Decrease in the Roughness of Vapor-Deposited Gold Surfaces Induced by Surface Mobility[†]

J. L. Zubimendi,[‡] M. E. Vela,[‡] R. C. Salvarezza,[‡] L. Vázquez,[§] J. M. Vara,^{||} and A. J. Arvia^{*,‡}

Instituto de Investigaciones Físicoquímicas Teóricas y Aplicadas (INIFTA), Casilla de Correo 16, Sucursal 4, 1900 La Plata, Argentina, Instituto de Ciencia de Materiales, Consejo Superior de Investigaciones Científicas, Departamento de Física Aplicada C-XII, Universidad Autônoma de Madrid, 28049 Madrid, Spain, and Instituto de Ciencia de Materiales, Consejo Superior de Investigaciones Científicas, Departamento de Química Física Aplicada C-II, Universidad Autónoma de Madrid, 28049 Madrid, Spain

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The evolution of the roughness exponent of vapor-deposited thin gold films (the object) on smooth glass substrate at 298 K has been observed by scanning tunneling microscopy. This evolution was enhanced by keeping the films at 323 K in contact with aqueous $0.5 \text{ M} \text{ H}_2\text{SO}_4$ containing different KCl concentrations for 24 h. It implies a decay of the surface roughness for lengths larger than the grain size of the object leading to better ordered surface structures. Results are compared to those obtained during film grown by increasing the temperature of the glass substrate from 298 to 673 K. The analysis of surface roughness data indicated an enhancement of surface atom diffusion in ordering the gold surface structure, by either anion adsorption or deposition temperature.

Introduction

Surface roughness reflects physical processes operating on solid surfaces during their growth and evolution toward the equilibrium form. To understand the dynamics of growing interfaces,¹ both continuum and atomistic models have been proposed. The main physical processes used for modeling roughness are intrinsic noise in surface growth, surface tension, and surface atom diffusion.² When the surface dynamics depends only on local processes, objects with a self-affine fractal surface with compact nonfractal mass are formed.¹ In this case, ξ , the interface width, scales with L_s , the substrate size, according to $\xi \propto$ $L_{s^{\alpha}}$, where α is the characteristic surface roughness exponent which becomes the relevant quantity characterizing the surface properties. Thus, values of α ranging between 0 and 1 are expected for 3D systems according to the three above mentioned physical contributions to surface dynamics.³

Recently, data derived from scanning tunneling microscopy (STM) imaging4-6 were used for the fractal characterization of gold film surfaces prepared by vapor deposition on a smooth glass substrate at the growth rate v = 30 nm s⁻¹, and substrate temperature $T_s = 298$ K. Then it was concluded that the surface of these films

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- [§] Departamento de Física Aplicada C-XII.
- "Departamento de Química Física Aplicada C-II.
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behaves as a self-affine fractal with roughness exponents, α (I) $\simeq 0.90$ for $L_s < d_s$, and α (II) $\simeq 0.36$ for $L_s > d_s$, where $d_{\rm s}$ is the average columnar size.⁴⁻⁶ Although the value $\alpha(II) \simeq 0.36$ appears to be consistent with the predictions of the stochastic Kardar, Parisi, and Zhang (KPZ) model,⁷ this theory does not incorporate the effect of grain formation. Thus, in principle, this coincidence appears to be fortuitous. On the other hand, the value $\alpha(I) \simeq 0.9$ for $L_{\rm s} < d_{\rm s}$, indicates a smooth textured columnar surface which can be produced by the surface mobility of gold atoms. Smoothening leads to the disappearance of irregularities within a diffusion length close to $d_{\rm s}$.^{4–6} In this case, the value of $\alpha(I) \simeq 0.9$ exceeds the value of α expected from continuum models incorporating surface relaxation,⁸ although that figure is consistent with values of α derived from simple growth models in which particles are randomly deposited onto a substrate and relax to kink sites maximizing the number of saturated bonds.^{9,10} Accordingly, it appears that surface atom diffusion changes the surface properties of deposited films from fractal to nonfractal.

A considerable increase in surface metal atom mobility can be produced during film formation by increasing the temperature of the substrate or by the adsorption of foreign particles. A strong surface rearrangement has been observed for electrodeposited metal films in contact with aqueous solutions containing adsorbable anions.¹¹

In this work the surface atom diffusion contribution to surface roughness development is analyzed during the growth of thin gold films by changing T_s , the substrate temperature, and for postdeposition surface roughness evolution by the adsorption of specific solution constituents. The disappearance of the columnar structure with $\alpha \simeq 0.4$ and the appearance of faceted surface structures imply values of changing gradually from $\alpha \simeq 0.4$ upward as the contribution of surface atom diffusion increases.

