

Monitoring Data of Electrokinetic Stabilisation Method for Soft Clay using EKG Electrode

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Keywords: Electrokinetic Stabilisation, Soft Clay, Electrokinetic Geosynthetic, Electroosmosis.

Abstract. Electrokinetic Stabilisation (EKS) method has the potential to overcome problems on highly compressible clay. This study presents the monitoring results from an experimental study of EKS on soft clay soil. Inactive kaolinite clay, inert electrode and distilled water (DW) were used as a pure system mechanism before any chemical stabilisers will be used in this research. Therefore, this monitoring data will provide a baseline study to improve efficiency of the EKS approach for ground improvement application. The test model were used inert electrodes of Electrokinetic Geosynthetic (EKG) developed at the Newcastle University to apply a constant voltage gradient of 50 V/m across a soil sample of 400 mm length. The distilled waters were used at the pore electrolyte fluid compartments and supplied under zero hydraulic gradient conditions for periods of 3, 7 and 14 days. Throughout, monitoring data of electric current for all treatments were measured. Results showed that the electric current trend in this pure system was attributed to the electrochemical effects in the clay-water electrolyte system.

Introduction

The application of deep chemical ground improvement using the Electrokinetic Stabilisation (EKS) method has the potential to overcome problems on highly compressible soft soil [1,2,3,4]. This technique also has the potential to be applied to treat contaminated soils for geoenvironmental application [5,6]. Electrokinetic stabilisation of clay soils using calcium-based stabilizers [2,3], lime [7] and other types of stabilizers [8,9] have been used widely for geotechnical application. In some cases, these methods are suitable to stabilise soil under existing structure [1,8]. Therefore, it is expected that chemical improvement of soils in situ using electrokinetic stabilisation will play a very important role in a future not only as an alternative method for problematic soil whether under existing foundation or as part of a new development.

Current research [1] conducted at the University of Birmingham, successfully proved the concepts and established the practical possibilities of improving the classical chemical modification and stabilisation reactions in situ. At the same time this work overcoming problems caused by electrode degradation [10]. However, further investigation need to be investigated to assess what happens when appropriate selected chemical stabilisers are added to clay soils, both when mixed and then transported using electric currents. For that reason, the present study developed a new approach to treat soft clay soils in order to modify and improve their properties. By providing a better understanding of the 'pure' relatively inert system, which has no chemicals introduced, this hopefully will provide a baseline study to improve efficiency of the EKS approach when applied to soft compressible soils. Thus this technique has potential to enhance key strength and stiffness gain under foundations to ensure this is controlled in the most efficient way.

Materials and Methods

Development of Experimental Apparatus. An experimental programme was conducted to assess the efficiency of EKS when applied to soft clay soil. The EKS model apparatus consisted of tank made of non-conductive material (PVC-U sheet) to prevent short circuiting. The PVC-U sheet has 12 mm thick with internal dimensions of 450 x 220 x 550 mm. The tank consists of main compartment of internal dimension 370 x 220 x 550 mm to locate soil samples for treatment and two small compartments that had internal dimension of 40 x 220 x 550 mm to supply the distilled waters (DW) into the soil. A schematic diagram of the EKS testing model is shown in Figure 1.

