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## Comparison of Methods for Estimating Carbon Evasion and Export Associated with a Coal Mine Discharge

Kyle Lee

*West Virginia University*, [kjlee@mix.wvu.edu](mailto:kjlee@mix.wvu.edu)

Dorothy J. Vesper

*West Virginia University*, [dorothy.vesper@mail.wvu.edu](mailto:dorothy.vesper@mail.wvu.edu)

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# Comparison of methods for estimating carbon evasion and export associated with a coal mine discharge

Kyle Lee & Dorothy J Vesper  
West Virginia University

## Introduction

The evasion of CO<sub>2</sub> from terrestrial waters plays a role in the global cycling of carbon but there are few datasets that have an accurate accounting of the flux. It has been shown that discharges from coal mines can have elevated concentrations of CO<sub>2</sub> due to sulfuric acid-driven dissolution of carbonate rock.

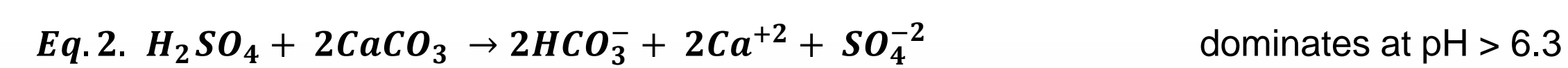
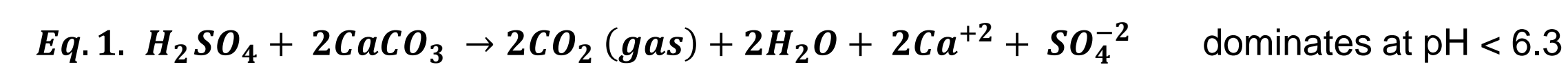
In this study we compared three methods for calculating the export dissolved inorganic carbon (DIC) and the evaluation of CO<sub>2</sub> from an abandoned-mine discharge in West Virginia. In general, the methods based on direct measurement of DIC and CO<sub>2</sub> were within an order-of-magnitude of the diffusive-flux model.



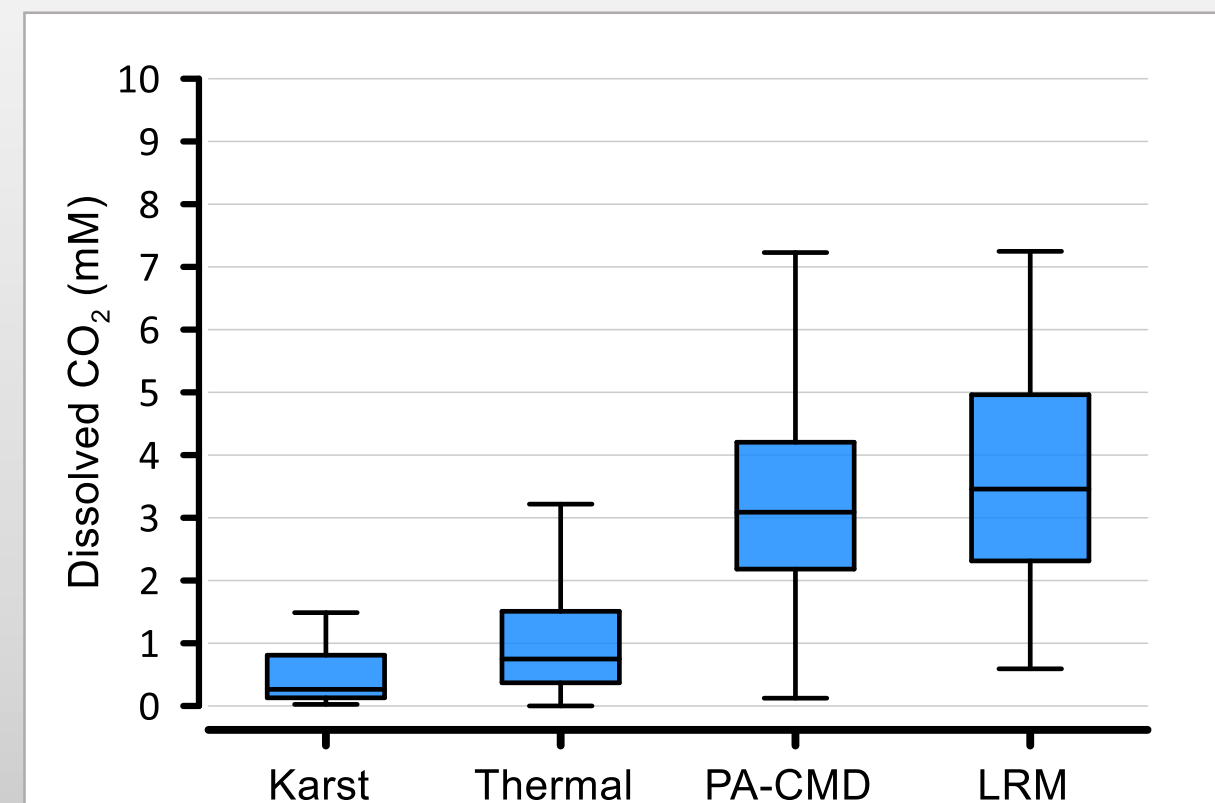
Surveying the stream channel at the LRM site. Looking upstream toward the portal.