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[‡] INIFTA.

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Figure 1. STM image of a vapor-deposited gold film grown at v = 0.1 nm s⁻¹ and $T_s = 298$ K: (a, top) three-dimensional image; (b, bottom) cross section of the image shown in part a.

Experimental Section

Gold deposits were grown in an evaporator chamber on smooth glass substrates previously cleaned in an ultrasonic bath by sequentially using water, trichloroethylene, acetone, and ethanol.^{4–6} The maximum deviation of the incident particles from the substrate normal was 9°. The experimental conditions at the evaporator chamber were as follows: pressure $P = 10^{-4}$ Torr (air as residual gas), average deposit growth rate $\langle v \rangle = 0.1$ nm s⁻¹, and 298 K < $T_{\rm s}$ < 673 K. The average film thickness was $\langle h \rangle \cong 200$ nm as has been estimated by the weight method. Previous results for this system⁵ showed that for $\langle h \rangle \geq 170$ nm, the root mean square roughness of the deposit surface approached a steady state.⁵

A number of gold films grown at 298 K were kept immersed in aqueous $0.5\,M\,H_2SO_4 + xM\,KCl, 0 \leq x \leq 0.5\,M$, at 323 K under purified nitrogen saturation for 24-h periods to observe the influence of the electrolyte composition on roughness relaxation. The experimental conditions were selected on the basis of previous results on electrodeposited gold films.^{11}

A McAllister STM operating in air was used to investigate the surface morphology of gold deposits. The STM was calibrated by imaging highly oriented pyrolytic graphite (HOPG). Tips were

made from 0.25 mm diameter platinum—iridium wires directly by cutting. Although different tips were used to minimize tip geometric artifacts, no influence of the tip shape on our experimental data was found. STM measurements were made using a 0.05-V bias voltage with the tip (+) at 1–2 nA constant current. Data were acquired in a fully automated worksation and stored as digitized images with 200 × 200 pixels. Occasionally, images with 400 × 400 pixels were taken with similar results.

Usually, STM measurements involve the uncertainty as to whether the tip scans over a parallel or a tilted plane of the sample. The presence of a tilted plane introduces an overestimation in the value of α as resulted from computer-generated surfaces with added planes. STM plane removal options in the software have been successfully used to remove the tilted plane.¹² Therefore, STM data were analyzed after the instrument plane fitting and substracting procedure.⁶

Results

STM images of vapor-deposited gold film surfaces resulting from those runs made at v = 0.1 nm s⁻¹ and T_s

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Figure 2. STM images of a vapor-deposited gold film grown at v = 0.1 nm s⁻¹ and $T_s = 298$ K after the film was kept in contact with 0.5 M H₂SO₄ for 24 h at 323 K: (a, top) cross section of an image 549 × 549 nm²; (b, bottom) three-dimensional view of the same image.

= 298 K (Figure 1a) show rounded top columns with $\langle d_s \rangle$ \approx 20 nm, branched voids, and small columns which agglomerate in larger rounded elements. The cross sections of these STM images (Figure 1b) correspond to jagged profiles¹³ due to the height fluctuation of columnar elements.^{4–6} These fluctuations arise from shadowing between neighboring growing columns.⁴ As already reported, small rounded top column surfaces are rather smooth.⁴

On the other hand, after immersion in 0.5 M H_2SO_4 at 323 K for 24 h, the topography of gold films undergoes some changes as columns are aligned following preferred directions (Figure 2a). However, cross sections of these STM images (Figure 2b) show surface profiles very similar to those recorded before immersion in 0.5 M H_2SO_4 at 323

K for 24 h. After immersion in $0.5 \text{ M H}_2\text{SO}_4 + 0.01 \text{ M KC1}$ at 323 K for 24 h, the rounded shaped contour of columns tends to disappear and geometric crystals dominate the surface topography (Figure 3). This effect become more evident by increasing the KCl concentration in the solution. For instance, after immersion in aqueous 0.5 M H₂SO₄ + 0.5 M KCl at 323 K for 24 h (Figure 4a), the columnar structure of the gold film is completely suppressed and replaced by aligned geometric crystals following well-defined directions which dominate the surface topography. These crystals are 15–20 nm in average size, and their cross section, as derived from STM images (Figure 4b), shows profiles with a smoother texture in relation to those observed for gold films before their immersion in KCl-containing solutions.