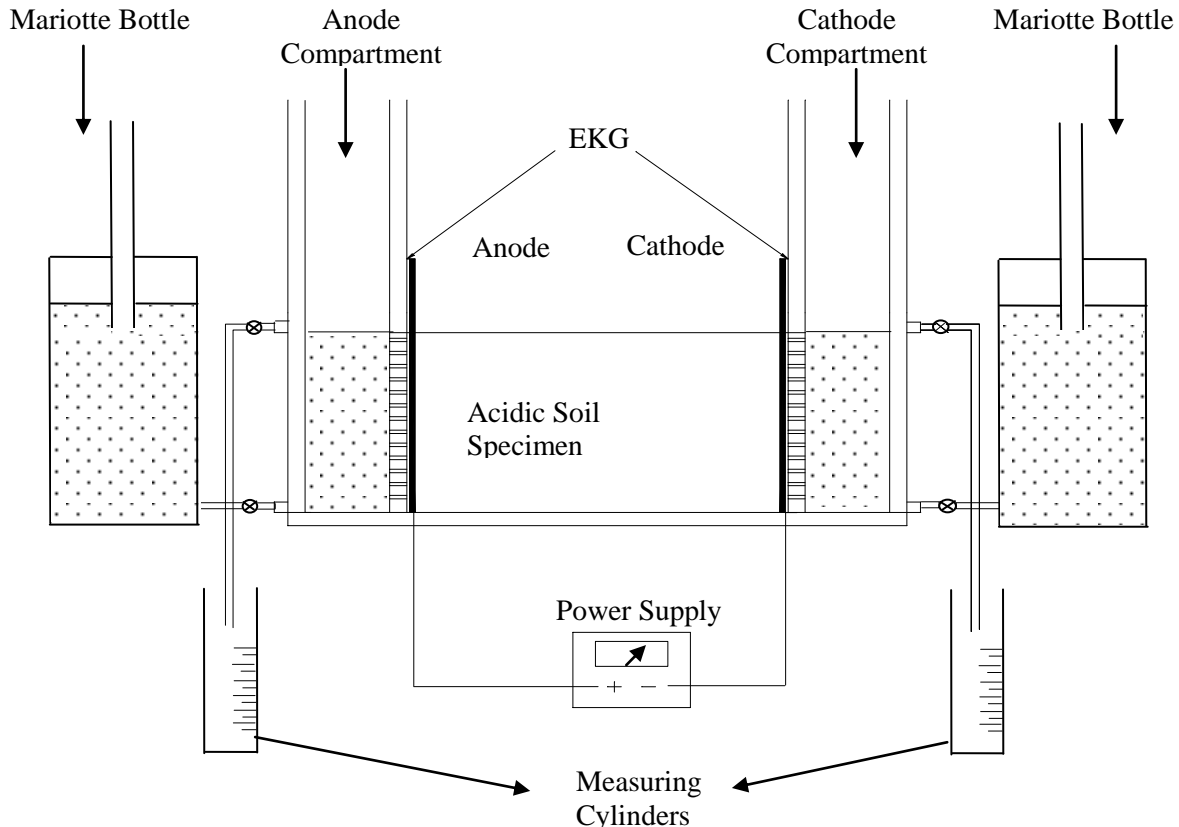


Figure 1: Schematic Diagram of EKS Testing Model

A DC power supply obtained by Thurlby Thandar Instruments (model TS3022S) was used to supply power to the electrodes. A constant voltage gradient of 50 V/m was used for all electrokinetic treatment for 3, 7 and 14 days so that the effects of electric current with time could be assessed.

Soil Specimen Preparation. A consistent method of soil specimen preparation was employed throughout this work for all treatment periods. The slurry sample was prepared by mixing the clay soils with deionised water to achieved 90% water content. The water content of slurry was chosen based on 1.5 time liquid limit (LL) to produce homogeneous sample. Approximately, 10.66 kg of deionised water was poured into the mixer bowl and then 11.84 kg of dry English China clay was gradually added to ensure consistency of mixing. The slurry sample was then mixed using Hobart mechanical mixer and blended thoroughly for 30 minutes. Two mixtures of slurry samples need to be filled in the tank with total of amount of soil and water was 23.68 and 21.32 kg, respectively, achieving a water content of 51 % for all test samples.

The initial height of the slurry is 386 mm for all tests samples. The load was applied to consolidate the slurry sample until it reached final height of about 271 mm using hydraulic jack to achieve moisture content of 51%. The first consolidation pressure was set at approximately 10 kPa for about 24 hours to prevent soil particle migration via the drainage. The subsequent surcharge pressures were gradually increased in four increments over a 4 days period until final height was reached. After consolidation thin solid plastic walls were removed at both end of soil compartment and the gaps left were then inserted by the Electrokinetic Geosynthetic (EKG) electrodes. The soil samples were reconsolidated under the same loading as the last consolidation (67.9 kPa) for about 24 hours before the loading was released upon EKS testing. Subsequently, the electric current were turned and monitored throughout the test durations. This was performed to make sure a good contact between soils and electrodes. Physical and chemical properties of the treated soil were carried out along the soil profiles that were divided into eight sections from anode to cathode.

Results and Discussions

Basic Properties of Soil. The basic physical and chemical properties of clay soil were investigated first through laboratory programme in accordance to British Standard as shown in Table 1. Then, a laboratory investigation of EKS approach was carried out which was developed at University of Birmingham. Commercially available EKG electrodes were used to prevent the degradation of electrode especially at the anode that requires replacement during the treatment. In order to avoid the complexity of the system with different clay minerals and chemical changes that associate with coupled flows, pH gradient and electrolysis and redox reactions, a relatively ‘pure’ and inert form of kaolin (English China Clay) was used in this study supplied by WBB Devon Clays. In addition, kaolin clay was selected in this study due to its low activity, low adsorptive capacity and high electro-osmotic transport efficiency compared with other clays.

