathrm{mg}, 0.24 \mathrm{mmol})$ and $\mathrm{PPh}_{3}(82$ $\mathrm{mg}, 0.31 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was added a solution of DEAD ( $51 \mathrm{mg}, 0.29 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$. The mixture was allowed to warm to room temperature over 1 h and then stirred for an additional 9 h . The reaction was quenched with EtOH ( 1 mL ) and concentrated in vacuo. Purification of the residue by column chromatography ( $20 \%$ EtOAc/hexane) gave the tetrahydrobenzoxepine 21 as a white solid ( $43 \mathrm{mg}, 68 \%$ ).
m.p. $105-106^{\circ} \mathrm{C}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ /hexane $)$.
$\mathrm{R}_{\mathrm{f}}=0.40$ ( $20 \% \mathrm{EtOAc} /$ hexane) .
$[\alpha]_{\mathrm{D}}^{20}-40.0\left(c 0.15, \mathrm{CHCl}_{3}\right)$.
IR (neat): $3300(\mathrm{NH}), 2976,2930,1713(\mathrm{C}=\mathrm{O}), 1450,1366,1236$, $1224,1170,760 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.28-7.26(\mathrm{~m}, 1 \mathrm{H}), 7.20(\mathrm{td}, J=$ $7.7,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.05(\mathrm{td}, J=7.5,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.01(\mathrm{dd}, J=7.5$, $1.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.29(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.91(\mathrm{t}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H})$, 4.30 (app d, $J=11.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.75$ (app t, $J=11.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.30-2.10 (m, 2 H$), 1.88-1.81(\mathrm{~m}, 1 \mathrm{H}), 1.79-1.72(\mathrm{~m}, 1 \mathrm{H}), 1.44$ ( $\mathrm{s}, 9 \mathrm{H}$ ).
${ }^{13} \mathrm{C}$ NMR (125.8 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=159.3,155.0,135.5,129.3$, $128.9,124.2,122.0,79.3,73.8,53.9,30.8,28.4\left(3 \times \mathrm{CH}_{3}\right), 26.7$.
HRMS (ESI): Exact mass calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{NO}_{3}[\mathrm{M}+\mathrm{H}]^{+}: 264.1594$, found: 264.1595.
HPLC: Chiralpak AD-H column (98:2 hexane: $i-\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 280 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=12.5 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=15.6$ $\min ; 90 \%$ ee.
(3S,4R)-3-Methyl-4-phenyl-3-(prop-2-en-1-yl)-[1,2,5]-thiadiazolidine-1,1-dioxide (24)
To a solution of the imine 23 (prepared as described previously ${ }^{8}$ by the reaction of imine $\mathbf{1 2}$ with allyltrifluoroborate $\mathbf{2 c}$, but using chiral diene ent-L1) ( $200 \mathrm{mg}, 0.80 \mathrm{mmol}$ ) in THF ( 32 mL ) at -20 ${ }^{\circ} \mathrm{C}$ was added DIBAL ( 1.0 M in THF, $3.2 \mathrm{~mL}, 3.2 \mathrm{mmol}$ ) over 2 min . The mixture was warmed gradually to room temperature over 2 h and stirred for a further 13 h . The reaction was quenched carefully with 1 M HCl solution until the pH value of the mixture reached 3. The mixture was diluted with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ and EtOAc $(30 \mathrm{~mL})$, and the organic layer was separated. The aqueous layer was extracted with EtOAc $(2 \times 30 \mathrm{~mL})$, and the combined organic layers were washed with brine $(30 \mathrm{~mL})$, dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and concentrated in vacuo. Purification of the residue by column chromatography ( $33 \% \mathrm{EtOAc} /$ hexane) gave the cyclic sulfamide 24 as a colorless amorphous solid ( $182 \mathrm{mg}, 90 \%$ ).
$\mathrm{R}_{\mathrm{f}}=0.26$ ( $30 \% \mathrm{EtOAc} /$ hexane) .
$[\alpha]_{\mathrm{D}}^{20}-75.8\left(c 0.42, \mathrm{CH}_{3} \mathrm{OH}\right)$.