## DIC and CO<sub>2</sub> in Coal Mine Drainage

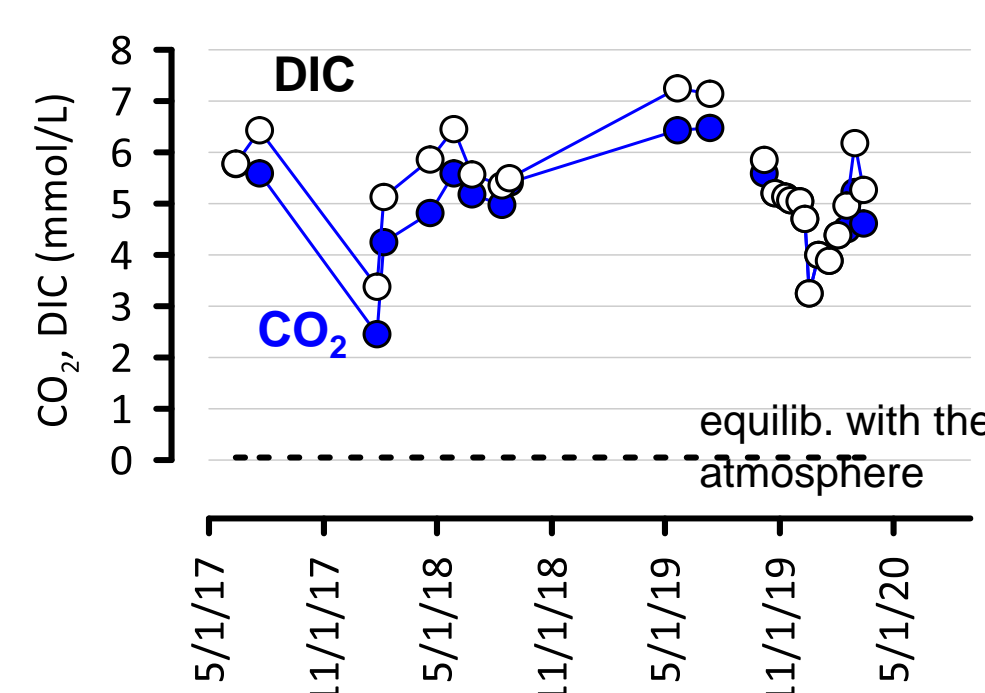
In mine waters, pyrite (FeS<sub>2</sub>) oxidizes and releases sulfuric acid (H<sub>2</sub>SO<sub>4</sub>). If the acid reacts with limestone (CaCO<sub>3</sub>), it produces DIC in the form of CO<sub>2</sub> gas (Eq. 1) or bicarbonate (HCO<sub>3</sub><sup>-</sup>; Eq. 2)



The DIC is the sum of the carbonate species (CO<sub>2</sub> + HCO<sub>3</sub><sup>-</sup> + CO<sub>3</sub><sup>-2</sup>); the dominant species is controlled by the pH of the system.



Dissolved CO<sub>2</sub> concentrations in different water types. Data from Mass and Wicks 2017 (typical karst, thermal), Cravotta 2008 (PA-coal mine drainage), and this study (LRM). The data clearly show that mine waters can have elevated concentrations of CO<sub>2</sub>.



CO<sub>2</sub> and DIC concentrations at the source portal at the Lamberts Run site (LRM). Although CO<sub>2</sub> concentrations change over time they are not solely linked to discharge. The pH is the primary control on the relative concentrations of DIC and CO<sub>2</sub>. DIC in the CO<sub>2</sub> form is lost as gas to the atmosphere while DIC in the HCO<sub>3</sub><sup>-</sup> form is exported offsite in the dissolved form.

