

W-doped TiO₂ as electron transport layer for high performance solution-processed perovskite solar cells

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Abstract

In this work, W-doped TiO₂, was successfully fabricated, as electron transport layer (ETL) in perovskite solar cells (PSCs) *via* low-temperature solution-processed method, whose outstanding performances were verified. The experimental results reveal that, W⁶⁺ was doped into the TiO₂ lattice successfully. The improved ETL exhibit significantly improved on conductivity, transport and extraction of photo-generated carriers and less trap-state density as compared with the pristine TiO₂ films. Meanwhile, strengthening the velocity of carrier and collection of sufficient charge can efficiently increase short-circuit current (J_{SC}) which favor an improved fill factor (FF) and a higher power conversion efficiency (PCE). Our optimized cation-anion-mixed PSC based on W-doped TiO₂ has achieved an efficiency of 18.85%. This improved PCE is almost 28.1% higher than that of the pure TiO₂. This study provides a promising approach to design metal-doped TiO₂ for fabricating high-performance perovskite solar cells.

Keywords: Perovskite solar cells; W-doped TiO₂; low temperature; Electron transport layer

1. Introduction

Organic–inorganic hybrid halide perovskite solar cells (PSCs) have been considered as a promising photovoltaic device which may replace the traditional silicon solar cells for solar energy harvesting since its adjustable bandgap, low rate of recombination, redundant electron-hole diffusion length and simple fabrication procedure[1-3]. TiO₂ are generally employed as the electronic transporting layer (ETL) in the N-I-P structure of PSCs. From 2009 to now, a rapidly rise of power conversion efficiency (PCE) for PSCs (from 3.8% to 25.2%) has been witnessed[4-7]. In PSCs, transporting electrons, blocking holes and avoiding them recombination are the major functions of the ETL [8-10]. Metal oxides like TiO₂, ZnO, SnO₂, WO_x and WO_x-TiO_x *etc.*, were frequently used in previously reported literatures[11-16]. In Hao's team, several optimized ETL were employed in getting the high-efficiency and stable performance PSCs, for instance, polyelectrolyte-doped SnO₂[17], coupled-ZnO layer[18], heterojunction-contact NiO[19]. Among the commonly used ETL in PSCs, TiO₂ has demonstrated superiorities such as excellent electron conductivity, well matching with the energy band of perovskite layer and ease in fabrication[20-25]. Therefore, it is quite essential to fabricate excellent TiO₂ layer for high-efficiency PSCs. However, TiO₂ has to be sintered at a high-temperature (> 450 °C) to obtain crystallinity and high carrier mobility[26], which complicates the preparation technology and confines the development of the flexible device. To address these issues, several novel low-temperature (<150 °C) deposition methods have been developed to fabricate TiO₂ ETL

such as chemical bath deposition (CBD)[27, 28], atomic layer deposition[29-31] and magnetron sputtering[32, 33]. Among them, Aswani Yella *et al.* have successfully fabricated the rutile TiO₂ through low-temperature (<100 °C) solution process[34]. In Ma's report, NaCl was doped on TiO₂ to passivate the interface of ETL and perovskite through solution-processed method at 70 °C[35]. MXene and bis-PCBM have been utilized to optimize the PSCs' device efficiency and stability by Ye *et al.* by solution process[36, 37]. Recently, a two-fluid spray coating method has been developed by Wu *et al.* to synthesize nanostructured TiO₂ under 150 °C[38].

TiO₂, on the other hand, is widely known as the n-type semiconductor which can be conducted by the oxygen vacancies formed in the preparation process. Nonetheless, owing to the lower migration rate of carriers and deep level defects in TiO₂, electrons formed in the perovskite layer can not transport efficiently and the proposed devices always suffer with a low performance. Doping metal ions is a valid way to boost the carrier concentration of semiconductors and reduce deep level defects[39]. To date, metal ions such as lithium[40, 41], magnesium[42-44], zinc[45], aluminum[46] and rubidium[47] have been reported to be doped into the TiO₂ lattice accordingly[48]. Niobium-doped TiO₂ as the ETL will increase the conductive path and strengthen the contact of interface in PSCs[49]. Yttrium-doped TiO₂ and Tin-doped TiO₂ have been used in PSCs to enhance electron extraction and transport and suppress the carrier recombination in interface[50, 51]. Tungsten-doped TiO₂ was utilized as photoanode

in PSCs and produced a PCE of 10.6%[52]. Even if these strategies can ameliorate the defects as mentioned above, their preparation methods are all suffered from a high annealing temperature (300-500 °C), which largely limited the manufacture of large-area devices and the applications on flexible substrates. Therefore, seeking one solution method to prepare optimized TiO₂ at low temperature is imperative and certainly a worthwhile subject.