Finally, for gold films prepared at v = 0.1 nm s⁻¹ and $T_s = 673$ K, STM images also show faceted surfaces (Figure 5a) with a texture smoother (Figure 5b) than that of other

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Figure 3. Three-dimensional STM image of a vapor-deposited gold film grown at v = 0.1 nm s⁻¹ and $T_s = 298$ K after the film was kept in contact with 0.5 M H₂SO₄ + 0.01 M KCl for 24 h at 323 K.

gold films studied in this work. In this case, the crystal size largely exceeds the size of those grown at $T_{\rm s} = 298$ K.

Discussion

The analysis of STM imaging data resulting from gold film surfaces was based on the dynamic scaling method.^{1,4–6} Accordingly, $\xi_{\rm stm}$, the root mean square roughness measured by STM on STM scans of length $L_{\rm s}$, scales as

$$\xi_{\rm stm}(L_{\rm s},h) \propto L_{\rm s}^{\alpha} f(x) \tag{1}$$

where $f(x) = \langle h \rangle / L_s^{\gamma}$, $f(x) = \text{constant for } x \to \infty$, and $f(x) = x^{\alpha/\gamma}$ for $x \to 0$, and ξ is defined as

$$\xi_{\rm stm}(L_{\rm s}) = [(1/N) \sum [h(x_{\rm s}) - \langle h \rangle]^2]^{1/2}$$
(2)

where $h(x_i)$ is the deposit height measured along the *x*-direction at the position x_i , and $\langle h \rangle$ is the average height of the sample formed by *N* points. After a certain critical time or thickness is reached, eq 1 becomes

$$\xi_{\rm stm}(L_{\rm s}) \propto L_{\rm s}^{\alpha} \tag{3}$$

According to eq 3 the value of α can be obtained from a log $\xi_{\rm stm}$ vs log $L_{\rm s}$ plot. For $T_{\rm s} = 298$ K, log $\xi_{\rm stm}$ vs log $L_{\rm s}$ plots exhibit two straight line portions with a crossing point at log $L_{\rm s} = 1.7$ and a saturation region close to the image size (Figure 6a). The slopes of the straight lines are α (I) = 0.77 ± 0.07 for log $L_{\rm s} < 1.7$, and α (II) = 0.37 ± 0.06 for log $L_{\rm s} > 1.7$. It should be noted that for computer-simulated fractals, data covering 4 to 5 orders of magnitude are required for logarithmic fitting, but for experimental systems, this goal is less ambitious due to specific limitations. Then, log $\xi_{\rm stm}$ vs log $L_{\rm s}$ linear plots covering at least 1 order of magnitude or thereabouts are considered acceptable from experimental data.

As already reported,⁵ the capability of the dynamic scaling method for rough surface analysis was tested for several computer-generated self-affine fractal surfaces

with preset values of α_{th} , built on 200 \times 200 and 256 \times 256 square grids, using the successive random addition algorithm.¹⁴ A good agreement between α_m , the measured value of α , and α_{th} was found in the 0.2 < α_{th} < 0.6 range, whereas for $\alpha_{th} > 0.6$, α_m became smaller than α_{th} . A better agreement with the data can be obtained by increasing substantially the number of points in the square grid, for instance, to 1000×1000 pixels. However, the $200\ \times\ 200$ grid involves a short STM imaging time minimizing drift effects. Thus, from the compromise between grid size and image drift, it is advisable to select a small grid to minimize drift effects and use the α_m vs α_{th} plot to make a correction to the α values measured from STM images. Then, after correction $\langle \alpha(I) \rangle$ and $\langle \alpha$ -(II)), the averaged values of α (I) and α (II) resulting from 20 different STM images taken in the 20 nm $< L_{\rm s} < 550$ nm range are $\langle \alpha(I) \rangle = 0.92 \pm 0.07$ and $\langle \alpha(II) \rangle = 0.40 \pm$ 0.05. Similarly, for gold deposits grown at $T_s = 298$ K and kept in contact with 0.5 M H_2SO_4 at 323 K, the log ξ_{stm} vs log L_s plots yielded after correction $\langle \alpha(I) \rangle = 0.90 \pm 0.05$ and $\langle \alpha(II) \rangle = 0.43 \pm 0.04$.