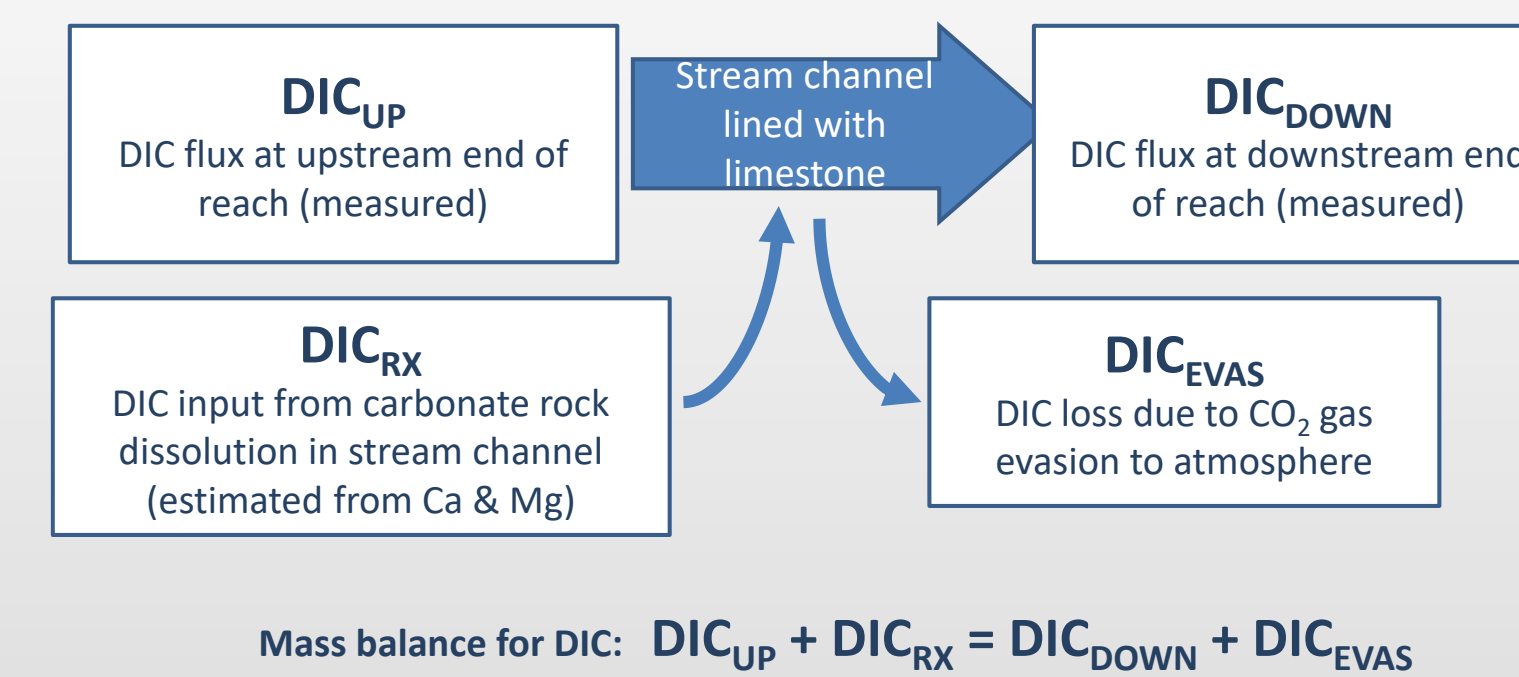
## Methods for measuring CO<sub>2</sub> evasion and export from mine drainage

Flux was estimated using three methods:

Method #1. Source Flux is calculated from the discharge and the concentrations at the portal. CO<sub>2</sub> and DIC concentrations are measured directly in the field following procedures developed by Vesper and Edenborn (2012, 2015).

$$\text{Source Flux (mass/time)} = \text{CO}_2 \text{ concentration (mass/vol)} \times \text{discharge (volume/time)}$$

Method #2. Stream Flux is calculated over a stream reach and considers both upstream and downstream chemistries. This is based on the mass balance of DIC between two points in the stream.



Method #3. The diffusive flux model estimates rates evasion based on transport of CO<sub>2</sub> through the water column following the approach used by Maas and Wicks (2016). In this method, the flux is calculated as follows:

$$F_{CO_2} = k K_O (P_{CO_2w} - P_{CO_2a})$$

Where  $k$  is the gas transfer velocity of CO<sub>2</sub> in cm h<sup>-1</sup>,  $K_O$  is the solubility constant of CO<sub>2</sub> in water in mol m<sup>-3</sup> atm<sup>-1</sup>;  $P_{CO_2w}$  is the partial pressure of CO<sub>2</sub> in water ( $P_{CO_2w}$ ) and air ( $P_{CO_2a}$ ) in atm.