In this work, one low-temperature solution method to obtain the crystallized TiO₂ film doped by W⁶⁺ was designed, which can extract more carriers on the interface between ETL and perovskites. According to the previous report, tungsten has been widely used in optimization on the performance for PSCs[52-59]. Among them, W-doped TiO₂ ETL and WO_x/TiO₂ bilayer have presented smoothy ETL with high crystallinity and enhanced conductivity. Although these demonstrated enhancements in TiO₂ optimized by W make this effective ETL for efficient PSCs, such as boost the carrier concentration and passivate the trap states in TiO₂, most of them are suffered from the high-temperature fabrication process. As shown in **Fig. 1**, in contrast with these previous studies, the solar cells in this work have shown excellent performance, given the short circuit current (J_{sc}), power conversion efficiency (PCE) and fabrication temperature[15, 41, 52-58]. In addition, optimized PSC with an 0.3 mol% W-doped TiO₂ as ETL and the mixed perovskite material based on Cs_{0.05}(MA_{0.17}FA_{0.83})_{0.95}Pb(I_{0.83}Br_{0.17})₃[60] as light absorption layer has achieved a better performance, including the J_{sc} increasing

from 22.73 mA/cm² to 24.26 mA/cm² and the fill factor (FF) from 58.37% to 69.22%.

As a consequence, a substantially PCE of 18.85% has been achieved.

2. Experimental section

2.1 Materials

Tungsten chloride (WCl_6 , 99.9% metals basis) was purchased from Sigma Aldrich. Dimethylsulfoxide (DMSO), γ -butyrolactone (GBL) and chlorobenzene were bought from Alfa Aesar. PbI_2 , Li-TFSI, Spiro-OMeTAD, 4-tert-butyl pyridine and methylammonium iodide (MAI) were bought from Xi'an Polymer Light Technology Corporation.

2.2. Fabrication of perovskite solar cells

The fabrication process of PSCs starts with commercially available F-doped SnO_2 (FTO) substrates with a sheet resistance of $8 \Omega^{-2}$ and more than 82% optical transmission in the visible light range, which were pre-cleaned. With deionized water, acetone and isopropyl alcohol, the cleaned substrates were treated in the UV-ozone for 15 min so as to improve the formation of TiO_2 layer. To prepare the precursor solution, the $TiCl_4$ solution was diluted into deionized water to 200 mM at 0 °C. WCl_6 was dissolved into ethyl alcohol to prepare solution with different mole ratios (0.1%, 0.3%, 0.5%), then they were dropped into the diluted $TiCl_4$ solution, respectively, to form the precursor solution. Next, the cleaned FTO substrates were immersed to the precursor solution at 70 °C for 1 h. Finally, the substrates deposited with TiO_2 were

cleaned with deionized water and ethanol and then dried at 120 °C for 1h.

To prepare the perovskite active layer, FAI, PbI₂, MABr and PbBr₂ (1:1.1:0.2:0.22 M ratio) were stirred in a mixture of DMF: DMSO: N-methyl-2-pyrrolidone (NMP) (the volume ratio is 16:4:1) in a dry gas environment. CsI was added in the DMSO solution (the volume ratio is 5:95), then it was put into the first mixed solution at 60 °C for around 2 h. Then, the mixed perovskite precursor solution was spin-coated on TiO₂ layer by a two-step process involving a low-speed spinning at 1000 rpm for 10 s and a high-speed spinning at 6500 rpm for 20 s, respectively. Chlorobenzene antisolvent was dropped on substrates at the 25th second. Next, substrates were annealed at 100 °C for 1 h. The spiro-OMeTAD layer was spin-coated on top of the perovskite film as the hole transport layer at 3000 rpm for around 30 s. Gold electrode (80 nm) was eventually deposited through the Trovato thermal evaporation at 7×10^{-6} Torr, and the area of this device is 0.04 cm².