For gold deposits grown at $T_s = 298$ K after being maintained in contact with $0.5 \text{ M H}_2\text{SO}_4 + 0.01 \text{ M KCl}$, the log $\xi_{\rm stm}$ vs log $L_{\rm s}$ plots also yielded two straight lines, although in this case after correction it resulted in $\langle \alpha(I) \rangle$ = 0.90 for log $L_{\rm s}$ < 1.5, and $\langle \alpha({\rm II}) \rangle$ = 0.49 ± 0.06 for log $L_{\rm s} > 1.5$. This increase in the value of $\langle \alpha({\rm II}) \rangle$ suggests that smoothening of the original topography has occurred due to the presence of Cl⁻ ions in the solution. The effect of these ions as promoters of gold surface smoothening was confirmed by increasing the KCl concentration in the solution. Thus, for gold deposits grown at $T_s = 298$ K and kept in contact with 0.5 M $H_2SO_4 + 0.5$ M KCl at 323 K for 24 h, the log ξ_{stm} vs log L_{s} plot exhibits two straight lines with $\alpha(\text{I}) = 0.74 \pm 0.07$ for log $L_{\text{s}} < 1.7$, and $\alpha(\text{II}) \cong$ 0.54 ± 0.05 for log $L_{\rm s} > 1.7$ (Figure 6b). In this case, from the analysis of 20 different STM images covering the 20nm $< L_{\rm s} < 550$ -nm range, after correction, it results in $\langle \alpha(I) \rangle = 0.88 \pm 0.07$ and $\langle \alpha(II) \rangle = 0.56 \pm 0.05$.

It should be noted that gold deposits grown at 298 K and aged in different electrolyte solutions at 323 K for 24 h exhibit a crossover point in the log ξ_{stm} vs log L_s plots covering the 20-nm < L_s < 40-nm range, i.e. at L_s values

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Figure 4. STM images of a vapor-deposited gold film grown at v = 0.1 nm s⁻¹ and $T_s = 298$ K after the film was kept in contact with 0.5 M H₂SO₄ + 0.5 M KCl for 24 h at 323 K: (a, top) three-dimensional image; (b, bottom) cross section of the image shown in part a.

Table 1.	Corrected	Average	Values o	fα Res	ulting at
Dif	ferent Elec	trolyte C	ompositi	on and	Ts

<i>T</i> _s , K	eletrolyte solution	temperature, K	$\langle \alpha(I) \rangle$	$\langle \alpha(II) \rangle$
298	none		0.92 ± 0.07	0.40 ± 0.05
298	H_2SO_4	323	0.90 ± 0.07	0.43 ± 0.04
298	H_2SO_4	323	0.90 ± 0.07	0.49 ± 0.06
	+ 0.01 M KCl			
298	H_2SO_4	323	0.88 ± 0.07	0.56 ± 0.06
	+ 0.5 M KCl			
673	none		0.93 ± 0.07	

which are close to the value of d_s . Then, the value $\langle \alpha(I) \rangle \cong 0.9$ obtained for $L_s < 20-40$ nm should be assigned to the smooth surface domains of either columns or geometric crystals forming the deposits, whereas values of $\langle \alpha(II) \rangle$, which are always lower than those of $\langle \alpha(I) \rangle$ for $L_s > 20-40$ nm, have to be related to height fluctuations involving either columns or crystals.

The interpretation of the physical origin of the two regions in the log ξ_{stm} vs log L_s plots which are revealed by changing the lateral correlation length, i.e. the microscope resolution, can be supported by scaling computer-simulated surfaces resulting from an ensemble of columns fluctuating in height with known values of α_s . The surface of each column is represented by a smooth rounded cusp of diameter $d_{\rm s}$. For this surface model the $\log \xi$ vs $\log L$ plots exhibit one or two linear regions depending on the L/d_s ratio. For $L/d_s \gg 1$ only one region with slope α_s is observed. Otherwise, for $L/d_s \leq 20$ two linear regions with $\alpha \rightarrow 1$ for $L < d_s$ and $\alpha = \alpha_s$ for $L_s > d_s$ d_s are observed. Finally, for $L/d_s \rightarrow 1$ only the region with $\alpha \simeq 1$ results from the log ξ vs log *L* plots. Note that α $\simeq 1$ is consistent with a smooth rounded cusp surface. Thus, from these simulations it can be concluded that the two regions in the log ξ vs log *L* plots correspond to truly different scaling ranges.



Figure 5. STM images of a vapor-deposited gold film grown at v = 0.1 nm s⁻¹ and $T_s = 673$ K: (a, top) three-dimensional image; (b, bottom) cross section of the image shown in part a.