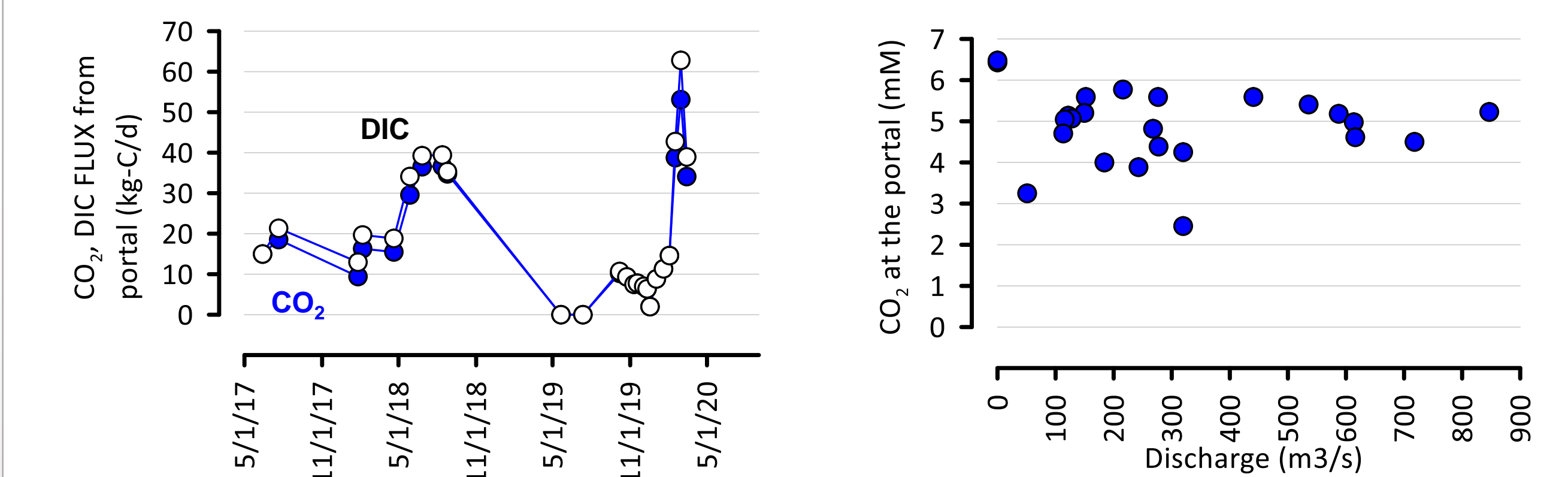
$K_O$  is calculated based on temperature and salinity. The value for  $k$  is calculated as