2.3. Instruments

X-ray diffraction (XRD) patterns were analyzed by Rigaku Ultima IV with a 0.1541 nm Cu Ka line. Thermo Scientific Escalab 250Xi was used to obtain the X-ray photoelectron spectroscopy (XPS) and Ultraviolet photoelectron spectroscopy (UPS) information. V-730 Jasco spectrometer was employed to obtain the absorption spectra. The morphology of the sample was measured by the scanning electron

microscopy (SEM) and atomic force microscopy (AFM) with SU-8010 HITACHI and Quanta 200 FEG, FEI. The current density-voltage (J-V) measurements were got by 300 W xenon lamp (Newport 66984). Keithley 2400 system and 100 mW/cm² AM 1.5 G standard Newport sun simulator. Incident photo-to-current conversion efficiencies (IPCE) was employed by Zolix, Solar Cell Scan100. Steady-state photoluminescence (PL) and time-resolved photoluminescence (TRPL) data were analyzed by FLS1000 spectrometer. Electrochemical impedance spectroscopy (EIS) was tested by a Donghua DH7000 electrochemical workstation. Film surface roughness were

3. Results and discussion

Firstly, we placed the FTO substrates into the prepared precursor solution. After cleaning and drying, the W-doped TiO₂ with different concentrations was coated on the FTO substrates. As shown in **Fig. 2a**, XRD was utilized to analyze the crystallinity of prepared W-doped TiO₂ films and confirm that TiO₂ is characterized by the rutile phase[61]. From tetragonal structure of TiO₂ [JCPDS Card No. 21-1276], diffraction peaks at 25.3°, 37.7°, 48.0°, 53.9°, 62.8° and 68.8° can be ascribed to the (101), (004), (200), (105), (213) and (116) crystal planes[61]. Apparently, all peaks shown are similar no matter how many dopants are doped in TiO₂ and characteristic peaks of WO₃ are not appeared. Further more, the crystallinity of TiO₂ did not transform after doped

with tungsten. Since the W^{6+} has the similar ionic radius with Ti^{4+} , the W^{3+} was hence doped into the lattice of TiO_2 by the lattice replacement, which is well consistent with the previous studies on W-doped TiO_2 [62, 63]. Besides, there is no obvious peak shift and change of full width in half-maximum of rutile phase diffraction (101), which means that the different doping contents do not change the rutile crystal and crystallinity heavily (**Fig. 2b**). **Fig. 2c and d** show the SEM images of the pristine and 0.3 mol% W-doped TiO_2 . In Fig. S1, atomic force microscopy (AFM) was carried out to characterize the surface morphology. The root mean square (RMS) roughness of W-doped TiO_2 (RMS:17.8 nm) exhibits less hole than pristine TiO_2 (RMS:13.7 nm). A more compact TiO_2 layer is formed after W doping, which suggests that the W doping contributes to the formation of a dense and smooth morphology with a lower recombination site, and an efficient carrier transport ability can be expected accordingly.

XPS was used to analyze the chemical element of W in TiO_2 . **Fig. 3a** shows the electron binding energy of O 1s, Ti 2p and 0.3 mol% W-doped TiO_2 sample and proves that the existence of W element in 0.3 mol% W- TiO_2 . In **Fig. 3b and c**, detail information of Ti 2p and O 1s spectra are then exhibited. Binding energy peaks at 458.6 and 458.7 eV are corresponding to $Ti^{4+}2p_{3/2}$ of pure and doped TiO_2 , respectively. The Ti 2p peak clearly shows a slight up-shift after optimization since the different electronegativity among titanium atoms (1.54) and tungsten atoms (2.36), indicating negative charge transfer to tungsten in Ti-O-W bond[64]. Hence, the core level binding energy of Ti 2p

will be increased. Moreover, considering the similar radius of W^{6+} (0.605 Å) and Ti^{4+} (0.60 Å), W^{6+} ions can displace Ti^{4+} easily, forecasting that W is involved in the lattice of TiO_2 [63, 65]. The peaks at 531.26, 529.8 eV for undoped TiO_2 and 531.39, 529.75 eV for 0.3 mol% W-doped TiO_2 , respectively. After deconvolution of O 1s in Fig. S2b, 0.3 mol% W-doped TiO_2 shows four peaks. The peaks at 532.07 and 531.27 eV are assigned to O_2 molecules and -OH assembled on surface, where the peaks at 530.14 and 529.55 eV are correspond to oxygen vacancies and Ti-O-W bound after doping tungsten ions[66]. Some people alleged the peak at 530.14 eV is characteristic of the W-O bound[65, 67]; Nevertheless, others claimed it should be the -OH assembled on the surface[68]. Hence, the metal-oxygen increase is consistent with XPS analysis, and this slight higher chemical shift indicates W doping can influence the chemical states. The XPS result in **Fig. 3d** exhibits the W^{6+} 4f signals in pristine and 0.3 mol% W-doped TiO_2 . Fig. S2c reveal two deconvoluted peaks of W $4f_{5/2}$ (37.6 eV) and W $4f_{7/2}$ (36.4 eV), indicating the W^{6+} appears in the W-doped TiO_2 layer[69]. Bind energy difference of two peaks is caused by Ti 3p level at 36.9 eV[70]. In other words, the presence of W is clearly shown in XPS and undetectable tungsten-associated phase found in XRD analysis which further indicates the W element is doped into the lattice of TiO_2 successfully.