Table 1: Properties of English China Clay

Particle Size Distribution :	
Silt	49 %
Clay	51 %
Atterberg Limit :	
Liquid Limit	60 %
Plastic Limit	34 %
Permeability :	
Falling Head Test	2.2×10^{-8} m/s
Oedometer Test	2.6×10^{-10} m/s
Average specific gravity (G_s) :	2.63
Compression index (C_c) :	0.365
pH value :	5.1
Conductivity :	19.6 μ S/cm
Surface Area :	8 – 10 m^2 /gm

Monitoring Data. Figure 2 shows monitoring of electric current for 3, 7 and 14 days of DW-DW system. It was observed that electric current of these 3 different periods shows similar trend except for 14 days treatment period which has lower value of about 9 mA at the beginning of test, compared with 3 days and 7 days treatment periods which has value of 12 and 13 mA, respectively.

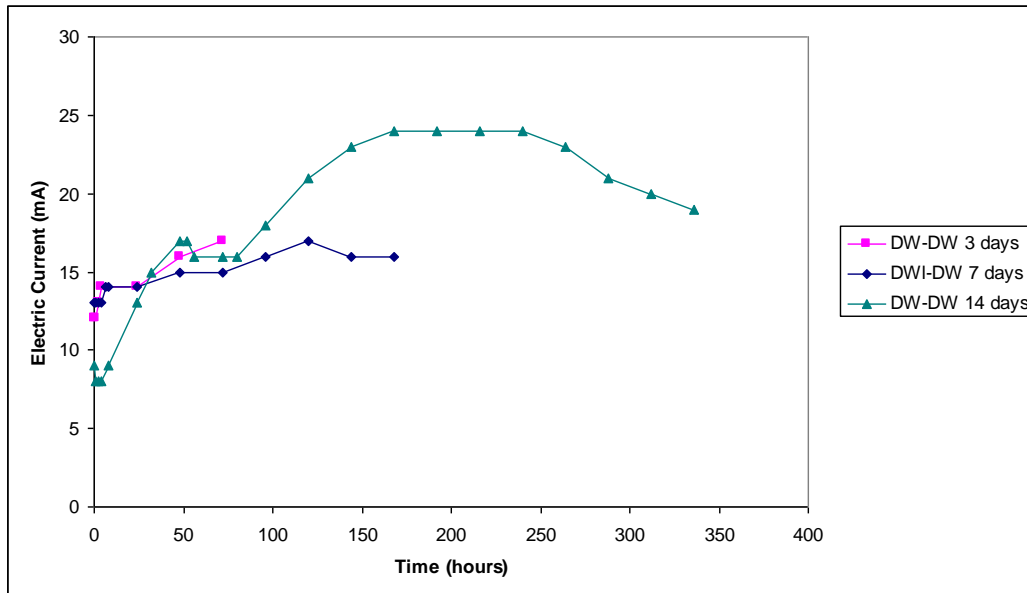


Figure 2: Electric Current of DW-DW system with time.

After 24 hours duration of EKS testing, the electric current values for 3, 7 and 14 days treatment period were reached at about 13 to 14 mA. The value of record current maintained these values, with similarly seen until 96 hours with both 7 days and 14 days tests. However, after this time for the 14 days test current rose, peaking at around 170 hours at 24 mA before dropping over the last 4 days (96 hours) to a value of 19 mA. The electric current trend in this pure system was attributed to the electrochemical effects in the clay-water electrolyte system which varied across the samples with time, thus affected the current profiles during tests.

The electric current variations with treatment time for the same type of electrode are comparable with Liaki [10] who reported that the current initially increase at about 14 mA, while it then drops within the first 100 hours of treatment, as resistant of the materials increases and the voltage was set at the constant level. However, it was observed a large variation especially at the beginning which reported by Liaki [10] with different treatment time. This was due to the soil electrode contact which varied upon insertion of electrode during sample preparation which causes large variation especially at the beginning. Therefore, the improvement of the procedures and design configuration made for the current research has overcome the problem related to the soil-electrode interfaces. It should be note that after consolidation thin solid plastic walls were removed at both end of soil compartment and the gaps left were then inserted by the EKG electrodes. In order to make sure a good contact between soils and electrodes, the soil samples were subjected to load under the same loading of the last consolidation for about 24 hours before electric current were turned on.

Conclusion

The results present and discuss in this study are comparable with previous study performed by Liaki [10]. However, the improved version of the procedures and design considerations have been made in this study to solve the problem faced by many researchers [7,8,9,10] and to achieve the objectives of this study before any chemical stabilisers will be introduced into the system. The results from monitoring data showed that there are several important mechanisms relating to the electrochemical effects and these can be drawn from this study using a pure system. The electric current profiles are influenced by the ionic concentration of the pore fluid within the soil matrices. Therefore, the addition of distilled water at both anolyte and catholyte compartments had caused increase of the electric current at the beginning of the test, but then decreased with treatment time indicating that a high resistivity zone forms due to precipitation of metal hydroxide at the vicinity of the cathode from the clay mineral itself and pore fluid. This phenomenon is also caused the fluid flow to stop, thus decreasing the electric current across the soil sample.

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