$$k = C(S_c/600)^{-0.5}$$

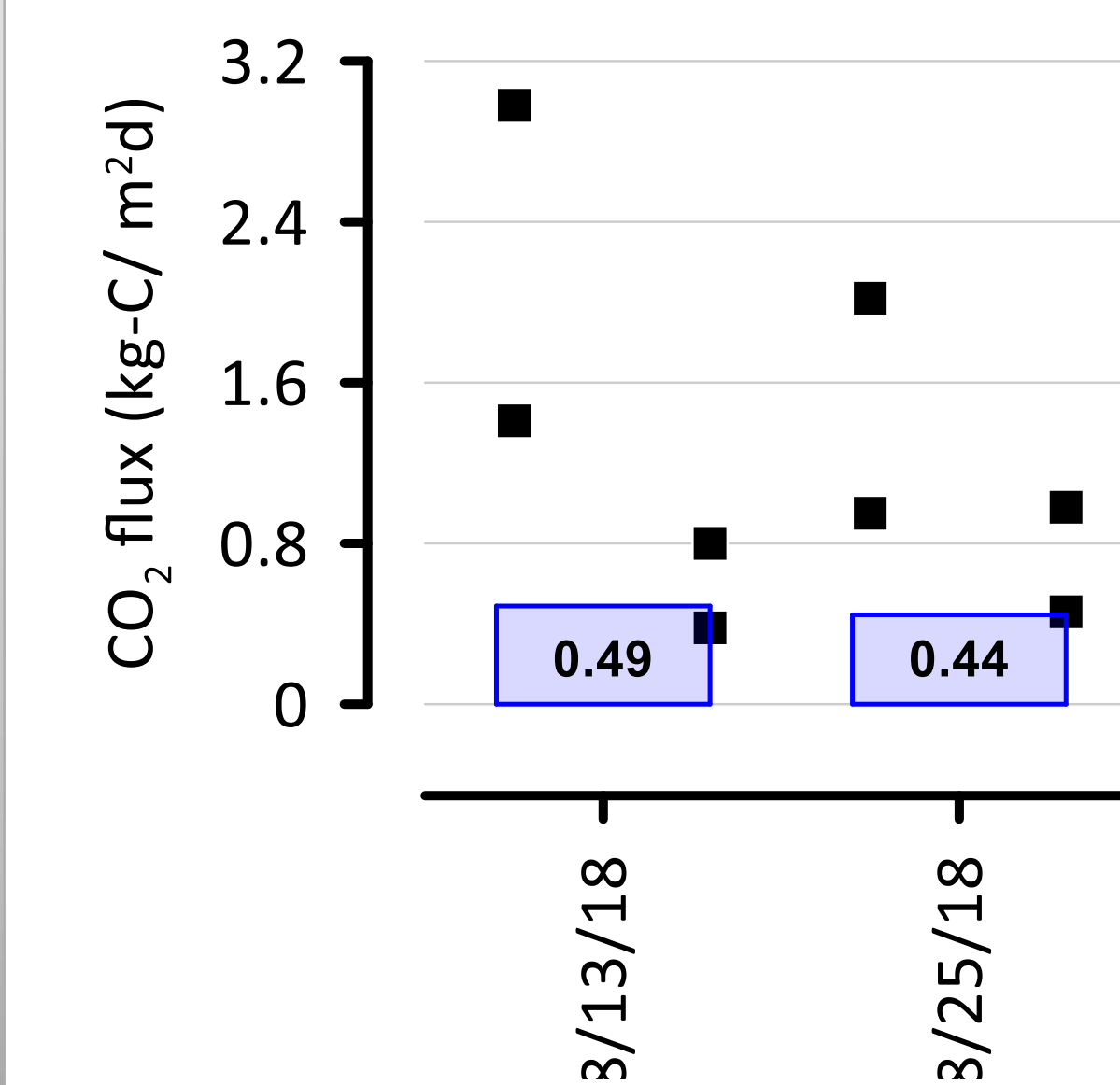
Where  $C$  is a constant (1.91 for wind speeds less than 3 m/s).  $S_c$  is the Schmidt number which is a function of temperature. The Schmidt number depends on the viscosity, diffusivity and density of the solution along the boundary layer.

Comparison: Methods #2 and #3 can be compared against each other by normalizing the method #2 flux by the area of the stream surface between the upstream and downstream locations used for the mass balance. To estimate this area the stream reach was surveyed and the stream width measured at 1 meter intervals.

## Comparison of values obtained for CO<sub>2</sub> loss from mine discharge at LRM



The flux of CO<sub>2</sub> and DIC from the portal at LRM (left). Discharge and concentrations do not vary with each other (right) and therefore the temporal patterns of flux differ from those of concentration.



Methods #2 (mass balance, blue bar chart) and #3 (diffusion model, squares) were compared for two datasets collected in August 2018. Method #2 flux is calculated over the entire 0-40 m stream reach based on measured values of CO<sub>2</sub> and DIC. The black squares provide the high and low flux estimates obtained from the diffusion model – calculated for both the upstream and downstream end of the reach.

The Method #3 diffusion model generates higher CO<sub>2</sub> fluxes than those based on the mass flux from the detailed measured concentrations.

Method	Advantages	Limitations
<b>Method 1. Source Flux</b>	Single point measurement	Direct measurement of CO <sub>2</sub> is most accurate at low-pH and high-CO <sub>2</sub> waters; does not account for degassing at the surface
<b>Method 2. Stream Flux</b>	Mass balance can be used to identify shifts in chemical species and types of DIC loss	CO <sub>2</sub> and DIC concentration vary with stream reach, depth and cross-section. Need to know the area of the reach.
<b>Method 3. Diffusion model</b>	No direct measurements of CO <sub>2</sub> or DIC are needed; can be broadly applied	Does not account for changes over distance or final equilibrium CO <sub>2</sub>

## Summary

- CO<sub>2</sub> is high in coal mine water relative to other natural waters
- CO<sub>2</sub> concentrations change over time
- CO<sub>2</sub> evasion and flux can be calculated using different means but with some limitations
- Direct measurements of CO<sub>2</sub> are critical for low-pH high-pH waters; Although CO<sub>2</sub> can be estimated from the DIC and pH, high-CO<sub>2</sub> water typically lose mass due to degassing prior to analysis.
- Limestone used to remediation the mine water can dissolve and release DIC into the system

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