To evaluate the function of W dopant on the direct-current conductivity (σ_0) of TiO_2 layer[71], we fabricated a device structured with FTO/ TiO_2 layer/Au. The I–V curves

are displayed in **Fig. 4a** and σ_0 of various TiO₂ can be obtained using Eq. (1)

$$I = \sigma_0 A D^{-1} V \quad (1)$$

A represents the area of device (4 mm²) and D represents for the TiO₂ thickness (70 nm). The σ_0 was calculated and shown in **Table S1**. It is clearly seen the σ_0 has increased with the increase of W doping ions. This result ascribed to the higher shift of Fermi energy for doping W element which will increase the amount of free electrons since forming additional oxygen vacancies on TiO₂. In **Fig. 4b**, to explore the carrier mobility of TiO₂ film, the current density voltage (J-V) curves of simple devices (FTO/pristine and 0.3 mol% W-doped TiO₂/PCBM/Ag) were obtained by Mott. Gurney Eq. (2)[72]

$$J = 9\varepsilon_0\varepsilon_r\mu_e V^2 / 8L^3 \quad (2)$$

J represents the current density, ε_r represents the dielectric permittivity of TiO₂, ε_0 is the vacuum permittivity, V stands for the device voltage, μ_e and L stand for electron mobility and thickness of the ETL. After calculation, the μ_e of pure TiO₂ and 0.3 mol% W-doped TiO₂ are 5.34×10^{-4} and 8.85×10^{-4} , respectively. It can manifest that W-doped method can enhance the electron mobility of TiO₂ film. The optical absorption spectra are generally used to show the energy gap and optical properties. A strong absorption appeared less 400 nm since the (O²⁻ to Ti⁴⁺) ligand to metal charge transition[73] has shown in **Fig. 4c**. Blue shift of the absorption edge proves a large band gap formed, indicating it was enlarged after doping W element. E_g can be calculated directly through the Tauc plot in **Fig. 4d**. W-doped TiO₂ exhibits an increase from 3.03 to 3.1 eV,

compared with the undoped device. It reveals that the replacement of Ti^{4+} by W^{6+} results in a negative shift of conduction band minimum (CBM)[74]. UPS is employed to analyze the valence electron structure. It is clearly found the Fermi level (E_f) decrease from pristine TiO_2 (4.16 eV) to 0.3 mol% W-doped TiO_2 (4.12 eV) in **Fig. S3**. This up shift is due to high binding power of tungsten to oxygen which is even dislodged causing the vacancies formed when doping the W^{6+} , so the open-circuit voltage (V_{oc}) of PSC is improved and the conductivity of TiO_2 layer will be enhanced. The cross-section SEM image is demonstrated in **Fig. 5a**. After calculation, the energy-level diagram of the solar cell with the position of E_f , valence band maximum (VBM) and CBM is shown in **Fig. 5b** and the corresponding data are analyzed in **Table S2**. W-doped TiO_2 has a lower CBM than that of undoped TiO_2 , assisting more direct injection electron generated from the perovskite into the TiO_2 conduction band. Hence, it will optimize electron accumulation and enhance transport between the TiO_2 and perovskite layer accordingly.

SEM images for top surface morphology of perovskite layer deposited on undoped TiO_2 and 0.3 mol% W-doped TiO_2 layer are shown in **Fig. 6a and b**. It is clearly seen that particles with porosity are aggregated on the ETL tightly and grain size becomes larger after doping W element. In **Fig. S4**, analyzed by software, distribution of the perovskite grain size is obtained from pristine and 0.3 mol% W-doped TiO_2 , respectively. Obviously, perovskite layer coated on the doped TiO_2 has a larger mean grain size (750