IR (neat): 3271 (NH), 2980, 1454, 1381, 1312, 1265, 1157, 922, $741,702 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.43-7.36(\mathrm{~m}, 5 \mathrm{H}), 5.67$ (dddd, $J=17.1,10.2,7.8,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.21-5.19(\mathrm{~m}, 1 \mathrm{H}), 5.14$ (ddd, $J$ $=17.1,3.0,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.90(\mathrm{~d}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.82(\mathrm{~d}, J=5.1$ $\mathrm{Hz}, 1 \mathrm{H}), 4.60(\mathrm{~s}, 1 \mathrm{H}), 2.49(\mathrm{dd}, J=13.7,7.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.59(\mathrm{dd}$, $J=13.7,6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.43(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (125.8 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=134.1,131.9,128.9,128.8$ (2 $\times \mathrm{CH}), 127.3(2 \times \mathrm{CH}), 121.2,70.0,65.0,39.7,24.2$.
HRMS (EI): Exact mass calcd for $\mathrm{C}_{12} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{~S}[\mathrm{M}]^{+}$: 252.0927, found: 252.0928.

HPLC: Chiralcel OD-H column (80:20 hexane: $i$ - $\mathrm{PrOH}, 0.8$ $\left.\mathrm{mL} / \mathrm{min}, 211 \mathrm{~nm}, 25^{\circ} \mathrm{C}\right) ; \mathrm{t}_{\mathrm{r}}($ major $)=13.9 \mathrm{~min}, \mathrm{t}_{\mathrm{r}}($ minor $)=28.1$ min; $97 \%$ ee.
(1R,2S)-2-Methyl-1-phenyl-4-butene-1,2-diamine (25)
A solution of cyclic sulfamide $24(101 \mathrm{mg}, 0.40 \mathrm{mmol})$ and ethylenediamine ( $267 \mu \mathrm{~L}, 4.00 \mathrm{mmol}$ ) in dioxane $(6 \mathrm{~mL})$ was stirred at $100{ }^{\circ} \mathrm{C}$ for 15 h . The reaction was cooled to room temperature and concentrated in vacuo. To the residue was added a 1.25 M solution of HCl in $\mathrm{MeOH}(2 \mathrm{~mL})$ and the resulting solution was stirred at room temperature for 2 h before being concentrated in vacuo. The residue was dissolved in $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL})$ and washed with EtOAc $(2 \times 10 \mathrm{~mL})$. The organic layers were discarded and the aqueous phase was basified with 3 M NaOH (2 $\mathrm{mL})$ and extracted with EtOAc $(3 \times 20 \mathrm{~mL})$. The combined organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and concentrated in vacuo to leave the diamine $\mathbf{2 5}$ ( $67 \mathrm{mg}, 88 \%$ ) as a pale yellow oil.
$\mathrm{R}_{\mathrm{f}}=0.15$ ( EtOAc ).
$[\alpha]_{\mathrm{D}}^{20}-30.0(c 0.20, \mathrm{MeOH})$.
IR (neat): $3400(\mathrm{NH}), 2964,1638,1603,1492,1452,1373,999$, 914, $704 \mathrm{~cm}^{-1}$.
${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.39-7.31(\mathrm{~m}, 5 \mathrm{H}), 5.89$ (ddt, $J$ $=17.5,10.2,7.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.12(\mathrm{ddt}, J=10.2,2.0,0.9 \mathrm{~Hz}, 1 \mathrm{H})$, 5.08 (ddt, $J=17.0,2.3,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.82(\mathrm{~s}, 1 \mathrm{H}), 2.13-2.03(\mathrm{~m}$, 2 H ), 1.65 (br s, 4 H$), 1.09(\mathrm{~s}, 3 \mathrm{H})$.
${ }^{13} \mathrm{C}$ NMR (125.8 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=142.6,134.3,128.3(2 \times \mathrm{CH})$, $127.9(2 \times \mathrm{CH}), 127.2,118.3,64.2,54.7,44.0,24.7$.
HRMS (EI): Exact mass calcd for $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{~N}_{2}[\mathrm{M}]^{+}$: 190.1465, found: 190.1465.

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