On the other hand, for gold deposits grown at v = 0.1 nm s⁻¹ and $T_{\rm s} = 673$ K, i.e. those films which exhibit a predominantly faceted topography, the log $\xi_{\rm stm}$ vs log $L_{\rm s}$ plots (Figure 6c) lead to a single straight line with $\alpha({\rm I}) \simeq 0.76 \pm 0.07$ and a small saturation region which appears when $L_{\rm s}$ is close to the STM image size. Then, from the analysis of 20 STM images for $T_{\rm s} = 673$ K, in the 20-nm $< L_{\rm s} < 550$ -nm range, after correction, it results in $\langle \alpha({\rm I}) \rangle = 0.93 \pm 0.07$.

The preceding $\langle \alpha \rangle$ values were confirmed by using $G(L_s,t)$, the height-to-height correlation function, resulting from the analysis of STM images.¹ In this case, the following equations were used to evaluate α :

$$G(L_{\rm s}) = [1/N\sum [h(x_1) - h(x_0)]]^2$$
(4)

and

$$G(L) \propto L_{\rm s}^{2\alpha}$$
 (5)

Again, $\langle \alpha \rangle$ values derived in this way are in good agreement

with those obtained by the dynamic scaling (Table 1).

Corrected values of $\langle \alpha \rangle$ resulting at different $T_{\rm s}$ and solution composition are assembled in Table 1. These results suggest a change in the surface characteristics of vapor-deposited gold films from fractal to nonfractal, which depends on the mobility of gold atoms at the surface. It is well-known that surface mobility of metal atoms can be enhanced either by increasing $T_{\rm s}$ or by the presence of adsorbates.¹⁵ In aqueous H₂SO₄ solutions at 323 K, the surface mobility of gold atoms is relatively small,¹⁶ and accordingly, the surface characteristics of vapor-deposited gold films remain close to those observed before aging the film in the electrolyte solution, at least for the interval of time covered by this work.

On the other hand, the addition of KCl to aqueous H_2SO_4 solutions at 323 K leads to an increase in the value of α . This means that in the KCl-containing solution the gold film surface smoothness is greater than that for those

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Figure 6. log ξ_{STM} vs log L_s plot from gold films grown at v = 0.1 nm s⁻¹: (a, top) $T_s = 298$ K; (b, middle) $T_s = 298$ K after the film was kept in contact with 0.5 M H₂SO₄ + 0.5 M KCl for 24 h at 323 K; (c, bottom) $T_s = 673$ K.

films without aging, or after aging in plain aqueous H₂SO₄

solution. This fact agrees with previously reported observations¹⁷ that small amounts of Cl⁻ ions in acid solution are able to eliminate surface roughness at metal surfaces.¹⁷

The values $\alpha(II) \cong 0.43$ and $\alpha(II) \cong 0.56$ obtained in KCl-free and 0.5 M KCl-containing H₂SO₄ aqueous solution, respectively, lie between $\alpha(II) \cong 0.4$ and $\alpha \cong 0.90$ obtained at the low and high T_s values. These data suggest that the surface diffusion of gold atoms partially removes surface irregularities for lengths greater than d_s , depending on the surface mobility induced by anion adsorption. This explanation is basically supported by the increase in D_s , the surface diffusion coefficient of gold atoms, by the presence of Cl⁻ ions in the solution. Thus, D_s changes from 10^{-14} cm² s⁻¹ in 0.5 M H₂SO₄ to 10^{-11} cm² s⁻¹ in H₂SO₄ + 0.05 M KCl.¹⁶

Values of α (II) shown in Table 1 indicate a continuous variation of α describing the roughness decay which leads to a faceted surface with $\alpha = 0.90$. However, for $t \rightarrow \infty$, the enhanced gold surface atom mobility assisted either by increasing T_s or by anion adsorption would tend to produce a completely smooth surface ($\alpha = 0$). Thus, the observed values of α (II) and α (I) $\cong 0.90$ would reflect transient regimes of the surface, and therefore, a crossover to the Edwards–Wilkinson scaling¹⁸ ($\alpha = 0$) would be expected.

Conclusions

Results shown in this work provide a quantitative approach to the role of gold surface atom diffusion in the change of fractal properties of gold film surfaces. Accordingly, the disordered structure produced by vapor deposition at $T_s = 298$ K and characterized by $\alpha \cong 0.4$ tends to disappear either by the presence of adsorbable anions from the electrolyte solution or by increasing the substrate temperature leading to better-ordered surface structures. The extent of surface rearrangement in aqueous solutions depends on the strength of anion adsorption and anion concentration in the electrolyte.

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