nm), compared to that based on pure TiO₂ device (500 nm), which are beneficial for the PSCs. The crystallinity of perovskite films are characterized by XRD in **Fig. 6c**. As clearly seen, the peaks at 14.11°, 28.44° and 31.88° can be assigned to the diffractions from (110), (220) and (310) crystal planes for perovskite. These peaks show a vigorous intensity of diffraction, demonstrating a better crystallinity perovskite film via doping W into TiO₂. The strong intensity is most probably ascribed to the high crystallinity and large grain size of the perovskite layer. Meanwhile, this appearance also corresponds to the enhanced absorption in 0.3 mol% W-doped TiO₂ sample (**Fig. 6d**). The absorption spectrum are obtained through multilayer film FTO/pristine TiO₂/perovskite and FTO/0.3 mol% W-doped TiO₂/perovskite. After W-doping, the light-absorbing promotion is discovered, owing to band gap and transmittance of TiO₂ become larger and stronger, thus the absorption of FTO/0.3 mol% W-doped TiO₂ film is enhanced and the corresponding performances will be improved.

Since the PSCs have the structure of a planar heterojunction, it reveals a multi-layer structure, including FTO/W-doped TiO₂ (70 nm)/Cs_{0.05}(FA_{0.83}MA_{0.17})_{0.95}Pb(I_{0.83}Br_{0.17})₃(450 nm)/Spiro-OMeTAD (250 nm)/Au (70 nm). **Fig. 7a** shows the incident photon-to-electron conversion efficiency (IPCE) curves of pristine and 0.3mol% W-doped PSC. The IPCE curve shows a similar trend, but the intensity is different. The IPCE spectrum intensity of W-doped TiO₂ as the electron transport layer device is significantly improved across the whole wavelength. In

comparison with the pristine TiO₂ (21.69 mA/cm²), doped TiO₂ (23.3 mA/cm²) has a higher integrated J_{sc} and corresponds to the J-V value. PSCs based on different concentrations of W-doped TiO₂ layer are fabricated and the performances of them are examined in **Table 1**. **Fig. 7b** shows the J-V curves of devices with different doping concentrations. Consequentially, the champion device, 0.3 mol% W-doped TiO₂ sample exhibits a better performance whose open-circuit voltage (V_{oc}) is 1.12 V, J_{sc} is 24.24 mA/cm², FF is 69.22%, and PCE is 18.85% when compared with the pristine TiO₂ sample. It is clearly noted that all the parameters are enhanced with the increased concentration of dopant since the increased density of active carrier, conductivity of the doped ETL and better E_g, E_f and CBM position of TiO₂. In terms of the substantial increase in J_{sc}, this improvement may be largely attribute to two factors: (1) The Fermi energy level has moved to conduction band side, which will improve the efficiency of electron injection; (2) With the higher conductivity of TiO₂ after doping W, more efficient charge will be extracted in the interface between TiO₂ and perovskite layer. Meanwhile, the higher V_{oc} may result from decreased charge recombination and modified charge gathering/separation. Interestingly, the performance of 0.5mol% W-doped TiO₂ is little worse than that of 0.3 mol% film, which could be ascribed to more electron defects formed and saturated electron barrier height owing to overdose doping. J-V curves were measured under reverse/forward scanning to analyze the hysteresis effect in **Fig. 7c**. **Table S3** summarizes the related parameters of the PSCs. It shows a hysteresis decrease after doping W in TiO₂. This improvement can be attributed to the

injection/separation of the effective carrier at the interface between ETL and perovskite layer. Therefore, W dopant plays as an important role in moderating the charge collection in the interface for ameliorating the hysteresis of PSCs.

The reproducibility of PSCs is studied through 35 individual devices based on the pristine and 0.3 mol% W-doped TiO₂. In **Fig. 8a and b**, the PCE value of W-doped device exhibits a more concentrated distribution when compared with undoped device. Statistical data of V_{oc} , J_{sc} , FF, and PCE are shown in **Fig. 8c, d, e and f**. In **Fig. S5**, we evaluated the device stability in the ambient air with a relative humidity of 36%. After 33 days exposure in the air, the W-doped device maintained about 80% of their initial PCE, which is higher than 60% of that for the control sample. The light stability was obtained in Fig. S6. In contrast, the PCE of 0.3 mol% W-doped TiO₂ PSC remained stable at more than half of its initial efficiency after 260 hours of exposure in a nitrogen atmosphere. This improved stability on PSCs is mainly attributed to the efficiently charge transfer between W-doped TiO₂ and perovskite layer, ameliorated conductivity of ETL, high quality of perovskite includes optimized crystallinity and large grain size by doping W dopant[75].

High efficiency perovskite solar cell based on tungsten-modified TiO₂ has been obtained. Then we use steady-state photoluminescence (PL) and time-resolved photoluminescence (TRPL) spectra to further explain the charge transfer dynamics

between perovskite and W-doped TiO₂, and analyze the underlying mechanism of efficiency improvement. In **Fig. 9a**, PL of perovskite films deposited on pure TiO₂ and 0.3 mol% W-doped TiO₂ are presented. Compared with the undoped sample, the photoluminescence intensity of the perovskite film on the W-doped TiO₂ was significantly reduced, shows more efficient PL quenching, indicating that electrons could be extracted and transported efficiently from the perovskite to the modified TiO₂[76]. Meanwhile, the doped device can effectively inhibit the carrier recombination at this interface. Furthermore, the TRPL curves (**Fig. 9b**) were shown to examine the charge excitation lifetime by equation (3):

$$f(t) = A_1 \exp\left(-\frac{t}{\tau_1}\right) + A_2 \exp\left(-\frac{t}{\tau_2}\right) \quad (3)$$

Where τ_1 represents the fast decay time, τ_2 stands for the slow decay time and A is the coefficient of decay amplitude[77-80]. All fitted parameters are listed in **Table S4**. A strongly PL quenching was observed, and the average charge lifetime (τ_{ave}) has decreased from 102.49 ns to 31.17 ns with tungsten treatment. Therefore, with better electron extraction between the perovskite and TiO₂, electrons from the perovskite layer can be transferred efficient, and non-radiative composite defects are significantly reduced[81]. These properties are attributed to the reduction of defect state density and the formation of a good interface between perovskite and W-doped TiO₂.

In order to quantitatively analyze the density of defect states in perovskite films, we prepared the following device structures: FTO/TiO₂ (undoped or doped)/ Perovskite

/PCBM/Ag. We measured the device current-voltage curves under dark conditions (**Fig. 9c**), and then calculated the defect density (N_t) by space charge limited current method (SCLC). N_t is calculated by equation N_t :

$$N_t = 2\varepsilon_r\varepsilon_0V_{TFL}/qL^2 \quad (4)$$

Where V_{TFL} is the trap-filled limit voltage, L stands for the thickness of perovskite layer (310 nm), e stands for the charge of electron and ε_r and ε_0 are the relative permittivity (28.8) and absolute permittivity (8.85×10^{-12} F/m), respectively[82]. It can be seen from the curve that the V_{TFL} decreases from 0.61V to 0.38 V; the calculated defect densities of pristine TiO_2 and 0.3 mol% W-doped device are $2.02 \times 10^{16} \text{ cm}^{-3}$ and $1.26 \times 10^{16} \text{ cm}^{-3}$, correspondingly. This result directly proves tungsten doping can effectively passivate the defect of perovskite layer. It is consistent with the photoluminescence result.

Electrochemical impedance spectroscopy (EIS) was utilized to examine the mechanism of charge transfer. In **Fig. 9d**, low frequency curve represents the charge composition, usually representing the composite impedance, while the high frequency curve is related to the carrier transport impedance[83, 84]. According to the simulation data, the series resistance of W-doped device decreases, which may be ascribed to the fill factor of the devices has increased. The series resistance (R_s) and recombination resistance (R_{rec}) were measured by EIS system in **Table S5**. A lower value of R_s and larger value of R_{rec} were obtained in W-doped device, indicating the charge transfer rate has increased and the electron conductivity were improved. Consequently, these results

confirm that tungsten doping is conducive to perovskite solar cell.

4. Conclusions

In summary, W-doped TiO₂ layer, as ETL in PSCs, was fabricated successfully through the low-temperature solution-process. The PCE of improved device is 18.85%, which is almost 28.1% higher than that of pristine TiO₂ device (14.71%). It can be seen that improved conductivity of film, upshift of the E_f and lower defect density will efficiently increase injection/separation of the carrier at the TiO₂/perovskite layer interface. Therefore, J_{sc} , FF and stability are subsequently increased accordingly since the improvement for doping W element. Consequently, the proposed method using lower-valency materials as dopant, fabricated in low temperature, might provide an effective way for high-performance PSCs technology and potential application in wearable low-cost electronics.

Acknowledgments

This research was funded in part by the Natural Science Foundation of the Jiangsu Higher Education Institutions of China Program (19KJB510059), the Suzhou Science and Technology Development Planning Project: Key Industrial Technology Innovation (SYG201924), and the Key Program Special Fund in XJTLU (KSF-P-02, KSF-T-03, KSF-A-04, KSF-A-05, KSF-A-07).

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